ARTICLE

Synthesis, antitumor, antibacterial and urease inhibitory evaluation of new piperazinyl *N***‑4 carbamoyl functionalized ciprofoxacin derivatives**

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Abstract

Background Quinolones are well known antibacterial chemotherapeutics. Furthermore, they were reported for other activities such as anticancer and urease inhibitory potential. Modifcation at C7 of quinolones can direct these compounds preferentially toward target molecules.

Methods Diferent derivatives of ciprofoxacin by functionalization at the piperazinyl *N-*4 position with *arylidenehydrazinecarbonyl* and *saturated heterocyclic-carbonyl* moieties have been synthesized and characterized using diferent spectral and analytical techniques. The synthesized compounds were evaluated for anticancer, antibacterial, and urease inhibitory activities.

Results Among the synthesized compounds derivatives **3f** and **3g** experienced a potent antiproliferative activity against the breast cancer BT-549 cell line, recording growth percentages of 28.68% and 6.18%, respectively. Additionally, compound **3g** revealed a remarkable antitumor potential toward the colon cancer HCT-116 cells (growth percentage 14.76%). Activity of compounds **3f** and **3g** against BT-549 cells was comparable to doxorubicin $IC_{50} = 1.84$, 9.83, and 1.29 μ M, respectively). Test compounds were less active than their parent drug, ciprofoxacin toward *Klebsiella pneumoniae* and *Proteus mirabilis*. However, derivative **4a** showed activity better than chloramphenicol against *Klebsiella pneumoniae* (MIC=100.64 and 217.08 µM, respectively). Meanwhile, many of the synthesized compounds revealed a urease inhibitory activity greater than their parent. Compound 3i was the most potent urease inhibitor with IC_{50} of 58.92 μ M, greater than ciprofloxacin and standard inhibitor, thiourea (IC₅₀=94.32 and 78.89 μ M, respectively).

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Conclusion This study provided promising derivatives as lead compounds for development of anticancer agents against breast and colon cancers, and others for optimization of urease inhibitors.

Graphical abstract

Keywords Ciprofoxacin · Carbamoyl · Anticancer · Antibacterial · Urease inhibitors · Molecular docking

Introduction

Among bioactive molecules, fuoroquinolones represent an important drug class that made a revolution in the feld of antibacterial chemotherapy. Thanks to their broad spectrum of activity and favorable pharmacokinetic profle, a wide range of contagious diseases could be managed using fuoroquinolone drugs including Gram-positive, Gram-negative, anaerobic, atypical, and mycobacterial infections [[1,](#page-14-0) [2\]](#page-14-1). In addition to clinically used drugs that constitute diferent generations, several new fuoroquinolone derivatives with up-and-coming activity are now under clinical investigation [\[3,](#page-14-2) [4](#page-14-3)]. Other derivatives with potent antibacterial activity were also described in diferent researches, for example ciprofoxacin hydrazones such as compound **I** experienced anti-mycobacterial activity comparable to or more than isoniazid [[5\]](#page-14-4).

Quinolones were additionally reported for anticancer activity with the frst member in clinical use, voreloxin, in addition to quarfoxin, which failed in phase III of clinical trials due to some pharmacokinetic problems [\[6](#page-14-5)[–8](#page-14-6)]. Numerous other candidates with promising antitumor potential were reported in many literature studies [[9\]](#page-14-7). Structural features for anticancer fuoroquinolones have been well determined, with the most important sites for modifcation at C3 carboxylic acid functionality and at position 7 of the ring core. Substitution at C7 exhibited a major infuence on fuoroquinolone activity including preferential binding to bacterial type II topoisomerases, gyrase or topoisomerase IV, as well as physicochemical properties and hence, distribution and access to target molecules [\[10](#page-14-8), [11\]](#page-15-0). Introducing aromatic or heteroaromatic moieties at position 7 can also increase the affinity of quinolone molecules toward mammalian enzymes and potentiate the antiproliferative tendency [[12,](#page-15-1) [13\]](#page-15-2). Among the prepared compounds in this direction, hydrazone derivatives of ciprofloxacin and norfloxacin such as compounds **II (QNT11)**, **III** and I**V** aforded potent antitumor activity $[14–16]$ $[14–16]$ $[14–16]$ $[14–16]$.

On the other hand, urease had emerged as an attractive molecular target for diferent health benefts, with the advantage that it is not expressed by human host cells [[17,](#page-15-5) [18\]](#page-15-6). This enzyme is produced by numerous microorganisms and represents a crucial factor for their growth and survival [\[19](#page-15-7)]. Urease (or urea amidohydrolase EC 3.5.1.5) is a nickel containing metalloenzyme that acts by converting urea into ammonia and carbamates, which in turn hydrolyzed to ammonia and carbonic acid with a net result of elevating pH of the medium [[20–](#page-15-8)[22\]](#page-15-9). Such catalytic activity plays a vital role for pathogenesis, growth, and maintenance of diferent microorganisms like *Helicobacter pylori* and *Proteus mirabilis* [\[23\]](#page-15-10). *Helicobacter pylori* urgently need urease to alkalinize surrounding medium in the gastric juice to achieve colonization. Moreover, urease was evidenced for prognosis and long lasting of *H. pylori* infection as well as its consequent peptic ulcer and even gastric carcinoma [[24,](#page-15-11) [25](#page-15-12)]. Urease also rises pH of urine to a favorable level for growth of some pathogens and represents a leading cause for renal injury and urinary calculi during *P. mirabilis* and other urease positive bacterial infections [[26\]](#page-15-13). Furthermore, urease was found to be a virulence factor in hepatic coma, encephalopathy in addition to diferent chronic diseases such as atherosclerosis and rheumatoid arthritis [\[27,](#page-15-14) [28\]](#page-15-15). Hence, hindrance of urease activity can be useful for cure of diferent infectious diseases and other health conditions. Targeting both bacteria and urease in the same time was proposed to be a good strategy for treatment of some GIT diseases [\[29](#page-15-16)]. Some fuoroquinolone drugs exhibited anti-urease activity such as norfloxacin [[30\]](#page-15-17), ciprofloxacin, sparafloxacin [\[31](#page-15-18)], metal complexes of sparafoxacin [[32\]](#page-15-19), ciprofoxacin capped nanoparticles [\[33](#page-15-20)] in addition to different ciprofloxacin analogs like compounds **V** and **VI**. Compound **V** experienced anti-urease along with anti *P. mirabilis* activities more than acetohydroxamic acid (AHA) and *N*-acetylciprofloxacin, respectively [[34\]](#page-15-21). Meanwhile, compound **VI** showed urease inhibitory activity more than the parent drug, ciprofoxacin and thiourea as a standard inhibitor [[35\]](#page-15-22).

Based on the above fndings, the present study aimed at the synthesis, characterization, and investigation of diferent C7 modified ciprofloxacin analogs, bearing urea like scaffold (**Series I and II**) that may be able to bind to urease and hence, can target both urease enzyme and urease producing bacteria. Also, compounds of **Series I** contain aryl moieties that may enhance anticancer activity. The synthesized compounds were tested for both anticancer and antibacterial potential. Moreover, urease inhibitory activity and molecular docking onto urease protein (PDB: 1E9Y) [[36\]](#page-15-23) have been carried out.

Materials and methods

Chemical synthesis and characterization

All chemicals used for preparation of the target compounds were of the commercially available analytical grade quality. Reaction progress was monitored using TLC (Kieselgel 60 G F254 precoated plates, E. Merck, Dermastadt, Germany). Spots were detected by exposure to UV lamp (Spectroline CM-10, Seattle, USA) at *λ* 254 and 365 nm. Melting points were determined on Stuart SMP1 electrothermal melting point apparatus (Stuart Scientifc, Stafordshire, UK) and were uncorrected. IR Spectra were recorded with Bruker Alpha Platinum-ATR FTIR spectrometer (Bruker Corporation, Germany), applying attenuated total refection technique (ATR) and were expressed as cm⁻¹. ¹H-NMR And ¹³C-NMR spectra were recorded on BRUKER Avance III400 MHz spectrophotometer (Bruker AG, Switzerland) at 400 MHz for ${}^{1}H$ and 100 MHz fo 13 C. TMS was used as an internal standard and $CDCl₃$ or $DMSO-d₆$ as a solvent. Chemical shift (*δ*) values are expressed in parts per million (ppm) and

coupling constants (*J*) in Hertz (Hz). Signals are designated as follows: *s* singlet, *d* doublet, *t* triplet, *q* quartet, *m* multiplet, *brs* broad singlet. Mass spectroscopy was performed using DI-50 unit of Shimadzu GC/MS-QP 5050A apparatus. Elemental analyses were carried out at the regional center for mycology and biotechnology, Al-Azhar University, Cairo, Egypt.

Synthesis of intermediate 1 (1‑cyclo‑ propyl‑6‑fuoro‑7‑{4‑[(4‑nitrophenoxy)carbonyl] piperazin‑1‑yl}‑4‑oxo‑1,4‑dihydroquinoline‑3‑car‑ boxylic acid)

A mixture of ciprofoxacin (0.994 g, 0.003 mol), 4-nitrophenyl chloroformate (0.605 g, 0.003 mol) and pyridine (0.356 g, 0.0045 mol) in acetonitrile (10 mL) was heated at refux for 5 h, then cooled to room temperature and poured to 0.1 M HCl. The formed precipitate was collected, washed successively with water and dried [[41\]](#page-15-24).

Yield 0.968 g (65%); canary yellow powder, mp: 290–92 °C; ¹ H-NMR (400 MHz, CDCl3) *δ* 14.95 (s, 1H, COOH), 8.82 (s, 1H, C2–H), 8.32 (d, *J*=8.8 Hz, 2H, Ar–H), 8.10 (d, *J*=12.0 Hz, 1H, C5–H), 7.47–7.35 (m, 3H, C8-H & 2Ar–H), 3.99–3.86 (m, 4H, piperazine-4H), 3.64–3.52 (m, 1H, cyclopropyl-H), 3.50–3.38 (m, 4H, piperazine-4H), 1.34–1.14 (m, 4H, cyclopropyl-H); MS *m/z* calcd for $C_{24}H_{21}FN_{4}O_{7}$ [M⁺]: 496.14, found: 496.57; Anal. calcd for $C_{24}H_{21}FN_{4}O_{7}$: C, 58.06; H, 4.26; N, 11.29; found: C, 58.34; H, 4.35; N, 11.43.

Synthesis of intermediate 2 (1‑cyclopro‑ pyl‑6‑fuoro‑7‑[4‑(hydrazinecarbonyl)pipera‑ zin‑1‑yl]‑4‑oxo‑1,4‑dihydroquinoline‑3‑carboxylic acid)

A mixture of ciprofloxacin carbamate intermediate **1** (0.992 g, 0.002 mol) and hydrazine hydrate (0.200 g, 0.004 mol) in ethanol (10 mL) was heated at refux for 5 h. After the reaction was completed, the mixture was cooled to room temperature and the formed precipitate was fltered off, washed with ethanol and dried $[42]$ $[42]$.

Yield 0.623 g (80%); off white powder, mp: 248–50 °C; ¹H-NMR (400 MHz, DMSO-*d*₆) *δ* 15.25 (brs, 1H, COOH), 8.68 (s, 1H, C8–H), 8.09 (s, 1H, NH), 7.93 (d, *J*=12.8 Hz, 1H, C5–H), 7.58 (d, *J*=7.6 Hz, 1H, C8–H), 3.87–3.79 (m, 1H, cyclopropyl-H), 3.58–3.46 (m, 4H, piperazine-4H), 3.36–3.24 (m, 4H, piperazine-4H), 1.37–1.29 (m, 2H, cyclopropyl-H), 1.24 (s, 2H, NH₂), 1.23–1.14 (m, 2H, cyclopropyl-H); MS m/z calcd for $C_{18}H_{20}FN_5O_4$ [M⁺]: 389.15, found: 389.07; Anal. calcd for C₁₈H₂₀FN₅O₄: C, 55.52; H, 5.18; N, 17.99; found: C, 55.68; H, 5.29; N, 18.27.

General procedure for synthesis of target com‑ pounds 3a–i

A mixture of hydrazide intermediate **2** (0.001 mol) and the appropriate aldehyde (0.003 mol) in methanol (15 mL) containing few drops of glacial acetic acid was heated at refux for 2 h then cooled to room temperature. The found precipi-tate was filtered off, washed with methanol and dried [[43](#page-15-26)].

(*E/Z***)‑7‑[4‑(2‑Benzylidenehydrazinecarbonyl)pipera‑ zin‑1‑yl]‑1‑cyclopropyl‑6‑fuoro‑4‑oxo‑1,4‑dihydro‑ quinoline‑3‑carboxylic acid 3a**

Yield 0.435 g (90%); off white powder, mp: 242–44 °C; IR (KBr) 3332 (NH str), 3097 (aromatic C–H str), 2957 (aliphatic C–H str), 1727 (carboxylic C=O str), 1668 (amidic C=O str) 1626 (quinolone C=O str); ¹H-NMR (400 MHz, DMSO-*d*6) *δ* 15.03 (brs, 1H, COOH), 10.42 (s, 1H, NH), 8.67 (s, 1H, C2–H), 8.17 (s, 1H, =CH), 7.93 (d, *J*=13.2 Hz, 1H, C5–H), 7.68–7.57 (m, 3H, C8–H & 2Ar–H), 7.47–7.34 (m, 3H, Ar–H), 3.90–3.79 (m, 1H, cyclopropyl-H), 3.76–3.67 (m, 4H, piperazinyl-H), 3.45–3.35 (m, 4H, piperazinyl-H), 1.38–1.30 (m, 2H, cyclopropyl-H), 1.25–1.17 (m, 2H, cyclopropyl-H); 13C-NMR (100 MHz, DMSO-*d*6) *δ* 176.86, 166.29, 154.98, 153.47 (d, *J*_{C–F}=249.0 Hz), 148.43, 145.49 (d, *J*_{C–F}=10.1 Hz), 143.39, 139.66, 137.92, 135.46, 129.61, 129.15, 126.89, 119.34, 111.49 (d, *J*_{C-F}=23.0 Hz), 107.21, 49.82, 44.21, 36.33, 8.06; MS *m/z* calcd for $C_{25}H_{24}FN_{5}O_{4}$ [MH⁺]: 478.19, found: 478.38; Anal. calcd for $C_{25}H_{24}FN_{5}O_{4}$: C, 62.88; H, 5.07; N, 14.67; found: C, 62.74; H, 5.11; N, 14.79.

(*E/Z***)‑7‑{4‑[2‑(4‑Chlorobenzylidene)hydrazinecarbonyl] piperazin‑1‑yl}‑1‑cyclopropyl‑6‑fuoro‑4‑oxo‑1,4‑dihydro‑ quinoline‑3‑carboxylic acid 3b**

Yield 0.471 g (92%); pale yellow powder, mp: 234–36 °C; IR (KBr) 3333 (NH str), 3062 (aromatic C–H str), 2955 (aliphatic C–H str), 1716 (carboxylic C=O str), 1665 (amidic C=O str) 1627 (quinolone C=O str); ¹H-NMR (400 MHz, DMSO-*d*6) *δ* 15.01 (brs, 1H, COOH), 10.50 (s, 1H, NH), 8.68 (s, 1H, C2–H), 8.16 (s, 1H, =CH), 7.93 (d, *J*=13.1 Hz, 1H, C5–H), 7.70–7.57 (m, 3H, C8–H & 2Ar–H), 7.48 (d, *J*=7.9 Hz, 2H, Ar–H), 3.89–3.79 (m, 1H, cyclopropyl-H), 3.76–3.66 (m, 4H, piperazinyl-H), 3.46–3.33 (m, 4H, piperazinyl-H), 1.40–1.29 (m, 2H, cyclopropyl-H), 1.25–1.16 (m, 2H, cyclopropyl-H); ¹³C-NMR (100 MHz, DMSO- d_6) δ 176.89, 166.31, 154.84, 153.47 (d, *J*_{C–F}=248.8 Hz), 148.47, 145.49 (d, *J*_{C–F}=11.8 Hz), 142.06, 139.68, 134.42, 134.05, 129.25, 128.50, 126.90, 119.34, 111.50 (d, *J*_{C–F}=23.0 Hz), 107.23, 49.81, 44.19, 36.33, 8.06; MS *m/z* calcd for

 $C_{25}H_{23}CIFN_5O_4 [M^+]$: 511.14, found: 511.86; Anal. calcd for $C_{25}H_{23}CIFN_5O_4$: C, 58.65; H, 4.53; N, 13.68; found: C, 58.51; H, 4.47; N, 13.54.

(*E/Z***)‑1‑Cyclopropyl‑6‑fuoro‑7‑{4‑[2‑(4‑methylbenzylidene) hydrazinecarbonyl]piperazin‑1‑yl}‑4‑oxo‑1,4‑dihydroquino‑ line‑3‑carboxylic acid 3c**

Yield 0.319 g (65%); pale beige powder, mp: 240–42 °C; IR (KBr) 3326 (NH str), 3043 (aromatic C–H str), 2919 (aliphatic C–H str), 1727 (carboxylic C=O str), 1670 (amidic C=O str) 1626 (quinolone C=O str); ¹H-NMR (400 MHz, DMSO-*d*6) *δ* 15.05 (brs, 1H, COOH), 10.34 (s, 1H, NH), 8.67 (s, 1H, C2–H), 8.13 (s, 1H, =CH), 7.92 (d, *J*=13.2 Hz, 1H, C5–H), 7.60 (d, *J* = 6.1 Hz, 1H, C8–H), 7.53 (d, *J* = 7.6 Hz, 2H, Ar–H), 7.23 (d, *J*= 7.6 Hz, 2H, Ar–H), 3.90–3.78 (m, 1H, cyclopropyl-H), 3.75–3.65 (m, 4H, piperazinyl-H), 3.44–3.35 (m, 4H, piperazinyl-H), 2.34 (s, 3H, Ph–CH3), 1.39–1.29 (m, 2H, cyclopropyl-H), 1.24–1.17 (m, 2H, cyclopropyl-H); ¹³C-NMR (100 MHz, DMSO- d_6) δ 176.86, 166.30, 155.03, 153.46 (d, *J*_{C–F}=249.0 Hz), 148.42, 145.49 (d, *J*_{C–F}=11.5 Hz), 143.52, 139.66, 139.29, 132.75, 129.75, 126.87, 119.27, 111.48 (d, *J_{C-F}*=23.2 Hz), 107.40, 107.01, 49.82, 44.21, 36.33, 21.38, 8.06; MS *m/z* calcd for $C_{26}H_{26}FN_5O_4$ [M⁺]: 491.20, found: 491.56; Anal. calcd for $C_{26}H_{26}FN_5O_4$: C, 63.53; H, 5.33; N, 14.25; found: C, 63.46; H, 5.29; N, 14.37.

(*E/Z***)‑1‑Cyclopropyl‑6‑fuoro‑7‑{4‑[2‑(4‑methoxyben‑ zylidene)hydrazinecarbonyl]piperazin‑1‑yl}‑4‑oxo‑1,4‑dihy‑ droquinoline‑3‑carboxylic acid 3d**

Yield 0.320 g (63%); pale beige powder, mp: 148–50 °C; IR (KBr) 3255 (NH str), 3075 (aromatic C–H str), 2972 (aliphatic C–H str), 1716 (carboxylic C=O str), 1654 (amidic C=O str) 1622 (quinolone C=O str); ¹H-NMR (400 MHz, DMSO- d_6) δ 15.10 (brs, 1H, COOH), 10.26 (s, 1H, NH), 8.62 (s, 1H, C2–H), 8.11 (s, 1H, =CH), 7.93 (d, *J*=13.0 Hz, 1H, C5–H), 7.81 (d, *J*=6.6 Hz, 1H, C8–H), 7.67–7.53 (m, 2H, Ar–H), 7.06 (d, *J*=6.8 Hz, 2H, Ar–H), 4.06–3.94 (m, 1H, cyclopropyl-H), 3.84 (s, 3H, Ph–OCH3), 3.76–3.64 (m, 4H, piperazinyl-H), 3.46–3.34 (m, 4H, piperazinyl-H), 1.41–1.29 (m, 2H, cyclopropyl-H), 1.27–1.16 (m, 2H, cyclopropyl-H); ¹³C-NMR (100 MHz, DMSO- d_6) *δ* 176.88, 166.31, 162.20, 160.72, 155.14, 148.46, 145.55 (d, *J*_{C–F}=13.1 Hz), 143.42, 139.68, 130.41, 128.39, 127.14, 114.90, 111.50 (d, *J*_{C–F}=23.1 Hz), 107.42, 107.04, 55.88, 49.84, 44.22, 36.33, 8.06; MS m/z calcd for $C_{26}H_{26}FN_5O_5$ $[M + 2H^+]$: 509.21, found: 509.60; Anal. calcd for $C_{26}H_{26}FN_5O_5$: C, 61.53; H, 5.16; N, 13.80; found: C, 61.37; H, 5.13; N, 13.91.

(*E/Z***)‑1‑Cyclopropyl‑6‑fuoro‑7‑{4‑[2‑(2‑hydroxyben‑ zylidene)hydrazinecarbonyl]piperazin‑1‑yl}‑4‑oxo‑1,4‑dihy‑ droquinoline‑3‑carboxylic acid 3e**

Yield 0.469 g (95%); off white powder, mp: 234–36 °C; IR (KBr) 3318 (NH str), 3034 (aromatic C–H str), 2950 (aliphatic C–H str), 1728 (carboxylic C=O str), 1672 (amidic C=O str) 1625 (quinolone C=O str); ¹H-NMR (400 MHz, DMSO-*d*6) *δ* 15.08 (brs, 1H, COOH), 11.42 (s, 1H, OH), 10.72 (s, 1H, NH), 8.67 (s, 1H, C2–H), 8.36 (s, 1H, =CH), 7.92 (d, *J*=12.6 Hz, 1H, C5–H), 7.61 (d, *J*=6.0 Hz, 1H, C8–H), 7.41 (d, *J*=6.4 Hz, 1H, Ar–H), 7.30–7.19 (m, 1H, Ar–H), 6.99–6.83 (m, 2H, Ar–H), 3.91–3.79 (m, 1H, cyclopropyl-H), 3.77–3.65 (m, 4H, piperazinyl-H), 3.50–3.35 (m, 4H, piperazinyl-H), 1.34 (s, 2H, cyclopropyl-H), 1.27–1.14 (m, 2H, cyclopropyl-H); ¹³C-NMR (100 MHz, DMSO- d_6) δ 176.87, 166.29, 157.63, 154.22, 153.96 (d, *J*_{C-F}=250.5 Hz), 149.33, 148.45, 145.45, 145.06, 139.66, 130.81, 129.78, 119.55, 119.39, 116.76, 111.50 (d, $J_{C-F} = 22.7$ Hz), 107.40, 107.07, 49.73, 43.87, 36.33, 8.07; MS *m/z* calcd for $C_{25}H_{24}FN_{5}O_{5}$ [M + H⁺]: 494.18, found: 494.17; Anal. calcd for $C_{25}H_{24}FN_5O_5$: C, 60.85; H, 4.90; N, 14.19; found: C, 60.68; H, 4.93; N, 14.16.

(*E/Z***)‑7‑{4‑[2‑(4‑Bromobenzylidene)hydrazinecarbonyl] piperazin‑1‑yl}‑1‑cyclopropyl‑6‑fuoro‑4‑oxo‑1,4‑dihydro‑ quinoline‑3‑carboxylic acid 3f**

Yield 0.490 g (88%); pale beige powder, mp: 236–38 °C; IR (KBr) 3363 (NH str), 3078 (aromatic C–H str), 2867 (aliphatic C–H str), 1715 (carboxylic C=O str), 1652 (amidic C=O str) 1627 (quinolone C=O str); ¹H-NMR (400 MHz, DMSO-*d*6) *δ* 14.86 (s, 1H, COOH), 10.51 (s, 1H, NH), 8.67 (s, 1H, C2-H), 8.14 (s, 1H, =CH), 7.92 (d, *J*=13.2 Hz, 1H, C5–H), 7.67–7.51 (m, 5H, C8–H & 4Ar–H), 3.89–3.78 (m, 1H, cyclopropyl-H), 3.77–3.66 (m, 4H, piperazinyl-H), 3.45–3.34 (m, 4H, piperazinyl-H), 1.40–1.29 (m, 2H, cyclopropyl-H), 1.25–1.16 (m, 2H, cyclopropyl-H); 13 C-NMR (100 MHz, DMSO-*d*6) *δ* 176.88, 166.30, 154.83, 153.46 (d, *J*_{C–F}=248.4 Hz), 148.45, 145.48 (d, *J*_{C–F}=11.9 Hz), 142.12, 139.67, 134.78, 132.16, 128.75, 122.68, 119.29, 111.50 (d, *J*C–F=23.0 Hz), 107.40, 107.03, 49.79, 44.18, 36.33, 8.06; MS m/z calcd for $C_{25}H_{23}BrFN_{5}O_{4}$ [M-H⁺]: 554.08, found: 554.59; Anal. calcd for $C_{25}H_{23}BrFN_{5}O_{4}$: C, 53.97; H, 4.17; N, 12.59; found: C, 53.87; H, 4.19; N, 12.52.

(*E/Z***)‑1‑Cyclopropyl‑6‑fuoro‑7‑{4‑[2‑(naphthalen‑1‑ylmeth‑ ylene)hydrazinecarbonyl]piperazin‑1‑yl}‑4‑oxo‑1,4‑dihyd‑ roquinoline‑3‑carboxylic acid 3g**

Yield 0.401 g (76%); white powder, mp: 262–64 °C; IR (KBr) 3253 (NH str), 3034 (aromatic C–H str), 2856 (aliphatic C–H str), 1688 (carboxylic C=O str), 1669 (amidic

C=O str) 1625 (quinolone C=O str); ¹H-NMR (400 MHz, DMSO-*d*6) *δ* 15.06 (brs, 1H, COOH), 10.53 (s, 1H, NH), 8.86–8.76 (m, 2H, Ar–H), 8.68 (s, 1H, C2–H), 8.03–7.89 (m, 3H, C5-H, =CH & Ar–H), 7.84 (d, *J* = 7.1 Hz, 1H, Ar–H), 7.67–7.53 (m, 4H, C8–H & 3Ar–H), 3.89–3.81 (m, 1H, cyclopropyl-H), 3.80–3.71 (m, 4H, piperazinyl-H), 3.48–3.38 (m, 4H, piperazinyl-H), 1.39–1.29 (m, 2H, cyclopropyl-H), 1.25–1.17 (m, 2H, cyclopropyl-H); 13 C-NMR (100 MHz, DMSO-*d*6) *δ* 176.87, 166.30, 154.95, 153.48 (d, *J*_{C–F}=249.6 Hz), 148.44, 145.49 (d, *J*_{C–F}=11.1 Hz), 143.22, 139.66, 136.93, 135.65, 134.06, 130.57, 130.04, 129.15, 127.43, 127.11, 126.56, 125.99, 124.70, 119.29, 111.50 (d, *J*C–F=23.1 Hz), 107.22, 49.83, 44.20, 36.33, 8.06; MS *m/z* calcd for $C_{29}H_{26}FN_5O_4$ [M⁺]: 527.20, found: 526.99; Anal. calcd for $C_{29}H_{26}FN_5O_4$: C, 66.02; H, 4.97; N, 13.28; found: C, 65.91; H, 4.94; N, 13.36.

(*E/Z***)‑1‑Cyclopropyl‑6‑fuoro‑7‑{4‑[2‑(furan‑2‑ylmethylene) hydrazinecarbonyl]piperazin‑1‑yl}‑4‑oxo‑1,4‑dihydroquino‑ line‑3‑carboxylic acid 3h**

Yield 0.313 g (67%); beige powder, mp: 206–08 °C; IR (KBr) 3333 (NH str), 3091 (aromatic C–H str), 2846 (aliphatic C–H str), 1711 (carboxylic C=O str), 1672 (amidic C=O str) 1624 (quinolone C=O str); 1 H-NMR (400 MHz, DMSO-*d*6) *δ* 15.03 (brs, 1H, COOH), 10.38 (s, 1H, NH), 8.67 (s, 1H, C2–H), 8.08 (s, 1H,=CH), 7.91 (d, *J*=13.1 Hz, 1H, C5–H), 7.75 (s, 1H, furan C5–H), 7.60 (d, *J*=6.8 Hz, 1H, C8–H), 6.76 (s, 1H, furan C3–H), 6.59 (s, 1H, furan C4–H), 3.89–3.79 (m, 1H, cyclopropyl-H), 3.74–3.64 (m, 4H, piperazine-4H), 3.44–3.34 (m, 4H, piperazine-4H), 1.38–1.30 (m, 2H, cyclopropyl-H), 1.24–1.17 (m, 2H, cyclopropyl-H); ¹³C-NMR (100 MHz, DMSO- d_6) δ 176.85, 166.29, 153.48 (d, *J*_{C–F}=256 Hz), 150.55, 148.42, 145.46 $(d, J_{C-F}=12.3 Hz)$, 144.59, 139.66, 133.81, 131.31, 120.46, 119.26, 112.15, 111.47 (d, *J*_{C–F}=21.2 Hz), 107.38, 107.00, 49.77, 44.11, 36.32, 8.05; MS m/z calcd for C₂₃H₂₂FN₅O₅ [M⁺]: 467.16, found: 467.59; Anal. calcd for $C_{23}H_{22}FN_{5}O_{5}$: C, 59.10; H, 4.74; N, 14.98; found: C, 59.28; H, 4.87; N, 14.83.

(*E/Z***)‑1‑Cyclopropyl‑6‑fuoro‑4‑oxo‑7‑{4‑[2‑(thiophen‑ 2‑ylmethylene)hydrazinecarbonyl]piperazin‑1‑yl}‑1,4‑dihy‑ droquinoline‑3‑carboxylic acid 3i**

Yield 0.406 g (84%); beige powder, mp: 234–36 °C; IR (KBr) 3337 (NH str), 3083 (aromatic C–H str), 2876 (aliphatic C–H str), 1710 (carboxylic C=O str), 1671 (amidic C=O str) 1625 (quinolone C=O str); ¹H-NMR (400 MHz, DMSO-*d*₆) *δ* 15.06 (brs, 1H, COOH), 10.37 (s, 1H, NH), 8.67 (s, 1H, C2–H), 8.39 (s, 1H,=CH), 7.92 (d, *J*=13.2 Hz, 1H, C5–H), 7.60 (d, *J* = 7.0 Hz, 1H, C8–H), 7.55 (d, *J*=4.2 Hz, 1H, thiophene C5–H), 7.32 (s, 1H, thiophene

C3–H), 7.10 (s, 1H, thiophene C4–H), 3.89–3.79 (m, 1H, cyclopropyl-H), 3.74–3.63 (m, 4H, piperazinyl-H), 3.44–3.34 (m, 4H, piperazinyl-H), 1.39–1.29 (m, 2H, cyclopropyl-H), 1.25–1.16 (m, 2H, cyclopropyl-H); 13 C-NMR (100 MHz, DMSO-*d*6) *δ* 176.86, 166.29, 154.81, 153.45 (d, *J*_{C–F}=248 Hz), 148.43, 145.47 (d, *J*_{C–F}=10.4 Hz), 140.37, 139.66, 138.84, 132.95, 129.33, 128.05, 127.90, 119.34, 111.48 (d, J_{C-F} =23.0 Hz), 107.20, 49.79, 44.16, 36.32, 8.06; MS m/z calcd for $C_{23}H_{22}FN_5O_4S$ [M⁺]: 483.14, found: 383.13; Anal. calcd for $C_{23}H_{22}FN_5O_4S$: C, 57.13; H, 4.59; N, 14.48; found: C, 57.32; H, 4.66; N, 14.61.

General procedure for synthesis of target com‑ pounds 4a–c

A mixture of ciprofloxacin carbamate intermediate **1** (0.496 g, 0.001 mol) and the corresponding amine [either as 0.003 mol in 10 mL dioxane, for compounds **4a** and **4b** or neat using 2 mL of morpholine (0.023 mol), for compound **4c**] was heated at refux for 3 h, then cooled to room temperature and poured to 0.1 M HCl (20 mL). The obtained precipitate was collected, washed successively with water and dried.

1‑Cyclopropyl‑6‑fuoro‑4‑oxo‑7‑[4‑(pyrrolidine‑1‑carbonyl) piperazin‑1‑yl]‑1,4‑dihydroquinoline‑3‑carboxylic acid 4a

Yield 0.274 g (64%); beige powder, mp: 278–80 °C; ¹H-NMR (400 MHz, CDCl₃) δ 14.81 (brs, 1H, COOH), 8.63 (s, 1H, C2–H), 7.87 (d, *J*=13.0 Hz, 1H, C5–H), 7.28 (d, *J*=6.4 Hz, 1H, C8–H), 3.53–3.42 (m, 5H, cyclopropyl-H & piperazinyl-H), 3.41–3.30 (m, 4H, piperazinyl-H), 3.33–3.22 (m, 4H, pyrrolidinyl-H), 1.85–1.75 (m, 4H, pyrrolidinyl-H), 1.38–1.27 (m, 2H, cyclopropyl-H), 1.23–1.12 (m, 2H, cyclopropyl-H); ¹³C-NMR (100 MHz, CDCl₃) δ 177.00, 166.83, 162.39, 153.63 (d, *J*_{C–F}= 251.5 Hz), 147.38, 145.81 (d, *J*_{C–F}=9.8 Hz), 139.07, 119.85, 112.35 (d, *J*_{C–F}=23.4 Hz), 108.10, 104.93, 49.65, 48.36, 45.88, 35.31, 25.52, 8.21; MS m/z calcd for $C_{22}H_{25}FN_4O_4$ [M⁺]: 428.19, found: 428.16; Anal. calcd for $C_{22}H_{25}FN_4O_4$: C, 61.67; H, 5.88; N, 13.08; found: C, 61.90; H, 6.04; N, 12.79.

1‑Cyclopropyl‑6‑fuoro‑4‑oxo‑7‑[4‑(piperidine‑1‑carbonyl) piperazin‑1‑yl]‑1,4‑dihydroquinoline‑3‑carboxylic acid 4b

Yield 0.354 g (80%); pale yellow powder, mp: 286–88 $°C$; ¹H-NMR (400 MHz, CDCl₃) δ 14.83 (brs, 1H, COOH), 8.67 (s, 1H, C2–H), 7.90 (d, *J*=12 Hz, 1H, C5–H), 7.33 (d, *J*=8.0 Hz, 1H, C8–H), 3.58–3.35 (m, 5H, cyclopropyl-H & piperazinyl-H), 3.32–3.23 (m, 4H, piperazinyl-H), 3.22–3.14 (m, 4H, piperidinyl-H), 1.61–1.47 (m, 6H, piperidinyl-H), 1.40–1.28 (m, 2H, cyclopropyl-H), 1.22–1.10 (m, 2H, cyclopropyl-H); 13C-NMR (100 MHz, CDCl3) *δ* 177.09, 166.84,

163.91, 153.68 (d, *J*_{C–F}= 251.8 Hz), 147.44, 145.82 (d, *J*_{C–F}=9.1 Hz), 139.07, 120.05, 112.49 (d, *J*_{C–F}=23.4 Hz), 108.24, 104.97, 49.64, 47.79, 46.70, 35.29, 25.76, 24.64, 8.24; MS m/z calcd for $C_{23}H_{27}FN_4O_4$ [M⁺]: 442.20, found: 442.51; Anal. calcd for $C_{23}H_{27}FN_4O_4$: C, 62.43; H, 6.15; N, 12.66; found: C, 62.67; H, 6.34; N, 12.90.

1‑Cyclopropyl‑6‑fuoro‑7‑[4‑(morpholine‑4‑carbonyl) piperazin‑1‑yl]‑4‑oxo‑1,4‑dihydroquinoline‑3‑carboxylic acid 4c

Yield 0.338 g (76%); pale beige powder, mp: $268-72$ °C; ¹H-NMR (400 MHz, CDCl₃) δ 14.81 (brs, 1H, COOH), 8.64 (s, 1H, C2–H), 7.90 (d, *J*=12 Hz, 1H, C5–H), 7.30 (d, *J*=6.0 Hz, 1H, C8–H), 3.70–3.58 (m, 4H, morpholinyl-H), 3.52–3.40 (m, 5H, cyclopropyl-H & piperazinyl-H), 3.33–3.22 (m, 8H, piperazinyl-H & morpholinyl-H), 1.38–1.28 (m, 2H, cyclopropyl-H), 1.18–1.08 (m, 2H, cyclopropyl-H); 13C-NMR (100 MHz, CDCl3) *δ* 176.97, 166.78, 163.52, 153.63 (d, *J*_{C–F}= 251.7 Hz), 147.41, 145.68 (d, *J*_{C–F}=9.9 Hz), 139.03, 119.99, 112.38 (d, *J*_{C–F}=23.6 Hz), 108.11, 105.04, 66.60, 49.56, 47.27, 46.52, 35.32, 8.22; MS m/z calcd for $C_{22}H_{25}FN_{4}O_{5}$ [M⁺]: 444.18, found: 444.06; Anal. calcd for $C_{22}H_{25}FN_4O_5$: C, 59.45; H, 5.67; N, 12.61; found: C, 59.31; H, 5.80; N, 12.67.

Biological investigation

Evaluation of anticancer activity

NCI anticancer screening

According to the protocol of Drug Evaluation Branch of the National Cancer Institute (NCI), Bethesda, USA (described in details at [http://www.dtp.nci.nih.gov\)](http://www.dtp.nci.nih.gov) [\[44–](#page-15-27)[46\]](#page-15-28), the process of in vitro screening utilized about 60 diferent human tumor cell lines (NCI-60 cell lines panel) representing leukemia, melanoma and cancers of the lung, colon, brain, ovary, breast, prostate, and kidney. Briefy, compounds were added to the cancer cell lines at a concentration of 10 μM and incubated for 48 h, then cellular growth was terminated by adding the protein binding dye, sulforhodamine B (SRB). Anticancer activity for each compound was recorded as the growth percent of cells treated with the test compound as compared with control untreated ones. Growth percentages were measured spectrophotometrically.

In vitro MTT cytotoxicity assay

Cell line cells were obtained from American Type Culture Collection. Cells were cultured using DMEM (Invitrogen/Life Technologies) supplemented with 10% FBS (Hyclone), 10 μg/mL of insulin (Sigma), and 1% penicillin–streptomycin. All of the other chemicals and reagents were from Sigma. Cells (cells density $1.2-1.8 \times 10,000$ cells/ well) were plated in a volume of 100 μ L complete growth medium $+100 \mu L$ of the tested compound per well in a 96-well plate for 24 h before the MTT assay.

The MTT method for monitoring in vitro cytotoxicity [\[47\]](#page-15-29) is well suited for use with multi-well plates. For best results, cells in the log phase of growth were employed and final cell number did not exceed 106 cells/cm². Each test included a blank containing complete medium without cells. Cultures were removed from incubator into a sterile work area. Each vial of MTT [M-5655] was reconstituted to be used with 3 mL of medium without phenol red and serum. Reconstituted MTT were added to an amount equal to 10% of the culture medium volume. Cultures were returned to incubator for 2–4 h. After the incubation period, cultures were removed from incubator and the resulting formazan crystals were dissolved by adding an amount of MTT solubilization solution [M-8910] equal to the original culture medium volume. Dissolution was enhanced by gentle mixing in a gyratory shaker or occasionally by pipetting up and down (trituration) to completely dissolve the MTT formazan crystals. Absorbance was measured spectrophotometrically at a wavelength of 570 nm. The background absorbance of multiwell plates was measured at 690 nm and subtracted from the 450 nm measurement. Tests performed in multiwell plates were read after transferring contents to appropriate size cuvettes for spectrophotometric measurement.

Screening of antibacterial activity

Antibacterial activity of test compounds were determined according to the agar cup difusion method [\[48](#page-15-30)] using *Klebsiella pneumoniae* and *Proteus mirabilis* as representatives for Gram-negative urease producing bacteria. These strains represent common contaminants of the environment and involved in human and animal diseases. Bacterial strains were kindly provided by the Assiut University Mycological Centre (AUMC). To prepare inocula for bioassay, bacterial strains were individually cultured for 48 h in a universal tube 20 mL nutrient broth medium. Test was done in 10 cm sterile plastic Petri plates in which microbial suspension (1 mL/ plate) and 15 mL of nutrient agar medium were poured. After solidifcation of the media, 5 mm diameter cavities were cut in the solidifed agar (three cavities/plate) using sterile cork borer. Test compounds dissolved in dimethyl sulfuxide (DMSO) at a concentration of 2000 μM were pipetted in the cavities (50 µL/cavity). Cultures were then incubated at 28 °C for 48 h. Results were read as the diameter (in mm) of inhibition zone around cavities. For determination of minimum inhibitory concentrations (MICs), chemical compounds giving positive results were diluted with DMSO to prepare series of two fold descending concentrations (2000–31.25 μM) and similarly assayed as mentioned

Evaluation of urease inhibitory activity

before till the concentration that gave no activity.

This study was performed using indophenol method, which depends on color production relative to ammonia liberated from the enzyme catalytic activity on urea [[38\]](#page-15-31). An assay mixture consists of 1 mL buffer (phosphate pH 6.7, 50 mM; EDTA, 2 mM; sodium salicylate, 400 mM and sodium nitroprossid, 10 mM), 30 μ L of enzyme (30,000 U/L), 10 μ L of test compounds at diferent concentrations and 10 μL urea (50 mg/dL) was incubated at 37 °C for 10 min. End point was achieved by adding 200 μL of colouring reagent (sodium hypochlorite 140 mM and dodium hydroxide 150 mM), making a fnal volume of 1.25 mL and the mixture was reincubated for additional 10 min at 37 °C. Experiments were carried out in triplicate and thiourea was used as a standard inhibitor. A mixture prepared and treated as before but inhibitors were omitted was used as a control. Absorbance was measured at 578 nm and percentages of inhibition were calculated using the formula: 100 -(optical density $_{test}$ / optical density_{control}) \times 100 [[40,](#page-15-32) [49\]](#page-15-33).

Molecular docking study

Docking simulation study was performed using Molecular Operating Environment (MOE®) version 2008.10 (Chemical Computing Group Inc., Montreal, Canada) [\[50\]](#page-15-34). Test compound **3i** was docked onto the binding pocket of active site of urease obtained from protein data bank (PDB: 1E9Y) [\[36](#page-15-23)]. Preparation of the compound for docking was achieved via building its 3D structure by MOE and database formation. Test compound was subjected to a conformational search, and all conformers were subjected to energy minimization, that was performed with MOE until a RMSD gradient 0.01 kcal mol⁻¹ A°⁻¹ with MMFF94X force-field.3D. Preparation of the protein was performed by 3D protonation, removal of unwanted water and fnally surfaces and maps were taken before docking of test compounds on the enzyme active site. Flexible ligand–rigid receptor docking of the most stable conformers was done with MOE-DOCK using triangle matcher as the placement method, London dG as the scoring function, and refnement of the results was achieved using force feld energy. Docking results had appeared in a DBV window (dock.mdb). The **S** feld, that the docking poses are ranked by the MM/GBVI binding free energy calculation is identical to the E_refne score. Database browser was used for comparing docking poses of the ligand in the co-crystallized structure. Thirty of the most stable docking models for the ligand were retained with the best scored conformation.

Results and discussion

Scheme 1 Synthesis of target compounds **3a**–**i** and **4a**–**c**

Chemistry

Protocols for synthesis of the target compounds are illustrated in Schemes [1.](#page-7-0) Ciprofoxacin was converted to the corresponding carbamate **1** by heating at refux with 4-nitrophenyl chloroformate in acetonitrile containing pyridine as a base. Conventional utilization of ethyl chloroformate gave ethyl carbamate of ciprofoxacin with a poor reactivity toward nucleophilic attack by amines. This was consistent with the previously reported data regarding tertiary carbamates [[37\]](#page-15-35). However, 4-nitrophenol is known for its better nucleofugality than ethanol and consequently, 4-nitrophenyl carbamate of ciprofoxacine **1** could be efectively reacted with diferent nucleophiles. Compound **1** was heated at refux with hydrazine hydrate in ethanol to afford the hydrazide intermediate **2**. Condensation of compound **2** with diferent aromatic aldehydes in methanol gave the target compounds **3a**–**i**. Alternatively, ciprofloxacin carbamate intermediate **1** was reacted with the appropriate cyclic amines to aford the titled derivatives **4a**–**c**. Diferent spectroscopic and analytical tools were applied for identifcation of the synthesized compounds including IR, 1 H-NMR, 13 C-NMR, Mass spectroscopy, and elemental analyses. Compounds **3a**–**i** were characterized by two singlet signals (1H each) at $\delta \sim 10.40$ and 8.16 ppm assigned as amidic NH and N=CH protons, respectively. ¹³C-NMR Spectra of compounds **3a**–**i** revealed three carbonyl signals appeared at *δ* ~ 176.87, 166.30, and 155.14–154.22 ppm, assigned to C=O of the quinolone nucleus, carboxylic group and that of carbamoyl bridge, respectively. Similarly, derivatives **4a–c** showed the same signals at $\delta \sim 177.00$, 166.80, and 163.91–162.39 ppm, respectively. Elemental analysis and MS furtherly confrmed the assigned structures of the synthesized compounds.

Biological evaluation

Evaluation of anticancer activity

NCI anticancer screening

All of the target compounds **3a**–**i** and **4a**–**c** were selected for in vitro anticancer screening according to the applied rules for compounds selection by drug evaluation branch of the National Cancer Institute, Bethesda, USA. Selected compounds were tested at a single concentration of 10 μM, using sixty diferent tumor cell lines known as NCI-60 cell line panel, representing nine types of human cancers including both solid and liquid tumors. Compounds were applied at determined concentration and incubated for 28 h then growth was terminated by adding a protein binding dye, sulforhodamine B (SRB). Antitumor activity was recorded as the growth percent of cells treated with test compound in comparison with control untreated ones.

Compound **3a** revealed a moderate antitumor activity against only leukemia SR cell line with a growth percentage of 54.41%. No other considerable cell growth suppression was observed except with non-small cell lung cancer NCI-H522 and renal cancer UO-31 cell lines, where growth percentages of 78.04 and 74.00%, respectively, were recorded (Supplementary data). On the other hand, derivative **3b** showed a weak anticancer activity, where the most pronounced cell growth inhibition noted was against leukemia MOLT-4, non-small cell lung cancer NCI-H522, renal cancer UO-31, and breast cancer BT-549 cell lines with growth percentages 82.92, 82.03, 81.15, and 85.64%, respectively (Supplementary data). Similarly, a weak cell growth inhibition was indicated with compounds **3c** (Supplementary data). Activity of interest exerted by compound **3c** was against leukemia MOLT-4, CNS cancer SF-268, SNB-75, melanoma LOX IMVI, renal cancer CAKI-1, and UO-31 cells (growth percentages 82.45, 84.21, 83.64, 85.16, 83.45, and 72.23%, respectively). Compounds **3d** and **3e** revealed also a weak activity, where the most intense cell growth inhibition was observed toward the renal cancer UO-31 cell line with growth percentages 84.34 and 84.11%, respectively (Supplementary data).

It is worth to note that compound **3f** experienced a remarkable cell growth inhibition against the breast cancer BT-549 cell line, recording a growth percentage of 28.68%. A moderate activity was also observed by compound **3f** against Leukemia CCRF-CEM, melanoma LOX IMVI, UACC-62, renal cancer UO-31, and breast cancer T-47D cells (growth percentages 62.87, 58.54, 49.35, 64.83, and 67.67%, respectively). Inversely, the activity against other cell lines was weak, Table [1.](#page-9-0) Compound **3g** revealed also a potent antitumor activity against colon cancer HCT-116 and breast cancer BT-549 cell lines (growth percentages 14.76 and 6.18%, respectively). The compound exhibited also a moderate activity against melanoma LOX IMVI cells with a growth percentage of 67.78%. However, it showed a weak cell growth inhibition with some of the other cell lines, Table [1.](#page-9-0) Compounds **3h** and **3i** showed a weak antitumor activity against some of the NCI panel cell lines (Supplementary data).

On the other hand, compound **4a** experienced a weak anticancer activity (Supplementary data), where the most pronounced cell growth inhibition recorded was against nonsmall cell lung cancer NCI-H522, melanoma UACC-62, and renal cancer UO-31 cell lines (growth percentages 76.36, 74.90, and 79.73%, respectively). However, compounds **4b** and **4c** revealed no signifcant antitumor potential, where the highest activity was obtained by compound **4b** against CNS cancer SNB-75 and renal cancer UO-31 cells with growth percentages of 87.92 and 86.42%, respectively (Supplementary data).

From the above results, it is obvious that most of the tested compounds experienced moderate to weak activity against the tested cell lines. Relatively, compounds **4a**–**c** having no aromatic or heteroaromatic moiety at position 7 showed a very weak activity, this fnding is consistent with that reported in literatures [[9\]](#page-14-7). The most promising antitumor results were obtained with compound **3f** toward breast cancer BT-549, derivative **3g** against breast cancer BT-549 and colon cancer HCT-116 as well as analog **3a** on leukemia SR cell lines with growth percentages of 28.68, 6.18, 14.76, and 54.41%, respectively (Fig. [1](#page-10-0)).

In vitro MTT cytotoxicity assay

Cytotoxic activity of compounds **3f** and **3g** was evaluated against two specifc cell lines, namely BT-549 and MCF10a, using doxorubicin (DOX) as a reference drug. These two cell lines were selected based on the sensitivity of the tested cells shown by the previously mentioned NCI results. The growth inhibition is expressed as the median growth inhibitory concentration (IC_{50}) , which corresponds to the concentration required for 50% inhibition of cell viability. Compounds **3f** and **3g** revealed IC_{50} comparable to the standard drug, doxorubicin toward the breast cancer BT-549 cell line (1.84, 9.83, and 1.29 µM, respectively), where derivative **3f** was nearly equipotent to standard. On the other hand, the test compounds were less toxic than doxorubicin toward the nontumorigenic MCF10a breast cell line (Fig. [2\)](#page-11-0).

Evaluation of antibacterial activity

Standard agar cup difusion method was applied for antibacterial screening of the target compounds **3a–i** and **4a–c** along

Table 1 Growth percentages of diferent cancer cell lines for compounds **3f** and **3g** at 10 µM

with intermediates **1** and **2**, using two Gram-negative urease producing bacterial strains, *Proteus mirabilis* and *Klebsiella pneumonia*. Chloramphenicol and the parent drug, ciprofoxacin were used as positive controls. Compound **4a** revealed activity against *Klebsiella pneumonia* more than that of the reference drug, chloramphenicol ($MIC = 100.64$ and 217.08 µM, respectively). However, the antibacterial activity of this derivative was less than the parent drug,

ciprofoxacin (MIC=40.16 µM). Compound **4c** showed anti-*Klebsiella pneumonia* activity only at the highest concentration used, 2 mM, while other derivatives revealed no activity at 2 mM. On the other hand, only compound **3e** exhibited activity against *Proteus mirabilis* at the initial concentration used, 2 mM. Meanwhile, other derivatives experienced no activity up to the highest concentration applied. Antibacterial screening results are shown in Table [2](#page-11-1).

O O

OH O

O

Fig. 1 Some important biologically active quinolones and the designed **series I** and **II** O N

N

 N^N H

 $N_{\rm c}$ \sim $N_{\rm N}$

O

OH

HN

O

F

O

N H N

O

N N

F

V

F

Series I

H N O N

Ar N

Heteroyclic Aromatic

Evaluation of urease inhibitory activity

Indophenol test for detection of ammonia (Weatherburn method) [[38](#page-15-31)] was applied for evaluation of urease inhibitory activity of the synthesized compounds along with ciprofoxacin, using thiourea as a reference enzyme inhibitor [[39\]](#page-15-36). This enzyme assay depends on the colored indophenol produced by the reaction of phenolic compounds and chlorine with ammonia liberated as a result of the enzyme catalytic activity on urea. Reduction in color intensity of samples treated with the test compounds and standard inhibitor at diferent concentrations was measured in comparison with untreated control one. Percentages of enzyme inhibition were calculated using the formula:

100-(optical density_{test}/optical density_{control}) \times 100 [[40](#page-15-32)].

Assay results revealed that most of the tested compounds have a urease inhibitory activity better than the parent drug, ciprofoxacin (CIP) and comparable to or more than the standard thiourea (THR, Fig. [3](#page-12-0)). Compounds **3i** and **4a** were the most potent urease inhibitors revealing IC₅₀ of 58.92 and 73.40 μ M, respectively (78.89 μ M for

thiourea). These compounds can be considered as potential candidates for further more investigation.

Molecular docking

MOE program was utilized to study the docking of the test compounds on the active site of urease enzyme (Fig. [4a](#page-13-0)–c). Ddocking reliability was validated using the known X-ray structure of *Helicobacter pylori* urease (PDB: 1E9Y) [\[36\]](#page-15-23) in complex with acetohydroxamic acid (AHA). Molecular docking study for the most active compounds **3i** and **4a** showed the ability of such derivatives to bind strongly with the bi-nickel center of the urease enzyme as indicated by their binding energy values. Compounds **3i** and **4a** revealed binding scores better than that of the standard ligand, AHA (binding energy − 78.70, − 104.16, and − 33.45 kcal/mol, respectively). Figure [4a](#page-13-0) shows the binding mode of AHA with *H. pylori* urease, revealing coordination with the bi-nickel center of the enzyme and formation of a hydrogen bond with His221. Binding mode of compound **3i** (Fig. [4](#page-13-0)b) denotes that the **Fig. 2** Cytotoxicity of compounds **3f** and **3g** against BT-549 and MCF10a cell lines expressed as IC_{50} $(n=3)$

Table 2 Antibacterial activity of the tested compounds

a Not determined

carbonyl oxygen of carbamoyl moiety coordinates with Ni3002 and forms two hydrogen bonds with His248 and His274. As a hydrogen bond donor, NH group additionally participates in hydrogen bonding with Asp362. In compound **4a** (Fig. [4](#page-13-0)c), the carbonyl oxygen of urea moiety coordinates with Ni3002 and makes two hydrogen bonds with His248 and His274.

Conclusion

as IC_{50} in $\mu M (n=3)$

Diferent *N*-4 carbamoyl piperazinyl derivatives of ciprofoxacin were synthesized and biologically investigated. However, most of the tested compounds showed a moderate to weak anticancer activity, the 4-bromophenylhydrazone derivative, **3f** experienced a potent antitumor activity against the breast cancer BT-549 cell line with a growth percentage of 28.68%. Increasing bulkiness among hydrazone series via introducing a naphthyl moiety (compound **3g**) markedly improved activity against breast BT-549 and colon HCT-116 cancer cell lines, where growth percentages of 6.18 and 14.76%, respectively, were recorded. MTT assay indicated cytotoxicity of compounds **3f** and **3g** comparable to doxorubicin against the breast cancer BT-549 cell line $(IC_{50} = 1.84, 9.83, \text{ and } 1.29 \mu M, \text{ respectively})$; however, the test compounds experienced a reduced cytotoxicity toward the non-cancerous MCF10a cells $(IC_{50} = 21.00, 29.1,$ and

18.7 µM, respectively). The prepared compounds showed reduced antibacterial activity than their parent drug, ciprofoxacin. Meanwhile, among saturated heterocyclic derivatives, compound **4a** revealed activity against *Klebsiella pneumoniae* better than the standard drug used, chloramphenicol (MIC = 100.64 and 217.08 μ M, respectively) indicating that a fve membered ring cyclic amine is better than six membered ones for antibacterial activity. On the other, the majority of the newly synthesized ciprofoxacin analogs were found to have a urease inhibitory activity more than their parent drug and comparable to thiourea, where the thienyl hydrazone derivative **3i** showed a promising activity with IC₅₀ of 58.92 μ M (78.89 μ M for standard, thiourea). Computational study indicated the newly developed carbamoyl functionality can participate efficiently in binding to Ni ion at the urease active site, which may interpret the positive impact of such modifcation on the in vitro urease inhibitory activity.

Fig. 4 2D and 3D docking of AHA (**a**), compounds **3i** (**b**) and **4a** (**c**) with *H. pylori* urease

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Compliance with ethical standards

Conflict of interest The authors declare that there is no confict of interest.

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