

Enhancing bioactivity and stability of polymer-based material-tissue interface through coupling multiscale interfacial interactions with atomic-thin TiO₂ nanosheets

Rongchen Xu^{1,2,§}, Xiaodan Mu^{1,§}, Zunhan Hu^{3,§}, Chongzhi Jia¹, Zhenyu Yang⁴, Zhongliang Yang¹, Yiping Fan¹, Xiaoyu Wang^{1,5}, Yuefeng Wu⁶, Xiaotong Lu⁶, Jihua Chen⁴ (✉), Guolei Xiang⁶ (✉), and Hongbo Li¹ (✉)

¹ Department of Stomatology, The First Medical Center, Chinese PLA General Hospital, Beijing 100853, China

² Department of Stomatology, The Third Medical Center, Chinese PLA General Hospital, Beijing 100039, China

³ Department of Stomatology, Kunming Medical University, Kunming 650500, China

⁴ National Clinical Research Center for Oral Diseases & Shaanxi Key Laboratory of Stomatology, Department of Prosthodontics, School of Stomatology, The Fourth Military Medical University, Xi'an 710032, China

⁵ Department of Stomatology, The Strategic Support Force Medical Center, Beijing 100101, China

⁶ State Key Laboratory of Chemical Resource Engineering, College of Chemistry, Beijing University of Chemical Technology, Beijing 100029, China

[§] Rongchen Xu, Xiaodan Mu, and Zunhan Hu contributed equally to this work.

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ABSTRACT

Stable and bioactive material–tissue interface (MTF) basically determines the clinical applications of biomaterials in wound healing, sustained drug release, and tissue engineering. Although many inorganic nanomaterials have been widely explored to enhance the stability and bioactivity of polymer-based biomaterials, most are still restricted by their stability and biocompatibility. Here we demonstrate the enhanced bioactivity and stability of polymer-matrix bio-composite through coupling multiscale material–tissue interfacial interactions with atomically thin TiO₂ nanosheets. Resin modified with TiO₂ nanosheets displays improved mechanical properties, hydrophilicity, and stability. Also, we confirm that this resin can effectively stimulate the adhesion, proliferation, and differentiation into osteogenic and odontogenic lineages of human dental pulp stem cells using *in vitro* cell–resin interface model. TiO₂ nanosheets can also enhance the interaction between demineralized dentinal collagen and resin. Our results suggest an approach to effectively up-regulate the stability and bioactivity of MTFs by designing biocompatible materials at the sub-nanoscale.

KEYWORDS

material–tissue interface, TiO₂ nanosheets, pulpo-dentinal complex, biomaterial, resin composite

1 Introduction

Inorganic nanomaterials have been widely explored as structural reinforcements and functional components in bio-medical polymer-matrix composites (PMCs) for wound healing, sustained drug release, tissue engineering scaffold, etc. [1–4]. Such PMCs can protect tissues against mechanical forces (e.g., bite force, blood pressure, and friction) [5–8] and biochemical stimulations (e.g., bacterial, enzyme, and virus) [9–11], reduce infection risks and tissue degradation [12, 13], and provide mechanical supports [12–14]. Material–tissue interfaces (MTFs) between exogenous materials and bio-components (such as protein, collagen, and cell) can critically provide delivering channels for biological factors [15], protect collagen fibrils against degradation [16], and stimulate cellular responses like adhesion, proliferation and differentiation [17]. In general, material–tissue interactions (MTI) involve two scales of interfaces, atomic-level organic–inorganic interfaces in PMCs and microscale interfaces between PMCs and cells/tissues. Manipulating the physicochemical properties and bio-

functions of PMCs can basically optimize the properties of such multiscale interfaces to enhance MTIs.

Interfacial adhesion is the primary property affecting the stability, physicochemical properties, biocompatibility, and bio-function of biomedical materials [18, 19]. This interfacial property can be generally enhanced through increasing the strength and density of interfacial bonds [20], which can be further controlled through varying the sizes and surface areas of filler particles, surface modification, physical treatments, and introducing coupling agents [21–23]. For biomedical PMCs, size and dimension of nanomaterials can critically affect interfacial adhesion and physicochemical performances [24]. For example, inspired by the liner ordered microstructures of bone, one-dimensional (1D) nanostructures such as hydroxyapatite (HAP) nano fibers have been widely explored in bone tissue engineering [25, 26]. Over last decade, two-dimensional (2D) nanomaterials, such as graphene, black phosphorous, and layered double hydroxides (LDHs), have also been explored as a new type of reinforcements for biomedical PMCs, because their large specific

Address correspondence to Jihua Chen, jhchen@fmmu.edu.cn; Guolei Xiang, xianggl@mail.buct.edu.cn; Hongbo Li, hongbo_li@sina.com

surface areas can provide more bonding sites with organic matrix and tissues [27–29]. Moreover, their high surface reactivity can strongly graft functional groups to interact with specific target and perform specific biofunctions like antibiosis, anti-virus, and stimulate cell responses. However, the practical medical applications of most materials are still limited by their surface bonding activity, stability, and biotoxicity. Therefore, exploring biocompatible 2D nanomaterials with enhanced interfacial adhesion and bioactivity is still on demand for developing promising biomedical materials.

TiO₂ nanostructures have been explored as promising fillers to improve the mechanical strength and biological activity of PMCs owing to high chemical inertness and biocompatibility, and inexpensive costs [30–32]. However, their applications are still limited by stability in PMCs. Herein we demonstrate the enhanced performances of TiO₂ nanosheets (TiO₂ NSs) to PMCs in cell-nanomaterial and collagen-nanomaterial interfacial interactions by human pulpo-dental complex. The complexity and diversity of oral microbiota, the masticatory force, and the anatomical structure of pulpo-dental complex make the dentin–resin interface as a challenging system to explore the effects of MTI, and we find that TiO₂ NSs can improve the hydrophilicity and mechanical properties of adhesive resin (AR) and form stable hybrid resin matrix surface. Adhesive resin composites (ARCs) modified with such TiO₂ NSs can promote the adhesion, proliferation, and osteogenic and odontogenic differentiation capacity of human dental pulp stem cells (hDPSCs) from pulp. Furthermore, TiO₂ NSs also improve the bonding stability between exogenous adhesive resin composites and human demineralized dentin collagen, leading to more stable material–collagen hybrid layers (Fig. 1(a)).

2 Experimental section

TiO₂ NSs were prepared following a solvothermal method [33–35]. Then the TiO₂-ARCs were prepared through mechanical agitation for 10 min and ultrasonic blending for 10 min. Table 1 presents detailed formula. Degree of conversion (DC), elastic module, hardness, stability, hydrophilicity, and roughness of resin were evaluated. Then, intact non-carries human third molars were extracted and collected from the donors after signing informed

consent to extract hDPSCs and be prepared as the dental bonding model. The protocol was approved by the Institutional Review Board of the last author's institute (#IRB-S2021-655-01). After isolation and identification of hDPSCs, the effect of TiO₂-ARCs on cell adhesion, proliferation, and differentiation was evaluated using cell counting kit-8 (CCK-8), scanning electron microscopy (SEM) observation, staining, quantitative real-time polymerase chain reaction (RT-qPCR), and Western blot analysis (WB). Furthermore, the interaction stability between demineralized dentin collagen and TiO₂-ARCs was tested through the interface strength test, fracture mode analysis, interface observation, and nano-leakage. Detail procedure of all tests and the statistic strategies are listed in the Electronic Supplementary Material (ESM).

3 Results and discussion

3.1 Preparation of TiO₂ NSs and TiO₂-ARCs

TiO₂ NSs were prepared following a solvothermal method by hydrolyzing TiCl₄ in ethylene glycol (EG), in which EG acts as both solvent and capping ligand. Transmission electron microscopy (TEM) image shows soft 2D nanosheet structure (Fig. 1(b)). X-ray diffraction (XRD) pattern can be indexed to monoclinic TiO₂(B) (PDF card 74-1940, Fig. 1(c)). These results indicate high purities in both morphology and phase of the TiO₂ NS.

TiO₂-ARCs were prepared with TiO₂ nanostructures and adhesive resin matrix. The resin matrix phase composes of commonly used resins in dentistry [36], including bisphenol-A glycol dimethacrylate (Bis-GMA) and hydroxyethyl methacrylate (HEMA) (Fig. 1(d)). Camphorquinone (CQ) and ethyl 4-(di-methylamino) benzoate (EDMAB) were used as photoinitiation reagents. Resin matrixes without TiO₂ and filled with TiO₂ nanoparticles (NPs) were used as blank and positive control groups. All the adhesive resins can transform from liquid to solid under ultraviolet (UV) irradiation (Fig. 1(e)), yielding organic–inorganic surfaces. The pure resin is transparent, while the composites modified with TiO₂ nanostructures are opaque. TEM image shows that TiO₂ NSs were well dispersed in the composites (Fig. 1(f) and Fig. S1 in the ESM).

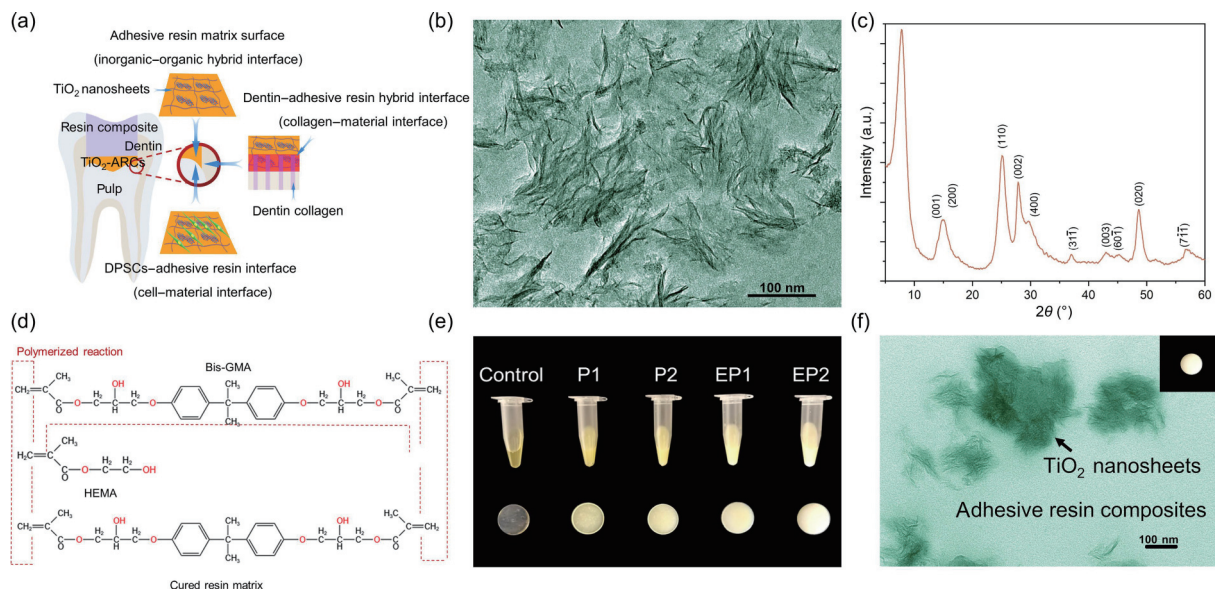


Figure 1 Composition and structure of TiO₂-ARCs. (a) Schematic diagram showing the multiscale interfaces in dental adhesive with TiO₂-ARCs. (b) TEM image and (c) XRD pattern of atomically thin TiO₂ NSs. (d) Monomers of adhesive resin composites. (e) ARCs images of different groups before and after polymerization. (f) TEM image showing the dispersion of TiO₂ NSs in ARCs.

Table 1 The formula of each tested adhesive resin composites

| Groups | Formula of each group (g) | | | | | |
|--------|---------------------------|------|------|-------|----------------------|----------------------|
| | Bis-GMA | HEMA | CQ | EDMAB | TiO ₂ NPs | TiO ₂ NSs |
| Ctrl | 6.6 | 3.3 | 0.05 | 0.05 | — | — |
| P1 | 6.6 | 3.3 | 0.05 | 0.05 | 0.025 | — |
| P2 | 6.6 | 3.3 | 0.05 | 0.05 | 0.05 | — |
| EP1 | 6.6 | 3.3 | 0.05 | 0.05 | — | 0.025 |
| EP2 | 6.6 | 3.3 | 0.05 | 0.05 | — | 0.05 |

3.2 Physicochemical properties of TiO₂-ARCs

We first studied the modification effects of TiO₂ NSs on the physicochemical properties of ARCs, which could basically regulate the stability and bioactivity of MTFs [37, 38]. The properties include DC, surface elastic module, hardness, thermal stability, water solubility (WSL), water sorption (WSp), surface wettability, and roughness [39–43].

DC represents liquid–solid phase transformation capacity of resin matrix [23], which is measured through attenuated total reflection-Fourier transform infrared (ATR-FTIR) spectrum. The values of all groups increase with light-curing time, and become stable within 80 s (Fig. 2(a) and Table S1 in the ESM). Addition of TiO₂ nanomaterials can lower DC values, which is inversely proportional to TiO₂ amounts. After light-curing for 80 s, experimental group 2 (EP2) shows the lowest DC value of 53.06 ± 0.52 (%) ($p < 0.05$). This trend results from the light-blocking effect of TiO₂, because TiO₂ can absorb and scatter UV lights in ARCs. TiO₂ NSs can more effectively reduce DC than TiO₂ NPs, suggesting TiO₂ NSs can more strongly inhibit polymerization. The higher specific surface area and reactivity of NSs may be the primary cause of this phenomenon. Polymerization of resin matrix is induced from the chain-growth reaction. After illumination, the free radicals of resin matrix are activated, then the electron transition and the rate of molecular bombardment are promoted, triggering chain reaction.

The addition of TiO₂ nanomaterials can enhance the mechanical strength of adhesive resin matrixes. At the same concentration, resin surface modified with TiO₂ NSs shows higher elastic module and hardness compared to TiO₂ NPs counterparts (Figs. 2(b) and 2(c), and Fig. S2 in the ESM). EP2 with 0.5 wt.% TiO₂ NSs shows the highest elastic module of 5.99 ± 0.29 GPa and hardness of 0.33 ± 0.02 GPa. TiO₂ NSs fillers insert into resin matrix molecular chains and compress their spaces, thus inhibiting their motions. Furthermore, the interactions between TiO₂ NSs and resin matrix molecules construct lock nodes, forming synergetic crosslinking interactions across multiscale interfaces, which can further enhance mechanical properties [44]. The interaction between TiO₂ NSs fillers and resin matrix molecular further leads to improved thermostability and lower solubility of the composites. Differential scanning calorimetry (DSC) curves show increased peak temperatures rise with the amounts of TiO₂ NSs. EP2 shows the highest peak temperature at 557.2 °C and the lowest WSL (Figs. 2(d) and 2(e)).

TiO₂ NSs can increase the hydrophilicity of resin surfaces, which is critical for bioactivity. The addition of TiO₂ nanostructures can increase water sorption rates and decrease contact angles of H₂O. The WSp of EP2 is significantly higher than others ($p < 0.05$) (Fig. 2(f)). The contact angles are 80.44° ± 0.46° for control group (Ctrl), 61.98° ± 1.36° for group 1 (P1), 57.60° ± 0.74° for group 2 (P2), 56.72° ± 0.68° for experimental group 1 (EP1), and 51.78° ± 0.76° for EP2 (Fig. 2(g) and Table S2 in the ESM). This is because TiO₂ nanostructures can increase the attraction interactions with H₂O through chemisorption and

physisorption [45]. In particular, the larger surface area of TiO₂ NSs can provide more interaction sites with water molecules to improve hydrophilicity. In addition, roughness can also regulate surface hydrophobicity [45, 46]. We find that all groups do not show significant differences in surface roughness (Fig. 2(h) and Fig. S3 in the ESM), thus TiO₂ nanomaterials do not effectively modify the homogeneity of resin matrixes. Therefore, the increased hydrophilicity mainly results from the interaction of TiO₂ with H₂O rather than roughness.

3.3 Regulation capacity of cell-resin interface to cell bio-behaviors

Resin matrix has been widely applied in dentistry and orthopedics as adhesive, pulp capping, and scaffold reagents [46–49]. We use the adhesive resin matrixes as stem cell niche to investigate the regulatory capacity of TiO₂ NSs on cell bioactivity. Considering the potential clinical applications, hDPSCs were extracted from non-carries intact human teeth and cultured on targeted resin surfaces under proliferative conditions. Primary and continuous cells with fibroblast-like morphology were observed under optical microscope (Figs. 3(a) and 3(b)). NF-H staining (Fig. 3(c)) and Oil Red O staining (Fig. 3(d)) results demonstrated that the cultured cells have potential for neurogenic and adipogenic differentiations. The positive expression of stromal cell antigen 1 (STRO-1) suggested that the cultured cells have stemness (Fig. 3(e)). Therefore, the isolated cells are hDPSCs with multi-differentiation potential.

3.3.1 Cell adhesion and proliferation

Cell adhesion and proliferation are significant for regulating cell fates like differentiation and apoptosis [50, 51]. We use CCK-8, cytoskeleton staining, SEM, and immunofluorescence staining to evaluate regulation effects of physicochemical and mechanical properties of substrate surfaces on cell adhesion and proliferation *in vitro*. CCK-8 results show that TiO₂ NSs can significantly improve hDPSCs proliferation on resin matrix surfaces (Fig. 3(f)), also illustrate that the TiO₂-ARCs exhibit biocompatibility. The ratios of living cells in all groups increase with incubation time. Figure 3(g) shows the F-actin of hDPSCs in EP2 was more organized and extended than that in other groups. These extended actin fibers suggest that targeted hDPSCs can exert traction force on the substrates, which can increase adhesion strength. Figure 3(h) shows hDPSCs cultured on TiO₂-ARCs of EP2 exhibit more spreading pseudopods than other groups to provide more binding sites with exogenous substrates. This further verifies the introduction of TiO₂ NSs can significantly promote the hDPSCs adhesion.

Living-cell staining further verifies the up-regulation capacity of TiO₂ NSs on hDPSCs proliferation and shows that all cells display the typical long-spindle morphology (Fig. 3(i)). The amount of cells adhered on EP2 is about two times of that on P2, suggesting that TiO₂ NSs can more effectively improve cell activity than TiO₂ NPs. Cells adhere onto material surfaces through the interaction

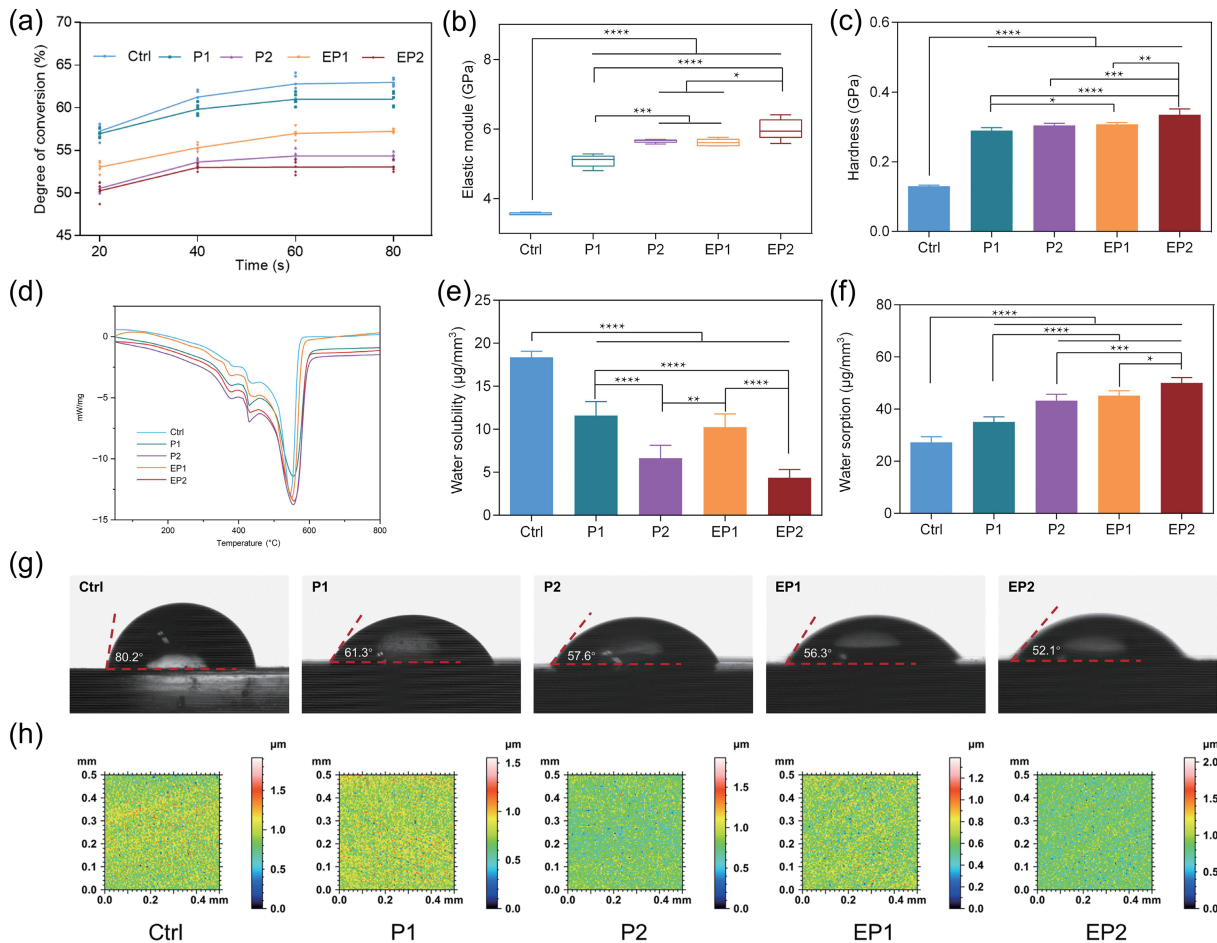


Figure 2 Physicochemical properties of TiO₂-ARCs. (a) Time-dependent degree of conversion of adhesive resins. (b) Elastic module, (c) surface hardness, (d) DSC curves, (e) water solubility, (f) water absorption, (g) water contact angles, and (h) two-dimensional optical profile of roughness of adhesive resins. Ctrl is blank control, and P1 and P2 are positive control with TiO₂ NPs. EP1 and EP2 are TiO₂-ARCs modified with TiO₂ NSs (mean ± SD; **p* < 0.05, ***p* < 0.01, ****p* < 0.001, and *****p* < 0.0001).

between adhesion receptor on cell membrane and adhesion ligands on material surfaces. When surface exhibits moderate hydrophilicity, adhesion related molecules and proteins are more easily recognized by receptors and activate the adhesion motifs [52]. Also, higher elastic module and hardness can up-regulate cell adhesion. Cells in adhesion processes can generate tractional forces with substrate surface. Surface with harder elastic module and hardness can resist the tractional forces and the bio-active structure such as protein will not detach and promote cell adhesion [52]. Our results are consistent with the previous reports that cell adhesion and proliferation can be improved when the surface exhibits moderate hydrophilicity and harder rigidity [52, 53].

Furthermore, the application strategy is another regulator. Different with our results, some reports have demonstrated that TiO₂ nanostructures have adverse effect on cell bioactivity fate [54–56]. They suggested that TiO₂ nanostructures might express cytotoxicity because they produce reactive oxygen species (ROS). Also, when TiO₂ nanostructure was absorbed on cell lipid bilayer, it might change the membrane permeability and lead to cell death. Herein, the *in vitro* results illustrated that TiO₂ nanostructures presented no toxicity, also demonstrated bioactivity when they were used as fillers.

3.3.2 Cell differentiation

Considering the natural bio-effect of hDPSCs and the potential clinical application of resin materials in dentistry, we evaluated the osteogenic/odontogenic differentiation capacity of hDPSCs *in*

vitro. The introduction of nano-TiO₂ can generate more alkaline phosphatase (ALP) and mineralize nodules at cell–resin interfaces (Figs. 4(a)–4(d)). These phenomena suggest that TiO₂ nanostructures can accelerate the induction of hDPSCs differentiation to osteogenic/odontogenic lineages. Related protein expression was convinced by WB test. Cell–resin interface modified with TiO₂ NSs can more strongly induce hDPSCs to differentiate to osteogenic/odontogenic lineage (Figs. 4(e) and 4(f)). We used RT-qPCR to further evaluate the gene expression related to osteogenic/odontogenic differentiation (Fig. 4(g)). Results showed hDPSCs cultured on resin surface of EP2 group exhibited significant gene upregulations on the 14th day (*p* < 0.05). Also, the degree of gene expression exhibited nano-TiO₂ concentration-dependent property.

The improved differentiation is highly related to enhanced mechanical properties of resin surfaces, including elastic module and hardness. It has been reported that mechanical strength of substrates' surface can regulate cell fates [57, 58]. The increase of surface mechanical strength can transfer the differentiation lineage of stem cells from neurogenic or adipogenic to osteogenic differentiation [40]. Also, the stretching and unfolding of the actin fibers mean that hDPSCs can expose more binding domains for the differentiation-related proteins or signaling molecules, such as Yes-associated protein (YAP), TAZ, FAK, and MAPK [59, 60]. These molecular sensors can regulate osteogenic/odontogenic differentiation through Hippo, ERK 1/2, and JNK MAPK pathways [59, 61]. Meanwhile, the increase of hDPSCs proliferation provides more candidates for differentiation and

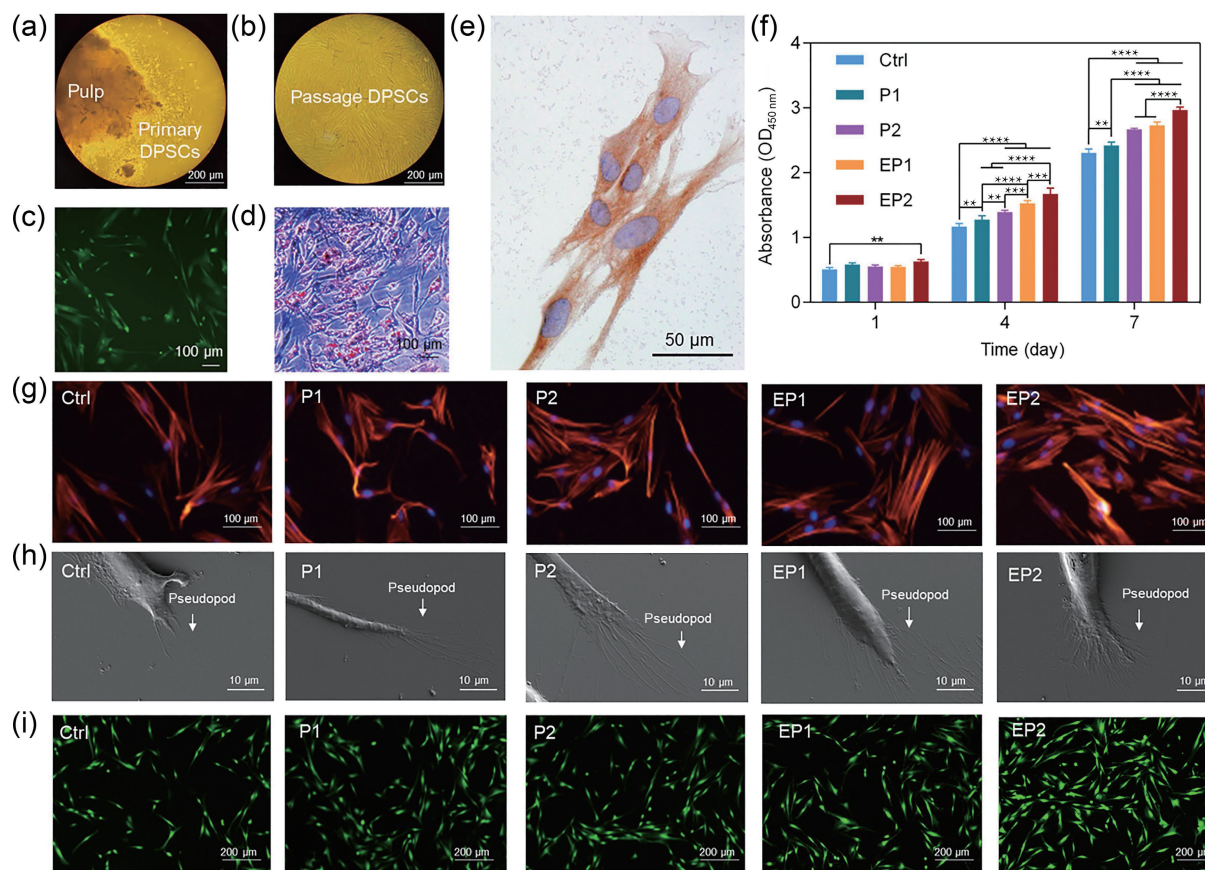


Figure 3 Regulation capacity of TiO₂-ARCs to adhesion and proliferation of hDPSCs. (a) The primary culture of hDPSCs. (b) The subculture culture of hDPSCs. (c) Immunofluorescence staining of NF-H. (d) Adipogenic differentiation of hDPSCs. (e) Immunohistochemical staining of STRO-1. (f) Proliferation of hDPSCs cultured on different resin discs by CCK-8. (g) Cytoskeleton staining of hDPSCs. (h) SEM observation of hDPSCs. (i) Cell viability of hDPSCs (mean \pm SD; ** $p < 0.01$, *** $p < 0.001$, and **** $p < 0.0001$).

makes more osteogenic/odontogenic genes and protein expressions in the unit space of resin surfaces. While, considering the potential application in clinic, further *in vivo* researches are needed to verify the bio-effect of TiO₂ NSs on MTI.

3.4 Regulation capacity of TiO₂-ARCs to human dentin–resin hybrid interface

To explore the effect of nanomaterial dimension on MTI stability in multiple bionic environments, we designed a dentin–resin interface model based on micro-mechanical interlocking theory [62]. Exogenous ARCs prepared as described earlier were used to construct the nano–bio interface. Nano demineralized dentin collagen framework was prepared by acid etching. Then, nano-resin penetrates into the demineralized collagen network to form hybrid layer and resin tags (RT) to physically contact with collagen fibrils. After photo-curing procedure, resin components transfer from liquid to solid, graft, and wrap onto collagen fibrils. The resin nanocomposite fills the collagen network space to yield MTF (Fig. 5(a)).

Interface stability is manifested by the integrity of interface structures and bonding strengths. We firstly quantify the immediate micro-tensile strength to test whether the addition of TiO₂ nanostructure could adjust interface immediate mechanical behaviors. The results show that the micro-tensile strengths of all groups are similar ($p > 0.05$) (Figs. 5(b) and 5(c)). This indicates that TiO₂ nanostructure cannot change the interface immediate strengths. We further adopted matrix collagenase aging and thermocycling aging to mimic *in vivo* environments to explore the potential effect of TiO₂ nanostructure on interface stability (Fig. 5(d)). We find that all groups present lower micro-tensile

strengths (Figs. 5(b) and 5(c), and Table S4 in the ESM). Strength of Ctrl separately dropped 52.4% and 49.4% after collagenase and thermocycling aging. While, with the introduction of TiO₂ nanostructure, the micro-tensile strength after aging increased. For example, EP2 exhibited the highest micro-tensile strength at 23.66 ± 2.31 MPa after collagenase aging and 27.35 ± 2.42 MPa after thermocycling, significantly higher than others ($p < 0.05$). Also, fracture mode analysis shows that the proportion of mixed fracture (with resin-dentin hybrid layer) decreases with the introduction of TiO₂ nanostructure at all conditions (Fig. 5(e) and Fig. S4 in the ESM), further indicating that NSs can improve the interface stability.

Compared to the control group, thicker hybrid interface forms after modifying adhesive resin with TiO₂ nanostructure (Fig. 5(f)). This means that the resin matrix modified with TiO₂ nanostructures has more bind sites with demineralized dentinal collagen which can increase the interface stability. Meanwhile, we used 50 wt.% ammoniacal silver nitrate solution [Ag(NH₃)₂NO₃(aq)] to evaluate the structural integrity of interface [63]. The silver ions can penetrate into the defect sites of interface. Figure 5(g) shows the silver deposition of EP2 is the lowest compared to other groups, suggesting that EP2 shows the best integrity of nano–bio interface.

Enhanced hydrophilicity of resin matrix is also considered as one important reason to improve resin–demineralized dentin collagen interaction. Collagen surface is composed with abundant hydrophilic groups such as –OH and –NH₂ [64]. Hence, the improvement of wettability means resin can more fully interact with demineralized collagen and wrap more collagen fibrils to increase micro-interlocking strength. Secondly, with the improvement of resin elastic module, average elastic module of

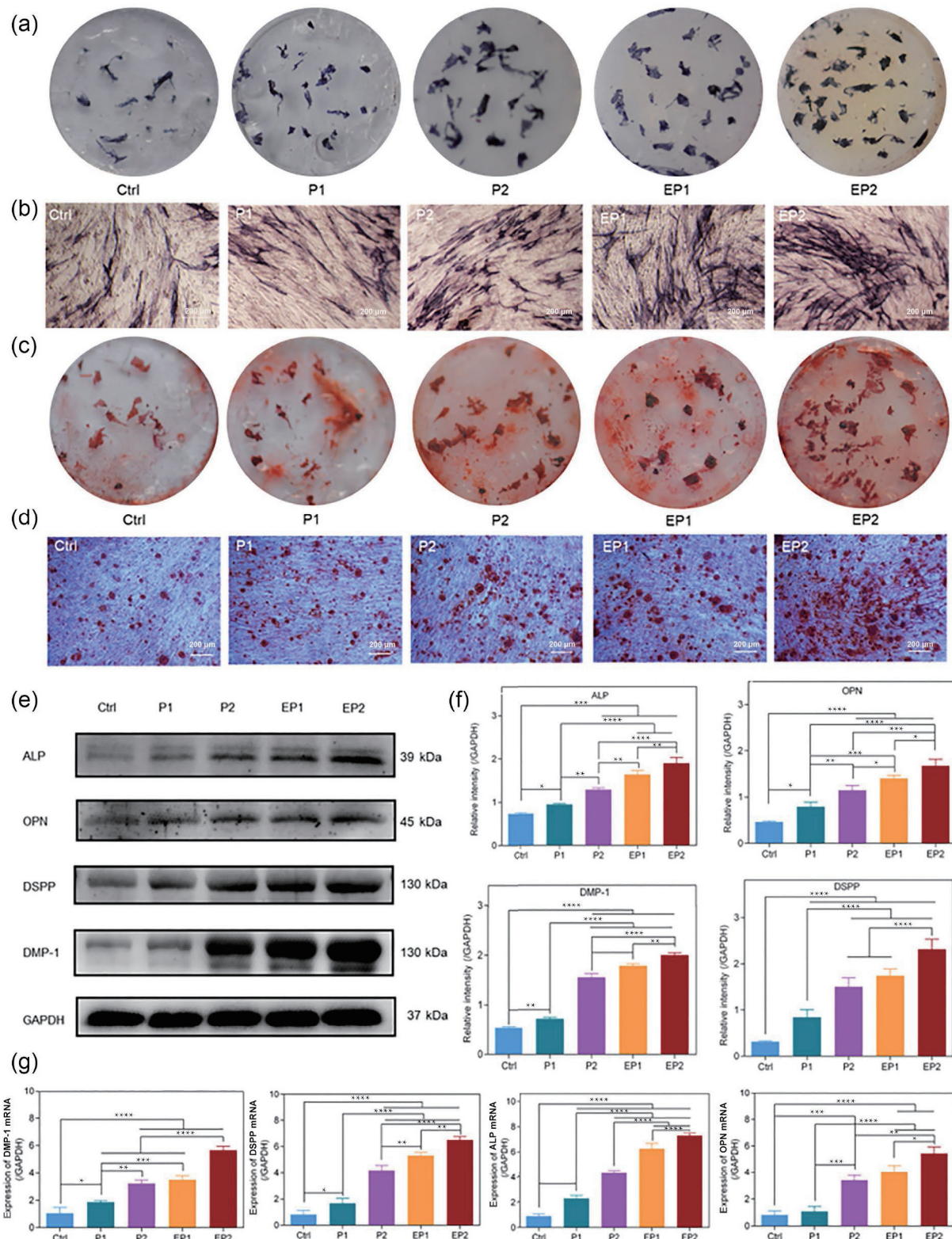


Figure 4 Differentiation capacity of hDPSCs regulated with TiO₂-ARCs. (a) The overall images of ALP staining. (b) The zoomed-in images of ALP staining. (c) The overall images of Alizarin red staining. (d) The zoomed-in images of Alizarin red staining. (e) WB of osteogenic/odontogenic-related proteins on the 14th day. (f) Quantitative assay of ALP, OPN, DMP-1, and DSPP protein expression on the 14th day. (g) Quantitative assay of osteogenic/odontogenic-related mRNA expression on the 14th day (mean ± SD; **p* < 0.05, ***p* < 0.01, ****p* < 0.001, and *****p* < 0.0001).

resin-dentin hybrid layer can improve. We have found that the enhanced elastic module of hybrid layer is benefit for interface stability [65].

Roughness can also regulate the interaction between collagen fibers and resins. With the increase of roughness, the micro-interlocking strength can be enhanced. However, it is also benefit for the adhesion of bacteria that can dissociate the collagen three-

helix structure, making the interface degradation [38]. Hence, the homeostasis of roughness after TiO₂ incorporation can also control the bacterial adhesion and has no adversely influence on the interface stability.

Worthwhile, it is still a contradiction whether the addition of nano materials could enhance the nano-bio interface stability. The incompatibility between nano materials and matrix may lead to

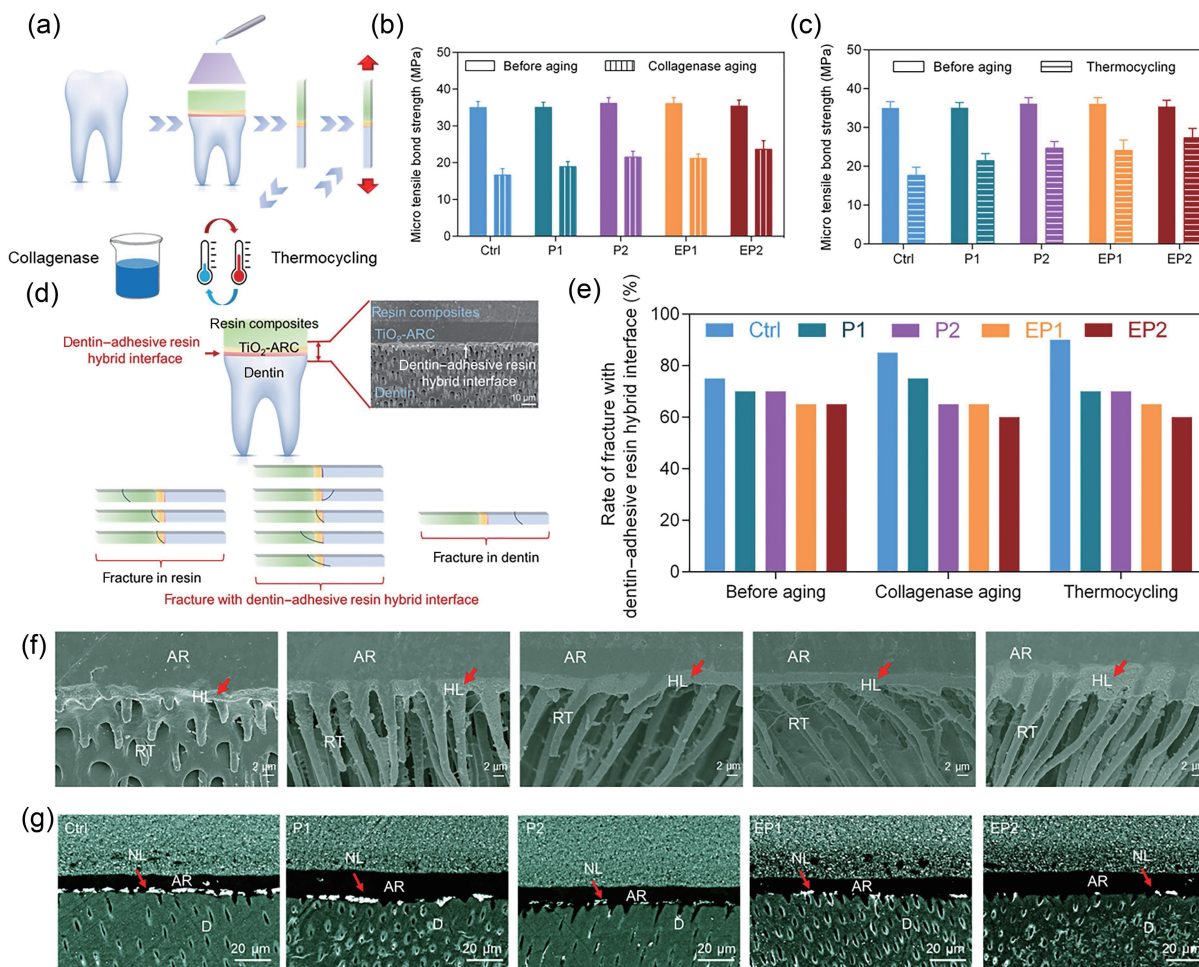


Figure 5 The effect of TiO₂-ARCs on dentin-resin hybrid interface stability. (a) Schematic diagram of construction of human dentin-resin hybrid interface. (b) Interface stability before and after collagenase aging. (c) Interface stability before and after thermocycling aging. (d) Schematic diagram of fracture mode. (e) Rate of fracture with dentin-adhesive resin hybrid interface before and after aging. (f) SEM of human dentin-adhesive resin hybrid interface (HL: dentin-adhesive resin hybrid interface). (g) Nanoleakage of dentin-adhesive resin hybrid interface (NL: nanoleakage at the bottom of hybrid interface; and D: dentin).

the existence of stress point and result in the uneven delivery of mechanical wave, while others demonstrate that the addition of void filler is significant for resin-dentin collagen bonding stability [66]. Our results are consistent with the latter. This suggests that TiO₂ NSs are useful to improve material-collagen interface and the results can be regarded as circumstantial evidence to prove the resin surface modified with TiO₂ NSs are homogeneous. The introduction of TiO₂ NSs can build stronger and more stable tissue-substrate hybrid layer.

4 Conclusions

In summary, we have demonstrated that the biomedical performances of polymer matrix composites can be effectively enhanced through tuning material-tissue interactions with TiO₂ NSs in sub-nanoscale. The key lies in exploring nanostructures that optimize the multiscale interfacial properties, including organic-inorganic hybridization interface, material-cell interface, and material-collagen interface. Our results show that atomically thin TiO₂ NSs are promising void fillers to improve the mechanical strength, hydrophilicity, stability, of resin matrixes. The material in sub-nanoscale can up-regulate the bioactivity of resin surface, provide a platform for hDPSCs adhesion, proliferation, and differentiation, and improve the interactions between resin matrix and dentinal collagen. This kind of nanocomposites has potential to improve the biological effect and interaction stability with bio-tissue of tissue engineering materials and drug delivery.

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Electronic Supplementary Material: Supplementary material (further details of fabrication and characterization of TiO₂ NSs and TiO₂-ARCs, the bioactivity evaluation of TiO₂-ARCs on hDPSCs, and the measurement of interaction with demineralized dentin collagen) is available in the online version of this article at <https://doi.org/10.1007/s12274-022-5153-1>.

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