Synthesis and Study of Bismuth-Containing High-Silica Glass by the IR Spectroscopy Method1

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Abstract—Bismuth-containing high-silica glasses (BCHSGs) based on the matrices of high-silica nanopo rous glasses have been synthesized. The structure of BCHSG plates has been investigated by means of the IR spectroscopy method in the frequency range $4000-400$ cm⁻¹. The bands corresponding to vibrations of the rous glasses have been synthesized. The structure of BCHSG plates has been investigated by means of the IR spectroscopy method in the frequency range 4000–400 cm⁻¹. The bands corresponding to vibrations of the Bi–O bond Bi-O bonds in structural units [BiO₃] and [BiO₆] were found on the spectral transmission curves of the synthesized BCHSGs. Besides, peaks corresponding to the presence of the α -Bi₂O₃ phase in the glass were revealed. It has been established that, depending on the BCHSG thermal treatment and the bismuth concen the sized BCHSGs. Besides, peaks corresponding to the presence of the α -Bi₂O₃ phase revealed. It has been established that, depending on the BCHSG thermal treatment and the tration in glass, changes of the glass st

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INTRODUCTION

Glasses doped with bismuth are characterized by luminescence in a broad spectral range from 450– 750 [1–8] to 1000–1700 nm [3–5, 8–11]. Different methods of glass fabrication are known: for instance, powder-in-tube technology [5, 11, 12]; modified chemical vapor deposition (MCVD) [5, 13, 14]; method of growing single crystals of solid solutions [15, 16]; method of oxide melt quench hardening (charge melting) [17, 18]; and fabricating bismuth containing composites based on porous silicate glass $[3, 4, 6-8, 19, 20]$. One should mention that the latter method is power- and resource-saving in comparison to traditional ones.

Glasses doped with bismuth are used to create new tunable broadband radiation sources [3, 4, 20–23], fiber optical waveguides, lasers and amplifiers [11, 13, 24–26], amplifiers for the second telecommunication transparency window $(1.2-1.35 \,\mu\text{m})$ [24, 27], fibers with the transmission band width from 1.3 to 1.6 μm [3], and 3D active micro- and nanosized pho tonic integrated circuits [4], etc.

In view of the intensive development of the advanced optical and information technologies, it is a matter of urgency to create materials with nanostruc tured elements, which are known to provide their unique physical properties. For example, bismuth containing high-silica glasses (BCHSGs) are charac terized by the unique luminescent properties in a

broad spectral range from blue-green to red lumines cence and, further, to near infrared luminescence [6– 8, 19].

The results obtained from studying the spectral luminescent properties of synthesized BCHSGs dem onstrated that, depending on the synthesis conditions, one observes the appearance of thermally induced bis muth centers and changes in the glass structure [6–8, 19]. The presence of bismuth in different valence states in BCHSGs is indicated by the data of X-ray power diffraction (XRPD) and the results of studies of the glass spectral-optical properties (spectra of absorption, luminescence, and luminescence excita tion). Additional information on the structure of BCHSGs can be obtained by the results of IR spec troscopy, which was the subject of the present work.

METHODOLOGY

The synthesis of BCHSGs, in particular, bismuth containing porous glasses (BCPGs) and quartz-like glasses (BCQGs) in the form of plane-parallel plates of a thickness of 1.5 mm, was carried out in accor dance with the procedure described in general in [6, 8, 19]. In the present work, BCHSGs were fabricated by the impregnation of plates of high-silica nanoporous glasses (HSNPGs) at room temperature for 0.5–72 h in a 0.5 M aqueous solution of bismuth nitrate pre pared based on a 2 M solution of $HNO₃$ followed by thermal treatment at *T* from 700 to \geq 870°C. According to the chemical analysis data, the synthesized BCPGs and BCQGs contain (as analyzed, wt %) $(0.12-0.21)$ Na₂O, $(2.90-3.43)$ B₂O₃, $(95.13-95.61)$

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Fig. 1. IR transmission spectra of glasses not containing bismuth: (1) porous glass, (2) quartz-like glass ($T = 870^{\circ}$ C) [19].

SiO₂, (0.97–1.60) Bi₂O₃, and ≤0.10 Al₂O₃. In the course of IR studies, HSNPG and quart-like glass (QG) samples not containing bismuth were used as standards [8].

IR transmission spectra of the glass samples were fabricated at room temperature in the range 4000– 400 cm–1 using a SPECORD M-80 spectrophotome ter (Carl Zeiss JENA). The samples were measured in the form of tablets pressed from mixtures of glass pow ders with KBr. The experimental spectra were pro cessed using the program Origin Lab 8.6 32Bit. The plot smoothening was performed by the FFT Filter method, unlike [6], in which the Savitzky–Golay method was used. To identify structural elements, the data in the literature [28–41] were used.

EXPERIMENTAL

The transmission spectral dependences for HSNPG and QG not containing bismuth and of BCHSGs in the IR range (Figs. 1–3) demonstrate 16 fundamental bands: 3688–3672, 3492–3472, 3432– 3412, 2940–2912, 2820–2808, 2736–2720, 2340– 2308, 2152–2140, 2104–2080, 2016–2000, 1872– 1860, 1664–1644, 1396–1384, 1108–1084, 872–852, and $468-456$ cm⁻¹. The bands observed at $3688-$ 3672, 3492–3472, 3432–3412, 2736–2720, 2340– 2308, 2152–2140, 2104–2080, 2016–2000, and $1664-1644$ cm⁻¹ are caused by the stretching vibrations of hydroxyl groups and water, asymmetric stretching ν*as* for OH, and symmetric stretching vibra tions of hydroxyl groups and water, asymmetric
stretching v_{as} for OH, and symmetric stretching vibra-
tion v_s for H-O-H-groups [28–32]. The band at $3432-3412$ cm⁻¹ is assigned to vibrations of the struction v_s for H-O-H-groups [28–32]. The band at 3432–3412 cm⁻¹ is assigned to vibrations of the structure Si–O–Si and the stretching Si–OH [33]. The bands at 2940–2912, 2820–2808, and 2736–2720

 cm^{-1} correspond to hydrogen bonds [28]. The weak band at $1872-1860$ cm⁻¹ is caused by the symcm⁻¹ correspond to hydrogen bonds [28]. The weak band at $1872-1860$ cm⁻¹ is caused by the symmetric bending and stretching of Si-O bonds [34]. weak band at $1872-1860 \text{ cm}^{-1}$ is caused by the symmetric bending and stretching of Si-O bonds [34].
Asymmetric stretching vibrations of B-O bonds in the structural units $[BO_3]$ are observed at 1396–1384 cm⁻¹ [30, 35]. The band at $1108-1084$ cm⁻¹ corresponds to the degenerate asymmetric stretching vibration of ν*as* groups $[BO_3]$ [31] and the asymmetric vibrations of the tetrahedron $[SiO₄]$ in the polymerised network groups $[BO_3]$ $[31]$ and the asymmetric vibrations of
the tetrahedron $[SiO_4]$ in the polymerised network
with the dominant bridge bonds $Si-O-Si$ [36]. Asymwith the dominant bridge bonds $Si-O-Si$ [36]. Asymmetric vibrations v_{as} of the tetrahedron [SiO₄] and with the dominant bridge bonds Si-O-Si [36]. Asymmetric vibrations v_{as} of the tetrahedron [SiO₄] and stretching vibrations of the B-O bonds in the strucstretching vibrations of the B-O bonds in the structural units $[BO_4]$ are assigned with the band 872– 852 cm^{-1} [31, 35]. The glass spectra (Figs. 1–3) contain the band at $468-456$ cm⁻¹ corresponding to the 852 cm⁻¹ [31, 35]. The glass spectra (Figs.
tain the band at $468-456$ cm⁻¹ correspone
bending vibrations of the Si-O bonds [37].

The broad and strong band observed in the range 1496–1484 cm–1 (Fig. 1, curve *1*; Fig. 3, curve *1*) is caused by the asymmetric vibrations of B—O bonds in the trigonal structural units $[BO_3]$ [35]. The weak band observed at 808–792 cm–1 (Fig. 1, curve *2*; Fig. 2, curves *2*, *5*; Fig. 3) corresponds to the symmetric vibrations of the $SiO₄$ and $AlO₄$ groups in tetrahedral positions [37, 38]. The strong band at $716-704$ cm⁻¹ (Fig. 1, curve *1*; Fig. 2, curve *3*) is due to bending vibrations of B-O-B bonds in $[BO_3]$ groups $[30, 39]$. The intensive absorption in the range $604-568$ cm⁻¹ confirms the presence of structural units $[BO_3]$ and $[BO₄]$ in (Figs. 1–3) [31, 37] and may indicate the presence of $[AIO_6]$ octahedra.

The band at $940-936$ cm⁻¹ appears for BCQG (Fig. 3, curve 2), which is associated with the cross-
linking of the glass structure by Bi-O-Si bridges [36]. (Fig. 3, curve *2*), which is associated with the cross-

Fig. 2. IR transmission spectra of bismuth-containing porous glasses. Impregnation time: (*1*) 0.5, (*2*) 7, (*3*) 24, (*4*) 48, (*5*) 72 h.

Fig. 3. IR transmission spectra of bismuth-containing high-silica glasses (*1*) BCPG (impregnation for 48 h, $T \le 700^{\circ}$ C), (2) BCQG (impregnation for 48 h, $T \geq 870^{\circ}$ C).

Earlier, this band was observed for the samples impreg nated for 72 h and thermally treated at $T \geq 870^{\circ}$ C and $T \ge 1500^{\circ}$ C [19]. In Fig. 3 (curve *1*), one reveals the band at $728-724$ cm^{-1} , corresponding to vibrations of $T \ge 1500^{\circ}\text{C}$ [19]. In Fig. 3 (curve *1*), one reveals the band at 728–724 cm⁻¹, corresponding to vibrations of the γ (Bi₍₃₎–O) bonds and the symmetric stretching vibrations in the $[BiO_3]$ groups $[40, 41]$. The band in the range $688-660 \text{ cm}^{-1}$ (Fig. 2, curves 1, 2, 4, 5; Fig. 3, curve *1*) is caused by the vibrations of the the range 688–660 cm⁻¹ (Fig. 2, curves 1, 2, 4, 5;
Fig. 3, curve 1) is caused by the vibrations of the
 γ (Bi₍₃₎–O) and Bi–O bonds of different lengths in dis- γ (Bi₍₃₎–O) and Bi–O bonds of different lengths in distorted $[BiO_6]$ structural units [40, 41]. Besides, this

band indicates the presence of phase $\alpha - Bi_2O_3$ in BCPG (impregnation for 48 h, $T \le 700^{\circ}$ C) [40].

CONCLUSIONS

Bismuth-containing high-silica glasses have been synthesized based on matrices of high-silica nanopo rous glasses. The synthesized glass structure was inves tigated by means of the method of IR spectroscopy.

Vibrations of Bi–O bonds in $[BiO_3]$ and $[BiO_6]$ structural units were identified in the synthesized sam ples. Besides, the bands corresponding to the presence of the α -Bi₂O₃ phase were found. It has been estab-lished that cross Bi–O–Si bonds are formed in the of the α -Bi₂O₃ phase were found. It has been estabstructure of bismuth-containing high-silica glass ther mally treated at temperatures of ≥870°C.

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