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# **Experimental study on macro spray, combustion and emission characteristics of biodiesel and diethyl carbonate blends**

**L. Zhao1,2 · X. Zhu2 · R. Zhao3 · J. Tian1 · D. Qian2 · Q. Lin1**

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#### **Abstract**

Biodiesel has a higher viscosity and poorer atomization efect compared to traditional diesel fuel, resulting in lower combustion efficiency. This study investigated the effects of exhaust gas recirculation and a diethyl carbonate additive on the combustion and emission of biodiesel blends. Biodiesel was blended with DEC (90% volume biodiesel+10% volume DEC: B90DEC10). Experiments were conducted at various brake mean efective pressures (BMEP: 0.2, 0.6, 1.0, 1.6, and 2.0 MPa) and EGR rates (0, 5, and 15%) on a direct injection and high pressure common rail diesel engine that met Euro VI emission standards. The results revealed that the viscosity and atomization efects of B90DEC10 were improved, and the premixed combustion period was longer and the heat release rate (HRR) was higher compared to traditional diesel. The maximum HRR of B90DEC10 was 20% higher than that of diesel at 0.6 MPa. Compared to diesel, the particulate matter emissions of B90DEC10 were reduced by 92%, and total hydrocarbons and carbon monoxide emissions were significantly reduced. However, there was an increase of 8–20% in nitrogen oxide emissions. At an EGR rate of 15%, the  $NO<sub>x</sub>$  emissions of B90DEC10 were reduced by 45%, while having minimal impact on PM emissions and brake thermal efficiency. Therefore, it was concluded that B90DEC10 is a potential alternative to diesel, and the combustion performance and emissions control were optimized when using a higher EGR rate (15%).

**Keywords** Biodiesel · Diethyl carbonate · Spray · Combustion · Emission

## **Introduction**

Biodiesel is a form of renewable energy, and its raw materials are mainly vegetable oils, animal fats, and certain fatty algae. Although there are numerous biodiesel production processes, transesterifcation is an easy, cheap, and widely used method to produce biodiesel. These attributes make it conducive to the wide application of biodiesel (Bhan et al. [2022](#page-14-0)). The use of biodiesel can reduce pollution emissions

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18255107780@163.com

- Department of Thermal Science and Energy Engineering, University of Science and Technology of China, 96 Jinzhai Road, Hefei 230026, Anhui, China
- <sup>2</sup> Anhui Jianghuai Automobile Group Co. LTD, 99 Ziyun Road, Hefei 230601, Anhui, China
- <sup>3</sup> Hefei Technology College, Dazhong Road, Hefei 230012, Anhui, China

and relieve the pressure on fossil fuel supplies. Previous studies have confirmed that biodiesel has an  $R-(C=O)$ O–R' group, which can restrain the formation of soot precursors (Zhu et al. [2016;](#page-15-0) Chandrasekar et al. [2022\)](#page-14-1). The main components of biodiesel are unsaturated fatty acids, which increase the cetane number, viscosity, and density compared to diesel. In the spray characteristics, compared to diesel, biodiesel has a longer spray tip penetration and higher injection velocity due to its higher density. Shao et al. (Shao et al. [2022\)](#page-14-2) found that increasing the injection velocity of fuel can achieve a lower turbulent viscosity ratio and higher vortices, which can improve the combustion performance, achieve higher combustion efficiency, and reduce nitrogen  $\alpha$  oxide (NO<sub>x</sub>) emissions. Due to its higher viscosity, biodiesel has a weaker atomization effect after injection, resulting in inadequate mixing of biodiesel and air and a reduction in combustion efficiency (Sharma et al. [2022;](#page-14-3) Xie et al. [2015](#page-14-4)).

Adding oxygen-containing compounds to biodiesel can reduce its viscosity. The most commonly used oxygenated compounds are alcohols, esters, and ethers. However, methanol and ethanol are short-chain alcohols and cannot be mixed



 $\boxtimes$  Q. Lin

directly with biodiesel. Chaitanya and Mohanty ([2022\)](#page-14-5) found that adding 30% pentanol to waste plastic oil (WPO) resulted in a 6.6% increase in brake thermal efficiency (BTE), a decrease of 0.06 kg/kWh in brake-specifc fuel consumption (BSFC), a reduction of 127 ppm in  $NO<sub>x</sub>$  emissions, a 7 ppm increase in total hydrocarbon (THC) emissions, and a 0.03% increase in carbon monoxide (CO) emissions compared to unmodifed WPO. Although long-chain alcohols are soluble in biodiesel, their calorific value and oxygen content decrease, and their viscosity and surface tension increase when they introduced as additives(Kuszewski [2019](#page-14-6); Karabektas and Yilancilar [2022\)](#page-14-7). Ethers are also a common oxygen additive. Although the cetane value of ethers are large, ethers are volatile, explosive, and toxic, which results in safety risks during their transportation and use. Raju et al. [\(2020\)](#page-14-8) found that adding 12% diethyl ether to mango seed methyl ester (MSME20) resulted in a signifcant improvement in brake thermal efficiency (BTE). Additionally, there was a notable decrease in CO, THC,  $NO_x$ , and particulate matter emissions, with reductions of 10.68, 33.33, 10.33, and 27.72%, respectively at maximum load. Carbonate esters have a high oxygen content and cetane number, as well as low viscosity, and excellent miscibility with biodiesel, making them a highly promising oxygen-containing additive. Aguado-Deblas et al. ([2020\)](#page-13-0) analyzed the combustion and emission of DEC mixed with vegetable oil. The study found that a combination of DEC and vegetable oil could completely replace diesel. Although the heating value and output power were lower than those of diesel, the particulate matter (PM) was only 5–8% of diesel, producing almost zero emissions. Hellier et al. [\(2013](#page-14-9)) combined respectively a mixture of dibutyl carbonate (DnBC), diethyl carbonate (DEC), and dimethyl carbonate (DMC) with *n*-malethane at a ratio of 30%, and found that these blends reduced soot and CO emissions when compared to diesel. Bridjesh and Kannaiyan Geetha ([2020\)](#page-14-10) mixed DEC with WPO-diesel blends and found that the additives not only optimized the combustion process but also reduced particulate matter emissions. Additionally, Wu et al.  $(2024)$  $(2024)$  found that lower CO and NO<sub>x</sub> emissions could be obtained by mixing diesel with waste tire pyrolysis oil at volume ratios of 85 and 15%. It was also found that supplying hydrogen through the intake manifold during combustion could increase the calorifc value of the mixture, further improving torque and power, and reducing fuel consumption, therefore providing a promising alternative to diesel.

Previous studies have shown that exhaust gas recirculation (EGR), as an important means of engine combustion process control and  $NO<sub>x</sub>$  emissions reduction, can substantially reduce combustion temperature and inhibit the formation of  $NO<sub>x</sub>$  emissions (Zhang and Balasubramanian [2015](#page-14-12); Zhou et al. [2020](#page-15-1)). However, as the EGR rate increases, particularly when it surpasses 40%, there is a signifcant



increase in the emissions of CO, THC, and soot mentioned previously, and the optimal EGR rate is generally considered to be between 5 and 15% (Zhang et al. [2021a](#page-14-13); Huang et al. [2022\)](#page-14-14). When the EGR rate is less than 5%, there is less impact on the engine performance (Rami Reddy et al. [2021](#page-14-15); Sun et al. [2020](#page-14-16)). Agnese et al. (Magno et al. [2016](#page-14-17)) found that a slight increase in EGR could reduce  $NO<sub>x</sub>$ , while simultaneously not increasing PM emissions from biodiesel.

Diethyl carbonate is an oxygenated additive that is non-toxic, has excellent miscibility with biodiesel, and has a high cetane number. In the present study, we examined the spray characteristics of biodiesel, diesel and biodiesel blends mixed with DEC at a 10% volume ratio. Additionally, we investigated the combustion and emission characteristics on a direct injection engine using various EGR rates. The results of this study hold signifcant value for the engineering applications of biodiesel as a potential alternative to diesel fuel, as well as for DEC as a suitable oxygen-containing additive.

## **Materials and methods**

## **Test fuels**

The fuels used in the experiment were diesel (D100), biodiesel (B100), and biodiesel-DEC blends. Biodiesel was obtained from soybean oil through a transesterifcation process. The biodiesel-DEC blends had a volume fraction ratio of 90% biodiesel blended with 10% DEC (B90DEC10).

The thermophysical properties of fuels are displayed in Table [1](#page-1-0).

#### **Experimental setup**

The engine used in this study was a 4-cylinder, high pressure common rail and direct injection (CRDi) diesel that met the Euro VI emission standards. It had a displacement of 2.0 L. The

<span id="page-1-0"></span>**Table 1** The thermophysical properties of the test fuels (25 ℃)

Fuel	Properties				
	D <sub>100</sub>	DEC	<b>B100</b>	B90DEC10	
Density $(g/cm^3)$	0.83	0.938	0.848	0.857	
Viscosity (mPa.s)	3.11	0.78	5.90	4.62	
Surface tension (mN/m)	28.3	25.6	30.5	29.4	
Fire point $(^{\circ}C)$	71	46	83	72	
Heating value (MJ/kg)	42.5	21.1	37.5	35.9	
Cetane value	55	58	52	52.6	
Latent heat of vaporization 260 (kJ/kg)		360	300	304	
Oxygen content $(\%)$	0	40.7	10	13	
Flash point $(^{\circ}C)$	55	25	175	58	



<span id="page-2-0"></span>**Fig. 1** Experimental schematic diagram



**Fig. 2** Actual experimental setup diagram

<span id="page-2-1"></span>engine had a rated output power of 95 kW and rated speed of 3200 r min−1. The intake was turbocharged and intercooled, and the engine exhaust was circulated into the cylinder through the EGR valve and cooler. Table [2](#page-3-0) displays the fundamental parameters of the diesel engine. In Fig. [1,](#page-2-0) a schematic diagram of the experimental setup is depicted. Additionally, Fig. [2](#page-2-1) illustrates a diagram of the actual experimental setup.

During the test, a Schneider water-cooled electric dynamometer was utilized to measure the output torque and speed of the diesel engine, and an AVL puma-open engine measurement and control system was used to regulate the testing procedure. A fuel consumption analyzer (AVL 7351 CME) was used to monitor the fuel consumption during the test. A cylinder pressure sensor (KISTLER 6056A) was installed in the cylinder near the center of the combustion chamber to monitor the combustion pressure in the cylinder. The cylinder pressure sensor (KISTLER 6056A) was used to collect the cylinder pressure charge signal. These signal was frstly amplifed by the charge amplifer in the combustion analyzer and then passed through the data acquisition card, data interface module and data conversion module to generate the cylinder pressure data. A Horiba MEXA-7100D tail gas measurement system was used to measure CO, THC,  $NO<sub>x</sub>$ , and other conventional gas emissions in tail gas emissions, and an opaque opacimeter was used to analyze PM in the tail gas emissions. The cylinder pressure was determined by an AVL combustion analyzer at an interval of 1°CA. The DEWESoft analysis software of the combustion analyzer could calculate the instantaneous heat release rate and CA50 according to the collected cylinder pressure data. For all equipment, the specifc models and their accuracy are shown in Table [3.](#page-3-1)



<span id="page-3-0"></span>

Air intake type	Turbocharger with WGT	
Injection type	<b>CRDi</b>	
Cooling type	Water	
Number of cylinders	4	
<b>Stroke</b>	4	
Displacement $(L)$	2.0	
Bore $\times$ stroke (mm $\times$ mm)	$86\times86$	
Compression	16.5:1	
Rated power (kW)	95	
Rated speed $(r.min-1)$	3200	
Max. torque $(Nm)/speed$ (r min <sup>-1</sup> )	$320/(1400 - 2400)$	
Injector hole diameter (mm)	0.15	

<span id="page-3-1"></span>**Table 3** Type, production, and accuracy of measured parameters

Parameter	Type	Manufacturer	Accuracy
Electric dynamometer	129,106	Schneider	$+0.1$ Nm
Fuel consump- tion instrument	7351 CME	AVL.	$\pm 0.01$ kg/h
Combustion analyzer	Indicom V2.5	AVL.	
Opacimeter	483	AVL.	$\pm 0.001$ mg/m <sup>3</sup>
<b>Emission Test</b> Systems	$MEXA-7100D$	Horiba	$\pm 1$ ppm
Angle encoder	<b>TRD-GK3600-</b> RZ.	Koyo	500 P/R
Crankshaft Posi-0281002315 tion Sensor		<b>Bosch</b>	$+0.1$ °CA
Exhaust gas temperature sensor	DARTS200	Delphi	+1 K
Cylinder pres- sure transducer	6056A	<b>KISTLER</b>	$+0.1$ MPa

<span id="page-3-2"></span>**Table 4** Test conditions of the engine



#### **Experimental method**

The engine's test conditions are displayed in Table [4.](#page-3-2) During the test, the engine speed was set at 1600 r/min, and the engine loads were varied at 10, 30, 50, 80, and 100%. The converted to the brake mean efective pressures (BMEPs) for



each load were recorded as 0.2, 0.6, 1.0, 1.6, and 2.0 MPa. The test was conducted without making any modifcations to the diesel engine's structure. The frst test was conducted using diesel (D100) to determine the reference value. Before each fuel was replaced, we emptied the fuel remaining in the fuel tank and fuel consumption instrument, and cleaned the fuel pipe by running stably for 0.5 h under idling conditions.

After passing through the EGR valve and cooler, the combustion exhaust entered the intake manifold and mixed with fresh air before entering the cylinder. The amount of exhaust gas could be regulated by adjusting the EGR valve opening as required. The EGR rates were determined by calculating the mass of recirculated exhaust gases entering the cylinder  $(m_{EGR})$ in relation to the total intake mass entering the cylinder  $(m<sub>intake</sub>)$ :

$$
EGR = \frac{m_{EGR}}{m_{\text{int}} \,_{ake}} \times 100\%
$$
 (1)

The instantaneous heat release rate (HRR) was calculated by integrating the combustion pressure and volume in the cylinder with the crank angle. The formula for this calculation was as follows:

$$
\frac{dQ}{d\theta} = \frac{\gamma - 1}{\gamma} P \frac{dV}{d\theta} + \frac{1}{\gamma - 1} V \frac{dP}{d\theta}
$$
(2)

where  $\theta$  is the crank angle; V and P represent the instantaneous working volume and in-cylinder combustion pressure, respectively;  $\gamma$  is a constant that represents the adiabatic index of mixture gases (fxed at 1.35).

## **Results and discussion**

#### **Macro spray characteristics**

In order to compare and analyze the breakup and atomization of the fuels, their spray characteristics were tested in



<span id="page-4-0"></span>**Fig. 3** Defnition of the spray characteristic parameters

a constant volume device using the Schlieren method (Li et al. [2018\)](#page-14-18). The fuel injection system was a high-pressure common rail system. The injection pressure for the test was 120 MPa, the ambient pressure was 2 MPa, and the ambient temperature was room temperature. Figure [3](#page-4-0) shows the defnition of the macro spray characteristics parameters. The spray tip penetration was the vertical distance from the bottom of the spray to the nozzle (Zhao et al. [2023](#page-15-2)). A triangle was formed between the bottom of the nozzle and the spray width at half of the spray tip penetration, the two sides of the triangle were tangent to the outer edge of the spray. The angle formed by the two sides was referred to as the spray cone angle, and it was determined using a trigonometric function.

Figure [4](#page-4-1) displays the spray morphology resulting from the fuel spray. The spray morphology of diesel was a symmetrical triangle. The spray morphology of biodiesel was elongated, which was directly related to the high density. This was because a high density can increase the initial kinetic energy after injection and increase the spray tip penetration (Fu et al. [2019](#page-14-19)). When biodiesel was mixed with DEC in a 10% volume ratio, the tail morphology of B30DEC10 showed more signifcant perturbation and difusion than the unmodifed biodiesel. This was because DEC reduces the viscosity of blended fuel and intensifes the fragmentation of blended fuel droplets under the efect of air entrainment.

Figure [5](#page-5-0) displays the fndings of quantitative analysis on the spray characteristics of various fuels. It is evident that biodiesel exhibits a greater spray tip penetration. Specifcally, the spray tip penetration of biodiesel was 3 mm greater than that of diesel at 1.4 ms. This was because the densities of biodiesel were greater than that of diesel,

Fuels Times	D100	<b>B100</b>	B90DEC10
0.1 <sub>ms</sub>		ı	
0.3ms			
0.5 <sub>ms</sub>			
0.7ms			
0.9ms			
1.1ms			
1.3ms			
1.5 <sub>ms</sub>			

<span id="page-4-1"></span>**Fig. 4** Spray morphology of the test fuels

resulting in an increased the momentum of the fuel. The spray tip penetration of B90DEC10 was slightly lower than that of biodiesel, but it was still greater than that of diesel. This is because although B90DEC10 has a highest density, its surface tension and viscosity are lower than those of biodiesel, which promotes B90DEC10 to be broken up more easily and the momentum leaving the nozzle is decreased (Zhang et al. [2021b\)](#page-15-3).

Biodiesel exhibited the smallest spray cone angle, while B90DEC10 had a larger spray cone angle. This was due to the fact that when DEC was mixed with biodiesel, the surface tension and viscosity of the blended fuel were decreased, leading to improved droplet breaking ability and atomization quality. The atomization of biodiesel was poor, resulting in a lower projection area compared to diesel. Due to its low viscosity and high density, B90DEC10 exhibited increased spray tip penetration and spray cone angle compared to biodiesel. Additionally, the spray projection area was 2% larger than that of biodiesel. When analyzing Schlieren images, the threshold of the gray value is defned as 30, and the image region with a gray value  $<$  30 is regarded as the liquid core area (Yi et al. [2018](#page-14-20)). The percentage of the liquid core area to the spray area was defned as the spray concentration (Zhao et al. [2023](#page-15-2)). In Fig. [5,](#page-5-0) Biodiesel had the lowest spray concentration, which indicated that the difusion





<span id="page-5-0"></span>**Fig. 5** Quantitative analysis on the macroscopic spray characteristics of various fuels

ability of biodiesel after injection was weaker than for the other fuels. However, the average spray concentration of B90DEC10 was 4.38% higher than that of biodiesel. This indicated that adding a 10% volume ratio of DEC to biodiesel could enhance atomization quality and improve combustion efficiency.

### **Combustion characteristics**

Figure [6](#page-6-0) illustrates the comparison of in-cylinder combustion pressure and heat release rate (HRR) at various loads. Due to the pilot injection, a small wave peak occurred before peak of the instantaneous heat release rate. The start of combustion (SOC) was latest for B90DEC10, while diesel was the earliest. This was because B90DEC10 had the lowest cetane number and highest latent heat of vaporization (Liang et al. [2021](#page-14-21); Qi et al. [2021\)](#page-14-22). The maximum HRR of B90DEC10 was 20% higher than that of diesel at 0.6 MPa. It is due to that the air/fuel ratio in the cylinder was relatively



smaller at low loads, and B90DEC10 can provide more oxygen atoms for combustion, which was benefcial for promoting the combustion (Natesan et al. Mar [2019\)](#page-14-23). Under medium to high loads, the intake air volume in the cylinder increased, so the maximum HRR of B90DEC10 was similar to that of diesel after 1.0 MPa. Biodiesel and B90DEC10 had the highest fuel injection quantity and also had inherent oxygen rich properties, leading to an increase in the premixed combustion period (Alruqi et al. [2023](#page-13-1)). B90DEC10 released more heat during premixed combustion due to its longer premixed combustion period. The peak in-cylinder combustion pressure is primarily determined by the rate of combustion during the premixed combustion period (Raju et al. [2020](#page-14-8)). The low viscosity of diesel engines improved the uniformity of the diesel and air mixture, resulting in a faster combustion rate and higher peak cylinder pressure during the initial stage of combustion. Moreover, diesel had the largest heating value, resulting in the largest peak incylinder combustion pressure. Although the peak in-cylinder



<span id="page-6-0"></span>**Fig. 6** Comparison of in-cylinder combustion pressure and the HRR at various loads

combustion pressure of B90DEC10 was 0.2–3% lower than that of diesel, it was 0.3–2% higher than that of biodiesel. Obviously, the lower viscosity of B90DEC10 promoted the atomization of fuel and improved the combustion efficiency.

Figure [7](#page-7-0) illustrates the comparison of in-cylinder combustion pressure and heat release rate at various EGR rates. At an EGR rate of 5%, there were no signifcant decrease in the maximum heat release rate (HRR) and the peak in-cylinder combustion pressure for both biodiesel and





<span id="page-7-0"></span>**Fig. 7** Comparison of in-cylinder combustion pressure and the HRR at various EGR rates

B90DEC10. However, the maximum HRR and the peak in-cylinder combustion pressure of diesel fuel decreased by 3 and 6%, respectively. This was due to the diferent thermophysical properties of the fuels. Firstly, the oxygen content of biodiesel and B90DEC10 could compensate for the insufficient concentration of oxygen in the combustion chamber when using 5% EGR rate. In addition, their low cetane number and poor fammability also prolonged the ignition delay, resulting in a more uniform mixture of fuel/ air and more conducive to promoting combustion. However, diesel does not contain oxygen, and EGR has a more obvious inhibitory efect on diesel combustion, which reduced the premixed combustion period and the HRR. When the EGR rate reached 15%, the peak in-cylinder combustion pressure of biodiesel and B90DEC10 decreased by 4 and 6%, respectively; and the maximum HRR decreased by 9 and 7%, respectively (compared with EGR0). In contrast, the maximum HRR and the peak in-cylinder combustion pressure of diesel decreased by 14 and 7% respectively, which was a higher reduction rate than that of B90DEC10. This



was because the amount of exhaust gas entering into the combustion chamber increased as the EGR rate increased, which signifcantly hindered combustion. However, oxygencontaining fuels can alleviate the hindrance of the increasing EGR rate.

Figure [8](#page-8-0) shows the variation of the CA50 with various EGR rates and loads. The CA50 was defned as the time at which 50% of the total heat release during combustion occurred. This metric is commonly used to analyze the center of gravity of the combustion process. Due to its high viscosity and low evaporability, biodiesel had a slower combustion rate, resulting in a delayed CA50. The CA50 of B90DEC10 was 0.3–0.7°CA earlier than that of biodiesel. This was due to the fact that B90DEC10 possesses the qualities of lower viscosity and higher oxygen content, which are advantageous in promoting combustion and accelerating the combustion rate. As a result, it leads to an earlier timing of CA50. As the EGR rate increased, the combustion rate decreased and the CA50 was delayed. For instance, at an EGR rate of 5%, the CA50 of B90DEC10 was delayed



<span id="page-8-0"></span>**Fig. 8** Variation of the CA50 with various loads and EGR rates



<span id="page-8-1"></span>**Fig. 9** Comparison of BSFC at various loads and EGR rates

by 0.2°CA; However, at an EGR rate of 15%, the CA50 of B90DEC10 was delayed by 0.5 CA.

Figure [9](#page-8-1) displays a comparison of brake-specifc fuel consumption (BSFC) at various EGR rates and loads. The BSFC of B90DEC10 and biodiesel were 15 and 20% higher than that of diesel, respectively. This was because diesel had the highest heating value, followed by B100, and DEC had the lowest heating value. As the EGR rate increased, the BSFC of fuels increased slightly. Figure [10](#page-9-0) displays a comparison of BTE at various load and EGR rates. Diesel had the highest BTE, and the BTE of B90DEC10 was 1.1–2.8% higher than that of biodiesel. Compared to biodiesel, B90DEC10 has a lower viscosity and improves the atomization efect. This diference promotes the full mixing of B90DEC10 with air, resulting in improved combustion. At the same time, the

addition of DEC to biodiesel increased the oxygen content of mixture and promoted full combustion. Moreover, the CA50 of B90DEC10 was advanced and became closer to the top dead center, making expansion work more efective.

The exhaust gas temperature before the turbine (EGT) is crucial indicator for analyzing the combustion performance of a diesel engine. According to Fig. [11](#page-9-1), the exhaust gas temperature before the turbine follows the order of diesel, biodiesel and B90DEC10. This was because diesel had the highest heating value and a good uniformity in its mixing with air, resulting in sufficient combustion. In contrast, biodiesel had a high viscosity and poor atomization efect, resulting in insufficient combustion. Due to B90DEC10 having the lowest heating value, its exhaust gas temperature before the turbine was also the lowest.











<span id="page-9-0"></span>**Fig. 10** Comparison of BTE at various loads and EGR rates



<span id="page-9-1"></span>**Fig. 11** variation of EGT before the turbine with various loads and EGR rates

<span id="page-9-2"></span>**Table 5** Regulatory limits of Euro VI for heavy-duty diesel engine

<b>WHSC</b>	PМ	$NO_{\rm v}$	THC	CO.
Unit	g/kWh	g/kWh	g/kWh	g/kWh
Regulatory limits	0.01	0.4	0.13	1.5

#### **Emission characteristics**

The PM,  $NO<sub>x</sub>$ , THC, and CO emissions were measured in real-time during the combustion process. The emission variation patterns of diesel, biodiesel, and biodiesel/DEC were studied. Table [5](#page-9-2) shows the regulatory limits of Euro VI for heavy-duty diesel engine according to the World Harmonized Stationary Cycle (WHSC). In this paper, the measured



emission results are compared with the regulatory limits of Euro VI.

Figure [12](#page-10-0) displays a comparison of PM emissions at various loads and EGR rates. Biodiesel and B90DEC10 had lower PM emissions than diesel. B90DEC10 had the lowest PM emissions, with emissions being 92% lower compared to diesel. This was because both DEC and biodiesel were oxygenated fuels, which promoted the oxidation of PM. The mixture also contained low levels of aromatic hydrocarbons, which were beneficial to reduce the formation of PM precursors (Serhan et al. [2018\)](#page-14-24). Compared to diesel, the PM emissions of biodiesel were approximately 80% lower at 30% load, and B90DEC10 generally had the lowest PM emissions. When the EGR rate reached 15%, there was a notable increase in PM emissions from both diesel and biodiesel.





<span id="page-10-0"></span>**Fig. 12** Comparison of the PM emissions at various loads and EGR rates

However, there was no notable increase in PM emissions from B90DEC10, as its oxygen content had reached 13%. Therefore, when B90DEC10 was used instead of diesel, PM emissions were far below the regulatory limit of the Euro VI standard.

Figure [13](#page-10-1) displays a comparison of  $NO<sub>x</sub>$  emissions at various fuels and EGR rates. High temperature and oxygenrich environment within the cylinder are the primary factors that contribute to the formation of  $NO<sub>x</sub>$  (Chen et al. [2019](#page-14-25)). Due to its higher oxygen content, biodiesel emitted 1.1–1.2 times more  $NO_x$  emission than diesel (Ayhan et al. [2020](#page-14-26)). The  $NO<sub>x</sub>$  emissions of B90DEC10 were 8–20% larger than those of diesel. When the BMEP was lower than 1.6 MPa, the  $NO<sub>x</sub>$  emissions of B90DEC10 were comparable to those of biodiesel, with no signifcant diference observed. This

was because DEC has a high latent heat of vaporization, and its evaporation needs to absorb large amounts of heat during combustion, which in turn lowers the temperature in combustion chamber. This decrease in temperature helps to suppress the formation of  $NO<sub>x</sub>$  emissions. When the BMEP was 2.0 MPa, there was a noticeable rise in  $NO<sub>x</sub>$  emissions from B90DEC10. This is due to that higher combustion temperatures were more likely to promote the generation of  $NO<sub>x</sub>$  emissions from oxygenated fuels under high load. We are aware that that  $NO<sub>x</sub>$  emissions can be effectively reduced through EGR. For instance, When the EGR rate was increased to 5%, there was a reduction of 12–22% in  $NO_x$ emissions. When the EGR rate was further increased to 15%, there was a significant reduction of  $45\%$  in NO<sub>x</sub> emissions. Additionally, when the EGR rate reached 15%, both B100



<span id="page-10-1"></span>**Fig. 13** Comparison of the NO<sub>x</sub> emissions at various loads and EGR rates





<span id="page-11-0"></span>**Fig. 14** Comparison of THC emissions at various loads and EGR rates

and B90DEC10 showed a noticeable increase in PM emissions, which breaked the trade-off balance between PM and  $NO<sub>x</sub>$  emissions (Oztürk and Can [2022](#page-14-27)).

Figure [14](#page-11-0) displays a comparison of THC emissions at various loads and EGR rates. Hydrocarbons are mainly produced in the wall quenching area, or in areas where the mixture is too lean or too rich (Venu et al. [2019](#page-14-28)). The mixture being too rich can lead to a high temperature in combustion chamber, resulting in an increase in THC emissions. On the other hand, if the mixture is too lean, it can lead to a low temperature in combustion chamber, resulting in higher THC emissions. The THC emissions were lowest for biodiesel because the higher oxygen content of biodiesel was conducive to their oxidation, and biodiesel contained almost no aromatic hydrocarbons. Additionally, the cracking of hydrocarbons was reduced in the biodiesel. Biodiesel had been found to produce THC emissions that were 20–30% lower than those of diesel at low loads. Additionally, the THC emissions of biodiesel fell below the regulatory limit. However, although the oxygen content was highest in B90DEC10, the THC emissions of B90DEC10 were larger than those of biodiesel at medium and low loads. Compared to biodiesel, the THC emissions increased by 20–45%. This was because the latent heat of vaporization of B90DEC10 was larger than for the other fuels, which significantly reduced the combustion temperature and led to an increase in the size of the quenching area in the cylinder (Babu and Anand [2017\)](#page-14-29). Additionally, the density of B30DEC10 was larger, and the combustion fame may strike the wall. The wall surface can rapidly cool the combustion fame in a process called cold shock or quenching, which will generate active free radicals in the combustion fame slow or even halt the chemical reaction, resulting in increased THC emissions. However, the THC emissions of B90DEC10 were lower than



those of biodiesel at high loads. This was due to the high oxygen content of B90DEC10 promoting the oxidation of THC. Additionally, the thermal conditions in the cylinder were effectively improved and the combustion temperature was higher with the increasing load, which was conducive to the oxidation of THC (Raju et al. [2020\)](#page-14-8). When the EGR rates were below 15%, low EGR rates had little efect on THC emissions.

Figure [15](#page-12-0) displays a comparison of CO emissions at various loads and EGR rates. The main reasons for the increase in CO emissions were the lower temperature or insufficient concentration of oxygen during the combustion reaction of the fuel (Esakki et al. [2022;](#page-14-30) Jayabal et al. [2020\)](#page-14-31). The combustion temperature and air intake in the engine cylinder both increased as the load increased, which promoted the oxidation of CO and decreased CO emissions. Diesel had the lowest CO emissions at low loads  $(BMEP=0.2 MPa)$ , and biodiesel had 45% higher CO emissions than diesel. When DEC was added to biodiesel, the CO emissions were higher than those of biodiesel. This was because CO emissions were related to the oxygen concentration during the combustion reaction, and an excessive oxygen concentration led to an increase in CO emissions. After adding DEC to biodiesel, the oxygen content of the fuel increased to 13%, resulting in an excessive air/fuel ratio and increasing the CO emissions (Chaitanya and Mohanty [2022](#page-14-5)). The latent heat of vaporization is another important infuencing factor. The high latent heat of evaporation of B90DEC10 means that it needs to absorb more heat compared to other fuels during the evaporation process. This results in a lower temperature of combustion in the cylinder, leading to an increase in CO emissions. When the BMEP was greater than 0.6 MPa, diesel produced the highest CO emissions. This was because the pressure and temperature in the combustion chamber



<span id="page-12-0"></span>**Fig. 15** Comparison of CO emissions at various loads and EGR rates



<span id="page-12-1"></span>**Fig. 16** Trade-off relationship between NO<sub>x</sub> and PM emissions

were higher, and the oxygen concentration in the combustion chamber was the important factor afecting CO emissions. The CO emissions increased with the increase of EGR rate. When the EGR rate was 15%, CO emissions increased by 12% on average at low loads. This was because EGR reduced the temperature and oxygen concentration in combustion chamber and inhibited the oxidation of CO. The efects of EGR on  $NO<sub>x</sub>$  emission reductions were more effective and signifcant than the increase in THC and CO emissions.

Figure [16](#page-12-1) illustrates the trade-off relationship between  $NO<sub>x</sub>$  and PM emissions. The trade-off curve applies to conventional diesel engines. Particulate matter is easily oxidized in the oxygen-rich and high temperature environment of the cylinder, but these are also favorable conditions for the generation of  $NO<sub>x</sub>$  (Wu et al. [2022](#page-14-32)). Generally, the concentration of oxygen and the temperature in



combustion chamber decreases with the increase of EGR rate. This effectively inhibits the generation of  $NO<sub>x</sub>$  emissions, but it also leads to an increase in PM emissions. In this study, the use of B90DEC10 could disrupt this PM- $NO<sub>x</sub>$  trade-off. As the EGR rate increased, there was a little increased in PM emissions. Exhaust gas recirculation had a significant impact on reducing  $NO_x$  emissions, and  $NO_x$ emissions decrease linearly with the increase of EGR rate. According to the Zelovich mechanism, the main chemical reaction for generating NO<sub>x</sub> is:  $O_2 + N \rightarrow NO + O$ ,  $O + N_2 \rightarrow NO + N$ , while the chemical reaction at high temperatures is:  $N_2 + O_2 \rightarrow 2NO$ ,  $NO + O_2 \rightarrow NO_2$ . The rate of these chemical reactions increases exponentially with an increase in temperature. Exhaust gas recirculation introduced  $CO<sub>2</sub>$  and other inert gases into the cylinder, diluting the local oxygen concentration in the mixture. At the same time, the specific heat capacity of  $CO<sub>2</sub>$  and other inert gases was larger, which reduced the combustion temperature and inhibited the generation of  $NO<sub>x</sub>$  emissions. When the EGR rate was increased to 15%, the PM emissions of diesel increased by 100%, whereas the PM emissions of B90DEC10 remained almost unchanged. For B90DEC10, the EGR rate of 15% achieved a better PM-NO<sub>x</sub> trade-off.

Figure [17](#page-13-2) illustrates the relationship between BTE and CO emissions at various EGR rates. While increasing the EGR rate could effectively reduce  $NO<sub>x</sub>$  emissions, it also decreased the concentration of oxygen in the combustion chamber. This could limit combustion efficiency, resulting in reduced thermal efficiency and increased CO emissions. Therefore, the relationship between thermal efficiency and CO emissions could be used to determine an appropriate EGR rate (Natesan et al. Mar [2019](#page-14-23)). Diesel and biodiesel exhibited the highest thermal efficiency at an EGR rate of 5%, with only a minimal increase in CO emissions. Therefore, an EGR rate of 5% was more appropriate. If the





<span id="page-13-2"></span>**Fig. 17** The relationship between BTE and CO emissions at various EGR rates

EGR rate reaches 15%, B90DEC10 has the higher thermal efficiency and the lower CO emission compared to diesel and biodiesel. Because the EGR rate of 15% was conducive for reducing  $NO<sub>x</sub>$  emissions, the most appropriate EGR rate for use with B90DEC10 was considered to be 15%.

## **Conclusion**

The spray, combustion and emission characteristics of diesel, biodiesel and biodiesel/DEC mixtures were studied and compared in an unmodifed CRDi diesel engine that met European VI emission standards. The following conclusions could be obtained through the above research.

The viscosity of B90DEC10 was reduced, and the atomization efect was improved compared to the other fuels. The maximum HRR of B90DEC10 was 20% greater than that of diesel at efective pressures below 0.6 MPa. The duration of the premixed combustion period was longer and the heat release rate (HRR) during this period was higher compared to diesel at efective pressures above 1.0 MPa. The BSFC of B90DEC10 was 2–5% greater than that of biodiesel. However, the BTE of B90DEC10 was 1.1–2.8% greater than that of biodiesel. If the EGR rate reached 15%, the peak in-cylinder combustion pressure and the maximum HRR decreased by 5 and 6%, respectively, and BTE was slightly reduced. The 15% EGR rate had minimal efect on engine thermal efficiency when using B90DEC10.

The oxygen content of fuel had a crucial role on the particulate matter emission. B90DEC10 produced the lowest PM emissions, with a 92% reduction compared to diesel, and disrupted  $PM-NO<sub>x</sub>$  emissions trade-off relationship. The  $NO<sub>x</sub>$  emissions of B90DEC10 were 8–20% larger than those of diesel. Exhaust gas recirculation signifcantly reduced  $NO<sub>x</sub>$  emissions. If the EGR rate reached 15%,  $NO<sub>x</sub>$  emissions of B90DED10 decreased by 45%. Interestingly, there was no signifcant increase in PM emissions observed. Under low load, the B90DEC10 had higher THC and CO emissions compared to diesel. Under medium and high loads, the pressure and temperature in the combustion chamber were enhanced, and the oxygen content of B90DEC10 was benefcial for the oxidation of THC and CO emissions, resulting in the lowest THC and CO emissions among the test fuels. B90DEC10 is therefore a potential alternative to diesel, with the best performance and emissions obtained when using a higher EGR rate (15% EGR).

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**Data Availability** All data, models, or code that support the fndings of this study are available from the corresponding author upon reasonable request.

#### **Declaration**

**Conflict of interest** The authors have no fnancial or proprietary interests in any material discussed in this article.

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