



An integrated typological, technological and provenance investigation of Late Hellenistic colour-coated pottery from Nea Paphos, Cyprus

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Abstract

The current paper contributes new evidence to the ongoing discussion of the location and organisation of production of Hellenistic colour-coated ware (CCW), through interdisciplinary investigation of morphological style, technology and provenance of relevant pottery from a Late Hellenistic deposit found in a well in the Agora of Nea Paphos on Cyprus. Through naked eye examination, four major macroscopic groups were identified based on vessel typology and fabrics. Subsequently, 53 representative samples were selected and analysed through a combination of refiring tests, chemical analysis, thin section petrography and scanning electron microscopy. The results indicated that the sheer majority of this pottery, despite some compositional and technological variability, reflects the use of similar raw materials compatible with the geology of Paphos. Therefore, during the Late Hellenistic period, most of the CCW consumed in Paphos were locally produced. The observed variation in the assemblage can be associated with several factors, such as natural diversity of raw materials, a number of co-existing pottery workshops, technological choices made by the potters and to a lesser extent post-depositional processes. This integrated research has a profound impact on studies of production and consumption of the Late Hellenistic pottery on Cyprus.

Keywords Cyprus · Nea Paphos · Colour-coated ware · Late Hellenistic period · Chemical analysis · Petrographic analysis

Background and research questions

The colour-coated ware (CCW) is a class of fine pottery commonly used around sites in the Eastern Mediterranean from the beginning of the Hellenistic period until the Early Roman times (late fourth c. BCE–first c. CE; Table 1). It was originally defined and named by John W. Hayes in his publication of

Hellenistic and Roman ceramics from the House of Dionysos in Nea Paphos (1991). Generally, CCW is characterised by the presence of a matt or semi-lustrous slip, whose colour can vary from red to black even within the same vessel. It is associated with tableware and includes a variety of forms, such as various types of bowls, plates and jugs, as well as unguentaria and lamps. It is widely accepted that the same or very similar shapes of CCW were produced at several places in the Eastern Mediterranean (e.g. Fenn and Römer-Strehl 2013: 377; Drougou 2014: 146). Based on archaeological evidence (Hayes 1991; Papuci-Władyka 1995) and chemical analyses (Picon and Blondé 2002), it has been suggested that Nea Paphos was one of the production places of CCW. The objective of this paper is to investigate further this suggestion addressing questions of provenance, technology and production organisation of the Late Hellenistic CCW pottery from Paphos,

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Table 1 Relative and absolute chronology of Nea Paphos with indication of the pottery found in the well deposit from the Agora and the CCW production

Period	Chronology	Important events	Pottery
Early Hellenistic	Late fourth–late third c. BCE	Late fourth–early third c. BCE: foundation of Nea Paphos	Late fourth c. BCE: the beginning of the CCW production
Middle Hellenistic	Late third–late second c. BCE	Late third–early second c. BCE: Nea Paphos becomes the capital of the island	
Late Hellenistic	Late second–late first c. BCE	58 BCE: the Romans annexed Cyprus	Late second–third quarter of the first c. BCE: pottery from the well deposit at the Agora in Paphos
Roman	Late first c. BCE–293 c. CE	17 or 15 BCE: earthquakes destroyed Nea Paphos	First c. CE: the end of the CCW production

through the combined typological, compositional and technological studies of a ceramic assemblage from a closed deposit.¹ The latter, related to a sealed context from a well at the Agora in Nea Paphos, includes pottery dated from the late second to the beginning of the third quarter of the first c. BCE. It is particularly valuable as it provides securely and closely dated evidence on CCW production and consumption for a key period in the history of the site.

The site and the pottery

Nea Paphos, located in southwestern Cyprus, at the crossroads of the main sea routes between East and West, was an important commercial and administrative hub for many centuries. It is believed that the city was founded either by Nicocles, the last king of Palaipaphos, in the late fourth c. BCE (Nicolaou 1966; Młynarczyk 1990: 66–105; Młynarczyk 1996: 193) or by Ptolemy I in the early third c. BCE (Bekker-Nielsen 2000; Balandier 2011: 376). The new city was divided into the residential (Maloutena, Ktisto), administrative (Fanari Hill) and commercial (Agora, harbour) districts (Fig. 1a). At the end of the third or in the early second c. BCE, it became the headquarters of the Governor of the island (Maier and Karageorghis 1984: 230; Młynarczyk 1990: 121–122). Not only was

¹ This study is part of a PhD research project focused on the characterisation and contextualisation of CCW in Paphos. In the framework of this project, a few thousands of CCW potsherds and whole vessels from two excavations in Paphos, the Agora and the Maloutena residential area, were studied macroscopically. As a result, 16 macroscopic groups (MG 1–MG 16) were defined, and 164 samples were selected for further chemical and petrographic analysis, see: Marzec 2017. In a previous paper (Marzec et al. 2018), MG 1 and MG 2 have been discussed. They date to the Early Hellenistic period (late fourth–ca. mid-second century BCE) and are represented by 35 samples, the majority of which was unearthed in the Maloutena residential area (33 out of 35 samples). The current paper presents the macroscopic groups: MG 9, MG 10, MG 11 and MG 12, dated to the Late Hellenistic period (late second–late first c. BCE) and represented by 53 samples (see Table 2) from one closed deposit in the Agora, which did not include any examples representing MG 1 and MG 2. Moreover, MGs discussed in the current paper are neither chemically nor petrographically related to MG 1 and MG 2.

the city an administrative centre of the island but also a location of strategic importance in which the Ptolemaic fleet was stationed to be supplied as its geographic location allowed easy access to natural resources. The dense forests and the mineral ores located in the nearby mountains were particularly attractive to the Lagids and influenced the socio-economic and cultural development of Nea Paphos in the Hellenistic period (Maier and Karageorghis 1984: 230–231; Green 1990: 30, 125, 194; Młynarczyk 1990: 129–130).

In Nea Paphos, CCW comprises the most common class of fine pottery recovered at the excavations in all districts through the entire Hellenistic period (Table 1). It appears in a number of different forms and fabrics, possibly indicating its varied origin. The stylistic study has so far shown that the CCW frequency as well as the diversity in the shape of the vessels and macroscopic appearance of fabrics increased significantly in the Late Hellenistic period, which dates from the late second to the late first c. BCE.² In 58 BCE, the Romans annexed Cyprus, but the significant changes in the pottery assemblage reflecting the new political situation are seen in contexts dated after an earthquake, which destroyed the city in 17 or 15 BCE. In those contexts, in terms of fine ware pottery, CCW occurs rarely and only includes semi-coarse echinus bowls and plates with a rolled rim, while eastern sigillata is the predominant pottery class.

The archaeological context

The ancient Agora in Nea Paphos comprised a square enclosed by porticoes from the east, south and west sides, which were built in the late second or the early first c. BCE. The excavations at the Agora have been carried

² Marzec E, Kajzer M (forthcoming) Hellenistic Table Ware Pottery (from the late 4th to the 1st century BC). In: Papuci-Władyka E (ed.) PAPHOS AGORA 1. The results of the Jagiellonian University Interdisciplinary Research in Nea Paphos (2011–2015). Kraków

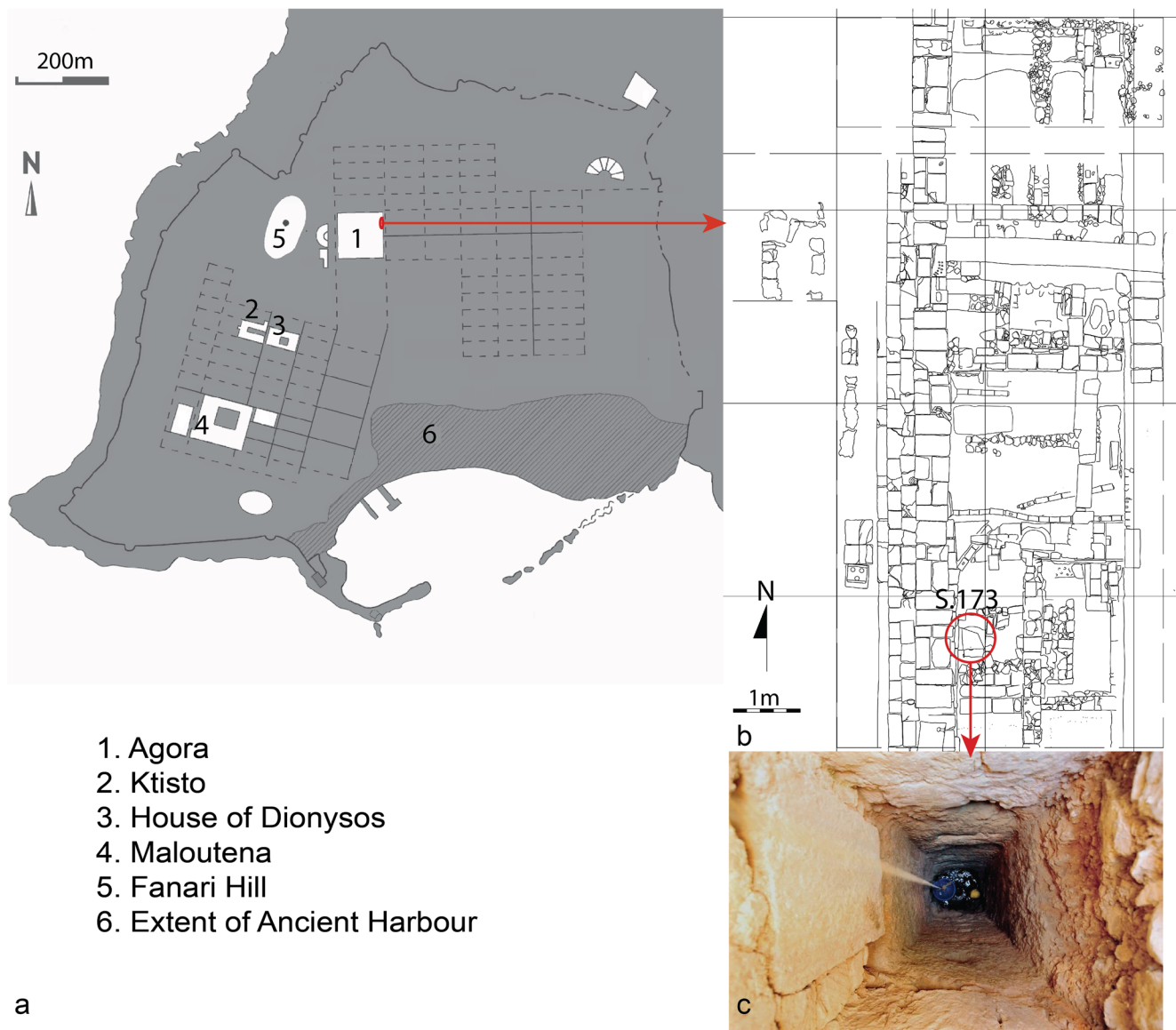


Fig. 1 **a** Plan of Nea Paphos, based on J. Młynarczyk's research (1990) with modifications by the Paphos Agora Project. **b** Trench II, illustrated by K. Rosińska-Balik, 2014, © Paphos Agora Project. **c** The well, structure 173, photographed by R. Słaboński

out since 2011 in four trenches (I–IV).³ One of them (Trench II) uncovered the area of the eastern portico, comprising a part of the crepidoma, the stylobate and rooms within the vicinity. The construction of the rooms dates to the times of Augustus, after the earthquake that hit the city in 17 or 15 BCE, but the presence of earlier layers in some parts of the trench suggests that the portico was originally built in the Late Hellenistic period. The Late Hellenistic closed deposit discussed in this paper was found in the well (Structure 173) discovered in Room 13 (Fig. 1). The well was partially carved into the bedrock. Its upper east side adjoined the crepidoma and the other three were constructed with limestone blocks.

³ Miszk Ł. (forthcoming) Stratigraphy of the Agora. In: Papuci-Władyka E (ed.) PAPHOS AGORA 1. The results of the Jagiellonian University Interdisciplinary Research in Nea Paphos (2011–2015). Kraków

The well had a depth of 7 m and was filled with objects from top to bottom. Moreover, water was encountered at the depth of 5.95 m and excavation required pumping equipment. The deposit yielded numerous nearly complete vessels and thousands of pottery sherds. This assemblage comprises different categories of ceramics such as fine wares, plain wares, cooking wares and amphorae. The pottery found in the well deposit dates from the late second until the third quarter of the first c. BCE. However, the presence of Eastern Sigillata A form 23, Eastern Sigillata D forms P18A and P20, as well as a coin of Ptolemy of Cyprus (80–58 BCE) suggests that the deposition took place between the end of the first quarter and the beginning of the third quarter of the first c. BCE. There is no evidence of later contamination or disturbance. Other finds belonging to the deposit, such as metal objects, glass vessels,

terracotta figurines and lamps, have confirmed this date.⁴ Fine pottery of various wares and fabrics is the most abundant category of finds in this deposit. Apart from CCW, Eastern Sigillata A and Eastern Sigillata D, the assemblage also includes the so-called Red Slip and banded pottery.

Methodology

The first stage of the research combined the macroscopic study of fabric and shape. The fabric examination was based on the classification of the ceramic assemblage into groups according to the following parameters: frequency, size and colour of inclusions, frequency, size and shape of voids, colour of the fresh break, fired core types (according to Gauß and Kiriati 2011: 408, Fig. 30), hardness, feel of the surface, texture of fresh break and colour of slip. The classification and description system adopted in this study were proposed by Orton and Hughes, and it is presented in *Pottery in Archaeology* (2013: 275–285). Fabrics were examined by naked eye and with a $\times 10$ magnifying glass in natural light. Hardness was assessed through the use of fingernail and knife, while the feel of the surface was established through finger touch. The macroscopic examination of fabric was complemented by the typological analysis, which was undertaken exclusively on diagnostic sherds. It followed the functional classification proposed by S. Rotroff (1997) with some modifications.

The macroscopic analysis was carried out on nearly a thousand of individual CCW vessels, recovered in the well deposit. Based on the two variables (fabric and vessel shape), the assemblage was classified into eight macroscopic groups (MG). The sheer majority of CCW was associated with four of them, MG 9, MG 10, MG 11 and MG 12, which are assumed to represent the potential local production (Hayes 1991; Papuci-Władyka 1995). Fifty-three samples representing their variability were selected for laboratory analysis to test the macroscopic classification consistency and provide further evidence on the provenance and manufacturing technology of the four most common groups (Tables 2 and 3). Other four MGs, represented only by single vessels or few sherds and classified as imports, were not sampled as the research was focused on the local production.

All selected samples were subjected to refiring tests at the same temperature and under the same conditions (1000 °C, 1 h soaking time, oxidising conditions) in order to ‘remove’ colour variation introduced by firing rather than raw materials composition and to provide a more solid ground for the macroscopic grouping (Whitbread 1995: 390–391; Kiriati et al.

2011: 69–70). The colours of fully oxidised clay pastes and slips were recorded using the Munsell Soil Color Charts (1990) and compared directly with the sherds before refiring. Then, all samples were analysed through wavelength-dispersive X-ray fluorescence (WD-XRF) spectroscopy in order to characterise their elemental composition. The ceramic samples were carefully cleaned to remove any surface contamination and decoration layers and subsequently ground to a fine powder. Fused glass beads were prepared, using 1 g of the previously dried and then ignited sample and 6 g of a mixture of lithium metaborate/lithium tetraborate with lithium bromide as non-wetting agent. The analysis was carried out on a WD-XRF spectrometer, BRUKER S8-TIGER with a Rh excitation source, using a custom calibration based on 43 certified reference materials (Georgakopoulou et al. 2017). Twenty-six major and trace elements were determined: Na, Mg, Al, Si, P, K, Ca, Ti, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Rb, Sr, Y, Zr, Ba, La, Ce, Nd, Pb, and Th. Subsequently, the WD-XRF data were subjected to statistical multivariate analyses using the R software. The statistical treatment followed the approach proposed by J. Buxeda i Garrigós (1999). Based on the results of the refiring tests, macroscopic and elemental analysis, 15 samples representing the observed variation were selected for petrographic analysis. Covered standard thin sections were prepared and examined using a Zeiss Axioskop 40 polarising microscope at magnifications ranging between $\times 25$ and $\times 500$. The description followed an adjusted version of the description system proposed by Whitbread (1995).

Based on the combination of macroscopic and laboratory analyses, the relation between the MGs was defined and a final grouping was established. In order to provide more detailed information concerning surface treatment and firing regimes, ten samples, analysed already through thin section petrography, and representing variability observed in relation to surface treatment and firing, were chosen for scanning electron microscopy (SEM) (Tite and Maniatis 1975; Maniatis and Tite 1981). The fresh breaks of the selected samples were carbon-coated and examined with a FEI, Quanta Inspect D8334 SEM in secondary electron and backscattering mode. Semi-quantitative analysis for major elements was carried out using an attached energy-dispersive X-ray fluorescence spectrometer (EDS). Additionally, geological sampling was carried out in the area of Paphos in order to explore the availability and the characteristics of the locally available potential raw materials for pottery manufacture. Four geological samples were collected: alluvial sediments from the Ezousa river and Agriokalami stream, bentonite clay from Armou and clay rich sediment from excavations at the ancient Agora in Nea Paphos (Fig. 2). Four briquettes were formed from each sample. One briquette was left unfired and the others were fired for 1 h (soaking time) at 700 °C, 900 °C and 1100 °C, respectively, under

⁴ Papuci-Władyka E (ed.) (forthcoming) PAPHOS AGORA 1. The results of the Jagiellonian University Interdisciplinary Research in Nea Paphos (2011–2015). Kraków

Table 2 Selected CCW specimens representing four MGs and type of analysis employed in the study

MG	Sample	Form	Refiring 1000 °C	WD- XRF	Thin section petrography	SEM
9 / n = 18	A.14	Jug with cylindrical neck and ovoid body	√	√	√	
	A.18	Plate with rolled rim	√	√	√	√
	A.19	Plate with rolled rim	√	√		
	A.30	Plate with rolled rim	√	√		
	A.36	Jug with pear-shaped body	√	√		
	A.37	Lagynoids	√	√		
	A.40	Jug with funnel-shaped mouth and globular body	√	√		
	A.47	Jar with two handles	√	√		
	A.50	Echinus bowl	√	√		
	A.51	Echinus bowl	√	√		
	A.53	Plate with rolled rim	√	√		
	A.54	Echinus bowl	√	√		
	A.56	Plate with rolled rim	√	√		
	A.61	Echinus bowl	√	√		
	A.63	Large plate with rolled rim	√	√	√	√
	A.65	Large echinus bowl	√	√		
	A.64	Plate with rolled rim	√	√	√	
	A.69	Echinus bowl	√	√	√	√
10 / n = 10	A.12	In-turned rim bowl	√	√	√	
	A.15	Large echinus bowl	√	√		
	A.16	Bi-conical jug	√	√		
	A.17	Echinus bowl	√	√		
	A.21	Plate with rolled rim	√	√	√	√
	A.41	Jug with cylindrical neck and ovoid body	√	√		
	A.42	Large echinus bowl	√	√	√	√
	A.43	Echinus bowl	√	√		
	A.48	Fishplate with overhanging rim	√	√		
	A.49	Echinus bowl	√	√		
11 / n = 17	A.03	Echinus bowl	√	√	√	
	A.05	Platter	√	√		
	A.08	Plate with rolled rim	√	√		
	A.22	Plate with rolled rim	√	√		
	A.23	Plate with rolled rim	√	√		
	A.24	Platter	√	√	√	√
	A.38	Large plate with rolled rim	√	√	√	√
	A.39	Jug with trefoil-mouth	√	√		
	A.44	Jug with cylindrical neck and ovoid body	√	√	√	√
	A.45	Plate with rolled rim	√	√		
	A.57	Echinus bowl	√	√		
	A.58	Hemispherical cup	√	√		
	A.59	Echinus bowl	√	√		
	A.60	Echinus bowl	√	√		
	A.62	Plate with rolled rim	√	√		
	A.66	Large echinus bowl	√	√	√	
	A.68	Echinus bowl	√	√		

Table 2 (continued)

MG	Sample	Form	Refiring 1000 °C	WD- XRF	Thin section petrography	SEM
12	A.07	Lekane	√	√	√	√
/	A.09	Large plate with rolled rim	√	√		
<i>n</i> = 8	A.10	Large plate with rolled rim	√	√		
	A.13	Plate with rolled rim	√	√		
	A.29	Platter	√	√		
	A.46	Echinus bowl	√	√	√	√
	A.52	Echinus bowl	√	√		
	A.67	Plate with rolled rim	√	√		

oxidising conditions using a muffle furnace. All fired briquettes were examined through thin section petrography, additionally those that were fired at 700 °C and 900 °C were analysed by WD-XRF (methodology as in Kiriati et al. 2011).

Geology

The area of Paphos is drained by three major rivers, the Ezousa, Xeropotamos and Dhiarizos, which have their sources in the Troodos Mountains (Fig. 2). The rocks of the Paphos area are associated with three geological formations. The Troodos Ophiolite complex consisting of ultramafic rocks is exposed in the highlands, north and east from Paphos. The Mamonia complex crops out in the vicinity of the city and includes the oldest sedimentary and volcanic rocks in this part of Cyprus, as well as ultramafic and mafic plutonic rocks, and metamorphic rocks. The Circum Troodos Sedimentary Succession is built with the autochthonous sediments and is divided into several formations. The oldest sediments in this zone are of volcanoclastic origin, rich in Campanian bentonitic clays. The rest of the Paphos region, including the city of Paphos, lie directly on Quaternary sediments, overlying the Circum Troodos Sedimentary Succession. They are composed of biocalcarenes, sandstones, sandy marls, conglomerates, silts, clays and gravels (Malpas and Xenophontos 1987; GSD 1995; Silantyev et al. 2005a, b).

Results and discussion

As above-mentioned, the majority of the CCW vessels found in the Late Hellenistic well deposit were classified into four MGs (Fig. 3). These MGs seem to be similar to each other in terms of vessel shape and style, but they were separated due to fabric variation related to the size and frequency of inclusions,

colour, hardness, and the feel of the surface. MG 9 is characterised by a fine and hard fabric with a smooth feel of the surface. In contrast, the soft fabric and the powdery feel of the surface are two distinguishing features of MG 10. It seems that vessels assigned to this MG are related to the so-called *Pink Powdery Ware* known from the excavations on Geronisos Island, which is located 18 km north of Paphos (Młynarczyk 2005, 2010). MG 11 has a hard, pink or brown fabric with a smooth feel of the surface. The vessels are fine to semi-coarse with frequent white inclusions and lime popping. The fabric of vessels assigned to MG 12 is also hard with frequent white inclusions and lime popping, but it is characterised by a rough feel of the surface and lighter colours of the fresh break. Table 3 provides the detailed characterisation of the MGs and presents their distinctive features.

The shape analysis indicates the presence of 23 forms representing the discussed MGs in the macroscopically studied deposit. The entire assemblage is represented by the following functional categories: drinking cups, other vessels for wine service, vessels for food service and vessels for other purposes (Table 4). It is important to note that echinus bowls (drinking cups) and plates with a rolled rim (vessels for food service) are the most common in the deposit and appear in all MGs. Among other vessels, various types of jugs are common, mainly in relation to MG 9. Other shapes that are not very common and have not been included in the sample selection are lids, rounded-mouth juglets, olpai, strainer-jugs, kantharoid skyphoi and fusiform unguentaria.

All selected 53 samples were subjected to refiring tests. After refiring, the colour of the fresh break and slip, within each MG and the whole assemblage, became more homogeneous. However, differences were still visible. The refired chips representing MG 9 and MG 11 show very similar range of body colours in the fresh break: pink (7.5YR 7/4, 8/4), reddish yellow (5YR 6/6) and yellowish red (5YR 5/6). MG 11 contains a few samples with a very pale brown (10YR 7/4) colour of body. This colour was also observed in the fresh breaks of all samples classified to MG 12 and of a few samples of MG

Table 3 The characterisation of the four MGs discussed here. The definition of terms used in the table: frequency: rare—less than 2%, few—2–15%, common—15–30%; frequent—30–50%, size: fine—smaller than 0.25 mm, medium—0.25–0.5 mm, coarse—0.5–1 mm; hardness: soft—can be easily scratched with a fingernail, hard—can be scratched with a knife; feel of surface: smooth—no irregularities can be felt; powdery—leaves powder on the finger; rough—with irregularities; texture of fresh break: smooth—flat, without ripple marks, hackly—with more inclusions; large and angular irregularities. The descriptions follow the system proposed by Orton and Hughes (2013: 277–282); fired core types according to Gauß and Kiriatzi (2011: Fig. 30); colours were recorded using the Munsell Soil Color Charts (1990)

	MG 9	MG 10	MG 11	MG 12
Frequency, size and colour of inclusions visible at the fresh break	Rare, fine to medium - Common fine to medium white - Common fine black - Common fine to medium red - Few fine sparkling	Rare to few, fine to medium - Common fine to medium white - Few to common fine to medium black - Few to common fine to medium red - Rare fine sparkling	Rare to few, fine to coarse - Frequent fine to coarse white - Common to frequent fine black - Rare to few fine red - Rare fine sparkling	Few, fine to coarse - Frequent fine to medium white - Common to frequent fine to medium black - Common fine to coarse red
Frequency, size and shape of voids visible at the fresh break	Rare - Frequent to few fine to medium rounded - Frequent coarse thin, rather angular and elongated shape	Rare - Frequent fine to coarse rounded - Frequent fine to coarse thin, rather angular and elongated shape	Rare - Frequent fine to coarse rounded - Frequent fine to coarse thin, rather angular and elongated shape	Rare to few - Frequent fine to coarse rounded - Frequent fine to coarse thin, rather angular and elongated shape
Colour of the fresh break	Pink (5YR 7/4, 7.5YR 7/4, 8/4), light reddish brown (5YR 6/4), very pale brown (10YR 7/3), reddish yellow (5YR 6/6)	White (10YR 8/2), light grey (10YR 7/2), very pale brown (10YR 7/3), pink (7.5YR 7/4), light reddish brown (5YR 6/4), light yellowish brown (10YR 6/4)	Pink (7.5YR 8/4), light reddish brown (5YR 6/4), reddish yellow (7.5YR 7/6), very pale brown (10YR 7/3, 7/4), light grey (10YR 7/1)	White (10YR 8/2), Very pale brown (8/3, 7/3), pink (7.5 YR 7/4)
Fired core	(1) Fully oxidised (2) Not fully oxidised (8) Not fully reduced	(1) Fully oxidised (6) Changing conditions (9, 10) Not fully reduced	(1) Fully oxidised (2) Not fully oxidised (9) Not fully reduced	(1) Fully oxidised (4) Not fully oxidised (10) Not fully reduced
Hardness	Hard	Soft	Hard	Hard
Feel of surface	Smooth	Powdery	Smooth	Rough
Texture of fresh break	Smooth	Smooth	Smooth	Hackly
Colour of slip	Red (2.5YR 4/8, 5/6, 5/8), yellowish red (5YR 5/6), brown (7.5YR 5/4); zones across the surface: red (2.5YR 4/2), reddish black (2.5YR 4/6), reddish brown (5YR 4/3), black (2.5YR 2.5/0)	Dark grey (2.5YR 4/0), weak red (2.5YR 4/2), strong brown (7.5YR 4/6), reddish brown (5YR 4/4), dark reddish grey (5YR 4/2), yellowish red (5YR 5/6, 6/6)	Red (2.5YR 5/6), dark grey (5YR 4/1), reddish brown (5YR 5/4), very dark grey (7.5YR 3/0), reddish yellow (7.5YR 6/6), dark greyish brown (10YR 4/2)	Brown (7.5YR 5/4), dark grey (7.5YR 4/0), yellowish red (5YR 5/6), pale yellow (5Y 7/3), yellowish brown (10YR 5/4)

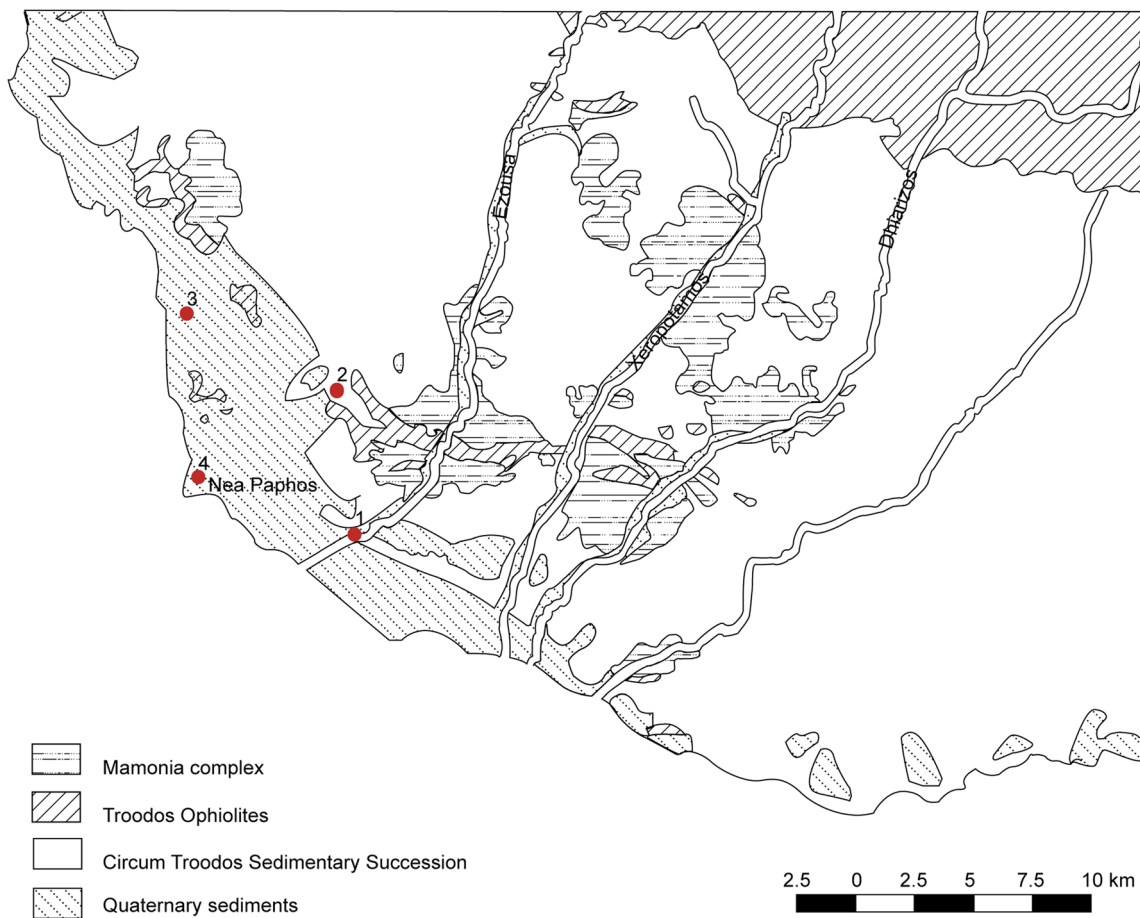


Fig. 2 The geological map of Paphos (based on GSD 1995). The locations of geological samples are indicated with red dots and numbers: 1 sediment from the Ezousa river, 2 bentonite clay from Armou, 3 sediment

from Agriokalami stream, and 4 clay rich sediment from excavations at the ancient Agora in Nea Paphos

10. The colour of the slips is homogenous across all MGs; it is represented by various hues of red (2.5YR 4/6, 4/8, 5/6) or reddish brown (2.5YR 4/4, 5/4; 5YR 5/3, 5/4), while the black or very dark grey zones seen on the surface of some vessels in most cases disappeared upon refiring.

The dataset with the elemental compositions of the assemblage consisting of 53 samples (Table 5) was analysed statistically. The compositional variation matrix for 26 elements was calculated in order to assess the total variation and the variability of each element (Buxeda i Garrigós 1999). The value of the total

Fig. 3 Photos of echinus bowls representing the four main MGs identified in the current assemblage: **a** sample A.61 (MG 9), **b** sample A.43 (MG 10), **c** sample M.60 (MG 11), and **d** sample A.46 (MG 12)



Table 4 The typological classification of the vessels representing the four MGs and the petrographic subgroups (PG 1A and PG 1B) discussed in this study

















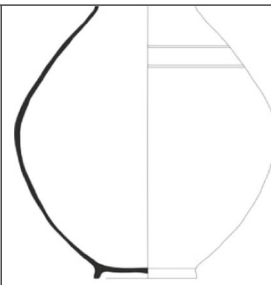
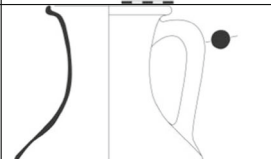
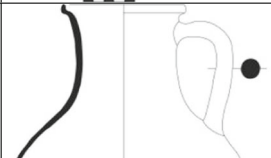
Sample	Form	Illustration	MG	PG
Drinking cups				
A.03	echinus bowl		11	1A
A.17	echinus bowl		10	
A.43	echinus bowl		10	
A.46	echinus bowl		12	1B
A.49	echinus bowl		10	
A.50	echinus bowl		9	
A.51	echinus bowl		9	
A.52	echinus bowl		12	
A.54	echinus bowl		9	
A.57	echinus bowl		11	
A.59	echinus bowl		11	
A.60	echinus bowl		11	
A.61	echinus bowl		9	
A.68	echinus bowl		11	
A.69	echinus bowl		9	1A
A.58	hemispherical cup		11	
Other Vessels for wine service				
A.14	jug with cylindrical neck and ovoid body		9	1B
A.41	jug with cylindrical neck and ovoid body		10	
A.44	jug with cylindrical neck and ovoid body		11	1B

Table 4 (continued)

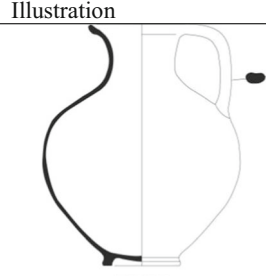
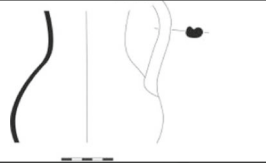
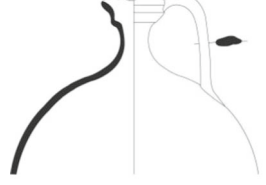
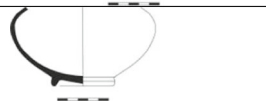

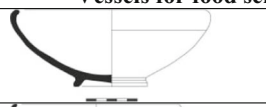








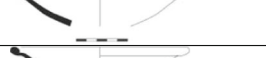

Sample	Form	Illustration	MG	PG
A.40	jug with funnel-shaped mouth and globular body		9	
A.36	jug with pear-shaped body		9	
A.39	jug with trefoil-mouth		11	
A.37	lagynoids		9	
A.16	bi-conical jug		10	
Vessels for food service				
A.15	large echinus bowl		10	
A.42	large echinus bowl		10	1B
A.66	large echinus bowl		11	1A
A.65	large echinus bowl		9	
A.12	in-turned rim bowl		10	1A
A.08	plate with rolled rim		11	
A.13	plate with rolled rim		12	
A.18	plate with rolled rim		9	1B
A.19	plate with rolled rim		9	
A.21	plate with rolled rim		10	1A
A.22	plate with rolled rim		11	

Table 4 (continued)

Sample	Form	Illustration	MG	PG
A.23	plate with rolled rim		11	
A.30	plate with rolled rim		9	
A.45	plate with rolled rim		11	
A.53	plate with rolled rim		9	
A.56	plate with rolled rim		9	
A.62	plate with rolled rim		11	
A.64	plate with rolled rim		9	1B
A.67	plate with rolled rim		12	
A.10	large plate with rolled rim		12	
A.09	large plate with sloping rim		12	
A.38	large plate with sloping rim		11	1A
A.63	large plate with sloping rim		9	1A
A.48	plate with overhanging rim		10	
A.05	platter		11	
A.29	platter		12	
A.24	platter		11	1A
A.07	lekane		12	1A
Vessels for other purposes				
A.47	jar with two handles		9	

Table 5 Chemical compositions of 53 samples discussed in this paper; oxides, LOI (loss on ignition) and sum are expressed in wt% and elements in ppm

Sample	Na ₂ O	MgO	Al ₂ O ₃	SiO ₂	P	K ₂ O	CaO	TiO ₂	V	Cr	Mn	Fe ₂ O ₃	Co	Ni	Cu	Zn	Rb	Sr	Y	Zr	Ba	La	Ce	Nd	Pb	Th	LOI	Sum
A03	1.60	4.27	13.83	49.56	536	1.28	13.73	0.70	173	94	947	8.41	27	65	153	104	42	556	20	77	296	22	29	12	53	3	6.34	100.05
A05	0.86	3.33	12.82	50.66	1079	1.69	16.07	0.68	105	96	1108	6.32	24	89	118	94	66	590	26	114	710	37	58	29	31	6	6.75	99.61
A07	1.80	4.75	13.32	50.48	773	1.60	14.67	0.77	130	204	899	8.59	26	111	112	112	46	814	20	102	204	8	25	12	33	5	3.58	99.92
A08	0.77	3.44	14.35	57.65	811	1.84	10.72	0.75	138	126	1080	6.95	26	100	87	95	73	495	27	127	439	34	64	25	23	8	3.22	100.08
A09	0.99	2.90	11.82	47.69	916	1.52	19.44	0.59	102	102	1265	6.05	24	81	118	90	52	804	23	105	451	21	36	19	25	7	8.39	99.82
A10	0.95	3.29	12.54	51.48	928	1.56	16.70	0.61	113	115	1385	6.77	25	98	122	79	53	632	21	104	374	21	35	28	38	5	5.62	99.94
A12	0.81	2.80	10.93	46.56	907	1.67	18.63	0.60	123	112	886	5.74	24	71	138	60	43	1283	23	100	717	26	47	13	234	10	11.33	99.54
A13	1.20	4.21	12.30	44.26	882	1.47	19.62	0.67	104	148	1037	7.06	24	86	228	104	50	816	23	96	306	23	33	20	63	7	8.55	99.73
A14	0.68	3.24	14.57	56.81	758	2.24	10.83	0.79	116	112	855	6.86	26	102	90	82	87	627	25	202	513	32	62	27	97	10	3.36	99.77
A15	0.84	2.79	10.45	47.64	1439	1.31	17.32	0.60	112	90	1263	5.63	23	80	96	93	45	689	23	98	468	22	46	25	26	8	12.32	99.35
A16	1.03	3.11	11.65	50.99	818	1.73	15.16	0.65	109	85	1201	6.36	23	67	119	68	51	999	23	108	525	26	42	15	370	8	8.76	99.90
A17	0.97	4.48	12.61	53.58	525	1.67	8.25	0.64	124	124	837	7.08	26	103	197	94	43	744	20	102	239	15	33	24	216	7	9.57	99.19
A18	0.61	2.86	12.97	51.17	810	1.68	12.91	0.62	108	130	945	6.28	25	84	108	103	57	810	24	104	613	29	45	24	23	7	10.37	99.88
A19	1.25	3.71	13.97	58.35	814	1.86	8.94	0.71	144	103	1160	7.30	28	86	158	115	62	533	26	120	341	28	54	28	91	7	3.50	99.97
A21	1.03	3.60	14.39	52.44	1056	1.96	12.90	0.83	148	130	954	7.62	26	86	111	107	64	584	25	116	299	27	50	24	30	7	4.65	99.81
A22	1.12	3.63	13.81	52.83	895	1.76	13.85	0.74	115	119	1161	7.30	28	91	124	107	66	559	26	119	495	24	51	26	32	7	4.63	100.06
A23	0.60	3.03	12.06	49.61	1117	1.80	16.58	0.62	119	128	925	5.74	21	91	109	112	66	841	22	111	722	37	60	24	28	5	9.34	99.83
A24	0.96	3.76	14.23	57.14	724	1.92	10.04	0.72	132	111	1237	7.20	27	76	110	91	69	604	24	114	443	30	50	25	111	7	3.65	100.02
A29	1.27	4.55	11.39	43.13	1082	1.40	21.36	0.65	115	178	916	6.76	20	195	108	101	46	817	21	88	346	9	33	12	38	6	8.97	99.88
A30	0.59	2.78	11.80	45.44	864	2.01	18.63	0.60	108	102	1211	5.55	24	111	123	128	66	1019	25	110	1246	38	60	23	50	7	12.16	100.10
A36	0.84	3.89	15.26	56.26	923	2.05	9.26	0.75	144	112	1274	7.66	29	95	146	136	75	575	24	117	443	26	52	36	51	12	3.08	99.48
A37	0.84	3.68	14.91	57.60	908	1.90	9.99	0.76	130	117	1114	7.38	29	92	111	105	73	431	25	125	417	25	55	26	28	9	2.39	99.82
A38	1.68	4.24	12.92	47.68	812	1.51	15.01	0.72	187	140	905	7.81	28	65	101	100	40	940	20	92	358	11	24	12	52	4	8.01	99.99
A39	0.68	3.60	12.85	46.99	971	1.76	18.30	0.63	138	111	990	6.34	26	98	116	107	67	597	23	102	662	26	51	29	24	8	8.01	99.57
A40	0.85	3.24	14.24	53.73	1329	1.88	9.82	0.72	139	103	943	7.07	25	85	147	116	62	495	23	119	371	26	59	22	33	10	7.66	99.62
A41	0.77	3.42	13.78	49.61	1300	2.04	16.82	0.71	83	126	964	6.46	26	120	97	106	75	764	25	126	1128	28	63	31	34	8	5.48	99.60
A42	0.83	3.14	12.91	53.29	901	1.95	14.07	0.69	119	105	1069	6.31	28	87	107	69	70	1019	24	119	976	32	57	22	107	7	6.26	99.93
A43	0.91	3.45	14.07	54.24	1077	1.99	13.40	0.74	148	116	1667	7.08	28	103	113	81	73	848	24	125	668	29	57	28	225	11	3.06	99.48
A44	0.61	2.67	11.24	48.45	1016	1.48	18.75	0.62	108	101	1031	5.51	23	98	86	101	60	635	26	106	643	31	49	22	25	10	10.39	100.13
A46	0.62	2.58	10.18	36.85	715	1.37	13.56	0.55	93	98	740	4.88	19	69	73	76	55	533	20	92	414	30	46	13	44	5	29.18	100.08
A47	0.95	3.75	14.77	54.86	3144	1.95	9.62	0.73	198	145	935	8.80	28	135	164	102	63	618	22	111	520	30	56	23	40	8	3.05	99.11
A48	1.09	3.61	12.31	51.49	762	1.86	14.33	0.67	88	113	1104	6.08	25	98	165	71	62	984	23	128	432	28	64	28	300	10	7.47	99.36
A49	0.62	2.97	12.72	49.05	970	2.06	17.62	0.65	118	120	889	5.83	22	91	101	76	74	1201	26	117	1230	39	66	27	166	8	8.01	100.06
A50	0.56	2.98	12.44	48.72	785	1.54	16.01	0.59	106	99	928	5.97	23	84	94	97	59	762	21	99	623	22	44	26	28	8	10.63	99.84
A51	0.75	2.78	11.51	49.65	766	1.41	15.18	0.64	104	90	1059	5.61	23	84	93	98	55	581	23	112	475	30	50	27	33	6	11.86	99.76
A52	1.33	4.38	9.42	37.41	663	0.94	22.68	0.53	124	153	785	6.00	20	74	111	89	23	832	18	63	147	7	17	13	14	7	16.65	99.64
A53	1.86	4.84	14.29	48.74	949	1.61	11.92	0.96	204	107	941	9.77	30	58	106	113	40	616	20	86	106	11	22	20	47	3	5.32	99.65
A54	0.79	3.20	13.89	54.12	617	1.76	9.63	0.70	129	95	924	6.85	24	82	281	111	59	487	23	114	390	24	53	29	75	9	8.46	99.74
A56	0.78	3.04	13.50	57.36	877	2.06	10.52	0.73	130	107	1172	6.31	26	74	131	83	76	840	26	134	569	29	53	30	141	5	5.05	99.80
A57	0.96	3.09	12.06	51.96	973	1.72	14.95	0.66	116	131	1336	5.74	34	75	83	63	53	1203	23	109	830	15	44	19	98	7	8.35	100.00
A58	1.15	3.61	14.43	58.88	1277	1.71	8.41	0.74	145	107	1384	7.55	28	94	113	113	59	415	24	116	350	26	38	24	27	8	2.20	99.11
A59	0.81	3.34	14.50	57.67	977	1.91	10.24	0.78	128	159	1160	7.05	28	102	96	98	74	406	25	133	379	29	58	30	26	8	3.01	99.70
A60	0.67	3.57	14.96	55.89	814	2.06	12.36	0.74	123	145	911	7.27	27	104	105	94	79	684	22	125	711	30	53	27	47	8	1.93	99.87

Table 5 (continued)

Sample	Na ₂ O	MgO	Al ₂ O ₃	SiO ₂	P	K ₂ O	CaO	TiO ₂	V	Cr	Mn	Fe ₂ O ₃	Co	Ni	Cu	Zn	Rb	Sr	Y	Zr	Ba	La	Ce	Nd	Pb	Th	LOI	Sum
A61	1.00	3.32	10.33	40.36	741	1.14	19.18	0.54	133	96	849	5.85	20	60	99	96	38	643	20	78	259	15	34	16	25	3	17.58	99.62
A62	1.00	3.37	13.68	53.21	1523	1.80	13.63	0.77	111	145	1078	7.03	23	99	117	97	67	601	28	125	558	24	52	31	41	8	4.60	99.77
A63	1.01	4.45	11.91	48.28	642	1.39	13.23	0.62	126	160	968	6.53	23	83	85	86	43	589	23	98	474	20	38	14	12	6	11.57	99.34
A64	0.82	2.84	9.42	35.09	1465	1.04	26.48	0.49	108	73	1083	5.34	20	58	115	92	36	992	20	77	626	22	27	11	17	4	18.03	100.04
A65	0.77	3.19	12.22	45.63	936	1.52	15.79	0.59	131	101	949	6.39	25	89	122	122	50	828	21	94	598	26	38	23	35	6	13.35	99.88
A66	1.41	4.03	13.61	51.63	983	1.89	12.94	0.97	144	177	1198	8.19	27	77	120	111	49	656	24	110	197	17	41	19	26	6	4.77	99.83
A67	1.43	4.03	12.62	48.49	887	2.11	15.12	1.06	128	147	866	7.93	24	69	109	68	42	1253	19	112	512	14	39	12	424	6	6.80	100.06
A68	0.95	2.61	10.40	43.89	1175	1.18	18.67	0.53	107	108	1070	5.54	20	127	124	100	47	614	20	95	624	14	38	16	24	8	15.54	99.76
A69	1.00	3.00	12.09	51.15	622	1.68	11.76	0.63	117	137	992	6.48	29	80	181	111	49	787	20	109	269	20	42	20	80	6	11.34	99.49

variation is comparably high ($vt = 1.95$) and reflects polygenic population (Buxeda i Garrigós and Kilikoglou 2003). The evenness chemical variability graph showed that Pb introduces a major part of the chemical variability, due to high concentrations measured in a few samples, which appears to be a result of contamination. For this reason, it was excluded from the dataset. Apart from Pb, the elements excluded from the dataset were Cu, Na, P and Ba, not least since their variability may be related to the original use or post-depositional conditions (e.g. Picon 1991; Buxeda i Garrigós 1999; Schwedt et al. 2004). After removing these elements, the total variation is lower ($vt = 0.84$) and it may represent monogenic population. Furthermore, the results of testing of the similarity or dissimilarity of all samples towards the groups' average compositions allowed for establishing final chemical cluster comprising 51 samples and two loners (A.52 and A.53). It is important to note, however, that the samples gathered together do not form a very tight compositional group, but rather constitute a loose cluster, which is characterised by a relatively high total variation: $vt = 1.80$ for all elements and $vt = 0.75$ without Pb, Cu, P, Na and Ba, and fairly high relative standard deviations for the most elements (Table 6).

Fifteen samples, representative of the macroscopic and chemical variability of the assemblage, were selected for thin section petrography. All samples were classified to a single petrographic group because of overall similarity in mineralogy. However, two subgroups were defined on the basis of grain size distribution and frequency of inclusions. The thin sections classified into subgroup PG 1A (A.03, A.07, A.12, A.21, A.24, A.38, A.63, A.66 and A.69) are characterised by semi-coarse fabric, while samples of subgroup PG 1B are finer (A.14, A.18, A.42, A.44, A.46 and A.64). It is important to mention that dividing the samples into subgroups was difficult, because of the continuum seen in the frequency of inclusions, sorting and grain size distribution.

The overall fabric group is characterised by a calcareous clay matrix and inclusions of ophiolitic rocks, mudstone, carbonates and low and high calcareous Tcfs (Fig. 4). Fragments of serpentinite, various volcanic and plutonic rocks, as well as their constituent minerals, such as plagioclase feldspar, hornblende, pyroxenes, epidote group minerals, opaque minerals, chlorite and mica, form a suite of inclusions typical of ophiolitic rocks. Moreover, quartz, chert, limestone and calcite were identified. Carbonates include microfossils (foraminifera, ostracods and radiolarians) and micritic pellets. In most cases, the colour of the matrix is associated with its optical activity. When the micromass is yellow, orange or orange brown in PPL and yellowish orange, yellowish brown or red brown in XP, it is optically active to moderate. When the micromass is brown or greenish brown in PPL and dark brown or dark greenish brown in XP, it appears optically inactive. Slip layers have been identified in few samples (thickness 10–16 μm). It seems that the slip was made of very fine, iron-rich clay.

Table 6 Mean chemical composition (oxides are expressed in wt% and elements in ppm) and relative standard deviation (rsd) of MG 9, MG 10, MG 11, MG 12, the final fabric group, a group of sugar moulds and molasses jugs from Kouklia (Waksman 2014; Tables 2 and 4), as well as chemical composition of the briquettes made of the geological samples and fired at 700 °C and 900 °C

	MG 9, <i>n</i> = 18		MG 10, <i>n</i> = 10		MG 11, <i>n</i> = 17		MG 12, <i>n</i> = 8		Final fabric group, <i>n</i> = 51		Sugar moulds and molasses jugs, <i>n</i> = 5 (Waksman 2014)		Geological samples							
	Mean	rsd	Mean	rsd	Mean	rsd	Mean	rsd	Mean	rsd	Mean	rsd	Ezousa river		Armou		Agriokalami stream		Paphos, Ancient Agora	
													700	900	700	900	700	900	700	900
Na ₂ O	0.89	33	0.89	16	0.98	33	1.20	29	0.94	30	1.21	27	0.97	0.88	0.60	0.62	0.50	0.48	0.45	0.44
MgO	3.38	17	3.33	15	3.51	14	3.84	21	3.44	16	3.69	10	3.45	3.60	2.33	2.35	3.65	3.64	1.92	1.95
Al ₂ O ₃	13.00	13	12.58	10	13.24	9	11.70	11	12.84	11	15.07	10	10.45	10.94	11.79	12.00	6.53	6.85	16.76	17.03
SiO ₂	50.74	12	50.88	5	52.31	8	44.97	12	50.69	10	55.78	4	52.80	55.93	72.05	73.61	39.67	43.91	58.21	58.60
P	997	58	975	27	985	23	856	16	974	39	651	19	669	701	975	1007	591	595	461	484
K ₂ O	1.71	19	1.82	13	1.70	14	1.50	21	1.71	16	1.92	23	1.11	1.21	0.95	0.97	0.86	0.92	2.61	2.67
CaO	13.31	34	14.85	20	13.95	22	17.89	19	14.39	26	12.75	36	15.37	17.35	5.14	5.18	28.53	32.16	4.93	5.51
TiO ₂	0.68	16	0.68	10	0.71	13	0.68	25	0.68	15	0.72	10	0.52	0.55	0.50	0.52	0.34	0.36	0.95	0.97
V	132	21	117	18	131	17	114	12	125	18	183	16	120	127	93	93	80	80	143	142
Cr	110	19	112	13	126	18	143	26	120	21	149	20	281	244	75	80	1225	1232	304	354
Mn	1017	12	1083	23	1115	15	987	23	1064	17	890	16	1282	1459	461	470	1304	1573	2096	2120
Fe ₂ O ₃	6.78	17	6.42	10	6.89	13	6.75	17	6.70	13	8.39	12	6.13	6.21	3.78	3.85	3.92	4.06	8.73	8.87
Co	25	12	25	8	26	13	23	12	25	12	nd	8	25	26	13	17	23	26	25	26
Ni	86	22	91	18	92	18	98	43	91	24	89	8	120	127	35	40	187	175	126	126
Cu	131	36	124	27	110	16	123	37	122	30	nd	11	130	116	62	66	65	120	59	62
Zn	105	14	82	20	99	12	90	17	96	17	84	11	114	167	73	73	59	61	101	100
Rb	58	24	60	22	61	19	46	22	59	21	52	21	45	50	36	36	30	34	101	96
Sr	678	45	911	25	648	30	813	26	733	30	544	31	368	409	170	173	576	627	222	231
Y	23	9	24	7	24	10	21	8	23	10	nd	8	21	20	35	37	18	19	56	56
Zr	112	24	114	10	110	13	95	16	111	17	107	8	97	99	94	94	77	78	256	255
Ba	492	48	668	51	524	34	344	36	528	45	393	58	217	235	48	84	229	218	365	349
La	25	25	27	23	26	31	17	50	25	29	22	26	14	21	14	22	17	25	54	46
Ce	47	24	52	20	48	23	33	26	47	23	51	18	37	42	47	36	42	42	103	103
Nd	24	25	24	24	23	26	16	36	23	28	nd	18	21	20	28	25	20	11	46	49
Pb	50	67	171	70	43	61	85	162	78	116	nd	39	39	35	16	25	17	17	28	29
Th	7	35	8	17	7	25	6	15	7	27	nd	6	6	5	4	4	1	4	13	12

nd not determined

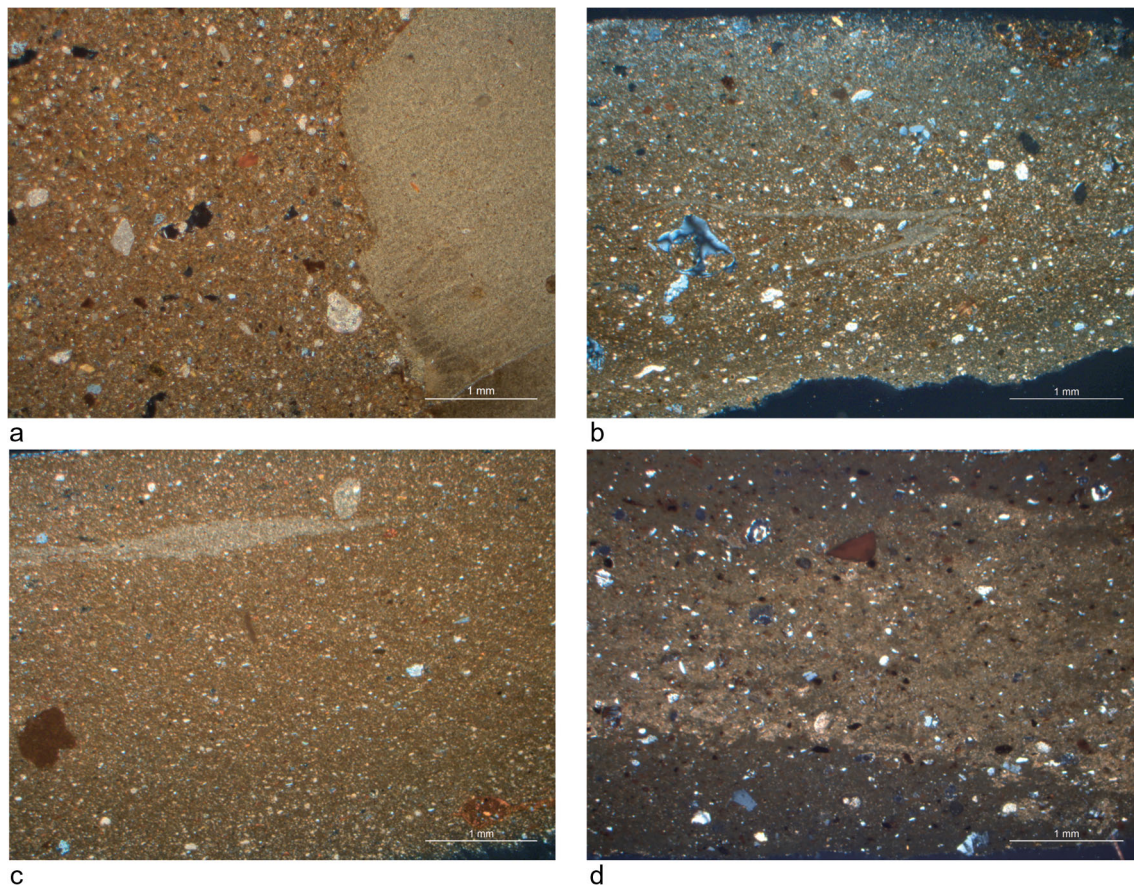


Fig. 4 Thin section photomicrographs, in crossed polarised light (XPL). **a** Sample A.69 (MG 9). **b** Sample A.12 (MG 10). **c** Sample A.44 (MG 11). **d** Sample A.46 (MG 12)

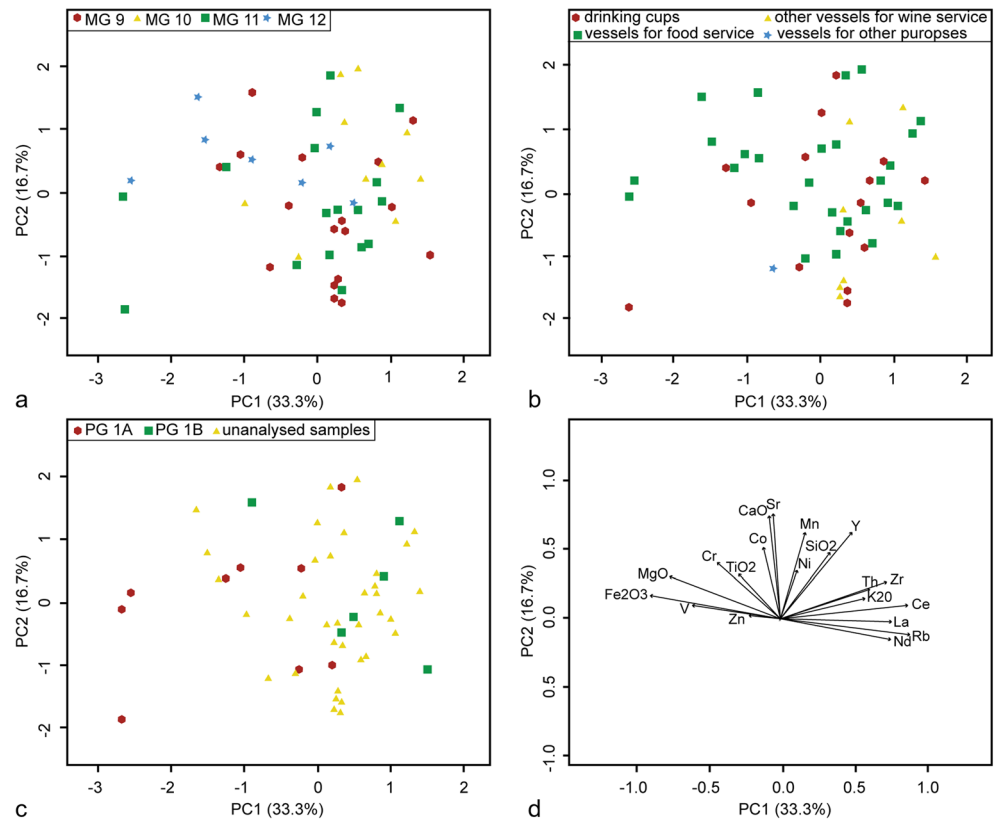
The presence of various Tcfs types within the samples may suggest natural or intentional (anthropogenic) clay mixing (Fig. 4). The varied colours and optical activity both observed across all samples, as well as within single ones, indicate that the analysed vessels were fired under variable conditions and in a range of temperatures from low to high. Despite the differences within the group, the overall similarity of the mineralogy of inclusions indicates the use of raw materials associated with the same geological environment. The mineralogy of this fabric group is closely comparable to the geological sample collected from the Ezousa river, which is characterised by the presence of ophiolite components, such as intermediate to basic igneous rocks and serpentinite, as well as carbonates, microfossils, chert and mudstone. This similarity suggests that locally occurring raw materials were used for the production of the pottery assigned to this group.

In order to examine possible patterns within this group, chemical data were correlated to MGs, typology and the results of petrographic analysis (Fig. 5). Principal component analysis (PCA) was performed on the logratio transformed chemical data using Al_2O_3 as divisor since this element introduces the lowest variability. The analyses

were carried out on the concentrations of the following 21 elements: Mg, Al, Si, K, Ca, Ti, V, Cr, Mn, Fe, Co, Ni, Zn, Rb, Sr, Y, Zr, La, Ce, Nd and Th. The results indicate that there is no correlation between the chemical composition of the samples and MGs (Table 6; Fig. 5a) nor typology (Fig. 5b). Instead, the origin of the variability within the cluster can be explained by both technological and taphonomic factors. The petrographic subgroups seem to show a pattern (Fig. 5c). Most of the samples classified to finer variant (PG 1B) tend to be distributed on the right side of the plot, suggesting that the high chemical variability may be related to the size and frequency of inclusions.

The observed variability of Cu and Pb may be related to the presence of metal objects in the deposit. It is worth noting that around 350 iron, bronze and lead objects were distributed from top to bottom of the well. Some samples (A.12, A.16, A.17, A.43, A.48, A.49, A.56 and A.67) show high Pb concentrations, reaching 424 ppm in sample A.67, and this appears to be directly linked to their local burial environment since all these vessels were found in contact with lead objects. Other elements which led to the high total variation of the 51 samples are Ca and Sr.

Fig. 5 The principal component plot of the WD-XRF data including 21 elements for the 51 samples. **a** In correlation with the MGs. **b** In correlation with the typology. **c** In correlation with the petrographic subgroups (PG 1A and PG 1B). **d** Projections of discriminating elements



Variability in these may be related to post-depositional processes (leaching or deposition in pores) or to technological parameters. For the latter, it is crucial to note that potential evidence for clay mixing has been observed both in thin sections and under the SEM (where Ca-rich nodules were observed). The latter analysis also revealed the presence of unevenly distributed baryte inclusions. These are likely the origin of the relatively high variability in Ba observed. The mixing of different materials would have caused further perturbations in the samples' elemental compositions, in particular if recipes and/or source materials themselves also show variation, likewise, if mixing of the materials had not resulted in complete homogenisation (Arnold et al. 1991), as is indicated by observations on thin sections and under the SEM.

Overall, the chemical composition of this group is comparable to Byzantine sugar moulds and molasses jugs from Kouklia (Waksman 2014), which also present high variability. This similarity suggests that this chemically heterogeneous group of pottery could have been produced in the region of Paphos. The chemical composition of one geological sample collected from an exposure by the Ezousa river further supports the assumption of local production (Table 6). Moreover, on the basis of chemical and mineralogical analysis, Roger H. King (1987: 207) already suggested that

prehistoric and Iron Age pottery discovered in the area of Palaipaphos is similar in composition to clay from the upper Ezousa river.

Ten samples were examined under the SEM, revealing a range of microstructures, with some samples not showing any vitrification while others were extensively vitrified (Fig. 6). The degree of vitrification combined with the atmosphere of firing and high CaO content indicates that these pots were fired in a wide range of temperatures. Overall, the macroscopic grouping of the pottery into four MGs appears to relate mainly to differences in firing rather than to the composition of the clay paste (Table 7).

In all samples analysed, the slip layers appear to be very thin and do not exceed 16 μm in thickness. The slip layers are composed of a finer material than the ceramic bodies. No significant chemical variations were observed between the ceramic bodies and slip layers. The latter present a slight depletion of CaO and MgO with an enrichment of K_2O in the surface layers.

The examination of the samples under the SEM led to the identification of high calcareous clay pellets (up to 42% of CaO) as well as a range of carbonate materials (e.g. secondary calcite infilling in voids and microfossils), features which were already seen in thin sections (Fig. 7). Their varied presence, as discussed above, may have contributed to the relatively high variability of CaO and Sr within this group.

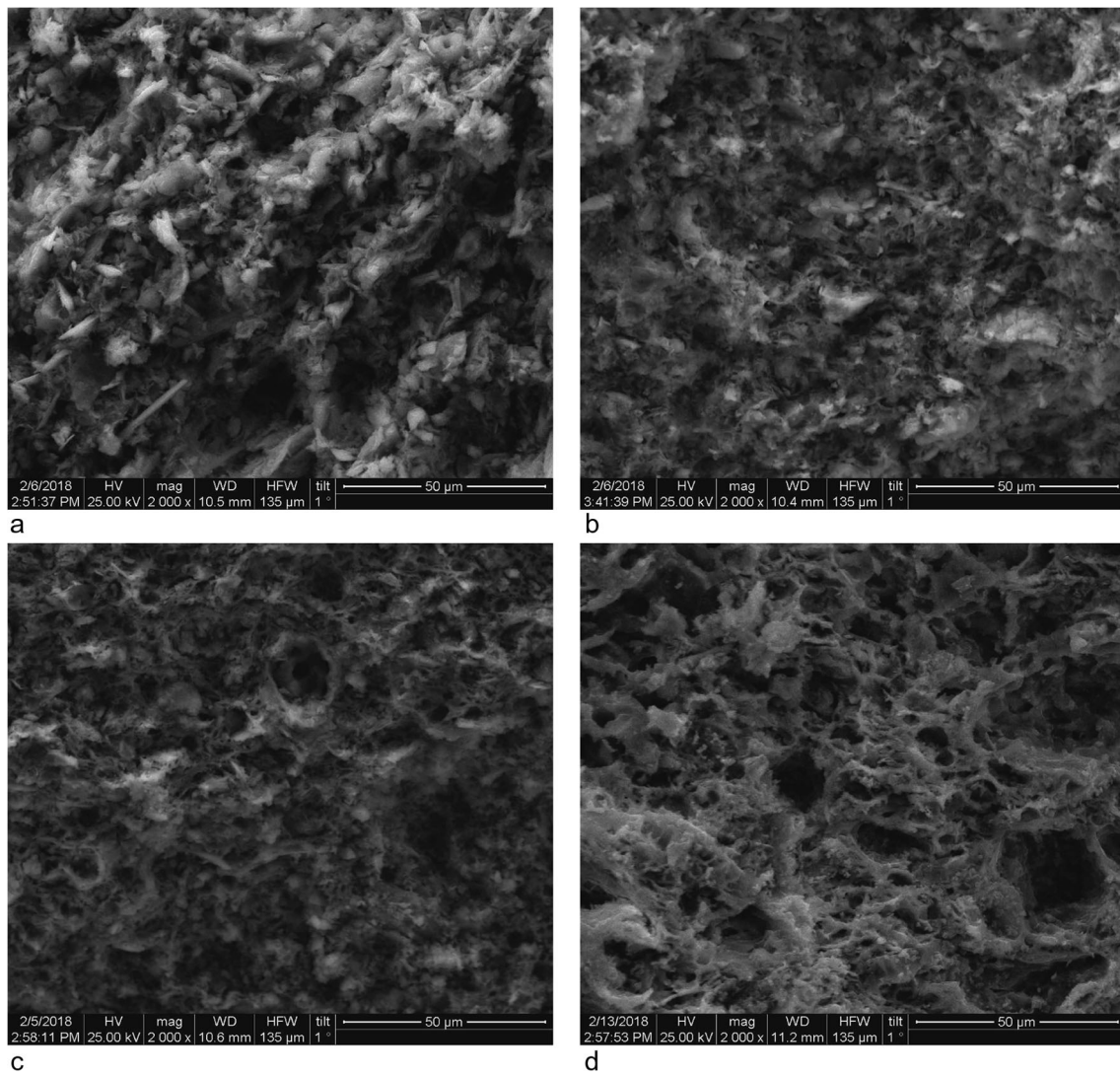


Fig. 6 The SEM micrographs (in secondary electron mode) showing the range of the degree of vitrification of the ceramic body. **a** Sample A.18 (MG 9) showing initial vitrification. **b** Sample A.21 (MG 10) showing no vitrification to initial vitrification. **c** Sample A.44 (MG 11) showing initial vitrification to extensive vitrification. **d** Sample A.07 (MG 12) showing extensive vitrification

Table 7 The results of SEM examination combined with the results of macroscopic, thin section and WD-XRF analyses

Sample	MG	Shape	Slip colour	Firing atmosphere	Optical activity	CaO (wt%)	Vitrification of body
A.18	9	Late with rolled rim	Red	O-	MD	12.91	IV
A.63	9	Large plate with sloping rim	Red	O-	AC	13.23	NV–IV
A.69	9	Echinus bowl	Red	O	AC	11.76	NV–IV
A.21	10	Plate with rolled rim	Brown	M	AC	12.90	NV–IV
A.42	10	Large echinus bowl	Brown	M	AC	14.07	NV–IV
A.24	11	Lekane	Black	M	MD	10.04	IV–V
A.38	11	Large plate with sloping rim	Red	O	MD	15.01	IV–V
A.44	11	Jug with cylindrical neck and ovoid body	Red	O	MD	18.75	IV–V
A.07	12	Lekane	Black/dark brown	M	IA	14.67	V
A.46	12	Echinus bowl	Black/dark brown	M	IA	13.56	V

Firing atmosphere (based on macroscopic analysis): O = oxidised, O- = not fully oxidised, M = mixed; optical activity (based on thin section petrography): AC = active, MD = moderate, IA = inactive. The CaO concentrations were assessed based on WD-XRF data. Vitrification: NV = no vitrification, IV = initial vitrification, V = extensive vitrification

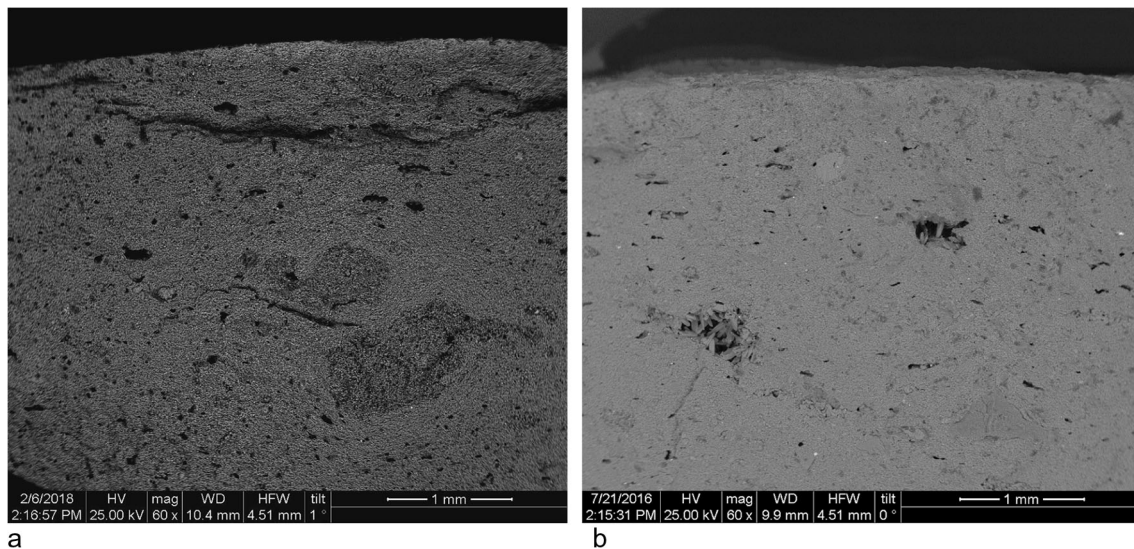


Fig. 7 The SEM micrographs (in backscattering mode) showing **a** high calcareous clay pellets, sample A.18 and **b** secondary calcite infilling voids, sample A.63

Summary and conclusions

The results of the integrated studies indicate that most of the analysed CCW vessels (51 out of 53) from the Late Hellenistic well deposit unearthed in the Paphos-Agora form one, albeit somewhat variable, fabric group, which was made from raw materials associated with the same geological unit(s), potentially from the same broader area. This fabric group includes samples, which at the outset of the research had been classified into four major typologically and stylistically related MGs, separated due to variation related to the inclusion frequency, colour, hardness, the feel of the surface and the texture of the fresh break. However, the results of chemical analysis indicated that they constitute a loose cluster, which does not exhibit any meaningful internal patterns in terms of either MGs or vessels types. This suggests that despite the observed variability, the MGs appear to be related to the similar provenance or raw material selection. This finding is supported by the petrographic analysis, which shows that all analysed samples are very similar and they belong to a single petrographic group, showing internal (continuous) variation of the frequency, sorting and grain size distribution of inclusions. This fabric group is characterised by the presence of ophiolite components and more specifically fragments of intermediate to basic igneous rocks and serpentinite, mudstone, carbonates, as well as Tcfs. The mineralogical and elemental composition of the vessels ascribed to this fabric group is similar to the composition of at least one clay sample that was collected from a section of the Ezousa river, which is around 5 km distant from the ancient town. Despite this similarity, it seems unlikely that this sample represents a source consistently exploited for such a large scale pottery production, but it suggests that the raw materials used for CCW are

compatible with the geology of this area. Further indication for the local origin of this pottery is provided by the observation that its chemical composition is similar to the composition of the Byzantine sugar moulds and molasses jugs considered to have been produced in the vicinity of Nea Paphos (Waksman 2014). It seems that most of the analysed samples represent locally produced pottery. This is in accordance with the results of John Lund's (2015: 158–160) work on the circulation of pottery in Hellenistic and Roman Cyprus, which indicated that the majority of pottery was manufactured locally.

The macroscopic and compositional variability of this group may be related to natural diversity of raw materials and technological/anthropogenic factors, introduced by a number of potters/workshops in the same area and by technological choices made by the potters. Finally, it seems to also be partly related to post-depositional processes.

It is worth mentioning that Hellenistic pottery from Knidos shows similar variability (Schneider 2000; Kögler 2011). Patricia Kögler (2011: 167; 2014: 168) argued that the production of low-quality fine ware pottery in Knidos in the Late Hellenistic period was linked to increasing demand, which she identified through the manufacture of wider range of types and an increase in quantity, as well as through a wider distribution than in the Early Hellenistic period. She suggested that the decrease in quality during the Late Hellenistic period was because workshops manufactured more vessels in the same amount of time than before and spent less time at each stage of production. Changes of a similar nature were also observed in other production centres in the Eastern Mediterranean. Susan Rotroff (1997: 11–12), for example, noted a decline in the potter's art at Athens in the Late Hellenistic period (for more examples Lund 2015: 219–220).

Assuming that the group of pottery presented here was indeed locally produced, similar explanation can be proposed for Nea Paphos. The importance of the city increased during the Late Hellenistic period, when it became the capital of the island at the end of the third or in the early second c. BCE (Lund 2015: 238). The development of Nea Paphos and its hinterland may have created increased demand for goods, including table ware pottery, which was easily broken and would have needed to be replaced quite often. Higher demand would have likely led to increased production, which one might speculate to have resulted in decreased quality and lower technological standardisation of the local production. These could be related with high-speed production (Hruby 2014) or increase in the numbers of workshops with less skilled potters (Albero Santacreu 2014: 257). Both assumptions are supported by low fabric standardisation and by signs of careless forming and surface treatment, often resulting in the production of non-symmetrical vessels, with some ‘defects’, such as deformed rim or base, and vessels covered only partially with a thin slip. Moreover, the relative lack of technological standardisation is also reflected in the observed variability in firing conditions. Variability in colour and optical activity of the matrix observed in thin sections, as well as the degree of vitrification, from no vitrification to extensive, assessed through SEM analysis, indicate that these pots were fired in a wide range of temperatures in not consistently controlled atmosphere. As this pottery comes from the same homogenous deposit at the Agora, its low quality and lack of technological standardisation could have been related with a specific context of pottery consumption, such as a feast, an event or series of events that would have encouraged production of such scale (Albero Santacreu 2014: 272). However, one can see similar variability in CCW of the same fabric group from other contexts in Paphos (Marzec 2017). A ‘marked decline in quality and finish’ in CCW of the first c. BCE was also observed by John Hayes (1991: 27) within the material from the House of Dionysos in Nea Paphos. Moreover, the preliminary results of ongoing studies focused on cooking pottery from this site show similar production pattern. It seems that this is the phenomenon of mass produced pottery characterised by low quality and low technological standardisation, increasingly observed in the Eastern Mediterranean during the Late Hellenistic period.

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