



# Microstructure and Mechanical Property Correlation of Mg-Si Alloys

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## Abstract

The present work aims at understanding the microstructure and mechanical property correlation of hypo (Mg-0.5, 0.7, 1.15 wt% Si) and hyper (Mg-2, 4, 6, 8 and 10 wt% Si) eutectic binary Mg-Si alloys. The microstructures of hypoeutectic alloys consist of  $\alpha$ -Mg and lamellar Mg-Mg<sub>2</sub>Si eutectic phases; whereas the microstructures of hypereutectic alloys consist of primary Mg<sub>2</sub>Si, Chinese script eutectic Mg<sub>2</sub>Si and  $\alpha$ -Mg phases depending upon the Si content. Significant increase in hardness values of these alloys was observed as a function of Si content which is due to the presence of hard Mg<sub>2</sub>Si phase. In the hypereutectic alloys, the average microhardness of primary Mg<sub>2</sub>Si was about ten times higher than that of  $\alpha$ -Mg. Compressive behaviour, tensile properties (UTS, YS and %E) at the room temperature (RT) as well as at 150 °C, elevated temperatures (ET) were analyzed and reported as follows: compressive strength of Mg-Si alloys has improved with the increase of Si content, Mg with 2 to 4 % Si shows good tensile properties at RT and ET. Alloys containing high Si content show a marginal reduction in strength at high temperatures due to the presence of the thermally stable Mg<sub>2</sub>Si.

**Keywords** Mg-Si alloys · Casting and solidification · Mg<sub>2</sub>Si · Hardness · Mechanical properties

## 1 Introduction

Nowadays, the design and development of ‘safer, lighter and cleaner’ vehicles are the subjects of worldwide energy-efficient research. In recent years, the aerospace and automobile industries have shown an increasing demand for novel thermo-mechanically reliable light materials for the overall weight-reduction which in turn limit the exhaust CO<sub>2</sub> emissions as per the legislative requirements [1–8]. Magnesium, with a density of 1.7 g/cc, which is two-third of the density of aluminium (2.7 g/cc) and one fourth of the density of steel (7.8 g/cc), is the lightest of the structural materials [1]. Mg alloys have an excellent specific strength, stiffness, dimensional stability, high

damping capacity and recyclability. Most of the commercially used Mg alloys such as AZ91 (Mg-Al-Zn), AM50/60 (Mg-Al-Mn) and AS21/41 (Mg-Al-Si) have reasonably high Al content and are the most economically competitive alloys [9–17]. They offer a combination of excellent castability, good corrosion resistance, reasonable room temperature strength and high ductility [18]. Their use in the automobile and electronic industries is restricted to relatively low-temperature applications. They are also unsuitable for power train applications in the automobile industry due to their poor mechanical properties at ET [19–21]. This is due to the presence of Mg<sub>17</sub>Al<sub>12</sub> intermetallic in these alloys, which limits their strength and creep properties at ET [22, 23]. Hence, the development of low Al or Al free Mg alloys suitable for ET applications have been seriously considered and research attempts are underway.

Mg-alloys added with elements such as Ag [24, 25], Zr [26], Y [27] and rare earth (RE) [28–30] show good mechanical strength and creep resistance at ET. However, the use of such elements is uneconomical for fabricating high performance structural components. This situation opens the door for the development of low cost, thermally stable Mg-based alloys [8]. Silicon (Si) being a cheap element has long been used as an alloying element in Mg alloys to improve the ET properties [31]. Currently, the Mg-Si alloys containing in-situ intermetallic Mg<sub>2</sub>Si have shown great potential as structural

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materials [32]. The maximum solid solubility of Si in Mg is only 0.003 at%, and Si atoms react with Mg atoms and precipitate as  $Mg_2Si$  intermetallic. The stable  $Mg_2Si$  that exists at stoichiometric composition exhibits high melting temperature (1085 °C), low density ( $1.99 \times 10^3 \text{ kg/m}^3$ ), high hardness ( $4.5 \times 10^9 \text{ N/m}^2$ ), low thermal expansion coefficient ( $7.5 \times 10^{-6} \text{ K}^{-1}$ ) and reasonably high elastic modulus (120 GPa) which contributes to the improvement of mechanical properties, heat resistance and wear resistance of the Mg-Si alloys [33–37].

Several researchers have reported that the addition of Si elements in Mg-Al alloys such as AS21 and AS41 alloys, as the minor alloying element for die casting applications. These alloys are developed in the 1970 s and AS42 alloy was produced in the late 1980's. Although these alloys were significantly creeping resistant, they did not meet the requirements on transmission parts due to their poor casting and mechanical properties [38].  $Mg_2Si$  intermetallic has many advantageous properties, but the binary Mg-Si based alloys for sand or permanent mold cast applications were not explored.

Elevated temperature mechanical properties and high strain rate superplasticity in Mg-Si alloys were also investigated by Mabuchi et al. [32, 39, 40]. They have studied the Mg-Si alloys with various amount of Si additions and reported that the alloys with low Si contents ( $< 10 \text{ wt}\%$ ) showed higher values of the ultimate tensile strength than that of an alloy having higher Si contents ( $\geq 10 \text{ wt}\%$ ) between 20 and 100 °C. Moreover, they have also reported that rapid solidified (RS) Mg-Si alloys show higher strength of 500 MPa at RT and superplastic behaviour at a high strain rate of  $10^{-1} \sim 1 \text{ S}^{-1}$  at 500 °C [40]. However, a complete understanding of the structural property correlation of Mg-Si alloys is essential for the further development of these alloys. Therefore, the present work is taken up with the aim to investigate the microstructural and mechanical properties of various Mg-Si alloys.

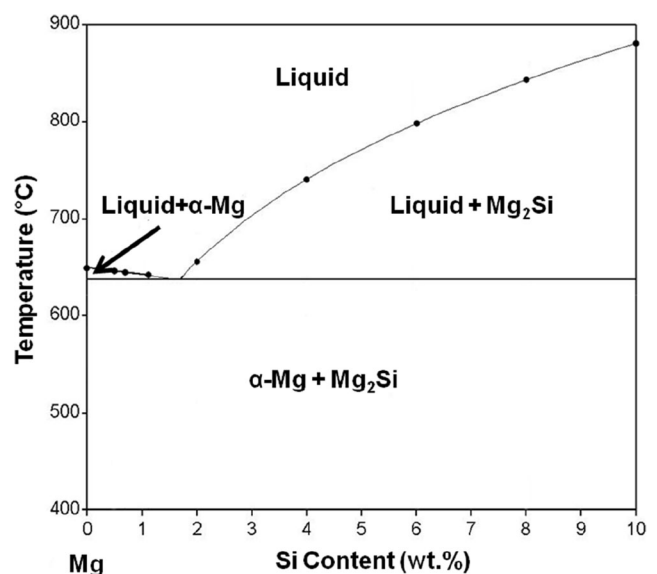
**Table 1** Holding temperature and time for various Mg-Si alloy composition

Mg-Si alloy	Holding temperature (°C)	Holding time (Min.)
Pure Mg	720	15
Mg-0.5Si	725	30
Mg-0.7Si	730	45
Mg-1.15Si	735	60
Mg-2Si	750	75
Mg-4Si	840	135
Mg-6Si	880	195
Mg-8Si	930	255
Mg-10Si	980	315

## 2 Experimental Details

Industrially pure magnesium ingot (99.7 %) and silicon lumps (99.5 %) were used as starting materials to prepare the Mg-Si alloys. Various compositions of these materials were melted separately in a low carbon steel crucible using a resistance furnace under a protective flux atmosphere. With reference to the Mg-Si phase diagram, different processing temperatures (750 to 890 °C) were maintained depending on the amount of Si addition for the complete dissolution of Si. For better dissolution, similar size range Si lumps were added together as a sandwich layer with Mg. Using this procedure, dissolution time and temperature of Mg-Si alloys were standardized. The melt was stirred which helped in dissolving the Si lumps uniformly in the melt. The holding temperature was always kept at 100 °C above the liquidus temperature for the respective compositions of Mg-Si alloys, which was obtained from the Mg-Si phase diagram (Fig. 1). The holding time was also standardized as 15 min for every 0.5 % Si addition. The standardized time and temperature were obtained by carrying out a number of trial experiments. The holding temperature and time for various Mg-Si alloy compositions are given in Table 1.

Later, the melt was poured into a preheated (350°C) rectangular steel mold having dimensions 250mm×200mm×30mm. Castings of pure Mg and Mg alloys containing different Si (0.5, 0.7, 1.15, 2, 4, 6, 8 and 10 wt%) contents were prepared. Samples for material characterization such as optical microscopy, hardness, tensile, compressive were machined out from the same location of all the castings as per the ASTM Standards. For microstructural analysis, samples were etched using picric acid-based etchant (Picric acid – 6 g, Acetic acid – 5 ml, Ethanol – 100ml, Distilled water – 10 ml). Phase analysis and



**Fig. 1** Expanded Mg-Si phase diagram up to 10 wt% [29]

microstructural observation were carried out using X-ray diffractometer (PHILIPSPW1710) and optical microscopy (Leitz-Metalloplan). Quantitative microstructural analysis was carried out using a Leica 2001 image analyzer in conjunction with the optical microscope. The fields of observation were selected randomly at different locations of the sample. The sizes of different phases were measured manually. For each case, at least ten fields were analyzed from a single specimen and the average value is reported. To identify the type of precipitates on various castings the microstructural, tensile fracture and compressive fracture samples surface were examined using a JEOL, JSM 35 C scanning electron microscope with the operating voltage of 15–30 KeV.

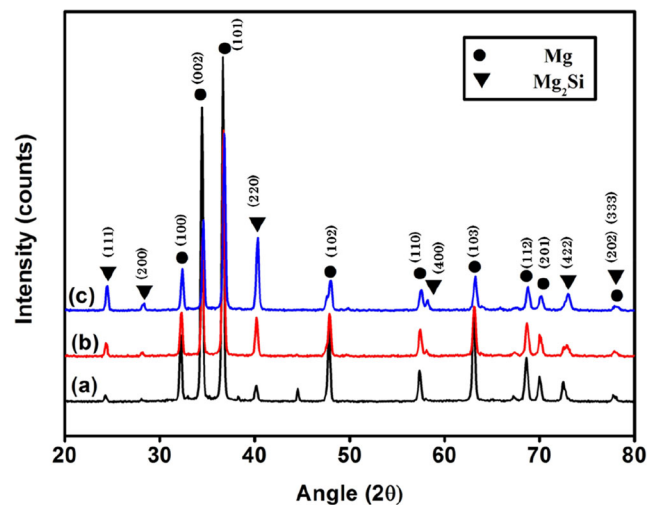
Hardness measurements were carried out using Vickers hardness tester (Model: LV700, LECO Corporation, Michigan, U.S.A) to identify the hardness of the primary phase, eutectic region and the matrix of the Mg-Si alloys. Vickers hardness (macro) measurements were carried out using an indentation load of 5 kgf allowing a dwell time of 15 s for each indentation. The Vickers microhardness measurements were carried out on the polished samples using the fully automated CLEMEX microhardness tester. A 25gf constant load was maintained throughout the experiments with a 15 s dwell time. Average of ten indentations was taken for each specimen, and the scatter values were reported. RT and ET tensile tests were carried out using a computer-controlled INSTRON 8801 universal testing machine. Samples for tensile testing were prepared in accordance with the ASTM E8 standards. For HT testing, the samples were heated on an oven attached to the INSTRON machine and kept for 15 min after the required temperature was reached so as to ensure the homogenization of temperature throughout the sample. The compressive testing of the Mg-Si alloys were conducted at RT on a computer-controlled INSTRON 8801 Universal Testing Machine using the crosshead speed of 0.5 mm/min. Samples were machined as per ASTM E9-09 standards.

## 3 Result and Discussion

### 3.1 XRD Analysis

Figure 2 shows the typical XRD patterns of Mg-Si alloys (Mg-1.15Si, Mg-6Si and Mg-10Si alloys). Only representative XRD patterns of hypo (Mg-1.15Si) and hypereutectic (Mg-6Si, Mg-10Si) alloys are presented in this figure. XRD patterns of all the alloys show the peaks of Mg and Mg<sub>2</sub>Si phases. The absence of Si peak indicates that all the added Si lumps completely reacted in the Mg melt forming Mg<sub>2</sub>Si phase during solidification.

Table 2 gives the relative intensity of the major representative Bragg reflection of the Mg<sub>2</sub>Si phases in Mg-Si alloys. It



**Fig. 2** XRD analysis of Mg-Si alloys (a) Mg-1.15Si, (b) Mg-6Si and (c) Mg-10Si

was found that with an increase in Si additions the relative intensity of all the peaks of Mg<sub>2</sub>Si increases. This increase in intensity indicates that the quantity of Mg<sub>2</sub>Si increases with an increase of Si content addition.

### 3.2 Microstructure

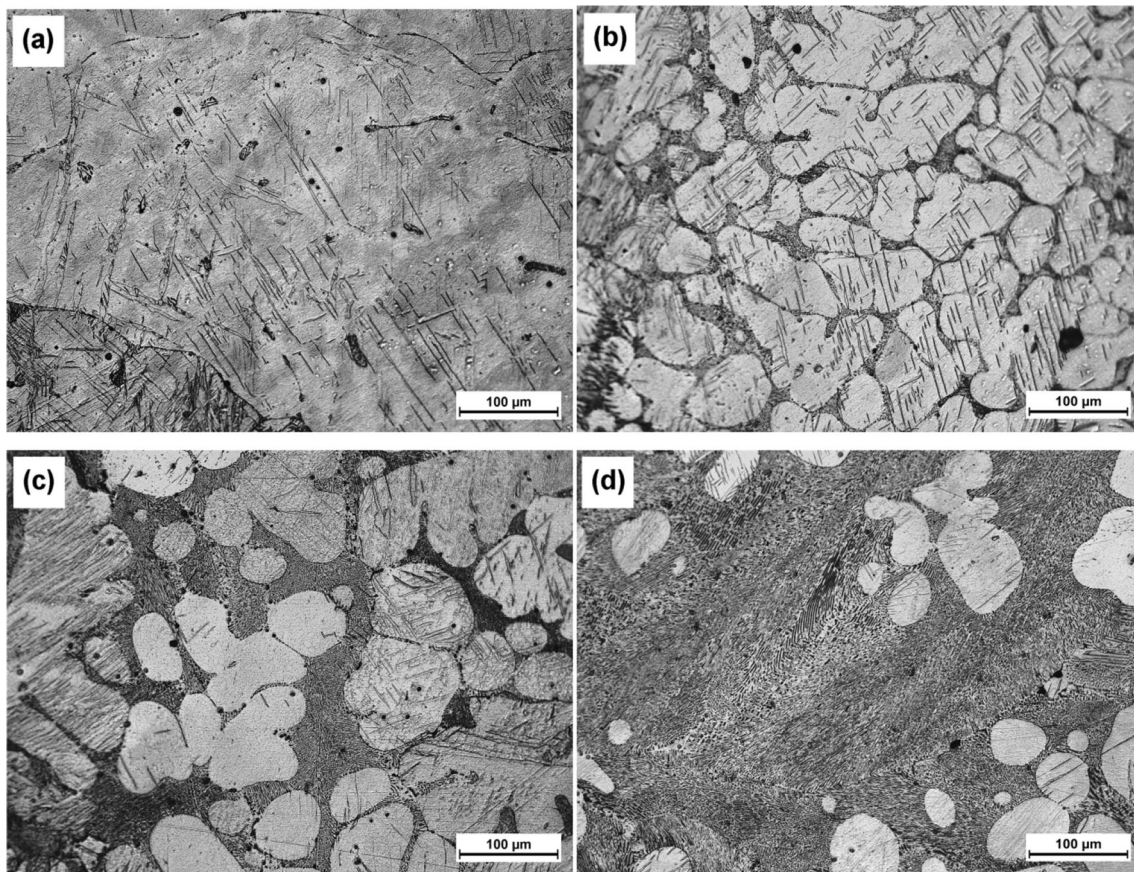
#### 3.2.1 Hypoeutectic Mg-Si Alloys

Figure 3 presents the optical micrographs of pure Mg and different hypoeutectic Mg-Si (0.5, 0.7, 1.15Si) alloys. The microstructure of pure Mg (Fig. 3a), consists of single-phase Mg with a coarse grain structure. With 0.5 Si addition to Mg (Fig. 3b), the  $\alpha$ -Mg grows as a dendritic structure and the eutectic forms in the inter-dendritic region. Similarly, Mg-0.7Si (Fig. 3c) and Mg-1.15Si (Fig. 3d) alloys also show dendritic  $\alpha$ -Mg with lamellar eutectic structure.

To identify the phases present in the microstructure, as per the Mg-Si binary diagram [41], the eutectic equilibrium reaction on the Mg-rich side,  $L \rightarrow \text{Mg} + \text{Mg}_2\text{Si}$ , occurs at 637.6 °C at 1.34 wt% Si [42]. With an increase in Si content, the amount of the eutectic Mg<sub>2</sub>Si phase increases and becomes coarse and continuous.

**Table 2** Relative intensity of major representative Bragg diffraction of Mg<sub>2</sub>Si

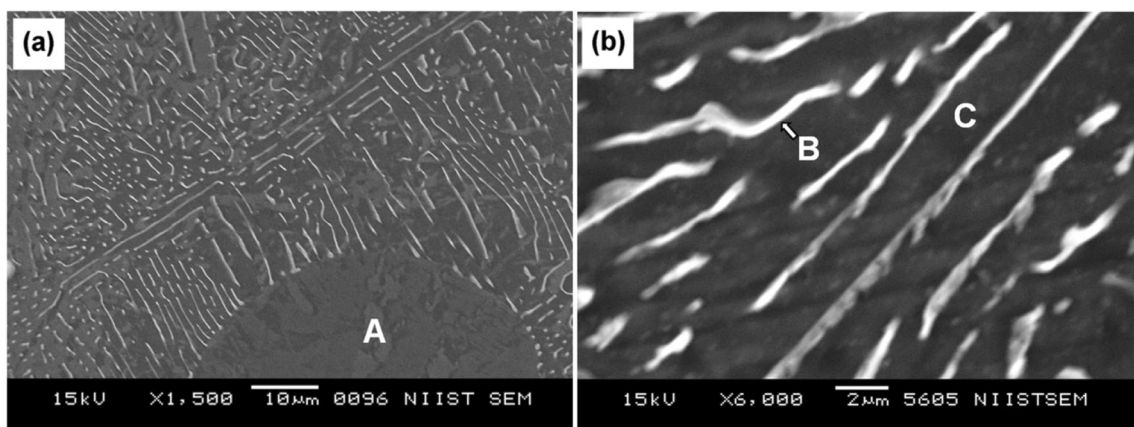
Sample detail	Relative intensity % ( $I/I_0$ ) of ( $hkl$ )						
	(111)	(200)	(220)	(222)	(400)	(420)	(422)
Mg-1.15Si	1.84	0.56	4.65	-	-	27.18	-
Mg-6Si	5.96	1.1	16.89	0.53	2.79	21.81	4.91
Mg-10Si	13.76	3.21	40.35	0.33	5.61	18.61	9.29



**Fig. 3** Microstructures of (a) pure Mg, (b) Mg-0.5Si, (c) Mg-0.7Si and (d) Mg-1.15Si

SEM-EDS analyses on the different regions were carried out in Mg-1.15Si alloy (Fig. 4) and the results are shown in Table 3. Region A consists of 99.87 at% Mg and 0.13 at% Si, and therefore, it is confirmed as  $\alpha$ -Mg phase. The region B has higher Si content (20.17 at%) and hence it is confirmed as eutectic  $Mg_2Si$ . The region C is the inter-lamellar region which has 4.53 at% Si and is eutectic  $\alpha$ -Mg.

Similarly, the amount of  $\alpha$ -Mg gets reduced as the Si content increases. In the hypoeutectic Mg-Si alloys, the eutectic grows in a lamellar manner from the solid  $\alpha$ -Mg phase surface. As the  $\alpha$ -Mg phase solidifies, excess Si diffuses on a short distance laterally where it is incorporated in the Si-rich  $Mg_2Si$  phase. Similarly, the Mg atoms rejected ahead of the  $Mg_2Si$  diffuse to the tip of



**Fig. 4** SEM micrograph of Mg-1.15Si alloy indicating regions A, B and C selected for EDS analyses. (a) Low magnification micrograph and (b) High magnification micrograph

**Table 3** EDS analyses of regions A, B and C marked in Fig. 4 (Weight and atomic fraction %)

Element/Regions	MgK		SiK	
	wt%	at%	wt%	at%
A	99.85	99.87	0.15	0.13
B	79.83	83.49	20.17	16.51
C	95.48	96.06	4.53	3.94

the adjacent  $\alpha$ -Mg lamella and hence a lamellar eutectic forms [43].

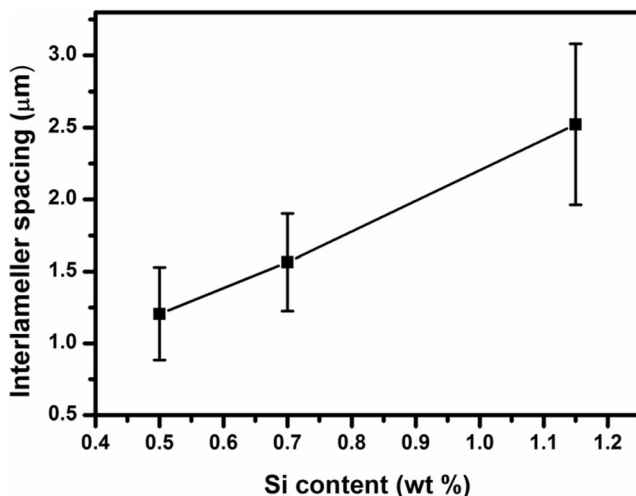
The interlamellar spacing of eutectic Mg-Mg<sub>2</sub>Si regions is measured using image analyses software and is shown in Fig. 5. It can be seen that interlamellar spacing increases with an increase in Si content in the hypoeutectic alloys.

### 3.2.2 Hypereutectic Mg-Si Alloys

Figure 6 shows the microstructures of different hypereutectic Mg-Si alloys (Mg-2 to 10Si). In general, the microstructures of these hypereutectic alloys consist of black polygonal/dendritic shaped particles surrounded by white regions containing needle/Chinese script like phases.

Figure 7 represents the SEM-EDS analyses carried out on different regions A, B, and C to identify the phase present in the alloy. The EDS analyses results of the marked regions are presented in Table 4.

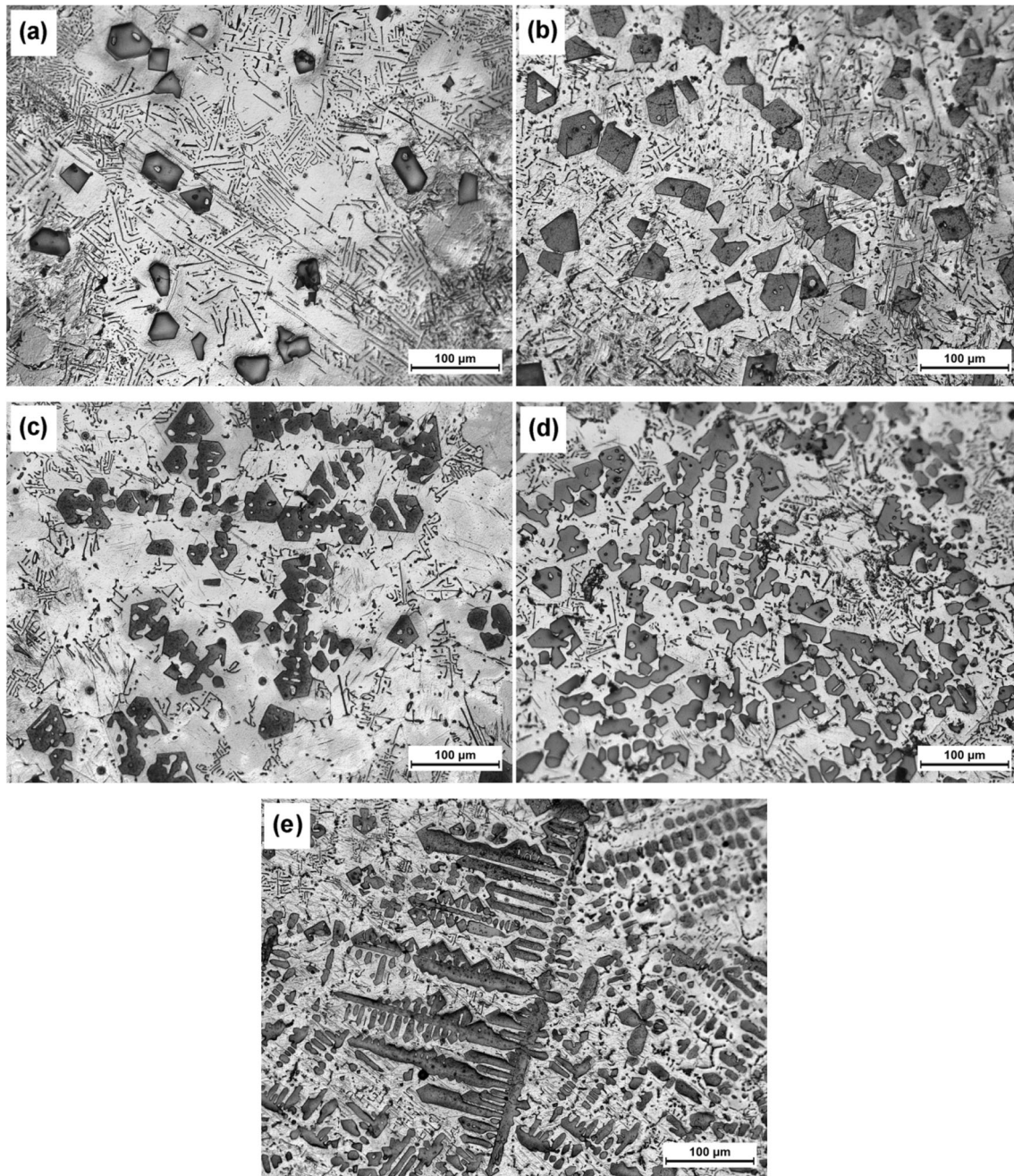
The region A consists of 34.27 at% Si and 65.73 at% Mg, which suggests that it is the primary Mg<sub>2</sub>Si phase. The primary  $\alpha$ -Mg (region B) consists of a very little amount of Si (0.13 at%). Region C has 22.09 at% Si and it is known as eutectic Mg<sub>2</sub>Si. However, the Si content present in the eutectic Mg<sub>2</sub>Si is less than that of the primary

**Fig. 5** Interlamellar spacing of different hypoeutectic Mg-Si alloys

Mg<sub>2</sub>Si. This is due to the occurrence of an error during EDS measurements; the error occurred because the penetration depth of the electron beam is higher than the thickness of the eutectic Mg<sub>2</sub>Si. From the result, it can be concluded that the black colour polygonal phase is primary Mg<sub>2</sub>Si and the white surrounding region is  $\alpha$ -Mg. It is apparent from the microstructure (Fig. 6a) that the needle/lamellar shape phase is eutectic Mg<sub>2</sub>Si. It is observed that most of the  $\alpha$ -Mg is seen around the primary Mg<sub>2</sub>Si phase and the size of the primary Mg<sub>2</sub>Si varies from 5 to 25  $\mu$ m. When the Si addition increases to 4 %, the size and volume of the primary Mg<sub>2</sub>Si also get increased (Fig. 6b). The morphology of eutectic Mg<sub>2</sub>Si changes from the long needle/lamellar to Chinese script type structure. When Si addition is further increased to 6 % most of the primary Mg<sub>2</sub>Si is converted to dendritic structure (Fig. 6c). In eutectic phase the Chinese script morphology of Mg<sub>2</sub>Si is more apparent. As the Si content increases further to 8 %, and 10 % the primary Mg<sub>2</sub>Si become coarser and dendritic (Fig. 6d and e). In addition, the amount of eutectic Mg<sub>2</sub>Si also decreases with an increase in Si content in hypereutectic alloys. The size of the primary Mg<sub>2</sub>Si particle in Mg-Si alloys was measured using image analysis software (Fig. 8). The average of the longest and shortest dimension of the particles was considered for the size measurement because the particles have no definite shape.

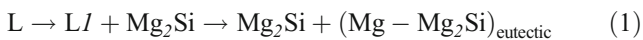
In case of dendrites (Mg-6 to 10 Si alloy), the primary dendritic arm size is taken as the size of the Mg<sub>2</sub>Si. Difficulties were encountered during the measurement of the dendrite size as the microstructure consists of different dendrite lengths, and hence large deviation in size was observed on high Si alloys (Fig. 6d and e). The volume percentage of the primary Mg<sub>2</sub>Si estimated using image analysis software and the theoretical percentage obtained from Pandat software are shown in Fig. 9. The results clearly show that the Mg<sub>2</sub>Si content increases with an increase in Si content in the alloys which is in line with the microstructural observation.

Figure 10 shows the schematic representation of the microstructural development of hypereutectic Mg-Si alloys during solidification. According to the Mg-Si binary equilibrium phase diagram, the hypereutectic Mg-Si alloys would solidify resulting in a microstructure comprising of primary Mg<sub>2</sub>Si and eutectic Mg-Mg<sub>2</sub>Si. The solidification path is indicated as follows [39]. During solidification, the first phase precipitating from the liquid melt is the primary Mg<sub>2</sub>Si and then the rest of the melt solidifies as a binary eutectic structure consisting of  $\alpha$ -Mg and eutectic Mg<sub>2</sub>Si. The schematic diagram of primary and eutectic growth of Mg<sub>2</sub>Si is presented in Fig. 11a-c. During growth, the primary Mg<sub>2</sub>Si consumes Si atoms from the surrounding liquid and hence the Si depleted liquid forms around it and reaches the composition of  $\alpha$ -Mg. When the temperature drops to 650  $^{\circ}$ C, the  $\alpha$ -Mg solidifies surrounding

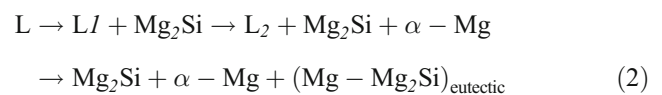


**Fig. 6** Microstructures of hypereutectic Mg-Si alloys (a) Mg-2Si, (b) Mg-4Si, (c) Mg-6Si, (d) Mg-8Si and (e) Mg-10Si

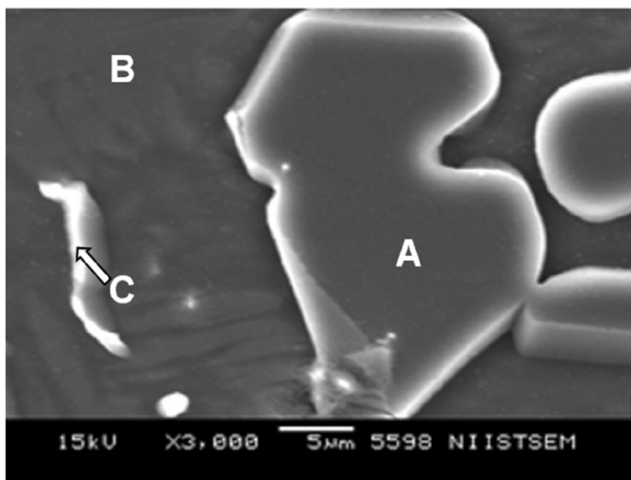
the primary  $Mg_2Si$ . The remaining liquid melt, which is near or equal to the eutectic composition, will solidify as eutectic  $Mg_2Si$  at  $637\text{ }^\circ\text{C}$  as shown in Eq. 1.



The formation of above microstructure feature is mainly attributed to the solidification of the alloy under relatively high cooling rate, leading to deviation from the equilibrium diagram during solidification. Hence, under non-equilibrium condition, such as in the case of ingot metallurgy, the solidification path is indicated as in the Eq. 2 [44].



Moreover with respect to Si addition, the morphology of the primary  $Mg_2Si$  also varies. From Figs. 10 and 11, it is noted that, for low Si-containing hypereutectic alloys, solidification starts at a lower temperature. For instance, in the case of Mg-2Si alloy, the primary  $Mg_2Si$  start solidifying at  $657\text{ }^\circ\text{C}$ . However, at higher Si content (Mg-10Si), the alloys starts solidifying at very high temperatures ( $880\text{ }^\circ\text{C}$ ) as

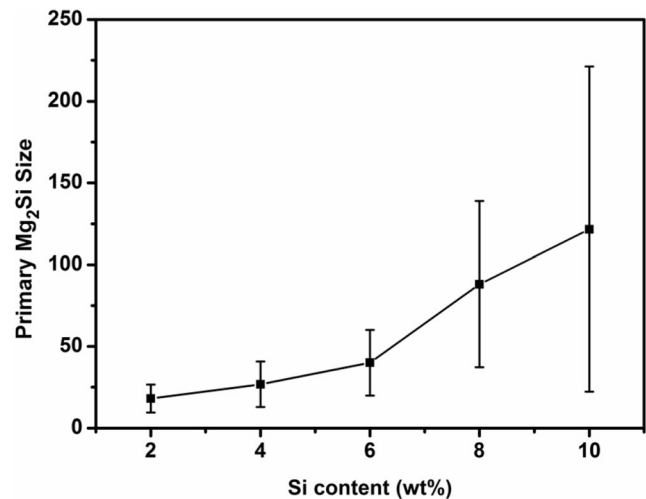


**Fig. 7** SEM micrograph of Mg-2Si indicating regions A, B, and C selected for EDS analyses

compared to the lower Si added alloys. Therefore at higher addition the primary  $Mg_2Si$  gets more time to solidify thus showing the coarse dendritic type morphology. The morphology change of the primary  $Mg_2Si$  depends on the Si content in the alloy. Porter and Easterling, [43] have reported that other than dendritic morphology, primary  $Mg_2Si$  exhibits three typical morphologies: perfect, hopper and truncated octahedron. Because of the random cutting angles, a variety of polygonal outlines of primary  $Mg_2Si$  in the polished sections such as triangle, square, trapezoid and hexagon are observed [45]. Theoretically, the growth of primary  $Mg_2Si$  crystals follow a typical faceted growth pattern in conventional solidification conditions, owing to its high entropy of fusion and large Jackson's factor [45, 46]. The final morphology of the crystal is determined by the crystal structure, the crystal growth conditions and process. The internal factors lead to equilibrium crystal form with minimum total surface free energy. Various morphologies are developed when different external factors also influence crystal forms and force the crystal to deviate from the equilibrium form. The crystal shape is derived from the competition of these internal and external factors, and its growth is related to the growth kinetics, such as interface property, capillarity, heat and mass transfer. The eutectic  $Mg_2Si$  forms towards the end of solidification at 637 °C and it grows as a needle or Chinese script type morphology in the

**Table 4** EDS analyses of region A, B and C

Regions marked in Fig. 7	Mg		Si	
	wt%	at%	wt%	at%
A	62.41	65.73	37.59	34.27
B	99.85	99.87	0.15	0.13
C	75.63	77.91	24.37	22.09



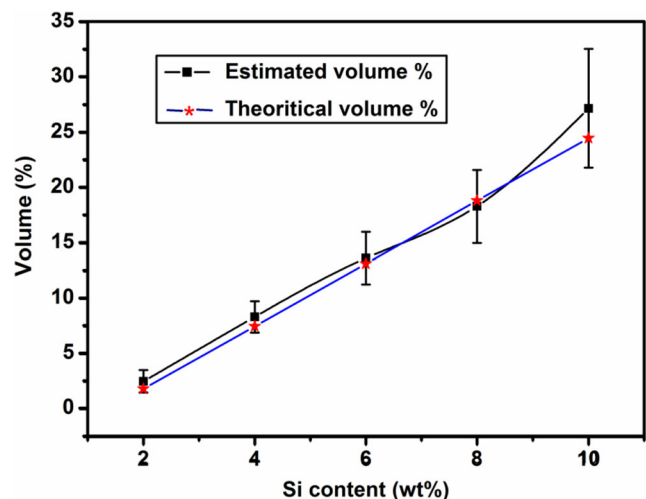
**Fig. 8** Variation in size of the primary  $Mg_2Si$  particle with Si content

hypereutectic alloys. This is in sharp contrast to the lamellar type of eutectic growth seen in the case of the hypoeutectic Mg-Si alloys. The morphological change of eutectic constituent in hypereutectic alloys is probably due to the high range of non-equilibrium solidification (from liquidus to solidus) and insufficient amount of Si atom for coupled eutectic growth at the final stage of solidification.

### 3.3 Mechanical Properties

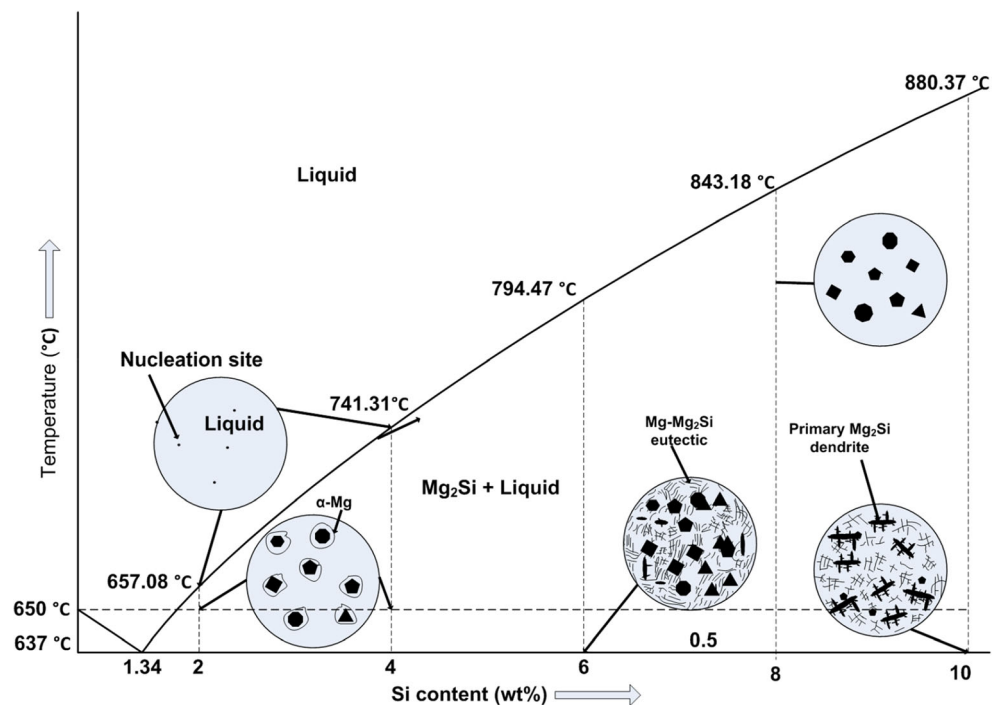
#### 3.3.1 Macrohardness

Figure 12 reveals the Vickers hardness of pure Mg and Mg-Si alloys. The hardness value steadily increases with an increase in Si content. The hardness of pure Mg increases from 30 HV5 to 72 HV5 with 10Si addition. The enhancement with an increase in the Si additions in hardness is due to the presence of hard  $Mg_2Si$  intermetallic phase. Moreover, increasing the



**Fig. 9** Theoretical and estimated volume percentage of  $Mg_2Si$  in hypereutectic Mg-Si alloys

**Fig. 10** Schematic representation of the microstructural development of hypereutectic Mg-Si alloys during solidification



volume of Mg<sub>2</sub>Si with respect to Si addition increases the hardness. While comparing with hypo, the hypereutectic Mg-Si alloys have higher hardness, which is due to the presence of the hard and brittle primary Mg<sub>2</sub>Si phase in the α-Mg matrix.

### 3.3.2 Microhardness

Indentation microhardness testing is a convenient technique to investigate the mechanical properties of a local, small volume of material. The Vickers microhardness (HV) is defined as in Eq. 3:

$$HV = 1.8544P/d^2 \quad (3)$$

Where P is the applied load and d is the indentation size. Figure 13 shows the Vickers microhardness of α-Mg, eutectic Mg-Mg<sub>2</sub>Si and primary Mg<sub>2</sub>Si phases present in both the hypo and hypereutectic Mg-Si alloys. It is noticed that the average hardness of the primary Mg<sub>2</sub>Si (350 HV0.025) is about 10 times higher than that of α-Mg (32 HV0.025).

The hardness of α-Mg and eutectic Mg-Mg<sub>2</sub>Si phase in different alloys do not change much in spite of the change in the eutectic morphology of the alloys. A relatively large variation in hardness is observed on the primary Mg<sub>2</sub>Si phase alloy, which is mainly due to the encountered difficulties during measurement.

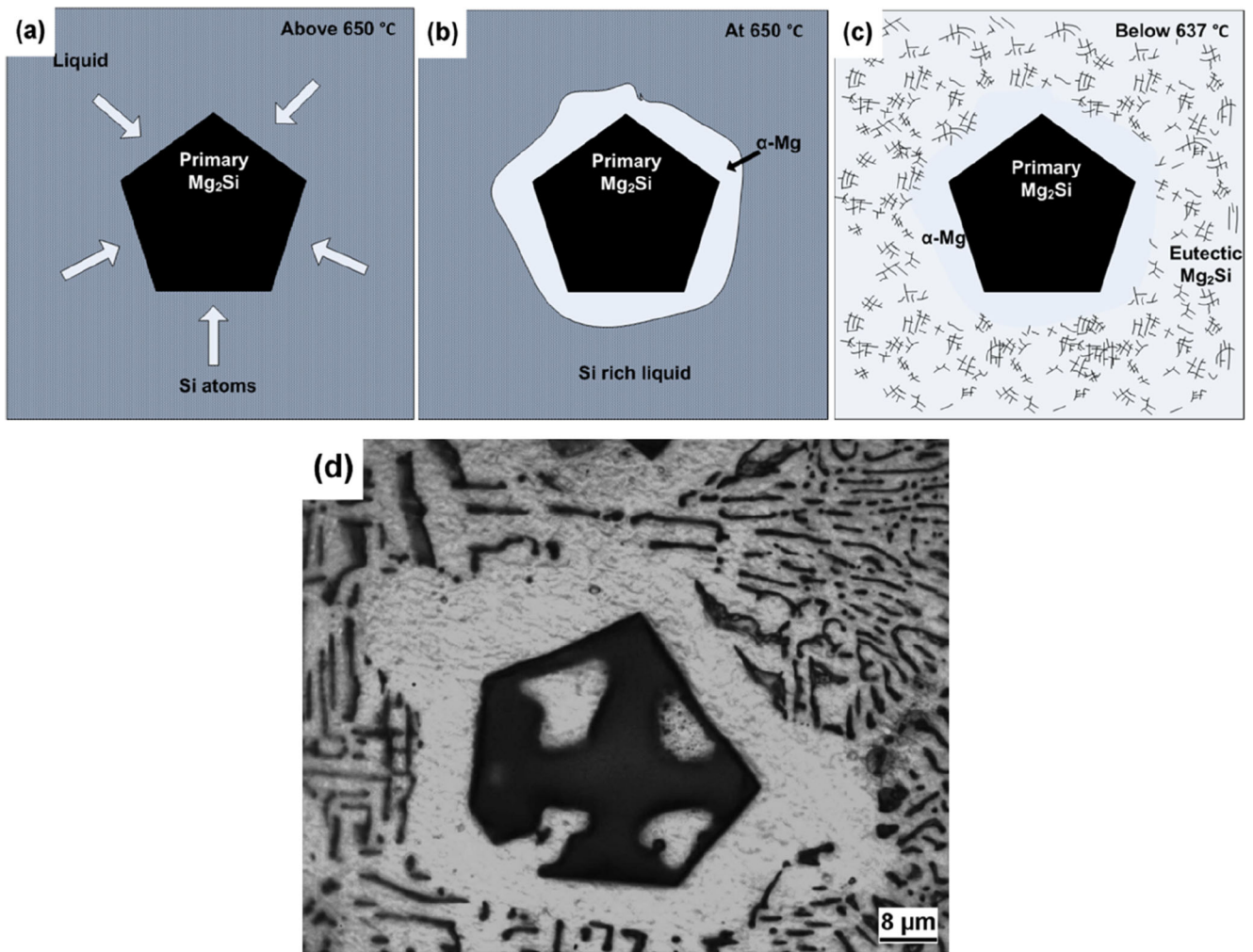
Micro indentation of the microhardness test are indicated in Fig. 14(a to d). Figure 14(a) represents the presence of low hardness α-Mg which is similar to pure Mg; the edges of the indentation are not sharp which indicate its ductility. The

eutectic Mg-Mg<sub>2</sub>Si region shows comparatively higher hardness than α-Mg in both the hypo and hypereutectic alloys (Fig. 14b). In hypereutectic alloys, the primary Mg<sub>2</sub>Si phase exhibits the highest hardness among all the other constituents. Figure 14(c), (d) shows the primary Mg<sub>2</sub>Si before and after indentation. For the same load on primary Mg<sub>2</sub>Si, the indentation mark observed is very small and it produces crack at the corners indicating its brittle nature. The crystal structure of the stoichiometric Mg<sub>2</sub>Si is diamond cubic, positioned by Si atoms, whereas the tetrahedral interstices are occupied by Mg atoms. Due to the restricted number of active slip systems Mg<sub>2</sub>Si reveals an inherent brittleness up to the brittle-ductile transition temperature of about 450°C [47].

### 3.3.3 Tensile Properties

**Room Temperature (RT)** The ultimate tensile strength (UTS), yield strength (YS) and the ductility (%E) values of different Mg-Si alloys at RT are shown in Fig. 15. The UTS values increase up to 2 % Si addition; beyond that, it decreases slightly (from 148 MPa to 139 MPa) in UTS is noticed with Mg-4Si alloy. However, the UTS value of Mg-6Si reduces drastically to 97 MPa which is even less than that of pure Mg (111 MPa) and thereafter, they do not change much. Even though the increase in YS up to 0.7Si is not significant, a gradual increase is noticed beyond that. The maximum YS is obtained for Mg-6Si alloy (77 MPa) and beyond that marginal reduction is observed. However, the YS of Mg-10Si alloy are higher than that of pure Mg. The ductility, as measured by the



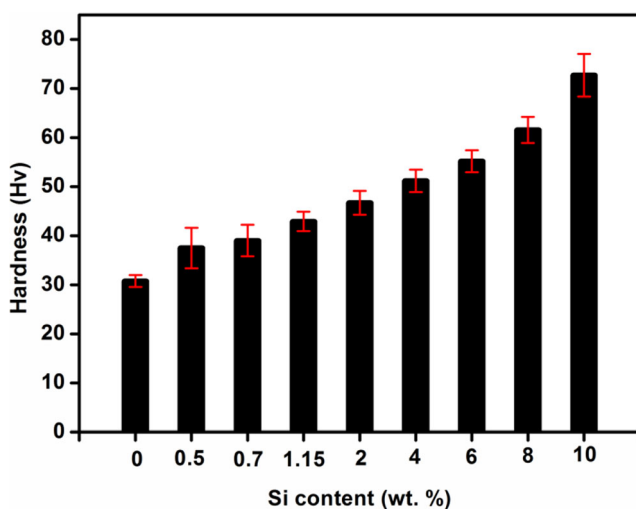


**Fig. 11** Schematic representations of (a) primary  $Mg_2Si$  growth above  $650\text{ }^\circ\text{C}$ , (b)  $\alpha\text{-Mg}$  growth surround the primary  $Mg_2Si$  at  $650\text{ }^\circ\text{C}$  (c)

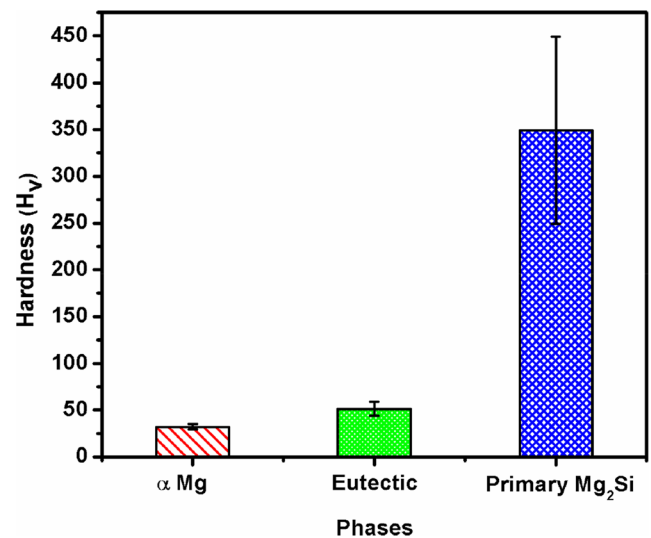
Eutectic  $Mg_2Si$  growth at  $637\text{ }^\circ\text{C}$  and (d) Typical microstructure  $Mg\text{-}4Si$  alloy

%E gradually decreases with Si additions. The ductility of  $Mg\text{-}10Si$  is 1.94 % whereas the ductility of pure Mg is 11.78 %.

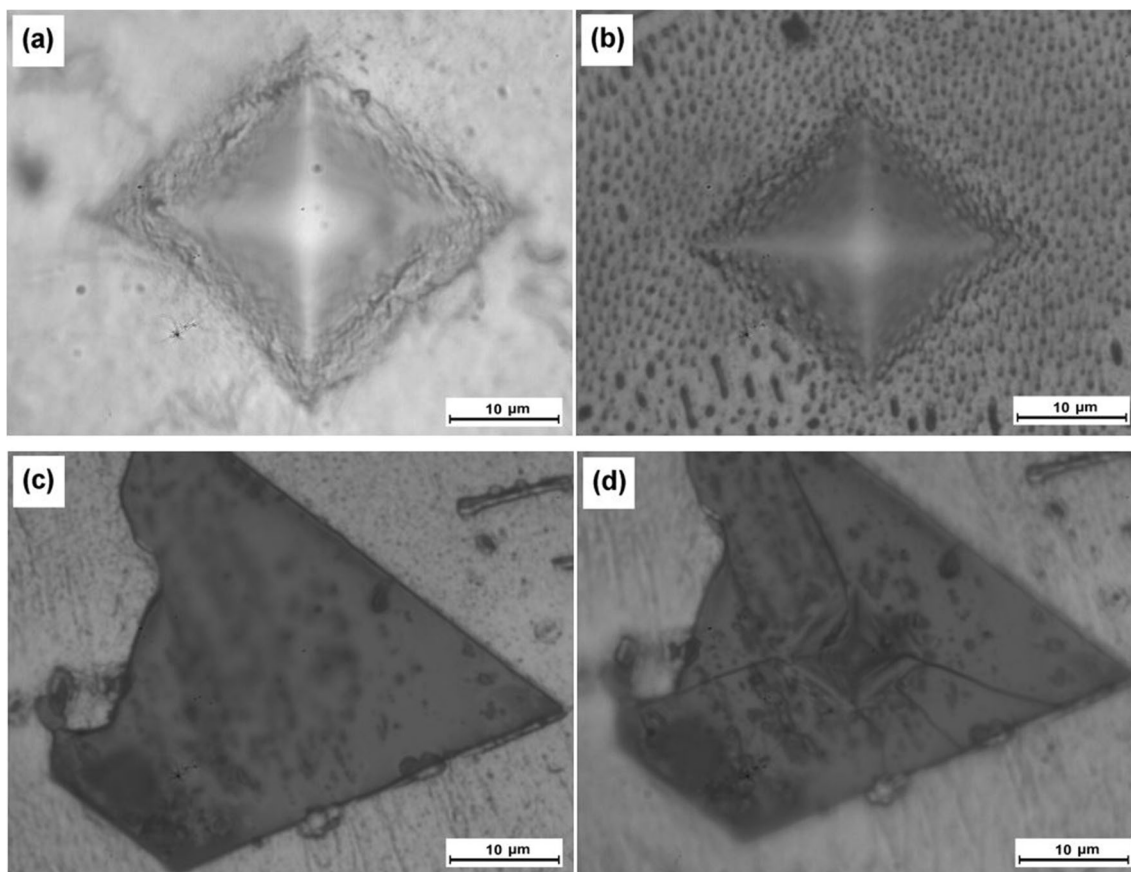
The overall observation shows that Mg with 2 and 4 % Si provide better tensile properties. The uniform distribution of



**Fig. 12** Hardness of  $Mg\text{-}Si$  alloy with respect to Si content



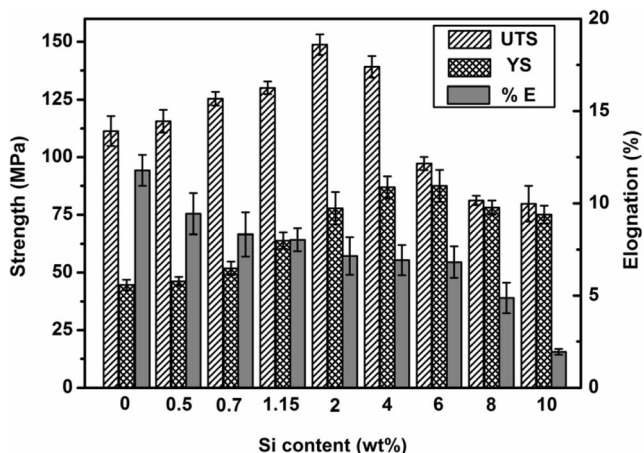
**Fig. 13** Micro hardness ( $H_v$ ) of  $Mg\text{-}Si$  alloy



**Fig. 14** Micro indentation on various phases in Mg-Si alloys (a) On  $\alpha$ -Mg halos, (b) On Mg-Mg<sub>2</sub>Si eutectic (c) Primary Mg<sub>2</sub>Si phase before indentation and (d) Primary Mg<sub>2</sub>Si after indentation

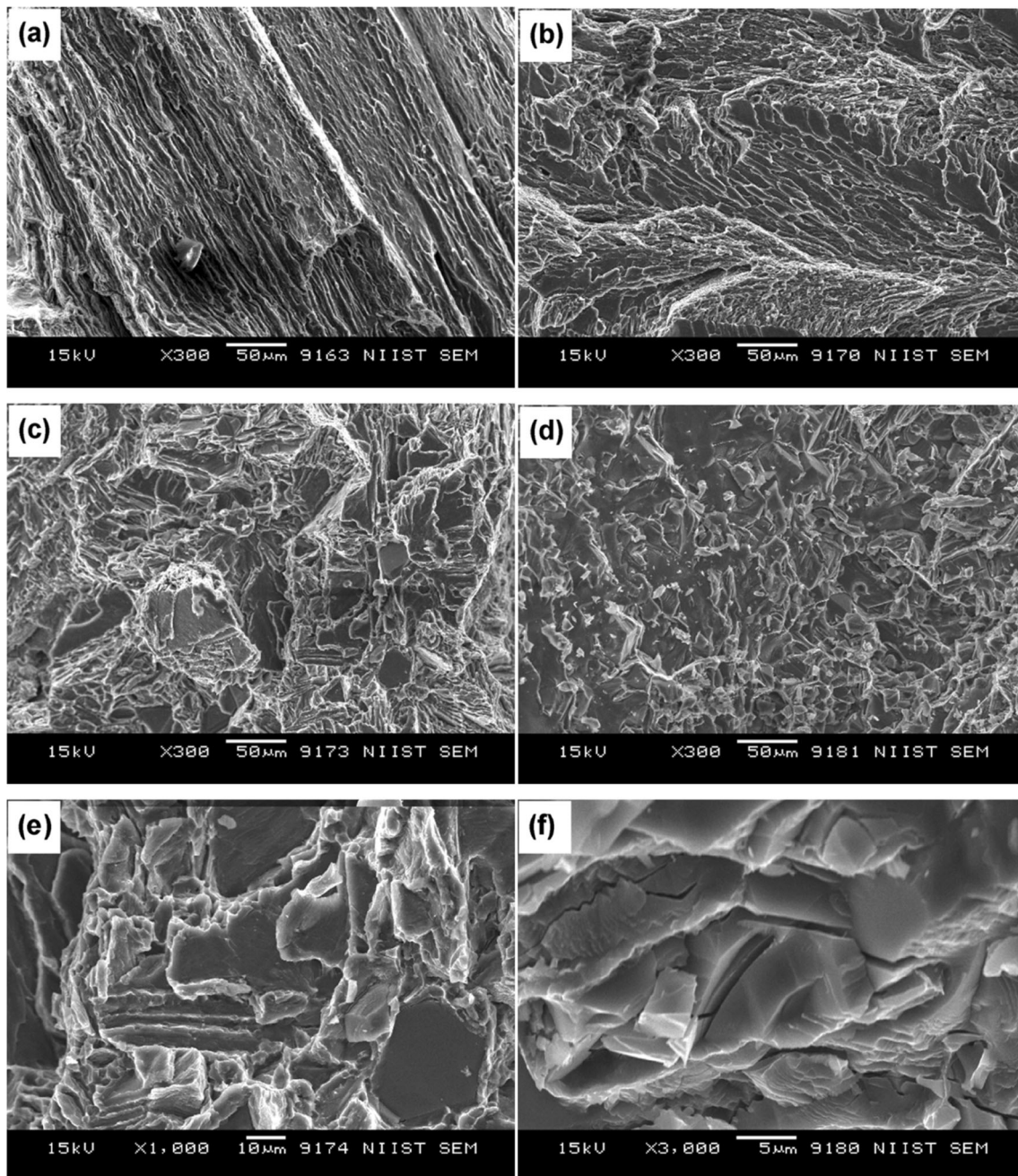
the fine eutectic and primary Mg<sub>2</sub>Si phases distributed in the Mg matrix is attributed to the improvement in tensile properties. In general, the increase in UTS and YS in hypoeutectic alloys (up to 1.15Si) with an increase in Si content is attributed to the increase in the eutectic volume (Fig. 3b-d).

The presence of the fine primary Mg<sub>2</sub>Si, in addition to a eutectic constituent in hypereutectic alloys, provides high strength properties in Mg-2Si alloys. The reduction in strength



**Fig. 15** Tensile properties of Mg-Si alloys tested at RT

properties with Mg-4Si and addition beyond that is due to the existence of high volume coarse dendrite shape primary Mg<sub>2</sub>Si phase (Fig. 6c-e). The main factors that form fracture morphology in engineering materials are the relation between the external loading, cohesive force levels and the internal stresses in the crystallographic lattice at different ranges of interaction [48]. Figure 16 presents the typical RT tensile fracture surfaces of Mg-(0, 1.15, 6 and 10) Si alloys. The microfracture mechanisms observed on the tensile fractured surfaces using SEM reveals long cleavages indicating the brittle type of fracture in pure Mg. With HCP structure, the failure of Mg alloys is expected to be brittle through cleavage or quasi-cleavage fracture, as the slip systems in HCP at RT are limited. Fractograph of hypoeutectic Mg-1.15Si alloy (Fig. 16b) shows cleavage with river type patterns which indicates the brittle type of fracture. Energy is consumed for forming two new surfaces and to overcome the work on the cleavage planes. The river pattern represents the cracks propagating in the cleavage planes which require more energy. The commonly observed connections of the cleavage steps reflect the tendency for minimizing the fracture work along the crack path. This alloy contains mostly eutectic, consisting of an alternate lamellar layer of hard Mg<sub>2</sub>Si and soft  $\alpha$ -Mg. Initially crack nucleates in the hard lamellar Mg<sub>2</sub>Si and then



**Fig. 16** Tensile fractograph of Mg-Si alloys tested at RT (a) pure Mg, (b) Mg-1.15Si (c)Mg-6Si, (d) Mg-10Si, (e) Cellular fracture surface of Mg-6Si alloy and (f) Brittle  $Mg_2Si$  dendrite in Mg-10Si alloy

propagates through the lamellar. The sharp edges or ends of the brittle  $Mg_2Si$  phases are preferred crack initiation and stress concentration sites [28, 49].

Figure 16c presents the fractograph of Mg-6Si alloy which shows many fractured particles of the primary  $Mg_2Si$ . Highly deformed regions around the brittle  $Mg_2Si$  particle, which are probably  $\alpha$ -Mg, are also observed. The fracture planes of almost all the coarse primary  $Mg_2Si$  particles exhibit clear cleavage without river pattern indicating a rapid fracture deriving from their intrinsic brittleness and pre-cracked structure, which is

known as cellular fracture surface. On the cellular fracture surface, the features of both brittle and ductile fracture are present simultaneously [50, 51], which are seen in the high magnified fractograph presented in Fig. 16e. It is typical on the polyphase material, where the microstructure components have different mechanical behaviour. Due to the strong cohesion at the interfaces between  $\alpha$ -Mg and  $Mg_2Si$ , the matrix is deformed under local active stress [51]. The formation of this kind of fracture demands higher energy input compared with brittle cracks. According to Emamy et al. [52], the energy is absorbed during plastic deformation, causing an

activation of the slip systems in successive micro-regions, which is evident from the high yield strength in Mg-6Si alloy. The fractograph of Mg-10Si shown in Fig. 16d reveals many large sheared  $Mg_2Si$  particles which suggest that these particles play a major role during the fracture. The relatively clean exposed surfaces of the particles point to the fact that they have cleaved during the fracture process. Multiple cracked  $Mg_2Si$  particles noticed in Fig. 16d and f suggests that the  $Mg_2Si$  dendritic particles provide the sites for fracture initiation because of their large dendritic morphology. The fracture of the  $Mg_2Si$  is due to the stress concentration around it. Once the crack nucleates, it propagates at a faster rate, which reduces the ductility [39].

**Elevated Temperature (ET)** Tensile properties of Mg-Si alloys at 150°C are shown in Fig. 17. In general, the strength properties of the alloys decrease with an increase in the testing temperature from RT to ET especially with low Si containing alloys (Figs. 15 and 17). However, in the high Si content (Mg-6, 8 and 10Si) alloys, the reduction in strength is very marginal. For instance, the average UTS of Mg-2Si alloy decreases from 148 MPa to 123 MPa (16.8 %) when the testing temperature increases from RT to ET, whereas the UTS of Mg-10Si reduces only from 79 MPa to 74 MPa (6.33 %). Normally in RT, the HCP structured Mg alloys exhibit brittle failure due to the limitations in slip systems; only basal slip is active at RT. However, at ET additional slip systems such as prismatic and pyramidal slip planes are active. Cross slip takes place through these planes at high temperatures and introduces more ductility [53].

The ET tensile fractographs are presented in Fig. 18. The improvement in the ductility of the alloys obtained at ET is correlated to the fracture surface. Figure 18a shows the fractograph of pure Mg tested at ET which significantly varies from that of the RT fractograph (Fig. 16). The ET fractograph of pure Mg has large amount of deformation with dimples

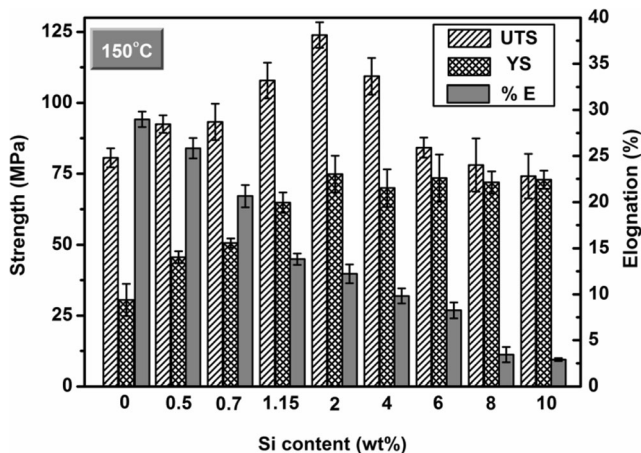


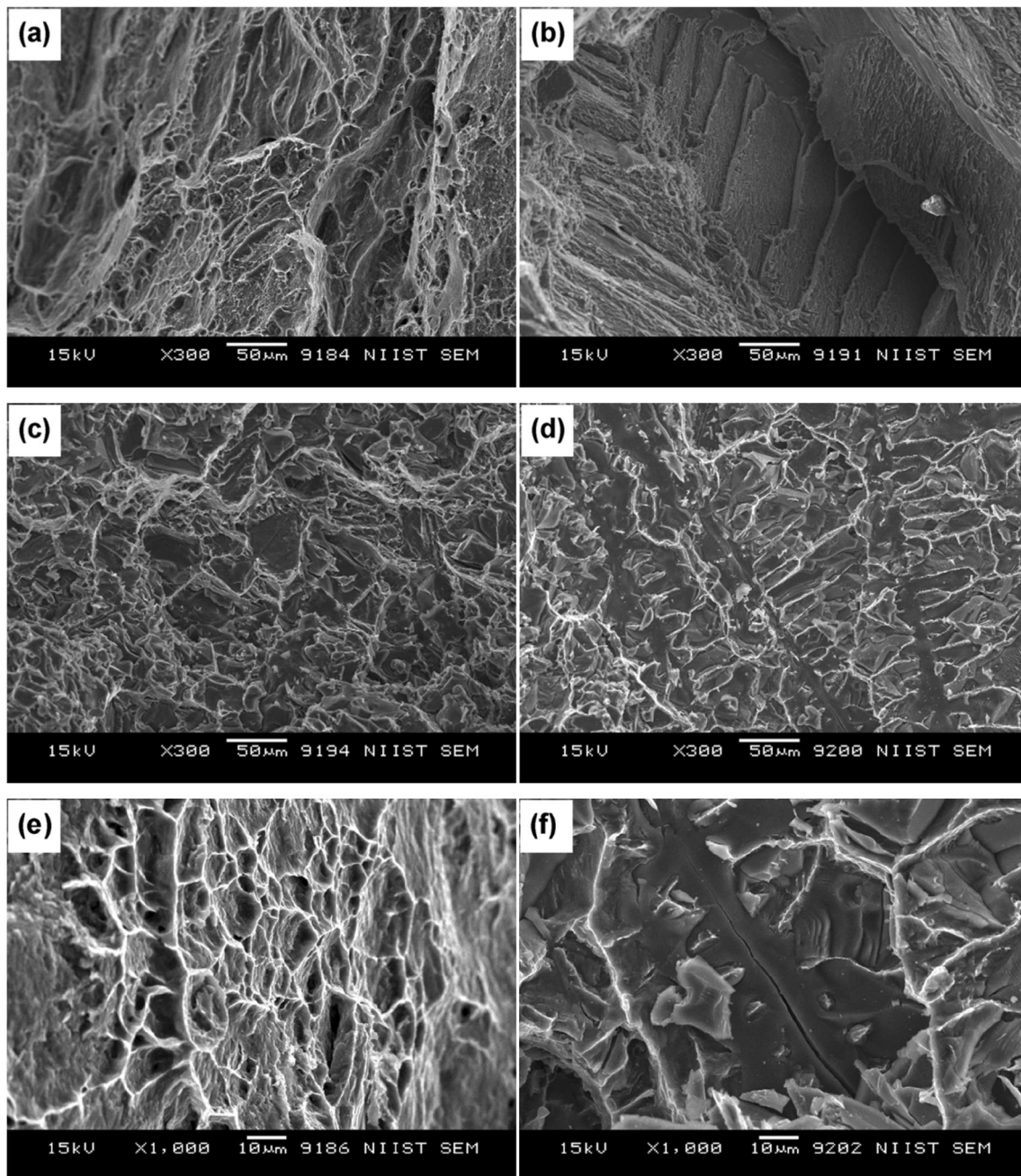
Fig. 17 Tensile properties of Mg-Si alloys tested at ET

along with few cleavages. The high magnification fractograph (Fig. 18c) of pure Mg at elevated temperature also shows more dimples and cleavage plane along the deformation zone. Figure 18b shows the ET fractograph of Mg-1.15Si which shows that, more cleavage facets are connected with tearing ridges and shallow dimples which ensures high elongation values obtained at 150 °C temperature. This is due to the introduction of additional slip planes such as pyramid and prismatic planes through which cross-slip takes place at ET. With 6Si addition, well-defined cleavage planes with  $Mg_2Si$  particle cracks and pullouts are observed in Fig. 18c, which is almost the same as RT fracture at 6Si alloy. When the Si content increases (Mg-10Si), the plastic zone in the Si added alloy is reduced due to the presence of coarse and brittle  $Mg_2Si$  particles, which makes the alloy more brittle (Fig. 18d). Also, decohesion of the  $Mg_2Si$  dendrites from the matrix has occurred in a few regions indicating that the interface between the matrix and  $Mg_2Si$  is relatively weak. Compared to fractograph of RT tested samples, more number of dendritic decohesion are observed at ET (Fig. 18d). Figure 18f shows the high magnified micrograph of cracked  $Mg_2Si$  in Mg-10Si alloy. The image clearly shows that the large crack observed along the dendrite, indicating that alloy failed in brittle manner even at ET.

### 3.3.4 Compressive Properties

Compression properties of different Mg-Si alloys were investigated and the ultimate compressive strength (UCS) values with varying Si content are shown in Fig. 19. The UCS of the alloy is found to increase significantly with an increase in Si content. For instance, the UCS of Mg-10Si is 283 MPa which is 35 % higher than that of pure Mg. This is due to the presence of the hard intermetallic  $Mg_2Si$  phase which shares the major load from the matrix. The fractographs of different Mg-Si alloys tested under compression load are presented in Fig. 20. Most of the Mg-Si alloy specimens tested in compression failed at an angle of 45° to the axis of loading which is in line with earlier observation [54].

In general, the fracture surfaces are shallow and flat, unlike those of tensile fracture. Figure 20a shows the compressive fracture surface of pure Mg at ambient temperature, which reveals a planar and smooth fracture surface, interspersed with some shallow ridges [55]. Smooth fracture surface indicates that pure Mg exhibits relatively more ductility compared to Si-added alloys [56]. This is also evident from the large spaced shear bands present on the fracture surface of pure Mg [56–58]. Shear bands are regions of distortion, where a portion of grain has rotated towards another orientation to accommodate the applied strain. When these regions extend across many grains they are called shear bands. Exhaustive shear



**Fig. 18** Tensile fractographs of Mg-Si alloys tested at ET (150 °C) (a) pure Mg, (b) Mg-1.15Si, (c) Mg-6Si, (d) Mg-10Si, (e) Ductile dimples of pure Mg and (f) cracking of Mg<sub>2</sub>Si dendrite

along these bands is a common type of failure in pure metal [59].

The fractograph of Mg-1.15Si alloy shows shear bands with comparatively less spacing (Fig. 20b) (compared to that of pure Mg), which results in less ductility. The fracture surface of Mg-6Si is shown in Fig. 20c, in which the matrix is smeared by the Mg<sub>2</sub>Si particle when being compressed. Particle fracture is not commonly observed it exhibits features that are consistent with the particles, having moved considerable distances before the final fracture [54]. In high Si samples (Mg-10Si) it is found that, fracture occurred after the

maximum flow stress has reached without the softening stage; the surface experiences a typical brittle fracture. Microcracks are gradually generated and then propagated into the matrix or through the interface between Mg<sub>2</sub>Si phase and Mg matrix. Finally, with a further increase in the load, the cracks propagate and cause total failure of the alloy. The higher levels of strength are attributed to different strengthening mechanisms, such as load transfer to the hard dispersoid phase, enhanced dislocation density due to differential CTEs between matrix and secondary phase, and strengthening arising from constrained plastic flow [54].

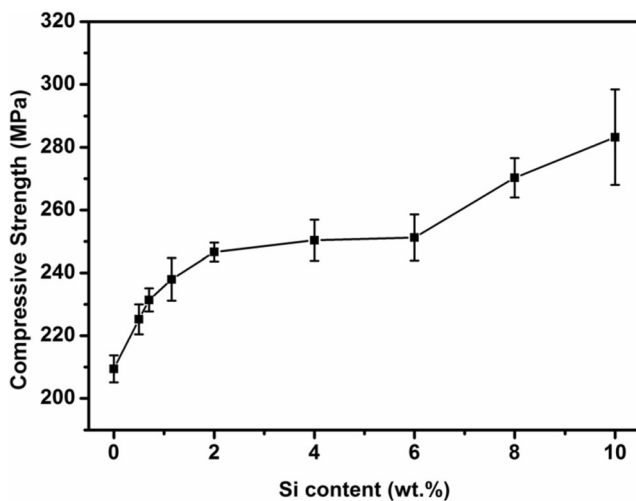


Fig. 19 Compressive strength of Mg-Si alloy with respect to Si content

The continuous interface observed in the fracture surface is also responsible for the increase in compressive strength. The fracture of Mg with high Si content occurs severely in  $Mg_2Si$  rich regions, which is due to the high local stresses caused by a restriction of the plastic zone surrounding neighbouring particles. This can be strongly influenced by

the mutually interactive factors of the following: (a) local plastic constraint, (b) particle size and (c) degree of particle agglomeration. In addition, the increased local stress arising from a high volume fraction of particles facilitates rapid linkage [55]. Compared to tensile properties, the compressive properties show a different trend with Si additions. UTS of Mg-Si alloys increases up to 2 % Si addition and then decreases with further Si additions: whereas, the UCS values of the Mg-Si alloys are found to increase with the increase in Si content. The reason for the discrepancy is attributed to the difference in the deformation modes during tensile and compression. An increase in Si addition leads to more volume of  $Mg_2Si$  particles in the Mg matrix which increases the stress concentration resulting in the formation and propagation of cracks at the interface between the  $Mg_2Si$  and matrix during tensile loading. In contrast, crack formation and propagation are retarded during compression loading.

## 4 Conclusions

The microstructure and mechanical properties of pure Mg and different hypo as well as hypereutectic Mg-Si (Mg-0.5, 0.7,

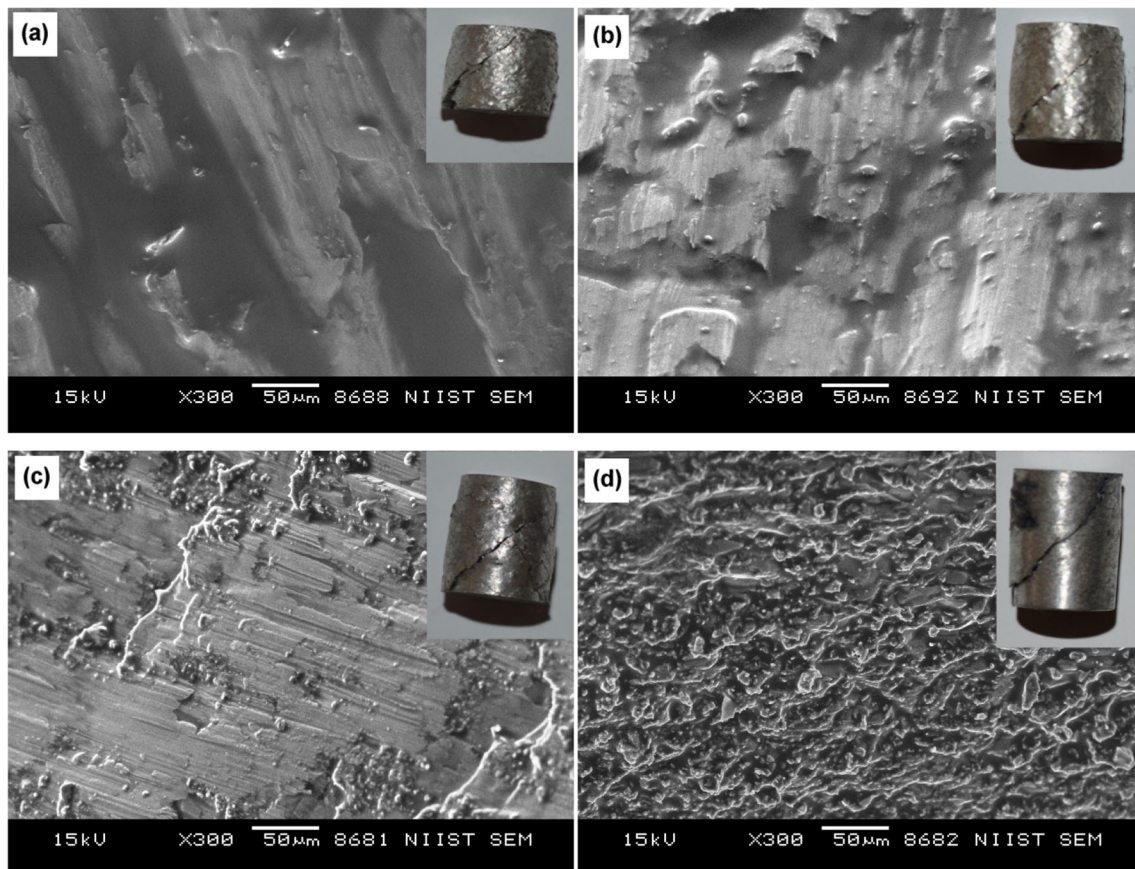


Fig. 20 Compressive fractographs of Mg-Si alloys (a) pure Mg, (b) Mg-1.15Si, (c) Mg-6Si and (d) Mg-10Si

1.15, 2, 4, 6, 8, and 10Si) alloys were investigated and the conclusions were as follows.

1. The microstructures of hypoeutectic Mg-Si alloys (Mg - 0.5, 0.7 and 1.15Si) were consisting of  $\alpha$ -Mg dendrites and lamellar Mg<sub>2</sub>Si phase. With the increase of Si content, the amount of the eutectic Mg<sub>2</sub>Si phase increased and became coarser and continuous. However, the amount of  $\alpha$ -Mg phase decreases as the Si content increases.
2. The microstructures of hypereutectic Mg-Si alloys (Mg - 2, 4, 6, 8 and 10Si) were having the phases such as primary Mg<sub>2</sub>Si,  $\alpha$ -Mg and eutectic Mg<sub>2</sub>Si. With the increase of Si content, the morphology of eutectic Mg<sub>2</sub>Si changes from a long needle, Chinese script type structure, into a small tiny needle shape structure. The amount of eutectic Mg<sub>2</sub>Si decreases with an increase in Si content in the hypereutectic alloys. Similarly, the volume percentage of primary Mg<sub>2</sub>Si increases with an increase in Si content in the alloys and its morphology changes from polygonal to coarse dendritic structure.
3. The hardness value increases steadily with an increase in Si content. The enhancement in hardness was due to the presence of hard Mg<sub>2</sub>Si intermetallic phase with the Si additions. The microhardness measurement revealed that  $\alpha$ -Mg has less hardness, comparable to that of pure Mg. The eutectic Mg-Mg<sub>2</sub>Si region yields comparatively higher hardness than  $\alpha$ -Mg in both the hypo and hypereutectic alloys. In the hypereutectic alloys, the primary Mg<sub>2</sub>Si phase exhibits the highest hardness among all other constituents. The average microhardness of primary Mg<sub>2</sub>Si was about 10 times higher than that of  $\alpha$ -Mg.
4. The UTS value increases up to 2 % Si addition, beyond that it decreases. The results show that Mg with 2 and 4 % Si exhibit, better tensile properties. The maximum YS was obtained at 6 % Si and beyond that marginal reduction was observed. The uniform distribution of the fine eutectic and primary Mg<sub>2</sub>Si in the Mg matrix was attributed to the improvement in strength properties. The reduction in strength obtained with 4Si addition and beyond was due to the presence of the high volume coarse dendritic primary Mg<sub>2</sub>Si phase.
5. The elevated temperature strength properties of these alloys decrease with an increase in the test temperature from RT to 150°C, especially with low Si-containing alloys. In high Si content (Mg - 6, 8 and 10Si) alloys, the reduction in strength is very marginal which is due to the presence of the high volume of thermally stable primary Mg<sub>2</sub>Si phase in high Si containing alloys.
6. The ultimate compressive strength (UCS) values of Mg-Si alloys were found to increase significantly with Si content. This increase is due to the presence of the hard

intermetallic Mg<sub>2</sub>Si phase which shared the major load from the matrix.

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Research involving human participants and/ or animals: Not Applicable.

**Data Availability** All the original data are available with the corresponding author.

**Code Availability** Not applicable.

## Declarations

**Conflict of Interest** The authors declare that they have no conflicts of interest.

**Ethics Approval** Not applicable.

**Consent to Participate** Not applicable.

**Consent for Publication** Publishers can publish images and results there are no restrictions.

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