

Rapid Assessment of Total Phenolic and Anthocyanin Contents in Grape Juice Using Infrared Spectroscopy and Multivariate Calibration

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Received: 22 August 2016 / Accepted: 7 November 2016 / Published online: 19 November 2016 © Springer Science+Business Media New York 2016

Abstract Grape juice is rich in bioactive compounds that are known for beneficial effects, such as prevention of cancer and diabetes. The determination of the content of these compounds has been carried out following several protocols which generated toxic waste. In this context, infrared spectroscopy in near (NIR) and mid- (MIR) regions were applied for the determination of total phenolic and anthocyanin content (TPC and TAC, respectively) in grape juice for being a rapid green methodology. Calibration models were built using partial least squares regression (PLSR) to predict TPC and TAC in grape juice. MIR and NIR had a similar satisfactory performance to predict TAC presenting low RMSEP (4.22 mg/100 mL and 4.44 mg/100 mL). In TPC prediction, MIR presented a RMSEP (0.21 mg GAE/mL) slightly better in comparison to the one presented by NIR (0.37 mg GAE/mL). Errors can be considered acceptable for the determination of TPC and TAC in grape juice, qualifying both infrared techniques to replace traditional wet methodologies, with the advantage of being environmentally friendly.

Keywords Grape juice \cdot NIR \cdot MIR \cdot Phenolic compounds \cdot Anthocyanins

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Introduction

Owing to its resistance to many fungal diseases and a high productivity, the main cultivar applied in grape juice processing in Brazil is cv. Isabel (*Vitis vinifera* × *Vitis labrusca*). Nevertheless, Isabel grapes present light color, requiring blending with other varieties of grape, as the hybrids BRS Cora and BRS Violet (Lima et al. 2015; Yamamoto et al. 2015). Grapes, as well as the juices produced from them, are known for their significant concentration of bioactive compounds, such as phenolic acids (e.g., gallic acid), flavonoids (e.g., proanthocyanidins), and anthocyanins (Ribeiro et al. 2015; Eshghi, Salehi and Karami 2014; Lima et al. 2015).

Grape juice and derivatives are very important commodities in the international juice trade. In the USA, wine, grape, and derivatives generate \$162 billion dollars a year; in Brazil, the consumption of grape juice has grown by 400% in 5 years (from 10 million liters in 2007 to 50 million liters in 2012). Grape juice is used not only as beverage but also as ingredient for other foodstuffs. Furthermore, grape juice meets the growing consumer demand for a product without preservatives but with color, flavor, high nutritional qualities, high concentration of minerals, vitamins, and phenolic compounds (Hui 2008; Dani et al. 2016; Santos 2014).

Phenolic compound content in grape juice is affected by culture and post-harvest conditions, or according to the grape species used in juice production. The mean compounds found in grapes, the anthocyanins, are related to a reduction in oxidative stress; they can help in the prevention of diseases as cancer, arteriosclerosis, diabetes, and neurological diseases. In addition, anthocyanins have antihyperglycemic and antiinflammatory effects; besides that, they are extremely important in sensory attributes as color (Dal Magro et al. 2016; Eshghi, Salehi and Karami 2014; Xu et al. 2010; Moreno-Montoro et al. 2015; Lambri et al. 2015).

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The determination of bioactive compounds as phenolic compounds and anthocyanins has been performed by many different methodologies. Spectrophotometric methods are widely applied in total phenolic compound quantification; an estimation by measurement of absorption at 280 nm and the Folin-Ciocalteu method are the most used (Lorrain et al. 2013). Liquid chromatography, coupled or not with mass spectrometry, was also used to quantify these compounds in many researches (Díaz-de-Cerio, Gómez-Caravaca, Verardo, Fernández-Gutiérrez and Segura-Carretero 2016; Moreno-Montoro et al. 2015; Dal Magro et al. 2016). The main problem, when such methodologies are used in the determination, is the extraction of compounds, which is mainly performed by liquid-liquid or solid-liquid extraction (Lorrain et al. 2013), resulting in the generation of wastes which are toxic for both environment and operator. Moreover, the fact that phenolic compounds are photosensitive leads to the need of using amber glassware and/or manipulation under reduced brightness to maintain the compound stability during chemical analysis.

In recent years, a number of new analytical approaches are being used to monitor food quality, such as vibrational spectroscopy, including infrared spectroscopy (near and mid), Raman spectroscopy, and hyperspectral/multispectral imaging. These techniques present some advantages such as the following: non-invasive and non-destructive, require a small amount of sample, do not require extraction, do not use toxic reagents, and can be used online (Lohumi et al. 2015; Ropodi, Panagou and Nychas 2016). Near infrared (NIR) spectra $(10,000-4000 \text{ cm}^{-1})$ provide information from overtones and their combinations, while mid-infrared (MIR) spectra (4000–400 cm⁻¹) give information from fundamental molecular vibrational frequencies. Therefore, the spectra obtained by MIR are straightforward (de Oliveira et al. 2014; Shiroma and Rodriguez-Saona 2009). Despite the differences between the analysis of NIR and MIR spectra, both require the application of chemometrics to develop models and to obtain other data (Hell et al. 2016).

Several works have been developed applying NIR and MIR spectroscopies to evaluate quality and to quantify different chemical compounds in food products, food authenticity, and adulterations. Infrared analysis has been successfully used to quantify total phenolic compounds and condensed tannins contained in grape seed, to evaluate ripeness of white grape, and to a qualitative and quantitative evaluation of grape berries at various stages of development (de Oliveira et al. 2014; Bag, Srivastav and Mishra 2011; Shiroma and Rodriguez-Saona 2009; Hell et al. 2016; Martelo-Vidal and Vázquez 2014; Mendes et al. 2015; Li et al. 2013; Ignat et al. 2012; Wang and Xie 2014; Kyraleou et al. 2015; Giovenzana et al. 2015; Musingarabwi et al. 2016).

This article reports a study that aimed at applying the techniques NIR and MIR for the determination of total phenolic compounds and anthocyanin contents in grape juices consumed in Brazil as alternatives for routine analysis, thus reducing the risk of accidents, the use of toxic reagents, work in sample preparation, and time spent on analysis.

Materials and Methods

Materials

A total of 65 commercial samples of grape juice were acquired in a local market and analyzed during this study. Sixty-two samples were products of different brands and lots, and three of them were obtained by lyophilization (Terroni Equipamentos Científicos, Brazil, Série LS). The commercial sample of grape juice with the highest phenolic content was concentrated by lyophilization during 30, 60, and 90 min. This process was carried out in order to ensure greater variability of the results to construct a calibration model with large values.

Total Anthocyanin Content Analysis

The total content of anthocyanins in 65 samples of grape juice was analyzed. The extraction was carried out mixing the grape juice with a solution of HCl 0.1 mol/L at room temperature (25 °C) for 30 min in a dark room. The samples were centrifuged at 3600 rpm/10 min to remove the solid particles, and the supernatant was collected according to the reference method (AOAC method 2005-02) with modifications (AOAC 2006; Giusti and Wrolstad 2001). The absorbance was recorded with a spectrophotometer (Hitachi High-Technologies, Japan, U-2900) in wavelengths of 520 and 700 nm, for both solutions, pH 1.0 and pH 4.5.

Total Phenolic Compound Content

The total phenolic content (TPC) was determined in the 65 samples of grape juice according to Singleton and Rossi (1965). The extract of the samples was obtained according to Paz et al. (2015). A volume of 500 μ L of sample extract, 75 μ L of water, 250 μ L of Folin-Ciocalteu, 1 mol/L reagent, and 1250 μ L of sodium carbonate was placed in the cuvette for 30 min in a dark room. The absorbance was determined at 700 nm in a spectrophotometer (Hitachi High-Technologies, Japan, U-2900). To calculate the phenolic concentration in samples, a calibration curve was done using gallic acid (GA) as standard antioxidant; the results were expressed in gallic acid equivalents (GAEs).

Spectra Acquisition

About 2 g of each sample was placed in a petri dish, and the liquid reflector (PerkinElmer, Waltham, USA, part number L118-0503) was placed on top of the sample and carefully

pushed against the glass before scanning. Transflectance measurements were conducted using near infrared reflectance accessory (PerkinElmer, Waltham, USA, model NIRA) in spectrometer FT-NIR (PerkinElmer, Waltham, USA, model Spectrum 100N) in the range of 10,000–4000 cm⁻¹, with 4 cm⁻¹ of resolution. All data were obtained in triplicate, and the mean was used in subsequent calculations.

All spectrum data of Fourier transformation mid-infrared (MIR) were determined using the attenuated total reflectance (ATR-FTIR) technique (Cary 630 FTIR Spectrometer, Agilent Technologies, USA). They were collected with 64 scans, 4 cm^{-1} of resolution, in a wavenumber range of 4000 to 400 cm⁻¹.

Chemometrics

Spectrum data were processed using the PLS toolbox for MatLab version 5.2 (The MathWorks Inc. 2009; Eigenvector Research Inc. 2010). For noise reduction and scattering correction, preprocessing was applied to it. The method of data smoothing based on the Savitzky-Golay algorithm, mean center, and standard normal variate (SNV) was used in this work. Cross-validation was used for optimization of the number of latent variables (LVs) in each model, and the number of LVs chosen was four and seven, for TAC and TPC, respectively.

Outliers were identified and excluded based on Student residual values and leverage in order to improve the quality of the calibration model.

The samples were divided into calibration and validation sets to perform external validation, where 75% of samples were used in the calibration set and 25% represent the validation set.

The partial least squares regression (PLSR) was the methodology chosen for building the calibration model. To evaluate the quality of the obtained models, the coefficient of determination (R^2) and the relative mean square errors of calibration (RMSEC) and prediction (RMSEP) were calculated.

Results and Discussion

Chemical Analysis

A summary of the total phenolic content (TPC) and total anthocyanin content (TAC) in grape juice samples evaluated is presented in Table 1. Range, mean, and coefficient of variation are given for each analysis. A phenolic content study of red grapes from the south of Brazil (*V. labrusca* L.) determined values of 2015.00 mg GAE/L for TPC and 420.01 mg/L for TAC, using conventional wet methodology. Those values were superior to the ones founded in this work, and these differences may occur since phenolic compound content is dependent on various factors as culture conditions or post-

Table 1 Total polyphenol and anthocyanin contents in grape juice

| Analysis | Range | $Mean \pm SD$ | CV |
|--------------------------------|------------|---|------|
| Total anthocyanins (mg/100 mL) | 43.56–2.67 | $\begin{array}{c} 14.50 \pm 0.343 \\ 1.29 \pm 0.07 \end{array}$ | 3.11 |
| Total polyphenols (mg GAE/mL) | 4.01–0.29 | | 0.56 |

n = 65 samples

harvest conditions, or according to the grape species used in juice production as well as the methodology applied to determine TPC and TAC. Other studies with anthocyanin and phenolic contents in grape were done, most of them in grapes V. vinifera or hybrids used in wine production (Fraige et al. 2014; Lambri et al. 2015; Toaldo et al. 2015). Grape juice showed higher mean content of TPC when compared to apple juice (1.295 mg GAE/mL and 0.098 mg GAE/mL) and similar TAC content to cranberry juice (13.64 mg/100 mL). Moreover, some studies suggest that the anthocyanins of grape juice present higher antioxidant activity than some vitamins as E and C, a feature that enhances the nutritional quality of the product (Knockaert, Pulissery, Colle, Van Buggenhout, Hendrickx and Loey 2012; Sharma et al. 2015; Fraige et al. 2014; Islam et al. 2016; Bazinet, Brianceau, Dubé and Desjardins 2012).

Overview and Pretreatment of Spectrum Data

The raw spectrum data obtained in MIR and NIR are represented in Fig. 1a, b, respectively. In MIR, larger peaks at 3200 and at 1600 cm⁻¹ associated with O-H stretch are observed, appearing as a result of the amount of water present in the sample. The fingerprint region about 1200–900 cm⁻¹ was determinant in the posterior PLS calibration model. This region is usually associated with C-O and C-C stretching and C-O-C and C-O-H, which could be related with esters, carbohydrates, and organic acids (Shiroma and Rodriguez-Saona 2009; Musingarabwi et al. 2016).

Analyzing the NIR spectrum data, two maximum peaks at $\cong 6900$ and $\cong 5200$ cm⁻¹ are observed as notorious. These peaks refer to O-H first overtone and in combination, and their higher position in the spectrum is related to the huge amount of water present in grape juice (Shiroma and Rodriguez-Saona 2009; Xie et al. 2009).

The preprocessing performed in all NIR and MIR spectrum data is detailed in Table 2. As a consequence of the light scattering presented by NIR spectrum data, it was necessary to apply the SNV algorithm for baseline correction. In order to remove the noise presented in the MIR spectrum data, the Savitzky-Golay method of data smoothing was used. The spectral region used in MIR calibration is near the fingerprint region, and qualitative and quantitative evaluation of grape berries was previously used. The rest of the spectral data were



Fig. 1 Raw spectrum data by MIR (a) and NIR (b)

hindered because of the huge water amount (Musingarabwi et al. 2016).

MIR × NIR

The statistical data for MIR and NIR calibration models are shown in Table 3 and Fig. 2. TAC provided a satisfactory coefficient of determination (R^2_c) for NIR (0.84) and MIR (0.81). Observing the values of RMSEP, RMSEC, MIR, and NIR, they were quite similar, with RMSEP = 4.22 mg/100 mL and RMSEC = 4.31 mg/100 mL for MIR and RMSEP = 4.44 mg/100 mL and RMSEC = 4.09 mg/100 mL for NIR. However, in the model, seven latent variables (LV) for NIR prediction and only five LV for the MIR model were necessary.

Chen et al. (2015) performed the prediction of TAC in grapes using NIR hyperspectral imaging and presented a more relevant RMSEP (1.29 mg/100 g) using 16 LVs, which is a number twice as large as the one used in this paper to build the NIR model. However, the use of too many LVs may result in an overfitting model. On that account, the use of less LVs is more suitable. The value of RMSEP found in this article was better when compared to Xiaowei et al. (2014) who developed a prediction model of TAC in flowering tea using NIR

Table 2 Preprocessing applied in MIR and NIR spectrum data

| | MIR | |
|--------------------|-------------------------|------------------------------------|
| | Preprocessing | Spectral range (cm^{-1}) |
| Total anthocyanins | Smoothing + mean center | 1747.42-829.11 |
| Total polyphenols | Smoothing + mean center | 1747.42-823.52 |
| | NIR | |
| | Preprocessing | Spectral range (cm ⁻¹) |
| Total anthocyanins | SNV + mean center | 10,000-8450 |
| Total polyphenols | Mean center | 10,000–4000 |



combined with ant colony optimization models with RMSEP of 12.25 mg/100 g; nevertheless, they used only three LVs.

Martelo-Vidal and Vázquez (2014) determined values of RMSEC between 54.88 and 35.94 mg/L in calibration models constructed to quantify the anthocyanins including malvidin in red wines; as reference method, the authors used HPLC that has lower errors compared to spectrophotometric analysis, therefore contributing to a final calibration model with a RMSEC lower when compared to those achieved in this paper.

TPC was determined with adequate accuracy. Comparing NIR and MIR models, a slightly better RMSEC was obtained by the NIR model (0.171 mg GAE/mL) and a better RMSEP in the MIR model (0.21 mg GAE/mL). Both techniques presented favorable coefficients of determination ($R_c^2 \ge 0.90$) and comparable number of LVs. Therefore, both techniques can be defined as efficient in determining TPC.

Table 3 Statistics for calibration and prediction of anthocyanin andpolyphenol contents by MIR and NIR

| | | MIR | | | |
|--------------------------------|-------------|----------------------------|-----------------|-----------|--------------------|
| | $N^{\rm a}$ | LV^{b} | $R^{2}_{c}^{c}$ | $RMSEC^d$ | RMSEP ^e |
| Total anthocyanins (mg/100 mL) | 65 | 5 | 0.81 | 4.31 | 4.22 |
| Total polyphenols (mg GAE/mL) | 65 | 5 | 0.90 | 0.26 | 0.21 |
| | | NIR | | | |
| | Ν | LV | $R^2_{\rm c}$ | RMSEC | RMSEP |
| Total anthocyanins (mg/100 mL) | 64 | 7 | 0.84 | 4.09 | 4.44 |
| Total polyphenols (mg GAE/mL) | 63 | 4 | 0.96 | 0.17 | 0.37 |
| | | | | | |

^a Total number of samples

^b Number of latent variables

^c Coefficient of determination of calibration set

^d Root mean square error of calibration

e Root mean square error of prediction

TAC Predicted (mg/100mL)

TPC Predicted (mgGAE/mL)

0

0





0

1 2 3 4 TPC Measured mgGAE/mL 0 1 2 3 4 TPC Measured (mgGAE/mL)

Hu et al. (2016) performed the determination of phenolic content in chocolate by ATR-FTIR and presented satisfactory RMSEP (5.08 mg GAE/g DFW) and RMSEC (3.21 mg GAE/g DFW). Kyraleou et al. (2015) used FT-IR to determine total phenolics contained in grape seeds and reported values of 0.97 for R^2 , RMSEC of 4.03 mg catechin/g DW, and RMSEP of 6.49 mg catechin/g DW; all errors presented by the authors were superior than those found in this work.

The content of total phenolic was determined by Frizon et al. (2015), using PLS regression and NIR in yerba mate (*Ilex paraguariensis*) with RMSEC of 16.07 mg 5CQA/g and RMSEP of 12.12 mg 5CQA/g; the errors were superior to the ones found in the NIR calibration model in this work. TPC was also determined by Páscoa et al. (2014) in milled red grape pomace using NIR and a PLS calibration model, presenting a RMSEC of 1.77 mg GAE/g sample and RMSEP of 2.12 mg GAE/g sample, which were superior than those achieved with grape juice.

Viegas et al. (2015) obtained a RMSEP = 0.022 mg/mLanalyzing TPC in wax jambu fruit using NIRS and PLS; this result presented a lower predicting error when compared to the ones determined in this paper. However, to achieve this RMSEP, the authors used nine LVs, while the model constructed to analyze grape juice used only four LVs.

Considering that the models used to predict TPC and TAC in grape juice presented good performance, with low values of RMSEP and RMSEC, other parameters commonly used to evaluate the quality of this product, as soluble solids, total acidity, total sugars, and insoluble solids, could also be calibrated by NIR or MIR.

Conclusion

Both techniques, NIR and MIR, presented quite similar results in the prediction of total phenolic and anthocyanin contents in grape juice. However, the MIR spectra are more accessible to interpret than the NIR spectra, making it possible to obtain a chemical knowledge of the system. The MIR spectra needed less pretreatments than the NIR spectra to achieve the best performance possible. It is important to consider that both spectroscopy techniques presented adequate performance, revealing a possible replacement of wet chemical analyses with the advantage of producing no toxic waste, little time consumption, no danger to the operator, and without any trouble in maintaining the stability of the phenolic compounds during the analytical process.

Acknowledgements The authors thank the Fundação de Amparo à Pesquisa do Estado de São Paulo (FAPESP, process 2015/15848-0) and the Brazilian National Counsel of Technological and Scientific Development (CNPq) for the financial support.

Compliance with Ethical Standards

Funding This study was funded by the Fundação de Amparo à Pesquisa do Estado de São Paulo (FAPESP), grant number 2015/15848-0.

Conflict of Interest Elem T. S Caramês declares that she has no conflict of interest. Priscila D. Alamar declares that she has no conflict of interest. Ronei J. Poppi declares that he has no conflict of interest. Juliana A. Lima Pallone declares that she has no conflict of interest.

Ethical Approval This article does not contain any studies with human participants or animals performed by any of the authors.

Informed Consent Not applicable.

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