#### ORIGINAL PAPER

# Optical and Optoelectronic Properties Enhancement of Porous Silicon Treated with Indium Oxide



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#### Abstract

This paper presents the results of the effect of Indium oxide  $(In_2O_3)$  on the structural, optical, and optoelectronic properties of porous silicon (PS). The results show an important improvement of optoelectronic property of PS coated with  $In_2O_3$  as a antireflective thin film. The treated PS with  $In_2O_3$  thin film was thermally annealed at various temperatures to improve the efficiency of the photovoltaic cells. The deposition of  $In_2O_3$  onto the PS sample was performed by simple immersion method. Surface morphology and chemical composition modification of the samples  $In_2O_3/PS$  were analyzed by SEM, EDX, and FTIR spectroscopy. Total reflectivity of  $In_2O_3/PS$  layers decreases significantly compared to as-prepared PS owing to an improvement in the light absorption. The PS treated with In<sub>2</sub>O<sub>3</sub> shows a significant enhancement in the minority carrier lifetime ( $\tau_{\text{eff}}$ ) indicating an improved surface quality in comparison with the untreated PS. Photoluminescence (PL) measurements of PS layer treated with  $In_2O_3$  has revealed an increase in PL-intensity and a bleu shift in emission PL band as a function of annealing temperature. The PL and  $\tau_{\text{eff}}$  enhancement are due to the surface passivation improvement and the decrease of recombination process.

Keywords Porous silicon . Indium oxide . Reflectivity . Photoluminescence . Lifetime

- The effective minority carrier lifetime increases from 1.85 μs to 16.4 μs after treatment of PS with  $In<sub>2</sub>O<sub>3</sub>$ .

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# 1 Introduction

Porous silicon (PS) thin film has received considerable attention due to its large range of applications such as solar cells [\[1](#page-6-0)], opto-electronic [\[2](#page-6-0)–[5](#page-6-0)], chemical and biological sensors [\[6](#page-6-0)–[11\]](#page-6-0), and biomedical devices [\[12](#page-6-0)]. However, the surface of PS layer can be oxidized in natural atmosphere which may change their chemical structure and optical properties [\[13](#page-6-0), [14\]](#page-6-0). To remedy this problem, many authors have indicated that the deposition of conducting material such as Cu, Ni, Au, Fe, Ag, Li, and Zn [[15](#page-6-0)–[22\]](#page-6-0) in the pores of PS layer enhance their photoluminescence and electrical properties. These different metals were deposited on PS layers by using various methods including thermal evaporation, sputtering, electrochemical, electrolyses and immersing plating. Various chemical and physical methods have been used to fabricate  $In_2O_3$  thin films, such as laser deposition [[16](#page-6-0)], atomic layer deposition [[17\]](#page-6-0), electrochemical deposition [\[18\]](#page-6-0), sputtering [\[19](#page-6-0)], and chemical vapor deposition  $[20]$  $[20]$  $[20]$ . These different deposition methods have been widely used for thin film preparation due to accurate control of deposition parameters [[21](#page-6-0), [23](#page-6-0)–[25\]](#page-6-0). Howeover, only specified thin film area can be obtained due to

 $-$  In<sub>2</sub>O<sub>3</sub> coated PS decreases the reflectivity significantly from 30 % to 2.4  $\%$ .

<sup>-</sup> FTIR analysis showed a substitution of Si-H bonds to Si-O-Si and Si-O-In after the thermal annealing.

<sup>-</sup> Annealing temperature of  $In_2O_3/PS$  composite at 300 °C leads to a noticeable improvement in PL intensity.

<sup>-</sup> A blueshift was obtained in the PL spectra due to the increase in the concentration of oxygen atoms and the quantum confinement effect of the oxidized silicon nanocrystals after thermal oxidation.

<span id="page-1-0"></span>the limitation of equipements of each deposition technique [\[14](#page-6-0)–[21\]](#page-6-0). In contrast, thin film deposition via chemical routes such as chemical bath, spin-coating and immersing methods are competitive alternatives to the conventional physical deposition techniques. It has been reported that the immersing of PS layer is a suitable method for large-area deposition of thin films, due to it is low-cost and the fact that it is very easy and more practical in comparison with other methods [\[15](#page-6-0)–[18,](#page-6-0) [23,](#page-6-0) [24](#page-6-0)].

Indium oxide  $(In_2O_3)$  is an attractive semiconducting material due to its wide band-gap (3.7 eV) and their high transmittance in the visible range  $(>90\%)$  [\[25\]](#page-6-0), which are used for many applications in nano-devices, such as solar cells [[26\]](#page-6-0), optoelectronic [\[27\]](#page-6-0), and gas sensors [\[28](#page-6-0), [29](#page-6-0)].

Several investigations of the origin and mechanism of light emission of  $In_2O_3$  thin film in glass and silicon substrates have been carried out by PL studies. But, the influences of this film on the photoluminescence and optoelectronic properties of PS layer have not been studied yet.

This paper examine the passivation of PS by immersion in indium solution and the thermal annealing effect on the physical properties of PS layer treated with  $In_2O_3$ , specially the reflectivity, the photoluminescence and the minority carrier lifetime.

# 2 Experimental Details

The study has been carried out on p-type boron doped monocrystalline silicon substrates with a cristallographic orientation of (100) and a resistivity of 1–2  $\Omega$ .cm. PS layers were fabricated by a two-step anodizing process using electrochemical anodization in a solution of  $HF(20\%)/C_2H_5OH$  (1:1) under stirring at a constant current density of  $5 \text{ mA/cm}^2$  for an etching time of 3 min. The initial PS formed on the silicon substrate was removed by chemical etching using a NaOH solution and followed by rinsing in bi-distilled water and drying under  $N_2$ . The second electrochemical anodization step was performed in a solution of  $HF(40\%)/C_2H_5OH$  (1:1) under stirring at a current density of 15 mA/cm<sup>2</sup> during 7 min in order to obtain a PS layer with a homogeneous pores structure. The formed PS layers were cleaned in a double deionized water in order to remove the residual acid and dried under  $N_2$ . In a typical experiment, two drops of 0.02 M indium (III) chloride butahydrate (InCl<sub>3</sub> 4H<sub>2</sub>O dissolved in 0.12 M hydrochloric acid solution) aqueous solution were added onto a PS layer at room temperature. The  $In_2O_3/PS$  multilayers were obtained by depositing solution dropwise onto PS substrates under 2500 rpm for 1 min by a spin coater and then dried at a heater plate at 100 °C for 10 min to evaporate the solvents. Finally, the samples were annealed at 200, 300, and 500 °C in an infrared furnace under air atmosphere for 90 min.

Scanning Electron Microscope (ZEISS Ultra Plus) was used to study the structures of PS layer before and after treatment with indium oxide. EDX analysis was performed to identify the elements composition of PS and  $In_2O_3/PS$  layers. The composition of PS and  $In_2O_3/PS$  samples was analyzed by Fourier transfer infrared (FTIR) spectroscopy in the 2200~400  $cm^{-1}$ spectral range using a 560 Nicolet-Magna spectrometer. The optical reflectivity spectra of samples was examined in the wavelength range 300–1200 nm using a UV/VIS/NIR Perkin-Elmer (Lambda 950) spectrophotometer. Photoluminescence measurements were carried out at room temperature and were registered on a PL (LS 55) Perkin-Elmer spectrometer using a Xe light source with a  $\lambda_{\text{exc}}$  of 447 nm. Furthermore, the effective minority carrier lifetime was identified using the photoconductivity method by a Sinton WTC-120 set-up.

# 3 Results and Discussions

# 3.1 SEM and EDX Analyses

Figure 1a shows the SEM image of untreated PS sample. From this image, we observe a nonuniform distribution of





Fig. 1 SEM images of top view of a PS layer and annealed  $In_2O_3/PS$ sample at 200 °C

micropores with a ravines form on the surface of PS layer. Figure [1b](#page-1-0) shows SEM image of the treated PS layer with  $In_2O_3$ . This figure exhibits a homogeneous distribution of  $In_2O_3$  clusters with some voids percentage that covered all the surface of PS layer.

In this study, EDX analysis was performed to report the elements composition presented in the samples of PS layer and  $In_2O_3/PS$  (see Table 1). Figure 2a and b exhibit the EDX pattern of PS and  $In_2O<sub>3</sub>/PS$  layers, and the relative elements composition of the samples were depicted in Table 1 below their corresponding spectra. In Fig. 2a, the spectrum of PS sample presents the O and Si elements, which confirms that the layer is made of silicon oxide forming the PS. In Fig. 2b, EDX spectrum of  $In_2O_3/PS$  sample shows the expected elements: Si, In, and O. However, the PS surface treated with indium oxide mainly consisted of Si presented by a high intensity peak centered at 1.88 keV. The characteristic peaks of In and O elements are present in the treated sample, which are attributed to indium oxide deposition on PS.

#### 3.2 FTIR Spectroscopic Analysis

In this work, FTIR spectroscopy was ridiculed to reveal the chemical bond structure of  $In_2O_3/PS$  $In_2O_3/PS$  $In_2O_3/PS$  composites. Figure 3 shows the FTIR spectra of PS, as-deposited and annealed samples of In<sub>2</sub>O<sub>3</sub>/PS at different temperature (from 200 to 500  $^{\circ}$ C). The vibration bonds at about 1080 cm<sup>-1</sup> and 1182 cm<sup>-1</sup> correspond to stretch modes of Si-O–Si and Si-O, respectively. The large vibration band in the range  $610–660$  cm<sup>-1</sup> is a mixture of stretching mode Si–Si and wagging mode Si–H<sub>n</sub>  $(n = 1; 2)$ . The vibration bonds centered at 842 cm<sup>-1</sup> and 977 cm<sup>-1</sup> are attributed to scissors mode  $Si-H<sub>2</sub>$  and wagging mode O<sub>3</sub>Si–H, respectively. All the peaks obtained in Fig. [3](#page-3-0) are in compliance with the previous data mentioned in the literature [\[30](#page-6-0)–[35](#page-6-0)]. Important changes in the FTIR spectrum after deposition of  $In_2O_3$  on PS can be seen in comparison with the untreated porous layer. These changes are pronounced particularly in Si–H and Si–O absorption peaks. We observe an increase in the peak intensity of the stretching mode Si–O–Si located at 1182 cm<sup>-1</sup>. Moreover, the peaks intensity of Si–H<sub>x</sub> (x = 1; 2) at 2083 and 2114 cm<sup>-1</sup> decrease in the case of as-deposited  $In_2O_3$  on PS layer. After annealing  $In_2O_3/PS$ , the intensity of Si–O bonds related asymmetric stretching modes and located in the range  $1000-1300$  cm<sup>-1</sup> increase significantly, while the peaks intensity of  $Si-H<sub>x</sub>$  located at

**Table 1** Elemental composition of PS and  $In_2O<sub>3</sub>/PS$  layers

Sample element	PS		$In_2O_3/PS$		
	O K	Si K	O K	In L	Si K
Weight %	22.72	77.28	23.13	20.23	56.64
Atomic $%$	28.18	71.82	39.35	8.24	52.41



Fig. 2 Typical EDX spectra of PS and  $In<sub>2</sub>O<sub>3</sub>/PS$  layers

2083 cm<sup>-1</sup> and 2114 cm<sup>-1</sup> have decreased progressively as a function of annealing temperature. The increase in the intensity and width of the absorption peak related Si–O–Si and the decrease of the peaks of  $Si-H_n$  are presumably caused by the hydrogen atoms replacement with oxygen and/or Indium atoms. As reported in the literature, the metal–oxygen–silicon bonding is located in the wavenumber range  $300-700$  cm<sup>-1</sup> [\[36](#page-6-0), [37](#page-7-0)].

The new vibration mode that appeared at 784  $cm^{-1}$  after deposition of  $In_2O_3$  on PS, corresponds to the O-Si-O bending mode [[38](#page-7-0)]. The vibration mode observed at 598  $cm^{-1}$  after annealing can be attributed to the formation of In–O bond [\[39\]](#page-7-0).

After the deposition and annealing temperature of  $In_2O<sub>3</sub>/$ PS, the FTIR spectra shows the appearance of new peak centered at 460 cm−<sup>1</sup> , which can be the result of interaction between the  $In_2O_3$  and PS interfaces. Therefore, the peak centered at 460 cm−<sup>1</sup> can be assigned to In-O-Si bonding. FTIR analysis indicates that Si–O bonds take place instead of the unstable Si–H bonds after deposition of  $In_2O_3$  onto the PS

<span id="page-3-0"></span>Fig. 3 FTIR spectra of the treated PS layer with indium oxide at different temperature



layer resulting in a surface oxidation. Thermal treatment of  $In_2O_3/PS$  composite at high temperatures under air also enhances the surface. Therefore, the increase in the absorption peaks intensity of Si-O-Si and In-O-Si can be attributed to the increase in the concentration of oxygen atoms into the  $In_2O<sub>3</sub>/$ PS composite as a function of annealing temperature.

## 3.3 Total Reflectivity Analysis

The reflectivity spectra of untreated PS, as-prepared and annealed In<sub>2</sub>O<sub>3</sub>/PS at various temperature (200 °C, 300 °C, and 500 °C), are saved in the wavelength region 300– 1200 nm, as shown in Fig. [4](#page-4-0). As well-known, the reflected optical field with multiple passes is principally related to the contribution of front and backside surfaces through the antireflective film. Several published works focused on the study of different nanostructured materials, such as metals and metals oxide as antireflective structures for the efficiency enhancement of photovoltaic cells. For instance, Bouzoura et al, F. Habubi et al., and Ahmadivand et al. have obtained the following result: the  $SnO<sub>2</sub>$ ,  $Zn<sub>3</sub>P<sub>2</sub>$ , and Ag with an ultrathin structures provide an average reflectance of 24%, 25.35%, and 26%, repectively, for a wavelength range of 300–1200 nm [[40](#page-7-0)–[42\]](#page-7-0). In an analogous condition, the metal oxide  $In_2O_3$  nanocrystals as a antireflective coating yield ∼22.6% of average reflectance for a similar wavelength range. On the basis of these

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Fig. 4 Reflectivity spectra of PS, as-deposited  $In_2O_3$  on PS, and annealed In2O3/PS at different temperatures (200, 300 and 500 °C)

investigations, we note that the metals oxide films provide a high reflectance percentage compared to the metals or metalsalloys. So, in the beggining we examine the absorption efficiency of Indium oxide film in c-Si and then, we study the annealing effect on the  $c-Si/In_2O_3$  nanocrystals to yield an efficient antireflective nanostructure. Annealing temperature has a significant effect on the surface morpholgy and optical property of the  $In_2O_3/PS$  films. Therefore, the annealing process activates the diffusion of  $In_2O_3$  inside the PS layer [\[43](#page-7-0)–[45](#page-7-0)], which may lead to enhancing the surface passivation quality of the PS.

The porosity of the PS layer reduces the reflection loss, allowing the light trapping within the cell by internal multireflection. Furthermore, the treated PS layer with Indium oxide shows a significant reflection loss in the visible wavelength range. The sample annealed at 200 °C exhibits the most excellent reduction of surface reflectivity from 30 to 2.4% in the visible light range 400–800 nm. The significant reduction in the reflectivity of annealed  $In<sub>2</sub>O<sub>3</sub>/PS$  compared to the untreated PS and other samples annealed at 300 and 500 °C may be the result of textured surface enhancement. Therefore, this antireflection coating can be used as a light trapping to improve the efficiency of the photovoltaic cells.

#### 3.4 Minority-Carrier Lifetime Measurements

To study the influence of  $In_2O_3/PS$  on the electronic quality of the silicon substrate, we measured the effective carrier lifetime  $(\tau_{\text{eff}})$ , before and after heat treatment. Figure 5 shows the enhancement of the  $\tau_{\text{eff}}$  as a function of the annealed In<sub>2</sub>O<sub>3</sub>/PS samples. For PS sample, the obtained results indicate a slight decrease of the  $\tau_{\text{eff}}$  compared to that for the c-Si substrate due to the increase of defects onto the porous layer. The treated PS with  $In_2O_3$  shows an improvement in the effective lifetime



Fig. 5 Minority carrier lifetime measurement after each process: untreated PS, and  $In_2O_3/PS$  before and after heat treatment

from 1.85 μs (for PSi) to 16.4 μs (for annealed  $In_2O_3/PS$  at 200 °C). The further increase in annealing temperature causes the degradation of effective minority carrier lifetime to 7.2 μs at 500 °C. The improvement in the effective lifetime could be attributed to surface passivation and the minimization of the dangling bonds density at the c–Si surface. This enhancement is mainly due to the coordination of In atoms to the Si atoms through the oxygen (In–O–Si bonds) as indicated in the FTIR analysis.

#### 3.5 PL Spectroscopic Analysis

The Photoluminescence (PL) measurements of PS and  $In_2O<sub>3</sub>/PS$  layers were carried out at a room temperature with 447 nm wavelength from a Xe lamp source. PL spectra of PS layer, as-deposited and annealed  $In_2O_3/PS$ samples at various temperatures (from 200 to 500 °C) as a function of photon energy are showed in Fig. [6](#page-5-0). As shown in this figure, the untreated PS sample exhibits a PL band centered at 1.86 eV. The PL spectra of asprepared and annealed  $In_2O_3/PS$  samples exhibit orangered emission bands centerted between 1.98 eV (626 nm) and 1.88 eV (659 nm). The origin of this orange-red emission is attributed to the quantum confinement of silicon oxide nanocrystals forming the PS layer and the interface state between indium oxide and the silicon oxide films, as shown in the FTIR analysis. In Fig. [6](#page-5-0), the PL result shows a significant improvement after depositing In<sub>2</sub>O<sub>3</sub> onto PS and annealing In<sub>2</sub>O<sub>3</sub>/PS at 200  $\degree$ C and 300 °C compared to PS sample. This reveals an increase in PL-band intensity and a bleu shift from 1.88 eV (for as-prepared In<sub>2</sub>O<sub>3</sub>/PS) to 1.98 eV (for annealed In<sub>2</sub>O<sub>3</sub>/PS at 500 °C). The origin of these changes can be attributed

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Fig. 6 PL spectra of PS, as-deposited and annealed  $In_2O<sub>3</sub>/PS$  at 200, 300, and 500 $\degree$ C

to the increase in the concentration of oxygen atoms incorporated into  $In_2O_3/PS$  and the quantum confinement effect of the oxidized silicon nanocrystals. The enhancement of the PL spectra of  $In_2O_3/PS$  structure is due to the surface passivation improvement and the decrease of recombination activities. Afterward, the PL intensity decreases significantly after sample annealing at 500 °C, due to the outflowing of impurities up to the surface of Si nanocrystals. The large decrease of the PL intensity at this annealing temperature can be related to defect states at Si and Si oxide interface, resulting the diffusin of impurities and the formation of nonradiative recombination sites. These higher temperatures promote the decrease in the void concentration of assembly  $In_2O<sub>3</sub>/PS$ , leading to the obtaining of a compact microstructure  $[46]$ . The shift to higher energy can also be attribute to the trapped electrons at the located states owing to  $Si = O$  bond of PS layer [\[47](#page-7-0)–[49\]](#page-7-0) and the substitution of hydrogen atoms by the oxygen atoms [[49](#page-7-0), [50\]](#page-7-0) and/or the oxidized indium forming In-O-Si according to the FTIR analysis.

From the above PL and FTIR discussions, we may deduce that the enhancement in the peaks intensity is related to an increase in the number of Si–metals (In) bonds. It is shown that annealing temperature at 200 °C and 300 °C has a significant influence on the peaks intensity of PL and FTIR, which is in good agreement with the carrier lifetime  $(\tau_{\text{eff}})$  and indicates the presence of a passivation process of the PS structure. Table 2 presents a comparative data of FTIR, Reflectivity, PL, minority carrier lifetime, and conversion efficiency characteristics of PS treated with different materials [\[22](#page-6-0), [43,](#page-7-0) [44](#page-7-0), [51](#page-7-0), [52\]](#page-7-0). This table shows similar physical characteristics for different materials used for Si solar cells improvement. Furtermore, from the values of conversion efficiency of these materials, we can conclude that the efficiency of the photvoltaic cell (PS treated with the Indium oxide) does not exceed 10%. Therefore, the  $In_2O_3$  films can be used as a suitable antireflection and passivation coating for c-Si solar cells.

# 4 Conclusion

This paper have exposed the influence of thermal annealing on the optical and optoelectronic properties of  $In_2O<sub>3</sub>/PS$ layers. In<sub>2</sub>O<sub>3</sub>/PS samples thermally annealed at 200  $^{\circ}$ C present a low average reflectance of 2.4%. FTIR spectroscopy measurements have shown the increase of Si–O–Si and Si-O-In bonds intensity with the increase of annealing temperature of  $In_2O_3/PS$  film, and found to be a replacement of Si-H bonds. A modification in the surface morphology of PS layer after treatment with  $In_2O_3$  is observed. Annealing temperature of In<sub>2</sub>O<sub>3</sub>/PS composite at 200  $^{\circ}$ C leads to a noticeable improvement in carrier lifetime (from 1.85 μs to 16.4 μs). The paper has also demonstrated that the treatment of PS with  $In_2O_3$  causes an improvement in PL intensity. The treatment of PS layer with  $In_2O_3$  acts as an efficient antireflection coating. The thermal annealing of  $In_2O_3/PS$  sample leads to dangling bonds passivation at the surface of PS layer.

Material	Reflectivity $R(\%)$	Peak intensity of $Si-O-Metal$ (a.u.)	Photoluminescence peak intensity (a.u.)	Minority carrier lifetime $\tau_{\text{eff}}(\mu s)$	Conversion efficiency $\eta(\%)$
$In_2O_3$ [This work]	2.4	0.4	0.099	16.4	
$V_2O_5$ [43]		1.1	0.026	134.74	12.5
LiBr $[44]$		0.05	0.03	18.2	
$ZnO$ [22]	4	$\overline{\phantom{0}}$			20
NiO[51]		0.7	0.04	—	$\qquad \qquad \  \, -$
$Er_2O_3[52]$		0.4	0.17	56	7.9
$Al_2O_3$ [53]	9	-		90	12

Table 2 Physical parameters of PS treated with different materials as a antireflective coating

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