### **RESEARCH**



# **Comparative Study of Entropies in Silicate and Oxide Frameworks**

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#### **Abstract**

Silicate and oxide frameworks are pervasive materials with remarkable structural complexity and tunability, offering a wide range of applications in catalysis, gas storage, drug delivery, electronics, and environmental remediation. Topological indices, which are mathematical representations of molecular structure, and Shannon entropy, a measure of information content, have emerged as powerful tools for studying the structural characteristics of these frameworks. In this study, we investigate the effectiveness of topological indices and entropy levels in revealing the structural characteristics of silicate and oxide frameworks. We formulate topological expressions for newly developed hybrid indices derived from geometric, harmonic, and Zagreb indices and conduct a scaled bond-wise comparative analysis between the two frameworks.

**Keywords** Silicate and oxide frameworks · Degree and degree-sum indices · Information function · Entropies

# **1 Introduction**

Silica minerals, the most prevalent minerals in Earth's crust, hold immense geological significance, exhibiting unique versatility as they emerge from diverse environments, ranging from high-temperature igneous settings to low aquatic conditions [\[1](#page-10-0)]. Among these minerals, quartz stands out as one of the most widely utilized and recognized [\[2\]](#page-10-1). Silica minerals, primarily composed of silicon dioxide  $(SiO<sub>2</sub>)$ , serve as a subset of the larger class of silicate minerals. They form a broader and more widespread category, incorporating silicon, oxygen, and additional metallic elements such as aluminum, iron, magnesium, potassium, sodium, and calcium [\[3,](#page-10-2) [4](#page-10-3)]. Their prevalence and complexity are essential for understanding plate tectonics, mineral formation, and the earth's geological history. The basic structural unit of silicate minerals is the silicon-oxygen tetrahedron  $(SiO<sub>4</sub>)$ , which provides the basis for the vast diversity of silicate structures, ranging from isolated tetrahedra to intricate three-dimensional frameworks [\[5](#page-10-4)[–7\]](#page-10-5). Silicate materials, especially zeolites, are used in environmental applications such as water purification and air pollution control. They contribute to reducing environmental pollutants and enhancing water quality [\[8\]](#page-10-6). Oxide frameworks constitute a diverse class of materials primarily composed of oxygen (O). Metal oxides like titanium dioxide and cerium oxide exhibit heterogeneous catalysis, influencing reactions in environmental cleanup and industrial processes [\[9](#page-10-7)[–13\]](#page-10-8).

In recent years, researchers have been exploring 2 dimensional, silicate, and oxide frameworks in various areas to broaden the scope of their applications  $[14–33]$  $[14–33]$  $[14–33]$  where Fig. [1](#page-1-0) shows the 2-D view of the  $SiO<sub>4</sub>$  tetrahedron. Depending on the arrangement of tetrahedra, various silicate structures can be identified, including chain silicates, sheet silicates, framework silicates, and cyclic silicates [\[5\]](#page-10-4). The unit block of silicates is formed by placing six units of  $SiO<sub>4</sub>$  in a cyclic order as shown in Fig. [2a](#page-1-1), and the removal of oxygen atoms in silicates gives the oxide unit as shown in Fig. [2b](#page-1-1).

Topological indices serve as essential tools in mathematical chemistry, providing quantitative measures of molecular structure and offering a graph-theoretical approach to characterize the structural complexities of molecules. They represent a mathematical concept derived from molecular graphs, gaining prominence in QSAR/QSPR studies. These indices encode structural information and connectivity patterns within molecules, facilitating predictions of biological activities and physicochemical properties [\[34](#page-10-11)[–40](#page-11-0)]. In QSAR and QSPR studies, topological indices serve as valuable tools for assimilating and predicting complex molecular behaviors,

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Fig. 1 2-dimensional view of SiO<sub>4</sub> tetrahedron

<span id="page-1-0"></span>with applications in drug design, environmental chemistry, and materials engineering [\[40](#page-11-0)[–45](#page-11-1)]. Topological indices are classified into three main types, including degree, distance, and eigenvalues of graphs. Exploring topological indices for silicate and oxide frameworks is essential for interpreting the structural features of these materials and their wide-ranging properties. Numerous research articles have already examined both distance-based [\[16,](#page-10-12) [17](#page-10-13)] and degree-based indices [\[18](#page-10-14)[–31](#page-10-15)]. However, to our knowledge, no work has been reported on entropy indices.

Shannon entropy is a promising tool for describing the information content of molecules in the field of molecular analysis [\[46](#page-11-2)[–48\]](#page-11-3). This approach offers the advantages of deriving a numerical value, facilitating easier comparisons

<span id="page-1-1"></span>**Fig. 2** Unit blocks (a) silicate (b) oxide

among different molecules, and obviating the need for cumbersome, high-dimensional, and computationally intensive matrix processing. Their ability to quantify information content has drawn significant interest across various fields in recent years [\[49](#page-11-4)[–54](#page-11-5)]. This paper explores the formulation of topological expressions for recently proposed hybrid indices based on geometric, harmonic, and Zagreb degree-based indices, emphasizing their efficacy in determining bond-wise entropy measures along with comparison between silicate and oxide frameworks.

## **2 Computational Techniques**

We provide the graph theoretical parameters to describe the topological indices and entropies. Our study primarily focuses on topological indices, including geometric, harmonic, and Zagreb, along with their hybrid counterparts. These indices serve to quantify the entropies of silicate and oxide frameworks. We represent these chemical frameworks as connected graphs, where vertices symbolize atoms and edges represent chemical bonds between two atoms. The atoms and bonds of frameworks are generally grouped into sets  $V(G)$  and  $E(G)$ , respectively, for a chemical graph  $G$ .

The vertex degree of  $a \in V(G)$ , denoted as  $d_G(a)$ , represents the count of neighboring vertex members associated with vertex *a*. Additionally, we define the degree-sum of the vertex *a* as  $s_G(a)$ , which is determined by adding the degrees of vertex members within the neighborhood of *a*. That is,  $s_G(a) = \sum_{n=1}^{\infty} a_n$ *p*∈*NG* (*a*)  $d_G(p)$  in which we used *NG*(*a*) = {*p* ∈ *V*(*G*) | *pa* ∈ *E*(*G*)}. Let  $d_{(p,q)}(G)$  =  $|\{ij \in E(G) : d_G(i) = p \text{ and } d_G(j) = a\}|$  and  $s_{(p,a)}(G) = |\{ij \in E(G) : s_G(i) = p \text{ and } s_G(j) = a\}|.$ 



Consequently, the total number of edges in *G* would be structured into distinct partition classes in accordance with symmetrical criteria of  $d_{(p,q)}(G)$  and  $s_{(p,q)}(G)$ , with these partition classes referred to as  $D(G)$  and  $S(G)$ , respectively.

The degree and degree-sum metrics of *G* can be readily converted into numerical indices through the utilization of a designated index function denoted as  $\chi$ . This function is formulated by employing two distinct mathematical operations, as outlined below [\[30,](#page-10-16) [31,](#page-10-15) [35,](#page-10-17) [55](#page-11-6)[–58\]](#page-11-7).

$$
\chi^{d}(G) = \sum_{d_{(p,a)}(G) \in D(G)} d_{(p,a)}(G) \chi(p, a)
$$
  

$$
\chi^{d*}(G) = \prod_{d_{(p,a)}(G) \in D(G)} d_{(p,a)}(G) \chi(p, a)
$$
  

$$
\chi^{s}(G) = \sum_{s_{(p,a)}(G) \in S(G)} s_{(p,a)}(G) \chi(p, a)
$$
  

$$
\chi^{s*}(G) = \prod_{s_{(p,a)}(G) \in S(G)} s_{(p,a)}(G) \chi(p, a)
$$

The bond additive and scalar multiplicative indices defined above, which correspond to self-powered index functions, are presented as follows.

$$
\chi^{dp}(G) = \sum_{d(p,a)(G) \in D(G)} d_{(p,a)}(G) \chi(p,a)^{\chi(p,a)}
$$

$$
\chi^{dp*}(G) = \prod_{d_{(p,a)}(G) \in D(G)} d_{(p,a)}(G) \chi(p,a)^{\chi(p,a)}
$$
  

$$
\chi^{sp}(G) = \sum_{d_{(p,a)}(G) \times (R, a) \times (P, a)} \chi^{(p,a)}
$$

$$
\chi^{sp}(G) = \sum_{s_{(p,a)}(G) \in S(G)} s_{(p,a)}(G) \chi(p,a)^{\chi(p,a)}
$$

$$
\chi^{sp*}(G) = \prod_{s_{(p,a)}(G) \in S(G)} s_{(p,a)}(G) \chi(p,a)^{\chi(p,a)}
$$

The index function  $\chi(p, a)$  for geometric, harmonic, and Zagreb, along with their respective hybrid indices is given below [\[38\]](#page-10-18).

- Geometric  $G(p, a) = \sqrt{pa}$
- Harmonic  $H(p, a) = \frac{2}{p+1}$
- **Bi-Zagreb**  $BM(p, a) = p + a$ <br>• **Bi-Zagreb**  $BM(p, a) = p + a + pa$
- Tri-Zagreb  $TM(p, a) = p^2 + a^2 + pa$
- Geometric Harmonic  $GH(p, a) = \frac{\sqrt{pa}(p+a)}{2}$
- Geometric Bi-Zagreb  $GBM(p, a) = \frac{\sqrt{pa}}{p+a+1}$
- Harmonic Bi-Zagreb *HBM*(*p*, *a*) =  $\frac{2}{(p+a+pa)(p+a)}$
- Harmonic Tri-Zagreb  $HTM(p, a) = \frac{2}{(p^2 + a^2 + pa)(p + a)}$
- Bi-Zagreb Geometric  $BMG(p, a) = \frac{(p+a+pa)}{\sqrt{pa}}$
- Bi-Zagreb Harmonic  $BMH(p, a) = \frac{(p+a+pa)(p+a)}{2}$
- Tri-Zagreb Geometric  $TMG(p, a) = \frac{p^2 + a^2 + pa}{\sqrt{pa}}$
- Tri-Zagreb Harmonic  $TMH(p, a) = \frac{(p^2 + a^2 + pa)(p + a)}{2}$

The above described index functions could be considered as the non-negative real valued structural information function  $\chi$  on  $E(G)$  in order to calculate the entropies of a graph *G* with degree and degree-sum metrics. Let  $E(G)$  =  ${c_1, c_2, ..., c_r}$ . The graph entropy of *G* is determined as follows:

$$
I_{\chi}(G) = -\sum_{i=1}^{r} \frac{\chi(c_i)}{\sum_{j=1}^{r} \chi(c_j)} \log(\frac{\chi(c_i)}{\sum_{j=1}^{r} \chi(c_j)})
$$
  
=  $\log(\sum_{i=1}^{r} \chi(c_i)) - \frac{1}{\sum_{i=1}^{r} \chi(c_i)} \log(\prod_{i=1}^{r} \chi(c_i)^{\chi(c_i)})$ 

As discussed in series of papers in recent years [\[51](#page-11-8)[–53](#page-11-9), [59](#page-11-10)– [61](#page-11-11)], the substitution of the multiplicative component with a scalar multiplicative index has been considered. Therefore,

$$
I_{\chi}(G) = \log(\chi(G)) - \frac{1}{\chi(G)} \log(\chi^{p*}(G))
$$

The significance of entropy generally depends on the specific system being considered. In a thermodynamic context, smaller entropy suggests a more ordered and structured state, while in information theory, it implies that information is more predictable or less uncertain.

# **3 Results and Discussion**

The foundation of silicate frameworks comprises  $(SiO<sub>4</sub>)$ tetrahedra, which combine in diverse ways to create various peripheral configurations, including chain, cyclic, hexagonal, rhombic, and trapezium shaped networks of silicates. In our study, we analyze the prevalent hexagonal framework, resembling honeycomb benzene systems where each bond in this system is replaced by a tetrahedron.

We use the notation  $SL_n$  to represent silicate frameworks of dimension *n*. As mentioned earlier, the oxide frameworks  $(OX_n)$  are obtained as a byproduct of silicates, where each silicon and its associated bond are deleted. The three dimensional silicate and oxide frameworks are shown in Figs. [3](#page-3-0) and [4.](#page-3-1)

The number of vertices and edges for silicate and oxide frameworks are ordered in the sets as  $\{3(5n^2 + n), 36n^2\}$ and  $\{3(3n^2 + n), 18n^2\}$ , respectively. Silicate and oxide frameworks have been extensively covered in several papers [\[18](#page-10-14)[–31](#page-10-15)] for computing various degree-based indices through bond partitions. We will utilize these partitions to derive the entropies for the first time and conduct a comparative analysis between them.



<span id="page-3-0"></span>**Fig. 3** Silicate framework SL3

The bond partitions of silicate and oxide frameworks induced from degree parameters are given as  $d_{(3,3)}(SL_n)$  =  $6n$ ,  $d_{(3,6)}(SL_n) = 18n^2 + 6n$ ,  $d_{(6,6)}(SL_n) = 18n^2 - 12n$ , and  $d_{(2,4)}(OX_n) = 12n, d_{(4,4)}(OX_n) = 18n^2 - 12n$ , respectively. Similarly, we tabulated the degree-sum bond distributions for silicate and oxide frameworks in Tables [1](#page-3-2) and [2.](#page-3-3)

The degree and degree-sum index expressions for silicate and oxide frameworks can be represented by

$$
\chi^{\{d,s\}}(G) = \begin{cases} \chi^d(G), \\ \chi^s(G) \end{cases}
$$



<span id="page-3-1"></span>Fig. 4 Oxide framework OX<sub>3</sub>

<span id="page-3-2"></span>**Table 1** Degree-sum partition of silicate frameworks

Bond $X-Y$	Degree-sum $s_{SL_n}(X) - s_{SL_n}(Y)$	Number of Bonds in $SL_n$			
$Si-O$	$15 - 15$	6п			
	$15 - 24$	24			
	$15 - 27$	$24(n-1)$			
	$18 - 27$	$12(n - 1)$			
	$18 - 30$	$18n^2 - 30n + 12$			
0–0	$24 - 27$	12			
	$27 - 27$	$3(4n - 6)$			
	$27 - 30$	$12(n - 1)$			
	$30 - 30$	$18n^2 - 36n + 18$			

where  $G \in \{SL_n, OX_n\}$ . We now ready to compute the degree and degree-sum indices for the topological function  $\chi$ , in which  $\chi \in \{G, H, BM, TM, GH, GBM, HBM, HTM,$ *BMG*, *BMH*, *TMG*, *TMH*}. The indices are calculated using the following equations:

For degree type,

$$
\chi^{d}(\text{SL}_{n}) = d_{(3,3)}(\text{SL}_{n}) \chi(3,3) + d_{(3,6)}(\text{SL}_{n}) \chi(3,6)
$$
  
+  $d_{(6,6)}(\text{SL}_{n}) \chi(6,6)$   
=  $6n\chi(3,3) + (18n^{2} + 6n)\chi(3,6)$   
+  $(18n^{2} - 12n)\chi(6,6)$ ,

and for degree-sum type,

$$
\chi^{s}(SL_{n}) = s_{(15,15)}(SL_{n}) \chi(15,15) + s_{(15,24)}(SL_{n}) \chi(15,24)
$$
  
+ $s_{(15,27)}(SL_{n}) \chi(15,27) + s_{(18,27)}(SL_{n}) \chi(18,27)$   
+ $s_{(18,30)}(SL_{n}) \chi(18,30) + s_{(24,27)}(SL_{n}) \chi(24,27)$   
+ $s_{(27,27)}(SL_{n}) \chi(27,27) + s_{(27,30)}(SL_{n}) \chi(27,30)$   
+ $s_{(30,30)}(SL_{n}) \chi(30,30)$   
=  $6n\chi(15,15) + 24\chi(15,24) + 24(n-1)\chi(15,27)$   
+12(n-1)\chi(18,27) + (18n<sup>2</sup> - 30n + 12)\chi(18,30)  
+12\chi(24,27) + 3(4n-6)\chi(27,27) + 12(n-1)\chi(27,30)  
+ (18n<sup>2</sup> - 36n + 18)\chi(30,30).

<span id="page-3-3"></span>**Table 2** Degree-sum partition of oxide frameworks

Degree-sum $s_{\text{OX}_n}(X) - s_{\text{OX}_n}(Y)$	Number of Bonds in $OX_n$			
$8 - 12$	12			
$8 - 14$	$12(n - 1)$			
$12 - 14$	12			
$14 - 14$	$3(4n - 6)$			
$14 - 16$	$12(n - 1)$			
$16 - 16$	$18n^2 - 36n + 18$			

 $\sqrt{ }$  $\int$ 

 $7(15n^2 + 2n)/5$ , 9(36857178520121*n*<sup>2</sup>

**Result 1** Let SL*<sup>n</sup>* be the silicate frameworks of dimension *n* where  $n > 1$ .  $(108+54\sqrt{2})n^2+(18\sqrt{2}-54)n,$ 

$$
\begin{cases}\n(108+54\sqrt{2})n^2 + (18\sqrt{2}-54)n, & 7. HBM^{[d,s]}(\text{SL}_n) = \begin{cases}\n(1633318697598976(5+\sqrt{15})n^2 & 143270459145262n \\
+10555311626496(12\sqrt{5}) & 6392926539773\n\end{cases}\\
1. G^{[d,s]}(\text{SL}_n) = \begin{cases}\n-633318697598976\sqrt{0} & 8. HTM^{[d,s]}(\text{SL}_n) = \begin{cases}\n4(16n^2+3n)/7, & 4(1603143349481n^2 - 14897288976\sqrt{0} \\
-633318697598976\sqrt{0} & 8. HTM^{[d,s]}(\text{SL}_n) = \begin{cases}\n(163318697598976\sqrt{0} & 8. HTM^{[d,s]}(\text{SL}_n) = \begin{cases}\n(1631\sqrt{2}+144)n^2 + (27\sqrt{2}-66)n, & 4(1603143349481n^2 - 148972888826794705 \\
+1266637395197952\sqrt{2} & 8. HTM^{[d,s]}(\text{SL}_n) = \begin{cases}\n(81\sqrt{2}+144)n^2 + (27\sqrt{2}-66)n, & 6855826794705 \\
-17592186044416 & 6855826794705\n\end{cases}\\
2. H^{[d,s]}(\text{SL}_n) = \begin{cases}\n(21n^2+4n)/3, & 9. \quad BMG^{[d,s]}(\text{SL}_n) = \begin{cases}\n(3749+16\sqrt{15})n^2 & -378\sqrt{15}\n\end{cases} & -178807888323216\n\end{cases}\\
3. B M^{[d,s]}(\text{SL}_n) = \begin{cases}\n(23937+2-756n, & 9. \quad BMG^{[d,s]}(\text{SL}_n) = \begin{cases}\n(234\sqrt{2}+648)n^
$$

We apply the bond partitions of oxide frameworks to derive the topological indices using the equations provided below. For degree type,

$$
\chi^{d}(\text{OX}_{n}) = d_{(2,4)}(\text{OX}_{n}) \chi(2,4) + d_{(4,4)}(\text{OX}_{n}) \chi(4,4)
$$
  
=  $12n\chi(2,4) + (18n^{2} - 12n)\chi(4,4),$ 

and for degree-sum type,

$$
\chi^{s}(\text{OX}_{n}) = s_{(8,12)}(\text{OX}_{n}) \chi(8, 12) + s_{(8,14)}(\text{OX}_{n}) \chi(8, 14)
$$
  
+
$$
s_{(12,14)}(\text{OX}_{n}) \chi(12, 14) + s_{(14,14)}(\text{OX}_{n}) \chi(14, 14)
$$
  
+
$$
s_{(14,16)}(\text{OX}_{n}) \chi(14, 16) + s_{(16,16)}(\text{OX}_{n}) \chi(16, 16)
$$
  
= 
$$
12\chi(8, 12) + 12(n - 1)\chi(8, 14) + 12\chi(12, 14)
$$
  
+
$$
3(4n - 6)\chi(14, 14) + 12(n - 1)\chi(14, 16)
$$
  
+
$$
(18n^{2} - 36n + 18)\chi(16, 16).
$$

**Result 2** Let  $OX_n$  be the oxide frameworks of dimension *n* where  $n > 1$ .

1. 
$$
G^{[d,s]}(OX_n) =\begin{cases} 72n^2 + (24\sqrt{2} - 48)n, \\ 288n^2 + (48\sqrt{7} + 48\sqrt{14} - 408)n + 48\sqrt{6} \\ -48\sqrt{7} - 48\sqrt{14} + 24\sqrt{42} + 36 \end{cases}
$$
  
\n2.  $H^{[d,s]}(OX_n) =\begin{cases} (9n^2 + 2n)/2, \\ (45045n^2 + 19942n + 2861)/40040 \end{cases}$   
\n3.  $B M^{[d,s]}(OX_n) =\begin{cases} 432n^2 - 120n, \\ 5184n^2 - 3024n + 216 \end{cases}$   
\n4.  $T M^{[d,s]}(OX_n) =\begin{cases} 864n^2 - 240n, \\ 13824n^2 - 8016n + 408 \end{cases}$   
\n $\begin{cases} 288n^2 + (72\sqrt{2} - 192)n, \\ (10133099161583616n^2 + (1161084278931456\sqrt{7} - 1161084278931456\sqrt{7} + 3482709361307204)/ \\ + 3482709361307204)/ \\ 2199023255552 \end{cases}$   
\n6.  $GBM^{[d,s]}(OX_n) =\begin{cases} (21n^2 + (12\sqrt{2} - 14)n)/7, \\ (191486536n^2 + 5626(12192) \\ \sqrt{7} + 6432\sqrt{14} - 42545)n \\ + 23689056\sqrt{42} - 36186432 \\ \sqrt{14} - 68592192\sqrt{7} + 79235808 \end{cases}$ 

7. *HBM*<sup>{d,s}</sup> (OX<sub>n</sub>) =   
\n
$$
\begin{cases}\n(84n^2 + 16n)/7, \\
(191486536n^2 + 84969478n + 3804919)/47871634\n\end{cases}
$$
\n8. *HTM*<sup>{d,s</sup>} (OX<sub>n</sub>) =   
\n
$$
\begin{cases}\n(42n^2 + 8n)/7, \\
(265475847n^2 + 111022130n + 19214171)/176983898\n\end{cases}
$$
\n
$$
\begin{cases}\n108n^2 + (42\sqrt{2} - 72)n, \\
(79798155897470976n^2 + 13405245765844992\sqrt{14}n + 14144117579710464\sqrt{7}n + 14144117579710464\sqrt{7}n - 112308515707551744n - 14144117579710464\sqrt{7}n + 14284855068065792\sqrt{6} + 53102495442325371)/\n\end{cases}
$$
\n9. *BMG*<sup>{d,s}</sup> (OX<sub>n</sub>) =   
\n
$$
\begin{cases}\n1728n^2 - 648n, \\
82944n^2 - 64848n + 7272\n\end{cases}
$$
\n
$$
10. \quad BMH
$$
<sup>{d,s}</sup> (OX<sub>n</sub>) =   
\n
$$
\begin{cases}\n1728n^2 - 648n, \\
82944n^2 - 64848n + 7272\n\end{cases}
$$
\n
$$
\begin{cases}\n216n^2 + (84\sqrt{2} - 144)n, \\
3(17732923532771328n^2 + 2973079441506304\sqrt{14}n + 3272146604261376\sqrt{7}n - 2973079441506304\sqrt{14} - 32721466
$$

We now provide the multiplicative self-powered degree as well as degree-sum indices of silicate frameworks for determining the numerical values of entropies. We denote  $SD =$ {(3, 3), (3, 6), (6, 6)} and *SS* = {(15, 15), (15, 24), (15, 27), (18, 27), (18, 30),(24, 27), (27, 27), (27, 30),(30, 30)}. Let  $\alpha(SL_n) = \prod_{(p,a)\in SD} \chi(p,a) \chi(p,a)$  and  $\beta(SL_n) = \prod_{(p,a)\in SS} \chi(p,a)$  $\chi(p, a)^{\chi(p, a)}$ . Therefore, the mathematical expressions for silicate frameworks are given by

- $\alpha \chi^{dp*}(\mathrm{SL}_n) = \alpha(\mathrm{SL}_n)(1944n^5 648n^4 432n^3)$
- $\chi^{sp*}(\mathrm{SL}_n) = \beta(\mathrm{SL}_n)(23219011584n^9 1896219279$ <sup>36</sup>*n*8+673351335936*n*7−1358312177664*n*6+<sup>1702727</sup> <sup>516160</sup>*n*<sup>5</sup> <sup>−</sup> <sup>1358312177664</sup>*n*<sup>4</sup> <sup>+</sup> <sup>673351335936</sup>*n*<sup>3</sup> <sup>−</sup> <sup>189621927936</sup>*n*<sup>2</sup> <sup>+</sup> <sup>23219011584</sup>*n*)

<span id="page-6-0"></span>**Table 3** Entropies based on  $\chi^{d}$ (SL<sub>*n*</sub>) and  $\chi^{s}$ (SL<sub>*n*</sub>) of silicate frameworks

Similary, for oxide frameworks, we denote  $OD =$  $\{(2, 4), (4, 4)\}, \ OS = \{(8, 12), (8, 14), (12, 14), (14, 14),$ (14, 16), (16, 16)},  $\alpha$ (OX<sub>n</sub>) =  $\prod_{(p,a)\in OD} \chi(p,a)^{\chi(p,a)}$ and  $\beta$ (OX<sub>n</sub>) =  $\prod_{(p,a)\in O}$   $\chi(p,a)$ <sup> $\chi(p,a)$ </sup>. Therefore, the mathematical expressions for oxide frameworks are given by

- $\alpha \chi^{dp*}(\text{OX}_n) = \alpha(\text{OX}_n)(216n^3 144n^2)$
- $\bullet$   $\chi^{sp*}(\text{OX}_n) = \beta(\text{OX}_n)(4478976n^5 24634368n^4 +$ <sup>53747712</sup>*n*<sup>3</sup> <sup>−</sup> <sup>58226688</sup>*n*<sup>2</sup> <sup>+</sup> <sup>31352832</sup>*<sup>n</sup>* <sup>−</sup> <sup>6718464</sup>)

As we can observe, the entropy formula involves incorporating mathematical expressions of topological indices and self-powered topological indices. The resulting mathematical expressions are longer in terms of dimension *n*. Therefore, we calculate the numerical entropy values for

degree and degree-sum index expressions for silicate and oxide frameworks at some fixed dimensions *n*, which are provided in Tables [3](#page-6-0) and [4.](#page-7-0)

Data scaling is an essential preprocessing step in various machine learning and statistical algorithms. Its primary objective is to transform the features of a dataset into a comparable scale, thereby preventing any single feature from unduly influencing the learning process of the models. The necessity for data scaling emerges from the distinct units, magnitudes, and ranges characterizing features within a dataset, potentially impeding the performance of models. As seen from our entropy calculations, the entropies of silicate and oxide frameworks have been computed based on the total number of bonds within those frameworks, which are not of equal quantity. Hence, we calculate the bond-wise entropy by scaling the total entropy through division by the number



<span id="page-7-0"></span>**Table 4** Entropies based on  $\chi^d$ (OX<sub>*n*</sub>) and  $\chi^s$ (OX<sub>*n*</sub>) of oxide frameworks



<span id="page-7-1"></span>



<span id="page-8-0"></span>**Fig. 5** Comparative graphs of scaled degree entropies between  $SL_n$  and  $OX_n$ 





<span id="page-8-1"></span>**Table 6** Scaled entropy values of silicate and oxide frameworks based on degree-sum indices

$\chi^s$	SL <sub>4</sub>	OX <sub>4</sub>	SL <sub>5</sub>	$OX_{5}$	SL65	OX <sub>6</sub>	SL <sub>7</sub>	OX <sub>7</sub>
G	0.01652	0.02879	0.01111	0.01950	0.00801	0.01414	0.00607	0.01076
H	0.00308	0.00705	0.00291	0.00609	0.00249	0.00504	0.00209	0.00417
BМ	0.02220	0.03847	0.01477	0.02576	0.01057	0.01851	0.00795	0.01398
T M	0.02401	0.04182	0.01594	0.02792	0.01138	0.02002	0.00855	0.01509
GН	0.02209	0.03836	0.01470	0.02550	0.01052	0.01833	0.00793	0.01385
<b>GBM</b>	0.00280	0.00613	0.00276	0.00560	0.00240	0.00474	0.00203	0.00397
$H$ $B$ $M$	0.00721	0.01391	0.00521	0.01002	0.00394	0.00757	0.00309	0.00593
HT M	0.00415	0.00889	0.00348	0.00711	0.00284	0.00568	0.00232	0.00461
BMG	0.01665	0.02923	0.01119	0.01978	0.00807	0.01433	0.00611	0.01090
BMH	0.02781	0.04779	0.01839	0.03178	0.01309	0.02272	0.00981	0.01708
T M G	0.01846	0.03258	0.01236	0.02194	0.00888	0.01584	0.00671	0.01200
TMH	0.02962	0.05114	0.01956	0.03394	0.01390	0.02422	0.01041	0.01819

<span id="page-9-0"></span>





of bonds in both silicate and oxide frameworks. That is, for the index function  $\chi$  and  $G \in \{SL_n, OX_n\}$ ,

Scaled Entropy of  $G = \frac{I_X(G)}{|E(G)|}$ 

Table [5](#page-7-1) provides a detailed comparison between silicate and oxide frameworks, highlighting their bond-wise degree entropies values and emphasizing the consistent higher values of bond-wise entropies in  $OX_n$  compared to  $SL_n$  as shown in Fig. [5.](#page-8-0) The trend persists for bond-wise degreesum entropies, where the scaled entropies are provided in Table [6](#page-8-1) and illustrated in Fig. [6.](#page-9-0)

This comparative analysis serves as a crucial tool in unveiling the relative disorder or randomness within these systems. It provides a quantitative measure, enabling the assessment and ranking of their respective complexities, thereby aiding predictions of stability under varied conditions. Ultimately, this comparative entropy analysis enhances the understanding of silicate and oxide frameworks and their implications in structural properties, facilitating applications in material

design and property optimization across various scientific and industrial fields.

# **4 Conclusion**

In this paper, we have investigated topological indices and entropy measures to comprehend the structural characteristics of silicate and oxide frameworks. We have derived the topological expressions for recently proposed indices and conducted a scaled entropy analysis between these two frameworks. Our formulation of topological expressions coupled with scaled entropy has revealed a higher entropy in oxide frameworks relative to their silicate counterparts. This observation highlights the intricate structural arrangements and versatile properties of silicate and oxide frameworks, providing valuable insights for future advancements.

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**Availability of data and materials** No data are used for the research.

#### **Declarations**

**Ethics Approval** Not Applicable

**Consent to participate** Not Applicable

**Consent for publication** All authors have approved the manuscript and given consent for publication.

**Competing interests** The authors declare no competing interests.

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