

Subcellular Localization of Manganese in Two Green Microalgae Species with Different Tolerance to Elevated Mn Concentrations

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Abstract—Green microalgae *Lobosphaera* sp. IPPAS C-2047 and *Micractinium simplicissimum* IPPAS C-2056 were examined for the first time for cell tolerance to elevated concentrations of manganese applied in the form of MnCl₂. Analyses of cell photosynthetic activity by chlorophyll fluorometry and the dynamic patterns of absorbance changes of cell suspensions at the peak of chlorophyll absorption revealed differential tolerance of two microalgae species to manganese. The acute toxicity assayed in 4-day treatments became apparent at manganese concentrations equal to or higher than 1 g/L for *M. simplicissimum* cells and at concentrations above 10 g/L for *Lobosphaera* sp. Transmission electron microscopy and energy-dispersive X-ray spectroscopy were used to study the subcellular distribution of manganese in microalgal cells under elevated nontoxic concentrations of manganese in the medium. The results with *Lobosphaera* sp. established the lack of individual manganese-containing inclusions on cell surfaces and in the cell interior; the intracellular distribution of manganese was dispersed with elevated accumulation of this element in the region of thylakoids and plastoglobules. The occurrence of manganese and phosphorus in plastoglobules was found for the first time. Apparently, these compartments become accessible for accumulation of Mn and P upon the translocation of thylakoid components during stress-induced disassembling of their structure. The cells of *M. simplicissimum* were able to oxidize the exoplasmic Mn²⁺ with the formation of manganese nanoparticles in the intercellular matrix as well as on the cell surface and within the cell walls. In addition, the manganese permeating into the cells was shown to compartmentalize in vacuoles and bind to the polyphosphate granules.

Keywords: *Lobosphaera*, *Micractinium simplicissimum*, analytical electron microscopy, intracellular localization, microalgae, manganese nanoparticles, tolerance, heavy metals

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INTRODUCTION

Biotechnologies based on the use of oxygenic phototrophic microorganisms—cyanobacteria and microalgae—are becoming widely spread in the production of biomass enriched with target products, such as proteins, carbohydrates, polyunsaturated fatty acids, minerals, vitamins, antioxidants, and mycosporin-like amino acids [1]. These microorganisms can also be used for purification of water areas and wastewater from heavy metals, biogenic elements, and organic compounds as well as for the creation of biosensors to assess the pollution of aquatic environments [2, 3].

Oxygenic phototrophic microorganisms, primarily microalgae, are frequently used in the field of environmental protection because they are suitable to trace the impact of adverse factors on individual cells and

cell populations over many generations and to unveil the long-term consequences of such an impact [4, 5]. Studies on the sensitivity of microalgae to various treatments require the search for algal species serving as pollution indicators as well as the species capable of rapid absorption, accumulation, and bioinactivation of pollutants in water bodies (bioremediation).

The problem of environmental pollution with heavy metals is becoming increasingly urgent [6]. Current technologies for wastewater treatment actively apply oxygenic phototrophic microorganisms whose cells can accumulate high concentrations of many elements, including heavy metals [7]. Manganese is one of the most abundant heavy metals in natural environments; it is a dangerous pollutant of aquatic ecosystems and wastewater [6]. The most common causes of metal ingress into wastewater include the enrichment of oxidized manganese ores, the manufacture of galvanic cells and ceramics, and organic syntheses. It should be noted that manganese is widely used as a component of fuel additives to increase the octane number [8].

Abbreviations: Chl—chlorophyll; EDX—energy dispersive X-ray spectroscopy; OD—optical density (absorbance); PAR—photosynthetically active radiation; PSII—photosystem II; PSA—photosynthetic activity; STEM—scanning transmission electron microscopy; TEM—transmission electron microscopy.

When heavy metals permeate into the cells, they react with functional groups of proteins and other compounds. These interactions may provide a detoxification mechanism (formation of metallothioneins and phytochelatins) and, at the same time, they can lead to numerous metabolic disorders underlying the high toxicity of heavy metals [9]. There is evidence for toxic effects of high manganese concentrations on plant cells [10]. However, there are hardly any studies in which the toxicity tests were combined with the analysis of intracellular manganese distribution in microalgae.

The search for microalgal cultures tolerating high manganese concentrations, the determination of concentration ranges at which microalgae retain high metabolic activity, and the studies of manganese intracellular distribution have both practical and fundamental importance since they expand our knowledge of the mechanisms for adaptation of microalgal cells to this heavy metal.

In this work, the influence of high manganese concentrations on photosynthetic activity of green microalgae *Micractinium* and *Lobosphaera* was studied to assess the ranges of their tolerance. In addition, the subcellular distribution of the toxicant was characterized by means of energy dispersive X-ray spectroscopy (EDX).

MATERIALS AND METHODS

Algal Strains and Culture Conditions

Suspension cultures of green microalgae *Lobosphaera* sp. IPPAS C-2047 (hereafter *Lobosphaera* sp.) and *Micractinium simplicissimum* IPPAS C-2056 (hereafter *M. simplicissimum*) were used. Microalgae were subcultured at regular intervals and grown in 750-mL glass flasks filled with BG-11 medium [11]. The culture vessels were mounted on an Innova 44 incubator shaker (New Brunswick, United States) and stirred at a speed of 150 rpm under constant temperature of 25°C and continuous illumination (40 $\mu\text{mol photons}/(\text{m}^2 \text{ s})$ of photosynthetically active radiation (PAR)).

For cultivating microalgae under excessive amounts of Mn^{2+} (hereafter manganese), the cells in the preculture were precipitated by centrifugation (3000g, 5 min) on a shaker (Eppendorf, Germany) and resuspended in BG-11 medium to a density of 30 mg chlorophyll/L. The chlorophyll (Chl) content in cell samples was assayed as previously described [12]. Solutions of manganese chloride (MnCl_2) in BG-11 medium were prepared separately for each given concentration. The incubation of microalgae at elevated manganese concentrations was performed using 12-well plates (Eppendorf, Germany). Cultivation in plates was carried out in accordance with the recommendations for determining the culture tolerance to various pollutants [13, 14]. One milliliter of a concentrated cell suspension and 2 mL of BG-11 medium with a given manganese concentration were added to the wells of the

plates. Thus, each well of the plate contained 3 mL of cell suspension of the same density (10 mg Chl/L) and different concentrations of manganese.

Cell suspensions of *Lobosphaera* sp. were incubated in media containing the following concentrations of manganese: 0.25, 0.5, 1.0, 5.0, 10.0, 15.0, and 25.0 g/L. The suspensions of *M. simplicissimum* cells were incubated in media with 0.05, 0.1, 0.5, 1.0, 3.0, and 5.0 g/L manganese. The above ranges of manganese concentrations were based on preliminary experiments that showed substantially different levels of lethal Mn^{2+} concentrations for these microalgae in acute toxicity tests. The cells of *Lobosphaera* sp. exhibited higher tolerance of manganese concentrations compared to *M. simplicissimum*. The control cultures were grown in a standard BG-11 medium containing 0.005 g/L manganese. The Mn-treated cells were cultured in a luminostat under continuous irradiance of 40 $\mu\text{mol photons}/(\text{m}^2 \text{ s})$ PAR and 23°C for 4 days.

To analyze the extracellular and subcellular distribution of manganese at elevated Mn concentrations in outer media using energy dispersive X-ray spectroscopy (EDX), microalgal cells were sedimented by centrifugation (3000g, 5 min) and resuspended in BG-11 medium to a density of 10 mg Chl/L. The suspensions of *Lobosphaera* sp. and *M. simplicissimum* were then supplemented with 5.0 and 0.5 g/L of manganese, respectively. The cells were cultured for 5 days under the conditions described above.

Determining Optical Density of Microalgal Suspensions

The physiological condition of microalgae was monitored directly in the wells of cell culture plates using an Infinite M200PRO spectrophotometer (Tecan, Austria). For this purpose, the difference in optical density (OD) of cell suspensions at 678 nm (the peak of chlorophyll absorption in cell cultures) and 750 nm (the spectral region of non-specific absorption) was determined. The OD was first recorded in 3 h and then after 2 and 4 days of cultivation in media with elevated manganese concentrations.

Determining Photosynthetic Activity

The photosynthetic activity (PSA) of algal cells was assessed by pulse-modulated chlorophyll fluorometry [15] using a FluorCam FC 800-C fluorometer (Photon Systems Instrument, Czech Republic). The PSA was assessed from the maximum photochemical quantum yield of photosystem II: $\phi(\text{PSII}) = F_v/F_m$ [16]. Microalgal cells were dark-adapted for 10 min prior to PSA measurements.

Transmission Electron Microscopy (TEM)

The ultrastructural organization of microalgae was inspected using conventional TEM of ultrathin

(<70 nm) sections. The elemental analysis was performed by analytical TEM, namely, the EDX of semithin (200–250 nm) sections. The cells were sedimented by centrifugation (3000g, 5 min) and fixed according to a standard protocol [17]. The pellets were successively treated with 2% (v/v) glutaraldehyde solution in 0.1 M cacodylate buffer (pH 6.8–7.2 depending on pH of the culture) for 30 min and then stained with 1% solution (w/v) of osmium tetroxide in the same buffer for 4 h at room temperature.

The fixed specimens were dehydrated in a series of ethanol–water solutions with increasing ethanol concentrations from 30 to 96% (v/v) followed by the triple dehydration with 100% ethanol (Sigma, United States). The last dehydration procedure was combined with contrasting by 2% uranyl acetate solution (w/v) in absolute ethanol (Sigma, United States) for 1 h at room temperature. The samples were embedded in Araldite epoxy resin (Sigma–Aldrich, United States) and polymerized at 56°C. The sections of a given thickness were obtained by means of an LKB-8800 ultramicrotome (LKB, Sweden). The sections were mounted on copper grids for electron microscopy that were covered with an ultrathin formvar resin support (Ted Pella, United States). To study the cell ultrastructure by TEM, the sections were additionally contrasted with lead citrate solution [18]. Images were obtained using a JEM-1011 electron microscope (JEOL, Japan). The dimensions of structures on TEM micrographs were determined using the Fiji (ImageJ) software v. 20200708-1553 (NIH, United States).

Energy-dispersive X-ray spectroscopy (EDX) of semithin sections was performed as previously described [19] with a JEOL-2100 analytical electron microscope (JEOL, Japan) equipped with a bright-field detector for operation in the scanning TEM (STEM) mode (JEOL, Japan). The microscope was also equipped with an X-Max silicon drift detector having an active crystal area of 80 mm² (Oxford Instruments, United Kingdom). Energy-dispersive X-ray spectra from the selected points or zonal regions of the specimen were recorded in the X-ray energy range from 0 to 10 keV using the bright-field STEM mode. The signal acquisition time for one spectrum was 120 s. Energy-dispersive spectra were recorded and processed in the “Point & ID mode” using the INCA program (Oxford Instruments, United Kingdom). The resulting spectra were displayed for the energy range of 0.15–7.5 keV that comprises the most intense peaks of all biologically significant elements, including Mn.

Statistical Processing of Results

All experiments were performed in three biological replicates. Significant differences between the treated and control samples were revealed with the Student's *t*-test. Differences were considered significant at $P \leq 0.05$. The results were processed using the

Microsoft Excel program. Figures show the mean values and standard deviations of the means.

RESULTS

Assessment of Culture Tolerance

Tolerance of microalgae cultures was assessed according to the standard bioassay procedure as recommended by state regulations, GOST P 54496-2011 [20]. In our experiments, we determined the manganese concentrations in the range from minimally active to toxic for algal cells. According to the GOST rules, the decrease of analyzed parameters in microalgae cells by 20% compared to normal physiological values indicates a nontoxic concentration of the pollutant. The decrease of the analyzed parameter by more than 50% with respect to the control value is considered as a manifestation of acute toxicity.

After 3-h incubation of *Lobosphaera* sp. in the BG-11 medium with the highest manganese concentration of 25.0 g/L and after incubation of *M. simplicissimum* in the presence of 3.0–5.0 g/L manganese, the optical density of cell suspensions decreased by 31–32%, which indicated the approaching manganese toxicity for both cultures (Fig. 1a). It should be noted that, at external manganese concentrations ≤ 15.0 g/L, the OD of *Lobosphaera* sp. suspensions decreased by no more than 20%, which indicates an extremely high tolerance of the studied culture to this heavy metal.

During the next 2 days of cultivation of *Lobosphaera* sp. in the BG-11 medium supplemented with 1.0–25.0 g/L manganese, the OD of the suspension decreased significantly (by 37–50%) with respect to the control value (Fig. 1b). For *M. simplicissimum* cells, the decrease in OD by 38–42% was observed at manganese concentrations above 3.0 g/L (Fig. 1e). On the fourth day, the OD of the *Lobosphaera* sp. cultures continued to decrease at all tested manganese concentrations as compared with that on the 2nd day (Figs. 1b, 1c). The OD of *M. simplicissimum* cell suspension on the fourth day of incubation at manganese concentrations up to 0.5 g/L was almost the same as on the second day, while it continued to decrease at manganese concentrations of 1.0–5.0 g/L (Figs. 1e, 1f).

Fluorometric studies of photosynthetic apparatus in microalgae cells at increasing manganese concentrations showed that the photosynthetic activity of *Lobosphaera* sp. diminished by more than 30% at manganese concentrations above 10 g/L, whereas a similar reduction in *M. simplicissimum* cells was attained at concentrations above 1.0 g/L (Fig. 2).

Data in Table 1 illustrate the comparative toxicity of manganese to *Lobosphaera* sp. and *M. simplicissimum* cells, based on the decrease in photosynthetic activity and OD (in percents) at various Mn concentrations with regard to the same parameters of cell suspensions in the standard medium. The results show that manganese concentrations up to 5.0 g/L had no

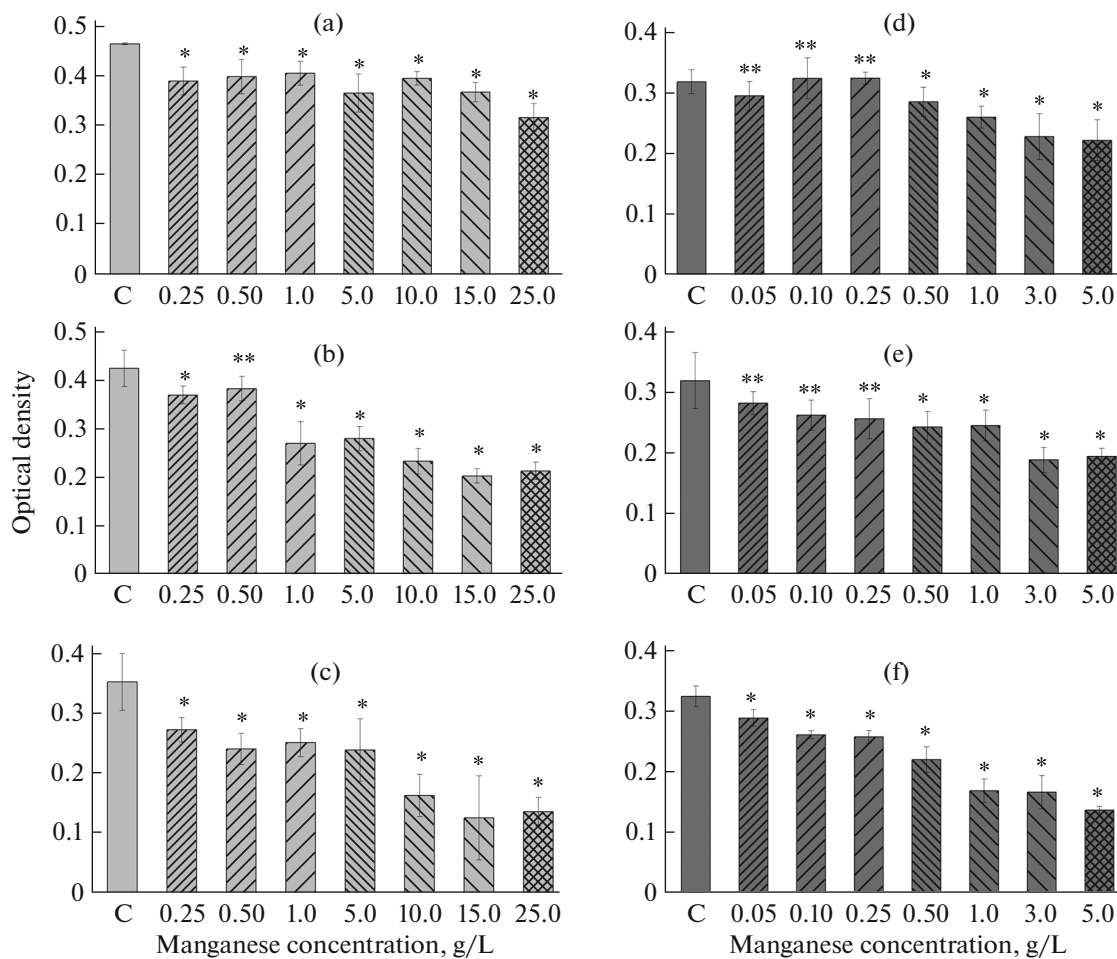


Fig. 1. Optical densities of cell suspensions of (a, b, c) *Lobosphaera* sp. IPPAS C-2047 and (d, e, f) *Micractinium simplicissimum* IPPAS C-2056 after cultivation of microalgae for (a, d) 3 h, (b, e) 2 days, and (c, f) 4 days in the BG-11 medium containing 0.005 g/L manganese (Control, C) or elevated Mn concentrations. Data are presented as mean values and standard deviations for three biological replicates. Significant differences were revealed by the Student's *t*-test (* $P \leq 0.05$, ** $P > 0.05$).

acute toxic effect on *Lobosphaera* sp., indicating a high tolerance of this culture to the tested heavy metal. On the other hand, at manganese concentrations ranging from 10.0 to 25.0 g/L, the OD decreased by $\geq 55\%$ and photosynthetic activity reduced by 32–54% depending on Mn level in the medium (Figs. 1, 2, Table 1). Similar reductions of analyzed parameters in *M. simplicissimum* cells were observed at manganese concentrations of ≥ 1.0 g/L, which indicates a higher sensitivity of this species to elevated amounts of manganese.

Subcellular Distribution of Manganese in Microalgae Cells

The ultrastructural organization of the studied microalgae corresponds to that of representatives of related taxa. The main features of these algae include unicellularity, the presence of a single nucleus and a single parietal chloroplast. Each chloroplast contains a pyrenoid surrounded by a starch sheath having the form of an egg shell in *M. simplicissimum* or many elongated

grains in *Lobosphaera* sp. (Figs. 3a, 3b). The cell wall in both cultures consists of 2–3 layers differing in electron density, but there is no three-lamellar structure. The cytoplasm contains regular organelles, such as ribosomes, mitochondria, elements of the Golgi apparatus, and vacuoles. The storage structures comprise starch grains, the cytoplasmic oleosomes, vacuolar inclusions, and plastoglobules in the chloroplast. Cultivating the algae in media with an elevated nontoxic manganese concentration (5.0 g/L for *Lobosphaera* sp. and 0.5 g/L for *M. simplicissimum*) produced no principal changes in the cell ultrastructure. However, analytical microscopy revealed genus-specific subcellular distribution of manganese in the studied microalgae.

Analysis of semi-thin sections of *Lobosphaera* sp. cells grown on a medium with 5.0 g/L Mn did not reveal the formation of individual manganese-containing inclusions in any cellular (sub)compartments (cell wall, chloroplasts, oleosomes, vacuoles, cytosol). All EDX spectra plotted in the range of 0.15–7.50 keV (Fig. 4)

featured the lines characteristic of X-ray emission of carbon ($K\alpha = 0.28$ keV) and oxygen ($K\alpha = 0.53$ keV) that are the main elements of organic compounds occurring in biological samples and the epoxy resin used for the preparation of specimens. The spectra also revealed a peak ($L\alpha = 0.93$ keV) characteristic of copper, the main component of electron microscopy grids on which specimen sections were mounted before TEM analysis, as well as the peak of osmium ($K\alpha = 1.91$ keV) that was used for cell fixation. Only the EDX spectra recorded in the regions of thylakoids (Figs. 3c, 4a) and plastoglobules (Figs. 3d, 4b) contained small peaks at 5.90 and 6.49 keV corresponding to the characteristic emission of manganese. It should be noted that the spectra obtained in the area of plastoglobules of *Lobosphaera* sp. also contained the characteristic peak of phosphorus with $K\alpha = 2.01$ keV (Fig. 4b).

In the EDX spectra obtained from other cell sub-compartments, e.g., the cell wall (Fig. 4c) and starch grains (Fig. 4d), no discernible peak of manganese X-ray emission was recorded. It is possible that the level of characteristic Mn emission in the cytosol and vacuolar inclusions was very weak and did not exceed 10–12 relative units.

Analysis of sections of *M. simplicissimum* cells revealed the accumulation of electron-dense nanoparticles 10–20 nm in size on the cell surface (Figs. 3e, 3g), in the outer layer of cell wall (Fig. 3f), and in the intercellular matrix (Figs. 3e, 3g). In the latter case, the size of nanoparticles sometimes reached 40–60 nm. The EDX spectra recorded from nanoparticle agglomerates (Figs. 4e, 4f) contained three different intensity peaks characteristic of manganese emission ($L\alpha = 0.64$ keV, $K\alpha = 5.89$ keV, and $K\alpha = 6.49$ keV). The sufficiently high intensity and occurrence of all peaks of the manganese emission provide reliable evidence for the enrichment of analyzed nanoparticles with this element. Judging from EDX spectra of *M. simplicissimum* (Fig. 4g), the vacuolar inclusions were characterized by the predominance of phosphorus, P ($K\alpha = 2.01$ keV) and nitrogen, N ($K\alpha = 0.39$ keV); they also featured small peaks corresponding to the emission of Ca ($L\alpha = 3.69$ keV) and Mn ($K\alpha = 5.90$ keV). Analysis of other compartments of *M. simplicissimum* cells (cytosol, starch grains, thylakoids, pyrenoid, lipid globules, nucleus, etc.) did not reveal manganese lines in the EDX spectra.

DISCUSSION

One of the most important tasks of microalgae eco-physiology is to study the cell responses to heavy metal ions whose elevated concentrations have a toxic impact on diverse physiological processes. This problem has obvious practical importance in view of ever-increasing environmental pollution with heavy metals and due to the use of microalgae for bioindication and biotesting of aquatic environments [21]. It is also of fundamental significance for clarifying the mecha-

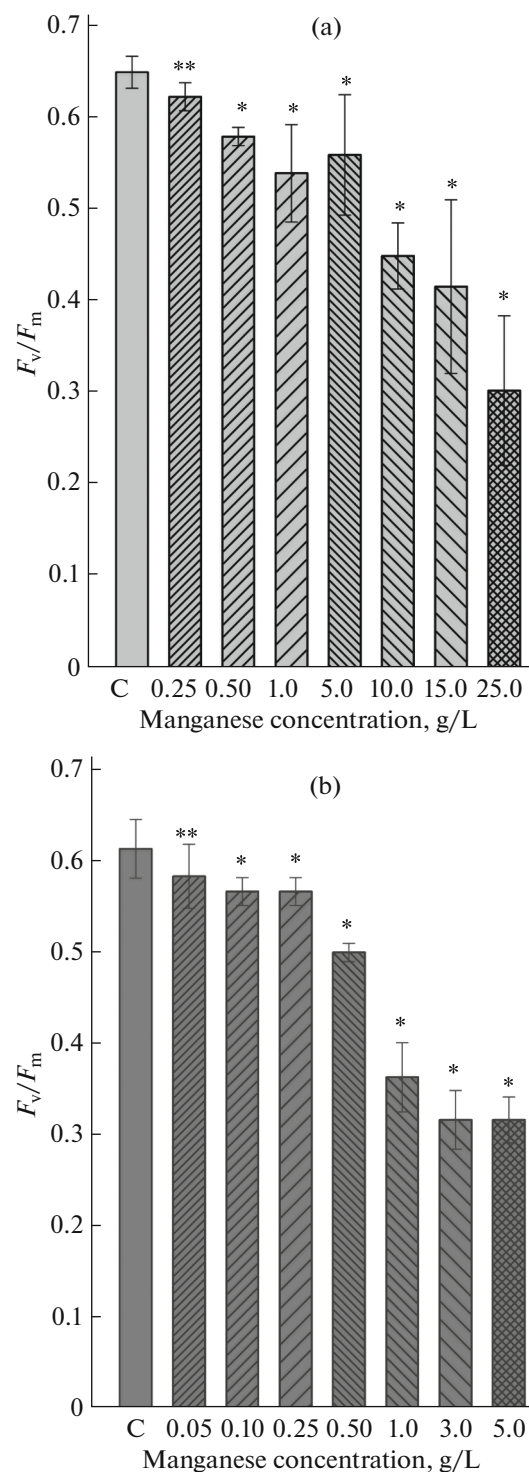


Fig. 2. Photosynthetic activity of (a) *Lobosphaera* sp. IPPAS C-2047 and (b) *Micractinium simplicissimum* IPPAS C-2056 cells after 4-day culturing in BG-11 media containing 0.005 g/L manganese (Control, C) or elevated Mn concentrations. Data are presented as mean values and standard deviations for three biological replicates. Significant differences were revealed by the Student's *t*-test (* $P \leq 0.05$, ** $P > 0.05$).

Table 1. Changes in optical density at the peak of Chl absorption and in photosynthetic activity of cell suspensions of *Lobosphaera* sp. IPPAS C-2047 and *Micractinium simplicissimum* IPPAS C-2056 after 4-day cultivation at elevated manganese concentrations

Mn ²⁺ concentration, g/L	Relative decrease of parameters with respect to the control values, %		Presence/absence of toxicity
	optical density of suspension	photosynthetic activity (F_v/F_m)	
<i>Lobosphaera</i> sp.			
0.25	21.8 ± 3.5	5 ± 1	Nontoxic
0.5	31.0 ± 2.8	11 ± 2	Nontoxic
1.0	27.9 ± 3.3	17 ± 2	Nontoxic
5.0	31.4 ± 5.7	14 ± 3	Acute toxicity is unproven
10.0	55.0 ± 4.2	32 ± 3	Toxic
15.0	62.6 ± 7.0	37 ± 5	Toxic
25.0	60.7 ± 4.6	54 ± 6	Toxic
<i>M. simplicissimum</i>			
0.05	6.1 ± 1.5	5 ± 1	Nontoxic
0.10	15.1 ± 0.9	8 ± 2	Nontoxic
0.25	16.2 ± 0.7	8 ± 1	Nontoxic
0.50	28.4 ± 1.9	18 ± 2	Acute toxicity is unproven
1.0	42.1 ± 2.1	40 ± 2	Acute toxicity is unproven
3.0	45.9 ± 3.8	53 ± 3	Toxic
5.0	55.5 ± 1.3	55 ± 4	Toxic

nisms of adaptation and tolerance of microalgae and for studying the bioremoval of heavy metals from water by microalgae that accumulate large quantities of these metals in their biomass.

The photosynthetic apparatus of microalgae is extremely sensitive to the presence in the environment of pollutants, including heavy metals [16]. The study of the photosynthetic apparatus's condition in microalgae by fluorometry is one of the most popular methods for analyzing PSA [15].

The results of our study revealed the decline of PSA at increasing concentrations of manganese in the medium, which indicates the toxic action of elevated concentrations of this metal on the photosynthetic apparatus in both cultures. The decrease in OD of cell suspensions in the region of maximum Chl absorption might be caused, on the one hand, by stress-induced decline in intracellular Chl content reflecting the reduction of the photosynthetic apparatus and, on the other hand, it might be due to the decreased number of cells in suspensions.

The results of this study showed that cultures of *Lobosphaera* sp. and *M. simplicissimum* are characterized by differential tolerance to elevated manganese concentrations in the outer medium.

The upper tolerance limits of manganese concentrations are known to vary for different cultures of oxygenic phototrophic microorganisms [8] that are char-

acterized by different adaptability to high concentrations of metals. The adaptive capacity of cells is determined by their potential to restrict the entry of metal ions from the medium and by their ability to bind or sequester metal ions in the extracellular polymeric matrix or cell surface structures [22].

The results of our experiments showed that *M. simplicissimum* cells are resistant to external manganese concentrations in the range from 0.05 to 0.5 g/L. This conclusion stems from an insignificant decrease in OD of the suspension compared to the control values on the second day of Mn treatment, the lack of accelerated OD decrease on the fourth day, and from the persistent PSA levels. By contrast, the OD of suspensions decreased continuously throughout the experiment at concentrations above 1.0 g/L. The observed declines in OD and PSA indicate the relatively low tolerance of *M. simplicissimum* culture to these manganese concentrations. Cultivation of *Lobosphaera* sp. in a medium containing 0.25–5.0 g/L manganese showed that the OD of suspensions and PSA of cells on the fourth day of treatment were actually independent of the manganese concentration in the medium, while culturing at higher Mn concentrations considerably overwhelmed the adaptive capacity of the culture.

The toxicity of high manganese concentrations for plant cells is predominantly ascribed to oxidative stress, the disturbance of chloroplast oxidative balance

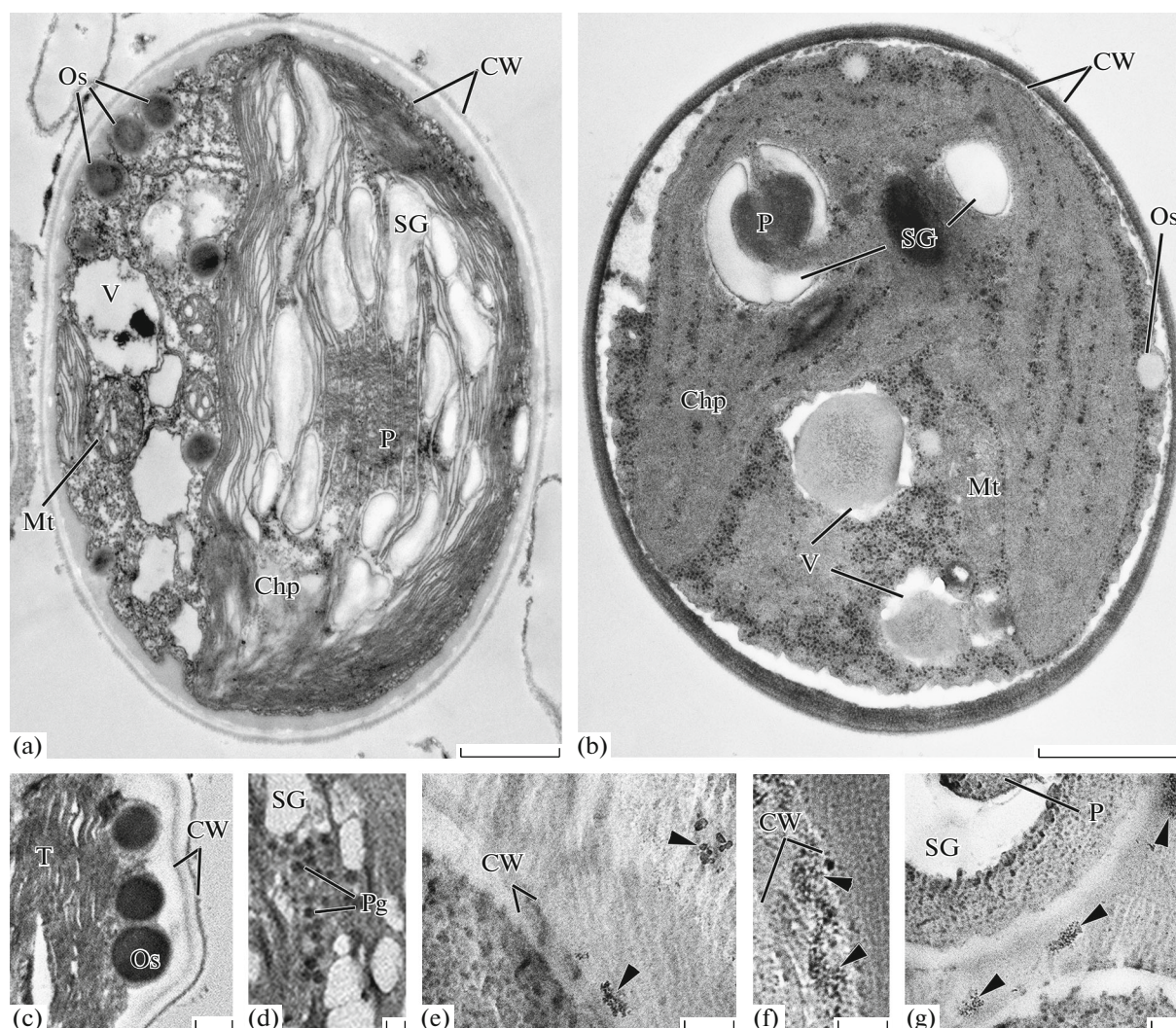


Fig. 3. Ultrastructural organization in microalgae (a) *Lobosphaera* sp. IPPAS C-2047 and (b) *Micractinium simplicissimum* IPPAS C-2056 (TEM images of ultrathin sections); (c–g) representative images of cell regions analyzed by EDX (STEM images of semi-thin sections): (c) chloroplast thylakoids of *Lobosphaera* sp. IPPAS C-2047, (d) plastoglobules in *Lobosphaera* sp., (e, g) nanoparticles on the cell surface and in the intercellular matrix of *M. simplicissimum* IPPAS C-2056, (f) nanoparticles in the cell wall. V—vacuole; SG—starch grain; CW—cell wall; M—mitochondrion; P—pyrenoid; Os—oleosome; Chl—chloroplast; T—thylakoids. Nanoparticles are indicated by arrowheads. Scale bars: (a, b) 1 μm ; (c–g) 0.2 μm .

in particular. The excess of manganese is known to enhance the activity of Mn-superoxide dismutase [23]; it may also inhibit cell growth and induce chlorosis and tissue necrosis [10, 24]. The different tolerance of *Lobosphaera* sp. and *M. simplicissimum* to elevated manganese concentrations might be associated with dissimilar mechanisms of acclimation, including the differential subcellular distribution of manganese in their cells.

In microalgae *Lobosphaera* sp., no formation of individual manganese-containing inclusions was found. The highest intensity of characteristic manganese emission was observed in thylakoids and plastoglobules, while the manganese signal in the cytosol and vacuoles was indiscernible. Other intracellular com-

partments and the cell wall contained manganese in undetectable amounts. The observed pattern seems to indicate a dispersed intracellular distribution of manganese and its selective accumulation in chloroplasts.

Manganese is a cofactor for a several enzymes and is present in chloroplasts as part of the Mn cluster of PSII oxygen-evolving complex [25]. Examination of sections of *Lobosphaera* sp. cells cultured at normal Mn concentrations demonstrated the absence of the manganese signal in all compartments, including chloroplasts, which is most likely due to limitations of EDX spectroscopy for detecting low concentrations of this element.

Judging from the available publications, the characteristic X-ray emissions of manganese and phos-

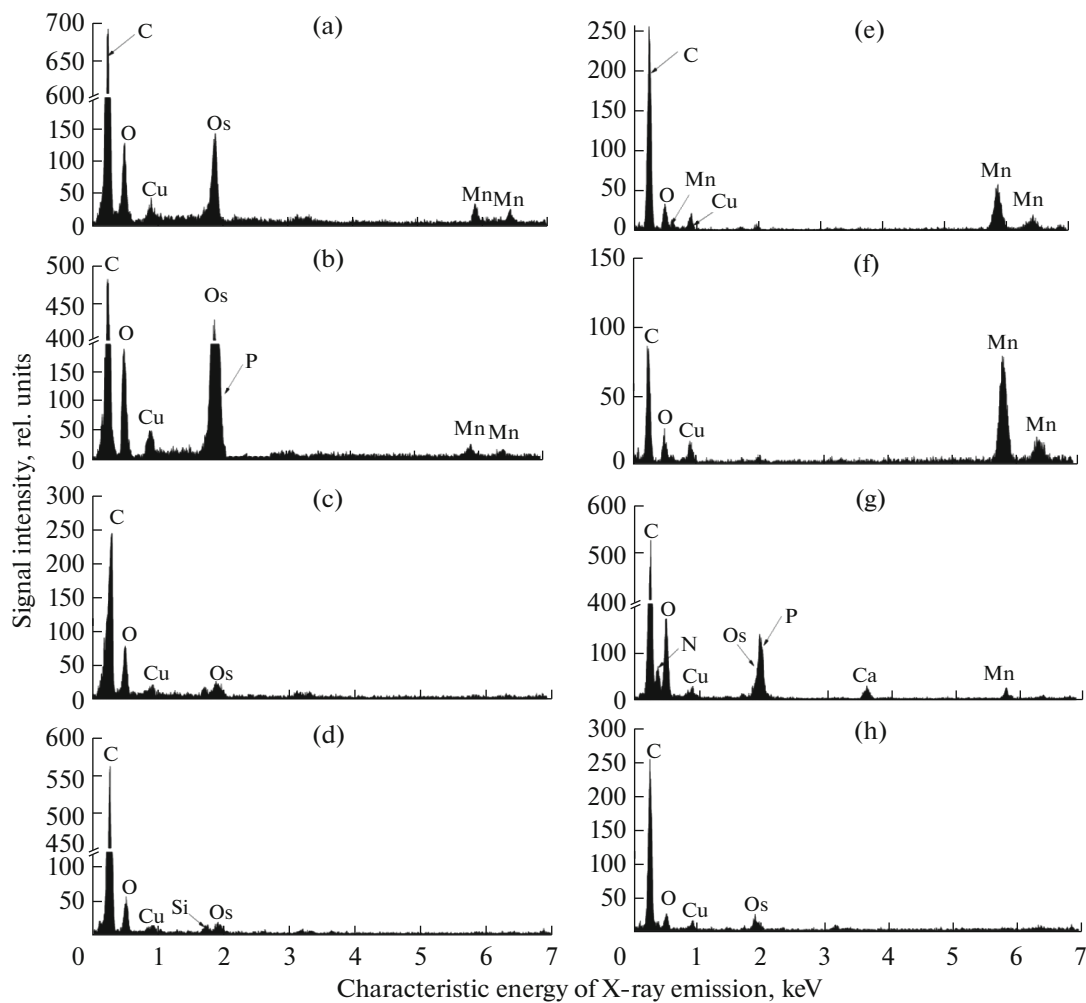


Fig. 4. Representative spectra of energy-dispersive X-ray spectroscopy for microalgae cells of (a–d) *Lobosphaera* sp. IPPAS C-2047 and (e–h) *Micractinium simplicissimum* IPPAS C-2056 in the areas of (a) thylakoids, (b) plastoglobules, (c) cell wall, (d) starch grain, (e) nanoparticle agglomerates on the cell surface, (f) cell wall, (g) vacuolar inclusions, and (h) pyrenoid.

phorus in plastoglobules have been revealed for the first time. We suppose that the detection of manganese and phosphorus in plastoglobules is determined by the ingress and accumulation of these elements in plastoglobules upon the rearrangement of thylakoid components during their disassembly under stress. Plastoglobuli are lipoprotein microcompartments integrated into plastidial metabolic functions, including the adaptation to variable environmental conditions [26]. Our results clearly demonstrate the previously unknown properties of plastoglobuli and expand the understanding of their phosphorylation networks and their potential under stress conditions.

In *M. simplicissimum* culture, the manganese nanoparticles were found to accumulate on the cell surface, in cell walls, and interstitial spaces (Figs. 3, 4). It is known that some types of bacteria and fungi are able to form manganese nanoparticles upon the oxidation of Mn^{2+} to Mn^{4+} and can produce insoluble man-

ganese oxide [27]. The exoplasmic manganese oxidation and the formation of nanoparticles were also described for some microalgae species, including a representative of the genus *Micractinium* [8]. Manganese oxidation occurs not only on cell surface structures but also in the extracellular matrix and cultivation medium. These extracellular reactions are facilitated, firstly, by the increase in external pH during photosynthetic carbon fixation and, secondly, by the involvement of superoxide radicals produced by extracellular proteins that are excreted by some microorganisms, including microalgae [8, 28]. The formation of manganese nanoparticles in the medium, on the cell surface, and in cell walls can be regarded as an adaptive strategy of microalgae cells in response to an increased amount of manganese in the medium, which restricts the Mn penetration into the cell.

Current nanobiotechnologies aimed at the synthesis of manganese nanoparticles by microbial cells are

being actively advanced [29]. Manganese nanoparticles exhibit electrocatalytic, magnetic, and fluorescent properties and possess antibacterial and antifungal activity. Therefore, they can be used in the production of supercapacitors and lithium rechargeable batteries as well as in medicine, analytical chemistry, detoxification of dangerous micropollutants, and other fields [29].

Analysis of *M. simplicissimum* sections showed the lack of characteristic peaks of manganese X-ray emission in intracellular subcompartments, except for vacuoles. In vacuoles, manganese was found in the composition of phosphorus-containing inclusions, which are similar in elemental composition to polyphosphate granules [19]. The accumulation of phosphorus reserves in such a form is known for many microalgae [19, 30, 31], including a representative of the genus *Lobosphaera* [32, 33]. Sanz-Luque et al. [31] explained the tolerance of microalgae to heavy metals Cd and Pb as a result of binding of these elements to vacuolar polyphosphate granules, which prevents the interference of toxicants in cell metabolism. With an example of *Chlamydomonas reinhardtii*, manganese was shown to be largely located in vacuoles, together with phosphorus and calcium [34]. Although manganese accumulation depended on the content of cellular polyphosphates, only a small amount of Mn^{2+} formed stable complexes with polyphosphates.

The tolerance of *M. simplicissimum* cells to elevated concentrations of manganese might be determined by several factors. First, the cells can perform exoplasmic oxidation and accumulation of insoluble forms of manganese. Second, manganese penetrating into the cells is compartmentalized in vacuoles or binds to polyphosphate granules or other ligands. The culture tolerance is compromised at manganese concentrations above 1.0 g/L; this might be due to the increased amount of manganese nanoparticles in cell walls, which impairs the functional transporters delivering nutrients into the cell and excreting cellular metabolites [35]. The high tolerance of microorganisms to heavy metals is known to rely on their ability to induce synthesis of specific metal-binding proteins and cysteine-rich peptides, such as metallothioneins and phytochelatins [36]. The formation of metallothioneins and phytochelatins has also been described for members of the Trebouxiophyceae family, in particular, for some species of *Chlorella* and *Micractinium* [9]. The mechanisms of cell tolerance to manganese also include its interaction with negatively charged groups of surface structures in microalgae cells [37]. The metal-binding sites in the cell wall of microalgae can be represented by protein thiol groups; amino acids; and the carboxyl groups of pectins, proteins, and uronic acids.

Thus, our study revealed the distinctions in tolerance of *Lobosphaera* sp. and *M. simplicissimum* to the increased amounts of manganese. This can be explained by different mechanisms underlying the cell

resistance to this heavy metal and by specific features of the subcellular manganese distribution in cells of the studied cultures.

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COMPLIANCE WITH ETHICAL STANDARDS

This article does not contain any studies with human participants or animals performed by any of the authors.

CONFLICT OF INTEREST

The authors declare that they have no conflicts of interest.

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