

Determination of Triazole Fungicide Residues in Fruits by QuEChERS Combined with Ionic Liquid-Based Dispersive Liquid-Liquid Microextraction: Optimization Using Response Surface Methodology

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Abstract A rapid, efficient, and environmentally friendly method using quick, easy, cheap, effective, rugged, and safe (QuEChERS) combined with ionic liquid-based dispersive liquid-liquid microextraction (QuEChERS-IL-DLLME) prior to high-performance liquid chromatography coupled with photodiode array detection (HPLC-PDA) has been developed for the determination of six triazole fungicides (triazolone, triadimenol, epoxiconazole, flusilazole, tebuconazole, and diniconazole) in various fruits (pear, apple, and grapefruit). Several parameters affecting the extraction efficiency in IL-DLLME, such as type and volume of ionic liquid and acetonitrile volumes and extraction time, were investigated by single factor experiments. Then, the extractant volume, dispersant volume, and extraction time were optimized using response surface methodology (RSM). The optimal values were determined to be within an extractant volume of 63.7 μL , a dispersant volume of 0.43 mL, and an extraction time of 1.7 min, respectively. Under the optimum conditions, an

excellent linearity with determination coefficient higher than 0.997 was obtained. The average recoveries in three concentration levels (0.2, 0.5, and 1 mg kg^{-1}) ranged from 63.8 to 119.1 %, respectively, and the relative standard deviations (RSDs) from 1.1 to 12.6 %. The limits of detection (LODs) ($S/N = 3$) and limits of quantification (LOQs) ($S/N = 10$) for the six triazole fungicides ranged from 3.4 to 26.8 $\mu\text{g kg}^{-1}$ and 9.8 to 50.3 $\mu\text{g kg}^{-1}$, respectively. The proposed method was successfully applied for the determination of trace amounts of triazole fungicides in various fruits including pear, apple, and grapefruit.

Keywords Ionic liquid (IL) · Dispersive liquid-liquid microextraction (DLLME) · QuEChERS · Triazole fungicides · Response surface methodology · Fruits

Introduction

Pesticides have gained extensive applications to control and improve the quality of agricultural products in modern agriculture, which include insecticides, fungicides, herbicides, and other (LeDoux 2011). Among the major fungicides, triazole fungicides have been widely used in fruits, vegetables, and grain crops during cultivation and storage, thanks to their excellent protective, curative, and eradicator power against a wide spectrum of crop diseases (Kahle et al. 2008; Zhang et al. 2012). However, they are powerful endocrine disruptors and have been demonstrated to change the liver function, decrease kidney weight, and alter urinary bladder structure (Guducu et al. 2011). Therefore, it is necessary to develop sensitive and selective methods for the analysis of triazole residues usually present in trace amounts. Potential analytical techniques include high-performance liquid chromatography (HPLC) with UV, diode array detection (DAD), VWD, and

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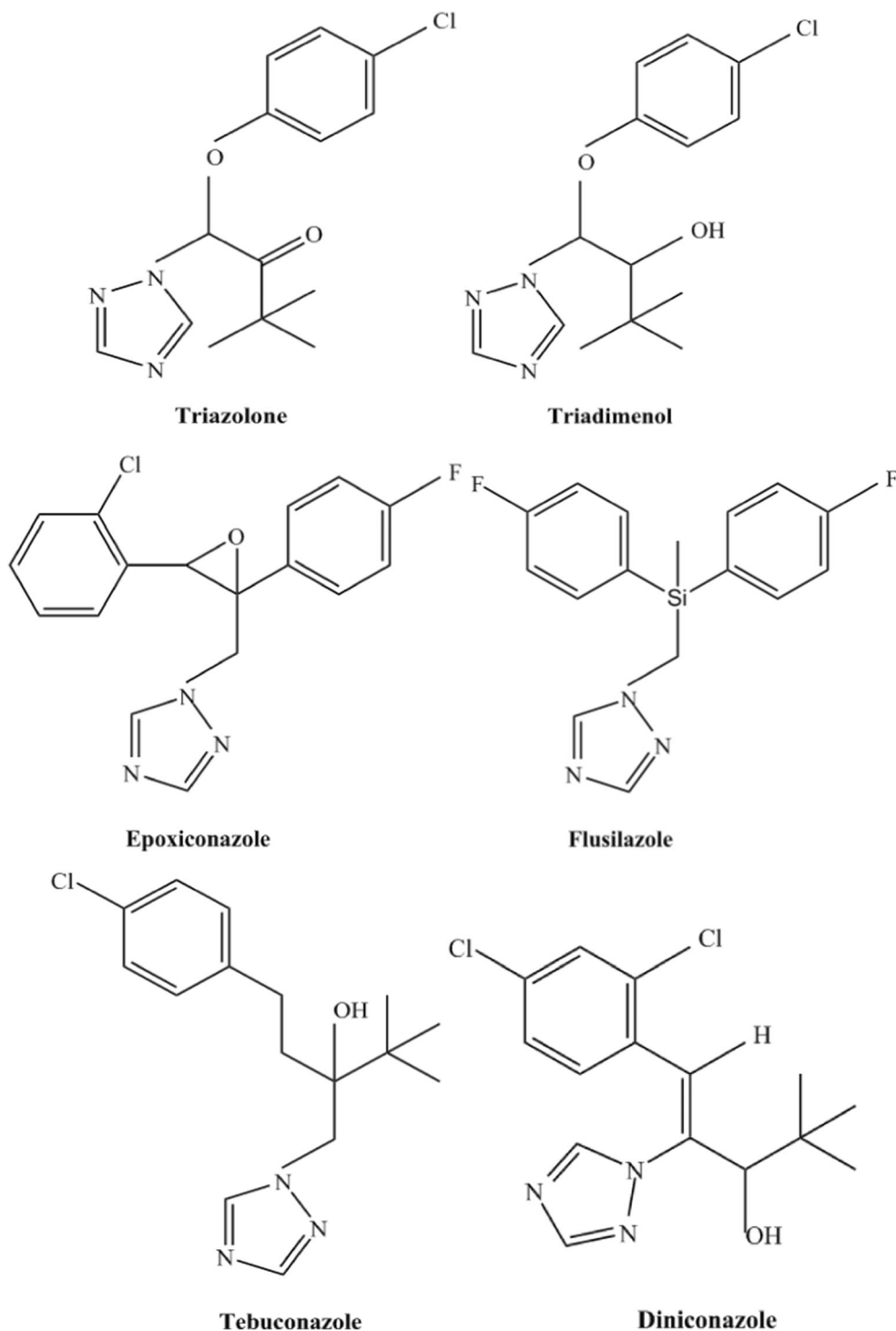
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photodiode array (PDA), respectively (Bordagaray et al. 2013; Bordagaray et al. 2014; Gao et al. 2012; Luo et al. 2013; Ye et al. 2012; Zhang et al. 2015), HPLC with tandem mass spectrometry (HPLC-MS/MS) (Li et al. 2013; Zhang and Xu 2014) and capillary electrophoresis (CE) with DAD (Rodriguez et al. 2001), gas chromatography (GC) with FID (Farajzadeh et al. 2010; Farajzadeh et al. 2011; Farajzadeh

et al. 2012; Farajzadeh et al. 2013; Freitas et al. 2014; Sarafraz-Yazdi et al. 2012), gas chromatography-mass spectrometry (GC-MS) (Farajzadeh et al. 2012; Freitas et al. 2014; Sarafraz-Yazdi et al. 2012), and gas chromatography-tandem mass spectrometry (GC-MS/MS) (Li et al. 2011).

Quick and effective sample preparation coupled with a reliable analytical technique is imperative. Liquid-liquid

Fig. 1 Chemical structures of six triazole fungicides



extraction (LLE) (Rezaee et al. 2006) and solid-phase extraction (SPE) (Sharif et al. 2006) are the most common sample preparation methods widely used for residue analysis. Recently, a growing number of studies have focused on two kinds of microextractions termed as liquid-phase microextraction (LPME) (Psillakis and Kalogerakis 2003) and solid-phase microextraction (SPME) (Kataoka et al. 2000), based on miniaturization of conventional LLE and SPE, respectively. As a novel LPME, DLLME has been recognized as a very popular preparation technique due to the simplicity of operation, time-saving, low cost, and high enrichment factor (Rezaee et al. 2006). However, hazardous solvents such as halohydrocarbon were frequently used as extraction solvents in the conventional DLLME. To overcome this problem, some low toxic and green solvents such as low-density alcohols and ionic liquids (ILs) have been successfully used as extraction solvents (Leong et al. 2014). Unfortunately, the lack of purification for samples with more complex matrices, such as fruits and vegetables, has caused DLLME to be limited to those with simpler matrices, specifically water and a few fruit juices (Zhang et al. 2014).

At present, “quick, easy, cheap, effective, rugged, and safe” (QuEChERS) sample preparation is the most common technique for multi-residue pesticide analysis in food, especially fruits and vegetables (Anastassiades et al. 2003). Although this technique has rapid cleanup ability, its poor enrichment capacity can lead to higher detection limits, i.e., lower sensitivity, compared with other techniques. Researchers proposed a new method comprised of DLLME preconcentration after QuEChERS extraction (Cunha and Fernandes 2011; Zhang et al. 2014; Zhao et al. 2007). Coupling these techniques takes advantages of the benefits of both methods while reducing some of their drawback. As a novel coupling sample preparation technique, QuEChERS-DLLME has been used for extracting and

enriching contaminant residues in not only water and fruit juices but also more complex matrices such as fruits and vegetables compared with DLLME.

To the best of our knowledge, there is no report on the extraction and enrichment of triazole residues in fruit using QuEChERS-DLLME or QuEChERS-IL-DLLME method. In this paper, a simple, rapid, and environmentally friendly method using QuEChERS-IL-DLLME followed by HPLC was applied for the determination of six triazole fungicides in fruits (pear, apple, and grapefruit). Figure 1 shows their structures. Several experimental parameters have been optimized by response surface methodology, and the optimized method was successfully applied to real samples.

Materials and Methods

Chemicals and Reagents

Triadimenol (98.0 % purity), epoxiconazole (98.5 % purity), flusilazole (98.0 % purity), tebuconazole (99.0 % purity), and diniconazole (99.0 % purity) were from Dr. Ehrenstorfer GmbH (Germany). Triazolone (99.7 % purity) was purchased from Sigma-Aldrich (USA).

HPLC-grade methanol, acetonitrile, 1-butyl-3-methylimidazolium hexafluorophosphate ($[C_4MIM][PF_6]$, >98 % purity), 1-hexyl-3-methylimidazolium hexafluorophosphate ($[C_6MIM][PF_6]$, ≥98 % purity), and 1-octyl-3-methylimidazolium hexafluorophosphate ($[C_8MIM][PF_6]$, >98 % purity) were from CNW (Germany). Sodium chloride and anhydrous magnesium sulfate were analytical reagent obtained from Sinopharm Chemical Reagent (Shanghai, China).

The standard stock solution of six triazole fungicides was prepared at the concentration of 100 mg L⁻¹ in methanol and stored in a glass volumetric flask at

Table 1 Analytical performance optimized for the determination of the six triazole fungicides

Fungicides	Linear range ($\mu\text{g L}^{-1}$)	Linear equation	Determination coefficient (R^2)	LOD ($\mu\text{g L}^{-1}$)	Precision (% RSD)			
					Intra-day ($n = 3$)		Inter-day ($n = 3 \times 3$)	
					Retention time	Peak area	Retention time	Peak area
Triazolone	10–10,000	$y = 22,673.08x - 914.63$	0.999	5	0.07	2.3	0.03	1.8
Triadimenol	30–10,000	$y = 23,061.65x + 1252.10$	0.997	10	0.04	2.2	0.02	2.4
Epoxiconazole	30–10,000	$y = 25,288.45x + 72.95$	0.999	10	0.04	2.3	0.03	1.0
Flusilazole	30–10,000	$y = 31,825.37x - 595.73$	0.999	10	0.04	1.5	0.02	0.7
Tebuconazole	30–10,000	$y = 18,367.42x - 661.91$	0.998	10	0.04	2.1	0.02	1.3
Diniconazole	10–10,000	$y = 32,025.16x - 1374.30$	0.997	5	0.05	0.8	0.02	0.8

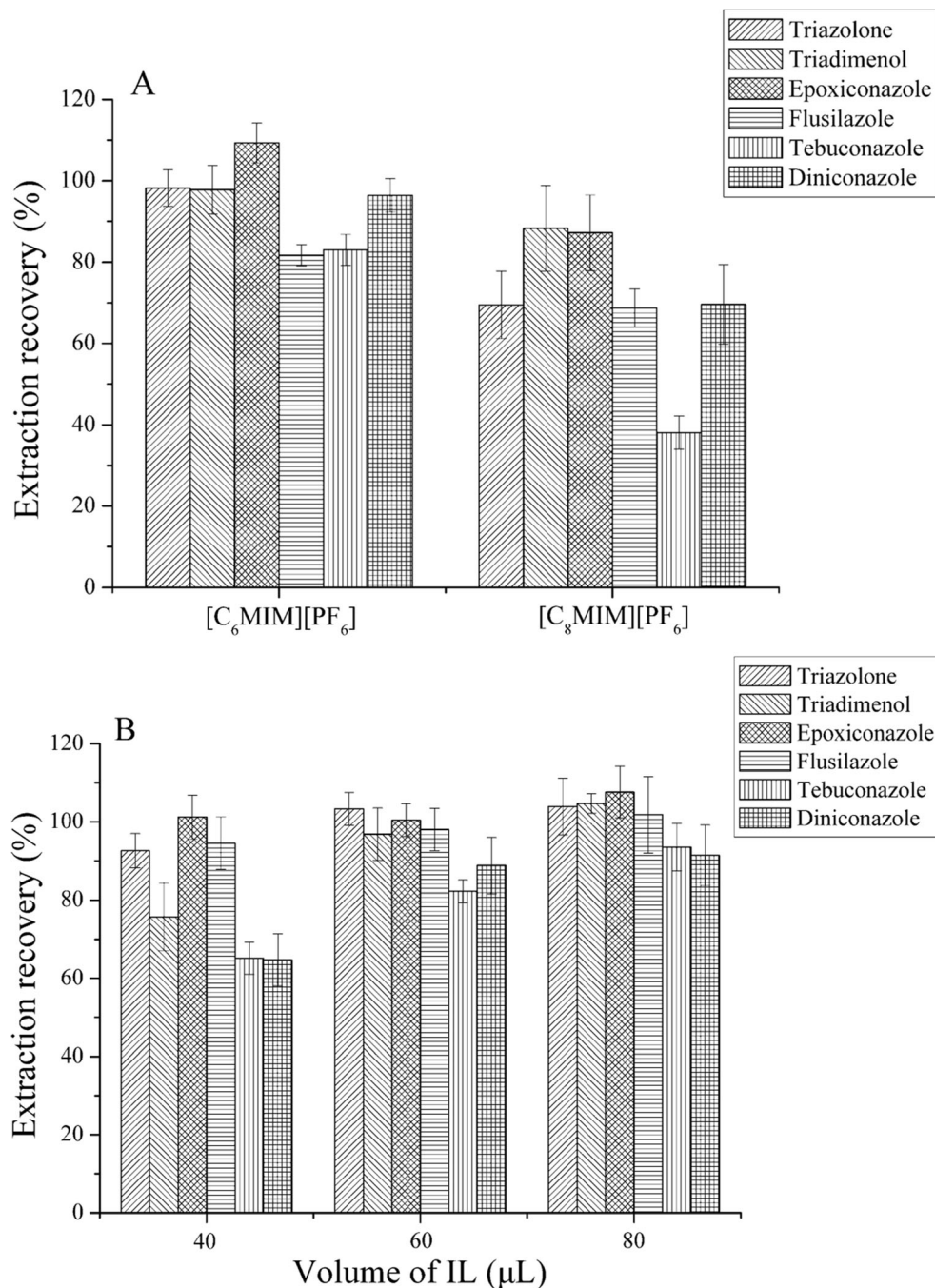
–50 °C. Standard working solutions at a series of concentrations were prepared by the dilution of aliquots of the stock solution with methanol and stored at 4 °C in a freezer. Deionized water (18 M Ω cm resistivity) from a Milli-Q Advantage A10 SP Reagent Water System (Millipore, Bedford, MA, USA) was used throughout.

Apparatus

A Waters Alliance e2695 Separations Module high-performance liquid chromatography (Waters Co.,

Milford, MA, USA) equipped with a PDA detector. Data were obtained and processed using Empower chromatography workstation. The separations were carried out in an XBridgeTM-C₁₈ column (Waters Co., Milford, MA, USA) (4.6 × 150 mm, 5 μ m). A CL31R refrigerated centrifuge (Thermo Scientific, USA) and GENIUS 3 vortex agitator (IKA, Germany) were used for the sample preparation. A Waters Quattro-Premier XE UPLC-MS/MS (USA) equipped with an Acquity BEH C₁₈ column (2.1 × 50 mm, 1.7 μ m) was used to perform the confirmation.

Fig. 2 Effects of **a** extractants and **b** their volumes on the recoveries of analytes in QuEChERS-IL-DLLME ($n = 3$). Five grams of spiked pear at 0.5 mg kg⁻¹



HPLC Analysis

The chromatographic separation was carried in gradient elution with a mobile phase of (A) water and (B) methanol as follows: 0 min, 60:40; 0.2 min, 60:40; 15 min, 40:60; 25 min, 60:40. The column temperature was 30 °C; the injection volume was 10 μL , and the flow rate was 1 mL min^{-1} . Selected as monitor wavelength for six analytes was 220 nm.

Validation Study

A test mixture with standard triazoles at a series of concentrations of 0.01, 0.1, 0.2, 0.5, 2, and 10 mg kg^{-1} (triazolone and diniconazole) and 0.03, 0.1, 0.2, 0.5, 2, and 10 mg kg^{-1} (the other four triazoles) was prepared in pure methanol not blank extract and analyzed under optimized conditions to determine linearity. Instrument precision and repeatability (intra- and inter-day variation) were determined using three replicates of

Fig. 3 Effects of **a** acetonitrile volumes and **b** extraction time on the recoveries of analytes in QuEChERS-IL-DLLME ($n = 3$). Five grams of spiked pear at 0.5 mg kg^{-1}

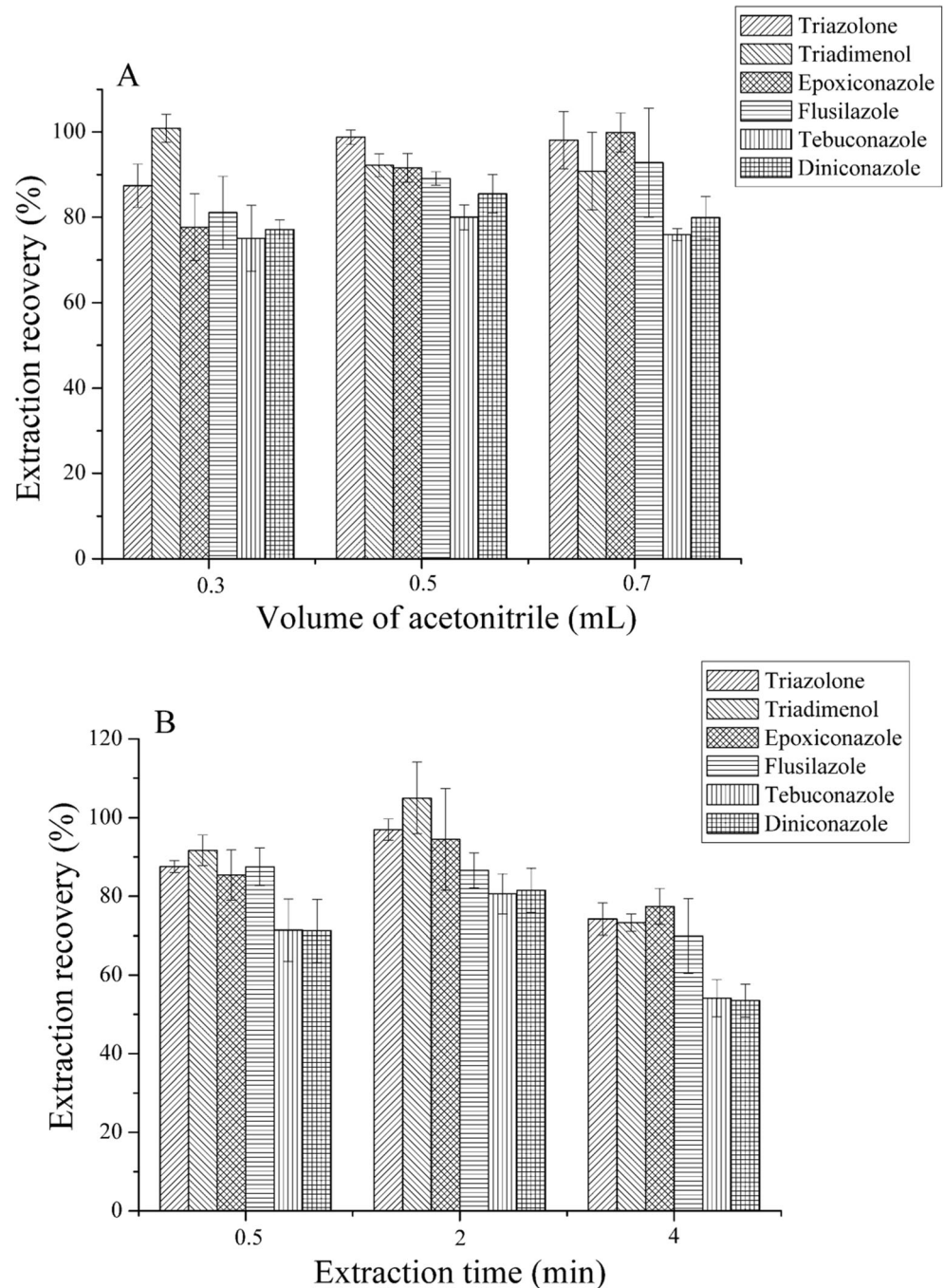


Table 2 Experimental variables and levels in the Box-Beknhen design matrix

Variables	Levels				
	Low (-1)	Central (0)	High (+1)		
(X_1) Extractant volume (μL)	50	60	70		
(X_2) Dispersant volume (mL)	0.3	0.4	0.5		
(X_3) Extraction Time (min)	1	2	3		
	Runs	X_1	X_2	X_3	Mean recovery (%)
	1	60	0.3	3	83.2
	2	50	0.5	2	93.1
	3	70	0.5	2	97.1
	4	60	0.4	2	99.2
	5	60	0.3	1	85.4
	6	60	0.5	3	84.5
	7	60	0.4	2	101.4
	8	70	0.4	3	94.0
	9	70	0.4	1	95.5
	10	50	0.4	1	94.2
	11	60	0.5	1	95.6
	12	50	0.4	3	88.4
	13	60	0.4	2	99.7
	14	70	0.3	2	86.3
	15	50	0.3	2	85.3

the standard working solution (0.5 mg L^{-1}). The precision was expressed as relative standard deviation (RSD, %).

A detailed study of matrix effects was performed by comparison of standards prepared in solvent and in matrix, a common way to test matrix effects. The differences were calculated using following equation (Botero-Coy et al. 2015). A positive difference value indicated matrix-induced signal enhancement, whereas a negative difference indicated signal suppression. We assumed that no relevant matrix effect occurred when differences were within $\pm 20\%$.

Recovery was performed by spiked blank samples (pear, apple, and grapefruit) at three different concentration levels (0.2 , 0.5 , and 1 mg kg^{-1}) with six replicates. Precision was expressed as RSD.

UPLC-MS/MS Confirmation

UPLC-MS/MS confirmation was performed according to the previously reported procedure (Zhang et al. 2015).

Samples

Pear, apple, and grapefruit were purchased from local supermarkets. Samples were homogenized before extraction to remove the sediments. Pear and apple were prepared in the form of whole fruit, while grapefruit pulp (2 kg each) was separate carefully into peel. A representative portion of these samples (200 g each) was chopped and

homogenized in a food chopper (HR 2095, Philips Electronics Co., Hong Kong, China).

QuEChERS-DLLME Procedure

The QuEChERS procedure described below was followed for extraction and cleanup (Zhang et al. 2014): (1) Weigh

Table 3 Estimated regression model of the relationship between the response variable (Y) and the independent variables (X_1 – X_3)

Source	Sum of squares	df	Mean square	F value	p value
Model	513.88	9	57.1	20.52	0.002
X_1	17.7	1	17.7	6.36	0.053
X_2	113.25	1	113.25	40.7	0.0014
X_3	53.05	1	53.05	19.06	0.0072
X_1X_2	2.25	1	2.25	0.81	0.4097
X_1X_3	4.62	1	4.62	1.66	0.2538
X_2X_3	19.8	1	19.8	7.12	0.0445
X_1^2	13.33	1	13.33	4.79	0.0802
X_2^2	221.77	1	221.77	79.7	0.0003
X_3^2	98.88	1	98.88	35.54	0.0019
Residual	13.91	5	2.78		
Lack of fit	11.25	3	3.75	2.82	0.2726
Pure error	2.66	2	1.33		
Cor total	527.79	14			

5.00 ± 0.01 g of sample into a 50-mL fluorinated ethylene propylene (FEP) centrifuge tube. (2) Add 5.00 mL acetonitrile into each tube to all samples and shake vigorously by hand for 1 min. (3) Keep the tubes in a refrigerator at least for 15 min at -20 °C. (4) Add 2.0 g anhydrous MgSO₄ and 0.5 g NaCl and shake vigorously by hand for 1 min. (5) Centrifuge at 10,000 rpm for 5 min. (6) Decant 1.00 mL extracts (upper layer) into the centrifuge tube containing 50 mg PSA. (7) Cap the tubes well and vortex them for 1 min. (8) Centrifuge at 4000 rpm for 5 min. (9) Transfer 0.43 mL extracts (upper layer) into a centrifuge tube, add 63.7 µL of [C₆MIM][PF₆] (as extraction solvent), and vortex for 1 min.

The DLLME procedure described below was followed for enrichment: (1) Weigh 5.00 g ± 0.01 g of deionized water into a sharp-bottom 15-mL FEP centrifuge tube. (2) Inject the above mixture quickly into water with a syringe to form cloudy solution. (3) Whirl vigorously for 1.7 min. (4) Centrifuge at 4000 rpm for 5 min. (5) Remove the sedimented phase and mix with methanol (1:1, v/v) in a trace intubation.

Results and Discussion

Optimum Separation Conditions

Six triazoles can be separated within 26 min. Although peaks of triadimenol 1 and 2 partially overlapped, they were together integrated without difficulty because the quantitative method depended on peak area sum of triadimenol enantiomers.

Table 1 summarizes the linearity, limits of detection (LODs), and reproducibility of peak area and retention time. The linearity of the method was tested using five different concentrations within the range of 10–15,000 µg L⁻¹ (triazolone and diniconazole) and 30–15,000 µg L⁻¹ (the other four triazoles), executing at least three replication injections. The results reveal a satisfactory linearity for all the analytes with the correlation coefficients (*R*²) higher than 0.997 in linear regression equation. Data on the regression equations are listed in

Fig. 4 3D response surfaces showing the effect of the different factors on the response (*Y*)

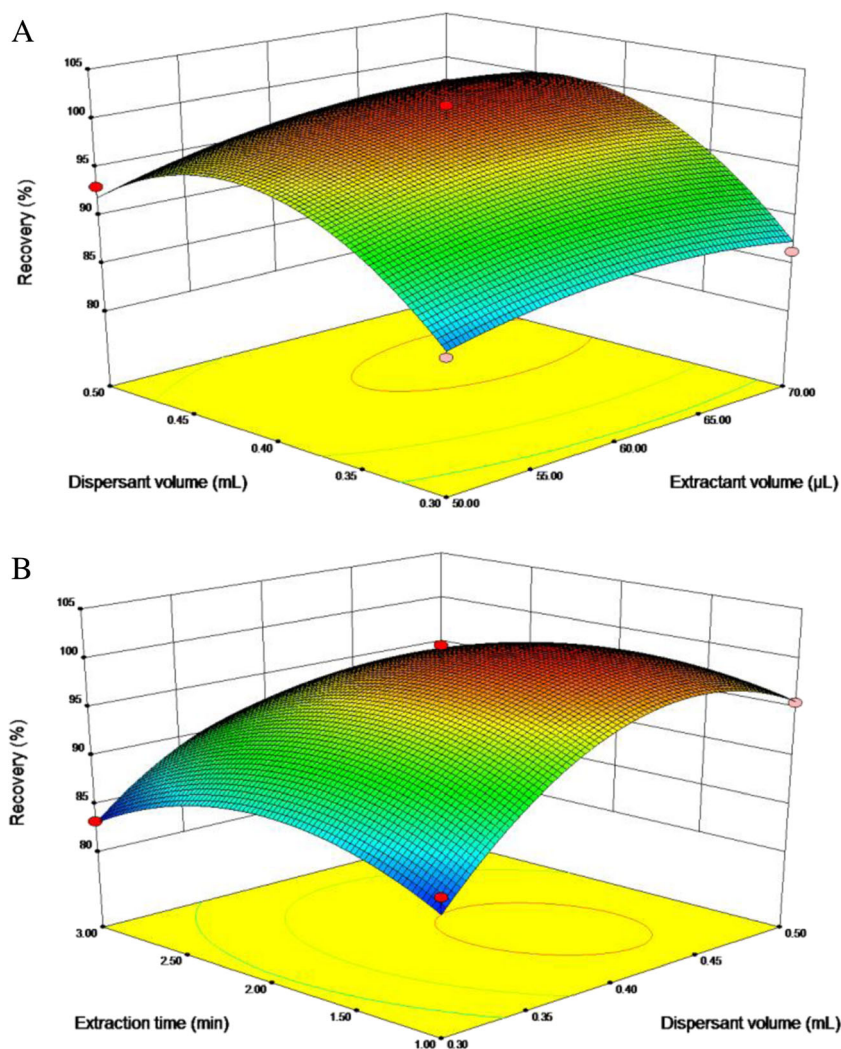


Table 4 Accuracy and precision obtained after QuEChERS-IL-DLLME of three spiked samples

Analytes	Added (mg kg^{-1})	Mean recoveries (%) / RSDs (%), $n = 6$			LOD ($\mu\text{g kg}^{-1}$)	LOQ ($\mu\text{g kg}^{-1}$)
		Pear	Apple	Grapefruit		
Triazolone	0.2	96.0/2.9	99.6/1.5	87.1/7.2	6.7	15.5
	0.5	94.8/6.8	98.2/3.8	100.3/4.6		
	1	118.3/7.6	113.3/4.1	92.5/12.4		
Triadimenol	0.2	99.2/4.1	110.0/10.8	108.8/8.8	26.8	50.3
	0.5	97.7/11.8	104.7/4.4	114.2/7.7		
	1	93.5/12.6	116.5/5.9	115.0/1.1		
Epoconazole	0.2	96.4/7.5	98.0/5.6	84.2/12.6	6.5	16.4
	0.5	90.0/9.6	92.5/9.3	83.8/10.3		
	1	105.6/7.2	119.1/1.6	89.7/11.1		
Flusilazole	0.2	84.9/12.5	63.8/1.7	95.3/5.6	5.1	12.6
	0.5	79.4/1.7	107.5/9.7	112.3/7.2		
	1	100.7/4.0	117.1/9.4	97.7/7.7		
Tebuconazole	0.2	64.8/8.0	63.9/2.4	76.1/12.6	15.7	37.2
	0.5	70.4/8.2	89.6/6.8	79.2/5.4		
	1	85.0/9.1	103.1/6.4	85.8/11.9		
Diniconazole	0.2	87.6/9.5	94.2/4.9	100.1/6.5	3.4	9.8
	0.5	87.6/12.5	101.8/1.3	84.5/1.3		
	1	84.9/9.2	91.9/4.6	81.9/7.8		

Table 1. The LODs ($S/N = 3$) of the six triazoles were between 5 and $10 \mu\text{g L}^{-1}$. The precision (RSDs) of the proposed method in terms of peak area for six replicate injections was 0.8–2.4 %. The RSDs in terms of retention time were between 0.02 and 0.07 %.

Optimization of QuEChERS-DLLME

Since the extraction solution of acetonitrile obtained after QuEChERS was used for dispersive solvent, the types and

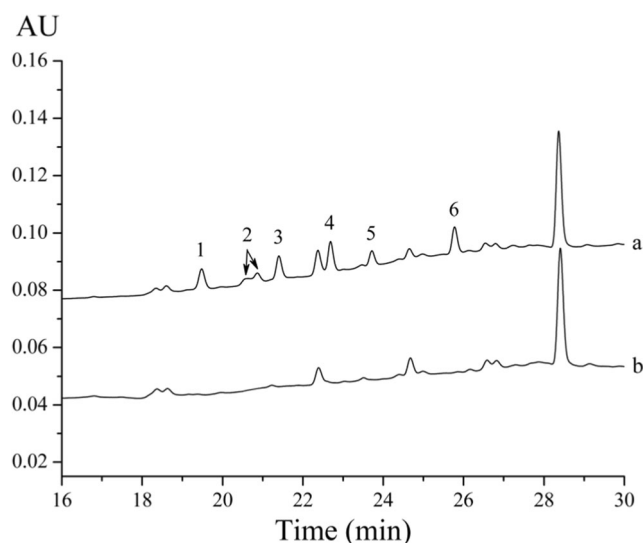


Fig. 5 Chromatograms **a** obtained from a pear spiked at 0.5 mg kg^{-1} after QuEChERS-IL-DLLME procedure under the optimal conditions and **b** obtained from the pear blank. Peak identification: triazolone (1), triadimenol (2), epoxiconazole (3), flusilazole (4), tebuconazole (5), and diniconazole (6)

volumes of extraction solvents were the only parameters to be optimized (Zhang et al. 2014). Several parameters affecting the extraction efficiency in IL-DLLME, such as type and volume of ionic liquid and acetonitrile volumes and extraction time, were investigated by single factor experiments.

The appropriate extraction solvent in IL-DLLME should meet several requirements including low water solubility, low volatility, and high extraction capability of analytes. In our study, three common ILs including $[\text{C}_4\text{MIM}][\text{PF}_6]$, $[\text{C}_6\text{MIM}][\text{PF}_6]$, and $[\text{C}_8\text{MIM}][\text{PF}_6]$ were used. Unfortunately, when $[\text{C}_4\text{MIM}][\text{PF}_6]$ was used, no sediment phase was found at the bottom of the tube after centrifugation, which was due to the higher solubility of $[\text{C}_4\text{MIM}][\text{PF}_6]$ than the other two ILs. Figure 2a showed that the better extraction recoveries were obtained using $[\text{C}_6\text{MIM}][\text{PF}_6]$ compared with $[\text{C}_8\text{MIM}][\text{PF}_6]$. Therefore, $[\text{C}_6\text{MIM}][\text{PF}_6]$ was used in subsequent experiments.

A series volumes of $[\text{C}_6\text{MIM}][\text{PF}_6]$ were evaluated for enrichment as follows: 5 g of pear was spiked with the standard solution at 0.5 mg kg^{-1} , 0.5 mL of acetonitrile was used for dispersant, and the extractant volume was changed from 40 to 80 μL in the interval of 20 μL . Observably, the extraction recoveries of the analytes were improved with the increase of the volume of $[\text{C}_6\text{MIM}][\text{PF}_6]$ (Fig. 2b). When $[\text{C}_6\text{MIM}][\text{PF}_6]$ was increased from 40 to 80 μL , the recoveries of the analytes were gradually enhanced. The maximal recoveries were mainly obtained at 80 μL . Thereby, 80 μL of $[\text{C}_6\text{MIM}][\text{PF}_6]$ was selected as the optimal extraction solvent volume.

In order to investigate the effect of dispersant volume, acetonitrile was varied from 0.3 to 0.7 mL in the interval

of 0.2 mL while the extractant solvent ($[C_6MIM][PF_6]$) was kept at 80 μ L. As shown in Fig. 3a, with the increase of acetonitrile from 0.3 to 0.5 mL, the extraction efficiency of most analytes increased gradually while the extraction efficiency of most analytes dropped down slightly above 0.5 mL. Therefore, 0.5 mL of acetonitrile was selected as the optimum dispersant volume to obtain an acceptable recovery.

In DLLME process, an appropriate extraction time is an important stage at which the extraction solvent is well dispersed into the sample solution, meanwhile enlarging the contacting area between the analytes and solution. The effect of extraction time was studied in the range from 0.5 to 4 min (Fig. 3b). The results revealed that the extraction efficiency increased from 0.5 to 2 min and then decreased slowly from 2 min. So, an extraction time of 2 min was chosen.

Optimization of Extraction Conditions by Response Surface Methodology

Response surface methodology (RSM) was carried out in order to establish regression equations between the dependent variables (the average recovery of six analytes) and three effectively independent variables, i.e., extractant volume (X_1), dispersant volume (X_2), and extraction time (X_3). Table 2 shows the variables and levels in the Box-Bekhen design matrix. The lower and upper levels for each variable were selected after running preliminary experiments for the variable.

The analysis of variance for RSM was carried out using the Design-Expert program and is shown in Table 3. In this experiment, the coefficients of X_2 , X_3 , X_2X_3 , X_2^2 and X_3^2 were statistically significant ($p < 0.05$), while the other coefficients were not statistically significant. The results were fitted with a second-order polynomial equation. The values of the regression coefficients were calculated, and the response variable and the test variables are related by the following second-order polynomial equation:

$$Y = 118.725 + 1.914X_1 + 657.125X_2 + 20.575X_3 + 0.750X_1X_2 + 0.108X_1X_3 - 22.250X_2X_3 - 0.019X_1^2 - 775.000X_2^2 - 5.175X_3^2$$

Compared with the single factor experiments, the response surface methodology is more accurate. The RSM, which is based on the single factor experiments, further optimizes the conditions. Moreover, the optimum conditions of single factor experiments which came from the designed conditions could not indicate the interactions between two parameters. The 3D response surface plot shown in Fig. 4 indicates the interactions of each two factors on the variation tendency of the average recovery of six analytes. We can get the significance of each parameter from the contour lines of the 3D response surfaces in Fig. 4.

The optimum conditions predicated by RSM were the extractant volume of 63.7 μ L, the dispersant volume of 0.43 mL,

Table 5 Comparison of the proposed methods and some DLLME methods for the determination of triazole fungicides in water and fruit juice

Instrument detector	Sample preparation	Analyte	Sample	LOD	Recovery (%)	Ref.
GC-FID GC-MS	DLLME	Diniconazole, tebuconazole	Water, grape juice	0.3–5.0 μ g L ⁻¹	74.0–99.0	Farajzadeh et al. 2012
GC-FID	AA-LLME, DLLME	Penconazole, hexaconazole, diniconazole, tebuconazole, triticonazole	Water	0.2–1.1 μ g L ⁻¹ 1.9–5.9 μ g L ⁻¹	92–105, 92–104	Farajzadeh et al. 2013
GC-FID	SEV-DLLME	Penconazole, hexaconazole, tebuconazole, diniconazole, triticonazole, difenconazole	Water, apple, grape juices	0.09–1.04 μ g L ⁻¹	–	Farajzadeh et al. 2011
GC-FID GC-MS	SBSE-DLLME	Penconazole, hexaconazole, diniconazole, tebuconazole, triticonazole, difenconazole	Water, apple, grape juices	0.53–24.0 μ g L ⁻¹	71–116	Farajzadeh et al. 2010
HPLC-DAD	DLLME	Hexaconazole, triadimefon, tebuconazole, penconazole	Water	8.5–29.0 μ g L ⁻¹	88.7–103.7	Luo et al. 2013
HPLC-UV	DLLME	Triadimefon, uniconazole, tebuconazole	Water	0.9–1.2 μ g L ⁻¹	90.6–105.3	Ye et al. 2012
HPLC-VWD	TC-IL-DLLME	Myclobutanil, tebuconazole	Water	0.3–0.8 μ g L ⁻¹	84.6–102.0	Gao et al. 2012
HPLC-PDA	VA-IL-DLLME	Triazolone, triadimenol, epoxiconazole, flusilazole, tebuconazole, diniconazole	Peach, apple, orange juices	0.4–6.7 μ g L ⁻¹	71.0–104.5	Zhang et al. 2015
HPLC-PDA	QuEChERS-IL-DLLME	Triazolone, triadimenol, epoxiconazole, flusilazole, tebuconazole, diniconazole	Pear, apple, grapefruit	3.4–26.8 μ g kg ⁻¹	63.8–118.3	Proposed method

AA air-assisted, SEV silylated extraction vessel, SBSE stir bar sorptive extraction, TC-IL temperature-controlled ionic liquid, FID flame ionization detector, VWD variable wavelength detector, DAD diode assay detector

and the extraction time of 1.7 min. To validate the predicted model, experiments were performed under the modified conditions.

Method Validation

Table 4 showed the average recovery value of the studied triazole fungicides ranging from 63.8 to 119.1 % (RSDs, 1.1–12.6 %). The LODs ($S/N = 3$) of the six triazoles were between 3.4 and 26.8 $\mu\text{g kg}^{-1}$. The LOQs ($S/N = 10$) of the six triazoles were between 9.8 and 50.3 $\mu\text{g kg}^{-1}$. The enrichment factors were in a range of 10.4 to 14.2 (pear, apple, and grapefruit). Figure 5 shows the chromatogram from the spiked pear at 0.5 mg kg^{-1} of each triazole fungicide obtained after QuEChERS-IL-DLLME. Obviously, QuEChERS-IL-DLLME is a very simple and effective method for preconcentrating triazole fungicides in fruit. Additionally, there was no interference peak in the typical chromatogram of blank pear after QuEChERS-IL-DLLME.

Table 5 summarizes the details of the proposed method and some other microextraction methods which were applied for triazole fungicide determination in water and fruit juices. Compared with most of the existing reports, more triazole fungicides were analyzed in this study. Most importantly, it was the first report on the extraction and enrichment of triazole residues in fruits using QuEChERS-IL-DLLME method.

Conclusions

In this study, a rapid, simple, and environmentally friendly QuEChERS-IL-DLLME method followed by HPLC-PDA was established to detect six triazole fungicides in pear, apple, and grapefruit. IL was selected as extraction solvent, and less-toxic organic solvent was used compared with the conventional DLLME process. Response surface methodology was used for the optimization of the extraction parameters affecting the extraction efficiency. The results of the method evaluation confirmed the method calibration, precision, and accuracy. The proposed method was successfully applied in the analysis of triazole fungicides in pear, apple, and grapefruit with satisfactory recoveries.

Compliance with Ethical Standards

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Conflict of Interest Yaohai Zhang declares that he has no conflict of interest. Yan Zhang declares that she has no conflict of interest. Jiyun Nie

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Ethical Approval This article does not contain any studies with human participants or animals performed by any of the authors.

Informed Consent Not applicable.

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