ORIGINAL ARTICLE

# Effect of potassium chloride and calcium chloride induced stress on in vitro cultures of Bacopa monnieri (L.) Pennell and accumulation of medicinally important bacoside A

M. L. Ahire · S. Laxmi · P. R. Walunj · P. B. Kavi Kishor · T. D. Nikam

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Abstract Growth, osmotic adjustment, antioxidant enzyme defense and principle medicinal component bacoside A was studied in in vitro raised shoots of Bacopa monnieri under different concentrations of KCl and CaCl<sub>2</sub>  $(0, 50, 100, 150)$ or 200 mM). Significant reduction was observed in shoot number per culture; shoot length, fresh weight, dry weight and tissue water content (TWC) when shoots were exposed to increasing KCl and CaCl<sub>2</sub> concentrations  $(50-200 \text{ mM})$ as compared to control. Minimum damage to the membrane as assessed by malondialdehyde (MDA) content was noticed in control in contrast to sharp increase in KCl and CaCl<sub>2</sub> stressed shoots. Higher amounts of free proline, glycine betaine and total soluble sugars (TSS) accumulated in KCl and CaCl<sub>2</sub> exposed shoots compared to the controls. Among different concentrations of KCl and  $CaCl<sub>2</sub>$ , increasing concentration of  $CaCl<sub>2</sub>$  showed more increase in osmolyte accumulation.  $Na<sup>+</sup>$  content decreased with increasing concentrations of KCl and CaCl<sub>2</sub>. Accumulation of  $K^+$ increased significantly in KCl (50–100 mM) stressed shoots as compared to control, while it decreased in  $CaCl<sub>2</sub>$  treated shoots indicating that it prevents the uptake of  $K^+$  ions.  $Ca^{2+}$ accumulation significantly increased with increasing concentrations of  $CaCl<sub>2</sub>$  up to 150 mM but decreased at higher concentrations. Shoots treated with KCl and CaCl<sub>2</sub> (0– 100 mM) showed higher antioxidant enzyme (SOD, CAT, APX and GPX) activities but KCl suppressed the activities

M. L. Ahire  $\cdot$  S. Laxmi  $\cdot$  P. R. Walunj  $\cdot$  T. D. Nikam ( $\boxtimes$ ) Department of Botany, University of Pune, Pune 411 007, India e-mail: tdnikam@unipune.ac.in

M. L. Ahire

Department of Botany, Yashavantrao Chavan Institute of Science, Satara 415 001, India

P. B. K. Kishor Department of Genetics, Osmania University, Hyderabad 500 007, India

at higher concentrations. Accumulation of bacoside A was enhanced with an increase in KCl and CaCl<sub>2</sub> concentration up to 100 mM. It appears from the data that accumulation of osmolytes, and elevated activities of antioxidant enzymes play an important role in osmotic adjustment in shoot cultures of Bacopa and the two salts tested have a positive effect on bacoside accumulation.

Keywords Antioxidant enzymes . Bacopa monnieri . Bacoside A . In vitro growth responses . Osmolyte accumulation . Salinity . Shoot cultures

# Abbreviations

- APX Ascorbate peroxidase BA 6-Benzyladenine CAT Catalase GPX Guaiacol peroxidase MDA Malondialdehyde MS Murashige and Skoog NBT Nitroblue tetrazolium chloride SOD Superoxide dismutase TSS Total soluble sugars
- TWC Tissue water content

# Introduction

The stress factors such as salt, drought, low temperature, flooding, heat, oxidative stress, heavy metal toxicity and pathogenic stress adversely affect the plant growth and productivity. These stresses affect almost every aspects of plants physiology which leads to a series of morphological, physiological, biochemical and molecular changes. It is estimated that every year about 6 million arable lands are lost from agriculture use. On the other hand existing agricultural land fails to fulfill the food, feed, fodder and

industrial raw material requirement of ever growing population. The new land is becoming unavailable for the cultivation of crops. Rapid increase in population pressure are leading to greater utilization of remaining meager fresh water supplies for drinking, living, and even less for agriculture. In this complex scenario, it has become crucial to consider development of non-conventional technologies which might more effectively utilize degraded, marginal and saline lands for agriculture by using the 'so-called' poor water. Despite advances in increasing plant productivity and resistance to a number of pests and diseases, improving salt tolerance in crop plants remains elusive. Furthermore, limited success in increasing the yield stability of crop plants grown in saline soil might be due to lack of understanding about salinity and other abiotic stresses that affect the cell division, differentiation and expansion which have a sustainable impact on plant growth and development. Salinity is one of the major abiotic stresses affecting more than 800 million hectares of arable land throughout the world (Munns and Tester [2008\)](#page-12-0). Depletion of cellular water content due to abiotic stresses such as drought, high soil salinity and temperature extremes is responsible for the greatest agricultural losses (Jaleel et al. [2007](#page-11-0); Lokhande et al. [2011](#page-11-0)). Upon exposure to stresses many plants show changes in chla/chlb ratio (Lin and Kao [1998;](#page-11-0) Lutts et al. [1996](#page-11-0)); membrane permeability and/or increase in lipid peroxidation (Dhindsa and Matowe [1981\)](#page-11-0). There are many cellular mechanisms by which plants ameliorate the effects of environmental stresses such as accumulation of compatible osmolytes, most commonly amino acids and quaternary ammonium compounds (Bohnert and Jensen [1996;](#page-11-0) Kavi Kishor et al. [2005](#page-11-0)). The increase in cellular osmolarity resulted due to accumulation of osmolytes is accompanied by the influx of water into, or reduced efflux from cells which helps to provide the neces-sary turgor for cell expansion (Cuin and Shabala [2005](#page-11-0); Meloni and Martínez [2009](#page-11-0)). Both  $K^+$  and  $Ca^{2+}$  are involved in photosynthesis and in protein synthesis regulation and their depletion may lead to serious inhibition of these processes (Santos et al. [2001\)](#page-12-0). Similarly, if they are accumulated in higher concentrations, they may inhibit photosynthesis and other vital activities. Maintenance of membrane integrity and the selective uptake of essential ions like  $K^+$  and  $Ca^{2+}$  are related with the acquisition of salt tolerance (Salama et al. [1994\)](#page-12-0).

Plants prevent the damaging effects of free radicals by evolving several non-enzymatic and enzymatic mechanisms that efficiently scavenge ROS (Irigoyen et al. [1992;](#page-11-0) Shalata and Tal [1998\)](#page-12-0). Enzymatic protection is partly performed by SOD that eliminates superoxide radicals and by catalases and peroxidases that degrade  $H_2O_2$  which influence the levels of lipid peroxidation (Santos et al. [2001](#page-12-0)). Recent studies have demonstrated that activities of these antioxidant enzymes and levels of antioxidant molecules increase and are correlatable to various environmental stresses (Hernández et al. [2000](#page-11-0); Sekmen et al. [2007\)](#page-12-0). Such a correlation was observed between NaCl induced salt stress tolerance and antioxidative responses in different plant systems (Vaidyanathan et al. [2003](#page-12-0); Benavente et al. [2004;](#page-11-0) Jogeswar et al. [2006](#page-11-0); Lokhande et al. [2011](#page-11-0)). In recent years, medicinal and aromatic plants or products are on an increasing demand in agro alimentary, perfumes, pharmaceutical and natural cosmetic industries (Baatour et al. [2010](#page-10-0); Tarchoune et al. [2012\)](#page-12-0). As in cultivated species, growth and yield of medicinal and aromatic plants are affected by environmental constraints such as salinity and drought (Baatour et al. [2010\)](#page-10-0). However, little information is available about the physiological basis and effects on secondary metabolite accumulation under different abiotic stress conditions in medicinal plants.

In Indian medicine, certain herbs have been used traditionally as brain or nerve tonics. Bacopa monnieri (L.) Pennell (Scrophulariaceae) is commonly known as 'Brahmi' or 'Nir Brahmi'. It is one of the sources of the *medhya rasayan* drugs (that counteracts stress and improves intelligence and memory) of Ayurveda. Recently, B. monnieri was placed second in a priority list of the most important Indian medicinal plants evaluated on the basis of their medicinal importance, commercial value and potential for further research and development. According to the National Medicinal Plant Board (NMPB), Government of India, annual demand of Bacopa is increasing day by day due to the popularity of the Bacopa based drugs.

Efforts were made earlier to induce the stress in vitro using NaCl (Ali et al. [1999;](#page-10-0) Debnath [2008\)](#page-11-0), mannitol (Debnath [2008\)](#page-11-0), heavy metals like copper (Ali et al. [1998a\)](#page-10-0), cadmium, zinc (Ali et al. [2000\)](#page-10-0) and aluminium (Ali et al. [1998b](#page-10-0)). But these authors did not identify the stress tolerance mechanisms, antioxidative enzyme responses and accumulation of bacoside in Bacopa monnieri. While some plants accumulate only proline and sugars under stress, others accumulate glycine betaine in addition to proline. Our results indicate that Bacopa accumulates glycine betaine also. Potassium  $(K^+)$  plays an important role in the regulation of osmotic potential of plant cells and calcium  $(Ca^{2+})$  is the structural element essential in the synthesis of cell wall. The calcium is said to antagonize the uptake of excess  $K^+$ . Many soils have an excess of certain elements, particularly sodium (Na<sup>+</sup>),  $K^+$  or Ca<sup>2+</sup> (Taiz and Zeiger [2010](#page-12-0)) which results in severe stress and subsequently decline in productivity. Detailed understanding of the basic mechanisms involved in plant salt tolerance is an important prerequisite to improve the performance of plants in saline soils (Binzel and Reuveni [1994](#page-11-0)). Since plants are affected not only by  $Na^+$  salts but also by  $K^+$ salts, in the present study, the effects of KCl and  $CaCl<sub>2</sub>$ stress on growth of shoot cultures and nutritional imbalance were studied in Bacopa monnieri. The effect of KCl and

CaCl<sub>2</sub> stress on lipid peroxidation, osmolyte accumulation, antioxidant enzyme activities (SOD, CAT, APX, GPX) and accumulation of medicinally active component bacoside A in in vitro shoot cultures were carried out.

## Materials and methods

Plant material, establishment of in vitro cultures and salt stress treatment

The shoot cultures of *B. monnieri* were established using nodal explants and were maintained in liquid Murashige and Skoog's (MS [1962\)](#page-12-0) medium containing 1 mg/l 6 benzyladenine (BA) (Ahire et al. [2012a\)](#page-10-0). For analysis of in vitro stress studies, shoots of Bacopa were subjected to the treatment of different concentrations of KCl and CaCl<sub>2</sub> separately in liquid MS medium supplemented with 1 mg/l BA. About 4 shoots (5 nodes each/bottle) were inoculated in glass bottles containing 50 ml of MS liquid medium fortified with 1 mg/l BA and respective KCl and  $CaCl<sub>2</sub>$ (0–200 mM) concentrations separately. Cultures were incubated under controlled conditions at  $25\pm2$  °C temperature,  $60\pm10$  % relative humidity and 8 h photoperiod (PFD) 40  $\mu$ mol m<sup>-2</sup> s<sup>-1</sup>) provided by white fluorescent tubes (Philips, India).

## Determination of mineral nutrients

Shoots were washed with distilled water and excess water was soaked on blotting paper followed by drying at 60 °C for 48 h in an oven. Dried shoots were ground to powder and 200 mg powder from each treatment was soaked in 10 ml of 35 %  $(v/v)$  HNO<sub>3</sub> (Qualigens, Mumbai, India) overnight at room temperature followed by acid digestion at 100 °C till the acid was evaporated and finally the residue was dissolved in 30 ml of distilled water. Samples were filtered using Whatman filter paper No. 1 (Whatman International Ltd., Maidstone, England). Cations such as  $Na<sup>+</sup>$ ,  $K<sup>+</sup>$  and  $Ca<sup>2+</sup>$  were measured by atomic absorption spectrophotometer (AA-7000, Labindia Analytical Instruments Pvt. Ltd., Mumbai, India). Standard solutions of Na<sup>+</sup>, K<sup>+</sup> and Ca<sup>2+</sup> were purchased from Qualigens, Mumbai, India and were used for quantification of ion content.

#### Growth analysis and tissue water status

Growth analysis was carried out after 28 days of incubation. Morphological observations like number of shoots per culture and shoot length (cm) were recorded. Tissue water status was determined by measuring fresh (FW) and dry weights (DW). Shoots were harvested and excess medium

was soaked on blotting paper and FW of shoots was measured. Shoots were kept in an oven at 60 °C separately till constant weight was obtained and noted as DW. FW and DW of the shoots obtained from each treatment were used to determine the water status, which is expressed as percentage tissue water content (TWC %) calculated using the equation as described by Lokhande et al. ([2011](#page-11-0)).

 $TWC\% = [(FW - DW)/FW] \times 100$ 

Determination of chlorophyll, lipid peroxidation and osmolyte content

Chla, Chlb and total chlorophyll contents in the control and treated shoots were estimated as per the standard method described by Arnon [\(1949\)](#page-10-0). The level of lipid peroxidation was measured in terms of malondialdehyde (MDA) content as described by Heath and Packer ([1968](#page-11-0)), with some modifications as described by Lokhande et al. ([2011\)](#page-11-0). Glycine betaine was estimated by the periodide colorimetric method according to Grieve and Grattan ([1983\)](#page-11-0) as described by Lokhande et al. [\(2010](#page-11-0)) and proline content was estimated as per Bates et al. ([1973\)](#page-10-0).

Determination of antioxidant enzyme activities

Treated and control fresh samples (500 mg) were homogenized in 5 ml of ice cold 50 mM sodium phosphate buffer (pH 7.0) containing 0.1 mM EDTA and 1  $\%$  (w/v) polyvinylpyrrolidone with chilled mortar and pestle. The homogenate was filtered with single layered cheese cloth and centrifuged at 10,000 rpm for 20 min at 4 °C. Appropriate aliquot/dilution of the supernatant was used as a crude enzyme(s) for determination of antioxidant enzyme activities. Soluble protein content in the enzyme extract was determined according to Lowry et al. [\(1951](#page-11-0)) using bovine serum albumin as standard. Total superoxide dismutase (SOD) enzyme (EC 1.15.1.1) activity was assayed according to Becana et al. ([1986\)](#page-10-0) by inhibition of the photochemical reduction of nitroblue tetrazolium (NBT). The reaction mixture (1 ml) containing 50 mM phosphate buffer (pH 7.0) and 0.1 mM EDTA to which an oxygen-generating system containing 14.3 mM methionine, 82.5 μM NBT, and 2.2 μM riboflavin, prepared freshly in situ, was added. Reaction was initiated by adding 25 μl of crude enzyme. The entire system was kept 30 cm below the light source (six 15 W fluorescent tube light) for 30 min. Reaction was stopped by switching off the tube light. For light blank, all the reactants without enzyme extract was incubated in light as for the samples, whereas all the reactants along with 25 μl enzyme extract were incubated in dark for dark blank. Reduction in NBT was measured by monitoring the change in absorbance at 560 nm. SOD activity is expressed as μKat of SOD  $mg^{-1}$  protein. Catalase (CAT) enzyme (EC 1.11.1.6) activity was measured by following the decomposition of hydrogen peroxide  $(H_2O_2)$  as described by Cakmak and Marschner [\(1992](#page-11-0)) with minor modifications. Activity was measured in a reaction mixture (1 ml) containing 50 mM phosphate buffer (pH 7.0) and 300 mM  $H_2O_2$ . The reaction was initiated by adding 50 μl enzyme extract and the activity was determined as a result of  $H_2O_2$ decomposition by monitoring the decrease in absorbance at 240 nm ( $\varepsilon$ =36 mM<sup>-1</sup> cm<sup>-1</sup>) for 2 min at an interval of 15 s. The slope of readings between the time interval considered as  $\Delta A$  and enzyme activity is expressed as  $\mu$ Kat of CAT  $mg^{-1}$  protein. Ascorbate peroxidase (APX) enzyme (EC 1.11.1.11) activity was determined according to Nakano and Asada ([1981](#page-12-0)). The reaction mixture (1 ml) contained 50 mM phosphate buffer (pH 7.0), 0.5 mM ascorbate and 0.1 mM  $H_2O_2$ . Reaction was started by adding 50 μl of crude enzyme. Ascorbate oxidation was monitored for 1 min by measuring the decrease in absorbance at 290 nm at every 15 s ( $\varepsilon$ = 2.8 mM<sup>-1</sup> cm<sup>-1</sup>). Enzyme activity is expressed as  $\mu$ Kat of APX mg−<sup>1</sup> protein. Guaiacol peroxidase (GPX) enzyme (EC 1.11.1.7) activity was assayed according to Hemeda and Klein ([1990\)](#page-11-0). The reaction mixture (1 ml) contained 50 mM phosphate buffer (pH 7.0), guaiacol, 200 mM  $H_2O_2$  and 10  $\mu$ l enzyme extract. The reaction was started by adding 200 mM  $H_2O_2$ . The increase in absorbance due to oxidation of guaiacol ( $\varepsilon$ =26.6 mM<sup>-1</sup> cm<sup>-1</sup>) was monitored at 470 nm. Enzyme activity is expressed as  $\mu$ Kat mg<sup>-1</sup> protein.

## Estimation of bacoside A

The dried samples (control and treated) were ground to fine powder and used for estimation. Bacoside-A was extracted as per Watoo et al. ([2007](#page-12-0)) with some modifications as described by Parale et al. ([2010\)](#page-12-0). One g of powder was soaked in 10 ml of distilled water for 2 h and the mixture was squeezed. Plant residue was extracted thrice in 20 ml of 95 % (v/v) ethanol for 24 h on a rotary shaker at 80 rpm at 25 °C. The pooled filtrate was evaporated to dry under vacuum. Dry residue was dissolved in methanol at a concentration of 1 mg/ml and used for HPTLC analysis. The standard bacoside-A was procured from Natural Remedies, Bangalore, India and dissolved in HPLC grade methanol at the concentration of 1 mg/ml and used to plot the standard curve. HPTLC was performed on  $20 \times 10$  cm aluminum foil plates coated with 200 μm layer of silica gel 60 F254 (E. Merck, Germany). Plates were prewashed with methanol and dried in an oven for 5 min at 105 °C temperature. The standard and samples were loaded with Linomat 5 semiautomatic applicator (CAMAG, Switzerland) fitted with 100 μl syringe as bands 8 mm in width. The delivery speed of the syringe was 90 nl/s. HPTLC plates were developed in a twin-trough chamber (20×10×4 cm; CAMAG, Switzerland). Chamber saturation was carried out using  $20 \times 10$  cm Whatman filter paper for 20 min. The plates were developed in toluene : ethanol : methanol : glacial acetic acid (4:3:3:1 v/v), until 80 mm from the lower edge of the plate. After development, the plates were dried using hair dryer. The dried developed plates were immersed for 2 s in 5 % methanolic sulphuric acid. After immersion, the plates were dried in an oven for 5 min at 100 °C. Documentation of the plates was carried out using CAMAG TLC Scanner III system at 500 nm with winCATS software 1.4.3.

#### Data analysis

A completely randomized design (CRD) was used in all experiments. The experiments were repeated at least thrice. The data were subjected to analysis of variance (ANOVA) followed by Duncan's multiple range tests (DMRT) at  $P \le 5$  %.

## **Results**

Effect of KCl and CaCl<sub>2</sub> stress on growth

In the present investigation, shoots of Brahmi were exposed to different salt (KCl and CaCl<sub>2</sub>:  $0$ ,  $50$ ,  $100$ ,  $150$  and 200 mM) concentrations. Significant reduction in number of shoots per culture was observed with increasing salt concentrations. Maximum number of shoots  $(138.0 \pm 5.7)$ was observed in control after 28 days of incubation and least number of shoots per culture  $(15.9 \pm 1.8)$  was recorded at 200 mM KCl stress. Similarly, shoot length, shoot fresh and dry weights decreased significantly with increasing concentrations of KCl and  $CaCl<sub>2</sub>$  (Table [1\)](#page-4-0). Tissue water content (TWC %) decreased with increasing KCl and  $CaCl<sub>2</sub>$ concentrations followed by fresh and dry weights of tissues (Fig. [1;](#page-4-0) Table [1\)](#page-4-0). Among KCl and  $CaCl<sub>2</sub>$ , higher concentration of KCl showed more pronounced effect on the number of shoots per culture.

Influence of KCl and  $CaCl<sub>2</sub>$  stress on chlorophyll content and lipid peroxidation

Decrease in chla, chlb and total chl was observed with increasing KCl and  $CaCl<sub>2</sub>$  $CaCl<sub>2</sub>$  $CaCl<sub>2</sub>$  concentrations (Table 2). Maximum chla (9.16±0.68 mg g<sup>-1</sup> FW), chlb (3.96± 0.31 mg g<sup>-1</sup> FW) and total chl (13.61±0.97 mg g<sup>-1</sup> FW) were recorded in control shoots. A sharp decrease was noticed with increasing KCl and  $CaCl<sub>2</sub>$  concentrations. Considerable reduction in chla (3.63±0.5 mg g<sup>-1</sup> FW), chl*b*  $(1.36\pm0.2 \text{ mg g}^{-1} \text{ FW})$  and total chl  $(4.99\pm0.7 \text{ mg g}^{-1} \text{ FW})$ was noticed at 200 mM CaCl<sub>2</sub> concentration. Progressive

Concentration (mM)	No. of shoots per culture		Shoot length (cm)		Fresh weight $(g)$		Dry weight $(g)$	
	<b>KCl</b>	CaCl <sub>2</sub>	KCl	CaCl <sub>2</sub>	<b>KCl</b>	CaCl <sub>2</sub>	<b>KCl</b>	CaCl <sub>2</sub>
$\overline{0}$	$138.0 \pm 5.7^{\rm a}$	$138.0 \pm 5.7^{\rm a}$	$13.0 \pm 1.4^a$	$13.0 \pm 1.4^{\rm a}$	$13.0 \pm 1.4^a$	$13.0 \pm 1.4^a$	$0.75 \pm 0.03^{\text{a}}$	$0.75 \pm 0.03^{\text{a}}$
50	$111.4 \pm 7.2^b$	$117.2 \pm 6.4^b$	$02.8 \pm 0.1^{\rm b}$	$03.2 \pm 0.1^{\rm b}$	$03.7 \pm 0.6^b$	$04.5 \pm 0.2^b$	$0.48 \pm 0.03^b$	$0.54 \pm 0.04^b$
100	$059.3 \pm 6.3$ <sup>c</sup>	$065.7 \pm 5.6^{\circ}$	$01.9 \pm 0.1$ <sup>c</sup>	$02.3 \pm 0.2$ <sup>c</sup>	$02.2 \pm 0.2$ <sup>c</sup>	$03.6 \pm 0.4^{\text{cd}}$	$0.33 \pm 0.02^{\circ}$	$0.41 \pm 0.03$ <sup>c</sup>
150	$028.1 \pm 3.0^{\rm d}$	$0.36 \pm 4.3^d$	$01.5 \pm 0.1$ <sup>d</sup>	$01.7 \pm 0.1$ <sup>d</sup>	$01.7 \pm 0.2$ <sup>d</sup>	$03.2 \pm 0.1$ <sup>d</sup>	$0.28 \pm 0.01$ <sup>d</sup>	$0.36 \pm 0.02^d$
200	$015.9 \pm 1.8$ <sup>e</sup>	$019.9 \pm 2.1$ <sup>e</sup>	$0.9 \pm 0.1$ <sup>e</sup>	$00.9 \pm 0.1$ <sup>e</sup>	$01.5 \pm 0.1^e$	$01.9 \pm 0.1$ <sup>e</sup>	$0.20 \pm 0.01^e$	$0.24 \pm 0.01^e$

<span id="page-4-0"></span>Table 1 Effect of different concentrations of KCl and CaCl<sub>2</sub> on shoot number, shoot length, fresh and dry weights in *Bacopa monnieri* 

Data presented in the table are mean  $\pm$  SE scored at 28 days after inoculation from 7 replicates. Mean followed by same letters within columns are not significantly different at P≤0.05 level by DMRT

decrease in chlorophyll content was observed with increasing concentrations of KCl and  $CaCl<sub>2</sub>$  more so with  $CaCl<sub>2</sub>$ . Lipid peroxidation in terms of MDA content was enhanced significantly with increasing KCl and  $CaCl<sub>2</sub>$  concentration from 0 to 200 mM (Fig. [2\)](#page-5-0). In shoots treated with KCl, MDA content increased up to 150 mM level but decreased thereafter (Fig. [2](#page-5-0)). About 1.51 times more MDA content was observed in the shoots treated with 150 mM KCl (Fig. [2](#page-5-0)). At lower CaCl<sub>2</sub> concentrations (50 and 100 mM), MDA content was not high. Concentration of MDA increased at higher CaCl<sub>2</sub> concentrations (150 and 200 mM). Higher lipid peroxidation (2.16 times more MDA content over control) was observed at [2](#page-5-0)00 mM  $CaCl<sub>2</sub>$  (Fig. 2) indicating more damage of membrane at higher KCl and CaCl<sub>2</sub> concentrations due to lipid peroxidation.

## Osmolyte accumulation in response to salt stress

A progressive increase in free proline content was recorded in KCl treated shoots as compared to control (Fig. [3](#page-6-0)). Elevation in proline content was 3.77 times higher at  $200 \text{ mM KCl concentration (Fig. 3).}$  $200 \text{ mM KCl concentration (Fig. 3).}$  $200 \text{ mM KCl concentration (Fig. 3).}$  In CaCl<sub>2</sub> treated shoots, proline content increased up to 150 mM (6.21 times more over control) but decreased at 200 mM concentration. Accumulation of glycine betaine in the shoots of Brahmi was low in control, whereas a steep increase was recorded with an increase in  $50-200$  mM KCl and  $CaCl<sub>2</sub>$  stresses (Fig. [4\)](#page-6-0). About 3.46 and 4.04 times more glycine betaine content was observed over control in shoots treated with 200 mM KCl and CaCl<sub>2</sub> respectively. Treatment of KCl and CaCl<sub>2</sub> resulted in higher accumulation of total soluble sugar (TSS) in shoots of Bacopa as compared to the controls (Fig. [5](#page-7-0)). Increased TSS content up to 150 mM levels of KCl was noticed though it declined slightly thereafter. Similarly,  $CaCl<sub>2</sub>$  increased the TSS and 3.64 times more over control was observed at 200 mM (Fig. [5](#page-7-0)). Among the different concentrations of KCl and  $CaCl<sub>2</sub>$ , increasing concentration of  $CaCl<sub>2</sub>$  showed more increase in osmolyte accumulation.

## Effect of KCl and  $CaCl<sub>2</sub>$  on the accumulation of ions

Salt stress  $(0-200 \text{ mM KCl}$  and  $CaCl<sub>2</sub>$ ) promoted the accumulation of cations in the shoots of Bacopa in contrast to the control (Table [3](#page-7-0)). Decrease in  $Na<sup>+</sup>$  concentration was ob-served with increasing KCl and CaCl<sub>2</sub> levels (Table [3](#page-7-0)).

concentrations of KCl and CaCl<sub>2</sub> on percentage tissue water content in shoots of Bacopa monnieri. Each value represents mean of three replications and vertical bars indicate SE. The bars with different letters on the same color columns are significantly different at  $p \leq 0.05$ 



Concentration (mM)	Chla (mg $g^{-1}$ FW)		Chlb (mg $g^{-1}$ FW)		Total Chl (mg $g^{-1}$ FW)	
	<b>KCl</b>	CaCl <sub>2</sub>	<b>KCl</b>	CaCl <sub>2</sub>	<b>KCl</b>	CaCl <sub>2</sub>
$\overline{0}$	$9.16 \pm 0.7^a$	$9.16 \pm 0.7^a$	$3.96 \pm 0.3^a$	$3.96 \pm 0.3^{\text{a}}$	$13.61 \pm 1.0^a$	$13.61 \pm 1.0^a$
50	$5.99 \pm 0.2^b$	$5.57 \pm 1.0^b$	$2.70 \pm 0.2^b$	$2.35 \pm 0.4^b$	$8.69 \pm 0.3^{b}$	$7.91 \pm 1.4^b$
100	$5.63 \pm 0.4^{\rm bc}$	$5.25 \pm 0.6^{\circ}$	$2.71 \pm 0.1^{\rm b}$	$1.89 \pm 0.1$ <sup>c</sup>	$8.34 \pm 0.5$ <sup>c</sup>	$7.14 \pm 0.5^b$
150	$5.36 \pm 0.3$ <sup>c</sup>	$4.97 \pm 0.4$ <sup>d</sup>	$2.58 \pm 0.1$ <sup>c</sup>	$1.97 \pm 0.2$ <sup>c</sup>	$8.21 \pm 0.4^{\circ}$	$6.94 \pm 0.8$ <sup>c</sup>
200	$4.53 \pm 0.2^d$	$3.63 \pm 0.5^e$	$2.40\pm0.1$ <sup>d</sup>	$1.36 \pm 0.2^d$	$6.93 \pm 0.2^d$	$4.99 \pm 0.7$ <sup>d</sup>

<span id="page-5-0"></span>Table 2 Effect of different concentrations of KCl and CaCl<sub>2</sub> on chlorophylla, chlorophyllb and total chlorophyll contents in *Bacopa monnieri*. Values were taken after 28 days of culture

Each value represents mean ± SE of three replications. Means followed by same letters within columns are not significantly different at P≤0.05 level by Duncan's multiple range test. DMRT was applied to each variety separately

Contrary to this,  $K^+$  concentration was enhanced significantly in KCl-stressed shoots as compared to the control (Table [3\)](#page-7-0), while the opposite effect was seen in  $CaCl<sub>2</sub>$ treated shoots. Accumulation of  $K^+$  increased by 1.72 times at 100 mM KCl level and decreased thereby.  $Ca^{2+}$  accumulation slightly increased in the shoots when they were exposed to 0–100 mM KCl but declined at higher concentrations. However,  $Ca^{2+}$  accumulated significantly when exposed to different concentrations (50–150 mM) of  $CaCl<sub>2</sub>$  (Table [3\)](#page-7-0), maximum being 7.85 times at 150 mM  $CaCl<sub>2</sub>$  (Table [3](#page-7-0)).

Effect of KCl and  $CaCl<sub>2</sub>$  stress on antioxidant enzyme activities

KCl and CaCl<sub>2</sub> stresses (0–100 mM) stimulated the antioxidant enzyme (SOD, CAT, APX and GPX) activities in shoots of Bacopa. Increasing concentration of KCl and  $CaCl<sub>2</sub>$  suppressed the activities (Fig. [6\)](#page-8-0). Both KCl and  $CaCl<sub>2</sub>$  (50–100 mM), increased the SOD activity (Fig. [6a](#page-8-0)). SOD specific activity was 1.35-folds higher at 150 mM KCl while it displayed almost the same activity at 100 mM

stress on lipid peroxidation (as measured by MDA content) in shoots of Bacopa monnieri. Each value represents mean of three replications and vertical bars indicate SE. The bars with different letters on the same color columns are significantly different at  $p<0.05$ 

CaCl<sub>2</sub>. Activity of CAT was also enhanced by 1.2-folds at 100 mM KCl and 1.4-folds at 100 mM CaCl<sub>2</sub> concentration (Fig. [6b\)](#page-8-0). The trend declined at higher concentrations of KCl and  $CaCl<sub>2</sub>$  (150 and 200 mM). Control shoots (without KCl and  $CaCl<sub>2</sub>$  in the medium) displayed lowest APX activity in contrast to treated tissues. While shoots cultured on medium containing 100 mM KCl showed 2.63-folds higher APX activity than controls (Fig. [6c\)](#page-8-0), activity was 1.94-folds higher in 100 mM CaCl<sub>2</sub> treated tissues. A steep increase in GPX activity was also recorded as the KCl and  $CaCl<sub>2</sub>$  concentrations increased up to 100 mM (Fig. [6d\)](#page-8-0). About 1.65-folds and 2.8-folds higher GPX activity was recorded in shoots treated with 100 mM KCl and CaCl<sub>2</sub> respectively (Fig. [6d\)](#page-8-0).

#### Bacoside A accumulation in response to  $KCl$  and  $CaCl<sub>2</sub>$  stress

In the present investigation, effect of KCl and  $CaCl<sub>2</sub>$  stress on the accumulation of the principle medicinal compound bacoside A was studied. Content of bacoside A was enhanced with an increase in KCl and  $CaCl<sub>2</sub>$  concentrations in the medium up to 100 mM. Highest bacoside A  $(2.37\pm$ 



concentrations of KCl and CaCl<sub>2</sub> on proline concentration in shoots of Bacopa monnieri. Each value represents mean of three replications and vertical bars indicate SE. The bars with different letters on the same color columns are significantly different at  $p<0.05$ 

<span id="page-6-0"></span>

0.29 mg  $g^{-1}$  DW) was noticed in the shoots cultured on 100 mM CaCl<sub>2</sub> (Fig. [7](#page-9-0)). Higher concentration of KCl and  $CaCl<sub>2</sub>$  (150 and 200 mM) resulted in a drastic decline in the bacoside A content in shoots of Bacopa.

# Discussion

Salinity is one of the major environmental constraints inducing a wide range of effects at the cellular and wholeplant levels. It causes significant yield reductions in different plant species (Belkheiri and Mulas [2013\)](#page-10-0). The ability to maintain turgor induced by salt stress may preserve the metabolic processes, and thus the growth of a plant (Martinez et al. [2004](#page-11-0)).  $K^+$  is the only important monovalent cation, that represents the major inorganic constituent essential for all higher plants (Tester and Davenport [2003](#page-12-0); Lv et al. [2012](#page-11-0)). It contributes to many physiological processes like osmotic pressure and ionic strength, cell elongation, growth of shoot and root, stomatal movement (Maathuis and Amtmann [1999](#page-11-0); Tester and Davenport [2003\)](#page-12-0); due to its relatively low charge:mass ratio, resulting in a small hydration shell and it has a low tendency to modify protein

concentrations of KCl and CaCl<sub>2</sub> on accumulation of glycine betaine in shoots of Bacopa monnieri. Each value represents mean of three replications and vertical bars indicate SE. The bars with different letters on the same color columns are significantly different at  $p \leq 0.05$ 

conformation (Maathuis and Sanders [1996](#page-11-0); Rascio et al. [2001](#page-12-0)).  $K^+$  is also essential for enzyme activation, protein synthesis and photosynthesis (Belkheiri and Mulas [2013\)](#page-10-0). Calcium on the other hand is an essential plant nutrient required for structural roles in the cell wall and membranes as a counter-cation for inorganic and organic anions in the vacuole. Besides, it is necessary for cell elongation and known to activate a number of enzymes such as phospholipase D, lecithinase, ATPase and amylase (Hepler [2005](#page-11-0)). It plays an important role in various physiological processes by acting as intracellular messenger in the cytosol and second messenger in the transduction of exogenous signals (Marschner [1995](#page-11-0)). The continuous supply, uptake, transportation, and metabolism of different ions seem to vary from the initiation of organs and meristematic tissue towards plant growth as has been pointed out by Kanashiro et al. [\(2009](#page-11-0)). The increase in Cl<sup>−</sup> concentration in the medium leads to a reduction of nitrate levels in tissues of several plant species (Gratten and Grieve [1994\)](#page-11-0). Similar observations were made by Santos et al. [\(2001](#page-12-0)) in KCl-stressed sunflower plants and calli.

Like many other abiotic stresses, salt stress inhibits plant growth. Salt stress may promote the synthesis of several

![](_page_6_Figure_10.jpeg)

concentrations of KCl and CaCl<sub>2</sub> on total soluble sugar content in shoots of Bacopa monnieri. Each value represents mean of three replications and vertical bars indicate SE. The bars with different letters on the same color columns are significantly different at  $p<0.05$ 

<span id="page-7-0"></span>![](_page_7_Figure_2.jpeg)

osmolytes necessary to combat stress thus relocating the valuable resources (Sabir et al. [2012](#page-12-0)). When shoots of Brahmi were exposed to different concentrations (0, 50, 100, 150 and 200 mM) of KCl and CaCl<sub>2</sub>, significant reduction in number of shoots, fresh and dry weight of shoots per culture and TWC were noticed (Table [1](#page-4-0); Fig. [1](#page-4-0)). At low concentrations of KCl and  $CaCl<sub>2</sub>$ (50–100 mM), shoot growth was less affected compared to 150 and 200 mM. Similarly, KCl stress decreased sunflower plant and calli growth (Santos et al. [2001](#page-12-0)). CaCl<sub>2</sub> dihydrate  $(CaCl<sub>2</sub> 2H<sub>2</sub>O)$  also decreased the growth of the plants.  $CaCl<sub>2</sub>$  in excess (3.35 mM) decreased both the fresh and dry mass of Aechmea blanchetiana plantlets (Kanashiro et al. [2009](#page-11-0)). Sabir et al. [\(2012](#page-12-0)) made identical observations in in vitro differentiating shoots of Withania somnifera Dunal. Among the KCl and  $CaCl<sub>2</sub>$  stresses,  $CaCl<sub>2</sub>$  showed more deleterious effect than KCl. Increasing concentration of KCl displayed more pronounced effect on the number of shoots per culture. Both growth and relative water content of salt-treated shoots of Withania somnifera were decreased with increasing salt concentration (Sabir et al. [2012](#page-12-0)). The growth reduction may be due to the decrease in the turgor. The saline solution establishes a

water potential imbalance between the apoplast and symplast that leads to decrease in turgor (Sangwan et al. [1994;](#page-12-0) Bohnert et al. [1995\)](#page-11-0). When the turgor is reduced below the threshold level of the cell wall, it results in the reduced growth responses (Sabir et al. [2012\)](#page-12-0). If the water potential difference is greater than it can be compensated for by turgor loss, then cellular dehydration starts. This growth cessation finally leads to the low dry weight accumulation (Molassiotis et al. [2006](#page-11-0); Sabir et al. [2012\)](#page-12-0).

Decrease in chla, chlb and total chl was observed with increasing KCl and  $CaCl<sub>2</sub>$  $CaCl<sub>2</sub>$  $CaCl<sub>2</sub>$  concentrations (Table 2). Salinity induces various biochemical and physiological responses in plants and affects almost all plant functions including chlorophyll content, photosynthesis, growth and development (Aghaleh et al. [2009\)](#page-10-0). Chlorophyll content is directly related to the growth and productivity of the plant. High salt concentration results in decreased chlorophyll pigments; and this might be due to the interference of salt ions in chlorophyll biosynthesis. High salt concentration creates an imbalance in ion homeostasis which restrains the iron to the protoprophyrin molecule. This results in decreased synthesis of chlorophyll pigments (Agastian et al. [2000](#page-10-0); Sabir et al. [2012](#page-12-0)). Lipid peroxidation as measured by MDA content

**Table 3** Effect of different concentrations of KCl and CaCl<sub>2</sub> on Na<sup>+</sup>, K<sup>+</sup> and Ca<sup>2+</sup> ion concentration in shoots of *Bacopa monnieri*. Values were taken after 28 days of culture

Concentration (mM)	$Na+$ (mmol $g-1 DW$ )		$K^+$ (mmol $g^{-1}$ DW)		$Ca^{2+}$ (mmol $g^{-1}$ DW)	
	<b>KCl</b>	CaCl <sub>2</sub>	<b>KCl</b>	CaCl <sub>2</sub>	KCl	CaCl <sub>2</sub>
$\overline{0}$	$0.90 \pm 0.03^{\text{a}}$	$0.90 \pm 0.03^{\text{a}}$	$09.13 \pm 0.2^d$	$9.13 \pm 0.2^a$	$0.34 \pm 0.02^b$	$0.34 \pm 0.02^e$
50	$0.74 \pm 0.03^b$	$0.34 \pm 0.01^b$	$12.34 \pm 0.2$ <sup>c</sup>	$9.54 \pm 0.2^a$	$0.39 \pm 0.02^{ab}$	$1.36 \pm 0.04$ <sup>c</sup>
100	$0.58 \pm 0.02^{\circ}$	$0.28 \pm 0.02$ <sup>bc</sup>	$15.69 \pm 0.3^{\text{a}}$	$8.52 \pm 0.3^b$	$0.42 \pm 0.02^a$	$1.86 \pm 0.04^b$
150	$0.54 \pm 0.02$ <sup>c</sup>	$0.24 \pm 0.02$ <sup>c</sup>	$14.73 \pm 0.3^{\rm b}$	$3.53 \pm 0.3$ <sup>c</sup>	$0.37 \pm 0.01^{ab}$	$2.67 \pm 0.12^a$
200	$0.37 \pm 0.03$ <sup>d</sup>	$0.15 \pm 0.01$ <sup>d</sup>	$14.48 \pm 0.2^b$	$2.41 \pm 0.1$ <sup>d</sup>	$0.10 \pm 0.02$ <sup>c</sup>	$0.73 \pm 0.0n$

Each value represents mean  $\pm$  SE of three replications. Means followed by same letters within columns are not significantly different at P ≤0.05 level by Duncan's multiple range test. DMRT was applied to each variety separately

<span id="page-8-0"></span>Fig. 6 Effect of different concentrations of KCl and CaCl<sub>2</sub> on activity of antioxidant enzymes in shoots of Bacopa monnieri. Shoots were transferred to media containing  $0-200$  mM KCl and CaCl<sub>2</sub> and the SOD  $(a)$ , CAT  $(b)$ , APX  $(c)$ and GPX (d) activities were measured after 28 days of culture. Each value represents mean of three replications and vertical bars indicate SE. The bars with different letters on the same color columns are significantly different at  $p \le 0.05$ 

![](_page_8_Figure_3.jpeg)

stress on bacoside A accumulation in shoot biomass of Bacopa monnieri. Each value represents mean of three replications and vertical bars indicate SE. The bars with different letters on the same color columns are significantly different at  $p<0.05$ 

<span id="page-9-0"></span>![](_page_9_Figure_2.jpeg)

increased with increasing KCl and  $CaCl<sub>2</sub>$  concentration from 0 to 200 mM (Fig. [2\)](#page-5-0). At higher KCl and  $CaCl<sub>2</sub>$ concentrations (150 and 200 mM), membrane damage was severe due to lipid peroxidation. These observations are in agreement with the hypothesis that the amount of MDA content is a direct sign of oxidative stress caused by damage to the lipid molecules of the cell membrane (Erturk et al. [2007\)](#page-11-0). Similar results were recorded by Santos et al. ([2001\)](#page-12-0) in the KCl treated sunflower plants and calli. An increase in lipid peroxidation was recorded in suspension cultures of C. roseus under salt stress (Elkahoui et al. [2005\)](#page-11-0). Increase in MDA content was also noticed in KCl and  $CaCl<sub>2</sub>$  treated in vitro differentiating shoots and calli of Withania somnifera (Sabir et al. [2012\)](#page-12-0).

The first response of cells during salt stress is the imbalance of osmotic potential due to excess salt. To adjust the osmotic potential of the cell, a suitable osmolyte or osmoprotectant molecules are required, which accumulate in cytosol. In the present study, a progressive increase in free proline content, glycine betaine and TSS were observed (Figs. [3,](#page-6-0) [4,](#page-6-0) and [5](#page-7-0)). Proline serves as a storage sink for carbon and nitrogen and as a free-radical scavenger (Chinnusamy et al. [2005](#page-11-0); Flors et al. [2007\)](#page-11-0). It stabilizes sub-cellular structures (membranes and proteins) and buffers cellular redox potential under stress (Kavi Kishor et al. [2005;](#page-11-0) Chinnusamy et al. [2005\)](#page-11-0). Glycine betaine and trehalose act as osmoprotectants by stabilizing quaternary structures of proteins and highly ordered states of membrane. Hence, these organic osmolytes are known as osmoprotectants and thus may alleviate salt induced damages (Bohnert and Jensen [1996;](#page-11-0) Chen and Murata [2002](#page-11-0)). Sugar is also considered as an osmoprotectant and reported to accumulate during salinity stress (Kovacik et al. [2009](#page-11-0)). Soluble sugars that are altered by abiotic stresses may also act as signaling molecules (Chaves and Oliveira [2004](#page-11-0)) and also interact with hormones as part of the sugar-sensing and signaling network in plants (Sabir et al. [2012\)](#page-12-0). Synthesis of

proline, glycine betaine and other osmolytes is an energy dependent process and consumes large number of ATP molecules (Raven [1985\)](#page-12-0), thus their synthesis affect the plant growth (Lokhande et al. [2011](#page-11-0)).

Salt stress is associated with complex traits, which include osmotic stress, specific ion effect, ion imbalances and nutrient deficiency. Accordingly, salt stress affects various physiological and biochemical mechanisms related to plant growth and development (Pitman and Lauchli [2002;](#page-12-0) Ahire et al. [2012b\)](#page-10-0). In the present study,  $Na<sup>+</sup>$  accumulation decreased with increasing concentrations of KCl and  $CaCl<sub>2</sub>$  in the medium (Table [3](#page-7-0)).  $K^+$  concentration increased significantly in the KCl-stressed shoots as compared to the control up to 100 mM KCl level. High  $Ca^{2+}$  accumulation was noticed when cultures were treated with 150 mM CaCl<sub>2</sub>. At still higher concentrations of both KCl and CaCl<sub>2</sub>, K<sup>+</sup> and  $Ca^{2+}$  ion accumulations declined. Thus, this study indicates that the shoots of *Bacopa* accumulate the  $K^+$  and  $Ca^{2+}$  at low concentrations of salt stress but may exclude them at higher KCl and  $CaCl<sub>2</sub>$  concentrations in the medium. Different plant species develop different adaptive mechanisms either to exclude salt from their cells or to tolerate it within the cells by sequestering in the vacuoles (Kozlowski [1997](#page-11-0); Munns [2002](#page-12-0); Parida and Das [2005](#page-12-0)). The uptake of large amounts of salt by the plant leads to the increase of osmotic pressure in the cytosol.

Salt stress leads to oxidative stress through an increase in reactive oxygen species (ROS), such as hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), superoxide  $(O_2^{\text{-}})$  and hydroxyl (OH<sup>\*</sup>) radicals. ROS can modulate normal cellular metabolism through oxidative damage to lipids, proteins and nucleic acids (Imlay [2003\)](#page-11-0). Plants have developed defensive antioxidative system, including low-molecular mass antioxidants as well as antioxidative enzymes such as SOD, CAT, APX and GPX. In the present study, KCl and  $CaCl<sub>2</sub>$  stresses (0–100 mM) significantly increased the antioxidant enzyme (SOD, CAT, APX and GPX) activities in shoots of Bacopa

<span id="page-10-0"></span>(Fig. [6a, b](#page-8-0), c, d). Higher activity of SOD is an indication of efficient detoxification of superoxide radical (Abraham and Dhar 2010) and results in formation of  $H_2O_2$ . The excess  $H_2O_2$  produced by SOD in response to salt stress is removed by both CAT, and APX (Parida et al. [2004](#page-12-0); Lokhande et al. [2011](#page-11-0)). The removal of  $H_2O_2$  in microbodies is mainly carried by the catalases (Scandalios et al. [1997\)](#page-12-0) and catalyzes either the direct decomposition of  $H_2O_2$  or the oxidation by  $H_2O_2$  of substrates. APX catalyzes the dismutation of  $H_2O_2$  through the ascorbate-glutathione cycle (Shigeoka et al. [2002;](#page-12-0) Jithesh et al. [2006\)](#page-11-0). Ascorbate is the most important reducing substrate for  $H_2O_2$  detoxification in the plant cells and APX uses ascorbate to reduce  $H_2O_2$  to water (Noctor and Foyer [1998](#page-12-0); Sergio et al. [2012](#page-12-0)). SOD generated  $H<sub>2</sub>O<sub>2</sub>$  is eliminated by GPX (Rios-Gonzales et al. [2002](#page-12-0)). Similar increase in the antioxidative enzyme activities under increased salt stress was reported in several other plants like rice seedlings (Kumar et al. [2009\)](#page-11-0), Sesuvium protulacastrum shoot cultures (Lokhande et al. [2011](#page-11-0)), brinjal seedlings (Ahire et al. 2012b); shoot and callus culture of Withania somnifera (Sabir et al. [2012](#page-12-0)). The study indicates that Bacopa has efficient antioxidative defensive mechanism to ameliorate the salt stress induced damages at lower salt concentrations.

Accumulation of bacoside A was enhanced with an increase in KCl and CaCl<sub>2</sub> concentrations in the medium up to 100 mM (Fig. [7\)](#page-9-0). Treatment of 100 mM CaCl<sub>2</sub> resulted in higher accumulation of bacoside A. But, KCl and CaCl<sub>2</sub> at higher concentrations suppressed its accumulation (Fig. [7](#page-9-0)). Salt stress exerts an unfavorable effect on the growth (Jaleel et al. [2007\)](#page-11-0), but increases secondary metabolite accumulation during biotic and abiotic stress conditions (Wahid and Ghazanfar [2006](#page-12-0); Cheng et al. [2007;](#page-11-0) Ghorpade et al. [2011](#page-11-0)). Increase in bacoside A content might be due to increased water stress which can regulate large number of transcripts including phenylpropanoid metabolic pathway genes. Calcium may act as a second messenger and thus trigger the expression of several genes associated with the biosynthetic pathway of bacoside A. Bacopa might synthesize the triterpene saponin to prevent the membrane damage. The low amount of saponins at higher concentrations of Cu provided an intrinsic defense to resist Cu-induced oxidative damage in P. ginseng plants (Ali et al. 2006). Similarly, the decrease in bacoside A accumulation at higher concentrations of salts might be due to inhibition of the metabolic pathway enzymes associated with its biosynthesis.

In conclusion, the findings presented here demonstrate that KCl and CaCl<sub>2</sub> induced salt stresses affected the number of shoots per culture, shoot length, shoot fresh and dry weights, tissue water content and lipid peroxidation (MDA content). Increased  $K^+$  and  $Ca^{2+}$  ions were observed under KCl and  $CaCl<sub>2</sub>$  stresses respectively. Antioxidant enzyme activities (SOD, CAT, APX and GPX) increased in shoots of Bacopa with enhanced salt levels (up to 100 mM). Bacoside A content increased but only under moderate stress  $(100 \text{ mM KCl and CaCl}_2)$ . The studies presented in this investigation provide an impetus for conducting growth adaptability responses of Bacopa monnieri in field conditions under different salt stress conditions and also to enhance bacoside A concentration.

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