

# Origin and Evolution Characteristics of Geothermal Water in the Niutuozen Geothermal Field, North China Plain

**Shufang Wang\*** (王树芳)

*Key Laboratory of Engineering Geomechanics, Institute of Geology and Geophysics, Chinese Academy of Sciences, Beijing 100029, China; Graduate University of Chinese Academy of Sciences, Beijing 100049, China; Beijing Institute of Hydrogeology and Engineering Geology, Beijing 100195, China*

**Zhonghe Pang** (庞忠和)

*Key Laboratory of Engineering Geomechanics, Institute of Geology and Geophysics, Chinese Academy of Sciences, Beijing 100029, China*

**Jiurong Liu** (刘久荣), **Pei Lin** (林沛)

*Beijing Institute of Hydrogeology and Engineering Geology, Beijing 100195, China*

**Sida Liu** (刘思达)

*School of Water Resources and Environment, China University of Geosciences, Beijing 100083, China*

**Ming Yin** (殷铭)

*China Institute of Geo-Environment Monitoring, Beijing 100081, China*

**ABSTRACT:** Statistical study of analyses of water from 43 samples from geothermal wells, three groundwater wells, and one sample of local rainwater along with rainwater data from the Global Network of Isotopes in Precipitation has been used to identify the origin and evolution of geothermal water in the Niutuozen (牛驼镇) geothermal field and estimate the renewability rate of its geothermal resource. The results show that the geothermal waters of the Jixianian Wumishanian dolomite reservoir and the Ordovician limestone reservoir are of Cl-Na type, the geothermal water of the Pliocene Minghuazhen (明化镇) Formation sandstone reservoir are Cl-Na type and HCO<sub>3</sub>-Na type and the groundwater of the Quaternary aquifer is HCO<sub>3</sub>-Na and HCO<sub>3</sub>-Na-Mg-Ca type. A linear relationship between silica concentration and temperature indicates that higher temperature probably enhances concentration of silica in Jixianian geothermal water.  $\delta^{18}\text{O}$  shift in Wumishanian geothermal water averaged 1.57‰, and was less than 1‰ in the other geothermal waters. The minimum and maximum <sup>14</sup>C ages of Wumishanian geothermal water are 17 000 and 33 000 years from north to the south of the Niutuozen geothermal field. Geothermal water and Quaternary groundwater belong to different groundwater systems with no hydraulic connections. Although the geothermal field receives some recharge from the Yanshan and Taihang mountains outside the northern and western boundaries of the geothermal field respectively, the renewability rate of geothermal

---

This study was supported by the National Basic Research Program of China (No. 2010CB428806) and Beijing Municipal Science and Technology Project (No. D07050601510000).

\*Corresponding author: shufangwang111@163.com

© China University of Geosciences and Springer-Verlag Berlin Heidelberg 2013

Manuscript received June 14, 2013.

Manuscript accepted October 22, 2013.

water are 17 000 and 33 000 years from north to the south of the Niutuozen geothermal field. Geothermal water and Quaternary groundwater belong to different groundwater systems with no hydraulic connections. Although the geothermal field receives some recharge from the Yanshan and Taihang mountains outside the northern and western boundaries of the geothermal field respectively, the renewability rate of geothermal

water is on the scale of 10 000 years.

**KEY WORDS:** hydrogeochemistry, isotope, geothermal water, origin and evolution, Niutuozen geothermal field.

## INTRODUCTION

Geothermal resources can provide clean green energy and have recently received considerable attention. The Niutuozen Geothermal Field is located in the Niutuozen Uplift in North China (Fig. 1) and its geothermal resources have been developed for more than 30 years (Chen, 1988). Geothermal water is extensively used for space heating in Xiongxian, Gu'an and Baxian to construct smoke-free cities. There is an urgent need for further research on geothermal energy potential, reservoir temperature and water chemistry because of rapidly increasing demand (Dotsika et al., 2010; Wu and Ma, 2010; Gao et al., 2009; Liu H et al., 2009; Zhang et al., 2007; Liu J R et al., 2002; Giggenbach et al., 1995; Wang and Shen, 1993; Wang J Y et al., 1993; Wang Y X et al., 1993; Giggenbach, 1992, 1988; Dai et al., 1988). Chemical and isotopic components record the origin and evolution of geothermal water which reflect geothermal energy potential (Shen and Wang, 2002; Bi, 1998; Chen, 1998; Wang and

Shen, 1993; Wang Y X et al., 1993). Twenty years ago discussion of the origin and evolution of geothermal resources throughout the north China Basin was based on relatively few geothermal water samples (Yao, 1995; Chen, 1988; Zhou, 1987; Geothermal Group, 1983) and did not achieve modern levels of accuracy. The present work includes new sampling sites and presents modern analyses for major ions,  $\delta^2\text{H}$ ,  $\delta^{18}\text{O}$ , tritium and  $^{14}\text{C}$  ages of 43 samples from geothermal wells, three samples from groundwater wells and one rain water sample along with 14 rain water analyses obtained from the Global Network of Isotopes in Precipitation (GNIP).

## GEOLOGICAL SETTING AND GEOTHERMAL RESOURCES

### Geological Settings

The Niutuozen Geothermal Field is located in the Niutuozen uplift zone in the northern part of the Jizhong graben in the North China Basin bounded by

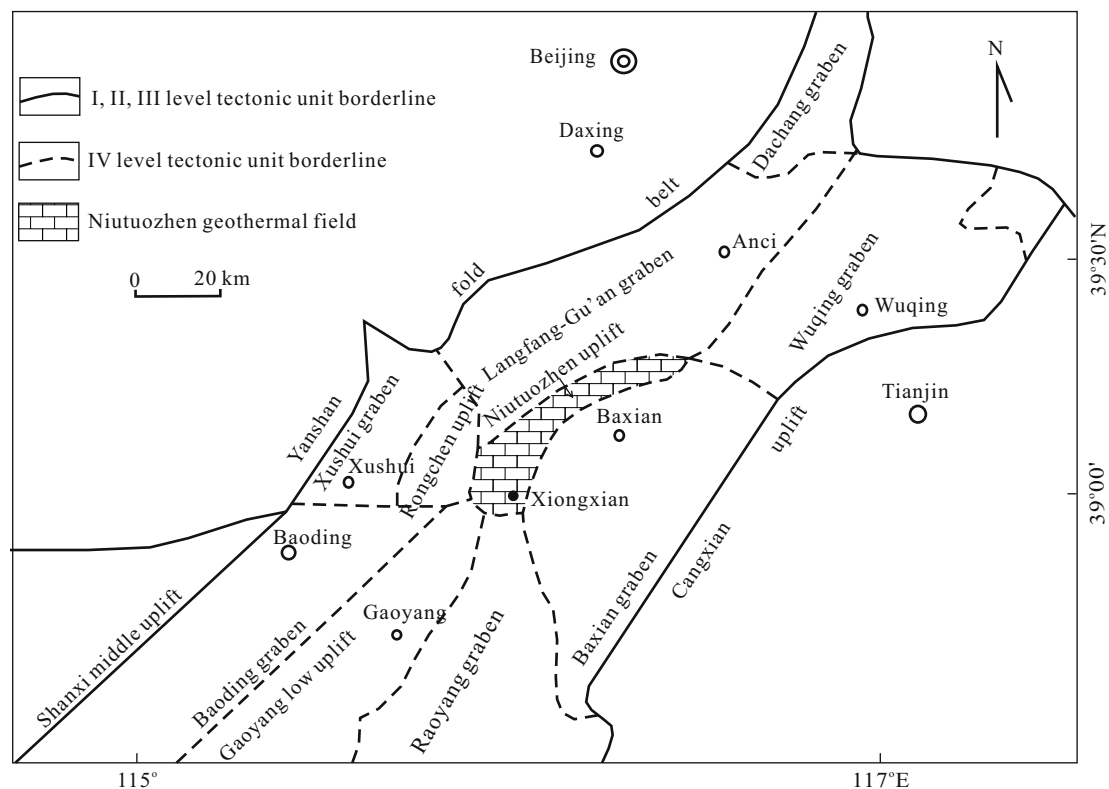


Figure 1. Location of the Niutuozen uplift.

the Yanshan Mountains and the Taihang Mountains. The basin began developing by vertical crustal movement before the Late Triassic Period, followed by mountain building from Jurassic to Early Tertiary and Late Miocene subsidence. NE-SW striking horsts and grabens formed alternately from west to east. The Niutuozen uplift (or horst) is bounded by the Niudong fault, Niunan fault, Rongcheng fault and Dax-

ing fault initiated during Late Jurassic to Cretaceous earth movements and finally during Himalayan movements. The uplift continued to move up to the surface until the Late Oligocene when it became mature and was eroded. The Minghuazhen Formation was deposited over the entire horst by the Late Miocene (Fig. 2) and it was finally buried locally under Quaternary sediments.

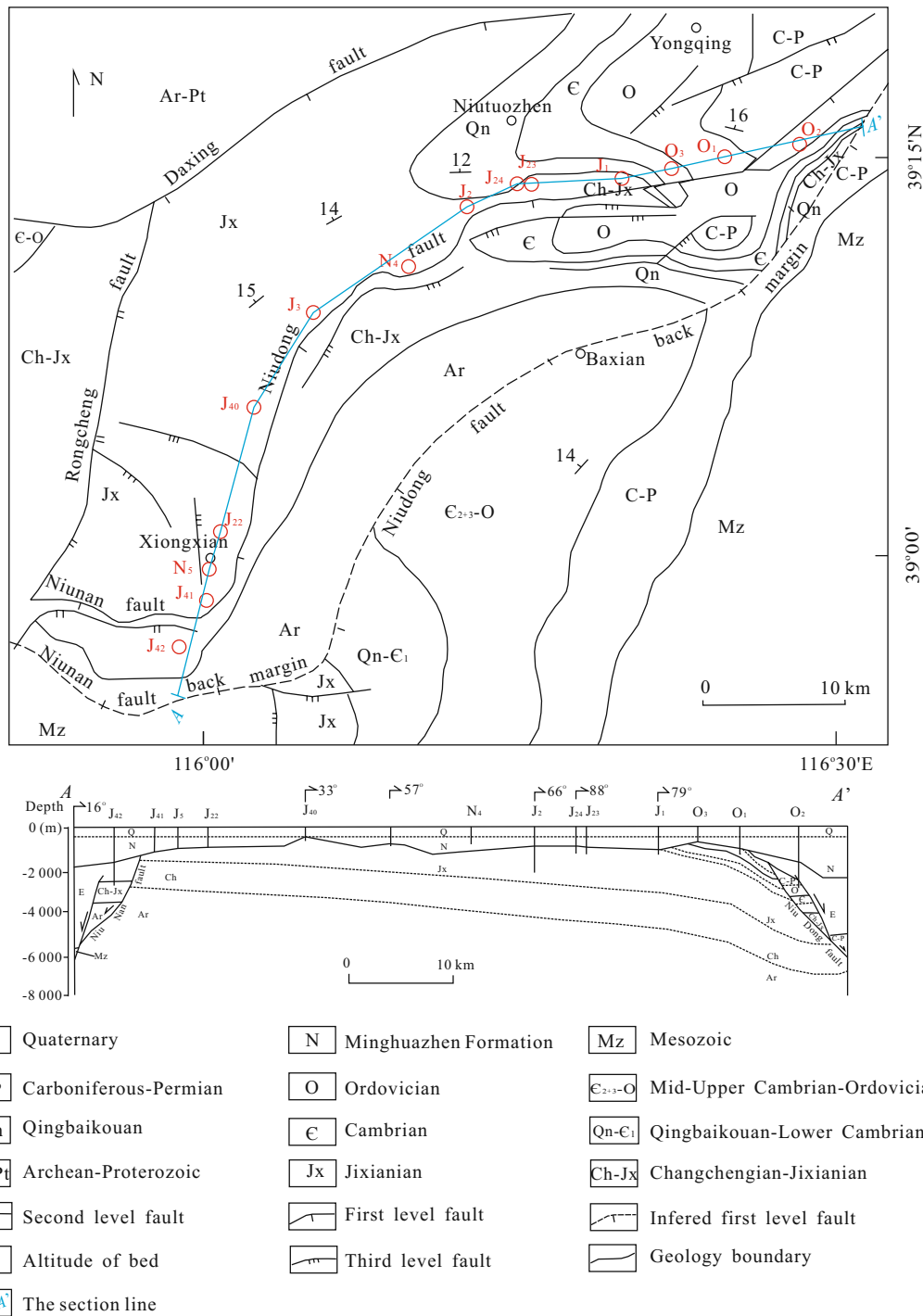


Figure 2. Pre-Cenozoic geological map and cross-section of the Niutuozen geothermal field.

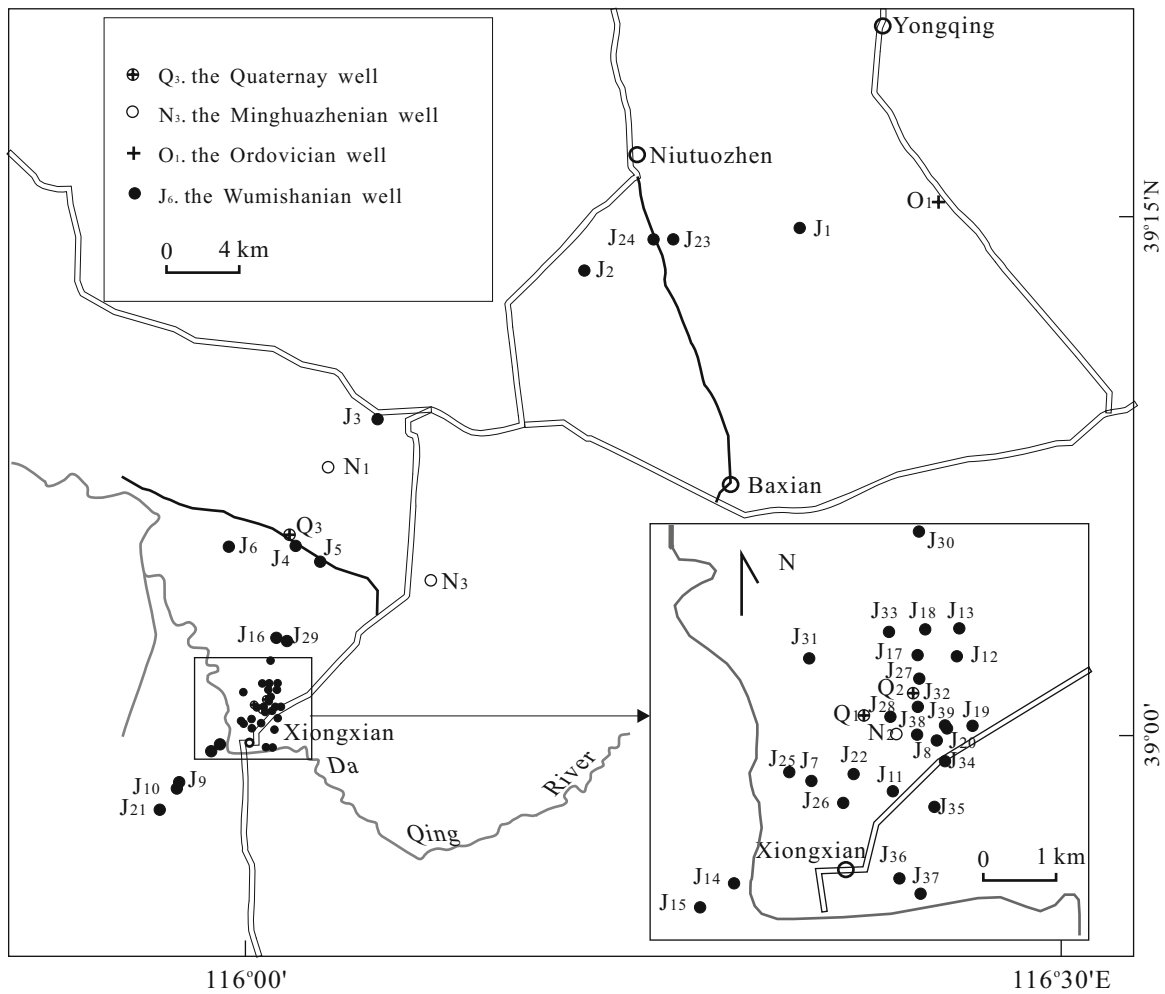
The Neogene Minghuazhen Formation sandstone with an average effective thickness of 225 m and Jixianian Wumishanian Formation dolomite with a thickness more than 1 000 m are the main reservoirs in the Niutuozen geothermal field. Quaternary clay with a thickness of 400 m is the main caprock of the Minghuazhen Formation reservoir, and mudstone of the Minghuazhen Formation and Quaternary clay more than 493 m thick are the caprocks of the Wumishanian dolomite reservoir.

The geothermal gradient in the cap rock of the Niutuozen uplift is more than 3.0 °C/100 m, reaching a maximum of 11.5 °C/100 m in the axis of the uplift (Chen, 1988). The gradient in the Wumishanian reservoir is generally between 1.65 and 2.55 °C/100 m (Chen, 1988; Geothermal Group, 1983). Maximum wellhead pressure was measured at  $3.74 \times 10^5$  Pa in 1970s (Chen, 1988), but has decreased to its present level of less than  $1.01 \times 10^5$  Pa with drastic develop-

ment of the geothermal water and no reinjection. Wellhead temperature ranges from 59 to 84 °C. The permeability of the Minghuazhen Formation Reservoir is between  $3.36 \times 10^{-14}$  and  $2.39 \times 10^{-12}$  m<sup>2</sup> and of the Wumishanian reservoir is between  $2.10 \times 10^{-11}$  and  $6.15 \times 10^{-10}$  m<sup>2</sup>.

**METHODS**

Samples were taken from a Quaternary cold water aquifer, the Minghuazhen Formation sandstone geothermal reservoir, an Ordovician limestone geothermal reservoir and the Wumishanian dolomite geothermal Reservoir and rainwater in the Niutuozen geothermal field in order to compare geochemical characteristics between different waters (Fig. 3). The samples were taken by inserting a sampling tube against the pressure of a water discharge main. Temperature and pH were measured in-situ, other analysis were made in laboratory. K, Na, Li and a wide range



**Figure 3. Sampling sites.**

of metal ions were analysed by atomic absorption; total acidity, total hardness and most anions were analysed by titration;  $\text{NH}_4^+$ ,  $\text{Fe}^{2+}$ ,  $\text{Fe}^{3+}$ , metaboric acid, metasilicic acid, metaphosphoric acid and some halide ions were analysed by spectrophotometry; and  $\text{NO}_3^-$  was analysed by ultraviolet spectrophotometry; selenium, metaarsenic acid and mercury were analysed by atomic fluorescence; and total dissolved solids were analysed by gravimetry.

Stable isotopes ( $^{18}\text{O}$ ,  $^2\text{H}$ ) were analysed by isotope ratio mass spectrometry and Picarro L1102-i liquid water isotope analyzer and  $^{14}\text{C}$  samples were analyzed by Liquid Scintillation Counter.

## RESULTS AND DISCUSSION

### Hydrogeochemistry

Geothermal waters from the Wumishanian and Ordovician reservoirs are of Cl-Na type, and those of the Minghuazhen Formation reservoir are of Cl-Na type and  $\text{HCO}_3^-$ -Na type (Table 1, Fig. 4). Wumisha-

nian geothermal water has higher total dissolved solid (TDS) ranging from 2 300 to 3 000 mg/L, than Ordovician reservoir water (2 206 mg/L) and Minghuazhen Formation (561–1 427.7 mg/L) geothermal water.

### Symbolic Components

Silica ( $\text{SiO}_2$ ) is the most significant symbolic component of geothermal water (Liu et al., 2002) and all water samples from Wumishanian geothermal reservoir are plotted in Fig. 5. The concentration of silica in Wumishanian geothermal water increases from 40 to 80 mg/L with temperature increase from 60 to 85 °C demonstrating a good linear relationship ( $R^2=0.842$ ) between silica concentration and temperature. The linear relationship indicates that temperature instead of lithology probably enhances the enrichment of  $\text{SiO}_2$  in such a low-temperature geothermal system because the geothermal water is stored in the Jixianian dolomite geothermal reservoir (Li, 1982; Browne, 1978).

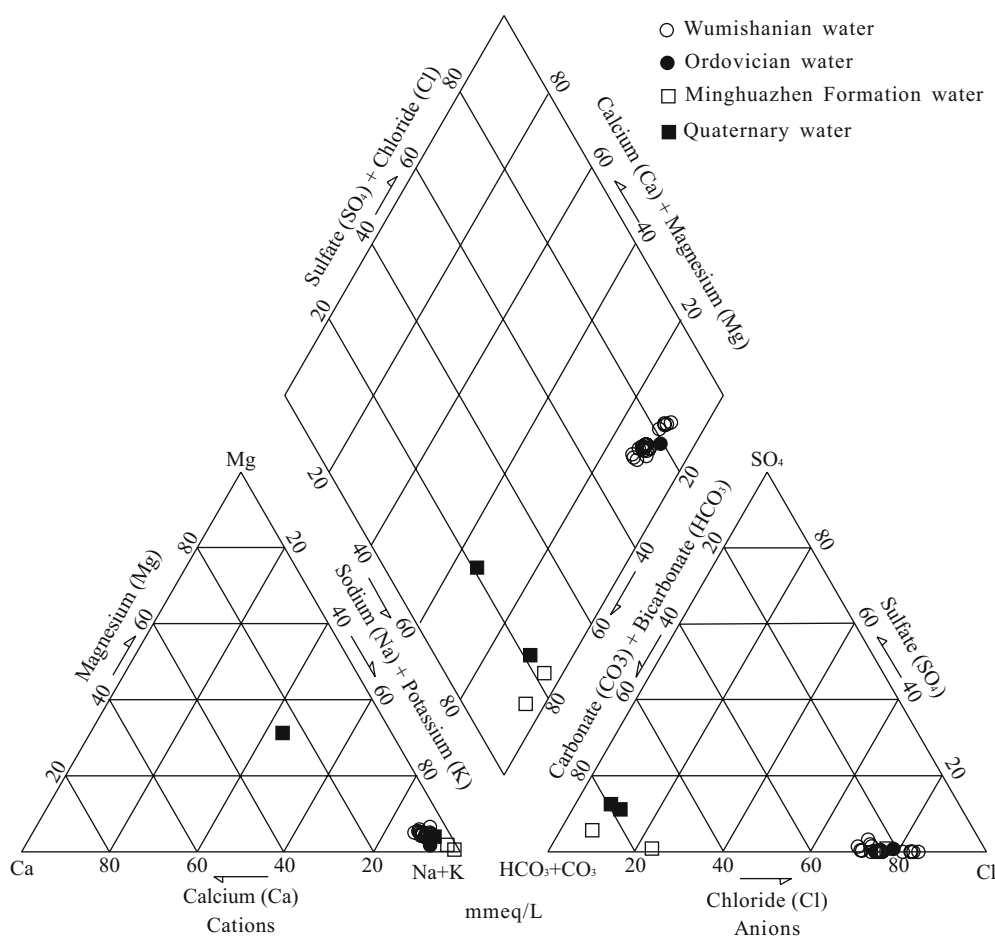


Figure 4. Piper diagram of compositions of geothermal water and groundwater.

**Table 1 The chemical components of the samples**

No.	Name	Water source	K <sup>+</sup>	Na <sup>+</sup>	Ca <sup>2+</sup>	Mg <sup>2+</sup>	Cl <sup>-</sup>	SO <sub>4</sub> <sup>2-</sup>	HCO <sub>3</sub> <sup>2-</sup>	TDS	T	δ <sup>2</sup> H	δ <sup>18</sup> O	Tritium	<sup>14</sup> C age	Other
			(mg/L)							(°C)	(‰)	(TU)	(a)			
1	J1	Jxw	58.0	910	39.08	21.88	1 400	0.5	442.4	2 886	83	-77.40	-9.29		17 211±874	a
2	J2	Jxw	57.0	900	39.08	25.53	1 365	0.2	469.8	2 868	85	-86.90	-9.11		22 912±640	a
3	J3	Jxw	58.2	910	40.08	27.96	1 347	0.6	485.1	2 879	74	-79.30	-9.18			a
4	J4	Jxw	57.0	900	46.09	25.28	1 365	0.0	485.1	2 889	78	-77.60	-9.18			a
5	J5	Jxw	56.0	880	32.06	35.25	1 294	0.0	524.7	2 833	70	-81.90	-9.08		22 185±365	b
6	J6	Jxw	57.8	838	55.11	31.00	1 143	0.5	701.7	2 929	60	-79.00	-9.08			a
7	J7	Jxw	58.0	832	56.11	24.07	1 179	0.2	655.9	2 814	76	-76.80	-8.98	<0.5	25 704±1 566	a
8	J8	Jxw	62.8	897	56.70	27.70	1 220	<0.2	669.0	2 940	59					a
9	J9	Jxw	66.2	911	54.10	22.10	1 201	<0.2	646.0	2 910	83					b
10	J10	Jxw	70.3	867	51.70	23.10	1 140	<0.5	647.0	2 800	83					b
11	J11	Jxw	63.3	848	55.10	25.90	1 160	<0.5	671.0	2 830	72					b
12	J12	Jxw	53.8	858	58.10	27.70	1 160	1.4	677.0	2 850	65					b
13	J13	Jxw	55.9	897	60.21	26.69	1 123	68.3	687.1	3 005	65					c
14	J14	Jxw	51.4	865	60.10	22.60	1 110	6.7	631.0	2 820	81	-75.30	-8.70			c
15	J15	Jxw	57.8	824	52.10	24.30	1 140	8.6	622.0	2 790	79					c
16	J16	Jxw	57.7	825	52.09	24.20	1 140	8.5	621.0	2 781	66					c
17	J17	Jxw	47.8	743	61.01	23.83	981	8.0	676.8	2 622	64					c
18	J18	Jxw	53.8	865	49.06	20.27	1 080	39.7	654.7	2 853	62					a
19	J19	Jxw	50.0	812	55.77	22.33	984	6.4	668.1	2 683	64					a
20	J20	Jxw	42.4	709	49.91	23.97	938	25.5	660.5	2 533	64					a
21	J21	Jxw	58.0	870	52.10	20.66	1 179	1.3	631.5	2 821	80	-79.30	-8.84	<0.5	33 536±3 697	a
22	J22	Jxw										-77.90	-8.66			a

## Continued

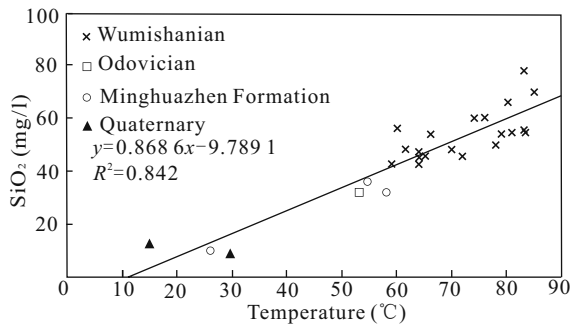
No.	Name	Water source	Water source							TDS	T (°C)	$\delta^2\text{H}$ (‰)	$\delta^{18}\text{O}$ (‰)	Tritium (TU)	$^{14}\text{C}$ age (a)	Other
			$\text{K}^+$	$\text{Na}^+$	$\text{Ca}^{2+}$	$\text{Mg}^{2+}$	$\text{Cl}^-$	$\text{SO}_4^{2-}$	$\text{HCO}_3^{2-}$							
23	J23	Jxw									-75.10	-8.91			a	
24	J24	Jxw									-78.00	-9.13			a	
25	J25	Jxw									-74.10	-8.70			d	
26	J26	Jxw									-73.70	-8.90			d	
27	J27	Jxw									-73.90	-8.80			d	
28	J28	Jxw									-72.90	-8.60			d	
29	J29	Jxw									-73.70	-8.90			d	
30	J30	Jxw									-73.70	-8.70			d	
31	J31	Jxw									-73.30	-8.70			d	
32	J32	Jxw									-73.70	-8.90			d	
33	J33	Jxw									-73.10	-8.60			d	
34	J34	Jxw									-73.70	-8.70			d	
35	J35	Jxw									-73.60	-8.80			d	
36	J36	Jxw									-72.80	-8.60			d	
37	J37	Jxw									-73.50	-8.80			d	
38	J38	Jxw									-73.70	-8.80			d	
39	J39	Jxw									-72.20	-8.50			d	
40	O1	O	24.0	702	47.09	10.33	968	0.9	445	2 206	53	-75.80	-9.87		12 461 ± 419	a
41	N1	Nim	10.0	449	28.06	11.55	573	12.6	339	1 428	58	-77.17	-10.64		27 171 ± 384	a
42	N2	Nim	14.9	299	5.60	1.20	112	<0.5	603.0	1 050	54.5	-78.92	-10.02	<0.5	26 900 ± 650	a
43	N3	Nim	1.0	146	4.01	1.82	20	20.77	366.1	561	26	-78.00	-11.18		25 142 ± 400	a
44	Q1	Q	0.8	103	4.61	3.04	18	32.29	244.1	413	29.5	-89.20	-11.90			a
45	Q2	Q	0.5	67	34.07	23.10	20	42.17	320.3	508	15	-71.50	-10.44			a
46	Q3	Q										-66.36	-8.79	67.8		a

**Continued**

No.	Name	Water source	K <sup>+</sup> Na <sup>+</sup> Ca <sup>2+</sup> Mg <sup>2+</sup>				Cl <sup>-</sup>	SO <sub>4</sub> <sup>2-</sup>	HCO <sub>3</sub> <sup>2-</sup>	TDS	T (°C)	δ <sup>2</sup> H (‰)	δ <sup>18</sup> O (‰)	Tritium (TU)	<sup>14</sup> C age (a)	Other
			K <sup>+</sup>	Na <sup>+</sup>	Ca <sup>2+</sup>	Mg <sup>2+</sup>										
47	Xiongxian	Rain									-58.79	-8.06			a	
48	Shijiazhuang	Rain									-52.9	-7.44			e1986	
49	Shijiazhuang	Rain									-48.4	-7.06			e1987	
50	Shijiazhuang	Rain									-43.1	-6.8			e1988	
51	Shijiazhuang	Rain									-57.4	-8.02			e1989	
52	Shijiazhuang	Rain									-62.1	-8.36			e1990	
53	Shijiazhuang	Rain									-63.5	-7.97			e1991	
54	Shijiazhuang	Rain									-63.1	-8.67			e1996	
55	Shijiazhuang	Rain									-54.3	-7.39			e1997	
56	Shijiazhuang	Rain									-39.7	-6.5			e1998	
57	Shijiazhuang	Rain									-51.3	-7.79			e1999	
58	Shijiazhuang	Rain									-60.2	-9.1			e2000	
59	Shijiazhuang	Rain									-56.9	-8.13			e2001	
60	Shijiazhuang	Rain									-46.4	-7.22			e2002	
61	Shijiazhuang	Rain									-52.5	-7.58			e2003	

a. From "Geothermal Survey Report of the Niutuozhen Geothermal Field in Hebei Province"; b. from Beijing Institute of Hydrogeology and Engineering Geology; c. from Shaanxi Green Energy Geothermal Development Co. Ltd.; d. from Beijing Institute of Hydrogeology and Engineering Geology, analyzed by Laboratory of Water Isotopes and Water-Rock Interaction, Institute of Geology and Geophysics, Chinese Academy of Sciences; e. the isotope data of rainfall in Shijiazhuang meteorological stations, 207 km SW from Xiongxian, in a representative position in North China Plain were available from the Global Network of Isotopes in Precipitation (GNIP). The GNIP Database was accessible at <http://www.iaea.org/water>. The laboratory and management institute was the Institute of Hydrogeology & Engineering Geology (Shijiazhuang, Hebei, China) (1986 means sampling year); f. the Quaternary groundwater is HCO<sub>3</sub><sup>-</sup>-Na and HCO<sub>3</sub><sup>-</sup>-Na-Mg-Ca type (Fig. 4) with TDS between 413.3 to 507.6 mg/L. The SO<sub>4</sub><sup>2-</sup> concentration of the geothermal water was measured to be extremely low (some <0.2 mg/L), indicative of desulphurization in a confined structure.



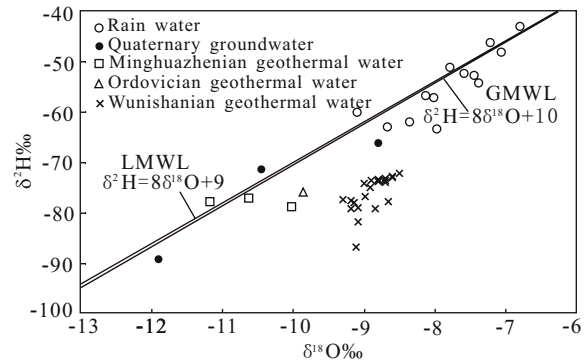


**Figure 5. Relationship between well head temperature and SiO<sub>2</sub> concentration**

**Origin of geothermal water**

Craig (1961) found that there was  $\delta^{18}\text{O}$  shift when geothermal water is compared with the Global Meteoric Water Line, which was brought about by  $^{18}\text{O}$  exchange between geothermal water and rock under high temperature conditions ( $>300\text{ }^\circ\text{C}$ ). He found that change of  $\delta^2\text{H}$  due to depletion of H in the rock was negligible compared with  $\delta^{18}\text{O}$ . Wen et al. (2010), Ma et al. (2008), Zhang et al. (2006), Pang (2004), Zhao et al. (2001), Wang Y X et al. (1993), Sun et al. (1992) and Pang et al. (1990a, b) have also used isotopic methods to study origin and mixing of geothermal water in different geothermal fields.

Stable isotope data for all geothermal waters (Fig. 6) show a  $\delta^{18}\text{O}$  range from  $-8.66\text{‰}$  to  $-11.90\text{‰}$ . It is obvious that the geothermal water from the Ordovician and the Minghuazhen Formation geothermal reservoir and from the Quaternary groundwater fall near the GMWL. However, samples from the Wumishanian geothermal reservoir demonstrate an average  $1.57\text{‰}$   $^{18}\text{O}$  shift, slight compared with samples from a Tibet high temperature geothermal system (Zhou et al., 2009). Truesdell and Hulston (1980) found that the

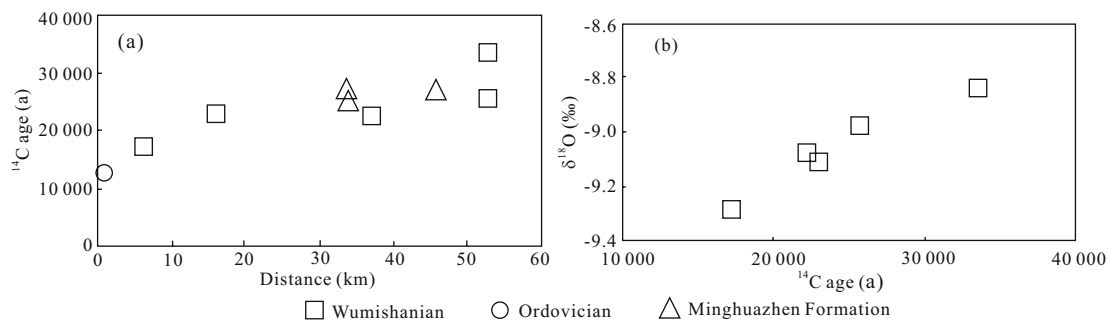


**Figure 6.  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  of different groundwaters.**

mineral components of reservoir rock are a significant factor affecting  $\delta^{18}\text{O}$  shift and that maximum shift occurred in carbonate rock systems. Exchange of  $^{18}\text{O}$  between deeply circulating meteoric water and hot carbonates in deep-seated aquifers for a long time (Zhou et al., 2004, 2001) may explain the slight  $\text{O}^{18}$  shift.

**Evolution of geothermal water**

In order to study the evolution characteristics of geothermal water along a flow line, Well O1 in the northernmost part of the geothermal field was taken as origin point. A plot of  $^{14}\text{C}$  age along the flow line shows a progressive increase indicating age increasing from north to south (Fig. 7a). A similar trend was also found in the studies of Chen (1988) and Geothermal Group (1983). There is a significant positive linear correlation between  $^{14}\text{C}$  age and  $\delta^{18}\text{O}$  (Fig. 7b) which is good evidence that the extent of  $^{18}\text{O}$  exchange between geothermal water and reservoir rock was enhanced with the increasing residual time of geothermal water.



**Figure 7. The relationship of  $^{14}\text{C}$  age with distance and  $\delta^{18}\text{O}$ .**

**DISCUSSION**

The high concentration of Cl and Na in the Wumishanian and Ordovician geothermal water indicates that the water experienced long time lixiviation from the recharge area during the process of flowing from north to south in the geothermal field (Wang et al., 1995), similar to the geothermal water in the geothermal fields of Tianjin and Beijing (Hu et al., 2007) and confirmed by the <sup>14</sup>C age of the geothermal water.

The remarkably differences in geochemical characteristics between Quaternary, Minghuanian, Ordovician and Wumishanian geothermal waters suggests that they derive from different groundwater systems (Zhou, 1987). The Quaternary groundwater system is mainly recharged by local modern rain and groundwater recharge from mountains to the north and west (Chen et al., 2010). By contrast, geothermal water of the Minghuazhenian, Ordovician and Wumishanian reservoirs was recharged by paleo- precipitation at least 12 000 years ago (Table 1).

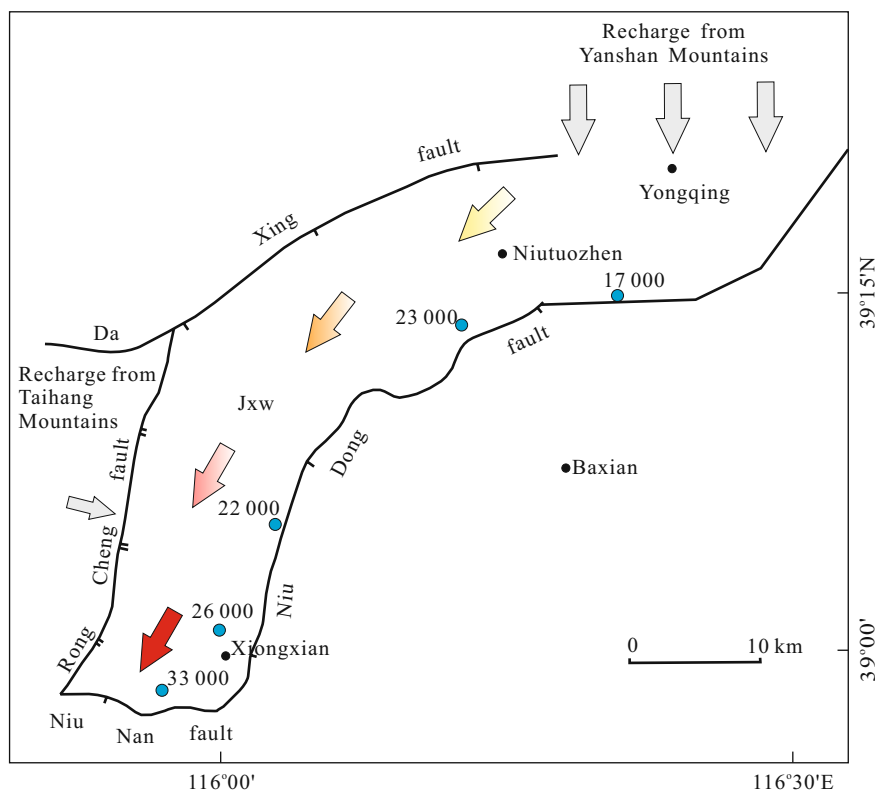
Its tritium isotope concentration 67.8 TU indicates that some Quaternary groundwater receives modern rain recharge. The δ<sup>18</sup>O and δ<sup>2</sup>H values of -8.79‰ and -66.36‰ of Quaternary geothermal water

are close to modern rain water but its tritium isotope value <0.5 TU indicates that this water has not been recharged by modern rain, in agreement with the <sup>14</sup>C results (Table 1).

<sup>14</sup>C ages indicate that the Wumishanian geothermal water flows from north to south at an average velocity of 3.6 m/a in the Niutuozen geothermal field (Fig. 8).

With increasing time for water-rock <sup>18</sup>O exchange processes, <sup>18</sup>O in the geothermal water is progressively enriched along the flow line from north to south as demonstrated by the good linear relationship between δ<sup>18</sup>O and <sup>14</sup>C age in Fig. 7b.

According to research by Yan and Yu (2000) and Zhou (1987), the Niutuozen geothermal field receives recharge both from the Yanshan Mountains in the north (about 100 km from the northern part of the geothermal field) and the Taihang Mountains in the west (about 50 km from the west boundary of the geothermal field) and moving Wumishanian geothermal water from the Yanshan Mountains may be diluted by water from the Taihang Mountains. Water from the Taihang Mountains is generally younger than water from Yanshan Mountains because of relative



**Figure 8. A conceptual evolution model of Wumishanian geothermal water.**

shorter flow distance. It could also be inferred that the water in the south of the geothermal field should be older than 33 000 year if it was only recharged by water from the north without being diluted by water from the west.

## ACKNOWLEDGMENTS

We would like to express our thanks to our workmates that made the measurements of geochemistry of geothermal water samples in the Laboratory of Beijing Institute of Hydrogeology and Engineering Geology. The authors are grateful to Water Isotopes and Water Rock Interaction laboratory for stable isotopes ( $^{18}\text{O}$ ,  $^2\text{H}$ ) analysis and State Key Laboratory of Earthquake Dynamics, Institute of Geology, China Earthquake Administration for  $^{14}\text{C}$  samples analysis.

## REFERENCES CITED

- Bi, E. P., 1998. Geochemical Modeling of the Mixing of Geothermal Water and Reinjection Water: A Case Study of Laugaland Low-Temperature Geothermal Field in Iceland. *Earth Science—Journal of China University of Geosciences*, 23(6): 631–634 (in Chinese with English Abstract)
- Browne, P. R. L., 1978. Hydrothermal Alteration in Active Geothermal Fields. *Annual Review of Earth and Planetary Sciences*, 6(A78-38764 16-42): 229–250, doi:10.1146/annurev.ea.06.050178.001305
- Chen, M. X., 1988. Geothermal Resources in North China. Science Press, Beijing. 1–218 (in Chinese)
- Chen, Z., 1998. Modeling Water-Rock Interaction of Geothermal Reinjection in the Tanggu Low-Temperature Field, Tianjin. *Earth Science—Journal of China University of Geosciences*, 23(5): 631–634 (in Chinese with English Abstract)
- Chen, Z. Y., Qi, J. X., Zhang, Z. J., et al., 2010. Isotope Hydrogeology Application in North Typical Basin. Science Press, Beijing. 1–146 (in Chinese)
- Craig, H., 1961. Isotopic Variations in Meteoric Waters. *Science*, 133(3465): 1702–1703, doi:10.1126/science.133.3465.1702
- Dai, Z. H., Guo, Y. X., Yang, J. C., 1988. Discussion on Geochemical Anomalies in Zhangzhou Geothermal Surveying Area. *Earth Science—Journal of China University of Geosciences*, 13(3): 255–262 (in Chinese with English Abstract)
- Dotsika, E., Poutoukis, D., Raco, B., 2010. Fluid Geochemistry of the Methana Peninsula and Loutraki Geothermal Area, Greece. *Journal of Geochemical Exploration*, 104(3): 97–104, doi:10.1016/j.gexplo.2010.01.001
- Gao, B. Z., Li, X. M., Nie, R. P., 2009. Hydrochemical Properties of Geothermal Fluids in Ordovician Reservoirs of Tianjin and Main Affecting Factors. *Acta Geoscientica Sinica*, 30(3): 369–374 (in Chinese with English Abstract)
- Geothermal Group, 1983. A Preliminary Investigation on the Characteristics of a Geothermal Field and the Conditions for Its Formations in the Northern Part of the North China Plain. *Bulletin of the 562 Comprehensive Geological Brigade, Chinese Academy of Geological Sciences*, (4): 109–126 (in Chinese with English Abstract)
- Giggenbach, W. F., 1988. Geothermal Solute Equilibria. Derivation of Na-K-Mg-Ca Geoindicators. *Geochim. Cosmochim. Acta*, 52(12): 2749–2765, doi:10.1016/0016-7037(88)90143-3
- Giggenbach, W. F., 1992. Isotopic Shifts in Waters from Geothermal and Volcanic Systems along Convergent Plate Boundaries and Their Origin. *Earth Planet. Sci. Lett.*, 113(4): 495–510, doi:10.1016/0012-821X(92)90127-H
- Giggenbach, W. F., Stewart, M. K., Sano, Y., et al., 1995. Isotope and Geochemical Techniques Applied to Geothermal Investigations. *IAEA-TECDOC*, 788: 209–231
- Hu, Y., Gao, B. Z., Jin, B. Z., et al., 2007. Geochemical Characteristics of the Geothermal Fluids and Formation Mechanism in Tianjin. *Geological Survey and Research*, 30(3): 213–218 (in Chinese with English Abstract)
- Li, X. L., 1982. Silica in Groundwater—Formation of Micro-Mineralization Silica Acid, Silica Acid-Bicarbonate Type Acidic Phreatic Water and Weak Mineralization Basic Siliceous Geothermal Water and Its Geochemical Significance for the Transportation of Uranium. *Journal of East China Geological Institute*, (1): 86–92 (in Chinese)
- Liu, H., Zhang, G. P., Jin, Z. S., et al., 2009. Geochemical Characteristics of Geothermal Fluid in Tengchong Area, Yunnan Province, China. *Acta Mineralogica Sinica*, 29(4): 496–501 (in Chinese with English Abstract)
- Liu, J. R., Pan, X. P., Yang, Y. J., 2002. Long-Term Geochemistry Changes of Geothermal Water from a Geothermal Well in the Urban Geothermal Field, Beijing. *Geoscience*, 16(3): 318–321 (in Chinese with English Abstract)
- Ma, Z. Y., Yu, J., Li, Q., et al., 2008. Environmental Isotope Distribution and Hydrologic Geologic Sense of Gua-

- nzhong Basin Geothermal Water. *Journal of Earth Sciences and Environment*, 30(4): 396–401 (in Chinese with English Abstract)
- Pang, Z. H., Fan, Z. C., Wang, J. Y., 1990a. Isotope Evidence for Geothermal Water Genesis and Seawater Involvement in Zhengzhou Basin, Southeast China. *Geochimica*, (4): 296–302 (in Chinese with English Abstract)
- Pang, Z. H., Fan, Z. C., Wang, J. Y., 1990b. The Study on Stable Oxygen and Hydrogen Isotopes in the Zhangzhou Basin Hydrothermal System. *Acta Petrologica Sinica*, (4): 75–84 (in Chinese with English Abstract)
- Pang, Z. H., 2004. International Research Cooperation and Development Assistance in Isotope Hydrology. *Hydrogeology and Engineering Geology*, (2): 114–116 (in Chinese)
- Shen, Z. L., Wang, Y. X., 2002. Review and Outlook of Water-Rock Interaction Studies. *Earth Science—Journal of China University of Geosciences*, 27(2): 127–133 (in Chinese with English Abstract)
- Sun, Z. X., Li, X. L., Shi, W. J., 1992. Isotopic Hydrogeochemistry of Mid-Low Temperature Geothermal Water in Jiangxi Province. *Journal of East China Geological Institute*, 15(3): 243–248 (in Chinese with English Abstract)
- Truesdell, A. H., Hulston, J. R., 1980. Isotopic Evidence on Environments of Geothermal Systems, Chapter 5. In: Fritz, P., Fontes, J. C. eds., *Handbook of Environmental Isotope Geochemistry*, Vol. 1. Elsevier, Amsterdam. 179–226
- Wang, D. C., Zhang, R. Q., Shi, Y. H., et al., 1995. Principles of Hydrogeology. Geological Publishing House, Beijing. 1–160 (in Chinese)
- Wang, J. Y., Xiong, L. P., Pang, Z. H., 1993. Low-Medium Temperature Geothermal System of Convective Type. Science Press, Beijing. 1–240 (in Chinese with English Abstract)
- Wang, Y. X., Shen, Z. L., 1993. Hydrogeochemistry of the Magmatic Fluid. *Earth Science—Journal of China University of Geosciences*, 18(4): 504–506 (in Chinese with English Abstract)
- Wang, Y. X., Sun, L. F., Chen, D. L., et al., 1993. A Comparative Study of Hydrochemistry on Geothermal Mineral Waters in Xinzhou Basin and Baikal Rift Zone. *Earth Science—Journal of China University of Geosciences*, 18(5): 661–670 (in Chinese with English Abstract)
- Wen, Y. H., Wang, N. A., Zhu, X. F., et al., 2010. Hydrochemistry and Origin of the Wushan Geothermal Field, Gansu. *Journal of Natural Resources*, 25(7): 1186–1193 (in Chinese with English Abstract)
- Wu, K. J., Ma, C. M., 2010. Geochemical Characteristics of Geothermal Water in Zhengzhou City. *Geotechnical Investigation & Surveying*, (3): 45–49 (in Chinese with English Abstract)
- Yan, D. S., Yu, Y. T., 2000. Assessment and Development of the Geothermal Resources of Jing, Jin and Ji Oil Area. China University of Geosciences Press, Beijing. 1–179 (in Chinese)
- Yao, Z. J., 1995. Paleoclimate Record of Geothermal Water for Last 0.03 Ma in North China. *Earth Science—Journal of China University of Geosciences*, 20(4): 383–388 (in Chinese with English Abstract)
- Zhang, B. M., Wang, X. Y., Lin, J. W., 2006. Isotopes Characteristics Analysis of Geothermal Water of Tianjin Geothermal Field. *West-China Exploration Engineering*, 119(3): 85–88 (in Chinese)
- Zhang, X. L., Liang, X., Sun, J., 2007. Hydrochemical Characteristic and Modeling of the Mixture Action in Qicun Geothermal Field. *Hydrogeology and Engineering Geology*, (6): 95–99 (in Chinese with English Abstract)
- Zhao, P., Mack, K., Duo, J., et al., 2001. Noble Gases Constraints on the Origin and Evolution of Geothermal Fluids from the Yangbaja in Geothermal Field, Tibet. *Acta Petrologica Sinica*, 17(3): 497–503 (in Chinese with English Abstract)
- Zhou, R. L., 1987. The Activity of Deep Underground Water in the Northern Part of the North China Plain and Its Effect on the Geothermal Field. *Bulletin of the 562 Comprehensive Geological Brigade Chinese Academy of Geological Sciences*, (6): 17–35 (in Chinese with English Abstract)
- Zhou, X., Chen, M. Y., Zhao, W. M., et al., 2001. Modeling of a Deep-Seated Geothermal System near Tianjin, China. *Ground Water*, 39(3): 443–448
- Zhou, X., Fang, B., Shen, Y., et al., 2004. Hydrogeochemistry and Origin of Thermal Groundwater in Bedrock Aquifers in Tianjin, China. *Journal of China University of Geosciences*, 15(1): 110–114
- Zhou, X., Fang, B., Zhou, H., et al., 2009. Isotopes of Deuterium and Oxygen-18 in Thermal Groundwater in China. *Environmental Geology*, 57(8): 1807–1814