



Enhanced bioconversion of grass straw into bioethanol by a novel consortium of lignocellulolytic bacteria aided by combined alkaline-acid pretreatment

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Received: 19 October 2023 / Revised: 1 January 2024 / Accepted: 4 January 2024
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Abstract

Grass straw represents an attractive and renewable resource for the production of value-added products like bioethanol. However, due to its recalcitrant structure and heterogeneity, efficient biodegradation of the grass straw remains a challenge for biorefinery. In view of this objective, the efficiency of the bacterial consortium was evaluated on grass straw pretreated with mild alkaline acid solutions. The results revealed that the pretreatment of the grass straw significantly enhanced the saccharification efficiency of the bacterial consortium. The Fourier transform infrared spectroscopy (FTIR), and X-ray diffraction (XRD) analyses revealed that chemical pretreatment altered the structure of the grass straw, removing the majority of the lignin and hydrogen bonding between cellulose and hemicellulose. The field emission scanning electron microscopy (FESEM) showed an altered surface of the pretreated grass straw with pores and tunnels. During saccharification, the highest reducing sugars ($224.33 \pm 1.15 \mu\text{g/mL}$) were released for 3% grass straw loadings using 2.5% of the bacterial consortium after 14 days of incubation, signifying the synergism of bacterial strains for lignocellulose bioconversion. Subsequently, the highest ethanol yield of $3.59 \pm 0.85\%$ (v/v) was attained at pH 5.5 on the 4th day of fermentation by *Saccharomyces cerevisiae* using 8% of yeast culture/inoculum. The overall obtained results highlight the significance of the cellulolytic bacterial consortium towards lignocellulosic biorefinery by secreting grass straw hydrolyzing cellulases and hemicellulases, signifying its valorization as feedstock as well as ethanol fermentation.

Keywords Bacterial consortium · Chemical pretreatment · Biomass processing · Grass straw · Lignocellulose valorization · Biorefinery

Abbreviations

| | |
|-------|-----------------------------------|
| LCB | Lignocellulosic biomass |
| GS | Grass straw |
| UT | Untreated |
| CT | Chemically pretreated |
| UT-GS | Untreated Grass straw |
| CT-GS | Chemically pretreated Grass straw |

| | |
|-------|---|
| BMS | Berg minimal salt media |
| CrI | Crystallinity index |
| SHF | Separate hydrolysis and fermentation |
| FTIR | Fourier transform infrared spectroscopy |
| EDS | Energy dispersive X-ray spectrometer |
| XRD | X-ray diffraction |
| FESEM | Field emission scanning electron microscopy |

Highlights

- Structural and chemical changes were observed after mild alkaline acid pretreatment of GS.
- Biomass loading and consortia volume significantly enhance reducing sugar production.
- Bioethanol production was evaluated by optimizing the effects of various parameters.
- Optimum concentration of ethanol ($3.59 \pm 0.85\%$) was obtained at 3% of biomass loading.
- The enzymatic hydrolysis improved the ability of CT-GS to produce of bioethanol.

1 Introduction

As the energy demand–supply gap widens, energy shortage due to limited fossil fuel resources has become a global issue. Nowadays, feedstock-based bioethanol is a promising alternative to fossil fuels. It is a renewable energy source that can reduce environmental damage caused by fossil fuels [1]. The feedstocks used to produce bioethanol can be categorized as sucrose-based feedstocks (sugar beet, sugar cane, and fruits, etc.), starch-based feedstocks (barley,

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wheat, rice, corn, and potatoes, etc.), and lignocellulosic biomass (LCB) based (wood, straw, husks, and grasses) [2, 3]. Among these renewable resources, LCB, such as grass straw (GS), offers a low-cost raw material with high availability for various industrial applications [4, 5]. The global production of straw-based lignocellulosic biomass, including GS, is approximately 7 billion tons [6] and generally consists of ~29–50% of cellulose, ~25–43% of hemicellulose, and 5–18% of lignin [7, 8] while major portion being burnt in open fields, augmenting greenhouse gas emissions [9]. A careful disposal strategy of straw lignocellulose would not only prevent secondary pollution but would also allow its utilization for the biofuel industry [10]. The GS is a renewable resource that can be utilized as an alternate and sustainable energy resource to meet the future demands of energy [11]. Within the realm of lignocellulose resources, grasses, primarily classified as monocots, or Poaceae (formerly Gramineae), attract attention for a variety of reasons. Compared to wood, the lignin content of grasses is often low, but they contain a higher amount of hydrolysable and fermentable polysaccharides. Noteworthy for their annual life cycle, grasses can serve as excellent examples of renewable resources [12]. Consequently, extensive research endeavors persistently focus on unraveling the unique structural and chemical attributes of grasses with a focus on optimizing methodologies for harnessing their considerable potential in lignocellulosic biorefinery [13, 14]. The bioconversion of GS is a suitable feedstock for biofuel production with minimum greenhouse gas emissions [15]. However, the recalcitrance of the GS impedes bioconversion into monomer sugars, thereby hindering its valorization for biorefinery. This recalcitrance is clearly attributed to its complex structure and chemical composition caused by hydrogen bonding between major components such as lignin, hemicellulose, and cellulose. Additionally, an essential factor affecting the degradability of GS is its degree of polymerization and crystallinity [16, 17]. The complex and recalcitrant nature of LCB is caused by the exceptional crystalline structure, high degree of polymerization (up to 10,000 units), and presence of a complex network of inter- and intra-molecularly hydrogen-bonded hydroxyl groups in cellulose [18]. Although hemicellulose is a heteropolymer of 5- and 6-carbon sugars, it does not significantly contribute to the recalcitrance of LCB [19]. The presence of reducing sugars in both cellulose and hemicellulose increases their industrial values as a key resource of several commercially useful chemicals [20]. In contrast, lignin is a complex and heterogeneous compound composed of methoxylated phenylpropanoid units such as sinaphyl alcohol, coniferyl alcohol, and p-coumaryl alcohol. Lignin is majorly responsible for the recalcitrance of LCB, thus impeding its industrial significance. Consequently, LCB requires a pretreatment to degrade the lignin structure and expose the cellulose and

hemicellulose polymers to enzymatic actions [21, 22]. Moreover, the removal of lignin also promotes enzyme–substrate interaction and improves saccharification efficiency as well as sugar hydrolysis [23]. A mild pretreatment of the LCB with inorganic solvents like acids and alkalis is preferred due to the high conversion rate (above 90%) of sugars and fast reactions [24]. The bioconversion of lignocellulosic GS into bioethanol is carried out through thermochemical as well as biochemical routes. However, the microorganisms mediated biodegradation offers an environment friendly, strategic and economic approach from the sustainability viewpoints. Microorganisms have evolved physiological traits to obtain energy from plant biomass, breaking down cell walls and releasing sugar monomers which can be utilized for their own metabolism and fermentation purposes [25]. Among microbes, bacteria are the most prevalent biological creatures in nature due to their ability to decompose natural polymers like cellulose [26]. Cellulolytic bacteria possess the ability to produce specific cellulases, enabling their conversion to valuable biofuels. The innovative enzyme combinations, microbial consortia, and bioengineering are crucial for successful biomass degradation [27, 28].

Despite decades of research, the systematic pretreatment and degradation of LCB, including GS, continues to be a major obstacle for twenty-first-century bio-refineries. The conversion of LCB into bioenergy involves employing pretreatment techniques such as physical, chemical, and biological methods. However, these methods have certain drawbacks because they are costly and cumbersome. Additionally, the development of highly efficient biocatalysts is a prerequisite for the production of biofuels [29–32]. The physical, chemical, and physico-chemical approaches for the utilization of waste biomass have proven ineffective due to their associated challenges and environmental issues; thus, a promising strategy of biological hydrolysis mediated by microbes is foreseen as a potential and promising alternative [33].

The enzymatic saccharification of LCB, facilitated by specific gut bacteria of animals, including snail emerges as a potentially attractive strategy from the sustainability perspective. The industrial use of lignocellulosic value-added products should consider enzymatic saccharification as a feasible strategy. Designing biologically distinct pathways to hydrolyze the LCB for a single cell is difficult. In contrast, bacterial consortia containing multiple species can balance two or more tasks simultaneously through synergistic mechanisms [34]. Therefore, splitting the biochemical functions by creating synthetic microbial consortia for a single-step cellulose conversion, where different microbial species and communities with specialized roles collaborate to enhance the bioconversion efficiency, seems to be a promising strategy [27]. This has inspired a fast-growing interest in engineering microbial consortia for biotechnology development. Thus, to increase the valorization of LCB, there is a dire

need to screen novel LCB, degrading microorganisms and creating effective microbial consortia [35].

In light of this objective, the present study aims to describe the saccharification of the chemically treated GS (CT-GS) by a bacterial consortium previously isolated from the gastro-intestinal tract of a giant African land snail [36]. The GS was pretreated with mild alkaline and acid solutions to enhance the overall hydrolytic efficiency of the consortium. The significance of pretreatment is that it was carried out at room temperature, and enzymatic digestibility was obtained through delignification of GS while rapid solubilization of the hemicellulose fraction improved hydrolysis of cellulose. Furthermore, the substrate (CT-GS) concentration and volume of consortia were optimized to improve the saccharification and fermentation potential of the bacterial consortium as well as yeast.

2 Materials and methods

2.1 Reagents and lignocellulosic substrates

The reagents like Sodium hydroxide (NaOH), D-glucose (C₆H₁₂O₆), 3,5-dinitrosalicylic acid (DNSA; C₇H₄N₂O₇), ammonium sulfate ([NH₄]₂SO₄), yeast extract, peptone, agar–agar, Potassium dichromate (K₂Cr₂O₇), Luria–Bertani broth (LB), disodium phosphate, were procured from the HiMedia (HiMedia Lab Pvt. Ltd., Mumbai, India) while the absolute ethanol, sulfuric acid (H₂SO₄), hydrochloric acid (HCL) were purchased from Thomas Baker (Thomas Baker Chemicals, Pvt. Ltd. Mumbai, India) and the SRL chemicals (Sisco Research Laboratories Pvt. Ltd. Mumbai, India), respectively. All chemicals and reagents were of the highest purity and analytical grade and prepared in sterile Milli-Q water unless otherwise mentioned. The lignocellulose substrates such as GS used in the study were collected from the Campus area (Latitude: 16.7050° N, Longitude: 74.2433° E) of Shivaji University, Vidyanaigari, Kolhapur (MS), India.

2.2 Phylogeny of the microorganisms

The cellulose-degrading bacterial strains included *B. tequilensis* G9, *Achromobacter xylosoxidans* R2, *B. subtilis* R10, and *Bacillus* sp. R27 that were previously isolated from the gastro-intestinal tracts of giant African land snail, *Achatina fulica* [37]. The bacteria were identified through 16S rDNA gene sequencing using bacteria-specific primers 27F and 1492R [38]. The phylogenetic relationships were inferred through the alignment with nucleotide sequences of the closely related bacterial species retrieved from the GenBank. The phylogenetic trees were constructed by the neighbor-joining method using the Kimera-2 model in the MEGA X software, followed by annotation in the iTOL program

(<https://itol.embl.de/>) with 500 bootstrap values. A total of ten different consortia were formulated with these bacteria and evaluated for effective degradation of GS [36]. Based on our previous study, the most efficient bacterial consortium was further characterized here to valorize the enzymatic hydrolysis of alkaline acid-treated GS for the liberation of reducing sugars and their fermentation into bioethanol.

2.3 Pretreatment and preparation of substrates

The collected GS was cleaned and chopped with scissors to remove the soil and other impurities. Fifty grams (50 g) of the chopped GS was washed with distilled water (DW), and then treated with an alkaline solution of NaOH (1% in Milli-Q water w/v) for 24 h. Next, the substrates were continuously rinsed with DW to neutralize the pH. Once the pH of the filtrate was near neutral (~7), the substrates were subjected to acid treatment by immersing the GS in H₂SO₄ solution (10% in Milli-Q water v/v) for 24 h. Subsequently, the pretreated GS was rinsed repeatedly with DW till the pH of the filtrate was adjusted to pH 7.0. Thereafter, the pretreated substrates were air dried (at room temperature) at 30 ± 2 °C, ground in a pulverizer, and then sieved to achieve particles of uniform size (≤ 1 mm). The substrates were stored and preserved at room temperature until further use. The pretreated GS used as a substrate in the study is hereafter referred to as CT-GS. The visual macroscopic evaluation of GS after each stage of the pretreatment is shown in Fig. 1. The effect of the alkali and acid treatments was evident from the change in color as well as the morphology of the substrates. The overall methodological approach used in this study is presented in Fig. 1.

2.4 Characterization of the substrates

The untreated substrates (UT-GS) as well as CT-GS were characterized and assessed to determine the changes in chemical composition and structure of the substrates due to pretreatment processes. The change in functional groups was detected by Fourier transform infrared (FTIR) spectrometry using transmittance mode over the range of 400–4000 cm⁻¹ on an FTIR spectrometer (Bruker Alpha Shimadzu, Japan). For FTIR analysis, the samples were mixed with 200 mg of potassium bromide powder (KBr) and compressed to form a pellet (1–1.5 mm thickness) by applying pressure of 6–7 tons. Furthermore, the surface properties and crystallinity patterns of the UT-GS and CT-GS were detected through X-ray powder diffraction (XRD) at a scan rate of 12° min⁻¹ and 2θ = 5°–90° using Bruker's AXS Analytical Instruments (Bruker Pvt. Ltd., Germany). The crystallinity indices (CrI) of UT-GS and CT-GS were calculated using the following equation:

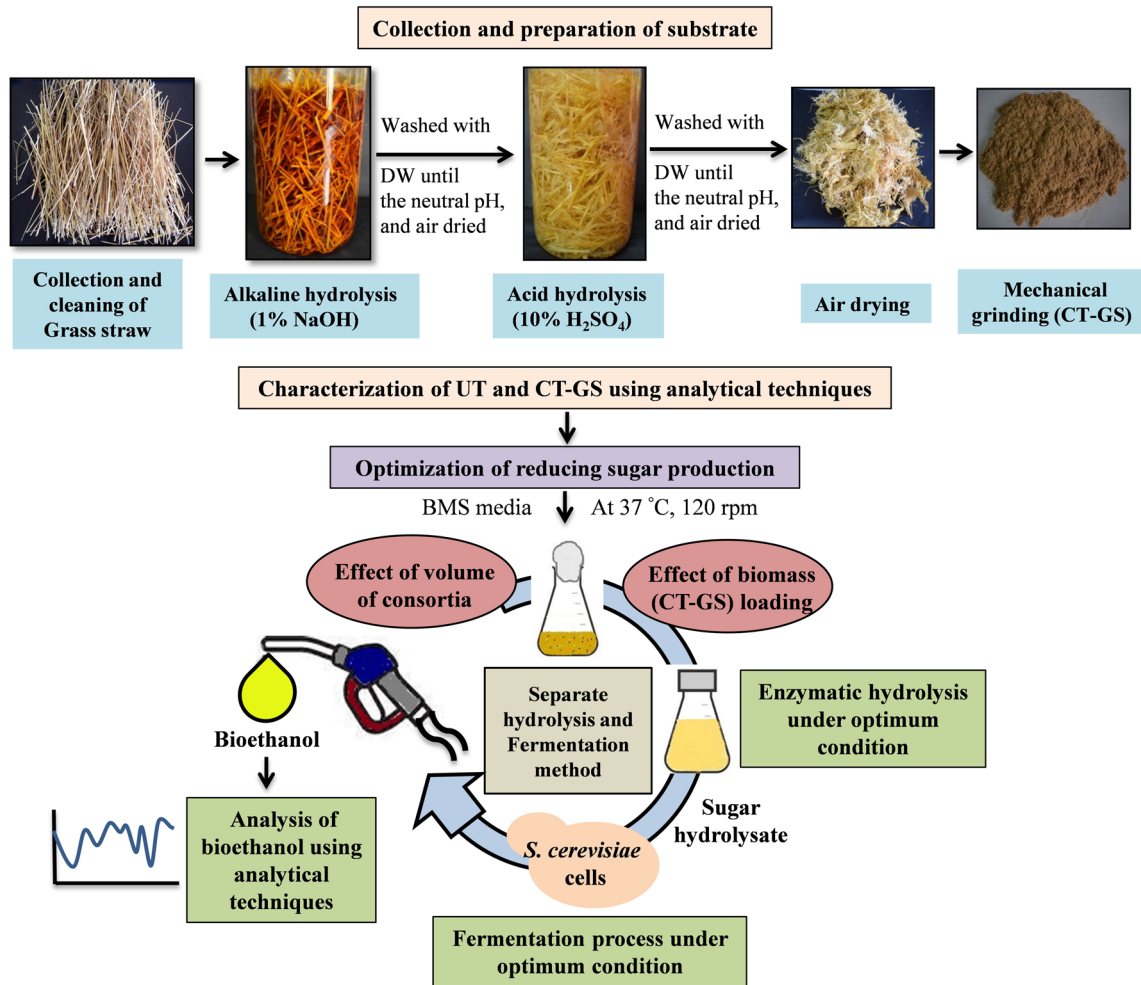


Fig. 1 An overview of the methodological approach used for the experimentation of the present study

$$\text{CrI} = (I_{002} - I_{001})/I_{002} \times 100 \quad (1)$$

where I_{002} (2θ) and I_{001} (2θ) is the intensity of crystalline peaks and the amorphous peaks in XRD patterns respectively.

Similarly, the elemental composition of biomass was investigated and compared by energy dispersive X-ray (EDS) spectrometer using AztecLive EDS analysis software (Oxford Instruments, UK). The surface morphology of the UT-GS and CT-GS substrates was analyzed by field emission scanning electron microscopy (FESEM) using MIRA3 FESEM (TESCAN Bron, Czech Republic).

2.5 Effect of biomass loading and volume of consortia on reducing sugar production

The influence of biomass loading on sugar production using bacterial consortia was determined by varying the biomass loadings. To achieve this, the 250 mL Erlenmeyer flasks

containing 100 mL of Berg minimal salt media (BMS) culture media were supplemented with varying biomass concentrations ranging from 0.5 to 4.0%. The media were inoculated with 1% consortium and incubated for 30 days in a rotary incubator shaking at 120 rpm at 37 °C. During the incubation period, sample aliquots of 2 mL were collected every 2 days to test the optimum concentration of the substrate required for maximum hydrolysis of CT-GS. The substrate hydrolysis was monitored by measuring the concentration of reducing sugars. To this end, sampled aliquots were centrifuged at 10,000 rpm for 10 min at 4 °C, and the supernatants obtained were assayed to estimate reducing sugar using the DNSA method [39]. In addition, the effect of the volume of consortium on substrate hydrolysis was also tested by using different volumes ranging from 0.5 to 4% of consortium in BMS medium. After inoculation, the culture flasks containing varying proportions of the bacterial consortium were incubated at 37 °C for 30 days in a rotary shaker. Similar to the influence of biomass loadings, sugar production was estimated by the DNSA method.

2.6 Bioethanol production

In this method, CT-GS was subjected to enzymatic saccharification by using bacterial consortium in BMS medium at the optimum conditions as determined in previous sections. The enzymatic hydrolysate of GS was concentrated to 5%, reducing sugar using a rotary evaporator (at 80 °C). The fermentation medium comprised of (g/L): 40; yeast extract, 5.0; (NH₄)₂SO₄, 10.0; KH₂PO₄, 4.5; and MgSO₄·7H₂O, 1.0. The medium was inoculated with 4% (v/v) of 48-h-old seed culture of *S. cerevisiae*, then incubated at 30 °C and 120 rpm for fermentation. The experiments were carried out for 7 days at 30 °C. Ten milliliters of the medium was sampled at regular intervals, centrifuged (10,000 rpm for 10 min), and filtered through 0.2 μm syringe filters for estimation of residual sugar contents and ethanol. The residual sugar was estimated by DNSA method [39] as prescribed earlier [36].

The supernatant obtained during the process of optimization was distilled using a distillation apparatus, and bioethanol was separated via the distillation process. For this, 3 mL of fermented sample was transferred to a round bottom flask and diluted with 30 mL Milli-Q water. The round bottom flasks were attached with condenser and collection flasks. The whole setup was placed on a heating mantle, and the water was circulated through the condenser. The temperature was maintained at 78–80 °C [Due to the difference in boiling point of water (100 °C) and ethanol (78.6 °C)] in the heating mantle. The vaporous was allowed to pass through the outlet port attached to the top. The evaporated fraction was then distilled, and the condensed sample was collected. Furthermore, the samples were subjected to distillation process, and then distillate was analyzed for ethanol content by the potassium di-chromate method. The ethanol was estimated calorimetrically [40] by dissolving 33.83 g of K₂Cr₂O₇ in 500 mL of DW followed by the addition of 325 mL of concentrated H₂SO₄ (36 N). The volume of solution was adjusted to 1000 mL with Milli-Q water, with a final normality of 0.23 N K₂Cr₂O₇ in solution. Finally, 25 mL of K₂Cr₂O₇ reagent (0.23 N) were mixed with 20 mL of distillate, and the samples were kept in a water bath at 60 °C for 20 min for the development of color. Afterwards, the reactions were cooled, followed by the measurement of optical density at 600 nm in a UV–Visible spectrophotometer (Eppendorf AG 22331 Biospectrometer, Hamburg, Germany).

2.7 Effect of physico-chemical parameters on ethanol production

The optimization of process conditions is paramount to obtain maximum ethanol yield. Different process parameters such as pH, temperature, yeast inoculum size, and fermentation period, were optimized for efficient bioconversion of CT-GS sugar hydrolysate into ethanol. The enzymatic hydrolysate of CT-GS was concentrated to a reducing sugar concentration

of 5% by evaporation in a rotary evaporator (80 °C). In order to study the effect of variation in pH of the medium, it was varied from 4.0 to 6.0 (increasing by 0.5 pH) with 1N hydrochloric acid (HCL) or alkali (1N NaOH) at RT under stirring conditions. Then media were inoculated with 6% (v/v) of *S. cerevisiae*, followed by incubation at 30 °C (120 rpm). Furthermore, the effect of temperature on ethanol fermentation was carried out by incubating the culture media at different temperatures, such as 25, 30, 35, and 40 °C, keeping pH and other conditions constant and optimum. Since the size of the yeast inoculum also significantly influences the rate of fermentation and ethanol yield, its optimization was carried out separately by varying the concentrations of *S. cerevisiae* in the range of 2 to 10% (v/v) in individual flasks and incubated at optimum pH and temperature.

Similarly, the fermentation of the reducing sugar hydrolysate was determined by varying fermentation periods ranging from 0 to 7 days, keeping all other parameters constant. During optimization, the culture supernatants (10 mL) were collected aseptically every 24 h for 7 days and centrifuged at 10,000 rpm for 10 min. The supernatants obtained were filtered through 0.2 μm syringe filters to collect the filtrate and subjected to a distillation process as mentioned in Sect. 2.6. The condensed samples were tested for bioethanol with K₂Cr₂O₇ reagent test [40]. The standard curves were plotted by using different concentrations ranging from 0 to 10% of the molecular grade ethanol solution. The cell biomass concentration was determined by the dry weight method [41]. For cell dry weight, 10 mL sample was collected and centrifuged at 10,000 rpm for 10 min in a pre-weighed tube, and then settled solids were washed with 10 mL of Milli-Q water. The tubes were dried overnight at 70 °C for 24 h and weighed again with weighing balance.

2.8 Confirmative analysis of bioethanol by gas chromatography-tandem mass spectroscopy and FTIR

The bioethanol obtained after separate hydrolysis and fermentation (SHF) process was analyzed by gas chromatography (GC–MS/MS) (Shimadzu, Japan) using SH-Rxi-5sil MS column (30 m × 0.25 μm). The column temperature during the program was 70 °C with the run time of 26 min. The injector temperature was maintained at 250 °C. The carrier gas was helium in the gas chromatography with the sample quantity of 1 μL at flow rate of 42 mL/min with split ratio of 1/25. The system was operated at electron ionization energy of 70 eV with 44 cm/sec of linear velocity. After comparing the respective retention times and mass spectra analyses with those of NIST14 library data (GC–MS system) and existing literature, the component identity was finally clarified. Furthermore, the bioethanol was confirmed using FTIR spectroscopy analysis to study the basic functional groups present in the sample.

2.9 Statistical analysis

All the experiments were performed in triplicates. To compare the data between UT-GS and CT-GS, a one-way analysis of variance (ANOVA) was conducted. Subsequently, multiple comparisons were carried out using the Tukey post-hoc test. A $p < 0.05$ was evaluated to consider statistical significance.

3 Results

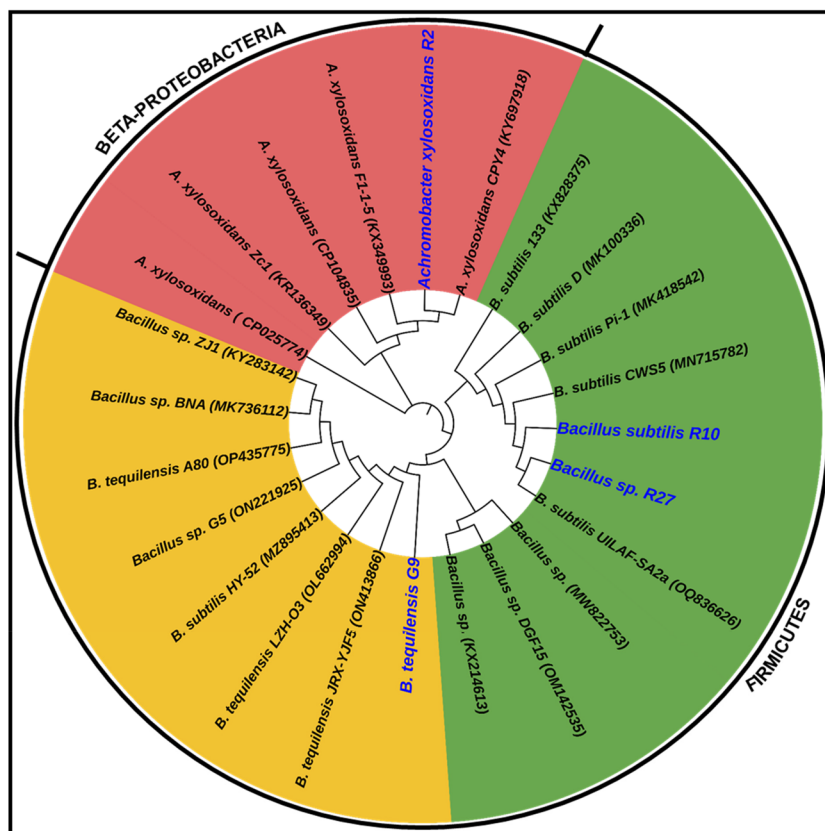
3.1 Phylogenetic analysis of the consortium

The 16S rDNA sequencing revealed that bacterial consortium was composed of gram-positive (*Bacillus* spp.) as well as Gram-negative bacteria (*Achromobacter xylosoxidans*). Among the four bacterial strains, only one strain, *Achromobacter xylosoxidans* R2, was gram-negative, belonging to the family Alcaligenaceae, while all other strains were affiliated with Gram-positive genera of *Bacillus*. The molecular homology inferred that the potential cellulose-degrading bacteria were *B. tequilensis* G9, *Achromobacter xylosoxidans* R2, *B. subtilis* R10, and *Bacillus* sp. R27. The phylogenetic tree (Fig. 2) demonstrated a $> 99\%$ similarity score of members of the bacterial consortium with closely related NCBI relatives.

3.2 Effect of the pretreatment process on the composition of grass straw

FTIR and XRD analysis The FTIR spectra of the untreated and chemically treated GS, as shown in Fig. 3a and Table 1, revealed noticeable changes in the composition of the substrates due to pretreatment processing. The FT-IR spectra of the UT-GS and CT-GS were dominated by the peaks at 3345 and 3399 cm^{-1} , reflecting the stretching vibrations of O–H stretching (hydrogen bond) [42] while the absorption bands at 2897 and 2354 cm^{-1} represented the CH, CH₂, or CH₃ stretching [43] and C–H stretch, C–H deformation of cellulose [44]. The wave numbers with peak intensities at 2132 and 2049 cm^{-1} were associated with C=C stretch vibration of cellulose. The absence of the vibration peak at 1736 cm^{-1} in the CT-GS, was obvious for C–O stretching of methyl ester and carboxylic acid in pectin or acetyl groups in hemicelluloses [45]. Similarly, the intensity of the bands at 1432, 1374, 1375, 1646, and 1515 cm^{-1} was found to be associated with C–H in-plane deformation, C=C aromatic skeletal vibration, and aromatic ring vibration of lignin whereas decrease in intensity of both the peaks (at 1374 and 1515) in the CT-GS revealed the removal of lignin after the chemical treatment. The absorption peak that was absorbed at 1465 cm^{-1} reflected the C–H deformation of lignin [42]. The intensity of the prominent peak in IR spectra of UT-GS at 1315 cm^{-1} attributed to

Fig. 2 A phylogenetic analysis based on the 16S rDNA gene sequencing of the bacterial strains composing the lignocellulolytic consortium



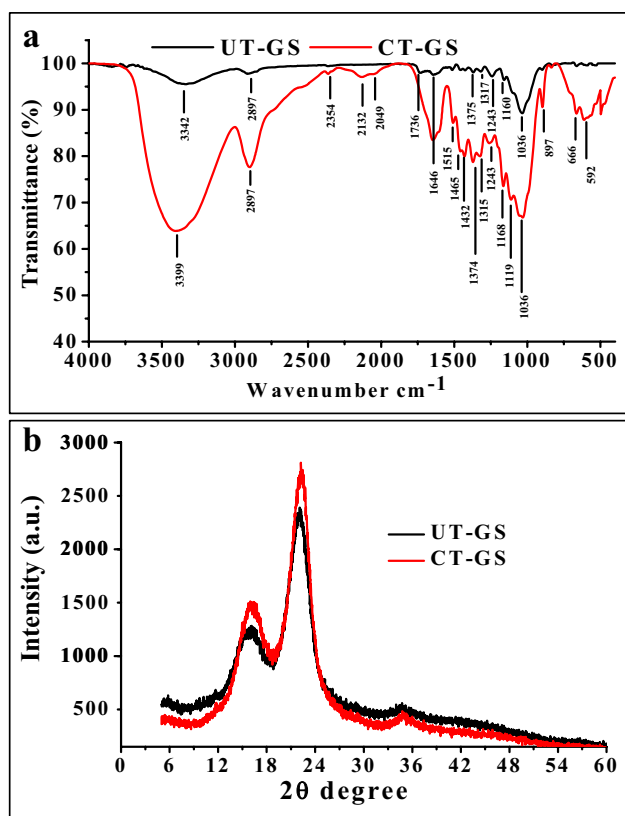


Fig. 3 Characterization of UT-GS and CT-GS. **a** FTIR, **b** XRD

$-\text{CH}_2$, C-H , or C-O of cellulose was found to increase after pretreatment (1317 cm^{-1}) [46]. The absorbance at 1243 cm^{-1} was associated with the C-O stretching in xylene of hemicellulose [47] while peaks at 1160 and 1168 cm^{-1} were assigned to C-O-C asymmetrical stretching in cellulose and hemicellulose [46]. The increase of the band intensities at 1036 and 897 cm^{-1} in the CT-GS indicated the aromatic C-H in-plane deformation [46] and typical β -1,4-glycosidic linkages of cellulose [48]. A detailed account of the changes observed in the chemical composition of the pretreated substrate is listed in Table 1, highlighting the effect of the combined alkaline-acid pretreatment on the bond's lignocellulosic material.

The X-ray diffraction crystallography of UT-GS and CT-GS exhibited two major peaks at 2θ around 22° and 16° representing crystalline cellulose and amorphous compounds, respectively (Fig. 3b). When compared to UT-GS, an increase in peak height at 22° was clearly observed in CT-GS, suggesting the higher exposure of crystalline cellulose. The CrI of UT-GS and CT-GS were 44.49 and 47.78%, respectively (Table 2).

EDS and FESEM analysis of biomass The EDS analyses indicated that UT-GS contained carbon (45.87%) and oxygen (27.19%) as dominant elements, indicating the carbon and oxygen atoms typically originated from the natural fiber (Fig. 4a). After pretreatment, the EDS spectra showed a higher percentage of carbon (78.39%) than oxygen (9.74%) in the CT-GS substrate (Fig. 4b). The morphological changes in

Table 1 FTIR-based identification of the altered functional groups in the GS due to alkali-acid pretreatment

| Frequency cm^{-1} | Functional group with type of vibrations | Corresponding biomass component | References |
|----------------------------|--|---------------------------------|------------|
| 3300–3400 | O–H stretching (hydrogen bond) | Cellulose | [32] |
| 2897 | CH, CH_2 , or CH_3 stretching | Cellulose | [33] |
| 2354 | C–H stretch, C–H deformation | Cellulose | [34] |
| 2132 | C=C stretch vibration | Cellulose | [35] |
| 2049 | C=C stretch vibration | Cellulose | [35] |
| 1736 | C–O stretching of methyl ester and carboxylic acid in pectin or the acetyl group | Hemicellulose | [35] |
| 1646 | C=C aromatic skeletal vibration | Lignin | [32] |
| 1515 | Aromatic ring vibration | Lignin | [37] |
| 1465 | C–H deformation | Lignin | [32] |
| 1432 | C–H in-plane deformation | Lignin | [37] |
| 1374, 1375 | C–H deformation | Cellulose | [32] |
| 1315, 1317 | $-\text{CH}_2$, C–H or C–O of cellulose | Cellulose | [36] |
| 1243 | C–O stretching in xylene | Hemicellulose | [37] |
| 1160, 1168 | C–O–C asymmetrical stretching | Cellulose and hemicellulose | [36] |
| 1036 | aromatic C–H in-plane deformation | Cellulose | [36] |
| 897 | β -1,4-glycosidic linkages | Cellulose | [38] |
| 666 | C–OH bending | Cellulose | [35] |
| 592 | C–OH bending | Cellulose | [35] |

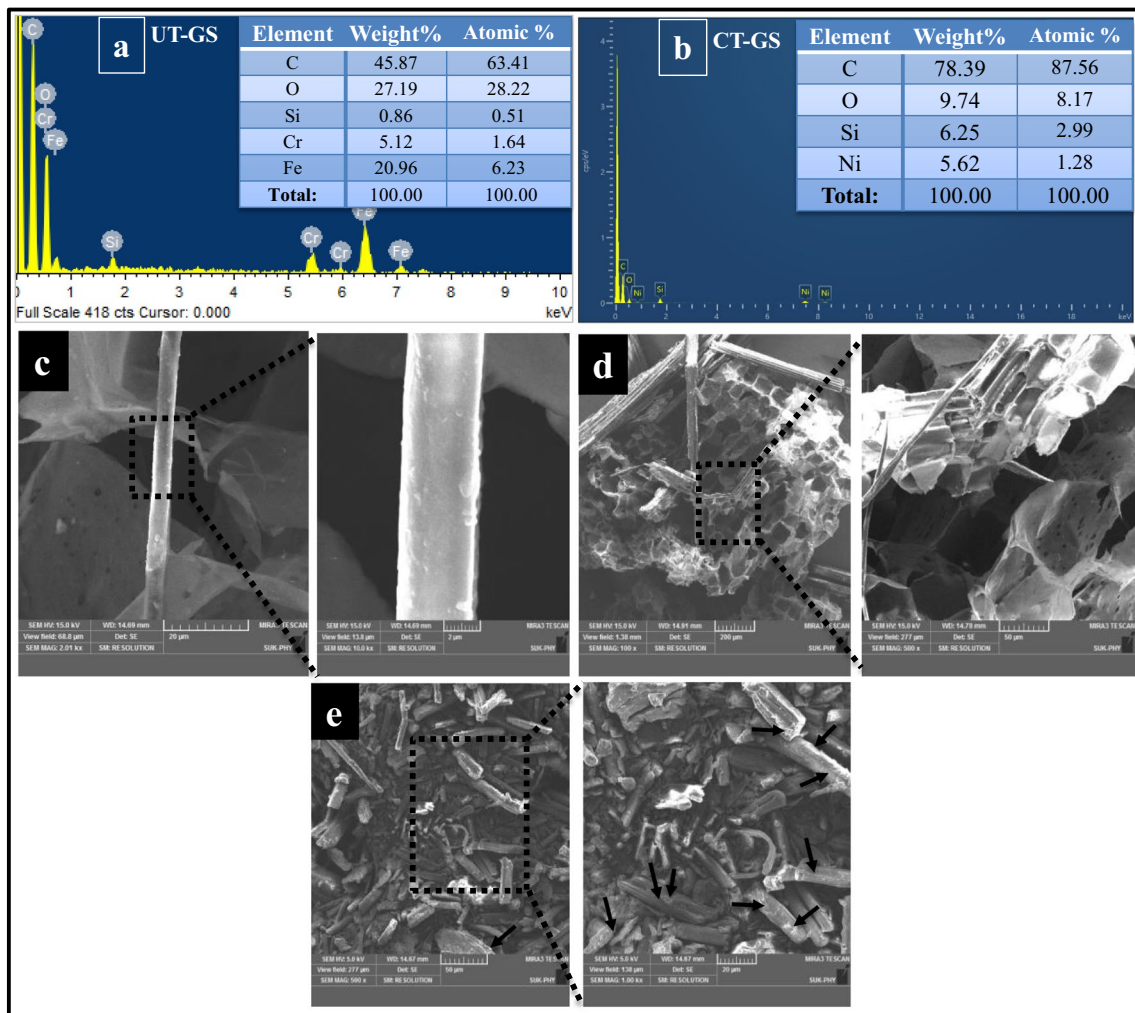
Table 2 The crystalline index (CrI) of UT-GS and CT-GS estimated from the XRD analyses

| LCB | I_{002} (2 θ) intensity of crystalline peak | I_{001} (2 θ) intensity of amorphous peak | CrI (%) Of LCB |
|-------|---|---|----------------|
| UT-GS | 22.21 | 16.12 | 44.49 |
| CT-GS | 22.21 | 16.12 | 47.78 |

the CT-GS substrate investigated through FESEM analysis (Fig. 4c) showed well-organized and unaffected surface of the untreated substrate, UT-GS. The smooth surfaces on the fibers indicated the lignin coverage, while the closed entire structure and intact plant cell wall with vascular structure indicated the recalcitrant nature observed in Fig. 4d. In contrast, the distorted surface and structure of the CT-GS revealed significant alteration caused by the chemical pretreatment. The surface of CT-GS turned rough and disorganized (Fig. 4e), which can be attributed to the hydrolysis of the amorphous regions of lignocellulose caused by the pretreatment processes.

3.3 Effect of biomass loading on reducing sugar production

In this study, different dosages of CT-GS were used to determine how biomass loading influences hydrolysis to produce reducing sugar. Figure 5a shows the interactive effect of biomass loading and incubation time on reducing sugar yield. Overall, a low yield of reducing sugar (5.71 ± 0.32 – 13.38 ± 0.20 $\mu\text{g}/\text{mL}$) was observed at lower biomass loadings (0.5–1%) after 4 days of incubation. However, the reducing sugar yield was found to consistently increase with higher biomass loadings (0.5–3%) up to 12 days. When the biomass loadings were further increased (3.5–4%), a decreasing trend in the reducing sugar production (159.69 ± 1.2 – 130.78 ± 1.0 $\mu\text{g}/\text{mL}$) was observed (Fig. 5a). In our study, a gradual increase in reducing sugar production from 57.91 ± 0.12 – 229.38 ± 1.42 $\mu\text{g}/\text{mL}$ for 0.5–3% biomass loading was observed. Overall, the lowest sugar production was observed on 4th day of incubation

**Fig. 4** The EDS of UT-GS (a) and CT-GS (b), and FE-SEM analysis of UT-GS (c and d) and CT-GS (e)

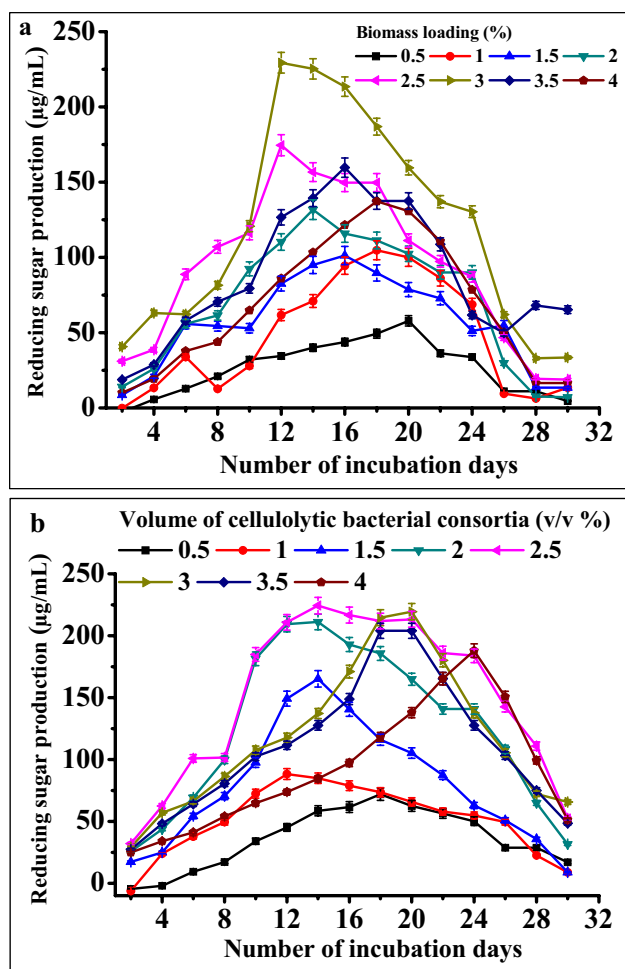


Fig. 5 The interactive effect of biomass loading (CT-GS) (a) and volume of cellulolytic bacterial consortia (b) on reducing sugar production over the incubation time of 30 days

with 0.5% biomass (Fig. 5a). The maximum reducing sugar yield of 229.38 ± 1.42 µg/mL at 3% biomass loading was obtained on 12th day of incubation while its lowest sugar production (33.03 ± 0.11 µg/mL) was found on 28th day of incubation (Fig. 5a). The reducing sugar production from 16 to 30 days of incubation tends to reduce drastically in case of all biomass loadings. Thus, 3% (w/v) of CT-GS of biomass was the optimum substrate loading for the present system with 12 days of incubation for the production of reducing sugar of 229.38 ± 1.42 µg/mL.

3.4 Effect of consortia volume on reducing sugar production

To study the effect of volume of cellulolytic bacterial consortia, inoculum volumes ranging from 0.5 to 4% (v/v) were examined for reducing sugar productions (Fig. 5b). Maximum reducing sugar production of 224.33 ± 1.15 µg/

mL occurred with 2.5% (v/v) cellulolytic bacterial consortia after 14 days of incubation; however, further increase up to 4% (v/v) resulted in reduced sugar yield of 214.60 ± 1.0 to 187.94 ± 0.93 µg/mL from 18 to 24th day of incubation (Fig. 5b). The lowest inoculum size of 0.5% produced 71.94 ± 0.76 µg/mL of reducing sugar after 18 days of incubation, which could be attributed to the slow growth of the bacteria and reduced conversion of the substrates.

3.5 Optimization of bioethanol production

In this study, several process variables, such as pH, temperature, inoculum size, and fermentation period, were optimized. To this end, when the pH of culture media was varied in the range of pH 4–6 and studied by monitoring daily over the period of 7 days, the rate of ethanol production was found increased gradually from 0.88 ± 0.1 to $2.68 \pm 0.9\%$ (v/v) with increase in pH from 4 to 5.5 and declined thereafter up to $2.34 \pm 0.41\%$ (v/v) at pH 6 (Fig. 6a). The maximum ethanol production ($2.68 \pm 0.94\%$) (v/v) was attained at pH 5.5 after 4 days of fermentation (Fig. 6a).

Similarly, when the effect of temperature in the range of 25–40 °C (with an interval of 5 °C) on the ethanol production was studied, considerable variation in ethanol production with respect to incubation time was observed. The strain grew vigorously at 30 °C, yielding a higher ethanol yield of $2.10 \pm 0.92\%$ (v/v) after 5 days of incubation (Fig. 6b). After 5th day, a continuous decrease in the ethanol production ($0.14 \pm 0.25\%$) (v/v) was observed up to 7th day at all tested temperatures, being lowest at 35 °C. The optimization of the inoculum size of yeast is ideal to reduce the cost of cellulosic ethanol production.

To investigate the effect of inoculum size of the yeast on the production of ethanol, various volumes of yeast culture ranging from 2 to 10% (v/v) were added to the sugar hydrolysate-containing culture media. Figure 6c shows that the ethanol production increased from 1.60 ± 0.82 to $2.07 \pm 0.92\%$ (v/v) at 2 to 6% yeast inoculum, respectively. The ethanol production was enhanced ($3.02 \pm 0.94\%$) (v/v) with increasing amounts of the yeast culture of 8% after 5 days of incubation (Fig. 6c).

During the initial days of fermentation, cell biomass and ethanol concentrations showed an irregular trend, probably due to the adaptation of the yeast cells to the media conditions (Fig. 7). Thereafter, cell biomass grew quickly as a result of the fast utilization of nitrogen and glucose. A sharp increase in cell biomass and ethanol was observed between 1 and 3 days of incubation. After 3 days, biomass concentration remained largely steady, and fermentation was observed nearly at steady conditions. A prolonged incubation decreased both the cell biomass and ethanol production. In

Fig. 6 Optimization of ethanol production. **a** pH, **b** temperature, and **c** inoculum (*S. cerevisiae*) size

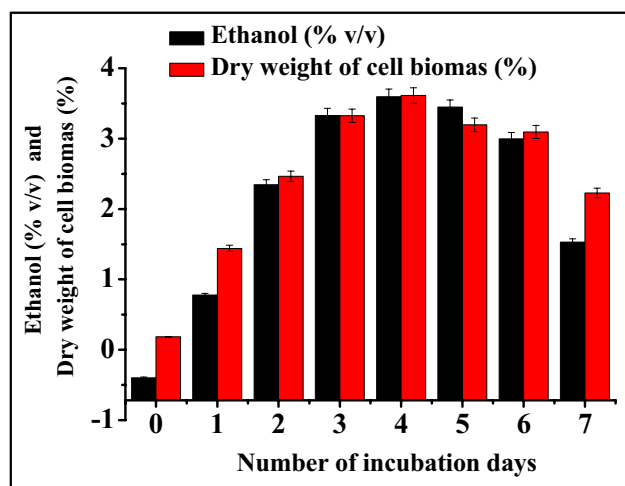
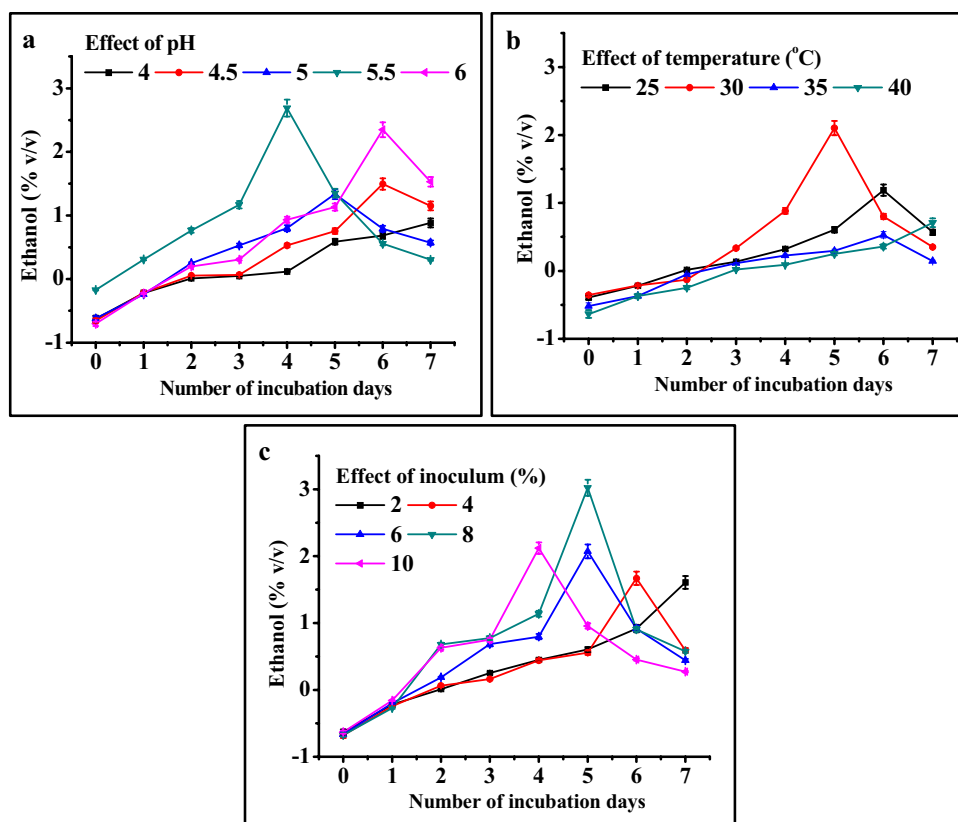


Fig. 7 Time-dependent ethanol production and dry weight of cell biomass

terms of the incubation period, the cell biomass increased till 4th day, thereafter showed a decreased growth, which could be due to the limited nutrients and prolonged fermentation, thereby causing reduced ethanol yields (Fig. 7). However, it was observed that $3.59 \pm 0.85\%$ (v/v) maximum bioethanol production from CT-GS sugar hydrolysate was estimated within 4 days of incubation (Fig. 7).

Table 3 GC–MS/MS-based identification of the products generated during fermentation

| Peaks | Rt (min) | Area (%) | Retention index | Name of compound |
|-------|----------|----------|-----------------|----------------------|
| 1 | 1.045 | 96.54 | 463 | Ethanol |
| 2 | 1.256 | 1.17 | 597 | 1-Propanol, 2-methyl |
| 3 | 1.631 | 2.28 | 697 | 1-Butanol, 3-methyl |

3.6 Characterization of bioethanol

The chromatographic profile of the compounds derived from fermentation is shown in Figure S1 and the identified peaks are presented in Table 3. The peaks elucidating the fermentation were assigned to 1-Propanol, 2-methyl, and 1-Butanol, 3-methyl, with retention time (Rt) of 1.256 and 1.631 respectively. The peaks corresponding to ethanol showed Rt of 1.045 (Fig. S1). Thus, the mass spectrum of bioethanol monitored at 31 and 45 mz^{-1} ion fragments were the most abundant fragments for ethanol (Fig. S2). The FTIR results for the analyzed sample revealed typical bioethanol absorption bands (Fig. 8 and Table. S1). The broad absorption band at 3309 cm^{-1} corresponded to the O–H stretching vibration of alcohols, confirming the presence of the alcohol group [49]. The minor peak identified between 2914 cm^{-1} and 2853 cm^{-1} indicated the

O–H stretching and C–H-based alkane, respectively [50], whereas the weak intensity peaks at 2113 cm^{-1} could be attributed to $-\text{C}=\text{C}-$ (alkyne) [51]. In addition, the sharp peak at 1657 cm^{-1} corresponded to the water bands, which represents the bending vibration of the H–O–H angle and the stretching vibration of the O–H groups [52]. The small peaks at 1457 cm^{-1} represented the CH_2 and CH_3 groups [53]. In addition, C–O stretch at 1106 cm^{-1} confirmed the bonds in the structure of ethanol [54]. Furthermore, the characteristic bands that appeared around 1048 cm^{-1} were associated with the stretching vibration of C–OH [49] (Fig. 8).

4 Discussion

4.1 Phylogenetic analysis of the consortium

To date, several lignocellulose-degrading bacterial consortia have been isolated from the different environments [55, 56] but not from the gut system of animals, particularly snails. To this end, here we have formulated a bacterial consortium with efficient lignocellulose biodegradation capabilities from the gastro-intestinal tract of a giant African land snail, *Achatina fulica*, that predominantly feeds on plant matter [4, 37]. Moreover, the interest in the use of bacterial consortia for the valorization of lignocellulose has gained scientific interest in recent years [57, 58]. A significant feature of the consortia-mediated efficient biodegradation of lignocellulose could be attributed to the synergism and metabolic cooperation of these bacterial strains and their enzymes within the gut environments [59].

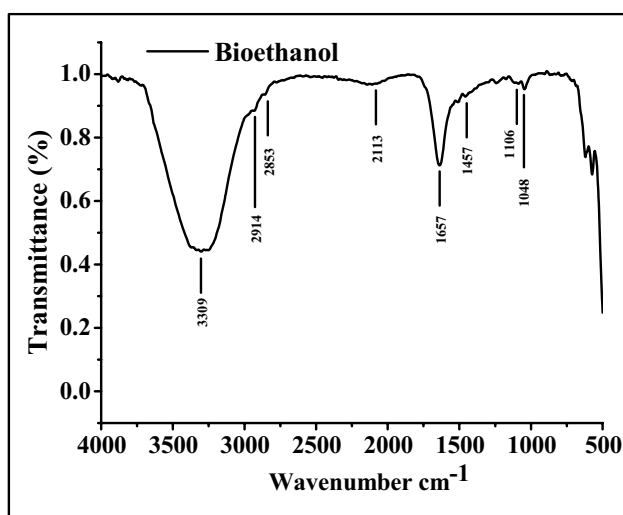


Fig. 8 FTIR spectra of the bioethanol produced from CT-GS after treatment with bacterial consortia

4.2 Effect of the pretreatment process on the composition of grass straw

The FTIR results indicate that pectin and hemicelluloses can be successfully removed using a combined alkaline-acid pretreatment. Therefore, it could be inferred that the significant differences observed between the chemical composition of UT-GS and CT-GS substrates were caused by the combined chemical pretreatment processes. The increase in CrI is proportional to the increase of crystalline cellulose in the total solid due to the removal of partial lignin, besides the amorphous contents of cellulose, hemicellulose, and other compounds [42, 60]. Moreover, lignin solubilization due to alkaline and acid treatments resulted in increased porosity and disintegration of the fiber matrix. In addition, the hydrogen bonding pattern between cellulose, hemicelluloses, and lignin components of the lignocellulosic polymer was typically disrupted by the pretreatment process [61]. Generally, the alkaline treatments are known to separate lignin from hemicellulose via OH radicals and dissolve hemicellulose and cellulose through the removal of hydrogen bonds between them, which in turn increases the degradation rate of the hemicellulose and lignin–complexes [62, 63]. This is well in agreement with our FTIR studies, where CT-GS showed functional groups with their corresponding biomass component (Fig. 3a). Thus, the main goal of pretreatment was to make the lignocellulosic fibers available for bacterial hydrolysis by increasing the porosity and exposing the cellulosic part for effective bioconversion. In other words, the morphological changes to CT-GS revealed the structure with increased exterior surface area to improve the enzymatic hydrolysis of the CT-GS by the consortium.

4.3 Effect of biomass loading on reducing sugar production

Biomass loading is one of the important variables that influence the reduction of sugar yield during enzymatic hydrolysis of LCB. The increasing substrate concentration also increases yield and hydrolysis rate; however, high substrate concentrations might hinder the overall hydrolysis process and adversely affect the yield due to feedback inhibition [64, 65]. The reduction in saccharification rate (reducing sugar production) at higher biomass loadings can be attributed to the limited stirring caused by a decrease in the aqueous phase and end product inhibition by the substrate concentration [24, 66].

The results of the current investigation are in congruence with the previous studies [65] that stated the concentration of reducing sugar increases proportionately with substrate loadings of 1 to 3% (w/v). However, the rise in the concentration of reducing sugar was insignificant when the substrate concentration was increased to 4%. Some studies have

shown that beyond certain concentrations and substrate loading, the rate of feedback inhibition by end products like cellobiose and glucose also increases, leading to reduced sugars production [67]. Our observations were in congruence with earlier studies [68, 69]. Since water serves as a reactant in hydrolysis reactions like the breakdown of glycosidic linkages, it is a crucial component for the diffusion of enzymes and sugar production in the lignocellulose bioconversion [70]. In the case of high-solid hydrolysis, lack of free water prevents diffusion of the end-products, while engaging the active sites of the enzymes with preoccupied substrates, thereby causing a strong inhibitory effect [71]. In the present study, after reaching the optimal conditions, reducing sugar yield started to fall after 20 days of incubation. Such effect can be attributed to end product inhibition caused by a high concentration of the product and mass transfer limitations within the reaction mixture due to the high viscosity of the slurry [70, 72].

4.4 Effect of consortia volume on reducing sugar production

The saccharification in terms of the concentration of reducing sugars released increased significantly during the initial period of incubation, being highest on the 14 days of culturing. After 14–24 days, the reducing sugar content was gradually reduced for 3–4% of the consortium. A reduction in the release of reducing sugars after 14 days could have occurred due to the consumption of the carbon source, leading to a lack of the available substrates for bacterial metabolism. In addition, the reduced sugars could be a result of the carbohydrate catabolism of the bacterial cells wherein they release the five- and six-carbon sugars from the cellulose and hemicellulose chains via enzymatic hydrolysis. These pentoses and hexoses are subsequently taken up and metabolized in the Embden-Meyerhof-Parnas pathway into two units of pyruvate, which then undergo oxidation to produce energy for the bacteria besides releasing CO₂ [73]. Furthermore, it is plausible to state that reduction in the number of bacteria during hydrolysis due to limited substrates or completion of the log phase of might have led to reduced sugar productions [74].

4.5 Optimization of bioethanol production

During fermentation, a number of variables influence the growth rate of yeast and ethanol yield [75]. The pH of the culture medium is paramount as it influences bacterial contamination, ethanol production, yeast growth, as well as generation of intermediate products. The observed production of ethanol from sugar hydrolysate was in congruence with previous studies, as yeast survival and development are largely impacted by pH levels between pH 2.75 to 4.25.

Since the ideal pH for fermentation processes of *S. cerevisiae* is 4.0–6.0 [76], our results are well in agreement with other reports that demonstrated bioethanol production at pH 5.5 with various substrates such as pretreated corn cob [77], pineapple peel [78], and *Moringa oleifera* seed husk [79].

As high temperatures are known to denature the tertiary structure of proteins and render them inactive, they become unfavorable for cell development and enzymes that govern microbial activity and, obviously for fermentation processes [80]. Our observations are aligned with other studies that also reported maximum fermentation of *Saccharum spontaneum* grass [81], wheat straw [82], rice straw [83], and bamboo hydrolysates [84] at temperatures between 20 to 33 °C by *S. cerevisiae* [85]. In the present study, the fermentation of CT-GS hydrolysate by *S. cerevisiae* was optimal at 30 °C, which is well in agreement with previous studies. Previously, many researchers have observed higher ethanol yields at 30 °C from the sugar hydrolysate of pretreated sugarcane leaves [86], wheat bran [87], and wheat straw [82] by using the SHF process.

In the context of the biorefinery, the bioethanol production obtained from the CT-GS is considerably higher when compared to the fermentation of wheat bran by *S. cerevisiae* [87]. Our results complement the observation reported by Farkas and colleagues, who used 10% yeast and found 4.5% bioethanol production from pretreated wheat bran after 7 days with a microbial consortium. The synthesis of ethanol is directly proportional to the inoculum size of the yeast. In other words, this states that increased cell concentration fastens the fermentation, allowing the cells to grow quickly and convert the available carbohydrates into bioethanol [88].

It could be observed that within first 3 days after inoculation, ethanol production as well as cell biomass showed a significant increase. Therefore, in the ethanol fermentation process, the value of ethanol produced reflects the number of living cell biomass. In addition, microorganism growth is influenced by fermentation duration. Shorter fermentation times result in ineffective fermentation due to insufficient microbial activity, whereas longer durations show harmful effects on microbial development. On the other hand, at lower temperatures, complete fermentation may be accomplished by extending the fermentation duration, resulting in the lowest ethanol production [88].

To date, few studies have investigated the suitability of various LCB substrates such as wheat straw [89], *Moringa oleifera* seeds husk [79], rice straw [42], bamboo [84], sugarcane leaves [86], sugarcane bagasse [90], rice husk [91], barley straw [92], rye straw [92], and Indian bamboo [93] for the production of bioethanol (Table S2). These studies indicate that the amount of sugar generated, the components employed in the fermentation process, and the amount of ethanol produced ultimately depend on the type of LCB used. As mentioned, the current study is in line with earlier

ones because majority of them have shown that the optimum bioethanol production was obtained in the range of 3–4 days of incubation period using sugar hydrolysate of different biomasses [42, 77, 94–96], etc. The high production of bioethanol within 3–4 days could be the result of the dissolved lignin and hemicellulose contents by the pretreatment processes before the enzymatic hydrolysis, leaving the readily usable and accessible sugars available for enzyme action and fermentation. Moreover, a combined pretreatment with dilute alkaline and acid solutions is largely appreciated for decomposition and reduced crystallinity of the lignocellulose besides the removal of enzyme inhibitors [20, 97]. Therefore, from the economic and practical viewpoints, a combined and mild pretreatment of the GS are recommended by the present study to achieve maximum bioethanol potentials through SHF processes.

Since *S. cerevisiae* has several important characteristics, including the capacity for rapid fermentation, genetic stability, resistance to low pH conditions, and tolerance to ethanol, and it is widely employed in many other biotechnological applications [98]. Recently, Douroua and his co-workers extracted and hydrolyzed polysaccharides of pomegranate residues to produce ethanol. The *S. cerevisiae* was reported to convert pomegranate residues derived reducing sugars into 9.7 g/L of bioethanol at pH 5.0 using 2.8 g/L of cell dry mass by solid-state fermentations [99]. On the contrary, the current study reports $3.59 \pm 0.85\%$ (v/v) of maximum bioethanol production at pH 5.5 on the by *S. cerevisiae* using 8% of yeast culture/inoculum by SHF process. In another study, oleaginous yeast strains were reported for lipids production using *Eucalyptus globulus* (hardwood), *Pinus radiata* (softwood), and rice hull hydrolysate [100, 101] as promising feedstocks for biodiesel production. The authors further observed that the yeast, *Meyerozyma guilliermondii* was able to consume total sugar within 50 to 90 h [101]; however, sugars present in rice hulls hydrolysate had a higher inhibitory effect on the fungal growth [100]. The fact that lignin and hemicellulose were dissolved by pretreatment procedures prior to enzymatic hydrolysis of CT-GS may have contributed to the high production of bioethanol in just 3–4 days by making the easily utilizable and accessible sugars available for fermentation and enzyme activities. Therefore, we state that the cellulolytic bacterial mediated degradation of CT-GS presents an economical and sustainable based strategy for the production of value-added products.

4.6 Characterization of bioethanol

In order to ascertain the potential products generated during the SHF process, we used GC–MS/MS that revealed the dominance of ethanol, which was in congruence with an agreement with a recent study [102]. Furthermore, FTIR was employed to assess the existence of functional groups

of bioethanol obtained after the fermentation process. The FTIR spectrum showed that the characteristic peaks observed correspond to the alcohol group. The presence of the methyl groups was supported by the fact that alkaline pretreatments of LCB increase the production of methane in the fermentation media [62, 103]. The main advantage of SHF is characterized by two distinct processes: enzymatic hydrolysis of cellulose into fermentable sugars and the fermentation of the released sugars to produce bioethanol. In SHF, both hydrolysis and fermentation processes are performed at optimum conditions, which helps to achieve maximum yield of the products, besides avoiding the inhibitory action among the involved microbial species. Therefore, higher hydrolysis yields of fermentable sugars and better fermentation product yields often represent the outcomes of these independent optimization processes. Correspondingly, we report the successful production of bioethanol through SHF processes added by mild alkali-acid pretreatments of the GS to obtain higher ethanol concentration. Hence, this study showcased the utilization of straw-based agricultural wastes as a promising and renewable resource to produce bioethanol through a sustainable and green chemistry approach.

5 Conclusion

A successful and sustainable lignocellulosic biorefinery demands technological breakthroughs to address challenges like the requirement of multienzyme cocktails and the removal of lignin-blocking compounds. The primary shortcomings of existing microbial pretreatment techniques for lignocellulose materials include slow response, speed, and difficulties in practical application due to the recalcitrance imposed by lignin contents. In this study, the combination of biological and chemical methods proved significant for the efficient degradation of LCB and its subsequent fermentation into ethanol. A coupling effect can be produced by alkali-acid pretreatment and microbial consortia degradation. For GS, a combined treatment approach followed by consortia-mediated enzymatic hydrolysis proved to be an efficient method to produce higher concentrations of reducing sugars that can be fermented into bioethanol via separate saccharification and fermentation strategies using *S. cerevisiae* as the fermentation catalyst. The changes in the operational pH, temperature, and inoculum size caused changes in ethanol production, signifying that optimization is important for higher yields. In conclusion, the combined chemical pretreatment enhanced the GS saccharification and fermentation, suggesting a potential strategy for higher ethanol yields. However, a thorough economic and process design is necessary to establish an industrially appropriate production strategy that will alleviate our energy problems by generating more ethanol in a steady manner.

Supplementary Information The online version contains supplementary material available at <https://doi.org/10.1007/s13399-024-05289-8>.

Acknowledgements This research study was carried out as a part of the DBT BULDER-SUK programme sanctioned to Shivaji, University, Kolhapur, MS, India.

Author contribution Priyadarshani S. Sadalage: project administration, methodology, investigation, validation, visualization, writing—original draft and editing. Mudasir A. Dar: formal analysis, data curation, writing—review and editing. Ana Cláudia Paiva-Santos: formal analysis. Kiran D. Pawar: conceptualization, supervision, validation and writing—review and editing.

Data availability All data generated or analyzed during this study are included in this published article [and its supplementary information files].

Declarations

Ethics approval This study does not contain any experiments with human participants or animals.

Consent for publication All the authors agree with consent to publication, and the current article does not contain data from any individual person.

Conflict of interest The authors declare no competing interests.

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