



A self-powered gas sensor based on PDMS/Ppy triboelectric-gas-sensing arrays for the real-time monitoring of automotive exhaust gas at room temperature

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ABSTRACT A new self-powered active gas sensor for real-time monitoring of automotive exhaust gas was devised. The pipe-shaped device was fabricated from polydimethylsiloxane/polypyrrole (PDMS/Ppy) triboelectric gas-sensing unit arrays. The gas-sensing units can actively convert the mechanical energy of gas flow into a triboelectric current. The output current signal depends on the species and concentrations of the target chemical gases (CO, NH₃, NO) in the gas flow, and thus can be used as a sensing signal. The device consists of seven gas-sensing units with different Ppy derivatives. As the different sensing units respond to the gases in different ways, the device can differentiate between gas species. The working mechanism is attributed to the coupling effect between the triboelectric effect of PDMS/Ppy and the gas-sensing properties of Ppy. The device can be installed in the tailpipe of an automobile, and can thus analyze the exhaust gas in real time without the need for any external electrical power. The results of the present study spur a new research direction for the development of automotive exhaust gas monitoring systems, thus playing an important role in the detection of air pollution.

Keywords: self-powered, gas sensing, triboelectrification, automotive exhaust gas, coupling effect

INTRODUCTION

In modern society, automobiles are an essential means of transportation [1–3]. There are about half a billion automobiles on the roads every day. This makes people's lives more convenient but also generates large amounts of pollutant gases, such as carbon monoxide (CO), nitric oxide (NO), ammonia (NH₃), and volatile organic com-

pounds (VOCs) [4–12]. None of NH₃, CO, or NO can be detected by the human senses. It is very important to have a means of selectively and rapidly detecting automotive exhaust gas in the atmosphere [13–15]. However, traditional automotive exhaust gas sensors incur problems associated with their large size, while their bulky integrated power units prevent real-time monitoring [16–24]. The power supply unit embedded in a gas sensor has become a bottleneck to the application of conventional sensors [25–27]. If the power supply unit is large enough to provide sufficient energy, the sensor will likely be big and heavy. On the other hand, a small power supply unit would have a minimal capacity and thus would have to be frequently recharged, such that the collection of vital data would be missed. Moreover, the maintenance costs associated with the power supply units would increase with the usage of the sensors [28]. Therefore, a self-sustainable small gas sensor, which could be easily installed in an automobile with no need of an external power source, would be a major contribution to the development of intelligent vehicles.

Recently, self-powered gas-sensing systems have been developed by integrating gas sensors and triboelectric nanogenerators to solve the problem of providing the sensor with a reliable power supply [29–33]. Moreover, in our previous work, triboelectrification and gas-sensing processes were combined into a single device (a self-powered active gas sensor). The triboelectric output of the device under applied mechanical deformation was dependent on the gas atmosphere [34,35]. The output triboelectric signal can act as both the power for driving the device and the gas-sensing signal [36,37]. This self-pow-

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ered active gas sensor is very suitable for the real-time monitoring of automotive exhaust gas, but requires the development of a new materials system and device structure. Given the wide variety of automobile tailpipe shapes, a flexible pipe-shaped device with gas-sensing arrays would be a good configuration. Various polypyrrole (Ppy) derivatives with different chemical/physical properties and charge characteristics can be easily modified by using different dopants and surfactants in the synthetic process, which can be selected to realize the monitoring of different gases in multiplexed sensing arrays [38–42].

In the present study, a new type of self-powered gas sensor for the real-time monitoring of automotive exhaust gases was developed. Based on the triboelectric/gas-sensing coupling effect, polydimethylsiloxane/Ppy (PDMS/Ppy)-nanostructure gas-sensing arrays are realized as a pipe-shaped device. Driven by the gas flow, the sensing unit arrays can output a series of triboelectric current signals dependent on the gas species and concentration in the gas flow. When installed in an automobile tailpipe, the device can monitor and calibrate the exhaust gas in real-time without any external electrical power source. This new self-powered gas sensor could be applied to future intelligent vehicles.

EXPERIMENTAL

Fabrication of self-powered gas sensor

The fabrication of self-powered gas sensor started with the creation of a pattern on copper foil using a photolithography method. The Cu foil was cleaned using deionized water and ethanol for several times. Then, 3 mL of positive photoresist was spin-coated onto the Cu foil at 200 rpm for 15 s, and then at 1500 rpm for 40 s. Then, the Cu foil was dried at 100°C for 150 s and fixed in a photolithography machine by being exposed to UV light for 2.4 s, thus producing the desired pattern. After that, the prepared Cu foil was immersed in developer for 60 s, and then washed with deionized water and dried in N₂ flow. Then, the prepared Cu foil was hardened at 120°C for 150 s. The photolithography process was performed in the dark. Next, the Cu foil with the photoresist pattern was wet-etched by being soaked in etching liquid (0.5 mol L⁻¹ sodium persulfate aqueous solution) at 60°C for 50 s to form patterned Cu-network arrays. Then, the Cu network was spin-coated with a PDMS mixture and cured at 90°C for 30 min in a vacuum oven. The PDMS mixture was obtained by mixing an elastomer with a cross-linker at a mass ratio of 10:1, followed by ultrasonic

treatment for 15 min to eliminate bubbles. The Cu network, covered with PDMS, was then wet-etched again in the etching liquid to create sufficient space between the PDMS and Cu network for friction and to provide space for the deposition of Ppy. Next, an inductively coupled plasma method was used to dry-etch the surface of the PDMS film with ICP-100A (Potentlube Technology Co., Ltd). Then, seven kinds of Ppy derivatives were deposited on the Cu network by electrochemical polymerization, forming seven independent gas-sensing units. The precursor solution contained 0.2 mol L⁻¹ dopants and 0.1 mol L⁻¹ pyrrole monomer. Sodium benzene sulfonate, sodium dodecyl benzene sulfonate, sodium dodecyl sulfate, oxalic acid, sodium oxalate, naphthalene sulfonic acid, and camphor sulfonic acid were used as the dopants. A Pt wafer, Ag/AgCl electrode, and Pt wire served as the working, reference, and counter electrodes, respectively, during the electrochemical polymerizing process. Finally, a Cu film was deposited on the rear surface of the PDMS film by means of electron-beam evaporation, to act as the opposite electrode.

Characterization and measurements

For the tests, the flexible gas sensor was affixed to the inner wall of a vehicle tailpipe. All the sample gases were purchased from Shenyang Xindongji Chemical Gas Co., Ltd. The concentrations of the sample gases (NO, NH₃, and CO) were 1000 ppm. The flow of the target gas and air was provided by a gas mixing system (MTI, GSL-3F). The gas flow rate was measured by a gas flow meter, and held at 20.57 m s⁻¹ unless mentioned otherwise. The output current of the device was measured by a low-noise current preamplifier (SR570, Stanford Research Systems). All the sensing measurements were conducted at room temperature (26°C). The morphology and microstructure of the material were investigated using a scanning electron microscope (SEM; Hitachi S4800).

RESULTS AND DISCUSSION

The design of the self-powered gas sensor for automotive exhaust gas is shown in Fig. 1. The NO, NH₃, and CO in automotive exhaust gas can all affect the human respiratory system, eyes, and skin, as shown in Fig. 1a. As these kinds of gases cannot be detected by the human senses, there is a need for a means of rapidly detecting automotive exhaust gas in the atmosphere. Our new self-powered automotive exhaust gas sensor is driven by the gas flow and generates triboelectric-sensing signals for the detection of toxic gases. The output current signals are immediately sent to a data-collecting system so that the

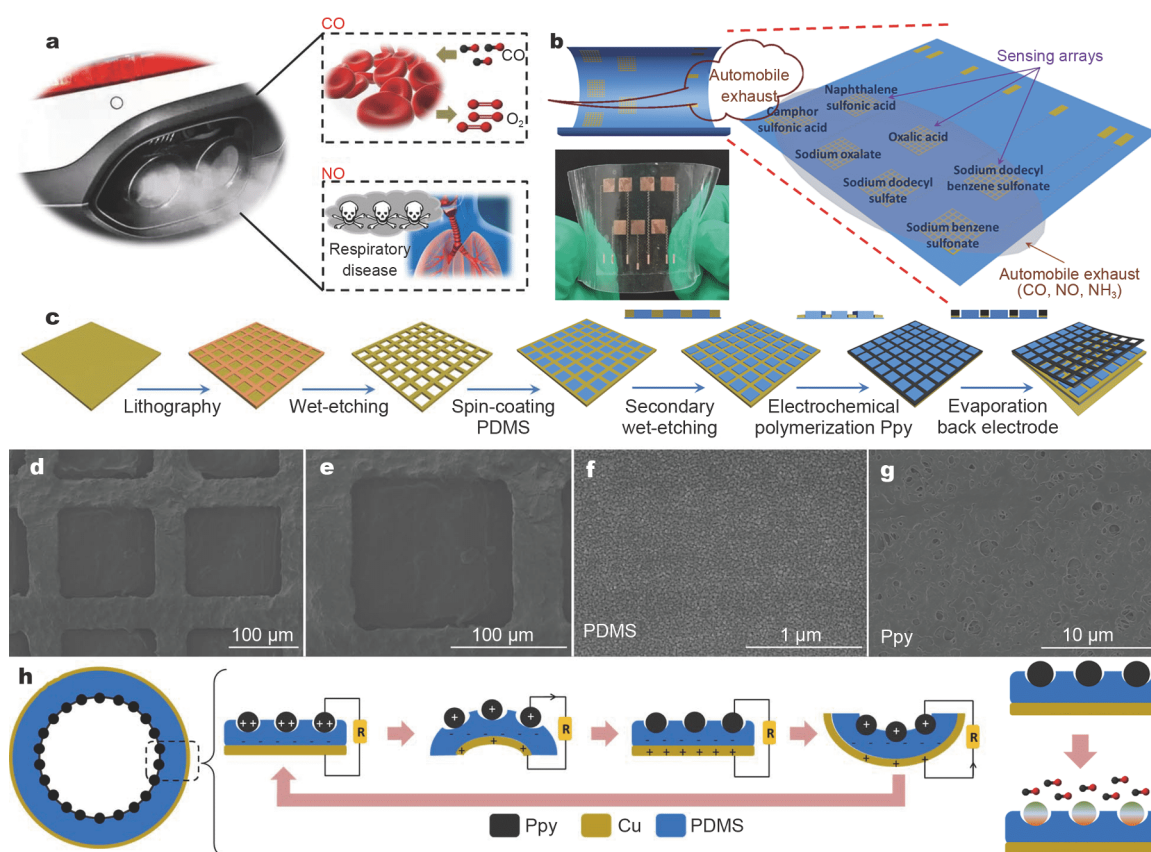


Figure 1 Experimental design of self-powered gas sensor. (a) Damage to human body caused by automotive exhaust gas. (b) Design of the gas sensor for exhaust gas analysis. Inset shows an image of the device. (c) Fabrication of gas sensor. SEM images of (d, e) a single gas-sensing unit, (f) PDMS and (g) Ppy-NSA. (h) Working mechanism.

automotive exhaust gases can be monitored in real time. Fig. 1b shows the experimental design of the self-powered gas sensor for the real-time monitoring of automotive exhaust gas. The device is rolled up and installed in the automobile tailpipe. The device is itself pipe-shaped, such that it can easily be driven by the exhaust gas flow. When the exhaust gas passes over the device, current signals are generated by the sensing units due to the vibration of the device (triboelectrification). The multiplexed sensing unit arrays (different Ppy derivatives) output different current signals for specific gas species. The inset in Fig. 1b shows how the device can bend, thus enabling real-time gas sensing in a vehicle tailpipe. To better demonstrate the flexibility, a device without a Cu back electrode is shown.

Fig. 1c shows the fabrication of the self-powered exhaust-gas sensor. The device mainly consists of three layers: Ppy on a Cu network, a PDMS layer, and a Cu back electrode. The Ppy nanostructures are deposited on the Cu network and act as the functional (sensing/triboelectrification) materials. The PDMS nanostructured

layer acts as both the negative triboelectric material and the flexible substrate. The Cu film on the back of the PDMS layer acts as the other electrode. The size of the device is typically $5\text{ cm} \times 6\text{ cm}$, and each internal sensing unit (of which there are seven) is $0.5\text{ cm} \times 0.5\text{ cm}$. Here, seven different Ppy derivatives (doped with different dopants or surfactants) are synthesized independently on different sensing units. The sensing units are labeled Ppy-SBS, Ppy-SDBS, Ppy-SDS, Ppy-OA, Ppy-SO, Ppy-NSA, and Ppy-CSA, depending on the dopants (sodium benzene sulfonate, sodium dodecyl benzene sulfonate, sodium dodecyl sulfate, oxalic acid, sodium oxalate, naphthalene sulfonic acid, and camphor sulfonic acid, respectively). SEM images of each sensing unit on the self-powered exhaust gas sensor are shown in Fig. 1d and e. The detailed structures are highly uniform. Fig. 1f shows the SEM image of the PDMS layer. The surface of the PDMS layer is treated using an inductively coupled plasma. Fig. 1g shows the morphology of the Ppy-NSA. The rough surfaces of the PDMS and Ppy increase the

active surface area contributing to the contact/triboelectrification.

The working mechanism of the self-powered automotive exhaust sensor is schematically illustrated in Fig. 1h. The mechanism is based on the coupling between the triboelectric effect of the PDMS/Ppy nanostructures and gas-sensing ability of the Ppy [43–45]. There is a gap between the PDMS layer and Ppy on the Cu network. This provides sufficient space for friction between the PDMS layer and Ppy. The gap width changes when the sensor is applied with vibration caused by gas flow. Due to the difference in the electron affinities, surface charges are generated by a triboelectrification process when the Ppy comes into contact with the PDMS layer [46,47]. In the initial state, the Ppy is fully in contact with the PDMS, which induces the transfer of electrons from the Ppy to the PDMS through a triboelectrification effect [48,49]. As a result, positive charges accumulate on the surface of the Ppy, while negative charges accumulate on the surface of the PDMS. At this time, the triboelectric charges on the surfaces are stable and there is no movement of the electrons in the external circuit. When the sample gas flows over the device, it causes the device to vibrate, and the PDMS and Ppy separate, leading to an imbalance in the triboelectric charges. The negative charges on the PDMS surface will transfer to the Ppy surface through the external circuit. When the PDMS comes into contact with the Ppy again, the decrease in the interlayer distance allows the electrons to return from the Ppy to the PDMS electrode, thus creating a reverse current. The electrostatic induction generates output current signals until the PDMS and Ppy completely come into contact with each other again. Thus, the alternative current can be observed in the external circuit while the device is vibrating. The triboelectrification effect between the Ppy and PDMS is affected by the surface chemical/physical state of the frictional materials [34,47]. The changed Ppy chemical state will influence the surface charge carrier density of the Ppy, thus resulting in different output currents according to the gas atmosphere. In this case, the interaction between the gas molecules and Ppy is multiform [50–52]. For reducing gases like NH_3 , there is a reduction in the charge carrier density. Since the majority carrier (hole) density decreases due to the electron-donating nature of the NH_3 , it results in a decrease in the conductivity and an increase in the resistance. For an oxidizing gas like NO, there is an increase in the charge carrier concentration, since NO is electron-accepting in nature and increases the conductivity of the material, resulting in the decreased resistance of the Ppy and an

increase in the output current.

Fig. 2 shows that the self-powered automotive exhaust gas sensor can be easily driven by the gas flow and convert mechanical energy into triboelectric current. Our experiments were conducted at room temperature (26°C) and 40% relative humidity (RH). In Fig. 2a, the output short-circuit current of the Ppy–NSA sensor unit is about 20 nA when the gas flow velocity is 20.57 m s^{-1} . Fig. 2b shows that the output open-circuit voltage of the Ppy–NSA sensor unit is about 0.5 V. The triboelectric current output from the sensor unit can be affected by the gas flow velocity, as shown in Fig. 2c. At gas flow velocities of 11.75, 14.69, 17.63, and 20.57 m s^{-1} , the output current of the Ppy–NSA sensing unit is 6.672, 10.061, 12.817 and 19.994 nA, respectively. The relationship between the output current and the gas flow velocity is shown in Fig. 2d. The current output by the sensor unit increases with the velocity of the gas flow. Fig. 2e shows the continuity and stability of the sensor unit. The output current signal remains at around 25 nA for 2 h. The exhaust gas sensor has good stability and repeatability, which would allow it to be applied to the continuous testing of a vehicle over an extended duration. The power density of the sensor unit is 40 nW cm^{-2} . The output of the unit can charge a $1\text{-}\mu\text{F}$ capacitor to 0.49 V in 700 s (black curve in Fig. 2f), a $10\text{-}\mu\text{F}$ capacitor to 0.22 V in 700 s (red curve in Fig. 2f), and a $22\text{-}\mu\text{F}$ capacitor to 0.14 V in 700 s (blue curve in Fig. 2f), such that it could power some small electronic devices or be applied to energy storage.

The triboelectric gas-sensing performance of individual sensing units was tested with different gases, as shown in Fig. 3. The experiment was conducted in a 20.57-m s^{-1} gas flow and at room temperature (26°C). Fig. 3a shows the output current signal of the Ppy–OA sensing unit in NH_3 gas flow of different concentrations. Upon setting the concentrations of the NH_3 in the gas flows to 0, 50, 100, 150, 200 and 250 ppm, the triboelectric currents output by the sensing unit were 12.817, 11.354, 10.235, 9.138, 8.317, and 6.959 nA, respectively. The output triboelectric current thus decreased with an increase in the NH_3 concentration. As shown in Fig. 3b, there is an approximately linear relationship between the output current and the NH_3 concentration. Fig. 3b also shows the relationship between the response and the NH_3 concentration. The response ($R\%$) of a Ppy–OA sensing unit to NH_3 can be roughly defined as:

$$R\% = \frac{I_{\text{am}} - I_{\text{a}}}{I_{\text{a}}} \times 100\%, \quad (1)$$

where I_{a} and I_{am} represent the maximum output currents

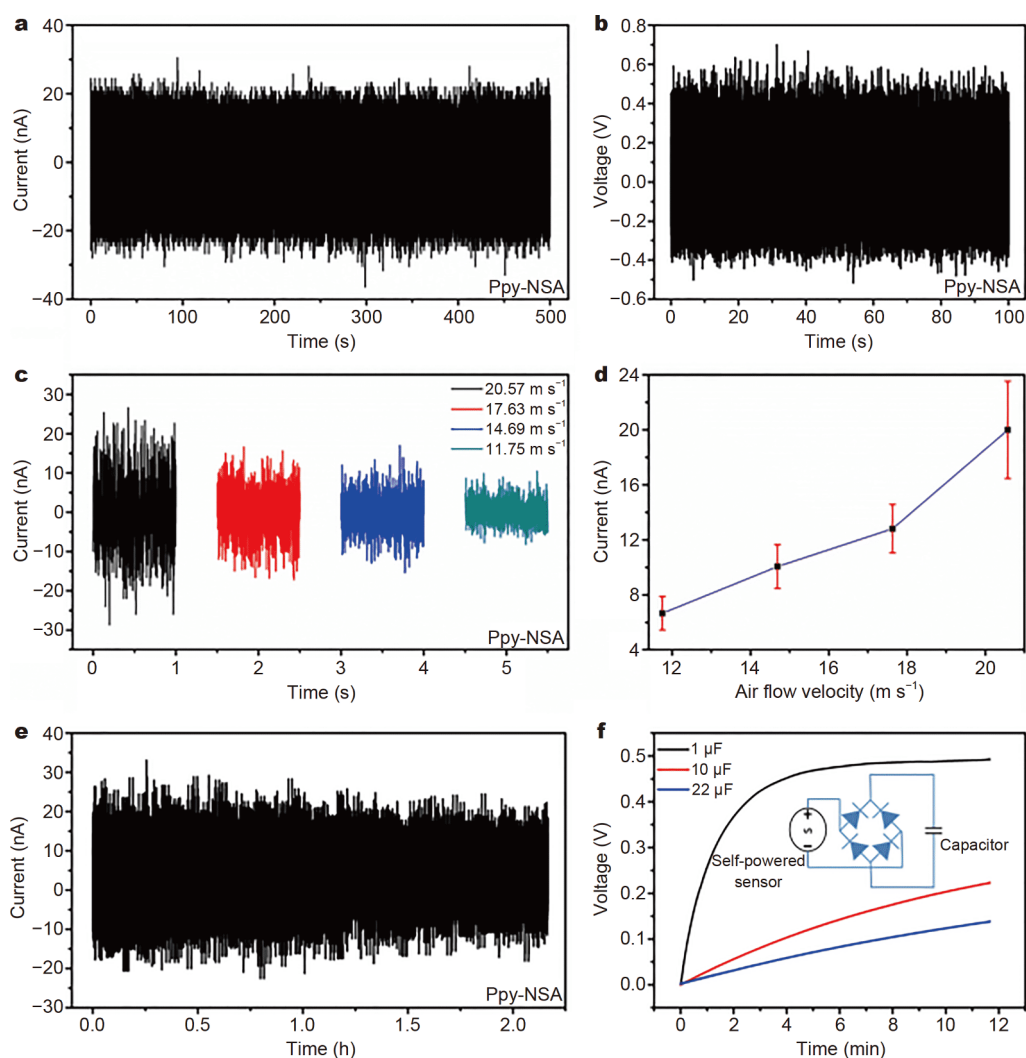


Figure 2 Triboelectric output performance of Ppy-NSA sensing unit. (a) Output current signal of unit under a constant gas flow. (b) Output voltage signal of the sensor. (c) Output current of the device under different gas flow velocities. (d) Relationship between gas flow velocity and output current signal. (e) Long-term stability of sensor. (f) Charging curves for 1-, 10-, and 22- μF capacitors when being charged by the unit. The inset shows the circuit diagram.

in the period for air and different concentrations of NH_3 gas, respectively. For NH_3 concentrations of 50, 100, 150, 200 and 250 ppm, the response was -11.42% , -20.15% , -28.71% , -35.11% , and -45.70% , respectively. In addition, the Ppy-OA sensing unit is also highly sensitive to NO gas flows, as shown in Fig. 3c and d. For NO gas flows of 0, 50, 100, 150, 200, and 250 ppm, the current output by the sensing unit is 9.650, 10.362, 11.113, 12.346, 13.446, and 14.667 nA, respectively, while the corresponding responses are 7.38%, 15.15%, 27.94%, 39.33%, and 51.98%. The triboelectric gas-sensing performance of Ppy-CSA sensing unit is shown in Fig. 3e-h. For NH_3 gas flows of 0, 50, 100, 150, 200, and 250 ppm, the tribo-

electric current output from the sensing unit is 28.275, 26.629, 23.550, 17.725, 17.000, and 15.839 nA, respectively, while the corresponding response is -5.82% , -16.71% , -37.31% , -39.88% , and -43.98% . For NO gas flows of 0, 50, 100, 150, 200, and 250 ppm, the short-circuit current of the sensing unit is 29.916, 47.228, 65.966, 74.008, 91.804, and 114.569 nA, respectively, while the corresponding response is 57.87%, 120.51%, 147.39%, 206.88%, and 282.98%. These excellent responses show that the self-powered gas sensor could be applied to the monitoring of automotive exhaust gas.

The sensing units doped with different dopants/surfactants have different selectivities for the detection of

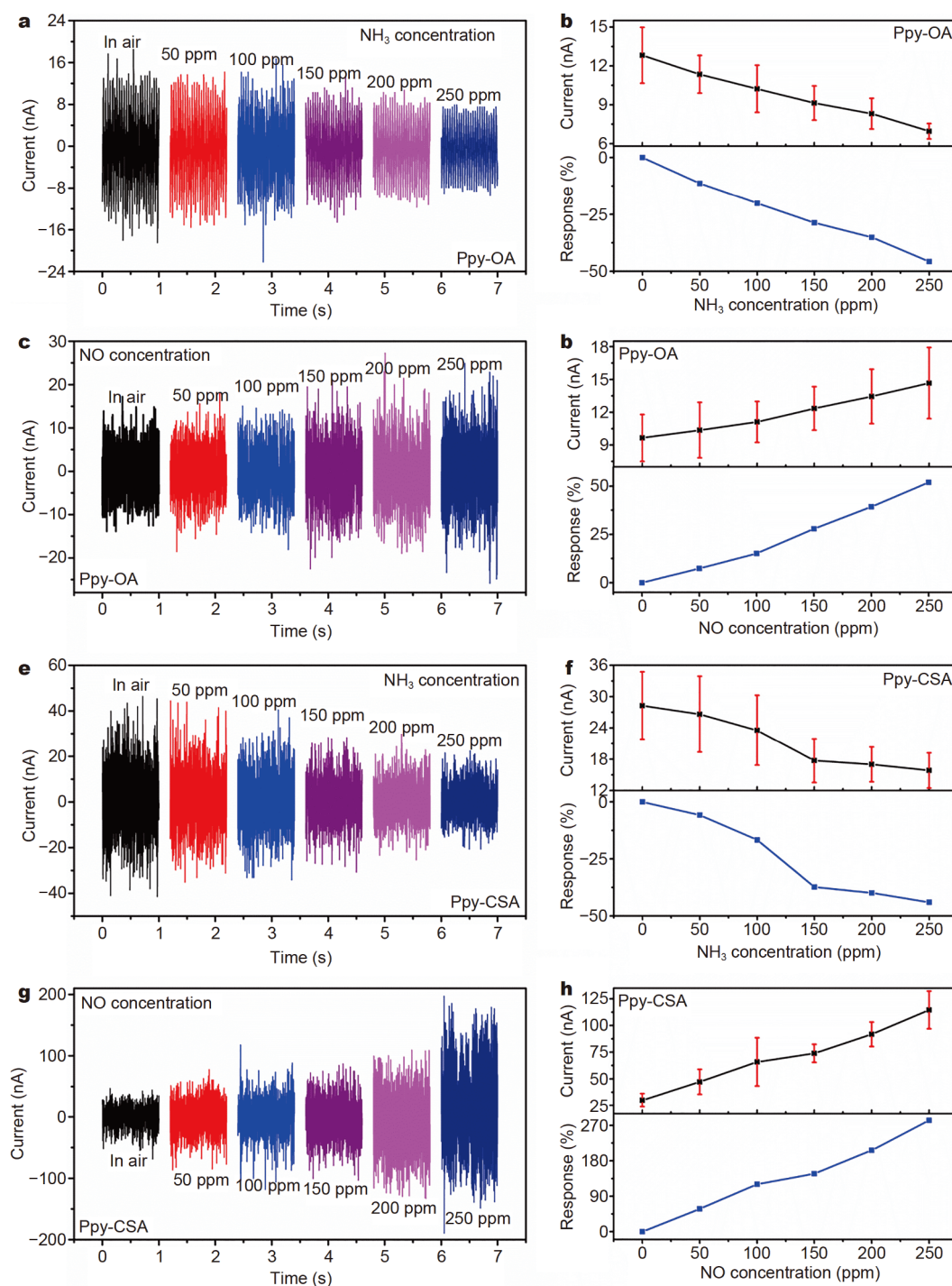


Figure 3 Triboelectric-gas-sensing performance of two sensing units. (a–d) Sensing performance of Ppy–OA sensing unit in NH₃ and NO gas flows. (e–h) Sensing performance of Ppy–CSA sensing unit in NH₃ and NO gas flows.

specific gases. Therefore, the sensing units can distinguish chemically diverse gases at the same time and analyze the various chemical components of automotive exhaust. Here, seven sensing units (Ppy–SDBS, Ppy–SDS, Ppy–

SBS, Ppy–OA, Ppy–SO, Ppy–NSA, and Ppy–CSA) are designed to output current signals for three target gases (CO, NO, and NH₃), as shown in Fig. 4. The responses of the seven sensing units are –11.70%, –5.16%, –22.02%,

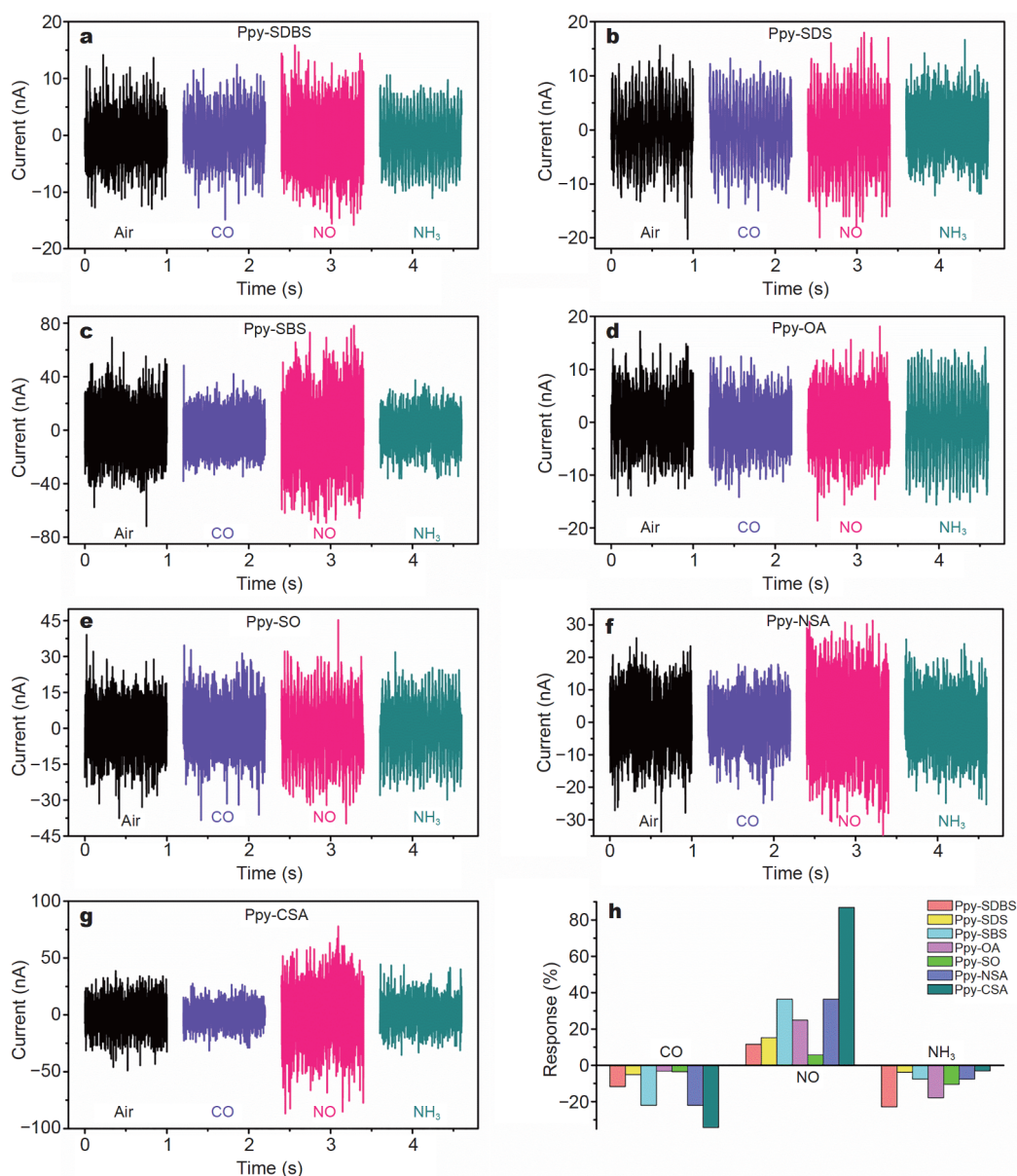


Figure 4 Triboelectric-gas-sensing performance of seven sensing units in different gases. (a–g) Output currents from seven sensing units in 50-ppm CO, 50-ppm NO, and 50-ppm NH₃ gas flows, respectively. (h) Responses of seven sensing units.

−3.15%, −3.53%, −22.02%, and −34.23%, respectively, for a 50-ppm CO gas flow. For a 50-ppm NO gas flow, the responses are 11.55%, 15.19%, 36.42%, 24.84%, 5.64%, 36.42%, and 86.91%, respectively. For a 50-ppm NH₃ gas flow, the responses are −22.87%, −3.84%, −7.55%, −17.90%, −10.49%, −7.55%, and −3.12%, respectively. These results point to the potential of the self-powered gas sensor for the monitoring of automotive exhaust gas.

Fig. 5 shows the real-time continuous sensing profiles of the sensing units against specific gases (response/re-

covery processes). The response times of Ppy-CSA for CO, Ppy-SBS for NO, and Ppy-NSA for NH₃ are about 11, 31, and 13 s, while the recovery times are about 5, 30, and 12 s, respectively. These results indicate that the device offers an effective means of monitoring automotive exhaust gas in real time. The response of the device (Ppy-NSA sensing unit) for CO at different gas flow velocities is shown in Fig. 6. In Fig. 6a, when the gas flow rate is 20.57, 17.63, 14.69 and 11.75 m s^{−1}, the output current is 19.99, 12.82, 10.06 and 6.67 nA for air, and 14.83, 9.84,

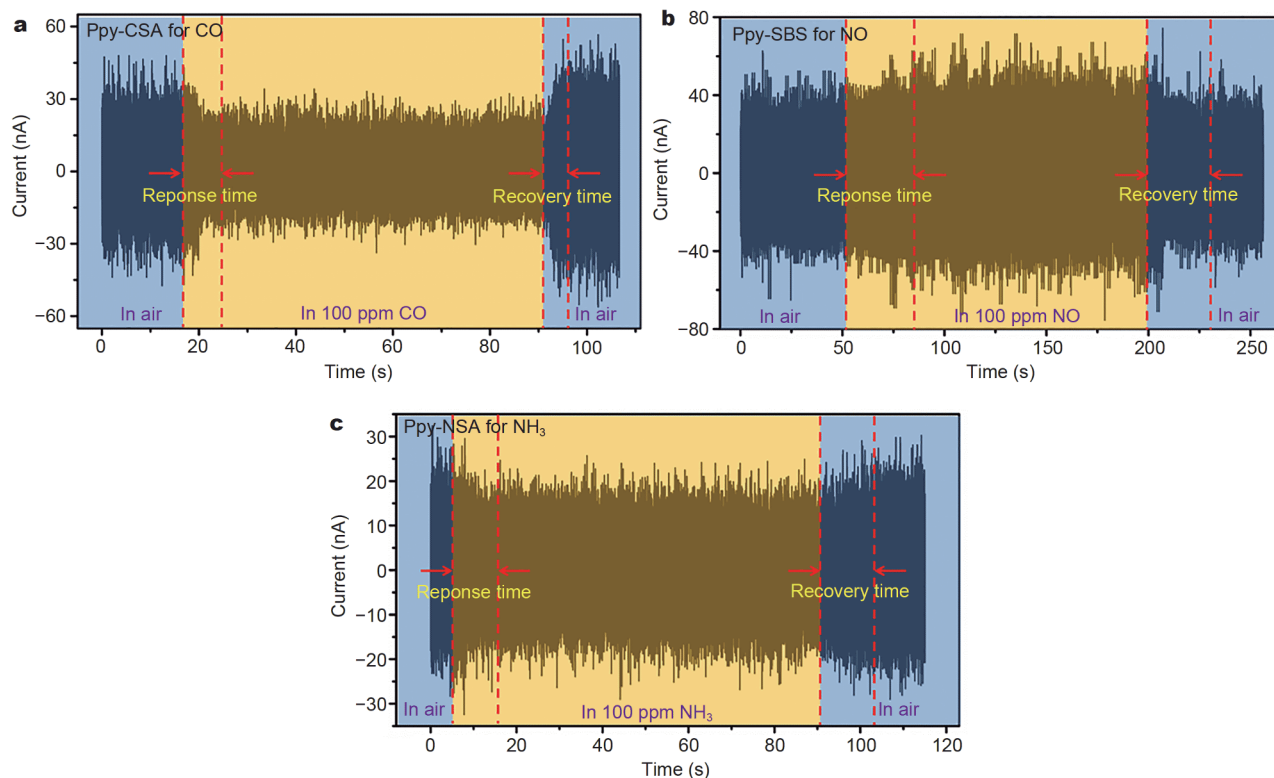


Figure 5 Response/recovery processes of different sensing units. (a) Ppy-CSA sensing unit for CO. (b) Ppy-SBS sensing unit for NO. (c) Ppy-NSA sensing unit for NH₃.

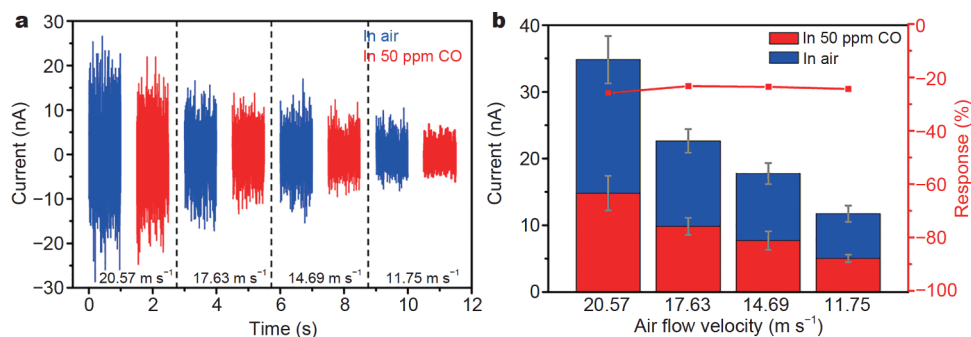


Figure 6 Sensing performance of Ppy-NSA sensing unit under different gas flow velocities. (a) Output current under different gas flow velocities for 50-ppm CO and in air. (b) Response of Ppy-NSA sensing unit under different gas flow velocities.

7.70 and 5.05 nA for a 50-ppm CO gas flow, respectively. The device response for different gas flow rates is almost the same, as shown in Fig. 6b (the response is -25.83% , -23.26% , -23.50% , and -24.33%). The results show that the sensing performance of the device is highly stable for different gas flow velocities, confirming a wide range of operating conditions.

In practical gas-monitoring applications, the effect of RH on the gas response of the device is very important.

Fig. 7 shows the influence of the RH on the NH₃ response of the Ppy-CSA sensing unit. The output current signals of the Ppy-CSA sensing unit under air flow and 50-ppm NH₃ under different RH conditions (40%, 60%, 80%, and 100% RH) are shown in Fig. 7a. Fig. 7b shows that the output current decreases as the RH increases both in air and NH₃. When the PDMS and Ppy come into contact and then separate, electrons transfer from Ppy to PDMS at the interface. As the humidity gradually increases, the

surface of the Ppy will adsorb water molecules, leading to weak triboelectric effect and a reduction in the charge transfer [53–55]. As shown in Fig. 7c, the NH_3 response increases with the humidity, even at high humidity levels. These results point to the potential for applying the unit to gas flows with a high humidity.

The present study devised a self-powered gas sensor for the real-time monitoring of automotive exhaust gas without an external electricity source, as shown in Fig. 8. A schematic diagram of the device is shown in Fig. 8a. The automotive exhaust gas is collected by a gas-mixing system, which can change the gas atmosphere (pure air and real automotive exhaust gas). The gas sensor can monitor the changes in the automotive exhaust gas in real time, and output current signals without an external power source. The current signals are sent to a data collector/computer which analyzes the exhaust gas. The automotive exhaust gases from two different vehicles (a sports utility vehicle (Fig. 8b, c) and a sedan ((Fig. 8d, e)) were tested using the gas sensor under a gas flow velocity of 20.57 m s^{-1} (the gas flow velocity was maintained by the gas-mixing system). The output current from the

Ppy–NSA sensing unit for pure air was measured first. When the air was switched to the exhaust gas from the sports utility vehicle, the output current decreased from ~ 25 to $\sim 20 \text{ nA}$ in $\sim 40 \text{ s}$, as shown in Fig. 8c. When the air was switched to the exhaust gas from the sedan, the output current increased from ~ 25 to $\sim 40 \text{ nA}$ in $\sim 150 \text{ s}$, as shown in Fig. 8e. These results imply that the concentrations of exhaust gas from the two kinds of vehicles are totally different. The competition mechanism which is available for the gas-sensing performance is expressed in the output current signals. The rising and falling current signals indicate that the exhaust gas from the sports utility vehicle contains more CO and NH_3 than NO, while the exhaust gas from the sedan contains more NO than CO and NH_3 . These different gas-monitoring results demonstrate that the device could be applied to real-time automotive exhaust gas monitoring without any external power.

CONCLUSIONS

A self-powered gas sensor was developed for the real-time monitoring of automotive exhaust gas. The device can be

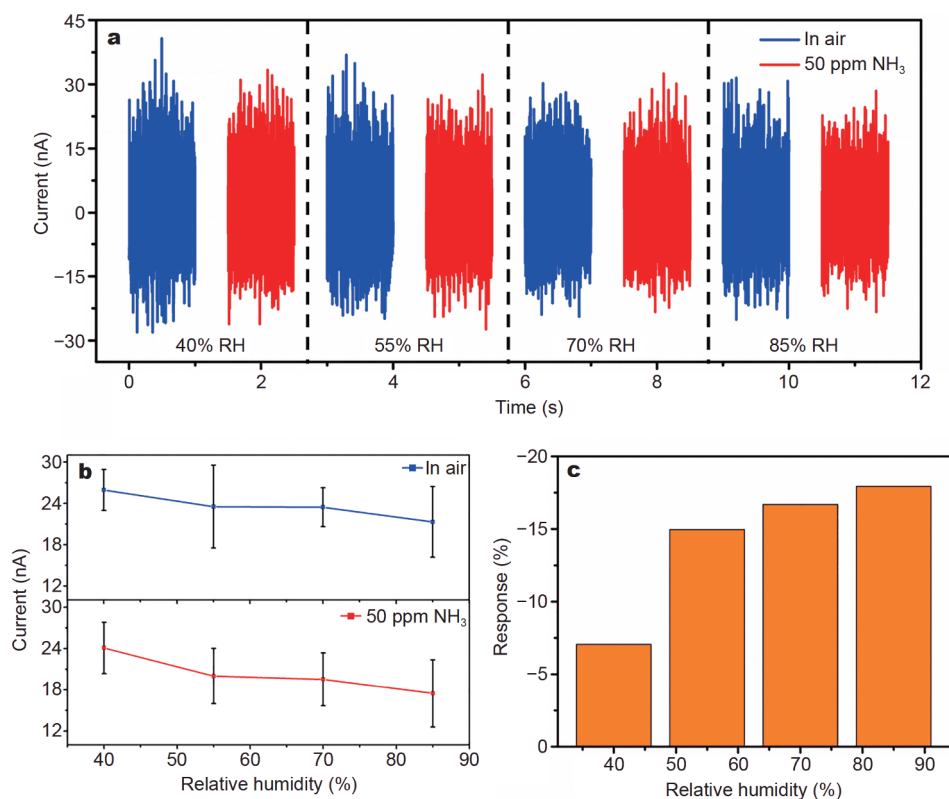


Figure 7 Sensing performance of Ppy–CSA sensing unit under different gas flow velocities. (a) Output current from the Ppy–CSA sensing unit for air and 50-ppm NH_3 gas flows under different conditions (40%, 55%, 70%, and 85% RH). (b) Relationship between output current and RH for air and 50-ppm NH_3 gas flow. (c) Relationship between response and RH.

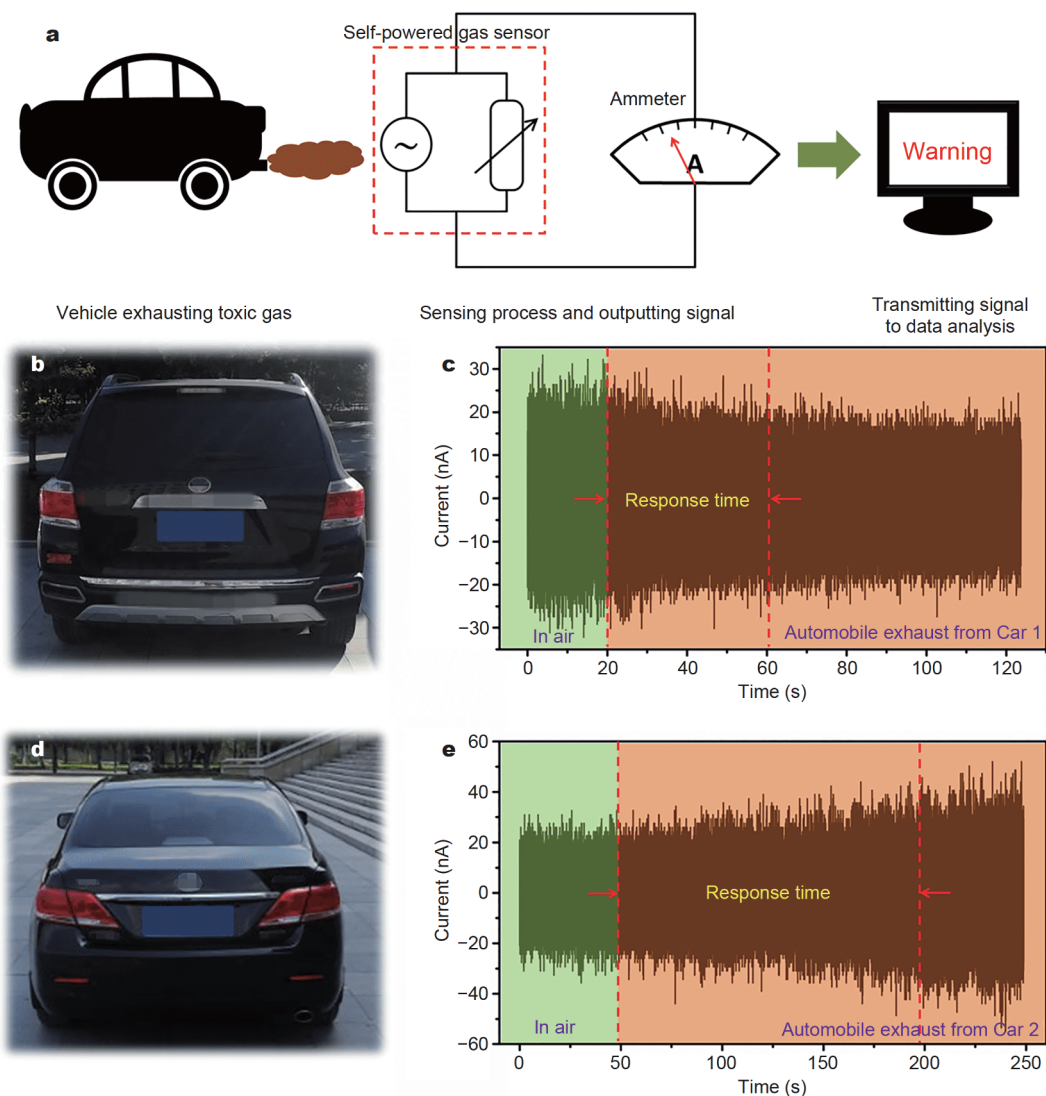


Figure 8 Practical application of self-powered automotive exhaust gas sensor. (a) Schematic diagrams of self-powered automotive exhaust gas sensor. (b, d) Vehicles used for the tests. (c, e) Current signals output by the device for automotive exhaust gases.

driven by the gas flow, while the output triboelectric current signals provide sensing data without the need for any external power supply. The mechanism is based on the triboelectric/gas-sensing coupling effect between Ppy derivatives and PDMS. The application of the device to the analysis of automotive exhaust gas was demonstrated using two actual vehicles. This new device should promote the development of self-powered automotive exhaust gas monitoring systems and could be applied to intelligent vehicles in the future.

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Conflict of interest The authors declare that they have no conflicts of interest.



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基于PDMS/Ppy摩擦气体传感阵列的自供电式气体传感器用于室温实时监控尾气

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摘要 本文提出了一种用于汽车尾气实时监测的新型自供电式气体传感器。这种管形器件由聚二甲基硅氧烷/聚吡咯(PDMS/Ppy)摩擦-气体传感单元阵列组成。摩擦-气体传感单元可以直接将气流的机械能转换为电能并输出摩擦电流。输出电流的信号强度取决于气流中目标化学气体(CO、NH₃、NO)的种类和浓度,因此这种输出电流可以同时作为传感信号。这种自供电式气体传感器将纳米发电机和气体传感器集成于单一器件,在一个物理过程中同时实现摩擦发电和气体传感,进而实现了自供电式气体传感器。该装置可固定在汽车排气管处,无需外接电源即可对废气进行实时分析。这一工作为汽车尾气监测系统的发展开辟了新的研究方向,将在大气污染的检测方面起到重要作用。