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# Hydrogen dissociation degree in the Grimm-type glow discharge measured by optical emission spectroscopy

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**Abstract.** The hydrogen dissociation degree is measured by means of optical emission spectroscopy in the cathode sheath of cylindrical abnormal Grimm-type DC glow discharge operating in the hydrogen–argon mixture at low pressure. The degree of dissociation, ranging from 80% in the close vicinity of the cathode and decreasing to 60% towards the negative glow, is obtained from the intensity ratio of the hydrogen Balmer H<sub> $\gamma$ </sub> line and the diagonal Fulcher- $\alpha$  molecular bands. The cathode sheath thickness is extracted from the electric field strength distribution measured using Stark spectroscopy of the hydrogen Balmer H<sub> $\alpha$ </sub> line. An electron temperature of 2.7 eV in the negative glow region is estimated by observing the intensity ratio of the diagonal Fulcher- $\alpha$  bands and the H<sub> $\alpha$ </sub> line.

### 1 Introduction

The cathode sheath (CS) of glow discharges (also called cathode fall or cathode dark space) is a low-intensity region next to the cathode surface, where electrons from the cathode gain sufficient energy for the collisional ionization of atoms. This results in the separation of the opposite charges and additional flux of charged particles, electrons towards the bright negative glow region, and ions and charge-exchange generated neutrals towards the cathode. The cathode sheath is also the region where a major voltage drop takes place to create a macroscopic electric field, accelerate charged species, and further sustain the glow discharge [1–4]. Therefore, the processes occurring in this region are most relevant for the applications in spectroscopic analysis and surface processing (e.g. thin film deposition, plasma etching, and depth profiling), as well as in the development of the plasma-based ion and negative hydrogen ion sources [2, 3, 5, 6]. In latter applications, the degree of dissociation plays an important role. The processes leading to hydrogen dissociation in the discharge can be investigated knowing other relevant discharge parameters, and optical emission spectroscopy (OES) provides a simple, non-intrusive, and reliable way to measure them. In our previous work [7], we used OES to measure electric field strength in the cathode sheath and gas temperature distribution in hydrogen and hydrogen–argon plane cathode Grimmtype DC glow discharge source (GDS).

In this paper, we present the results of the hydrogen degree of dissociation in the cathode sheath and electronic temperature in the negative glow of a Grimmtype GDS in a hydrogen-argon mixture at low pressure, described in detail in [8] and references therein. Throughout the CS region, we used the intensity ratio of the hydrogen Balmer  $H_{\nu}$  line and the Fulcher- $\alpha$ molecular band system, as proposed in [9, 10]. The intensity of the Fulcher- $\alpha$  system is determined using all three branches (P, Q, and R) of  $d^3\Pi_u$ ,  $\nu' = 0, 1, 2, 3$  $\rightarrow a^{3}\Sigma^{+}_{q}, \nu'' = 0, 1, 2, 3$  system. The border between the cathode sheath and negative glow (NG) region is determined using previously employed techniques for the measurement of the electric field strength distribution from the spectral profiles of the  $H_{\alpha}$  line [7]. To complete this study we also estimated the electron temperature in the negative glow, using the intensity ratio of the  $H_{\alpha}$  line and Fulcher- $\alpha$  system.

# 2 Experimental

Let us begin by mentioning some important details regarding the modified Grimm GDS used in this study, whose detailed description is given earlier [11, 12]. The modification of the standard Grimm lamp [13, 14] with a plane cathode adjacent to the hollow cylindrical anode is made following the design of Ferreira et al. [15], by inserting the cathode into the cylindrical anode. This allows, besides conventional OES from the end of the

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discharge, in the direction of the discharge axis (socalled end-on), optical observations parallel to the cathode surface through the longitudinal slots in the anode body (side-on). In this way, different zones of the discharge, namely CS and NG, can be optically observed merely by translating the lamp along the discharge axis, resulting in the information on axial distributions of various plasma parameters.

For the particular conditions of this experiment, the discharge with copper cathode is established in the 95/5 per cent hydrogen/argon mixture at a flow of  $300 \text{ cm}^3/\text{min}$ . At a pressure of 4.5 mbar, the currentstabilized power supply sustained the discharge at 12 mA and 775 V. For side-on observations of the radiation intensity axial distributions and Stark splitting of hydrogen Balmer lines, the source is translated in 0.125 mm steps for the H<sub>a</sub> line recordings and 0.250 mmfor the Hy and Fulcher- $\alpha$  system, see schematics in Fig. 1. For hydrogen Balmer lines recording, a plastic linear polarizer parallel with the discharge axis (i.e. electric field direction) is placed between the source and the detection system to select  $\pi$ -polarized Stark components. The light from the particular position along the discharge axis is focused with unity magnification onto the 20  $\mu$ m entrance slit of the 2 m spectrometer with 0.37 nm/mm reciprocal dispersion in the second diffraction order of a 651 g/mm reflection grating, resulting in a Gaussian instrumental profile of 0.014 nm full width at half maximum. Thermoelectrically cooled CCD at -10 °C (Hamamatsu, 2048  $\times$  506 pixels, 12  $\times$  $12 \ \mu m$  pixel size) is used as a light detector, and the data are digitally collected and processed. For intensity measurements, the spectrometer-detector system is calibrated by a standard tungsten lamp.

## 3 Results and discussion

# 3.1 Distribution of the electrical field strength in the cathode sheath

To conduct our study on the hydrogen dissociation degree distribution in the CS region from the total intensities of the Fulcher- $\alpha$  system and hydrogen Balmer H<sub> $\gamma$ </sub> line, we needed to determine the thickness and boundary of the CS region first. To this end, we used the criteria of the external electric field decrease from the maximum near the cathode surface towards zero at the border with the NG region. Stark polarization spectroscopy of hydrogen Balmer H<sub> $\alpha$ </sub> line is used to measure the electric field strength distribution in the CS.

To obtain the electric field value at a given point in the CS, we recorded the  $H_{\alpha}$   $\pi$ -polarized line profile, see Fig. 2a, and fitted it with the model function (1) from [7]. The model takes into account the entire  $H_{\alpha}(\pi)$  symmetrical Stark splitting manifold, as well as the small central peak which originates from the part of the discharge protruded through the side-on observation anode slot into the field-free region. The fitting procedure takes into account the entire profile of the Stark-split line and yields precise information on the local electric field strength, see also [16]. It should be noted here that the  $H_{\nu}$  profile, used in this work for the measurement of the hydrogen dissociation degree, is also Stark-split in the CS, see example in Fig. 2b, and the information on the local electric field could be extracted from the  $H_{\nu}$   $\pi$ -polarized line profile by using a simple spectroscopic formula [11]:

$$E[V/cm] = 2.51 \times 10^4 \times \Delta \lambda_{p-p} [nm], \qquad (1)$$



Fig. 1 The central part of modified Grimm GDS and optical set-up for side-on OES



Fig. 2 The examples of side-on recorded  $\pi$ -polarized experimental profiles of the **a** H<sub> $\alpha$ </sub> line with its best fit (solid line) by the model function [7]; **b** H<sub> $\gamma$ </sub> line with indicated peak separation  $\Delta\lambda_{p-p}$ , Eq. (1). The corresponding values of electric field strength *E* and distance *d* from the cathode are given in the legends. **c** Electric field strength axial distribution in the CS region obtained from the H<sub> $\alpha$ </sub> line together with experimental uncertainties. For better visibility, the uncertainties are enlarged five times. Experimental conditions are given in Sect. 2

where  $\Delta\lambda_{p-p}$  is the peak separation, measured from the experimental profile as a distance between convolved Stark components, shifted equally to the red and blue wavelength side, see Fig. 2b. However, due to the low intensity and weaker resolution of Stark components, the electric field values obtained from the H<sub> $\gamma$ </sub> line are systematically lower, leading to a discrepancy of about 15% in comparison with the H<sub> $\alpha$ </sub> results. By polynomial fitting of the electric field distribution obtained from the axial recording of the H<sub> $\alpha$ </sub>( $\pi$ ) profile, we estimated the CS thickness of 1.75 mm, see Fig. 2c.

Although according to the simplest predictions the electric field distribution in the CS region of a glow discharge should be a linear function of the distance to the cathode, numerous experimental findings showed that this need not be the case (see, e.g. [16-19]). The departure from the linearity proved to be a consequence of a series of intertwined causes which have to be analysed in a kinetic approach like, e.g. the iterative kinetic model [18, 19].

As stated in Sect. 3.1, the values of the electric field strength E are obtained as one of the fitting parameters to the experimental profiles of the mentioned model function suitable for the  $H_{\alpha}$  spectral line profile. The uncertainty for this (and all other) model function fitting parameter(s) we determined following [20] via Monte Carlo simulations of the synthetic data sets using the experimental uncertainties of the recorded profiles. Within this approach, we took that the uncertainty of each experimental point from a recorded profile is  $\varepsilon_{\rm tot} = \sqrt{\varepsilon_{\rm sys}^2 + \varepsilon_{rnd}^2}$ , where  $\varepsilon_{\rm sys}$  is the systematic error introduced by the detector, while  $\varepsilon_{rnd}$  is the statistical error calculated at the 95% confidence level from the standard deviation of a series of ten profile measurements at the observed wavelength.

#### 3.2 Fulcher-α diagonal band spectra

Figure 3a illustrates the potential energy curves of the excited states in the H<sub>2</sub> molecule (electronic and vibrational levels) belonging to the Fulcher- $\alpha$  excitation and emission processes, according to data given in [21, 22]. The electron impact on H<sub>2</sub> causes  $d^3\Pi_u$ ,  $\nu' = 0-3 \leftarrow$  $X^{1}\Sigma^{+}_{q}$ ,  $\nu = 0$  transitions, while the radiative transitions  $d^3\Pi_u$   $(\nu', N') \rightarrow a^3\Sigma^+{}_g (\nu'', N'')$  constitute Fulcher- $\alpha$  emissions. For the H<sub>2</sub> $d^3\Pi_u$  state,  $\Lambda$ -doubling causes the rotational levels to split into two components, denoted as  $d^3\Pi_u^-$  and  $d^3\Pi_u^+$ , as shown in Fig. 3b. The spontaneous emission  $d^3\Pi_u^- \to a^3\Sigma_g^+$ results in Q-branch, while  $d^3\Pi_u^+ \to a^3\Sigma_g^+$  constitutes P and R branches, see Fig. 3b. Given the perturbation of the  $d^3\Pi^+_{\ u}$  state in interaction with  $e^3\Sigma^+_{\ u}$ , the transition probabilities for P and R branches differ from Hönl-London factors [23-25], and the O-branch of the Fulcher- $\alpha$  hydrogen band is therefore considered more suitable for, e.g. temperature distribution measurements in the CS region [7]. Vibrational term values in certain electronic states are given in [26]

Figure 4 shows experimental spectra of P, Q, and R branch lines of the Fulcher- $\alpha$  diagonal bands with a)  $\nu' = \nu'' = 0$  and b)  $\nu' = \nu'' = 1$ , respectively, measured in the cathode sheath of our discharge. Clear spectra with distinguished rotational lines (wavelengths from [22]) of P, Q, and R branches in the visible 590–645 nm wavelength region, are obtained. The intensities of all branch lines of the  $d^3\Pi_u, \nu' \rightarrow a^3\Sigma^+_{\ g}, \nu'' (\nu' = \nu'' = 0, 1, 2, 3)$  electronic transition are used in this study, see also [7].

#### 3.3 Degree of dissociation

Not many studies can be found about the hydrogen dissociation degree in the literature. Most of the studies are carried out in pure hydrogen discharges or argon-hydrogen mixtures at significantly different discharge conditions. For example, for the microwave discharge source, the degree of dissociation of hydrogen is found in the range of 30-90% [27], while this value is near 100% for the cascaded arc plasma [28]. For the typical glow discharge/optical emission spectroscopy (GD-OES) conditions in the argon-hydrogen mixture, the estimated value for the degree of dissociation is 67% [29]. It was also reported that the hydrogen dissociation rate in an argon-hydrogen glow discharge is considerably lower than in a pure hydrogen discharge for pressures above 5 Torr [30], which is the pressure range above our experimental conditions.

The degree of dissociation is the ratio of the dissociated fraction and the total amount of the substance.



Fig. 3 a Potential curves of the hydrogen molecule with its vibrational states belonging to the Fulcher- $\alpha$  system transitions; b Rotational structure of the Fulcher- $\alpha$  bands system of H<sub>2</sub> for  $\nu' = \nu'' = 0$ . The data are taken from [21, 22]



Fig. 4 Several rotational lines in the emission spectra of P, Q, and R-branches of the **a**  $d^3\Pi_{\rm u}, \nu' = 0 \rightarrow a^3\Sigma^+{}_{\rm g}, \nu'' = 0$  and **b**  $d^3\Pi_{\rm u}, \nu' = 1 \rightarrow a^3\Sigma^+{}_{\rm g}, \nu'' = 1$  band. Experimental conditions are given in Sect. 2

For the hydrogen discharges, it is the ratio between the dissociated hydrogen and the total sum of atomic and molecular hydrogen. The density of the atomic hydrogen is associated with Balmer lines emission, while the molecular density corresponds to the emission of molecular radiation (Fulcher- $\alpha$  band transition:  $d^3\Pi_u \rightarrow a^3\Sigma^+_{\rm g}$ , v' = v'' = 0–3,  $\lambda = 590$ –645 nm). Some earlier studies [9, 10] indicate that the hydrogen dissociation degree can be estimated from the total intensity ratio of the hydrogen Balmer H<sub> $\gamma$ </sub> line and Fulcher- $\alpha$  system.

The spontaneous emission intensity is the number of photons emitted by a unit volume per second over all directions. The intensity of a spectral line of transition is proportional, in an optically thin case, to the product of the population density of the upper state and the transition probability for spontaneous emission. According to collisional radiative models, the population density depends on the plasma parameters  $T_{\rm e}$  and

 $n_{\rm e}$ , and is proportional to the population coefficients which describe the coupling of the population density to the ground state of the particle. The effective emission rate  $X^{\rm eff}$  is the product of population coefficients of the particle and the transition probability for spontaneous emission. Finally, the intensities of atomic and molecular spectral lines are proportional to the densities of atoms [H] or molecules [H<sub>2</sub>] multiplied by the corresponding effective emission rates  $X^{\rm eff}$ . Thus, for our experimental conditions, we considered the intensity ratio of  $H\gamma/H_2$  suitable for the determination of the density ratio between hydrogen atoms and molecules [H]/[H<sub>2</sub>].

The intensity ratio, according to [9, 10], is:

$$\frac{I_{\gamma}}{I_{\text{Ful.}}} = \frac{X_{\gamma}^{\text{eff}}}{X_{H_2}^{\text{eff}}} \frac{[\text{H}]}{[\text{H}_2]},\tag{2}$$



Fig. 5 a Total intensities of the  $H_{\gamma}$  line and Fulcher- $\alpha$  bands (v' = v'' = 0, 1, 2, 3) and b Dissociation degree of hydrogen against the distance from the cathode through the CS region. Experimental conditions are given in Sect. 2

where  $X_{\gamma}^{\text{eff}}$  and  $X_{H_2}^{\text{eff}}$  denote the effective emission rates of the H<sub>\gamma</sub> line and Fulcher- $\alpha$  system, respectively, calculated following the collisional radiative model [9], while [H] and [H<sub>2</sub>] stand for the number densities of hydrogen atoms and molecules, respectively. With the high spectral resolution of our optical detection system, we were able to measure the total intensity of the Fulcher- $\alpha$ band and calculate the ratio  $\frac{I_{\gamma}}{I_{H_2}}$ . Moreover, according to [9] and references therein, the ratio of the emission rate coefficients  $\frac{X_{H_2}^{\text{eff}}}{X_{H_2}^{\text{eff}}}$  is barely dependent on the electron temperature  $T_e$  and electron density  $n_e$  in the range  $5 \cdot 10^{16} \text{ m}^{-3} - 5 \cdot 10^{18} \text{ m}^{-3}$  and approximately equals 1.3, according to Fig. 5 from [9]. Using this value for the  $\frac{X_{\gamma}^{\text{eff}}}{X_{H_2}^{\text{eff}}}$ , the ratio  $\frac{[\text{H}]}{[\text{H}_2]}$  is obtained. Knowing the relationship between the neutral particle density ratio and the degree of dissociation [10, 31], we estimated the degree of dissociation in the cathode sheath as follows:

$$\mathbf{D} = \frac{[H]}{[H] + 2[H_2]} = \frac{\frac{[H]}{[H_2]}}{\frac{[H]}{[H_2]} + 2}.$$
 (3)

Figure 5 presents our results for total intensities of the hydrogen Balmer  $H_{\gamma}$  line and Fulcher- $\alpha$  system (P, Q, and R branches) throughout the CS region. The results show that the degree of dissociation varies from 60 to 80% and decreases towards the end of the CS region.

#### 3.4 Electron temperature

To understand the behaviour and to model gas discharges and low-density plasmas, it is required to know the values of discharge parameters like, e.g. electron number density, electron temperature and the electron energy distribution. In Grimm-type glow discharge, earlier measurements performed by a Langmuir probe in argon and neon found excitation temperatures in the range from 0.6 to 0.85 eV for Ar I and 0.5 to 0.9 eV for Ne I [32]. Furthermore, measured excitation temperatures of Ar II were in the range of 3.5–4.4 eV [32], indicating the existence of at least two distinguishable electron groups. Generally three groups of electrons are assumed to exist in a Grimm GDS: the primary group consisting of highly energetic electrons, the group of secondary electrons possibly split into many subgroups. and the group of thermalized electrons. The thermalized and one secondary electron group encountered in the Langmuir probe measurements are reported to have an average energy of 0.5 eV and 4.5 eV, respectively [32]. Another study of the Grimm-type glow discharge using Langmuir probe [33] reports the typical electron temperature of 0.45 eV.

In the negative glow plasma of our Grimm GDS we estimated the electron temperature using the intensity ratio of the Fulcher- $\alpha$  system and hydrogen Balmer line  $H_{\alpha}$ . This spectroscopic method is proposed in [34], assuming that the electron energy distribution function is close to Maxwell's. The main mechanisms of excitation of the  $d^3\Pi_u$  state are the electron impact excitation of the hydrogen molecule from the ground state  $(d^3\Pi_u)$  $\leftarrow X^1 \Sigma_a^+$ ) and the radiative decay into the  $a^3 \Sigma_a^+$ . These two processes are dominant in comparison to cascade processes from higher excited states, collisional deexcitation, and other secondary processes for Fulcher- $\alpha$  system in low-temperature plasmas, see, e.g. [34, 35]. The  $H_{\alpha}$  emission is mainly due to dissociative excitation and dissociative ionization from  $H_2$  molecules by electron impact. For this process, the threshold energy is 13.47 eV while the maximum of its cross section  $\sigma_{\rm em}$ is  $(4.2 \pm 0.8) \ 10^{-22} \ {\rm m}^2$  at the electron energy of 15.6 eV, which is close to the threshold of excitation; semiempirically (7  $\pm$  2)  $10^{-22}$  m² [35]. The cross sections for the excitation of atomic and molecular lines have different dependencies on electron energy [35, 36], see Fig. 6a. The emission cross section  $\sigma_{H_2}$  has a sharp peak near the threshold energy and decreases rapidly as the excitation energy increases while the  $\sigma_{H_{\alpha}}$  increases gradually.

The intensity of the Fulcher- $\alpha$  can be expressed as  $I_{H_2} = \text{const. } n_e n_{H_2} \langle \sigma_{H_2} v \rangle$  while for  $H_{\alpha}$  stands: $I_{H_{\alpha}} = \text{const. } n_e n_{H_2} \langle \sigma_{H_{\alpha}} v \rangle$ , with the rate coefficients  $\langle \sigma_{H} v \rangle$  derived from the emission cross sections  $\sigma_{H}$  using the Maxwellian distribution function:

$$\langle \sigma_H v \rangle = 1.2 \ 10^8 \pi^{-\frac{1}{2}} \int_{\Delta E}^{\infty} E \sigma_H T_e^{-\frac{3}{2}} e^{-\frac{E}{T_e}} \mathrm{d}E \qquad (4)$$

where E and  $\Delta E$  are the excitation and threshold energy, respectively, and  $T_{\rm e}$  is the electron temperature. The ratio of intensities can be expressed as a relationship of the rate coefficients  $I_{H_2}/I_{H_{\alpha}} = \langle \sigma_{\rm H_2} v \rangle / \langle \sigma_{\rm H_{\alpha}} v \rangle$  and consequently as a function of the electron temperature. Therefore, the intensity ratio of



**Fig. 6** a Excitation functions of the Fulcher- $\alpha$  system and Balmer H<sub> $\alpha$ </sub> line in the energy range 0–100 eV. The data are taken from [35, 36]. **b** The intensity ratio of the Fulcher- $\alpha$ system to the H<sub> $\alpha$ </sub> line as a function of electron temperature. The data are taken from [34]

the Fulcher- $\alpha$  system  $(d^3\Pi_u - a^3\Sigma_g)$  to the H<sub> $\alpha$ </sub> emission is sensitive to the electron temperature, see Fig. 6b.

The estimated value of the electron temperature  $T_{e}$ in the negative glow in our discharge conditions is 2.7 eV, while the estimated gas temperature is around 900 K, see details in [7]. The addition of hydrogen to an argon discharge has a significant effect on its analytical characteristics inducing, e.g. a change in the electron energy distribution function [29]. This is a possible explanation for  $T_{\rm e}$  being so much higher in an Ar- $H_2$  discharge than in pure Ar. From the above results, one can consider that the measurement of the intensity ratio of Fulcher- $\alpha$  system to  $H_{\alpha}$  emission may be useful for the estimation of the electron temperature in hydrogen discharge at low pressures. Finally, it should be underlined that the theory applied is based on the assumption of direct excitation of spectral transitions by electron impact and dissociative excitation of hydrogen molecules. These processes mostly dominate in a weakly ionized hydrogen plasma at low pressures.

#### 4 Summary and conclusion

This study aims to estimate the degree of dissociation in the CS region and electron temperature in a Grimm-type glow discharge. The intensity of all three branches (P, Q, and R) of the Fulcher- $\alpha$  system,  $d^3\Pi_u$ ,  $\nu' = 0, 1, 2, 3 \rightarrow a^3\Sigma^+_{g}, \nu'' = 0, 1, 2, 3$  is used in combination with intensities of H $\gamma$  and H $_{\alpha}$  hydrogen Balmer lines to obtain results. All lines of P, Q, and R branches of the Fulcher- $\alpha$  system are located in the visible 590–645 nm wavelength region, with sufficient resolution and a high signal-to-noise ratio. The boundary of the cathode sheath is found using Stark spectroscopy for the determination of the electric field strength distribution from the H<sub> $\alpha$ </sub> line.

To determine the degree of dissociation in the cathode sheath region we used the intensity ratio of the hydrogen Balmer H<sub> $\gamma$ </sub> line and Fulcher- $\alpha$  system as suggested in earlier studies [9, 10]. Obtained value varies between 80 and 60% within the cathode sheath region and is comparable with previous results [27–30].

Electron temperature in the negative glow region is estimated from the intensity ratio of the Fulcher- $\alpha$  system and hydrogen Balmer H<sub> $\alpha$ </sub> line [34]. The estimated electron temperature T<sub>e</sub> in the negative glow in our discharge conditions is 2.7 eV, while the estimated gas temperature [7] is around 900 K.

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# Author contribution

MMV contributed to conceptualization, methodology, investigation, and writing. GLM contributed to conceptualization, methodology, investigation, and writing. IRV contributed to conceptualization, methodology, and writing—review. DS contributed to conceptualization, methodology, and supervision.

**Data Availability Statement** This manuscript has no associated data or the data will not be deposited. [Authors' comment: The datasets supporting the conclusions of this article are available from the corresponding author upon reasonable request.]

#### Declarations

**Conflict of interest** The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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