

Research

Mechanical properties of geopolymer concrete incorporating supplementary cementitious materials as binding agents

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Abstract

This research investigates the environmental impact of cement production by exploring eco-friendly geopolymer binders as alternatives. Geopolymer concrete, developed using silica and alumina-rich precursors such as pozzolanic materials, achieves high compressive strength, up to 43.6 N/mm² with a 16 M concentration and integrated steel fibers. Utilizing manual mixing and industrial by-products, the study pioneers cast-in-situ geopolymers with innovative curing techniques. The paper presents experimental results on the engineering properties of geopolymer concretes of 40 MPa, cured at 100 °C and 60 °C. The study systematically varies binder content, examining proportions of fly ash, GGBS, metakaolin, and silica fume, along with different mix ratios and molar concentrations. Key findings include increased compressive strength with higher NaOH concentration, peaking at 35.2 N/mm² and 34.22 N/mm² for 14 M mixes at 7 and 28 days, and 40.29 N/mm² for 16 M mixes at 7 days. Optimal results were observed at higher curing temperatures, especially with 14 M and 16 M compositions at 100 °C. The study recommends mechanized mixing for efficiency and calls for further investigation into the microstructure and chemistry of geopolymers to advance sustainable construction practices. This research represents a significant step towards eco-conscious building materials, reducing the environmental impact of the construction industry.

Keywords Compressive strength · Environmentally friendly · Fly ash · GGBS · Metakaolin · NaOH · Na₂SiO₃ · Silica fume · Sustainable

1 Introduction

The production of Ordinary Portland Cement (OPC) significantly contributes to greenhouse gas emissions, as it releases a substantial amount of carbon dioxide (CO₂) into the atmosphere. For every ton of OPC manufactured, one ton of CO₂ is emitted, making OPC production a major environmental concern [1]. As OPC ranks as the second most commonly used material globally, just after water, the development of sustainable cement substitutes is imperative. This can be achieved by combining natural resources like kaolin with the cementitious qualities of industrial byproducts such as fly ash and ground granulated blast furnace slag (GGBS) [2, 3]. Geopolymer concrete (GPC) emerges as a promising alternative to traditional cement due to its environmentally benign nature. Named for Davidovits, "geopolymer" refers to an alternate cementitious material that resembles ceramic. Unlike OPC, the polymerization process of geopolymers does not release greenhouse gases. Geopolymers are produced by mixing alkaline solutions with pozzolanic compounds or

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aluminosilicate sources [4]. Common materials like fly ash and GGBS, which are rich in silica and alumina, can replace cement in GPC, thereby reducing CO₂ emissions and enhancing mechanical strength and durability [5, 6].

Numerous studies have explored the properties and performance of GPC made from supplementary cementitious materials (SCMs): Palomo et al. [7] Investigated GPC using Class F fly ash and tested various ratios of alkaline activator to fly ash. The mixtures activated with sodium hydroxide and sodium silicate achieved compressive strengths exceeding 60 MPa after curing for 24 hours at 65 °C. Xu and van Deventer [8] found that an ideal ratio of 0.33 between alkali solution and alumina-silicate yielded a maximum compressive strength of 19 MPa after a 72-hour curing period at 35°C. Hardjito and Rangan [9] studied GPC composition by varying sodium hydroxide concentrations from 8M to 16M and adjusting the sodium hydroxide to sodium silicate ratio. Higher NaOH molarity and Na₂SiO₃ to NaOH ratio enhanced compressive strength, reaching 67 MPa after 24 hours of curing at 60 °C. Januarti Jaya Ekaputri et al. [10] examined the mechanical properties of GPC made from Jawa Power Paiton fly ash. The highest compressive strength of 48.59 MPa was achieved with a 10 M activator solution and a sodium silicate to sodium hydroxide ratio of 1.5. Tabassum et al. [11] found that different sodium hydroxide solution concentrations have distinct effects on geopolymer concrete mixtures. Rovnanik [12] investigated how curing temperature and duration affect metakaolin-based geopolymer, finding higher temperatures hasten dense structure formation and geopolymerization. Liew et al. [13] explored curing conditions' impact on metakaolin geopolymer pastes, emphasizing heat curing's necessity. Therefore, Geopolymer concrete requires heat curing. Low temperatures hinder geopolymerization, reducing mechanical properties. Geopolymer concrete's strength qualities were shown to be enhanced by an ideal NaOH content of 12 M. A maximum compressive strength of 40.21 MPa was reached by the concrete after 28 days [14]. Achieving sufficient strength also necessitates proper curing. Various studies have demonstrated that the compressive strength of fly ash-based geopolymer concrete (GPC) specimens cured in an oven is higher than that of those cured under ambient conditions [15]. The majority of research has demonstrated that geopolymer concrete is typically produced using sodium hydroxide solution molarities within the range of 8 M to 16 M. Optimal strengths, as indicated by various studies, is generally noted within the concentration range of 12 M to 16 M. [16–18].

Despite the extensive research on GPC, there is a need to consolidate and build upon the existing knowledge to develop more practical and cost-effective formulations. The core problem addressed in this study is to identify the optimal mix designs and curing conditions that maximize the strength of GPC. The research aims to fill gaps in understanding the interplay between different SCMs and alkaline activators, as well as the influence of curing regimes on the mechanical properties of GPC. The promise of GPC made by SCMs is found in both its enhanced compressive strength and environmental advantages. This research aims to improve the development of high-performance and sustainable GPC, thereby reducing carbon emissions in the construction sector, by drawing on and expanding upon the findings of earlier studies.

This experimental research focused on the process of making the geopolymer concrete, which is notable for its exceptional characteristics, including strong adhesion, uniformity during mixing, and high slump levels. Initially, the workability of geopolymer concrete decreases during manual mixing due to its high viscosity, primarily because the alkaline-to-geopolymer solids ratio drops according to the mix design. However, by adjusting the alkaline/binder (a/b) ratio to 0.42 and incorporating additional alkaline liquid, the workability of the concrete can be significantly improved. Moreover, a slight increase in the alkali solution content during mixing enhances the desired slump value across different mix proportions of geopolymer concrete. Introducing extra activator to the mix results in a higher concentration of alkali solution, reducing viscosity and cohesion during manual mixing, ultimately facilitating the achievement of necessary workability and strength, especially in higher grades of concrete. Previous studies suggest that sufficient strength also requires curing, as various investigations have demonstrated that the compressive strength of fly ash-based geopolymer concrete (GPC) specimens cured in an oven is higher than that of those cured under ambient conditions. This underscores the importance of controlled curing processes to optimize the mechanical properties of geopolymer concrete.

The primary objective of this study is to explore the compressive strength characteristics of geopolymer concrete by examining how adjustments in the composition of binders impact its performance. The study systematically investigates various combinations of fly ash and GGBS at different proportions, ranging from 10% to 20% for fly ash and 40% to 60% for GGBS, while keeping metakaolin constant at 10%. Silica fume is also included in proportions ranging from 18% to 28%, with the curing procedure involving oven temperatures between 60°C and 100°C. To comprehensively explore the effects on compressive strength, the study considers three different mix proportions and two different molar concentrations (14 and 16 M). By analyzing these different compositions, the researchers aim to gain insights into optimizing the properties of geopolymer concrete for enhanced compressive strength. The implications of this study are significant for advancing the field of geopolymer concrete technology, contributing to the development of more efficient and effective mix designs that exceed current performance standards and

Table 1 Composition of FA, GGBS, MK, and SF in terms of weight percentage

Materials	SiO ₂	Al ₂ O ₃	CaO	Mgo	Cl	Fe ₂ O ₃	Na ₂ O	SO ₃	LOI
Fly ash	30.53	30.53	–	1.4	0.029	30.53	0.46	0.64	0.96
Slag	34	17.4	33.34	8.76	0.029	1.96	–	0.024	0.46
Metakaolin	51	45	0.02	0.07	–	0.35	0.13	–	0.50
Silicafume	99.92	0.031	0.00	0.00	–	0.012	0.004	–	0.001

Table 2 Attributes of Sodium Silicate

Material	Composition of chemicals by weight (wt/wt)			Specific gravity
Na ₂ SiO ₃	Na ₂ O 16.25%	SiO ₂ 33%	H ₂ O 50.75	1.67

providing a deeper understanding of the interactions between different binder components for more sustainable and high-performance concrete solutions.

2 Experimental work

2.1 Materials used

This study utilized various types of binders as the primary alumina-silicon source materials for geopolymer concrete. These binders included class F fly ash (in accordance with IS 3812-2003 [19]) and ground granulated blast furnace slag (GGBS) (in accordance with IS 16714:2018 [20]), with specific gravities of 2.00 and 2.87, respectively, sourced from Suyog Elements Pvt Ltd in Baruch, Gujarat, India. Additionally, metakaolin from AJ Corporation in Mumbai, India, and silica fume from Astra Chemicals in Chennai, Tamil Nadu, India, with specific gravities of 2.6 and 2.64, respectively, were used. The chemical compositions of the fly ash, GGBS, metakaolin, and silica fume are detailed in Table 1. Coarse aggregate with a maximum size of 4.75 mm, a specific gravity of 2.53, and a 24-hour water absorption rate of 4.32% was employed. Fine aggregate, with a maximum size of 600 microns, a specific gravity of 2.63, a 24-hour water absorption rate of 0.6%, and a fineness modulus of 2, was also used. Both aggregates conform to the standards outlined in IS 383-2016 [21] and meet the criteria for zone II.

This investigation utilized commercially available sodium hydroxide (NaOH) pellets with a purity of 98%. Additionally, we utilized liquid sodium silicate (Na₂SiO₃), commonly known as Waterglass, which is easily obtainable in the market. The detailed chemical composition of sodium silicate is presented in Table 2. It is essential to note that the choice of both sodium hydroxide and sodium silicate was made based on their availability and well-established properties in relevant applications. Demineralized (DM) water is recommended for diluting sodium hydroxide (NaOH). The use of DM water in the mixing process eliminates mineral impurities, resulting in a cleaner and more effective final sodium hydroxide solution.

In this experimental study, a unique variety of fiber known for its outstanding resilience and lasting quality, namely brass-coated micro steel fiber, is employed. This fiber adheres to the standards outlined in ASTM A-820 Type-1 [22]. The dimensions of the fiber utilized in this research are 0.26 mm in diameter and 13 mm in length, exhibiting a straight configuration. Further specifications of the steel fiber are provided in Table 3.

The Conplast SP550, known for its exceptional water-reducing properties, is widely used, especially in micro silica concrete applications. It adheres to the IS: 9103:1999(2007) [23], and ASTM-C-494 Type 'G' [24] standards. Characterized by its brown liquid form and easy dispersal in water, it possesses a specific gravity of 1.24. This additive, containing Sulphonated Naphthalene Superplasticizer, plays a vital role in the current study. Its adaptability and adherence to industry norms make it a valuable choice for enhancing concrete performance. Here, fig. 1 presents photographs of the materials used in the experiment.

Table 3 The Characteristics of the Steel Fiber

Test	Specification ASTM A-820 type—1	Results
Carbon	1.00%	0.71%
Silicon	0.30%	0.20%
Manganese	1.20%	0.53%
Sulphur	0.03%	0.005%
Phosphorous	0.03%	0.021%
Tensile strength of wire	2750-3050 MPa	2815–3018



Fig. 1 Photographs of materials used in the experiment. **a** Fly ash. **b** GGBS. **c** Metakaolin. **d** Silica Fume. **e** Sodium Hydroxide Pellets. **f** Sodium Silicate Gel. **g** Brass Steel Fiber. **h** Superplasticizer. **i** Coarse Aggregates. **j** Fine Aggregate

2.2 Alkaline liquid

Alkaline liquids are typically formulated by blending a solution of sodium hydroxide with sodium silicate at room temperature. As these two solutions combine, they undergo a reaction known as polymerization, resulting in the release of a substantial amount of heat. It is advisable to allow the mixture to stand for approximately 24 hours. This resting period ensures that the alkaline liquid, serving as a binding agent, is fully prepared [25].

2.3 Preparation of alkaline liquid

2.3.1 Sodium hydroxide

Two separate concentrations of sodium hydroxide pellets are dissolved in water, specifically at 14 and 16 molars. It's strongly recommended to prepare the sodium hydroxide solution at least 24 hours before use. Moreover, if this preparation exceeds a 36-hour duration, the solution tends to transition into a semi-solid state. Thus, it's crucial to utilize the prepared solution within this prescribed time limit [26–28].

2.3.2 Molarity calculation

Consider two concentrations of sodium hydroxide (NaOH) solution: 14 and 16 mol per liter. This translates to 560 g (14×40) and 640 g (16×40) of NaOH solids per liter of water for each molarity, with 40 representing NaOH's molecular weight. It's important to emphasize that water remains the primary constituent in both alkaline solutions. Notably, the NaOH concentration directly influences the quantity of solid NaOH in each solution, with the 16 Molar solution containing a greater mass compared to the 14 Molar solution.

The correct method involves adding 560 g of sodium hydroxide solids gradually to a specified amount of water, such as 500 ml. After ensuring complete dissolution, the volume of Sodium Hydroxide Solution (SHS) is measured to confirm it reaches one liter. If the solution falls short of this volume, additional water is added to reach exactly one liter. Conversely, if the SHS exceeds one liter, 560 g of sodium hydroxide solids are added to a smaller volume of water than previously used, and the process is repeated [29].

2.4 Mixing, casting and curing

A framework and code of practice exist for conventional concrete mixes, but not for geopolymer concrete. Thus, creating a geopolymer concrete mix must be based on conventional mix design concepts. Various mix proportioning methods are used to achieve the necessary concrete strength, considering the task, material properties, availability, field conditions, and requirements for durability and workability. Rangan [25] proposed a fly ash-based technique for geopolymer concrete, while Anuradha et al. [26] provided updated guidelines based on the Indian standard code. In this experimental geopolymer concrete mix was created utilizing the mix design technique specified in IS 10262-2019 [30]. In this study, the geopolymer mixing procedure encompassed five sequential stages. Initially, an alkaline solution was prepared by dissolving sodium hydroxide (NaOH) solids in demineralized water to achieve the desired concentration. This solution was then blended with sodium silicate solution before a 24-hour period prior to casting. Secondly, coarse and fine aggregates were meticulously mixed with the binder manually to create a well-blended dry mixture. The third step involved combining the prepared alkaline solution with a superplasticizer. In the fourth step, the liquid component was gradually incorporated into the dry mix, and the mixing process persisted for approximately 10-15 minutes until a uniform concrete mix was obtained. Finally, steel fibers were introduced and continuously mixed for 3-5 minutes. The workability of the freshly mixed concrete was assessed using the slump cone test, similar to that used for cement concrete. After the flow test, the fresh concrete was placed in the mold according to IS 1199-1959 [31]. Then the concrete was promptly poured into 150mm x 150mm x 150mm molds, followed by compaction using a tampering rod with approximately 45-50 blows to ensure proper compaction. After a 24-hour curing period, the specimens were demolded and subjected to further curing in a hot air oven for an additional 24 hours at various temperatures, alongside being maintained at ambient temperature (25–27°C) until testing. The curing temperatures differ according to the raw material utilized for fly ash-based geopolymer concrete, curing occurs

at 60°C [26, 32]. Furthermore, to determine the compressive strengths, geopolymer concrete cubes were tested according to the guidelines specified in IS 516-1959 [33].

Table 4 provides a detailed breakdown of the compositions of the binder mixes, expressed as percentages, and specifies the molar concentrations for three distinct mix designations: Geopolymer -1 (GP1), Geopolymer -2 (GP2), and Geopolymer -3 (GP3). Each mix designation is characterized by concentrations of 14 M and 16 M. In the experimental phase, a total of 72 cubes were cast, with twelve specimens allocated to each mix designation. These cubes underwent curing at various temperatures, as depicted in fig 2. Following an initial 24-hour curing period in an oven, the casted cube specimens were transferred to a laboratory environment and left at room temperature until the conclusion of the testing day, in accordance with the procedure outlined in Fig 3. Subsequently, compression tests were conducted on the cube specimens of geopolymer concrete using a testing machine with a capacity of 2000 KN shown in fig 4. Notably, the results reported represent the average strength derived from three cube measurements. Furthermore, Table 5 presents a summary of the proportions of binder mix designs, detailing the quantities in kilograms per cubic meter (Kg/

Table 4 The standards for blending ratios to enhance compressive strength

MIX	(% of geopolymer precursors)				Super plasticizer	Steel Fiber	Na ₂ SiO ₃ / NaOH	Alkaline tobinder Ratio
	Fly ash	GGBS	Metakaolin	Silica Fume				
GP1	20%	40%	10%	28%	1.2%	2%	1	0.42
GP2	10%	60%	10%	18%	1.2%	2%	1	0.42
GP3	15%	53%	10%	20%	1.2%	2%	1	0.42

Fig. 2 Specimens oven-cured at (a) 60 and (b) 100 °C for a 24-h duration

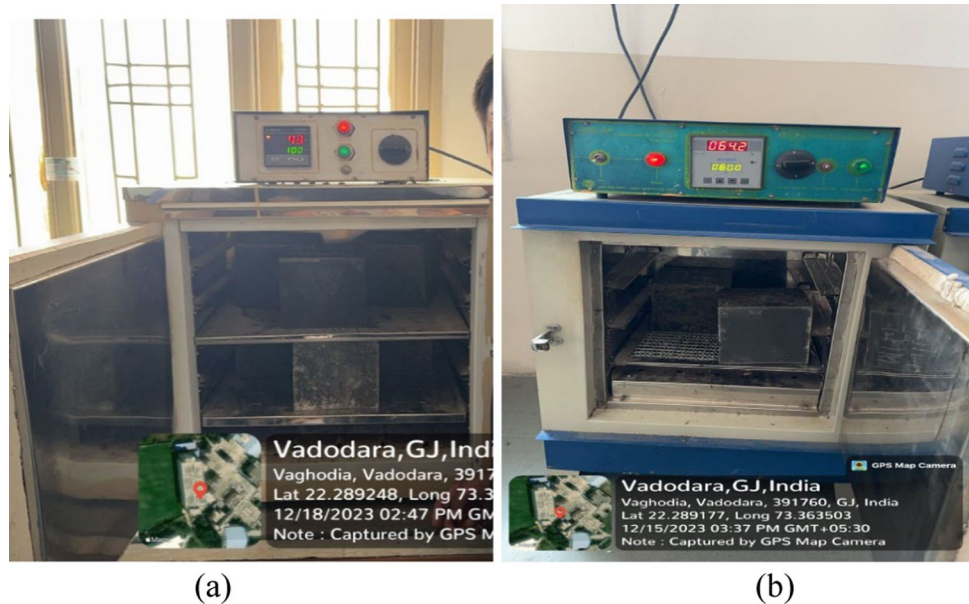


Fig. 3 Cube cured under ambient conditions



Fig. 4 Cube specimen compression testing (a) under applied loading conditions (b) during cube failure conditions

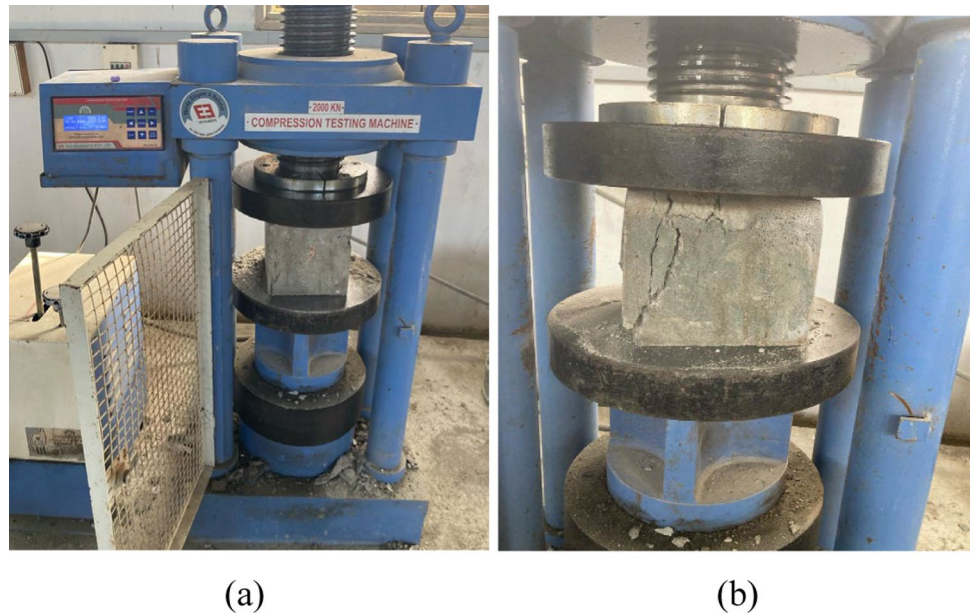


Table 5 The mix composition as measured in kilograms per cubic meter (kg/m^3)

Parameters	ASM Weight in (Kg/m^3)			
	Fly ash	GGBS	Metakaolin	Silica fume
M1 (mix)	153.409	306.81	76.70	214.722
M2 (mix)	76.70	460.225	76.70	130.068
M3 (mix)	115.056	406.533	76.70	153.409
Fine aggregates	530.02			
Coarse aggregates	725.38			
Super plasticizer	9.205			
Steel Fibers	15.34			
Sodium hydroxide (NaOH)	55.01			
Sodium silicate (Na_2SiO_3)	12381.28			
Water	213			

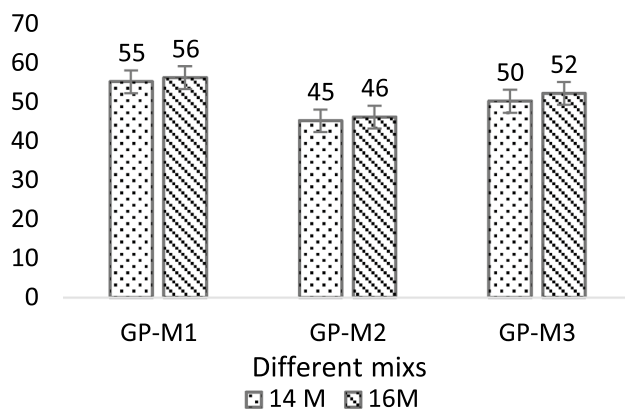
m^3). This comprehensive overview aids in understanding the precise formulation of the geopolymer concrete and its experimental parameters.

3 Results and discussion

3.1 Workability

The changes in workability for the different mixes are shown in Fig. 5. The data reveal that the slump value decreased with an increase in GGBS content, consistent with previous studies [34]. Mix no. GP1 exhibited the highest workability with a slump value of 55 mm. Mixes GP2 and GP3 showed the smallest variation in workability values for both molars of GPC. There was no additional water added to the alkaline solution during the GPC casting process. However, adding more GGBFS affected the workability of the trial mixes [16]. Das et al. [35] has identified that the low workability is due to irregular shapes of fly ash and angular GGBFS. These shapes lead to significant particle interlocking, thereby decreasing workability. Greater amounts of CaO in GGBS accelerate hydration and the formation of C-A-S-H/C-S-H gel, leading to quicker setting times and decreased workability [36–38]. When metakaolin was added or replaced by other materials, the slump value decreased due to its plate-like shape, which required more water or superplasticizers. The increased surface area and high fineness of slag also contributed to this need for additional water or superplasticizers to maintain

Fig. 5 Workability of various mixtures



workability [39]. With higher silica fume content, the slump in geopolymer concrete decreased. Their high viscosity results in low workability, making them more cohesive and viscous than OPC concrete [40, 41].

3.2 Compressive strength

Following a thorough examination conducted over a period of 7 days, it was discerned that the highest levels of strength were attained under specific conditions. For instance, in the case of the 14 M concentration, the GP3 mixture displayed a remarkable strength of 35.2 N/mm² after 7 days, while the GP2 mixture exhibited a slightly lower but still notable strength of 34.22 N/mm² after 28 days. The findings suggest that increasing the GGBS content in the GPC mixtures resulted in higher compressive strength. This improvement is due to the significant production of calcium silicate hydrate gel [42]. Similarly, for the 16 M concentration, the GP2 mixture demonstrated a notable compressive strength of approximately 40.29 N/mm² after 7 days, while the GP1 mixture showcased an even higher strength of 43.6 N/mm² after 28 days. Okoye et al., conducted research on the impact of silica fume on the compressive strength of geopolymer concrete. The study revealed that the strength consistently increased with the addition of silica fume, achieving the maximum improvement at a 40% addition, which was the highest amount tested in the experiment [43]. It is crucial to note that all these mixtures underwent initial curing at a temperature of 100°C, indicating a standardized initial condition. This controlled environment ensures consistency in the experimental setup, allowing for accurate comparison and analysis of the results. The compressive strength results for 14 molar mixtures at 7 and 28 days are shown in Fig. 6, illustrating the early-stage performance of each mixture. Meanwhile, Fig. 7 present the compressive strength results for 16 molar mixtures at 7 and 28 days, providing insights into the longer-term performance of the mixes. By encompassing data from the three provided mixes (GP1, GP2, and GP3) across two different molar concentrations (14M and 16M), these figures provide a comprehensive

Fig. 6 14 Molar concrete compressive strength at 7 and 28 days for GP1, GP2, GP3 Mix

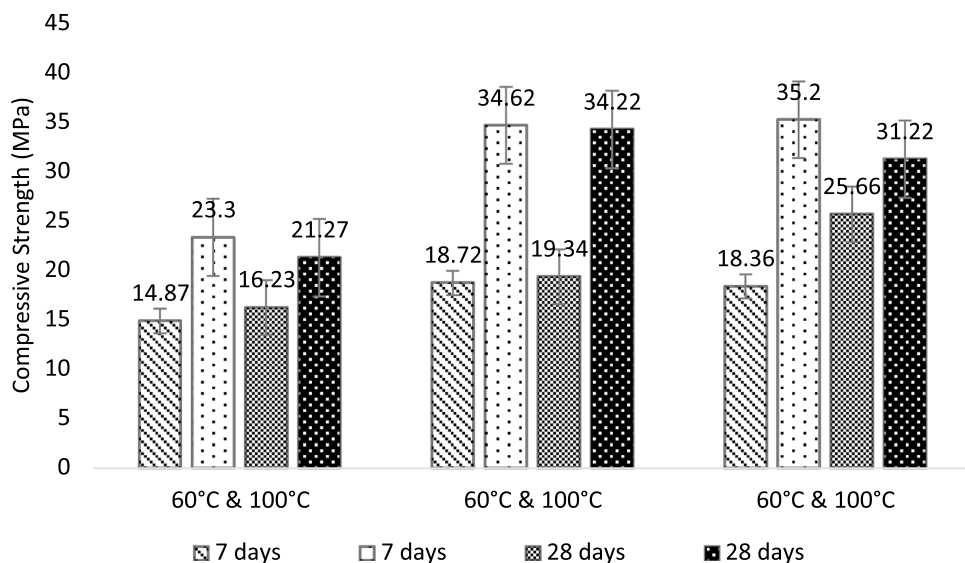
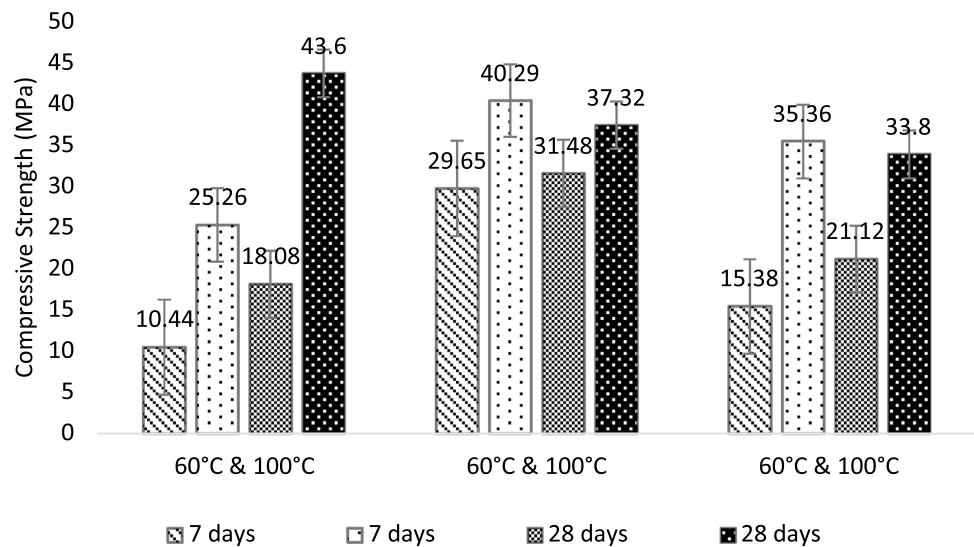


Fig. 7 16 Molar concrete compressive strength at 7 and 28 days for GP1, GP2, GP3 Mix



overview of the experimental findings. This comprehensive approach facilitates a deeper understanding of how varying factors such as mix composition and molar concentration impact the compressive strength of the materials over time.

When the GP2 blend with a molarity of 16 was subjected to a curing temperature of 60°C, it exhibited exceptional performance characteristics. Specifically, it achieved an impressive compressive strength of 29.65 N/mm² within a remarkably short period of just 7 days. This rapid development of strength highlights the effectiveness of the curing process at this temperature. Furthermore, even after the initial 7-day period, the GP2 mixture, still maintained at a molarity of 16, continued to display its strength. Over the course of a 28-day curing period, it further improved its compressive strength, eventually reaching a peak value of 31.48 N/mm². This sustained enhancement in strength suggests that the curing process not only initiates rapid development but also facilitates continued improvement in the material's mechanical properties over time.

Conversely, the GP1 mix, despite sharing the same molarity (16 Molar), displayed notably lower performance, registering a compressive strength of only 10.44 N/mm² within the initial 7 days. This value notably lagged behind the corresponding results for both different molarities during the same period. Moreover, for the GP1 mix with a slightly lower molarity of 14 M, the compressive strength recorded after 28 days was similarly inferior, reaching only 16.23 N/mm². In summary, the GP2 blend, particularly with a molarity of 16, exhibited superior compressive strength characteristics compared to the GP1 mix under similar conditions, showcasing its potential for robust performance in concrete applications.

Figures 5, 6 illustrate the changes in compressive strength for 14 M and 16 M NaOH concentration solutions. Both figures indicate that increasing the molarity of the NaOH solution leads to a slight rise in compressive strength. Notably, the 16 M mixture, containing 40% GGBFS and 28% silica fume, shows higher compressive strength compared to the 14 M mixture. While the compressive strength of 16 M NaOH solution mixtures increase steadily, the 14 M NaOH mixtures exhibit erratic increments. The higher molarity may facilitate more Al atoms receiving electrons from Na atoms, leading to increased sialate bond formation [44, 45].

4 Conclusions

4.1 Findings

This research focused on developing eco-friendly geopolymer concrete (GPC) using fly ash, GGBS, metakaolin, and silica fume. The study identified optimal mix designs and curing conditions to maximize the compressive strength of GPC. Specifically, the highest compressive strengths were achieved under certain conditions: for 14 M NaOH, the GP3 mixture reached 35.2 N/mm² after 7 days, and the GP2 mixture reached 34.22 N/mm² after 28 days. For 16 M NaOH, the GP2 mixture achieved 40.29 N/mm² after 7 days, and the GP1 mixture reached 43.6 N/mm² after 28 days. The GP2 blend, with 16 M NaOH and a 60°C curing temperature, achieved 29.65 N/mm² in 7 days and 31.48 N/mm² in 28 days, while the GP1 mix performed poorly, reaching only 10.44 N/mm² in 7 days and 16.23 N/mm² in 28 days. Initial

curing at 100 °C was essential for consistency. The research demonstrated that GPC offers significant environmental benefits by reducing carbon emissions and supports sustainable construction practices.

4.2 Research Limitations

The study was conducted under controlled laboratory conditions, which may not fully capture real-world applications. It primarily addressed mechanical properties, environmental impacts, and practical scalability. The specific focus on high molarity NaOH solutions and selected binder contents may not represent the full spectrum of possible formulations.

4.3 Recommendations for Future Research

Future studies should focus on consolidating and expanding existing knowledge to develop more practical and cost-effective GPC formulations. Investigations should include a broader range of SCMs, mix proportions, and curing methods, such as ambient curing. Further research should explore the long-term durability and environmental benefits of GPC, optimizing formulations for high-performance and sustainable construction applications. Understanding the microstructural and chemical interactions in GPC will further enhance its development and practical use.

4.4 Implications

Despite the extensive research on GPC, there is a need to consolidate and build upon the existing knowledge to develop more practical and cost-effective formulations. The core problem addressed in this study is to identify the optimal mix designs and curing conditions that maximize the strength of GPC. The research aims to fill gaps in understanding the interplay between different SCMs and alkaline activators, as well as the influence of curing regimes on the mechanical properties of GPC. This study highlights the potential of GPC to significantly reduce carbon emissions in the construction sector, offering a viable alternative to traditional Portland cement. The findings underscore the promise of GPC, particularly in its enhanced compressive strength and environmental advantages, thus contributing to the development of high-performance, sustainable GPC. This work lays the foundation for further advancements in eco-friendly building materials, promoting a more sustainable approach in the construction industry.

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Data availability The data supporting this research is enclosed within this paper and does not need to be referred from an external source.

Declarations

Competing interests The authors declare no competing interests.

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