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REVIEWS

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Semiconducting quantum dots: Modification and applications in biomedical science

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ABSTRACT Semiconductor quantum dots (SQDs) have received much attention due to their high quantum yield (QY), tunable emission spectrum, and excellent photostability. These unique optical properties endow SQDs with excellent biomedical application prospects, including biomedical imaging, drug delivery, clinical diagnosis, photodynamic therapy, DNA hybridization, and RNA profiling. This review introduces the classification of QDs and provides a brief description of the characteristics of QDs under each classification. Taking the type II B-VI A QDs as an example, inorganic and organic modification methods, and the corresponding advantages and disadvantages are summarized and discussed. Controlled modification approaches make them exhibit different functions in the bioimaging and drug delivery fields. The typical or classic instances are also listed to present the highlights of the applications of SQDs in the biomedical field. Based on these, this review raises a variety of possible challenges and perspectives of SQDs in biomedical applications in the future.

Keywords: semiconducting quantum dots, modification, imaging, drug delivery, biomaterials

INTRODUCTION

Semiconductor quantum dots (SQDs) have attracted great attention in recent decades, due to their unique optoelectronic properties [1], such as strong absorption, size-dependent photo-luminescent emission, high quantum yield (QY), and high optical stability. Different types of QDs exhibit functional characteristics in a variety of fields, including sensing [2], optoelectronic devices [3], biomedicine [4], and energy [5].

The exploration of SQDs has always been full of chal-

lenges. For example, most of the early QDs investigated were oil-soluble [6] and easy to agglomerate after drying, resulting in fluorescence quenching. Thus, there are considerable application difficulties in the biological fields. With the advent of water-soluble QDs [7], QDs could exhibit stable illumination in biological samples. Although traditional SQDs have excellent optical properties, the biosafety problems due to the presence of heavy metals such as cadmium containing QDs, cannot be ignored [8]. The emergence of QDs with high biocompatibility and safety, such as carbon QDs (CQDs), graphene QDs, silicon QDs [9], and black phosphorus QDs (BPQDs) [10], has accelerated the application of QDs in the biomedical field. QDs can simultaneously achieve diversified applications in the same area, such as BPQDsbased [11,12] multifunctional nanoplatforms for bioimaging, photothermal therapy, and photodynamic therapy. However, although technical breakthroughs in the research of QDs have been widely demonstrated, there are still many gaps in basic research. A variety of intrinsic problems, such as the luminescence mechanism and the surface state of QDs, are still inconclusive. These problems directly affect the precise modification, controllable modification, service mechanism, and performance improvement of QDs. Consequently, many QDsrelevant studies have achieved unique phenomena, but they were unable to provide a definitive theoretical explanation for the experimental results.

Although there are many unsolvable uncertainties in the research of QDs, they can still have more application values based on existing investigations. We provide a brief summary of the characteristics and application directions of different types of QDs in the section of

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CLASSIFICATION AND APPLICATION. Among these QDs, type II B-VI A QDs were selected as detailed introduction objects in the modification approaches (section of MODIFICATIONS). Based on the understanding of current modification methods for QDs, the section of BIOMEDICAL APPLICATIONS reviews the application of functionalized SQDs in bioimaging and drug delivery. The highlights and existing problems in related research are summarized and the prospects are presented in each section.

CLASSIFICATION AND APPLICATION

The concept of QDs was put forth in the 1990s [13]. In general, QDs are defined as nanocrystals with a particle size of 10 nm in three space dimensions. Strictly speaking, QDs are a kind of nanocrystals less than or equal to their own exciton Bohr radius in three space dimensions. According to the elemental composition, QDs can be classified as shown in Table 1.

ZnTe [14], ZnSe [15], ZnS [16], ZnO [17], CdTe [18], CdSe [19], CdS [20], HgTe [21], HgSe [22], and HgS [23] QDs are type II B-VI A QDs. The SQDs have attracted broad attention in the sensing [24], imaging [25], and biomedical [26] fields, because of their adjustable and robust fluorescence emission [27]. Chemical modification, doping [28], and core/shell technology [29] for SQDs have been developed and further adapted to practical applications. However, the biotoxicity, solubility, and light scintillation properties have been still criticized.

Type III A-V A QDs, especially InP QDs [30], usually present more stable optical properties than those of type II B-VI A QDs, because the characteristics of chemical bonds within these materials mostly appear as covalent bonds. However, this type of QDs has to overcome the problem of low QY by coating other materials with a wider energy gap, such as in InP/ZnS [31] and InP/ZnS QDs [32]. Type IV A-VI A QDs including PbS [33] and PbSe [34] QDs have smaller energy gaps. For this reason, their fluorescent emission region is mainly located in the near-infrared and mid-infrared spectral ranges. This kind of QDs can make use the most of visible light *via* multiple exciton effects, which are usually applied in the solar cell field [35].

In 2004, CQDs were officially named and reported to be intensely photoluminescent both in solution and solid *via* simple surface passivation [36]. Since then, CQDs are well known as the type IV A QDs, because of their excellent physicochemical properties, fluorescence stability, and biocompatibility. Graphene QDs of this type date back to the year of 2008 [37]. Graphene and graphene-

Table 1 The classification of QDs

Туре	Example
I B-VI A	Cu ₂ S
I B-VII A	AgBr
II B-VI A	ZnTe, ZnSe, ZnS, ZnO, CdS, CdSe, CdTe, HgTe, HgSe, HgS
III A-V A	AlSb, AlAs, AlP, GaSb, GaAs ,InGaAs, InAs, InP, InN
IV A-VI A	PbS, PbSe, PbTe
IV A	C, Si, Graphene
V A	Black P
I B-III A-VI A	CuInS ₂ , CuInSe ₂ , AgInS ₂
Pdots	Semiconducting polymer dots
TMDCs	TiS ₂ , TiSe ₂ , TaS ₂ , MoS ₂ , MoSe ₂ , WS ₂ , WSe ₂ , ReS ₂
MXene	$M_{n+1}X_nTz$ (M = transition metals; X = C and/or N; n = 1-3; Tz = F ⁻ , O ²⁻ and OH ⁻ etc.)
Perovskite	$MPbX_3$ (M = Cs or CH_3NH_3 ; X = Cl, Br or I)

based composite materials [38] have been widely used in the field of optics [39] and devices [40] due to their unique properties. However, the fluorescence lifetime of QDs is usually very short, and the fluorescent signal is easily interfered by biological background fluorescence.

BPQDs have emerged in recent years [41,42]. We presented a gentle top-down method to prepare BPQDs in solution in 2015 [43]. In the following year, a solvothermal method to synthesize BPQDs was reported, and the nonlinear optical properties of BPQDs were revealed for the first time [44]. BPQDs have been successfully applied to ultrafast laser technology [45,46]. Other materials, such as antimonene [47,48] and bismuthene [48,49], have also attracted the attention of researchers.

Ternary I B-III A-VI A QDs, such as CuInS₂, CuInSe, and AgInS₂ QDs, usually possess tunable bandgaps. Their fluorescence can be continuously adjusted from the red to the near-infrared region [50]. Due to their outstanding optical properties, QDs are applied in many fields, including biosensing, biomedical imaging [51], solar cells [52], and photovoltaic devices [53]. More importantly, the long fluorescence lifetime and low biological toxicity provide an absolute advantage for using QDs in the biological systems. After being further synthesized into core/shell structures, QDs can realize *in vivo* whole animal fluorescence imaging [54].

Semiconductor polymer QDs (Pdots) generally refer to a portion of semiconductor polymer nanoparticles that is equivalent in size (<20 to 30 nm) to inorganic semiconductor QDs. Pdots with advantages such as small size, high brightness, fast radiation transition rate, good light stability, and low biological toxicity, are very suitable for bioimaging [55], sensing [56], and oncotherapy [57].

Transition metal dichalcogenide (TMDC) QDs are also a class of materials with multifunctional properties. One such example is molybdenum disulfide (MoS_2) QDs. The QDs are a type of transition metal disulfide nanomaterial with a unique two-dimensional structure similar to graphene. It has an adjustable emission spectrum, high QY, and is easily functionalized. The good biocompatibility makes TMDC QDs widely used as catalysts [58], fluorescent probes [59], biosensors [60] and tumor treatment platforms [61].

There is an important emerging material named MXene dots derived from MXene sheets. Compared with sheets, the advantage of QDs is their ultra-small size and fluorescence properties, which make them promising candidate materials in biomedical [62,63], bioimaging [64], biosensing [65], catalysis, [66] and optical [67,68] fields. MXene QDs are bio-friendly materials rich in surface functional groups. They may be assembled with other functional materials to produce applied materials.

In recent years, inorganic perovskite QDs (IPQDs) have gradually emerged in the fields of light-emitting diodes (LED) [69], photo-detectors [70], QD lasers [71], and solar cells [72]. The room-temperature synthesis, superior photo-luminescence, underlying origins, and potentials in lighting and displays have been reported [73]. For better efficiency and stability, coated IPQDs have been prepared and used in white LEDs [74].

We did not introduce all QDs, such as Cu₂S [75] and AgBr [76] QDs, listed in Table 1. As summarized above, different types of QDs have unique highlights and advantages. Once QDs are put into practical applications, whether for forming core/shell materials or doping with other elements, further optimization of the QDs is inevitable. Among various applications based on QDs, biomedicine is considered the most directly related research field to humans. If QDs are applied in biomedical fields, fluorescence lifetime, optical stability in aqueous solution and biocompatibility of the QDs would become the primary considerations.

MODIFICATIONS

In terms of optical performance, it is necessary to control the defect states and reduce the carrier trap to significantly increase luminous efficiency and stability. If SQDs are to be employed for biological applications, more requirements, including water solubility, biocompatibility, non-specific cellular uptake [77], active targeting [78], and long fluorescence lifetime, need to be considered. Consequently, surface modification becomes very compulsory to optimize the performance of SQDs. There is an instance [79] (Fig. 1) that shows the effect of simple surface modification on the stability of ZnS-coated CdSe QDs. QDs modified with –OH show improvement in stability and dispersion within hypertonic conditions. However, all QDs are stable in nonelectrolyte solutions. Under weak alkaline conditions, all modified QDs remain stable for 30 min. However, only amine groups-modified QDs are stable in acidic conditions. The results provide references for further surface modification of QDs for applications in biomedical fields.

Organic modification

The organic modification of SQDs refers to organic ligands [80], organic polymers [81], proteins, and other



Figure 1 Physical and chemical stabilities of different modified QDs in buffers. The samples were excited at a 365-nm wavelength. The QDs from 1 to 5 are modified by -COOH, -OH/COOH, -OH, $-NH_2/OH$ and $-NH_2$, respectively. The symbols, +, – and ±, stand for aggregated, not aggregated and partially aggregated, respectively. Reprinted with permission from Ref. [79], Copyright 2004, American Chemical Society.

organics modifiers. Organic ligand modification can optimize some characteristics of SQDs. The trapping states on the surface of SQDs, including electron traps and hole traps, cannot be simultaneously passivated by organic ligands. Besides, water molecules, oxygen and illumination can easily cause erosion and degradation of the organic ligands on the surface of SQDs. Other organic modification approaches have other problems. After surface modification with tert-butyl-N-(2-mercaptoethyl)-carbamate, CdSe QDs showed a strong fluorescent emission with several hundred nanoseconds of fluorescence lifetimes. Unfortunately, these SQDs can only be used in the aerated methanolic solution [82]. It is reported that the surface properties decide the cytotoxicity of QDs [83]. Modification of CdSe QDs with sodium dodecyl sulfate decreased the cytotoxicity, which became useless as the incubation time and concentrations of the modifier increased. Water-soluble CdSe ODs modified with polymer exhibited strong yellow fluorescence emission that was observed both in vivo and in vitro. The QDs remained stable in aqueous solution over 30 days, suggesting the potential for bio-labeling and diagnostic purposes [84]. Even so, the average size of the modified SQDs was around 150 nm, which is too large to enter cells via endocytosis. Modified SQDs depend on active transport to reach the inside of cells. Cell recognition is inevitable, which reduces the efficiency of the SQDs. Besides, with increase in particle size, cell viability decreases [85].

Inorganic modification

Doping with metal ions and the formation of a core/shell structure can be classified as inorganic modifications. For optimal performance, elements such as Co²⁺, Ni²⁺ [86], Mn^{2+} [87], and Cu^{2+} [88] are doped into SQDs to improve their intrinsic properties. CdSe/ZnS QDs were synthesized and characterized as early as 1997 [89]. After that, core/shell SQDs emerged as a new trend. The core/shell SQDs are divided into three categories. In type-I QDs, such as CdSe/ZnS QDs [90], the energy levels of the conduction band and the valence band of the shell materials are both higher than that of the core materials. The type-I approach is adopted to increase stability against photobleaching and fluorescence QY via improving surface passivation [91]. In type II, the energy level of the conduction or valence band of the shell materials is greater than that of the core materials, for instance, in CdTe/CdSe [92] and CdSe/ZnTe QDs [93]. The emission wavelength of SQDs will show a substantial redshift, and it is possible to shift to the near-infrared region by adjusting the thickness of the shell. In reverse type I, the bandgap of the core is greater than that of the shell, for example, in CdS/CdSe [94] and ZnSe/CdSe QDs [95]. Typically, the second and third types of core/shell SQDs need to be coated again to obtain a broader bandgap of the shell, and multiple shell structures will improve the optical properties, including enhancing the photostability, increasing the luminous efficiency, and restraining photobleaching.

The above advantages of core/shell SQDs mainly focus on the improvements in optical performance. The core/ shell structure also facilitates the applications of SQDs in biological systems. The concentrations of Cd²⁺ on the surface of mercaptopropionic acid (MPA)-coated CdS and CdSe/ZnS QDs were further investigated to compare the differences in cytotoxicity effects between organic and core/shell modifications [96]. The CdSe/ZnS QDs started to release Cd²⁺, causing the poisoning of fibroblasts cells at a concentration of $5.9\pm1.3 \mu mol L^{-1}$. However, the concentration of mercaptopropionic acid-coated CdS needed to release Cd^{2+} was as low as $0.65\pm1.2 \ \mu mol \ L^{-1}$. The core/shell SQDs have certain advantages in reducing biological toxicity. With the protection of the shell, SQDs become relatively safe and can be used in the biological systems. The CdTe/ZnSe core/shell QDs have been prepared to directly detect DNAs in the extracts of human cancer and normal cells. They were shown to be highly efficient even at an extremely low concentration of DNA [97]. To promote the practical applications, fast and large-scale preparation of core/shell QDs has also been studied. The "flash" synthesis method of CdSe/CdS core/ shell QDs has been successfully developed. No more than 3 min is needed to grow a thick shell up to 6.7 nm [98]. Water-soluble AgInSe₂/ZnS and CuInSe₂/ZnS core/shell QDs on a multigram scale have been prepared using thioglycolic acid and gelatin as the dual stabilizers in an electric pressure cooker as the reaction vessel [99]. Core/ shell SQDs can also be designed to reduce signal interference from bioluminescence via near-infrared emis-Water-soluble CdSeTe/ZnS core-shell sions. QDs protected by 2-(dimethylamino)ethanthiol were synthesized and incorporated into a chitosan film to form SQDs/chitosan films. Electrodes modified with SQDs/ chitosan films have been applied to detect concentrations of cholesterol in biological samples [100].

For application, SQDs need to be modified by many functional groups. Therefore, some surface modifications are considered to connect SQDs and functional groups. Fig. 2a shows water-soluble silica-coated CdSe/ZnS QDs [101]. The silanized QDs not only maintained the optical



Figure 2 Methods for attaching functional units to the surface of SQDs. (a) Silica-coating. Reprinted with permission from Ref. [101], Copyright 2001, American Chemical Society. (b) Coordination. Reprinted with permission from Ref. [102], Copyright 2016, American Chemical Society. (c) Esterification. Reprinted with permission from Ref. [103], Copyright 2017, Elsevier B.V. (c) Ligand exchange. Reprinted with permission from Ref. [104], Copyright 2019, Royal Society of Chemistry. (e) Phase transfer. Reprinted with permission from Ref. [105], Copyright 2016, Elsevier B.V.

characteristics of the original CdSe/ZnS cores, but their silica bead helped to introduce biological functional groups. Even in buffers that simulate physiological en-

vironments, silica-coated CdSe/ZnS QDs also displayed high stability. Coordination is also a simple method to connect functional groups. As shown in Fig. 2b, lanthanides have strong coordination ability, which makes them self-assemble to the surface of SQDs [102]. On this basis, other functional groups, such as amino acids and proteins, can be further modified on the surface of SQDs *via* coordination. In addition, esterification (Fig. 2c) [103], ligand exchange (Fig. 2d) [104], or phase transfer (Fig. 2e) [105] are also common approaches for the surface functional modification of SQDs.

Sometimes, different shells with a modified group [106] may be coated onto the core to achieve precise experimental purposes. An *N*-acetyl-*L*-cysteine (NAC)-capped CdTe/CdS/ZnS core/shell/shell QDs-based "OFF-ON" fluorescent biosensor was constructed to sensitively and specifically test *L*-ascorbic acid [107]. The detection limit could reach 1.8×10^{-9} mol L⁻¹ under optimum conditions. The approach was successfully applied to human urine samples, three synthetic samples, and vitamin C tablets.

Mainly classic biological modification approaches for SQDs were introduced above. These modifications make SQDs more compatible with the needs of biomedical applications. Moreover, lanthanide (Ln) is an ideal synergistic material. Ln ions have a long fluorescence lifetime, which is easy to distinguish from short-lived biological background fluorescence via time-resolved techniques [108]. The significant Stokes shift of Ln [109] can effectively improve the resolution of the detection signal. More importantly, there is an energy transfer between Ln ions and the original materials. Based on the energy transfer, the fluorescence intensity of Ln ions or original materials will be sharply increased. The combination of Ln ions and SQDs via Förster resonance energy transfer (FRET) is considered an exceptional tool for biological analysis [110]. The absorption of SQDs is strong and spectrally broadband, but their fluorescence lifetime is quite short. There is no broad absorption of Ln ions; however, the fluorescence lifetime of Ln ions can last up to milliseconds. After doping, the Ln-SQDs materials combine the optical advantages of Ln ions and SQDs, which is more conducive to the applications in bioimaging, biosensing, and other biomedical fields. The attraction is that both Ln ions and SQDs can be used as energy acceptors and energy donors. FRET can occur when the excited donor fluorophore and the ground-state acceptor fluorophore are in close proximity (generally 1-20 nm) [111]. FRET can occur not only from Ln ions to SQDs but also from SQDs to Ln ions. Eu³⁺ was incorporated into a CdS lattice via a simple methodology based on the direct precipitation of CdS nanocrystals with 1-thioglycerol as a capping molecule. A previous study revealed an effective energy transfer process from CdS to

Eu³⁺ and indicated that the energy transfer process involved surface localized states that could be mediated by a thermally activated exchange mechanism [112]. A facile sol-gel method was used to prepare Eu³⁺-doped ZnS na-nocrystals. Even though Eu³⁺ did not appear in the ZnS lattice, an energy transfer from ZnS to Eu³⁺ occurred [113]. It did not matter whether the Ln ions entered the QDs lattice or not, and the energy transfer phenomenon could occur when the Ln ions and QDs were spatially close. Ln ions working as the energy donors and CdSe/ ZnS QDs serving as energy acceptors have also been reported [114]. The energy transfer phenomenon can be observed by QD emission increases of 1000-fold with simultaneous Ln emission quenching. As introduced, the cooperation between Ln ions and SQDs is very flexible. Certain complementarities of Ln ions and SQDs in optical properties can be combined by FRET. The direction of energy transfer between the Ln ions and the QDs can be selected according to the application needs when designing functional materials.

In summary, water solubility and low biotoxicity are the primary factors considered in biomedical applications. Presently, the water solubility of SQDs can be solved by simple surface modification. The design of the core/shell structure of SQDs may reduce biotoxicity, but unfortunately, the reduction ratio is limited. Once the concentration of QDs is increased, the risk of metal leakage is inevitable. To ensure low biological toxicity, some surface modification is necessary to generate SQDs with certain selectivity and specificity. The modified SQDs showed excellent performances for in vitro experiments. New problems, however, must be addressed when SQDs are used in biological systems. The introduction of modifiers may increase biological toxicity and the particle sizes of the modified SQDs are usually very large, leading to further problems. SQDs with high selectivity and specificity may have relatively high biotoxicity. With larger QD particle sizes, the survival rate of cells decreases. These limit the application of SQDs in the field of biosensing and bioimaging to a certain extent. Without considering the toxicity of the modifier and the particle size of the SQDs, there is good tissue-penetrating ability of near-infrared SQDs. Ln-modified SQDs can also reduce the interference of bioluminescence via a timeresolved fluorescence technique. Additionally, the rapid and large-scale synthesis of SQDs can gradually push SQD technology from scientific research to practical applications. In general, SQDs can meet the needs of biomedical applications within a specific range.

Some studies may need to be considered. The con-

centration of dopants in the as-prepared SQDs should be precisely controlled, and the particle sizes of SQDs need to be minimized to ensure low toxicity. The fluorescence intermittency of SQDs must be overcome if the SQDs are used for single-molecule tracking. The investigation of the biological metabolism of SQDs cannot be ignored, especially in different cells, tissues, and even organs.

BIOMEDICAL APPLICATIONS

Biological imaging

Table 2 [115–131] summarizes the SQD materials for bioimaging. One of the unique properties of SQDs is tunable emission wavelengths [132]. Therefore, some reports have summarized the optimal emission peaks for

Table 2 Use of SQDs for biological imaging

SQDs, while others have reported emission ranges of SQDs. The particle sizes referred to in Table 2 are the results derived from transmission electron microscopy images, and they are not hydrodynamic diameters. The particle sizes and hydrodynamic diameters are quite different. For example, in S12, the particle size and hydrodynamic diameter are about 5.6 and 10 nm in ethanol, respectively. Also, even the same QDs have different hydrodynamic diameters in various solvents. Still, for S12, the hydrodynamic diameters of the QDs in ethanol and water are 10 and 49.2 nm, respectively. Strictly speaking, the QYs in Table 2 are not comparable because the test methods, instruments, and reference materials are different from each other. However, they are still listed to provide a general understanding of QDs.

No.	QDs	Emission wavelength (nm)	Size (nm)	Object	Remarks	QY	Refs.
S1	CdTe	510-670	1.4±0.30 3.3±0.30	HepG2 ^a HO-8910PM ^b	Glutathione-capped Folic acid-conjugated	42.0%	[115]
S2	CdTe	530-630	2.7	SGC-7901 ^c	Ribonuclease-A-conjugated	75.6%	[116]
S3	Cu ⁺ -doped CdS	586	3.47 ± 0.44	KB^{d}	-	21.9%	[117]
S4	CdTe/CdSe	700-820	5-8	MCF-7 ^e Mouse	Multivalent polymer ligands	66.0%	[118]
\$5	CdSe/CdS	610	6	Bel-7404 ^f Tissue	-	30.0%	[119]
S6	CdS/ZnS	500	41.2±5.5	Hela ^g /HepG2/A549 ^h	Polyethyleneimine-Coated	10.3%	[120]
S7	CdSe/ZnS	538/657	4.8	MGC803 ⁱ	Tumor targeted Dual antibody-conjugated	_	[121]
S8	CdTe/ZnTe	520-700	3	RAW264.7 ^j Panc-1 ^k	Anti-claudin 4 target Histological studies	52.0%	[122]
S9	ZnSe/ZnS	382	3	RAW264.7 SCG7901 ¹	-	70.6%	[123]
S10	Mn-doped ZnS	600	5.3±1.0	MCF-7	Folic acid-chitosan stabilized	-	[124]
S11	ZnS:Mn/ZnS	595	$4.4{\pm}0.7$	$T47D^m$	Folic acid-conjugated	22.1%	[125]
S12	AgInSe ₂ /ZnS	700-820	5.0-6.0	MCF-7 Mouse	-	40.0%	[126]
S13	ZnCuInS/ZnS	585-715	5.6±0.4	NIH-3T3 ⁿ Mouse	Hydrophilic ligand and PEG	26.0%	[127]
S14	Gd-doped CuInS/ZnS	670	3.0	U87 MG° Mouse	310 ns Tumor targeted Fluorescence/ magnetic resonance	_	[128]
S15	CdTe/CdSe/ZnSe	e 670	4.0 ± 0.8	Tissue	-	4.80%	[129]
S16	CdTeSe/CdZnS	700-800	5.8	Hela P4	PEG-phospholipid	25.0%-30.0%	[130]
S17	Gd-doped ZnO	550	4	HeLa	-	34.0%	[131]

a) HepG2: human hepatic carcinoma cell line; b) HO-8910PM: human ovarian cancer cell line; c) SGC-7901: human gastric cancer cells; d) KB: human oral epidermoid carcinoma cell; e) MCF-7: human breast adenocarcinoma cell line; f) Bel-7404: human hepatoma cell; g) Hela: human cancer of cervix cell line; h) A549: human lung carcinoma cell line; i) MGC803: gastric cancer cell line; j) RAW264.7: mice macrophages cell line; k) Panc-1: human pancreatic cancer cells; l) SCG7901: human gastric carcinoma cell line; m) T47D breast cancer cells; n) NIH-3T3: mouse embryonic fibroblast cell line; o) U87MG: malignant glioma cell.

The SQDs materials shown in Table 2 enable bioimaging in a wide variety of cells, tissues, and organs, and fluorescence emission of the SQDs from the visible to the near-infrared regions can be achieved. Glutathione-capped CdTe QDs (S1) possesses size-dependent emission ranging from 510 to 670 nm, and the fluorescence emission of ribonuclease-A-conjugated QDs can be adjusted from 530 to 630 nm. Thus, different modifications can directly affect the optical properties of QDs. S1, at sizes of 1.40±0.30 and 3.30±0.30 nm, displays 510 and 670 nm fluorescence emissions, respectively. These two QDs can be further conjugated with folic acid for targeted fluorescence imaging. HepG2 and HO-8910PM have been selected for imaging objects because they are known to express abnormally high levels of folate receptors on their cell surface. S7 not only has targeted imaging capabilities in vitro but can also realize targeted imaging in vivo. BRCAA1 monoclonal antibody and Her2 antibody were chosen to conjugate with the QDs. The BRCAA1 antigen is a specific protein for the intracellular epitope of histone deacetylase complex subunit SAP180, expressed in the cytoplasm of the breast cancer cell line MCF-7 and gastric cancer cell line MGC803 [133], and HER2 is confirmed to be overexpressed in gastric cancer and some other aggressive diseases [134]. As expected, the cytoplasm of MGC803 cells, where the BRCAA1 protein is mainly expressed, is evenly labeled by the red signals of BRCAA1 monoclonal antibody-modified QDs, and the membranes of MGC803 cells are labeled by the green signals of HER2 antibody-modified QDs. Similar [135] and other celltargeted imaging studies are receiving increased attention.

CdS QDs are used in both S3 and S6, which form metal doping and core-shell structure materials. Inevitably, the particle size of S6 is greater than that of S3. However, not all core-shell materials have larger sizes. For example, the particle sizes of S4 and S5 are almost the same as those of S3. The performance differences caused by the metal doping and core-shell modifications are still noticeable. The optimal emission peaks for S3 and S6 are 586 and 500 nm, respectively. Compared with S6, although S3 has no passivation shell, the QY still reaches 21.82%. As a core-shell material, S6 can easily be further functionalized with folic acid. It can be used not only as a sensitive and selective fluorescence probe for folate receptors, but also to realize the targeted imaging of Hela cells. Some metaldoped SQDs can also be biologically modified, like the corresponding core-shell SQDs in S10 and S11. S10 uses Mn-doped ZnS QDs, and S11 uses Mn-doped ZnS as the core and with a ZnS shell modification on the outside. Although the optimal emission peaks of the two SQDs are approximately the same, and each one could be folatefunctionalized for cell-targeted imaging, there are still differences in some properties. The ZnS shell provides effective passivation and elimination of surface defects on the ZnS:Mn core surface, which is demonstrated by the high QY (22%) of the core/shell SQDs compared with the ZnS:Mn QDs (2.7%).

Both S4 and S8 are core-shell structure QDs with CdTe as the core. However, different shell materials endow QDs with distinct optical properties. The fluorescence emissions of S4 are tunable between 700 and 820 nm, and the fluorescence emission range of S6 is from 520 to 700 nm. Similarly, when the shell material is the same, and the core material is different, the optical properties of the SQDs are also greatly different. As shown in S6, S7, S9, S12, and S13, their optimal fluorescence emission peaks are very distinct. Thus, the materials for the core and the shell can be chosen according to the actual wavelength requirements of the cell imaging.

Near-infrared light not only can be clearly distinguished from the background fluorescence of the biological sample itself but also has stronger tissue penetration [136], so the preparation and modification of near-infrared imaging SQDs materials have also become the focus of attention. CdTe, CdSe and other SQDs have been studied because of their long fluorescence emission wavelengths, although, the emission of CdSe-based QDs is only from 650 to 700 nm, which is still below the optimal near-infrared range of 700-900 nm for in vivo imaging. This does not prevent it from combining with other materials to form near-infrared emission SQDs, such as CdTe/CdSe core/shell QDs (S4), with emissions ranging from 700 to 820 nm. Some multi-element materials have also been designed as QDs for near-infrared cell imaging, such as AgInSe₂/ZnS (S12) and CdTeSe/CdZnS (S16) QDs. In addition to designing QDs as the core/shell structures mentioned above, the materials can also be designed as core/shell/shell structures. As shown in S15, the optimal emission peak position of CdTe/CdSe/ZnSe QDs is 670 nm which is obviously larger than that of the original CdTe core (530 nm), and the fluorescence emission can be tuned from 550 to 800 nm. If the mercaptoundecanoic acid amount is reduced in the synthesis, the emission peak will be further red-shifted for 57 nm.

Ln-doped SQDs for imaging are relatively special because imaging includes not only fluorescence imaging but also magnetic resonance imaging. Besides, the characteristic feature of Ln-doped QDs is their long lifetime; the lifetime of S14 can be as long as 310 ns. As shown in Fig. 3, S14 can perform fluorescence imaging and mag-



Figure 3 *In vivo* fluorescence imaging (a) and T_1 -weighted imaging (b) of U87 tumor-bearing mice after tail-vein injection of S14 at various time intervals (pre, 2, 4, 6 and 18 h). Reprinted with permission from Ref. [128], Copyright 2015, American Chemical Society.

netic resonance imaging in vivo. It is worth noting that the fluorescence signals nearly disappeared 18 h of postinjection, indicating that most of the SQDs could be gradually removed from the body. Compared with conventional contrast agents (Gd-DTPA), SQDs can realize stronger imaging intensity and relatively long circulation time in the body, which provides a longer imaging time window for the technicians. In addition, SQDs can present better permeability and retention effects than Gd-DTPA, although the Gd doped-ZnO QDs performed the same as S14. However, the nature of the host material itself causes its fluorescence emission range to be significantly different from that of S14. Although the emission peak position of S17 could be adjusted by the doping amount of Gd, the optimal emission peak position can only vary between 525 and 575 nm, far from 670 nm (S14). The magnetic resonance imaging resolution was also significantly lower than that of S14.

In addition to the animal cell imaging mentioned above, SQDs are also widely used in the field of microbial imaging [137]. Lin *et al.* [138] made *Staphylococcus aureus* work as a living substrate to prepare CdSe QDs with high brightness, strong light stability, and uniform brightness. The fluorescent bacterial probe still retained the ability to be recognized and phagocytosed by macrophages. This live bacterial fluorescent probe could be used as a multifunctional tool for biological research on *Staphylococcus aureus*. Mir *et al.* [139] prepared AgInS@ZnS QDs with a high fluorescence lifetime (> 300 ns) and high QY (49%) for stable imaging in *Candida albicans*. The QDs could also be applied as a carrier to deliver photosensitizers for cancer diagnosis and photodynamic therapy. Hong *et al.* [140] used the conjugation of 4-formylbenzoate modified CdSe QDs to 6-hydrazinonicotinate acetone hydrazone-modified viruses to achieve fluorescent labeling of the H9N2 virus. This labeling method not only made the labeling efficiency as high as 92%–93%, but also maximized the retention of viral infectivity. This labeling strategy can be applied to both enveloped and non-enveloped viruses.

In general, many types of SQDs can be used for imaging, and they can be metal-doped SQDs, core-shell SQDs, metal-doped core-shell SQDs, or other forms of SQDs. The overall fluorescence emission range covers visible to near-infrared light. The types of imaging objects include animal cells, tissues, organs, bacteria, fungi and viruses. Although the biotoxicity of SQDs has been criticized, it is undeniable that the excellent optical properties of SQDs make them exhibit satisfactory imaging capabilities in most cell samples. According to the specific or overexpressed substances in the nucleus and cytoplasm of some tumor cells, SQDs can be functionalized to achieve targeted imaging of cells or tissues. On this basis, according to the size-dependent effect of the SQDs, SQDs with different fluorescence emission wavelengths can be prepared, which can realize dual-light cell imaging of the same type SQDs by various targeted modifications. The SQDs achieve not only fluorescence imaging but also magnetic resonance imaging. Compared with existing commercial contrast agents, although many studies on SQDs need to be performed before clinical application, they show some absolute advantages in animal experiments. As summarized, SQDs have performed well in the imaging field. But many problems still need to be solved. The current research on SQDs used for imaging has mainly focused on material preparation, characterization, cytotoxicity testing, and general in vitro/in vivo imaging/ targeted imaging. Many aspects of QD imaging still need to be explored.

Next, the development of SQDs in the field of imaging should closely combine with research on the immunity and metabolism of cells, tissues, and organs. Based on this work, targeted imaging can be explored to guide actual clinical medicine. For example, targeted imaging techniques can be used to outline tumor boundaries for tumor resection in non-precise areas. After this technology is mature, tumor resection can be performed at precise sites. Some tumor surgery in specific, defined locations, including brain tumor resection, can damage normal brain cells if a small error occurs in the resection, which may cause irreversible lifelong bodily injury. However, it is difficult to accurately distinguish tumor cells from normal cells with the naked eve, meaning that surgeries are extremely difficult. Once targeted imaging technology is mature enough to outline the tumor boundaries for such surgeries, it will significantly increase the success rate of the operation. Conservative treatments such as chemotherapy can be gradually replaced by active surgical procedures on a large scale. Existing in vivo imaging techniques can only display the planar conformation of the imaged area, and the actual volume of the imaged area cannot be determined. Therefore, it is impossible to provide accurate pre-judgment for the above-mentioned surgeries. However, the imaging technology of QDs has not matured. On the contrary, related research on imaging is just beginning.

Drug delivery

Although drug-delivery-related research on QDs has not entered the clinical research stage, they have brought new hope for the cure of major human diseases, especially cancer. Table 3 [26,141–152] lists the current drugdelivery materials based on SQDs. The various compositions and modifications make them exhibit different

Table 3 Use of SQDs for drug delivery

characteristics in drug-delivery studies.

The blue-light emitting ZnO QDs mentioned in s1 have been combined with chitosan and folic acid for tumortargeted doxorubicin (DOX) delivery, and the drugloading efficiency reached 75%. With the hydrophilicity and cationic charge characteristics, chitosan increases the stability of the QDs. Folic acid is used to weaken the electrostatic interaction between DOX and QDs in the ZnO-QDs-chitosan-folate carriers and the binding affinity between DOX and the carriers is the primary factor influencing the DOX-releasing response. The carrier shows the characteristic of pH insensitivity. The s2 also reports a ZnO QD-based DOX delivery platform. Dicarboxyl-terminated polyethylene glycol (PEG) is introduced to render the carrier stable in physiological fluids, and the hyaluronic acid is conjugated to the carrier as a targeting ligand. This DOX delivery platform is pHsensitive, and the release of DOX is triggered by the dissolution of the ZnO QDs and dissociation of DOXmetal complex under acidic conditions. Furthermore, the acidic environment provided by tumors can completely biodegrade the ZnO QD carriers. After dissolution, the carriers alone display antitumor activity by preferentially killing cancer cells. Both s1 and s2 are ZnO QD-based DOX-delivery carriers. However, their DOX-releasing

No.	QDs	Modifier	Remarks	Drug	Object	Refs.
s1	ZnO	Chitosan Folic acid	pH non-sensitive	DOX	-	[26]
s2	ZnO	Hyaluronic acid PEG	pH responsive	DOX	A549	[141]
s3	CdTe	Silica mesoporous Selective release		DOX	Hela	[142]
s4	CdTe	Polyethylene glycol Regulate the protein express		DOX	PRMI 8226 ^a	[143]
s5	CdSe/ZnS	– Antibacteria		Qe	BGC-823 ^b	[144]
s6	CdSe/ZnS	GEM nanovectors	Dual enzymatic reactions	GEM	BxPC-3 ^c	[145]
s7	ZnO@polymer	-	pH-Triggered	DOX	U251 ^d	[146]
s8	ZnO,Au@polymer	Copolymer	Cooperation effect	CPT	Hela	[147]
s9	Mn:ZnS	Folic acid Chitosan	Attenuate tumor resistances	5-FU	4T1	[148]
s10	ZnSe:Mn/ZnS	Folate Hybrid silica F127 micelles	Linearly release behavior	PTX	_	[149]
s11	CdSe/ZnS	β-CD L-arg	Produce ROS	DOX	HepG2	[150]
s12	CuInS ₂ /ZnS	SiO ₂ nanohybrids	Efficient assistance of drug delivery	Pt(IV)	MCF-7	[151]
s13	$Fe_{3}O_{4}@ZnO:Er^{3+},Yb^{3+}@(\beta-CD)$	-	Microwave-triggered drug release	Etoposide (VP-16)	MCF-7	[152]

a) PRMI 8226: multiple myeloma line; b) BGC-823: gastric carcinoma; c) BxPC-3: human pancreas adenocarcinoma cell; d) U251: human glioblastoma cells.

triggers and induction of tumor cell apoptosis are quite different. In addition to the drug release influences mentioned above, the drug release behavior is affected by many factors, such as surface properties, the degradation rate, particle size, the interaction force of drug binding to the surface, and the rates of dehydration and hydration of the polymers [153].

The CdTe QDs introduced by s3 are coated by modified mesoporous silica nanoparticles, which lead to pore size improvements with better encapsulation and controlled release. Also for the consequent release inside the cell, DOX was observed to only be released while the carriers were inside the cells. The pH-triggered CdTe QD delivery carrier (PEG-CdTe-DOX) mentioned in s4 was modified by PEG, which facilitated intracellular drug accumulation. The PEG-CdTe-DOX was found to enhance the cytotoxic effects to tumor cells *via* inducing apoptosis *via* the caspase-mediated apoptotic pathway. That is, the anti-tumor activity of DOX could be improved by PEG-CdTe-DOX *via* regulation of the protein expression of apoptosis-associated genes.

The s5 and s6 represent CdSe/ZnS QD-based delivery carriers. The Qe/CdSe/ZnS nanoparticles (s5) are more cytotoxic than quercetin (Qe) and Qe/CdSe/ZnS, which is attributed to the sustained-release of Qe from the Qe/ CdSe/ZnS nanoparticles. This sustained-release may enhance the Qe uptake by the tumor cells for Qe loaded into the amine-enriched CdSe QDs. Besides, the Qe/CdSe/ZnS nanoparticles displayed effective antibacterial activities via disrupting the bacterial cell wall and membrane. The s6 shows dual-enzyme-sensitive gemcitabine (GEM) nanovectors which are synthesized via the conjugation of matrix metalloproteinase-9 (MMP-9)-detachable PEG, cathepsin B-cleavable GEM, and targeting the ligand CycloRGD ligand to the CdSe/ZnS QDs. GEM nanovectors will accumulate in tumor tissue via enhanced permeability and retention effects. Then, the overexpressed MMP-9 in tumor tissue removes the PEG corona in the tumor microenvironment, which leads to the exposure of RGD. RGD will guide the rest of GEM nanovectors to locate towards the cancer cells. After internalizing into pancreatic cancer cells, the release of GEM can be further promoted by the elevated lysosomal cathepsin B. The dual enzymatic reactions help GEM nanovectors achieve prolonged circulation time.

By comparing s1 with s6, it can be seen that even the same carrier materials will exhibit various forms of functions after different functional modifications. As introduced, the drug cytotoxicity, drug release, drug residence time, and other characteristics can be adjusted according to the actual situation. In addition, functional modifications will also promote drug absorption. Qe (s5) plays an important role in inflammation, antibacterial and anticancer activities, and aging. Unfortunately, the water-solubility of Qe is extremely low, which poses a challenge to drug absorption. After modification of the QDs, the as-prepared nanoparticles possessing high water-solubility can be used in bacteria and tumor cells.

After the above comparison between the same basic carriers, the varied service modes for in vitro or in vivo SQD drug-loading systems will be introduced. The s7 is a pH-responsive ZnO@polymer-DOX drug delivery system. The system is very stable in aqueous solutions at pH 7.0 and rapidly decomposes at pH 6.0, which ensures the safety and minimizes healthy concerns. The drug delivery mechanism mainly involves the cellular internalization of ZnO@polymer-DOX, the biodegradation in endosomes/ lysosomes, and the subsequent penetration of DOX into the nucleus. 5-Fluorouracil (5-FU), an anti-cancer drug used to induce thymidine synthase overexpression and cancer cell resistance was chosen to load into the chitosan and folic acid-modified Mn:ZnS QDs in s9, and 4T1 cells were selected to explore the drug delivery mechanism. The architecture of the reticuloendothelial system organs can be significantly affected by 4T1 cells that cause lymphocyte depletion and adversely affect the normal liver and kidney metabolic functions. 5-FU delivery systems with targeting specificity can protect the entire reticuloendothelial system by suppressing the proliferative tendency of 4T1 cells, including the disruption of the cell growth rate. Besides, the delivery systems have been found to attenuate possible tumor resistances to 5-FU.

The s11 shows the DOX delivery carriers of CdSe/ZnS QDs modified by β -cyclodextrin (β -CD) and L-arginine (L-arg). With the presence of β -CD, DOX can efficiently go through the cell membranes, and the specific binding of L-arg makes the HepG2 cells more sensitive to the DOX delivery carriers. It is thought that the DOX delivery carriers efficiently induce apoptosis, which may relate to the production of reactive oxygen species (ROS). This mechanism was confirmed by ROS analysis. 2',7'-Dichlorodihydrofluorescein diacetate (H2DCF-DA) is sensitive to the oxidation formed from Fenton reactions of H₂O₂ [154]. The QDs, DOX, and DOX-loaded QDs were respectively incubated with DCF in HepG2 cells. The green fluorescence of DCF was observed only in the DOX-loaded QD sample, indicating that the intracellular ROS levels in cells can be significantly increased by treatment with DOX-loaded QDs.

The drug delivery system presented in s12 was prepared

by silica nanohybrids integrated with $CuInS_2/ZnS$ QDs and magnetite nanocrystals. The PEG and amine groups were grated with the outside SiO₂ shell, which provides better biocompatibility and allows subsequent bioconjugation. The Pt (IV) was used as the model drug and conjugated with SiO₂ nanohybrids *via* the NH₂ functional groups, which was then reduced in the intracellular milieu to generate cytotoxic Pt (II) by reductive elimination of the axial ligands [155].

Fig. 4 (s10) shows that the hydrophobic ZnSe:Mn/ZnS core-shell QDs and the anticancer drug paclitaxel (PTX) were co-loaded into folate-conjugated hybrid silica nanocapsules via F127 micelles (abbreviated as QPNCPs), and the optical properties of the composite nanocapsules were dependent upon the F127:QDs mass ratio-dependent behavior. The linear release behavior of PTX from the QPNCPs was observed within 12 h, which provided a certain amount of PTX concentration in the blood circulation system. Besides, the QPNCPs greatly enhanced the solubility of PTX by up to 630 times. The sustained release performance and enhanced solubility were attributed to the hydrophobic PPO block of the F127 molecules and the rigid structure introduced by the SiO₂ shell. Another nanocarrier (s8) could also achieve the controlled release of hydrophobic anticancer drugs. The core-shell targeted delivery nanocarrier of camptothecin (CPT) is composed of ZnO QDs-conjugated gold nanoparticles (Au NPs) as the core and an amphiphilic hyperbranched block copolymer as a shell that include a poly-*L*-lactide (PLA) inner arm and an outer arm of FAconjugated sulfated polysaccharide from *Gynostemma pentaphyllum* Makino (GPPS-FA). The controlled release behavior follows a diffusion-controlled mechanism, and the GPPS on the surface of the carrier makes the delivery nanocarrier more effective in killing tumors *via* the cooperation between the carriers and CPT.

The drug delivery carriers summarized above all use *in* vivo factors as a trigger for drug release. The s13 introduced a kind of nanocarrier, which is a noninvasive, externally controlled drug-delivery system. The nanocarrier (abbreviated as $Fe_3O_4@ZnO:Er^{3+}$, $Yb^{3+}@\beta-CD$) was synthesized with a Fe_3O_4 core and a $ZnO:Er^{3+}, Yb^{3+}$ shell, and the chemically inert β -CD cavity stored etoposide (VP-16) molecules using hydrophobic interactions (Fig. 5). The $ZnO:Er^{3+}, Yb^{3+}$ shell is a good microwave absorber and its excellent microwave thermal response property is used in the microwave-triggered VP-16 release. The release profile can be adjusted by the duration and number of cycles of microwave application.

In summary, we reviewed some representative SQDbased drug delivery systems. Although SQDs as basic carrier materials do not play a decisive role in the efficacy of the entire drug-delivery system, some QDs can synergize with drugs to increase the drug toxicity. Func-



Figure 4 Schematic assembly of ZnSe: $Mn/ZnS-PTX/SiO_2$ nanocapsules templated by F127 micelles; the fluorescence spectra and images (the inset photographs, excited at 375 nm) of QNCPs; the cumulative release of PTX in QPNCPs within 24 h. Reprinted with permission from Ref. [149], Copyright 2016, Elsevier Inc.



Figure 5 Schematic illustration of $Fe_3O_4@ZnO:Er^{3+},Yb^{3+}@\beta-CD$ nano-composites and the controlled release of VP-16 by microwaves. Reprinted with permission from Ref. [152], Copyright 2014, Elsevier B.V.

tional modifications of the carriers can be used to develop the drug delivery systems into different service models. For the loaded drug, a drug with low solubility can gain considerable solubility by incorporating it into a drugloading system. Some drugs can achieve sustained release by being coated in a hydrophobic cavity of the carrier. In other modifications, such as with PEG, the material can afford the carriers better biocompatibility. The porous structure of silica materials allows for the regulation of the loading amount of the drug. An identification group, such as folic acid, can provide the corresponding targeting property to the carriers. Also, the modification not only determines the trigger of the drug release for the carriers, but also affects the way the drug induces apoptosis in tumor cells.

We reviewed SQD-based materials in the field of drugloaded delivery research and summarized their respective strengths and highlights based on typical or classic instances. However, there are still many common problems in SQD-based delivery systems that need to be improved and solved. For example, among many anticancer drugs, DOX with a broad antitumor spectrum can be used in the chemotherapeutic treatment of many human malignancies, such as breast cancer, bladder cancer, multiple myeloma, lymphoma, and small cell lung cancer [156]. Therefore, DOX is used as a loaded drug in many cases, but DOX is a type of drug that works after intercalating into DNA [157]. DOX interacts with DNA by inserting and inhibiting the biosynthesis of macromolecules [158]. Topoisomerase II used to untie the DNA supercoil structure is inhibited by this effect. After topoisomerase II unfolds the DNA preparing for replicating, DOX will stabilize topoisomerase II to prevent the DNA doublehelix from recombining, thereby stopping the replication process. Therefore, if DOX cannot assuredly enter the nucleus smoothly, then DOX can only act on the cells in the disappeared period of the nuclear membrane and nucleolus. It is not easy to carry drugs into the nucleus. There are thousands of nuclear protein complexes on the nuclear membrane. These nuclear pores and specific nuclear importin combine to form a cryptographic substance exchange system [158]. Synthetic nanomaterials have difficulties entering the nucleus due to the lack of specific binding functions. Nuclear targeting studies of some other materials can provide some reference for QD drug delivery. The nuclear localization signal (NLS) extracted from SV40 large T is used to deliver Au nanoparticle with 25 nm to cell nuclei. Therefore, it is conceivable to modify the delivery system with NLS and some tumor identifiers simultaneously. The tumor identifier would be used to locate the drug delivery system to the tumor cells. After the drug delivery system enters the interior of the cell, the NLS would bring it to the cell nucleus. This dual-targeting location approach is applicable to drugs inducing cell apoptosis, such as DOX, via entering into the nucleus to act on DNA. Based on the above, the dosage of the drug would be greatly reduced, which would help to prevent the development of drug resistance in the tumor cells. DOX can also be combined with substances produced by the cells themselves, such as colyone, which is a type of molecule signaling the termination of cell proliferation. It not only increases the specificity of the drug-loading system but also reduces the rejection of cells to foreign substances.

As far as current research is concerned, pH is the main trigger for drug release. Other triggers include enzymes, microwaves, light sources, and magnetic fields. Also, signal molecules (generated during excess cells proliferation), local high temperature or low temperature, and ultrasound can all be attempted to be used as the triggers for drug release. But no matter which trigger method is used, it will lead the drug delivery system to continue to release the drug once it starts. However, this continuous drug release method may not be applicable for some clinical uses where the induction of cell apoptosis is not needed. For example, in some osteosynthesis procedures, it is necessary to implant the drugs promoting bone growth into the fracture. However, each patient's recovery ability is different. Local bone hyperplasia will happen if the administration is excessive. In this instance, the dose of drug administered must be controlled according to the actual situation, which raises new challenges for drug delivery research. One challenge is to build the real on-off drug delivery system. That is, the drug delivery system can control the initiation and termination of the drug release, even the amount of dose per unit time can be adjusted according to the trigger intensity. Another challenge is that these cases have a long recovery period and require long-term administration. As far as current long-term drug studies are concerned, it is still difficult to extend the administration time to several months. In the introduction, we introduced the advantages of combination of lanthanide materials and QDs in detail, but the functional materials relating to these are still relatively few. The combination of lanthanide metalorganic framework (MOF) materials and QDs may provide a real possibility for true long-term administration. Because the coordination number of the lanthanide ions is more than that of conventional ions, it is conducive to combination with more drug molecules. Besides, MOF materials can provide loadable spaces for the drug.

In the above fluorescence imaging studies, some assumptions for the detection of tumor spread are proposed, which can be combined with drug delivery research. This will not be discussed in detail here. Fluorescence imaging technology can also provide early warning for drug-delivery systems, and it is expected to present different fluorescence chromaticity at various stages of tumor cell growth. If this technique is achieved, different drugs can be simultaneously loaded into the delivery system, which can release specific drugs at homologous wavelengths of different fluorescent chromaticities.

CONCLUSIONS AND OUTLOOK

In this review, we introduced the SQDs prepared by different modifications for different biomedical applications. In each part, we have summarized and analyzed the advantages and disadvantages of the existing research, and on this basis, some outlooks on the prospects in the corresponding fields are raised. However, there are far more studies of SQDs than these.

For short-term goals, SQDs can be combined with other cutting-edge materials to make full use of their properties. At present, the combination of semiconductor materials with MXene, black phosphorus, TMDCs, and other materials has shown strength in the fields of optics [159], catalysis [160], and sensing [161]. SQDs can look for opportunities to optimize the materials used in these studies. For example, BPQDs have good biocompatibility and capabilities for photothermal and photodynamic therapy. However, the stability of BPQDs is poor, and they are easily oxidized. SQDs have good stability and optical performance. Therefore, nanoparticle composites can be obtained by doping water-soluble SQDs with BPQDs. The new material is expected to combine the advantages of the two materials. By the improvement of these materials, the application of SQDs in the field of biomedicine can be further enriched. SQDs will not only be used for imaging, sensing, and drug delivery. SQDbased materials can be doped into nanofibers due to their advantages of size and excellent biocompatibility, and new composite materials can be made into scaffolds by 3D printing. Such scaffolds will have better cell affinity and compatibility than traditional materials and can be used as support materials for the treatment of bone and cardiovascular diseases.

In the future, SQDs researches should focus on the two keywords of precision and intelligence. As shown in Fig. 6, the growth and luminescence mechanism of materials should be clarified first in fundamental research on SODs. Clear mechanisms will provide authoritative theoretical reference for the preparation of materials, making it possible to control the synthesis and modification of the materials. Then, the physical and chemical properties of the nanomaterials can also be correspondingly adjusted. Based on these, the door of SQD biotechnology will indeed be opened. In biological application research on SQDs, primary research on the stability, compatibility, and toxicity of SQDs in biosystems still need to be strengthened. In particular, studies on toxicity should not stop at the level of testing cell viability. Studies on the toxicity and metabolism of cells, tissues, and organs after SQDs enter the organism should be part of systematic research.

In functional studies, sensing, imaging, diagnostic, and therapeutic research should be tightly integrated with intelligent electronic devices and communication technologies. For example, in the diagnosis and treatment of cancer, attempts can be made to load functionalized SQDs into specific nanorobots. Nanorobots can enter the body via injection or other routes and reach the lesion through the blood circulating. When the nanorobots are specifically fixed in the lesion, the course of the disease can be monitored at any time. After pairing with an in vitro electronic device, the nanorobot will emit a signal to make the external electronic device sound an alarm if the lesion is abnormal. With the help of portable electronic devices, patients can discretionary decide to make the nanorobots release functional SQDs. The patients can also terminate the process after receiving feedback to prevent damage to normal cells. After servicing its purpose, the robot will be degraded and excreted from the body as they receive instructions. Unused nanorobots can continue to monitor in real-time at the lesion to prevent the recurrence of the disease. In this way, the biological application of SQDs can truly move toward precision and intelligence.



Figure 6 Prospect of biological researches on SQDs.

Although some clinical prospects cannot be realized in a short time, they still provide a specific direction for the future development of SQDs in the biomedical fields. SQD materials are expected to perform better in the cross-disciplinary field of nanobiomedicine in the future.

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Conflict of interest The authors declare that they have no conflict of interest





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半导体量子点:修饰及其在生物医学中的应用

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摘要 半导体量子点由于其高量子产率、可调谐的发射光谱和优 异的光稳定性而受到人们的广泛关注. 这些独特的光学特性赋予 半导体量子点在生物医学领域中出色的应用前景, 譬如量子点可 在生物医学成像、药物输送、临床诊断、光动力疗法、DNA杂交 和RNA分析等领域中得以应用.本综述介绍了量子点的分类,并简 要描述了每类量子点的特征. 以II B-VI A型量子点为例, 分别讨论 和总结了其无机和有机修饰方法及相应的优缺点. 基于可控的修 饰方法,量子点在生物成像和药物传递领域表现出不同的功能.本 文还列出了量子点的一些典型或经典实例,用于介绍半导体量子 点在生物医学领域中的应用亮点. 以此为基础, 本综述提出了半导 体量子点在未来生物医学领域中的各种潜在挑战和应用前景.