## **REVIEW**



# **Recent progress for silver nanowires conducting flm for fexible electronics**

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# **Abstract**

Silver nanowires (AgNWs), as one-dimensional nanometallic materials, have attracted wide attention due to the excellent electrical conductivity, transparency and fexibility, especially in fexible and stretchable electronics. However, the microscopic discontinuities require AgNWs be attached to some carrier for practical applications. Relative to the preparation method, how to integrate AgNWs into the fexible matrix is particularly important. In recent years, plenty of papers have been published on the preparation of conductors based on AgNWs, including printing techniques, coating techniques, vacuum fltration techniques, template-assisted assembly techniques, electrospinning techniques and gelating techniques. The aim of this review is to discuss diferent assembly method of AgNW-based conducting flm and their advantages.

# **Graphic abstract**

Conducting flms based on silver nanowires (AgNWs) have been reviewed with a focus on their assembly and their advantages.



**Keywords** Silver nanowires conducting flns · Flexible electronics · Printed electronics · Electronic industry



Extended author information available on the last page of the article

### **Introduction**

In the past year, the global semiconductor industry has faced a huge challenge of insufficient production capacity. There are many factors afecting chip production, one of which is the work stoppings and production reductions of chip caused by COVID-19 and natural disasters (the Renesas Electric fre in Japan, the heavy snow in Texas, USA), on the other hand, the price of all kinds of chip raw materials continue to rise, such as polyvinyl chloride (PVC) prices rose by 60%, glass prices rose by 40%, copper prices rose by 48%. In addition, the production capacity of high-end chips that rely on advanced manufacturing process is insufficient. Only Samsung and Taiwan Semiconductor Manufacturing Co., Ltd. (TSMC) have mastered the chips manufacturing process below 7 nm, so the production capacity of chips with advanced manufacturing process is insufficient. The development of more advanced manufacturing processes and the mass production of high-end chips are far from assured. The shortage of chips has led to a shortage of CPUs, graphics cards, electronic devices and other commodities and a massive price hike, and has had a huge impact on the global electronics industry. This also reveals the following two problems. The current electronic industry is extremely dependent on the traditional electronic materials. The manufacturing techniques of traditional electronic materials, represented by silicon-based semiconductors, are about to encounter a bottleneck. Thus, there is an urgent need for new materials to make the electronics industry to leapfrog.

Throughout the past 100 years, the electronics industry has been developed rapidly [\[1](#page-12-0)[–4](#page-12-1)]. Research in the field of electronics is becoming more and more thorough [[5–](#page-12-2)[13](#page-12-3)]. There has been a rapid rise in the literature about electronics and the publications on electronics from Scopus as shown in Fig. [1.](#page-1-0) The electronics industry has grown at a rapid rate since the 1940s. The electronics-related research papers have grown at a sequential growth rate of more than 1100% in



<span id="page-1-0"></span>**Fig. 1** Literature search results of diferent years in Scopus with electronics as the keyword, data from Scopus

1940s. The main reason is the development of electronic materials. The semiconductor material represented by Si promotes the development of computers. The development of Si chips (integrated circuits) has had the greatest impact on the development of the electronics industry. From the electronic tube computer (1946) covering more than 100  $m<sup>2</sup>$ , to the mobile phone that can be held in one hand, everything depends on the advanced integrated circuit manufacturing process. With the invention of the transistor in Bell Laboratories in 1947, the electronics industry ushered in its frst milestone. In 1958, the integrated circuit (IC) was invented. Since then, there has been a new outbreak in the electronics industry. As a result, research on electronic technology increased exponentially throughout the 1960s, with a sequential growth rate of 3,300%. In 1964, Moore's Law was proposed by Moore, who predicted that the number of transistors on a chip would double every 18 months [[14](#page-12-4)]. From the very beginning of the electron tube to the transistor to the small-scale IC to the present large-scale IC, Si seems to have become the natural material suitable for the IC because of the appropriate electrical properties, large reserves, mature purifcation process, stable chemical properties and other factors. The development of the electronic industry depends on the development of Si-based semiconductor materials, but at present Si material is about to face its physical limit, and the development of new materials is bound to bring revolutionary changes to the electronic industry [\[15](#page-12-5)].

On the other hand, the development of transparent conductive flm (TCF) also plays a catalytic role in the develop-ment of the mobile electronics industry [\[16](#page-12-6)]. Among them, indium tin oxide (ITO) is a kind of TCF usually prepared by magnetron sputtering (MSP) [[17](#page-12-7)]. Because of its high transmittance, good conductivity and stability, it has an irreplaceable position in the felds of touch panel, liquid crystal display, solar cell and so on [\[18](#page-12-8)[–21](#page-13-0)].

With the development of ITO and its application on mobile phones, the sales of touch screen mobile phones have a leapfrog growth. Demand for electronics is shifting from stationary desktop computers to portable mobile devices. Some tasks that once needed to be handled on a computer have been gradually handled by mobile devices, and the direction of the electronics industry is gradually shifting toward lightweight and portable development. However, the problems faced by ITO mainly focus on high cost of raw materials. A large number of rare metal indium is used in the preparation of ITO. On one hand, the price is relatively high; on the other hand, the reserve of indium is limited, resulting in the high price of ITO. Compared with AgNWs, graphene and CNTs, the transmittance and conductivity of ITO is lower. ITO cannot withstand bending and stretching.

The development of nanomaterials has pushed the revolution of electronic industry [\[22](#page-13-1)[–25\]](#page-13-2). More advanced processes have been developed in nanomaterials in the feld of IC. The manufacture process of  $MoS<sub>2</sub>$  transistors with the gate length of 1 nm was developed, this process prompted small gate length, which would integrate more transistors on an integrated circuit resulting in excellent performance of the device, the advantages of this process even exceeded that of the Si-based IC processes [[26,](#page-13-3) [27](#page-13-4)]. Che et al*.* prepared a transparent electrode by polyaniline (PANI)-grafted CNTs exhibiting excellent conductivity and transmittance [[28](#page-13-5)]. Compared with ITO, PANI-grafted CNTs transparent electrode also has a high fexibility, resulting in the development of fexible wearable device. All these advantages lead to the rapid development of nanomaterials and there is a trend to replace traditional materials [[29–](#page-13-6)[32](#page-13-7)].

With the further development of electronic technology, in many practical application areas, electronic devices need to be attached to the surface of living organisms to detect movement or body conditions, such as fexible displays, flexible supercapacitors, electronic skins, etc. [[33](#page-13-8)[–51](#page-13-9)]. To more closely combine electronic products with the human body and pursue the comfort of electronic devices, electronic devices need to be light, fexible, bendable and stretchable enough to adapt to body movements [[52,](#page-13-10) [53](#page-13-11)]. And it needs to maintain stable work during the deformation process. Obviously, traditional rigid electronic materials will no longer be suitable for the fexible electronic feld. Since 2018, the commercialization of fexible electronics has gradually started from mobile phones. Major manufacturers have launched flexible fold-screen mobile phones and promoted them vigorously.

Excellent electrical property, fexiblity, and lightweight make nanomaterials be the potential materials to replace traditional rigid materials in fexible electronics [[54](#page-13-12)[–66](#page-14-0)]. Carbon-based nanomaterials, such as graphene and CNTs stand out among the nanomaterials due to the unique structure of high transparency, high specific surface area and excellent mechanical properties and it show an irreplaceable position in the felds of electronics, energy storage and so on [[67–](#page-14-1)[95](#page-15-0)]. But its rather poor uniformity in large scales limit its applications [[96](#page-15-1)]. In addition, the conductivity of carbon-based materials (100–10,000  $\Omega$ /sq) is relatively poor compared to a traditional metallic conductors. The comparison between the conductivity and transparency of common nanomaterials and ITO is shown in Table [1](#page-2-0).

In recent years, silver nanowires (AgNWs) as a onedimensional (1D) metallic nanostructural materials has attracted much attention and shows a potential trend to replace traditional electronic materials [\[97–](#page-15-2)[110](#page-15-3)]. AgNWs are a kind of silver-based metallic nanowires, which have been the focus of attention for several years due to not only the excellent electrical conductivity  $(6.3 \times 10^7 \text{ S m}^{-1})$ and thermal conductivity (429 W m<sup>-1</sup> K<sup>-1</sup>) of AgNWs in electronic devices but also outstanding conductivity, transparency, thermal properties and mechanical properties

<span id="page-2-0"></span>**Table 1** Comparison of properties between ITO and diferent transparent conductive flms

	Sheet resistance/ Transmittance conductivity		Year	References
Graphene	550 S $\cdot$ cm <sup>-1</sup>	70%	2008	[124]
Graphene	$10 S/cm^{-1}$		2011	$\lceil 125 \rceil$
Graphene	$350 \Omega/sq$	90%	2009	[126]
<b>CNTs</b>		85%	2004	$\lceil 127 \rceil$
<b>CNTs</b>	$160 \Omega/sq$	87%	2006	[128]
AgNWs	$20 \Omega/sq$	80%	2010	$\lceil 123 \rceil$
<b>ITO</b>	$304 \Omega/sq$	81%	2016	[129]

afforded by their high aspect ratio  $[111–122]$  $[111–122]$ . Because of the excellent electrical properties and good operability in solution, AgNWs act as conductors or circuits printed in some fexible electronics and it show the transmittance of 80% which have even match with ITO in some studies. AgNWs have become a potential material to replace ITO (Table [1](#page-2-0)). And due to the characteristics of fexibility, low resistance, good operability, etc., AgNWs have surpassed ITO in some applications. Thus, the AgNWs are proved much suitable for solar cells, which is because of the large diference between difusive transmittance and specular transmittance compared with ITO. This phenomenon is caused by the light scattering efects [[123](#page-15-6)].

Another reason for the widespread use of AgNWs for its convenient preparation [[130](#page-16-0)]. AgNWs can be synthesized by a variety of simple techniques, such as ultraviolet irradiation method, hydrothermal method and solution-based synthesis, etc. [[131–](#page-16-1)[133](#page-16-2)]. Zhou et al. prepared AgNWs by an ultraviolet irradiation method  $[134]$  $[134]$ . The mixed solution of  $AgNO<sub>3</sub>$  and PVA was put under UV light for 48 h and then the AgNWs (silver nanorods) with a diameter of 15–20 nm and a length of 350 nm were got. Xu et al. prepared AgNWs by hydrothermal method [[132\]](#page-16-4). Hexamethylenetetramine aqueous was dropped into  $AgNO<sub>3</sub>$ aqueous and then the gemini surfactant was added, after hydrothermal reaction at 100 ℃, the AgNWs were formed. The length of the AgNWs prepared in this method could reach 50 μm. Since the preparation method of AgNWs is more convenient, AgNWs have been already planned for mass production. This would also signifcantly reduce the cost of AgNWs, facilitating their replacement with traditional electronic materials.

All these properties make AgNWs be promising materials for the preparation of fexible stretchable electronic devices. However, the microscopic discontinuities make AgNWs attach to some carrier for practical applications. Relative to the preparation method, how to integrate AgNWs into the fexible matrix is particularly important. This review mainly discusses the assembly of conductive flms of AgNWs, the



fabrication methods of AgNW-based conducting flm are still discussed in more details below [\[135,](#page-16-7) [136\]](#page-16-8).

# **Assembly of AgNW‑based conducting flm**

AgNWs could be dispersed in many solvents such as ethanol, isopropanol and water with good solubility, so the AgNW-based conducting flms are often manipulated in a solution process [[137](#page-16-9)[–143](#page-16-10)]. The AgNWs solution is transferred to the fexible matrix to fabricate conducting flms by techniques such as printing, coating, vacuum fltration, template-assisted assembly and gelation [\[144](#page-16-11)[–149\]](#page-16-12).

### **Coating techniques**

Coating technology is widely used in the manufacture of thin flms. In terms of the preparation of conductive flms of AgNWs, the dispersion solution of AgNWs is usually scraped, dropped or sprayed on the surface of the substrate evenly, and then the solvent evaporation. After coating process, AgNWs network is formed on the substrate to form conductive pathway. Diferent coating methods often lead to diferent uniformity, electrical conductivity and transparency of the flm, so we will discuss them as follows.

### **Meyer rod coating**

Meyer rod coating is a kind of bar coating with the Meyer rod (a kind of metal bar wound around the metal wires). During the Meyer rod coating process, AgNWs solution is poured onto the surface of substrate, and then excess solution is scraped fat by Meyer rod (Fig. [2a](#page-4-0)). The thickness of AgNWs flm determines the conductivity and transparency of the flm, and the thickness could be tuning by varying the diameter of the metal wires around the bar.

Hu et al. prepared AgNWs electrodes by the Meyer rod coating process (Fig. [2](#page-4-0)b) [[123](#page-15-6)]. The AgNWs ink was frst poured onto the PET substrates and then coated by a Meyer rod. After drying by the infrared lamp, the AgNWs were formed uniformly on the polyethylene terephthalate (PET) substrates (Fig.  $2c$  $2c$ ). And then a 20 nm thick Teflon was attached to surface of the AgNWs flm to protect the electrodes. The electrodes exhibit excellent conductivity and transmittance, compared with ITO (Fig. [2](#page-4-0)d).

The distribution of AgNWs is uneven leading to a poor uniformity of conductivity of AgNWs conducting flm when large-scale conducting flms were made. Zhang et al. solved this problem through a dynamic heating process [\[150](#page-16-13)]. To improve the uniformity of the conducting flm, the prepared flm was moving repeatedly under a xenon lamp in 0.05 m/s instead of the fxation. The temperature gradient of dynamic heating was more uniform in the dynamic heating process



(Fig. [2e](#page-4-0), f). The uniformity of conductive flm prepared through the dynamic heating process was compared with traditional Meyer rod coating. 25 regions were partitioned in the PET substrate of  $20 \times 40$  mm. The conductivity of the 25 regions of the conducting flm prepared by dynamic heating and traditional process was measured separately. The uniformity of the conducting flm prepared by dynamic heating is better.

Another way to improve the conductivity of a conductive flm is through post-processing. For example, Song et al. also fabricated a transparent conducting flm by the Meyer rod coating method [[151\]](#page-16-14). AgNWs with the concentrations from 2 to 0.5 mM were coated by Meyer rod on glass or ITO substrate, respectively. Then the flms were sintered at 150, 200, 250, 300 and 400 ℃ (Table [2\)](#page-4-1). The sheet resistance was sharply reduced from 80 to 6  $\Omega$ /sq when the sintering temperature rose to 300℃. But the AgNWs networks were broken at 400 ℃, which led to an open circuit. This is because the AgNWs began to weld together from 200 ℃, and the PVP began to be decomposed from 250 ℃. The transmittance of the flms could reach 65% and it is very smooth at the near-infrared regions (from 1100 to 1300 nm), which is advantageous over ITO for solar cells. In addition, the contact resistance can also be efectively reduced by coating the surface of AgNWs with GO (Fig. [2g](#page-4-0)) [\[152](#page-16-15)]. As shown in Fig. [2](#page-4-0)h, the surface roughness of the conductive flm is signifcantly reduced after GO is covered. At the same time, the environmental stability of the conductive AgNWs flms covered with graphene is also improved (Fig. [2i](#page-4-0)).

#### **Spin coating**

Spin coating is the common process for preparing membranes in the laboratory. Spin coating is a process to form a uniform flm on revolving platform by centrifugal force [[153](#page-16-16)]. The volatile solvent evaporates during the rotation process and the flm is left. The advantages of spin coating lie in fast fabrication of uniform and small flms. However, it is hard to apply to large-scale industrial production due to the material loss during rotation and the thickness control is difficult. Lee et al. integrated AgNWs on the graphene surface through the spin coating technology, which efectively enhanced the thermal and electrical conductivity of graphene [\[154](#page-16-17)]. The transfer of electron/phonon energy carriers between polycrystalline organisms was efectively resolved, where AgNWs were equivalent to bridges (Fig. [3](#page-5-0)a, b).

Lang et al. prepared a transparent conductive flms by a spin costing [[155\]](#page-16-18). The polar organic–inorganic and the nonpolar organic–inorganic hybrid matrix were synthesized and then mixed with AgNWs. The mixed AgNWs ink was spun under diferent rotational speeds (from 300 to 5000 rpm). The thickness of the flm could be controlled by the rotational speed.



<span id="page-4-0"></span>**Fig. 2 a** Schematic diagram of fabrication process of the Meyer rod coating, **b** actual operation diagram of fabrication process of the Meyer rod coating, **c** AgNWs were formed uniformly on the PET substrates, **d** the electrodes exhibit excellent conductivity and transmittance, compared with ITO. Adapted with permission from Ref. [[123\]](#page-15-6). Copyright 2010 American Chemical Society. **e** The temperature distribution of static drying, **f** the temperature distribu-

<span id="page-4-1"></span>**Table 2** Sheet resistance under diferent sintering temperatures

Sintering temperature $({}^{\circ}C)$	150	200	-250	300	400
Rs of sample 4 $(\Omega/\text{sq})$	80				

### **Spray coating**

In the spraying process, the AgNW dispersion is rapidly atomized through the nozzle under the action of compressed gas. The AgNWs are then uniformly attached to the substrate to form conductive network. Compared with spin coating and Meyer rod coating, spray coating can produce a more uniform conductive flm.

AgNWs–exfoliated graphene (EG)-based transparent electrode with a low-surface roughness was prepared by Ricciardulli et al. through a spraying coasting process

tion of dynamic heating. Adapted with permission from Ref. [[150](#page-16-13)]. Copyright 2014 Yuan-Jun Song et al. **g** fowchart of preparation of AgNWs/graphene oxide (GO) composite conductive flms, **h** 3D AFM images of AgNWs flms and GO/AgNWs flms, **i** resistance changes of AgNWS and GO/AgNWS flms at 80 °C and 75% RH for 16 consecutive days. Adapted with permission from Ref. [\[151\]](#page-16-14). Copyright 2020 IOP Publishing

[[156](#page-16-19)]. In most cases, the drawback of contact resistance and rough surface of AgNW-based conducting flm is hard to avoid. The usual way to prove these problem is thermal annealing, high-force pressing and welding techniques [[157–](#page-16-20)[159](#page-16-21)]. However, exfoliated graphene was adopted to cover the surface of the AgNWs to reduce the contact resistance and surface roughness. In addition, the mechanical stability and chemical stability were also improved after introducing the EG. First, the AgNW solution was spray costed onto the substrates and then the EG fakes were also spray coated onto the surface of AgNW networks (Fig. [4\)](#page-5-1). Compared with spin coating and vacuum fltration, spray coating is a more suitable method, which always led to a randomly arrangement of EG and the vacuum fltration always led to vacancies. The spray coating is the most suitable method for this case.



<span id="page-5-0"></span>

<span id="page-5-1"></span>**Fig. 4 a** The fabrication process of the spray-coated AgNWs-EGbased electrode. **b** The SEM images of AgNW network coated by EG. **c** The change of sheet resistance during multiple bending of

**Drop casting**

When the thickness and uniformity of the film are not needed to be considered, drop casting is a better choice. The solution of AgNWs with a defnite concentration is dropped onto a substrate to evaporate the solvent. After evaporating, the AgNWs are left on the substrate. Or it also could be transferred to another matrix. But it is hard to ensure the uniformity of AgNW conducting flm during dropping, and the coffee rings may be introduced at the drying process. The thickness of flm is also uncontrollable.

Jiang et al. prepared a self-healing sensor intercalated with AgNWs by the drop casting process [[160](#page-16-22)]. The

AgNWs and AgNWs-Eg fim. Adapted with permission from Ref. [[156\]](#page-16-19). Copyright 2018 WILEY–VCH

AgNW solution was dropped at the tetrafluoroethylene (TFE) plate. The ambient temperature was then raised to 60 ℃ for 1 h to ensure the volatilization of solvent and the formation of AgNW film. Subsequently, a selfhealing polymer was attached to the AgNWs on the TFE plate and heated to 90 ℃ for 30 s, and then a pressure of 0.2 MPa was applied at 90 ℃ for 1 min to keep tightly contact of polymer and AgNW film (as shown in Fig. [5](#page-6-0)). The conductivity can be healed after damaging due to the discontinuous network of AgNWs which can can wriggle to the original location with the healing process of polymer substrate. Thus, no sintering or welding process is adopted.

<span id="page-6-0"></span>**Fig. 5** The fabrication process of AgNW-based sensor. Adapted with permission from Ref. [\[160](#page-16-22)]. Copyright 2019 WILEY–VCH



# **Printing techniques**

Flexible printing electronics refers to a technology that combines printing technology with flexible electronics  $[161–166]$  $[161–166]$  $[161–166]$ . By means such as screen printing and inkjet printing, the conductive ink is transferred to a fexible substrate to form a specifc pattern [[167,](#page-17-1) [168\]](#page-17-2). Flexible printed electronics have drawn much attention in recent years according to the large scale, low cost and versatile manufacturing. Compared with coating technology, fexible printing technology is suitable for the occasions with more refned patterns, such as fexible circuit board, fexible thin-flm transistors, etc. [\[169–](#page-17-3)[172\]](#page-17-4). The shape or pattern of conductor could also be precisely designed before fabricating. For example, Gao et al. realized the printing of piezoresistive pressure sensors on nanocellulose paper (NCP) as shown in Fig. [6](#page-7-0)a [\[173\]](#page-17-5). Cui et al. used the method of electrohydrodynamic (EHD) printing to achieve the minimum line width of 45 µm, as shown in Fig. [6](#page-7-0)b [\[174](#page-17-6)]. There have been a large number of fexible electronics fabricated by the printed electronics techniques including fexible displays, sensors and solar sell, etc. [\[175](#page-17-7)[–180\]](#page-17-8).

Printing technology involves several key technical points, i.e., the preparation of conductive ink, printing and postprocessing. According to diferent printing methods, printing technology can be divided into screen printing and inkjet printing.

#### **Screen printing techniques**

The screen printing technique is the most widely used printing techniques. The design of printing pattern is a unique feature of printing technology. In this process, a custommade screen with open areas of desired pattern is used as the print template. The AgNW ink is added on the custom-made screen. When a squeegee is scraped over the surface of the screen, the AgNW ink remains on the substrate through the open areas in the screen and the desired pattern forms. For example, Ali et al. used screen printing technology to design two kinds of patterns (i.e., straight and folded), and printed them on fexible thermoplastic polyurethane (TPU) to prepare fexible strain sensors (as shown in Fig. [6c](#page-7-0)) [\[181](#page-17-9)]. The size of the printed pattern is 28 mm long, 4 mm wide, and 0.8 mm line width, as shown in Fig. [6c](#page-7-0). A more elaborate structure was designed in the work of Liang et al. [\[182](#page-17-10)]. A  $10\times6$  thin-film transistor (TFT) array is screen printed on a flexible PET substrate with a resolution of up to 50  $\mu$ m, as shown in Fig. [6d](#page-7-0), e. Actually, the rheological behavior of ink afects the resolution of printing. The peak holding step (PHS) test was carried out to analyze the change in viscosity of three kinds of AgNW inks with diferent AgNW contents (i.e., 6.0%, 6.6% and 7.3%), respectively. The shear rates were varied to simulate the scraping process and the recovery process (Fig. [6e](#page-7-0)). Once the squeegee scrapes the ink through the screen, the ink could fow through the screen due to the decreasing viscosity of ink. While, when the force of the squeegee disappears, the ink does not recover to the original viscosity immediately. This gives the ink time to level out. In contrast, the high viscosity ink has less time to level out, which lead to coffee rings, as shown in Fig. [6f](#page-7-0). The low viscosity ink is inclined to achieve sharper print lines [[183\]](#page-17-11). Xu et al. used AgNW ink with a low concentration (for example, 2.5 mg/mL) and low viscosity (0.74 Pa·s at shear rates of  $10 \text{ s}^{-1}$ ) to screen print a 2.5 cm  $\times$  2.5 cm transparent electrode, and covered it with GO to ensure electrode stability [[184\]](#page-17-12). The transparency of the prepared transparent electrode reaches 83.5% and the square resistance reaches 11.9  $\Omega$ /sq. The authors applied it to a long-term wearable heart rate monitoring device.

#### **Inkjet printing techniques**

Inkjet printing is a common non-contact printing technology in our daily life [\[185\]](#page-17-13). Books, documents, pictures and so on can be printed by inkjet printing technique. Inkjet printing technique is crucial in printed electronics. Compared to screen printing technology, inkjet printing is more fexible in applications and is suitable for a variety of materials such as paper, glass, PET, TPU due to its non-contact





<span id="page-7-0"></span>**Fig. 6 a** The piezoresistive pressure sensors prepared by printing silver nanowires on NCP. Adapted with permission from Ref. [\[173](#page-17-5)]. Copyright 2019 American Chemical Society; **b** optical images of AgNW lines printed on PDMS. Adapted with permission from Ref. [[174\]](#page-17-6). Copyright 2009 Royal Society of Chemistry, **c** two patterns are printed by AgNWs on TPU, one of which is a straight line and the other is a wavy line (schematics and photographs). Adapted with

characteristics. It's worth noting that an ordinary commercial printer can complete the printing process of electronics.

Inkjet printing technique was demonstrated in Choi's work [[186\]](#page-17-14). The inkjet printing process can be easily realized through a desktop inkjet printer (HP Deskjet 1010), as shown in Fig. [7](#page-8-0)a. In this study, a mixture of single-walled carbon nanotubes (SWNTs), activated carbon (AC) powders and AgNWs were mixed to form printing ink. The mixed ink was subjected to the sonication-driven scission process to avoid printer nozzle blockage caused by the long AgNWs and then the ink was printed on paper to form fexible solid supercapacitors and it had a good electrical performance, in the circuit it could light up an LED lamp, as shown in Fig. [7b](#page-8-0), c. Cellulose nanofibril (CNF) was used as the primer layer sprayed onto A4 paper in advance to reduce the capillarity phenomenon of paper, which can improve the printing resolution. Generally, the surface of substrate plays an important role for printing electronics. For wetting substrates (such as A4 paper), the rough surface may induce

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permission from Ref. [[181\]](#page-17-9).Copyright 2018 Elsevier, **d** fexible TFT arrays prepared by screen printing of AgNWs, **e** The resolution of TFT Arrays can reach 50 µm, **f** the Rheological behavior of 3 kinds of AgNW inks and **g** the coffee rings of high viscosity ink appeared in the printing process. Adapted with permission from Ref. [\[182](#page-17-10)]. Copyright 2016 WILEY–VCH

the capillarity. The capillarity causes the ink to spread and reduces the resolution of printing. While for non-wetting substrates such as polyethylene terephthalate (PET), the poor adhesion for ink may lead to the formation of coffee ring. However the surface of the CNF primer layer is between wetting and non-wetting substrates, and there is no capillarity and cofee-ring appearance on the surface of CNF primer layer, as shown in Fig. [7d](#page-8-0) [\[186](#page-17-14)]. Similarly, Gao et al. used a nanocellulose paper with a smooth surface as a printing carrier [\[173](#page-17-5)]. Printing on NCPs, which have a smoother surface, yields sharper patterns than PET, as shown in Fig. [7](#page-8-0)e.

Compared with screen printing, because the ink needs to pass through the nozzle, inkjet printing has higher requirements for the size of the printed nanomaterial, the size of its requirements is usually long and its size is less than 1/50 of the nozzle aperture. However, for the metal nanowires, because of the one-dimensional structure, the requirements are more loose. For example, Finn et al. used 2.2 μm AgNWs to carry out inkjet printing through a 21.5 μm nozzle, and



<span id="page-8-0"></span>**Fig. 7 a** Inkjet printing on paper can be done using only commercial printers, **b** cyclic voltammetry curves of two printed supercapacitors of diferent sizes, **c** printed SCs are used to light a LED, **d** the infuence of substrate on the efect of inkjet printing. Adapted with permission from Ref. [\[186](#page-17-14)]. Copyright 2016 WILEY–VCH, **e** AgNWS

achieved a line width of  $1-10$  mm and a thickness of 2  $\mu$ m [\[185\]](#page-17-13).

Zhang et al. adopted electrohydrodynamic inkjet (E-Inkjet) printing to efectively improve the printing resolution, and the printing principle is shown in the Fig. [7](#page-8-0)f [\[187](#page-17-15)]. The print resolution of 30 μm is achieved by applying a voltage between the nozzle and the substrate, as shown in Fig. [7](#page-8-0)g.

### **Electrospinning**

Electrospinning process is the common way to fabricate 1D conductive nanofbers [[188](#page-17-16)[–198\]](#page-18-0). Electrospinning works in a similar way to e-Inkjet, where a strong electric feld between the nozzle and the printing substrate transfers the solution from the nozzle to the printing platform. Unlike

trace comparison diagram printed on PET surface and NCP surface. Adapted with permission from Ref. [[173](#page-17-5)]. Copyright 2019 American Chemical Society, **f** the printing principle of E-inkjet printing and **g** print trace obtained by E-inkjet printing. Adapted with permission from Ref. [[187\]](#page-17-15). Copyright 2021 Springer Nature

E-Inkjet, the solution used for electrospinning is usually a pre-polymerized solution of AgNWs mixed with a polymer. In this process, AgNWs are mixed with a polymer solution, sprayed into a strong electric feld to form small fbers and collected onto a substrate to form a conductive network.

Generally, the specifc surface area of nanowires is large, and it is easy to be oxidized when exposed in the air, thus afecting the performance. Conductive flms prepared by means of coating and printing need to be post-processed to isolate AgNWs from the environment. Electrospinning prevents oxidation by placing a polymer coating directly on the outside of AgNWs.

The electrospun AgNWs-polyvinylidene fuoride (PVDF) nanofbers were demonstrated by Cheon et al. [\[199](#page-18-1)]. The triboelectric nanogenerator (TENG) with AgNWs-PVDF has been fabricated by the electrospinning process. The AgNWs



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and PVDF were dissolved in the mixed solvent of N,N-dimethyl acetamide and acetone for 5 min of ultrasonic processing and 3 h of stirring to make sure the solution uniform. Then the electrospun process was performed (Fig. [8](#page-9-0)a, b,c). In the presence of an electrostatic feld, the polymer solution is emitted from the needle tip (Taylor cone) to a collector (Al foil) for 2 h. In the process, polymer fbers with the highly oriented crystalline polymer chains are formed. These nanofibers were collected and cut into  $2 \times 2$  cm<sup>2</sup> nanofibers mat (Fig. [8d](#page-9-0)). The addition of AgNWs makes it easier to trap the charge and induce the tribocharges. (Fig. [8e](#page-9-0)). After thermal treatment at 160℃ for 2 h, the PVDF-AgNWs nanofbers mat was treated as the electrodes of TENGs and served as the negative triboelectric potential. The compact spinning technique was used to prepare a core sheath silver nanowire/polyvinylpyrrolidone [[200](#page-18-2)]. Due to the presence of PVP coating, the nanofbers prepared have the properties of insulation. Pan et al. used the mixed solution of AgNWs and AgNPs to prepare a transparent conductive flm through the near-feld electrospinning process. The flm reached 70% transparency and 0.032  $Ω$ /sq square resistance, which was similar to that of ITO [[201](#page-18-3)].

## **Vacuum fltration**

In vacuum fltration process, AgNWs are prepared as dispersion and passed through the fltration funnel. Solvent flows over the membrane and AgNWs remain on the membrane to form an AgNWs network, which is then transferred to the substrate. Xu et al. prepared a transparent conductive flm based on AgNWs [\[202](#page-18-4)]. The light transmittance of the prepared transparent conductive flm can be adjusted from 65.6 to 87.5% and sheet resistance from 177.3 to 4.95  $\Omega$ /sq by controlling the weight density of the silver nanowire solution.

Shen et al. fabricate a AgNWs and tungsten oxide  $(WO<sub>3</sub>)$  based electrochromic electrode for supercapacitors by the vacuum fltration method [[203](#page-18-5)]. First, the PDMS was spin-coated on the fexible PET substrate and it was pre-cured. Subsequently, the AgNWs ethanol solution was diluted by deionized water and then vacuum fltered on a flter membrane. The AgNWs flm on the flter could be easily transferred onto the pre-cured PDMS. Then the  $WO_3$ was deposited on the fexible substrates with a 600 nm thickness. The composite electrode exhibits an areal capacitance of 13.6 mF/cm<sup>2</sup> and corresponding specific capacitance of 138.2 F/g when the scan rate is 10 mV/s. And it also shows the electrochromic property. The color could be changed between dark blue and transparent when a negative (dark blue) or positive (transparent) bias potential is applied, respectively. This is caused by the reaction of the transformation between  $WO_3$  and  $H_xWO_3$  (Fig. [9](#page-10-0)).

In general, the silver nanowires of AgNWs flms prepared by vacuum pumping and fltration are physically bonded, and the uniformity of deposition is difficult to be controlled, resulting in poor electrical conductivity. Mushtaq et al. deposited AgNWs in the poly(Ethersulfone) (PES) membrane frst, and then electroplating the pristine AgNWs membrane with a layer of silver in the posttreatment method [[204](#page-18-6)]. Electroplating treatment of the conductive flm has a better uniformity and conductivity.



<span id="page-9-0"></span>**Fig. 8 a** The schematic diagram of TENG, **b** the SEM images of PVDF-AgNWs and nylon, **c** the structure diagram of PVDF-AgNWs and nylon, **d** schematic illustration of electrospinning process, **e** the

surface potentials of PVDF and PVDF-AgNWs. Adapted with permission from Ref. [[199](#page-18-1)]. Copyright 2017 WILEY–VCH



<span id="page-10-0"></span>

#### **Template‑assisted assembly**

Template-assisted assembly refers to the method of using templates to assemble some specifc shapes or specifc structures. Template-assisted assembly technology can usually obtain a three-dimensional ordered onlooker structure.

Physical template assembly is the common templateassisted assembly. The template is usually a polymer with a specifc structure or a substance with a network structure, such as polyurethane, sponge, foam nickel, etc. PU is a commonly used template material in various polymer templates. On one hand, because PU is a common commodity foam material, on the other hand, because PU has a unique network structure. Ge et al*.* utilized commercial PU foam as a template to form a stretchable AgNWs conductor with a specifc binary network structure [\[205](#page-18-7)]. The PU was immersed in the ethanol solution of AgNWs, with the evaporation of ethanol, the AgNWs were left on the surface of the PU to

form a conductive network. The existence of this unique binary network enables PU/AgNWs to still exhibit a high conductivity of 19.2 S cm−1 under 50% strain. Another reason for choosing PU as a template is that it is easy to be removed. Wu et al*.* utilized PU as a template to assist in the preparation of graphene/AgNWs foam [[206](#page-18-8)]. The PU template is immersed in an aqueous solution of GO and AgNWs. GO is then reduced to graphene by dopamine and attached to the foam surface together with AgNWs. Then the PU foam is formed by pyrolysis at 700 °C. After removing the template, the material can still maintain the shape of the template (Fig. [10](#page-10-1)a).

In addition to polymers with specifc structures, ice is often used as a template. Ice-template method is the common template-assisted assembly method to fabricate 3D AgNWs networks. In this process, AgNWs aqueous solution is frozen to make the water condense into ice crystals, and then the ice crystals are sublimated by vacuum drying



<span id="page-10-1"></span>**Fig. 10 a** SEM image of graphene/AgNWs foam. Adapted with permission from Ref. [\[206](#page-18-8)]. Copyright 2014 American Chemical Society, **b** the schematic diagram of ice-templated process, **c** the SEM images of the 3D networks of AgNWs–GO skeleton under diferent magnifcations. Adapted with permission from Ref. [\[207\]](#page-18-9). Copyright 2018 WILEY–VCH, **d** schematic diagram of the capillary-assisted fuidic assembly process, **e** optical microscopy images of AgNWs

prepared by CFA technique, **f** spray-coating technique. Adapted with permission from Ref. [[208](#page-18-10)]. Copyright 2012 Royal Society of Chemistry, **g** three-phase interface assembly and its four-step principle display diagram and **h** the SEM images of the AgNWs conductive flm prepared by three-phase interface assembly. Adapted with permission from Ref.[[209\]](#page-18-11). Copyright 2010 WILEY–VCH



technology to form 3D AgNWs networks. Ice-templated method was adopted by Xue et al. to prepare hierarchical AgNWs–graphene host for Li-ion batteries. The viscous mixed aqueous solution of AgNWs and GO was bladecoated onto substrate and then dipped in liquid nitrogen [[207](#page-18-9)]. Small and orderly ice crystals were formed in the AgNWs–GO hydrogel. Subsequently, the freeze-drying process was carried, leaving the voids in the AgNWs–GO hydrogel during the sublimation of the small ice crystals. An ordered 3D network was formed (Fig. [10b](#page-10-1), c).

Regular alignment of AgNWs is realized through capillary-assisted fuidic assembly (CFA) (Fig. [10](#page-10-1)d) [[208\]](#page-18-10). The capillary array is used as a template for the self-assembly of AgNWs. When the AgNW solution fows through the capillary, the AgNWs will self-assemble along the direction of the capillary in the capillary. As the AgNWs fow out of the capillary, the AgNWs form a well-aligned conductive flm on the substrate (Fig. [10e](#page-10-1), f). The oriented AgNWs exhibit excellent anisotropic conductivity. In parallel, it exhibits far lower square resistance and higher transparency than randomly distributed AgNWs conductive flms.

In other studies, the template does not refer to a specifc shape of solid, but a specifc interface structure. A specifc water–oil-air interface is used as a template to achieve a special orderly oriented AgNWs flm assembly by Shi et al. [\[209\]](#page-18-11). A water dispersion of AgNWs was dropped on the chloroform solution to construct a template with a threephase interface structure (Fig. [10](#page-10-1)g). The AgNWs realize a process from the water–oil interface to the water–oil-gas interface and then to the water–gas interface, during which they self-assemble into a flm. The evaporation of chloroform at the oil–gas interface makes the pressure at the oil–gas interface lower than the water–oil interface, forming a pressure gradient  $(\Delta P)$ . The  $\Delta P$  causes the directional movement of AgNWs. The AgNWs assembled in this way have excellent regularity (Fig. [10](#page-10-1) h).

### **Gelation**

The gelation is a method to construct 3D nanofller network [[210–](#page-18-12)[212\]](#page-18-13). The AgNWs are frst mixed with the precursor solution of hydrogels. Then the gelation reaction is carried out to fx AgNWs in the hydrogels to make the conductive pathway form. In this method, AgNWs are often lapped together in a three-dimensional manner, and higher concentrations are usually required to achieve conductivity improvement. This often leads to poor transparency.

Lim et al. prepared stretchable conductors with AgNWs and alginate hydrogels by the gelation methods [[213\]](#page-18-14). The AgNWs and sodium alginate were mixed at 70 ℃, and then dried at 40 °C. 0.1 M CaCl<sub>2</sub> aqueous solution was then added for 20 min to make sure the cross-linking between the algi-nate chains happening (Fig. [11\)](#page-11-0). The conducting film exhibits excellent conductivity, which the impedance of the flm is extremely stable and maintains the low value of 8.5  $Ω$ . The conducting flm could be cut into any shape by laser according to the actual usage (Fig.  $11$ ). The conductivity of AgNWs conducting film  $(\geq 0.3 \text{ wt\% AgNWs})$  exhibits a stretch insensitivity when the strain is between 0 and 30%. However, the color of the flm shows opaque metallic silver, which makes the flm hard to be used in the optical feld.

# **Conclusions and prospects**

With the development of fexible electronic, AgNWs as an important component of fexible electronics devices have shown the potential to replace or even surpass traditional conductors. However, there are still some factors that limit



<span id="page-11-0"></span>**Fig. 11** The schematic diagram of the fabrication process of AgNW-based alginate hydrogels flm. The flm can be cut into any shapes by laser after forming



AgNW-based conducting flm to replace traditional electrodes. One is the synthesis of higher quality AgNWs, which have a higher aspect ratio and yield a ratio and must be suitable for mass production. The other and most critical point is to develop more appropriate fabrication techniques. The microscopic discontinuities make AgNWs be attached to some carrier for practical applications. Relative to the preparation method, how to integrate AgNWs into the fexible matrix is particularly important.

Although AgNW-based transparent conductive flms have surpassed traditional electronic materials in many properties, it is still challenging to replace traditional electronic materials. Although the preparation method of AgNWs has been relatively complete, the high cost of raw material silver makes the cost of printed circuits and other aspects high. Due to the large specifc surface area, the AgNW material is very prone to oxidation in the air, resulting in performance degradation. How to treat the surface of the conductive flm to prevent oxidation is a problem worthy of consideration. Although graphene is used for surface treatment in some studies, its high cost is still difficult to apply in actual places. The commonly used methods for post-treatment of AgNWbased transparent conductive flms are thermal annealing or chemical annealing, which often requires higher temperatures or expensive chemical reagents, which is also an obstacle to the commercialization of AgNWs.

In this review, we discussed the assembly techniques for obtaining the AgNWs conducting flm in fexible electronics. In these fabrication techniques, the printing techniques are suitable for detailed printed circuit. The coating techniques are suitable for large-scale preparation of conductive flms. The 3D fabrications including are suitable for 3D conducting networks. We believe that with the development of fexible technology, the manufacturing technology of AgNWs conducting flm will be further developed, and the commercialization of the AgNW-based fexible conducting flm is coming.

### **Declarations**

**Conflict of interest** We declare that we have no fnancial and personal relationships with other people or organizations that can inappropriately infuence our work, there is no professional or other personal interest of any nature or kind in any product, service and/or company that could be construed as infuencing the position presented in, or the review of, the manuscript entitled, "Recent Progress for Silver Nanowires Conducting Film for Flexible Electronics".

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