#### RESEARCH



# Conch Shell (*Turbinella pyrum*) Powder: A Potential Marine Biological Source of Calcium and Some Trace Minerals for Growing Crossbred Calves

Jagruti Bhagat<sup>1</sup> · Tapas Kumar Dutta<sup>1</sup> · Anupam Chatterjee<sup>1</sup> · Sushil Kumar Yadav<sup>1</sup> · Asif Mohammad<sup>1</sup> · Saroj Rai<sup>1</sup>

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#### Abstract

The present study was conducted to evaluate the effect of feeding conch shell (*Turbinella pyrum*) powder (either fresh or calcined) as a marine organic source of calcium (Ca) supplemented in the diet of crossbred calves on voluntary intake, growth performance, and blood biochemistry in growing crossbred Jersey calves. A growth trial of 90 days was conducted on 15 Jersey crossbred female calves (Av. weight,  $70.68 \pm 2.90$  kg; Av. age,  $197.73 \pm 12.40$  days), equally divided into three groups of 5 animals each, i.e., control ( $T_0$ ), treatment 1 ( $T_1$ ), and treatment 2 ( $T_2$ ). All animals were fed total mixed ration (TMR) prepared with a concentrate mixture, chaffed paddy straw, and green fodder at the ratio of 40:30:30 on DM basis. Calves under the control group were fed with TMR containing a standard mineral mixture having dicalcium phosphate (DCP) as a Ca source. Calves under  $T_1$  group were supplemented with TMR containing fresh conch shell powder (FCSP), and  $T_2$  calves were fed with TMR containing conch shell calcined powder (CSCP) as Ca source. We observed 11.66% increase (p < 0.01) in Ca concentration in CSCP compared to FCSP. The concentration of minerals like Mg, Co, Mn, and Fe was enhanced in CSCP compared to the FCSP. However, the calcination process of fresh conch shell powder (FCSP) reduced the concentration of Cu, and Zn. The Ca/P ratio was estimated as 2.11, 2.06, and 2.10 in T<sub>0</sub>, T<sub>1</sub>, and T<sub>2</sub> diets, which could be considered ideal for calf ration. Calves under  $T_1$ , and  $T_2$  groups consumed significantly (p < 0.001) greater amounts (g/kg W<sup>0.75</sup>) of DM and CP compared to T<sub>0</sub>. However, increased voluntary intake did not translate into increased body weight gain (kg), and feed conversion ratio (kg DMI/kg gain) in  $T_1$  and  $T_2$  groups in comparison to  $T_0$ . We observed similar blood glucose, urea, alkaline phosphatase (ALP), aspartate aminotransferase (AST), and alanine transaminase (ALT) concentration among the three treatments. Ca, and P levels in blood plasma were also identical among the three groups. The digestibility of Ca was increased significantly (p = 0.01) in FCSP (T<sub>1</sub>)- and CSCP (T<sub>2</sub>)-treated calves compared to control (T<sub>0</sub>) calves. Similarly, T<sub>1</sub> and  $T_2$  enhanced P digestibility compared to  $T_0$ . This first report with short-term experimentation depicted some promising scope for the use of locally available conch shell powder (fresh or calcined form) as a potential source of Ca for feeding to livestock, because these new sources of Ca did not affect intake, digestibility of Ca and P, growth performance, blood chemistry, and liver enzymes negatively in weaned crossbred calves.

Keywords Blood chemistry · Conch shell · Calcium · Calves · Calcined · Growth · Powder

☑ Tapas Kumar Dutta tkdcirg@gmail.com; Tapas.Dutta@icar.gov.in

Jagruti Bhagat jagrutibhagat8@gmail.com

Anupam Chatterjee anuchatterjee@gmail.com

Sushil Kumar Yadav sushil.yadav164@gmail.com Asif Mohammad mail.asif.m@gmail.com Saroj Rai drsaroj.rai@gmail.com

<sup>1</sup> Eastern Regional Station, ICAR-National Dairy Research Institute, Kalyani, West Bengal Pin - 741 235, India

#### Introduction

The total coastal length in India (including Andaman and Nicobar Islands and Lakshadweep) is 7516.6 km [1] which has plenty of sea/marine shells [2]. Bay of Bengal, Arabian Sea, and Indian Ocean, surrounding Indian subcontinent, are wealthier in biological resources of calcium (Ca) and other important minerals. The major sources of Ca include crustacean shells, fish bones, coral, shell fish, and seaweed. Sometimes, marine processing leftovers are considered unusable; however, it is a copious and cheaper source of Ca [3]. "Turbinella pyrum," a marine animal of "mollusc" group under the family Turbinellidae [4], is used as conch shells or Shankha (Chart 1) in India. Molluscs are a class of animals that have no bones and have a hard outer shell [5]. Hindu religious groups in India use Shankha and ladies' Bangles, commercially produced from conch shells (Chart 1) which are fished in southern coastal lines in India and Sri Lanka. Fished conch shells are processed in several small-scale industries located in the state of West Bengal, India. The conch shell powder is the by-product produced during the process of hand-crafting of Shankha, Bangles, and other valued products (Chart 1). The calcined product of conch shell powder is known as Shankha Bhasma which is traditionally used in Ayurvedic medicines to treat many ailments in human beings [6] like peptic ulcers, piles, cough, and certain gastrointestinal disorders [7]. Minerals accomplish various basic physiological functions within the tissues and organs; so, insufficiency or surplus of one or more of them may result into metabolic dysfunctions that may cause low productive and reproductive efficiency [8]. Ca is an essential macro-mineral required for appropriate bone health, integrity, and metabolism in livestock and human beings. Ca enters the circulation through the feed/ food chain or Ca supplemented through the mineral mixture, and a dynamic homeostatic balance is maintained between blood and bone Ca [9]. Growing calves need to be supplied with adequate amounts of Ca in order to accomplish acceptable performance, as well as sufficient bone mineralization. Lack of information about Ca availability in different feed resources may impair the accuracy of ruminant's ration formulation [10]. In recent years, Ca supplements of marine origins have gained considerable attention due to their copious reserves, higher biological safety, and activity [11, 12].

The production of meat (chicken, pork, chevon, mutton, and beef) and milk will accelerate to meet the increasing demand of human population. Hence, the requirements for different nutrients, including minerals, will also increase in the future. Animals can never fully utilize their dietary Ca and P, as some of them are inaccessible or lost during normal digestion and metabolic processes or have an unfavorable impact on their absorption due to a number of dietary or non-dietary variables [13]. The utilization of Ca and P in animals depends upon their bioavailability in feed resources [14]. However, an inappropriate diet of livestock and poultry birds may limit the bioavailability of Ca and P. For example, the presence of phytic acid in cereal grains and oxalic acid in some green fodder can cause Ca to precipitate as calcium phytate and calcium oxalate, which are insoluble compounds [15, 16]. An adequate supply of Ca in the diet could entirely be provided by animal-based and inorganic mineral salts, the most common sources of which include ground limestone, calcium chloride, mono- and dibasic calcium phosphate, and oyster shells [17]. However, the cost of mineral mixtures is a concern these days due to the increased cost of various mineral salts. Instead of purchasing expensive mineral salt, we may use locally available mineral resources of natural origin. For example, Ca from CaCO<sub>2</sub> can be obtained from marine shells or wastes such as crustaceans and bivalve shells. The seashells are composed of more than 95% inorganic minerals and a small quantity of organic matrix [18, 19]. Xu et al. [3] reported that the marine shell, an organic source of Ca, contained about 95% CaCO<sub>2</sub>. CaCO<sub>3</sub> in marine shells is commonly found as calcite and aragonite in adult individuals and often amorphous CaCO<sub>3</sub> (ACC) in young animals [20].

Conch shell powder, along with another P source, can replace the commonly used dicalcium phosphate (DCP) in the mineral mixture for feeding to livestock. Since no published reports are available on the use of conch shell (*Turbinella pyrum*) powder (either fresh or calcined) in the mineral mixture supplemented in the diet of livestock, the present study was conducted to assess the effect of replacing DCP with fresh conch shell powder or conch shell calcined powder (a marine organic source of Ca) on voluntary intake, growth performance, blood chemistry, and liver enzymes in growing crossbred Jersey calves.

# **Materials and Methods**

#### Site of the Experiment

The experiment was conducted in the Animal Nutrition Laboratory and Cattle Yard Complex of ICAR-National Dairy Research Institute, Eastern Regional Station, Kalyani, West Bengal (located in the lower Gangetic region of India) as per the approval of the University (20-M-AN-01 of 2020). The latitude and longitude position of the study site is 22°56'30"N and 88°32'04"E, respectively. The experiment was divided into two phases, i.e., I and II.

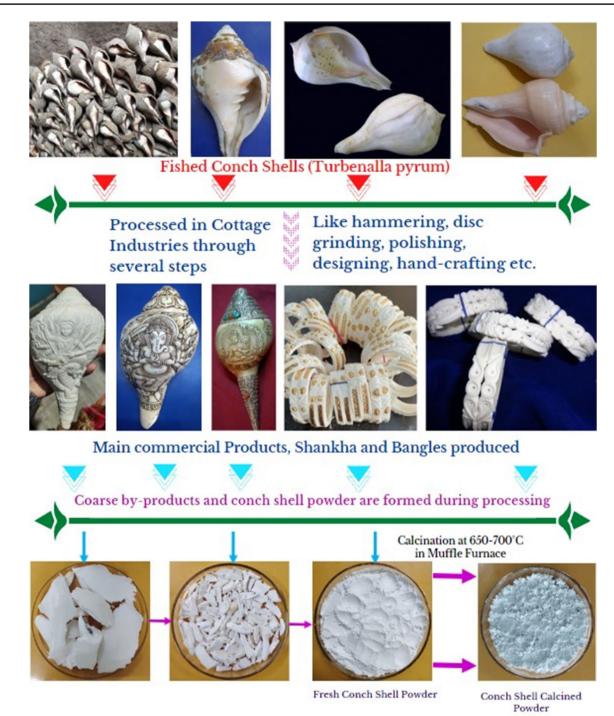


Chart 1 Processing of conch shells to different commercial products and generated by-products

# Phase-I: Physico-chemical Evaluation of Fresh Conch Shell Powder and Conch Shell Calcined Powder

The sufficient quantity of fresh conch shell powder (FCSP) was procured from Rupam Sankha Silpalaya (registered Industry), Kolmijore, Paschim Midnapur, West Bengal, India, for conducting the experiment. Conch shell calcined powder (CSCP) was prepared by ashing part of FCSP in the muffle furnace at 650–700 °C for 7–8 h. FCSP and CSCP were subjected to solubility test using different inorganic acids and water.

#### **Solubility Test**

The weighed samples (5 g) of FCSP and CSCP were allowed to dissolve in 25 ml of distilled water, normal tap water (source deep tube well), and concentrated acids like  $H_2SO_4$ , HCl, HNO<sub>3</sub>, and HClO<sub>4</sub>. The samples were kept

for dissolution for 18 h. The residual samples were filtered through repeated washing with double-distilled water on pre-weighed filter paper no. 1 until the sample present in the beaker was completely filtered. Then, the filter paper, along with undissolved part of the sample, was kept in a hot air oven until complete drying at  $80.0 \pm 5.0$  °C.

Solubility  $\% = 100 - \frac{(\text{wt. of filter paper + wt. of undissolved sample) - wt. of filter paper}{\text{wt. of sample}} \times 100$ 

#### **Estimation of Minerals**

Minerals such as Ca, Mg, Cu, Zn, Mn, Fe, and Co in FCSP, CSCP, feeds, fodders, fecal samples, and water samples were analyzed by using Atomic Absorption Spectrophotometer (Agilent 240AA model). P content in these feeds and mineral resources was evaluated as per the method described by O'Dell [21].

#### Evaluation of Density and pH

The density of both FCSP and CSCP was evaluated by using the following formula,

Density of powder  $(g/cm^3)$  = weight  $(g)/volume (cm^3)$ .

The pH of each powder was estimated by dissolving 10 g of each powdered sample in 100 ml of double distilled water. A digital pH meter (Elico, India) was used for the purpose.

# Phase-II: Assessment of Fresh Conch Shell Powder and Conch Shell Calcined Powder in Crossbred Calves

#### **Experimental Animals**

The experiment was conducted in 2022 on 15 healthy female Jersey crossbred calves (av. weight,  $70.68 \pm 2.90$  kg; av. age,  $197.73 \pm 12.40$  days) divided into 3 equal treatments (5 calves in each group), i.e.,  $T_0$ as control,  $T_1$ , and  $T_2$ . The average initial body weight and age of calves were statistically similar among the three treatments. All calves were housed individually in the well-ventilated experimental shed under similar management conditions. Ad libitum clean and fresh drinking water was offered twice daily to all animals at around 10.00 AM and 4.00 PM. The experimental shed and calves were cleaned regularly throughout the trial period.

#### **Experimental Feeding**

The growth trial in crossbred calves continued for 90 days. Total mixed rations (TMRs) (calculated CP 12% and TDN 65%) were prepared separately for each group using concentrate mixture containing separate mineral mixture (Table 1), chaffed paddy straw, and green fodder at the ratio of 40:30:30 on DM basis. We used seasonal oat fodder during the initial 2 months of the experiment, followed by maize fodder, to prepare TMRs for the last month of the experiment.

**Control group (T<sub>0</sub>):** Calves under the control group were fed ad libitum TMR (CP 12% and TDN 65%) with a mineral mixture containing dicalcium phosphate (DCP) as a source of Ca and P.

**Treatment group** ( $T_1$ ): Ad libitum TMR (CP 12% and TDN 65%) with a mineral mixture containing fresh conch

 Table 1
 Composition (%) of mineral mixture and total mixed rations (TMRs) for different experimental groups

Composition	T <sub>0</sub>	T <sub>1</sub>	T <sub>2</sub>
Composition of mineral mixture	* (%)		
Dicalcium phosphate	97.852	-	-
Fresh conch shell powder	-	42.945	-
Conch shell calcined powder	-	-	39.480
Diammonium phosphate	-	55.829	59.220
CuSO <sub>4</sub>	0.682	0.389	0.413
ZnSO <sub>4</sub>	1.355	0.773	0.820
MnSO <sub>4</sub>	0.070	0.040	0.043
CoSO <sub>4</sub>	0.040	0.023	0.024
Total	100.00	100.00	100.00
Composition of total mixed ratio	ons (% on DM	(basis)	
Concentrate mixture**	40	40	40
Green fodder	30	30	30
Paddy straw	30	30	30
Total	100	100	100

\*\*Concentrate mixture (CP 20.79%, TDN 74.56%) prepared with maize grain 35%, wheat bran 24%, groundnut cake 14%, mustard oil cake 24%, \*respective mineral mixture 2%, and common salt 1%

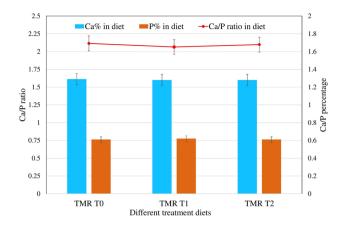


Fig. 1 Ca and P levels as well as Ca/P ratio in the diets as affected by different sources of calcium

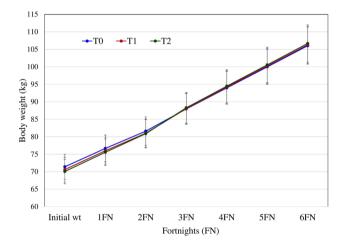


Fig. 2 Growth of crossbred calves affected by different sources of calcium

shell powder (FCSP) as a source of Ca and diammonium phosphate as a source of P.

**Treatment group**  $(T_2)$ : Ad libitum TMR (CP 12% and TDN 65%) with a mineral mixture containing conch shell calcined powder (CSCP) as a source of Ca and diammonium phosphate as a source of P.

Three experimental mineral mixtures were prepared using the ingredients mentioned in Table 1. The percentage of calcium and phosphorus was kept similar among all mineral mixtures (Fig. 1).

# Observations

#### **Chemical Composition of Feeds**

Throughout the experimental period, representative samples of feeds and residues were taken and analyzed in triplicate in the Animal Nutrition Laboratory for Proximate Principles [22] and cell wall fractionations [23].

#### Voluntary Intake of Nutrients, Body Weight, and Feed Conversion Efficiency

The body weight of experimental calves was recorded before the start of the experiment and during different phases (fortnightly) of the trial period by using the ISI-branded digital platform balance (Fig. 2). Fortnightly dry matter intake (DMI) by individual calf was measured by deducting the DM of the residue from the DM of the total feed offered. Similarly, organic matter intake (OMI) and crude protein intake (CPI) patterns were recorded in all experimental animals on fortnightly basis. Average daily body weight gain (ADG), feed conversion efficiency (FCE), and feed conversion ratio (FCR) were calculated based on the following formulae.

Average daily body weight gain $(ADG) =$	weight gain (kg)		
Average daily body weight gain (ADG) =	Number of days of observation		
	weight gain (kg)		
Feed conversion efficiency (FCE) = $\frac{1}{dry}$	matter intake (kg DMI) × 100		
average da	aily dry matter intake (kg DMI)		
Feed conversion ratio (FCR) = $\frac{average}{average}$ d	aily body weight gain (kgADG)		
-			

Similarly, FCE and FCR were also calculated based on the total crude protein intake (CPI) by the calves under different treatment groups.

#### **Blood Metabolites and Enzymes**

Blood samples were collected from the jugular vein of all experimental calves at 0, 45, and 90 days of feeding under aseptic conditions. Some relevant blood metabolites (urea, glucose), minerals (Ca, P), and liver enzymes (alkaline phosphatase (ALP), aspartate aminotransferase (AST), alanine transaminase (ALT)) were analyzed. Blood glucose was estimated using the commercially available glucose kit (GOD-POD) method (Span Diagnostic Ltd., India). Blood urea was estimated using the commercially available Urea Test Kit (urease Berthelot, endpoint assay) method (Span Arkray Healthcare Pvt. Ltd., India). Blood Ca and P were estimated using the commercially available Liquixx Calcium and Phosphorus test kit of Trans Asia Bio Medicals Ltd., India. Blood ALT was evaluated using the commercially available ALT Test Kit of Coral Clinical Systems (P) Ltd., India. We evaluated Blood AST using the commercially available AST Test Kit of Coral Clinical Systems (P) Ltd., India. Blood ALP activity was also measured by the method of pNPP Kinetic method using Diagnostic Kit manufactured by Arkray Healthcare Pvt. India.

#### **Digestibility Trial**

A digestion trial of 6-day collection period was conducted on 15 animals (5 experimental calves under each group) by total collection method after the completion of 90 days of the growth trial to study Ca and P intake, outgo through feces, and digestibility. The digestion trial was conducted following the detailed method described by Anil et al. [24]. Ca and P content of feeds offered, residues left, and fecal samples were evaluated as per the methods described in the earlier section.

#### **Statistical Analysis**

Data related to the chemical and mineral composition of different samples were statistically analyzed by using oneway analysis of variance (ANOVA) [25]. The intake of nutrients, growth performance, feed conversion efficiency, and blood parameters were analyzed by a mixed ANOVA with repeated measures, where treatment was a fixed factor (between subject effect), time period (within-subject) was a random factor having subject (calf) nested within it, and their (period×treatment) interaction was evaluated. A computerized IBM SPSS 26.0 software program was used for ANOVA. Tukey's HSD test was used to measure the differences of means. The levels of significance were considered at p < 0.05, and p < 0.01.

# Results

# Evaluation of Physicochemical Properties of FCSP and CSCP

The FCSP is white colored, while the CSCP is slightly grayish (Chart 1). The mean organic matter, total ash, and acid insoluble ash (AIA) of FCSP were estimated  $10.62 \pm 16.16$ ,  $89.17 \pm 1.79$ , and  $0.96 \pm 0.01$ , respectively. The density (g/cm<sup>3</sup>) of FCSP and CSCP was measured as  $0.25 \pm 0.0004$  and  $0.40 \pm 0.0004$ , respectively. The pH of FCSP and CSCP, suspended in doubled distilled water,

 Table 3
 Mineral composition of fresh conch shell powder (FCSP)

 and conch shell calcined powder (CSCP)

Minerals	FCSP	CSCP	Pooled mean $\pm$ S.E.M	p value
Ca (%)	34.89 <sup>a</sup>	38.96 <sup>b</sup>	$37.04 \pm 0.84$	p=0.005
P (%)	0.23 <sup>b</sup>	0.006 <sup>a</sup>	$0.08 \pm 0.05$	p < 0.001
Mg (ppm)	108.53 <sup>a</sup>	175.53 <sup>b</sup>	$144.95 \pm 7.51$	p < 0.001
Co (ppm)	22.74 <sup>a</sup>	23.53 <sup>b</sup>	$23.01 \pm 0.17$	p = 0.035
Mn (ppm)	7.41 <sup>a</sup>	10.31 <sup>b</sup>	$8.99 \pm 0.33$	p < 0.001
Fe (ppm)	101.83 <sup>a</sup>	196.23 <sup>b</sup>	$153.48 \pm 10.58$	p < 0.001
Cu (ppm)	12.57 <sup>b</sup>	4.22 <sup>a</sup>	$7.57 \pm 0.95$	p < 0.001
Zn (ppm)	779.46 <sup>b</sup>	307.18 <sup>a</sup>	$506.89 \pm 52.00$	<i>p</i> <0.001

The values with different superscripts (a, b) between two Ca sources differ significantly

was 9.80 and 12.02, respectively. The solubility of Carich powders (FCSP, and CSCP) was found to be highest (p < 0.001) in concentrated HCl followed by HNO<sub>3</sub>, HClO<sub>4</sub>, H<sub>2</sub>SO<sub>4</sub>, tap water, and distilled water (Table 2). However, the overall solubility of FCSP was higher (p < 0.001) than CSCP. The interaction of treatments × solvents was also significant (p < 0.001).

The concentration (%) of Ca was found to be significantly greater (p = 0.005) in CSCP compared to FCSP, wherein the P content was depressed (p < 0.001) in CSCP due to the calcination of FCSP (Table 3). We observed an 11.66% increase in Ca concentration in CSCP compared to FCSP. The concentration of minerals like Mg, Co, Mn, and Fe was enhanced in the CSCP compared to the FCSP (Table 3). However, the calcination process of FCSP reduced the concentration of Cu and Zn, which could be due to the partial vaporization of these essential minerals during the calcination process at high temperatures (> 650 °C).

# Evaluation of Chemical and Mineral Composition in Different Feeds

All feed ingredients (oat fodder, paddy straw, maize grain, wheat bran, groundnut cake (GNC), and mustard oil cake

Solvents	Solubility % FCSP	Solubility% CSCP	Pooled mean $\pm$ S.E.M	Significance
Distilled water	5.99	3.98	$4.99^{b} \pm 0.45$	Treatment effect (T), $p < 0.001$
Tap water	5.47	3.42	$4.44^{a} \pm 0.46$	Solvent effect (S), $p < 0.001$
HCl	99.21	99.11	$99.16^{f} \pm 0.03$	T×S effect, $p < 0.001$
HNO <sub>3</sub>	97.41	98.14	$97.77^{e} \pm 0.19$	
$H_2SO_4$	40.64	43.95	$42.29^{\circ} \pm 0.76$	
HClO <sub>4</sub>	98.10	94.46	$96.28^{d} \pm 0.82$	
Pooled mean $\pm$ S.E.M	$57.80^{B} \pm 10.21$	$57.18^{A} \pm 10.25$	$57.49 \pm 7.13$	

Table 2 Solubility % of fresh conch shell powder and conch shell calcined powder in different solvents

The values with different superscripts (A, B; and a, b, c, d, e, f) among different treatments and between two Ca sources differ significantly

Parameter	Oat fodder	Paddy straw	Maize grain	Wheat bran	GNC	MOC
Chemical compos	ition (%)					
DM	$24.30 \pm 0.002$	$90.18 \pm 0.11$	$94.93 \pm 0.25$	$96.38 \pm 0.02$	$92.98 \pm 0.29$	$91.85 \pm 0.13$
СР	$7.47 \pm 0.07$	$3.66 \pm 0.06$	$8.28 \pm 0.15$	$15.52 \pm 0.26$	$43.22 \pm 0.02$	$33.09 \pm 0.31$
OM	$92.15 \pm 0.02$	$85.13 \pm 0.03$	$96.86 \pm 0.01$	$96.28 \pm 0.13$	$95.58 \pm 0.05$	$92.10 \pm 0.03$
EE	$1.80 \pm 0.01$	$1.23 \pm 0.02$	$4.20 \pm 0.05$	$3.71 \pm 0.03$	$7.23 \pm 0.03$	$7.17 \pm 0.03$
ТСНО	$82.77 \pm 0.07$	$80.23 \pm 0.02$	$83.84 \pm 0.19$	$77.06 \pm 0.24$	$45.13 \pm 0.07$	$51.85 \pm 0.33$
Total ash	$7.85 \pm 0.02$	$14.87 \pm 0.02$	$3.14 \pm 0.01$	$3.71 \pm 0.13$	$4.41 \pm 0.05$	$7.89 \pm 0.028$
NDF	$66.23 \pm 0.13$	$78.28 \pm 0.02$	$22.87 \pm 0.003$	$30.33 \pm 0.13$	$27.39 \pm 0.04$	$25.41 \pm 0.009$
ADF	$38.17 \pm 0.02$	$53.35 \pm 0.31$	$5.85 \pm 0.10$	$9.60 \pm 0.01$	$14.28 \pm 0.12$	$16.27 \pm 0.08$
Hemicellulose	$28.05 \pm 0.14$	$24.92 \pm 0.31$	$17.01 \pm 0.10$	$20.71 \pm 0.12$	$13.11 \pm 0.14$	$9.14 \pm 0.08$
Cellulose	$31.46 \pm 0.02$	$44.43 \pm 0.06$	$4.61 \pm 0.13$	$4.74 \pm 0.04$	$11.2 \pm 0.01$	$11.40 \pm 0.03$
ADL	$6.76 \pm 0.02$	$8.58 \pm 0.07$	$1.25 \pm 0.21$	$4.86 \pm 0.05$	$3.08 \pm 0.11$	$4.87 \pm 0.06$
Mineral compositi	ion					
Ca (%)	$0.89 \pm 0.001$	$0.20\pm0.001$	$0.12 \pm 0.001$	$0.18 \pm 0.001$	$0.18 \pm 0.005$	$0.58 \pm 0.004$
P (%)	$0.29 \pm 0.002$	$0.05 \pm 0.002$	$0.32 \pm 0.006$	$0.53 \pm 0.004$	$0.54 \pm 0.005$	$0.68 \pm 0.001$
Mg (%)	$0.23 \pm 0.003$	$0.17 \pm 0.001$	$0.13 \pm 0.01$	$0.51 \pm 0.01$	$0.60 \pm 0.002$	$0.54 \pm 0.001$
Cu (ppm)	$13.45 \pm 0.02$	$2.75 \pm 0.08$	$1.03 \pm 0.004$	$10.26 \pm 0.02$	$12.86 \pm 0.05$	$6.93 \pm 0.05$
Zn (ppm)	$38.71 \pm 0.003$	$11.81 \pm 0.01$	$20.27 \pm 0.13$	$48.44 \pm 0.06$	$52.32 \pm 0.001$	$47.24 \pm 0.14$
Mn (ppm)	$56.33 \pm 0.01$	$62.33 \pm 0.01$	$6.85 \pm 0.08$	$40.61 \pm 0.17$	$31.29 \pm 0.04$	$58.37 \pm 0.02$
Fe (ppm)	$154.62 \pm 0.15$	$256.11 \pm 6.17$	$58.42 \pm 0.28$	$200.99 \pm 7.24$	$120.41 \pm 0.04$	$178.24 \pm 0.003$
Co (ppm)	$0.06 \pm 0.001$	$0.43 \pm 0.002$	$0.004 \pm 0.001$	$0.002 \pm 0.001$	$0.002 \pm 0.001$	$0.23 \pm 0.001$

Table 4 Chemical composition (On % DM basis) of different fodders and concentrate mixtures

OM organic matter, CP crude protein, EE ether extract, TCHO total carbohydrate, NDF neutral detergent fiber, ADF acid detergent fiber, ADL acid detergent lignin, GNC groundnut cake, MOC mustard oil cake

(MOC) used for the preparation of total mixed rations (TMRs) were analyzed for chemical composition, such as dry matter (DM), organic matter (OM), crude protein (CP), ether extract (EE), total carbohydrates (TCHO), total ash (TA), neutral detergent fiber (NDF), acid detergent fiber (ADF), hemicellulose, cellulose, and acid detergent lignin (ADL). Table 4 displays the estimated values of different feed ingredients. Oil cakes (GNC, and MOC) contained greater amounts of CP compared to other feed resources, whereas fodder resources (paddy straw, and oats fodder) had higher amounts of NDF and ADF. Ca concentration in oats fodder, and MOC was estimated to be higher than other feed ingredients (Table 4). However, P content was greater in MOC, GNC, and wheat bran than in other feed resources. Table 4 presents the detailed values of micro-minerals (Cu, Zn, Mn, Fe, and Co) of these feed resources.

The TMRs prepared as per the composition mentioned in Table 1 were analyzed for chemical and mineral composition (Table 5). We estimated similar DM, OM, CP, EE, TCHO, total ash, ADF, hemicellulose, and lignin among three TMRs, wherein NDF and cellulose concentration were greater in TMRs under  $T_1$  and  $T_2$  than  $T_0$ . Three TMRs had a similar concentration of Ca, and P. TMR  $T_1$  and TMR  $T_2$  contained greater amounts of Mg, Cu, Zn, Mn, and Fe compared to TMR  $T_0$ . Cobalt concentration was found identical among the three TMRs.

#### Voluntary Intake of Nutrients and Utilization of Ca and P

Throughout six fortnights, the total dry matter intake (kg/ day/calf) varied from 2.21 to 3.20 in the control ( $T_0$ ) group, 2.36 to 3.56 in  $T_1$ , and 2.33 to 2.97 in  $T_2$ , respectively. The average DM intake was significantly higher (p < 0.05) in T<sub>1</sub> and  $T_2$  compared to  $T_0$  (Table 6). There was a significant (p < 0.001) difference in total DMI (kg/d/calf) among different fortnights indicating the periodic effect, which was mainly due to gain in body weight (BW) by growing calves over different fortnights. DMI (kg/100 kg body weight and g/kg W<sup>0.75</sup>) was significantly (p < 0.05) greater in T<sub>1</sub> and T<sub>2</sub> groups compared to T<sub>0</sub> group, with no significant difference between  $T_1$  and  $T_2$  (Table 6). Similarly, the period effect (Fortnights) was also significant for DMI per unit of body weight (kg/100 kg BW and g/kg W<sup>0.75</sup>), whereas the interaction between treatment, and period  $(T \times P)$  showed (p > 0.05)non-significant effect for DMI. Calves under T1 (containing FCSP), and T<sub>2</sub> (containing CSCP) consumed a greater (p=0.012) amount of total OM compared to the control

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Table 5Chemical composition(On % DM basis) and mineralprofile of different total mixedrations (TMRs)

Parameter	TMR T <sub>0</sub>	TMR T <sub>1</sub>	TMR T <sub>2</sub>	Pooled mean $\pm$ S.E.M	p value
Chemical compos	ition (%)				
DM	43.70	47.03	43.28	$44.67 \pm 1.26$	0.469
OM	89.29	89.43	89.29	$89.34 \pm 0.04$	0.231
СР	12.04	12.13	12.13	$12.10 \pm 0.05$	0.765
EE	2.83	2.85	2.83	$2.83 \pm 0.01$	0.610
ТСНО	74.13	74.24	74.12	$74.40 \pm 0.06$	0.766
Total ash	10.71	10.57	10.71	$10.66 \pm 0.04$	0.231
NDF	57.04 <sup>a</sup>	57.21 <sup>b</sup>	57.32 <sup>c</sup>	$57.19 \pm 0.04$	<i>p</i> < 0.001
ADF	33.29	33.65	33.56	$33.60 \pm 0.12$	0.539
Hemicellulose	23.75	23.55	23.76	$23.69 \pm 0.12$	0.770
Cellulose	28.29 <sup>a</sup>	28.83 <sup>b</sup>	28.85 <sup>b</sup>	$28.66 \pm 0.09$	<i>p</i> < 0.001
Lignin	5.00	4.82	4.71	$4.84 \pm 0.12$	0.689
Mineral compositi	ion				
Ca (%)	1.29	1.28	1.28	$1.28 \pm 0.002$	0.583
P (%)	0.61	0.62	0.61	$0.61 \pm 0.009$	0.929
Mg (%)	0.25 <sup>a</sup>	0.26 <sup>b</sup>	0.26 <sup>c</sup>	$0.22 \pm 0.001$	<i>p</i> < 0.001
Cu (ppm)	16.36 <sup>a</sup>	20.17 <sup>b</sup>	21.44 <sup>c</sup>	$18.51 \pm 0.795$	<i>p</i> < 0.001
Zn (ppm)	86.72 <sup>a</sup>	89.90 <sup>c</sup>	88.09 <sup>b</sup>	$88.24 \pm 0.485$	0.001
Mn (ppm)	46.56 <sup>a</sup>	51.47 <sup>b</sup>	51.85 <sup>b</sup>	$49.96 \pm 0.863$	<i>p</i> < 0.001
Fe (ppm)	59.26 <sup>a</sup>	63.78 <sup>b</sup>	65.85 <sup>c</sup>	$62.96 \pm 0.978$	p<0.001
Co (ppm)	0.51	0.59	0.61	$0.57 \pm 0.029$	0.402

The values with different superscripts (a, b, c) among different treatments differ significantly

 $(T_0)$  (Table 6). The OMI (kg)/100 kg BW and OMI (g)/kg  $W^{0.75}$  ranged from 2.59 (T<sub>0</sub>) to 2.92 (T<sub>2</sub>) and 79.77 (T<sub>0</sub>) to 89.78 (T<sub>2</sub>). The values were statistically greater in  $T_1$  and  $T_2$  than in  $T_0$ . However, similar to DMI, the periodic effect on OM intake (kg/d/calf) was also statistically significant (p < 0.001), whereas OM intake per unit BW increased significantly (p < 0.001) as the experiment progressed, indicating a periodic effect. However, the  $T \times P$  effect was found to be non-significant (p > 0.05) for these parameters. We estimated significantly greater (p < 0.001) crude protein intake per unit body weight (CPI, g/100 kg BW and g/W<sup>0.75</sup>) in the treatments supplemented with FCSP  $(T_1)$  and CSCP  $(T_2)$ compared to the control  $(T_0)$  treatment, wherein calves under  $T_2$  consumed a higher (p = 0.016) amount of total CP than control  $T_0$ . Similar to DM intake, the periodic effect showed statistical significance (p < 0.001) only for total CPI, and CPI/kg  $W^{0.75}$ , whereas the T × P effect was non-significant.

Calves under different treatments consumed (g/day) similar total Ca through TMR diets (Fig. 3) offered during the digestion trial conducted at the last phage of the growth trial. However, Ca intake (g)/kg  $W^{0.75}$  was greater (p=0.021) in CSCP-treated calves ( $T_2$ ) compared to DCP-treated control calves ( $T_0$ ). Calves under  $T_1$  had similar values (Ca intake (g)/kg  $W^{0.75}$ ) with  $T_0$  and  $T_2$ . Ca outgo (g/day) through feces was identical among three treatments, whereas Ca digestibility increased significantly (p=0.01) in FCSP ( $T_1$ ) and

CSCP (T<sub>2</sub>) treated calves compared to control T<sub>0</sub>. During the same digestion trial, we observed similar total P intake among different treatments (Fig. 4). However, P intake (g)/ kg W<sup>0.75</sup> enhanced (p < 0.001) in CSCP-treated calves (T<sub>2</sub>) compared to FCSP (T<sub>1</sub>) and control calves (T<sub>0</sub>). Fecal outgo of P reduced significantly in T<sub>2</sub> in comparison to T<sub>0</sub>; however, the value of T<sub>1</sub> was similar to T<sub>0</sub> and T<sub>2</sub>. The addition of fresh and calcined powder of conch shell increased (p < 0.001) P digestibility in T<sub>1</sub> and T<sub>2</sub> compared to T<sub>0</sub> in the present study (Fig. 4).

#### Growth Performance and Feed Conversion Efficiency

The initial body weight (kg) was similar among  $T_0$ ,  $T_1$ , and  $T_2$  groups (Table 6). At the end of the 90-day trial, the corresponding final body weight (kg) was 106.03, 106.31, and 106.68 in the  $T_0$ ,  $T_1$ , and  $T_2$  groups, respectively. The statistical difference among the groups was non-significant (p > 0.05). Calves under  $T_1$  and  $T_2$  treatments gained 3.03, and 5.78% higher compared to  $T_0$  during the feeding period; however, the difference was non-significant. The pattern of fortnightly body weight of calves is expressed graphically in Fig. 2, which indicated no difference among different treatments. The average daily body weight gain (g/d/animal) ranged from 366.60 ( $T_0$ ) to 392.24 ( $T_2$ ); the difference was observed similar among the treatment groups. Furthermore,

<b>Table 6</b> Body weight change, voluntary	intake of different nutrients, and feed conversion efficiency	affected by different dietary treatments

Attributes	Treatmen	nts		Pooled Mean $\pm$ S.E.M	<i>p</i> value		
	T <sub>0</sub>	T <sub>1</sub>	T <sub>2</sub>		Treatment effect (T)	Period effect (P)	Interac- tion effect $(T \times P)$
Voluntary intake of nutrients							
Av. DM intake (kg/d)	2.71 <sup>a</sup>	2.95 <sup>b</sup>	2.97 <sup>b</sup>	$2.88 \pm 0.06$	0.044	< 0.001	0.981
Av. DM intake kg/100 kg BW	2.99 <sup>a</sup>	3.25 <sup>b</sup>	3.27 <sup>b</sup>	$3.17 \pm 0.03$	< 0.001	0.019	0.394
Av. DM intake (g/kg W <sup>0.75</sup> )	91.85 <sup>a</sup>	100.07 <sup>b</sup>	$100.54^{b}$	$97.49 \pm 0.91$	< 0.001	< 0.001	0.095
Av. OM intake (kg/d)	2.36 <sup>a</sup>	2.64 <sup>b</sup>	2.65 <sup>b</sup>	$2.55 \pm 0.06$	0.012	< 0.001	0.992
Av. OM intake kg/100 kg BW	2.59 <sup>a</sup>	2.91 <sup>b</sup>	2.92 <sup>b</sup>	$2.81 \pm 0.03$	< 0.001	0.013	0.626
Av. OM intake (g/kg W <sup>0.75</sup> )	79.77 <sup>a</sup>	89.47 <sup>b</sup>	89.78 <sup>b</sup>	$86.34 \pm 0.92$	< 0.001	< 0.001	0.394
Av. CP intake (kg/d)	336.30 <sup>a</sup>	370.56 <sup>ab</sup>	376.29 <sup>b</sup>	$361.05 \pm 7.89$	0.016	< 0.001	0.986
Av. CP intake kg/100 kg BW	369.81 <sup>a</sup>	$408.40^{b}$	414.14 <sup>b</sup>	$397.45 \pm 3.77$	< 0.001	0.072	0.594
Av. CP intake (g/kg W <sup>0.75</sup> )	11.37 <sup>a</sup>	12.56 <sup>b</sup>	12.74 <sup>b</sup>	$12.23 \pm 0.12$	< 0.001	< 0.001	0.295
Body weight gain							
Initial body weight (kg)	71.40	70.60	70.05	$70.68 \pm 2.90$	0.984	-	-
Final body weight (kg)	106.03	106.31	106.68	$106.34 \pm 4.22$	0.998	-	-
Total body weight gain (kg)	34.63	35.71	36.63	$35.64 \pm 1.70$	0.905	-	-
Fortnightly body weight gain (kg)	5.77	5.95	6.10	$5.94 \pm 0.19$	0.786	0.124	1.000
Av. daily body weight gain (g)	366.60	379.57	392.24	$379.47 \pm 11.25$	0.695	0.806	1.000
Feed conversion efficiency (FCE) a	and feed co	onversion r	atio (FCR	)			
FCE (ADG as % of DMI)	14.08	13.66	14.12	$13.95 \pm 0.44$	0.893	0.017	0.999
FCR (kg DMI/kg gain)	8.99	8.03	7.54	$8.19 \pm 0.60$	0.621	0.500	0.613
FCE (ADG as % of CPI)	0.11	0.11	0.11	$0.11 \pm 0.003$	0.840	0.020	0.999
FCR (kg CPI/kg gain)	47.77	49.81	51.59	$49.72 \pm 1.60$	0.644	0.129	1.000

Values with different superscripts (a, b) in a row are significantly different

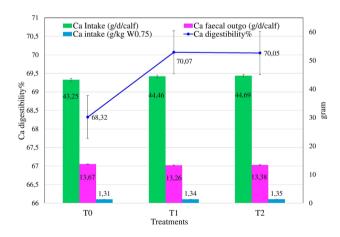


Fig.3 Effect of different Ca sources on Ca utilization pattern in calves under different treatments

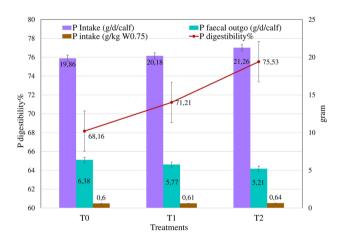


Fig. 4 Effect of different Ca sources on P utilization pattern in calves under different treatments

periodic (P), and interaction (P×T) effects were also nonsignificant (p > 0.05). Similarly, fortnightly body weight gain was also statistically similar among the three treatments.

Different sources of Ca in the present study did not alter the feed conversion efficiency (ADG as percent of DM intake), and feed conversion ratio (kg DMI/kg weight gain) in calves under three treatment groups (Table 6). Similarly, FCE (ADG as percentage of CPI) and FCR (kg CPI/kg weight gain) were also statistically identical among different treatments. However, the periodic effect was found significant (p < 0.05) for FCE (based on DMI and CPI), indicating

Days of feeding	T <sub>0</sub>	T <sub>1</sub>	T <sub>2</sub>	Pooled mean $\pm$ S.E.M	Significance
Blood glucose (mg	/dl plasma)				
0th days	49.06	51.59	51.30	$50.65 \pm 0.68$	Treatment effect (T), $p = 0.190$
45th days	49.86	47.17	51.67	$49.57 \pm 0.85$	Period effect (P), $p = 0.190$
90th days	50.81	51.88	52.14	$51.61 \pm 0.73$	$T \times P, p = 0.224$
Mean $\pm$ S.E.M	$49.91 \pm 0.79$	$50.22 \pm 0.80$	$51.70 \pm 0.68$	$50.61 \pm 0.45$	
Urea (mg/dl of plas	sma)				
0th days	16.90	16.36	17.49	$16.92^{\circ} \pm 0.25$	Treatment effect (T), $p = 0.867$
45th days	14.11	13.71	13.99	$13.94^{a} \pm 0.14$	Period effect (P), $p < 0.001$
90th days	15.94	16.49	15.52	$15.98^{b} \pm 0.25$	$T \times P, p = 0.088$
Mean $\pm$ S.E.M	$15.65 \pm 0.34$	$15.52 \pm 0.37$	$15.67 \pm 0.47$	$15.61 \pm 0.22$	
Calcium (mg/dl pla	isma)				
0th days	10.28	10.16	10.38	$10.27^{a} \pm 0.12$	Treatment effect (T), $p = 0.598$
45th days	10.82	11.32	11.38	$11.17^{b} \pm 0.09$	Period effect (P), $p < 0.001$
90th days	11.41	11.18	11.14	$11.24^{b} \pm 0.07$	$T \times P, p = 0.084$
Mean $\pm$ S.E.M	$10.84 \pm 0.17$	$10.88 \pm 0.15$	$10.97 \pm 0.13$	$10.90 \pm 0.09$	
Phosphorus (mg/dl	of plasma)				
0th days	5.49	5.61	5.56	$5.56 \pm 0.08$	Treatment effect (T), $p = 0.088$
45th days	5.14	5.70	5.71	$5.52 \pm 0.09$	Period effect (P), $p = 0.313$
90th days	5.72	5.62	5.64	$5.66 \pm 0.05$	$T \times P, p = 0.041$
Mean $\pm$ S.E.M	$5.45 \pm 0.08$	$5.65 \pm 0.08$	$5.64 \pm 0.05$	$5.58 \pm 0.04$	
ALP (IU/l serum)					
0th days	275.76	275.83	275.96	$275.85 \pm 3.72$	Treatment effect (T), $p = 0.994$
45th days	278.25	278.25	279.34	$278.61 \pm 3.27$	Period effect (P), $p = 0.778$
90th days	279.26	279.55	279.66	$279.49 \pm 3.46$	$T \times P, p = 1.000$
Mean $\pm$ S.E.M	$277.76 \pm 3.00$	$277.88 \pm 5.09$	$278.32 \pm 1.45$	$277.98 \pm 1.98$	
AST (IU/l serum)					
0th days	98.30	98.65	98.30	$98.42 \pm 1.11$	Treatment effect (T), $p = 0.906$
45th days	99.36	99.01	98.30	$98.89 \pm 1.64$	Period effect (P), $p = 0.944$
90th days	98.65	98.63	97.24	$98.18 \pm 1.40$	$T \times P, p = 0.999$
Mean $\pm$ S.E.M	$98.77 \pm 1.25$	$98.77 \pm 0.96$	$97.95 \pm 1.84$	$98.49 \pm 0.79$	
ALT (IU/l serum)					
0th days	20.86	21.57	21.04	$21.16 \pm 0.35$	Treatment effect (T), $p = 0.181$
45th days	20.29	21.30	21.06	$20.89 \pm 0.31$	Period effect (P), $p = 0.507$
90th days	20.96	21.97	21.43	$21.45 \pm 0.34$	$T \times P, p = 0.991$
Mean $\pm$ S.E.M	$20.70 \pm 0.34$	$21.61 \pm 0.34$	$21.18 \pm 0.28$	$21.17 \pm 0.19$	

 Table 7
 Blood glucose, urea, Ca and P, alkaline phosphatase (ALP), aspartate aminotransferase (AST), and alanine transaminase (ALT) concentration in animals fed with different TMRs

The values with different superscripts (a, b, c) among different treatments/period differ significantly

calves fed at younger age had greater FCE compared to older age. However, the interaction between treatment  $\times$  period (T  $\times$  P) was similar for FCE, and FCR.

#### **Blood Metabolites and Enzymes**

The blood glucose concentration (mg/dl of plasma) ranged from 49.06 to 50.81 in  $T_0$ , 47.88 to 51.88 in  $T_1$ , and 51.30 to 52.14 in  $T_2$  groups, respectively (Table 7). Ca from FCSP and CSCP did not alter the blood glucose concentration. Additionally, no statistically significant difference was recorded among different periods of blood sampling. Similarly, the interaction effect between treatment, and period  $(T \times P)$  was also observed to be non-significant. The average blood urea concentration was 15.65, 15.52, and 15.67 (mg/dl) in the T<sub>0</sub>, T<sub>1</sub>, and T<sub>2</sub> groups, indicating no significant difference among different treatments. However, the periodic effect was significant (p < 0.001). Furthermore, the interaction effect (T×P) was non-significant.

The Ca concentration (mg/dl of plasma) at the start of the experiment was estimated 10.28, 10.16, and 10.38 (mg/dl) in  $T_0$ ,  $T_1$ , and  $T_2$  groups, and it was 11.41, 11.18, and 11.14

(mg/dl of plasma) at the end of the experiment (Table 7). The average plasma Ca concentration (mg/dl of plasma) in calves of the present study ranged from 10.88 (T<sub>0</sub>) to 10.97 (T<sub>2</sub>). However, the difference (p > 0.05) was non-significant among the treated groups. A significant rise (p < 0.001) in blood Ca concentration was observed as the experiment progressed due to the feeding of dicalcium phosphate in T<sub>0</sub> and conch shell powders (fresh and calcined) in T<sub>1</sub> and T<sub>2</sub> as a source of Ca. However, the interaction effect (T×P) was similar for plasma Ca concentration. We observed average plasma P concentration (mg/dl plasma) ranging from 5.45 in T<sub>0</sub> to 5.65 in T<sub>2</sub>. The treatment effect (T), periodic effect (P), and interaction effect (T×P) were indistinguishable among the three treatments.

The average concentration (IU/l of plasma) of alkaline phosphatase (ALP) ranged from 275.76 ( $T_0$ ) to 279.66 ( $T_2$ ) in the study (Table 7). We observed no difference among the three treatments. The values for alkaline phosphatase level in blood were within the normal physiological range which indicated no adverse effect of new Ca source on the liver. The treatment effect (T), periodic effect (P), and interaction  $(T \times P)$  effect were comparable among the three treatments. The average concentration (IU/l plasma) of aspartate aminotransferase (AST) was observed to be non-significant (p > 0.05) among the different groups (T<sub>0</sub> 98.77 IU/l, T<sub>1</sub> 98.77 IU/l, and T<sub>2</sub> 97.95 IU/l). In addition, the periodic effect, and interaction  $(T \times P)$  effect were also statistically non-significant. We observed similar average concentrations (IU/l plasma) of alanine transaminase (ALT) in different treatment groups; the values were within the normal physiological range. Moreover, the periodic effect and interaction effect  $(T \times P)$  were also identical.

# Discussion

#### Physico-chemical Properties of FCSP and CSCP

The calcination process of fresh conch shell powder (FCSP) in the present study transformed the most stable polymorphs of CaCO<sub>3</sub> of conch shell (shankha) aragonite form into calcite form, along with the formation of portlandite and lime as the majority component [26]. Similar to our findings, properly ground ocean quahog clamp shells (OQCS), and surf clam shells (SCS) contain a higher percentage of Ca (>36%) [27]. Both FCSP and conch shell calcined powder (CSCP) had greater solubility in HCl, which is advantageous to livestock due to the acidic environment in the stomach (including abomasum in ruminants). Similarly, Rasheed et al. [7] concluded that conch shell powder was soluble in diluted HCl. Similar to our finding, Savita et al. [26] also reported pH values of 12.80 and 9.05 for conch shell calcined powder, and fresh conch shell powder. The acid-insoluble ash (AIA) content of FCSP corroborated the findings of Savita et al. [26] who observed 0.96% AIA in fresh conch shell powder. The calcination process increased the Ca concentration of conch shell powder; whereas it depressed the P concentration in the present study. High temperature (> 650 °C) during the calcination process in muffle furnace could have resulted in the reduction of P, Cu, and Zn in conch shell powder, whereas the level of Mg, Co, Mn, and Fe increased in this powder along with Ca due to such calcination process. The elevated levels of these essential minerals might be advantageous to the livestock to fulfill their nutritional requirements.

# Voluntary Intake of Nutrients and Utilization of Ca and P

The incorporation of FCSP and CSCP in the total mixed ration (TMR) of T<sub>1</sub> and T<sub>2</sub> resulted in similar Ca, and P concentrations with that of control T<sub>0</sub> TMR. The Ca/P ratio was estimated at 2.11, 2.06, and 2.10 in  $T_0$ ,  $T_1$ , and  $T_2$ , which could be considered ideal for calf ration (Fig. 1). The Ca:P ratio of 2:1 or greater was recommended by NRC [28] to avoid the formation of urinary calculi in sheep. Similarly, it was investigated that the Ca/P ratio in the complete ration plays an imperative role in plummeting the occurrence of mineral calculi in beef cattle, and that a good balanced diet could have an essential protective role in the development of this pathology [29]. The Ca/P ratio should be maintained 2:1 in cattle diet; any Ca/P imbalance may result in higher phosphate elimination through urine which is an imperative factor in the genesis of calculi [30]. Antonio Esteves et al. [16] also explained that the ratio of Ca/P has a great impact on the degree of absorption of both elements in animals. An excess of either causes amplification in fecal voidance by the development of insoluble Ca that is not obtainable for absorption. However, Mg, Cu, Zn, Mn, and Fe increased significantly due to the addition of FCSP and CSCP during the formulation of TMR in T<sub>1</sub> and T<sub>2</sub> in the present study. Voluntary intake of nutrients (DM, OM, and CP) as expressed per unit of body weight (per 100 kg body weight or kg  $W^{0.75}$ ) enhanced due to the replacement of DCP with FCSP, and CSCP in the diets of crossbred calves. Calves under three treatments fulfilled their nutritional requirements in terms of DM, and CP intake as recommended by NRC [31]. Anil et al. [24] also reported a similar DM intake pattern per 100 kg body weight of finisher female calf fed with TMR (DCP 7.85%, TDN 57.51%) under an intensive system of management.

We have not found any published reports on the effect of feeding conch shell powder or conch shell calcined powder in ruminants (calves or heifers or dairy cows) and non-ruminants (including poultry birds) to compare our present results. However, some reports are available on the use of different marine shells as a source of Ca. Similar to our findings, Finkelstein et al. [27] reported that the DMI

was lowest for cows fed aragonite and greatest for cows fed ocean quahog clam shell (Arctica islandica), and surf clam shell (Spisula solidissima). However, some researchers reported that the oceanic source of oyster shells could be used effectively in poultry birds as a source of Ca. Similar to our findings, Lana [32] discovered that broiler hens fed diets with 0.42% calcium from Charru Mussel Shell meal consumed more feed than the other groups fed different organic calcium sources. Similarly, Oso et al. [33] also reported that broilers fed a diet containing oyster shells as the Ca source had greater feed consumption, and body weight gain than broilers fed a diet having limestone. Daily intake of Ca significantly increased in poultry birds fed with oyster shells, and limestone compared to control [34]. Cambra-Lopez et al. [35] evaluated the P digestibility of the three commercial sources of DCP, and found that the digestibility of P was 80.1% (DCP1), 77.4% (DCP2), and 71.4% (DCP3), respectively. We have also recorded similar digestibility of P in the present study. Omole et al. [36] discovered that replacing dicalcium phosphate with oyster shell did not reduce feed intake but increased growth rate in finishing broiler. Mako et al. [37] observed that the amount of feed consumed by the poultry birds on diet 1 (oyster shells 54% and bone meal 46%) was greater than other treatment combinations. Similarly, the report by Olgun [38] showed that the supplementation of oyster shells up to one-third of the Ca requirement increased eggshell weight; however, it had an antagonistic effect on feed consumption and egg weight. This finding, however, conflicts with a study by Ahmed et al. [39] that claimed 24-week-old Bovan pullets treated with oyster shells had reduced feed intake than those treated with limestone. Badejo et al. [40] observed no significant (p < 0.05) difference between the oyster shell group, and the periwinkle shell group concerning feed consumption in the spent layer. CaCO<sub>3</sub> supplementation improved apparent starch digestibility, whereas the digestibility of other nutrients (DM, OM, CP, EE, and ADF) was not affected by CaCO<sub>3</sub> [41]. Santana et al. [42] investigated that Ca sources did not affect the digestibility coefficients in pigs.

#### **Growth Performance and Feed Conversion Efficiency**

Despite increased voluntary intake, the new marine source of Ca, either in fresh conch shell powder (FCSP) form or conch shell calcined powder (CSCP) form, did not alter the growth pattern in weaned crossbred calves compared with control calves fed with DCP. We did not find any published literature to compare the works on the usage of conch shell powder (fresh or calcined) as a potential source of Ca in ruminants. However, our findings are in close agreement with the finding of Santana et al. [42] who investigated that calcium supplied in the form of calcitic limestone, monodicalcium phosphate, calcinated bone flour, and oyster flour had no difference in growth performance of pigs. Also, similar results were obtained in brown laying hens; different Ca sources (100% fine limestone; 60% fine limestone and 40% large limestone; and 60% fine limestone and 40% oyster shell) had identical effects on body weight gain [43]. The FCE and FCR remained unaltered due to FCSP and CSCP treatments in calves. The results corroborated the findings of Olgun et al. [38], who reported no significant difference between the control (limestone) and the treatments (replaced with 1/3 eggshell, 1/3 oyster shell, 1/3 eggshell + 1/3 oyster shell, 2/3 eggshell, and 2/3 oyster shell) to modify FCR. The study by Badejo et al. [40] revealed that feed consumption and feed conversion ratio in the spent layer did not differ significantly among different Ca sources (bone meal, oyster shell, periwinkle shell, and limestone).

#### **Blood Metabolites and Enzymes**

Ca from three sources (DCP, FCSP, and CSCP) did not affect blood glucose, and blood urea concentration in growing calves. Similarly, the concentration of plasma Ca, and P was also similar among the three treatments. The results clearly indicated that marine sources of Ca in different forms (fresh and calcined) had no adverse effects on the concentration of glucose, urea, Ca, and P in blood plasma. The concentration of ALP, AST, and ALT did not differ among the three treatments, indicating no adverse effects on liver enzymes due to the feeding of new Ca sources. All these parameters were within the normal physiological range. Manowari and Singh [44] estimated almost similar blood glucose levels in crossbred heifers and cows. The mean blood urea value in normal healthy control livestock was reported as 26.76 mg/dl [45]. Reece and Hotchkiss [46] measured similar Ca, and P levels in calves fed with different feeding modules. Gading et al. [47] also observed identical blood Ca, and P concentrations in weaner calves. The phosphorus concentration in the present study corroborated the finding of Ramana et al. [48] who revealed that plasma P (mg%) concentrations in stall-fed heifers and cattle calves were from 5.03 to 5.18 mg% and 5.24 to 6.60 mg%, respectively. Brazilian male sheep fed a basal ration supplemented with different sources of Ca, namely limestone (L), alfalfa hay (AH), oyster shell meal (OSM), and citrus pulp (CTP) utilized inorganic sources of Ca in a greater way compared to other sources [10]. The average ALT values reported in this study were within the normal physiological range of cattle [49]. The typical ALT range is 11-40 IU/l of plasma [50]. Similarly, the normal activity of AST in the blood is 78 to 132 IU/l in healthy cattle [50]. Mamun et al. [51] reported a similar pattern of blood ALP, AST, ALT, and P levels in cattle reared in the tropical zone of Bangladesh. However, we estimated greater plasma Ca concentration in the present study compared to the study of Mamun et al. [51].

# Conclusions

Both fresh conch shell powder, and conch shell calcined powder are rich in Ca, Zn, Fe, and Co; however, CSCP is richer in Ca, Mg, Mn, and Fe, whereas FCSP contained a greater amount of Cu, and Zn. Despite enhanced voluntary intake of different nutrients, the new marine organic source of Ca, either in fresh or conch shell calcined powder form, did not modify the growth performance in weaned crossbred calves compared to control dicalcium phosphate-fed calves. Blood metabolites and liver enzyme profiles remained unaffected on supplementation of fresh or calcined conch shell powder. Short-term experimentation depicted some promising scope for the use of locally available conch shell powder (fresh/calcined) as a potential source of Ca for feeding to the livestock; however, long-term experimentation can ascertain the positive response of this marine-based organic Ca source in growing, pregnant, and lactating livestock as well as in poultry birds.

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Author Contribution TKD, and AC conceived and designed the experiment; JB, TKD, AC, and SR conducted experiment and formal analysis; AM, and SR analyzed the data; JB, TKD, and AC wrote the manuscript. All authors have read and agreed to the published version of the manuscript.

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**Data Availability** The data will be made available from the corresponding author on reasonable request.

# Declarations

**Ethical Approval** The animal feeding experiment was conducted as per committee approval of ICAR-National Dairy Research Institute (NDRI Deemed University), Karnal, Haryana, India (NDRI Reg. No. 20-M-AN-01, 2020), and accordingly, all standard Institutional and Government of India ethical protocols including animal right/welfare issues were followed during the entire animal trial period.

Competing Interests The authors declare no competing interests.

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