



# Microstructure evolution, crystallography, nanomechanical properties and wear-resistance of multi-component $M_2B$ in Fe–B alloy

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To investigate the basic characteristics of multi-component boride in Fe–B alloy, the samples were prepared by directional solidification, and then the microstructural evolution, crystallography, nanomechanical properties and wear-resistance of multi-component  $M_2B$  have been systematically studied. The results show that the boride grows in the sequence of Mo-rich  $\rightarrow$  Fe-rich  $\rightarrow$  Cr-rich  $M_2B$ . The Mo-rich  $M_2B$  has the body centered tetragonal (bct) structure with stoichiometry of  $Fe_{0.72}Cr_{0.27}Mo_{0.80}(B,C)$ , the Fe-rich  $M_2B$  possesses the bct structure with stoichiometry of  $Fe_{1.59}Cr_{0.38}Mo_{0.08}(B,C)$ , while the Cr-rich  $M_2B$  owns the body centered orthorhombic (bco) structure with stoichiometry of  $Fe_{1.48}Cr_{0.50}Mo_{0.02}(B,C)$ . Moreover, the Mo addition can facilitate more dislocations in the  $M_2B$  compared to the Cr addition. Thus, the Mo-rich  $M_2B$  has the best stiffness and toughness, followed by the Cr-rich  $M_2B$  and then the Fe-rich  $M_2B$ . Additionally, the Mo-rich and Cr-rich  $M_2B$  can possess higher wear-resistance relative to the Fe-rich  $M_2B$ .

## Introduction

Nowadays, considerable amounts of energy consumption and economic loss are caused by the friction and wear in industrial equipment, thus it's imperative to design environmentally and energy-saving wear-resistant materials [1–3]. Until now, iron-based wear-resistant material containing certain B and C contents was widely applied, because of the existence of intermetallic compounds (e.g.,  $Fe_2B$  [2],  $Cr_7C_3$  [3] or  $Fe_3C$  [4]). Compared to the  $Cr_7C_3$  or  $Fe_3C$ , the  $Fe_2B$  has higher hardness and heat stability, thus leading to better wear-resistance [5, 6]. As boron is difficult to dissolve into iron, it mainly segregates at grain boundaries to form  $Fe_2B$  [7–9]. Therefore, Fe–B alloy can be designed on the basis of the fundamental concept of replacing the carbide in iron with  $Fe_2B$ .

Thus far, Fe–B alloy has attracted more and more attention as a novel wear-resistant material. Ren et al. [10] reported that

increasing B content could increase the volume of  $Fe_2B$ , which effectively improved the wear-resistance of Fe–B alloy. Zhang et al. [11] revealed the effect of hot forging on the mechanical properties and wear-resistance of Fe–B alloy. The results showed that the toughness and wear-resistance of alloy increased considerably after hot forging. Yi et al. [12] optimized the cooling rate for designing Fe–B alloy with high-performance. The effect of cooling rate on microstructure, mechanical properties and residual stress of Fe–B alloy was investigated. Results showed that the critical cooling rate of martensite transformation may be optimal quenching rate of alloy. Yi et al. [6] has also discussed the effect of matrix microstructure on the abrasive wear behavior of Fe–B alloy. It had been proven that the Fe–B alloy with high  $V_m/V_p$  (where the  $V_m/V_p$  indicates the volume ratio of martensite to pearlite) could effectively subject to severe abrasive wear in a low contact stress. As a new wear-resistant material,

Fe–B alloy has been put into production, and achieved very good economic benefits. Scholars have developed Fe–B alloy-based rolls, mud pump impeller blades and glass molds [13, 14]. Such as, the service life of Fe–B alloy-based roller is about 8 times than that of ductile iron-based roller.

As mentioned above, Fe–B alloy can possess excellent wear-resistance, which is mainly related to the existence of Fe<sub>2</sub>B. During the wear process of Fe–B alloy, the Fe<sub>2</sub>B resists abrasive and protects metallic matrix from being shoveled off. However, the Fe<sub>2</sub>B possesses inherent brittleness because of the weak B–B bond in [002] direction [15, 16]. This will decrease the wear-resistance of Fe–B alloy due to a great susceptibility to cracking. That is, it is particularly necessary to toughen the Fe<sub>2</sub>B. Jian et al. [17] indicated that transition metal element Cr could replace partial Fe atom in Fe<sub>2</sub>B crystal, which could improve the toughness of M<sub>2</sub>B-type (M = Cr, Fe and Mo) boride, improving the wear-resistance of Fe–B alloy as a result. Similarly, Yi et al. [18] revealed that Mo addition could also form multi-component M<sub>2</sub>B in Fe–B alloy. Except for these, Xiao et al. [19–21] investigated the fundamental properties of M<sub>2</sub>B so as to provide a guidance for its real application. And the values of bulk modulus for Cr<sub>2</sub>B, Fe<sub>2</sub>B and Mo<sub>2</sub>B were 279.56, 249.73 and 289.43 GPa, respectively.

To sum up, the formation of multi-component M<sub>2</sub>B is significant for the wear-resistance of Fe–B alloy. Thus, it is imperative to study the basic characteristics of various multi-component M<sub>2</sub>B in the Fe–B alloy. However, few investigations on the microstructure evolution, crystallography, nanomechanical properties, and wear-resistance of multi-component M<sub>2</sub>B in Fe–B alloy have been reported. In this work, the microstructure evolution of multi-component M<sub>2</sub>B in Fe–B alloy is systematically analyzed using directional solidification method (DS). Accordingly, the crystallography, nanomechanical properties and wear-resistance of multi-component M<sub>2</sub>B were discussed in detail. Even more, the study aims to make contributions to further design of Fe–B alloy with high-performance.

## Result and discussion

### Microstructure evolution

Figure 1 shows phase diagram of the tested alloy during equilibrium solidification process. When the solidification temperature drops to 1316 °C (point 3), the δ-Fe precipitates from liquid melt firstly, owing to the transformation of L → δ-Fe. This is because the tested alloy has high content of ferrite stabilizer elements (Mo, Cr, Al, V), which leads to the formation of δ-Fe as

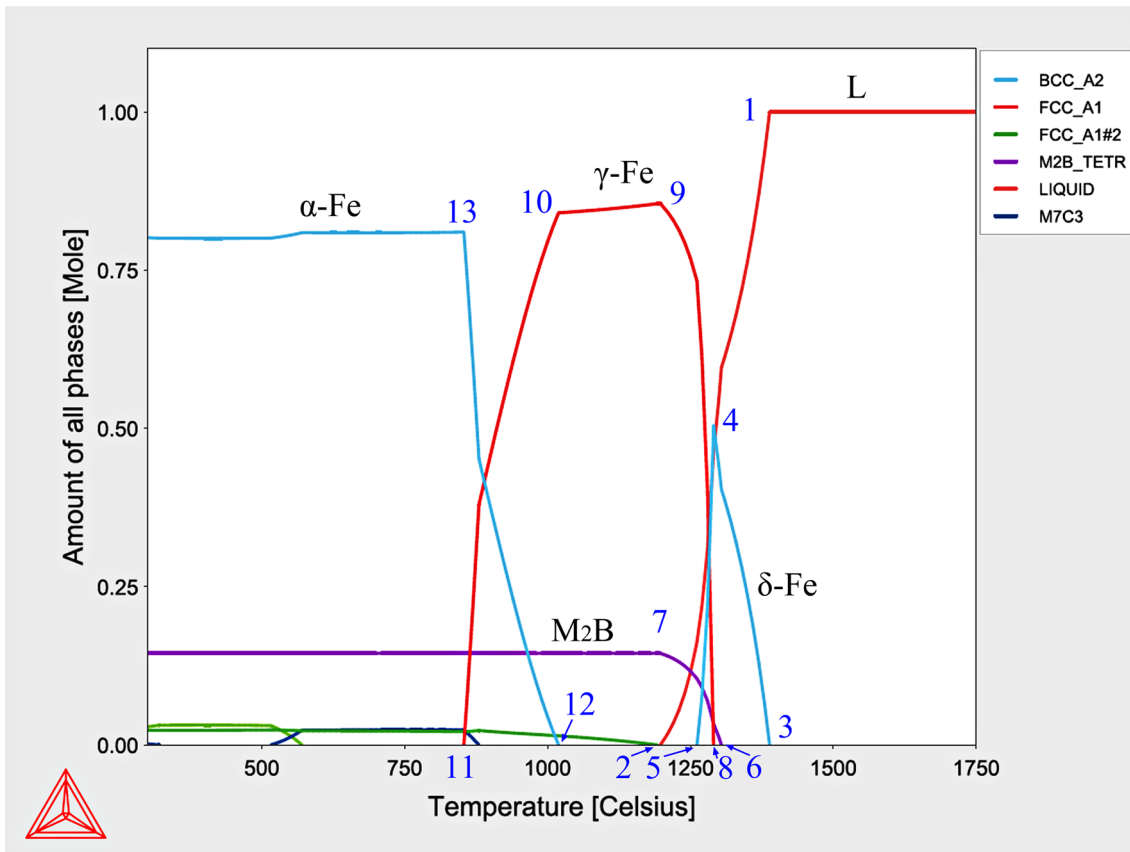
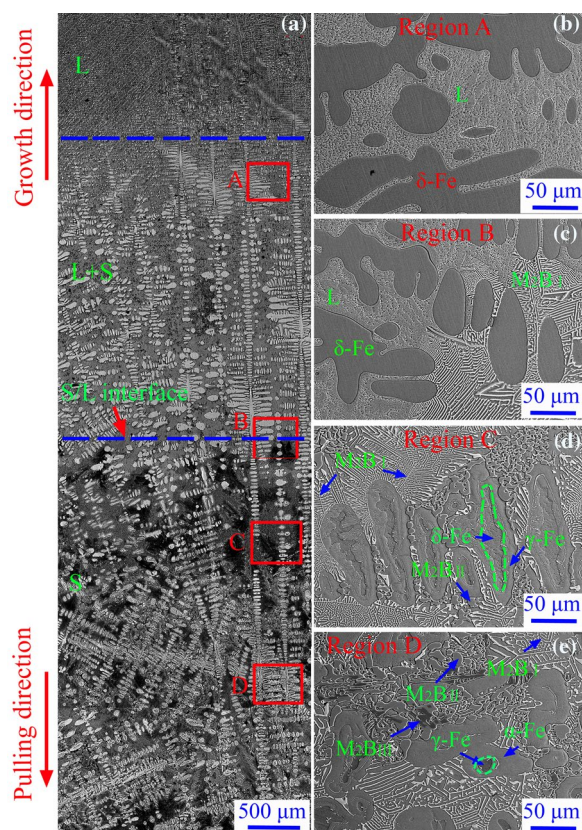


Figure 1: Thermo-calc calculation of the tested Fe–B alloy.



**Figure 2:** Macro- and micro-structures of the tested DS bar: (a) Longitudinal-section macrostructure; (b)–(e) Microstructures of the A, B, C and D regions.

primary phase [22, 23]. As the solidification temperature drops from 1213 °C (point 6) to 1081 °C (point 7), the alloying elements (e.g., Cr, Mo, V or B) are remained in residual liquid melt accompanied with the growth of  $\delta$ -Fe [24–26]. As a result, the transformations of  $L + \delta$ -Fe  $\rightarrow \gamma$ -Fe and  $L \rightarrow \gamma$ -Fe +  $M_2B$  occur, just as reported in previous researches [25–28]. With the further decrease of solidification temperature (point 12  $\rightarrow$  point 13), pearlitic transformation occurs, and then the  $\gamma$ -Fe is converted to  $\alpha$ -Fe at 869 °C.

Figure 2 shows the macro- and micro-structures of the studied DS bar. From the macrostructures, the DS bar can be divided into three parts [Fig. 2(a)]. The upper part is “liquid region (L)” caused by sudden power termination, the bottom part is “stable solidification region (S)”, and the middle part is “mushy zone (S + L)”. The relatively well-aligned grains grow along external coordinate  $c$ -axis. From the microstructures, the dendritic  $\delta$  phase forms in liquid melt firstly [Fig. 2(b)]. As solidification temperature drops, the  $M_2B$  forms subsequently because of the transformation of  $L \rightarrow \gamma$ -Fe +  $M_2B$ . Herein, the white-coralloid  $M_2B_I$  forms firstly [Fig. 2(c)], the light-coloured rod-like  $M_2B_{II}$  acts as following phase [Fig. 2(d)], and then the dark-coloured blocky  $M_2B_{III}$  appears [Fig. 2(e)]. Meanwhile, the metallic matrix

changes from  $\delta$ -Fe to  $\gamma$ -Fe and then to  $\alpha$ -Fe with the decrease of solidification temperature. The experimental results are in agreement with the calculated phase diagram.

To clarify alloying element distribution in the multi-component  $M_2B$ , the corresponding composition is examined using EPMA, as listed in Table 1. The stoichiometry of  $M_2B$  is calculated by the formula of  $M_x(B,C)_y$ . The  $x$  and  $y$  represent (Fe, Mo, Cr, etc.) at% and (B,C) at%, respectively. From Table 1, the  $M_2B_I$  is rich in Mo, and the stoichiometry is  $Fe_{0.72}Cr_{0.27}Mo_{0.80}(B,C)$ , closing to the  $M_{1.96}B$ -type boride [7–9, 29, 30]. The  $M_2B_{II}$  is rich in Fe, and the stoichiometry is  $Fe_{1.59}Cr_{0.38}Mo_{0.08}(B,C)$ , which tends to be the  $M_{2.09}B$  [7–9, 30, 39]. However, the  $M_2B_{III}$  is rich in Cr and deficient in Mo, possessing a stoichiometry of  $Fe_{1.48}Cr_{0.50}Mo_{0.02}(B,C)$ , closing to the  $M_{2.03}B$  [7–9, 30, 39]. According to the element periodic table, the atomic radius  $R_{Cr}$ ,  $R_{Mo}$  and  $R_{Fe}$  are 0.185, 0.201 and 0.172 nm, respectively, and the electronegativity  $X_p^{Cr}$ ,  $X_p^{Fe}$  and  $X_p^{Mo}$  are 1.66, 1.83 and 2.16, respectively. Therefore, the similar electronegativity and atomic radius result in that Mo or Cr atom can replace Fe atom in  $Fe_2B$  crystal. Combined with the microstructures, it can be known that the multi-component  $M_2B$  forms in a sequence of Mo-rich  $\rightarrow$  Fe-rich  $\rightarrow$  Cr-rich  $M_2B$ . This may be because the Gibbs free energies of various  $M_2B$  ( $M = Mo, Cr$  or  $Fe$ ) are  $G_{Mo_2B} < G_{Fe_2B} < G_{Cr_2B}$  [17–19]. Another factor identified in this work is the partition coefficient of alloying element, such as the partition coefficients are 0.51 and 0.34 for the Mo and Cr atoms, respectively [31].

### Crystal structure analysis

Figure 3 shows bright-field TEM micrographs and corresponding selected area diffraction patterns (SADPs) of the multi-component  $M_2B$ . From Fig. 3(a) and d, it can be observed that the rod-like Fe-rich  $M_2B$  is a bct structure with the lattice parameter of  $a = b = 0.5109$  nm,  $c = 0.4249$  nm and  $c/a = 0.83$  ( $C16$ ,  $CuAl_2$ -type structure). From Fig. 3(b) and e, the coral-loid-like Mo-rich  $M_2B$  is clearly distinguished and indexed to a bct structure ( $a = b = 0.5547$  nm,  $c = 0.4739$  nm and  $c/a = 0.85$ ). From Fig. 3(c) and (f), the blocky Cr-rich  $M_2B$  has a bco structure with lattice parameter of  $a = 1.4583$  nm,  $b = 0.7379$  nm and  $c = 0.4245$  nm. Moreover, compared with the  $Fe_2B$ -rich  $M_2B$ , the lattice constants of Cr-rich and Mo-rich  $M_2B$  have an obvious increase. These mean that the addition of alloying elements Cr and Mo can affect the crystal structure of  $Fe_2B$ . Except for these, Ma et al. [7–9] revealed that a possible orientation relationship (OR) between bco  $M_2B$  and metallic matrix was  $\langle 1-10 \rangle_{M_2B} // \langle 110 \rangle_{\alpha}$ , while a possible OR between bct  $M_2B$  and metallic matrix was  $\langle 002 \rangle_{M_2B} // \langle 110 \rangle_{\alpha}$ . Similarly, a possible OR between bct  $M_2B$  and metallic matrix detected by electron back-scattering diffraction was  $\langle 001 \rangle_{M_2B} // \langle 001 \rangle_{\alpha}$  [29].



**TABLE 1:** Electron microprobe analysis of the multi-component  $M_2B$  in Fe–B alloy.

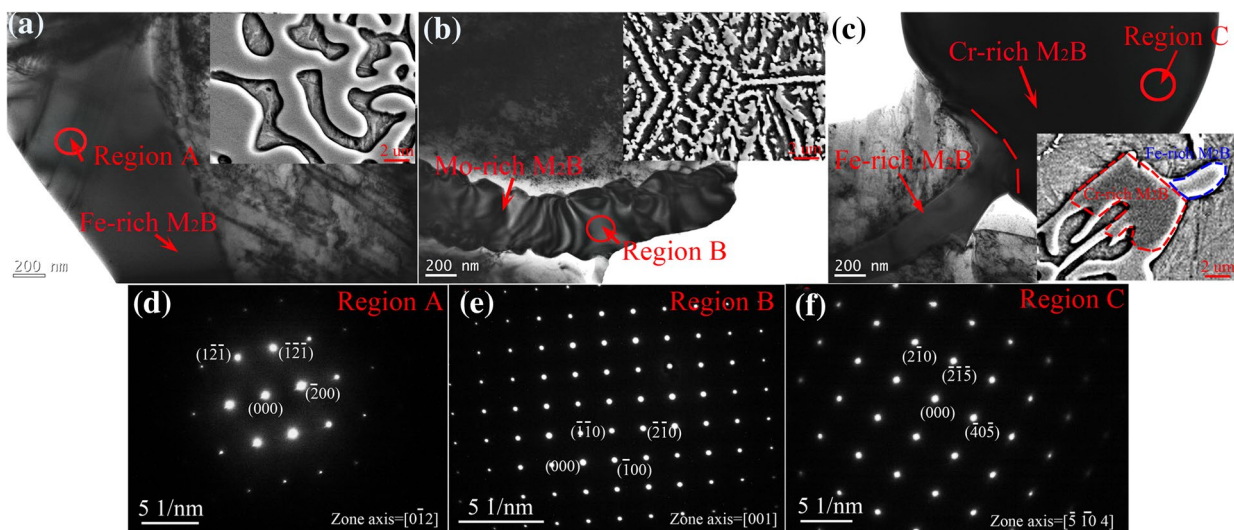
$M_2B$	Element (at%)							Calculated formula	
	B	C	Cr	Mn	Fe	V	Mo		
Mo-rich $M_2B$	21.97	10.11	12.00	0.66	51.17	1.31	2.78	$Fe_{1.59}Cr_{0.38}V_{0.04}Mo_{0.08}(B,C)$	$\sim M_{1.96}B$
Cr-rich $M_2B$	30.03	2.72	16.43	0.59	48.49	0.99	0.75	$Fe_{1.48}Cr_{0.50}V_{0.03}Mo_{0.02}(B,C)$	$\sim M_{2.03}B$
Fe-rich $M_2B$	31.86	4.81	6.23	0.33	24.06	5.76	26.95	$Fe_{0.72}Cr_{0.27}V_{0.17}Mo_{0.80}(B,C)$	$\sim M_{2.09}B$

In order to investigate the detailed substructure of multi-component  $M_2B$ , the high-resolution transmission electron microscope (HRTEM) and corresponding fourier-filtered transform (FFT) images are detected as shown in Fig. 4. From Fig. 4(a), the interplanar distance of the (404) plane for the Fe-rich  $M_2B$  is 0.612 nm, and nearly no obvious defects can be observed. From Fig. 4(b), the interplanar distance of the (002) plane for the Cr-rich  $M_2B$  is 0.235 nm, and several defects can be found in the  $M_2B$ . To be exact, the Burgers vector is marked by drawing a frame enclosing non-perfect zone. Two extra half-planes are inserted from the bottom and right sides of Burgers frame. This represents two full dislocations with Burgers. Moreover, the central area enclosed with circle can be indexed to an edge dislocation, which is marked by the “T”. From Fig. 4(c), more edge dislocations can be found in the Mo-rich  $M_2B$ . Several stacking faults can be found in the region A, and spatial distribution of dislocations can be observed in the region B. This reveals that the addition of Cr and Mo facilitates the formation of dislocation in the  $M_2B$ . Compared to the Cr addition, the Mo addition can induce the formation of more dislocations. Just as reported in previous works [17, 18], the Mo or Cr addition can affect the electron environment of B–B bond, which changes the density of surrounding B electrons,

leading to the formation of dislocation in  $Fe_2B$  crystal as a result.

### Nanomechanical properties

Figure 5 shows the  $P$ – $h$  curves of multi-component  $M_2B$  in the tested DS sample at the load of  $1 \times 10^{-2}$  N using nanoindentation. It can be found that pop-ins occur in the  $P$ – $h$  curves. And they occur in the Fe-rich, Cr-rich and Mo-rich  $M_2B$  when the load reaches  $4.62 \times 10^{-2}$ ,  $6.11 \times 10^{-2}$  and  $8.66 \times 10^{-2}$  N, respectively. Moreover, the hardness ( $H$ ) and Young’s modulus ( $E$ ) of  $M_2B$  are calculated according to the Oliver-Pharr method. The results are listed in Table 2. It can be known that, the  $H$  values of Fe-rich, Cr-rich and Mo-rich  $M_2B$  are 18.82, 21.22 and 24.86 GPa, respectively, and the  $E$  values of Fe-rich, Cr-rich and Mo-rich  $M_2B$  are 203.85, 225.77 and 255.57 GPa, respectively. The Mo-rich  $M_2B$  exhibits the best stiffness, followed by the Cr-rich  $M_2B$  and then the Fe-rich  $M_2B$ . Except for these, the “plasticity factor,  $\delta_A$ ” for quantifying the brittleness of  $M_2B$  are calculated. The  $\delta_A$  of Fe-rich, Cr-rich and Mo-rich  $M_2B$  is 0.25, 0.29 and 0.32, respectively. Compared to the Fe-rich  $M_2B$ , the Mo-rich and Cr-rich  $M_2B$  possess better toughness. As previous reports discussed, pure  $Fe_2B$  belongs to the bcc structure

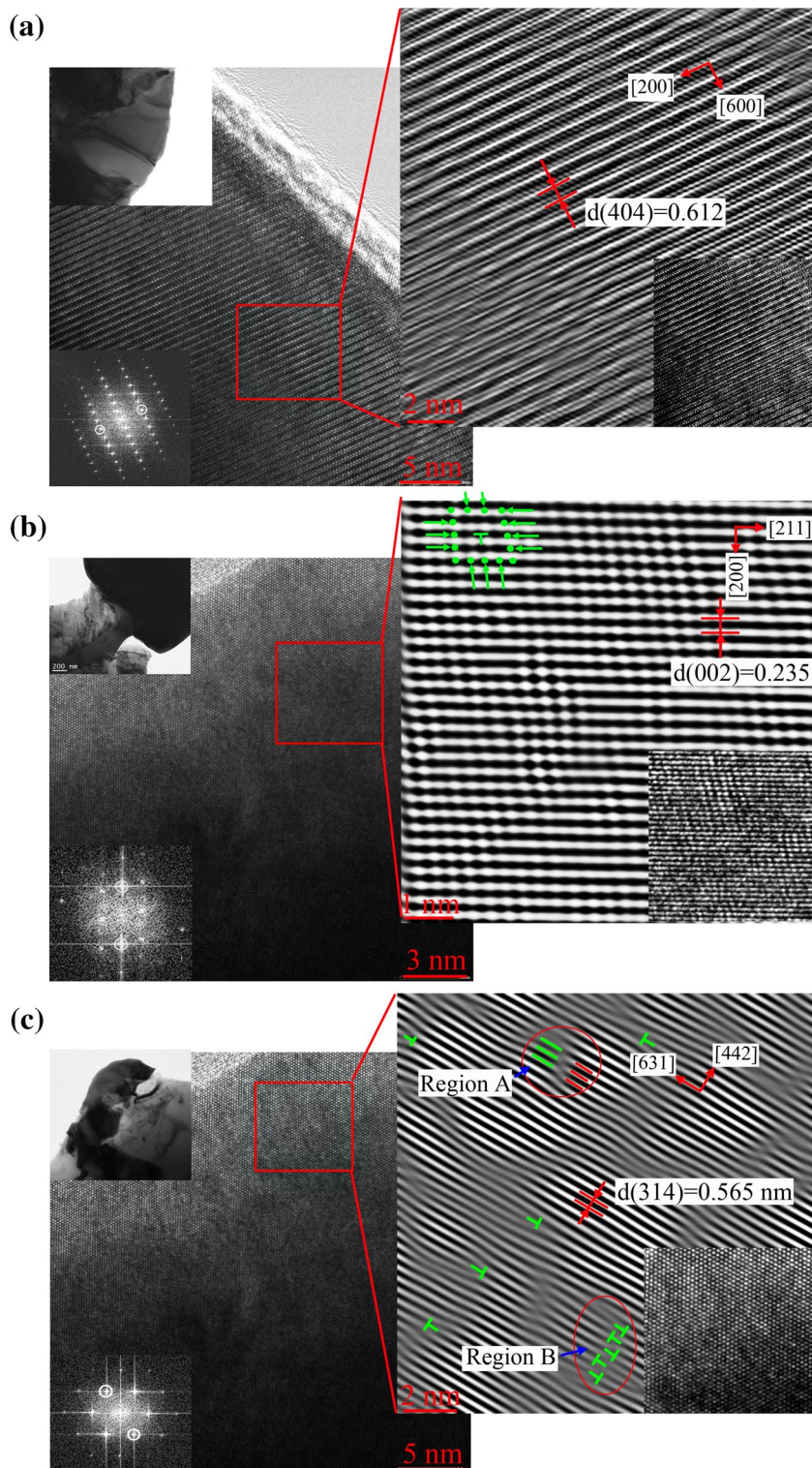


**Figure 3:** Bright-field TEM micrographs and corresponding selected area diffraction patterns (SADPs) of the multi-component  $M_2B$  in Fe–B alloy: (a), (b) and (c) show TEM micrographs of the Fe-rich, Mo-rich and Cr-rich  $M_2B$ , respectively; (d), (e) and (f) show corresponding SADPs of the Fe-rich, Mo-rich and Cr-rich  $M_2B$ , respectively.

(C16, CuAl2-type) and space group of  $I4/mcm$  (No. 140) [7–9, 19–21]. With the addition of alloying elements (Cr and Mo), the bcc  $Fe_2B$  changes to the bct or bco  $M_2B$ . This may affect the electron environment of B–B bond. As the B–B bond in  $M_2B$  crystal is weak in [002] direction, the change of electron

environment may strengthen the B–B bond, thus the  $\delta_A$  value of  $M_2B$  increase as a result. Furthermore, the formation of dislocation may also facilitate the improvement of  $H$  and  $E$  values in the  $M_2B$ . Just as shown in the reference [32], the formation of dislocation improved the strength of  $M_7C_3$ -type carbide owing

**Figure 4:** High-resolution transmission electron microscopy (HRTEM) images and Fourier-filtered transform (FFT) images of the multi-component  $M_2B$ : (a) Fe-rich  $M_2B$ ; (b) Cr-rich  $M_2B$ ; (c) Mo-rich  $M_2B$ .



to the increase of sub-grain boundaries. To some extent, the result reveals that the addition of Cr or Mo can improve the hardness and toughness of multi-component  $M_2B$ .

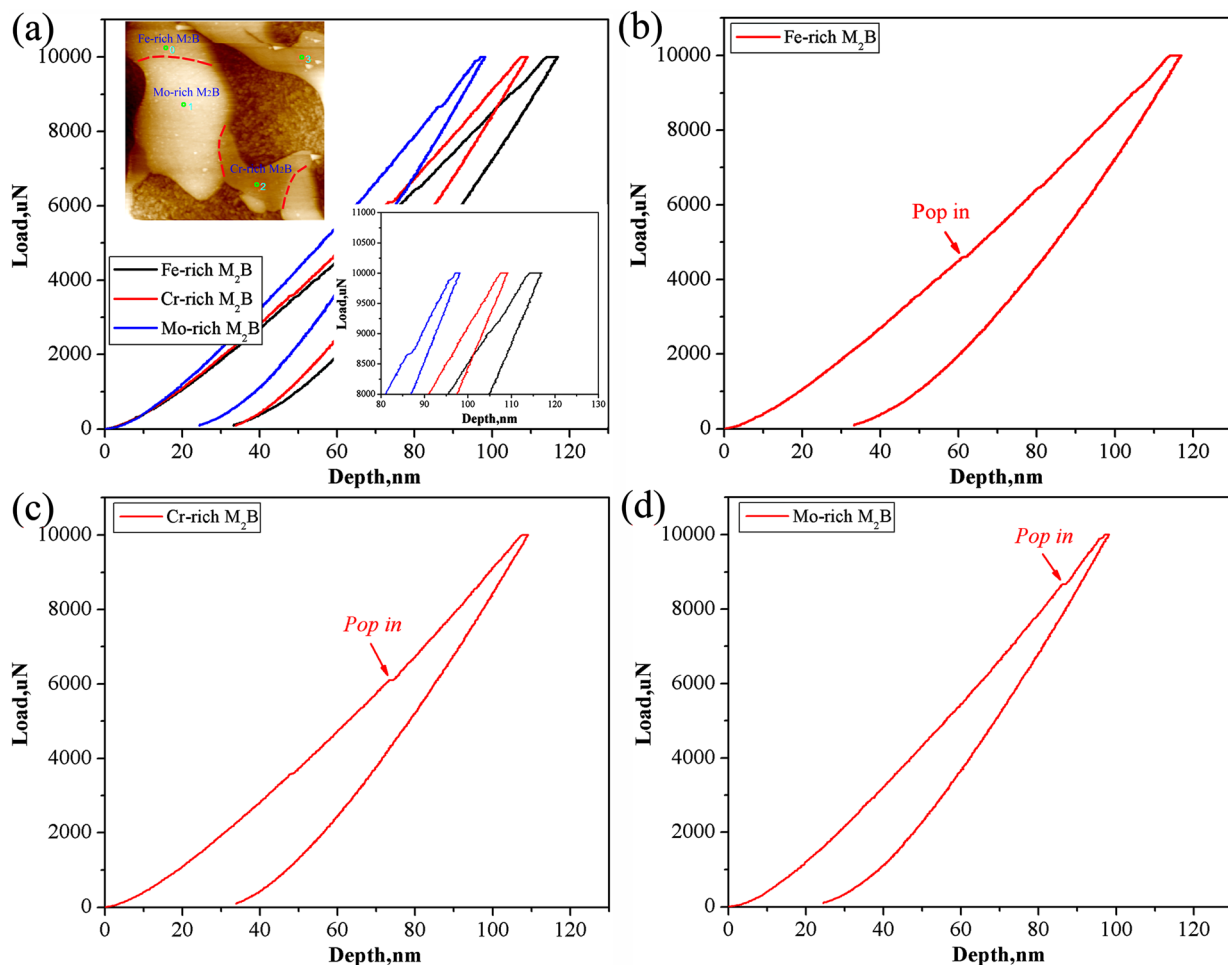
### Wear-resistance

Recently, nanoindentation was applied to evaluate wear-resistance of material [33–35]. In fact, the wear-resistance of material refers to its elastic strain, which can be calculated by the ratio of  $H/E$  [34]. Additionally, another parameter ( $H^3/E^2$ ) relating to yield pressure reflects the resistance to plastic deformation, which can also assess the wear-resistance of material [35]. In this work, the  $H/E$  and  $H^3/E^2$  of multi-component  $M_2B$  are calculated as shown in Table 3. Visibly, the Mo-rich  $M_2B$  has the largest  $H/E$  and  $H^3/E^2$ , followed by the Cr-rich  $M_2B$  and then the Fe-rich  $M_2B$ . To some extent, it indicates that the addition of Mo or Cr can improve the wear-resistance of multi-component  $M_2B$ .

To further verify above mentioned deduction, the worn morphologies of multi-component  $M_2B$  at different loads are observed using SEM, as shown in Fig. 6. When the normal load is 7 N [Fig. 6(a)], the Fe-rich and Mo-rich  $M_2B$  are shallowly scraped off, and there are nearly no obvious micro-cracks existing in the  $M_2B$ . As the normal load increases to 9 N [Fig. 6(b)], the Fe-rich  $M_2B$  occurs fracture, while the Mo-rich  $M_2B$  is lightly scratched. With the increase of normal load to 15 N [Fig. 6(c)], the Mo-rich and Fe-rich  $M_2B$  has been damaged, while the Mo-rich  $M_2B$  possesses higher wear-resistance

**TABLE 2:** Nanoindentation results (hardness, Young’s modulus and plasticity factor) for the multi-component  $M_2B$  of Fe–B alloy at the load of  $1 \times 10^{-2}$  N for 5 s.

Multi-component $M_2B$	$H$ (GPa)	$E$ (GPa)	$\delta_A$
Fe-rich $M_2B$	18.82	203.85	0.25
Cr-rich $M_2B$	21.22	225.77	0.29
Mo-rich $M_2B$	24.86	255.57	0.32



**Figure 5:**  $P$ – $h$  curves of the multi-component  $M_2B$  in Fe–B alloy through nanoindentation at the load of  $1 \times 10^{-2}$  N for 5 s.



relative to the Fe-rich  $M_2B$ . According to previous works [6, 10], the  $M_2B$  resists the abrasives and protects material from being shoved off during the wear process of Fe–B alloy. If the  $H/E$  and  $H^3/E^2$  of  $M_2B$  is low (e.g., Fe-rich  $M_2B$ ), the brittle  $M_2B$  can't resist the abrasives, as a result the  $M_2B$  tends to fracture easily. With higher  $H/E$  and  $H^3/E^2$ , the initiation of microcracks can be inhibited in the  $M_2B$ , owing to higher resistance to plastic deformation. Thus, the undamaged  $M_2B$  can resist the abrasive better.

## Conclusions

The multi-component  $M_2B$  is significant for the wear-resistance of Fe–B alloy. To study the basic characteristics of multi-component  $M_2B$ , the Fe–B alloy has been prepared using directional solidification, and then the microstructural evolution, crystallography, nanomechanical properties and wear-resistance of multi-component  $M_2B$  have been systematically investigated. The main conclusions are as follows:

- (1) During the solidification process, the multi-component  $M_2B$  of Fe–B alloy grows in the sequence of Mo-rich  $\rightarrow$  Fe-rich  $\rightarrow$  Cr-rich  $M_2B$ . The Mo-rich  $M_2B$  possesses white-coralloid structure, the Fe-rich  $M_2B$  has light-coloured rod-like structure, and the Cr-rich  $M_2B$  shows dark-coloured blocky structure.
- (2) In the Fe–B alloy, the Fe-rich  $M_2B$  has a stoichiometry of  $Fe_{1.59}Cr_{0.38}Mo_{0.08}(B, C)$  with the bct structure ( $a = b = 0.5109$  nm,  $c = 0.4249$  nm and  $c/a = 0.83$ ), the Mo-rich  $M_2B$  possesses a stoichiometry of  $Fe_{0.72}Cr_{0.27}Mo_{0.80}(B, C)$  with the bct structure ( $a = b = 0.5547$  nm,  $c = 0.4739$  nm and

$c/a = 0.85$ ), while the Cr-rich  $M_2B$  owns a stoichiometry of  $Fe_{1.48}Cr_{0.50}Mo_{0.02}(B, C)$  with the bco structure ( $a = 1.4583$  nm,  $b = 0.7379$  nm and  $c = 0.4245$  nm).

- (3) The addition of Cr or Mo leads to the formation of dislocation in multi-component  $M_2B$ , and the Mo addition can induce more dislocations relative to the Cr addition. This Cr/Mo addition facilitates the hardness and toughness of  $Fe_2B$ . The hardness  $H$  of Fe-rich, Cr-rich and Mo-rich  $M_2B$  is 18.82, 21.22 and 24.86 GPa, respectively, and the plasticity factor  $\delta_A$  of Fe-rich, Cr-rich and Mo-rich  $M_2B$  is 0.25, 0.29 and 0.32, respectively.
- (4) With the higher hardness and toughness, the initiation of microcracks can be inhibited in the  $M_2B$  owing to higher resistance to plastic deformation, thus the Mo-rich and Cr-rich  $M_2B$  possess higher wear-resistance relative to the Fe-rich  $M_2B$ .

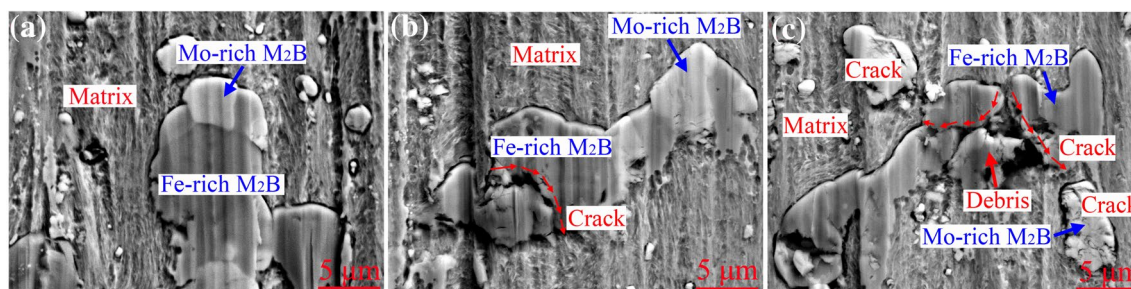
## Experimental procedures

### Directional solidification

The directional solidification has been widely applied to analyze microstructure evolution or growth mechanism of alloy [36–38]. In the present work, a Y-type block ingot was prepared using vacuum induction furnace, depending on a nominal chemical composition. Subsequently, some  $\phi 4$  mm  $\times$  10 mm bars were spark cut from the ingot. The directional solidification test was performed on a Bridgman-type directional furnace. The prepared bars were firstly loaded into an alumina tube crucible with inner diameter of 8 mm and length of 150 mm, followed by putting into induction furnace column, and then heated up to 1600 °C for 10 min. Finally, the tube crucible was quenched into cooling liquid with the withdrawal rate of 15  $\mu$ m/s for 20 min. The schematic of directional solidification and corresponding mushy zone is shown in Fig. 7. The final composition of sample was measured by X-ray fluorescence spectrometer (S8 TIGER), the result was listed as follows (wt%): B 1.80, C 0.41, Cr 5.36, Mo 8.34, Al 0.71, Si 1.03, V 0.97, Mn 0.62 and Fe in balance.

**TABLE 3:**  $H/E$  and  $H^3/E^2$  of the multi-component  $M_2B$  in Fe–B alloy.

Multi-component $M_2B$	$H/E$ ( $\times 10^{-2}$ )	$H^3/E^2$ (GPa $\times 10^{-1}$ )
Fe-rich $M_2B$	9.23	1.60
Cr-rich $M_2B$	9.40	1.87
Mo-rich $M_2B$	9.73	2.35



**Figure 6:** Worn morphologies of the multi-component  $M_2B$  in Fe–B alloy at the loads of 7 ~ 15 N, using SiC as abrasive: (a) 7 N; (b) 9 N; (c) 15 N.

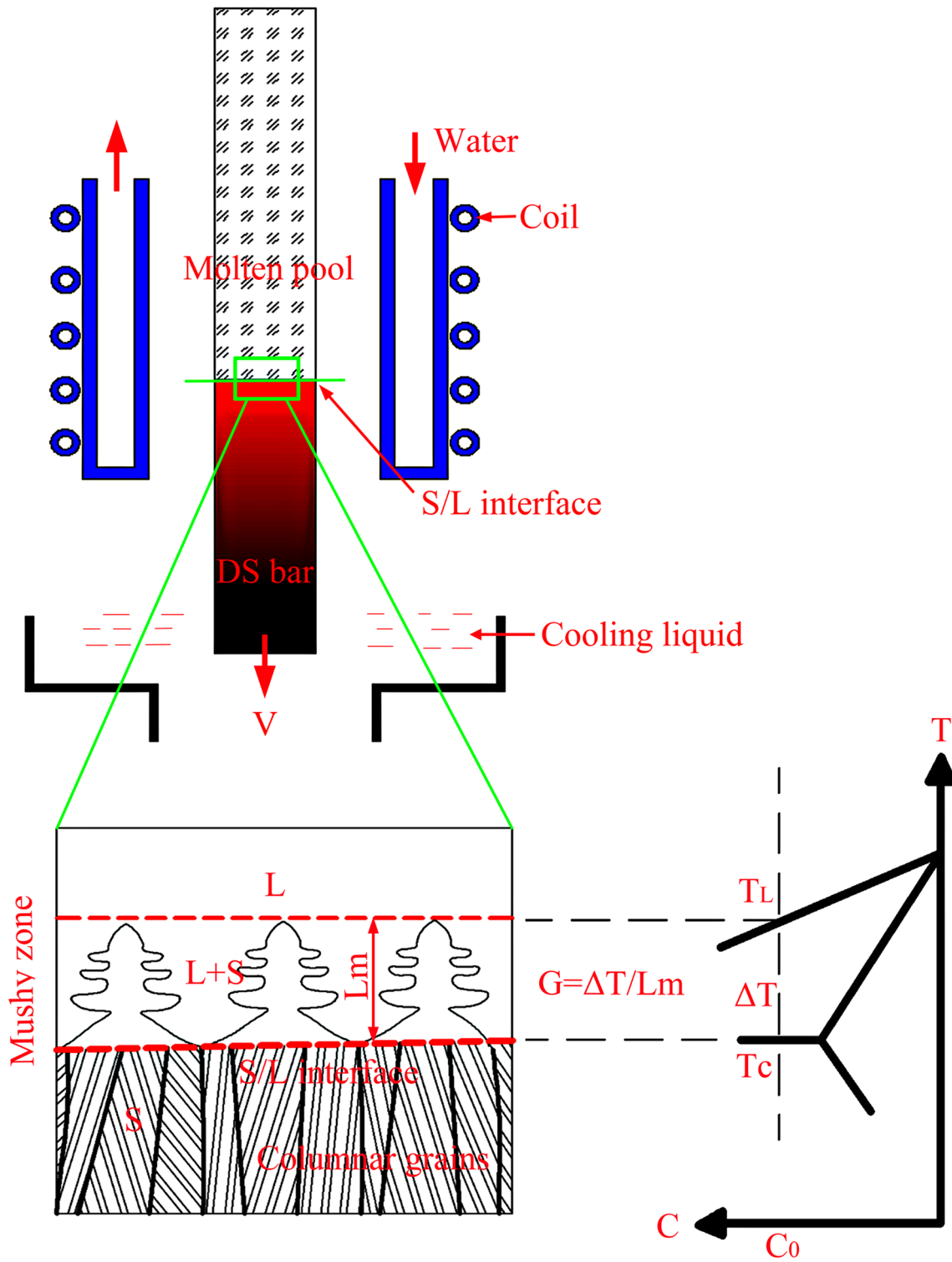


Figure 7: Schematic of directional solidification and corresponding mushy zone.



### Characterization and phase diagram calculation

After wire-cutting, grinding and polishing, the DS sample was etched with 4 vol% nital solution for metallographic characterization. Subsequently, the etched DS sample was observed by light microscopy (OM, LEICA DMI 5000M) and scanning electron microscope (SEM, JSM-7000F). To measure alloying element distribution, electron probe microanalysis (EPMA, JXA-8230) equipped with wavelength dispersive X-ray (WDX) was applied. The EPMA measurement was scanned at 10 keV and 100 nA with the scanning range of 40–75 Å and step size of 0.01 Å. High-resolution transmission electron microscope (HRTEM, JEM-F200) was used to analyze the crystallography of multi-component M<sub>2</sub>B. Thin foils with the thickness of 0.5 mm for transmission electron microscope observation were prepared using spark erosion machine. Subsequently, they were mechanically ground by hand to about 50 μm, followed by punching out some discs of 3 mm in diameter. The thin discs were twin-jet electropolished to electron transparency at 75% methanol, 5% perchloric acid and 20% glycerol maintained at 0 °C. Some thin foils were then thinned to about 30 μm for HRTEM observation, which was operated at 400 keV with an incidence angle of 4° for 30 min. The high-resolution images were gathered at a point-to-point resolution of ~1.7 Å, and analyzed subsequently using Digital Micrograph software.

Thermo-Calc calculation software, containing a sophisticated database (thermo-chemical databank) and programming interface package, is often utilized to calculate the equilibrium phase diagram [39]. In this work, the POLY-3 calculation (a type of calculation in Thermo-Calc software with TCFE-7 database.) was chosen. The selected parameters mainly included the temperature range of 500–1750 °C and the pressure of 150 kPa. Calculated criterion was the minimization of Gibbs free energy.

### Nanoindentation

The metallographic sample was prepared for nanoindentation on the UMIS instrument and triboindenter (from Hysitron Inc.) at 1 × 10<sup>-2</sup> N load for 5 s, following the ASTM E2546. The displacement (depth) of indenter was continuously monitored so as to record the load–displacement (*P*–*h*) curve. All measurements were performed in air currents.

The hardness (*H*) and Young’s modulus (*E*) of multi-component M<sub>2</sub>B were calculated using the Oliver–Pharr method via the Eqs. 1–4 [40].

$$H = \frac{P_{\max}}{Ah_c} \tag{1}$$

where *P*<sub>max</sub> is the maximum load, *A* is the contact area between material and indenter at the load of *P*<sub>max</sub>, and *h*<sub>c</sub> is the depth. The *h*<sub>c</sub> was calculated by the Eq. 2.

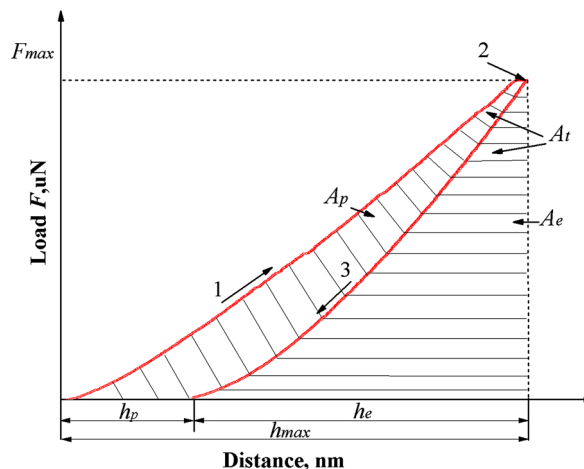


Figure 8: Schematic of the *P*–*h* curve for Berkovich indenter: (1) Loading curve; (2) Holding curve at the *P*<sub>max</sub>; (3) Unloading curve.

$$h_c = h_{\max} - \varepsilon \frac{P_{\max}}{S} \tag{2}$$

where  $\varepsilon$  is 0.75 for the Berkovich indenter, *S* is the initial unloading stiffness at the load of *P*<sub>max</sub>.

The *E* could be reduced by the Eqs. 3 and 4.

$$\frac{1}{E_{\text{eff}}} = \frac{(1 - \nu^2)}{E} + \frac{(1 - \nu_i^2)}{E_i} \tag{3}$$

$$E_{\text{eff}} = \frac{1}{\beta} \frac{\sqrt{\pi}}{2} \frac{S}{\sqrt{A}} \tag{4}$$

where *E<sub>i</sub>* is the modulus of indenter, and  $\nu$  and  $\nu_i$  are Poisson’s ratio of material and indenter, respectively. *E*<sub>eff</sub> is the effective elastic modulus.  $\beta$  is 1.034 for the Berkovich indenter.

Additionally, the plasticity characteristic of multi-component M<sub>2</sub>B (plasticity factor,  $\delta_A$ ) was calculated by the Eq. 5 [41].

$$\delta_A = \frac{A_p}{A_t} = 1 - \frac{A_e}{A_t} \tag{5}$$

where *A<sub>e</sub>* and *A<sub>t</sub>* are the areas under the unloading and loading curves, respectively (*A<sub>p</sub>* = *A<sub>t</sub>* – *A<sub>e</sub>*), as shown in Fig. 8.

### Abrasive wear behavior

Wear tests were carried out on a ML-100 type pin-on-disk apparatus with pin sample against SiC abrasive paper (240 mesh/60 μm, 2600 HV [17, 18]). The pin samples with the size of ϕ6 mm × 10 mm were spark cut from the DS sample. The tests were performed in a spiral track with main parameters of sliding distance of 6 m, rotating rate of 60 r·min<sup>-1</sup>, sliding rate of 4 mm·r<sup>-1</sup> and normal loads of 7–15 N.

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## Data availability

The datasets generated during and/or analyzed during the current study are available from the corresponding author on reasonable request.

## Declarations

**Conflict of interest** There are no conflicts to declare.

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