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Nanocomposites Based on Multi-Walled Carbon Nanotubes, Magnetite Nanoparticles, and Core—Shell Molecularly Imprinted Polymers in Piezoelectric Sensors for the Determination of Macrolide Antibiotics

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Abstract—A piezoelectric sensor with a recognition layer based on magnetic carbon nanocomposites, including multi-walled carbon nanotubes, magnetic Fe_3O_4 nanoparticles, and polymer nanospheres with molecular imprints of erythromycin and azithromycin, obtained by the "core-shell" method, is developed. Silicon dioxide particles are used as cores, on the surface of which a shell molecularly imprinted with macrolides is synthesized by free radical polymerization or the sol-gel method. SiO_2 particles are obtained by the Stober method by varying the ratio of reagents during the synthesis. The size of the cores and nanoparticles of molecularly imprinted polymers (MIP) is determined by atomic force microscopy, and the density and uniformity of the layer on the surface of magnetic carbon nanocomposites (MCNC) are determined by the piezoelectric quartz crystal microbalance method. The optimal ratio of the reagents (template : functional monomer : crossmonomer) is established by a spectrophotometric method during the synthesis of "core-shell" nanostructures by free radical polymerization. A thin shell of SiO₂ with imprints of an antibiotic based on organosilicon compounds used in the synthesis of the core is formed by the sol-gel method on the surface of the silicon dioxide core. The sensor recognition layer is formed under the action of an external magnetic field. The dependence of the analytical signal of the sensor based on MIP@SiO₂/MCNC on concentration is linear in the range $5-160 \,\mu\text{g/mL}$ for azithromycin and $10-160 \,\mu\text{g/mL}$ for erythromycin, and with a recognition layer based on SiO₂@SiO₂/MCNC, in the concentration range $20-400 \mu g/mL$ for erythromycin.

Keywords: piezoelectric sensor, molecularly imprinted core—shell polymers, multi-walled carbon nanotubes, magnetic nanoparticles, magnetic carbon nanomaterials, macrolide antibiotics **DOI:** 10.1134/S1061934823110035

Piezoelectric gravimetric sensors have proved themselves well in the determination of toxicants in mixtures of complex composition with high sensitivity and selectivity virtually without sample preparation [1–4]. Most often, antibodies, antigens, and aptamers are used as recognition elements in piezoelectric sensors; they are characterized by a high specificity of interaction and allow obtaining information about the course of a biochemical reaction in almost real time [5-8]. However, their instability during storage, exposure to organic solvents, and high concentrations of electrolytes, as well as the difficulty of preparation [9], stimulates the search for "synthetic" antibodies, which can be molecularly imprinted polymers (MIP), which have a number of undeniable advantages over natural receptors. Molecularly imprinted polymers can be obtained by the copolymerization of a func-

tional monomer and a cross monomer in the presence of template molecules (templates), which are most often the analytes. After the removal of the template molecule, an imprint is formed in the polymer matrix, which corresponds to the molecule to be determined in shape and chemical functionality, and the analyte can be selectively rebound to it [10, 11]. However, at present, there are no standard procedures for obtaining MIP with the necessary characteristics.

Molecularly imprinted polymers can be obtained by block polymerization and used after grinding and screening [12, 13]. However, regular-shaped imprinted polymers, especially in the nanoscale region, find wider application in sensorics [14, 15]. The MIP nanoparticles exhibit higher surface area to volume ratios; therefore, the imprints are more accessible to templates, which improves the kinetics of rebinding with template molecules. The most popular methods for the synthesis of MIP nanoparticles, such as micro- and miniemulsion polymerization, the precipitation method, and the core-shell method (coreshell, core-shell by grafting), carried out in two stages, should be noted [16-20]. At the first stage, a spherical core is formed, and at the second stage, a thin-layer polymer shell with imprints of analyte molecules is synthesized. In this case, the imprints are concentrated in a rather thin layer, which accelerates the rebinding of the template with the MIP. Magnetic Fe₃O₄ nanoparticles, chitosan nanoparticles, activated polystyrene beads, quantum dots, and SiO₂ nanoparticles are used as cores [21-23]. For the shell synthesis, microemulsion and free radical polymerization (FRP), as well as the sol-gel method (SGM) are most often used [24, 25]. An earlier study on the use of Fe_2O_3 particles as cores showed that the high mass of MIP particles contributed to narrowing the range of analyte concentrations determined using a gravimetric piezoelectric sensor [26].

In the development of piezoelectric sensors, the method of forming a recognition layer is of paramount importance. Previously, we proposed obtaining a recognition layer of an immunosensor based on magnetic carbon nanocomposites (MCNC) under the action of an external magnetic field [27]. The carbon nanotubes (CNT) present in the nanocomposite contributed to an increase in the specific surface area of the electrode, i.e., made it possible to place a larger number of recognizing molecules, and magnetic particles (MNP) significantly reduced and simplified the procedure for

preparing a piezoelectric sensor for analysis. In forming the receptor coating of a piezoelectric sensor based on MIP nanoparticles, it is necessary to limit the mass of polymer particles in order not to make the recognition layer heavier. More promising is the use of MIP obtained by the core—shell method based on light SiO₂ cores as recognition structures, which can be modified in various ways, thus changing and improving their properties [28]. This will reduce the cost of analysis, increase the range of the determined analyte concentrations, but at the same time retain a possibility of fixing the recognition layer under the action of magnetic forces.

The aim of this work was to study the conditions for the synthesis of polymer particles with molecular imprints of macrolides by the core—shell method with silicon dioxide cores and a possibility of their use in the recognition layer of a piezoelectric sensor based on magnetic carbon nanocomposites for the determination of antibiotics in liquid media.

EXPERIMENTAL

Materials and equipment. For the synthesis of polymers with molecular imprints we used the following substances: functional monomer—methacrylic acid (MAA) (Vekton, Russia), crosslinking monomer ethylene glycol dimethacrylate (EGDMA) (Aldrich Chemistry, United States), and polymerization initiator—azobisisobutyronitrile (Labtekh, Russia). Macrolides of the erythromycin series were used as templates (Scheme 1): erythromycin (Sintez, Russia), azithromycin (Production of medicines, Russia).



Scheme 1. Structural formulas of (a) erythromycin and (b) azithromycin.

The solvents were as follows: acetonitrile, toluene (Vekton, Russia), ethanol (Ekos1, Russia), acetic acid (NevaReaktiv, Russia), acetone (98%, Qumica, Spain), hydrochloric acid (density 1.198 g/cm³, Neva-Reaktiv, Russia). Triton X-100 (Sigma–Aldrich, United States) was used to transfer particles to a dispersed state. SiO₂ nanospheres were synthesized using tetraethoxysilane (**TEOS**) (Komponent-Reaktiv,

Russia), (3-aminopropyl)triethoxysilane (APTES) (Sigma–Aldrich, United States), and a 25% aqueous NH_3 solution (Khimreaktivsnab, Russia). The obtained SiO₂ particles were hydrophobized with polyvinylpyrrolidone (Aplichem, Germany).

Solutions of macrolide antibiotics in ethyl alcohol of a concentration of 40 μ g/mL were used for reembedding in MIP. To construct calibration depen-

BIZINA et al.



Fig. 1. Synthesis of nanoparticles with molecular imprints of macrolides on an example of erythromycin using free-radical polymerization and the sol-gel method.

dences, solutions with concentrations of 0.1, 0.5, 5, 10, 20, 40, 100, 160, 200, and 400 μ g/mL, obtained by diluting a concentrated solution with ethyl alcohol, were used.

The surface morphology of the nanoparticles was studied by atomic force microscopy (Solver P47-PRO scanning probe microscope, Nanotechnology-MDT, Russia) in the semicontact mode in air at a scanning rate of 1.34 Hz. The nature of functional groups on the surface of SiO₂ particles was determined by IR spectrometry (Iraffinity-1 FTIR spectrometer, Japan), as well as the nature of bonds formed during the synthesis of MIP by free radical polymerization and the sol-gel method.

Piezoelectric AT-cut resonators 8 mm in diameter with a natural oscillation frequency of 10 MHz, with gold electrodes located on both sides of the quartz plate (ZAO "ETNA", Russia) were used as sensors. The analytical signal was recorded on a CPNA-330 setup (ZAO "ETNA", Russia).

The ratio of the template, functional and crosslinking monomers was determined by the spectrophotometric method in the UV region of the spectrum (Fluorat-02-Panorama, Russia). Synthesis of nanoparticles with molecular imprints. Two procedures for the synthesis of nanoparticles with a core–shell structure were used: in the first case, a polymer shell was synthesized on the surface of a silicon dioxide core by the FRP method–SiO₂@MIP. The second method consisted in obtaining a shell on the surface of the silica core by the sol–gel method using siloxanes–SiO₂@SiO₂ (Fig. 1).

The synthesis of SiO_2 cores was carried out according to the Stober method. For work, we chose the procedures described in the literature, which differed in the ratio of reagents during the synthesis (Table 1).

The synthesis of a polymer shell by free radical polymerization requires the preliminary hydrophobization of the SiO₂ core. For this, 0.3 g of SiO₂ nanoparticles were added to 50 mL of an aqueous solution of polyvinylpyrrolidone (60 mg/mL) and the mixture was stirred with a magnetic stirrer at 25°C for 12 h. The resulting particles were separated, washed to pH 7, and dried at 70°C to a constant weight [31]. To synthesize a polymer shell on the surface of hydrophobized silicon dioxide cores, 0.06 g of a macrolide was dissolved in 18 mL of a porogenic solvent, an acetoni-trile-toluene mixture (3 : 1, by volume). Next,

Synthesis method	TEOS, mL	APTES, mL	$\rm NH_3$ solution, mL	Ethanol, mL	Reference
1	1.5	1.5	0.2	—	[29]
2	18	—	36	182	[30]
3	5	_	9	80	[27]

Table 1. Ratio of reagents in the synthesis of silicon dioxide cores

0.1 mmol of MAA was introduced, and the resulting mixture was kept for 2 h at 4°C to form the template—monomer prepolymerization complex. A 0.015-g portion of SiO₂ particles and 0.8 mmol of EGDMA were added, sonicated for 10 min, the polymerization initiator azobisisobutyronitrile (0.0985 g) was added, and the resulting solution was continuously stirred at 50°C. The resulting SiO₂@MIP particles were dried to a constant weight.

The sol-gel method consisted in the formation of macrolide surface imprints in a silica shell [32]. For this, 0.6 mL of TEOS, 9 mL of ethanol, and 0.9 mL of distilled water were mixed, 0.4 mL of a 25% ammonia solution was added, and the mixture was stirred with a magnetic stirrer for 1 h at 25°C. Next, 0.3 mL of APTES (as a functional monomer), 10 mg of erythromycin (template), and 0.5 mL of TEOS (crosslinker) were added, and the mixture was stirred for another 1 h. The particles were separated by centrifugation and washed with ethanol to remove template molecules.

Templates were not used in the synthesis of nonimprinted polymer particles ($SiO_2@NIP$) by the FRP and SGM methods.

Synthesis of composite nanomaterials and formation of a receptor layer. We used magnetic carbon nanocomposites described in [27]. Fe₃O₄ nanoparticles were immobilized on the surface of multi-walled carbon nanotubes synthesized at the Institute for Problems of Microelectronics Technology and High-Purity Materials, Russian Academy of Sciences, Chernogolovka [33]. The CNT–MNP ratio was chosen by piezoelectric quartz crystal microbalancing [34] taking into account the mass of the coating, because an overload led to a breakdown of the analytical signal of the sensor, and a decrease in the amount of MNP in the composite led to a decrease in the strength of binding to the sensor under the action of an external magnetic field. It was found that the optimal ratio was 3:1. Transfer to the dispersed state was carried out according to the scheme proposed earlier [35]: 1 mg of MCNC was added to 1 mL of a 1% aqueous solution of Triton X-100 and subjected to ultrasonic treatment for 1 h.

Recognizing elements (MIP@SiO₂ and SiO₂@SiO₂) were immobilized on the surface of MCNC by physical sorption; to do this, 1 mg of activated MCNC and 1 mg of MIP nanoparticles were added to 1 mL of a 1% aqueous solution of Triton X-100, heated to 90°C for 15 min, and sonicated for 1 h.

A receptor layer based on composite materials (MIP@SiO₂/MCNC and SiO₂@SiO₂/MCNC) was formed by dosing 2 μ L of the dispersion onto a sensor electrode preliminarily purified with a 1 mM hydro-chloric acid solution, acetone, and ethanol, placed above a neodymium magnet in the detection cell described in [27].

Evaluation of the quality of the receptor coating. The recognition layer based on MIP@SiO₂/MCNC or SiO₂@SiO₂/MCNC composites was characterized by the piezoelectric quartz crystal microbalance method similar to that described in [16, 17]. As an analytical signal, we used the change in the oscillation frequency of the resonator before and after the application of a nanocomposite dispersion, as well as before and after the desorption and sorption of template molecules.

The mass of the recognition coating $\Delta m_{\rm MIP}$ (µg) was calculated based on the Sauerbrey equation: $\Delta m_{\rm MIP} = (S\Delta F)/(2.3 \times 10^6 F_0^2)$, where $\Delta F = F_1 - F_2 (F_1)$ is the oscillation frequency of the pure sensor, Hz; F_2 is the oscillation frequency of the sensor with the MIP film, Hz); F_0 is the natural oscillation frequency of the quartz crystal, MHz; S is the electrode surface area (0.10066 cm^2) . The influence of nonspecific surface sorption was taken into account by measuring the analytical signal of the sensor ΔF_{an} (Hz) relative to the NIP: $\Delta F_{an} = \Delta F_{MIP} - \Delta F_{NIP}$. The mass of the removed $(\Delta m_{\rm MIP})$ and re-embedded template molecules $(\Delta m_{\rm EMB})$ was determined from the difference between the mass of the MIP layer before and after the removal of the templates. The molar concentration of macrolide surface imprints R (µmol/cm²) on the sensor electrode surface was determined as follows: R = $\Delta m_{\rm EMB}/SM$, where M is the molar mass of the template.

The selectivity of the recognition layer was calculated from the cross response coefficients: CR, $\% = \Delta F_{int}/\Delta F_{analyte}$, where $\Delta F_{analyte}$ and ΔF_{int} are the analytical signals of the sensor in solutions of the analyte and the interfering substance.

RESULTS AND DISCUSSION

Synthesis of polymer particles with molecular imprints of macrolides by the core-shell method. *Synthesis of cores.* The size and morphology of the silica particles used as cores are largely determined by the ratio of reactants during the synthesis. The procedures

that provide the formation of particles with active functional groups on the surface of the core were chosen as the basis [28–30]. The cores were synthesized according to the Stober method, including the hydrolysis and subsequent condensation of organosilicon compounds [36]. The TEOS and APTES used in the work ensured the formation of surface active hydroxyl and amino groups (Table 1). The average diameters of the synthesized SiO₂ cores were determined by atomic force microscopy (Fig. 2).

The influence of the SiO₂ particle diameter on the density and uniformity of the layer on the surface of magnetic carbon nanocomposites was established by the piezoelectric quartz crystal microbalance method. Thus, particles synthesized by method 2 and having a smaller diameter (77 \pm 3 nm) were more compactly arranged on the MCNC surface, forming a uniform layer with a sufficiently large surface area. When particles of larger diameters (157 \pm 7 and 148 \pm 7 nm) obtained by methods 1 and 3 were used, an accumulation of large morphologically inhomogeneous structures was observed on the MCNC surface, which sterically hindered interaction with the surface layer.

Synthesis of the polymer shell was carried out by free radical polymerization or the sol-gel method. In the first case, because of the use of hydrophobic functional and cross monomers in the preparation of the shell, preliminary hydrophobization of SiO_2 nanospheres with a solution of polyvinylpyrrolidone was carried out in order to enhance bonding between the polymer layer and the core, reduce particle aggregation, and increase their stability. In synthesizing the polymer shell by SGM, hydrophobization was not carried out, because the stages of the formation of the core and surface shell were carried out sequentially using the same reagents.

In the synthesis of MIP by the FRP method, MAA and EGDMA were used as functional and cross monomers, which proved themselves well earlier [18– 20]. An important condition for the successful synthesis is the choice of the ratio of reagents in the polymerization mixture that provides the maximum number of available surface imprints. The optimal ratios of template : functional monomer and template : functional monomer : cross monomer were determined taking into account the maxima in the absorption spectra in the UV region (180–380 nm) by varying the amount of the reagents in the polymerization mixture (Fig. 3).

The maximum absorbances were recorded in the absorption spectra at the following molar ratios of templates : functional monomer : cross monomer: for erythromycin 1:8:60, for azithromycin 1:4:30. Different ratios of reagents in the synthesis of MIP of individual macrolides are probably associated with differences in the structures of their molecules. In view of the fact that the nonspecific sorption of the template is possible on the surface of a silica core, and



Fig. 2. AFM image of SiO_2 particles synthesized according to method 2 and a histogram of particle size distribution.

molecular imprints are formed in a thin polymer layer, the concentration of the template used in the synthesis of MIP was additionally increased. It was found that the maximum signal in the absorption spectra in the UV region was observed at a template : SiO_2 ratio of 4 : 1; therefore, in synthesizing the shell by the FRP method, a polymerization mixture with the composition template : functional monomer : cross monomer 4 : 32 : 240 (for erythromycin) and 4 : 16 : 120 (for azi-thromycin) was used.

A comparison of the IR spectra of the macrolide and MIP obtained by the FRP method (Fig. 4a) showed a shift of the peak of carbonyl groups to the shorter wavelengths from 1747 to 1722 cm⁻¹, which indicated the formation of hydrogen bonds between methacrylic acid (functional monomer) and template molecules during the formation of the prepolymerization complex. Also, the higher intensity of the -C-OH peak in the MIP spectra as compared to NIP ones indicates the presence of the carbonyl group of the lactone ring of macrolides in the imprinted polymer (988 cm^{-1}). The broadening of this band in the MIP spectra compared to the spectrum of the antibiotic was noted, which confirms the involvement of the hydroxyl groups of the macrolide in the hydrogen bonds with the functional monomer.

Similar changes in IR spectra were also observed during the formation of the shell by SGM (Fig. 4b). The presence of the macrolide lactone ring was indicated by a peak at 950 cm^{-1} , which is characteristic of antibiotics of this series. The formation of bonds between the template and the shell was confirmed by



Fig. 3. Selection of the ratio of reagents in the synthesis of polymer shell by free radical polymerization for (a) erythromycin, (b) azithromycin.



Fig. 4. Infrared spectra of polymers with molecular imprints of macrolides on the example of erythromycin obtained by (a) free radical polymerization and (b) sol-gel method: (1) antibiotic, (2) molecularly imprinted polymer, and (3) non-imprinted polymer.

the broadening and shift of the band of the carbonyl group from 1747 to 1706 cm⁻¹ compared to the spectrum of the macrolide. A comparison of the spectra of the SiO₂ shell sorbed by the macrolide and the desorbed shell made it possible to prove the successful removal of the template molecules.

Characteristics of an affine piezoelectric sensor based on composite nanomaterials. For the repeated use of the recognition layer of a gravimetric piezoelectric sensor based on MIP nanoparticles, it is important to carry out regeneration that ensures the removal of the maximum number of template molecules from the molecular imprints and a stable polymer structure. Due to the fact that macrolides in general are poorly soluble in water, but are easily soluble in ethanol, acetone, and some other polar organic solvents, a possibility of using ethanol; an ethanol-acetonitrile mixture (1:3); and an acetic acid-ethanol mixture (1:6) as regenerating solutions was studied. Experimental data have shown that the use of acetonitrile-ethanol and acetic acid-ethanol mixtures led to a significant destruction of the polymer already after the first regeneration cycle. Only the use of ethanol made it possible to achieve satisfactory results in the desorption and resorption of macrolide molecules.

An most important characteristic is the duration of a contact of the receptor layer based on MIP particles with the regenerating solution, which ensures the most complete removal and subsequent incorporation of antibiotic molecules. Using the piezoelectric quartz crystal microbalance method, it was found that the complete recovery of erythromycin molecules by ethanol from the SiO₂@MIP polymer matrix occurred within 4 min, and the maximum incorporation of antibiotics into the formed imprints was after 3 min (Fig. 5), which indicates a high availability of the imprints in the MIP structure to template molecules. The duration of the complete extraction of the template from the SiO₂@SiO₂ structure was 10 min, while the maximum incorporation occurred in 6 min, which indicates the lower availability of the imprints to the template molecules in the polymer shell obtained by the sol-gel method.

Table 2 presents analytical characteristics of the piezoelectric sensor based on composite nanomaterials. It was noted that, during the synthesis, the mass of antibiotics (m_{template} , μg) incorporated in the SiO₂@SiO₂ or SiO₂@MIP structures significantly exceeded the similar characteristic obtained upon repeated embedding ($\Delta m_{\rm EM1}$ and $\Delta m_{\rm EM2}$), which is explained by the partial destruction of the matrix adjacent to the imprint during primary regeneration. However, as a result of subsequent procedures of the desorption-sorption of antibiotics into molecular imprints, the masses of the re-embedded template molecules, as well as the concentrations of surface imprints $(R_1 \text{ and } R_2)$, almost coincided, which indicates the formation of a constant number of available imprints on the particle surface and the stability of the developed sensor.

It was found that the sensor signal based on MIP@SiO₂/MCNC was linear in the range 5–160 μ g/mL for azithromycin and 10–160 μ g/mL for erythromycin. For the sensor based on SiO₂@SiO₂/MCNC, an increase in the upper limit of



Fig. 5. Change in the mass of polymer structures using an example of $SiO_2@MIP$ depending on the duration of (a) desorption and (b) sorption from a solution of erythromycin of a concentration of 40 µg/mL.

the determined erythromycin concentration to $400 \ \mu g/mL$ was observed, which indicates a higher concentration of surface imprints in the shell obtained by the sol-gel method.

The developed sensors were tested in the determination of macrolides in pork and chicken meat (Table 3). It was found that in all samples the concentrations of antibiotics did not exceed the maximum allowable level of the drugs (0.2 mg/mL) [37]. The RSD values indicate a high reproducibility of the measurement results.

The selectivity of the sensor based on molecularly imprinted particles was evaluated using the cross-reactivity coefficient CR (%) (Table 4). It was found that molecularly imprinted particles synthesized by the FRP and SGM methods have high selectivities with

Parameter	Coating based on	Coating based on SiO ₂ @SiO ₂ /MCNC	
	azithromycin	erythromycin	
m _{MIP} , μg	3.2 ± 0.3	3.0 ± 0.6	2.3 ± 0.7
$m_{\text{template}}, \mu \mathbf{g}$	1.6 ± 0.1	1.6 ± 0.3	1.3 ± 0.4
R, mmol/cm ²	0.021 ± 0.001	0.018 ± 0.004	0.018 ± 0.006
$\Delta m_{\rm EM1}, \mu { m g}$	0.7 ± 0.1	0.8 ± 0.2	1.1 ± 0.2
R_1 , mmol/cm ²	0.009 ± 0.003	0.011 ± 0.002	0.014 ± 0.003
$\Delta m_{\rm EM2}, \mu { m g}$	0.6 ± 0.2	0.8 ± 0.1	1.0 ± 0.2
R_2 , mmol/cm ²	0.009 ± 0.001	0.010 ± 0.003	0.014 ± 0.005
Calibration function equation	$\Delta f = 0.4c + 469$	$\Delta f = 0.6c + 385$	$\Delta f = 1.9c + 394$
Range of determined concentrations, $\mu g/mL$	5-160	10-160	20-400
Correlation coefficient	0.99	0.98	0.96
Limit of detection, µg/mL	0.5	0.5	5

Table 2. Characteristics of sensors based on MIP@SiO₂/MCNC and SiO₂@SiO₂/MCNC (n = 3, P = 0.95)

Table 3. Results (μ g/mL) of erythromycin determination in meat by the added-found method using a sensor with a recognition coating based on MIP@SiO₂/MCNC (n = 5, P = 0.95)

Added	Found	RSD, %			
Chicken meat					
25	25 ± 1	2			
50	51 ± 1	2			
100	101 ± 3	3			
Pork					
25	32 ± 2	5			
50	57 ± 2	3			
100	108 ± 2	2			

respect to antibiotics of other classes. However, the CR values for other macrolides exceed 70% or more, which suggests that MIP with individual macrolide imprints can be used to determine the amount of erythromycin antibiotics.

Sensors based on magnetic carbon nanocomposites and molecularly imprinted polymers synthesized by FRP and SGM methods can be recommended for determining the amount of macrolides in the presence of other antibiotics in liquid media.

CONCLUSIONS

A procedure for forming the receptor layer of a piezoelectric sensor based on magnetic carbon nanocomposites with molecularly imprinted core—shell polymers for the determination of macrolide antibiotics in liquid media was proposed. The influence of the diameter of SiO₂ cores on the density and uniformity of the laver on the surface of magnetic carbon nanocomposites was established. It was noted that the sensor based on MIP synthesized by free radical polymerization is characterized by a linear dependence of the analytical signal on the analyte concentration in the range $5-160 \ \mu g/mL$ for azithromycin and 10-160 µg/mL for erythromycin. The results of the determination of macrolides using a piezoelectric sensor coated on the basis of MIP@SiO₂/MCNC are comparable in sensitivity to the well-known method of analvsis by HPLC-UV with preliminary solid-phase extraction [38] (LOD = $0.45 \,\mu\text{g/mL}$) and voltammetric method [39] (LOD = $1.4 \,\mu g/mL$). The developed piezoelectric sensors ensure the determination of macrolides in a wider concentration range. In using MIP obtained by the sol-gel method, the upper limit of the analytical range increased to 400 µg/mL. The sensors were tested in the determination of macrolides in pork and chicken meat.

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CONFLICT OF INTEREST

The authors declare that they have no conflicts of interest.

Table 4. Evaluation of the selectivity of the determination of erythromycin antibiotics using piezoelectric sensors with a recognition layer based on composite nanomaterials

Template	Coating	Analyte	CR, %
Erythromycin	MIP@SiO ₂ /MCNC	Erythromycin	100
		Azithromycin	70
		Penicillin G	12
		Tetracycline	8
Azithromycin	MIP@SiO ₂ /MCNC	Erythromycin	75
		Azithromycin	100
Erythromycin	SiO ₂ @SiO ₂ /MCNC	Erythromycin	100
		Azithromycin	84
		Penicillin G	19
		Tetracycline	15
Azithromycin	SiO ₂ @SiO ₂ /MCNC	Erythromycin	88
		Azithromycin	100

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