REVIEW ARTICLE



Haemotoxic Effect of Lead: A Review

Rina Rani Ray¹

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Abstract Lead exposure is one of the major environmental issues to alter the health and well being of man and animals. Blood being the easy target for lead intoxication shows prominent effects and alteration of several hematological parameters may be regarded as a bio indicator of lead intoxication. Blood lead levels above 10 µg/dL have been associated with numerous manifestations. Anaemia is reported after acute and chronic exposure of lead in both workers of lead factory and in laboratory animals, accomplished either through impairment of heme biosynthesis or by enhanced rate of blood cell destruction. This is confirmed by marked reductions in blood haemoglobin level and haematocrit value and echinocytic transformation of normal erythrocytes after lead exposure. The number of total leucocytes tends to increase with the increase in basophils, eosinophils and lymphocytes possibly due to direct toxic action of lead on leucopoiesis in lymphoid organs. High levels of lead inhibit aggregation of platelets both in human and rat blood. Lead induced changes in the red blood membrane include the changes in lipids and proteins profile of some membrane-associated enzymes or in ions transport mechanisms. The mechanism of lead toxicity may be due, in part, to disruption of calcium-mediated processes. The chronic exposure to lead can result in a drastic changes in the cholesterol and phospholipid content, hexose, hexosamine and sialic acid levels and membrane acetyl cholinesterase, NADH dehydrogenase and Na^+-K^+ ATPase levels. Moderate exercise, intake of methionine, glycine, curcumin, methionine, carotene and

Rina Rani Ray raypumicro@gmail.com pectin enriched food and vitamin C could reduce the severity of lead toxicity.

Keywords Lead · Hemotoxic effect · Heme biosynthesis · Anemia · Erythrophagocytosis · Lead toxicity

Introduction

Lead (Pb) has been recognized as one of the most emerging hazardous heavy metals among environmental pollutants (ATSDR 2007). Tons of lead is discharged from electric battery manufacturing, lead smelting, and mining activities, as well as internal combustion engines fuelled with leaded petroleum and lead bearing products, forming the primary sources of lead contamination. A recent report holds that 33 % of over 370 water samples from the top 26 cities of India tested positive from harmful content of lead (over 10 parts per billion); resulting mainly from PVC pipes used in plumbing, sanitation and agriculture (The Lead Group Inc. 2010). Lead exposure from conventional use of vermillion, lead enriched toys and paints are increasingly aggravating the situation.

Lead has been mined and used by mankind for 6000 years, and the history of lead poisoning is nearly 2500 years old (Hernberg 2000). In early days Romans used lead compounds for glazing pottery, and metallic lead for cooking utensils and piping. In middle ages, lead was widely used both for industrial, domestic, and medicinal purposes. Certain folk remedies for digestive ailments have been found to contain very high levels of lead. In new age i.e. during fifteenth to eighteenth centuries, especially in Europe and later in America, the use of lead and its compounds gained increased use in pottery, piping, shipbuilding, window making, the arms industry, pigments, and later

¹ Department of Zoology, Bethune College, 181, Bidhan Sarani, Kolkata 700 006, India

book printing industries. This led to more exposure to lead compounds and subsequent entry of lead in food chain. Today, most bullets for shotguns, handguns, and rifles are made of a lead core surrounded by a copper or steel jacket to protect the lead from changing shape at high speeds. Lead is a key component in various cosmetics, the pigment in many paints, a sweet and sour condiment popular for seasoning and adulterating food; a wine preservative and reliable piping for the vast network plumbing and for making colourful toys for children. Lead is commonly used in painting and soldering and in making stained glass, jewellery, pottery, ammunition, or fishing sinkers. The addition of lead to gasoline is one of the greatest public health failures of the twentieth century. The current threshold for lead toxicity, defined as a blood lead level of 10 µg/dL, was adopted by the United States (US) Centers for Disease Control and Prevention (CDC) in 1991 and the World Health Organization in 1995.

In children the list of lead toxic effects includes intellectual deficits, reading problems, school failure and delinquent behaviour; all these effects have substantial public health impact and are apparently long lasting and irreversible. In adults it is associated with subsequent vascular mortality (Rossi 2008). Actually a large amount of lead and related elements like cadmium have been released into the environment due to mineral processing activities and have impacted water resources, soils, vegetables, and crops.

In some areas, this pollution is extremely hazardous to human health (Zhang et al. 2012). The most prominent effects are cognitive deficiencies in children, renal impairment, hypertension, cataracts, and reproductive problems such as miscarriage, stillbirth, and decreased fertility in men and women (Patrick 2006). Sciarillo et al. (1992), observed delinquent, antisocial and aggressive behaviours in human. According to Tanquerel des Planches (1850) abdominal pain and peripheral neuropathy are two significant symptoms of lead toxicity in man. This metal affects almost all major body organs, particularly the CNS. Various molecular, cellular and intracellular mechanisms including the generation of oxidative stress, ionic mechanism and apoptosis have been proposed to explain the toxicological profile of lead (Farhat and Khademi 2015).

Chronic exposure of lead compounds through inhalation or ingestion results in adverse effect on the blood, central nervous system, and cardio vascular system, kidneys and vitamin D metabolism. Children are particularly sensitive and exposure of lead can also result in reduced cognitive development and reduced growth. Adverse effect on human reproduction has also been reported. Lead compounds can be persistent in the environment and have the potential to accumulate in food chains. Lead is classified as possible human carcinogen based upon the fact that exposure of animals to high level of lead have produced cancer usually in kidneys.

One of the easy targets for lead toxicity is the hematological system (Mojzis and Nistiar 2001) and anemia is probably one of the most well-known toxic effects of lead. The haemolytic effect of lead is reported from the blood of smelter worker and from the individuals living around a lead releasing factory. The in vitro and in vivo studies with murine models also indicate the similar hemotoxic effect of lead.

This review attempts to elucidate the mechanism of action of lead and consequent haematological manifestations of lead toxicity in both human and in experimental animals.

Effect of Lead Toxicity on Non-human Animals

Since lead is a ubiquitous and naturally occurring metal, it can be mobilized and released into the environment at higher than background concentrations as a result of anthropogenic activity and exposure to elevated concentrations of lead from natural and human sources can be hazardous to wildlife (Scheuhammer and Norris 1995). Lead is one of the most studied toxicants, and found to be toxic to several physiological systems in vertebrates, including the nervous, renal, cardiovascular, reproductive, immune, and hematologic systems. In male Wistar rats, after 4 weeks, blood pressure responses to lead were biphasic. Low lead levels reduced blood pressure, dilated the carotid artery and increased cardiac output. At higher lead doses, rats had increased blood pressure (Wildemann et al. 2015).

The adverse effects are also well pronounced in lead exposed non-human vertebrates (Table 1), which are now being explored by the veterinarians and wildlife professionals to fully identify the lead exposure risks to mankind. Conservationists recognize that hunting, angling and shooting sports deposit thousands of tons of lead into the environment each year (Pokras and Kneeland 2009) and the flesh of the birds and animals hunted with lead enriched gun shots, become a source of lead poisoning of the human being. Lead can accumulate in invertebrates and plants, but its influence on food chain is yet to be discovered (Pattee and Pain 2003).

Absorption and Pharmacokinetics of Lead in Human

The pharmacokinetics of lead in humans is complex and its absorption is inversely proportional to chronologic age of the victim and the smaller the particle, the more completely the lead is absorbed. Thus, exposure to lead dust results in

Exposed animal	Effects	References
Herring Gull (Larus argentatus)	Problem in locomotion, food begging, feeding, treadmill learning, thermoregulation, and individual recognition. Abnormal feather	Burger and Gochfeld (2005)
Laysan Albatross (Diomedea immutabilis)	"Droop wing" due to peripheral neuropathy	Sileo and Fefer (1987) and Platt et al. (1999)
Turkey Vulture (Cathartes aura)		
Bald Eagles (Haliaeetus Leucocephalus	Various poisoning effects	Janssen et al. (1986), Mateo et al.(2001),
California Condors (<i>Gymnogyps</i> californianus)		Pokras and Chafel (1992) and Sidor et al. (2003)
Common Loon (Gavia immer)		
Golden eagles, bald eagles, ravens, turkey vultures, and pumas	Animals exposed to the fragments of spent lead ammunition show high level of lead in organs and tissues	Burco et al. (2012), Clark and Scheuhammer (2003), Craighead and Bedrosian (2008), Cruz-Martinez et al. (2012), Fisher et al. (2006), Kelly and Johnson (2011), Stauber et al. (2010) and Wayland and Bollinger (1999)
Egyptian vultures (<i>Neophron percnopterus</i>)	Males were more susceptible	Gangoso et al. (2009)
	Bone lead concentration increased with age, reflecting a bioaccumulation effect. The degree of mineralization decreased with increase in lead concentration	
Bald Eagle (<i>Haliaeetus leucocephalus</i>), Steller's Sea Eagle (<i>H. pelagicus</i>), White- tailed Sea Eagle (<i>H. albicilla</i>) and the critically endangered Californian Condor (<i>Gymnogyps californianus</i>)	Lead accumulation	Behrens and Brooks (2000), Church et al. (2006), Franson (1996), Kenntner et al. (2001) and Meretsky et al. (2000)
Domestic dogs and cats rodents and songbirds	Aggressive behaviours	Janssens et al. (2003) and Li et al. (2003)
Dogs	Neuromuscular signs include seizures, hyper- excitability, hysteria, ataxia, muscle spasms, paresis, paralysis, blindness	Zook et al. (1969)
Alligator mississippiensis	anorexia, weight loss, poor growth, lethargy and death	Camus et al. (1998)
Western fence lizard (Sceloporus occidentalis)	No effect on locomotor performance	Holem et al. (2006)
Domestic and free living birds	Physiological or behavioural changes	De Francisco et al. (2003)

higher absorption than exposure to the equivalent amount of lead from chips of lead paint. The organic lead compounds (i.e. tetraethyl lead) are readily absorbed through intact skin. The majority (nearly 100 %) of lead inhaled as vapour or fumes is absorbed directly through the lungs. Although the blood generally carries only a small fraction of the total lead body burden, it serves as the initial receptacle of absorbed lead and distributes (Fig. 1) lead throughout the body, resulting in a number of physiological manifestations and also making it available to other tissues or for excretion. The elimination half-life of lead in adult human blood has been estimated to be 1 month, whereas in children it may be as high as 10 months. Blood lead is also important because the blood lead level (BLL) is the most widely used measure of lead exposure. The less sensitive erythrocyte protoporphyrin (EP) assay is also used as a measure of blood lead.

Lead in Blood

After lead is absorbed from the gastrointestinal tract or the lungs, it enters the blood stream and since lead is not converted to any acceptable or assimilable form, it remains stored as body "burden" of lead and about 95 % of the total body stores of lead are found in bone. In blood, about 99 % of lead remains in the red blood cells (RBC) and only 1 % in plasma with a half life of 28 days (US Department of Health and Human Services 2000). Systolic blood pressure (5.32 %, p = 0.05) and diastolic blood pressure (5.87 %, p = 0.05) were significantly increased in the automobile workers as compared to controls (Dongre et al. 2011). Valentine et al. (1976) documented the report of lead-induced deficiency when sufficiently severe gives rise to findings similar to the hereditary disorder.

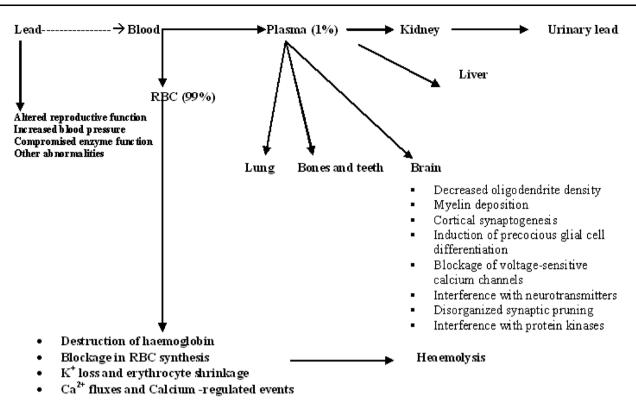


Fig. 1 Flow diagram showing a path of possible entry of lead in animal system and its prospective effects

Animals with lead-induced hypertension exhibited oxidative stress which was associated with mild up-regulation of superoxide-generating enzyme and NAD(P)H oxidase (Vaziri et al. 2003).

Mode of Action of Lead on Blood

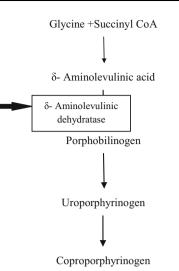
Lead induced anaemia is produced principally by two mechanisms: impairment of heme biosynthesis and increased rate of blood cell destruction (Schwartz et al. 1990). Hemolytic anaemia may be induced by the inhibition of haemoglobin synthesis (Piomelli et al. 1982; Gurer et al. 1998). Previously, lead-induced anemia was considered to be resulted from the inhibition of δ -aminolevulinic acid dehydratase participating in the heme biosynthesis. However, little is known whether lead could affect the destruction of erythrocyte. In their study, Jang et al. (2011) demonstrated that lead could accelerate the splenic sequestration of erythrocytes through phosphatidylserine (PS) exposure and subsequently increased erythrophagocytosis (Table 2).

Although the precise mechanisms of action of lead toxicity are not clear, in vitro studies demonstrated that lead can interfere with fundamental biochemical processes which are accomplished by binding to proteins and/or interference with calcium dependent processes (Goldstein

1993). Hence, the mechanism of lead toxicity may be due, in part, to disruption of calcium-mediated processes which leads to neurotoxicity (Pounds 1984; Simons 1993). Lead activates protein kinase C in human erythrocytes after exposure to lead in vitro (Hwang et al. 2001). Such activation is evident from the erythrocytes isolated from workers in the lead industry and this may be directly relevant to the neurotoxicity of lead. Although the effects of Pb²⁺ on Ca²⁺ fluxes and calcium-regulated events have been suggested as major mechanisms involved in Pb²⁺ toxicity (Bressler et al. 1999; Marchetti 2003; Toscano and Guilarte 2005), other potential mechanisms include the capacity of Pb²⁺ to affect cell membrane composition. According to Kempe et al. (2005) exposure to Pb^{2+} ions at concentrations relevant for toxicity activates K⁺ channels, leading to K⁺ loss and erythrocyte shrinkage and favoring phosphatidylserine exposure at the surface of the cell membrane. The affected erythrocytes are removed from circulating blood that in turn leads to the development of anemia after Pb²⁺ intoxication. The study of Golalipour et al. (2007) on effect of lead intoxication showed that hematologic effects of lead are reversible and after recovery period, most of parameters were close to normal in 36 male adult albino Wistar rats. In brief, lead interferes with calcium and iron metabolism by forming complexes with sulfhydryl groups and other ligands, remaining at the active sites of the associated enzymes (Selevan et al. 2003).

 Table 2
 Actions to be taken to prevent lead poisoning by measuring the blood lead level

Blood lead level (µg/dL)	Actions to be taken
<2	Average whole blood lead level for the general population
≥10	Public heath action level for children and pregnant women. Counseling to maintain lead levels below 10 µg/dL and as low as possible
10–14	Public heath action
15–19	Repeat screening, case management to abate sources
<u>≥</u> 25	Action level for adult case follow-up
20-44	Immediate medical evaluation, case management
45–49	Medical evaluation, chelation, case management
≥50	On an initial test, a 2 week follow-up blood lead test is indicated to confirm the lead level. Workers with an average lead levels of 50 µg/ dL or greater must be removed from working with lead
≥60	Workers with whole blood lead levels greater that 60 µg/dL must be immediately removed from further lead exposure if confirmed by a follow-up test
>69	Hospitalization, immediate chelation, case management



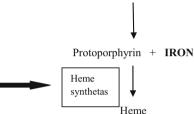


Fig. 2 Impairment of heme biosynthesis by lead. The *arrows* denote the lead action

Slobozhanina et al. (2005) opined that low concentrations of lead did not produce reactive oxygen species in erythrocytes in vitro, but can change the physicochemical state of proteins and lipids in erythrocyte and lymphocyte membranes and thereby changes the functions of the receptors and channels present at the plasma membrane level resulting in the alteration in the molecular composition of the intracellular space and cell functions.

Biomarker of Lead Toxicity in Blood

In early days, basophilic stippling of red blood cells was considered as a classic sign of lead poisoning which is replaced by measurement of blood lead and zinc protoporphyrin levels for biological monitoring of lead-exposed workers (Cheson et al. 1984). Wibowo et al. (1979) found a relationship between blood lead (Pb–B) and lead manganese (Pb–Mn) concentration and the interaction between lead and manganese is mediated through a relationship between Pb and Mn porphyrin interactions in erythrocytes.

In laboratories, the titre of the lead affected enzymes involved in haem biosynthesis (Fig. 2) is measured to quantify the effect of lead on blood. These key enzymes are delta aminolevulinic acid dehydratase (ALAD) and haem synthetase, impaired synthesis of which reduces the concentration of delta aminolevulinic acid dehydratase (BLL of 5–10 µg/dL) and elevation of protoporphyrin in the RBC (BLL > 20 µg/dL) respectively. Since δ -aminolevulinic acid dehydratase (ALAD) activity in lead workers was inhibited by lead sensitively and stoichiometrically, ALAD activity may be adopted as a reliable biomarker of lead toxicity in human (Chiu et al. 2013).

Roels and Lauwerys (1987) studied the relations of PbB to heme synthesis biomarkers and reported no evident threshold for inhibition of aminolevulinic acid dehydratase synthesis at Pb as low as 8–10 μ g/dL, while the threshold for increasing erythrocyte protoporphyrin levels was evident in the range of 15–20 μ g/dL, consistent with other studies, including two meeting criteria for inclusion in this report (Rabinowitz et al. 1986; Hammond et al. 1985). Moreover, The NAD⁺, ATP, and NADP⁺ concentrations in erythrocytes seem to be the most sensitive indicators of the effects of Pb²⁺ on cell energy metabolism even at low levels of exposure (Baranowska-Bosiacka and Hlynczak 2004).

Since Karita et al. (2005) found lead exposure in all the workers was negatively associated with Hb, Hct, and RBC, after controlling for age and working status, these parameters may be recognized as nonspecific but useful markers of lead toxicity. However, a marked change in the morphological, biochemical and numerical changes in blood cells resulted from chronic and acute exposure at lead.

Changes in Erythrocytes

Erythrocyte deformability (ED) is an important physiological factor which plays an essential role in delivery of oxygen to the tissues through bringing about significant changes in viscoelastic and geometric properties of the membrane accompanied by a change in internal viscosity (Mokken et al. 1992). Marked reductions in blood haemoglobin level, haematocrit value and erythrocyte count were observed after 1 and 2 weeks of intraperitoneal treatment of Swiss mice with lead nitrate (8 mg/kg for 5 consecutive days per week), as compared to the respective controls. Echinocytic transformation of many erythrocytes and the formation of a few microcytes, as revealed by both light and scanning electron microscopy, indicated the anaemia to be anisocytic and haemolytic in nature (Ray and Sarkar 2013). Such changes were confirmed by Ahyayauch et al. (2013), from a comparative study of chronically lead exposed erythrocytes, acutely lead exposed erythrocytes, and model membranes (LUV) concluded that in all cases lead increased membrane permeability and the overall membrane lipid order and brought about change in erythrocyte morphology, giving rise to macrocytic and echinocytic forms. Jang et al. (2011) also observed abnormal shape changes of erythrocytes and micro vesicle generation after exposure to Pb^{2+} . The erythrocytes of the lead exposed individuals exhibited anisocytosis, moderate to marked hypochromasia and polychromasia, where many erythrocytes were coarsely or finely stippled. Basophilic stippling of red blood cells (BSC), formed by the inclusions of aggregated ribosomes, often mimicking the siderotic (iron) granules, Heinz bodies or even reticulocytes, has been noted in lead intoxication since 1899 (Cheson et al. 1984) and was considered a classic sign of lead poisoning.

Wistar rats fed with lead and chlorpyrifos showed Anisocytosis, a characteristic feature of moderate to severe anemia. (Ambali et al. 2011). Al-Ali and Abdula (2007) in 24 Balb/c mice found a significant decrease in hemoglobin concentration after acute and chronic lead exposure, that may be resulted from the effect of lead on the haemopoietic system may cause anemia which may result from shortened life span of erythrocytes that due to increase of their membrane fragility, and in another hand, it may cause a decreased hemoglobin synthesis which may result from decreased levels of enzymes involved in heme synthesis.

Previously, lead-induced anemia was considered to be from the inhibition of δ -aminolevulinic acid dehydratase participating in the heme biosynthesis (Jang et al. 2011). They demonstrated that lead could accelerate the splenic sequestration of erythrocytes through phosphatidylserine (PS) exposure and subsequently increased erythrophagocytosis. They found that the oral administration of Pb^{2+} increased phosphatidylserine exposure on erythrocytes in rat in vivo, leading to reduction of hematocrit and hemoglobin.

Reduction in RBCs, PCV, Hb, MCV, MCH and MCHC following exposure to lead acetate in rats revealed microcytic hypochromic anemia due to effect of lead on activity of α aminolevulinic acid dehydratase (ALAD), key enzyme of heme synthesis. Pb²⁺ inhibited the conversion of coproporphyrinogen III to protoporphyrin IX leading to reduction in haemoglobin production and shortened life span of erythrocytes (Klassen 2001). Progressive destruction of RBCs due to binding of Pb²⁺ with RBCs, resulted in the increase of cell fragility and destruction (Rous 2000).

However, cytochemical tests indicated that lead treatment did not interfere with utilization of iron for haemoglobin synthesis and hence did not cause denaturation of haemoglobin in already formed erythrocytes (Ray and Sarkar 2013).

Dongre et al. (2011) made an experiment with 30 automobile workers with occupational lead exposure and same number of healthy people with non-occupational lead exposure of same age as control from same place for hematological analysis. They found decreased non-activated erythrocyte δ-aminolevulinic acid dehydratase $(\delta$ -ALAD) and activated δ -ALAD levels in automobile workers as compared to control subjects. But the ratio of activated/nonactivated δ-ALAD was significantly increased (43.83 %) in automobile workers as compared to controls. In automobile workers heamoglobin, hematocrit, mean corpuscle volume, mean corpuscle hemoglobin, mean corpuscle hemoglobin concentration, red blood cell count were significantly decreased.

Although according to some reports (Tola 1975; Valentine et al. 1982; Grandjean et al. 1989; Romeo et al. 1996) stating that lead exposure did not negatively correlate with Hb levels, and Hb tended to increase with the elevated blood lead in Israeli industrial workers (Froom et al. 1999). Makino et al. (1997) found a significant positive relationship between blood lead and Hb count among 1573 workers with BPbs of 0.05–1.88 µmol/L.

Erythrocyte microrheology changes were measured using cation-osmotic hemolysis (COH) in healthy rats and rats after 6 and 12 months of lead ingestion (Mojzis and Nistiar 2001), where by using COH, properties of two membrane constituents, spectrine membrane skeleton and membrane bilayer were studied. COH in rats after 12 months of lead ingestion was found significantly lower only in the area with lower ionic strength.

Ervthrocvte nucleotidase has been shown to be inhibited by low concentrations of lead in vitro (Paglia et al. 1975). Since blood and erythrocyte lead are virtually synonymous (Albahary 1972), the most lead-sensitive proteins of the erythrocyte remain affected so far the blood lead levels remain elevated. Severe lead-induced deficiency when gives rise to findings similar to the hereditary disorder, whereas pyrimidine-containing nucleotides are virtually absent in the erythrocytes of normal and reticulocyte-rich blood, 12 % of erythrocyte nucleotides in the blood of a patient with lead intoxication contained cytidine. Nucleotidase activity was about 25 % that in normal erythrocytes and 15 % or less of that expected in comparable reticulocyte-rich blood. The distribution of nucleotidase activity in patient erythrocytes is unknown and much more severe deficiency could have been present in subsets of the cell populations analyzed (Valentine et al. 1976). Genetically determined deficiency of this enzyme is associated with chronic hemolysis, marked basophilic stippling of erythrocytes on stained blood films, and unique intra erythrocytic accumulations of pyrimidinecontaining nucleotides.

Changes in Leucocytes

Acute and chronic feeding of lead nitrate resulted in the elevation of total WBC count (Shah and Altindog 2005; Al-Ali and Abdula 2007) with the increase in basophils, eosinophils and lymphocytes and females were more hematologically affected with lead compared to males (Al-Ali and Abdula 2007). Patients with a high level of blood lead were reported to have relatively high number of these types of white blood cells (Al-Temimi et al. 2002).

In automobile workers of Karnataka having regular exposure to lead, increased total white blood cell count as compared to the controls was found (Dongre et al. 2011). Analysis total leucocytes count and differential leukocyte count revealed leucopenia and lymphopenia in lead exposed male and female adult rats (Suradkar et al. 2009). Although no significant change was observed in neutrophil, eosinophil, basophil and monocyte count in any treatment groups, the total lymphocyte count decreased significantly. This was probably due to direct toxic action of lead on leucopoiesis in lymphoid organs or increased lysis due to presence of lead in the body (Avadheshkumar et al. 1998).

In Swiss albino mice, Ray and Sarkar (2013) observed a marked change in TC and DC of leucocytes, where TC of leucocytes was considerably decreased along with a fall in the DC of lymphocytes but with an increase in the number of neutrophils. This led the authors to conclude that chronic lead treatment might gradually exert a cytotoxic action on both lymphopoietic and myelopoietic tissues, resulting in a marked leucopenia in the treated mice.

Sharma et al. (2012) studied the alterations in different types of white blood cells (WBC) due to chronic lead acetate toxicity in neonates after feeding lead acetate to Swiss mice from 10th day of gestation to 21th day of lactation fed on a standard diet and water ad libitum and concluded that lead induced hematological disorders in white blood cells were neutrophilic degeneration, formation of immature cells, abnormal neutrophils, reactive and plasmacytoid lymphocyte, reactive monocyte etc. Lead also caused fluctuations in the number of various cell types at different stages of postnatal development.

Changes in platelets

López et al. (1992) found high levels of lead inhibit aggregation of platelets both in human and rat blood although rat platelets were about ten times more sensitive to the effect of lead than that of in human. According to Ohyashiki et al. (1991), the platelet membrane is highly vulnerable to oxidative stress than the erythrocyte membrane. The levels of platelet count (PLT), plateletcrit (PCT) and mean platelet mass (MPM) were significantly decreased and platelet distribution width (PDW), platelet large cell ratio (P-LCR) and mean platelet volume were increased with an increase in blood lead levels as a result of lead–lead exposure in 429 male workers, serving in acid battery plant process (Barman et al. 2014) this might be the cause of impairment of coagulation function after endothelial tissue injury in these workers.

The thrombocytopenia in male Wistar rats, orally administered with lead and *chlorpyrifos* might be due to oxidative damage to the platelet membranes. This resulted in the formation of lipid peroxides within the platelet membranes thereby provoking cellular lysis (Ambali et al. 2011) and drop in platelet count.

Biochemical Changes

It was found that lead induced changes in the red blood membrane lipids and proteins (Jehan and Motlag 1995; Bellom-Ohvi et al. 1996), in the activity of some membrane-associated enzymes (Jehan and Motlag 1995) or in ions transport mechanisms (Calderon-Sahnas et al. 1999). Furthermore, it was documented that chronic exposure to lead can result in an erythrocyte deformability decrease (Honguchi et al. 1991; Terayama 1993) resulting in anaemia. Jehan and Motlag (1995) found that lead induced drastic changes in the cholesterol and phospholipid content, hexose, hexosamine and sialic acid levels and activities of the erythrocyte membrane enzymes—acetyl cholinesterase (AChE), NADH dehydrogenase and Na⁺–K⁺ ATPase.

Valentine et al. (1976) in the blood of a 26-year-old Mexican–American spray painter, found although the total and differential leukocyte count and blood platelets were normal, but a moderately severe persistent anaemia with reticulocytosis ranging from 9.5 to 24 %. In lead intoxication, porphyrin was also found to accumulate in the erythrocytes, but in the form of zinc protoporphyrin (ZPP) (Lamola et al. 1975).

Lead intoxication is accompanied by a genetically determined acquired deficiency of erythrocyte pyrimidinespecific, 5'-nucleotidase as depicted by the influence of Pb ions on energy metabolism of erythrocytes in rats and consequent alterations in the concentration of high energy purine nucleotides and their derivative (Baranowska-Bosiacka et al. 2004). These rats were found to suffer from chronic hemolysis, marked basophilic stippling of erythrocytes on stained blood films, and unique intra erythrocytic accumulations of pyrimidine-containing nucleotides. The findings indicate that the hemolytic anemia and increased basophilic stippling characteristic of certain cases of lead intoxication might share a common etiology with essentially identical features of the genetically determined disorder (Valentine et al. 1976). Lead was found to cause a decrease of the high-energy purine nucleotide concentrations in erythrocytes and an increase in the products of their catabolism.

Effect of Acute and Chronic Exposure

Pb²⁺ effects on red blood cells after acute treatments reproduce faithfully the effects of chronic intoxication except that acute effects require lead concentrations 2–3 orders of magnitude higher. Liposomes exhibited a similar sensitivity to Pb²⁺ than erythrocytes in acute treatments. The parallelism between membrane effects of Pb²⁺ in the three systems under study reflects that the data are relevant to human intoxication by lead. In particular the observation that exposure to low (1–2 μ M) Pb²⁺ concentrations for extended periods of time has the same effects on membranes to that of acute treatments with 50–500 μ M Pb²⁺ supports the contention that the data in this paper can be at the origin of certain signs, e.g. anaemia, of lead intoxication in humans (Ahyayauch et al. 2013).

From an epidemiologic study in 1974 to assess the association between blood lead level and hematocrit in 579 1 to 5-year-old children living near a primary lead smelter Schwartz et al. (1990) found that blood lead levels ranged from 11 to 164 μ g/dL. They found a strong non-linear, dose–response relationship between blood lead level and hematocrit and this relationship was influenced by age but was independent of iron deficiency. These findings suggested that blood lead levels close to the currently

recommended limit value of 1.21 μ mol/L (25 μ g/dL) are associated with dose-related depression of hematocrit in young children.

Effect of occupational lead-exposure on the gene expression (Sod1) and activity (SOD) of superoxide dismutase, catalase and glutathione peroxidase (GPx, Gpx1) of leukocytes and erythrocytes of healthy male employees of a lead-zinc works resulted in increased expression of Sod1 and Gpx1 in erythrocytes whereas SOD activity in leukocytes was raised only in the subgroups having low exposure to lead (Kasperczyk et al. 2012). The environmental lead exposure in blood inhibits delta-aminolevulinic acid dehydratase activity (ALA-D) from whole blood, tested in 241 urban mothers and their newborns (Campagna et al. 1999).

Remedial Measures

Results of the experiments with Male Sprague–Dawley rats showed that lead administration reduced blood superoxide dismutase (SOD), glutathione peroxidase (GPx) and catalase (CAT) and increased Tumour Necrosis factor (TNF- α); in the controls, but in the exercise group, changes were not statistically significant. Although malondialdehyde (MDA) in both groups increased after lead injections but it was significantly lower in exercise group compared to the sedentary animals, from which Mohammadi et al. (2013) concluded that voluntary exercise might be considered as a preventive tool against lead-induced oxidative stress and inflammation. Modulation of cellular thiols for protection against reactive oxygen species (ROS) has been used as a therapeutic strategy against lead poisoning (Flora et al. 2012).

The role of amino acids and other nutrients in fighting lead poisoning were studied thoroughly by LEAD Group Incorporated ABN (2010). According to their report, other nutrients that have influence on lead level are methionine, glycine, curcumin, methionine, carotene and pectin, and nutrients found in garlic. From animal studies it was found that garlic could reduce blood and tissue lead levels, probably by the sulfur-based compounds essential for amino acid construction and antioxidant function, including methionine. Cysteine, at high concentration, reversed the effect of the metals. D-Penicillamine (D-P) was a chelator drug which is used for treatment of lead toxicity for several years but is no longer in use for its side effects. Chelation can be effective at reducing the total lead body burden and is critical in managing acute toxicity. The preferred chelating agent is Succimer. Ambali et al. (2011) opined that Vitamin C was able to mitigate the anemia induced by co-exposure to CPF and lead through the improvement of iron absorption (Naidu 2003) and reduction of lipoperoxidative damage of the erythrocyte membranes. Salawu (2010) opined that tomato could prevent some major adverse effects of lead intoxication on blood constituents.

Owolabi et al. (2012) studied the regenerative effect of *Moringa oleifera* extract on hematological parameters in lead intoxicated and control adult Wistar rat and found that the extract could reduce the severity of lead toxicity. Flora et al. (2012) pointed that the chelation therapy and newer therapeutic strategies, like nanoencapsulation, to treat lead induced toxic manifestations.

Genetic Susceptibility to Lead Intoxication

Onalja and Claudio (2000) reported the involvement of three very important polymorphic genes influencing susceptibility to lead exposure. Inhibition of erythrocyte δ-ALAD was considered as sensitive indicator of lead exposure. Kelada et al. (2001) described eight variants of this gene but with one polymorphism that yields two polymorphic alleles-ALAD-1 and ALAD-2, imparting susceptibility to lead toxicity. It was found that individuals homozygous for the ALAD-1 allele have a higher body burden of lead and may be at a higher risk of the long term effects of lead. In a case study of a 30-year old male working in a lead acid battery factory for over 6 years, Bijoor and Venkatesh (2007) found that despite very high levels of blood lead (82.8 and 47.5 µg/dL), the patient was asymptomatic with unaffected blood parameters. From detail analysis with PCR-based genotyping followed by an electrophoretic technique, Bijoor and Venkatesh (2007) concluded that some polymorphic genes of ALAD must be operative and confirmed the fact that the effect of lead and the appearance of signs and symptoms of lead poisoning are genetically determined.

Conclusion

Lead has a more severe effect on the blood-forming system in iron deficient people. Globally, governments have undertaken a broad spectrum of regulatory initiatives, both domestically and through international collaboration, to reduce human and ecological exposure to lead and other hazardous chemicals. While progress varies from country to country, it is generally acknowledged that collectively, these measures are contributing to improvements in human health.

Generally young children and expectant mothers, suffering from iron deficiency are more susceptible to lead intoxication than the adult men. Because the combination of iron deficiency and lead exposure cause more severe effects on the blood forming system than either condition alone, women and children tend to show more severe effects. Blood lead level was significantly higher in the exposed workers with anemia than in those without anemia, intake of iron enriched nutrients and food supplemented with sulphur containing amino acids may improve the situation. However, consciousness about the sources of lead poisoning is a pre requisite to avoid the health hazards associated with acute and chronic lead exposure.

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