



# The Effects of Carbon Dioxide Removal on the Carbon Cycle

David P. Keller<sup>1</sup> · Andrew Lenton<sup>2,3</sup> · Emma W. Littleton<sup>4</sup> · Andreas Oschlies<sup>1</sup> · Vivian Scott<sup>5</sup> · Naomi E. Vaughan<sup>6</sup>

Published online: 14 June 2018  
© The Author(s) 2018

## Abstract

Increasing atmospheric CO<sub>2</sub> is having detrimental effects on the Earth system. Societies have recognized that anthropogenic CO<sub>2</sub> release must be rapidly reduced to avoid potentially catastrophic impacts. Achieving this via emissions reductions alone will be very difficult. Carbon dioxide removal (CDR) has been suggested to complement and compensate for insufficient emissions reductions, through increasing natural carbon sinks, engineering new carbon sinks, or combining natural uptake with engineered storage. Here, we review the carbon cycle responses to different CDR approaches and highlight the often-overlooked interaction and feedbacks between carbon reservoirs that ultimately determines CDR efficacy. We also identify future research that will be needed if CDR is to play a role in climate change mitigation, these include coordinated studies to better understand (i) the underlying mechanisms of each method, (ii) how they could be explicitly simulated, (iii) how reversible changes in the climate and carbon cycle are, and (iv) how to evaluate and monitor CDR.

**Keywords** Climate change · Carbon dioxide removal (CDR) · Mitigation · Carbon cycle · Negative emissions · Carbon cycle feedbacks · Climate feedbacks

## Introduction

The Earth's climate and the carbon cycle are inherently linked. Carbon cycle processes determine the flow of carbon between reservoirs, (Fig. 1a). In the atmosphere, the carbon-containing gases carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>), along with water vapor, are the major greenhouse gases (GHGs). These GHGs absorb a proportion of the Earth's emitted long-wavelength radiation, thereby trapping heat. CO<sub>2</sub> in particular is a very long-lived greenhouse gas, whose atmospheric

concentration has been rising at unprecedented rates due to continued intensive fossil fuel use, land use change, and cement production.

Concentrations of atmospheric CO<sub>2</sub> have now exceeded 400 ppm, and are currently at levels not seen in at least the last 800,000 years [3]. This has led to detectable warming of the Earth, and changes in the global carbon cycle [3].

As atmospheric CO<sub>2</sub> levels continue to rise the likelihood of “severe, pervasive, and irreversible” impacts increases [4]. This was recognized by the United Nations Framework Convention on Climate Change, who facilitated the Paris Agreement on climate change [5] in which countries pledged Nationally Determined Contributions (NDCs) to deliver emissions reductions. However, the emissions reductions resulting from current NDCs appear to be insufficient to limit warming to “well below 2° C above pre-industrial,” the goal of the Paris Agreement [6]. Consequently, it is increasingly likely that some form of carbon dioxide removal (CDR) will be needed [7–10] to reach this goal.

CDR encompasses a range of methods aimed at reducing atmospheric CO<sub>2</sub> levels either by directly extracting CO<sub>2</sub> from the atmosphere or by deliberately enhancing land and ocean carbon sinks to increase removal of CO<sub>2</sub> from the atmosphere. While there is clearly an overlap between CDR and mitigation actions, here, we focus on widely discussed methods that propose the deliberate uptake of atmospheric CO<sub>2</sub>, not the

---

This article is part of the Topical Collection on *Carbon Cycle and Climate*

✉ David P. Keller  
dkeller@geomar.de

<sup>1</sup> GEOMAR Helmholtz Centre for Ocean Research Kiel, Düsternbrooker Weg 20, 24105 Kiel, Germany

<sup>2</sup> CSIRO Oceans and Atmosphere, Hobart, Australia

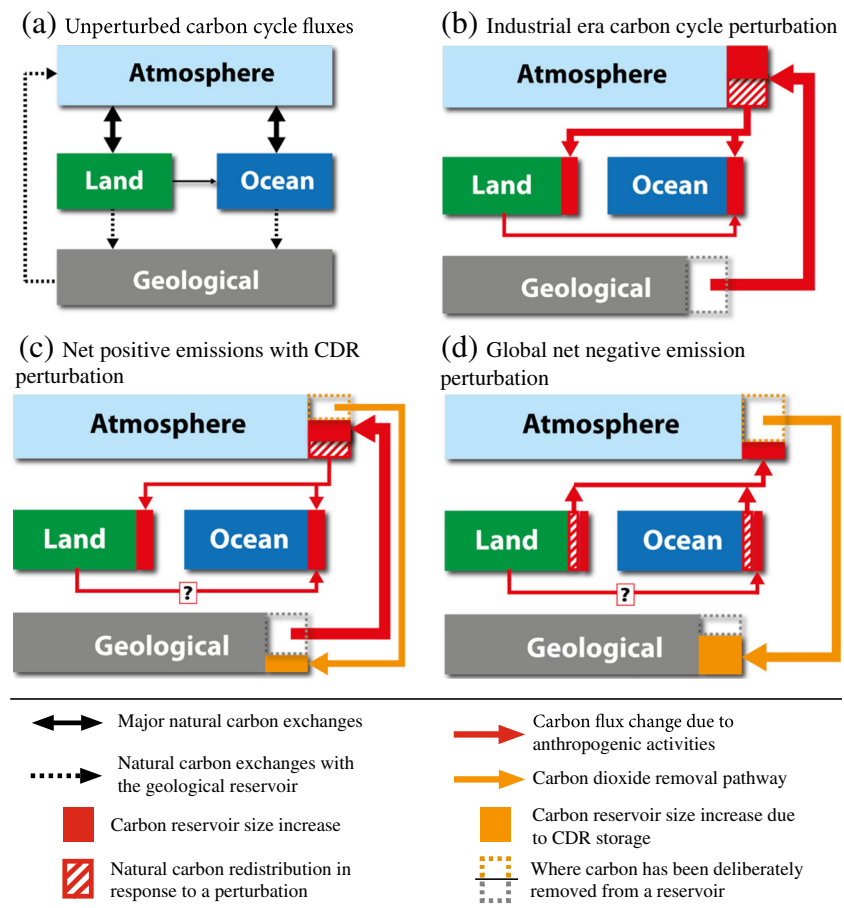
<sup>3</sup> Antarctic Climate and Ecosystems Cooperative Research Centre, Hobart, Australia

<sup>4</sup> College of Life and Environmental Sciences, University of Exeter, Exeter, UK

<sup>5</sup> School of GeoSciences, University of Edinburgh, Edinburgh, UK

<sup>6</sup> Tyndall Centre for Climate Change Research, School of Environmental Sciences, University of East Anglia, Norwich, UK

**Fig. 1** Schematic representation of the main carbon flows among atmospheric, land, ocean, and geological reservoirs for (a) the Earth before significant anthropogenic impacts; and how carbon flows have or may have changed due to anthropogenic activities such as (b) industrial era fossil fuel combustion, (c) when carbon dioxide removal (CDR) begins, but net CO<sub>2</sub> emissions are positive, and (d) when CO<sub>2</sub> is removed from the atmosphere, i.e., “net negative emissions.” Note that when net emissions are negative as in (d), it is still possible to have some emissions, but these are not depicted here. Carbon exchanges depicted in (a); black and dashed lines) also occur in b, c, and d. The question mark in the land to ocean carbon flux perturbation in c and d indicates that it is unknown how or if this carbon cycle perturbation will be affected by CDR. Adapted from [1•, 71•]



reduction of input of CO<sub>2</sub> (or avoided emissions). Furthermore, we focus on CDR at a sufficiently large scale to have global impact on the carbon flows in and out of the atmosphere (see Table 1).

A further distinction can be made between CDR methods that seek to accelerate the uptake of atmospheric CO<sub>2</sub> by enhancing natural sinks and methods that seek to engineer the removal and subsequent storage of CO<sub>2</sub> [92, 93]. Examples of the former include increasing carbon storage in biomass through expanding forest cover (afforestation) [36, 37•], and accelerating the rate of natural weathering of rocks, which chemically removes CO<sub>2</sub> from the atmosphere, by increasing their surface area (enhanced weathering) [1•, 68]. Examples of the latter include combining bioenergy with carbon capture and storage (BECCS) [1•, 94] and direct air capture (DAC) with storage [11]. BECCS uses biomass resources (e.g., energy crops or forestry residues) for energy conversion processes (e.g., combustion or gasification) and captures the CO<sub>2</sub> during these processes. DAC uses machines to separate CO<sub>2</sub> from the air. For both methods, the captured CO<sub>2</sub> can be geologically stored or used as a chemical feedstock for the manufacture of long-lived products.

At present, CDR methods and technology are immature and untested at scale [95], so our understanding of their

potential and impacts, including possible carbon cycle feedbacks, are reviewed from limited investigations. For example, many studies extrapolate from natural analogues or local applications, modeling and/or laboratory investigations, and in some cases a small number of pilot projects. However, these are insufficient for a full assessment of CDR and so there is a pressing need to advance research and, where relevant, the development of CDR technologies [96].

Fuss et al. [97] highlight several important CDR research priorities: (i) improving estimates of sustainable potentials, particularly for methods that require using large land areas or other limited resources; (ii) assessing the benefits and risks of different CDR methods to contribute towards climate stabilization; (iii) developing a governance model for CDR; and (iv) understanding the carbon cycle responses to CDR.

This article focuses on this last priority (iv) *understanding the carbon cycle responses to CDR*, specifically reviewing the latest research findings relating to CDR and the carbon cycle. The structure of this paper is as follows. We first describe the factors that control the response of the carbon cycle to perturbations; then collate and consider research findings into the overall effect of CDR on the carbon cycle with a description of the response to atmospheric CO<sub>2</sub> removal; and finally examine the possible carbon cycle impacts of specific CDR

**Table 1** Description of proposed carbon dioxide removal (CDR) methods where enough literature (>2 publications that investigate carbon cycle responses) exists to begin understanding how the carbon cycle may respond to large-scale (e.g., > 1Pg C) deployment

Method	General description	References
Direct air capture of CO <sub>2</sub> with storage	Technology that chemically or electro-chemically removes CO <sub>2</sub> from air and concentrates it for storage	[11, 13, 14, 15••, 16–21, 126]
Bioenergy with carbon capture and storage	Grow terrestrial vegetation* and use the biomass to create biofuels that can be burned in conjunction with carbon capture and storage technology	[1••, 60, 127, 128, 101, 102, 65, 26–35, 56]
Afforestation/reforestation	Plant or restore forests to increase CO <sub>2</sub> uptake (via primary production) and storage in biomass and soils	[1••, 22, 36, 37•, 42–53, 71•, 79, 70, 117]
Soil and land carbon management	Employ management practices, such as no-till agriculture, irrigation, cover crops, compost amendments, wetland restoration, and fire management, to increase C retention and storage in agricultural soils or managed natural lands	[57–59, 61–65, 67, 71•, 120]
Biochar	Pyrolyze terrestrial biomass* to form biochar and add it to soils where the C can remain sequestered (biochar is recalcitrant); biochar amendments may also enhance vegetation productivity and soil carbon storage	[23–25, 55, 61–65, 71•]
Enhanced weathering on land	Spread alkaline minerals on land to chemically remove CO <sub>2</sub> from the atmosphere in reactions that form ions, which are eventually transported to the ocean, or in some cases solid minerals (geological sequestration), may also enhance vegetation productivity and subsequently soil carbon storage	[1••, 68, 69, 73••, 54]
Ocean alkalization	Increase the alkalinity of the upper ocean to chemically increase the carbon storage capacity of seawater and thus, also increase CO <sub>2</sub> uptake	[37•, 69, 72, 73••, 74–84]
Ocean fertilization	Add micronutrients like iron or macronutrients like nitrogen and phosphorus to increase phytoplankton growth (CO <sub>2</sub> fixation) and ocean carbon storage via the biological pump (the transport of this fixed carbon into the deep ocean)	[37•, 85–90]
Artificial ocean upwelling	Use pipes or other methods to pump nutrient rich deep ocean water to the surface where it has a fertilizing effect; see ocean fertilization above	[37•, 40, 41, 91]

\*Bioenergy and biochar can also be created from marine micro- or macro-algal biomass. However, no literature exists on how the C cycle would respond to large-scale (e.g., > 1Pg C) marine biomass growth and harvesting for these purposes

methods with a focus on widely discussed and researched methods (Table 1). Other CDR methods, which have received less study (Table 2), are also briefly addressed. We conclude with a discussion of present unknowns and future research priorities.

## Factors Controlling the Carbon Cycle Response to Perturbations

The atmosphere, ocean, terrestrial biosphere, and geosphere hold a vast amount of carbon that is naturally exchanged between reservoirs (Fig. 1a). The exchanges of carbon between these reservoirs are mediated through biological, chemical, and physical mechanisms. Understanding how the carbon cycle has responded to past changes is key to understanding how it may respond to the ongoing anthropogenic perturbations. Here, we briefly describe the most relevant factors that control the carbon cycle response to anthropogenic perturbations. We focus only on those carbon reservoirs and processes that respond on timescales of up to a few centuries, thereby neglecting exchanges with the geological reservoir, except in cases of direct anthropogenic disturbance, e.g., fossil fuel use and engineered geological or geochemical storage of CO<sub>2</sub>.

## The Factors that Control the Response of the Ocean to Carbon Cycle Perturbations

The exchange of atmospheric CO<sub>2</sub> with the ocean is through air-sea CO<sub>2</sub> fluxes driven by the differences in the partial pressures (pCO<sub>2</sub>) between the atmosphere and surface ocean. The rate of the exchange depends on wind speed, temperature, and other factors, whereas the pCO<sub>2</sub> gradient (if present) sets the magnitude and the direction of the carbon flux. In the case that a perturbation acts to reduce oceanic pCO<sub>2</sub> to be lower than the atmosphere pCO<sub>2</sub>, e.g., when CO<sub>2</sub> is added to the atmosphere via fossil fuel combustion, oceanic carbon uptake is enhanced. Conversely, if atmospheric pCO<sub>2</sub> is reduced and is lower than that of the ocean, e.g., via DAC, this will lead to a transfer of carbon from the ocean into the atmosphere. As these CO<sub>2</sub> fluxes are driven by pCO<sub>2</sub> differences across the marine boundary layer, it is important to understand the processes that control how carbon is (re)distributed in the ocean, i.e., the factors that determine whether carbon stays near the surface or is sequestered into the deep ocean where it cannot readily be exchanged with the atmosphere.

The surface ocean pCO<sub>2</sub> is controlled through the physical and biological carbon pumps [111]. The biological carbon pump describes the photosynthetic uptake of carbon at the surface, i.e., conversion from inorganic to organic carbon,

**Table 2** Proposed carbon dioxide removal methods for which there is insufficient literature to assess their response (<2 publications) and confidently report on either the functional feasibility of the method (i.e.,

would it potentially work from an Earth system perspective) or on how the global carbon cycle would respond to the method.

Method	General Description	Not enough knowledge/publications on the technology	Not enough knowledge/publications on the global carbon cycle response	References
Biomass Burial	Harvest and bury terrestrial or marine biomass, which contains the carbon that vegetation has removed from the atmosphere during growth, in the deep ocean, coastal sediments, or on land	✓	✓	[108, 129, 103]
Artificial Downwelling	Artificially enhance the transport of carbon that has been taken up at the surface ocean into the deep ocean where it will be stored for hundreds to thousands of years	✓	✓	[106, 107]
Removal of CO <sub>2</sub> from Seawater with Storage	Technology that chemically or electro-chemically remove CO <sub>2</sub> from seawater and concentrate it for storage; in surface waters this will increase the oceanic uptake of CO <sub>2</sub> as seawater chemistry/physics compensates for its removal		✓	[132]
Coastal (Blue Carbon) Management	Plant and manage mangroves, wetlands, seagrass beds, or macroalgae to increase CO <sub>2</sub> uptake (via primary production) and burial in sediments	✓	✓	[66,98-100 109]
Carbon Product Utilization	Grow biomass (e.g., wood) and utilize it in structures to remove it from the active C-cycle or build with carbon-absorbing cements		✓	[129, 103, 131]
Cloud Alkalinization	Spray fine alkaline minerals into clouds to chemically remove CO <sub>2</sub> from the atmosphere in reactions that form ions, which then rain out and are eventually stored in the ocean; may also enhance vegetation productivity and subsequently soil carbon storage	✓	✓	[110]
Iron salt aerosol method	Add iron salt aerosols to the atmosphere to reflect incoming solar radiation, chemically or physically remove greenhouse gases and other aerosols, and have a fertilizing effect on land and in the ocean; perhaps also increase geological carbon storage?	✓	✓	[130]

and its subsequent transport (in particular via sinking particles) to the deep ocean. This sequestration of photosynthetically fixed carbon is inefficient with only a small proportion (10–15%) making it into the deep ocean (below the surface mixed layer that is in contact with the atmosphere) on annual timescales. The remaining 85–90% of the organic carbon is remineralized (i.e., broken down and transformed into an inorganic form via respiration). Photosynthesis is sensitive to changes in temperature, nutrients, and light availability. The rate of remineralization is sensitive to changes in temperature [112], the available quantity and quality of organic matter, and other factors such as oxygen levels and, for calcium-carbonate containing matter, seawater chemistry.

The physical carbon pump describes the physico-chemical processes that move dissolved inorganic carbon from the surface to the interior ocean. The magnitude of this pump is determined by CO<sub>2</sub> solubility, which depends on temperature (i.e., the warmer the ocean the less soluble CO<sub>2</sub> is in seawater) and carbonate chemistry, as well as physical ocean circulation, which transports the dissolved inorganic carbon. Ocean circulation can change in response to changes in wind stress, which particularly affects the upper ocean, and changes in buoyancy fluxes in response to warming and alterations in the hydrological cycle such as melting of ice, which can impact the formation of deep waters and the deep overturning circulation.

Although the general large-scale mechanisms of marine carbon cycling are well established, there remain processes that are less well understood and poorly quantified [111]. These lead to uncertainties in the present-day ocean CO<sub>2</sub> uptake ( $3 \pm 0.5$  Gt C year<sup>-1</sup>) [113], and in predicting the future evolution of the marine carbon cycle in response to climate and environmental change.

## The Factors that Control the Response of the Land to Carbon Cycle Perturbations

Biological processes primarily control the exchange of atmospheric CO<sub>2</sub> with the land, where the majority of the carbon is stored in soils and permafrost [3]. CO<sub>2</sub> is removed from the atmosphere by plant photosynthesis and primarily returned to the atmosphere by respiration and other processes such as fire. As long as primary production (GPP; i.e., gross photosynthetic carbon fixation) is greater than carbon losses due to respiration (autotrophic and heterotrophic) and processes such as fire, the land will be a carbon sink as it is today [113]. However, if a climate-carbon cycle perturbation such as an ocean-based CDR reduces GPP and increases or maintains rates of carbon losses that are higher than GPP, then the land could become a source of CO<sub>2</sub>. The factors that control the balance between terrestrial carbon uptake and loss are described below.

Photosynthesis rates are sensitive to changes in temperature and the availability of nutrients, light, and water. If nutrients, light, and water are available, the photosynthetic rate will increase with temperature until a maximum species-specific rate is reached, after which the photosynthetic rate decreases. Plant photosynthesis is also directly affected by the concentration of CO<sub>2</sub> in the atmosphere through the CO<sub>2</sub> fertilization effect. Higher concentrations of CO<sub>2</sub> allow more photosynthesis and reduced canopy transpiration, which in turn lead to increased plant water use efficiency and reduced fluxes of surface latent heat [3]. This climate-carbon cycle feedback (i.e., CO<sub>2</sub> fertilization) is thought to play a major role in determining how the land carbon sink responds to changes in atmospheric CO<sub>2</sub> [114]. Changes in the amount, density, or type of vegetation (e.g., by deforestation) will also alter photosynthetic carbon uptake. In addition, these changes may have physical climate feedbacks (e.g., through albedo changes that can cause warming or cooling), thereby having a further impact upon vegetation [115]. Past land use and land cover changes have already had a large effect on the climate and carbon cycle and will play a role in determining how the land responds to any future perturbations [116–118].

While the rate of autotrophic (plant) respiration is sensitive to the factors that control photosynthesis, heterotrophic (animal, fungi, protist, and non-photosynthetic bacteria) respiration is sensitive to changes in temperature, moisture, and the available quantity and quality of organic matter and nutrients [53, 119, 120]. Changes in vegetation may alter the organic matter available for heterotrophic consumption, thereby impacting respiration rates. Soil disturbance (e.g., from land use change and fire) can have a range of impacts on soil organic matter and can impact the net flux of carbon to the atmosphere.

The terrestrial carbon cycle and present-day CO<sub>2</sub> sink is less well constrained than that of the ocean with an estimated atmosphere to land flux of  $1.9 \pm 0.9$  Gt C year<sup>-1</sup> [113]. How the terrestrial carbon cycle will be impacted in the future in response to climate and land management changes, including the possible application of land-based CDR, is even more uncertain [121]. For example, in high CO<sub>2</sub> emission scenarios, some models suggest that the land will become a source of CO<sub>2</sub> while in others it remains a sink [122]. Major uncertainties remain in the projected changes in the availability of water and nutrients, but also in the physiological and, eventually, adaptive, and evolutionary responses of plants to enhanced atmospheric CO<sub>2</sub> and temperatures, as well as other environmental changes.

Carbon stored in permafrost is also sensitive to changes in the climate. If permafrost thaws, then it can be released/transformed into CO<sub>2</sub> and other greenhouse gases (in particular methane), though details in seasonal temperatures as well as the hydrological cycle may be crucial to determining the exact fate of the organic matter. At present, many aspects of these processes remain poorly understood [123].

Freshwater bodies also exchange carbon with the atmosphere across the air-water interface, with these exchanges driven by biological, chemical, and physical processes that are sensitive to many of the factors described above (e.g., temperature, light, hydrology, and nutrient and organic matter input). A significant amount of terrestrial carbon is also transported through freshwater bodies and eventually reaches the ocean if not remineralized or buried along the way [12, 124].

## The Response of the Carbon Cycle to Anthropogenic Perturbations

### Industrial Era Perturbations

The carbon cycle has responded to industrial era anthropogenic perturbations by redistributing carbon within and across reservoirs (Fig. 1b) in response to land use changes, other environmental disturbances (e.g., fires, nutrient additions, invasive species), and CO<sub>2</sub> emissions. Only around 42% of the CO<sub>2</sub> added to the atmosphere since the industrial revolution has remained in the atmosphere (termed the airborne fraction), with the remainder approximately evenly taken up by the land and ocean, respectively [113]. Carbon fluxes from the land to ocean have also increased [12].

### Atmospheric Carbon Dioxide Removal Perturbations

In this section, we review what has been learned about the response of the carbon cycle to CDR from idealized model simulations, which in most cases are analogous to technology that directly removes CO<sub>2</sub> from the atmosphere [11] and permanently stores it elsewhere (DAC; Table 1). Note that in the literature, CDR is often defined as the *gross* removal of carbon (e.g., 1 GtCO<sub>2</sub> removed/sequestered from the atmosphere and stored in geological reservoirs) rather than the *net* removal of carbon from the atmosphere (e.g., the net impact is much less when the response of the carbon reservoirs is accounted for). While understanding the gross sequestration potential of a method is important, understanding the net removal of CO<sub>2</sub> from the atmosphere is the key to evaluating the efficacy (defined here as a change in atmospheric CO<sub>2</sub> per unit CDR) of that method as a climate change mitigation measure.

The earliest idealized simulations of CDR, in models with a carbon cycle, prescribed either massive instantaneous CO<sub>2</sub> removals (negative pulses) or large decreases (e.g., 1% year<sup>-1</sup>) in atmospheric CO<sub>2</sub> [13, 14]. Following this, studies were conducted with more complex pathways of atmospheric CO<sub>2</sub> reduction [15•, 16, 17•, 18••] and additional components such as permafrost [19]. Some studies also focused on the response of a particular carbon reservoir such as the ocean [20, 21]. As a result of these studies, we have a growing

understanding of how the carbon cycle responds to atmospheric CDR. Models project that when net CO<sub>2</sub> emissions under CDR are positive, but start to decline, the land and ocean carbon sinks will begin to weaken and take up less CO<sub>2</sub> (Fig. 1c). Note that these responses may not be driven entirely by CO<sub>2</sub> forcing as other factors such as a changing climate also affect the strength of these sinks. At some point, as net CO<sub>2</sub> emissions decline, carbon sink uptake will exceed emissions input and the atmospheric CO<sub>2</sub> concentration will begin to decline. Atmospheric CO<sub>2</sub> will continue to decline once atmospheric CDR is deployed at a large enough scale that net CO<sub>2</sub> emissions become negative. At some point (years to centuries; and depending on the CDR rates), the oceans and land may become net emitters of CO<sub>2</sub> back to the atmosphere (Fig. 1c). Note that the land response is uncertain with some models suggesting that the land may never become a source of CO<sub>2</sub>. This uncertainty is not surprising given that there is also a large spread in the simulated terrestrial carbon cycle response to increasing atmospheric CO<sub>2</sub> [122] because of differences in model representations of system processes [125]. Also uncertain are the timescales of the response, as changing land and ocean sinks to sources does not occur instantaneously (e.g., transient climate change simulations have shown that the sinks may continue to respond to the prior emission trajectory for years to centuries before responding to CDR [15••, 18••]). More specifically, land responses may take years to decades if only some land components, e.g., vegetation, have been perturbed and potentially longer (centuries to millennia) if other components like the permafrost carbon pool have been perturbed. The ocean also generally takes a long time (decades to millennia) to respond. For all carbon reservoirs, the response depends on the rate and/or amount of CDR and the prior state of the climate and carbon cycle [17•, 18••].

This carbon cycle response tends to oppose CDR [15••, 18••]. For example, as illustrated in Fig. 4 of Keller et al. [126], instantly removing 100 Gt CO<sub>2</sub> from the atmosphere in Earth system models at a pre-industrial steady-state will only reduce the atmospheric CO<sub>2</sub> concentration by 100 Gt CO<sub>2</sub> immediately following the removal. After 100 years, atmospheric CO<sub>2</sub> is only ~25 Gt CO<sub>2</sub> lower because carbon is gradually released by the ocean and land in opposition to atmospheric CDR.

It is conceivable that CDR will be discontinued at some point, e.g., after atmospheric CO<sub>2</sub> has reached a desired level. The carbon cycle response to cessation will depend on the state of the climate and the cumulative amount of CDR. In overshoot emission scenarios simulations, models show that when CDR ceases and CO<sub>2</sub> emissions are zero, the ocean is likely to again become a carbon sink (if it had stopped being so in the first place) as the oceanic carbon pumps drive net oceanic pCO<sub>2</sub> to again become lower than the atmosphere pCO<sub>2</sub>. Because of slow ocean circulation, it will take several millennia before the deep ocean carbon cycle recovers from

past and any future anthropogenic perturbations [21]. The response of the land is less clear, with some simulations showing that it may continue to lose carbon to the atmosphere, albeit at a lower rate, or that it may again revert to a carbon sink [15••, 17•, 18••]. These responses may take years to decades on land, unless soil carbon pools have been perturbed in which case it may take centuries to millennia for any change of sign to occur.

Recent studies find that the efficacy of CDR depends on the scenario of removal and the state of the Earth system [15••, 17•, 18••]. When the background climate and carbon cycle state is the same (i.e., in equilibrium/at steady-state), more CDR results in stronger opposition (outgassing) from the ocean and land. CDR efficacy also changes over time in conjunction with co-occurring climate-carbon cycle feedbacks. If the background climate and carbon cycle state has been perturbed (i.e., is out of equilibrium), then the efficacy of CDR depends on the magnitude of perturbation (i.e., CDR efficacy is state dependent and the same level of CDR may be more or less effective in different scenarios). As described in Jones et al. [15••], a “perturbation airborne fraction” metric can be calculated to estimate the effectiveness of CDR for different background climate and carbon cycle states. In either

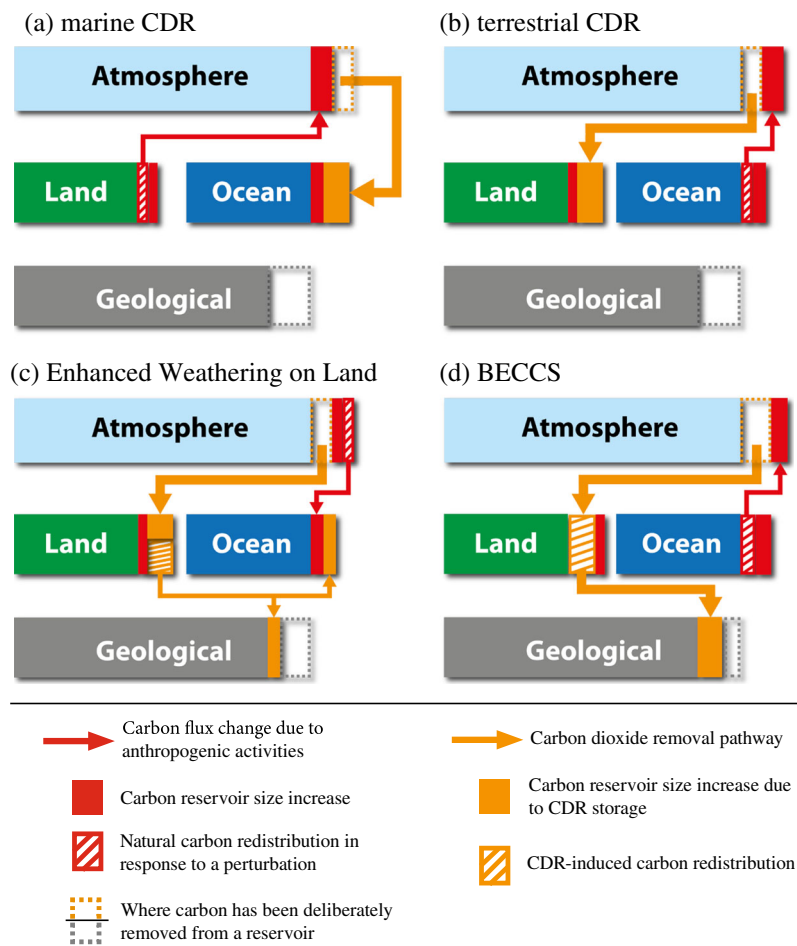
case (steady-state vs. a perturbed state), there are likely to be large regional differences within reservoirs due to the CDR carbon cycle responses, e.g., in terrestrial photosynthesis and respiration or oceanic air-sea gas exchange [18••, 20].

However, in the longer term (centuries to millennia), the carbon cycle response to the same cumulative amount of CDR (if permanently removed) appears to be pathway independent (if the background climate and carbon cycle state are the same), and determined only by cumulative CO<sub>2</sub> emissions [18••]. However, this does not account for the possibility of there being un-represented thresholds or tipping elements in the climate system [38].

## CDR Method-Specific Carbon Cycle Responses

Methods proposed to enhance ocean or land carbon sinks aim to increase the flux of carbon from the atmosphere into that reservoir (Tables 1 and 2). More technical CDR methods, including those that combine natural processes and technology, are designed to directly remove carbon from a reservoir and isolate it from the climate system, e.g., geological reservoir storage [2]. For almost all evaluated methods, this results

**Fig. 2** Schematic representation of how carbon flows among atmospheric, land, ocean, and geological reservoirs may change due to prolonged “net negative emission” carbon dioxide removal (CDR) from (a) marine sink enhancement methods such as ocean fertilization or alkalinity enhancement; (b) terrestrial sink enhancement methods such as afforestation/reforestation, biochar, or soil carbon management; (c) enhanced weathering on land; and (d) bioenergy with carbon capture and storage (BECCS). All carbon exchanges depicted in Fig. 1a also occur here. Note that when net emissions are negative, it is still possible to have some fossil fuel emissions, but these are not depicted here. Adapted from [1••, 71•]



in reduced carbon uptake or even carbon loss from another reservoir (Figs. 1 and 2). All methods will also either directly or indirectly redistribute carbon within reservoirs. For many methods, storage is an issue as sequestration sites, e.g., soils, may have limited capacity or only temporarily or inefficiently hold the carbon, i.e., the issue of permanence [2, 71•].

### Ocean-Based CDR

Artificial ocean alkalization (AOA), a proposed CDR method that aims to reduce surface seawater  $p\text{CO}_2$  and thereby enhance oceanic carbon uptake (Table 1), has recently been reviewed in several publications [69, 72, 73••]. Model simulations of AOA [37•, 39, 74–81] have found AOA to have, theoretically, a high carbon sequestration potential, and also explored how it might affect ocean carbonate chemistry. However, for the most widely proposed practical application of AOA using crushed olivine minerals, the chemistry is more complex than the simplified addition of alkalinity simulated by most models [82, 83]. Only a few studies have investigated the additional fertilizing effect of nutrient additions from olivine-based AOA [75, 76]. Albright et al. [84] recently explored AOA in a field setting at very small scale by adding alkalinity to a reef and found that it is possible to locally return seawater chemistry to a near pre-industrial state. Studies with Earth system models have all shown that in high  $\text{CO}_2$  emission scenarios, the terrestrial biosphere loses carbon, via climate-carbon cycle feedbacks, as atmospheric  $\text{CO}_2$  and temperatures decrease in response to increasing oceanic  $\text{CO}_2$  uptake [37•, 39] (Fig. 2a). Furthermore, the mean state of the ocean carbonate chemistry determines the magnitude of the Earth system response to AOA [80]. There appears to be little cessation effect as  $\text{CO}_2$  taken up by the ocean as a result of AOA remains in the ocean upon AOA termination [37•, 72, 77], although any olivine-related fertilizing effects are temporary [75].

Enhancement of the biological pump as a means of atmospheric CDR has centered on macro- and micro-nutrient ocean fertilization (Table 1). Keller et al. [37•] found that as in many earlier studies, there is limited potential for iron fertilization to draw down  $\text{CO}_2$ . Moreover, it has recently been observed in regions of natural iron fertilization that the carbonate counter pump (i.e., the biological precipitation of the carbonate shells of some species, followed by sinking of particulate inorganic carbon, which increases surface ocean  $p\text{CO}_2$  on centennial to millennial timescales) may reduce iron fertilization-induced C sequestration by 6–32% [85]. Oschlies et al. [86] and Keller et al. [37•] show that when iron is added during a high  $\text{CO}_2$  emission scenario, fertilization-induced atmospheric  $\text{CO}_2$  drawdown is initially opposed by a net loss of carbon from the terrestrial biosphere (Fig. 2a), via climate-carbon cycle feedbacks, as atmospheric  $\text{CO}_2$  and temperatures decrease. This phenomenon lasts until it is countered by a decreasing

sequestration efficacy (i.e., fertilization becomes less effective at enhancing oceanic carbon uptake) and increased terrestrial uptake as a result of  $\text{CO}_2$  fertilization. They also show that in the fertilized region there is a decline in pH and an increase in  $p\text{CO}_2$ . Carbon cycling is also affected in other regions of the ocean because of reduced nutrient availability following enhanced biological production. Furthermore, if ocean iron fertilization is terminated, carbon may remain in the ocean for a few decades [37•], but after 100 years up to a third of it could be potentially returned to the atmosphere [87].

Other studies have estimated the carbon sequestration potential of nitrogen and phosphorus fertilization [88–90]. While these macro-nutrient fertilization studies explore some of the factors that would determine carbon sequestration and estimate ocean carbon uptake, they do not fully explore the response of the global carbon cycle.

Artificial ocean upwelling (Table 1) as a biological pump enhancement has been investigated by Oschlies et al. [91] and reviewed by Bauman et al. [40] and Pan et al. [41]. Many of the impacts are the same as for ocean fertilization (see above) since it is the goal of this method to fertilize upper ocean biology with upwelled nutrients. However, upwelling also directly affects the carbon cycle since in most places the deep water that is pumped up contains more  $\text{CO}_2$  than the surface ocean, part of which may outgas into the atmosphere if the additional C is not consumed by phytoplankton. Deeper waters are also cooler than the surface ocean, and when pumped up have a cooling effect. In model simulations, this has been shown to increase terrestrial carbon uptake via reduced heterotrophic respiration at lower temperatures, i.e., there is no terrestrial loss of C as with other ocean-based CDR methods depicted in Fig 2a [37•]. Indeed, in model studies, artificial upwelling sequesters most (80% on centennial timescales) of the carbon drawn down from the atmosphere on land [88], and may therefore not be viewed primarily as ocean method. When artificial upwelling is stopped, the sequestered carbon is returned to the atmosphere within a few decades, and readjustment of the planetary energy budget leads to global mean temperatures even higher than in a world that had never experienced artificial upwelling [91].

### Terrestrial CDR

Although there are many studies on afforestation/reforestation (Table 1), very few of them quantify how the global carbon cycle responds from a CDR perspective. Most studies have instead focused either on carbon sequestration potential [42–44] and/or on the climatic trade-off between terrestrial  $\text{CO}_2$  uptake and biogeophysical change (e.g., in albedo) [1••, 37•, 45–50]. This trade-off determines how effective afforestation/reforestation is at cooling the climate and so impacts temperature-sensitive carbon cycle processes. Sonntag et al. [36, 79] recently showed that during simulated



reforestation climate-carbon cycle feedbacks, such as CO<sub>2</sub> fertilization, only slightly offset carbon sequestration and reductions in atmospheric CO<sub>2</sub>. Studies [37•, 51] have also investigated how atmospheric circulation and the water cycle would change with irrigated afforestation, which has carbon cycle implications. To date, the evidence indicates that strengthening the terrestrial carbon sink via afforestation/reforestation will weaken the ocean sink or even make it a CO<sub>2</sub> source as atmospheric CO<sub>2</sub> decreases (Fig. 2b) [22, 37•, 52, 60, 70]. Afforestation/reforestation also has limited carbon storage capacity and even after saturation (i.e., when the forest reaches a state of equilibrium and carbon uptake is balanced by loss), management must continue to prevent disturbances from releasing sequestered carbon [71•].

Much of what is known about how enhanced terrestrial weathering (Table 1) would affect the global carbon cycle is derived from our understanding of natural chemical weathering [54, 69, 73••] as well as recent modeling studies [68]. Weathering alters soil chemistry (e.g., increasing pH), which may increase nutrient availability, and can potentially add toxic trace metals to the environment. Soil composition may also be altered by the formation of minerals. These effects can impact vegetation, often positively (i.e., enhancing growth), and lead to changes in soil hydrology and carbon cycling. Nearby water bodies will be affected as well when weathering products are transported into them. Carbon sequestration estimates are poorly constrained, although likely low [1••] and with the exception of Taylor et al. [68] most studies do not account for global carbon cycle responses. To date, our understanding of the carbon cycle response to enhanced terrestrial weathering is that it removes CO<sub>2</sub> from the atmosphere and redistributes it to the land, ocean, and geological reservoirs (Fig 2c). This redistribution may initially cause the ocean carbon sink to weaken or even to change from a sink to a source of CO<sub>2</sub> as atmospheric CO<sub>2</sub> decreases [68]. However, if enough weathering products are transported to the ocean such that alkalinity increases, this will eventually enhance ocean carbon uptake (see section on AOA). Most sequestered carbon will ultimately be permanently stored in the ocean or geological formations and the method is not subject to reversal upon cessation [68, 73••].

The effects of biochar or other measures to manage soil carbon (Table 1) on the global carbon cycle are poorly known, with almost all studies limited to impacts on the terrestrial carbon cycle and GHG exchange with the atmosphere [61, 62]. These measures, which are too numerous to review individually here, affect the terrestrial carbon cycle and atmosphere mainly through biophysical (e.g., albedo, hydrological) and biogeochemical (e.g., nutrients, the release of other GHGs) processes [23, 61–65, 71•, 120]. Although biochar can also be made from marine biomass [55, 56, 66], no studies have investigated how transferring large amounts of carbon

from the ocean to land would impact carbon cycling. There have been many estimates of the carbon sequestration potential of these methods [24, 25, 57–59, 62, 65, 67, 71•], with most suggesting a low total potential. CDR through soil carbon management, and to a lesser extent biochar, is limited by terrestrial carbon storage capacity and even after saturation (i.e., when the system reaches a state of equilibrium and carbon uptake is balanced by loss) some management must continue or leakage could occur [71•]. To date, our understanding of the global carbon cycle response to biochar or soil carbon management suggests that strengthening the terrestrial carbon sink should weaken the ocean carbon sink or lead to outgassing as atmospheric CO<sub>2</sub> decreases (Fig. 2b).

BECCS (Table 1) is the CDR method most often assumed in climate change mitigation scenarios [26] and there are many recent carbon sequestration potential estimates [1••, 27–30, 60, 65]. However, as far as we are aware, global carbon cycle responses to BECCS have only been fully quantified by Muri [31], although there have been offline modeling efforts [60]. Research instead usually focuses either on biomass plantations (simulated with offline terrestrial models [29, 30, 32–34, 60]) or the ESM response to atmospheric CO<sub>2</sub> removal, where BECCS is treated as DAC, e.g., Jones et al. [15••]. In these DAC-like studies, biomass plantations may be prescribed, but the harvest products are returned to the terrestrial carbon pool as litter or crop residues, rather than being removed from the simulated carbon cycle. Consequently, these studies likely underestimate soil carbon losses resulting from bioenergy cropland expansion. The impacts of bioenergy plantations on soil carbon stocks are likely to be mixed and depend heavily on previous land cover [35]. Increases in soil carbon can remove CO<sub>2</sub> from the atmosphere, representing a carbon sink regardless of whether CCS is performed on the harvested biomass, e.g., as shown in simulations where biomass is grown to substitute bioenergy for fossil fuels [127]. However, carbon stored in soils is vulnerable to future changes in land use and does not represent the same permanence as geological CO<sub>2</sub> storage. All terrestrial biomass plantation studies have shown that there would be biogeophysical (e.g., albedo, hydrological) and biogeochemical (e.g., nutrients) changes that impacts the carbon cycle [1••, 28–30, 33, 34, 65]. From a global carbon cycle perspective, the net result of terrestrial-based BECCS is to move carbon from the atmosphere to the terrestrial biosphere and then to permanent geological storage (Fig. 2d). This will weaken the ocean carbon sink or even result in it switching to a CO<sub>2</sub> source as atmospheric CO<sub>2</sub> decreases. Note that there have also been proposals to utilize marine biomass (macroalgae or microalgae) as the BECCS fuel feedstock [101, 102, 128]. This would have different impacts on the carbon cycle than the use of a terrestrial feedstock, but these have not yet been quantified.

## CDR Carbon Storage and the Carbon Cycle

Ideally, any captured carbon will be permanently stored (e.g., in geological reservoirs), and this underpins all technical methods, e.g., BECCS and DAC (Fig. 2c, d). Permanent storage capacity for sequestered CO<sub>2</sub> appears adequate to technically match current fossil fuel reserves (i.e., the amount that could be emitted and countered by CDR) [2]. However, it is not clear whether storage can be accessed fast enough or is appropriately located to meet the mitigation demands for fossil fuels with carbon capture and storage and CDR, so the use of more temporary storage sites has also been investigated. Reith et al. [104] recently revisited the idea of using the deep ocean as a site to store carbon from DAC and quantified leakage rates and terrestrial and ocean carbon cycle responses. They found that their targeted atmospheric CO<sub>2</sub> reductions, which were equivalent to the amounts of CO<sub>2</sub> injected, were 16–30% lower on decadal to centennial timescales because of leakage and land and ocean carbon back fluxes that occurred in response to lowering atmospheric CO<sub>2</sub>. Responses like this need to be taken into account when considering any storage sites that are not permanent.

## Solar Radiation Management Effects on the Carbon Cycle

Solar radiation management (SRM), a term that describes methods that attempt to reduce climate warming by increasing the reflection of incoming short-wave solar radiation back into space, is not intended to reduce CO<sub>2</sub>. However, it may do so by impacting the carbon cycle via reduced surface temperatures, altered diffusive radiation, or by limiting carbon cycle feedbacks that amplify climate responses to emissions (i.e., feedbacks that increase atmospheric CO<sub>2</sub>). These effects are not explored here, but reviewed by Cao [105].

## Knowledge Gaps

There are many carbon cycle uncertainties for all the CDR methods discussed above. Uncertainties may be highest for methods that have been investigated in only a few studies, or are merely conceptual proposals (Table 2). Such methods include artificial downwelling [106, 107], coastal (blue carbon) sink management [66, 98–100, 109], cloud alkalization [110], burial of terrestrial biomass on land or in the deep ocean [103, 108, 129], iron salt aerosol additions [130], storing carbon in structural materials [103, 131], (e.g., by building with wood or carbon absorbing cements or utilizing it in other products), or extracting CO<sub>2</sub> directly from seawater [132]. There is also little knowledge about the effects that may emerge when any of these methods is scaled up to have globally significant impact. These include not only biogeochemical but also

ecological as well as economical feedbacks. Biological methods will directly have to account for not only possible impacts on ecology, but also on possible effects of CDR on climate trajectories and therefore the different capacity of species to adapt and evolve.

## Summary

When carbon is deliberately transferred to or removed from a reservoir, the carbon cycle responds by redistributing the carbon in other reservoirs (Figs. 1 and 2), as well as within them, as biological, chemical, and physical processes adjust to the changing quantities of carbon within the reservoir and steeper gradients at the reservoir interfaces. By reducing the greenhouse effect, which leads to cooling, CDR also triggers climate-carbon cycle feedbacks. Together, these responses affect the efficacy of CDR and mean that removing 1 Gt of CO<sub>2</sub> from the atmosphere will not ultimately reduce the atmospheric CO<sub>2</sub> concentration by 1 Gt. The short-term efficacy of any CDR method depends on the state of the climate system and the carbon cycle. For many methods, environmental factors (e.g., light or nutrient availability) are also important. However, in the long run (multi-millennial) for methods that can permanently sequester carbon, it appears that the carbon cycle response to the same cumulative amounts of CDR is, to first order, pathway independent (if the background climate and carbon cycle state is the same) and determined only by cumulative CO<sub>2</sub> emissions. Although, this does not account for the possibility of there being thresholds or tipping elements in the climate system [38].

In addition to the general response of the carbon cycle to CDR (i.e., reservoir scale redistribution), there are a wide variety of CDR method-specific carbon cycle impacts. Many of these significantly alter the carbon cycle at the site of deployment, but they can also have large regional or even global-scale impacts. Storage is an important issue for most methods as sequestration sites might only temporarily or inefficiently hold the carbon and/or have limited capacity [2].

Although CDR has been simulated in idealized scenarios that remove massive amounts of carbon in a short time, studies that apply more realistic constraints show that most sequestration rates are likely to be low and limited by many factors, such as land area or local environmental conditions (e.g., temperature, hydrology, chemistry) that impact reaction or growth rates [1•, 28, 29, 36, 37•, 68, 71•, 73••]. This does not mean that CDR could not potentially work, just that most methods are relatively slow acting and might take many decades to centuries to reduce atmospheric CO<sub>2</sub> to some desired level. The few methods that have the theoretical potential to rapidly remove more CO<sub>2</sub> remain technologically immature and would likely take decades to be fully developed and deployed at climatically relevant scales.

## Conclusion

CDR is increasingly and widely applied in global and regional decarbonization scenarios used to inform policy and economic discussions on climate change mitigation. As a result, improved understanding of the response of the Earth system to CDR is needed to inform the research and development, possible policy frameworks, and the proper accounting of CDR.

For anthropogenic CO<sub>2</sub> emissions, it is well established that only a fraction of the emissions effectively remains in the atmosphere because carbon cycle feedbacks adjust the carbon cycle to the anthropogenic perturbation by redistributing carbon among the interacting oceanic, terrestrial, and atmosphere reservoirs. Because of the different response times of terrestrial and oceanic carbon reservoirs, this partitioning changes with time and also depends on the state of the climate system. The same processes apply when CO<sub>2</sub> is removed from the atmosphere, yielding an airborne fraction of CO<sub>2</sub> removal, i.e., the perturbed airborne fraction [15••]. As a corollary, taking CO<sub>2</sub> out of the atmosphere will lead to backfluxes of CO<sub>2</sub> from the ocean and the land. The services that the land and ocean have provided in taking up a substantial portion of anthropogenic CO<sub>2</sub> emissions can be viewed as a “carbon debt” that will have to be started to be paid back if large-scale CDR is ever deployed.

The carbon cycle response to CDR and its impact on CDR efficacy will also have to be considered in the comparative assessment of different methods. Two options for estimating efficacy appear straightforward: measuring CDR in terms of (i) carbon stored (e.g., in biomass or chemically inert forms of carbon on land or in the ocean) or (ii) net changes in atmospheric CO<sub>2</sub>. The first option (i) is equivalent to measuring the local fluxes of CO<sub>2</sub> from the atmosphere, and thus corresponds to the way anthropogenic CO<sub>2</sub> emissions are measured. For monitoring and verification of CDR, accounting based on local measurements would, however, pose substantial challenges, especially if future leakage (lack of permanence) is possible. The latter option (ii) can more easily be quantified. However, attributing net changes in atmospheric CO<sub>2</sub> to any particular CDR method, especially if multiple methods were deployed, would be problematic. It is likely that aggregated methods will be required to verify and monitor any CDR deployment that, according to current understanding, will be required to meet internationally agreed climate goals. The carbon cycle responses discussed in this review will have to be carefully considered in assessments of CDR options and, if deployed, in the accounting process.

Our literature review shows that relatively few studies have quantified how the global carbon cycle responds to CDR. Far more studies quantify gross carbon sequestration than net atmospheric CO<sub>2</sub> reduction that accounts for the response of other reservoirs. This is likely because global carbon cycle

responses can only be evaluated with Earth system models that include these reservoirs, while gross carbon sequestration and local carbon cycle impacts can often be quantified with laboratory or field experiments and simpler models that resolve only parts of the full Earth system. Nonetheless, net CDR efficacy must be properly evaluated as it lies at the heart of the climate mitigation potential of CDR.

This review has made it clear that further and targeted research is urgently needed to better constrain the carbon cycle response to CDR. This research should focus on: (i) understanding the underlying mechanisms involved in each CDR method (i.e., improving the quantification of relevant biological, chemical, and physical processes); (ii) understanding and reducing the large model spread that occurs when the carbon cycle response to perturbations is simulated (e.g., reducing the uncertainty in the sensitivity of carbon fluxes and storage to environmental conditions); (iii) quantifying how reversible the climate and carbon cycle are with respect to perturbations by CO<sub>2</sub> emissions and CDR; and (iv) developing a methodology to evaluate and monitor CDR and its impacts on the carbon cycle and climate system, in order to allow for reliable accounting of CDR contributions to managing atmospheric CO<sub>2</sub> concentrations.

For (i) some understanding can be gained through more detailed simulations, however, in many cases laboratory, mesocosm, or field experiments will be needed to make advances. In combination with model assessment and intercomparison studies (such as the carbon dioxide removal model intercomparison project, CDRMIP; [126]), this will help to reduce uncertainties in simulating CDR (e.g., by providing better parameterizations and by developing more realistic models). Model intercomparison and perturbed parameter ensemble studies will also provide information to address (ii) and eventually our representation of CDR in Earth system simulations, thereby providing increasing confidence in evaluating reversibility (iii) using models. Such advances require well-funded and sustained international research programs.

Given the potential large impacts (side effects), costs, and deployment requirements (e.g., land area) of many methods, an interdisciplinary approach not limited to social sciences, international law, and environmental ethics is required to develop the relevant assessment methods and tools needed for (iv). Some of this research overlaps with what is already being done to address the World Climate Research Program Grand Science Challenges; however, many aspects of CDR research are unique and will require building new interdisciplinary research consortia and dedicated research programs.

**Funding Information** D. P. Keller and Andreas Oschlies received funding from the German Research Foundation’s Priority Program SPP 1689 “Climate Engineering” (projects ComparCE-2 and CDR-MIA; KE 2149/2-1). V. Scott, E.W. Littleton, and N.E. Vaughan received funding from the UK Greenhouse Gas Removal Programme (projects MERLiN: NE/P019749/1 and FAB-GGR: NE/P019951/1, respectively).

## Compliance with Ethical Standards

**Conflict of Interest** The authors state that they have no financial or personal relationships with any third party whose interests could be positively or negatively influenced by the article's content.

**Open Access** This article is distributed under the terms of the Creative Commons Attribution 4.0 International License (<http://creativecommons.org/licenses/by/4.0/>), which permits unrestricted use, distribution, and reproduction in any medium, provided you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made.

## References

Papers of particular interest, published recently, have been highlighted as:

- Of importance
- Of major importance

1. Smith P, Davis SJ, Creutzig F, Fuss S, Minx J, Gabrielle B, et al. Biophysical and economic limits to negative CO<sub>2</sub> emissions. *Nat Clim Chang* [Internet]. 2015;6:42–50. Available from: <http://www.nature.com/doi/10.1038/nclimate2870>. **A review of multiple CDR method sequestration potentials.**
2. Scott V, Haszeldine RS, Tett SFB, Oschlies A. Fossil fuels in a trillion tonne world. *Nat Clim Chang* [Internet]. Nature Publishing Group; 2015;5:419–23. Available from: <http://www.nature.com/doi/10.1038/nclimate2578>.
3. IPCC. Climate Change 2013: The physical science basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change. 2013.
4. IPCC. Climate change 2014: impacts, adaptation, and vulnerability. Part A: Global and Sectoral Aspects. Contribution of Working Group II to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change [Field, C.B., V.R. Barros, D.J. Dokken, K.J. Cambridge, United Kingdom and New York, NY, USA: Cambridge University Press; 2014.
5. UNFCCC. Paris Agreement of the 21st session of the Conference of Parties on climate change. 2016.
6. UNEP. The emissions gap report 2017. Nairobi; 2017.
7. Rogelj J, Fricko O, Meinshausen M, Krey V, Zilliacus JJJ, Riahi K. Understanding the origin of Paris agreement emission uncertainties. *Nat Commun* [Internet]. Nature Publishing Group; 2017;8:15748. Available from: <http://www.nature.com/doi/10.1038/ncomms15748>; <http://www.ncbi.nlm.nih.gov/pubmed/28585924>.
8. Sanderson BM, O'Neill B, Tebaldi C. What would it take to achieve the Paris temperature targets? *Geophys Res Lett*. 2016;1–10.
9. Parson EA. Opinion: climate policymakers and assessments must get serious about climate engineering. *Proc Natl Acad Sci* [Internet]. 2017;114:9227–30. Available from: <http://www.pnas.org/lookup/doi/10.1073/pnas.1713456114>.
10. Hansen J, Sato M, Kharecha P, von Schuckmann K, Beerling DJ, Cao J, et al. Young people's burden: requirement of negative CO<sub>2</sub> emissions. *Earth Syst Dyn* [Internet]. 2017;8:577–616. Available from: <https://www.earth-syst-dynam.net/8/577/2017/>.
11. Sanz-Pérez ES, Murdock CR, Didas SA, Jones CW. Direct capture of CO<sub>2</sub> from ambient air. *Chem Rev* [Internet]. 2016;116:11840–76. Available from: <http://pubs.acs.org/doi/abs/10.1021/acs.chemrev.6b00173>.
12. Regnier P, Friedlingstein P, Ciais P, Mackenzie FT, Gruber N, Janssens IA, et al. Anthropogenic perturbation of the carbon fluxes from land to ocean. *Nat Geosci* [Internet]. Nature Publishing Group, a division of Macmillan Publishers Limited. All Rights Reserved. 2013;6:597–607. Available from: <https://doi.org/10.1038/ngeo1830>.
13. Cao L, Caldeira K. Atmospheric carbon dioxide removal: long-term consequences and commitment. *Environ Res Lett* [Internet]. 2010 [cited 2015 Jan 9];5:024011. Available from: <http://stacks.iop.org/1748-9326/5/i=2/a=024011?key=crossref.5df1131fed99d0e3ec424b0060b26207>.
14. Boucher O, Halloran PR, Burke EJ, Doutriaux-Boucher M, Jones CD, Lowe J, et al. Reversibility in an earth system model in response to CO<sub>2</sub> concentration changes. *Environ Res Lett* [Internet]. 2012 [cited 2014 Sep 8];7:024013. Available from: <http://stacks.iop.org/1748-9326/7/i=2/a=024013?key=crossref.eb2eeb26bbd2088c8ec21123358988c9>.
15. Jones CD, Ciais P, Davis SJ, Friedlingstein P, Gasser T, Peters GP, et al. Simulating the earth system response to negative emissions. *Environ Res Lett* [Internet]. IOP Publishing; 2016;11:095012. Available from: <http://stacks.iop.org/1748-9326/11/i=9/a=095012?key=crossref.6b5747055a178d1c59ffa940adb33091>. **A multi-model comparison of the carbon cycle response to CDR. They introduce a new metric to measure and assess the effectiveness of negative emissions.**
16. Herrington T, Zickfeld K. Path independence of climate and carbon cycle response over a broad range of cumulative carbon emissions. *Earth Syst Dyn* [Internet]. 2014 [cited 2015 Jan 12];5:409–22. Available from: <http://www.earth-syst-dynam.net/5/409/2014/>.
17. Zickfeld K, MacDougall AH, Matthews HD. On the proportionality between global temperature change and cumulative CO<sub>2</sub> emissions during periods of net negative CO<sub>2</sub> emissions. *Environ Res Lett* [Internet]. IOP Publishing; 2016;11:055006. Available from: <http://stacks.iop.org/1748-9326/11/i=5/a=055006?key=crossref.636dc44f6dba4b8a51aeff076afe0be1>. **A study that systematically investigates the response of the carbon cycle and global mean temperature to CDR.**
18. Tokarska KB, Zickfeld K. The effectiveness of net negative carbon dioxide emissions in reversing anthropogenic climate change. *Environ Res Lett* [Internet]. IOP Publishing; 2015;10:094013. Available from: <https://doi.org/10.1088/1748-9326/10/9/094013>. **A study that investigates how the pathways of CDR effect the carbon cycle and climate response.**
19. MacDougall AH. Reversing climate warming by artificial atmospheric carbon-dioxide removal: can a Holocene-like climate be restored? *Geophys Res Lett* [Internet]. 2013 [cited 2013 Oct 18];40:5480–5. Available from: <http://doi.wiley.com/10.1002/2013GL057467>.
20. Vichi M, Navarra A, Fogli PG. Adjustment of the natural ocean carbon cycle to negative emission rates. *Clim Chang* [Internet]. 2013;118:105–18. Available from: <http://link.springer.com/10.1007/s10584-012-0677-0>
21. Mathesius S, Hofmann M, Caldeira K, Schellnhuber HJ. Long-term response of oceans to CO<sub>2</sub> removal from the atmosphere. *Nat Clim Chang* [Internet]. 2015;5:1107–13. Available from: <http://www.nature.com/doi/10.1038/nclimate2729>
22. House JI, Colin Prentice I, Le Quere C. Maximum impacts of future reforestation or deforestation on atmospheric CO<sub>2</sub>. *Glob Chang Biol* [Internet]. 2002;8:1047–52. Available from: <http://doi.wiley.com/10.1046/j.1365-2486.2002.00536.x>
23. Fan C, Chen H, Li B, Xiong Z. Biochar reduces yield-scaled emissions of reactive nitrogen gases from vegetable soils across

- China. *Biogeosciences*. 2017;14(11):2851–63. <https://doi.org/10.5194/bg-14-2851-2017>
24. (Han) Weng Z, Van Zwieten L, Singh BP, Tavakkoli E, Joseph S, Macdonald LM, et al. Biochar built soil carbon over a decade by stabilizing rhizodeposits. *Nat Clim Chang* [Internet]. 2017;7. Available from: <http://www.nature.com/doi/10.1038/nclimate3276>.
  25. Van Groenigen JW, Van Kessel C, Hungate BA, Oenema O, Powlson DS, Van Groenigen KJ. Sequestering soil organic carbon: a nitrogen dilemma. *Environ Sci Technol*. 2017;51:4738–9.
  26. Fuss S, Canadell JG, Peters GP, Tavoni M, Andrew RM, Ciais P, et al. Betting on negative emissions. *Nat Clim Chang* [internet]. Nature Publishing Group; 2014;4:850–3. Available from: <https://doi.org/10.1038/nclimate2392>.
  27. Tokimatsu K, Yasuoka R, Nishio M. Global zero emissions scenarios: the role of biomass energy with carbon capture and storage by forested land use. *Appl Energy* [Internet]. Elsevier Ltd; 2017;185:1899–906. Available from: <http://linkinghub.elsevier.com/retrieve/pii/S0306261915015275>.
  28. Boysen LR, Lucht W, Gerten D, Heck V, Lenton TM, Schellnhuber HJ. The limits to global-warming mitigation by terrestrial carbon removal. *Earth's Futur* [Internet]. 2017;1–12. Available from: <http://doi.wiley.com/10.1002/2016EF000469>.
  29. Boysen LR, Lucht W, Gerten D, Heck V. Impacts devalue the potential of large-scale terrestrial CO<sub>2</sub> removal through biomass plantations. *Environ Res Lett* [Internet]. IOP Publishing; 2016;11:095010. Available from: <http://stacks.iop.org/1748-9326/11/i=9/a=095010?key=crossref.244d75c1a47b210f79f0a1cb65588a46>.
  30. Boysen LR, Lucht W, Gerten D. Trade-offs for food production, nature conservation and climate limit the terrestrial carbon dioxide removal potential. *Glob Chang Biol*. 2017;23:4303–17.
  31. Muri H. The role of large-scale BECCS in the pursuit of the 1.5°C target: an Earth system model perspective. *Environ Res Lett* [Internet]. 2018;13(4):044010. <https://doi.org/10.1088/1748-9326/aab324>.
  32. Boysen LR, Brovkin V, Arora VK, Cadule P, De Noblet-Ducoudré N, Kato E, et al. Global and regional effects of land-use change on climate in 21st century simulations with interactive carbon cycle. *Earth Syst Dyn*. 2014;5:309–19.
  33. Heck V, Hoff H, Wiersenus S, Meyer C, Kreft H. Land use options for staying within the planetary boundaries—synergies and trade-offs between global and local sustainability goals. *Glob Environ Chang* [Internet]. 2018;49:73–84. Available from: <http://linkinghub.elsevier.com/retrieve/pii/S0959378017300249>
  34. Heck V, Gerten D, Lucht W, Popp A. Biomass-based negative emissions difficult to reconcile with planetary boundaries. *Nat Clim Chang* [Internet]. Springer US; 2018;8:151–5. Available from: <https://doi.org/10.1038/s41558-017-0064-y>.
  35. Beringer T, Lucht W, Schaphoff S. Bioenergy production potential of global biomass plantations under environmental and agricultural constraints. *GCB Bioenergy*. 2011;3:299–312.
  36. Sonntag S, Pongratz J, Reick CH, Schmidt H. Reforestation in a high-CO<sub>2</sub> world—Higher mitigation potential than expected, lower adaptation potential than hoped for. *Geophys Res Lett* [Internet]. 2016;43:6546–53. Available from: <http://doi.wiley.com/10.1002/2016GL068824>.
  37. Keller DP, Feng EY, Oeschles A. Potential climate engineering effectiveness and side effects during a high carbon dioxide-emission scenario. *Nat Commun* [Internet]. Nature Publishing Group; 2014 [cited 2014 Feb 26];5:1–11. Available from: <http://www.nature.com/doi/10.1038/ncomms4304>. **A comparative evaluation of different CDR measures that includes an assessment of carbon cycle feedbacks.**
  38. Lenton TM, Held H, Kriegler E, Hall JW, Lucht W, Rahmstorf S, et al. Tipping elements in the Earth's climate system. *Proc Natl Acad Sci* [Internet]. 2008;105:1786–93. Available from: <http://www.pnas.org/cgi/doi/10.1073/pnas.0705414105>.
  39. González MF, Ilyina T. Impacts of artificial ocean alkalization on the carbon cycle and climate in earth system simulations. *Geophys Res Lett* [Internet]. 2016;43:6493–502. Available from: <http://doi.wiley.com/10.1002/2016GL068576>
  40. Bauman S, Costa M, Fong M, House B, Perez E, Tan M, et al. Augmenting the biological pump: the shortcomings of geoengineered upwelling. *Oceanography* [Internet]. 2014;27:17–23. Available from: <https://tos.org/oceanography/article/augmenting-the-biological-pump-the-shortcomings-of-geoengineered-upwelling>
  41. Pan Y, Fan W, Zhang D, Chen J, Huang H, Liu S, et al. Research progress in artificial upwelling and its potential environmental effects. *Sci China Earth Sci* [Internet]. 2016;59:236–48. Available from: <http://link.springer.com/10.1007/s11430-015-5195-2>
  42. Kreidenweis U, Humpenöder F, Stevanović M, Bodirsky BL, Kriegler E, Lotze-Campen H, et al. Afforestation to mitigate climate change: impacts on food prices under consideration of albedo effects. *Environ Res Lett* [Internet]. 2016;11:085001. Available from: <http://stacks.iop.org/1748-9326/11/i=8/a=085001?key=crossref.498dab12c59b27f71805e8cdabaf36f1>.
  43. Humpenöder F, Popp A, Stevanovic M, Müller C, Bodirsky BL, Bonsch M, et al. Land-use and carbon cycle responses to moderate climate change: implications for land-based mitigation? *Environ Sci Technol*. 2015;49:6731–9.
  44. Popp A, Humpenöder F, Weindl I, Bodirsky BL, Bonsch M, Lotze-campen H, et al. Land-use protection for climate change mitigation. 2014;4:2–5.
  45. Cheng W, Moore JC, Cao L, Ji D, Zhao L. Simulated climate effects of desert irrigation geoengineering. *Sci Rep* [Internet]. Nature Publishing Group. 2017;7:46443. Available from: <http://www.nature.com/articles/srep46443>.
  46. Davies-Barnard T, Valdes PJ, Singarayer JS, Pacifico FM, Jones CD. Full effects of land use change in the representative concentration pathways. *Environ Res Lett* [Internet]. 2014;9:114014. Available from: <http://stacks.iop.org/1748-9326/9/i=11/a=114014>.
  47. Mykleby P, Snyder P, Twine T. Quantifying the tradeoff between carbon sequestration and albedo in mid and high latitude North American forests. *Geophys Res Lett*. 2017;2493–501.
  48. Naudts K, Chen Y, McGrath MJ, Ryder J, Valade A, Otto J, et al. Europe's forest management did not mitigate climate warming. *Science* (80-. ). [Internet]. 2016;351:597–600. Available from: <http://www.sciencemag.org/cgi/doi/10.1126/science.aad7270>.
  49. Wang Y, Yan X, Wang Z. Effects of regional afforestation on global climate. *J Water Clim Chang* [Internet]. 2015;6:191. Available from: <http://jwcc.iwaponline.com/cgi/doi/10.2166/wcc.2014.136>
  50. Bathiany S, Claussen M, Brovkin V, Raddatz T, Gayler V. Combined biogeophysical and biogeochemical effects of large-scale forest cover changes in the MPI earth system model. *Biogeosciences*. 2010;7:1383–99.
  51. Kemena TP, Matthes K, Martin T, Wahl S, Oeschles A. Atmospheric feedbacks in North Africa from an irrigated, afforested Sahara. *Clim Dyn* [Internet]. Springer Berlin Heidelberg. 2017;0:0. Available from: <http://link.springer.com/10.1007/s00382-017-3890-8>.
  52. Ridgwell A, Maslin MA, Watson AJ. Reduced effectiveness of terrestrial carbon sequestration due to an antagonistic response of ocean productivity. *Geophys Res Lett*. 2002;29:19–1–4.
  53. Zhang Q, Wang YP, Matear RJ, Pitman AJ, Dai YJ. Nitrogen and phosphorous limitations significantly reduce future allowable CO<sub>2</sub> emissions. *Geophys Res Lett* [Internet]. 2014;41:632–7. Available from: <http://doi.wiley.com/10.1002/2013GL058352>.

54. Taylor LL, Beerling DJ, Quegan S, Banwart SA, Sheffield S, Taylor LL. Simulating carbon capture by enhanced weathering with croplands : an overview of key processes highlighting areas of future model development. 2017.
55. Roberts DA, Paul NA, Dworjanyn SA, Bird MI, de Nys R. Biochar from commercially cultivated seaweed for soil amelioration. *Sci Rep* [Internet]. 2015;5:9665. Available from: <http://www.nature.com/articles/srep09665>.
56. Schenk PM, Thomas-Hall SR, Stephens E, Marx UC, Mussnugn JH, Posten C, et al. Second generation biofuels: high-efficiency microalgae for biodiesel production. *BioEnergy Res*. 2008;1:20–43.
57. Sommer R, Bossio D. Dynamics and climate change mitigation potential of soil organic carbon sequestration. *J Environ Manag* [Internet]. Elsevier Ltd; 2014;144:83–7. Available from: <https://doi.org/10.1016/j.jenvman.2014.05.017>.
58. Poeplau C, Don A. Carbon sequestration in agricultural soils via cultivation of cover crops—a meta-analysis. *Agric Ecosyst Environ* [Internet]. Elsevier B.V.; 2015;200:33–41. Available from: <https://doi.org/10.1016/j.agee.2014.10.024>.
59. Sperow M. Estimating carbon sequestration potential on U.S. agricultural topsoils. *Soil Tillage Res* [Internet]. Elsevier B.V.; 2016;155:390–400. Available from: <https://doi.org/10.1016/j.still.2015.09.006>.
60. Heck V, Gerten D, Lucht W, Boysen LR. Is extensive terrestrial carbon dioxide removal a ‘green’ form of geoengineering? A global modelling study. *Glob Planet Chang* [Internet]. Elsevier B.V.; 2016;137:123–30. Available from: <https://doi.org/10.1016/j.gloplacha.2015.12.008>.
61. Zhang D, Yan M, Niu Y, Liu X, van Zwieten L, Chen D, et al. Is current biochar research addressing global soil constraints for sustainable agriculture? *Agric Ecosyst Environ* [Internet]. Elsevier B.V.; 2016;226:25–32. Available from: <http://linkinghub.elsevier.com/retrieve/pii/S0167880916302146>.
62. Paustian K, Lehmann J, Ogle S, Reay D, Robertson GP, Smith P. Climate-smart soils. *Nature* [Internet]. Nature Publishing Group; 2016;532:49–57. Available from: <http://www.nature.com/doi/10.1038/nature17174>.
63. Ravi S, Sharratt BS, Li J, Olshevski S, Meng Z, Zhang J. Particulate matter emissions from biochar-amended soils as a potential tradeoff to the negative emission potential. *Sci Rep* [Internet]. Nature Publishing Group; 2016;6:35984. Available from: <http://www.nature.com/articles/srep35984>.
64. Jeffery S, Abalos D, Prodana M, Bastos AC, van Groenigen JW, Hungate BA, et al. Biochar boosts tropical but not temperate crop yields. *Environ Res Lett* [Internet]. 2017;12:053001. Available from: <http://stacks.iop.org/1748-9326/12/i=5/a=053001?key=crossref.bf9e05e5fb214cb193e8f1c2d66a7ac9>.
65. Canadell JG, Schulze ED. Global potential of biospheric carbon management for climate mitigation. *Nat Commun* [Internet]. Nature Publishing Group; 2014 [cited 2014 Nov 23];5:5282. Available from: <http://www.ncbi.nlm.nih.gov/pubmed/25407959>.
66. Duarte CM, Wu J, Xiao X, Bruhn A, Krause-Jensen D. Can seaweed farming play a role in climate change mitigation and adaptation? *Front Mar Sci* [Internet]. 2017;4. Available from: <http://journal.frontiersin.org/article/10.3389/fmars.2017.00100/full>.
67. Powlson DS, Stirling CM, Jat ML, Gerard BG, Palm CA, Sanchez PA, et al. Limited potential of no-till agriculture for climate change mitigation. *Nat Clim Chang* [Internet]. Nature Publishing Group; 2014;4:678–83. Available from: <http://www.nature.com/doi/10.1038/nclimate2292>.
68. Taylor LL, Quirk J, Thorley RMS, Kharecha PA, Hansen J, Ridgwell A, et al. Enhanced weathering strategies for stabilizing climate and averting ocean acidification. *Nat Clim Chang* [Internet]. 2015; Available from: <http://www.nature.com/doi/10.1038/nclimate2882>.
69. Hartmann J, West J, Renforth P, Köhler P, De La Rocha CL, Wolf-Gladrow D, et al. Enhanced chemical weathering as a geoengineering strategy to reduce atmospheric carbon dioxide, a nutrient source and to mitigate ocean acidification. *Rev Geophys*. 2013;51:113–49.
70. Arora VK, Montenegro A. Small temperature benefits provided by realistic afforestation efforts. *Nat Geosci*. 2011;4:514–8.
71. Smith P. Soil carbon sequestration and biochar as negative emission technologies. *Glob Chang Biol* [Internet]. 2016;22:1315–24. Available from: <http://doi.wiley.com/10.1111/gcb.13178>. **A review of soil carbon and biochar CDR.**
72. Meysman FJR, Montserrat F. Negative CO<sub>2</sub> emissions via enhanced silicate weathering in coastal environments *Biol Lett* [Internet]. 2017;13:20160905. Available from: <http://rsbl.royalsocietypublishing.org/lookup/doi/10.1098/rsbl.2016.0905>.
73. Renforth P, Henderson G. Assessing ocean alkalinity for carbon sequestration. *Rev Geophys* [Internet]. 2017;55:636–74. Available from: <http://doi.wiley.com/10.1002/2016RG000533>. **A comprehensive review of artificial ocean alkalization.**
74. Feng EY, Keller DP, Koeve W, Oschlies A. Could artificial ocean alkalization protect tropical coral ecosystems from ocean acidification? *Environ Res Lett* [Internet]. 2016;11:074008. Available from: <http://stacks.iop.org/1748-9326/11/i=7/a=074008?key=crossref.70dd72536bcafae0d7e0da2e421b7917>.
75. Hauck J, Köhler P, Wolf-Gladrow D, Völker C. Iron fertilisation and century-scale effects of open ocean dissolution of olivine in a simulated CO<sub>2</sub> removal experiment. *Environ Res Lett* [Internet]. IOP Publishing; 2016;11:024007. Available from: <http://stacks.iop.org/1748-9326/11/i=2/a=024007?key=crossref.7d5ab8a17f2b2687fa59e1d3b4b66c6f>.
76. Köhler P, Abrams JF, Völker C, Hauck J, Wolf-Gladrow DA. Geoengineering impact of open ocean dissolution of olivine on atmospheric CO<sub>2</sub>, surface ocean pH and marine biology. *Environ Res Lett* [Internet]. 2013;8:14009. Available from: <http://stacks.iop.org/1748-9326/8/i=1/a=014009>.
77. Ilyina T, Wolf-Gladrow D, Munhoven G, Heinze C. Assessing the potential of calcium-based artificial ocean alkalization to mitigate rising atmospheric CO<sub>2</sub> and ocean acidification. *Geophys Res Lett* [Internet]. 2013 [cited 2013 Nov 11];40:5909–14. Available from: <http://doi.wiley.com/10.1002/2013GL057981>.
78. Paquay FS, Zeebe RE. Assessing possible consequences of ocean liming on ocean pH, atmospheric CO<sub>2</sub> concentration and associated costs. *Int J Greenh Gas Control*. 2013;17:183–8.
79. Sonntag S, González MF, Ilyina T, Kracher D, Nabel JEMS, Niemeier U, et al. Quantifying and comparing effects of climate engineering methods on the Earth system. *Earth’s Futur*. 2018;6(2):149–68. <https://doi.org/10.1002/2017EF000620>.
80. Lenton A, Matear RJ, Keller DP, Scott V, Vaughan NE. Assessing carbon dioxide removal through global and regional ocean alkalization under high and low emission pathways. *Earth Syst. Dyn.* [Internet]. 2018; Available from: <https://www.earth-system-dynam-discuss.net/esd-2017-92/>.
81. Feng EY, Koeve W, Keller DP, Oschlies A. Model-based assessment of the CO<sub>2</sub> sequestration potential of coastal ocean alkalization. *Earth’s Futur* [Internet]. 2017;5:1252–66. Available from: <http://doi.wiley.com/10.1002/2017EF000659>.
82. Montserrat F, Renforth P, Hartmann J, Knops P, Leermakers M, Meysman FJR. Olivine dissolution in seawater: implications for CO<sub>2</sub> sequestration through enhanced weathering in coastal environments. *Environ Sci Technol*. 2017;51:3960–72.
83. Griffioen J. Enhanced weathering of olivine in seawater: the efficiency as revealed by thermodynamic scenario analysis. *Sci Total Environ*. 2017;575:536–44.
84. Albright R, Caldeira L, Hosfelt J, Kwiatkowski L, Maclaren JK, Mason BM, et al. Reversal of ocean acidification enhances net coral reef calcification. *Nature* [Internet]. Nature Publishing

- Group. 2016;531:362–5. Available from: <http://www.nature.com/doi/10.1038/nature17155>.
85. Salter I, Schiebel R, Ziveri P, Movellan A, Lampitt R, Wolff GA. Carbonate counter pump stimulated by natural iron fertilization in the Polar Frontal Zone. *Nat Geosci* [Internet]. 2014 [cited 2015 Jan 12];7:885–9. Available from: <http://www.nature.com/doi/10.1038/ngeo2285>.
  86. Oschlies A, Koeve W, Rickels W, Rehdanz K. Side effects and accounting aspects of hypothetical large-scale Southern Ocean iron fertilization. *Biogeosciences*. 2010;7:4017–35.
  87. Robinson J, Popova EE, Yool A, Srokosz M, Lampitt RS, Blundell JR. How deep is deep enough? Ocean iron fertilization and carbon sequestration in the Southern Ocean. *Geophys Res Lett*. 2014;41(7):2489–95. <https://doi.org/10.1002/2013GL058799>.
  88. Harrison DP. Global negative emissions capacity of ocean macro-nutrient fertilization. *Environ Res Lett* [Internet]. 2017;12:035001. Available from: <http://stacks.iop.org/1748-9326/12/i=3/a=035001?key=crossref.c48b784e5caabd045debab6fa4036e0d>.
  89. Lawrence MW. Efficiency of carbon sequestration by added reactive nitrogen in ocean fertilisation. *Int J Global Warming*. Vol. 6, No. 1, pp.15–33.
  90. Matear RJ, Elliot B. Enhancement of oceanic uptake of anthropogenic CO<sub>2</sub> by macronutrient fertilization. *J Geophys Res* [Internet]. 2004;109:C04001. Available from: <http://doi.wiley.com/10.1029/2000JC000321>.
  91. Oschlies A, Pahlow M, Yool A, Matear RJ. Climate engineering by artificial ocean upwelling—channelling the sorcerer’s apprentice. *Geophys Res Lett*. 2010;37
  92. Schäfer S, Lawrence M, Stelzer H, Born W, Low S, Aaheim A, et al. The European transdisciplinary assessment of climate engineering (EuTRACE): removing greenhouse gases from the atmosphere and reflecting sunlight away from Earth. 2015.
  93. National Research Council. Climate intervention [internet]. Washington, D.C.: National Academies Press; 2015. Available from: <http://www.nap.edu/catalog/18805/climate-intervention-carbon-dioxide-removal-and-reliable-sequestration>
  94. Kato E, Yamagata Y. BECCS capability of dedicated bioenergy crops under a future land-use scenario targeting net negative carbon emissions. *Earth’s Futur* [Internet]. 2014;2:421–39. Available from: <http://doi.wiley.com/10.1002/2014EF000249>.
  95. European Academies’ Science Advisory Council. Science advice for the benefit of Europe negative emission technologies: what role in meeting Paris agreement targets? EASAC Policy Rep. 2018.
  96. Williamson P. Emissions reduction: scrutinize CO<sub>2</sub> removal methods. *Nature* [Internet]. 2016;530:5–7. Available from: <http://www.nature.com/news/emissions-reduction-scrutinize-co2-removal-methods-1.19318>.
  97. Fuss S, Jones CD, Kraxner F, Peters GP, Smith P, Tavoni M, et al. Research priorities for negative emissions. *Environ Res Lett* [Internet]. 2016;11:115007. Available from: <http://stacks.iop.org/1748-9326/11/i=11/a=115007?key=crossref.0d8b5da77c87ad502a8129cad2731d44>.
  98. Mongin M, Baird ME, Hadley S, Lenton A. Optimising reef-scale CO<sub>2</sub> removal by seaweed to buffer ocean acidification. *Environ Res Lett* [Internet]. IOP Publishing; 2016;11:034023. Available from: <http://iopscience.iop.org/article/10.1088/1748-9326/11/3/034023>.
  99. Hill R, Bellgrove A, Macreadie PI, Petrou K, Beardall J, Steven A, et al. Can macroalgae contribute to blue carbon? An Australian perspective. *Limnol Oceanogr* [Internet]. 2015;2:n/a-n/a. Available from: <http://doi.wiley.com/10.1002/lno.10128>.
  100. Johannessen SC, Macdonald RW. Geoengineering with seagrasses: is credit due where credit is given? *Environ Res Lett* [Internet]. IOP Publishing; 2016;11:113001. Available from: <http://stacks.iop.org/1748-9326/11/i=11/a=113001?key=crossref.1db6e1f7810fa528a2d86b3f3b5281a0>.
  101. Moreira D, Pires JCM. Atmospheric CO<sub>2</sub> capture by algae: negative carbon dioxide emission path. *Bioresour Technol* [Internet]. Elsevier Ltd; 2016;215:371–9. Available from: <http://linkinghub.elsevier.com/retrieve/pii/S0960852416303546>.
  102. Greene CH, Huntley ME, Archibald I, Gerber LN, Sills DL, Granados J, et al. Geoengineering, marine microalgae, and climate stabilization in the 21st century. *Earth’s Futur* [Internet]. 2017;5:278–84. Available from: <http://doi.wiley.com/10.1002/2016EF000486>.
  103. Caldeira K, Bala G, Cao L. The science of geoengineering. *Annu Rev Earth Planet Sci* [Internet]. 2013 [cited 2014 Feb 20];41:231–56. Available from: <http://www.annualreviews.org/doi/abs/10.1146/annurev-earth-042711-105548>.
  104. Reith F, Keller DP, Oschlies A. Revisiting ocean carbon sequestration by direct injection: a global carbon budget perspective. *Earth Syst Dyn* [Internet]. 2016;7:797–812. Available from: <http://www.earth-syst-dynam.net/7/797/2016/>
  105. Cao L. The effects of solar radiation management on the carbon cycle. *Curr Clim Change Rep* [Internet]. Current Climate Change Reports; 2018;1–10. Available from: <http://link.springer.com/10.1007/s40641-018-0088-z>.
  106. Zhou S, Flynn P. Geoengineering downwelling ocean currents: a cost assessment. *Clim Chang*. 2005;71:203–20.
  107. Flynn PC, Zhou JS. Can geoengineering sustain critical ocean currents? In: Badescu V, Cathcart RB, editors. Macro-engineering Seawater Unique Environ. Arid Lowl. Water Bodies Rehabil. [Internet]. Berlin: Springer Berlin Heidelberg; 2011. p. 583–93. Available from: [https://doi.org/10.1007/978-3-642-14779-1\\_26](https://doi.org/10.1007/978-3-642-14779-1_26)
  108. Strand SE, Benford G. Ocean sequestration of crop residue carbon: recycling fossil fuel carbon back to deep sediments. *Environ Sci Technol*. 2009;43:1000–7.
  109. Duarte CM, Losada IJ, Hendriks IE, Mazarrasa I, Marbà N. The role of coastal plant communities for climate change mitigation and adaptation. *Nat Clim Chang* [Internet]. Nature Publishing Group; 2013 [cited 2013 Nov 7];3:961–8. Available from: <http://www.nature.com/doi/10.1038/nclimate1970>.
  110. Amirova S, Tulaikova T. The method for effective CO<sub>2</sub> purification in the atmosphere. The atmospheric CO<sub>2</sub> purification method. *Glob J Sci Front Res H Environ Earth Sci*. 2015;15
  111. Heinze C, Meyer S, Goris N, Anderson L, Steinfeldt R, Chang N, et al. The ocean carbon sink—impacts, vulnerabilities and challenges. *Earth Syst Dyn* [Internet]. 2015;6:327–58. Available from: <http://www.earth-syst-dynam.net/6/327/2015/>.
  112. Gillooly JF. Effects of size and temperature on metabolic rate. *Science* (80-. ). [Internet]. 2001;293:2248–51. Available from: <http://www.sciencemag.org/cgi/doi/10.1126/science.1061967>.
  113. Le Quéré C, Andrew RM, Canadell JG, Sitch S, Korsbakken JI, Peters GP, et al. Global Carbon Budget 2016. *Earth Syst Sci Data* [Internet]. 2016;8:605–49. Available from: <http://www.earth-syst-sci-data-discuss.net/essd-2016-51/>.
  114. Zhu Z, Piao S, Myneni RB, Huang M, Zeng Z, Canadell JG, et al. Greening of the Earth and its drivers. *Nat Clim Chang*. 2016;6:791–5. early online
  115. Bonan GB. Forests and climate change: forcings, feedbacks, and the climate benefits of forests. *Science* [Internet]. 2008 [cited 2013 Sep 16];320:1444–9. Available from: <http://www.ncbi.nlm.nih.gov/pubmed/18556546>.
  116. Ameth A, Sitch S, Pongratz J, Stocker BD, Ciais P, Poulter B, et al. Historical carbon dioxide emissions due to land use changes possibly larger than assumed. *Nat Geosci*. 2017;10:79–84.
  117. Pongratz J, Reick CH, Raddatz T, Caldeira K, Claussen M. Past land use decisions have increased mitigation potential of reforestation. *Geophys Res Lett*. 2011;38:1–5.

118. Houghton RA. Interactions between land-use change and climate-carbon cycle feedbacks. *Curr Clim Change Rep* [Internet]. *Current Climate Change Reports*; 2018;1–13. Available from: <http://link.springer.com/10.1007/s40641-018-0099-9>.
119. Hicks Pries CE, Castanha C, Porras R, Torn MS. The whole-soil carbon flux in response to warming. *Science* (80-. ). [Internet]. 2017;1319:eaal1319. Available from: <http://www.sciencemag.org/lookup/doi/10.1126/science.aal1319>.
120. Stockmann U, Adams MA, Crawford JW, Field DJ, Henakaarchchi N, Jenkins M, et al. The knowns, known unknowns and unknowns of sequestration of soil organic carbon. *Agric Ecosyst Environ* [Internet]. Elsevier B.V.; 2013;164:80–99. Available from: <https://doi.org/10.1016/j.agee.2012.10.001>.
121. Hewitt AJ, Booth BBB, Jones CD, Robertson ES, Wiltshire AJ, Sansom PG, et al. Sources of uncertainty in future projections of the carbon cycle. *J Clim* [Internet]. 2016;29:7203–13. Available from: <http://journals.ametsoc.org/doi/10.1175/JCLI-D-16-0161.1>.
122. Friedlingstein P, Meinshausen M, Arora VK, Jones CD, Anav A, Liddicoat SK, et al. Uncertainties in CMIP5 climate projections due to carbon cycle feedbacks. *J Clim* [Internet]. 2014 [cited 2014 Sep 8];27:511–26. Available from: <http://journals.ametsoc.org/doi/abs/10.1175/JCLI-D-12-00579.1>.
123. Schuur E a G, McGuire A. D, Schädel C, Grosse G, Harden JW, Hayes DJ, et al. Climate change and the permafrost carbon feedback. *Nature* [Internet]. 2015;520:171–9. Available from: <http://www.nature.com/doi/10.1038/nature14338>.
124. Bauer JE, Cai W-J, Raymond P A., Bianchi TS, Hopkinson CS, Regnier P a. G. The changing carbon cycle of the coastal ocean. *Nature* [Internet]. 2013 [cited 2013 Dec 5];504:61–70. Available from: <http://www.nature.com/doi/10.1038/nature12857>.
125. Koven CD, Chambers JQ, Georgiou K, Knox R, Negron-Juarez R, Riley WJ, et al. Controls on terrestrial carbon feedbacks by productivity versus turnover in the CMIP5 Earth System Models. *Biogeosciences* [Internet]. 2015;12:5211–28. Available from: <http://www.biogeosciences.net/12/5211/2015/>
126. Keller DP, Lenton A, Scott V, Vaughan NE, Bauer N, Ji D, et al. The Carbon Dioxide Removal Model Intercomparison Project (CDRMIP): rationale and experimental design for CMIP6. *Geosci Model Dev* [Internet]. 2018;1–72. Available from: <https://www.geosci-model-dev-discuss.net/gmd-2017-168/>.
127. Mayer D. Potentials and side-effects of herbaceous biomass plantations for climate change mitigation. Universität Hamburg; 2017.
128. N'Yeurt ADR, Chynoweth DP, Capron ME, Stewart JR, Hasan MA. Negative carbon via ocean afforestation. *Process Saf Environ Prot* [Internet]. Institution of Chemical Engineers; 2012 [cited 2014 Aug 26];90:467–74. Available from: <http://linkinghub.elsevier.com/retrieve/pii/S0957582012001206>.
129. Zeng N. Carbon sequestration via wood burial. *Carbon Balance Manag.* 2008;3(1):12.
130. Oeste FD, de Richter R, Ming T, Caillol S. Climate engineering by mimicking natural dust climate control: the iron salt aerosol method. *Earth Syst Dyn* [Internet]. 2017;8:1–54. Available from: <http://www.earth-syst-dynam.net/8/1/2017/>
131. Xi F, Davis SJ, Ciais P, Crawford-Brown D, Guan D, Pade C, et al. Substantial global carbon uptake by cement carbonation. *Nat Geosci* [Internet]. 2016;9:7–14. Available from: <http://www.nature.com/doi/10.1038/ngeo2840>
132. Eisaman MD, Parajuly K, Tuganov A, Eldershaw C, Chang N, Littau KA. CO<sub>2</sub> extraction from seawater using bipolar membrane electro dialysis. *Energy Environ Sci.* 2012;5:7346.