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Methane as a Substrate for Energy Generation Using Microbial Fuel Cells

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Abstract Methane (CH₄) is a well-known and abundant feedstock for natural gas, and is readily available from various sources. In thermal plants, the CH₄ generated from anthropogenic sources is converted into electrical energy via combustion. Microbial fuel cell (MFC) technology has proven to be an efficient strategy for the biological conversion of a many substrates, including biogas (CH₄), to electricity. MFC technology uses gaseous substrate along with an enriched and selective microbial consortium. Predominantly, methanotrophs and electrochemically active *Geobacter* were utilized in a syntrophic association on the anode of an MFC. This review focuses on the exploitation of CH₄ as a substrate for bioelectrogenesis via MFCs.

Keywords Microbial fuel cells · Greenhouse gases · Anaerobic methane oxidation · Reverse methanogenesis · Methanol

Methane (CH₄) and carbon dioxide (CO₂)—the key components of greenhouse gases (GHG)—have the potential to provide a promising platform for generating renewable and sustainable value-added products through biological and bioelectrochemical processes. The major emissions of CH₄ occur from natural and manmade anthropogenic activities.

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² Department of Chemical Engineering, College of Engineering, Qatar University, P O Box 2713, Doha, Qatar The natural sources include oceans, termites, and wetlands, which collectively account for approximately 36% of global CH₄ emission. The remaining portion of CH₄ emission (64%) majorly arises from human sources including the utilization, production, and transportation of fossil fuels. Minor quantities are also emitted from various sources, such as agriculture, landfill, and dairy farming (https:// www.epa.gov/ghgemissions/overview-greenhouse-gases# methane). This has led to doubling of atmospheric CH₄ levels over the last 150 years [1]. Besides, a comparative analysis of the impact of CH₄ and CO₂ on the environment as GHGs reveals that CH₄ is 25 times more dangerous than CO₂. This has led to the utilization of natural gas for electricity generation. However, current decline in power charges has led to a search of renewable and sustainable process for upgradation and valorization of CH₄. Therefore, the generation of value-added products utilizing CH₄ on-site can help in minimizing losses due to storage leaks and transportation [2]. Methane can be converted to liquid fuels or electricity via conventional technologies, such as chemical conversion or gas-turbine generator set. These technologies require a capital investment of billions of dollars as well as a large land area for operation. Direct use of CH₄ to generate electricity in fuel cells is challenging, owing to the requirement of high thermal operation (650-1100 °C) and instability of catalyst [1]. Thus, biological conversion of CH₄ to value-added products [methanol, electricity, or polyhydroxyalkanoates] seems an interesting and promising option [3, 4]. In addition, biological systems can offer adaptability/flexibility in scaling up of operation and provide an ability to integrate with the catalytic process that generates desired chemical products. However, there are several reviews available on the biological conversion of CH₄ to chemicals [5–7]. Therefore,

the present study provides a perspective on the conversion of CH_4 to bioelectricity.

The biological conversion of CH₄ to electricity has been challenging due to difficulties in microbial culturing, eventually affecting the process of anaerobic CH₄ oxidation. The first report on the use of CH₄ as a sole substrate was published in 1965, with pure cultures of Pseudomonas methanica [8]. Until 2011, no publications/reports were recorded on use of CH₄ by a similar process. In 2011, Girguis and Reimers, procured a US patent on the use of CH₄ as a feed for microbial fuel cells (MFCs) by using methanotrophs as a biocatalyst [9]. This turned the attention of the research fraternity to the bioelectrochemical conversion of CH₄ to electricity using MFCs. A recent study from the Pennsylvania State University, USA and National Institute of Cardiology, Mexico reported direct oxidation of CH₄ to electricity by reverse methanogenesis [10]. The anaerobic CH_4 -oxidizing bacteria (ANME) performed reverse methanogenesis. ANME strains were difficult to isolate due to their association with other bacteria that reduce metal oxides and other inorganic chemicals, such as nitrites and sulfates. Furthermore, ANME requires CH₄ activation without oxygen-based radicals and a longer growth time, even after adaptation for several days under laboratory conditions. Due to all these factors, the use of CH₄ in MFC has been challenging. Based on their previous work on a microbially engineered ANME strain (Methanosarcina acetivorans), Wood and his colleagues established a syntrophic association between M. acetivorans and Geobacter sulfurreducens [10]. They also analyzed the influence of undefined microbial inoculum, collected from a wastewater treatment plant. This mechanism generated sustainable and considerable amount of energy in the form of electricity. M. acetivorans synthesizes an Mcr (methylcoenzyme M reductase) enzyme and can convert CH₄ to acetate. G. sulfurreducens and other undefined microbial consortia can oxidize acetate to CO2 with simultaneous electricity generation. An eightfold improvement in power density (160 mW/m²; control-20 mW/m²) was recorded due to the presence of electroactive G. sulfurreducens species along with engineered M. acetivorans and sludge on the anode of double chamber MFC. Interestingly, the CH₄-fed MFC exhibited a columbic efficiency (CE) of 90%, which suggested that most of the electrons generated during CH₄ conversion are extracted in the form of current [10]. These values of CE were comparable with those obtained for conventional organic substrates in MFC [11]. Additional results exhibited a variable increase in current generation by use of external electron carriers. The use of cytochrome C and humic acids (0.5%) as mediators had shown 1.5- and 1.9-fold increase in power density, respectively, compared to control [12]. Further, an 11-fold increase in current generation was recorded by increasing the humic acid concentration from 0.5 to 3.3%, suggesting that internal electron carriers might be a rate-limiting factor in this system. However, further investigations need to be carried out to understand why *G. sulfurreducens* utilized extracellular electron transfer instead of using outer membrane or electrically conductive nanowires for pumping electrons to anode.

Similarly, other researchers have tried using CH₄ as an oxidant in MFC with major interest in decoupling of denitrifying anaerobic CH₄ oxidizing archaea (DAMO) [13]. This study demonstrated 25 mV of power generation. Interestingly, it provided an alternative for the successful separation of DAMO archaea to understand their physiological characteristics. After 45 days of operation, the MFC anode exhibited a 2.5-fold increase with DAMO archaea, with a 12-fold decrease in DAMO and simultaneous increase in ANME. Considerable increase in the abundance of the genera Geobacter and Ignavibacterium were observed on the anode. Recently, Chen and Smith had analyzed the use of CH₄ as a sole substrate in a single chamber air cathode MFC operated continuously [14]. MFC operation with a 16-h hydraulic retention time exhibited 85% CH₄ removal with a high power density of 62 mW/m². Reverse transcription-quantitative polymerase chain reaction analysis exhibited higher abundance of methanotrophs and the genus Geobacter. Instead of direct conversion of CH₄ to electricity in the anode chamber of MFC, few researchers had tried a two-stage system [15]. Initially CH₄ is converted to liquid fuels (methanol), and later, methanol is converted to electricity. The two-stage system generated a power density of 426 mW/m², which was found to be 6.8 times higher than single stage process (62 mW/m²). Methylophilus, Arcobacter, and Acetobacterium are the major genera found in methanol-fed MFC.

The long-term desire of anaerobic oxidation of CH₄ to produce electricity has been fulfilled. The above-discussed electrochemical studies open new possibilities in employing MFC as a biological post-treatment strategy for energy recovery and to mitigate GHG emission from anthropogenic sources (Fig. 1). CH₄ oxidation in MFC can also open a new approach for generation of value-added products and chemicals in microbial electrolysis cells (MECs). On the basis of current generation and the desired product, energy from MFC can be used for CO₂ reduction to generate alcohols and other carboxylic acids, as well as hydrogen (H₂), by using protons from the anode [16]. To further increase CH₄ conversion in MFC, the association of ANME with other electrochemical active bacteria (EAB) needs to be analyzed. This can include either the analysis of intermediates or extracellular electron transfer between ANME and EAB. The next-generation microbial sequencing tools, such as pyrosequencing or meta-omics, can help in better understanding of microbial communities

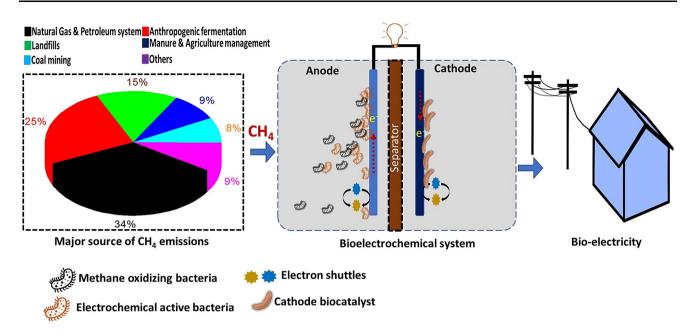


Fig. 1 Bioelectrogenic conversion of methane to electricity using microbial fuel cell with electroactive microbes

and gene expression, that are related to anaerobic oxidation of CH₄ in natural systems [17]. Meanwhile, MFC reactor configurations can be modified/developed for the use of gaseous substrates. In addition, the economic evaluation of MFC for electricity generation should be performed with CH₄.rich biogas (\sim \$2.6 per 1000 ft³).

Over the past few years, several aspects of MFC are being explored to enhance its performance. In this regard, the use of cheap natural gas for electricity seems interesting. However, the cost and energy conversion efficiency should be evaluated. The anode materials, such as gas diffusion electrodes, with specificity for methanotrophs should be tested to provide a better interface with syntrophic microbial community. In addition, there is a need to fabricate high conductive, scalable scaffolds, with better surface properties to enhance CH_4 oxidation by methanotrophs and EAB. However, there are still several bottlenecks, such as long-term operation and limiting the diffusion of CH_4 from the anode to cathode, which need to be evaluated to achieve a pilot scale of operation.

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