Synthesis and Antimicrobial Activity of *N*,6-Diaryl-4-methyl-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamides

V. L. Gein^{*a*}*, T. M. Zamaraeva^{*a*}, A. V. Balandina^{*a*}, and M. V. Dmitriev^{*b*}

^a Perm State Pharmaceutical Academy, Ministry of Healthcare of the Russian Federation, ul. Polevaya 2, Perm, 614990 Russia *e-mail: geinvl48@mail.ru

^b Perm State National Research University, Perm, Russia

Received April 14, 2016

Abstract—Three-component reactions of acetoacetic acid *N*-arylamides, aromatic aldehydes, and thiourea afforded *N*,6-diaryl-4-methyl-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamides. Structure and antimicrobial activity of the compounds obtained were examined.

Keywords: three-component reaction, *N*,6-diaryl-4-methyl-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamides, antimicrobial activity

DOI: 10.1134/S107036321612015X

Dihydropyrimidine-2-one(thione) derivatives are of interest due to wide spectrum of biological activity. Structure of these pyrimidine derivatives includes ester group at the position 5 of heterocycle [1–5]. Some of alkoxycarbonyl derivatives of pyrimidine show cardio-vascular [6, 7], antihypertensive [8], antiarrhythmic [9], anticancer [10–12], antioxidant [13, 14], anti-inflammatory [15], fungicidal [16, 17], and antimicrobial [18, 19] activity.

Data on the synthesis of 5-carbamoyl derivatives of dihydropyrimidine-2-ones and -2-thiones are scarce. It is known that biological activity is retained when changing ester moiety at the position 5 of hydrogenated pyrimidine ring with amide group [1-5].

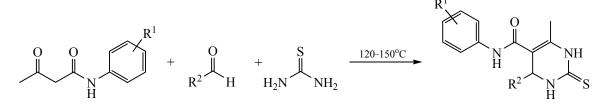
Currently principles of green chemistry are preferred that make promising the synthesis of pyrimidine derivatives without using expensive catalysts and toxic solvents.

In order to obtain new pyrimidine derivatives, study their spatial structure and antimicrobial activity, we synthesized new N,6-diaryl-4-methyl-2-thioxo-1,2,3,6tetrahydropyrimidine-5-carboxamides **1–12** by reacting acetoacetic acid N-arylamides with aromatic aldehydes and thiourea. The reactions were carried out without solvent under conditions analogous to those described in [20] (Scheme 1).

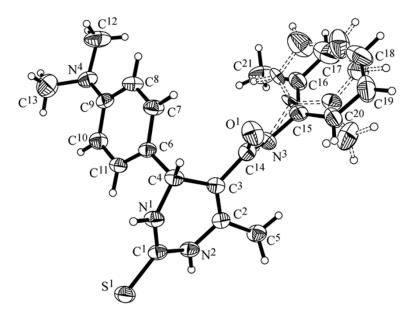
Compounds 1–12 were colorless crystalline substances, soluble in DMF, DMSO, as well as in acetic acid and ethanol on heating, and insoluble in water, benzene and toluene.

The IR spectra of compounds 1–12 contained the absorption bands due to stretching vibrations of amide

Scheme 1.



 $R^{1} = 2-CH_{3} (1-6), 2,4-(CH_{3})_{2} (7-9), 2-Cl (10-12); R^{2} = 4-FC_{6}H_{4} (1), 4-CH_{3}OOCC_{6}H_{4} (2, 9), 4-(CH_{3})_{2}NC_{6}H_{4} (3), 2,4-Cl_{2}C_{6}H_{3} (4, 10), 4-BrC_{6}H_{4} (5), 2-NO_{2}C_{6}H_{4} (6, 11), 3-pyridyl (7), 2-thienyl (8), 2-FC_{6}H_{4} (12).$



 $Crystal\ structure\ of\ 4-methyl-6-(4-dimethylaminophenyl)-N-2-methylphenyl-2-thioxo-1, 2, 3, 6-tetrahydropyrimidine-5-carboxamide\ 3.$

groups (1660–1688 cm⁻¹), N–H (3120–3400 cm⁻¹) and C=C bonds (1600–1630 cm⁻¹).

Compounds 1–12 showed characteristically in the ¹H NMR spectra a singlet of C⁴H₃ group (1.85–2.15 ppm), doublets of H¹ and H⁶ protons (5.20–6.02 and 9.05–9.39 ppm, $J_{1,6} = 1.9-3.0$ Hz), a singlet of amide NH proton (8.91–9.37 ppm), and a broad signal of N³H moiety (9.74–10.07 ppm). Signals from the corresponding aryl substituents were also present. In the mass spectrum of compound **5** we observed the molecular ion peak $[M]^+$ of m/z 417 and the fragment ion peaks of m/z 311 $[M - CH_3C_6H_4NH]^+$, 106 $[CH_3C_6H_4NH]^+$. These data confirmed the structure of the obtained compounds.

The crystal structure of compound 3 was determined by X-ray diffraction analysis (see figure) of a single crystal obtained by slow crystallization from ethanol. Compound 3 crystallizes in the centrosymmetric space group belonging to the monoclinic crystal system. The tetrahydropyrimidine ring adopts a distorted *boat* conformation; deviations of the C¹ and N^1 atoms from the $C^4C^3C^2N^2$ plane equal 0.395 and 0.559 Å, respectively. The dimethylaminophenyl moiety occupies pseudo-axial position. The carbonyl group is turned with respect to the $C^4C^3C^2N^2$ plane: the torsion angle $C^2C^3C^{14}O^1$ is $-137.1(3)^\circ$. The tolylcarbamoyl substituent is not planar: the torsion angles O¹C¹⁴N³C¹⁵ and $C^{14}N^3C^{15}C^{20}$ are 8.9(8)° and 50(1)°, respectively. The ortho-tolyl fragment is disordered by two positions with populations of 0.703(8) and 0.297(8) that indicates the presence of two conformers of compound 3 in the crystal. Molecules of 3 in the crystal are linked to each other through intermolecular hydrogen bonds to form infinite two-dimensional network parallel to the plane (1 0 0). Hydrogen bonding parameters are shown in Table 1.

Antimicrobial activity of compounds 1–12 was determined in relation to *St. aureus*, *E. coli*, *C. albicans* strains (Table 2). According to the data obtained, they posses weak antimicrobial activity.

 Table 1. Hydrogen bonds parameters in the molecule of 3

D–H···A	<i>d</i> (D–H), Å	<i>d</i> (H···A), Å	$d(\mathbf{D}\cdots\mathbf{A}),\mathbf{\dot{A}}$	Angle DHA, deg
$N^{1}-H^{1}\cdots N^{4} [x, -y + 1/2, z - 1/2]$	0.85(2)	2.22(2)	3.031(3)	160(2)
$N^2 - H^2 \cdots S^1 [-x + 1, -y, -z + 2]$	0.83(3)	2.56(3)	3.378(3)	169(3)
N ³ -H ³ ···O ¹ [x, $-y + 1/2, z + 1/2$]	0.82(2)	2.00(2)	2.819(3)	174(3)

RUSSIAN JOURNAL OF GENERAL CHEMISTRY Vol. 86 No. 12 2016

	MIC, µg/mL				
Compound	St. aureus	E. coli	C. albicans		
	ATCC 6538	ATCC 25922	ATCC 24433		
1	500	500	н/а		
2	500	500	1000		
3	500	1000	1000		
4	500	1000	500		
5	500	1000	—		
6	500	500	250		
7	1000	1000	125		
8	500	500	250		
9	500	500	1000		
10	1000	1000	500		
11	1000	1000	250		
12	500	500	1000		
Dioxidine	62.5-1000	3.9-62.5	_		
Chloramine B	500	250	_		
Fluconazole	-	-	8-32		

 Table 2. Antimicrobial activity of compounds 1–12

EXPERIMENTAL

IR spectra in mineral oil were recorded on a Specord M-80 spectrometer. ¹H NMR spectra of the solutions in DMSO- d_6 were registered on a Bruker 500 spectrometer (500.13 MHz) relative to internal TMS. Mass spectrum was obtained on a Finnigan MAT INCOS-50 instrument (ionization energy 70 eV). Elemental analysis was performed on a Perkin Elmer 2400 analyzer. Melting points were determined on a M-565 instrument.

X-ray diffraction analysis of compound 3 was performed on a Xcalibur R diffractometer equipped with a CCD-detector [Mo K_{α} -radiation, 295(2) K, ω scanning]. A correction for absorption was applied empirically using SCALE3 ABSPACK algorithm [21]. Crystals of compound 3, C21H24N4O3S, monoclinic, space group P21/c, unit cell parameters: a = 11.630(5), b = 19.765(6), c = 9.414(3) Å, $\beta = 101.47(3)^{\circ}$, $V 2120.6(14) \text{ Å}^3$, $d_{\text{calc}} = 1.192 \text{ g/cm}^3$, $\mu = 0.170 \text{ mm}^{-1}$, Z = 4. The structure was solved by the direct fullmatrix method and refined by anisotropic approximation for all non-hydrogen atoms. Hydrogen atoms were refined according to a *rider* model in isotropic approximation with dependent thermal parameters; the NH hydrogen atoms were localized from difference electron density synthesis and refined independently in an isotropic approximation. Atomic coordinates of the

disordered components were determined taking into account SAME soft limits. All calculations were performed using SHELX-97 [22] and OLEX2 [23] program packages. The final divergence factors: $R_1 =$ 0.0657, $wR_2 = 0.1243$ for 2288 reflections with $I > 2\sigma(I)$, $R_1 = 0.1609$, $wR_2 = 0.1659$ for all 4943 independent reflections, S = 1.002. The crystallographic data for compound **3** were deposited to the Cambridge Crystallographic Data Centre (CCDC 1471336).

4-Methyl-N-(2-methylphenyl)-6-(4-fluorophenyl)-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (1). A mixture of 0.01 mol of 2-methylacetoacetanilide, 0.01 mol of 4-fluorobenzaldehyde, and 0.01 moles of thiourea was heated at 120-150°C for 5-10 min until gas evolution ceased and the reaction mixture solidified. After cooling, the solid was treated with ethanol, filtered, and recrystallized from ethanol. Yield 2.56 g (72%), mp 170-172°C (EtOH). IR spectrum, v, cm⁻¹: 1616 (C=C), 1688 (CON), 3216 (NH). ¹H NMR spectrum (DMSO- d_6), δ , ppm: 1.89 s (3H, 4-CH₃), 2.12 s (3H, CH₃C₆H₄), 5.34 d (1H, CH, $J_{1,6} = 1.9$ Hz), 7.02–7.24 m (8H, CH₃C₆H₄, FC₆H₄), 9.10 s (1H, NH, amide), 9.30 d (1H, N¹H, $J_{1,6} = 1.9$ Hz), 9.88 br.s (1H, N³H). Found, %: C 64.10, 64.33; H 5.02, 5.19; N 11.71, 11.95. C₁₉H₁₈FN₃OS. Calculated, %: C 64.21; H 5.10; N 11.82.

Compounds 2–12 were prepared similarly.

6-(4-Carbomethoxyphenyl)-4-methyl-*N*-(2-methylphenyl)-2-thioxo-1,2,3,6-tetrahydropyrimidine-5carboxamide (2). Yield 3.08 g (78%), mp 234–236°C (EtOH). IR spectrum, v, cm⁻¹: 1630 (C=C), 1680 (CON), 1720 (CO), 3150, 3200, 3360 (NH). ¹H NMR spectrum (DMSO- d_6), δ , ppm: 1.90 s (3H, 4-CH₃), 2.12 s (3H, <u>CH₃C₆H₄</u>), 3.79 s (3H, <u>CH₃OOCC₆H₄</u>), 5.42 d (1H, CH, $J_{1,6} = 2.1$ Hz), 7.02–7.93 m (8H, CH₃C₆<u>H₄</u>, CH₃OOCC₆<u>H₄</u>), 9.14 s (1H, NH, amide), 9.36 d (1H, N¹H, $J_{1,6} = 2.1$ Hz), 9.93 br.s (1H, N³H). Found, %: C 63.67, 63.90; H 5.27, 5.45; N 10.49, 10.75. C₂₁H₂₁N₃O₃S. Calculated, %: C 63.78; H 5.35; N 10.62.

4-Methyl-*N*-(**2-methylphenyl**)-**6**-(**4-dimethylaminophenyl**)-**2-thioxo-1,2,3,6-tetrahydropyrimidine-5carboxamide (3).** Yield 2.21 g (58%), mp 238–240°C (EtOH). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 1.89 s (3H, 4-CH₃), 2.10 s (3H, <u>CH</u>₃C₆H₄), 2.82 s [6H, (<u>CH</u>₃)₂NC₆H₄], 5.20 d (1H, CH, *J*_{1,6} = 2.1 Hz), 6.57– 7.02 m [8H, CH₃C₆<u>H</u>₄, (CH₃)₂NC₆<u>H</u>₄], 8.95 s (1H, NH, amide), 9.18 d (1H, N¹H, *J*_{1,6} = 2.1 Hz), 9.74 br.s (1H, N³H). Found, %: C 66.17, 66.41; H 6.26, 6.45; N 14.59, 14.84. $C_{21}H_{24}N_4OS$. Calculated, %: C 66.29; H 6.36; N 14.72.

4-Methyl-*N*-(2-methylphenyl)-6-(2,4-dichlorophenyl)-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (4). Yield 2.76 g (68%), mp 262–264°C (EtOH). IR spectrum, v, cm⁻¹: 1610 (C=C), 1660 (CON), 3180, 3210, 3352 (NH). ¹H NMR spectrum (DMSO-*d*₆), δ, ppm: 1.91 s (3H, 4-CH₃), 2.08 s (3H, <u>CH</u>₃C₆H₄), 5.67 d (1H, CH, $J_{1,6} = 2.8$ Hz), 6.99–7.34 m (7H, CH₃C₆<u>H</u>₄, 2,4-Cl₂C₆H₃), 9.02 s (1H, NH, amide), 9.05 d (1H, N¹H, $J_{1,6} = 2.8$ Hz), 9.82 br.s (1H, N³H). Found, %: C 56.04, 56.27; H 4.13, 4.30; N 10.22, 10.47. C₁₉H₁₇Cl₂N₃OS. Calculated, %: C 56.16; H 4.22; N 10.34.

4-Methyl-6-(4-bromophenyl)-*N*-(**2-methylphenyl)**-**2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (5).** Yield 2.87 g (69%), mp 246–248°C (EtOH). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 1.91 s (3H, 4-CH₃), 2.11 s (3H, <u>CH</u>₃C₆H₄), 5.31 d (1H, CH, $J_{1,6} = 3.0$ Hz), 6.85–7.55 m (8H, CH₃C₆<u>H</u>₄, 4-BrC₆H₄), 9.13 s (1H, NH, amide), 9.20 d (1H, N¹H, $J_{1,6} = 3.0$ Hz), 10.00 br.s (1H, N³H). Found, %: C 54.68, 54.93; H 4.28, 4.45; N 9.98, 10.22. C₁₉H₁₈BrN₃OS. Calculated, %: C 54.81; H 4.36; N 10.09.

4-Methyl-*N*-(**2-methylphenyl**)-**6**-(**2-nitrophenyl**)-**2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide** (6). Yield 2.44 g (64%), mp 216–218°C (EtOH). IR spectrum, v, cm⁻¹: 1620 (C=C), 1670 (CON), 3180, 3280, 3400 (NH). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 1.86 s (3H, 4-CH₃), 2.10 s (3H, <u>CH₃C₆H₄), 6.02 d</u> (1H, CH, *J*_{1,6} = 2.8 Hz), 7.00–7.89 m (8H, CH₃C₆<u>H₄</u>, 2-NO₂C₆H₄), 9.07 d (1H, N¹H, *J*_{1,6} = 2.8 Hz), 9.16 s (1H, NH, amide), 10.02 br.s (1H, N³H). Found, %: C 59.56, 59.80; H 4.64, 4.83; N 14.52, 14.77. C₁₉H₁₈N₄O₃S. Calculated, %: C 59.67; H 4.74; N 14.65.

4-Methyl-6-(3-pyridyl)-*N*-(2,4-dimethylphenyl)-2thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (7). Yield 2.64 g (75%), mp 266–268°C (EtOH). IR spectrum, v, cm⁻¹: 1600 (C=C), 1660 (CON), 3200 (NH). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 1.85 s (3H, 4-CH₃), 2.12 s and 2.19 s [6H, (<u>CH₃)</u>₂C₆H₃], 5.35 d (1H, CH, *J*_{1,6} = 2.1 Hz), 6.86–7.63 m [7H, (CH₃)₂C₆<u>H</u>₃, C₅H₄N], 9.08 s (1H, NH, amide), 9.32 d (1H, N¹H, *J*_{1,6} = 2.1 Hz), 9.90 br.s (1H, N³H). Found, %: C 64.64, 64.87; H 5.62, 5.81; N 15.78, 16.03. C₁₉H₂₀N₄OS. Calculated, %: C 64.75; H 5.72; N 15.90.

4-Methyl-6-(2-thienyl)-*N*-(2,4-dimethylphenyl)-2thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (8). Yield 2.26 g (63%), mp 240–242°C (EtOH). IR spectrum, v, cm⁻¹: 1610 (C=C), 1660 (CON), 3120, 3210, 3320 (NH). ¹H NMR spectrum (DMSO- d_6), δ , ppm: 1.94 s (3H, 4-CH₃), 2.13 s and 2.22 s [6H, (<u>CH₃</u>)₂C₆H₃], 5.59 d (1H, CH, $J_{1,6} = 2.8$ Hz), 6.79–7.33 m [6H, (CH₃)₂C₆<u>H</u>₃, C₄H₃S], 8.91 s (1H, NH, amide), 9.39 d (1H, N¹H, $J_{1,6} = 2.8$ Hz), 9.90 br.s (1H, N³H). Found, %: C 60.37, 60.61; H 5.28, 5.45; N 11.62, 11.87. C₁₈H₁₉N₃OS₂. Calculated, %: C 60.48; H 5.36; N 11.75.

6-(4-Carbomethoxyphenyl)-4-methyl-*N*-(2,4-dimethylphenyl)-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (9). Yield 3.24 g (79%), mp 240–242°C (EtOH). IR spectrum, v, cm⁻¹: 1630 (C=C), 1680 (CON), 1720 (CO), 3200, 3150, 3380 (NH). ¹H NMR spectrum (DMSO-*d*₆), δ , ppm: 1.86 s (3H, 4-CH₃), 2.11 s and 2.19 s [6H, (<u>CH₃)</u>₂C₆H₃], 3.79 s (3H, <u>CH₃OOCC₆H₄), 5.41 d (1H, CH, *J*_{1,6} = 2.1 Hz), 6.86– 7.92 m [7H, (CH₃)₂C₆<u>H₃</u>, CH₃OOCC₆<u>H₄</u>], 9.06 s (1H, NH, amide), 9.34 d (1H, N¹H, *J*_{1,6} = 2.1 Hz), 9.92 br.s (1H, N³H). Found, %: C 64.42, 64.65; H 5.58, 5.75; N 10.14, 10.39. C₂₂H₂₃N₃O₃S. Calculated, %: C 64.53; H 5.66; N 10.26.</u>

4-Methyl-6-(2,4-dichlorophenyl)-*N*-(2-chlorophenyl)-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (10). Yield 3.50 g (82%), mp 256–258°C (EtOH). IR spectrum, v, cm⁻¹: 1620 (C=C), 1680 (CON), 3190, 3220, 3390 (NH). ¹H NMR spectrum (DMSO-*d*₆), δ, ppm: 2.10 s (3H, 4-CH₃), 5.66 d (1H, CH, $J_{1,6} = 1.9$ Hz), 7.06–7.49 m (7H, ClC₆H₄, 2,4-Cl₂C₆H₃), 9.23 d (1H, N¹H, $J_{1,6} = 1.9$ Hz), 9.37 s (1H, NH, amide), 9.99 br.s (1H, N³H). Found, %: C 50.55, 50.79; H 3.24, 3.39; N 9.73, 9.98. C₁₈H₁₄Cl₃N₃OS. Calculated, %: C 50.66; H 3.31; N 9.85.

4-Methyl-6-(2-nitrophenyl)-*N*-(2-chlorophenyl)-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (11). Yield 3.17 g (79%), mp 200–202°C (EtOH). IR spectrum, ν, cm⁻¹: 1620 (C=C), 1680 (CON), 3190, 3320, 3410 (NH). ¹H NMR spectrum (DMSO-*d*₆), δ, ppm: 2.15 s (3H, 4-CH₃), 6.01 d (1H, CH, $J_{1,6} = 1.9$ Hz), 7.06–7.89 m (8H, ClC₆H₄, 2-NO₂C₆H₄), 9.09 d (1H, N¹H, $J_{1,6} = 1.9$ Hz), 9.30 s (1H, NH, amide), 10.07 br.s (1H, N³H). Found, %: C 53.55, 53.80; H 3.65, 3.84; N 13.78, 14.04. C₁₈H₁₅ClN₄O₃S. Calculated, %: C 53.67; H 3.75; N 13.91.

4-Methyl-6-(2-fluorophenyl)-*N*-(2-chlorophenyl)-2-thioxo-1,2,3,6-tetrahydropyrimidine-5-carboxamide (12). Yield 3.15 g (84%), mp 210–212°C (EtOH). ¹H NMR spectrum (DMSO- d_6), δ , ppm: 2.11 s (3H, 4CH₃), 5.57 d (1H, CH, $J_{1,6} = 1.9$ Hz), 6.99–7.36 m (8H, ClC₆H₄, FC₆H₄), 9.20 d (1H, N¹H, $J_{1,6} = 1.9$ Hz), 9.30 s (1H, NH, amide), 9.91 br.s (1H, N³H). Found, %: C 57.41, 57.64; H 3.93, 4.10; N 11.06, 11.31. C₁₈H₁₅ClFN₃OS. Calculated, %: C 57.52; H 4.02; N 11.18.

Antimicrobial activity of the compounds obtained against strains of *St. aureus*, *E. coli* and *C. albicans* was determined by serial dilutions of the test compound solution with meat-peptone broth and Sabouraud liquid medium and examined activity. Bacterial load on 1 mL of culture liquid was 250 000 microbial cells. Control and test tubes were incubated at $36-37^{\circ}$ C for 18–20 h. Growth of bacterial cultures or inhibition due to the bacteriostatic action of the test compounds was registered. Minimum inhibitory concentration of the substance (MIC, µg/mL), which inhibits the growth of bacterial cultures, was determined. Dioxidine, chloramine B, and fluconazole were used as comparison drugs.

REFERENCES

- Kappe, C.O., *Eur. J. Med. Chem.*, 2000, vol. 35, p. 1043. doi 10.1016/S0223-5234(00)01189-2.
- Vdovina, S.V. and Mamedov, V.A., *Russ. Chem. Rev.*, 2008, vol. 77, no. 12, p. 1017. doi 10.1070/ RC2008v077n12ABEH003894
- Wan, J.-P. and Liu, Y., Synthesis, 2010, p. 3943. doi 10.1055/s-0030-1258290
- Kappe, C.O., *Multicomponent Reactions*, 2005, p. 95. doi 10.1002/3527605118.ch4
- Kappe, C.O., *Heterocycles*, 1997, vol. 45, no. 10, p. 1967. doi 10.3987/COM-97-7931
- Atwal, K.S., Swanson, B.N., Unger, S.E., Floyd, D.M., Moreland, S., Hedberg, A., and O'Reilly, B.C., *J. Med. Chem.*, 1991, vol. 34, p. 806. doi 10.1021/jm00106a048
- Chikhale, R.V., Bhole, R.P., Khedekar, P.B., and Brusari, K.P., *Eur. J. Med. Chem.*, 2009, vol. 44, p. 3645. doi 10.1016/j.ejmech.2009.02.021
- Jauk, B., Pernat, T., and Kappe, C.O., *Molecules*, 2000, vol. 5, p. 227. doi 10.3390/50300227
- 9. WO Patent 009392 A2, 2005.
- 10. Sharma, S.K., Kumar, P., Narasimhan, B., Ramasamy, K.,

Manic, V., Mishra, R.K., and Majeed, A.B.A., *Eur. J. Med. Chem.*, 2012, vol. 48, p. 16. doi 10.1016/ j.ejmech.2011.11.028

- Fewell, S.W., Smith, C.M., Lyon, M.A., Dumitrescu, T.P., Wipfb, P., Day, B.W., and Brodsky, J.L., *J. Biol. Chem.*, 2004, vol. 279, p. 51131. doi 10.1074/jbc.M404857200
- Wright, C.M., Chovatiya, R.J., Jameson, N.E., Turner, D.M., Zhu, G., Werner, S., Huryn, D.M., Pipas, J.M., Day, B.W., Wipf, P., and Brodsky, J.L., *Bioorg. Med. Chem.*, 2008, vol. 16, p. 3291. doi 10.1016/ j.bmc.2007.12.014
- Magerramov, A.M., Kurbanova, M.M., Abdinbekova, R.T., Rzaeva, I.A., Farzaliev, V.M., and Allakhverdiev, M.A., *Russ. J. Appl. Chem.*, 2006, vol. 79, no. 5, p. 787. doi 10.1134/S107042720605017X
- Zamanova, A.V., Kurbanova, M.M., Rzaeva, I.A., Farzaliev, V.M., and Allakhverdiev, M.A., *Russ. J. Appl. Chem.*, 2010, vol. 83, no. 2, p. 293. doi 10.1134/ S1070427210020205
- Tozkoparan, B., Yarin, M., Sarac, S., Ertan, M., Kelicen, P., and Demirdamar, G., *Arch. Pharm. Med. Chem.*, 2000, vol. 333, p. 415. doi 10.1002/1521-4184 (200012)333:12<415::AID-ARDP415>3.0.CO;2-E
- Kathiravan, M.K., Salake, A.B., Chothe, A.S., Dudhe, P.B., Watode, R.P., Mukta, M.S., and Gathwe, S., *Bioorg. Med. Chem.*, 2012, p. 5678. doi 10.1016/ j.bmc.2012.04.045
- Singh, O.M., Singh, S.J., Devi, M.B., Devi, L.N., Singh, N.I., and Lee, S.-G., *Bioorg. Med. Chem. Lett.*, 2008, vol. 18, p. 6462. doi 10.1016/j.bmcl.2008.10.063
- Shah, T.B., Gupte, A., Patel, M.R., Chaudhari, V.S., Patel, H., and Patel, V.C., *Ind. J. Chem. B*, 2010, vol. 49, p. 578.
- Zohdi, H.F., Rated, N.M., and Elnagdy, S.M., *Eur. J. Med. Chem.*, 2011, vol. 46, p. 5636. doi 10.1016/ j.ejmech.2011.09.036
- Gein, V.L., Zamaraeva, T.M., Zorina, A.A., and Vakhrin, M.I., *Russ. J. Org. Chem.*, 2009, vol. 45, no. 10, p. 1581. doi 10.1134/S1070428009100315
- CrysAlisPro, Agilent Technologies, Version 1.171.37.33 (release 27-03-2014 CrysAlis171 .NET).
- Sheldrick, G.M., Acta Crystallogr. (C), 2015, vol. 71, p. 3. doi 10.1107/S2053229614024218
- Dolomanov, O.V., Bourhis, L.J., Gildea, R.J., Howard, J.A.K., and Puschmann, H., *J. Appl. Crystallogr.*, 2009, vol. 42, p. 339. doi 10.1107/S0021889808042726