



# Dielectric and magnetic properties of BaFe<sub>11.7</sub>Ni<sub>0.3</sub>O<sub>19</sub> (BHNF) ceramic synthesized by chemical route

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## Abstract

Nickel-doped barium hexaferrite, BaFe<sub>11.7</sub>Ni<sub>0.3</sub>O<sub>19</sub> (BHNF), ceramic was prepared by a chemical route. The single phase formation of BHNF ceramic in the sintered material at 1200 °C for 6 h was confirmed by powder XRD. Hexagonal and round shape particles were observed by transmission electron microscopy (TEM), and the particle size was found to be 81 ± 10 nm. The average roughness and root mean square roughness were observed by atomic force microscopy (AFM) evaluated to be 57 and 73 nm, respectively. The zero field cooled (ZFC), field cooled (FC), and magnetic hysteresis curves indicate temperature dependent ferromagnetic behavior of BHNF ceramic. The dielectric constant ( $\epsilon'$ ) was determined to be  $1.7 \times 10^4$  at 423 K and 100 Hz. The high value of  $\epsilon'$  indicates the presence of semiconducting grains and insulating grain boundaries which was supported by internal barrier layer capacitance mechanisms.

**Keywords** Chemical route · FT-IR · Magnetic behavior · Dielectric properties

## Introduction

Barium hexaferrite with chemical composition, MFe<sub>12</sub>O<sub>19</sub> (M = Ba, Sr or Pb), have considerable attention in electronic applications such as permanent magnet, high-density magnetic recording media, microwave device, and telecommunications equipment. These important applications of this material are due to strong uniaxial crystalline anisotropy, high saturated magnetic polarization, large coercivity, and high resonant frequency with an excellent capability and corrosion resistance [1–3]. The large crystal anisotropy and high intrinsic coercivity of this material are responsible for high stability and electrical resistivity. The hexaferrite (BaFe<sub>12</sub>O<sub>19</sub>) have M-type structure which exhibits strong ferromagnetism among ferrites. It was widely used in electronic device, and their properties strongly depend on microstructure and morphology. These materials have high relevance in higher frequency and high resistance to the heat and corrosion than others due to

high coercivity with small particle size [4]. The magnetic and dielectric properties of ferrites have been strongly influenced by synthesis procedures and substitution of various cations such as divalent and trivalent [5]. The various synthesis methods for the fabrication of M-type hexaferrite ceramic has been used, such as ceramic method [6], citrate auto-combustion synthesis [7], co-precipitation [8], mechano-thermal treatment [9], sol-gel, microemulsion techniques [10], ball milling [11], wet mixing method [12], and hydro-thermal [13]. The substitution of divalent and tetravalent (Co<sup>2+</sup>, Ni<sup>2+</sup>, Ti<sup>4+</sup>) cations at the site of Fe<sup>3+</sup> and Ba have been carried out for the improvement of magnetic properties [14, 15]. These improvements have occurred because of their similarity in ionic radii and their electronic configuration. Their magnetic properties may be conjointly enhanced by the substitution of Fe<sup>3+</sup> with magnetic or non-magnetic ions which result in significant change in saturation magnetization because of modification within the net magnetic moment per molecules [16–19]. The substitution of Mn<sup>3+</sup>, La<sup>3+</sup>, Co<sup>2+</sup>-Zr<sup>4+</sup>, Ti<sup>4+</sup>-Mn<sup>4+</sup>, and Co<sup>2+</sup>-Ti<sup>4+</sup> ions have been decreased in saturation magnetization (Ms) and coercivity (Hc) [20]. This decrease is due to the presence of secondary phase which also inhibits the recording application of material. Most of the works have been carried out for the synthesis and characterization of M-type barium hexaferrite, for examples BaFe<sub>12</sub>O<sub>19</sub>, BaMe<sub>x</sub>Fe<sub>12-x</sub>O<sub>19</sub> (Me<sup>3+</sup> = Al, Cr, Bi, Sc), BaMe<sub>x</sub>Ir<sub>x</sub>Fe<sub>12-2x</sub>O<sub>19</sub>

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( $\text{Me}^{+2} = \text{Co}, \text{Zn}$ ),  $\text{BaZn}_x\text{Sn}_x\text{Fe}_{12-2x}\text{O}_{19}$ ,  $\text{BaCo}_x\text{Ru}_x\text{Fe}_{12-2x}\text{O}_{19}$ ,  $\text{BaMe}_x\text{Ti}_x\text{Fe}_{11.6-2x}\text{O}_{19}$  ( $\text{Me}^{+2} = \text{Co}, \text{Zn}$ ), and  $\text{BaMe}_x\text{Ti}_x\text{Fe}_{12-2x}\text{O}_{19}$  ( $\text{Me}^{+2} = \text{Co}, \text{Zn}, \text{Mn}$ ) [21].

In the present study, we studied the effect of  $\text{Ni}^{+2}$  substitutions for the improvement of electrical and magnetic properties of  $\text{BaFe}_{12}\text{O}_{19}$  ceramic synthesized by chemical route. The chemical route is more suitable than other synthesis methods for the preparation of nanostructured materials. It was observed that substituting  $\text{Fe}^{+3}$  by  $\text{Ni}^{+2}$  ions reduces coercivity and saturation magnetization which is a major factor for recording data [22].

## Experimental

### Material synthesis

Nickel-doped barium hexaferrite was synthesized by the chemical route using analytical grade chemicals,  $\text{Ba}(\text{NO}_3)_2$  (99% Merck, India),  $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$  (98% Merck, India), and  $\text{Ni}(\text{NO}_3)_2$  (99% Merck, India) as starting materials. The stoichiometric amount of metal nitrates was dissolved in deionized distilled water along with the appropriate amount of citric acid ( $\text{C}_6\text{H}_8\text{O}_7 \cdot \text{H}_2\text{O}$ , 99% Merck, India) as per equivalent to the metal ions. The resulting solution was heated on a hot plate with magnetic stirrer at 70–80 °C to evaporate excesses water. On further heating, a fluffy mass was obtained which burns with a sooty flame. The obtained porous ash was crushed into fine powder with the help of mortar and pestle. The resultant dry powder was calcined at 800 °C for 8 h in the electrical muffle furnace after that cylindrical pellets (13.0 mm diameters, 1.6 mm thicknesses) was prepared using 2 wt% polyvinyl alcohol (PVA) as a binder. These pellets were sintered at 1200 °C for 6 h and further used for different physiochemical characterizations such as XRD, SEM, TEM, EDX, and AFM.

### Characterization

The crystalline phase of sintered BHNF ceramic was identified by X-ray diffraction analysis (Rigakuminiflex 600, Japan) employing  $\text{Cu-K}\alpha$  radiation ( $\lambda = 1.54 \text{ \AA}$ ). The microstructure and elemental analysis of the BHNF were observed by scanning electron microscope (ZEISS, model EVO-18 research; Germany) and energy-dispersive X-ray spectroscopy (EDX, Oxford instrument; USA), respectively. Transmission electron microscope (TEM, FEI TECANI  $\text{G}^2$  20 TWIN; USA) was used for particle size determination. The surface morphology was examined by atomic force microscopy (NTEGRA Prima, Germany). The ferroelectric properties of the sintered BHNF ceramic were measured by ferroelectric tracer (Automatic P-E loop tracer, Marin India). Magnetic behavior and temperature dependence zero field cooled ( $M^{\text{ZFC}}$ ), field cooled ( $M^{\text{FC}}$ ) over

a temperature range 5–300 K and applied a magnetic field 2 T ( $\pm 2$  T) were recorded by Quantum Design MPMS-3 and SQUID VSM dc magnetometer respectively. The frequency and temperature dependence dielectric of silver coated pellets were carried out using LCR meter (PSM 1735, NumetriQ 4th Ltd. U.K.).

## Results and discussion

### Crystallinity and microstructural studies

XRD patterns of BHNF ceramic calcined at 800 °C for 8 h and sintered at 1200 °C for 6 h, respectively are shown in Fig. 1. The single phase formation of the hexagonal BHNF ceramic is confirmed at 1200 °C for heating up to 6 h. The diffraction patterns are matched with JCPDS card no. (78–0132) along with small impurity of  $\text{Fe}_2\text{O}_3$  (JCPDS card no. 89–7047). The average crystallite size is calculated by the Debye–Scherrer formula [23].

$$D = k\lambda/\beta \cos\theta \quad (1)$$

where,  $D$  is the crystallite size,  $k$  is a constant value taken as 0.96,  $\lambda$  is the wavelength of X-ray,  $\theta$  is the Bragg angle of peaks, and  $\beta$  represents full width at half maxima (FWHM). The average crystallite size of BHNF ceramic is found to be  $30 \pm 5$  nm. To understand the functional group of BHNF ceramic, FTIR spectrum recorded in the frequency range 500–2000  $\text{cm}^{-1}$  is shown in Fig. 2. There are two absorption peaks observed in the frequency range in between 500 and 800  $\text{cm}^{-1}$  which is associated with Fe–O stretching vibration bands due

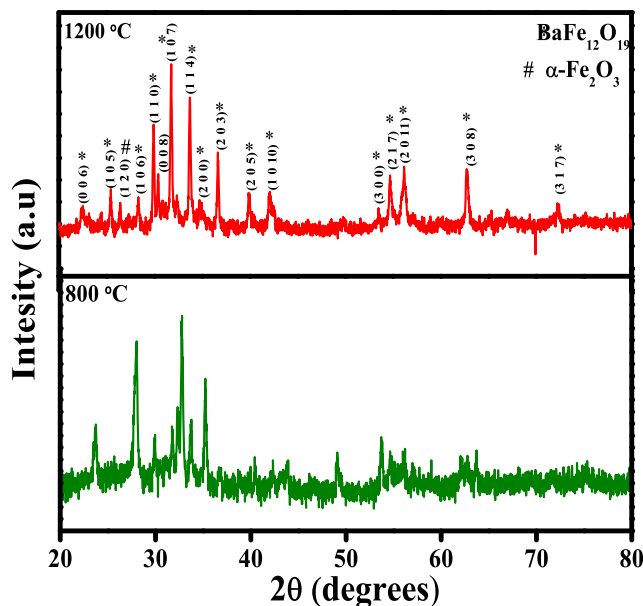
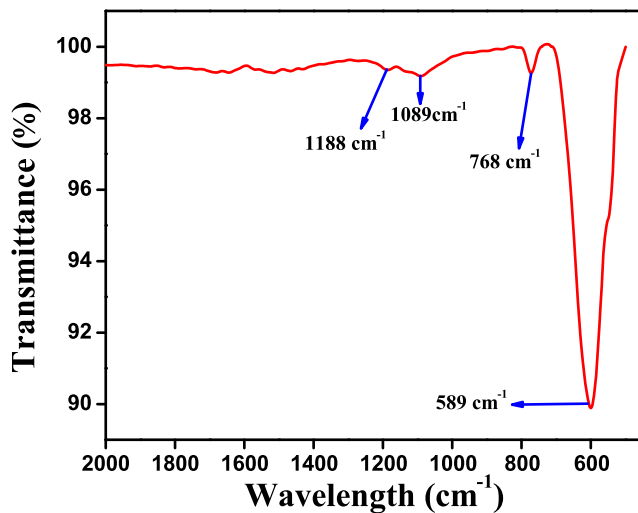


Fig. 1 Shows XRD diffraction patterns of  $\text{BaFe}_{11.7}\text{Ni}_{0.3}\text{O}_{19}$  ceramic a calcined at 800 °C for 8 h b sintered at 1200 °C for 6 h



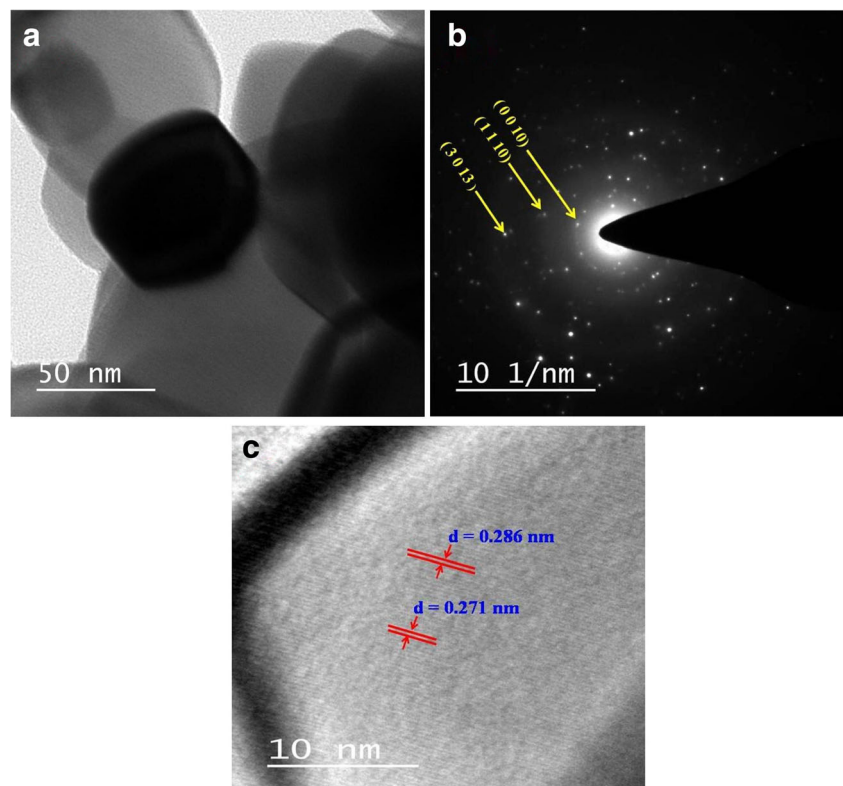
**Fig. 2** FTIR spectra of  $\text{BaFe}_{11.7}\text{Ni}_{0.3}\text{O}_{19}$  ceramic sintered at  $1200\text{ }^{\circ}\text{C}$  for 6 h

to octahedral and tetrahedral sites [24]. The absorption peaks are attributable to metal–oxygen–metal bonds such as Fe–O–Fe bonds, are also found in the frequency ranges  $1100\text{--}1500\text{ cm}^{-1}$ . The observed absorption peaks confirm the formation of hexaferrite [8].

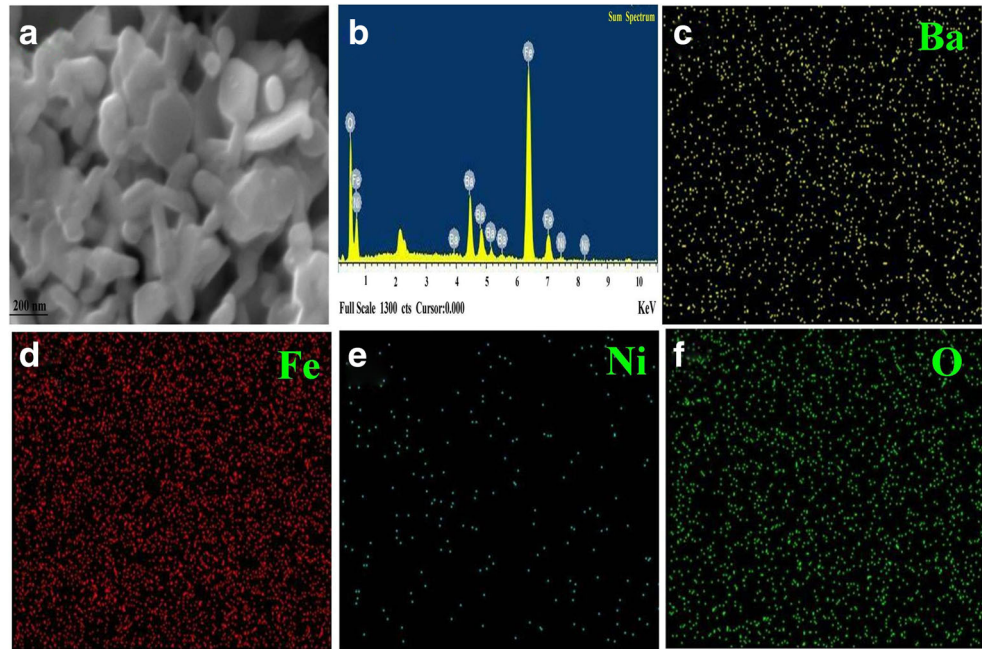
The bright field TEM images of BHNf ceramic sintered at  $1200\text{ }^{\circ}\text{C}$  for 6 h is shown in Fig. 3a, which

reveals the presence of hexagonal and round shape particles of size  $81 \pm 10\text{ nm}$ . With the help of SAED patterns, the zone axis containing  $(3\ 0\ 13)$ ,  $(1\ 1\ 10)$ , and  $(1\ 1\ 10)$ ,  $(0\ 0\ 10)$  planes are calculated to be  $(\bar{1}\bar{3}\ \bar{1}\bar{7}\ 3)$  and  $(10\ \bar{1}\bar{0}\ 0)$ , respectively as shown in Fig. 3b. Figure 3c shows the high resolution (HR-TEM) image showing the corresponding planes  $(1\ 0\ 7)$  and  $(1\ 1\ 2)$  having d-spacing  $0.271$  and  $0.286\text{ nm}$ , respectively. All the indexed planes obtained by HR-TEM and SAED pattern are in good agreement with XRD results which confirmed the hexagonal nature of the material. Figure 4a shows hexagonal plate-like grains which are separated by well-defined grain boundaries. The average grain size is found to be  $620 \pm 10\text{ nm}$  which is larger than crystallite as well as particle size due to crystallites are formed by combination of few particles whereas grains by agglomeration. The pH of the solution also affected on the morphology of the particles of the barium hexaferrite as a result of which particle size will increase with the decrease of pH which can ensure to decrease within the rate of growth or increasing the nucleation rate of the particles. The EDX spectra of BHNf ceramic indicates the presence of Ba, Fe, Ni, and O elements as shown in Fig. 4b. The atomic percentages of Ba, Fe, Ni, and O are found to be 21.74, 53.79, 3.23, and 21.24, respectively. Figure 4c–f shows energy dispersive mapping which indicates the

**Fig. 3** **a** Bright field TEM image **b** selected area diffraction (SAED) pattern **c** High-resolution TEM image of  $\text{BaFe}_{11.7}\text{Ni}_{0.3}\text{O}_{19}$  ceramic



**Fig. 4** **a** SEM image, **b** EDX spectra, and **c–f** elemental mapping of each elements corresponding to  $\text{BaFe}_{11.7}\text{Ni}_{0.3}\text{O}_{19}$  ceramic



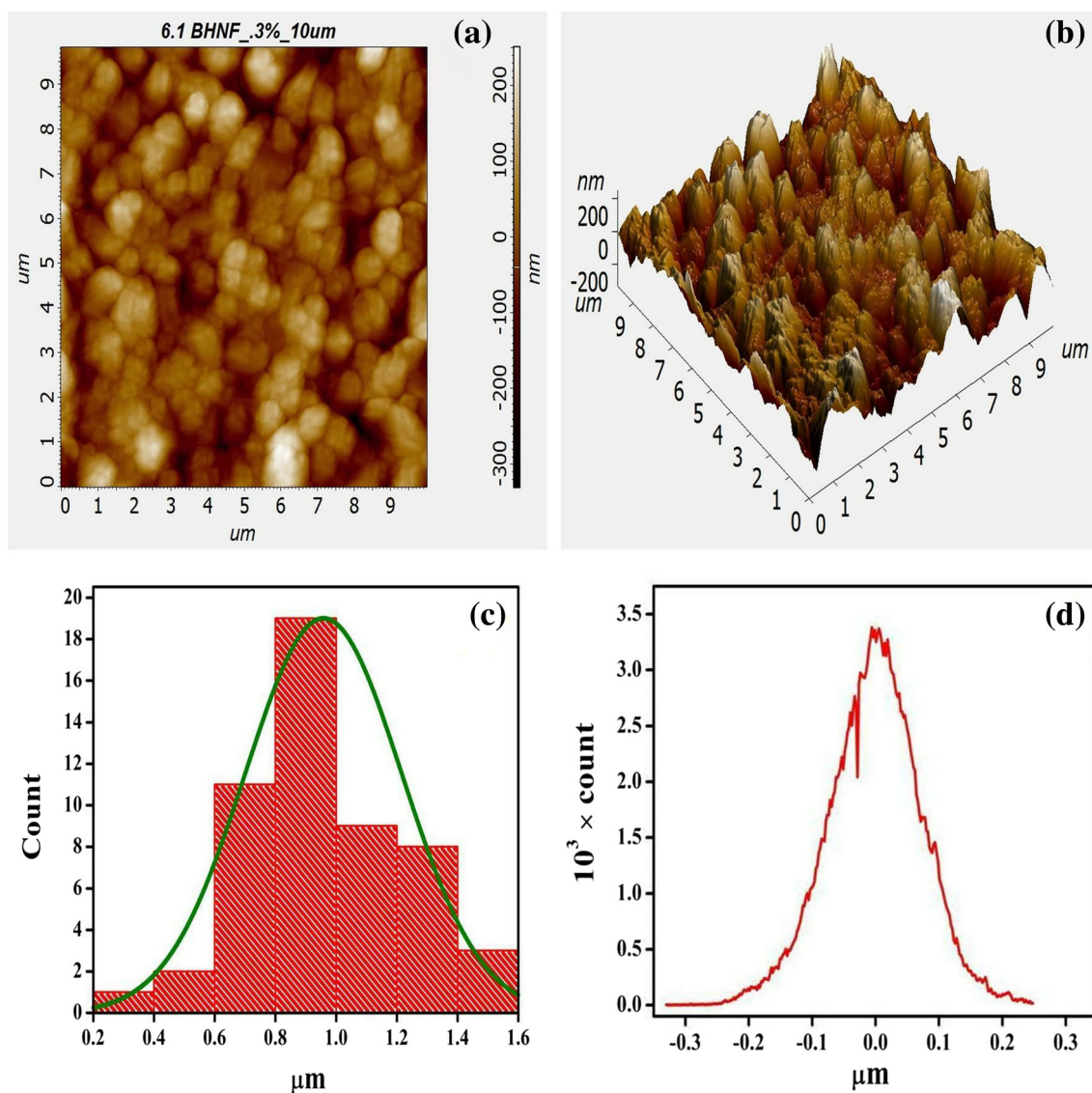
homogeneous distribution of the corresponding elements in BHNH ceramic.

Figure 5a represents the two-dimensional AFM image of BHNH ceramic sintered at 1200 °C for 6 h; the surface morphology reveals the homogenous distribution of grains is separated by grain boundaries. The three-dimensional image of BHNH ceramic is shown in Fig. 5b; after a thorough analysis of the grains, we observed the average roughness ( $R_a$ ), root mean square roughness ( $R_q$ ), maximum area peak height ( $S_p$ ), and maximum peak-valley depth are 57 nm, 73 nm, 250 nm, and 329 nm, respectively. Figure 5c shows the histogram of grain size which indicates that the majority of grains are in the range of 800–1000 nm, whereas the average grain size is found to be 630 nm which is supported by SEM. Figure 5d shows peak distribution curve for grain roughness of BHNH ceramic [25]. The polarization versus electric field (P–E) hysteresis loop of BHNH ceramic measured at room temperature is shown in Fig. 6 which indicates the absence of saturation in the P–E curve. The measured values of remnant polarization ( $P_r$ ) and coercivity ( $E_c$ ) are obtained from the curve and found to be  $0.0125 \mu\text{C}/\text{cm}^2$  and 3.46 kV/cm. The ferroelectricity of material is because of the distinction in ionic radii of host permittivity and impurity of cations.  $\text{Ni}^{+2}$  cations reduce the off-center effect of  $\text{FeO}_6$  octahedra and also suppress the oxygen vacancy which enhanced the ferroelectric behavior. The absence of saturation polarization ( $P_s$ ) is explained due to the lossy capacitor nature of these materials [26, 27].

## Magnetic studies

The variation of magnetization with temperature at constant applied magnetic field of 100 Oe of the BHNH ceramic is shown in Fig. 7a. The observed magnetization (ZFC and FC) increases gradually with decrease in temperature. The ZFC curve shows a hump at 172 K which indicates blocking temperature ( $T_b$ ) as shown in inset Fig. 7a. At this temperature, most of the spins are blocked in the material. The observed value of  $T_b$  is considerably low in comparison to the previously reported value [28]. This low value of  $T_b$  is favorable for biomedical applications in magnetic hyperthermia. Figure 7b represents the M–H hysteresis loop of BHNH ceramic at 5 K and 300 K recorded at  $\pm 2$  T. The saturation magnetization ( $M_s$ ) and remnant magnetization ( $M_r$ ) decreases with increasing temperature. This inverse behavior of magnetization with temperature is due to the existence of spin canting and superexchange interaction of  $\text{Fe}^{3+}\text{--O}^{2-}\text{--Fe}^{3+}$  on the substitution of  $\text{Ni}^{+2}$  ions in  $\text{BaFe}_{12}\text{O}_{19}$  ceramic [29]. The lower values of coercivity ( $H_c$ ) of the BHNH ceramic as compared with barium hexaferrite is due to the presence of unsymmetrical hexagonal particles as reported in earlier [30]. The decrease in coercivity of nickel-doped barium hexaferrite may be due to reduction in rate of the magneto-crystalline anisotropy ( $K$ ) with increasing the concentration of Ni [31]. The low values of “K” enhance the conductivity of the materials which is useful for sensor, transducers, and catalysts. The saturation magnetization is found to be 52.860 and 37.135 emu/g for 5 K and 300 K, respectively. The squareness ratio ( $M_r/M_s$ ) values for BHNH ceramic are near to 0.5 at room





**Fig. 5** AFM images of  $\text{BaFe}_{11.7}\text{Ni}_{0.3}\text{O}_{19}$  ceramic **a** two dimensional image for grain boundary, **b** three dimensional for surface roughness, **c** bar diagram for particle size distribution, and **d** depth histogram graph

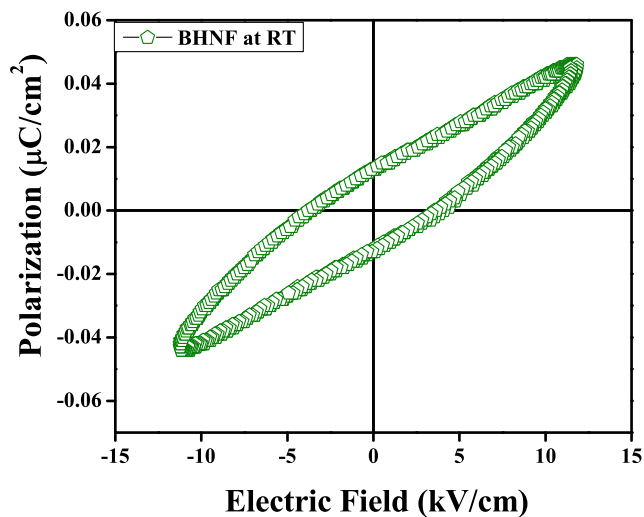
temperature. The lower value of  $M_r/M_s$  suggests that BHNF ceramic having a single domain, noninteracting particles which are randomly oriented, exhibit hard magnetic properties [32]. The magnetic parameters for BHNF ceramic obtained from hysteresis loop at two temperatures are mentioned in Table 1.

### Dielectric studies

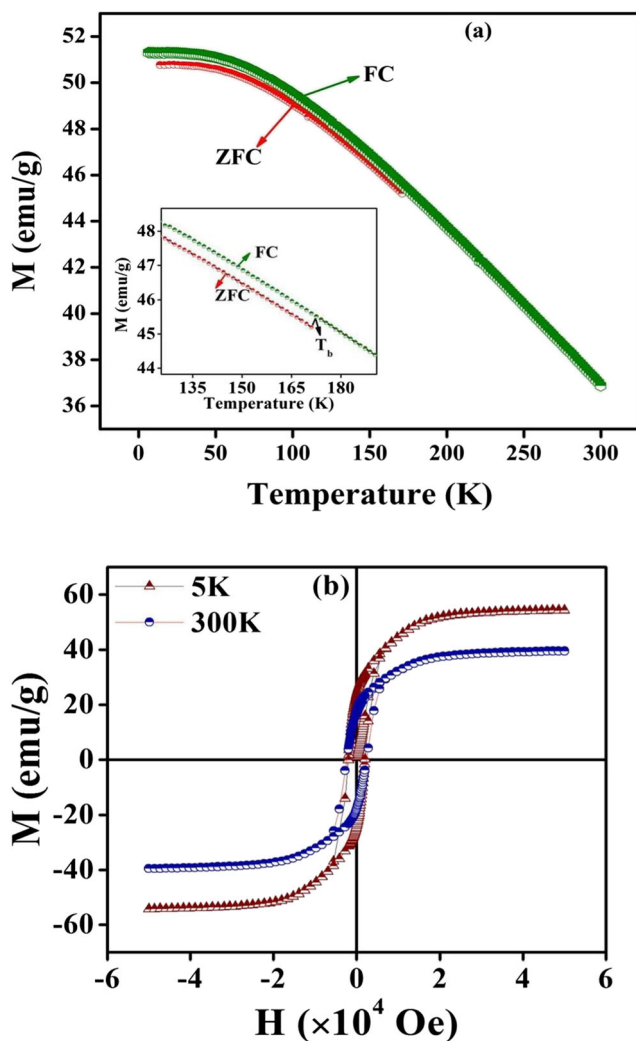
Figure 8 shows the dielectric constant ( $\epsilon'$ ) and dielectric loss ( $\tan \delta$ ) with the temperature at few selected frequencies. The relaxation peaks are observed at all measured frequencies and shifted towards higher temperature as shown in Fig. 8a, which may be due to the presence of oxygen vacancies in BHNF ceramic. The value of dielectric constant is found to be  $1.5 \times 10^4$  at 100 Hz [33]. Figure 8b shows temperature

dependent dielectric loss at few selected frequencies. The observed dielectric loss remains constant from the temperature (300–405 K) and increases with increasing temperature. This increment in  $\tan \delta$  with higher temperature and lower frequencies may be due to interfacial charge polarization at phase boundaries. The values of  $\tan \delta$  at room temperature are found to be 0.87 for 100 Hz, 1 kHz, 10 kHz, and 100 kHz, respectively.

The effect of frequency on the dielectric constant ( $\epsilon'$ ) and dielectric loss ( $\tan \delta$ ) is measured in the temperature range 303–423 K at an interval of 60 degrees are shown in Fig. 9. It is observed from Fig. 9a that the value of dielectric constant decreases with increasing frequency and is almost constant towards higher frequency, due to reversibility of electric field which takes place so rapidly. The value of  $\epsilon'$  is found to be  $1.7 \times 10^4$  at 423 K and 100 Hz. The higher value of dielectric



**Fig. 6** The polarization versus electric field (P–E) hysteresis loop of  $\text{BaFe}_{11.7}\text{Ni}_{0.3}\text{O}_{19}$  ceramic at room temperature



**Fig. 7** **a** The variation of magnetization with temperature at constant applied magnetic field of 100 Oe, **b** M–H hysteresis loop recorded at  $\pm 2$  T of the BHNf ceramic

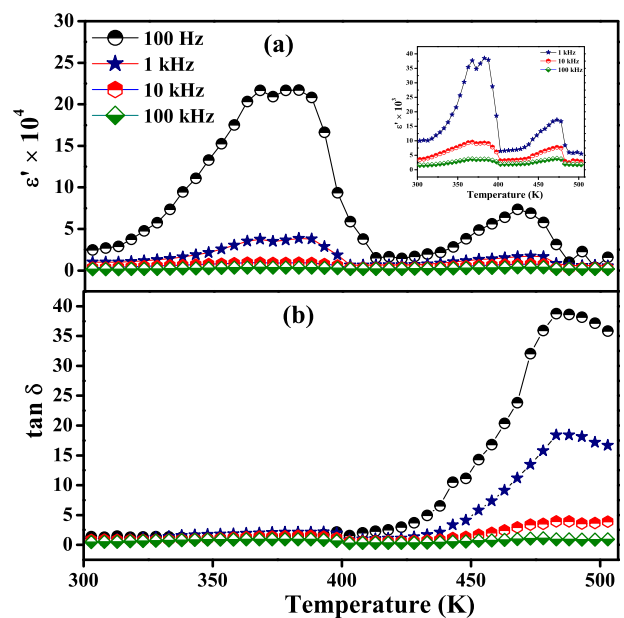
**Table 1** Magnetic parameters of  $\text{BaFe}_{11.7}\text{Ni}_{0.3}\text{O}_{19}$  (BHNf) ceramic at different temperature

T (K)	$M_s$ (emu/g)	$M_r$ (emu/g)	$M_r/M_s$	$H_c$ (A/m)
5	52.868	25.258	0.477	$1.55 \times 10^5$
300	37.135	18.233	0.491	$1.83 \times 10^5$

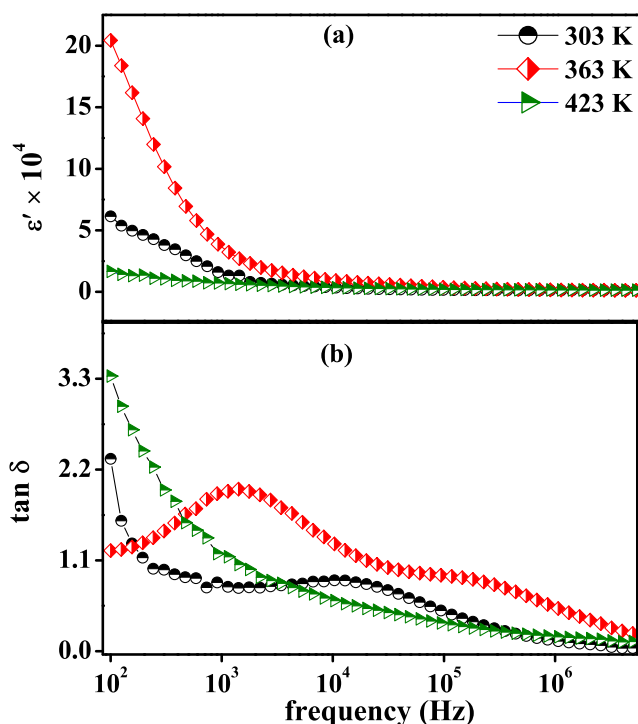
constant at lower frequency may be due to the interfacial charge polarization between semiconducting grains as well as insulating grain boundaries which is supported by internal barrier layer capacitance (IBLC) mechanism [34–36]. The temperature dependent dielectric loss ( $\tan \delta$ ) with frequency of BHNf ceramic is shown in Fig. 9b. The values of  $\tan \delta$  increase with the increase of temperature and are almost independent in higher frequency. The Debye-type relaxation peaks are occurring in low-frequency side indicates thermally active relaxation behavior which is suppressed at higher frequency  $10^5$  Hz [37–39]. The value of  $\tan \delta$  is found to be 3.3 at 423 K and 100 Hz.

## Conclusion

BHNf ceramic nanostructured materials have been successfully synthesized by a chemical route. XRD confirmed the formation of phase of BHNf at 1200 °C with a minor phase of  $\text{Fe}_2\text{O}_3$ . BHNf ceramic shows excellent dielectric and magnetic properties over a wide range of temperature. The value of dielectric constant ( $\epsilon'$ ) is found to be  $1.7 \times 10^4$  at 423 K and



**Fig. 8** **a** Dielectric constant ( $\epsilon'$ ) and **b** dielectric loss ( $\tan \delta$ ) as a function of temperature at few selected frequencies



**Fig. 9** Frequency dependent **a** dielectric constant ( $\epsilon'$ ) and **b** dielectric loss ( $\tan \delta$ ) at few selected temperatures

100 Hz. Ferromagnetic to paramagnetic transitions observed on increasing temperature. The less squareness ( $M_r/M_s$ ) ratio indicates the presence of monodomain crystal structure and justifies its suitability in various applications such as high magnetic recording media and permanent magnet.

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