


# Hydrothermal extraction of dietary fiber from pearl millet bran: optimization, physico-chemical, structural and functional characterization

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**Abstract** Pearl millet bran is rich source of dietary fiber and several other bioactive compounds and is an unexploited by-product of millet processing industries. The utilization of pearl millet bran for extraction of dietary fiber can be an effective method for its valorization. Hydrothermal extraction of dietary fiber from pearl millet bran is a simple eco-friendly technique in terms of minimal consumption of toxic solvents, increased extraction yield, high purity and considered as an economically viable technique. In the present investigation, extraction and optimization of dietary fiber from pearl millet bran was performed using hydrothermal technique. The highest yield of dietary fiber (74.5%, w/w) was obtained under optimized conditions of water to solid ratio (20:1), temperature (90 °C) and time (15 min). The extracted dietary fiber from pearl millet bran was further assessed for its physico-chemical, functional and structural properties. The studies of functional and physico-chemical properties presented the water holding capacity (6.50 g/g and 3.99 g/g), swelling power (2.0 g/g and 2.05 g/g), oil holding capacity (4.91 g/g and 2.42 g/g), solubility (70%), total phenolic content of 4.24 mg GAE/g and 4.32 mg GAE/g, DPPH reduction of 86.6% and 83.9%, respectively. The results indicated that pearl millet bran can act as rich source of dietary fiber with health enhancing properties and can be utilized as potential food component in preparation of functional food products.

**Keywords** Eco-friendly · Waste valorization · Hot water · Dietary fiber

## Abbreviations

DF	Dietary fiber
SDF	Soluble dietary fiber
IDF	Insoluble dietary fiber
DFC	Dietary fiber concentrate
HT	Hydrothermal treatment
BBD	Box–Behnken design
RSM	Response surface methodology
SEM	Scanning electron microscopy
XRD	X-ray diffraction
FTIR	Fourier transform infrared spectroscopy
TPC	Total polyphenolic content
WHC	Water holding capacity
OHC	Oil holding capacity
WS	Water solubility
SC	Swelling capacity
EA	Emulsifying activity
PCA	Principle component analysis

## Introduction

Millets are group of small seeded annual grasses and is growing extensively in the tropical and semi-arid region of the world (Sachdev et al. 2023). Pearl millet (*Pennisetum glaucum*) is a staple food in Asia, Africa and contributed to 40% of the world millet production. This was followed by foxtail millet (*Setaria italica*), proso millet (*Panicum miliaceum*) and finger millet (*Eleusine carocana*). Pearl millet grains are often referred to as “nutri-cereals” and “coarse grain” because of their higher fiber, mineral, protein, antioxidant activity. It constitutes of high functional

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and nutritional properties and provides a wide range of health benefits (Selladurai et al. 2022). The by-product of millet milling process, millet bran remains underexploited residue, despite the fact that it is rich in polyphenols, dietary fiber, minerals and gluten-free (Barbhai and Hymavathi 2022). Dietary fiber (DF) is defined as that part of a plant in food that is not digested and resistant to enzymatic reaction in the human small intestine, maybe fully or partial fermented in the large intestine, and consists primarily of polysaccharides and a trace amount of protein (Kaur et al. 2021). The DF can be divided into two categories: water-soluble fiber (SDF) and water-insoluble fiber (IDF). SDF and IDF can be found in varying amounts in a variety of food sources, including grains, fruits, vegetables, and legumes (He et al. 2022). DFs can result in beneficial effects, such as lowering the risk of intestinal disorders, diabetes, obesity, hypertension and constipation, regulates blood pressure, thus reducing the risk of heart diseases (Zhu et al. 2018). Dietary fiber is excellent for weight loss as it reduces appetite by adding bulk to the diet. In particular, SDFs (such as glucan, pectin, inulin and oligosaccharides) are readily fermented in the digestive tract as compared to IDFs (such as cellulose, some hemicelluloses and lignin). This has a significant effect on intestinal health of human beings by controlling the composition of gut microbes. The consumption of SDF by humans has a positive impact on metabolism and cholesterol, which might be helpful for the treatment of atherosclerosis and other heart diseases (Wei et al. 2022).

The millet bran contains high dietary fiber content, bioactive compounds and micronutrients. However, it is generally discarded as waste or used as a fodder, which doesn't make use of this potential dietary resource (Mustač et al. 2020). In order to use this unutilized resource, it is therefore of considerable potential to develop a highly efficient method for extracting dietary fiber from pearl millet bran. The hydrothermal (HT) extraction of dietary fiber is an economical and ecofriendly method besides it involves minimal chemicals, no sophisticated equipments and produces high purity dietary fiber. The hydrothermal treatment results in higher extraction yields by modifying the composition of dietary fibers by solubilizing their insoluble fraction and can improve the functionality of dietary fibers, making them more suitable for incorporation in food formulations (Garcia-Amezquita et al. 2020). The present study involves the use of underutilized by-product (bran) of pearl millet which is majorly grown around the world. The hydrothermal method was used as an effective and green solvent for the extraction of dietary fibers from pearl millet bran. The present study involves lower solvent consumption and lesser extraction time which makes the process more economical. Further, the optimization of various extraction variables was carried out to get the maximum yield of dietary fibers from

pearl millet bran and characterization (physico-chemical, functional and structural) was also studied.

## Materials and methods

### Materials

The pearl millet (*Pennisetum glaucum*) was procured from the certified centres at CCSHAU Agricultural Centre (Haryana) and University of Agricultural Sciences Bangalore, the seeds were cleaned to remove the foreign material. Pearl millet bran was obtained through parboiling the pearl millet grains. The whole parboiled-decorticated kernels were milled, polished and sieved through 60 mesh sieve to get bran powder of uniform particle size.

### Experimental design

#### *Extraction of dietary fiber*

The dietary fiber extraction from pearl millet bran was carried out by hydrothermal method (Elleuch et al. 2008; Cardenas-Toro et al. 2014). The pearl millet bran (1 g) was added into 20 mL distilled water at 100 °C for 5 min. Dietary fiber concentrates were recovered by centrifugation. The residue obtained was subjected to five successful rinsing (100 mL water at 40 °C) followed by centrifugation each time. The recovered residue was dried in an oven (40 °C) to collect the total dietary fiber concentrate (DFC) and yield was estimated using equation below:

$$\text{Dietary fiber concentrate}(\%, w/w) = \frac{(\text{Weight of residue})}{(\text{Weight of sample})} \times 100 \quad (1)$$

#### *Determination of total, insoluble and soluble dietary fiber*

The assessment of SDF and IDF content in pearl millet bran was carried out according to the procedure of Wang et al. (2022) with some modifications. The hydrothermally treated mixture was centrifuged (5000 g, 15 min), both supernatant and residue were collected. This was followed by centrifugation, residue was separated out, washed with distilled water, and left to dry to obtain IDF. The filtrate was separated out, mixed with ethanol (95%, four-fold volume) and left for precipitation (2 h). After precipitation, the obtained residue was rinsed with 100% ethanol and left to dry. Total dietary fiber was obtained as the sum of IDF and SDF content (Wang et al. 2022).

### Experimental design by response surface methodology

The effect of several extraction parameters including liquid to solid ratio (A) (10:1–30:1), temperature (B) (85–95 °C) and time (C) (10–20 min) on the yield of dietary fiber extraction (% w/w) was studied using Response Surface Methodology (RSM) with three factors Box–Behnken design (BBD). The data analysis and determination of regression equation coefficients was done using the statistical software Design Expert 13.0.5.0 (Statease Inc., Minneapolis, USA). Model fitting was confirmed by determination of  $R^2$  value and each term was tested statistically with F-values of  $p \leq 0.05$ . The analysis of variance (ANOVA) showed the significance of model.

### Microstructural characterization

#### Scanning electron microscopy

Scanning electron microscopy (JSM, 7610 F plus, JEOL, Japan) was used for the structural characterization of SDF and IDF extracted from pearl millet bran.

#### Fourier transform infrared spectroscopy

The functional groups of dietary fiber from pearl millet bran were observed using Fourier transform infrared spectroscopy (FTIR) spectrophotometer (RX-I Perkin Elmer Spectrum, USA). The absorbance range for FTIR was 4000 to 400  $\text{cm}^{-1}$ .

#### X-ray diffraction

The morphology of IDF and SDF samples extracted from pearl millet bran were recorded using XRD (PAN-analytic-X'pert PRO MRD, Almelo, Netherlands). The diffractograms were taken between 5° and 60°.

### Physico-chemical characterization

#### Total phenolic content

The Folin-Ciocalteu calorimetric method was used to determine the total phenolic content (TPC) of dietary fiber samples from pearl millet bran (Ainsworth and Gillespie 2007). Ethanol (50%) was used to prepare the extract (100  $\mu\text{L}$ ) of dietary fiber samples and it was mixed with sodium carbonate (750  $\mu\text{L}$ ; 7.5% w/v) and Folin-Ciocalteu reagent (750  $\mu\text{L}$ ; 10% v/v). The solution was placed at room temperature (30 min) and its absorbance was recorded (765 nm). To calculate TPC, the standard curve of gallic acid was used and the results were expressed as mg gallic acid equivalent per gram of dietary fiber sample (mg GAE/g).

### DPPH scavenging activity

To determine antioxidant activity of dietary fiber samples from pearl millet bran, the extract was prepared using distilled water followed by addition of ethanolic DPPH solution (0.1 mM, 1 mL) to extract. It was then placed in dark (30 min) and the absorbance was recorded (517 nm). The DPPH scavenging activity was calculated using the Eq. (1) (Nenadis and Tsimidou 2002)

$$\text{DPPH scavenging activity(\%)} = \frac{A_{\text{control}} - A_{\text{sample}}}{A_{\text{control}}} \times 100 \quad (2)$$

### Particle size

The particle size analyzer (Shimadzu corporation SALD-2300, Japan) was used for the determination of particle size of dietary fiber samples extracted from pearl millet bran.

### Functional characterization

#### Water holding capacity

Estimation of water holding capacity (WHC) of dietary fiber obtained from pearl millet bran was carried out by mixing the sample (0.5 g) with distilled water (20 mL) in a centrifuge tube. It was then placed at room temperature (30 min) followed by centrifugation (3000 g, 20 min) and separation of supernatant (Kurek et al. 2018). The centrifuge tube containing the residue was weighed and WHC was calculated using the Eq. (3).

$$\text{WHC(g/g)} = \frac{M_2 - M_1}{M_1} \quad (3)$$

where  $M_1$  is the weight of the sample,  $M_2$  is the final weight of the residue.

#### Oil holding capacity

Estimation of oil holding capacity (OHC) of dietary fiber obtained from pearl millet bran was carried out by mixing the sample (0.5 g) with appropriate amount of oil (10 mL) in a centrifuge tube. The mixture was placed at room temperature (30 min), centrifuged (3000 g, 20 min) and the supernatant was removed (Kurek et al. 2018). The centrifuge tube containing residue was weighed and OHC was estimated using Eq. (4)

$$\text{OHC(g/g)} = \frac{O_2 - O_1}{O_1} \quad (4)$$

where  $O_1$  is the sample weight,  $O_2$  is the final weight of the residue.

#### Water solubility

Water solubility (WS) of extracted dietary fiber from pearl millet bran was determined by adding the sample (0.3 g) in distilled water (10 mL) and stirring (100 °C, 30 min). The centrifugation (4000 g, 15 min) was done and supernatant collected was transferred to weighing petri plate and left to dry. The WS was determined using the Eq. (5) (Li et al 2022a, b)

$$WS(\%) = \frac{s_1}{s_0} \times 100 \quad (5)$$

where  $s_1$  was the weight of supernatant after drying,  $s_0$  was the weight of initial dried sample.

#### Swelling capacity

To determine the swelling capacity (SC) of dietary fiber from pearl millet bran, the IDF and SDF sample (0.5 g) placed in a test tube, then volume was noted. This was followed by the addition of water (10 mL) and then incubated for 18 h. The final volume achieved by the sample was measured using the Eq. (6) (Li et al. 2022a, b)

$$SC(\text{mL/g}) = (V_2 - V_1)/m \quad (6)$$

where  $V_1$  is volume of sample,  $V_2$  is sample volume after 18 h, and  $m$  is the weight of sample.

#### Emulsifying activity

To determine the emulsifying activity (EA) of dietary fiber from pearl millet bran, the sample solution (5 mL, 0.5%, w/v) was mixed with oil (5 mL). The mixture was then homogenized and centrifugation (4000 g, 5 min) was done. The emulsion activity was determined using the Eq. (7) (Panwar et al. 2023)

$$EA(\%) = \frac{\text{Volume of emulsified layer}}{\text{Total volume of solution}} \times 100 \quad (7)$$

#### Statistical analysis

The data analysis was done using the statistical software Design Expert 13.0.5.0 (Statease Inc., Minneapolis, USA). The principle component analysis (PCA) was analyzed using SPSS software at a level of significance ( $p < 0.05$ ).

## Results and discussion

The optimum yield of extracted dietary fiber from pearl millet bran was assessed. The physicochemical, functional and structural characteristics of the dietary fibre obtained was evaluated and mentioned below.

#### Extraction of dietary fiber from pearl millet bran

The total extraction yield of dietary fiber obtained from pearl millet bran isolated using the hydrothermal treatment was found to be 74.5% (w/w). The result obtained was comparable with past studies where it was reported that maximum yield of dietary fiber recovered from millet bran was 73.18 g/100 g (Liu et al. 2012). The effect of various process parameters (temperature 85–95 °C, time 10–20 min, and ratio of liquid to solid 10:1–30:1) on the dietary fiber extraction yield was selected on the basis of primary trials and examined. BBD was used to analyze the interactions between process variables and their effect on the yield of dietary fiber. The three factors BBD design and the responses expressed as IDF and SDF (% w/w) has been presented in Table 1. The extraction process variables including ratio of liquid to solid (A), temperature (B) and time (C) showed significant effect ( $p < 0.05$ ) on dietary fiber yield. The quadratic effect of all the process parameters and their interactive effect (AB, AC, and BC) was found significant. F-value was significant with  $R^2$  values of 0.996 and 0.997 for IDF and SDF, respectively. The Lack of Fit F-value of 0.439 and 1.55 implies the Lack of Fit is not significant relative to the pure error. There is a 33.29% chance that a “Lack of Fit F-value” this large happens because of noise. The results demonstrated the applicability of the mathematical model for prediction within the range of the selected experimental variables and was found to be appropriate for extraction of IDF and SDF present in pearl millet bran (Table 2).

#### Optimization of process variables for dietary fiber extraction

The interactive effect of different process variables such as temperature, time, and liquid to solid ratio on IDF and SDF yield has been presented in Fig. 1a–f. When the ratio of liquid to solid was increased from 10:1 to 20:1 at temperature up to 90 °C, the extraction yield of IDF was found to increase, however, subsequently its value was found to drop (Fig. 1a). This may be due to the higher concentration difference between the liquid (water) and solid (bran sample), which caused greater diffusion of dietary fiber into the solvent (Wang and Bai 2017). A similar effect of ratio of liquid to solid was observed in earlier studies on corn hull, where increase in ratio of liquid to solid up to 20:1 led to increase in extraction yield of dietary fiber (Wang et al. 2018). An

**Table 1** BBD experimental design for IDF and SDF extraction

Run No	Factor 1	Factor 2	Factor 3	Response 1	Response 2
	A: Solid/liquid ratio	B: Temperature (degree celsius)	C: Time (min)	R1 Insoluble dietary fiber % (w/w)	R2 Soluble dietary fiber % (w/w)
1	30	85	15	63.0	8.0
2	20	85	20	61.0	5.0
3	10	95	15	62.6	7.0
4	20	90	15	68.0	8.1
5	20	90	15	67.5	8.2
6	20	90	15	67.5	8.3
7	20	85	10	67.5	7.0
8	20	95	20	67.0	6.1
9	20	95	10	59.2	5.0
10	10	90	10	66.5	5.4
11	30	90	20	66.5	5.8
12	30	90	10	66.0	5.5
13	20	90	15	68.0	8.1
14	20	90	15	68.0	8.2
15	10	85	15	68.0	6.0
16	10	90	20	67.0	4.5
17	30	95	15	67.0	6.0

insignificant effect of time and ratio of liquid to solid on IDF yield was observed ( $p > 0.05$ ). The interactive effect of time and temperature on the IDF yield has been presented in Fig. 1c. The extraction yield of IDF increased, when the treatment time was increase from 10 to 15 min and then found to be decreased. The yield of IDF increased initially when the temperature increased from 85 to 90 °C, but after 90 °C temperature its value eventually decreased. Due to an increase in water solvating capacity at higher temperature, the yield of IDF increased with temperature (Benito et al. 2013).

The interactive effect of temperature and liquid to solid ratio on extracted SDF has been shown in Fig. 1d. The SDF yield increased with temperature from 85 to 90 °C and ratio of liquid to solid 10:1 to 20:1. The increase in solubility of the soluble substances during the hydrothermal treatment at high temperature resulted in the breakdown of millet bran hemicelluloses and release of soluble substances from the sample. The interactive effect of time and ratio of liquid to solid on the SDF yield has been presented in Fig. 1e. The heating time from 10 to 15 min result in increased extracted dietary fiber yield; however, longer duration of hydrothermal treatment led to the thermal disintegration of the soluble dietary fiber (Wang et al. 2018). A similar observation was reported by researchers where it was found that long extraction time resulted in degradation of  $\beta$ -glucan from waxy barley when pressurized hot water was used (Benito et al. 2013). The increase in SDF yield with increasing temperature from 85 to 90 °C may be due to higher solubility and

diffusion process (Fig. 1f). When the extraction temperature is too low, it is difficult to dissolve the soluble fibers from the plant cell wall (Kamal et al. 2023).

### Verification of the model

The different process variables (extraction time, liquid to solid ratio and temperature) were optimized and resulted in the highest yield of IDF and SDF at 90 °C temperature, liquid to solid ratio of 20:1 and treatment time of 15 min. At this optimized condition, the SDF and IDF yield was predicted to be 8.2% and 67.8%. The practical experiment gave the yield of IDF and SDF was 67% and 7.5%.

### Microstructural characterization

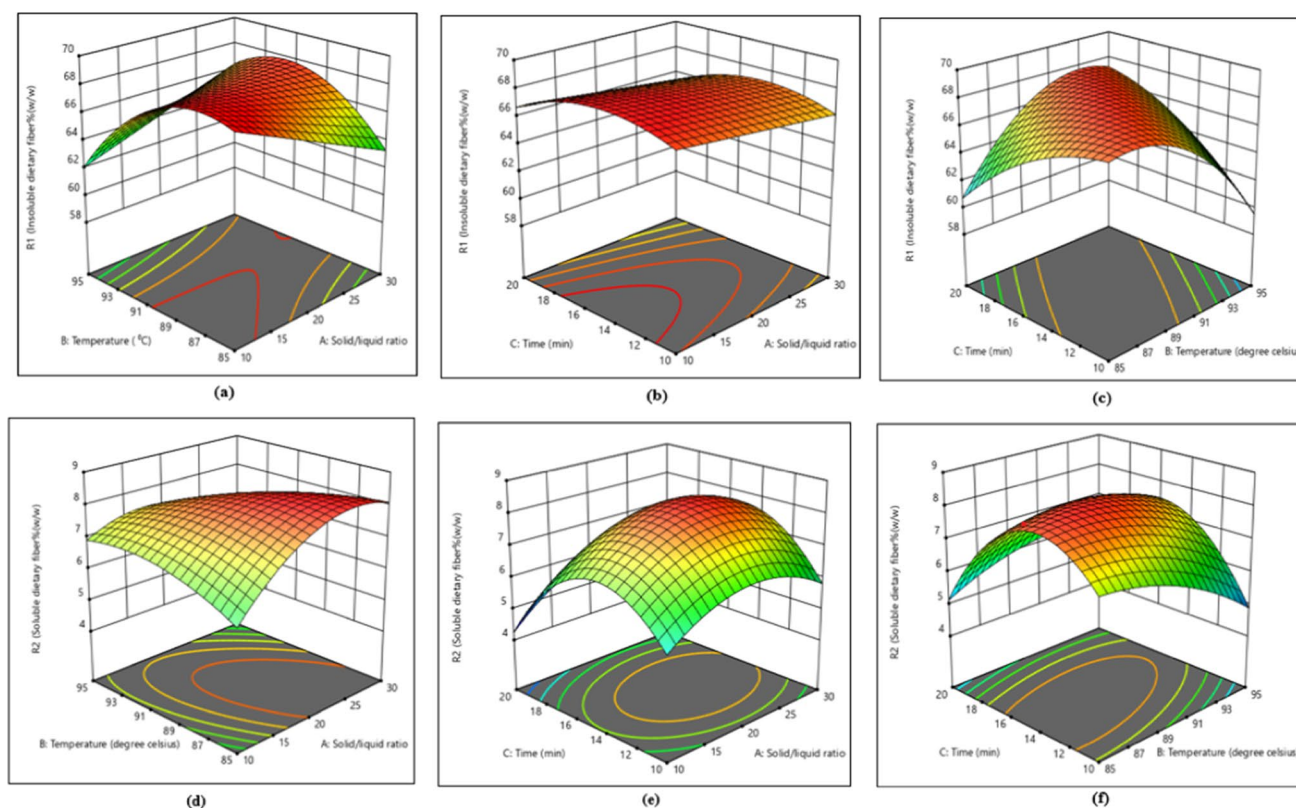
#### SEM analysis

The structural morphology of the dietary fiber samples from pearl millet bran were carried out by scanning electron microscopy indicated in Fig. 2a. SDF-HT sample shown irregular flakes of uneven size and dense surface structure. The IDF-HT revealed a comparatively compact surface and smooth area with fine constituent parts on the surface (Chu et al. 2019). The hydrothermal extraction caused degradation of the cellular structure of dietary fiber samples and resulted in development of clusters and fractures. This may have led to formation of porous structure

**Table 2** ANOVA for response surface model for IDF and SDF extraction from pearl millet bran

Source	IDF				SDF			
	Sum of Squares	df	Mean Square	F-value p-value	Sum of Squares	df	Mean Square	F-value p-value
Model	117.42	7	16.77	324.83 <0.0001	28.52	9	3.17	366.64 <0.0001
A-solid/liquid ratio	0.3200	1	0.3200	6.20 0.0345	0.7200	1	0.7200	83.31 <0.0001
B-temperature	1.71	1	1.71	33.14 0.0003	0.4512	1	0.4512	52.21 0.0002
C-time	0.6612	1	0.6612	12.81 0.0059	0.2813	1	0.2813	32.54 0.0007
AB	22.09	1	22.09	427.79 <0.0001	2.25	1	2.25	260.33 <0.0001
AC					0.3600	1	0.3600	41.65 0.0003
BC	51.12	1	51.12	990.03 <0.0001	2.40	1	2.40	277.98 <0.0001
A <sup>2</sup>					3.82	1	3.82	441.99 <0.0001
B <sup>2</sup>	31.53	1	31.53	610.69 <0.0001	0.9600	1	0.9600	111.08 <0.0001
C <sup>2</sup>	8.07	1	8.07	156.37 <0.0001	15.64	1	15.64	1809.96 <0.0001
Residual	0.4647	9	0.0516		0.0605	7	0.0086	
Lack of fit	0.1647	5	0.0329	0.4393 0.8045*	0.0325	3	0.0108	1.55 0.3329* Non-significant
Pure error	0.3000	4	0.0750		0.0280	4	0.0070	
Cor total	117.88	16			28.58	16		
Fit statistics	R <sup>2</sup> = 0.9961 Adjusted R <sup>2</sup> = 0.9930 Predicted R <sup>2</sup> = 0.9843				R <sup>2</sup> = 0.9979 Adjusted R <sup>2</sup> = 0.9952 Predicted R <sup>2</sup> = 0.9803			
	Adequate precision = 55.9414 Coefficient variation = 0.344%				Adequate precision = 51.4358 Coefficient variation = 1.41%			

\*Non-significant at 95% confidence level



**Fig. 1** Interactive effect of hydrothermal extraction **a, d** Temperature and solid to liquid ratio. **b, e** Time and solid to liquid ratio **c, f** Time and temperature on IDF and SDF from pearl millet bran

which corresponds to moderate cholesterol binding capacity and hydration properties of dietary fibers (Wen et al. 2017).

#### FTIR analysis

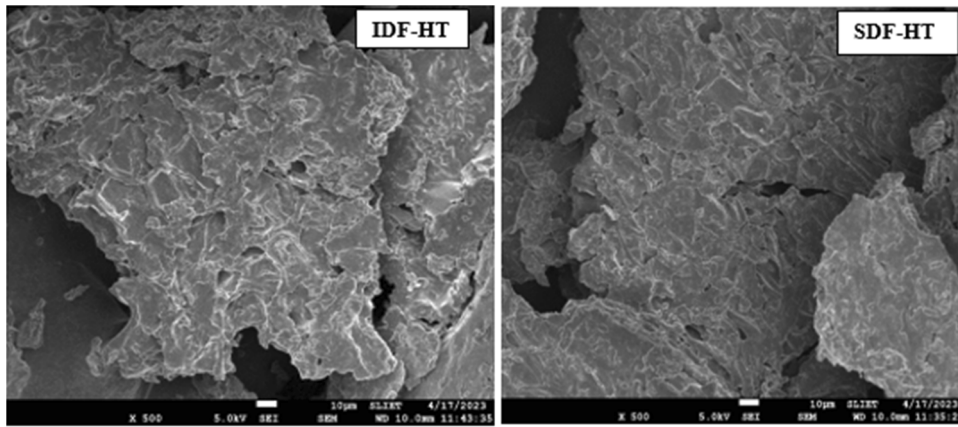
The organic functional groups were studied using FTIR spectrum and the molecular structures of IDF and SDF has been represented in Fig. 2b. At about  $3272\text{ cm}^{-1}$  and  $3268\text{ cm}^{-1}$ , both samples showed significant signals which were caused by the vibrations of H-bonds linked with the  $-\text{OH}$  groups of polysaccharides present in the samples. The intensity of absorption peaks at  $2800\text{--}3000\text{ cm}^{-1}$  indicated the C–H,  $-\text{CH}_2$  or  $-\text{CH}$  groups stretching bands of polysaccharides (Chu et al. 2019). The absorption bands near  $1634\text{--}1643\text{ cm}^{-1}$  represented O–H bending vibrations that was characteristic of adsorbed water. The band of absorption peak around  $1744\text{ cm}^{-1}$  corresponded to  $\text{C}=\text{O}$  stretching vibrations in the carboxyl groups ( $-\text{COOH}$ ) and suggested that uronic acids existed in the form. Weak intensity peaks near  $1200\text{--}1435\text{ cm}^{-1}$  represented the C–O stretching in hemicelluloses, cellulose and C–H bending vibrations (Kaur et al. 2021).

#### XRD analysis

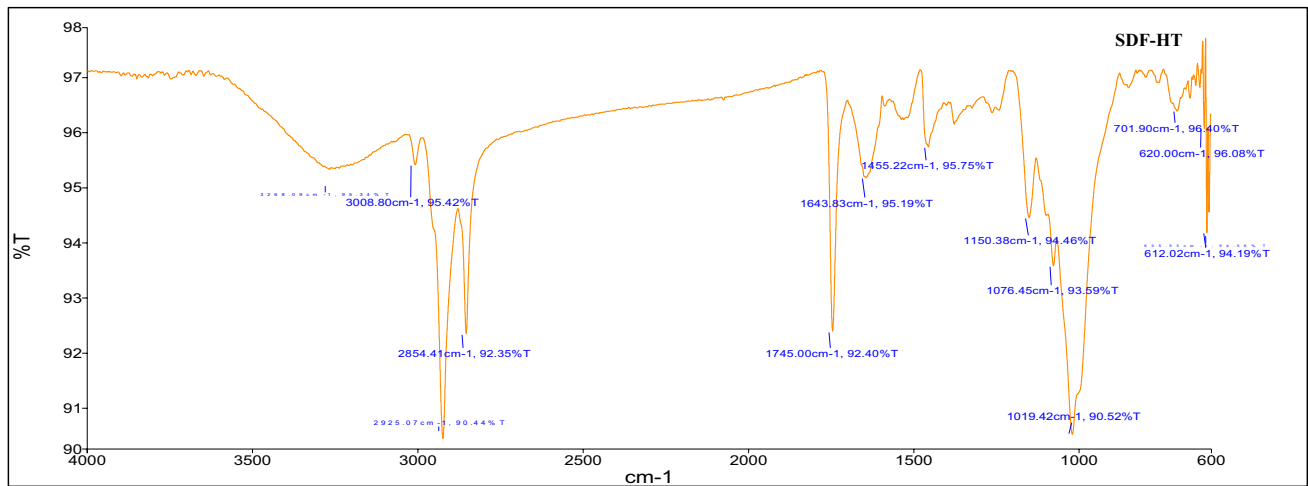
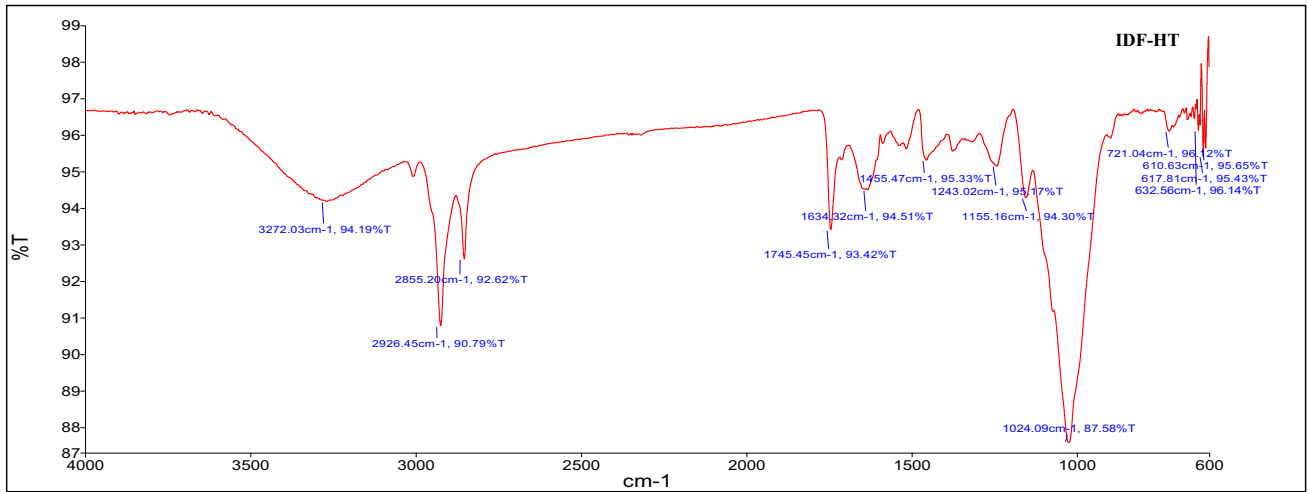
The morphological nature of dietary fiber samples extracted from pearl millet bran was analyzed using X-Ray diffraction. The crystal structure of the IDF and SDF (HT extraction) samples has been presented in Fig. 2c. Amorphous regions in dietary fiber may be characteristic of non-crystalline hemicellulose, cellulose, and lignin whereas, crystalline regions in most part might be due to the presence of cellulose. The SDF-HT sample had weak diffraction peaks which indicated that SDF-HT mainly existed in amorphous form and was difficult to crystallize (Wang et al. 2022). The results showed that the sharp diffraction peaks in IDF-HT observed in the range of  $15$  to  $25^\circ\text{C}$  indicated the presence of typical type I cellulose (Li et al. 2022a, b).

#### Physico-chemical characterization

TPC of IDF and SDF obtained from pearl millet bran using hydrothermal treatment was found to be  $4.24\text{ mg GAE/g}$  and  $4.32\text{ mg GAE/g}$ . The lower TPC content may be attributed to the decortication of millet husks and brans. The DPPH activity of IDF and SDF from pearl millet bran obtained was  $86.6\%$  and  $83.9\%$  (Li et al. 2022a,



(a)



(b)

**Fig. 2** Structural examination of dietary fiber extracted from pearl millet bran using hydrothermal extraction **a** SEM **b** FTIR **c** XRD

b). The higher DPPH antioxidant activity exerts positive effects on nutritional profile. The particle size analysis of IDF and SDF obtained from pearl millet bran using hydrothermal extraction indicated values of 181.6 and

161.7 µm, respectively. The particle size directly effects the functional properties of the dietary fibers (Jiang et al. 2022).



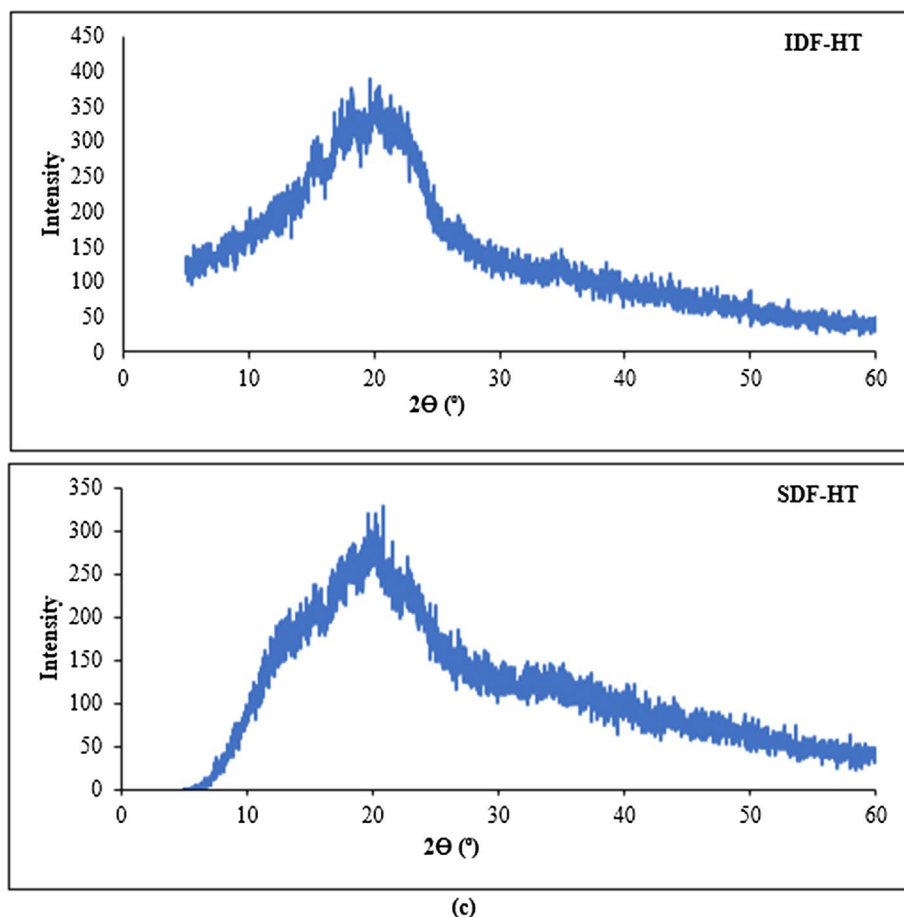


Fig. 2 (continued)

### Determination of functional properties

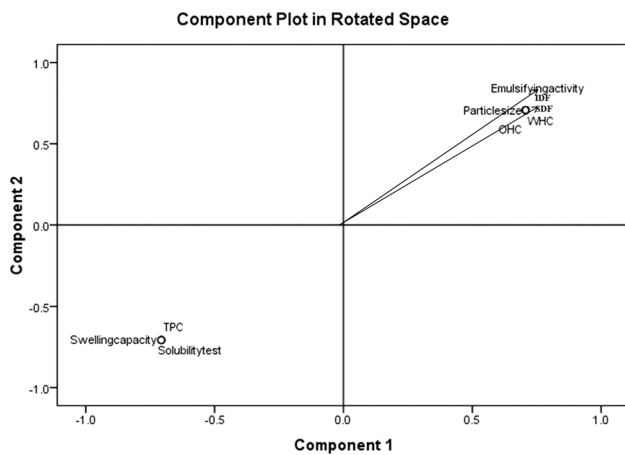
WHC is the capability of DF to capture enormous quantity of water and resulted in prevention of water from oozing out (Wang et al. 2022). WHC of IDF and SDF obtained from pearl millet bran using hydrothermal treatment was found to be 6.50 g/g and 3.99 g/g. Similarly, oil holding capacity for IDF and SDF sample obtained was 4.91 g/g and 2.42 g/g, respectively. WHC and OHC are associated with the surface characteristics such as thickness, viscosity, hydrophobic and hydrophilic groups of the dietary fiber components, respectively (Fernández-López et al. 2009). The solubility of pearl millet bran SDF was found to be 70% at 100 °C. Dietary fiber with high WHC makes it suitable in applications as functional food ingredient to alter the viscosity, texture and avoid syneresis (Elleuch et al. 2011). The SC for IDF and SDF sample from pearl millet bran using hydrothermal treatment obtained was 2.0 g/g and 2.05 g/g. The EA associated with the protein's ability to absorb water, oil and formation of emulsion (Devisetti et al. 2014). The EA for IDF and SDF sample from pearl millet bran obtained was 24% and 22.5%.

### Principle component analysis

The validation of differences among different properties of dietary fiber samples was studied using Principal Component Analysis (Fig. 3). In this case, it was observed that both samples IDF and SDF lie in the same quadrant depicting a strong correlation. Regarding the functional parameters, it was found that emulsifying activity, particle size, water holding capacity and oil holding capacity were a positive correlation with IDF and SDF whereas TPC, DPPH activity, solubility index, and swelling capacity indicate a negative correlation with both samples (Sharma et al. 2022).

### Conclusion

Hydrothermal method has been successfully utilized as an efficient ecofriendly technique for the extraction of dietary fiber from pearl millet bran with potential application at the industrial level. The highest yield of total dietary fiber (74.5%, w/w) was found with 20:1 liquid to solid ratio at 90 °C temperature after extraction time of 15 min. SEM



**Fig. 3** Principle component analysis of IDF and SDF samples from pearl millet bran

images indicated a considerable effect of hydrothermal treatment on dietary fiber. FTIR analysis of IDF and SDF indicated the presence of organic functional groups. XRD analysis of SDF and IDF from pearl millet bran revealed the presence of crystalline form. WHC, OHC and SC of dietary fiber samples from pearl millet bran can help to enhance the texture and rheological properties in food products. The study revealed that pearl millet bran contains considerable amount of dietary fiber and therefore, can be used as a potential food component in preparation of functional food products.

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**Author contributions** RK: Investigation, Methodology, Writing—original draft. PSP: Conceptualization, Writing—review & editing, Supervision, Project administration. BK: Data analysis. CSR: Supervision, Guidance.

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**Availability of data and materials** Not applicable.

**Code availability** Not applicable.

**Declarations**

**Conflict of interest** The authors declare no conflict of interest regarding the publication of this paper.

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