Carbon Dioxide and Methane Emissions from Mangrove-Associated Waters of the Andaman Islands, Bay of Bengal

Neetha Linto • J. Barnes • Ramesh Ramachandran • Jennifer Divia • Purvaja Ramachandran • R. C. Upstill-Goddard

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Abstract We estimated CO₂ and CH₄ emissions from mangrove-associated waters of the Andaman Islands by sampling hourly over 24 h in two tidal mangrove creeks (Wright Myo; Kalighat) and during transects in contiguous shallow inshore waters, immediately following the northeast monsoons (dry season) and during the peak of the southwest monsoons (wet season) of 2005 and 2006. Tidal height correlated positively with dissolved O₂ and negatively with pCO₂, CH₄, total alkalinity (TAlk) and dissolved inorganic carbon (DIC), and pCO₂ and CH₄ were always highly supersaturated (330-1,627 % CO2; 339-26,930 % CH4). These data are consistent with a tidal pumping response to hydrostatic pressure change. There were no seasonal trends in dissolved CH₄ but pCO₂ was around twice as high during the 2005 wet season than at other times, in both the tidal surveys and the inshore transects. Fourfold higher turbidity during the wet season is consistent with elevated net benthic and/or water column heterotrophy via enhanced organic matter inputs from adjacent mangrove forest and/or the flushing of CO₂-enriched soil waters, which may explain these CO₂ data. TAlk/DIC relationships in the tidally pumped waters were most consistent with a diagenetic origin of CO₂ primarily via sulphate reduction, with additional inputs via aerobic respiration. A decrease with salinity for pCO₂, CH₄, TAlk and DIC during the inshore transects reflected offshore transport of tidally pumped waters. Estimated mean tidal creek emissions were ~23–173 mmol $m^{-2} day^{-1} CO_2$

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J. Barnes · R. C. Upstill-Goddard (🖂)

and ~0.11–0.47 mmol m⁻² day⁻¹ CH₄. The CO₂ emissions are typical of mangrove-associated waters globally, while the CH₄ emissions fall at the low end of the published range. Scaling to the creek open water area (2,700 km²) gave total annual creek water emissions ~3.6–9.2×10¹⁰ mol CO₂ and $3.7–34\times10^7$ mol CH₄. We estimated emissions from contiguous inshore waters at ~1.5×10¹¹ mol CO₂year⁻¹ and 2.6×10^8 mol CH₄year⁻¹, giving total emissions of ~1.9×10¹¹ mol CO₂year⁻¹ and ~3.0×10⁸ mol CH₄year⁻¹ from a total area of mangrove-influenced water of ~3×10⁴ km². Evaluating such emissions in a range of mangrove environments is important to resolving the greenhouse gas balance of mangrove ecosystems globally. Future such studies should be integral to wider quantitative process studies of the mangrove carbon balance.

Keywords Mangrove ecosystems \cdot Tidal pumping \cdot pCO₂ \cdot CH₄ \cdot O₂ \cdot Talk \cdot DIC \cdot Emission fluxes

Introduction

Continental shelves cover only 8 % of the ocean surface but they play a major role in marine biogeochemistry. Large nutrient fluxes from rivers and upwelling and high rates of organic matter remineralisation arising from benthic–pelagic coupling lead to much higher primary productivity than in the open ocean (Wollast 1998; Muller-Karger et al. 2005). Consequently, they are considered overall net sinks for tropospheric CO₂, equivalent to perhaps as much as 25–60 % of the open ocean CO₂ sink (Borges et al. 2005). Conversely, near-shore upwelling regions, estuaries, coral reefs and salt marsh and mangrove waters together constitute a net source to the atmosphere, not only of CO₂ but also of CH₄ (Borges et al. 2005; Bange 2006). The emissions are both high and

N. Linto · R. Ramachandran · J. Divia · P. Ramachandran Institute for Ocean Management, Anna University, Chennai, India

Ocean Research Group, School of Marine Science and Technology, Newcastle University, Newcastle Upon Tyne, UK e-mail: rob.goddard@ncl.ac.uk

uncertain, the uncertainty reflecting a regional data bias favouring studies in northern temperate locations.

Mangrove ecosystems cover $\sim 1.4 \times 10^5$ km² of the global intertidal area, equivalent to ~0.7 % of the total area of tropical forest (Giri et al. 2011). Nevertheless, they are among the world's most productive biomes, accounting for ~11 % of the total terrestrial carbon flux and ~10 % of the terrestrial dissolved organic carbon flux, to the oceans (Jennerjahn and Ittekot 2002; Dittmar et al. 2006). Their overall status is net autotrophic (Alongi 2002), mangrove plant biomass both above and belowground accounting for around 15 % of organic carbon sequestration by contemporary marine sediments (Jennerjahn and Ittekot 2002). Even so, mangrove sediments and their overlying waters are widely considered to be net heterotrophic (Gattuso et al. 1998), reflecting large and variable organic carbon inputs from diverse sources such as mangrove and terrestrial detritus, microphytobenthos, phytoplankton and sea grasses (Bouillon and Boschker 2006). Consequently, mangrove surrounding waters have been shown to emit large amounts of CO2 and CH₄ to the atmosphere (e.g., Borges et al. 2003; Barnes et al. 2006; Ramesh et al. 2007; Bouillon et al. 2007b; Bouillon et al. 2008; Koné and Borges 2008). Indeed, Chen and Borges (2009) estimate mangrove surrounding waters to emit around 0.05 Pg C annually as CO₂, about 9–18 % of the total coastal source (estuaries, salt marshes and mangroves) of 0.28- $0.57 \text{ Pg C year}^{-1}$ and Barnes et al. (2006) suggest that mangrove surrounding waters could be the dominant coastal CH₄ source. Given a net shelf CO₂ sink of $\sim 0.33-0.36$ Pg C year⁻¹ (Chen and Borges 2009) quantifying mangrove emissions of gaseous carbon is important for better constraining the carbon budgets of tropical continental margins.

This paper analyses the seasonal (southwest and northeast monsoons) distributions of pCO_2 , dissolved CH_4 and O_2 , total alkalinity (TAlk) and dissolved inorganic carbon (DIC) over 24-h cycles in two pristine mangrove creeks and in adjacent inshore surface water transects in the Andaman Islands, Bay of Bengal. In addition, annual CO_2 and CH_4 emissions were derived from these mangrove waters and using these and other data, the global carbon balance of mangrove ecosystems was briefly re-examined.

Materials and Methods

Study Area

The Andaman Islands are an archipelago of more than 550 islands (524 uninhabited) covering 6,408 km² of the southeastern Bay of Bengal and extending north-south for 360 km between latitudes $6-14^{\circ}$ N and $92-94^{\circ}$ E (Fig. 1). Subsoils derive mainly from argillaceous and algal limestone and climate is humid tropical with a relative humidity of 70– 90 %. Temperature is rather constant with minima of 24– 25 °C and maxima of 27–30 °C during all months; there is no discernable winter season (Singh et al. 1987). Mean annual precipitation is ~314 cm and monthly precipitation is maximal (~60 cm) at the peak of the south west monsoon (June– September) and minimal (<5 cm) between January and April (Singh et al. 1987). There is only one river, the Kalpong, which discharges to north-west North Andaman Island (Selvam 2003). Around 86 % of the total land area is covered by rainforest, some of which has been cleared for commercial plantation and agricultural use, leading to localised decreases in soil pH and organic content (Dagar et al. 1995).

Mangrove forest, which covers 892 km² (~0.6 % of the global total) and fringes around one quarter of the 1,900-km coastline, is among the world's most pristine. Mangrove species diversity and biomass is high, reflecting the high rainfall amount and frequency. Some 24 mangrove species are present, of which Rhizophora apiculata, Rhizophora mucronata and Ceriops tagal dominate (Selvam 2003). Around 255 km² of the mangrove forest is described as "dense" with the remainder listed as "moderately dense" or "open" (Forest Survey of India 2005). Mangrove clearance for agriculture, aquaculture and domestic and industrial development is significant and is increasing, causing soil erosion and the loss of organic matter and nutrients from some locations (Singh et al. 1986). Mangrove-associated open water totals at $\sim 2,700 \text{ km}^2$; this is around three times the area occupied by mangrove forest and is 2 % of the global estimate of 0.15×10^6 km² for open mangrove waters (Borges et al. 2005). The tidal range in Andaman open mangrove waters is 1.90 m (Selvam 2003), and the diurnal range in salinity is around 5-25 at the peak of the wet season (July-August) and 10-34 at the end of the dry season (March-April). Local times of sunrise and sunset are around 05.10 and 17.30 hours respectively, throughout the year.

Sampling and Analytical Techniques

Surveys of 24-h duration were conducted, commencing at midday (around 7 h after sunrise) with hourly surface water sampling for pH, TAlk, dissolved CH₄ and O₂, turbidity, salinity and temperature, at fixed locations in two tidal mangrove creeks: Wright Myo ($11^{\circ}47'27.7''$ N, $92^{\circ}42'24.3''$ E), South Andaman, which is 8 km long and 20–40 m wide dependant on tidal state and Kalighat ($13^{\circ}07'30.7''$ N; $92^{\circ}56'$ 48.2" E), North Andaman, which is 10 km long and has a mean width of ~100 m (Fig. 1). Wright Myo was surveyed at the beginning of the inter-monsoon period immediately following the northeast monsoon, on 14–15 April 2005 and 21–22 April 2006 ("dry season") and during the peak of the southwest monsoon, on 22–23 August 2005 and 24–25 August 2006 ("wet season"). During 2004, this site was surveyed similarly for CH₄ and N₂O, as reported elsewhere (Barnes et al. 2006).



Fig. 1 Locations of Kalighat and Wright Myo tidal surveys and the surface water transects: *TI* Kalighat to Mayabander, *TII* Mayabander to Interview Island via Austin creek, *TIII* Wright Myo to the Andaman Sea and *TIV* Wandoor National Park to Chidiyatapu

Kalighat was surveyed on 11–12 April 2005 (inter-monsoon: "dry season") and on 17–18 August 2005 (southwest monsoon: "wet season"). Additionally, wet and dry season surface water transects were carried out for pH, TAlk and CH₄ across ambient salinity ranges between mangroves and adjacent shallow near shore waters: TI and TII during April and August 2005 and TIII and TIV during both April and August 2005 and April and August 2006 (Fig. 1). TI covered ~18 km between Kalighat Creek (water depth ~1 m) and Mayabander (water depth ~10 m), TII covered ~28 km from Mayabander to Interview Island (maximum water depths ~5–10 m) via Austin creek (coral reef; maximum water depth ~20 m), TIII was a 9-km survey along and just offshore of Wright Myo (maximum water depth ~3-5 m) and TIV covered ~47 km from Wandoor National Park to Chipiyatapu (open water coral reef and fringing mangroves; maximum water depth ~ 15 m), through a channel connecting the Bay of Bengal to the Andaman Sea (Fig. 1).

Water samples were collected manually from 0.2 m depth (Richter & Wiese 2.5-L water sampler). Subsamples for dissolved CH₄ analysis were transferred into 100 mL glass bottles via silicon tubing with the careful exclusion of air bubbles, poisoned with 0.1 mL saturated HgCl₂ and sealed to leave no headspace. Analysis of CH₄ was by single-phase equilibration gas chromatography (GC) with a routine analytical precision (1 σ) of ±1 % (Upstill-Goddard et al. 1996). Method calibration was with certified gas standards of 6.74 and 12.5 ppmv CH₄ in N₂ (National Physical Laboratory,

New Delhi, India). Considering logistical constraints, pH was not determined in situ. Instead, pH and TAlk subsamples were decanted into 100 mL pre-combusted glass bottles, poisoned with 0.1 mL of saturated HgCl₂ solution and sealed. Following transfer to the laboratory, they were stored in a refrigerator at 4 °C and subsequently allowed to equilibrate to room temperature (25 °C) prior to analysis. Corrections to in situ temperature were made as appropriate. The total time between sampling and analysis was always less than 4 days. TAlk was determined by electro-titration (Gran 1952), and pH was measured with a combination electrode (ORION) calibrated with NBS standard pH buffers. Typical reproducibility (1 σ) of TAlk and pH measurements was, respectively, $\pm 4 \ \mu eq \ kg^{-1}$ and ± 0.005 . CO2Sys software (Pierrot et al. 2006) was used to calculate pCO₂ from pH, TAlk, temperature and salinity, adopting a protocol for waters of varying salinity (Frankignoulle and Borges 2001) and using the dissociation constants of Millero et al. (2006). On samples from three of the tidal surveys (Kalighat and Wright Myo, August 2005; Wright Myo, April 2006), cross-checking was made on these pCO₂ estimates by incorporating a methaniser (nickel catalyst at 250 °C in H₂) in the GC system to convert CO₂ to CH₄. Linear correlation of the data yielded: pCO₂ (GC)=0.936 pCO₂ (pH-TAlk)+397.8 (r^2 =0.941, n=72). The offset is primarily a result of progressive divergence in the liquid junction potential between buffer and sample towards higher salinities (Frankignoulle and Borges 2001). Consequently, its effect is minimal at high pCO₂ and low salinity, typically being around 0.25 % at pCO2>6,000 µatm but progressively increasing to around 3.6 % at pCO₂=3,000 µatm and around 13 % at pCO2<2,000 µatm. Nevertheless, the likely range in pCO₂ was considered to be sufficiently wide so as to preclude these inherent errors from materially affecting our subsequent data interpretation. For consistency, only those survey pCO₂ data derived from pH and TAlk are discussed in this paper as these are the only complete pCO₂ data set for our samples. Salinity, temperature, turbidity and dissolved O2 were determined in situ with a pre-calibrated multi-parameter probe (Horiba W-22.23). Precisions (1σ) were ± 0.01 salinity, ± 0.2 °C, ± 3 % turbidity and ± 0.01 mg O₂L⁻¹. Probe measurements of dissolved O2 were routinely cross-checked by collecting two samples in every five for standard Winkler titration. The two methods gave satisfactory agreement: O₂ $(\text{probe})=0.992 \text{ O}_2 \text{ (Winkler)}-0.02 \text{ } (r^2=0.959, n=61). \text{ Wind}$ speed was recorded with a hand-held cup anemometer at 1 m height (Lutron LM8000; resolution $\pm 0.1 \text{ m s}^{-1}$); 1 min means were subsequently converted to equivalent 10 m values (Amorocho and DeVries 1980). Water depths and mid-depth water current velocities were measured hourly (open channel flow meter, Valeport, UK) for all tidal surveys. In addition, gas bubbles effusing from Wright Myo creek sediments on 21 April 2006 were collected. Samples were captured over a 2-h period spanning the first low tide (1700-1900 hours), in a gas-tight syringe with an integral luer needle connected to an inverted glass funnel with a septum at the base of its stem. Samples were transferred to gas-tight vacutainers for subsequent gas chromatographic analysis of CO_2 and CH_4 as described above.

Sea-to-air emission fluxes (F) of CO₂ and CH₄ were estimated using $F = k_{\rm w} L \Delta p$, where $k_{\rm w}$ is the gas transfer velocity (in centimetres per hour), L is the Ostwald solubility coefficient and Δp is the partial pressure difference (in atmospheres) across the air-water interface. Values of L were from Weiss (1974) for CO₂ and Weisenburg and Guinasso (1979) for CH₄. k_w was estimated from the 10-m-derived wind speeds using the empirical k_w —wind speed relations of Clark et al. (1995) and Borges et al. (2004) for CO₂. Both were derived for use in shallow coastal waters where tidal current interaction with the seabed generates turbulence additional to that due to wind speed alone; the latter formally includes expressions for tidal current velocities and water depth, which is considered to be more applicable here than some other more commonly used relations (e.g., Liss and Merlivat 1986; Wanninkhof 1992). k_w for CO₂ was scaled to $k_{\rm w}$ for CH₄ using Schmidt numbers given in Wanninkhof (1992).

Results and Discussion

Creek Water Carbon During Tidal Surveys

Means and ranges of hourly pCO₂, CH₄, TAlk, DIC, O₂, salinity and turbidity are given in Table 1. Mean CH₄ did not differ significantly between dry and subsequent wet season surveys or in the case of Wright Myo, between consecutive years (mean of all samples, 349±125 nmol L⁻¹; 15,960±5,930 % saturation). Similarly, there was no significant seasonal variation in pCO₂ at Wright Myo (mean of all samples, 3,409±919 µatm; 903±243 % saturation). Previously, no seasonal variability in either CH₄ or N₂O at Wright Myo (Barnes et al. 2006) was found to be consistent with a negligible thermal effect on methanogenesis or nitrification/denitrification rates. Mean water temperatures were: dry season, 28.0±1.0 °C (range, 25.5-30.2 °C); wet season, 27.6±0.9 °C (range, 26.0-29.4 °C). By contrast, pCO₂ at Kalighat was around twice as high in the wet season than in the dry season and dissolved O₂ saturations were correspondingly lower (Table 1). These data are consistent with a wet season increase in net benthic and/or water column heterotrophy at Kalighat, perhaps reflecting enhanced organic matter inputs from adjacent mangrove forest and/or the flushing of soil waters enriched in CO₂ by respiration, as inferred previously for mangrove surrounding waters (Koné and Borges 2008). The observed three- to fourfold higher turbidity during the wet seasons (Table 1) is consistent with an enhanced organic matter flux and would

Table 1 Means	(all: <i>n</i> =24)	and ranges (in brack	ets) of biogeochem	nical variables me	sasured hourly in And	laman mangrove c	reek waters			
Site/date	Season	pCO ₂ (µatm)	pCO ₂ (% saturation)	CH_4 (nmol L^{-1})	CH ₄ (% saturation)	TAlk (mmol L^{-1})	DIC (mmol L ⁻¹)	O_2 (% saturation)	Salinity	Turbidity (mg L^{-1})
Kalighat										
April 2005	Dry	$\begin{array}{c} 2,180{\pm}382 \\ (1,704{-}3,392) \end{array}$	574 ± 101 (448–893)	421 ± 106 (154-603)	$9,283\pm4,248$ (7,956–26,419)	2.61 ± 0.15 (2.41–2.81)	2.58 ± 0.16 (2.32-2.80)	75±8 (60−92)	19.0 ± 8.7 (8-34)	42.0 ± 8.1 (31–63)
August 2005	Wet	$\begin{array}{c} 4,145\pm1,038\\ (2,970{-}6,140)\end{array}$	$_{(782-1,616)}^{1,090\pm273}$	340 ± 155 (30-577)	$14,490\pm6,434$ (1,402–24,638)	2.47 ± 0.11 (2.16-2.66)	$\begin{array}{c} 2.58 {\pm} 0.59 \\ (2.23 {-} 2.82) \end{array}$	51 ± 13 (16-75)	4.3 ± 2.5 (1.2-10.2)	133 ± 129 (26-453)
Wright Myo										
April 2005	Dry	$2,870\pm684$ (1,246-4,217)	755 ± 180 (647 $-1,010$)	414 ± 118 (198–585)	$\begin{array}{c} 19,624\pm5,223\\ (9,813-26,930)\end{array}$	2.25 ± 0.15 (1.98–2.49)	2.32 ± 0.15 (1.98–2.50)	44±5 (38−57)	25.0±5.0 (17−34)	56 ± 29 (18-134)
April 2006	Dry	$3,671\pm796$ (2,720-6,004)	990 ± 209 (716 $-1,580$)	284 ± 108 (146-583)	$15,683\pm5,991$ (8,059–30,636)	$\begin{array}{c} 2.54{\pm}0.04 \\ (2.41{-}2.60) \end{array}$	$\begin{array}{c} 2.70 {\pm} 0.61 \\ (2.52 {-} 2.63) \end{array}$	69 ± 11 (42-86)	29.7±0.3 (29.1–30.2)	67 ± 20 (24-109)
August 2005	Wet	$3,366\pm821$ (1,500-5,072)	886±216 (395–1,335)	357±113 (168–583)	$\begin{array}{c} 17,156\pm6,572 \\ (8,449-26,572) \end{array}$	0.90 ± 0.38 (0.37-2.09)	$\begin{array}{c} 0.86 {\pm} 0.37 \\ (0.84 {-} 2.10) \end{array}$	52±17 (23–94)	8.4 ± 7.32 (0.8–28.8)	219±118 (102-568)
August 2006	Wet	$3,686\pm1,025$ (1,930–5,490)	970±270 (508−1,445)	283±75 (147–418)	$15,644\pm4,170 (7,290-23,286)$	$1.75 {\pm} 0.19 \\ (1.45 {-} 2.02)$	$1.84{\pm}0.17 \\ (1.45{-}2.02)$	51±11 (30-72)	14.7 ± 4.5 (7.3–21.0)	263 ± 106 (97-428)

also tend to suppress photosynthesis, consistent with low dissolved O_2 during the wet seasons (Table 1). Zhai et al. (2005) similarly found lowest dissolved O_2 during the Pearl Estuary wet season, which was ascribed to elevated heterotrophy, both in contiguous mangrove sediments and their surrounding waters. The lack of seasonal contrast at Wright Myo implies that the effects of any such processes on CO_2 dynamics are insignificant there.

Figures 2, 3 and 4 show temporal trends in carbon system components, dissolved O₂ saturation, salinity and tidal height at Kalighat and Wright Myo. In all surveys tidal height was correlated positively with dissolved O₂ and negatively with pCO₂, CH₄, TAlk and DIC, notwithstanding the correlations for TAlk and DIC being comparatively weak during the Kalighat wet season (Fig. 2). For most surveys the coincidence of maximal pCO₂ and minimal O₂ only broadly matched light-dark periods of photosynthesisrespiration. Most notably, during the Kalighat dry season the data were around 12 h out of phase, maximal O_2 and minimal pCO_2 occurring just prior to midnight (Fig. 2). Overall, the data are consistent with a dominance of tidal pumping, in which mangrove sediment pore waters depleted in O₂ and enriched in the by-products of organic matter oxidation (CO₂, CH₄ and TAlk), seep into creek waters in response to the cyclic rise and fall in hydrostatic pressure (Ovalle et al. 1990), thereby obscuring quasi-coincident cycles in pCO₂ and O₂ driven by photosynthesisrespiration. We previously observed tidal pumping dominance of the diel cycles of CH₄ and N₂O at Wright Myo (Barnes et al. 2006) and Lara and Dittmar (1999) observed tidal pumping of nutrients in Brazilian mangrove waters. Tidal pumping of DIC is also well established in mangrove creeks (Borges et al. 2003; Kristensen et al. 2008; Zablocki et al. 2011; Maher et al. 2013) and has been shown to dominate resulting pCO₂ distributions and CO₂ emissions to air (Borges et al. 2003; Bouillon et al. 2007c; Koné and Borges 2008). The dominance of tidal pumping in the mangrove CO₂ budget was recently demonstrated unequivocally through correlations of pCO₂ with ²²²Rn, a natural tracer of dissolved pore water inputs (Santos et al. 2012; Maher et al. 2013). Hydrostatic pressure control of mangrove creek biogeochemical variables is thus now firmly established.

At Kalighat, the inverse relationship between CH_4 and tidal height was partially disrupted by heavy rainfall during the first 4 hours of the wet season survey; dissolved CH_4 was suppressed at low water (Fig. 2). This effect was also observed in an earlier study at Wright Myo and ascribed to the suppression of methanogenesis by O₂ penetration of the sediment in the rain drops and by additional dilution with this freshwater flux (Barnes et al. 2006). This mechanism could also enhance the potential for aerobic respiration, consistent with the higher pCO₂ observed. At Kalighat Fig. 2 Tidal variation of water height (*filled circles*), salinity (*open triangles*), pCO_2 (*open* squares), dissolved CH₄ (*open* diamonds), percent O₂ saturation (*open circles*), DIC (*open* squares) and TAlk (*open circles*) at Kalighat Creek during: **a** April 2005 (dry season) and **b** August 2005 (wet season). Shaded areas correspond to hours of darkness. Tidal heights are all relative to the mean of the lowest and highest tidal elevations observed during the survey



 pCO_2 was also higher during nighttime tidal minima than during daytime tidal minima, by ~20 % during the dry season survey and ~14 % during the wet season survey

Fig. 3 Tidal variation of water height (filled circles), salinity (open triangles), pCO₂ (open squares), dissolved CH4 (open *diamonds*), percent O₂ saturation (open circles), DIC (open squares) and TAlk (open circles) at Wright Myo creek during: a April 2005 (dry season) and b August 2005 (wet season). Shaded areas correspond to hours of darkness. Tidal heights are all relative to the mean of the lowest and highest tidal elevations observed during the survey



Fig. 4 Tidal variation of water height (filled circles), salinity (open triangles), pCO₂ (open squares), dissolved CH₄ (open diamonds), percent O₂ saturation (open circles), DIC (open squares) and TAlk (open circles) at Wright Myo creek during: a April 2006 (dry season) and b August 2006 (wet season). Shaded areas correspond to hours of darkness. Tidal heights are all relative to the mean of the lowest and highest tidal elevations observed during the survey



Porewater production of CO_2 and CH_4

Net pore water production of CO₂ and CH₄ reflects several competing diagenetic reactions. Organic matter oxidation via

aerobic respiration, denitrification and the reduction of nitrate, sulphate and iron and manganese oxides produces CO_2 . Methanogenesis can occur via CO_2 reduction and from the fermentation of acetate and other low molecular weight compounds that also produces CO₂, and CH₄ is oxidised by sulphate (Berner 1980). Carbonate dissolution and precipitation lead to CO₂ removal and addition respectively, and benthic primary producers are a CO₂ sink (Berner 1980). Despite the complexity, creek water TAlk vs. DIC relationships (Fig. 5) contain important information on the likely dominant diagenetic pathways in creek pore waters. During individual seasons, TAlk and DIC were highly correlated with essentially identical relationships at both sites, but there was a significant seasonal TAlk: DIC offset (wet season, TAlk=0.90 DIC+0.10, r^2 =0.99; dry season, TAlk=0.92 DIC+0.20, $r^2=0.97$; both, n=72). An important detail is that TAlk and DIC concentrations at Wright Myo were lower overall than at Kalighat during both seasons, which could reflect the relative intensities of tidal pumping at the two sites. During the wet season when these differences were greatest (Fig. 5), tidal amplitudes at Kalighat were approximately twice those at Wright Myo (Figs. 2, 3 and 4). An additional possibility is some inter-site variability in organic matter decomposition rates. Alongi et al. (2001) recognise aerobic respiration and sulphate reduction as the major mechanisms of organic matter oxidation in mangrove sediments. However, the first step in organic matter diagenesis is hydrolysis of particulate organic carbon (POC), which yields TAlk but no other DIC (Krumins et al. 2013). While this could account for the observed TAlk intercepts (Fig. 5) and would imply a higher supply rate of reactive POC to mangrove sediments during the dry season (larger TAlk intercept), such an interpretation conflicts with our observations of higher turbidity during the wet season (Table 1), which was interpreted above to reflect higher wet season fluxes of organic carbon, consistent with increased pCO₂ and decreased O₂ deriving from sediment pore waters and a suppression of photosynthesis. Clearly, some remineralisation of



Fig. 5 TAlk vs. DIC during the tidal surveys at Kalighat Creek and Wright Myo

POC also could additionally have occurred in the overlying water but our data do not allow any evaluation of this. Interestingly, other work strongly implies such seasonal variability to be insignificant (Bouillon et al. 2007b; Maher et al. 2013). An alternative explanation is that the observed TAlk vs. DIC seasonality (Fig. 5) reflects at least partly, seasonal changes in aerobic respiration and denitrification, the only primary or secondary diagenetic reactions for which ΔDIC exceeds Δ TAlk (Krumins et al. 2013). Higher aerobic respiration during the wet season is consistent with the correspondingly higher pCO₂ and lower O₂ observed (Fig. 2). There is evidence for sediment denitrification at some mangrove sites (Alongi et al 1998, 2000) and a ratio of 0.8 for $\Delta TAlk/\Delta DIC$ during denitrification (Krumins et al. 2013) is close to that observed in this study (Fig. 5). Even so, denitrification is thought to be a relatively minor contributor to organic matter degradation in most mangrove ecosystems (Alongi et al. 2000; Koné and Borges 2008). Consistent with this, although a substantial pore water source of N₂O in Wright Myo creek waters was previously found, based on concurrent observations of NO3⁻ and NO2⁻, this study considered it more likely that this was a consequence of nitrification rather than denitrification (Barnes et al. 2006). Nitrification is a net consumer of TAlk (Krumins et al. 2013) but otherwise leaves DIC unaltered. Iron and manganese reduction have also been found to be important in some Asian mangroves (Alongi et al. 1998; Kristensen et al. 2000) but for both $\Delta TAlk/\Delta DIC$ is far larger than we observed at Wright Myo or Kalighat. Carbonate dissolution can be a large contributor of TAlk but has a Δ TAlk/ Δ DIC ratio of 2 (Krumins et al. 2013). The major diagenetic pathway most consistent with the TAlk and DIC data is sulphate reduction, for which $\Delta TAlk/\Delta DIC$ is 1 (Krumins et al. 2013). Similar conclusions were reached by Koné and Borges (2008) based on Talk/DIC stoichiometry in waters surrounding Vietnamese mangroves. Having considered all the evidence, it can thus be concluded that the most likely major diagenetic pathway of organic matter oxidation in Andaman mangrove sediments is sulphate reduction, with a variable secondary role for aerobic respiration, broadly in line with previous conclusions for mangroves (Alongi et al. 2001; Koné and Borges 2008.

Carbon system components in surface water transects

TAlk vs. salinity or DIC vs. salinity relationships in surface water transects contain information on the potential lateral transport of "tidally pumped" creek waters. TAlk and DIC were linearly correlated in all transects (data not shown). TAlk vs. salinity is shown in Fig. 6. All dry season transects showed a small overall decrease in TAlk with increasing salinity, consistent with a weak but measurable influence of creek water TAlk in even the most offshore samples. By Fig. 6 TAlk vs. salinity during the inshore water transects during: **a** dry (April 2005 and 2006) and **b** wet seasons (August 2005 and 2006)



contrast, the wet season data were much more variable. Although in TI (Kalighat to Mayabander, Aug 2005) an overall decrease in TAlk with salinity was evident, all other wet season transects followed an opposing trend (Fig. 6). Notably, surface (10 m) TAlk concentrations in the Andaman Sea (November) are in the range 2.12–2.16 mmol kg⁻¹ (Sarma and Narvikar 2001), essentially identical to much of the wet season data for salinities above 15 (Fig. 6). Evidently, the wet season creek water outflow signal of TAlk was largely masked by these values.

Overall pCO_2 and CH_4 decreased with salinity in all transects (Fig. 7). Means and ranges of pCO_2 were generally highest during the wet seasons whereas means and ranges of CH_4 showed no marked seasonality (Table 2). Highest pCO_2 and CH_4 was observed in TIII (Fig. 7; Table 2), which was in closest proximity to adjacent mangrove creek waters at Wright Myo (Fig. 1). Indeed, seasonality in these transects

reflected that at Wright Myo, during 2005 pCO₂ was overall lower during the dry season transect than at other times and CH₄ showed little temporal variation (Fig. 7; Table 2), consistent with the results of the tidal surveys (Table 1). Whereas CH₄ was highly supersaturated at all salinities, CO₂ was consistently supersaturated only between Kalighat and Mayabander (TI) and along Wright Myo (TIII), during both the wet and dry seasons (Fig. 7; Table 2). Between Mayabander and Austin Creek (TII), mild CO2 undersaturation at high salinities was observed during April 2005 (dry season), and between Wandoor and Chipiyatapu (TIV), CO2 was always supersaturated only during the August 2006 (wet season) transect. Indeed, TIV had the overall lowest pCO₂ and CH₄ values of any transect (Table 2), consistent with its being the furthest removed from the influence of mangrove creeks. Undersaturation was observed in one TIV sample during April 2006 but during both the wet and dry

Fig. 7 Variation of a pCO_2 and b dissolved CH_4 with salinity in surface water transects: TI (*open squares*), TII (*open diamonds*), TIII (*closed circles*) and TIV (*open circles*). Note the different scales for different seasons



season TIV transects of 2005 $\rm CO_2$ was predominantly undersaturated throughout.

Evidently, creek water tidal pumping can significantly influence the seasonal distributions of pCO_2 and CH_4 in mangrove-fringing near-shore waters, which like the creeks themselves, are year-round sources of tropospheric CH_4 . By contrast, the observed distributions of CO_2 presumably reflect the balance of net photosynthesis versus creek water inputs. A significant imprint of tidally pumped CO_2 and CH_4 in mangrove fringing waters is not surprising given that offshore advection of DIC is a likely major export pathway for mangrove carbon (Bouillon et al. 2007c; Maher et al. 2013). Shelf regions are often considered to be net annual CO_2 sinks (Chen and Borges 2009). For tropical shelves adjacent to extensive

	TO CONVINIO III) CASI		יוסווו ווו פווואווואוואמאיאי	ore water name cele				
Transect/season	pCO ₂ (µatm)	pCO ₂ (saturation %)	CH ₄ (nmol L ⁻¹)	CH ₄ (saturation %)	TAlk (mmol L ⁻¹)	DIC (mmol L ⁻¹)	Temperature (°C)	Salinity
I: Kalighat to Mayabar	ıder							
April 2005 Dry	955 ± 592 (381–2,035)	251 ± 156 (100-536)	249 ± 88 (84 -376)	$11,482\pm3,970 \\ (3,958-17,024)$	2.37 ± 0.04 (2.33–2.51)	2.10±0.11 (1.96–2.27)	29.3 ± 0.4 (28.7–30.5)	31.4 ± 2.0 (29.6–34.0)
August 2005 Wet	$1,848\pm1,508 \\ (454-4,411)$	486 ± 396 (119-1,160)	253 ± 84 (86-378)	$\begin{array}{c} 10,176\pm3,035 \\ (3,761{-}14,488) \end{array}$	2.09 ± 0.46 (1.27–2.59)	2.06 ± 0.51 (1.27–2.62)	26.8 ± 0.5 (26.0–27.5)	14.9 ± 9.8 (2.9-32.7)
II: Mayabander to Aus	tin Creek							
April 2005 Dry	486±152 (292–761)	128 ± 40 (77–200)	74 ± 41 (28–190)	$3,466\pm1,897$ (1,303-1,897)	2.33±0.05 (2.25–2.42)	2.00 ± 0.05 (1.41–2.10)	28.9 ± 0.9 (25.8–29.4)	24.4 ± 1.3 (32.8 -36.0)
August 2005 Wet	762 ± 314 (320–1,613)	201 ± 83 (84-424)	81 ± 42 (28–168)	$3,422\pm1,829$ (1,256-7,342)	1.42 ± 0.19 (0.96 -1.69)	$1.36\pm0.16 \\ (0.97-1.62)$	26.4±0.6 (25.5–27.5)	21.3 ± 4.8 (13.5–29.8)
III: Wright Myo								
April 2005 Dry	$1,996\pm265 \\ (1,710-2,653)$	525±70 (450–698)	290 ± 78 (187-459)	$12,892\pm3,240 \\ (8,436-17,608)$	2.46 ± 0.02 (2.43–2.49)	2.42±0.03 (2.37–2.47)	28.8 ± 0.4 (27.9–29.3)	26.6 ± 3.1 (20.0 -30.0)
August 2005 Wet	$3,548\pm263$ (3,285-4,066)	934 ± 69 ($864-1,070$)	275±79 (157–459)	$11,406\pm2,893 \\ (6,539-17,451)$	$1.46\pm0.37\\(0.71{-}1.79)$	1.55 ± 0.36 (0.71–1.79)	26.1 ± 0.6 (25.0–27.0)	$19.8 \pm 7.3 \\ (4.8 - 31.7)$
April 2006 Dry	$2,925\pm329$ (2,266–3,398)	770±87 (596–894)	290 ± 78 (187-459)	$13,597\pm3,640 \\ (8,709-21,599)$	2.44 ± 0.05 (2.34-2.52)	2.46 ± 0.08 (2.30-2.51)	31.1 ± 0.8 (30.0–32.8)	30.4 ± 0.5 (29.5–31.2)
August 2006 Wet	$2,937\pm802$ (1,713 $-3,948$)	773±211 (451–1,039)	262 ± 85 (194-455)	$11,480\pm3,894 \\ (8,435-20,304)$	$1.99\pm0.14 \\ (1.55-2.10)$	2.05 ± 0.13 (1.66-2.16)	29.1 ± 0.3 (28.2–29.4)	22.0±2.8 (14.2–25.4)
IV: Wandoor to Chipiy	atapu							
April 2005 Dry	318±54 (224−412)	84 ± 14 (59-108)	127 ± 39 (68–195)	$5,897\pm1,771$ (3,183–9,049)	$2.31 {\pm} 0.05 \\ (2.19 {-} 2.39)$	$1.48\pm0.07 \\ (1.84-2.14)$	28.9 ± 0.6 (28.4-30.5)	33.2±0.8 (32.0–34.0)
August 2005 Wet	462 ± 292 (273-1,304)	121±77 (72–343)	124 ± 39 (73-189)	$5,476\pm1,587$ (3,340-7,360)	$1.90\pm0.12 \\ (1.54-2.02)$	$\begin{array}{c} 1.68 {\pm} 0.11 \\ (1.47 {-} 1.84) \end{array}$	26.4±0.7 (25.5–27.5)	30.4 ± 2.6 (26.9–33.6)
April 2006 Dry	497 ±76 (277–607)	131 ± 20 (73-160)	124 ± 31 (62–176)	$5,832\pm1,502$ (2,905–8,322)	2.20 ± 0.02 (2.18–2.25)	$1.90 {\pm} 0.06 \\ (1.67 {-} 1.94)$	30.5 ± 1.2 (29.4–31.6)	32.0±0.2 (31.5–32.3)
August 2006 Wet	$1,199\pm138 \\ (838-1,415)$	316 ± 36 (220-372)	140 ± 41 (75–209)	$\begin{array}{c} 6,338 \pm 1,846 \\ (3,411 - 9,505) \end{array}$	2.30 ± 0.05 (2.20-2.42)	2.17 ± 0.06 (1.97–2.24)	29.0 ± 0.2 (28.6 -29.6)	$\begin{array}{c} 28.8 \pm 1.0 \\ (26.2 - 29.6) \end{array}$

Table 2 Means and ranges (in brackets) of CO_2 , CH_4 and ancillary measurements in inshore water transects

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mangroves with year-round CO₂ export via tidal pumping, this view may require further scrutiny.

Emission fluxes of CO2 and CH4

Considering the several potentially important physical and biogeochemical controls of air–sea gas exchange, selecting an appropriate k_w parameterisation is problematic; those in most common use are based on wind speed alone and result in considerable uncertainty (Upstill-Goddard 2006). Two parameterisations specifically applicable in shallow coastal waters where tidal current–seabed interaction generates significant turbulence are Clark et al. (1995) and Borges et al. (2004). Although the latter formally quantifies tidal current velocities and water depth, neither parameterisation has proven universally applicable (Upstill-Goddard 2006). Both were therefore applied to the present data, resulting in an uncertainty range that may be considered typical of such estimates (Upstill-Goddard 2006).

Daily mangrove creek water emissions of CO₂ and CH₄ derived from the Clark et al. (1995) and Borges et al. (2004) relations (Table 3) are the sums of individual hourly estimates over the full 24-h tidal surveys (all, n=24). Daily emissions derived from surface transect data (Table 4) are the means of individual daily estimates for each sampling station (all, n=15); these emissions were only obtained using Clark et al. (1995) as no water depth or current data are available. Variability in wind speeds was low across all six creek surveys (Table 3); hence the values of k_w derived from Clark et al. (1995) did not vary significantly. Given the almost identical Schmidt numbers of CO₂ and CH₄ (Wanninkhof 1992) $k_{\rm w}$ derived from Clark et al. (1995) was also essentially identical for both gases in all six surveys: CO_2 , 1.31 ± 0.08 cm h⁻¹; CH₄, 1.31 ± 0.07 cm h⁻¹ (*n*=120). By contrast, k_w estimates based on Borges et al. (2004) were somewhat more variable: CO_2 , 4.08±1.56 cm h⁻¹; CH₄, 4.13±1.63 cm h⁻¹, the difference reflecting changes in water depths and current velocities over tidal cycles. Variations in k_w evidently exert only weak control of Andaman creek CO2 and CH4 emissions, which over tidal cycles are primarily functions of changing hydrostatic pressure and porewater concentration gradients. At Kalighat wet season CO2 emissions derived using Clark et al (1995) were more than twice as high as dry season CO_2 emissions whereas at Wright Myo the seasonal differences were much smaller. By contrast, wet season CO₂ emissions derived using Borges et al (2004) were more than twice as high as dry season emissions at both sites (Table 3). Koné and Borges (2008) reported a similar but larger (fivefold) seasonal trend in CO2 emissions from Kiên Vàng mangroves in Vietnam, which they attributed to higher net benthic and/or water column heterotrophy during the rainy season, consistent with our earlier explanation for the higher wet season pCO₂ at Kalighat. That the CH₄ data do not support this may be explained by our earlier observation of a significant suppression of mangrove creek CH₄ emissions at Wright Myo immediately following rainfall (Barnes et al. 2006). Consistent with observed simultaneous changes in inorganic nitrogen speciation and turbidity, this behaviour was ascribed to CH₄ oxidation during direct disturbance of the creek sediment surface by raindrops (Barnes et al. 2006). The seasonal variation in Andaman CH₄ emissions was much smaller than for CO₂ overall, irrespective of the k_w formulation used (Table 3). No significant difference between wet and dry season CH₄ emissions at Wright Myo (Barnes et al. 2006) was previously found.

Table 5 summarises previously published emissions of CO_2 and CH_4 from mangrove surrounding waters. In so far as the authors are aware, only three other studies have reported simultaneous such estimates for CO_2 and CH_4 (Ramesh et al. 2007; Bouillon et al. 2007b; Kristensen et al. 2008). Ramesh et al. (2007) summarise our earlier emissions estimates for Wright Myo and Kalighat and it is

Table 3 Emissions fluxes of CO_2 and CH_4 (in millimoles per square meter per day) at Kalighat and Wright Myo

CO ₂ emission (mr	mol m ⁻²	day ⁻¹)			CH ₄ emission (mmo	$1 \text{ m}^{-2} \text{ day}^{-1}$)
		Wind speed (m s^{-1})	Clark et al. (1995)	Borges et al. (2004)	Clark et al. (1995)	Borges et al. (2004)
Kalighat						
April 2005	Dry	$0.45 {\pm} 0.28$	22.7	49.3	0.17	0.38
August 2005	Wet	$0.31 {\pm} 0.31$	51.9	172.8	0.14	0.47
Wright Myo						
April 2005	Dry	0.40 ± 0.15	30.2	49.3	0.18	0.29
April 2006	Dry	0.20 ± 0.15	37.3	59.5	0.12	0.19
August 2005	Wet	0.29 ± 0.21	39.4	112.3	0.15	0.44
August 2006	Wet	$0.17 {\pm} 0.22$	38.7	113.8	0.11	0.34
Average of all site	s (n=6)		36.7±9.8	92.8±49.2	$0.15 {\pm} 0.03$	$0.35 {\pm} 0.10$

Table 4 Emissions fluxes of CO_2 and CH_4 (in millimoles per square meter per day) during surface water transects estimated with the Clark et al. (1995) relation

	Transect/season		Wind speed (m s^{-1})	CO_2 emission (mmol m ⁻² d ⁻¹)	CH_4 emission (mmol m ⁻² day ⁻¹)
TI	April 2005	Dry	1.45 ± 0.80	7.1	0.14
	August 2005	Wet	2.37±1.72	25.1	0.19
TII	April 2005	Dry	$1.18 {\pm} 0.83$	1.1	0.04
	August 2005	Wet	$2.33 {\pm} 0.95$	8.2	0.07
TIII	April 2005	Dry	1.23 ± 0.30	20.7	0.14
	April 2006	Dry	$1.04 {\pm} 0.34$	24.3	0.14
	August 2005	Wet	$0.26 {\pm} 0.16$	46.3	0.12
	August 2006	Wet	1.62 ± 1.59	39.2	0.19
IV	April 2005	Dry	$2.57{\pm}1.03$	-1.3	0.10
	April 2006	Dry	2.50 ± 1.90	2.2	0.13
	August 2005	Wet	3.29±1.19	2.6	0.10
	August 2006	Wet	$1.10 {\pm} 0.53$	10.1	0.07
Average	of all transects (n	=12)		14.7 ± 14.6	$0.12 {\pm} 0.05$

noteworthy that these are lower than our present estimates, by up to an order of magnitude in the case of CH_4 (Tables 3 and 5). Based on these data, emissions of CO_2 and CH_4 from mangrove creek waters evidently exhibit high inter-annual variability. In other earlier work at Wright Myo we also estimated CH_4 emissions over tidal cycles, in this case using static and floating chambers (Barnes et al. 2006). Our current estimates (Table 3) are at the lower end of this earlier range (Table 5), which is perhaps not surprising given that flux chambers can tend to give larger fluxes than those based on formulations of k_w (Barnes et al. 2006).

The emission estimates for CO_2 , both for the tidal creeks and the surface water transects (Tables 3 and 4), appear typical of mangrove associated waters globally (Table 5) whereas the estimated CH₄ emissions (Tables 3 and 4) are towards the low end of the published range (Table 5). However, it should be noted that the CH₄ data in Table 5 span two orders of magnitude because of the inclusion of both anthropogenically impacted and pristine sites. Highest CH₄ emissions, from for example the Jiulonjiang estuary (Alongi et al. 2005) and SE Indian (Purvaja and Ramesh 2001) and Advar estuary mangroves (Ramesh et al. 2007), tend to reflect urbanisation and/or large inputs of domestic sewage. Kristensen et al. (2008) also found higher CH₄ emissions from an anthropogenically impacted site (Mtoni) than from a pristine site (Ras Dege) in Tanzania although the difference was less pronounced than in these other studies (Table 5). Our new data for Kalightat and Wright Myo (Table 3) support our previous contention that mangrove ecosystems could dominate coastal CH₄ emissions worldwide and that these may have been seriously underestimated in the past (Barnes et al. 2006; Upstill-Goddard et al. 2007), but they do not materially affect previous estimates of mangrove water emissions of either CO₂ or CH₄ globally (Table 5). Worth noting, however, are the observations of substantial ebullition at both Kalighat and Wright Myo during the tidal surveys. These imply that ebullition, which is not quantified in Table 5, may make an important contribution to total emissions, especially for CH₄. This is because bubbles collected at Wright Myo (21 April 2006; dry season) were around 1 % by volume CO_2 (26 times ambient air) and 5 % by volume CH₄ (28000 times ambient air). Studies of CH₄ ebullition in subtropical estuaries (Shalini et al. 2006; Rajkumar et al. 2008), a tropical wetland (Marani and Alvala 2007) and a hypertrophic temperate lake (Casper et al. 2000) estimated ebullition contributions $\sim 60-90$ % to total CH₄ emissions. While these results imply that the CH₄ emissions presented in Table 5 could be rather conservative, accurately estimating ebullition fluxes is a substantial challenge (Upstill-Goddard 2006). Consequently, accurately quantifying the ebullition contribution to mangrove water CO2 and CH4 emissions requires more detailed study.

Scaling the emission fluxes listed in Table 3 to a creek open water area of~2,700 km² gives total Andaman creek water emissions of 3.6×10^{10} mol $CO_2 year^{-1}$ and 3.7×10^7 mol $CH_4 year^{-1}$ based on the k_w formulation of Clark et al. (1995). Corresponding emissions derived using the k_w formulation of Borges et al. (2004) are somewhat larger: 9.2×10^{10} mol CO₂ year⁻¹ and 3.4×10^{8} mol CH₄ year⁻¹. Taking an additional account of the contribution from surrounding inshore waters (Table 4) is problematic because the full area extent of the CO₂ and CH₄ outflow plumes is not known, but based on the surveys conducted, these could be estimated to be around ten times the creek area. If so and taking account of their average lower emissions per unit area (around 42 and 71 % respectively, of the creek water values for CO₂ and CH₄; Tables 3 and 4), using k_w derived from Clark et al (1995), the additional emissions could be as much as 1.5×10^{11} mol

Table 5	Previously	published	estimates	of CO ₂	and C	CH ₄ em	issions	from	mangrove	surrounding	waters
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Country	Location	CO_2 emission (mmol m ⁻² day ⁻¹)	$CH_4 emission (mmol m^{-2} day^{-1})$	Reference
Andaman Islands	Wright Myo		0.43–0.84	Barnes et al. (2006)
Andaman Islands	Wright Myo	65.2	0.01	Ramesh et al. (2007)
Andaman Islands	Kalighat	57.3	0.01	Ramesh et al. (2007)
Australia	Queensland		0.02–0.53	Kreuzwieser et al. (2003)
Bahamas	Normans Pond	13.8±8.3		Borges et al. (2003)
Bermuda	Mangrove Bay	59.8±17.3		Zablocki et al. (2011)
Bermuda	Ferry Reach	5.5±1.3		Zablocki et al. (2011)
Brazil	Itacuracá creek	113.5±104.4		Ovalle et al. (1990)
China	Jiulonjiang Estuary ^a		0-0.1	Alongi et al. (2005)
India	Adyar mangroves ^a	48.7	5.2	Ramesh et al. (2007)
India	Gaderu creek	56.0±100.9		Borges et al. (2003)
India	Godavari Delta	70.2±127.0		Bouillon et al (2003)
India	Moorimgamga creek	23.2±10.1		Ghosh et al. (1987)
India	Muthupet/Cauvery Delta	87.2	0.04	Ramesh et al. (2007)
India	Pichavaram/Cauvery Delta		0.5–1.6	Purvaja et al. (2004)
India	Pichavaram/Cauvery Delta	16.6	0.05	Ramesh et al. (2007)
India	Saptamukhi creek	56.7±37.4		Ghosh et al. (1987)
India	Sundarbans		0.4	Biswas et al. (2004)
India	Sundarbans		0.02–0.14	Biswas et al. (2007)
India	Pichavaram		5.4–20.3	Purvaja and Ramesh (2001)
Kenya	Kidogoweni	71.0±43.0		Bouillon et al (2007b)
Kenya	Kinondo creek	52.0±42.0		Bouillon et al (2007b)
Kenya	Tana river delta	58.0±45.0		Bouillon et al (2007a)
Papua New Guinea	Nagada creek,	43.6±33.2		Borges et al. (2003)
Tanzania	Mtoni ^a	34.5±43.3		Bouillon et al (2008)
Tanzania	Mtoni ^a	3.0-40.0	0.07-0.35	Kristensen et al (2008)
Tanzania	Ras Dege	34.2±27.2		Bouillon et al (2007c)
Tanzania	Ras Dege	1.0-80.0	0.01-0.07	Kristensen et al (2008)
Thailand	Ranong Province		0.20–0.8	Lekphet et al. (2005)
USA	Florida Bay	4.6±5.4		Millero et al. (2001)
USA	Florida mangrove		0.02	Harriss et al (1998)
USA	Shark River, Florida	43.8±52.1		Koné and Borges (2008)
Vietnam	Main channel (dry) ^a	27.1±12.0		Koné and Borges (2008)
Vietnam	Main channel (wet) ^a	81.3±31.8		Koné and Borges (2008)
Vietnam	Kiên Vàng (dry) ^a	32.2±39.4		Koné and Borges (2008)
Vietnam	Kiên Vàng (wet) ^a	154.7±159.1		Koné and Borges (2008)
Vietnam	Tam Giang (dry) ^a	141.5±117.8		Koné and Borges (2008)
Vietnam	Tam Giang (wet) ^a	128.5±110.0		Koné and Borges (2008)

^a Recognised as significantly anthrogenically impacted by the authors

 $CO_2 year^{-1}$ and 2.6×10^8 mol $CH_4 year^{-1}$, giving total emissions of ~ 1.9×10^{11} mol $CO_2 year^{-1}$ and ~ 3.0×10^8 mol CH_4 -year⁻¹ from a total area of mangrove influenced water of ~ 3×10^4 km². Using the k_w formulation of Borges et al. (2004) would undoubtedly yield larger total emissions; however, the lack of water depth and current data for the inshore water transects precludes this.

Mangrove Water Emissions and the Mangrove Carbon Budget

To consider the scale of gaseous carbon emissions from mangrove-associated waters in the context of other compartments of the mangrove carbon budget is instructive. Combining the emissions estimates (Tables 3 and 4) with those compiled in Table 5, an average emissions of ~55±40 mmol CO_2m^{-2} day⁻¹ and ~1.2±3.0 mmol CH_4m^{-2} d⁻¹ from mangrove-associated waters can be estimated globally. Our figure for CO₂ is close to an earlier estimate of \sim 59±52 mmol CO_2m^{-2} day⁻¹ derived from a slightly smaller dataset (Bouillon et al. 2008). Water column emissions of gaseous carbon are a subset of total sediment emissions and attributing fractional contributions to the total from inundated and exposed sediments individually is problematic because of the intertidal nature of the system. However, a compilation by Bouillon et al (2008) of CO₂ emission estimates derived for exposed mangrove sediments using benthic chambers gave a global average of $61\pm46 \text{ mmol } \text{CO}_2\text{m}^{-2} \text{ day}^{-1}$. Given the uncertainties that arise from spatio-temporal variability and measurement methodologies, this is identical to their figure for mangrove waters derived from pCO₂ data, which prompted them to set total global mangrove CO₂ emissions from sediments/water column at ~ 60 ± 45 mmol m⁻² day⁻¹, equivalent to $\sim 42 \pm 31$ Tg C year⁻¹. Our estimates for CO₂ and CH₄ do not materially affect this figure considering the inherent uncertainties arising from the high spatio-temporal variability of tidal and inter-tidal mangrove waters and given the above indication that more than 98 % of the total gaseous carbon flux is CO₂.

Additional mangrove carbon sinks include burial and lateral export. Recent work in an Australian mangrove creek provides the first direct estimate of the relative carbon export contributions from DIC, DOC and POC (Maher et al. 2013). This showed DIC export to be at least an order of magnitude greater than DOC export and that POC was consistently imported, at rates comparable to or exceeding, those of DOC export. Given the associated uncertainties the DIC: DOC export ratio is consistent with an earlier synthesis (Bouillon et al. 2008) that set global mangrove carbon sinks (in teragrammes C per year) at: burial, 18.4; DIC export, 178 \pm 165; DOC export, 24 \pm 21; and POC export, 21 \pm 22. Together, these sinks were considered sufficient to balance mangrove net primary production (Bouillon et al. 2008). Clearly, mangrove water emissions of gaseous carbon are a significant term in the mangrove carbon budget, apparently of the same order as DOC and POC fluxes. However, the available data are limited and the associated uncertainties are very large. Importantly, other potentially relevant mechanisms of CO₂ and CH₄ emission remain unaccounted for; for example, crab burrows, which are not routinely included in sediment efflux measurements, may be important conduits for enhanced CO_2 loss (Kristensen et al. 2008), and there is evidence that plant pneumatophores could at least make a significant contribution to total CO₂ emissions (Bouillon et al. 2008) while they may even dominate CH₄ emissions (Purvaja et al. 2004). Moreover, there are no reliable estimates of global mangrove ebullition rates for either CO₂ or CH₄, even though these might well be quantitatively significant.

Conclusions

Our estimates of CO2 and CH4 emissions from Andaman mangrove tidal creeks and adjacent inshore waters are within previously published ranges for these systems, consistent with a significant contribution to the mangrove sink for global carbon. Evaluating such emissions in a range of mangrove environments is important to resolving the greenhouse gas balance of mangrove ecosystems globally; however, future such studies should be integral to much wider quantitative process studies of the overall mangrove carbon balance. Improving our current understanding of mangrove carbon dynamics is important, not only for reducing current uncertainties, but for predicting the system response to large-scale changes such as mangrove replanting and clearance, both of which are becoming increasingly important aspects of mangrove "management" and for which the response through gaseous carbon emissions may not be well anticipated.

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