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# **One‑pot, three‑component synthesis and in vitro antibacterial evaluation of novel 3‑amino‑N‑benzyl‑1‑aryl‑1H‑benzo[f] chromene‑2‑carboxamide derivatives**

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**Abstract** A series of novel 3-amino-N-benzyl-1-aryl-1Hbenzo[f]chromene-2-carboxamides have been synthesized by the reaction of N-benzyl-2-cyanoacetamide with 2-naphthol and aromatic aldehydes in the presence of piperidine as a base. The chemical structures of all the compounds were confrmed by spectroscopic methods as well as elemental analysis. In addition, the antibacterial activities of some target compounds against Gram-negative and Gram-positive bacteria in vitro were determined. Among these compounds, 4b, 4c, 4h and 4i exhibited antibacterial efects. The advantages of the present work reaction are clean reaction, metalfree transition, environmentally friendly approach to prepare diferent benzo[f]chromene-2-carboxamide derivatives, good yields, low cost, readily available starting materials and in addition does not need cumbersome procedure.

**Keywords** Multi-component reactions (MCRs) · 1H-benzo[f]chromene-2-carboxamides · Antibacterial activity

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### **Introduction**

Multi-component reactions (MCRs) have proven to be a unique and versatile tool in organic and medicinal chemistry due to their ability to synthesize diverse chemical libraries of pharmaceutical agents with higher efficiency and atom economy in a single step from three or more diferent starting materials reacting. Additionally, multi-component protocols have the several advantages such as experimental simplicity and synthetic efficiency over conventional chemical reactions [[1](#page-6-0)]. Therefore, designing of new MCRs for the synthesis of diverse groups of compounds, especially the ones that are biologically active, has gained great attention in green organic synthesis [[2–](#page-6-1)[4\]](#page-6-2). Hence, MCRs gained signifcant popularity as a pivotal theme in the synthesis of many important heterocyclic 'drug-like' libraries such as chromene derivatives currently. Chromone (4H-benzopyran-4-one) is a heterocyclic compound found primarily in plants but also found in other sources [[5\]](#page-6-3). Simple chromones and their analogues are a large class of heterocycles that have attracted a great deal of interest for a long time either from a biosynthetic and synthetic point of view or owing to their interesting biological activities.

Especially, among various chromene derivatives, 2-amino-4H-chromenes have attracted great interest due to their wide range of biological and pharmaceutical properties such as antibacterial antimicrobial [[6\]](#page-6-4), antiviral [\[7](#page-6-5)], mutagenicity  $[8]$ , sex pheromone  $[9]$  $[9]$  and antitumor  $[10]$  $[10]$ activities. A few exemplary biologically active 4H-chromene derivatives are collected in Fig. [1](#page-1-0).

Recently, the synthesis of 2-amino-4H-chromenes has been achieved by the condensation of 1-naphthol, malononitrile and aromatic aldehydes in the presence of a catalyst, such as methanesulfonic acid [\[11\]](#page-6-9), tetrabutylammonium bromide [[12\]](#page-6-10), TiCl<sub>4</sub> [[13\]](#page-6-11), K<sub>2</sub>CO<sub>3</sub> [\[14](#page-6-12)], heteropolyacid [\[15](#page-6-13)],

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<span id="page-1-0"></span>



basic alumina [\[16\]](#page-6-14), 1-butyl-3-methyl imidazolium hydrox-ide [[17](#page-6-15)], KF-Al<sub>2</sub>O<sub>3</sub> [\[18](#page-6-16)], iodine and K<sub>2</sub>CO<sub>3</sub> [\[19](#page-6-17)], diazabicyclo-[2.2.2]octane [\[20\]](#page-6-18), Mg/Al hydrotalcite [\[21](#page-6-19)], piperidine [[22](#page-6-20)], montmorillonite KSF [\[23\]](#page-6-21), nanosized magnesium oxide  $[24]$  $[24]$  $[24]$ , NaHCO<sub>3</sub>  $[25]$  $[25]$  $[25]$ , hexadecyl trimethylammonium bromide [[26](#page-6-24)], cetyltrimethylammonium chloride [[27\]](#page-6-25), cetyltrimethylammonium bromide [[28](#page-6-26)] and N, N-dimethylaminoethyl-benzyldimethylammonium chloride [[29](#page-6-27)]. Due to the important aforementioned properties of chromene derivatives, considerable attentions have been focused on the synthesis of new 3-amino-N-benzyl-1-aryl-1H-benzo[f] chromene-2-carboxamide derivatives.

Herein, the synthesis of new chromene-2-carboxamide by the reaction of N-benzyl-2-cyanoacetamide with 2-naphthol and aromatic aldehydes has been described (Scheme [1\)](#page-1-1).

## **Experimental section**

#### **Chemicals**

All chemicals were either prepared in our laboratory or purchased from Merck and Aldrich Chemical Corporation and were used without any further purifcation. All reactions

<span id="page-1-1"></span>**Scheme 1** Three-component synthesis of 3-amino-N-benzyl-1-aryl-1H-benzo[f]chromene-2-carboxamides

were monitored by TLC, petroleum–ethyl acetate (3:1). Melting points were determined with a hot-plate microscope apparatus.

#### **Apparatus**

IR spectra were recorded in KBr, using a BRUKER FT-IR spectrophotometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on Bruker 500 MHz and 300 MHz spectrometers using  $DMSO, CDCl<sub>3</sub>$  and TMS as solvent and internal standard, respectively. Elemental analyses for C, H and N were performed on a 204Q CHN (Heraeus, USA).

## **General procedure for the synthesis of 3‑amino‑N‑benzyl‑1‑aryl‑1H‑benzo[f] chromene‑2‑carboxamide 4**

N-benzyl-2-cyanoacetamide (1) was synthesized from benzylamine and ethyl cyanoacetate following the reported procedure  $[30]$  $[30]$  (Table [1](#page-2-0)).

A mixture of *N*-*benzyl*-*2*-*cyanoacetamide* (1 mmol), aromatic aldehydes (1 mmol), 2-naphthol (1 mmol) and piperidine (0.1 mmol) in absolute EtOH (5 ml) was refuxed for an appropriate time as shown in Table [2](#page-2-1). The



<span id="page-2-0"></span>**Table 1** Catalyst screening and reaction condition optimizations<sup>a</sup>





<sup>a</sup> Reaction condition: 1 (1 mmol), 2 (benzaldehyde, 1 mmol), 3 (1.1 mmol) and 4 (1 mmol) <sup>b</sup> Isolated yields

<span id="page-2-1"></span>**Table 2** Chromene-2-carboxamides **4** from N-benzyl-2-cyanoacetamide **1**, 2-Naphthol **3** and Aromatic aldehydes **2**

$\overline{4}$	Ar	Time (h)	Yield <sup>a</sup> $(\% )$	M.p. $(^{\circ}C)$
4a	Ph	3	75	$125 - 126$
4b	$4-CIC6H4CHO$	2.5	70	$183 - 185$
4c	4-BrC <sub>6</sub> H <sub>4</sub> CHO	2.5	72	$127 - 128$
4d	$4-NO_2C_6H_4CHO$	$\overline{c}$	80	$227 - 230$
4e	$3-NO_2C_6H_4CHO$	$\overline{c}$	75	148-149
4f	$4-CH_3C_6H_4CHO$	3.5	60	133-134
4g	$4-OCH3C6H4CHO$	$\overline{4}$	68	$227 - 230$
4h	2, 4-Cl,ClC <sub>6</sub> H <sub>4</sub> CHO	$\mathfrak{D}$	78	146-148
4i	3-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub> CHO	$\overline{4}$	65	$101 - 103$
4j	4-OHC <sub>6</sub> H <sub>4</sub> CHO	$\overline{4}$	70	192-194

a Isolated yield of pure product

reaction progress was monitored by TLC. After completion of the reaction, the mixture was cooled to r.t and then  $H<sub>2</sub>O$  (5 mL) was added. The solid product was filtered, washed with cold distilled water (2 mL) to obtain essentially pure products. The solid products were recrystallized from ethanol.

The structure of the products was characterized by IR,  ${}^{1}H$ and 13C NMR spectra along with elemental analysis data.

**3‑Amino‑N‑benzyl‑1‑phenyl‑1H‑benzo[f] chromene‑2‑carboxamide (4a):** White solid, Yield 75%; mp: 125–126 °C. IR (KBr) ( $v_{\text{max}}$ , cm<sup>-1</sup>): 3425(NH), 1683 (C=O). Anal. calcd. for  $C_{27}H_{22}N_2O_2$  (406.48): C, 79.78; H, 5.46; N, 6.89; % found: C, 79.70; H, 5.54; N, 6.84, % d<sub>H</sub>  $(400 \text{ MHz}, \text{CDCl}_3)$  8.76 (s, 2H, NH<sub>2</sub>), 7.66–7.86 (m, 4H, CH arom), 7.53–7.66(m, 3H, CH arom), 7.44–7.53 (d, *J* = 8 Hz, 2H, CH arom), 7.29–7.35 (m, 5H, CH arom and NH), 6.96  $(t, J = 7.5 \text{ Hz}, 1H, CH$  arom), 6.94 (m, 1H, CH arom), 6.84  $(t, J = 7.5 \text{ Hz}, 1H, \text{CH}$  arom), 5.20 (s, 1H, CHPh), 4.15 (d,  $J = 5$  Hz, 2H, CH<sub>2</sub>NH). d<sub>C</sub> (100 MHz, CDCl<sub>3</sub>) 37.18, 44.16, 87.40, 11931, 122.71, 122.99, 123.44, 124.26, 125.43, 125.67, 126.54, 128.01, 128.27, 128.43, 128.45, 128.58, 128.86, 128.91, 128.97, 129.58, 130.56, 131.05, 131.45, 134.84, 143.35, 163.38, 177.39.

**3‑Amino‑N‑benzyl‑1‑(4‑chlorophenyl)‑1H‑benzo[f] chromene‑2‑carboxamide (4b):** White solid, Yield 70%; mp: 183–185 °C. IR (KBr) (νmax, cm−1): 3420 (NH), 1675 (C=O). Anal. calcd. for  $C_{27}H_{21}CIN_2O_2$ (440.92): C, 73.55; H, 4.80; N, 6.35% found: C, 73.44; H, 4.90; N, 6.25.  $d_H$  (300 MHz, CDCl<sub>3</sub>) 8.15 (s, 2H, N<u>H</u><sub>2</sub>),

7.85–7.87 (d, *J* = 7 Hz, 2H, CH arom), 7.64–7.69 (m, 3H, CH arom), 7.36–7.47 (m, 6H, CH arom), 7.20–7.36  $(m, 5H, CH$  arom and NH $), 5.30$  (s, 1H, CHPh), 4.40 (d,  $J = 5$  Hz, 2H, CH<sub>2</sub>NH). d<sub>c</sub> (75 MHz, CDCl<sub>3</sub>) 35.35, 55.06, 87.50, 114.49, 121.95, 128.28, 128.74, 128.80, 130.45, 130.50, 130.98, 131.52, 134.41, 135.06, 135.81, 139.99, 147.38, 150.77, 152.79, 153.30, 154.38, 155.77, 164.41, 175.06.

**3‑Amino‑N‑benzyl‑1‑(4‑bromophenyl)‑1H‑benzo[f] chromene‑2‑carboxamide (4c):** White solid, Yield 72%; mp: 127–128 °C. IR (KBr) ( $v_{\text{max}}$ , cm<sup>-1</sup>): 3435(NH), 1670 (C=O). Anal. calcd. for  $C_{27}H_{21}BrN_2O_2$  (484.08): C, 66.81; H, 4.36; N, 5.77% found: C, 66.79; H, 4.11; N, 5.88%.  $d_H$  $(500 \text{ MHz}, \text{DMSO}, \text{d}_6)$  8.26 (s, 2H, NH<sub>2</sub>), 7.86–7.87 (d,  $J = 8$  Hz, 3H, CH arom), 7.61–7.66 (m, 5H, CH arom and N<u>H</u>), 7.42 (s, 1H), 7.39–7.40 (d,  $J = 8$  Hz, 3H, CH arom), 7.30–7.33 (t, *J* = 7.5 Hz, 2H, CH arom), 7.24–7.27 (t, *J* = 7.5, 2H, CH arom), 5.30 (s, 1H, CHPh), 4.49 (d,  $J = 5$  Hz, 2H, CH<sub>2</sub>NH). d<sub>C</sub> (125 MHz, DMSO, d<sub>6</sub>) 35.21, 58.04, 87.51, 114.33, 121.95, 128.33, 128.87, 128.92, 129.79, 130.01, 130.11, 130.98, 131.46, 131.58, 135.14, 135.87, 137.21, 137.46, 147.61, 150.79, 152.80, 153.37, 163.46, 179.46.

**3‑Amino‑N‑benzyl‑1‑(4‑nitrophenyl)‑1H‑benzo[f] chromene‑2‑carboxamide (4d):** White solid, Yield 80%; mp: 227–230 °C. IR (KBr) ( $\nu_{\text{max}}$ , cm<sup>-1</sup>): 3405 (NH), 1680 (C=O). Anal. calcd. for  $C_{27}H_{21}N_3O_4$  (451.15): C, 71.83; H, 4.69; N, 9.31; % found: C, 71.68; H, 4.40; N, 9.78%.  $d_H$ (300 MHz, DMSO, d<sub>6</sub>) 8.42 (s, 2H, NH<sub>2</sub>), 7.86–7.87 (d,  $J = 8$  Hz, 2H, CH arom), 7.61–7.66 (m, 4H, CH arom and N<u>H</u>), 7.42–7.44 (d, *J* = 8 Hz, 3H, C<u>H</u> arom), 7.40 (m, 3H, CH arom), 7.30–7.33 (t, *J* = 7.5 Hz, 2H, CH arom), 7.24– 7.27 (t,  $J = 7.5$  Hz, 2H, CH arom), 5.32 (s, 1H, CHPh), 4.32 (d,  $J = 5$  Hz, 2H, C<sub>H<sub>2</sub>NH). d<sub>C</sub> (125 MHz, DMSO,</sub> d6) δ 33.88, 42.33, 88.33, 113.57, 113.63, 117.21, 117.74, 118.42, 119.30, 129.72, 129.78, 129.88, 131.03, 131.07, 133.78, 133.86, 133.99, 134.06, 134.74, 156.05, 156.09, 179.90.

**3‑Amino‑N‑benzyl‑1‑(3‑nitrophenyl)‑1H‑benzo[f] chromene‑2‑carboxamide (4e):** White solid, Yield 75%; mp: 148–149 °C. IR (KBr) ( $v_{\text{max}}$ , cm<sup>-1</sup>): 3420(NH), 1677 (C=O). Anal. calcd. for  $C_{27}H_{21}N_3O_4$  (451.15): C, C, 71.68; H, 4.40; N, 9.78%; % found: C, 71.38; H, 4.47; N, 9.80%.  $d_H$  (500 MHz, DMSO,  $d_6$ ) 8.89 (s, 2H, NH<sub>2</sub>), 7.52–7.57  $(m, 6 H, CH)$  arom), 7.48–7.50  $(m, 5H, CH)$  arom and NH $),$ 7.28–7.29 (t, *J* = 8 Hz, 2H, CH arom), 6.93–6.98 (m, 2H, CH arom),  $6.58-6.60$  (t,  $J = 7.5$  Hz, 1H, CH arom),  $5.37$  (s, 1H, CHPh), 4.57 (d,  $J = 5$  Hz, 2H, CH<sub>2</sub>NH). d<sub>C</sub> (75 MHz, DMSO, d<sub>6</sub>) 35.59, 57.22, 89.50, 112.89, 121.94, 123.38,

123.92, 125.09, 131.26, 131.63, 132.93, 132.99, 135.32, 136.04, 139.57, 147.95, 150.77, 152.77, 153.45, 163.30, 172.77.

**3‑Amino‑N‑benzyl‑1‑p‑tolyl‑1H‑benzo[f] chromene‑2‑carboxamide (4f):** White solid, Yield 60%; mp: 133–134 °C. IR (KBr) ( $v_{\text{max}}$ , cm<sup>-1</sup>): 3425(NH), 1685 (C=O). Anal. calcd. for  $C_{28}H_{24}N_2O_2$  (420.5): C, 79.98; H, 5.75; N, 6.66% found: C, 79.99; H, 5.85; N, 6.65%. d<sub>H</sub> (500 MHz, DMSO,  $d_6$ ) 8.20 (s, 2H, NH<sub>2</sub>), 7.91–7.94 (t,  $J = 7$  Hz, 2H, CH arom), 7.69–7.80 (m, 6H, CH arom), 7.55–7.67 (m, 4H, CH arom and NH), 7.40–7.44 (m, 1H, CH arom), 7.31–7.34 (t, *J* = 7.5 Hz, 1H, CH arom), 7.25–7.28 (t, *J* = 7.5 Hz, 1H, CH arom), 7.09–7.10 (d, *J* = 8 Hz, 1H, CH arom), 5.24 (d, *J* = 5.5 Hz, 1H, CHPh), 4.27 (d, *J* = 5 Hz, 2H, CH<sub>2</sub>NH), 2.20 (s, 3H).  $d_C$  (125 MHz, DMSO,  $d_6$ ) 31.75, 37.36, 86.79, 101.50, 111.07, 121.08, 123.22, 124.58, 124.98, 125.32, 125.80, 126.31, 128.31, 128.80, 128.99, 129.27, 130.58, 139.79, 140.37, 154.19, 155.30, 179.90.

**3‑Amino‑N‑benzyl‑1‑(4‑methoxyphenyl)‑1H‑benzo[f] chromene ‑2‑carboxamide (4 g):** White solid, Yield 68%; mp: 227–230 °C. IR (KBr) ( $v_{\text{max}}$ , cm<sup>-1</sup>): 3400 (NH), 1685 (C=O). Anal. calcd. for  $C_{28}H_{24}N_2O_3$  (436.5): C, 77.04; H, 5.54; N, 6.42% found: C, 77.15; H, 5.44; N, 6.53%.  $d_H$  $(500 \text{ MHz}, \text{DMSO}, \text{d}_6)$  8.50 (s, 2H, NH<sub>2</sub>), 8.03–8.04 (d,  $J = 8$  Hz, 2H, C<u>H</u> arom), 7.85–7.86 (d,  $J = 8$  Hz, 2H, C<u>H</u> arom), 7.61–7.69 (m, 2H, CH arom), 7.39–7.41 (d, *J* = 9 Hz, 3H, CH arom), 7.29–7.32 (t, *J* = 8 Hz, 1H, CH arom), 7.24– 7.27 (t, *J* = 7.5 Hz, 3H, CH arom), 7.14–7.17 (t, *J* = 7 Hz, 3H, CH arom), 5.39 (s, 1H, CHPh), 4.31 (d, *J* = 5.5 Hz, 2H, CH<sub>2</sub>NH), 3.45 (s, 3H, OCH<sub>3</sub>). d<sub>c</sub> (125 MHz, DMSO, d<sub>6</sub>) 34.62, 46.32, 56.79, 86.82, 111.68, 121.11, 123.24, 124.50, 124.62, 125.03, 126.63, 126.86, 128.03, 128.34, 128.45, 128.61, 129.00, 130.33, 130.92, 139.35, 139.55, 154.13, 156.19.

**3‑Amino‑N‑benzyl‑1‑(2, 4‑dichlorophenyl)‑1H‑benzo [f] chromene‑2‑carboxamide (4 h):** White solid, Yield 78%; mp: 146–148 °C. IR (KBr) ( $v_{\text{max}}$ , cm<sup>-1</sup>): 3425(NH), 1683 (C=O). Anal. calcd. for  $C_{27}H_{20}Cl_2N_2O_2$  (475.37): C, 68.22; H, 4.24; N, 5.89% found: C, 68.29; H, 4.13; N, 5.81%.  $d_H$  $(300 \text{ MHz}, \text{DMSO}, d_6)$  8.35 (s, 2H, NH<sub>2</sub>), 7.59–7.78 (d, *J* = 7.5 Hz, 4H, CH arom), 7.46–7.51 (t, *J* = 7 Hz, 4H, CH arom), 7.20–7.25 (t, J = 7 Hz, 4H, CH arom), 7.13– 7.19 (m, 3H, CH arom and NH), 5.33 (s, 1H, CHPh), 4.36 (d,  $J = 4.5$  Hz, 2H, CH<sub>2</sub>NH).  $d_C$  (125 MHz, DMSO,  $d_6$ ) δ 36.70, 44.00, 87.04, 103.04, 110.54, 113.04, 104.02, 116.77, 119.42, 122.54, 124.72, 130.25, 133.21, 134.85, 140.04, 145.51, 147.16, 152.20, 153.09, 158.20, 160.09, 164.90, 165.09.

<b>Strains</b>	4h (mm)	4i (mm)	4b dia, of clear zone dia, of clear zone dia, of clear zone dia, of clear zone dia, of clear zone (mm)	4c (mm)	<b>SXT</b> (mm)	Flu dia. of clear zone $(mm)$			
S. aureus PTCC 1112		7.5			25				
M. luteus PTCC 1110	7.5			9	30				
B. cereus PTCC 1015				9	20				
P. aeruginosa ATCC 27853			10	8.5	23				
E. coli PTCC1330			12		28				
$E.$ faecalis*				10	24				
K. pneumonia*			8	8	19				
C. albicans PTCC 5027	8			9.5		32			

<span id="page-4-0"></span>**Table 3** In vitro antimicrobial activity of the compounds, 50 mg/ml (IZ), trimethoprim/sulfamethoxazole (SXT) and fuconazole (Flu) were used as positive control

*SXT* trimethoprim/sulfamethoxazole, *FL* fuconazole

\* Clinical sample

<span id="page-4-1"></span>**Table 4** In vitro antimicrobial activity of the compounds, (MIC and MBC, mg/ml)



\* Clinical sample

**3‑Amino‑N‑benzyl‑1‑(3‑methoxyphenyl)‑1H‑benzo[f] chromene‑2‑carboxamide (4i):** Yellow solid, Yield 65%; mp: 101–103 °C. IR (KBr) ( $v_{\text{max}}$ , cm<sup>-1</sup>): 3420 (NH), 1680 (C=O). Anal. calcd. for  $C_{28}H_{24}N_2O_3$  (436.5): C, 77.04; H, 5.54; N, 6.42% found: C, 77.10; H, 5.49; N, 6.50%. d<sub>H</sub> (300 MHz, DMSO,  $d_6$ ) 8.32 (s, 2H, N<u>H</u><sub>2</sub>), 7.88–7.90 (d, *J* = 8, 2H, CH arom), 7.67–7.72 (t, *J* = 8 Hz, 2H, CH arom), 7.43–7.50 (m, 4H, CH arom), 7.35 (m, 3H, CH arom and NH), 6.85–6.89 (t,  $J = 8$  Hz, 3H, CH arom), 6.73–6.76 (d,  $J = 8$  Hz, 2H, CH arom), 5.20 (s, 1H, CHPh), 4.40 (d,  $J = 5$  Hz, 2H, CH<sub>2</sub>NH), 3.71(s, 3H, OCH<sub>3</sub>). d<sub>C</sub> (75 MHz, DMSO, d<sub>6</sub>) 36.78, 43.90, 57.09, 84.98, 102.98, 110.31, 113.04, 116.67, 119.02, 122.54, 124.78, 125.57, 130.15, 133.20, 134.84, 140.78, 145.57, 147.96, 152.37, 153.99, 158.27, 159.69, 165.90.

**3‑Amino‑N‑benzyl‑1‑(4‑hydroxyphenyl)‑1H‑benzo[f] chromene‑2‑carboxamide (4j):** White solid, Yield 70%; mp: 192–194 °C. IR (KBr) ( $v_{\text{max}}$ , cm<sup>-1</sup>): 3402, 1660 (C=O). Anal. calcd. for  $C_{27}H_{22}N_{2}O_{3}$  (422.48): C, 76.76; H, 5.25; N, 6.63% found: C, 76.87; H, 5.14; N, 6.60%. d<sub>H</sub> (300 MHz, DMSO,  $d_6$ ) 10.30 (s, 1H, OH), 8.38(s, 2H, NH<sub>2</sub>), 7.82–7.85

 $(d, J = 8 \text{ Hz}, 3\text{H}, \text{CH} \text{ arcm}),$  7.62–7.67  $(t, J = 7.8 \text{ Hz}, 2\text{H},$ CH arom), 7.38–7.45 (m, 3H, CH arom and NH), 7.29–7.30  $(d, J = 8$  Hz, 4H, CH arom), 6.99–7.01  $(d, J = 8$  Hz, 4H, CH arom), 5.28(s, 1H, CHPh), 4.28 (d,  $J = 5$  Hz, 2H, CH<sub>2</sub>NH). d<sub>C</sub> (75 MHz, DMSO, d<sub>6</sub>)  $\delta$  36.23, 55.12, 84.36, 112.75, 113.95, 115.27, 116.60, 119.35, 122.52, 124.75, 128.84, 132.91, 133.46, 135.45, 141.75, 145.84, 152.16, 153.14, 157.99, 158.42, 159.58, 160.54, 169.80.

#### **Determination of antimicrobial activity**

For studying of antimicrobial efect of IS16 and IS16 L were used *Escherichia coli* PTCC1330*, Staphylococcus aureus* PTCC 1112*, Pseudomonas aeroginosa* ATCC 27853 and *E. faecalis* which isolated from clinical samples. The antimicrobial efect of benzo[f]chromene derivatives on the microorganisms was assayed by agar well difusion method [[31](#page-6-29)]. Mueller–Hinton medium (Merck) and brain–heart infusion medium (Merck) were used as the test medium. Also, the minimum concentrations of the benzo[f]chromene to inhibit the microorganisms (MIC) were determined by

the micro-broth dilutions technique strictly following the National Committee for Clinical Laboratory Standards (NCCLS) recommendations [[32\]](#page-6-30). The inoculum was prepared using an overnight culture of each bacterium and yeast strains (18–24 h) adjusted to a turbidity equivalent to a 0.5 McFarland standard [[33\]](#page-6-31).

Petri plates containing 20-ml agar test plates were seeded with microorganism strains. Wells were cut and 50 μl of the benzo[f]chromene (50 mg/ml; DMSO was used as solvent) were added. The plates were then incubated at 37 °C for 24–48 h. The antibacterial activity was assayed by measuring the diameter of the inhibition zone formed around the well. DMSO as solvent was used as a negative control, whereas media with trimethoprim–sulfamethoxazole (SXT) were used as the positive controls. The experiments were performed in triplicate. Serial dilutions ranging from 100 mg/mL to 24.41 µg/mL were prepared in medium.

## **Results and discussion**

Our initial experiments were focused on a one-pot, threecomponent reaction of 2-naphthol, benzaldehyde and N-benzyl-2-cyanoacetamide using diferent catalysts under various conditions, and the results are listed in Table [1](#page-2-0). In the absence of any catalysts, the reaction did not proceed even after prolonged reaction time and no desired product was formed (Table [1,](#page-2-0) entry 1). But, in the presence of basic catalysts such as KOH,  $Et<sub>3</sub>N$  or pyridine the desired product was obtained in 10–55% yield (Table [1,](#page-2-0) entries 2–9). In addition, the above mentioned model reaction was conducted in the presence of piperidine under various conditions (Table [1,](#page-2-0) entries 10–13).

The best result was obtained using piperidine at refux condition (Table [1](#page-2-0), entry 12). Therefore, piperidine was chosen as a suitable catalyst.

We found that decreasing the temperature to 25  $\degree$ C (Table [1](#page-2-0), entry 13) led to the lower yield and longer reaction time.

In order to generalize the optimum conditions, diferent derivatives of 3-amino-N-benzyl-1-aryl-1H-benzo[f] chromene-2-carboxamide (4a–j) were prepared from the one-pot reaction mixture of N-benzyl-2-cyanoacetamide (1), appropriate aldehyde (2a–j) and 2-naphthol (3) in the presence of a catalytic amount of piperidine in ethanol under refux conditions. The results have been summarized in Table [2.](#page-2-1)

The effect of antibacterial activity of six out of ten of these new benzo[f]chromenes, namely 4a, 4b, 4c, 4d, 4h and 4i, against *E. coli* PTCC1330*, S. aureus* PTCC 1112*, Pseudomonas aeroginosa* ATCC 27853 and *E. faecalis* (which were isolated from clinical samples) was assessed by evaluating the presence of inhibition zone (IZ) and MIC and MBC values. Results showed that 4b, 4c, 4h and 4i exhibited antibacterial efects. Compounds 4h and 4i were efective on *S. aureus* PTCC 1112, whereas 4b was efective on *E. coli* PTCC1330. Also 4c could inhibit growth of *E. faecalis*. According to the results, it seems that the antibacterial efect of the ligands is more on Gram-positive bacteria (Table [3](#page-4-0)).

The MIC and MBC values for the benzo[f]chromenes (4a, 4b, 4h and 4i) were in the range of 100 mg/ mL–24.41 µg/mL. The results of our investigation showed that these compounds had bacterostatic and bactericide effects (Table [4](#page-4-1)).

<span id="page-5-0"></span>**Scheme 2** Proposed mechanism for synthesis of 3-amino-N-benzyl-1-aryl-1H-benzo[f] chromene-2-carboxamides



#### **The proposed mechanism**

A plausible mechanism for synthesis of 3-amino-N-benzyl-1-aryl-1H-benzo[f]chromene-2-carboxamide is shown in Scheme [2](#page-5-0).

The reaction is supposed to proceed in a stepwise fashion. The initial reaction is conducted via a Knoevenagel condensation between compound 1 and aromatic aldehyde 2 to form the intermediate 5 in the presence of piperidine, which suffers immediate addition of compound 3 to the C=C bond of 5. The concerted cyclocondensation of hydroxy and CN of the intermediate 6 was performed to furnish the corresponding products (4a–j).

#### **Conclusion**

In conclusion, new 3-amino-N-benzyl-1-aryl-1H-benzo[f] chromene-2-carboxamide derivatives from the reaction of *N*-*benzyl*-*2*-*cyanoacetamide* with aromatic aldehydes and 2-naphthol in the presence of piperidine as a base have been synthesized. Among these synthesized compounds, four compounds 4b, 4c, 4h and 4i exhibited potential antibacterial activity which may guarantee their future applications in a moderate antibiotic therapy.

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