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Isolation and identifcation of thermotolerant yeast strains producing bioethanol from agro‑food wastes

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Abstract

Bioethanol production from agro-food industry waste is one of the suitable alternatives to fossil fuels. In this study, six agro-food wastes were evaluated to select those with a high concentration of starch or fermentable sugars for bioethanol processing. Yeast's ability to produce biofuels by fermentation is afected by temperature. In this study, thermotolerant yeasts, obtained from natural sources in Algeria (soil and agro-food waste), were selected for their capacity to produce and tolerate bioethanol. Three wastes were selected, two of which had signifcant starch content. In order to get fermentable sugars, these wastes underwent chemical and enzymatic hydrolysis. The selected yeasts were identifed by morphological, physiological, biochemical, and molecular characterization. Enzymatic and acid hydrolysis of whole potato and durum wheat bran released (190 g/L and 130 g/L) and (138.40 g/L and 90.03 g/L) reducing sugars, respectively. Among the isolated strains, three were found to be able to produce bioethanol, namely Candida tropicalis, Candida glabrata, and Saccharomyces cerevisiae. These strains were identical to those stored in the data bank with 99%, 100%, and 100%, respectively. In addition, *C. glabrata* and *C. tropicalis* exhibited an ethanol tolerance of up to 14%, while *S. cerevisiae* tolerates up to 15%. Interestingly, enzymatic hydrolysis–treated potatoes produced a considerable amount of bioethanol after 48 h of fermentation by *S. cerevisiae* (7.525% (v/v)), *C. glabrata* (6.80% (v/v)), and *C. tropicalis* (4.50% (v/v)). Taken together, our fndings suggest that *S. cerevisiae* and whole potato waste could be considered good candidates for industrial bioethanol production at high temperatures.

Keywords Agro-food waste · Thermotolerant yeast strains · Bioethanol · Saccharomyces cerevisiae · Candida tropicalis · Candida glabrata

1 Introduction

World demand for energy is continuing to rise signifcantly as a result of the rapid growth in the human population, the expansion of agriculture and industry, and the growth of the transportation industry. Thus, the increment in the consumption of fuels derived from petrochemicals is becoming a serious problem, particularly in terms of air pollution, which has an impact on both human health and global warming. Therefore, it is deemed necessary to fnd

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alternatives for these fuels. In this context, producing energy from agricultural and agro-industrial activity wastes could be a potential solution. Indeed, these wastes are rich in renewable organic matter (biomass), and they make it possible to reduce dependence on oil and thus the environment pollution [\[1\]](#page-13-0). In light of all this, particular attention must be given to the better management of organic waste and particularly the by-products from agricultural residues. Hence, their use as a raw material for bioenergy production could make a good alternative to fossil fuels. Bioethanol can be produced by renewable natural resources such as plants containing sucrose (beets, sugar cane, etc.) or starch (wheat, corn, potatoes, etc.); accordingly, it represents a promising source of energy. The production of biofuel is thought to be infuenced by temperature, one of the factors that greatly afect yeast fermentation ability. The temperature in Algeria rises considerably and tends to increase with global warming, especially during the summer when average temperatures vary between 32 and 44 °C [[2\]](#page-13-1). Therefore,

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the selection of thermotolerant yeast strains is crucial to deal with climate change. Importantly, high-temperature fermentation has several benefts, such as lowering the cost of tank cooling, obtaining higher saccharifcation yields, and minimizing bacterial contamination [[3](#page-13-2)]. Heretofore, plant crops were devoted to produce ethanol, which has drawn criticism because it competes with its use for human and animal food in countries such as the USA and Brazil [\[4](#page-13-3)]. The production of bioethanol from sources other than those used for human or animal consumption has attracted considerable interest in scientifc studies [\[3](#page-13-2), [5,](#page-13-4) [6](#page-13-5)]. According to Tse et al. [\[7](#page-13-6)], agricultural and agro-food industry wastes present an opportunity to produce ethanol due to the abundant and the available supplies, biodegradability, high carbon and nutrient content, and their utility in managing the economic and the environmental issues of the wastes from industry [\[5](#page-13-4)]. In Algeria, food industries have generated a huge amount of starchy wastes such as durum wheat bran, potato, rinse water, cutting water, quenching water, and process water. These wastes are produced annually at 60 million tons, 10 million tons, 3.8 million m³, 3.3 million m³, 1.5 million m³, and $58,000 \text{ m}^3$, respectively. In this study, we sought to isolate and identify locally thermotolerant yeast strains from the soil of Bechar region as well as agro-food wastes. Moreover, we aimed to examine their ethanolic fermentation on some selected wastes at high temperature, in order to contribute to the recovery of agro-food wastes.

2 Materials and methods

2.1 Agro‑food waste selection

There are several raw materials used for the production of bioethanol. In this study, six wastes were chosen based on their abundance and availability (Table [1\)](#page-1-0). Except for whole potatoes provided by Chipsou factory in Rouiba, Algeria, all other waste products were used directly. The potatoes were cleaned using tap water and sliced into small pieces measuring 2 cm in length, 2 cm in width, and 4 mm in thickness using an electric slicer. The slices were dried in an oven for 9–10 h at a temperature of 60 °C. The

resulting powder was obtained through grinding and sieving with a 500-μm sieve. Finally, the powder was packaged in polyethylene bags and stored in a refrigerator at 4 °C until it was analyzed.

2.2 Physicochemical and biochemical analysis of agro‑food wastes

The physicochemical properties of agro-food wastes were analyzed to assess their potential for bioethanol production and biotransformation. AFNOR methods were used to measure the pH, total solids, ash, and fat contents of all the collected agro-food wastes [\[8](#page-13-7)]. The electrical conductivity at 20 °C was measured using a Jenway 4520 conductivity meter, as stated by Rodier [[9](#page-13-8)], while total nitrogen and protein amounts were determined using the Kjeldahl method [[8](#page-13-7)]. Also, the estimation of reducing sugars was conducted employing the 3,5-dinitrosalycilic acid (DNS) method, which was introduced by Miller [\[10](#page-13-9)]. In addition, glucose was measured enzymatically through the glucose oxidase–peroxidase GODPOD method developed by Trinder [\[11\]](#page-13-10). The starch content was carried out using a colorimetric method described by Uarrota et al. [\[12](#page-13-11)] with some modifcations.

Three milliliters of the starch solution was placed in a test tube, and 1 mL of an I/KI solution was added. The resulting mixture was then assessed for the intensity of its blue color using a spectrophotometer set at 600 nm against a reagent blank; the amount of starch present in the sample was calculated according to a standard curve prepared in the range of 10–50 mg of soluble starch per milliliter (mL), treated in the same manner as the test sample. Additionally, to investigate the chemical oxygen demand (COD) of both liquid and solid samples, the protocols recommended by Rodier [\[9\]](#page-13-8) and Noguerol–Arias et al. [\[13](#page-13-12)] were followed, respectively. Biological oxygen demand after 5 days (BOD5) of both liquid and solid samples was performed using methods described by Rodier [[9\]](#page-13-8) and Kolář et al. [\[14\]](#page-14-0), respectively. Mineral salts such as Na⁺, Ca²⁺, K⁺, Zn²⁺, and Mg²⁺ were quantified using atomic absorption spectrophotometry [[8\]](#page-13-7). The phosphorus was measured by the colorimetric method [\[15\]](#page-14-1), and chlorides were evaluated by titrimetry [[8\]](#page-13-7).

Table 1 Diferent sources of agro-food wastes

2.3 Isolation, selection, and identifcation of thermotolerant yeast strains producing bioethanol

2.3.1 Isolation of the thermotolerant yeasts

Various yeast strains were obtained from diverse sources, including soil and agro-food waste. Soil samples were gathered from two distinct regions, namely Bechar and Jijel, while agro-food waste samples were obtained from two local factories, which were TAFNA starch plant (Maghnia, Algeria) and EURL MARAVILLA (Chipsou) (Rouiba, Algeria). Approximately 100 g of soil was placed in a sterile plastic bag, while liquid waste was put into plastic bottles. The solid sample mainly potato was crushed and homogenized in a sterile mortar. One gram of substrate was dissolved in a sterile sodium citrate solution (2% w/v) previously heated to 45 °C in order to soften it, dissolve its constituents, and release the microbial cells [[16](#page-14-2)]. The yeasts were isolated on yeast peptone glucose (YPG) agar medium containing 10 g/L of yeast extract, 5 g/L peptone, 20 g/L glucose, 20 g/L agar, and 0.05 g/L chloramphenicol. The diferent cultures were isolated by conventional methods of decimal dilutions [\[17](#page-14-3)]. The inoculation was done on the surface by spreading 0.1 mL of each dilution in transverse streaks. The dishes were then incubated at 37 °C for 72 h, and pure cultures were stored at 4 °C.

2.3.2 The selection of thermotolerant yeasts

The isolated yeast strains were purifed and selected based on their ability to produce and tolerate high concentrations of bioethanol. A colony from a Petri dish was transferred to glass tubes containing 10 mL of YPG broth and incubated at 37 °C for 24 h. Five milliliters of the obtained culture was transferred into a 100-mL Erlenmeyer fask containing 20 mL of YPG broth and was further incubated at 37 °C for 18 h with stirring at 150 rpm [\[18](#page-14-4)]. The resulting active cells were centrifuged at 8000 rpm for 10 min, rinsed twice with sterile distilled water, and served as inoculum. The alcoholic fermentation was carried out in 250-mL Erlenmeyer fasks containing 100 mL of sterile synthetic YPG broth at pH 5 and was incubated at 37 °C for 48 h with the agitation of 150 rpm. The effects of temperature (37 \degree C, 40 \degree C, and 45 \degree C), pH (4, 5, and 6), and substrate concentration (glucose) (10%, 15%, 20%, 25%, and 30% w/v) were studied on bioethanol production. The ethanol content was determined at the end of each fermentation.

2.3.3 Ethanol tolerance test

Yeast suspensions were transferred into glass tubes containing 10 mL of YGP broth mixed with a variable volume of absolute ethanol in order to obtain diferent concentrations (0%, 5%, 10%, 11%, 12%, 13%, 14%, 15%, 16%, and 20% (v/v)) [[6,](#page-13-5) [19,](#page-14-5) [20](#page-14-6)]. The tubes were then incubated at 37 $^{\circ}$ C for 48 h. The cell concentration in each tube was measured at the beginning and the end of fermentation by determining the optical density at 600 nm, which is proportional to the number of cells [[21\]](#page-14-7).

2.3.4 Identifcation of strains producing bioethanol

The yeast strains were identified through conventional techniques outlined by Kurtzman and Fell [[22](#page-14-8)] according to their morphological, physiological, and biochemical characteristics. The cell morphology is examined using an optical microscope (MOTIC B3-220-PL). After incubation of yeasts for 48 h on YPG agar medium at 37 °C, a single colony of yeast was mixed in a droplet of sterile distilled water on glass slide and smeared until the smear dry off. The smear was then stained using diluted methylene blue dye, air dried, and observed under light microscope at magnification \times 100. Additionally, White et al.'s [\[23\]](#page-14-9) molecular approach was employed for further identifcation. The QIA fast R genomic extraction kit was frstly used to extract DNA from the isolated yeast strains. Then, polymerase chain reaction (PCR) was performed for the internal transcribed spacer (ITS) region of 26S rDNA using the following primers: ITS1 5′-TCCGTAGGTGAACCTGCGG-3′ and ITS4 5'-TCCTCCGCTTATTGATATGC-3'. The D₁/D₂ domain of the 26S rDNA was also amplifed by PCR using the following primers: LROR 5′ACCCGCTGAACTTAAGC3′ and LR6 5′CGCCAGTTCTGCTTACC3′. The amplifcation of DNA and sequencing were done in a fnal volume of 50 μL reaction volume (10 mM Tris HCl, 50 mM KCl, 1.5 mM MgCl2, 0.2 mM dNTPs, 1.25 IU Taq polymerase, 0.2 mM each primer, and 0.2 μL of DNA). The PCR was conducted in a DNA thermal cycler (Biometra, Germany). The reaction underwent 35 cycles, and the resulting products, along with their restriction fragments, were detected by electrophoresis on a 1.5% agarose gel. Furthermore, the obtained sequences were matched with those available in the National Center for Biotechnology and Information's (NCBI) GenBank using the Basic Local Alignment Search Tool (BLAST). The highest alignment score was then used to determine the species. Additionally, the neighbor-joining method was used to construct a phylogenetic tree, utilizing Mega 6 software, as described by Altschul et al. [[24\]](#page-14-10).

2.4 Production of bioethanol on selected agro‑food wastes

Three agro-food wastes were selected for bioethanol production based on their sugar content. Except quenching water, the potato powder and the durum wheat bran require

a pre-treatment by hydrolysis of starch to fermentable sugar mainly glucose which will be fermented into ethanol by alcoholic fermentation.

2.4.1 Starch hydrolysis in selected agro‑food wastes

The hydrolysis of agro-food waste was done by acid and enzymatic hydrolysis. The waste acid hydrolysis was performed as described by Taibi et al. $[25]$ with some adjustments. Fifty grams of both potato powder and durum wheat bran was suspended at 35% (w/v) of hydrochloric acid solution (1 M HCl) which was heated at 100 °C for 2 h. After cooling in the open air, the substrate was recovered by fltration. pH was adjusted using a solution of 5 M NaOH until it was equal to 5 in order to eliminate the fermentation inhibitors [\[26](#page-14-12)]. The enzymatic hydrolysis process consisted of two stages: liquefaction and saccharifcation. According to Taibi et al. [[26](#page-14-12)], liquefaction involved mixing 50 g of substrate with a phosphate buffer solution (Weight ratio 1:3) and 5 mL of 0.15% (w/v) alpha-amylase (30 U/mg) and heating the mixture at 100 °C for 10 min, followed by a 2-h liquefaction step at 90 °C and pH 6. The resulting solution was centrifuged, and the supernatant was analyzed for reducing sugars according to the method of Miller [\[10\]](#page-13-9). After liquefaction, the mixture pH was adjusted to 4.8 with 1 M HCl and cooled to 60 °C. Twenty milliliters of *Aspergillus niger* amyloglucosidase (120 U/mg) was added, and saccharifcation was carried out for 4 h at 60 °C. The enzyme was then inactivated by increasing the temperature to 100 °C, and samples were analyzed for reducing sugars as described above [\[26](#page-14-12)].

2.4.2 Fermentation conditions

Each pure isolated yeast was transferred from an agar culture colony to a glass tube containing 10 mL of YPG broth and incubated at 37 °C for 24 h. Then, 5 mL of this culture was added to 10 mL of sterilized YPG medium and stirred at 150 rpm for 18 h at 37 °C [[18](#page-14-4)]. Active cells were centrifuged at 8000 rpm for 10 min, washed twice with sterile distilled water, and used as an inoculum. The fermentations were carried out in Erlenmeyer flasks containing 100 mL of sterile fermentation medium with 10% (v/v) inoculum size, incubated at 37 °C for 48 h under agitation speed of 150 rpm. All experiments were conducted in triplicate.

2.4.3 Analytical methods

Cell growth was determined by measuring optical density at 600 nm with a UV–Visible Spectrophotometer Varian 50 Tablet, and cell numeration was carried out using a Thoma hemocytometer [\[27\]](#page-14-13). A standard absorbance curve was established as a function of the cell population number. The ethanol concentration expressed in alcoholic degree $\%$ (v/v) was measured using an ebulliometer of the Dujardin– Salleron type (Paris, France) based on the variances in boiling points between water and alcohol, as indicated by a reference table [28]. Moreover, the concentration of ethanol (g/L) was measured using headspace GC–FID to establish a calibration curve between the ethanol concentration and the alcoholic degree.

Ethanol productivity (Qp, g/L/h) was calculated using the following equation:

$$
Qp = P/t
$$

where P is the ethanol concentration (g/L) and t is the fermentation time (h) giving the highest ethanol concentration.

Ethanol yield (Yp/s, g/g) was calculated as the actual produced ethanol and expressed as g ethanol per g of utilized sugar.

The ethanol fermentation efficiency (Ey, $\%$) was calculated by the following equation:

 $Ey = (Yp/s/0.511) \times 100$

where Yp/s is the ethanol yield (g/g) and 0.511 is the theoretical maximum ethanol yield per unit of glucose from glycolytic fermentation (g/g).

2.5 Statistical analysis

All analyses were performed in triplicate, and results were expressed as average values \pm standard deviation. The data were analyzed for statistical signifcance using one-way and two-way analysis of variance (ANOVA), followed by Duncan's multiple range test ($p < 0.05$). The statistical software package XLSTAT v. 14 was used in the analysis of the experimental data.

3 Results and discussion

3.1 Physicochemical and biochemical characterization of agro‑food wastes

Based on the data presented in Table [2,](#page-4-0) various nutrient compounds in all the tested agro-food wastes showed diversity and richness. The variation in nutrient content may be linked to the origin and initial composition of the used material during the production process, as stated by Slopiecka et al. [[29](#page-14-15)]. Bioethanol is generated by the fermentation of simple sugars present in biomass as well as sugars resulting from a prior treatment by enzymatic or chemical hydrolysis of agro-food wastes containing high starch content [\[30](#page-14-16)]. In the present study**,** only three wastes among the tested agrofood wastes were selected as potential raw materials for bioethanol production regarding their important starch and sugar contents, namely quenching water with 50 g/L of sugar, whole potatoes with 60% of starch, and durum wheat bran with 41.50% of starch. These high amounts of starch and

sugar in these three wastes are generally linked to the initial composition of starchy raw materials (potatoes, wheat, and corn) used during the production process and its operating conditions [\[31,](#page-14-17) [32](#page-14-18)]. The high initial sugar concentration is an important infuencing parameter which has a direct efect on yeast cell growth and alcoholic fermentation rate [[6,](#page-13-5) [7](#page-13-6)]. Furthermore, selected agro-food wastes have a slightly acidic pH and a total nitrogen concentration greater than 0.50% (w/w), corresponding to a protein concentration of more than 4% (w/w). The availability of nitrogen and proteins in the culture medium has a direct efect on the yeast growth. Insufficient amounts of these nutrients can impede or reduce the speed of alcoholic fermentation, resulting in the formation of undesirable sensory compounds, such as sulfur derivatives [\[33\]](#page-14-19). The agro-food waste contains low fat content which varies from 0.4 to 3.65%. This variation depends on the type of agro-food waste and its constituents [\[29](#page-14-15)].

The presence of fatty acids in the fermentation medium increases the tolerance of yeasts to ethanol, which leads to a high production of ethanol [[34\]](#page-14-20). Additionally, our findings reported that biological oxygen demand concentrations over 5 days (BOD₅) vary from 903.80 to 7522 mg/L, while chemical oxygen demand concentrations (COD) present values between 907 and 14225 mg/L. The COD/BOD₅ ratio ranges from 1.53 to 2.48, demonstrating that COD values are higher than $BOD₅$ values in all tested samples. In general, these two parameters vary considerably depending on the type and characteristics of the waste [[35\]](#page-14-21). The high values of COD and BOD5 in our agro-food waste constitute a crucial environmental pollution factor which afects on the one hand the physical and chemical structure of the soil, and on the other hand, it reduces the aquatic life by increasing the concentration of dissolved oxygen [[36](#page-14-22)]. Furthermore, a high electrical conductivity was observed,

varying from 0.74 to 11.70 mS/cm depending on the presence of ions, their total concentration, their mobility, and their valence, as well as the temperature of the solution; these data demonstrate the abundance and richness of these discharges in various mineral salts [[37\]](#page-14-23). These results were confrmed by atomic absorption spectrometry, and a diversity of mineral elements was revealed in the samples, including Na, Ca, K, Zn, Mg, and P, which vary in the range of 0.01 (Zn) to 2.12 (K) g/100g. Indeed, these micronutrients are crucial for microbial development and the synthesis of bioethanol [[38](#page-14-24)]. They play an important role in the fermentation process and afect the ethanol yield [\[39\]](#page-14-25). Several studies have been carried out to highlight the importance of each element. Jones and Greenfield [[40\]](#page-14-26) cited that sodium (Na) improves the sugar assimilation by contributing to a good alcoholic fermentation, while magnesium (Mg) increases yeast tolerance to ethanol [\[41](#page-14-27)]. In addition, it has been demonstrated that magnesium in the fermentation medium afects cell growth, glucose consumption, and ethanol production; on the other hand, lack of magnesium triggers stress responses and reduces the expression of genes involved in energy metabolism [[42](#page-14-28)]. In the presence of zinc (Zn) , Ismail et al. $[43]$ $[43]$ $[43]$ have also noted a considerable improvement in ethanol tolerance. According to Alminderej et al. [\[44](#page-14-30)], calcium (Ca) enhances bioethanol production and yeast tolerance to high ethanol concentrations. Also, phosphorus can be provided to yeasts in the form of phosphate salts, which are required for the creation of adenosine triphosphate (ATP), phospholipids, and nucleic acids [[45\]](#page-14-31). According to Manuel Fernando et al [\[46](#page-14-32)], the addition of phosphate salt increased the ATP, which improved the efficiency of the fermentation process. Thus, the greatest improvement in ethanol production derived specifically from the increase in K^+ concentration and the reduction in acidity of the fermentation medium [[47](#page-14-33)]. In the same line, our fndings have proven that these wastes constitute attractive and nutrient-rich raw materials for yeast growth and biofuel production.

3.2 Isolation of yeast strains producing bioethanol

A total of thirty thermotolerant yeast strains were isolated from diferent natural sources (Table [3](#page-6-0)). Six of which were able to produce bioethanol on YPG broth at 37 °C. Strains Q1 and Q2 were isolated from quenching water, C1 from cutting water, P1 and P2 from whole potato, and B2 from the soil of Bechar. Our results are in line with previous studies, in which thermotolerant yeasts were able to grow and produce ethanol from 37 \degree C to 45 \degree C [[48–](#page-15-0)[50\]](#page-15-1) and even up to 50 °C [[51\]](#page-15-2). Consequently, the six isolates that produced ethanol at 37 °C were chosen to select the best-performing yeast strains in ethanol production and tolerance.

3.3 Selection of thermotolerant yeast strains producing bioethanol

The effects of temperature, pH, substrate concentration, and ethanol tolerance were studied in order to select the most efficient yeast strains that are capable of producing and tolerating large amounts of ethanol in the YPG broth.

3.3.1 Efect of the substrate concentration

Figure [1](#page-6-1) demonstrates that all isolated yeast strains tolerate a sugar concentration up to 25% (w/v), at which point strains Q1, B1, P1, C1, P2, and Q2 produce ethanol the most, with maximum values of 11.45%, 9.15%, 8.35%, 7.05%, 5.45%, and 3.80% (v/v), respectively. Moreover, the data suggests that as the substrate concentration increases, the ethanol production yield also increases until it reaches a maximum yield at a concentration of 25% (w/v). Our results are in concordance with those reported by Lin et al. [[27\]](#page-14-13); their data showed that the ethanol yield produced by *S. cerevisiae* starts to decline above a substrate concentration of 30%. Similarly, Osho [[21\]](#page-14-7) and Triwahyuni et al. [\[52](#page-15-3)] have conducted studies showing that high sugar concentrations (25% and 30% (w/v)) in the fermentation environment inhibit cell growth and ethanol production. At a concentration of 40% (w/v), cell growth is completely halted as water infltrates through the cytoplasmic membrane of the cells to balance intra- and extracellular concentrations. This decrease in production yield decrement could be attributed to the high concentration of sugars which creates osmotic stress in the yeasts [[53](#page-15-4)].

3.3.2 Efect of temperature

According to thermodynamic laws, temperature has an impact on biological processes. Indeed, yeasts can only work in a narrow range of "optimal" temperatures, and they cannot survive in a critical temperature, like all other living organisms [[54](#page-15-5)]. Figure [2](#page-7-0) shows that the optimum temperature for ethanol production seems to be 37 °C for all the tested yeast strains with maximum ethanol contents of 11.45%, 9.15%, 8.35%, 7.10%, 5.45%, and 3.80% (v/v) corresponding to strains Q1, B1, P1, C1, P2, and Q2, respectively. In addition, based on the aforementioned graph, we noted that any deviation from the optimum temperature, either increment or decrement, results in a decline in ethanol production yield for all the tested isolates. Eventually, at a temperature of 45 °C, ethanol production stopped entirely for all strains except for strain B1, which managed to produce a minimal amount of bioethanol at a concentration of 0.5% (v/v). This decrement in production can be justifed by the denaturation of the tertiary structure and inactivation of enzymes that regulate microbial activity and the fermentation process [[55](#page-15-6)]. Our fndings are in agreement with those demonstrated by

Fig. 1 Bioethanol production using the six isolated yeast strains at diferent substrate concentrations. Values represent the mean of three replicates and error bars represent the standard deviation

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Nuanpeng et al. [\[56](#page-15-7)] with a temperature of 37 °C as optimal for bioethanol production using *S. cerevisiae* DBKKUY–5. A number of publications have proven that temperature increment enhances the inhibiting effect of ethanol on cellular activities by lowering yield cell and ethanol production [[57,](#page-15-8) [58](#page-15-9)]. However, a very high temperature causes many changes in the cell that ultimately afect protein structures and function, accumulate denatured and aggregated biomacromolecules, and lead to growth inhibition or cell death [[59\]](#page-15-10). Zhang et al. [[60](#page-15-11)] showed that high temperature can not only trigger high DNA damage of *S. cerevisiae* but also trigger other biological abnormalities including loss of cell wall integrity. Moreover, Duhan et al. [[61](#page-15-12)] have reported a temperature of 35 °C as optimal for bioethanol production on potatoes using *S. cerevisiae* MTCC–170, whereas

Fig. 3 Bioethanol production using the six isolated yeast strains at diferent pH. Values represent the mean of three replicates, and error bars represent the standard deviation

Ünal et al. [[62\]](#page-15-13) have noted higher ethanol yield at 30 °C using *S. cerevisiae*. Generally, ethanol production depends on fermentation temperature which profoundly afects yeast growth and fermentation performance [\[63](#page-15-14)].

3.3.3 Efect of pH

pH plays an essential role in the production of bioethanol. Figure [3](#page-7-1) illustrates that the diference in pH afects the amount of the produced bioethanol. pH 5 gave the highest ethanol content for all six isolated yeast strains where they have maximum ethanol contents of 11.35%, 9.05%, 8.40%, 7.00%, 5.45%, and 3.80% (v/v) corresponding to strains Q1, B1, P1, C1, P2, and Q2, respectively. According to Darvishi and Abolhasan [[64](#page-15-15)], an increase in the rate of

ethanol production at pH 5 was noted. These results are in agreement with our fnding as well as those obtained by Narendranath and Power [[65\]](#page-15-16), and they stated that yeast is an acidophilic organism and grows in an acidic environment. Because acidic conditions promote greater yeast development and inhibit the growth of harmful bacteria, alcohol production in acidic environments is important [\[27\]](#page-14-13). A pH between 4 and 6 is considered as optimum pH where the yeast can grow better, depending to several parameters such as the temperature, the amount of oxygen, and the yeast strain. Moreover, it has been noted that an increase or a decrease in the pH on either side of the optimum value for all the yeast isolates leads automatically to a decrease in the ethanol production yield. In addition, this inappropriate pH also has a negative efect on yeast cells because it can afect the structure of the cell wall and modify the conformation of proteins protruding from the plasma membrane; at the same time, it also impacts the organization and function of cell membrane lipids and disrupts the function of proteins embedded in membranes [\[66\]](#page-15-17). Narendranath and Power [\[65\]](#page-15-16) have reported that the extracellular pH changes from the optimum pH may be too difficult for the cell to maintain a constant intracellular pH and the enzymes may not function normally. Also, if the enzymes are disabled, the yeast cell will not be able to grow and produce ethanol efficiently. Additionally, when the pH is lower than 4 or higher than 6, a more extended incubation period is required, and ethanol concentration decreases considerably [[67\]](#page-15-18). It has been also signaled that more acidic and basic media retard yeast metabolic pathways and inhibit cell growth, resulting in a lower ethanol yield [\[68\]](#page-15-19).

3.4 Ethanol tolerance

Figure [4](#page-8-0) illustrates the tolerance of several yeast strains to diferent ethanol concentrations. A heterogeneity of ethanol resistance levels was reported in the diferent species. Strain Q2 seems to be less resistant compared to the other tested strains. Indeed, the Q2 strain has the capacity to withstand ethanol concentrations up to 11% (v/v). Above this value, only the other strains can survive. The strains C1 and P2 tolerate a concentration up to 12% (v/v) compared to 14% (v/v) for the strains P1 and B1, while the strain Q1 has the highest ethanol resistance compared to other strains (15%). This variability in ethanol tolerance could be explained by the variation of the strain's biochemical composition. Numerous studies demonstrated the alteration of cellular lipid composition in response to ethanol exposure [[69,](#page-15-20) [70](#page-15-21)]. Furthermore, You et al. [\[69](#page-15-20)] and Archana et al. [[70](#page-15-21)] have reported that unsaturated fatty acids such as palmitoleic and oleic acids play an important role in overcoming the toxic efects of ethanol in *S. cerevisiae* cells, and when it is exposed to ethanol stress, the proportion of monounsaturated fatty acids increases with a concomitant decrease in saturated fatty acids. This unsaturated fatty acid composition could be a useful criterion to assess the potential ethanol tolerance of other microorganisms. Similarly, Boudjema et al. [[71](#page-15-22)] have advocated a range of ethanol tolerance between 11 and 14% (v/v) for their isolated yeasts. In the same line, Nguyen et al. [[72\]](#page-15-23) have mentioned a total cell growth inhibition when the ethanol concentration is between 11.8 and 13% (w/v) for four diferent strains (*C. tropicalis* OP010088, *Meyerozyma guilliermondii* OP010064, *C. tropicalis* OP010090, and *M. guilliermondii* OP010065). On the other hand, yeasts generally cannot tolerate an ethanol concentration superior than 16%. Yet, several studies by Hawaz et al. [[6\]](#page-13-5), Urano et al. [\[19](#page-14-5)], and Moneke et al. [\[20](#page-14-6)] have succeeded to isolate yeast strains supporting up to 20% ethanol. This variation in tolerance could be attributed to the diference in genetic makeup of the yeast isolates and their ecological origin [\[6](#page-13-5)]. The ethanol resistance mechanism involves many interactions between signal transduction pathways and regulatory networks, as well as a complex network at the genomic level. Transcription dynamics and profling results of important

Fig. 4 Ethanol tolerance of isolated yeast strains after 48 h of incubation at 37 °C. For each yeast strain, diferent letters indicate signifcant diferences $(p < 0.05)$ between the strains

gene sets, such as heat shock proteins, provided an understanding of the tolerance mechanism [[73\]](#page-15-24). According to our results, all selected yeast strains can produce bioethanol at signifcantly variable levels. Based on the one-way ANOVA statistical analysis of the data presented in Fig. [4](#page-8-0), the tolerance to ethanol is signifcantly diferent according to the yeast strains tested. The Q1, P1, and B1 strains are signifcantly distinguished from the other strains by their tolerance to high amounts of ethanol. Table [4](#page-9-0) shows the efficiency of bioethanol production by the six isolates at the diferent tested parameters (temperature, pH, and substrate concentration). It has been shown that the concentrations of bioethanol produced at all parameters are signifcantly diferent (*p* < 0.05) depending on the tested yeast strains. However, among these tested isolates, Q1, B1, and P1 strains produce signifcantly higher concentrations of bioethanol than the other isolates at all tested parameters. Thus, these three strains could be good candidates for the bioethanol production from agro-food wastes and are selected for further study.

3.5 Identifcation of yeast isolates

The isolated strains were confrmed molecularly by comparing the obtained PCR-amplifed sequences with those stored in GenBank of the National Center for Biotechnology and Information (NCBI) using the BLAST algorithm. Phylogenetic analysis of the isolated strains revealed that they belong to two diferent genera: Saccharomyces and *Candida*. Strain Q1 is closely related to *S. cerevisiae* CBS 2888 (KY109257.1) with similarity of 100% (Fig. [6\)](#page-10-0), whereas strains P1 and B1 seem close to *C. tropicalis* L2 (MK752673.1) and *C. glabrata* CBS 5278 (KY106475.1), respectively, with similarities of 99% and 100% (Fig. [7](#page-10-1)). Finally, these identifed yeast strains are named *S. cerevisiae* Z0730–8635I (Q1), *C. tropicalis* Z0730–8636I (P1), and *C. glabrata* Z0730–8637VS (B1), and they are integrated in GenBank with the following accession numbers: OP876823, OP876824, and OP876825, respectively. The primary identifcation of yeast isolates is based on the morphological characteristics of colonies grown on solid medium. Table [5](#page-10-2) reveals the presence of white, rounded, creamy, smooth, domed, and buttery texture colonies, showing pseudomycelium development only in P1 strain. Furthermore, microscopic examination demonstrated that all strain isolates have ovoid cell shapes with budding asexual reproduction (Fig. [5\)](#page-10-3).

The morphological identification of the strains was followed by physiological and biochemical characterization (Table [6](#page-11-0)). These results indicate a variation in sugars fermentation and assimilation from one strain to another. Interestingly, all our isolates were able to use glucose during their growth. Moreover, trehalose was also efficiently assimilated by all strains, whereas it was fermented only by P1 and B1 strains. On the other hand, all the isolates present a negative test for lactose and starch. Furthermore, only the P1 isolate showed an ability to assimilate xylose and adonitol which are not assimilable by the others, while unlike the other strains, the Q1 isolate was able to use rafnose. The selected yeast isolates seem to belong to two diferent genera of yeast such as Saccharomyces and *Candida*, based on comparisons between all of our fndings and the characteristics listed by Kurtzman and Fell [\[22](#page-14-8)], Boudjema et al. [[71](#page-15-22)], Silva et al. [\[74](#page-15-25)], and Zhang et al. [[75\]](#page-15-26). Indeed, isolate B1 was found to have the characteristics of *C. glabrata*. Interestingly, as stated

Ethanol concentration ($\%$ v/v) at diferent Isolate Q1 Q2 C1 P1 P2 B1 Temperature (°C) 25 8.16 ± 0.10^a 2.00 ± 0.08^f 3.95 ± 0.04^d 5.20 ± 0.16^c 3.20 ± 0.14^e 6.30 ± 0.00^b 30 10.20 ± 0.08^a 2.90 ± 0.08^f 6.60 ± 0.12^d 7.33 ± 0.23^c 4.70 ± 0.08^e 8.05 ± 0.00^b 37 11.45 \pm 0.07^a 3.80 \pm 0.16^f 7.10 \pm 0.08^d 8.35 \pm 0.22^c 5.45 \pm 0.12^e 9.15 \pm 0.00^b 40 5.60 ± 0.16^a 1.40 ± 0.04^e 3.60 ± 0.00^e 3.70 ± 0.04^e 3.10 ± 0.16^d 4.20 ± 0.12^b 45 $0.00 \pm 0.00^{\circ}$ $0.60 \pm 0.00^{\circ}$ pH 4 10.40 ± 0.00^a 2.10 ± 0.00^e 5.00 ± 0.10^d 6.50 ± 0.00^c 1.20 ± 0.04^f 7.80 ± 0.14^b 5 11.35 \pm 0.04^a 3.80 \pm 0.16^f 7.00 \pm 0.40^d 8.40 \pm 0.04^c 5.45 \pm 0.12^e 9.05 \pm 0.08^b 6 10.80 ± 0.08^a 2.00 ± 0.16^e 5.00 ± 0.04^d 7.60 ± 0.04^c 1.30 ± 0.08^f 8.50 ± 0.08^b Substrate concentration (glucose) (% w/v) 10 5.10 ± 0.08^a 1.20 ± 0.28^d 2.20 ± 0.16^c 2.80 ± 0.04^b 2.90 ± 0.08^b 4.80 ± 0.14^a 15 $6.85 \pm 0.21^{\circ}$ $1.00 \pm 0.08^{\circ}$ $3.00 \pm 0.16^{\circ}$ $4.80 \pm 0.18^{\circ}$ $3.67 \pm 0.23^{\circ}$ $5.80 \pm 0.16^{\circ}$ 20 9.10 ± 0.14^a 3.10 ± 0.14^f 5.25 ± 0.20^d 6.50 ± 0.12^c 4.50 ± 0.20^e 7.20 ± 0.08^b 25 11.45 ± 0.04^a 3.80 ± 0.16^f 7.05 ± 0.12^d 8.35 ± 0.12^c 5.45 ± 0.08^e 9.15 ± 0.00^b 30 10.80 ± 0.16^a 0.00 ± 0.00^f 3.66 ± 0.12^d 4.30 ± 0.08^c 3.00 ± 0.08^e 6.70 ± 0.12^b

Table 4 Ethanol production efficiency of thermotolerant yeast strains at different parameters

Values bearing diferent superscript letters within the same row are signifcantly diferent based on one-way ANOVA and Duncan's multiple range test at $p < 0.05$. Results are expressed in means \pm standard deviation

(–) Negative test

Fig. 5 Microscopic characteristics of isolated yeast strains by methylene blue staining cultured in YPG agar at 37 °C after 48 h (magnifcation ×100; Q1 and B1: scale bar = $10 \mu m$; P1: scale bar = $15 \mu m$)

 $Q1$

Fig. 6 Phylogenetic tree of strain Q1 based on the neighbor-joining method with a bootstrap analysis of 1000 replications

Fig. 7 Phylogenetic tree of both strains P1 and B1 based on the neighbor-joining method with a bootstrap analysis of 1000 replications

Fig. 8 The radar plot of ethanol yield obtained using three isolated yeast strains on agro-food wastes after acid and enzymatic hydrolysis: **a** whole potato, **b** durum wheat bran

Table 6 Biochemical characteristics of yeast strain isolates

(+) Positive growth reaction, (–) negative growth reaction, (w) weak growth reaction

by Risan [[76\]](#page-15-27), *C. glabrata* does not produce pseudohyphae at temperatures above 37 °C and has smooth, oval, creamcolored colonies. Risan [[76](#page-15-27)] and Devadas [\[77](#page-15-28)] have mentioned that *C. glabrata* only ferments and assimilates glucose and trehalose. In fact, the Q1 isolate was found to have the characteristics of *S. cerevisiae*. Kavanagh [\[78](#page-15-29)] stated that colonies of *S. cerevisiae* are typically creamy with an oval shape. Based on the fndings of Asyikeen et al. [\[79](#page-15-30)], Boudjema [[80\]](#page-15-31), and Moussan et al. [[81](#page-16-0)], isolate Q1 can be identifed as *S. cerevisiae* due to its ability to ferment sucrose, maltose, fructose, glucose, and raffinose but not lactose. This idea is confrmed by El Nemr [[82\]](#page-16-1) who has reported that *S. cerevisiae* cells are unable to ferment lactose due to a lack of lactase or β-galactosidase. Table [5](#page-10-2) reveals that the P1 isolate is highly similar to *C. tropicalis*, which is consistent with the fndings of other researchers. For instance, Ebabhi et al. [\[83](#page-16-2)] and Maikan et al. [[84](#page-16-3)] successfully isolated *C. tropicalis* strains from local food samples, which exhibited creamy, fat, and smooth colony characteristics. Similarly, Boudjema et al. [[71](#page-15-22)] and Shariq and Sohail [\[85](#page-16-4)] identifed a yeast strain with ovoid cells, mycelium formation, and fermentation capabilities for glucose, galactose, sucrose, xylose, maltose, and trehalose as *C. tropicalis*.

3.6 Production of bioethanol from selected agro‑food wastes

Table [7](#page-12-0) summarizes the results of bioethanol production from three selected agro-food wastes using the diferent three isolated yeast strains. On the basis of the statistical analysis of the data presented in Table [7,](#page-12-0) it has been shown that the bioethanol production is signifcantly varied according to the tested yeast strains and the selected agro-food wastes. This variation could be related to diferent factors, namely the substrate, the hydrolysis process, and the strain type used in the fermentation process [[67\]](#page-15-18). Bioethanol concentrations

Substrate	Type of hydrolysis	Yeast	Reduc- ing sugar (g/L)	Ethanol concentration		Ethanol	Ethanol yield	Fermentation
				$(\% \text{ v/v})$	(g/L)	productivity (g/L/h)	(g/g)	efficiency $(\%)$
Whole potato	Acid liquefac- tion	S. cerevisiae	130.00	6.60 ± 0.03 ^{ca}	52.60 ± 0.04 ^{ca}	1.09 ± 0.05 ^{ca}	0.40 ± 0.10^a	79.18 ± 0.01^a
		C. tropicalis	130.00	$3.10 \pm 0.06^{\circ}$	$25.00 \pm 0.05^{\circ}$	0.52 ± 0.02 ^c	0.19 ± 0.02 ^{ab}	37.63 ± 0.03^{ab}
		C. glabrata	130.00	$4.40 \pm 0.07^{\text{cb}}$	$35.25 \pm 0.07^{\text{cb}}$	$0.73 \pm 0.04^{\text{cb}}$	0.27 ± 0.11^a	$53.06 \pm 0.02^{\text{a}}$
	Enzymatic hydrolysis	S. cerevisiae	190.00	7.85 ± 0.10^a	62.46 ± 0.11^a	1.30 ± 0.01^a	$0.33 \pm 0.04^{\text{a}}$	64.33 ± 0.06^a
		C. tropicalis 190.00			$4.50 \pm 0.09^{\text{ac}}$ 36.04 \pm 0.09 ^{ec}	$0.75 + 0.07^{\text{ac}}$	0.19 ± 0.03^{ab}	37.12 ± 0.08^{ab}
		C. glabrata	190.00	$6.80 + 0.02^{ab}$	$54.17 + 0.02^{ab}$	$1.13 + 0.12^{ab}$	$0.29 + 0.10^a$	$55.79 \pm 0.13^{\circ}$
Durum wheat bran	Acid liquefac- tion	S. cerevisiae	90.03	3.05 ± 0.05 ^{da}	24.61 ± 0.04 ^{da} 0.51 ± 0.01 ^{da}		0.27 ± 0.12^a	53.49 ± 0.03^a
		C. tropicalis	90.03		$2.75 \pm 0.14^{\text{dc}}$ $22.24 \pm 0.13^{\text{dc}}$ $0.46 \pm 0.03^{\text{dc}}$		0.25 ± 0.01^{ab}	48.34 ± 0.06^{ab}
		C. glabrata	90.03		$3.18 \pm 0.08^{\text{db}}$ $25.63 \pm 0.08^{\text{db}}$ $0.53 \pm 0.07^{\text{db}}$		$0.28 + 0.09^a$	$55.71 \pm 0.02^{\text{a}}$
	Enzymatic hydrolysis	S. cerevisiae	138.40		6.70 ± 0.02^{ba} 53.39 \pm 0.02 ^{ba} 1.11 \pm 0.06 ^{ba}		$0.39 \pm 0.15^{\text{a}}$	75.49 ± 0.01^a
		C. tropicalis 138.40			$3.55 \pm 0.11^{\text{bc}}$ $28.55 \pm 0.09^{\text{bc}}$ $0.59 \pm 0.14^{\text{bc}}$		0.21 ± 0.01^{ab}	40.36 ± 0.07 ^{ab}
		C. glabrata	138.40		$5.90 \pm 0.15^{\circ}$ 47.08 \pm 0.03 ^b	$0.98 + 0.04^b$	$0.34 \pm 0.05^{\text{a}}$	$66.57 \pm 0.05^{\text{a}}$
Quenching water		S. cerevisiae	51.31		$1.70 \pm 0.06^{\text{ea}}$ 13.97 \pm 0.20 ^{ea}	$0.29 + 0.02^{\text{ea}}$	$0.27 \pm 0.02^{\text{a}}$	53.28 ± 0.13^a
		C. tropicalis 51.31		$1.30 \pm 0.10^{\circ}$	$10.81 \pm 0.01^{\circ c}$	$0.22 \pm 0.15^{\circ}$	0.21 ± 0.10^{ab}	41.23 ± 0.10^{ab}
		C. glabrata	51.31		$1.50 \pm 0.11^{\text{eb}}$ $12.39 \pm 0.08^{\text{eb}}$ $0.26 \pm 0.14^{\text{eb}}$		$0.24 + 0.09^a$	$47.26 \pm 0.03^{\text{a}}$

Table 7 Ethanol yield obtained using three isolated yeast strains on agro-food wastes after acid and enzymatic hydrolysis

Values bearing diferent superscript letters within the same column are signifcantly diferent based on two-way ANOVA and Duncan's multiple range test at $(p < 0.05)$. Results are expressed in means \pm standard deviation

and their productivity values vary signifcantly from 10.81 to 62.46 g/L and from 0.19 to 1.30 g/L/h, respectively. Additionally, from Table [7](#page-12-0) and Fig. [8,](#page-11-1) the highest ethanol concentrations are produced by *S. cerevisiae* and *C. glabrata*, at 7.85% and 6.80% (v/v), respectively; these results are in agreement with those cited by Techaparin et al. [[58\]](#page-15-9) which showed that the thermotolerant *S. cerevisiae* KKU–VN8 strain could produce the maximum amount of ethanol at a temperature of 37 °C from a nutritional substrate containing 15% (w/v) of glucose.

Unlike *S. cerevisiae*, several *Candida* strains are recognized for their inability to grow sufficiently and produce ethanol under anaerobic conditions [[86\]](#page-16-5). Importantly, our results indicate that *C. glabrata* was able to produce ethanol more efficiently and grow under anaerobic conditions and that its fermentation efficiency was not significantly different from that of *S. cerevisiae* ($p < 0.05$).

With regard to *S. cerevisiae*, the obtained results have demonstrated that this strain presents a signifcantly high yield (*p* < 0.05) compared to *C. tropicalis* and a signifcantly better ethanol productivity ($p < 0.05$) than the other tested strains (Table [7,](#page-12-0) Fig. [8\)](#page-11-1). These results support the widespread use of *S. cerevisiae* as the most efective starter in the industrial production of bioethanol [\[57\]](#page-15-8) because it tolerates high ethanol concentrations and a wide pH range which makes the process less susceptible to infections [[27,](#page-14-13) [67](#page-15-18), [87](#page-16-6)]. On the other hand, the *C. tropicalis* strain has demonstrated its ability to produce ethanol from agro-food waste, but at a lower concentration comparing to the other two strains. These results are in concordance with those cited by Mussato et al. [[88\]](#page-16-7), who

have obtained an ethanol yield of 5.45% (v/v) by *C. tropicalis* and 14.8 (v/v) by *S. cerevisiae*. Whole potatoes and durum wheat bran were used as raw materials, and they underwent acid and enzymatic hydrolysis to release fermentable sugars into the medium and avoid slow progress of fermentation due to insufficient intake of fermentable carbohydrates for yeast metabolism [\[7](#page-13-6)]. The amounts of bioethanol produced from the two agro-food wastes after acid and enzymatic hydrolysis ranged from 22.24 to 52.60 g/L and from 28.55 to 62.46 g/L, respectively. The initial concentration of substrate seems to have an impact on the production capacity, which explains the high yield of bioethanol production from potatoes in both processes. In addition, the statistical analysis shows that the fermentation of the enzymatic hydrolysates produces a higher bioethanol yield than that of the two tested residual acid hydrolysates. This diference could be attributed to the high amount of fermentable sugars that are released during the enzymatic hydrolysis comparing to acid hydrolysis which is characterized by releasing other inhibitor substances such as organic acids and furaldehydes (furfural and 5-hydroxymethyl furfural) [[89\]](#page-16-8). Despite their low concentrations, these substances are able to alter the metabolism, lengthen the lag phase, damage cell membranes, and acidify the cytoplasm, which results in the reduction of ethanol production and productivity [\[90\]](#page-16-9). On the other hand, enzymatic hydrolysis already has high yields of simple sugars (85–95%), generates few phenolic compounds and effluents to be treated, and does not cause corrosion problems [\[91\]](#page-16-10). Regarding our results, the additional amylases were crucial for the alcoholic fermentation of the three tested yeast strains. This was demonstrated

in the previous study of Vincent et al. [[92\]](#page-16-11) by the fact that *C. glabrata* did not produce ethanol from starch in the absence of these enzymes. The fermentation of the quenching water has been carried out. This contained 51.31 g/L of reducing sugars, and the production of bioethanol using the three tested yeast strains was between 10.81 and 13.97 g/L. Although this waste did not require any treatment contrary to other tested wastes, the rate of bioethanol production remains low. This could be explained by the low concentration of fermentable sugars that are released in this waste, which leads to low bioethanol production [[93\]](#page-16-12). Several studies have been conducted on the production of bioethanol from diferent substrates including sugarcane, sugar beet, and sweet sorghum. The most common starches are corn and wheat [\[3](#page-13-2), [44](#page-14-30), [62](#page-15-13)]. The variations of ethanol yields change considerably from one substrate to another, mainly due to differences in the conversion efficiency of the diferent raw materials (acid hydrolysis and enzymatic hydrolysis), the composition of the culture medium, the fermentation process, and the used yeast strain whether commercially or isolated locally.

4 Conclusion

In the study, thirty strains of yeast were isolated from two Algerian natural sources (soil and agro-food waste). Among them, three thermotolerant yeast strains were successfully selected, specifcally *S. cerevisiae* Q1, *C. tropicalis* P1, and *C. glabrata* B1. The strains, isolated from three agro-food wastes, were chosen for their ability to produce and to tolerate ethanol from three agro-food wastes. Interestingly, tolerance levels have been observed at high temperatures up to 40 °C and at ethanol concentrations up to 16% (v/v). The ethanol production capacities of these yeast strains were maximal at 37 °C. Moreover, whole potato wastes and durum wheat bran underwent acid and enzymatic hydrolysis of starch in order to release fermentable sugars. According to the results, the enzymatic hydrolysis made it possible to have a better yield of sugars, in particular from potato (190 g/L). The highest ethanol concentrations are obtained after 48 h of fermentation as follows: 7.525% (v/v), 6.80% (v/v), and 4.50% (v/v) on potatoes treated by hydrolysis enzymatic by *S. cerevisiae*, *C. glabrata*, and *C. tropicalis*, respectively. Taken together, our results provide important information regarding *S. cerevisiae* and whole potato waste for bioethanol production and represent a promising starting point for improving high-temperature industrial ethanol production processes in the near future.

Author contributions Younes Gherbi and Khaled Boudjema designed the study. Younes Gherbi performed the data collection and analysis. Younes Gherbi and Khaled Boudjema analyzed the data and wrote the manuscript. All the authors commented on previous versions of the manuscript. All the authors read and approved the manuscript.

Data availability Not applicable.

Declarations

Ethics approval This declaration is not applicable to this manuscript.

Competing interests The authors declare no competing interests.

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