



# Saccharification of water hyacinth biomass by a combination of steam explosion with enzymatic technologies for bioethanol production

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## Abstract

In the present work, bioethanol was produced by sugar fermentation obtained from water hyacinth using a novelty hybrid method composed of steam explosion and enzymatic hydrolysis, using hydrolytic enzymes produced by solid-state fermentation and water hyacinth as substrate. The highest activity, 42 U for xylanase and 2 U for cellulase per gram of dry matter, respectively, was obtained. Steam explosion pretreatment was performed at 190 °C for 1, 5, and 10 min, using water hyacinth sampled from the Maria Lizamba Lagoon, the Arroyo Hondo and the Amapa River. The highest amounts of reducing sugars of water hyacinth were obtained from the samples from the lagoon (5.4 g/50 g of dry matter) after 10 min of treatment. Steamed biomass was hydrolysed using the enzymes obtained by solid-state fermentation, obtained reducing sugars (maximum 15.5 g/L); the efficiency of enzymatic hydrolysis was 0.51 g of reducing sugars per gram of water hyacinth. Finally, reducing sugars were fermented using *Saccharomyces cerevisiae* for conversion to ethanol, with the highest ethanol concentration (7.13 g/L) and an ethanol yield of 0.23 g/g of dry matter.

**Keywords** Lignocellulosic biomass · Solid-state fermentation · Fermentable sugars · Hydrolytic enzymes · Alcoholic fermentation

## Introduction

Water hyacinth (WH) is an important aquatic plant high in hemicellulose and low in lignin content. It contains about 48% hemicellulose, 18% cellulose and 3% lignin, although the reported composition varies (Aswathy et al. 2010). The species is an important source of fermentable sugars for bioethanol production to substitute fossil fuels, and its residual biomass can also be converted into other value-added chemicals in a well-integrated biorefinery facility. Its further advantages are that it does not compete with food crops for arable land, proliferates in clear water and wastewater and is highly reproducible (Singh and Bishnoi 2013; Yan et al. 2015). However, the production of bioethanol from WH requires the recovery of fermentable sugars using pretreatments that destroy covalent bonds between hemicellulose and lignin but not between cellulose and hemicellulose within the complex physical mixture of lignocellulose (Sun et al. 2016). Hemicellulose and lignin are strongly intertwined and linked by covalent bonds that hinder saccharification, presenting a bottleneck for enzymatic hydrolysis or fermentation (Gütsch et al. 2012). Also, there are technical

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challenges in the pretreatment of biomass and in enzymatic saccharification, especially the sourcing of enzymes (Aswathy et al. 2010).

Regular ethanol production from lignocellulosic biomass is realised by three major steps: pretreatment to disrupt the recalcitrant structures and to facilitate polysaccharide accessibility due to the increase of the surface area, enhancing accessibility for enzymatic attacks in the hydrolysis step. Enzymatic hydrolysis is then performed to hydrolyse the polysaccharides into fermentable sugars, followed by their fermentation into bioethanol. Hydrothermal pretreatment as a steam explosion has been used as a promising method to enhance cellulose availability and hemicellulose recovery without requiring any chemicals; as an environmentally friendly method, it only uses compressed hot water as a solvent and maintains cellulose and hemicellulose availability for the enzymes, resulting in lower downstream detoxification costs compared to other pretreatment techniques (Batista et al. 2019; Pratto et al. 2020).

The steam explosion has been classified as a green and competitive technology as it only contains lignocellulosic feedstock and water, preventing corrosion problems and the formation of neutralisation sludge; this pretreatment has successfully been used in delignification and the removal of hemicellulose (Ibrahim et al. 2010; Oliveira et al. 2013; Martin-Sampedro et al. 2014a), facilitating hemicellulose removal and lignin transformation because of the increase in the surface area for cellulose hydrolysis (Singh et al. 2015). Hence, de-lignification can substantially improve biomass enzymatic saccharification. Pretreatment of WH via steam explosion is an effective delignification strategy (Das et al. 2015). Ferro et al. (2015) indicate that pretreatment enhances enzymatic accessibility of cellulose and increases the level of saccharification. In this context, we evaluated the steam explosion for subsequent enzymatic hydrolysis using hydrolytic enzymes produced by solid-state fermentation (SSF) for the recovery of fermentable sugars for bioethanol production, using *Saccharomyces cerevisiae*.

## Methods

### Microorganisms and inoculum

*Trichoderma harzianum* PBLA (Lopez-Ramirez et al. 2018) was used as inoculum for xylanase and cellulase production by SSF. *Saccharomyces cerevisiae* was used as inoculum in alcoholic fermentation. Both strains were provided by Plant Pilot of Solid-State Fermentation of the Autonomous Metropolitan University, Mexico. *Trichoderma harzianum* PBLA was grown in 250-mL Erlenmeyer flasks with potato dextrose agar (PDA) (BIOXON, Mexico) for seven days at 30 °C. Generated spores were inoculated in Erlenmeyer

flasks containing liquid YPD medium (20 g L<sup>-1</sup> of yeast extract, 20 g L<sup>-1</sup> of peptone, and 40 g L<sup>-1</sup> of glucose) and maintained at 30 °C for 24 h; mycelial growth was considered as inoculum in SSF.

*Saccharomyces cerevisiae* was propagated in culture medium with glucose (50 g L<sup>-1</sup>), K<sub>2</sub>HPO<sub>4</sub> (5 g L<sup>-1</sup>), (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (2 g L<sup>-1</sup>), MgSO<sub>4</sub>·7H<sub>2</sub>O (0.4 g L<sup>-1</sup>) and yeast extract (1 g L<sup>-1</sup>) and was maintained at 30 °C for 36 h. Biomass was used as inoculum for ethanol production. Both strains were conserved in distilled water.

### Sampling and collection of water hyacinth

The sampling area for WH (*Eichornia crassipes*) was defined on the Papaloapan Hydrological Basin reported by the National Commission for the Knowledge and Use of Biodiversity (CONABIO 2016). Samples were obtained from the Maria Lizamba Lagoon located at 18° 29' N 70° and 96° 01' 42" W, the Amapa River located at 18° 18' 88" N and 96° 18' 19" W, and the Arroyo Hondo River located at 18° 27' 31" N and 96° 20' 28" W.

### Water hyacinth conditioning

Once sampled, WH was washed to remove impurities. Subsequently, leaves, stem, and root were separated, followed by drying in the sun for approximately 60 h at an average temperature of 35 ± 2 °C. Finally, the dried biomass was cut into pieces of uniform size (1 cm) and stored until use as a substrate in SSF and steam explosion.

### Solid-state fermentation

Water hyacinth (from the Maria Lizamba Lagoon) was used to support hydrolytic enzyme production (xylanase and cellulase) in packed bed columns (2.5 cm in diameter and 20 cm long) with oxygen supply by forced aeration. Prior to SSF, WH was impregnated (pretreated) with H<sub>2</sub>SO<sub>4</sub> (0.25 M), homogenised and sterilised in an autoclave at 120 °C for 15 min. After sterilization, WH was impregnated with a culture medium containing the following macronutrients (g L<sup>-1</sup>): glucose (50), KH<sub>2</sub>PO<sub>4</sub> (5), NH<sub>4</sub>NO<sub>3</sub> (5), Urea (2), MgSO<sub>4</sub>·7H<sub>2</sub>O (0.42), CaCl<sub>2</sub> (1), NaCl (5) and peptone (5), as well as 1 mL of micronutrients containing (g/100 mL): FeSO<sub>4</sub>·7H<sub>2</sub>O (0.5), MnSO<sub>4</sub>·7H<sub>2</sub>O (0.61), ZnSO<sub>4</sub>·7H<sub>2</sub>O (0.1) and CoCl<sub>2</sub>·H<sub>2</sub>O (0.036) (Mekala et al. 2008). The inoculum was adjusted to 2 × 10<sup>7</sup> spores/mL at 65% of initial moisture (quantified in a gravimetric balance Ohaus, Model MB45), and the initial pH value was adjusted at 5.5. Columns were packed with the inoculated mixture, oxygen was supplied with water-saturated air at a flow rate of 50 mL/min, and the packed columns were maintained at 30 °C. The concentration of produced CO<sub>2</sub> (respirometry)

was measured online connecting the outflow air to a gas analyser (Ávila-Cisneros et al. 2014).

### Enzymatic extract

The enzyme extracts consisted of the recuperated mixed fermented medium from SSF with distilled water in a ratio of 1:10 (w/v), vortexed for 1 min (Lopez-Ramirez et al. 2018). The extract was separated by centrifugation at 10,000 rpm for 15 min at 4 °C. The supernatant was decanted and used as an enzyme source for xylanase and cellulase assays.

### Xylanase and cellulase activity

Xylanase activity was determined by mixing 0.1 mL of the enzymatic extract with 0.9 mL of 0.25% (w/v) birchwood xylan solution in 0.1 M sodium citrate buffer pH 5.2. The mixture was incubated at 40 °C for 15 min. The enzymatic reaction was stopped by adding 1.5 mL of 3,5-dinitrosalicylic acid (DNS); reducing sugars were estimated by the DNS method (Miller 1959) using xylose as standard. Cellulase activity was determined by mixing 0.1 mL of the enzymatic extract with 0.9 mL of 0.25% (w/v) solution of carboxymethyl cellulose in 0.1 M sodium citrate buffer pH 5.2; the mixture was incubated at 40 °C for 30 min. The enzymatic reaction was stopped by the addition of 1.5 mL of DNS; reducing sugars were estimated by the DNS method (Miller et al. 1960) using glucose as standard.

One enzymatic unit (U) was defined as the amount of enzyme required to liberate 1 µmol of reducing sugars from the substrate per minute. Enzyme activities are reported as units per gram of dry matter (U/g dm). All assays were performed in duplicate.

### Steam explosion

Pretreatment of WH was carried out using the emerging technology (steam explosion) at 190 °C with retention times of 1, 5, and 10 min. Batch processing was used for 50 g of WH samples.

### Enzymatic hydrolysis of water hyacinth pretreated with steam explosion

Enzymatic saccharification was carried out by mixing 20 g of exploited WH and 200 mL of liquid generated from the steam explosion with the enzymatic extract (cellulase and xylanase), adjusting the enzymatic activity in the reaction at 36 U/g of pretreated WH. The enzymatic reaction was performed in 500-mL Erlenmeyer flasks with gentle agitation (125 rpm) in a water bath; the pH was adjusted to 5.2 with 0.1 M citrate buffer and maintained at 50 °C for 48 h. Samples were taken every 12 h, and the release of reducing

sugars was monitored using the DNS method (Miller et al. 1960).

### Ethanol production using water hyacinth hydrolysed by a combination of methods

Ethanol production was carried out using the product of WH pretreated by steam explosion combined with the enzymatic method as a single carbon source. Alcoholic fermentation was performed by adding the culture medium containing yeast extract 1 g L<sup>-1</sup>, NH<sub>4</sub>SO<sub>4</sub> 2 g L<sup>-1</sup> and MgSO<sub>4</sub>·7H<sub>2</sub>O; inoculum concentration was adjusted at 1 × 10<sup>6</sup> cellules/mL and maintained for 72 h at 30 °C with agitation at 125 rpm. After fermentation, the broth was centrifuged at 10,000 rpm for 15 min and the obtained supernatant was filtered through 0.45-µm filters and used for ethanol quantification.

### Ethanol quantification

The ethanol content was analysed via the spectrometric method for ethanol quantification (Isarankura-Na-Ayudhya et al. 2007), using 0.1 M solution of potassium dichromate in 5 M sulphuric acid. The reaction was performed by mixing 300 µL of alcoholic samples with 3 mL of dichromate solution, maintained at room temperature for 30 min. Absorbance was measured at 590 nm, and ethanol concentration was determined using a standard curve of ethanol.

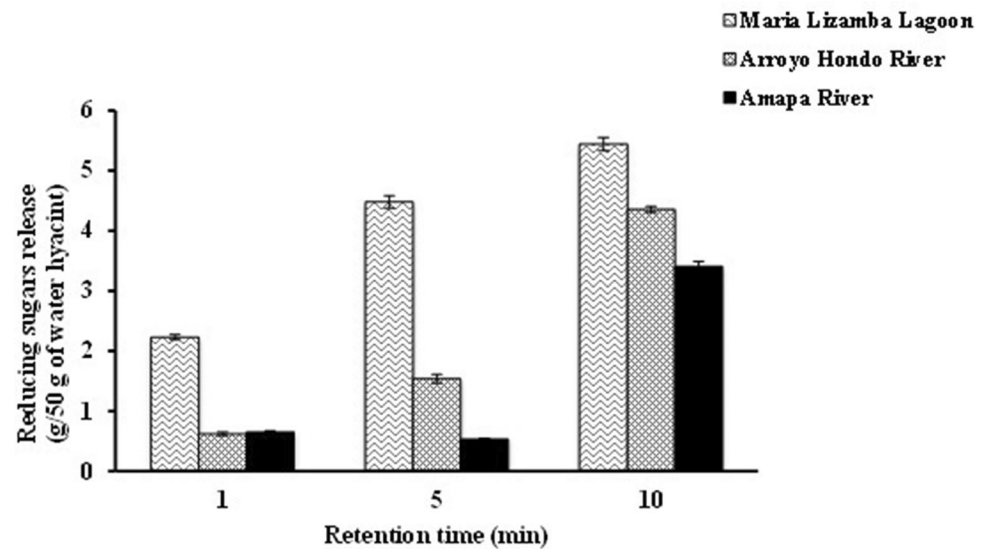
## Results and discussion

### Steam explosion

Results of the steam explosion of samples of WH obtained from the Maria Lizamba Lagoon, the Arroyo Hondo River and the Amapa River are shown in Fig. 1. Based on these results, the time of the steam explosion treatment is important, and maximum quantities of reducing sugars from WH sampled from all three sites were obtained after 10 min (maximum time assayed in this research) at 190 °C. It should be noted that the highest level of reducing sugars (fermentable sugars) was obtained from WH sampled from the Maria Lizamba Lagoon (5.4 g/50 g of water hyacinth), in contrast to the results obtained for the Arroyo Hondo River (4.3 g/50 g of water hyacinth) and the Amapa River (3.4 g/50 g of water hyacinth). Tukey's test showed a significant difference ( $p > 0.05$ ) between reducing sugar levels, and for this reason, WH from the Maria Lizamba Lagoon was used for hydrolytic enzyme production by SSF.

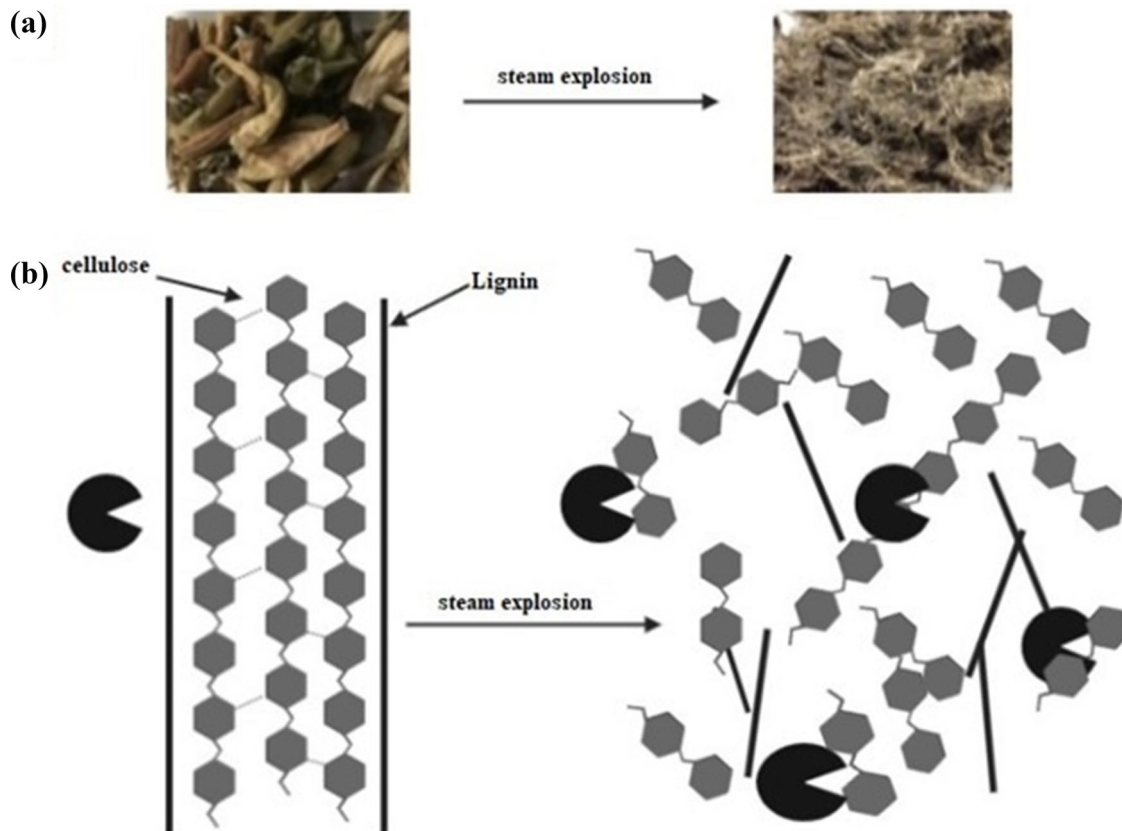
Steam explosion is classified as a green technology due to the reaction medium, which contains only lignocellulosic feedstock and water; this pretreatment has effectively been used in delignification and hemicellulose removal

**Fig. 1** Release of reducing sugars from samples of water hyacinth pretreated by steam explosion at 190 °C and different treatment times



(Martin-Sampedro et al. 2014a, b; Ibrahim et al. 2011; Oliveira et al. 2013) through lignin transformation and increases the surface area for cellulose hydrolysis (Singh et al. 2015). In our study, WH was obviously hydrolysed

and delignified after pretreatment by steam explosion, as can be seen in Fig. 1 (release of fermentable sugars) and Fig. 2a. Our results were similar to those found by Oliveira et al. (2013), who indicated that the steam explosion



**Fig. 2** Effect of steam explosion on lignin present in water hyacinth **a** Water hyacinth from the Maria Lizamba lagoon pretreated by steam explosion, **b** Schematic representation of the steam explosion effect on water hyacinth delignification

degraded lignin (Fig. 2b) or transformed/hydrolysed it by the explosion-produced rapid decompression and high temperatures to hemicelluloses (auto-hydrolysis) and free mono- and oligosaccharides. The sugar yields obtained in these work were lower than those obtained in other works, such a 19.7 g/100 g of dry olive leaves at 180 °C for 10 min (Romero-Garcia et al. 2016) and 46% of glucose after of two cycles of the steam explosion at 183 °C for 5 min for *Eucalyptus globulus* (Martin-Sampedro et al. 2014b). Similar results have been obtained for rice husk by Piñeros-Castro et al. (2011) at 190 °C for 10 min. However, it is important to mention that the lignocellulosic raw materials pretreated with steam explosion were different from those obtained from WH, since this is the first report.

On the other hand, it is important to note that Piñeros-Castro et al. (2011) indicated that in the steam explosion process, the amounts of furfural and hydroxymethylfurfural increase at exposure times of more than 10 min. The presence of these compounds decreases both the specific growth rate and ethanol production in alcoholic fermentation (Taherzadeh et al. 1999); similar results have been observed by Sun et al. (2016). For this reason, no longer periods were assayed in the present work.

### Hydrolytic enzyme production by SSF

The results of hydrolytic enzyme production in packed bed columns using *T. harzianum* PBLA as inoculum and WH from the María Lizamba Lagoon as substrate are shown in Figs. 3 and 4. The highest xylanase activity was obtained after 96 h (42 U/g of dry matter) of cultivation in SSF. However, after 72 h, a first activity peak occurred (40 U/g of dry matter). Maximum cellulase activity was observed after 68 h of cultivation (2 U/g of dry matter). These results indicate that the time required for maximum xylanolytic and cellulolytic activity was shorter than that for the production of hydrolytic enzymes by SSF, using the same support and fungus, reported by Arana-Cuenca et al. (2019); in their study,

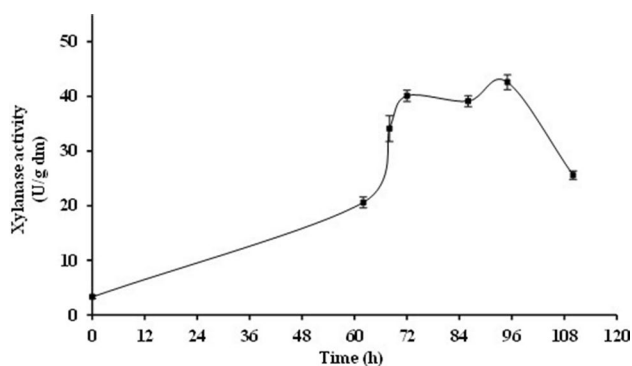


Fig. 3 Xylanase production profil by *T. harzianum* PBL4 during SSF

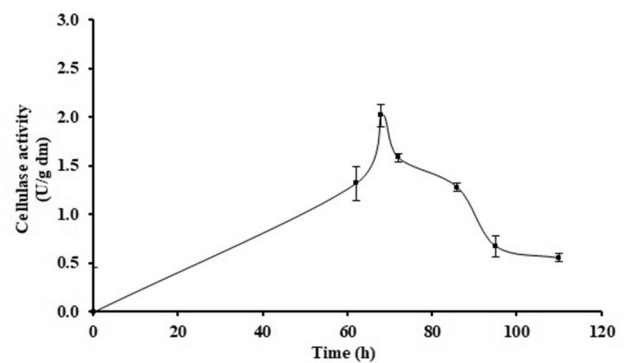


Fig. 4 Cellulase production profil by *T. harzianum* PBL4 during SSF

the highest xylanolytic and cellulolytic activity was observed after 108 h of cultivation, similar to the results reported by Lopez-Ramirez et al. (2018) (48 h) under similar conditions. The time needed to obtain the highest enzymatic activity in this work was shorter than that reported in other studies that used the same substrate in SSF. In a study by Deshpande et al. (2008), the time required to obtain the highest cellulase and xylanase activity using other supports and fungi was 9 days, while for cellulase activity, Zhao et al. (2011) reported 7 days under optimized conditions. The activity levels found in the present study were lower for cellulase and xylanase activity produced via WH as a substrate in SSF. It should, however, be noted that only a few reports are available on cellulase and xylanase production using WH as a substrate in SSF (Lopez-Ramirez et al. 2018; Arana-Cuenca et al. 2019).

Maximum CO<sub>2</sub> production occurred between 54 and 62 h in all analysed columns in SSF (Fig. 5). The respirometric results showed that xylanase and cellulase production started after reaching the maximum CO<sub>2</sub> production. Respirometric

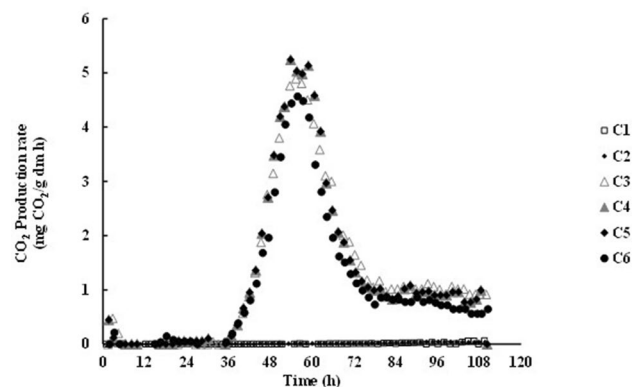


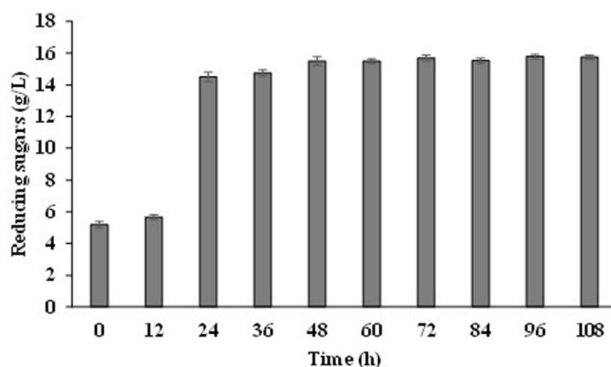
Fig. 5 Profile of CO<sub>2</sub> production rate by *T. harzianum* PBL4 in SSF performed in packed bed columns. C1 and C2 were SSF blanks (without *T. harzianum* PBL4), and C3 to C6 were SSF columns monitored over 110 h

analysis suggests that for the production of the hydrolytic enzyme by *T. harzianum* PBLA in SSF, it is necessary that the maximum rate of CO<sub>2</sub> production has been reached; approximately after 12 h, maximum xylanase (first maximum, Fig. 3) and cellulase activity (Fig. 4) will be obtained, indicating that the SSF must be stopped. Respirometry analysis is a useful tool for the microbial study of physiology and metabolism (Lopez-Ramirez et al. 2018; Pliego-Sandoval et al. 2012; Méndez-González et al. 2020).

### Enzymatic hydrolysis of water hyacinth pretreated with steam explosion

Lignocellulosic material, such as WH, may provide fermentable sugars for ethanol production. However, recalcitrant structures of WH biomass are difficult to convert. For that reason, it is necessary to use combinations of methods, as described by Singh et al. (2015), especially when enzymes are used for cellulose hydrolysis. In this case, pretreatment is required to make cellulose more accessible to the hydrolytic enzymes, facilitating its conversion to glucose (fermentable sugars). Martín-Sampedro et al. (2014b) indicated that the steam explosion facilitates enzymatic saccharification, and for this reason, we used enzymatic hydrolysis as a pretreatment method; additionally, hydrolytic enzymes (cellulases and xylanases) were produced via SSF using WH as a substrate because the specificity by substrates may be greatest.

Enzymatic hydrolysis of WH from the María Lizamba Lagoon pretreated by steam explosion showed that this process improved enzymatic saccharification, as can be seen in the enzymatic kinetics shown in Fig. 6. The results indicate that the maximum reducing sugars were obtained after 48 h of enzymatic reaction (15.5 g/L); however, after 24 h, a level of 14.5 g/L of reducing sugars was obtained. The efficiency of enzymatic hydrolysis was 0.46 and 0.51 g of reducing sugars per gram of WH (dry matter) after 24 and 48 h of the enzymatic reaction, respectively. The enzymatic efficiency

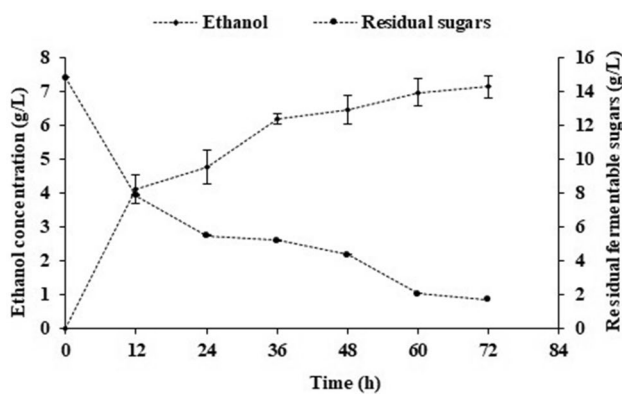


**Fig. 6** Kinetics of enzymatic hydrolysis of Maria Lizamba Lagoon water hyacinth biomass using xylanases and cellulases produced by SSF

obtained in this work is highly important; according to Zhang et al. (2016), the use of enzymes in the saccharification of water hyacinth is larger for ethanol production; however, when the hydrolytic enzymes were produced by SSF from WH (present work), the costs of ethanol production were reduced. Also, the efficiency of saccharification obtained in the present research (0.46 and 0.51) is higher than that obtained when WH was saccharified using acid pretreatment followed by enzymatic hydrolysis using commercial enzymes (0.40 g of reducing sugars per g of WH), even when the enzymatic reaction conditions were optimised (Zhang et al. 2016). The result obtained after 48 h of enzymatic hydrolyses corresponds to the theoretical maximum of reducing sugars from WH (0.51 g/g) reported by Xia et al. (2013). We conclude that the combination of steam explosion pretreatment and enzymatic hydrolysis with WH-produced enzymes is adequate to obtain higher amounts of fermentable sugars. The yields after 24 h of enzymatic reaction were similar to those obtained by Satyanagalakshmi et al. (2011), who reported 0.48 g/g after 24 h of incubation. However, it is important to mention that in the cited works, commercial enzymes for WH hydrolysis were used, and the results obtained by these authors after 48 and 72 h were similar to those obtained after 24 h in the present research. In contrast to previous findings, in our study, the efficiency of enzymatic reaction was superior after 48 h increasing reducing sugars in 5% compared to the results obtained after 24 h of the enzymatic reaction. The yields of reducing sugars were lower in our study than in the studies by Aswathy et al. (2010) (0.73 g/g) and Sukumaran et al. (2009) (0.71 g/g), who used enzymes produced by SSF, and in Das et al. (2015) (0.567 g/g), who used commercial enzymes. The amount of fermentable sugars was higher than that reported in other works, probably because enzyme production using WH as support by SSF increases the specificity for the same substrate. This is the first report that used enzymes produced by SSF, using the substrate for saccharification.

Comparing the results on fermentable sugars obtained from WH saccharified by steam explosion followed by enzymatic hydrolysis with those found for other lignocellulosic material, such as sugar cane bagasse (Bunterngsook et al. 2018; with 0.79 g/g), our results were lower. However, it should be noted that we used recombinant enzymes; however, our yield was higher than that reported for saccharified wheat bran, where a yield of reducing sugars of 0.19 g/g was obtained (Jiang and Guo 2016). It is important to mention that the yield of fermentable sugars depends on the recalcitrant strength of the substrate to be saccharified and on the amount of fermentable sugars they contain.

On the other hand, as mentioned by Singh et al. (2015) and Martín-Sampedro et al. (2014b), the steam explosion improved enzymatic saccharification. After the steam explosion, cellulose was more accessible to the enzymes.



**Fig. 7** Ethanol production by *Saccharomyces cerevisiae* using reducing sugars obtained from water hyacinth enzymatic saccharification

We used WH as the source of cellulases and xylanases for SSF to reduce ethanol production costs. This is the first report on this combination of pretreatments for saccharification, namely the steam explosion followed by enzymatic hydrolysis.

## Ethanol production

To verify the effect of pretreatments on ethanol production, reducing sugars obtained after enzymatic hydrolysis of WH were used as carbon sources for alcoholic fermentation by *Saccharomyces cerevisiae*. We obtained the highest ethanol concentration (7.13 g/L) after 72 h of fermentation, with an ethanol yield of 0.23 g/g of dry matter (WH) or 0.46 g/g of reducing sugars with 80% of reducing sugars consumed. The kinetics of ethanol production from WH biomass saccharified with SSF-produced enzymes can be seen in Fig. 7. The ethanol yield was higher than that reported by Ma et al. (2010) (0.192 g/g of dry matter), who used reducing sugars obtained from WH pretreated with acid after enzymatic hydrolysis. Likewise, the volumetric ethanol yield in our study was higher than those of previous studies, e.g. 3.49 g/L (Ganguly et al. 2013), 4.25 g/L, (Aswathy et al. 2010), 9.8 g/L (Singh et al. 2013), while it was lower than several previously found values (10.1 g/L (Mishima et al. 2008), 10.44 g/L (Das et al. 2016), 13.6 g/L (Ganguly et al. 2013)) and comparable with the 0.46 g/g of reducing sugar reported by Kumar et al. (2009), who used different pretreatments for WH saccharification. Ethanol production from cellulosic materials greatly depends on the disruption of the lignocellulosic structure; different pretreatments including acid, alkali, microwave, liquid hot water and compound pretreatments have been used for this purpose. For ethanol production using WH as a substrate, an effective method is an acid pretreatment (Zhang et al. 2016). However, the ethanol content obtained in the present work (7.13 g/L), when steam explosion followed by enzymatic hydrolysis was used, was higher

than that obtained (1.289 g/L) when acid hydrolysis followed by enzymatic hydrolysis was used (Zhang et al. 2016). Our results indicate that combined pretreatment may improve bioethanol production from WH. It is, however, important to indicate that this is the first report on the use of steam explosion and enzymatic saccharification (using enzymes produced in the same substrate). The yield of ethanol may be increased if *S. cerevisiae* is replaced by microorganisms that use hexoses and pentoses since the former can only use hexoses (Das et al. 2015; Ganguly et al. 2013).

Comparing our results with those obtained using other lignocellulosic materials showed that the yields of ethanol obtained in the present research (23%) are higher, even when the same methods of saccharification were used (steam explosion followed by enzyme hydrolysis). However, it should be noted that the enzymes used in previous studies were commercial enzymes. For example, in reports on ethanol production from bagasse of cane, cane leaves and sugarcane straw, ethanol yields were 16.2, 20.3 and 5.7%, respectively, as reported by Mokomele et al. (2018) and Pratto (2020).

In general, the processes for obtaining ethanol from lignocellulosic biomass present technical–economic problems such as energy costs and the costs of the enzymes used, as well as enzyme specificity: in this work, the cost of hydrolytic enzymes was kept low by producing them on the substrate to be hydrolysed, thus solving not only the cost problem, also obtaining highly specific enzymes (Madadi et al. 2017). Regarding the operating costs generated by the steam explosion, as commented by Pratto et al. (2020), the projected operating costs of an economic analysis in an ethanol production process from lignocellulosic biomass are reduced by up to 20%, even using steam explosion. This is because it is a process carried out for only a few seconds to a few minutes, requiring low amounts of energy (Tu et al. 2017; Kumari and Singh 2018; Kucharska et al. 2018).

## Conclusions

The combination of pretreatments proposed in the present investigation allowed better recovery of reducing sugars from WH than other proposed pretreatments. Also, the steam explosion allows higher efficiency of enzymatic hydrolysis by making the substrates more accessible for enzymes. Likewise, the recovered reducing sugars have a higher ethanol conversion rate than reported previously for water hyacinth. In this sense, SSF is an excellent process for hydrolytic enzyme production, reducing the costs of ethanol production.

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**Author contributions** RH conceptualization. LAF and NL and ECV did the analytical work and methodology. RH, LAF and EF Data analysis. MAL validation. RH and MAL helped in preparing and editing the MS.

## Compliance with ethical standards

**Conflict of interest** The authors declare that they have no conflict of interest in the publication.

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