



# Assessment of drinking water quality and various household water treatment practices in rural areas of Northern India

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## Abstract

Groundwater is an essential freshwater resource and contributes about two-thirds of the world's freshwater reserves. Groundwater contamination can occur due to faulty construction or leakage in pipes and somehow through surface contaminants like a human or animal fecal matter or other foreign substances. The present study was conducted to assess the physicochemical as well as the bacteriological quality of groundwater in the rural areas of the Kurukshetra district, northern India. The samples were collected and analyzed for various physicochemical (pH, EC, TDS, Cl, TH, Ca, and Mg) and the bacteriological [total coliforms (TC) and fecal coliforms (FC)] parameters. The samples were then treated and re-analyzed after applying various household practices like boiling, solar disinfection or SODIS, chlorination, and reverse osmosis (RO) technique. After treatment, the trend obtained from the analysis for removing bacteriological contaminants follows as RO  $\approx$  boiling > chlorination  $\approx$  SODIS. It indicates that RO followed by boiling is the best method for drinking water treatment at home which removes most of the microbes as well as EC and TDS to a large extent. However, chlorination and SODIS methods are also effective in removing contaminants.

**Keywords** Groundwater · Water quality · Boiling · Chlorination · Solar disinfection · Reverse osmosis

## Introduction

Drinking water can be called as 'safe' when it does not pose any significant health risk over a lifetime of consumption [World Health Organization (WHO), 2008]. Generally, borehole water or groundwater has fine bacterial quality. However, pathogens can be noticed due to infiltration of surface water, septic discharges, or leakage in sewer lines (Ahmed et al. 2011). Water scarcity has become a major problem, especially in the arid and semi-arid regions of India, which are receiving < 500 mm of the average annual precipitation (Keesari et al. 2014). The dependency on groundwater resources increased mainly because of the failure of monsoonal rains and the limited availability of surface water.

In India, water-borne diseases are responsible for the death of over 100,000 people annually. Providing safe water

to such an increasing population is a major challenge (WaterAid India 2019). India is at first rank worldwide with over 386,000 annual deaths attributable to diarrheal diseases [United Nations Children's Fund (UNICEF) and World Health Organization (WHO), 2009]. According to WHO/UNICEF (2010) Joint Commission Report, out of the 1.2 billion population in India, now 88% of people have access to improved water sources, as compared to 72% in 1990. Although, the report also says the presence of improved water sources does not ensure that the water is safe for drinking. Water experts have universally accepted that the significant risk related to the intake of water is the presence of microbial contamination by human or animal feces (WHO, 2004). Waterborne diseases such as diarrhea, cholera, and typhoid become an enormous burden due to a lack of safe water. Diarrhea itself is a prime reason for morbidity and mortality among children < five years of age and, mostly in low-income countries, diarrhea was the third prime cause of death in 2004 (WHO, 2009).

Piped water supplies used for drinking purposes should be provided with water protection and centralized treatment. However, worldwide about 780 million people are not having access to an "improved" drinking water source,

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and many available water sources are unsafe and at a far distance from the houses [Centers for Disease Control and Prevention (CDC), 2016]. Although households can access piped water at home, it may also get contaminated by some faults present in the distribution system. Drinking water can also get contaminated due to improper handling practices and unsafe storage in the household (Nath et al. 2006) and increased distance from the treatment reservoir to the distribution points (Sharma et al. 2013).

Water treatment at the household level is promoted widely as a suitable intervention, especially in developing countries, to reduce the waterborne disease burden in poor and remote rural areas (Islam et al. 2015; Boisson et al. 2013). Generally, water treatment is intended to improve the water source to make it and suitable for the end-user. Any system used for water treatment can remove organic or inorganic matter, which can be insoluble or particulate form, including microbes. The household water treatment (HWT) system imparts a medium to improve the drinking water quality. Today's desired treatment technologies comprise chemical, ceramic filters, disinfectants, solar disinfection (SODIS), or ultraviolet (UV) disinfection, as well as boiling process (Clasen 2009). If electricity is available in continuous supply at the household level, numerous technologies are available for water treatment. Examples of such treatment brands available in India include water purification by reverse osmosis (RO), like Kent Osmosis or other that uses a combination of several treatments like UV and boiling treatment, and sediment filtration, for example, Aqua Guard (Jain 2009). Although, in developing countries, especially rural and low-income people are often deprived of continuous electricity supply, and mostly these people are at higher risk of having an unsafe supply of water.

Many chemical disinfectants are available for water disinfection, such as chlorine dioxide, ozone, chlorine, and hydrogen peroxide. Globally, chlorination treatment is the most widely used technique to disinfect water. In many countries, it is used enormously because of its lower price, effectiveness against microbes, and implementation in any size operations (Tomás-Callejas et al. 2012; Luo et al. 2011). Chlorine is also associated with some disadvantages like the developing disinfection by-products (DBPs) in the water, residual chlorine with harmful impacts on aquatic life, and its low effectiveness against some viruses and protozoan parasites (Tomas-Callejas et al. 2012). Chlorine is most efficacious against bacteria such as *E. coli* as compared to parasites (Arnold and Colford 2007). In India, several studies showed resistance among users to use water treated with chlorine due to the unpleasant smell and taste, which are perceived longer (Firth et al. 2010; Gopal et al. 2009). Application of chlorine in low concentration may cause ineffective disinfection of water and in high dosage, chlorine may change the taste and smell resulting in a high concentration of DBPs.

These complexities lead to ineffective chlorination at the household level. Therefore, household chlorination is generally performed by tablets or a dilute solution with a fixed concentration of chlorine and these are mostly provided by government supplies.

Boiling water is probably the simplest and oldest method, which is mostly used to remove pathogens from water. Some authorities recommend that the water should be boiled for 1–5 min (CDC 2017). The water is brought to a rolling boil to kill the pathogens in the most effective way, especially in turbid waters or even at high altitudes (WHO, 2018). Field studies in India proved to be effective the boiling process in improving the microbiological quality of water (Clasen et al. 2008; Freeman et al. 2012). Approximately, 21.6% population in 67 middle- and low-income countries reported water boiling at home before drinking, which is four times higher than those reported drinking water chlorination or filtration (Rosa et al. 2010).

Natural solar (UV) radiation can be an effective method to heat water up to temperatures lower than boiling temperature. Especially the UV-A spectrum (320 to 400 nm) is the most essential component of solar radiation for inactivating microorganisms. The UV light damages the nucleic acids of the microbes and gets inactivated. It also prevents replication [United States Environment Protection Agency (USEPA), 2003]. Temperatures of 55 °C to more than 60 °C killed most of the pathogens rapidly due to the thermal effect. This high temperature represents pasteurization capable of inactivating nearly all enteric pathogens (enteric viruses, bacteria and parasites) within several minutes to hours (Skinner and Shaw 2004). The present study has the following objectives:

1. To analyze the untreated household water samples for physicochemical and bacteriological parameters.
2. To check the efficiency of various household treatment practices including boiling, chlorination, SODIS, and reverse osmosis in treating the raw water at home.

## Materials and methods

### Study area

Kurukshetra district is present in Haryana north–eastern region with an area covering 1530 km<sup>2</sup>. It is demarcated latitudinally in North with 29° 53' 00" and 30° 15' 02" and longitudinally in East with 76° 26' 27" and 77° 07' 57". In Haryana, 3.46 percent area is covered by the Kurukshetra district. The population of Kurukshetra is 9,64,231 as per the 2011 census, which is distributed in six blocks, namely Thanesar, Babain, Shahabad, Ladwa, Ismailabad, and Pehowa. Kurukshetra is covered by arid brown soil.

The Markanda river mainly irrigates the study area, and the river Sarasvati have also joined the river Ghaggar in Kurukshetra (Pehowa) [Ministry of Water Resources (MOWR), 2016]. The average annual rainfall of about 582 mm is recorded over the study area. However, the annual rainfall during the study (in 2015) was 390.8 mm (India Meteorological Department, 2015). According to the estimates of 2011 [Central Ground Water Board (CGWB), 2013], in major parts of the Kurukshetra district, the water level lies at > 30 m below ground level (bgl) and ranges from 20.18 to 32.64 m bgl during the pre-monsoon and from 21.80 to 34.41 m bgl in the post-monsoon season. The major concern for the Kurukshetra district is the depletion of its groundwater resources, as the water table now lies under the dark zone (CGWB, 2013). This study’s purpose was to analyze the microbiological quality of groundwater in nine villages of Kurukshetra district and to check the efficiency of different HWT practices to improve the quality of groundwater. The villages were selected from the study blocks Thanesar (T), Shahabad (S), and Ladwa (L).

### Sample collection

The sampling sites were finalized based upon the physico-chemical and bacteriological analysis of the 81 groundwater samples from different villages of the Kurukshetra district during pre-monsoon, monsoon, and post-monsoon season from 2015 to 2016. The results of one year study were already published elsewhere (Malan et al. 2020; Malan and Sharma 2018). However, the nine samples that were found to be most contaminated during the one year study were finally selected for this treatment experiment. The water samples were collected from the household borewells of different villages located in three blocks of Kurukshetra (Fig. 1) during September 2016 (monsoon season). The borewells depth ranged between 36 and 94 m bgl. Clean and sterilized glass bottles were used for sampling of groundwater. The sample bottles were filled after running the water for five minutes to avoid any external contamination. Sterile conditions were sustained throughout the collection and analysis of the samples. Sampling and analysis were performed as per the guidelines given by Bureau of Indian Standards (BIS 2012) and (WHO, 2011). The samples were analyzed for TC and

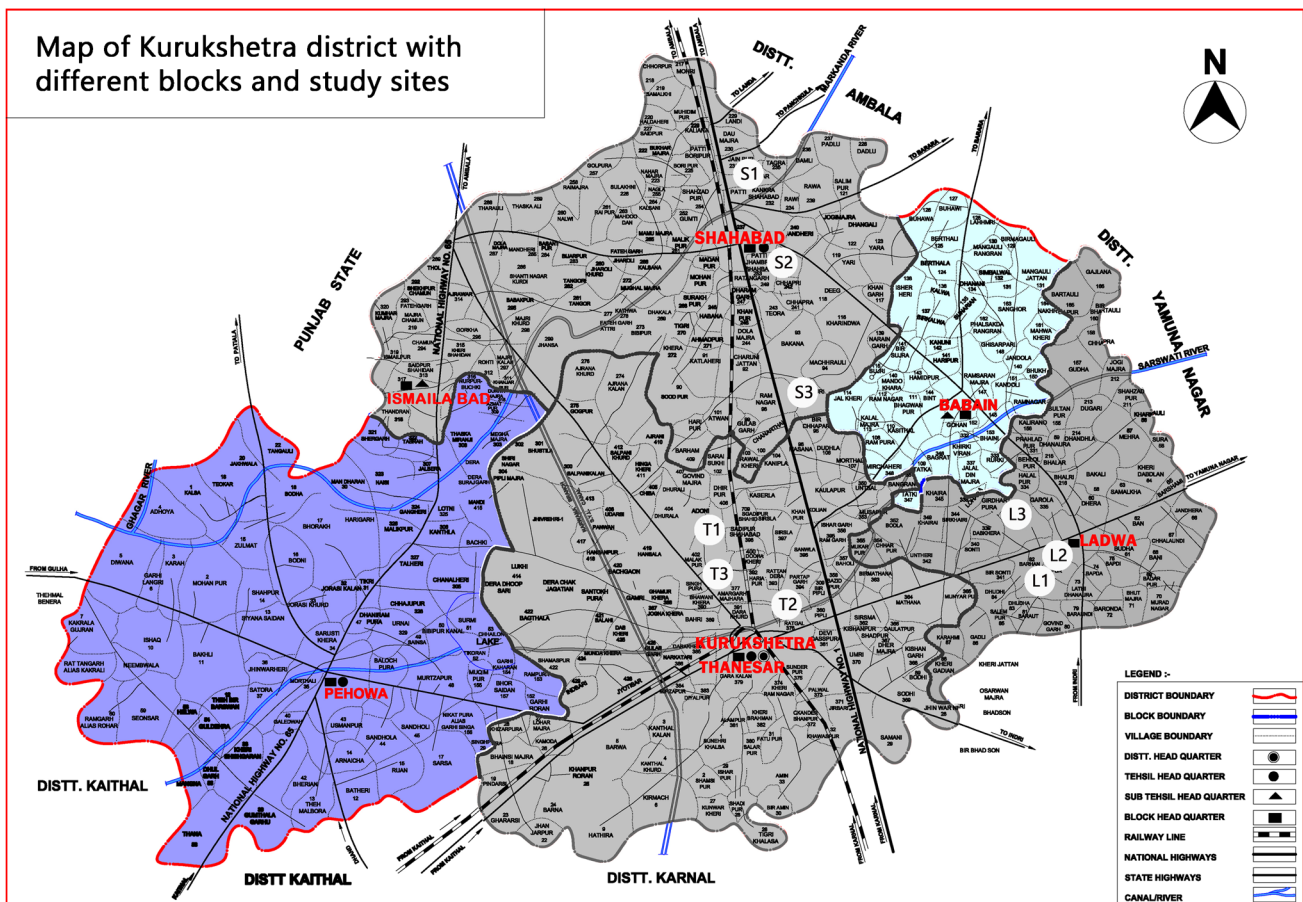


Fig. 1 Map of Kurukshetra district showing three study blocks (grey color) and nine study sites (depicted with white circles)

FC within six hours of collection using the most probable number (MPN) method. The samples were then treated with various HWT practices like boiling, chlorination, SODIS, and reverse osmosis technique and analyzed again after treatment. The chemicals and media were acquired from Hi-Media, Mumbai, India.

### Physicochemical analysis

The untreated and treated water samples were examined for different physicochemical parameters, including pH, total dissolved solids (TDS), electrical conductivity (EC), total hardness (TH), bicarbonate, chloride, calcium, and magnesium contents using the methods of American Public Health Association (APHA) (2005).

### Enumeration of TC and FC (MPN index/100 ml)

The TC and FC count in the samples were enumerated by MPN method as per APHA (2005). The TC were enumerated using the “five-tube assay” of the MPN technique by using MacConkey broth. The MPN index of water samples was evaluated from the MPN table (APHA 2005). The presence of coliforms in water samples was confirmed by the transfer of culture (a loopful) from a positive presumptive tube into a tube of “Brilliant Green Lactose Bile” (BGLB) broth with Durham tubes for the production of gas. Further, the pure colonies of the bacteria were isolated by streaking a loopful of broth from a positive BGLB tube onto “Eosine Methylene Blue” (EMB) agar plate. The identification of colonies on EMB agar were based on “Gram’s staining test” and some other biochemical tests, including indole, methyl-red, Voges-Proskauer (VP), and citrate utilization test, collectively known as IMViC test.

The positive BGLB tubes obtained after incubating at 44.5 °C confirmed the FC in groundwater samples. Eijkman test was also used for FC confirmation in which the MPN positive tubes from the presumptive test were inoculated in “Tryptone broth” tube for an indole production test.

### Treatment procedure

The HWT techniques like boiling, chlorination, solar disinfection (SODIS), and reverse osmosis (RO) were used to treat groundwater samples. One liter of water was boiled to a rolling boil in a metal container for approximately 10 min and then filtered before drinking (WHO, 2004). In the chlorination treatment method, the chlorine tablets supplied by Primary Health Centers (PHCs) to the villagers free of cost through Accredited Social Health Workers (ASHA) workers by the Government of India (GoI) were used. About 2 mg of chlorine tablet was used to treat 1 L of water and kept for

half an hour to complete the reaction. In SODIS method, direct sunlight is used to treat the water at home. Polyethylene terephthalate (PET) bottles as specified/recommended by WHO/UNICEF (2012), usually of 1-L capacity, were used to collect water. They were exposed to sunlight for a minimum of 6 h for total disinfection under optimum weather conditions (less than 50% cloudy). If the sky is more than 50% cloudy, the bottle water temperature does not exceed 42 °C, the bottle should be exposed for two consecutive days. The water with high turbidity should be filtered before using the SODIS method as it can affect the UV radiation. In reverse osmosis (commonly referred as RO), water is de-mineralized by flowing under pressure through a semi-permeable membrane. The treated water was collected from household RO systems of different brands and then analyzed in the laboratory to know the efficiency of various RO systems in the removal of contaminants as compared to other HWT practices.

## Results and discussion

### Physicochemical analysis of water sample before and after treatment

Based on physicochemical parameters, groundwater suitability for drinking in some of these villages has been discussed elsewhere (Malan and Sharma 2018) and the bacterial isolates and source identification was reported in another study (Malan et al. 2020). However, the present paper investigates the efficiency of various treatment facilities used by rural households. The result of the physicochemical analysis of groundwater samples was mentioned in Table 1. In raw water samples, the pH was ranged between 7.41 and 8.11 (Fig. 2), EC 337–1308  $\mu\text{S}/\text{cm}$  (Fig. 3), TDS 202–623 mg/l (Fig. 4),  $\text{HCO}_3^-$  200–416 mg/l (Fig. 5),  $\text{Cl}^-$  6.01–142.16 mg/l (Fig. 6), TH 44–252 mg/l (Fig. 7),  $\text{Ca}^{2+}$  24–224 mg/l (Fig. 8), and  $\text{Mg}^{2+}$  between 4.86 and 28 mg/l (Fig. 9). After applying various treatment practices, the pH was ranged between 6.88 and 8.84, EC 39.6–1400  $\mu\text{S}/\text{cm}$ , TDS 19.7–696 mg/l,  $\text{HCO}_3^-$  16–420 mg/l,  $\text{Cl}^-$  0–186.20 mg/l, TH 24–372 mg/l,  $\text{Ca}^{2+}$  0–200 mg/l, and  $\text{Mg}^{2+}$  between 2.92 and 57.35 mg/l (Figs. 2, 3, 4, 5, 6, 7, 8, 9).

On mean value basis, all the physicochemical parameters were observed within WHO's permissible limits (2011) except  $\text{Ca}^{2+}$  ion and TDS before disinfection. The studied parameters of treated water samples varied differently after treatment with different HWT practices. Maximum reduction in all physicochemical parameters was observed with RO treatment. There is no reduction in parameters values after treatment with chlorine except  $\text{Ca}^{2+}$ . The various

**Table 1** Physicochemical parameters of drinking water before and after treatment

Parameters	Untreated sample (n=09)	Boiled (n=09)	Chlorinated (n=09)	Solar disinfection (SODIS) (n=09)	Reverse osmosis (RO) (n=06)	WHO permissible limits (2011)
pH	7.78±0.25 (7.41–8.11)	7.91±0.22 (7.68–8.23)	7.84±0.14 (7.68±8.10)	7.91±0.60 (7.08–8.84)	7.05±0.15 (6.88–7.31)	6.5–8.5
EC	786.44±325.74 (337–1308)	720.33±414.07 (160–1396)	877.28±267.06 (499–1318)	755±348.93 (329–1400)	55.41±28.53 (39.6–113)	1500
TDS	373.11±153.51 (202–623)	376.89±159.91 (235–696)	428.78±152.69 (251–671)	360.67±136.73 (204–596)	24.2±8.15 (19.7–40.7)	500
HCO <sub>3</sub> <sup>-</sup>	319.11±72.66 (200–416)	271.56±40.91 (216–332)	328.89±70.62 (224–420)	338.22±68.18 (224–408)	35.33±13.48 (16–56)	500
Cl <sup>-</sup>	52.72±49.79 (6.01–142.16)	66.52±65.37 (8.01–186.20)	62.29±50.08 (12.01–154.17)	53.39±50.38 (4–158.17)	4.0±3.58 (0–10.01)	250
TH	180±71.90 (44.00–264.00)	176.89±66.48 (52–268)	208±82.85 (80–372)	174.22±58.24 (76–236)	36±9.46 (24–48)	600
Ca <sup>2+</sup>	118.22±79.70 (24.00–224.00)	63.56±34.44 (20–128)	108.44±60.51 (4.00–192)	125.33±63.56 (32–200)	4.0±4.38 (0–12)	75
Mg <sup>2+</sup>	15.01±9.26 (4.86–28.00)	27.54±15.11 (7.78–50.54)	24.19±13.65 (14.58–57.35)	11.88±8.16 (2.92–25.27)	7.77±2.53 (4.86–11.66)	50

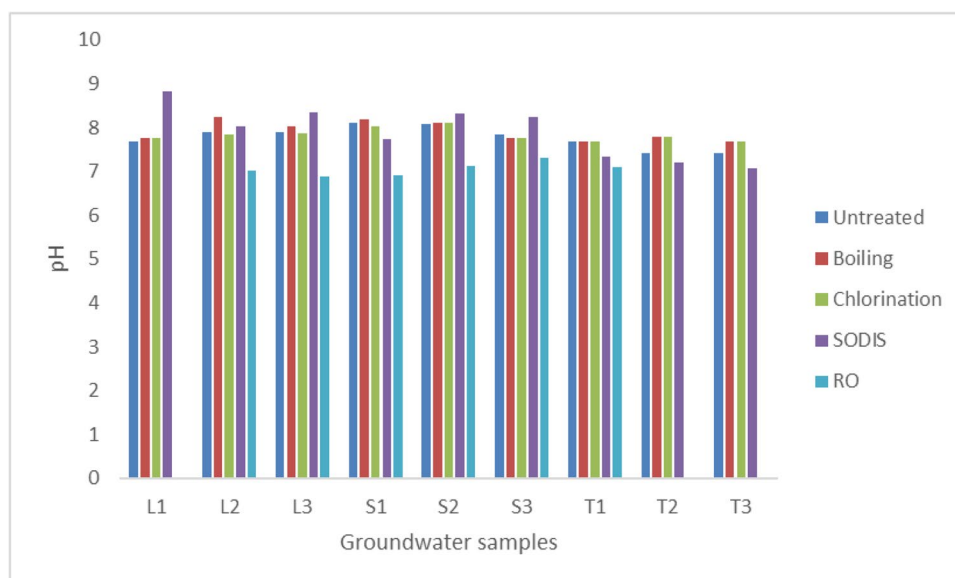
Values are expressed in mean±SD (the value in parentheses denotes the range of each parameter); all parameters are in mg/l except pH (no units) and EC (µS/cm)

treatment practices showed the overall trend of RO > boiling ≈ SODIS > chlorination for most of the physicochemical parameters except calcium ion. Studies have shown effective removal of the physico-chemical parameters by household treatment practices. A report from CDC (2008) stated that RO is effective in the removal of some aqueous salts like sodium, chloride, and common metal ions. A significant decrease was found in the total hardness of the samples after boiling at 100 °C for 10 min in Jaffna Peninsula of Sri Lanka (Wijeyaratne and Subanky 2017).

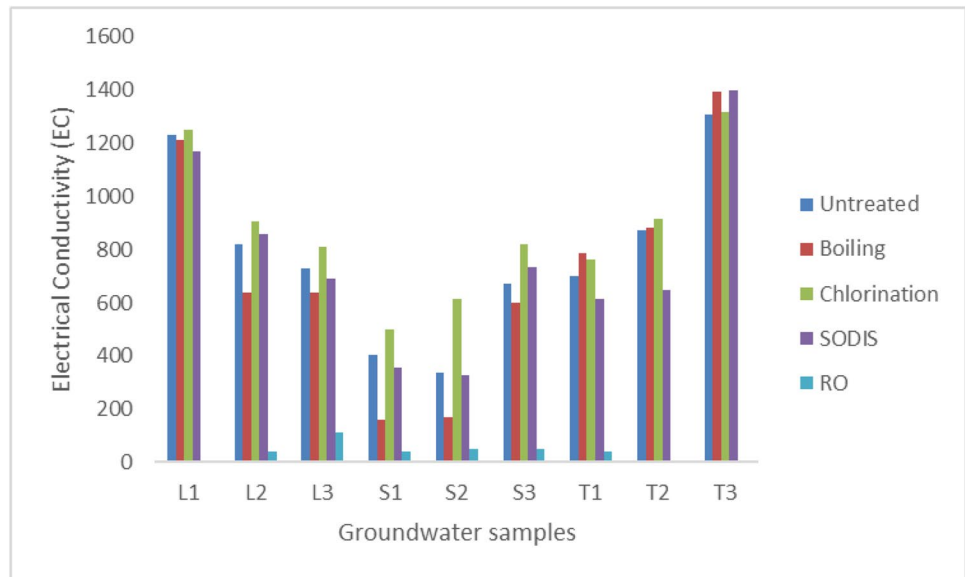
**Bacteriological analysis of water samples before and after treatment**

TC and FC count results in drinking groundwater samples before and after treatment are depicted in Tables 2 and 3, respectively. All study sites were found contaminated with bacteria. Among nine samples tested, all the groundwater samples showed TC count ranged between 11 and 48 with a mean value of 26.22±12.31, which was above the permissible limits of zero or nil as prescribed by WHO (2011). The

**Fig. 2** Efficiency of different household treatment practices for pH



**Fig. 3** Efficiency of different household treatment practices for EC



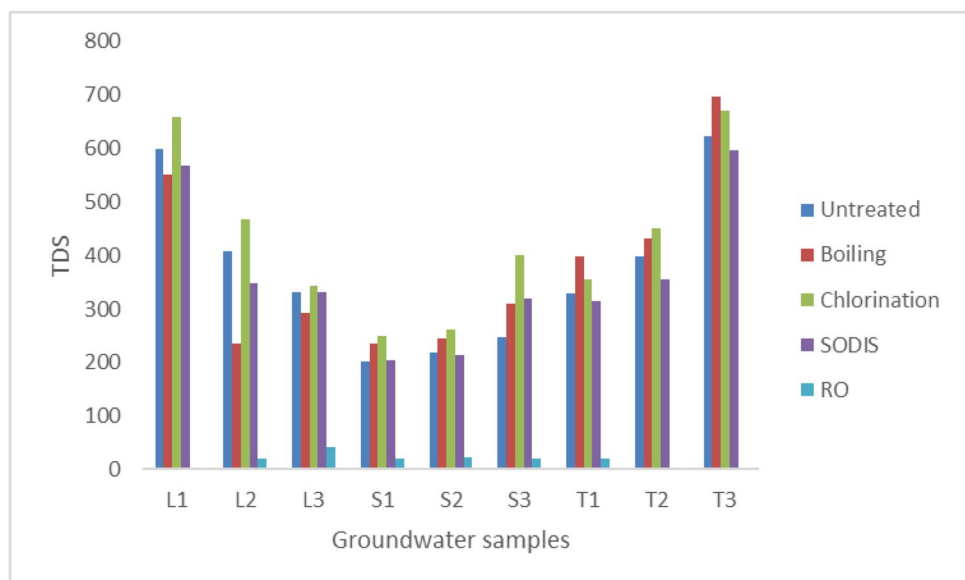
FC were comparatively observed in very less numbers and varied between 2–6.8 with a mean value of  $4.84 \pm 1.39$  and were also above the permissible limits (nil/100 ml of water) of WHO (2011) used for drinking purposes (Fig. 10). In general, no particular trend was observed between the water samples of different villages for physico-chemical as well as bacteriological parameters. However, in groundwater samples of Ladwa and Thanesar blocks, the coliform count was found more which may be either due to the improper sanitary conditions observed in the periphery of the households or may be due to underground leakage in septic tanks and from human or animal fecal matter contamination. In general, drinking water samples from households with animal

and solid wastes in their surroundings and poor sanitation practices were found to have more TC and FC count.

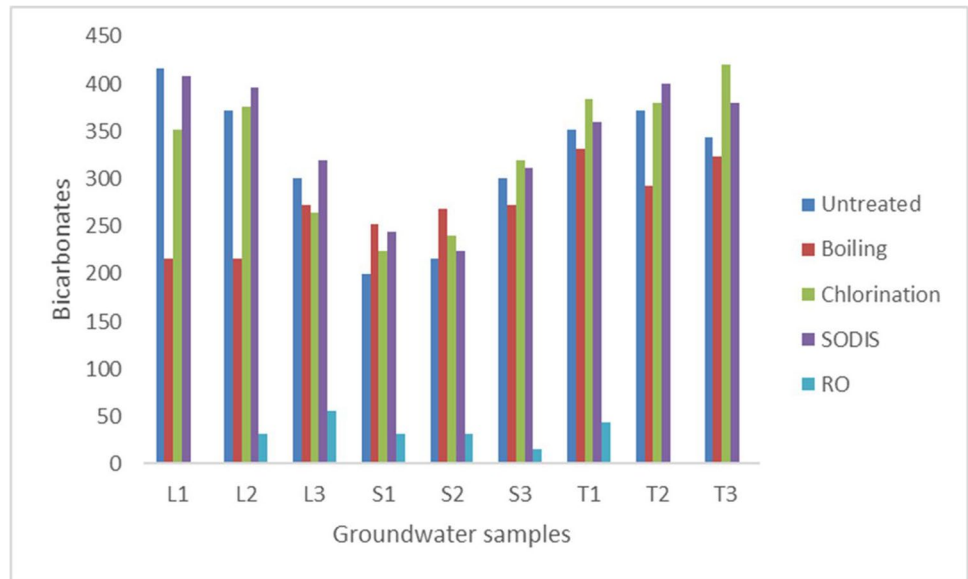
Various HWT practices used for total coliforms removal were found to be effective in order of RO  $\approx$  boiling  $>$  chlorination  $\approx$  SODIS. Both RO system and boiling were found to successfully remove about 87.0% of TC from the groundwater samples. However, chlorination and SODIS treatments removed 76.94% and 76.00% TC.

In the case of FC, all the treatment practices were shown to remove up to  $< 1.8$  to 2, which is considered almost nil as per MPN table APHA (2005); however, no specific trend was observed between different HWT practices. In a similar study by Clasen et al. (2008), the boiling method

**Fig. 4** Efficiency of different household treatment practices for TDS



**Fig. 5** Efficiency of different household treatment practices for bicarbonates



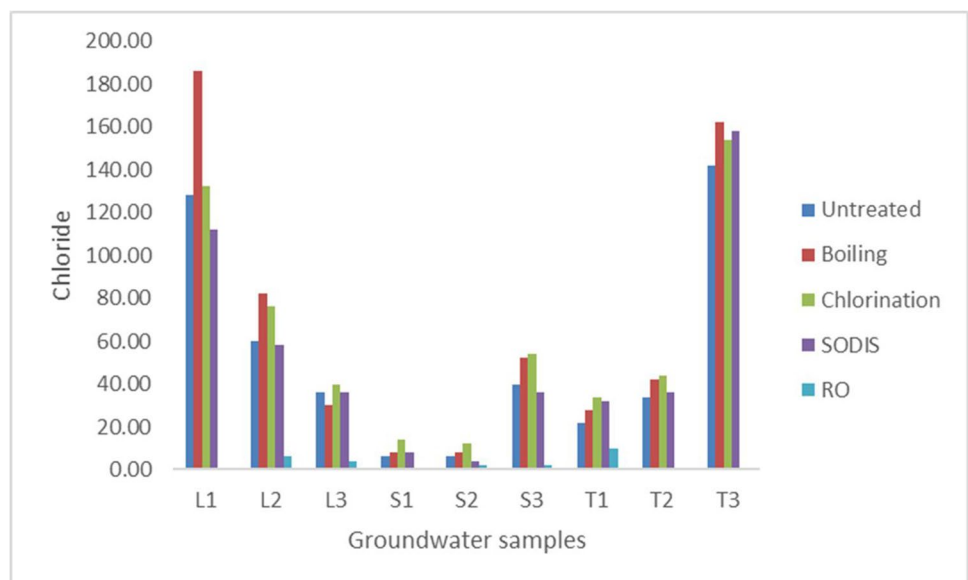
successfully reduced 99% of fecal coliforms from the Vasai and Nalasopara regions of India. The WHO (2016) in India reported the effectiveness of boiling followed by filtration, chlorination and SODIS treatment processes in removing viruses, bacteria, and protozoa from drinking water.

**Boiling treatment of water samples**

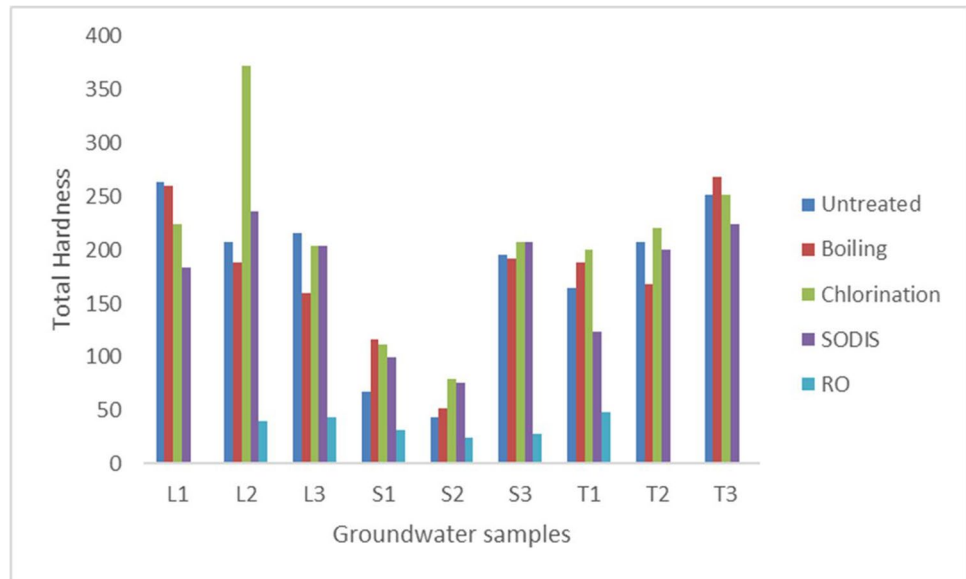
Following the present findings, heating water to a rolling boil is sufficient to inactivate or kill the pathogens (WHO, 2011). Boiling has reduced the concentration of most of the ions except chloride, magnesium, and TDS, which were found to be more in most of the samples after treatment. The increased concentration may be due to bicarbonate salts left after boiling, thereby increasing

TDS (WHO, 2004). Moreover, the non-carbonate hardness caused by the excess magnesium could not be removed by boiling (Sengupta 2013). Also, at acidic pH, the reaction between  $Cl^-$  and  $HO^{\cdot}$ , results in the formation of  $HOCl^-$  which promotes the production of  $Cl^{\cdot}$  (Jayson et al. 1973). In certain Asian countries, more than 90% of the population preferred boiling as a household drinking water treatment method (WHO, 2009). However, boiling is dependent on the availability of fuel required to achieve the boiling temperature up to 100 °C, which is added to its cost. In rural India, boiling water costs up to approximately US \$10.56 per annum (about 471.40 INR) by using liquid petroleum gas (LPG) gas for treating six liters of water per household/day (Firth et al. 2010). While wood biomass uses, the estimated annual cost of treating water

**Fig. 6** Efficiency of different household treatment practices for chloride



**Fig. 7** Efficiency of different household treatment practices for total hardness



through boiling per household in India remained only US \$1.66 (74.10 INR) per annum.

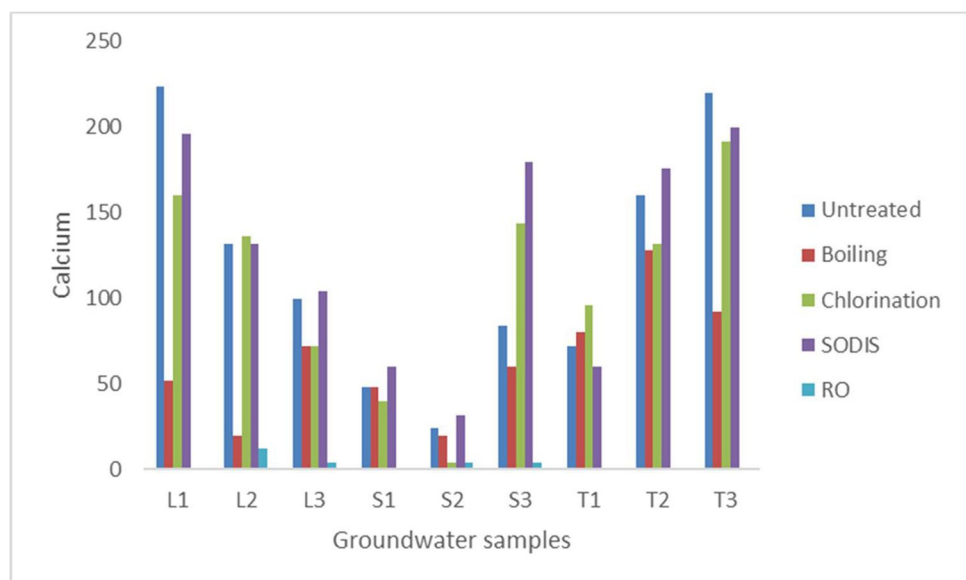
### Chlorination treatment of water samples

Chlorination is also much more effective in the removal of pathogens than physicochemical parameters, as investigated in the current study depicted in Table 2. The WHO standards (WHO, 2003) stated 2–3 mg/L and 5 mg/L as the minimum and maximum limits for adding chlorine in water, respectively, for satisfactory disinfection and to maintain the minimum residual chlorine. All parameters had shown a significant increase after chlorine disinfection except calcium ion. It may be due to the chemical nature of the chlorine, which adds a disadvantage to this method. After a retention

time of approximately 30 min, some byproducts formation occurred after various chemical reactions of chlorine with the organic substances naturally present in water. These compounds are known as disinfection byproducts (DBPs), such as trihalomethanes (carcinogenic) and halogenated acetic acids, which affect the chemical quality and somewhat physical properties like the taste and odor of water (Washington State Department of Health 2004). Some of the DBPs investigated as carcinogens and may cause an adverse impact on the reproductive and developmental health of humans (Chowdhury et al. 2009; Wei et al. 2010).

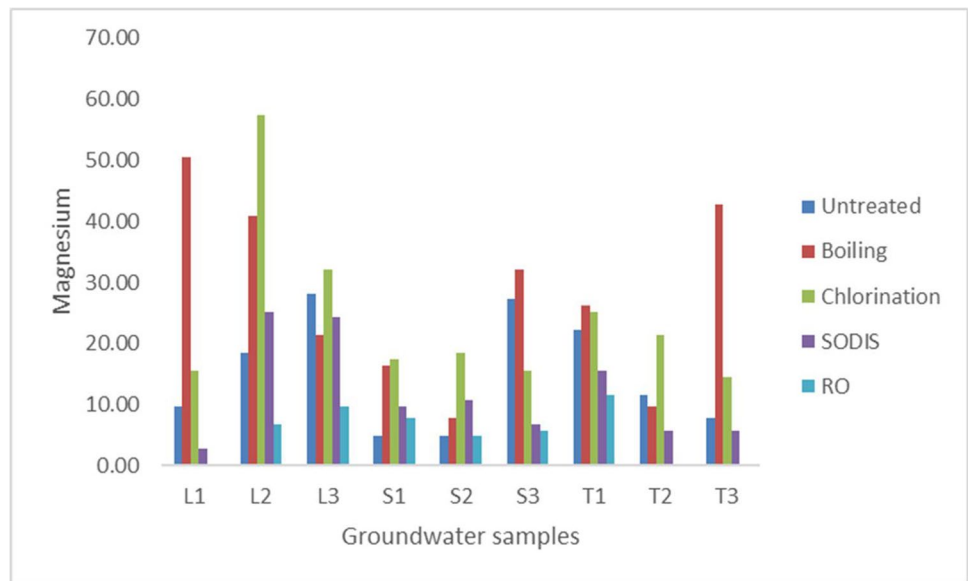
The chlorine left after complete disinfection and does not react with the water components is known as free or residual chlorine (RC) (0.3–0.5 mg/l) (Oram 2014a) which must be present in treated water to avoid the growth of pathogens

**Fig. 8** Efficiency of different household treatment practices for calcium





**Fig. 9** Efficiency of different household treatment practices for magnesium



(Bertelli et al. 2018). In the present study, RC concentration in treated water ranged from 1.42 to 4.25 mg/l and was nil in two samples, thereby not meeting the minimum criteria of 0.2–0.5 mg/l RC (WHO, 2011). It may be due to the maximum use of chlorine during the disinfection process. In Gwalior city (Madhya Pradesh, India), the average concentration of RC from all sampling locations of municipal piped water supply ranged from 0.08 to 0.98 mg/l, which was within the limits of WHO (Sharma and Rather 2015). While in rural Bangladesh, the average RC concentration was measured as 0.06 mg/l among the water samples of the

source pond, and all samples were below the WHO standard of 0.2–0.5 mg/l (Ahsan et al. 2017). Similarly, in piped distribution supply of France, RC was absent in 50% of the water samples (Bertelli et al. 2018). Hence, it is depicted from this study that the concentration of chlorine tablets added to treat water is not sufficient due to higher bacterial count and RC was not left behind in about 44% of samples. However, chlorination is a convenient method to treat a large amount of water if added to proper concentration. In India, chlorine tablets are distributed free of cost in the rural area of Haryana and can be a better option.

**Table 2** Total coliform count of groundwater samples before and after treatment

Block	Samples	Approx. depth of samples (meters)	Total coliforms in untreated and treated samples (MPN/100 ml)					WHO permissible limits (2011)
			Untreated samples	Boiling	Chlorination	SODIS	RO (Brand)*	
Ladwa	L1	68	17	2	5	4	—** (No RO)	Nil/100 ml of sample
Ladwa	L2	91	14	<2	4	4	<2 (Brand 1)	--do--
Ladwa	L3	76	23	4	7	6	4 (Brand 2)	--do--
Shahabad	S1	60	39	4	7	9	4 (Brand 3)	--do--
Shahabad	S2	76	17	2	4	4	2 (Brand 2)	--do--
Shahabad	S3	94	11	<2	4	5	2 (Brand 3)	--do--
Thanesar	T1	48	39	4	6	8	<2 (Brand 1)	--do--
Thanesar	T2	36	28	2	2	<2	— (No RO)	--do--
Thanesar	T3	41	48	6	9	9	— (No RO)	--do--
	Mean ± SD (Range)		(26.22 ± 12.31) 11–48	3.42 ± 1.39 (2–6)	5.33 ± 2.0 (2–9)	6.12 ± 2.08 (4–9)	3.42 ± 0.99 (<2–4)	Nil/100 ml of sample

Water samples are L1- Zainpur Jattan, L2- Braichpur, L3- Niwarsi, S1- Ramnagar, S2- Kishangarh, S3- Dhantori, T1- Adhon, T2- Kheri Markanda, T3- Bahadur Pura

\*Brand 1- RO+UV+UF, Brand 2- RO+UV, Brand 3- RO only, where, RO- Reverse osmosis, UV- Ultraviolet, UF- Ultrafiltration

\*\* Household did not have RO system

**Table 3** Fecal coliform count of groundwater samples before and after treatment

Block	Samples	Approx. depth of samples (meters)	Fecal coliforms in untreated and treated samples (MPN/100 ml)					WHO permissible limits (2011)
			Untreated samples	Boiling	Chlorination	SODIS	RO (Brand)*	
Ladwa	L1	68	4	< 1.8	< 1.8	2	—** (No RO)	Nil/100 ml of sample
Ladwa	L2	91	2	< 1.8	< 1.8	< 1.8	< 1.8 (Brand 1)	—do—
Ladwa	L3	76	4	< 1.8	2	2	< 1.8 (Brand 2)	—do—
Shahabad	S1	60	6.1	2	2	2	2 (Brand 3)	—do—
Shahabad	S2	76	4.5	2	< 1.8	2	< 1.8 (Brand 2)	—do—
Shahabad	S3	94	4	< 1.8	< 1.8	< 1.8	2 (Brand 3)	—do—
Thanesar	T1	48	6.8	< 1.8	2	2	< 1.8 (Brand 1)	—do—
Thanesar	T2	36	4	< 1.8	< 1.8	< 1.8	— (No RO)	—do—
Thanesar	T3	41	6.1	< 1.8	2	2	— (No RO)	—do—
	Mean ± SD (Range)		4.84 ± 1.39 (2–6.8)	2 ± 0 (< 1.8–2)	2 ± 0 (< 1.8–2)	2.33 ± 0 (< 1.8–2)	2 ± 0 (< 1.8–2)	Nil/100 ml of sample

Water samples are L1- Zainpur Jattan, L2- Braichpur, L3- Niwarsi, S1- Ramnagar, S2- Kishangarh, S3- Dhantori, T1- Adhon, T2- Kheri Mar-kanda, T3- Bahadur Pura

\*Brand 1- RO + UV + UF, Brand 2- RO + UV, Brand 3- RO only, where, RO- Reverse osmosis, UV- Ultra violet, UF- Ultra filtration

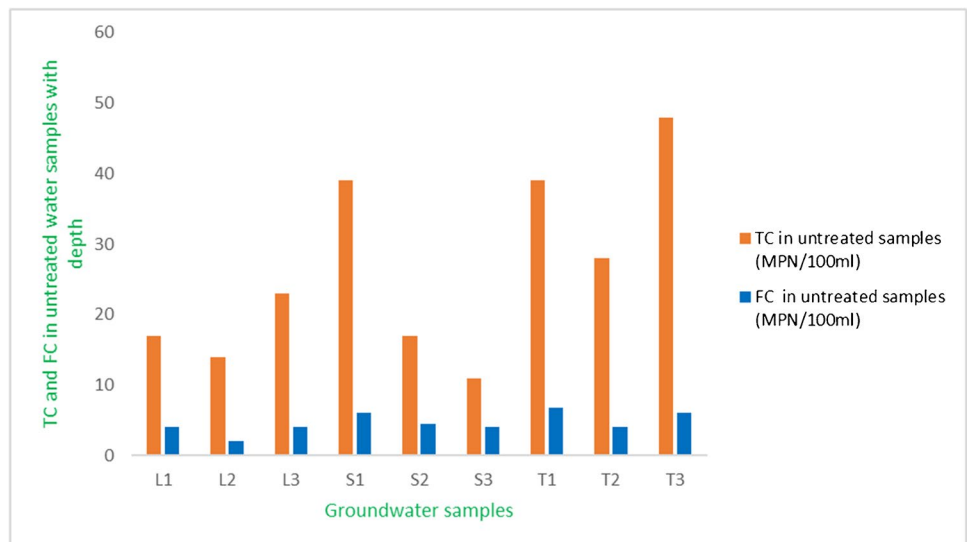
\*\* Household did not have RO system

### SODIS treatment of water samples

SODIS is a treatment method that treats raw water by solar energy (ultraviolet and infrared radiations). The duration of sun exposure for effective water treatment depends on factors such as season, cloud coverage, latitude, altitude, size of the bottle, and the turbidity of water. However, there is the risk of migration of plasticisers from the plastic bottle

into the water. Schmid et al. (2008) revealed maximum concentrations of 0.046 and 0.71 mg/L, respectively of di(2-ethylhexyl)phthalate (DEHP) and di(2-ethylhexyl)adipate (DEHA) plasticisers and were in the same range as reported in the commercially bottled water and hence safe to use. In the present study, SODIS has effectively reduced the EC and TDS, but the concentration of total hardness and cations and anions remained somewhat similar or unchanged

**Fig. 10** Presence of TC and FC in groundwater samples



**Table 4** Correlation matrix for physicochemical parameters and coliform counts between different treatments

	pH	EC	TDS	HCO <sub>3</sub>	Cl	TH	Ca	Mg	TC	FC
pH	1									
EC	.213	1								
TDS	.263	.968**	1							
HCO <sub>3</sub>	.408**	.791**	.822**	1						
Cl	.048	.878**	.861**	.509**	1					
TH	.236	.866**	.838**	.794**	.729**	1				
Ca	.128	.810**	.808**	.809**	.636**	.773**	1			
Mg	.210	.338*	.296	.225	.343*	.596**	-.048	1		
TC	-.002	.226	.208	.228	.096	.122	.281	-.163	1	
FC	.093	.058	.064	.163	-.040	-.047	.145	-.258	.880**	1

\*\* Correlation is significant at the 0.01 level (2-tailed)

\* Correlation is significant at the 0.05 level (2-tailed)

in some treated samples. It is a low-price technology for sunny regions where electricity is unavailable (Ray and Jain 2011). In India, about 40–75% reduction in estimated diarrheal cases has been observed among children after SODIS drinking water treatment (Rose et al. 2006; Rai et al. 2010). The major disadvantage is that a large amount of water cannot be treated at a time because it depends on the availability of PET bottles and the household's storage capacity. However, it is a cheaper method as compared to other HWT techniques because it is not dependent on electricity and fuel consumption.

### RO treatment of water samples

The RO process uses the reverse osmosis phenomenon and depends on the osmotic pressure difference between the pure water and the salt water, which removes the salts from water (Younos and Tulou 2005). RO system had reduced almost all ions to a very low level and maintained an optimum balance between cations and anions. This treatment process demineralizes water to a large level. Pre-treatment from RO can utilize various options, like microfiltration (MF), ultrafiltration (UF), and nanofiltration (NF). The basic difference in the membranes is based on the pore size, with NF being the smallest pore size and MF having the largest (Ray and Jain 2011). The RO systems generally remove heavy metals, phosphate, fluoride, nitrates, sodium, TDS, volatile organic compounds, and other pharmaceutical or agrochemical contaminants in a one-step procedure. Besides, this method also removes biological contaminants because of the very small pore sizes in the membranes without extra time or cost.

Although the RO method is very effective in treating water, there are some drawbacks like potentially high start-up prices, electricity requirements, effluent water handling, and the need to replace filters and membranes at a regular interval (Wimalawansa 2013). Another disadvantage of the

RO system is that it uses large amounts of water. About 75% or more of water is discarded along with the contaminants (Oram 2014b). Due to these reasons, three out of nine households in this study did not install RO system in their houses. Each of the two households was found to have different brands of RO systems like Brand 1 (RO + UV + UF), Brand 2 (RO + UV) and Brand 3 (RO only). All RO's were equally effective in treating bacteriological contaminants, but the system with brand 1 had reduced the coliforms up to <2 MPN/100 ml. It indicates that UF membrane is more functional in the removal of microbes as compared to other RO and UV systems. However, the estimated cost for RO treatment is based on the location (geography, quality of water and economic status of the households, etc.) and volume of water to be treated. In India, the cost of RO systems with the capacity of 10–15 L generally ranges from 8000 to 20,000 INR based on the functions available like simple reverse osmosis, UV, ultrafiltration, and taste enhancers etc. Based on electricity requirements, RO is the most expensive method, whereas boiling is dependent on fuel availability. Chlorination relies on the availability of chlorine tablets or solutions, while SODIS is weather dependent. Therefore, the adoption of the treatment method also depends on the accessibility of resources in rural settings.

In the current study, it was found that RO followed by boiling is the best household method to treat drinking water at home, which removes most of the microbes and EC and TDS to a large extent. In a similar study on water treatment practices conducted in an urban area of Burla in Odisha, India, and majority (63%) of the participants responded boiling to be the prime method for disinfecting the drinking water at home, accompanied by a membrane filter (13%) and chlorine tablets (6.8%) (Pradhan et al. 2018). Similar findings were reported in urban slums by Joshi et al. 2014 in Delhi and Beistline (2016) in Kolkata in India. In Gwalior city in Madhya Pradesh, India, boiling was used by 32% of households used, aqua guard (RO brand) by 18%

and 16.5% adopted alum treatment of municipal supplied drinking water (Sharma and Rather 2015). In India's rural population (in states namely West Bengal, Uttar Pradesh, Tamil Nadu, Rajasthan, and Andhra Pradesh), approx. 28.5% of people used boiling (traditional method), about 81% filtered by cloth or RO/Purifier (modern method) and about 15.2% had awareness regarding the addition of bleaching powder for water treatment (Rural marketing survey on safe water supply, 2012). The report further mentioned that most of the people preferred filtration with RO/local purifier or with simple cloth filtration and it was observed as the finest method for removing toxic contaminants from water. In a report from Zambia, a lower-middle-income country, HWTS was followed by only 34.9% of households and/out of which 25.9% households were using chlorination disinfection while 15.2% depended on boiling for the purification of their water (Rosa et al. 2016). Similar case studies were observed from semi-urban areas of Vasai and Nalasopara of India and peri-urban Cambodia (Clasen et al. 2008; Brown and Sobsey 2012). From the above discussion, boiling and filtration by RO are more practiced in rural areas than SODIS and chlorination.

### Statistical analysis

The significance between values of different treated parameters was calculated by one-way ANOVA and differences were considered significant at  $p < 0.05$ . The concentration of all the parameters was observed significant between different treatments except  $\text{HCO}_3$  ion. The values of different parameters after treatment were not observed significant along with different sites (Table 4).

The correlation analysis of parameters with different treatments is given in Table 4. It was found that EC and TDS were in high positive correlation with,  $\text{HCO}_3$ , Cl, TH, Ca, and with each other. Bicarbonate ion was positively correlated with TH and Ca. Total hardness was in positive correlation with Ca and Cl ions. A strong positive correlation was observed between TC and FC counts but they did not show any correlation with other physicochemical parameters (Table 4).

### Conclusions

All the analyzed rural groundwater samples were contaminated with the coliforms whereas except calcium ion, all physico-chemical parameters remained within the permissible limits. Various household treatment practices viz. boiling, chlorination, SODIS and RO were effective in reducing coliforms in groundwater samples. In comparison, boiling and RO were the most efficient treatment methods for microbe's removal from drinking groundwater. Further, boiling practice

is more preferred by the villagers because of its low cost as compared to RO systems. Also, it is not dependent on the availability of electricity which is a big problem in rural areas. After chlorination, chlorine residual should be checked regularly to know the proper working of the disinfection system. The use of the method depends on the availability of resources like fuel, electricity, chlorine tablets, weather, and the treatment method's cost per household. Therefore, the present study results depicted that all household methods are effective in treating drinking water and water should be treated at home before drinking to prevent various water-borne diseases, especially in rural settings.

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### Declarations

**Conflict of interest** The authors declare no competing interests.

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