scientific reports

OPEN



Insights into the prediction of the liquid density of refrigerant systems by artificial intelligent approaches

Huaguang Li^{1⊠} & Alireza Baghban^{2⊠}

This study presents a novel model for accurately estimating the densities of 48 refrigerant systems, categorized into five groups: Hydrofluoroethers (HFEs), Hydrochlorofluorocarbons (HCFCs), Perfluoroalkylalkanes (PFAAs), Hydrofluorocarbons (HFCs), and Perfluoroalkanes (PFAs). Input variables, including pressure, temperature, molecular weight, and structural groups, were systematically considered. The study explores the efficacy of both the multilayer perceptron artificial neural network (MLP-ANN) and adaptive neuro-fuzzy inference system (ANFIS) methodologies in constructing a precise model. Utilizing a comprehensive dataset of 3825 liquid density measurements and outlier analysis, the models achieved R² and MSE values of 0.975 & 0.5575 and 0.967 & 0.7337 for MLP-ANN and ANFIS, respectively, highlighting their remarkable predictive performance. In conclusion, the ANFIS model is proposed as an effective tool for estimating refrigerant system densities, particularly advantageous in scenarios where experimental measurements are resource-intensive or sophisticated analysis is required.

Abbreviations

HFEs	Hydrofluoroethers
RMSE	Root mean squared error
HCFCs	Hydrochlorofluorocarbons
PFAAs	Perfluoroalkylalkanes
HFCs	Hydrofluorocarbons
PFAs	Perfluoroalkanes
MLP	Multilayer perceptron
BPT	Back propagation training
ANN	Artificial neural network
GA	Genetic algorithm
CFCs	Chlorofluorocarbons
RBF	Radial basis function
RITE	Research institute of innovative technology for the earth
FL	Fuzzy logic
MP	Montreal Protocol
AARD	Average absolute relative deviation
HAs	Halogenated alkanes
STD	Standard deviation
GC	Group contribution
\mathbb{R}^2	Coefficient of determination
EOS	Equations of state
MF	Membership function
LM	Levenberg-Marquardt

MSE Mean Squared Error

¹Intelligent Manufacturing College, Qingdao Huanghai University, Qingdao 266427, Shandong, China. ²Process Engineering Department, National Iranian South Oilfields Company (NISOC), Ahvaz, Iran. ^{\Box}email: lihuaguang0917@126.com; Alireza_baghban@alumni.ut.ac.ir

RDP Relative deviation percentage

ANFIS Adaptive neuro-fuzzy inference system

To create an economical refrigeration cycle for low temperatures, it is imperative to possess a thorough understanding of the thermodynamic characteristics of refrigerant systems, such as liquid density¹⁻⁴. Despite the extensive amount of experimental data available in the literature, there are still incongruities among various datasets^{5,6}. Therefore, it is crucial to calculate the thermodynamic characteristics of these substances for application in any scenario where experimental data is unavailable^{7,8}.

Before the 1980s, in the refrigeration industry, the main refrigerants were hydrochlorofluorocarbons (HCFCs) and chlorofluorocarbons (CFCs)⁹⁻¹¹. Since on January 1, 1996 production or use of CFCs was barred by Montreal Protocol (MP) so the quantity of CFCs decreased briskly^{12,13}. Because of obstruction, which the MP placed on CFCs, HCFCs were replaced with CFCs in different industries¹⁴. Although HCFCs have decreased ozone layer damage, they still pose contamination issues and are planned to be removed by 2030. Hydro fluorocarbons (HFCs) because of very low ozone layer damage, coincidence to HCFCs and CFCs in terms of physical properties, limited existence in the atmosphere, minor flammability or nonflammable and economical were used as a substitution of CFCs and HCFCs¹⁵. Due to the serious global warming effect of HFCs, HFEs were introduced as a new generation of refrigerants by $RITE^{16,17}$. Also, there are some compounds that have the potential of using refrigerant fluids such as perfluoroalkylalkanes (PFAAs) and perfluoroalkanes (PFAs)¹⁸. We do not have detailed studies that predict the thermodynamic properties of refrigerants by theoretical methods; so every author uses a special equation and method to forecast the thermodynamic properties of the refrigerant systems^{19,20}. As the authors are cognizant, almost all of these attempts have been confined to the finite systems, and we don't have a systematical work for testing the qualification of each method²¹⁻²³. The liquid density of refrigerants calculated applying 14 correlations and 4 equations of state (EOS) by Nasrifar and Moshfeghian²⁴. Lugo et al. worked on aqueous solutions (secondary refrigerants) and suggested another method for calculating many of their thermophysical properties²⁵. By forecasting the density of the HAs, HFEs, and their chemical mixtures, Scalabrin and his colleagues suggested density with a three-parameter model dependent on corresponding states²⁶. Maftoon-Azad et al. inspected analytical EOS for predicting the density of HCFC and HFC refrigerants (compressed liquids)²⁷ and for predicting the volumetric behavior of just six refrigerants they used the Ihm-Song-Mason equation²⁸. In 2005, Goharshadi and Moosavi founded the density 3 HFEs and 11 HFCs and HCFCs^{29,30}. Also, they applied this EOS for finite refrigerants. Generally, the previous equations or correlations of the state some adjustable parameters in addition to critical constants. For modeling the various thermodynamic properties, ANNs (Artificial Neural Networks) can be a convenient substitution^{31,32}. A neural network is contrived from a large number of interconnected neurons that the organization of this neurons connection determines the structure of a network^{33–35}. To achieve an overall desired behavior of the network, the arrangement of the connection strengths is controlled by the learning algorithms^{36,37}. In other words, if you learn the relationship between output and input vectors, the ANN is a useful algorithm to prospect each function that has a limited number of discontinuities³⁸. Hence, for modeling of the nonlinear treatment of chemical properties, the ANN is an appropriate technique^{39,40}.

Recently, effective attempts have been made depending on the GC (group contribution) approach attached to the ANN model to develop the prediction models. Many researchers applied the ANN-GCM method for calculating the thermodynamic properties of various substances^{41,42}. Namely, some properties of ionic liquids have been estimated with ANN-GCM methods such as melting point, density, thermal decomposition temperature, glass transition temperature, heat capacity, surface tension and viscosity⁴³⁻⁵³. Furthermore, this approach can be applied to anticipate the temperature at which flammability limits occur for organic compounds, the solid vapor pressure of pure compounds, the enthalpy of sublimation for organic compounds (at 298 K), the flash point temperature, and the vaporization enthalpy of organic compounds. It is also useful for predicting the densities of hydrocarbon systems, the specific volume of polymeric systems, and forecasting the density of liquid alkali metals^{47,54-61}.

Since the last decade, for accurate prediction of the refrigerant systems thermodynamical properties, some bounded efforts have been made to extend ANN models. Chouai et al. utilized the ANN method for PVT depiction of R32, R134a, and R143a in the temperature range of 240 to 340 K and pressure up to 20 Mega Pa⁶². Also, Mohebbi et al. used GA-ANN (a neural network relied on the genetic algorithm) for 6 mixed and 14 pure refrigerants to estimate the saturated liquid density⁵. Their model can predict this parameter with a mean absolute error (%) of 3.53 and 1.46 for mixed and pure refrigerants, respectively. The latest research in this area was conducted by Moosavi and his colleagues, who used ANN-GCM model to predict density of refrigerants¹⁸. Their model has the ability to predict this parameter with average absolute deviation of 0.28 at testing phase.

The purpose of this study is to estimate the liquid densities for various refrigerant mixtures at a wide range of pressures and temperatures with three intelligent approaches that contain MLP-ANN, and ANFIS. To develop these models, a large data set is used. Then, different statistical methods are used to evaluate and analyze the obtained models. Additionally, an outlier analysis will be performed to identify suspected points.

Theory MLP-ANN

ANN is a method that is acquired to rely on biological NNs created by a collection of interconnected nodes that admitted as artificial neurons. This method has the proficiency in handling signals sent by links between the nodes. Principally, every artificial neuron uses a non-linear aggregate of inputs of the neuron for determining the outputs. As well as, a weight parameter could be used to decrease or increase the strength of signals at the connections. Commonly, for a node, we have three categories of the activation functions that used to obtain the output from a determined collection of inputs (Eqs. 1–3):

Linear function:

$$h(y) = ySPS: id:: \epsilon 1 \tag{1}$$

Sigmoid function:

$$g(y) = \frac{1}{1 + e^{-y}}$$
(2)

Hyperbolic tangent function:

$$k(y) = \frac{e^{y} - e^{-y}}{e^{y} + e^{-y}} SPS : id : : \epsilon 3$$
(3)

Furthermore, an effective factor in the definition of ANNs is the bias term. MLP-ANN includes hidden, input and output layers and it is a feed-forward type of ANN. The activation function of this kind of ANN's nodes is non-linear, and a back propagation training (BPT) approach is applied in this model^{63,64}.

ANFIS

Primarily, Zadeh introduced the notion of Fuzzy Logic (FL), which had the capability to arrange outputs on a spectrum from completely false to completely true. In contrast, classical logic is only able to arrange conclusions as either true or false⁶⁵. Consociate of linguistic rules of if–then and principals of fuzzy logic developed the model. Using the basics of fuzzy logic helps us to touch a transparent development process and outputs with more accuracy. Coupling ANN and fuzzy logic make it possible to get precise solutions for extraordinarily non-linear systems^{66,67}. ANFIS is created of integrating from fuzzy logic and ANN. Mamdani and Takagi–Sugeno are two structures for FIS^{68–70}. Logical explanation in progress of fuzzy if–then rules used in the first FIS type, but the next type of FIS generates the if–then rules rely on the performance of accessible empirical data. The Takagi–Sugeno type inference system used in the ANFIS method to demonstrate the non-linear reliance of variables⁷⁰. An if–then rule is applied in a generic ANFIS structure for Y_1 and Y_2 (input parameters) as follow (Eqs. 4–7):

if
$$Y_1$$
 is C_1 and Y_2 is D_1 then $g_1 = p_1 Y_1 + q_1 Y_2 + s_1$ (4)

if
$$Y_1$$
 is C_2 and Y_2 is D_2 then $g_2 = p_2 Y_1 + q_2 Y_2 + s_2$ (5)

if
$$Y_1$$
 is C_1 and Y_2 is D_2 then $g_3 = p_3 Y_1 + q_3 Y_2 + s_3$ (6)

if
$$Y_1$$
 is C_2 and Y_2 is D_1 then $g_4 = p_4 Y_1 + q_4 Y_2 + s_4$ (7)

In these equations, g shows the output parameters and respectively $C_i and D_i (i = 1, 2)$ are fuzzy sets for $Y_1 and Y_2$. Generally, this structure has 5 layers. The initial layer for fuzzification utilizes the membership function to convert input data into linguistic terms. In this investigation, the GM is employed and is defined as follows:

$$O_i^1 = \gamma(P) = \exp\left[-\frac{1}{2}\frac{(P-Q)}{\Lambda^2}\right]$$
(8)

In Eq. (8), Q represents the center of the Gaussian distribution, Λ^2 refers to the variance and O is the output of the layer. For getting the most accurate model, GM should be optimized. In the second layer, by computing the commonly referred to firing strength parameters, it becomes possible to assess the dependability of the preceding components (Eq. 9):

$$O_i^2 = v_i = \gamma_{Ai}(P)\gamma_{Bi}(Q) \tag{9}$$

Moreover, in the third layer, the normalization of estimated firing strengths has been carried out (Eq. 10):

$$O_i^3 = \overline{v_i} = \frac{v_i}{\sum_i v_i} \tag{10}$$

In Eq. (11) for output parameter the linguistic terms are defined (fourth layer):

$$O_i^4 = \overline{v_i}g_i = \overline{v_i}(p_iY_1 + q_iY_2 + s_i) \tag{11}$$

where q, r_s and p_i are linear parameters for optimization. Eventually, in the fifth layer all of the rules associated to an output will be appear in the following formula (Eq. 12):

$$O_i^5 = \sum_i \overline{\nu_i} g_i = \frac{\sum_i \nu_i g_i}{\sum_i \nu_i}$$
(12)

Model development Preprocessing procedure

This paper shows three strategies such as ANFIS and MLP-ANN are applied to estimate density rely on molecular mass (Mw), pressure (P), the structural groups and temperature (T). The structural groups used as input parameters are shown in Table 1. The computational tools and platforms employed for model development and evaluation were crucial components of our methodology. Specifically, MATLAB (version 2020) served as the primary computational environment for implementing and assessing the ANFIS and MLP-ANN strategies applied to estimate density. The choice of MATLAB was driven by its versatility, extensive toolboxes for neural network implementation, and widespread use in scientific research.

Furthermore, the dataset employed for modeling purposes, consisting of 3825 data points, is meticulously sourced and extensively referenced throughout the manuscript^{16,71–86}. These data points, carefully selected from reputable and relevant literature, form the foundation of our study and contribute to the robustness and validity of our modeling approach. Testing and training subsets got from collected data. 25% of data are applied for testing, and 75% of data are occupied for training the recommended models. Normalization of data was performed by Eq. (13)⁸⁷:

$$\Delta_{Norm} = 2 \frac{D - D_{min}}{D_{max} - D_{min}} - 1 \tag{13}$$

where the Δ is the parameter value. Norm, max, and min stand for the normalized, maximum, and minimum values, respectively. The normalized data spans from -1 to 1. Density is the output of the model, and the input variables are the other four parameters that were mentioned earlier.

Model development

MLP-ANN

Following parameter shows the general output parameter for this model (Eq. 14):

$$Y_{\Sigma} = \sum_{i=1}^{n} \left(W_{3i} \frac{1}{1 + e^{-(x_i W_i)}} \right) + b_3$$
(14)

In Eq. (14), $W_{i,3}$ and *n* are respectively, the weight vector for neurons and for output layer neurons, and the number of hidden layer neurons. Also, b_3 is the bias term.

Additionally, decreasing the differences between real and estimated data gives us the optimum output parameter (using the ANN structure). The minimization occurs with regulating weight and bias parameters. In this task, we utilize the error function determined in Eq. (15):

$$S_{\Sigma} = \sum_{j} \sum_{i} \left(r_i^j - o_i^{j,l} \right) \tag{15}$$

The notation r_i^j represents the ith actual output for the jth data point, while $\sigma_i^{j,l}$ denotes the output of the ith neuron in the first layer, where j is the data point index in the training dataset. Applying the Levenberg–Marquardt algorithm causes optimization. Moreover, Fig. 1 shows the performance of the utilized network relies on the MSE calculated data by using MLP-ANN.

ANFIS

Figure 2 shows the diagram of a generic ANFIS includes two variables as input. Training the proposed ANFIS takes place by utilizing a genetic algorithm (GA). Equation (16) determines the whole parameters of this model that depends on the number of variables (N_v) , the number of clusters (N_c) and the number of MF parameters (N_{MF}) :

$$P_{Tot} = N_c . N_v . N_{MF} \tag{16}$$

In this manuscript, the MF utilized is the GM function. Zand σ^2 are the MF parameters. Pressure (P), temperature (T), molecular mass (Mw), and the structural groups are input variables. So, for 480 ANFIS parameters, the total number was obtained. For the GA algorithm used in achieving the optimum parameters of this model,

$-CF_3$	CF ₂ Cl ₂	$-CF_2H$
$-CCl_2H$	CF ₃ Cl	$-CH_2F$
$-CF_2Cl$	CF ₄	– CFH –
- CFClH	CCl ₂ FH	- CF ₂ -
$-CCl_2F$	CF ₂ ClH	-CH2-
$-CH_3$	CF ₃ H	-0-
CCl ₃ F	CF ₂ H ₂	

Table 1. Categories of structural groups examined in MLP-ANN and ANFIS models.



Figure 1. MLP-ANN's performance based on Mean Squared Error across various iterations of the LM algorithm.



Figure 2. The ANFIS structure.

the cost function is the RMSE between the real and estimated data. Figure 3 denotes the RMSE values of each iteration.

Models' evaluation

For attaining to the precision of the predictive model's, root mean squared error (RMSE), coefficient of determination (R^2), mean squared error (MSE), standard deviation (STD) and AARD are acceptable statistical criteria's. Following equations are the mathematical definition of mentioned criteria (Eqs. 17–21)⁸⁸:

$$\text{EMeanSqErr} = \frac{1}{N} \sum_{i=1}^{N} \left(D_i^{exp.} - D_i^{cal.} \right)^2$$
(17)

$$AARD_{Perc} = \frac{100}{N} \sum_{i=1}^{N} \frac{\left| D_i^{exp.} - D_i^{cal.} \right|}{D_i^{exp.}}$$
(18)



Figure 3. Performance of ANFIS during the training stage employing the Particle Swarm Optimization approach.

STDev =
$$\left(\frac{1}{N-1}\sum_{i=1}^{N} \left(D_i^{exp.} - D_i^{cal.}\right)^2\right)^{0.5}$$
 (19)

$$\operatorname{RootMeanSqErr} = \left(\frac{1}{N} \sum_{i=1}^{N} \left(D_i^{exp.} - D_i^{cal.}\right)^2\right)^{0.5}$$
(20)

$$R_{Coef}^{2} = 1 - \frac{\sum_{i=1}^{N} \left(D_{i}^{exp.} - D_{i}^{cal.} \right)^{2}}{\sum_{i=1}^{N} \left(D_{i}^{exp.} - \overline{D^{exp.}} \right)^{2}}$$
(21)

where *D*^{*exp*} denotes the mean experimental output value (density), *exp*. and cal. are an abbreviation of the experimental and calculated values, and N denotes the quantity of data points.

Results and discussion

The presented strategies were hired to determine the density of various refrigerant systems by considering pressure (P), temperature (T), the structural groups, and molecular mass (Mw) as input parameters. More information about these intelligent models is brought in Table 2. Using the MLP-ANN model, density values are better estimated than other model (ANFIS). This fact has been proven by the statistical analysis given in the evaluation section of the models.

To assess the efficacy of the models employed in this study, we employed various graphical methods. Figure 4 illustrates the plots comparing experimental and predicted density values for each model. Notably, the MLP-ANN model emerges as the most precise performer in density prediction, demonstrating a superior alignment between predicted and actual values. This graphical representation provides a clear visual insight into the predictive accuracy of the models, with the MLP-ANN model showcasing particularly commendable performance.

Regression plots between experimental versus predicted density values are shown in Fig. 5. The best fitting lines are obtained by using linear regression between the real and estimated values (Fig. 5a,b).

The relative deviations of the estimated and real data are shown in Fig. 6. It is apparent that the least deviation is related to the MLP-ANN model.

As a matter of fact, this is because of the accumulation of data points around the zero line. MRE values for MLP-ANN and ANFIS are 4.751 and 5.068, respectively. Additionally, for understanding the capability of these models in the prediction of the density values the statistical error analyses are done. These analyses are brought in Table 3.

Outlier detection

The precision of the models put forward is significantly impacted by the actual data employed in the segment dedicated to model development⁸⁹. In order to ensure the robustness of our models, locating and removing a set of data points exhibiting distinct behavior from the rest of the dataset, referred to as outliers, is regarded as a crucial step in enhancing the reliability of models⁹⁰. The leverage analysis is used in addition to standardized residuals implementation to determine potential outliers. By plotting standardized residuals (R) versus hat values (H), William's plot, outliers are detected. This multifaceted approach allows us to thoroughly assess the data points that might disproportionately influence the model outcomes. Equation (22) is used to calculate diagonal

ANFIS					
Category	Value/remark				
MF	Gaussian				
Number of MF parameters	480				
Number of clusters	10				
Amount of data utilized for training	2926				
Amount of data utilized for testing	975				
Optimization technique	PSO				
Population size	85				
Iteration	1000				
C ₁	1				
C ₂	2				
MLP-ANN					
Number of input neuron layers	23				
Number of hidden neuron layers	20				
Number of output neuron layers	1				
Activation function for hidden layer	Logsig				
Activation function for output layer	Purelin				
Data used for training	2926				
Data used for testing	975				
Maximum iterations	1500				

Table 2. Additional information about models trained for density estimation.







Scientific Reports | (2024) 14:2343 |





Components of the hat matrix, which are expressed as hat values and are used in the identification of feasible/ suitable regions.

$$H = X \left(X^T X \right)^{-1} X^T \tag{22}$$

Taking into account n as the number of data points and k as the number of input parameters, X represents a $(n \times k)$ matrix. This matrix is instrumental in evaluating the influence of each data point on the model. Warning leverage and cut-off values on the horizontal and vertical axis make a squared area called the feasible region. Below equation gives the warning leverage:

$$H^* = \frac{3(k+1)}{n}$$
(23)

This calculated warning leverage is pivotal in setting thresholds to identify potential outliers. Typically, the threshold value for R is deemed to be 3. Values beyond the boundaries of the feasible region are treated as outlier data. By meticulously considering these calculated parameters, our approach offers a comprehensive method for identifying and addressing outliers in the dataset. Figure 7 shows William's plot. According to this Figure, 27 and 17 points of the ANFIS and MLP-ANN approaches are placed outside of the feasible region.

While the presented models, MLP-ANN and ANFIS, exhibit remarkable predictive performance in estimating the densities of various refrigerant systems, it is essential to acknowledge certain limitations in the current study. One limitation is the reliance on a specific dataset comprising 3825 data points. Although the dataset is meticulously sourced from reputable literature, its scope may not cover all possible scenarios and variations in refrigerant properties. Additionally, the models are developed based on the selected input parameters, including molecular mass, pressure, structural groups, and temperature. The exclusion of certain relevant parameters or the consideration of additional factors could potentially impact the models' generalizability to a broader range of refrigerant systems. Furthermore, the outlier analysis conducted in this study identified specific data points that deviate from the predicted trends. While these outliers were carefully addressed, their presence underscores the sensitivity of the models to anomalous data. Future research endeavors could explore ways to enhance the robustness of the models by incorporating more diverse datasets, exploring additional input parameters, and





Model	Dataset	R ²	MRE (%)	MSE	RMSE	STD
	Train	0.966	5.233	0.7598	0.8717	0.7574
ANFIS	Test	0.969	4.575	0.6555	0.8096	0.7006
	Total	0.967	5.068	0.7337	0.8096	0.7436
	Train	0.976	4.671	0.5260	0.7252	0.5987
MLP-ANN	Test	0.969	4.990	0.6522	0.8076	0.6881
	Total	0.975	4.751	0.5575	0.8076	0.6222

 Table 3. Assessing the effectiveness of the proposed models through statistical analysis.

implementing advanced outlier detection techniques. Moreover, continuous refinement of the models through ongoing validation against experimental data will contribute to their reliability in practical applications.

Conclusions

In summary, the prediction of densities for 48 refrigerant systems was facilitated through the utilization of two intelligent models, incorporating crucial parameters such as molecular mass (Mw), pressure (P), structural groups, and temperature (T). The superior performance of the MLP-ANN approach over the ANFIS model, demonstrated by consistently lower error values, has been highlighted by the findings. Transcending its theoretical significance, the research harbors practical implications of particular relevance to scientists and engineers engaged in the design of economically viable low-temperature refrigeration cycles. The accuracy demonstrated by the MLP-ANN model establishes it as a valuable tool, one that effectively guides the optimization of system performance and steers the development of energy-efficient and cost-effective refrigeration technologies. These implications, indicative of the broader impact, underscore the study's contribution to advancements in the field of refrigeration system design. The findings not only deepen the understanding of the intricacies involved but also actively contribute to the evolution of methodologies, offering insights that shape the trajectory of progress in refrigeration technology.



Figure 7. Identification of potentially questionable datasets for various models: (a) ANFIS, (b) MLP-ANN.

.....

Data availability

The data that support the findings of this study are available from the corresponding author, Alireza Baghban, upon reasonable request.

Received: 4 November 2023; Accepted: 25 January 2024 Published online: 29 January 2024

References

- 1. Linteris, G. T., Bell, I. H. & McLinden, M. O. An empirical model for refrigerant flammability based on molecular structure and thermodynamics. *Int. J. Refrig* 104, 144–150 (2019).
- Lorbek, L. et al. Two-phase flow patterns in adiabatic refrigerant flow through capillary tubes. Int. J. Refrig. 115, 107–116 (2020).
 Rausch, M. H. et al. Wetting behavior and interfacial tension of a refrigerant oil in air and refrigerant atmospheres. Int. J. Refrig
- 107, 225–233 (2019).
- Sun, Z. et al. Options of low Global Warming Potential refrigerant group for a three-stage cascade refrigeration system. Int. J. Refrig 100, 471–483 (2019).
- 5. Mohebbi, A., Taheri, M. & Soltani, A. A neural network for predicting saturated liquid density using genetic algorithm for pure and mixed refrigerants. *Int. J. Refrig.* **31**(8), 1317–1327 (2008).
- 6. Bell, I. H. et al. The hunt for nonflammable refrigerant blends to replace R-134a. Int. J. Refrig 104, 484-495 (2019).
- 7. Rashed, M. *et al.* A novel thermodynamic design model of a new HFO refrigerant single phase vapor jet cooling system. *Int. J. Refrig* **110**, 153–167 (2020).
- 8. Casari, N., Pinelli, M. & Suman, A. Analysis of CoolProp library for the assessment of uncertainty propagation for refrigerant fluids in state diagrams and thermodynamic properties. *Int. J. Refrig* 107, 214–224 (2019).
- Lorentzen, G. The use of natural refrigerants: a complete solution to the CFC/HCFC predicament. Int. J. Refrig 18(3), 190–197 (1995).
- 10. Spauschus, H. HFC 134a as a substitute refrigerant for CFC 12. Int. J. Refrig 11(6), 389-392 (1988).
- 11. McLinden, M. O. Thermodynamic properties of CFC alternatives: a survey of the available data. *Int. J. Refrig* 13(3), 149–162 (1990).
- Ramsdell, J. W. *et al.* Cumulative dose response study comparing HFA-134a albuterol sulfate and conventional CFC albuterol in patients with asthma. *Ann. Allergy* 81(6), 593–599 (1998).
- 13. Calm, J. M. The next generation of refrigerants-Historical review, considerations, and outlook. *Int. J. Refrig.* **31**(7), 1123–1133 (2008).
- 14. Kruse, H. European research concerning CFC and HCFC substitution. Int. J. Refrig 17(3), 149-155 (1994).
- Tsai, W.-T. An overview of environmental hazards and exposure risk of hydrofluorocarbons (HFCs). Chemosphere 61(11), 1539– 1547 (2005).
- Ohta, H. et al. Liquid-phase thermodynamic properties of new refrigerants: Pentafluoroethyl methyl ether and heptafluoropropyl methyl ether. J. Chem. Eng. Data 46(5), 1020–1024 (2001).

- Kunkelmann, C. & Stephan, P. Numerical simulation of the transient heat transfer during nucleate boiling of refrigerant HFE-7100. Int. J. Refrig 33(7), 1221–1228 (2010).
- Moosavi, M., Sedghamiz, E. & Abareshi, M. Liquid density prediction of five different classes of refrigerant systems (HCFCs, HFCs, HFEs, PFAs and PFAAs) using the artificial neural network-group contribution method. *Int. J. Refrig* 48, 188–200 (2014).
- 19. Coquelet, C., El Abbadi, J. & Houriez, C. Prediction of thermodynamic properties of refrigerant fluids with a new three-parameter cubic equation of state. *Int. J. Refrig* 69, 418–436 (2016).
- Goharshadi, E. K. & Moosavi, F. Prediction of the volumetric and thermodynamic properties of some refrigerants using GMA equation of state. *Int. J. Refrig* 30(2), 377–383 (2007).
- Alam, M. S. & Jeong, J. H. Calculation of the thermodynamic properties of R448A and R449A in a saturation temperature range of 233.15 K to 343.15 K using molecular dynamics simulations. *Int. Commun. Heat Mass Transfer* 116, 104717 (2020).
- Maalem, Y., Tamene, Y. & Madani, H. Modeling of the vapor-liquid equilibria properties of binary mixtures for refrigeration machinery. Int. J. Thermodyn. 26(2), 22–33 (2023).
- 23. Sarfaraz, B. et al. Thermodynamic evaluation of mixed refrigerant selection in dual mixed refrigerant NG liquefaction process with respect to 3E's (Energy, Exergy, Economics). Energy 283, 128409 (2023).
- Nasrifar, K. & Moshfeghian, M. Evaluation of saturated liquid density prediction methods for pure refrigerants. *Fluid Phase Equilib.* 158, 437–445 (1999).
- Lugo, R. et al. An excess function method to model the thermophysical properties of one-phase secondary refrigerants. Int. J. Refrig 25(7), 916–923 (2002).
- 26. Scalabrin, G. et al. A predictive density model in a corresponding states format. Appl. Pure Mixed Refrig. 26(1), 35-50 (2003).
- Leila, M.-A., Javanmardi, M. & Boushehri, A. J. An analytical equation of state for some liquid refrigerants. Fluid Phase Equilib. 236(1-2), 237-240 (2005).
- 28. Song, Y. & Mason, E. Statistical-mechanical theory of a new analytical equation of state. J. Chem. Phys. 91(12), 7840-7853 (1989).
- 29. Goharshadi, E. K. & Moosavi, F. Prediction of thermodynamic properties of some hydrofluoroether refrigerants using a new equation of state. *Fluid Phase Equilib.* 238(1), 112–119 (2005).
- Goharshadi, E. K., Morsali, A. & Abbaspour, M. New regularities and an equation of state for liquids. *Fluid Phase Equilib.* 230(1–2), 170–175 (2005).
- 31. Kardani, M. N. *et al.* Phase behavior modeling of asphaltene precipitation utilizing RBF-ANN approach. *Pet. Sci. Technol.* **37**(16), 1861–1867 (2019).
- Daneshfar, R. et al. Estimating the heat capacity of non-newtonian ionanofluid systems using ANN, ANFIS, and SGB tree algorithms. Appl. Sci. 10(18), 6432 (2020).
- Nabipour, N. et al. Estimating biofuel density via a soft computing approach based on intermolecular interactions. Renew. Energy 152, 1086–1098 (2020).
- Vanani, M. B., Daneshfar, R. & Khodapanah, E. A novel MLP approach for estimating asphaltene content of crude oil. Pet. Sci. Technol. 37(22), 2238–2245 (2019).
- 35. Daneshfar, R. et al. A neural computing strategy to estimate dew-point pressure of gas condensate reservoirs. Pet. Sci. Technol. 38, 706–712 (2020).
- Qu, S., Zhao, L. & Xiong, Z. Cross-layer congestion control of wireless sensor networks based on fuzzy sliding mode control. Neural Comput. Appl. 32, 13505–13520 (2020).
- 37. Zhu, Q. Research on road traffic situation awareness system based on image big data. IEEE Intell. Syst. 35(1), 18-26 (2019).
- Kardani, M. N. et al. Group contribution methods for estimating CO2 absorption capacities of imidazolium and ammonium-based polyionic liquids. J. Clean. Prod. 203, 601–618 (2018).
- Kardani, N. et al. Estimation of bearing capacity of piles in cohesionless soil using optimised machine learning approaches. Geotechnical and Geological Engineering 38(2), 2271–2291 (2020).
- Ghanbari, A. *et al.* Neural computing approach for estimation of natural gas dew point temperature in glycol dehydration plant. *Int. J. Ambient Energy* 41(7), 775–782 (2020).
- Su, W., Zhao, L. & Deng, S. Group contribution methods in thermodynamic cycles: Physical properties estimation of pure working fluids. *Renew. Sustain. Energy Rev.* 79, 984–1001 (2017).
- Kubic, W. L. Jr. et al. Artificial neural network based group contribution method for estimating cetane and octane numbers of hydrocarbons and oxygenated organic compounds. Ind. Eng. Chem. Res. 56(42), 12236–12245 (2017).
- 43. Lazzús, J. A. ρ-T-P prediction for ionic liquids using neural networks. J. Taiwan Inst. Chem. Eng. 40(2), 213-232 (2009).
- 44. Lazzús, J. A. A group contribution method to predict the melting point of ionic liquids. Fluid Phase Equilb. 313, 1-6 (2012).
- Gharagheizi, F., Ilani-Kashkouli, P. & Mohammadi, A. H. Computation of normal melting temperature of ionic liquids using a group contribution method. *Fluid Phase Equilb.* 329, 1–7 (2012).
- Aguirre, C. L., Cisternas, L. A. & Valderrama, J. O. Melting-point estimation of ionic liquids by a group contribution method. *Int. J. Thermophys.* 33(1), 34–46 (2012).
- Lazzús, J. A. A group contribution method to predict the glass transition temperature of ionic liquids. *Thermochim. Acta* 528, 38–44 (2012).
- Gharagheizi, F. & Abbasi, R. A new neural network group contribution method for estimation of upper flash point of pure chemicals. Ind. Eng. Chem. Res. 49(24), 12685–12695 (2010).
- 49. Lazzús, J. A. A group contribution method to predict the thermal decomposition temperature of ionic liquids. *J. Mol. Liq.* 168, 87–93 (2012).
- Gharagheizi, F. et al. Development of a group contribution method for estimating the thermal decomposition temperature of ionic liquids. Fluid Phase Equilib. 355, 81–86 (2013).
- 51. Farahani, N. et al. A simple correlation for prediction of heat capacities of ionic liquids. Fluid Phase Equilib. 337, 73-82 (2013).
- 52. Gharagheizi, F. *et al.* Development of a group contribution method for determination of viscosity of ionic liquids at atmospheric pressure. *Chem. Eng. Sci.* **80**, 326–333 (2012).
- 53. Mirkhani, S. A. et al. Prediction of surface tension of ionic liquids by molecular approach. J. Mol. Liq. 179, 78-87 (2013).
- 54. Lazzús, J. A. Estimation of solid vapor pressures of pure compounds at different temperatures using a multilayer network with particle swarm algorithm. *Fluid Phase Equilib*. **289**(2), 176–184 (2010).
- 55. Gharagheizi, F. A new group contribution-based model for estimation of lower flammability limit of pure compounds. J. Hazard. Mater. 170(2–3), 595–604 (2009).
- Gharagheizi, F., Ilani-Kashkouli, P. & Mohammadi, A. H. Estimation of lower flammability limit temperature of chemical compounds using a corresponding state method. *Fuel* 103, 899–904 (2013).
- Gharagheizi, F. et al. Prediction of vaporization enthalpy of pure compounds using a group contribution-based method. Ind. Eng. Chem. Res. 50(10), 6503–6507 (2011).
- Gharagheizi, F. et al. A group contribution model for determining the sublimation enthalpy of organic compounds at the standard reference temperature of 298 K. Fluid Phase Equilib. 354, 265–285 (2013).
- Moosavi, M. & Soltani, N. Prediction of hydrocarbon densities using an artificial neural network–group contribution method up to high temperatures and pressures. *Thermochim. Acta* 556, 89–96 (2013).
- Sabzevari, S. & Moosavi, M. Density prediction of liquid alkali metals and their mixtures using an artificial neural network method over the whole liquid range. Fluid Phase Equilib. 361, 135–142 (2014).

- Moosavi, M. & Soltani, N. Prediction of the specific volume of polymeric systems using the artificial neural network-group contribution method. *Fluid Phase Equilib.* 356, 176–184 (2013).
- Chouai, A., Laugier, S. & Richon, D. Modeling of thermodynamic properties using neural networks: Application to refrigerants. *Fluid Phase Equilib.* 199(1-2), 53–62 (2002).
- 63. Rosenblatt, F. Principles of Neurodynamics. Perceptrons and the theory of brain mechanisms (Cornell Aeronautical Lab Inc, 1961).
- 64. Rumelhart, D. E., Hinton, G. E. & Williams, R. J. Learning internal representations by error propagation (California Univ San Diego La Jolla Inst for Cognitive Science, 1985).
- 65. Zadeh, L. A. Fuzzy sets. Inf. Control 8(3), 338-353 (1965).
- Safari, H. et al. Assessing the dynamic viscosity of Na-K-Ca-Cl-H2O aqueous solutions at high-pressure and high-temperature conditions. Ind. Eng. Chem. Res. 53(28), 11488–11500 (2014).
- Zarei, K. *et al.* Predicting the heats of combustion of polynitro arene, polynitro heteroarene, acyclic and cyclic nitramine, nitrate ester and nitroaliphatic compounds using bee algorithm and adaptive neuro-fuzzy inference system. *Chemom. Intell. Lab. Syst.* 128, 37–48 (2013).
- 68. Jang, J.-S.R. et al. Neuro-fuzzy and soft computing-a computational approach to learning and machine intelligence. **86**(3), 600–603 (1998).
- 69. Nikravesh, M., Zadeh, L. A. & Aminzadeh, F. Soft Computing and Intelligent Data Analysis in Oil Exploration Vol. 51 (Elsevier, 2003).
- 70. Qu, R. *et al.* A survey of search methodologies and automated system development for examination timetabling. *J. Sched.* **12**(1), 55–89 (2009).
- 71. Morgado, P. et al. Systems involving hydrogenated and fluorinated chains: volumetric properties of perfluoroalkanes and perfluoroalkylalkane surfactants. J. Phys. Chem. B 115(50), 15013–15023 (2011).
- 72. Morgado, P. et al. Viscosity of liquid perfluoroalkanes and perfluoroalkylalkane surfactants. J. Phys. Chem. B 115(29), 9130–9139 (2011).
- Yokoyama, C. & Takahashi, S. Saturated liquid densities of 2, 2-dichloro-1, 1, 1-trifluoroethane (HCFC-123), 1, 2-dichloro-1, 2, 2-trifluoroethane (HCFC-123a), 1, 1, 1, 2-tetrafluoroethane (HFC-134a) and 1, 1, 1-trifluoroethane (HFC-143a). *Fluid Phase Equilib.* 67, 227–240 (1991).
- Holcomb, C. & Van Poolen, L. Coexisting densities and vapor pressures for R 143 from 314 to 401 K with new critical point property estimates. *Fluid Phase Equilib.* 100, 223–239 (1994).
- Tamatsu, T., Sato, H. & Watanabe, K. Measurements of pressure-volume-temperature properties of 1, 1, 2, 2-tetrafluoroethane. J. Chem. Eng. Data 37(2), 216–219 (1992).
- Defibaugh, D. R. & Moldover, M. R. Compressed and saturated liquid densities for 18 halogenated organic compounds. J. Chem. Eng. Data 42(1), 160–168 (1997).
- Defibaugh, D. R. et al. Thermodynamic properties of HFC-338mccq, CF3CF2CF2CH2F, 1, 1, 1, 2, 2, 3, 3, 4-octafluorobutane. J. Chem. Eng. Data 42(3), 488–496 (1997).
- Ribeiro, A. et al. Relative permittivities of 1, 1, 1, 2, 3, 3, 3-Heptafluoropropane (HFC-227ea), 1, 1, 1, 2, 3, 3-hexafluoropropane (HFC-236ea), and 1, 1, 1, 3, 3-pentafluorobutane (HFC-365mfc) in the liquid phase. J. Chem. Eng. Data 52(5), 2041–2049 (2007).
- 79. Yin, J. & Wu, J. Compressed liquid densities of 1, 1, 1, 3, 3-pentafluoropropane (HFC-245fa) and 1, 1, 1, 3, 3, 3-hexafluoropropane (HFC-236fa). Fluid Phase Equilib. **307**(1), 1–5 (2011).
- Widiatmo, J., Tsuge, T. & Watanabe, K. Measurements of vapor pressures and PVT properties of pentafluoroethyl methyl ether and 1, 1, 1-trifluoroethane. J. Chem. Eng. Data 46(6), 1442–1447 (2001).
- Bobbo, S. *et al.* Compressed liquid densities and saturated liquid densities of HFC-365mfc. *Fluid Phase Equilib.* 222, 291–296 (2004).
- Kayukawa, Y. *et al.* Thermodynamic property measurements for trifluoromethyl methyl ether and pentafluoroethyl methyl ether. J. Chem. Eng. Data 48(5), 1141–1151 (2003).
- Piñeiro, M. et al. PpT measurements of nonafluorobutyl methyl ether and nonafluorobutyl ethyl ether between 283.15 and 323.15 K at pressures up to 40 MPa. Int. J. Thermophys. 24(5), 1265–1276 (2003).
- Piñeiro, M. M. *et al.* Determination of high-pressure liquid density for n-perfluorohexane and n-perfluorononane. *Fluid Phase Equilib.* 220(1), 125–134 (2004).
- 85. Bamgbade, B. et al. Experimental density and PC-SAFT modeling of Krytox*(perfluoropolyether) at pressures to 275 MPa and temperatures to 533 K. Fluid Phase Equilib. 332, 159–164 (2012).
- WebBook, N. Thermophysical Properties of Fluid Systems. http://webbook.nist.gov/chemistry/fluid/ (access on 18 January 2015) (2016).
- 87. Esmaeili, M., Ahmadi, M. & Kazemi, A. Kernel-based two-dimensional principal component analysis applied for parameterization in history matching. *J. Pet. Sci. Eng.* 191, 107134 (2020).
- Esmaeili, M., Ahmadi, M. & Kazemi, A. A generalized DEIM technique for model order reduction of porous media simulations in reservoir optimizations. J. Comput. Phys. 422, 109769 (2020).
- 89. Leroy, A. M. & Rousseeuw, R. J. Mathematical Statistics. In Robust regression and outlier detection (Wiley, 1987).
- Hosseinzadeh, M. & Hemmati-Sarapardeh, A. Toward a predictive model for estimating viscosity of ternary mixtures containing ionic liquids. J. Mol. Liq. 200, 340–348 (2014).

Author contributions

All authors contributed in writing, modeling, software, data collecting, and analysis.

Competing interests

The authors declare no competing interests.

Additional information

Correspondence and requests for materials should be addressed to H.L. or A.B.

Reprints and permissions information is available at www.nature.com/reprints.

Publisher's note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons licence, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons licence, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons licence and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this licence, visit http://creativecommons.org/licenses/by/4.0/.

© The Author(s) 2024