



Determination and characterization of vitamin B₁₂ in the muscles and head innards of edible shrimp

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Received: 10 September 2019 / Accepted: 9 December 2019 / Published online: 8 January 2020
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Abstract

We determined the vitamin B₁₂ content in both the muscles and head innards of various shrimp species [*Argis lar* (Owen, 1839); Toge-zako shrimp, *Argis toyamaensis* (Yokoya, 1933); *Pandalopsis japonica*, Balss, 1914; *Pandalus eous* Makarov, 1935] using a microbiological assay based on *Lactobacillus delbrueckii* subsp. *lactis* ATCC7830. Approximately 2–4 µg vitamin B₁₂/100 g wet weight—a considerable amount—was detected in shrimp muscles. The shrimp head innards contained significantly higher levels of vitamin B₁₂ (~ 12–33 µg/100 g wet weight). Commercially available shrimp-innard products contained ~ 30 µg vitamin B₁₂/100 g wet weight. We purified vitamin B₁₂ compounds from the extracts of shrimp muscles and head innards using an immunoaffinity column. The muscle extract contained only one corrinoid compound, which was identified as vitamin B₁₂ using liquid chromatography–electrospray ionization/tandem mass spectrometry, whereas the shrimp head innards contained three corrinoid compounds, which included large amounts of vitamin B₁₂ and two smaller amounts of vitamin B₁₂-*d*-monocarboxylic acid and tentatively identified vitamin B₁₂ dicarboxylic acids.

Keywords *Argis lar* · *Argis toyamaensis* · *Pandalopsis japonica* · *Pandalus eous* · Vitamin B₁₂

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Introduction

Vitamin B₁₂ (B₁₂), also known as cobalamin, is a water-soluble vitamin that is one of a group of compounds containing a corrin ring. The lower ligand of B₁₂ is attached to the cobalt-coordinated corrin ring through the nucleotide loop containing 5,6-dimethylbenzimidazole as a base (Watanabe and Bito 2016). B₁₂ is synthesized by specific archaea and bacteria but not by animals or plants (Watanabe et al. 2014). The synthesized B₁₂ is accumulated mainly in higher predatory animals through the natural food chain; therefore, animal-derived foods contain considerable amounts of B₁₂ (Watanabe and Bito 2018a, b). In particular, fish and shellfish are reported to be important nutritional sources of B₁₂ for humans (Bourre and Paquette 2008; Scheers et al. 2014). Thaumarchaeota and cyanobacteria have recently been identified as the major authentic B₁₂ and pseudovitamin B₁₂ (pseudo-B₁₂) producers, respectively, in the oceans (Heal et al. 2017). Pseudo-B₁₂ is inactive in humans because it carries adenine in place of 5,6-dimethylbenzimidazole in the nucleotide moiety of the molecule (Watanabe et al. 1999).

To maintain an adequate supply of B₁₂ for the general population, more detailed information on biologically active B₁₂ compounds in seafoods is warranted. Therefore, we have purified and identified corrinoid compounds from various seafood species (Bito et al. 2018). Although most of the meats from the fish and shellfish that we tested contained B₁₂ but not pseudo-B₁₂ (Bito et al. 2018), pseudo-B₁₂ was detected in the edible portion of some sea snails (Teng et al. 2015).

Shrimp is one of the most consumed seafoods worldwide, and both aquaculture-produced and wild-caught shrimp are used to supply seafood products for human consumption (Butcherine et al. 2019). Shrimp is a good source of several types of nutrients, such as amino acids, polyunsaturated fats, and minerals (Kaymaci and Altun 2016); however, the detailed characteristics of B₁₂ have not been elucidated in this edible seafood. Although microbiological assays have determined that there is ~2 µg B₁₂/100 g wet weight in the raw muscle of edible shrimp (Ministry of Education, Culture, Sports, Science and Technology 2010), information on the B₁₂ content in the other edible portions, such as the shrimp head innards, are lacking; shrimp head innards are used to make shrimp sauces and other products in Asian countries. Moreover, it is unclear whether both shrimp muscles and head innards contain pseudo-B₁₂.

In the present study, we determined the B₁₂ contents of the shrimp head innards and muscles in four species of shrimp and purified and characterized corrinoid compounds from both of these edible portions using liquid chromatography–electrospray ionization/tandem mass spectrometry (LC/ESI–MS/MS). To the best of our knowledge, this is the first study to demonstrate that shrimp head innards contain substantial amounts of B₁₂ plus two unidentified B₁₂ compounds, which might be inactive for humans.

Materials and methods

Materials

The cyanocobalamin that was used as B₁₂ in this study was purchased from Sigma-Aldrich (St. Louis, MO). B₁₂-*b*-, *-d*-, and *-e*-monocarboxylic acids were prepared and their concentrations were determined on the basis of $\epsilon_{361} = 28.06 \times 10^3 \text{ M}^{-1} \text{ cm}^{-1}$ as described previously (Watanabe et al. 1992). The revised designation of these B₁₂ acid compounds were used according to Anton et al. (1980). *Lactobacillus delbrueckii* subsp. *lactis* ATCC 7830 was purchased from ATCC (Manassas, VA). A B₁₂ assay medium based on *L. delbrueckii* subsp. *lactis* ATCC 7830 was purchased from Nissui (Tokyo). Four types of fresh shrimp (Alaskan pink shrimp, *Pandalus eous* Makarov, 1935; Morotoge shrimp, *Pandalopsis japonica* Balss, 1914;

Kuro shrimp, *Argis lar* [Owen, 1839], and Toge-zako shrimp, *Argis toyamaensis* [Yokoya, 1933]) were purchased from local markets in Tottori, Japan. Products made from shrimp head innards were purchased from various seafood markets in Japan.

Extraction and assay of B₁₂ in shrimp muscles and head innards

B₁₂ was assayed using a microbiological method with *L. delbrueckii* subsp. *lactis* ATCC 7830, which has been adopted by the *Standard Tables of Food Composition in Japan* (Watanabe and Bito 2018a, b). After removing the shells from the fresh samples, the muscles and head innards were separated. Each muscle (2.0 g) or innard sample (1.0 g) was homogenized using a mortar and pestle and then added to 40 ml distilled water, 10 ml of 0.57 M acetic acid buffer (pH 4.5), and 0.4 ml of 0.05% (w/v) potassium cyanide (KCN). Total B₁₂ was extracted by boiling the solution for 30 min in a draft chamber. After cooling, 0.6 ml of 10% (w/v) metaphosphoric acid was added to the B₁₂ extract and the vessel was filled to 100 ml with distilled water. The prepared extract was filtrated through a 150-mm-diameter Whatman filter paper (GE Healthcare UK, UK). The filtrate was then divided into two portions of 25 ml each, the pH of one portion adjusted to 6.0, and the vessel filled to 50 ml with distilled water to use as extract A for the total B₁₂ assays. The pH of the remaining portion was adjusted to 11 and the portion autoclaved (MC-23; ALP, Tokyo) at 121 °C for 30 min. After cooling, the pH of the solution was readjusted to 6.0, and the vessel was filled to 50 ml with distilled water to use as extract B to determine the alkali-resistant factor.

A 2.5-ml assay mixture was created containing 0.01 ml of extract A, B, or standard B₁₂ solution (0, 0.5, 1.0, 2.0, 3.0, 4.0, or 5.0 µg/l B₁₂) and 1.25 ml B₁₂ basal medium for the assay (Nissui) prepared according to the manufacturer's protocol, and 1.24 ml distilled water. This mixture was placed in polypropylene tubes (13 × 100 mm; Bio-Rad Laboratories, Hercules, CA), vigorously shaken to mix it, and autoclaved at 121 °C for 5 min. After *Lactobacillus delbrueckii* subsp. *lactis* ATCC 7830 were precultured in a B₁₂ inoculum broth (Nissui) for 18 h, the bacterial cells were washed several times and diluted with sterile saline. The diluted bacteria solution was aseptically added to each assay mixture and allowed to stand for 16–18 h at 37 °C. To estimate bacterial growth, the turbidity of each assay mixture was measured at 600 nm using the ultraviolet–visible (UV-Vis) spectrophotometer (UV-2550; Shimadzu, Kyoto, Japan).

Lactobacillus delbrueckii subsp. *lactis* ATCC 7830 requires B₁₂ as an essential nutrient, but exhibits nucleotide and deoxyribonucleotide activity, which is known to be an alkali-resistant factor, as well as B₁₂. The correct B₁₂ values were calculated by subtracting the values for the

alkali-resistant factor from the total B₁₂ values. The recovery rate of B₁₂ from the extraction was calculated to be 105% after adding a known amount of authentic B₁₂ to a sample.

Analysis of B₁₂ compounds purified from the muscles or head innards of shrimp using LC/ESI-MS/MS

After ~30 g muscle or ~2 g head innards from Alaskan pink shrimp and Kuro shrimp, respectively, was homogenized using a mortar and pestle, the B₁₂ compounds were extracted by boiling at pH 4.5 in the presence of KCN, as described above. The B₁₂ compounds were partially purified from the extract using Sep-Pak Vac 20 cc (5 g) C18 cartridges (Waters, Milford, MA) that were washed with 20 ml 75% (v/v) ethanol and then equilibrated with 20 ml Milli-Q water (Merck Millipore, Burlington, MA). The extract was filtered through a 150-mm-diameter Whatman filter paper (GE Healthcare) and loaded onto the C18 cartridge, which was washed with 25 ml Milli-Q water (Merck Millipore). The B₁₂ compounds were eluted with 10 ml of 75% (v/v) ethanol. The eluate was evaporated to dryness under reduced pressure using the Integrated SpeedVac System ISS110 centrifugal concentrator (Savant Instruments, Holbrook, NY). The residue was dissolved in 1 ml Milli-Q water (Merck Millipore) and loaded into an EASI-EXTRACT vitamin B₁₂ immunoaffinity column (P80; R-Biopharm, Darmstadt, Germany), and the B₁₂ compounds were purified according to the manufacturer's instructions. The purified B₁₂ compounds were dissolved in Milli-Q water (Merck Millipore) and filtered through a Millex-LH membrane filter (Merck Millipore). Aliquots (5 µl) of the filtrate were analyzed using an ACQUITY UPLC H-Class Xevo G2-SQT (Waters). Each purified sample was injected into a 3-µm 2.1 × 100-mm InertSustain column (GL Science, Tokyo) equilibrated with 85% solvent A [0.1% (v/v) acetic acid] and 15% solvent B (methanol) at 40 °C. The B₁₂ compounds were eluted using a linear gradient of methanol (15% solvent B for 0–5 min, 15–90% solvent B for 5–11 min, and 90–15% solvent B for 11–15 min) at a flow rate 0.2 ml/min. The identification of authentic B₁₂ (*m/z* 678.2914 representing [M+2H]²⁺; retention time 9.55 min) was confirmed by comparing the observed molecular ions and their retention times.

To determine the relative content (percent) of B₁₂, B₁₂-*d*-monocarboxylic acid, and tentatively identified B₁₂-dicarboxylic acids in the head innards of both shrimp, the absorbance was measured at 361 nm. The relative content of each peak area against total peak area (i.e., the sum of the peak area of B₁₂ and B₁₂-acid compounds) was calculated. The recovery rate of B₁₂ from the extraction and purification was calculated to be 102% after adding a known amount of authentic B₁₂ to a sample.

High-performance liquid chromatography analysis of B₁₂-*b*-, *-d*-, and *-e*-monocarboxylic acids and unidentified B₁₂ compounds found in the head innards of Alaskan pink shrimp *P. eos*

B₁₂ compounds were purified from the head innard extract from the Alaskan pink shrimp using the B₁₂ immunoaffinity column as described above. The purified compounds were dissolved in 80 µl Milli-Q water (Merck Millipore), filtered through a Millex-LH membrane (Merck Millipore), and loaded into the Shimadzu HPLC system (SPD-10AV UV-Vis detector, SCL-10A VP system controller, DGU-20A₃ degasser, LC-10Ai liquid chromatograph, and CTO-20A column oven). A 35-µl aliquot of the purified compounds was loaded into a reversed-phase high-performance liquid chromatography column (ϕ 4.6 × 150 mm; Wakosil-II 5C18RS; FUJIFILM Wako Pure Chemical, Osaka, Japan), isocratically eluted with 20% (v/v) methanol solution containing 1% (v/v) acetic acid at 40 °C, and monitored by measuring the absorbance at 361 nm. The flow rate was 1 ml/min. Authentic B₁₂ and B₁₂-*b*-, *-d*-, and *-e*-monocarboxylic acids (each 1 µg) were analyzed under the same conditions.

Results

The B₁₂ contents of the muscles and head innards of four species of shrimp were determined using a microbiological assay based on *L. delbrueckii* subsp. *lactis* ATCC 7830 (Table 1). The shrimp muscles contained ~2.4–4.3 µg B₁₂/100 g wet weight, the values of which were identical to those described in the *Standard Tables of Food Composition in Japan* (Ministry of Education, Culture, Sports, Science and Technology 2010) but approximately eight times greater in the head innards than in the muscles. Shrimp-innard products A and B also contained ~28.5–30.1 µg B₁₂/100 g wet weight. The muscles, head innards, and whole bodies (except the shells) of the shrimp tested contained ~0.2, 0.3, and 0.5 µg B₁₂/fresh total body weight (g), respectively. These results suggest that shrimp is a source of B₁₂ for humans.

To evaluate whether the muscles and head innards of Alaskan pink shrimp contain B₁₂ or an inactive corrinoid, such as pseudo-B₁₂, corrinoids were purified using an immunoaffinity column and analyzed using LC/ESI-MS/MS. Authentic B₁₂ eluted as one ion peak with a retention time of 9.6 min (Fig. 1a). The mass spectrum of authentic B₁₂ indicated that a doubly charged ion with an *m/z* of 678.2894 [M+2H]²⁺ was prominent (Fig. 1b). The exact mass calculated from its formula of C₆₃H₈₈CoN₁₄O₁₄P was 1354.5674 g/mol, and the isotope-distribution data showed that B₁₂ was the major doubly charged ion under LC/ESI-MS/MS conditions. The MS/MS spectrum of authentic

Table 1 Vitamin B₁₂ content in the muscles and head innards of raw shrimp and products from shrimp head innards

	Vitamin B ₁₂ content				
	(µg/100 g wet weight)		(µg/total fresh body)		
	Muscles	Innards	Muscles	Innards	Whole body
Alaskan pink shrimp <i>Pandalus eous</i>	2.49 ± 0.24	28.24 ± 5.74	0.21 ± 0.04	0.36 ± 0.02	0.57 ± 0.06
Morotoge shrimp <i>Pandalopsis japonica</i>	2.61 ± 0.29	21.44 ± 4.52	0.18 ± 0.05	0.22 ± 0.06	0.40 ± 0.11
Kuro shrimp <i>Argis lar</i>	4.12 ± 0.53	33.21 ± 5.70	0.29 ± 0.07	0.41 ± 0.09	0.69 ± 0.16
Togezako shrimp <i>Argis toyamaensis</i>	4.33 ± 0.72	12.48 ± 4.75	0.30 ± 0.03	0.14 ± 0.07	0.44 ± 0.10
Shrimp head innards product A	–	30.06 ± 4.34			
Shrimp head innards product B	–	28.48 ± 6.66			

Data are presented as mean ± SD of three independent experiments ($n=3$)

B₁₂ indicated that its singly charged ions at m/z 359.1007 and m/z 997.4782 were attributable to the nucleotide and corrin ring moieties, respectively (Fig. 1c). The compounds purified from the muscles of Alaskan pink shrimp eluted as one ion peak with a retention time of 9.5 min (Fig. 2a), and its mass spectrum primarily comprised a doubly charged ion with m/z 678.2894 [M+2H]²⁺ (Fig. 2b). The MS/MS spectrum of the purified compounds with singly charged ions at m/z 359.1007 and m/z 997.4782 were identical to those of authentic B₁₂ (Fig. 2c).

The corrinoid compounds purified from the head innards of Alaskan pink shrimp eluted as three ion peaks with retention times of 9.4, 9.8, and 10.0 min (Fig. 3a), respectively. The MS spectrum of the ion peak with a retention time of 9.4 min showed that the doubly charged B₁₂ ion was formed at m/z 678.2894 [M+2H]²⁺ (Fig. 3b). The MS/MS spectrum of the latter peak was identical to that of authentic B₁₂ (Fig. 3e). The MS spectra of the ion peaks with a retention time of 9.8 min (compound A) and 10.0 min (compound B) showed that a major doubly charged ion was formed at m/z 678.7849 [M+2H]²⁺ and 679.2751 [M+2H]²⁺, respectively (Fig. 3c, d). The MS/MS spectra of compounds A and B with a singly charged ion at m/z 359.1007 were identical to that of authentic B₁₂; however, the MS/MS spectra of compounds A and B with singly charged ions at m/z 998.4603 and m/z 999.4493, respectively, were not identical to that of authentic B₁₂ at m/z 997.4782 (Fig. 3f, g). These results suggest that the head innards of Alaskan pink shrimp contained B₁₂ and two B₁₂ compounds, A and B, which were B₁₂ compounds with different corrin ring moieties that have one and two additional hydrogen atoms, respectively (Figs. 2, 3). Similar results were obtained for the other shrimp tested.

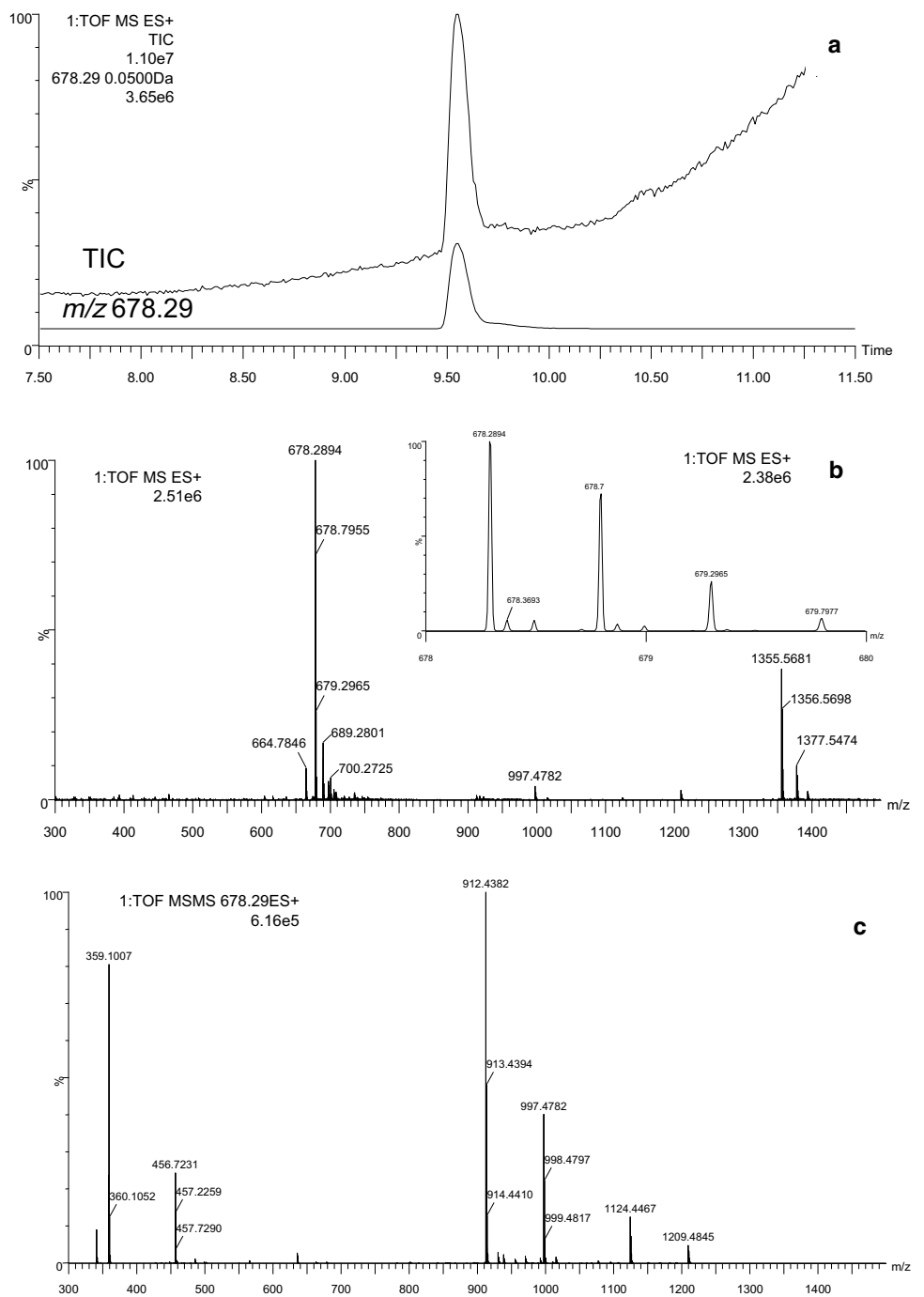
The chemical structure of B₁₂ and the molecular masses of the unidentified B₁₂ compounds were used to predict their identity. Compound A was hypothesized to be B₁₂-monocarboxylic acid while compound B was predicted to be B₁₂-dicarboxylic acid (Fig. 4). Immunoaffinity chromatography was used to purify corrinoids from

head innard extract obtained from Alaskan pink shrimp. This was followed by analysis using a C18-reversed phase HPLC column. HPLC studies showed that B₁₂, unidentified compound A and unidentified B compound eluted as three separate peaks with retention times of 9.8, 16.2, and 18.3 min, respectively (Fig. 5a). HPLC analysis of known forms of B₁₂ under the same conditions showed that authentic B₁₂-*b*-, *-d*-, and *-e*-monocarboxylic acids eluted as single peaks with retention times of 9.7, 13.5, 16.2, and 20.0 min, respectively (Fig. 5b–e). The retention time of B₁₂-*d*-monocarboxylic acid was identical to that of compound A (retention time of 16.2 min). When B₁₂-*d*-monocarboxylic acid was analyzed using LC/ESI–MS/MS, it eluted as an ion peak with a retention time of 9.6 min (Fig. 6a). This was similar to be the retention time observed for compound A (retention time of 9.8 min) (Fig. 3a). Thus, MS and MS/MS spectra of B₁₂-*d*-monocarboxylic acid (Fig. 6b, c) were identical to that of compound A (Fig. 3c, f).

However, the identity of compound B could not be established as the preparation methods for B₁₂-*bd*-, *-be*-, and *-de*-dicarboxylic acids have not been reported. At present, compound B can be hypothetically identified as one of these B₁₂-dicarboxylic acids.

To determine the relative B₁₂ content (%), B₁₂-*d*-monocarboxylic acid and B₁₂-dicarboxylic acid in the shrimp head innards of Alaskan pink shrimp and Kuro shrimp were selected because of their high B₁₂ contents compared with those of the other shrimp tested. The relative B₁₂ content (%) of each peak area against the total peak area (i.e., the sum of the peak area of B₁₂ and unidentified compounds) was calculated by measuring the absorbance at 361 nm (Table 2). Approximately 65% of the B₁₂ compounds identified in the head innards of Alaskan pink shrimp and 25% in the case of Kuro shrimp were derived from B₁₂-monocarboxylic and -dicarboxylic acids. The total B₁₂ content in shrimp head innard products A and B contained ~22% these B₁₂ acid compounds.

Fig. 1 Liquid chromatography–electrospray ionization/tandem mass spectrometry (LC/ESI–MS/MS) chromatograms of authentic B₁₂. **a** Total ion chromatogram (TIC) and mass chromatogram of authentic B₁₂ (*m/z* 678.29). **b** Mass spectrum of authentic B₁₂ (insert magnified spectrum from *m/z* 678 to 680). **c** MS/MS spectrum of the peak of authentic B₁₂ at *m/z* 678.2894



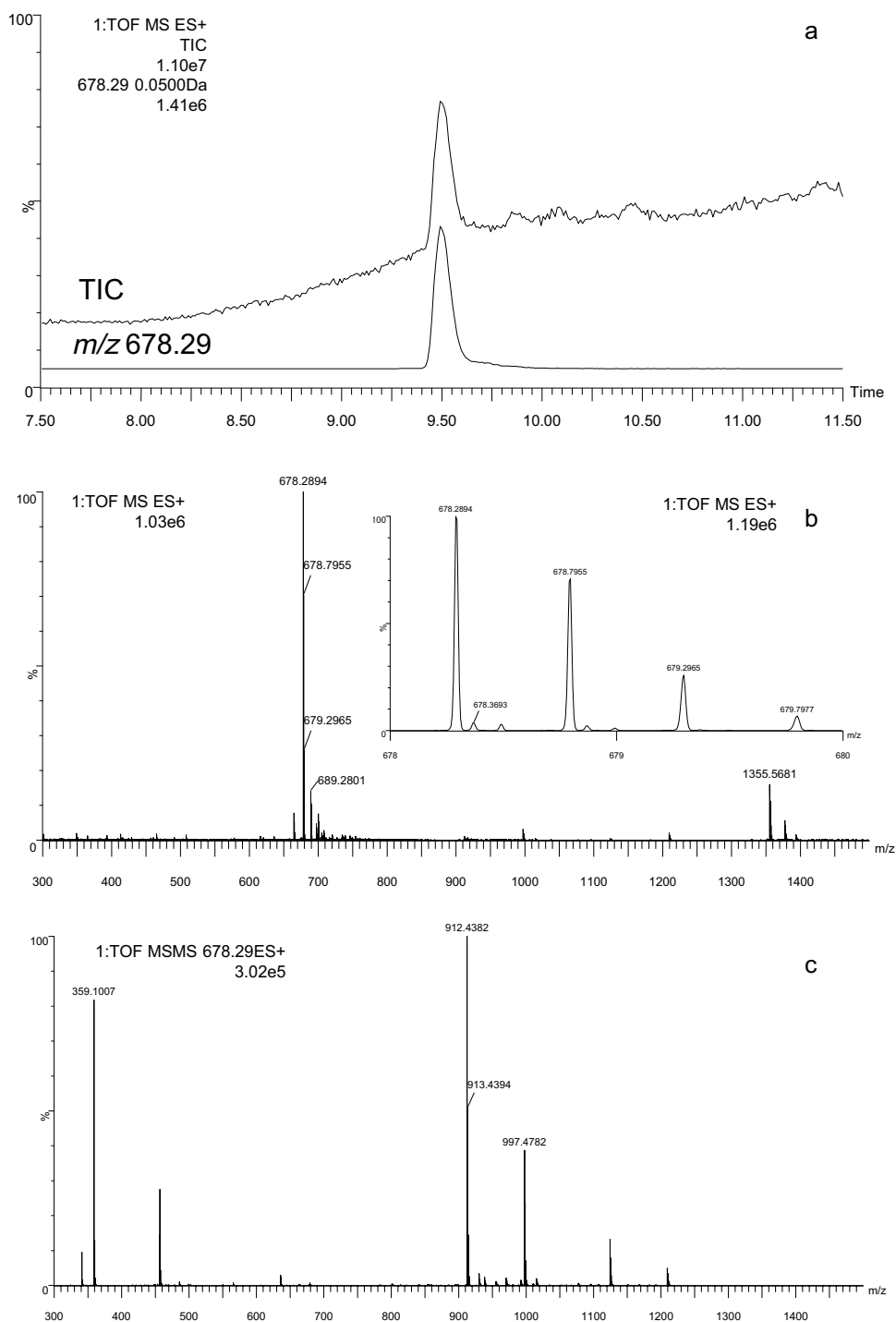
The ability of B₁₂-*d*-monocarboxylic acid to show B₁₂ activity was studied using the standard microbiological assay with *L. delbrueckii* subsp. *lactis* ATCC 7830. In particular, the effect of various doses of B₁₂ and B₁₂-*d*-monocarboxylic acid (0.01–0.05 pmol) on the growth of the lactic acid bacterium was studied (Fig. 7). At a concentration of 0.05 pmol, the activity shown by B₁₂-*d*-monocarboxylic acid was only 6% of that displayed by the authentic B₁₂. These results suggest that

B₁₂-*d*-monocarboxylic acid hardly functions as B₁₂ in this bacterium.

Discussion

Although shrimp is one of the most consumed seafoods worldwide and is a good source of several nutrients (Kaymaci and Altun 2016), the detailed characteristics of

Fig. 2 LC/ESI-MS/MS chromatograms of corrinoid compounds purified from muscles of *Pandalus eous* Makarov, 1935. **a** TIC and mass chromatogram of the corrinoid compounds (m/z 678.29). **b** Mass spectrum of the corrinoid compounds at 9.49 min (insert magnified spectrum from m/z 678 to 680). **c** MS/MS spectrum of the peak of the corrinoid compounds at m/z 678.2894. For abbreviations, see Fig. 1



B_{12} have not been elucidated in this seafood. Thus, we determined the B_{12} content in both the muscles and head innards of various shrimp species using a microbiological assay based on *L. delbrueckii* subsp. *lactis* ATCC7830. As shown in Table 1, approximately 2–4 $\mu\text{g } B_{12}/100 \text{ g}$ wet weight was detected in shrimp muscles. The shrimp head innards contained significantly higher levels of B_{12}

(~12–33 $\mu\text{g}/100 \text{ g}$ wet weight). Consumption of approximately eight to 14 shrimp muscles, six to seven shrimp head innards, and three to six whole shrimp bodies could supply the recommended dietary allowance of B_{12} for adults (2.4 $\mu\text{g}/\text{days}$) (Institute of Medicine 1998), which suggests that shrimp is a suitable source of B_{12} for humans.

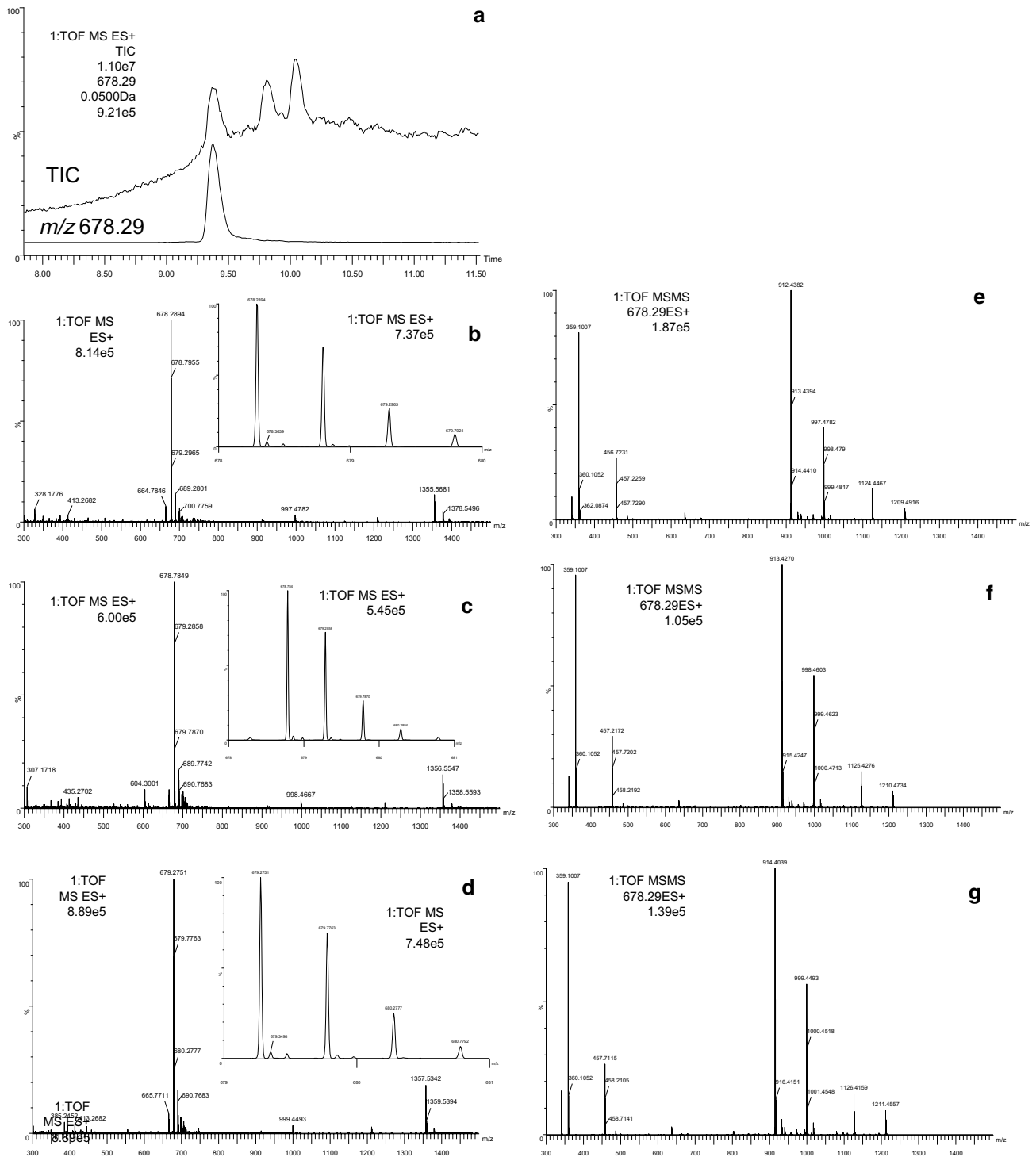


Fig. 3 LC/ESI-MS/MS chromatograms of corrinoid compounds purified from the head innards of *P. eous* Makarov, 1935. **a** TIC and mass chromatogram of the corrinoid compounds (m/z 678.29). **b–d** Mass spectra of the corrinoid compounds with retention times of 9.49 min (the magnified spectrum from m/z 678 to 680 is shown as an insert in **b**), 9.81 min (the magnified spectrum from m/z 678 to 681 is

shown as an insert in **c**), and 10.04 min (the magnified spectrum from m/z 679 to 681 is shown as an insert in **d**), respectively. **e–g** MS/MS spectra of the peaks of the corrinoid compounds with retention times of 9.49 min (m/z 678.2894), 9.81 min (m/z 678.7849), and 10.04 min (m/z 679.2751), respectively. For abbreviations, see Fig. 1

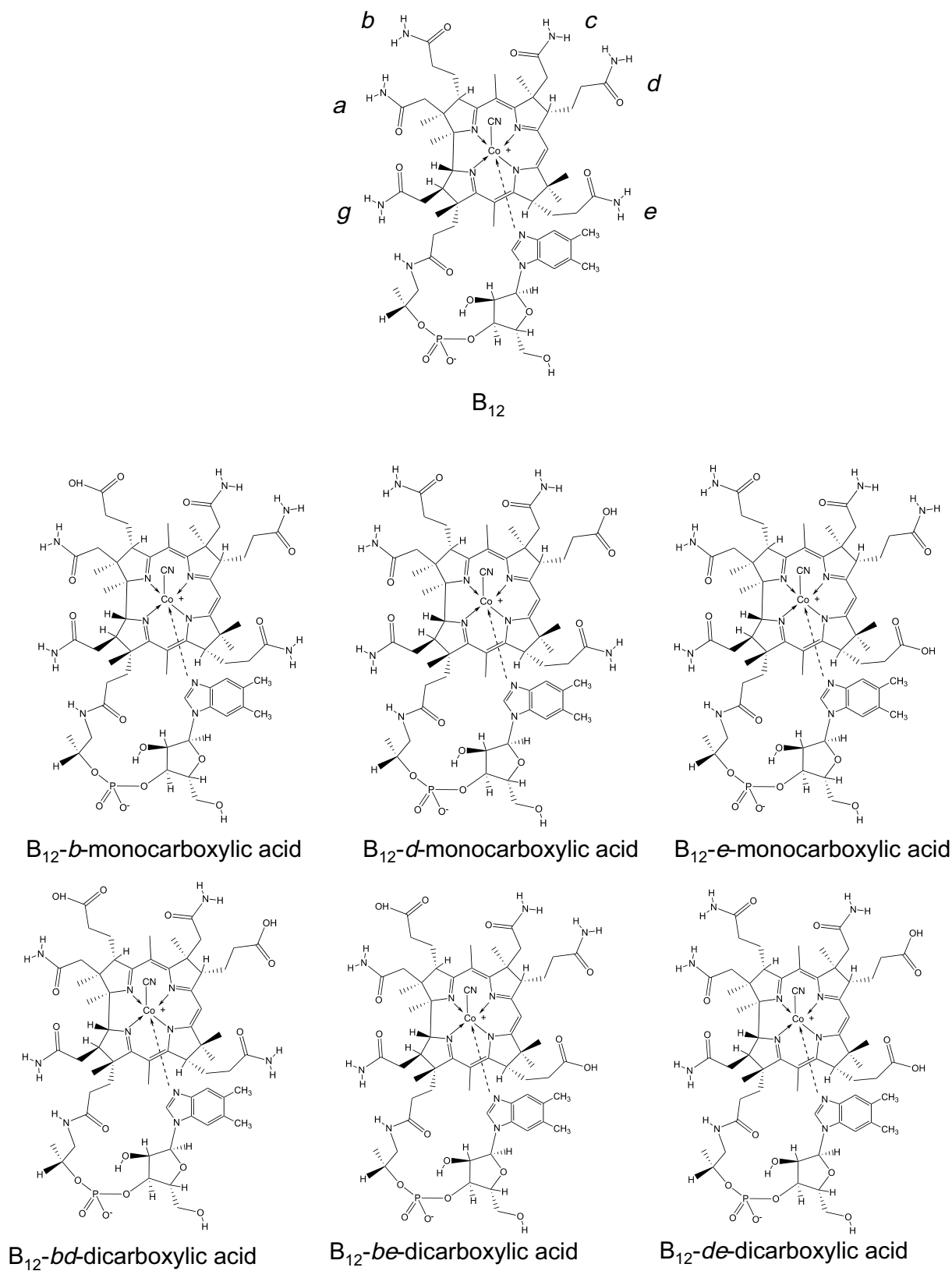


Fig. 4 Chemical structures of unidentified B₁₂ compounds A and B predicted using LC/ESI-MS/MS analysis. Compounds A and B were identified as B₁₂-monocarboxylic acid and a B₁₂-dicarboxylic acid, respectively. For abbreviations, see Fig. 1

Fig. 5 High-performance liquid chromatography chromatograms of **a** B₁₂ compounds present in head innard extract of *P. eos* Makarov, 1935, **b** B₁₂, **c** B₁₂-*b*-monocarboxylic acid, **d** B₁₂-*d*-monocarboxylic acid, and **e** B₁₂-*e*-monocarboxylic acid

We purified B₁₂ compounds from the extracts of shrimp muscles and head innards using an immunoaffinity column, and the purified B₁₂ compounds were identified using LC/ESI-MS/MS. The shrimp muscle extracts were characterized by the presence of only B₁₂ while the extracts obtained from shrimp head innards contained three corrinoid compounds. These head innard extracts contained large amounts of B₁₂ and smaller amounts of two unidentified B₁₂ compounds denoted as compound A and compound B. The HPLC and LC/ESI-MS/MS analyses identified compound A as B₁₂-*d*-monocarboxylic acid (Figs. 5, 6). The identity of compound B could not be established completely and it was hypothetically identified as one of the B₁₂-dicarboxylic acids because of the presence of additional masses of two hydrogen atoms for B₁₂ compounds having different corrin ring moieties. The content of the B₁₂-mono and -dicarboxylic acids varied from 17 to 65% of the total B₁₂ content in shrimp head innards and their products (Table 2), which suggested that these B₁₂ acid compound levels were dependent on the species sampled.

An intrinsic factor (IF) plays an important role in the gastrointestinal absorption of B₁₂ in humans. This protein is known to have high specific binding for authentic B₁₂; however, it lacks any significant affinity towards B₁₂-*b*-, *-d*-, and *-e*-monocarboxylic acids or B₁₂-*bde*-tricarboxylic acid (Kolhouse and Allen 1977). The low affinity of this IF protein for B₁₂ mono- and tricarboxylic acids explains the poor absorption of these compounds following their oral administration in rabbits (Kolhouse and Allen 1977). Saïdo et al. (1993) reported that oral and intravenous administration of B₁₂-*b*-, *-d*-, and *-e*-monocarboxylic acids did not show any improvement in the B₁₂ levels of B₁₂-deficient rats. These studies strongly suggest that both B₁₂-*d*-monocarboxylic acid and B₁₂-dicarboxylic acids found in shrimp head innards might not be absorbed by the human intestine. However, shrimp head innards can be still considered a source of B₁₂ for humans owing to their contents of B₁₂-mono and -dicarboxylic acids (17–65% of the total B₁₂ content).

Anton et al. (1980) reported the formation of B₁₂-mono- and -dicarboxylic acids on mild acid hydrolysis of B₁₂. These B₁₂ acids are derived from the propionamide side chains *b*, *d*, and *e*, which are more susceptible to hydrolysis than the acetamide side chains *a*, *c*, and *g* (Bernhauer et al. 1966) (Fig. 4). The presence of these B₁₂ acid compounds in shrimp head innards is quite unusual. It is possible that trace elements are absorbed from sea water and accumulate in shrimp head innards (Kaymaci and Altun 2016), and might react with B₁₂ to produce

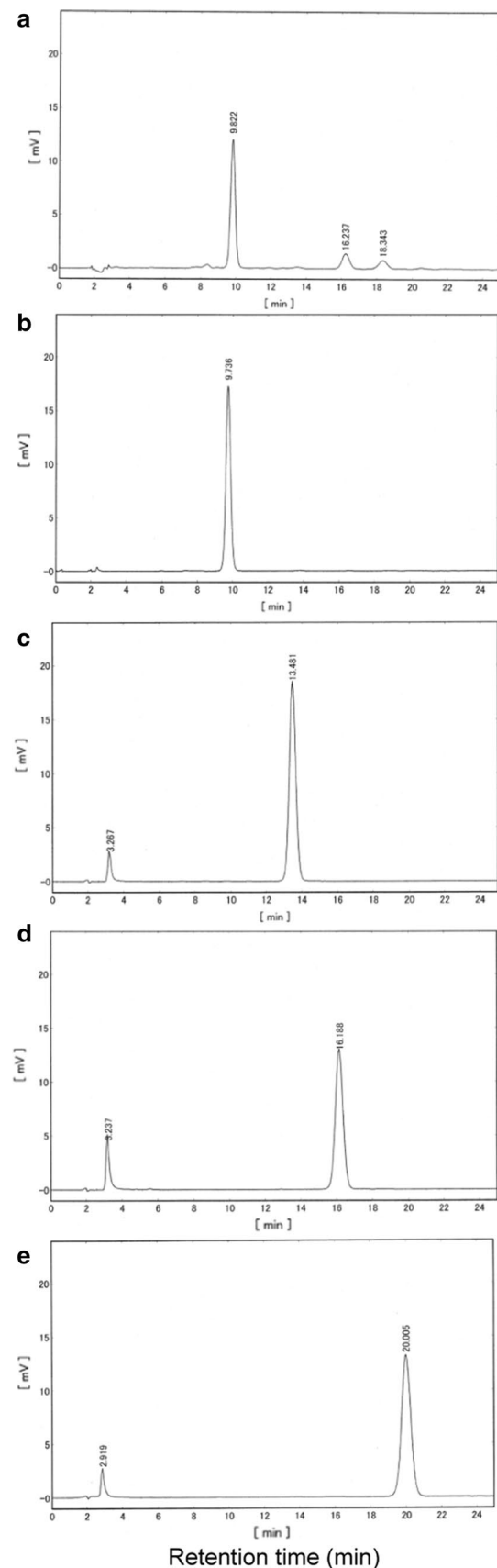


Fig. 6 LC/ESI-MS/MS chromatograms of B_{12} -*d*-monocarboxylic acid. **a** TIC and mass chromatogram of B_{12} -*d*-monocarboxylic acid (m/z 678.78), **b** mass spectrum of the corrinoid compounds with retention times of 9.6 min (insert magnified spectrum from m/z 678 to 681), **c** MS/MS spectrum showing the peak of the corrinoid compound with a retention time of 9.6 min (m/z 678.78). For abbreviations, see Fig. 1

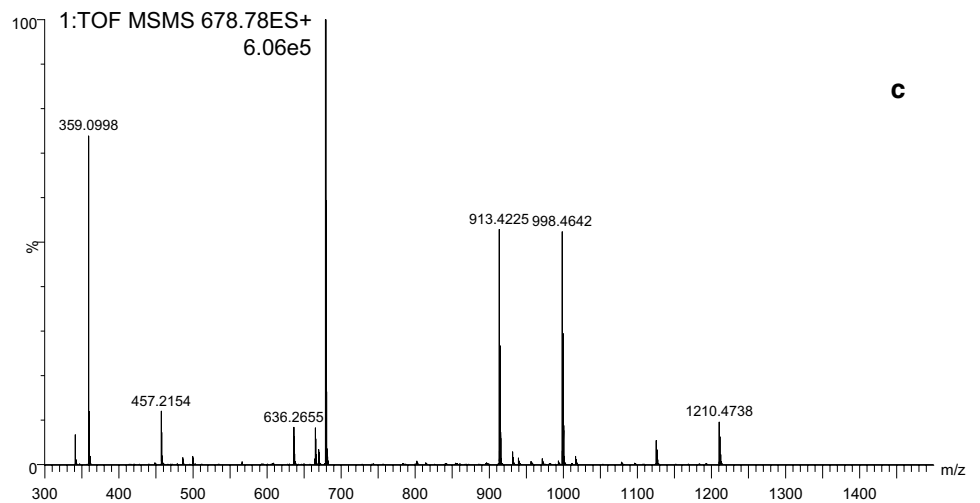
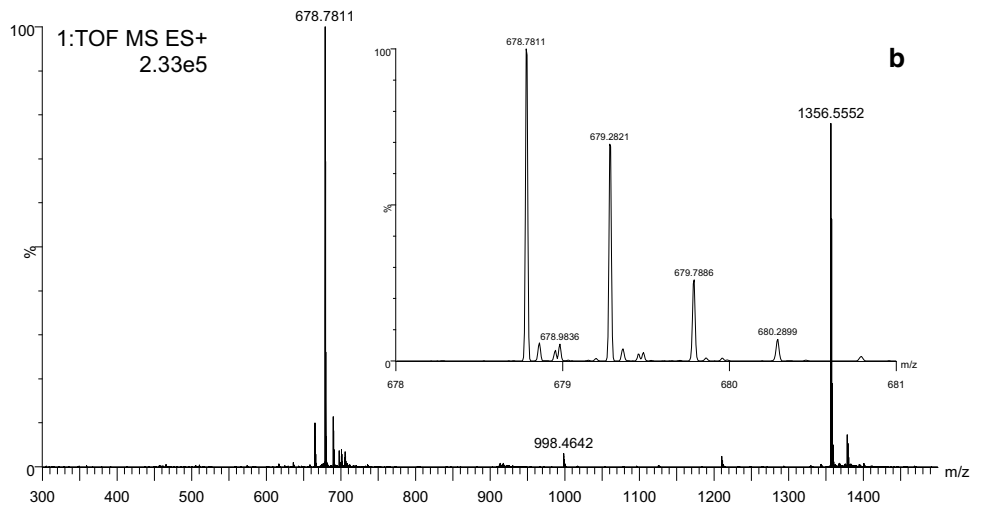
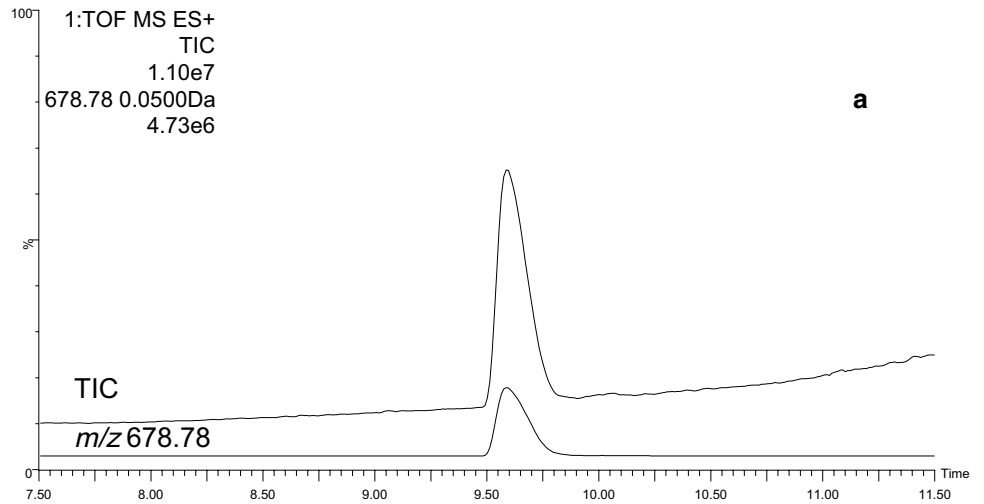


Table 2 Relative contents (%) of B₁₂ and compound A (B₁₂-*d*-monocarboxylic acid) and compound B (tentative B₁₂-dicarboxylic acids) in selected shrimp head innards and their products

	Relative content (%)		
	B ₁₂	Compound A B ₁₂ - <i>d</i> -monocarboxylic acid	Compound B tentative B ₁₂ -dicarboxylic acids
Alaskan pink shrimp <i>P. eous</i>	34.6 ± 0.1	24.6 ± 0.2	40.8 ± 0.3
Kuro shrimp <i>A. lar</i>	73.9 ± 0.3	11.0 ± 0.2	15.0 ± 0.1
Shrimp head innards product A	82.3 ± 0.6	11.5 ± 0.2	6.1 ± 0.4
Shrimp head innards product B	78.2 ± 1.3	13.9 ± 0.2	7.9 ± 1.1

Data are presented as mean ± SEM of triplicates

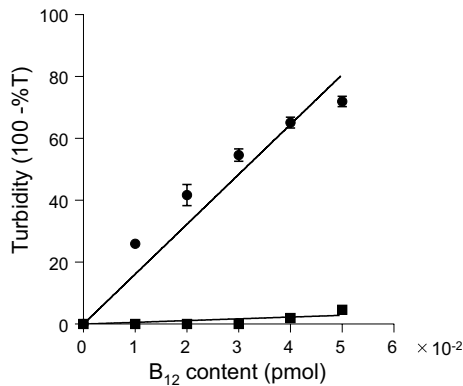


Fig. 7 Effect of authentic B₁₂ (black circle) and B₁₂-*d*-monocarboxylic acid (black square) on growth of *L. delbrueckii* subsp. *lactis* ATCC 7830 expressed as turbidity as a function of dose. The experiment was performed in triplicate and results are given as mean ± SEM

these acid compounds. However, no detailed mechanism has been reported so far to explain this hypothesis.

Euglena gracilis Z, a B₁₂-dependent alga, has been widely utilized in microbiological assays for B₁₂. Supplementation of growth culture medium with B₁₂-*d*-monocarboxylic acid increased the growth of *Euglena* to similar levels to those observed in authentic B₁₂-supplemented culture medium (Watanabe et al. 1992). *L. delbrueckii* subsp. *lactis* ATCC 7830 is a bacterium commonly used to determine the B₁₂ content of food. As shown in Fig. 7, B₁₂-*d*-monocarboxylic acid did not show any significant effect on the growth of this bacterium. There are no reports on the effect of B₁₂-dicarboxylic acids on the growth of *L. delbrueckii* subsp. *lactis* ATCC 7830. Since the structure of B₁₂ changed more in the case of B₁₂-dicarboxylic acid compared to B₁₂-*d*-monocarboxylic acid, it is possible that B₁₂-dicarboxylic acids are inactive in this bacterium. The possible inactivity of B₁₂-*d*-monocarboxylic and B₁₂-dicarboxylic acids in *L. delbrueckii* subsp. *lactis* ATCC 7830 suggests that these compounds might not affect the B₁₂ content of shrimp head innards.

Acknowledgements This work was supported by the Japan Society for the Promotion of Science KAKENHI grant number 16K07736 (to F. W.).

Author contributions N. O. and N. H. performed most of the experiments. Y. U. and S. T. analyzed the B₁₂ compounds using LC/ESI-MS/MS and interpreted the results. N. O., T. B., and F. W. designed the experiments, interpreted the results, and wrote the manuscript. All authors reviewed and commented on the manuscript and approved the final version.

Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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