# **Correlation of Arsenic Levels in Smokeless Tobacco Products and Biological Samples of Oral Cancer Patients and Control Consumers**

Sadaf S. Arain<sup>1</sup> • Tasneem G. Kazi<sup>1</sup> • Hassan I. Afridi<sup>1</sup> • Farah N. Talpur<sup>1</sup> • Atif G. Kazi<sup>2</sup> • Kapil D. Brahman<sup>1</sup> • Naeemullah<sup>1</sup> • Abdul H. Panhwar<sup>1</sup> • Muhammad A. Kamboh<sup>3</sup>

Received: 23 March 2015 / Accepted: 26 April 2015 / Published online: 15 May 2015 © Springer Science+Business Media New York 2015

Abstract It has been extensively reported that chewing of smokeless tobacco (SLT) can lead to cancers of oral cavity. In present study, the relationship between arsenic (As) exposure via chewing/inhaling different SLT products in oral cancer patients have or/not consumed SLT products was studied. The As in different types of SLT products (gutkha, mainpuri, and snuff) and biological (scalp hair and blood) samples of different types of oral cancer patients and controls were analyzed. Both controls and oral cancer patients have same age group (ranged 30-60 years), socio-economic status, localities, and dietary habits. The concentrations of As in SLT products and biological samples were measured by electrothermal atomic absorption spectrophotometer after microwaveassisted acid digestion. The validity and accuracy of the methodology were checked by certified reference materials. The resulted data of present study indicates that the concentration of As was significantly higher in scalp hair and blood samples of oral cancer patients than those of controls (p < 0.001). It was also observed that the values of As were two- to threefolds higher in biological samples of controls subjects, consuming SLT products as compared to those have none of these habits

Hassan I. Afridi hassanimranafridi@yahoo.com

Sadaf S. Arain ssadiashafi@gmail.com

Tasneem G. Kazi tgkazi@yahoo.com

Farah N. Talpur farahtalpur@hotmail.com

Atif G. Kazi dratifgulkazi@gmail.com

Kapil D. Brahman kr\_brahman@yahoo.com (p>0.01). The intake of As via consuming different SLT may have synergistic effects, in addition to other risk factors associated with oral cancer.

**Keywords** Arsenic · Oral cancer · Biological samples · Smokeless tobacco products · Atomic absorption spectrophotometry

# Introduction

Oral cancer is a common malignancy among people who have tobacco smoking and chewing habits [1]. The disease is characterized by a high rate of morbidity and mortality [2]. The etiology of oral cancer is multifactorial, major risk factors are tobacco and alcohol consumption [3]. Both tobacco smoking (cigarettes, cigars, and pipes) and chewing SLT products with and without other ingredients have been shown to increase the risk of developing oral cancer [3–5]. Cancers caused by SLT use often begin as leukoplakia or erythroplakia, which has a higher chance to becoming cancerous over the time [6].

Naeemullah naeemullah433@yahoo.com Abdul H. Panhwar haleem\_analyst@yahoo.com Muhammad A. Kamboh afzal82\_kamboh@yahoo.com

- <sup>1</sup> National Center of Excellence in Analytical Chemistry, University of Sindh, Jamshoro 76080, Pakistan
- <sup>2</sup> Liaquat University of Medical and Health Sciences, Jamshoro 76080, Pakistan
- <sup>3</sup> Department of Chemistry, Faculty of Science, Universiti Teknologi Malaysia, Johor Bahru, Malaysia

Tobacco plant (*Nicotiana tabacum*) is well-known for its capacity to concentrate toxic elements from its growing environment [7]. Tobacco is known to contain numerous classes of carcinogenic substance such as tobacco-specific nitrosamines, which are often regarded as a major factor in SLT-related carcinogenesis. The combined exposure of nitrosamines and other classes of organic and inorganic substances, including toxic metals enhances the carcinogenetic effects [8].

Chronic exposure to As and heavy metals has long been recognized to enhance the cancer incidence among exposed human populations. In fact, As and heavy metals are considered to be able to act not only as carcinogens but also as co-carcinogens that could activate certain chemical compounds [9, 10]. The As exposure may cause gastrointestinal irritation, decreased production of red and white blood cells, abnormal heart rhythm, damage of blood vessels, pins and needle sensation in hands and feet, as well as damage the internal organs [11–13].

The effect of As exposure on human health was observed in population of south and southeastern Asia, particularly in Bangladesh, Taiwan, India, and Pakistan [14–16]. Numerous studies have demonstrated that high exposure of As caused various cancers, chromosome aberrations, and oxidative stress [17–21]. The As toxicity causes skin cancer, mouth ulcerations, low hemoglobin, leukemia, acute renal failure, and nerve damages [22]. Human exposure to As via different routes such as water and foods can lead to diverse disease processes. However, intake of As from non-food sources are often overlooked although they may be a contributory factor in the development of disease and this requires further investigation [23].

In the last decade, toxic element, As, received much attention as humans may be exposed through occupational and environmental exposure [24]. In a previous study, it was reported that As levels in surface and underground water, vegetables, and tobacco were high and population of southern areas of Pakistan have high exposure to As from food and non-food items [25]. The study reported that chewing SLT products is a risk factor for As-related skin lesions in women [26]. The As in mammals causes lipid per oxidation, glutathione depletion, as well as protein and enzyme oxidation [27–29]. The As carcinogenicity include its ability to alter DNA methylation patterns, induce cell death and proliferation, inhibit DNA repair, and induce genetic damage [30–32].

Several analytical techniques, electrothermal atomic absorption spectrometry [33], and hydride generation atomic absorption spectrometry [34, 35] are used for the determination of trace levels of As with sufficient sensitivity. As the rate of oral cancer is increased in Pakistan, although many risk factors has been well characterized in its pathogenesis, while very common habit of chewing SLT products is also one of the main reasons.

The aim of present study was to evaluate and compare the concentration of As in different types of snuff (dry and moist), mainpuri and gutkha, available, and consumed in Pakistan. The As in scalp hair and blood samples of oral cancer patients and controls consumed different SLT products were also analyzed.

## **Materials and Methods**

#### **Reagents and Glassware**

Ultra-pure water obtained from ELGA labwater system (Bucks, UK). Concentrated nitric acid (65 %) and hydrogen peroxide (30 %) were obtained from Merck (Darmstadt, Germany). Working standard solutions of As were prepared prior to their use by stepwise dilution of certified standard solutions (1,000 ppm) Fluka Kamica (Buchs, Switzerland), with 0.2 mol/L of HNO<sub>3</sub>. Moreover, matrix modifier was employed to analyze As was prepared from Mg(NO<sub>3</sub>)<sub>2</sub> and 99.999 % Pd (Sigma, St. Louis, MO). All solutions were stored in polyethylene bottles at 4 °C. For the accuracy of methodology, certified reference materials (CRM), human hair BCR 397, Clincheck® control-lyophilized human serum and Virginia tobacco leaves (ICHTJ-cta-VTL-2) were used. Glasswares and polyethylene containers were soaked in 10% (v/v) HNO<sub>3</sub> for 24 h; washed with distilled water, finally with de-ionized water and dried in such a manner to ensure that no any contamination from glasswares occur.

#### Instrumentation

The determination of As was carried out by means of a double-beam Perkins-Elmer Atomic Absorption Spectrometer Model 700 (Norwalk, CT) equipped with the graphite furnace HGA-400, pyrocoated graphite tubes with integrated platform, an autosampler AS-800, and deuterium lamp as background correction system. A single element hollow cathode lamp used for As was operated at 7.5 mA with a spectral bandwidth of 0.7 nm. Portions of both standards/ samples and modifier transferred into autosampler cups, then 20  $\mu$ L (standard or sample volume 10  $\mu$ L and modifier 10  $\mu$ L in each case) was injected into the electrothermal graphite atomizer. The graphite furnace heating program was set for the drying, ashing, atomization, and cleaning steps as temperature ranges (°C)/time (s): 80-120/15, 300-600/15, 2,000-2, 100/5, and 2,100-2,400/2, respectively. A horizontal electrical shaker (220/60 Hz, Gallenkamp, England) was used for shaking the samples. The pH was measured by a pH meter (781pH meter, Metrohm). A PEL domestic microwave oven

(Osaka, Japan), programmable for time and microwave power from 100 to 900 W, was used for digestion of samples.

#### **Study Population**

A survey was carried out about the gutkha and mainpuri chewing, while snuff inhaling habits of both genders, age ranged 30-60 years, residing in the different cities of Pakistan. The data of hospital based case-control study population was collected from Nuclear Institute of Medicine and Radiotherapy (NIMRA) Jamshoro and Larkana institute of nuclear medicine and radiotherapy (LINAR), situated in different areas of Sindh, Pakistan, during 2011-2013 years, by collecting files and extracting important information about the oral cancer. During 1-year study period (2011), the information department of both hospitals recorded >5,200 cases of cancers of all types, and mouth cancer comprised of 2.7 % of the total. The oral cancer patients were divided into sub groups according to the different over found locations of oral cancer, Lips, tongue, cheeks, and pharynx (throat). Oral cancer patients and controls were further grouped according to their SLT chewing habits, not consumed any SLT product (NU), gutkha (GU), snuff (SU), and mainpuri users (MPU). Complete demographic information is listed in Table 1.

Physical examinations were performed to measure participant's weight, height, blood pressure, and biochemical data. The biochemical tests of oral cancer patients and controls were performed to estimate hemoglobin, red blood cells, packed cell volume, transferrin iron-binding capacity, mean corpuscular hemoglobin concentration, and volume in the blood.

 Table 1
 Characteristics of study subjects (30–60) age groups

Controls/oral cancer patients			GU <sup>b</sup>	SU °	MPU <sup>d</sup>
Male					
Controls		192	158	135	209
Oral cancer patients	Lips	61	54	47	55
	Tongue	43	51	29	34
	Cheeks	32	37	26	32
	Pharynx (throat)	31	23	30	26
	Total	167	165	132	147
Female					
Controls		136	120	107	98
Oral cancer patients	Lips	43	31	28	24
	Tongue	25	31	23	27
	Cheeks	21	15	22	19
	Pharynx (throat)	24	15	24	26
	Total	110	92	97	96

<sup>a</sup> Non-SLT users

<sup>d</sup> Mainpuri users

Criteria for the selection of patients was of biopsy proved oral squamous cell carcinoma prior to any treatment and they were not taking any mineral supplements during last 3 months. The criteria for selection of 1,155 referent subjects were same age group, socio-economic status, and dietary habits, being free of any cancer diagnosis and not taking any mineral supplement. The biochemical results are given in Table 2. The histological information is not given in this study. Prior to the biological samples collection, the controls have undergone a standard routine medical examination. This study was approved by ethical committee of Sindh University, working under the auspices of higher education commission of Pakistan.

## Sampling of SLT Products

A total of 23 brands of snuff (dry and moist), 11 brands of gutkha, and 12 brands of mainpuri were purchased from local markets of the different cities of Pakistan as per their availability over a 3-year period (January 2011–December 2013). The samples were packed in their original packing and placed in prewashed dried plastic bags separately and stored at 4 °C, until tested. Ten composite samples of each brand of snuff, gutkha, and mainpuri were prepared by homogenizing the mixture after removing the wrappers. Care was taken to avoid any source of contamination, and this preparation was carried out in a clean environment. All samples were dried at 80 °C. The dried samples were ground with agate mortar and pestle, sieved through nylon sieves with mesh sizes of 125  $\mu$ m, and then stored in the labeled sample bottles.

#### **Biological Samples**

Venous blood samples (5 mL) were collected by 7-mm heparinized lithium Vacutainer<sup>®</sup> tubes (Becton Dickinson). About 2 ml of venous blood samples were stored at -20 °C until elemental analysis. The scalp hair samples were taken from five different parts of the scalp (frontal, cranial, occipital, right, and left lateral). The first 5 cm of hair from the root were used for analysis. Hair samples were put into separate plastic envelopes for each participant, tightly sealed and attached with identification number of the participant and questionnaire. In the laboratory, hair specimens were further cut into pieces, approximately 0.2- to 0.3 cm, and washed four times with a 1:200 v/v dilution of Triton X-100, then rinsed three times with ultra-pure water and two times with acetone [36], then dried in an oven at 80–85 °C.

## **pH Determination of SLT Products**

Weighed 1 g sample of each brand of gutkha, mainpuri, moist and dry snuff, added 10 mL of ultrapure water in flask (100-mL capacity), and placed in an electrical shaker at 30 rpm for 30 min, then filtered the solution through

<sup>&</sup>lt;sup>b</sup> Gutkha users

<sup>&</sup>lt;sup>c</sup> Snuff users

Table 2 Clinical and biochemical characteristics of referents and different types of oral cancer patients

Parameters	Normal range	Controls	Cancer patients				
			Lips	Tongue	Cheeks	Pharynx	
Male							
Hb (mg/dL)	13–16	14.6+0.56	9.34±0.51	$7.62 \pm 0.72$	9.38+1.02	7.76±0.69	
RBC (×10 <sup>12</sup> /L)	4-6.6	6.3+0.2	$2.35 \pm 0.24$	$1.96 \pm 0.31$	$2.32 \pm 0.58$	2.25±0.34	
PCV (%)	40–54	52.5+1.3	$24.9 \pm 0.83$	$18.4 {\pm} 0.59$	24.6+1.93	16.8±0.69	
MCH (pg)	27–32	30.6+0.9	49.2±1.38	$54.8 \pm 1.33$	48.5+2.52	53.8±2.65	
MCHC (g/dL)	32–36	35.4+0.3	$18.9 {\pm} 0.91$	$17.9 \pm 0.75$	$15.5 \pm 0.82$	14.7±0.55	
MCV (fl)	76–94	93.4+0.5	$172 \pm 7.98$	$194 \pm 9.41$	167+8.39	215±9.27	
Serum Fe (µg/100 mL)	60–160	149+9.5	$106 \pm 7.55$	$90.8 {\pm} 6.58$	82.5+8.22	62.8±4.59	
TIBC (µg/100 mL)	280-400	365+35.1	164±16.8	$185 \pm 17.9$	192+7.37	$153 \pm 10.8$	
Female							
Hb (mg/dL)	11-14.5	$12.9 \pm 1.4$	$8.96 {\pm} 0.22$	$7.63 {\pm} 0.54$	$7.65 \pm 0.92$	$5.82 {\pm} 0.68$	
RBC (×10 <sup>12</sup> /L)	3.5-4.5	$4.10 \pm 0.32$	$1.95 \pm 0.37$	$1.83 {\pm} 0.26$	$2.01 \pm 0.18$	$1.60 \pm 0.12$	
PCV (%)	35–47	$44.2 \pm 2.6$	$18.5 \pm 1.04$	$16.2 \pm 0.71$	23.8+1.06	15.9±0.99	
MCH (pg)	27–32	$30.3 {\pm} 0.9$	46.9±1.62	45.8±1.83	$48.9 \pm 1.97$	$54.5 \pm 1.71$	
MCHC (g/dL)	32–36	35.2±0.7	$16.9 \pm 0.72$	$17.2 \pm 0.58$	$15.9 \pm 0.32$	15.1±0.62	
MCV (fl)	92.6±2.6	94.2±1.4	$165 \pm 9.62$	$185 {\pm} 6.98$	169+7.33	210±8.65	
Serum Fe (µg/100 mL)	60–160	135±13.2	92.6±5.21	89.6±9.74	79.2+8.55	67.8±5.92	
TIBC (µg/100 mL)	280-400	$362 \pm 35.2$	$170 \pm 12.6$	$187 \pm 9.29$	184+10.6	155±7.39	

Hb hemoglobin, RBC red blood cells, MCH mean corpuscular hemoglobin, MCHC mean corpuscular hemoglobin concentration, MCV mean corpuscular volume, PCV packed cell volume, TIBC transferrin iron-binding capacity

Whatman No. 42 filter paper and extracts was taken to determine the pH.

#### **Microwave-Assisted Acid Digestion**

Replicate six samples of each CRM (0.5 mL Clincheck<sup>®</sup> control-lyophilized human serum, 0.2 g of Virginia tobacco leaves and BCR 397 human hair) and duplicate samples of different types of SLT products (0.2 g), whole blood (0.5 mL) and scalp hair (0.2 g) were taken separately in polytetrafluoroethylene (PTFE) flasks (25 mL in capacity). Added 3 mL of a freshly prepared mixture of concentrated HNO<sub>3</sub> - H<sub>2</sub>O<sub>2</sub> (2:1, v/v), kept at room temperature for 10 min. Then placed the flasks in covered PTFE container and heated at 80 % of total power (900 W) for 3–4 min. The digested samples were diluted up to 10 mL with 0.1 mol/L concentrated HNO<sub>3</sub>. A blank extraction (without sample) was carried out through the complete procedure.

## **Statistical Analysis**

All statistical analyses were performed using computer program Excel X State (Microsoft Corp., Redmond, WA, USA) and Minitab 13.2 (Minitab Inc., State College, PA, USA). The data from triplicate samples of each composite samples were expressed as means  $\pm$  std. The Student's *t* test was used to assess the significant difference of As in certified and experimentally found values. The one-way ANOVA was used to assess the significance of differences between the concentrations of As observed in the biological samples of oral cancer patients and control subjects. A *P*<0.05 was considered significant difference.

## **Analytical Figures of Merit**

The concentration range of As for calibration curve reached from the quantification limit up to 50 µg/L. The detection and quantification limits, given by LOD= $3 \times s/m$  and LOQ= $10 \times s/m$ m respectively, where s is the standard deviation of ten measurements of a reagent blank and m is the slope of the calibration graph. The LOD and LOQ, calculated for As were 0.126 and 0.421 µg/L, respectively. The validity and efficiency of the MAD method was checked with certified values of human hair BCR 397, Clincheck<sup>®</sup> control-lyophilized human serum, Virginia tobacco leaves (ICHTJ-cta-VTL-2), and with those obtained from conventional wet acid digestion method on same CRM (Table 3). The microwave-assisted digestion method was less time-consuming, requiring <10 min to complete the digestion of samples. The mean values for As differed <1-2 % from the certified values. Non-significant

CDM $\overline{x} \pm s^{a}$	% Recovery <sup>b</sup>	MAD $\overline{x} \pm s$	Paired t test <sup>c</sup> $t_{\text{Experimental}}$	% Recovery	Certified values		
	Certified sample of	of human hair (µg/g)					
$0.305 \pm 0.02 \ (6.55)^d$	98.4	0.307±0.02 (6.51)	0.578	99	$0.31 {\pm} 0.02$		
	Certified sample of serum (µg/l)						
19.4±1.91 (9.84)	98.9	19.5±1.54 (7.89)	0.960	99.5	19.6±4.0		
	Virginia tobacco leaf ( $\mu g/g$ )						
0.944±0.04 (4.23)	99	0.946±0.08 (8.45)	0.651	99.3	$0.953 {\pm} 0.08$		

 Table 3
 Determination of arsenic in certified sample of human hair, serum and Virginia tobacco leaf by conventional (CDM) and microwave digestion method (MAD)

 $t_{\text{Critical}}$  at 95 % confidence limit=2.57

<sup>a</sup> Average value±confidence interval (P=0.05)

<sup>b</sup> % Recovery=[Experimental value]/[Certified value]×100

<sup>c</sup> Paired t test between certified values vs found values, degree of freedom (n-1)=5

<sup>d</sup> Values in parenthesis RSD

differences (p>0.05) were observed when comparing the values obtained by both methods (paired *t* test).

## Result

# Arsenic Concentration in Different Smokeless Tobacco Products

Multiple samples of different brands of each SLT products were analyzed and the mean concentrations along with the standard deviation for ten composite samples of each brand are provided in Table 4. While the range of As levels in brown and green moist snuff was found to be 0.574-1.46 and  $0.995-1.53 \ \mu g/g$ , respectively. In dry brown and black snuff products, the As levels were observed in the range of 0.733-1.04 and  $0.642-1.07 \ \mu g/g$ , respectively. The contents of As in different brands of mainpuri (n=12) was found in the range of  $0.246-0.622 \ \mu g/g$ .

#### **Demographic Characteristic of Study Population**

The controls and patients have informed that they consumed mainpuri, snuff, and gutkha for  $\geq 8.0 \pm 2.5$  years. Clinical characteristics including basic medical data were obtained from medical records with the help of paramedical staff. The occupational history included (jobs held for more than 1 year over the lifetime), the study subjects (patients and controls) were mostly drivers, working in workshops (automobile, battery recycling) and labors in construction buildings. The females are mostly house wife or working as maid and in garment factories. The exclusion criteria for patients and controls were smoking or drinking alcohol.

# Arsenic Concentration in Biological Samples of Controls and Oral Cancer Patients

The mean values of As in biological samples of oral cancer patients and control subjects are presented in Table 5. The resulted data indicated that the contents of As was significantly higher in scalp hair and blood samples among cancer patients (lips, tongue, cheeks, and pharynx) than those of controls (P < 0.001). The ranges of As in the scalp hair samples of male control subjects (NU, GU, SU, and MPU) were found at 95 % confidence intervals (CI) 0.92-0.98, 1.31-1.43, 1.20-1.38, and 1.47–1.56  $\mu$ g/g, respectively, were significantly lower as compared to resulted data obtained from oral cancer patients (p < 0.001). The same trend was observed in females. The As concentrations in blood samples of male and female control subjects (NU, GU, SU, and MPU) at 95 % CI 1.62-2.07, 2.54–2.70, 2.33–2.67, 2.70–3.02 µg/L and 1.58–1.97, 2.40-2.59, 2.23-2.40, 2.49-2.82 µg/L, respectively, were found to be significantly lower than patients who consumed different SLT products (p < 0.001) (Table 5).

The unpaired student *t* test between cancerous patients and controls at different degrees of freedom, were calculated for different probabilities. Our calculated  $t_{\text{value}}$  exceeds that of  $t_{\text{critical value}}$  at 95 % confidence intervals, which indicated that the difference between mean values of As in controls and cancerous patients (both gender) showed significant differences (p < 0.001).

# Discussion

The pH of all SLT products was highly basic, found in the range of 8.1–8.7, which favors the formation of tobacco-specific amines thus making the product potentially toxic. The production of nitrosamines is major contributors to the

**Table 4** Arsenic concentrations in gutkha, mainpuri, dry and moist snuff samples ( $\mu g/g$ )

Gutkha			Snuff			Mainpuri		
(G)	$\overline{x} \pm s^{a}$	$\mu g/10 \ g^b$	Moist snuff	$\overline{x} \pm s$	µg/10 g	MP	$\overline{x} \pm s$	$\mu g/10 g$
G1	0.398±0.03	3.68–4.34 <sup>c</sup>	BM1 <sup>d</sup>	0.611±0.03	5.74-6.35	MP1	0.753±0.05	6.83-7.95
G2	$0.520 {\pm} 0.04$	4.89-5.78	BM2	$1.05 {\pm} 0.05$	9.95-11.0	MP2	$0.466 {\pm} 0.04$	4.19-5.14
G3	$0.564 {\pm} 0.05$	5.23-6.12	BM3	$1.33 \pm 0.09$	12.4–14.6	MP3	$0.480 {\pm} 0.05$	4.19-5.33
G4	$0.592 {\pm} 0.03$	5.56-6.22	BM4	$0.712 {\pm} 0.07$	6.41-7.83	MP4	$0.760 {\pm} 0.04$	7.22-7.98
G5	$0.337 {\pm} 0.03$	3.12-3.68	BM5	$1.23 \pm 0.06$	11.7-13.1	MP5	$0.471 {\pm} 0.03$	4.38-5.14
G6	$0.440 {\pm} 0.02$	4.23-4.67	BM6	$1.02 \pm 0.06$	9.60-11.0	MP6	$0.661 {\pm} 0.06$	5.89-7.22
G7	$0.271 \pm 0.03$	2.46-3.01	BM7	$1.30 {\pm} 0.07$	12.1-13.9	MP7	$0.784{\pm}0.07$	7.03-8.55
G8	$0.473 \pm 0.04$	4.34-5.12	BM8	$1.00 {\pm} 0.03$	9.60-10.3	MP8	$0.831 {\pm} 0.04$	7.79–8.74
G9	$0.404 \pm 0.01$	3.90-4.23	GM9 <sup>e</sup>	$1.16{\pm}0.05$	11.0-12.1	MP9	$0.675 {\pm} 0.07$	5.89-7.41
G10	$0.348 {\pm} 0.03$	3.12-3.68	GM10	$1.46{\pm}0.06$	13.9–15.3	MP10	$0.594 {\pm} 0.03$	5.52-6.27
G11	$0.487 {\pm} 0.04$	4.45-5.34	GM11	$1.18{\pm}0.03$	11.4–12.1	MP11	$0.504 {\pm} 0.02$	4.76-5.33
			GM12	$1.08 {\pm} 0.09$	9.95-12.1	MP12	$0.637 {\pm} 0.05$	5.52-6.84
			GM13	$1.19{\pm}0.05$	11.4-12.4			
			GM14	$1.23 \pm 0.07$	11.4–13.1			
			$DB1^{f}$	$0.892 {\pm} 0.06$	8.22-9.62			
			DB2	$0.817 {\pm} 0.05$	7.64-8.86			
			DB3	$0.992 \pm 0.04$	9.46-10.4			
			DB4	$0.809 {\pm} 0.06$	7.33-8.86			
			DBK5 <sup>g</sup>	$0.711 {\pm} 0.07$	6.42-7.94			
			DBK6	$1.02{\pm}0.05$	9.77-10.7			
			DBK7	$0.840 {\pm} 0.04$	7.94-8.86			
			DBK8	$0.924 {\pm} 0.08$	8.55-10.4			
			DBK9	$0.855 {\pm} 0.06$	7.94–9.16			

<sup>a</sup> Average value±confidence interval (P=0.05)

 $^{b}$  Intake of As from all SLT products were based on  $\mu g/10~g$ 

<sup>c</sup> The intake of As via different types of SLT product are presented in range (minimum-maximum)

<sup>d</sup> Brown moist

e Green moist

<sup>f</sup>Dry brown

g Dry black

increased risk of chewing SLT products for cancer of upper digestive tract [8].

This case–control study was conducted to evaluate the possible association between As exposure via consumption of different types of SLT products and its altered levels in blood and scalp hair samples of oral cancer patients with related to controls of both gender. The As concentrations in mainpuri, snuff, and gutkha samples consumed by cancerous patients and controls were determined. The high level of As was observed in both types of snuff, while it was observed that the levels of As varies in biological samples of controls and patients, according to the types of SLT products consumed, but difference was not significant (p>0.05). In all SLT products, significant variation in elemental contents would be expected [37]. Though SLT is described as a group 1 carcinogen by the International Agency for Research on Cancer, little is known regarding bioavailability, absorption, and toxicological effects of toxic and carcinogenic inorganic substances from them. The resulted values of As in biological samples of oral cancer patients of both gender, confirms that chewing SLT products could be major risk factors for the oral disease. The resulted data indicated that in controls of both gender, who not consumed any SLT have two- to threefold lower levels of As in their biological samples as compared to those results obtained from controls consumed SLT products. The significant high levels of As was observed Arsenic Levels in Oral Cancer Patients

 Table 5
 The As concentrations in scalp hair and blood samples of controls and different types of oral cancer patients

Biological samples	Types of SLT	Controls	Different types of oral cancer patients				
	products consumed		Lips	Tongue	Cheeks	Pharynx (throat)	
Male							
Scalp hair( $\mu g/g$ )	NU <sup>a</sup>	$0.95 {\pm} 0.05$	$1.72 \pm 0.15$	$1.98 \pm 0.21$	$2.19 \pm 0.17$	$2.34 \pm 0.35$	
	$\mathrm{GU}^{\mathrm{b}}$	$1.37 \pm 0.13$	2.55±0.31	$2.79 \pm 0.26$	$2.92 \pm 0.29$	$3.45 {\pm} 0.38$	
	$SU^{c}$	$1.29 \pm 0.18$	2.41±0.36	$2.68 \pm 0.43$	$2.85 \pm 0.32$	$3.32 \pm 0.45$	
	$MPU^d$	$1.52 \pm 0.09$	$2.69 \pm 0.33$	$2.95 \pm 0.27$	3.24±0.55	$3.69 {\pm} 0.52$	
Blood(μg/L)	NU	$1.85 \pm 0.24$	$3.25 \pm 0.18$	$3.51 {\pm} 0.47$	$3.75 \pm 0.35$	$3.94{\pm}0.29$	
	GU	2.62±0.15	4.16±0.24	$4.35 {\pm} 0.35$	$4.56 {\pm} 0.48$	4.73±0.64	
	SU	$2.50 \pm 0.37$	$3.89 \pm 0.35$	$3.96 {\pm} 0.54$	$4.18 \pm 0.31$	$4.18 {\pm} 0.48$	
	MPU	$2.84{\pm}0.30$	4.35±0.51	$4.67 \pm 0.70$	$4.85 {\pm} 0.58$	$4.98 {\pm} 0.56$	
Female							
Scalp hair(µg/g)	NU	$0.90 {\pm} 0.08$	$1.65 \pm 0.19$	$1.95 \pm 0.24$	$2.10 \pm 0.15$	2.27±0.25	
	GU	1.35±0.12	$2.48 \pm 0.27$	2.75±0.23	$2.84{\pm}0.28$	$3.42 \pm 0.32$	
	SU	$1.24 \pm 0.20$	$2.34{\pm}0.15$	$2.64 \pm 0.19$	$2.73 \pm 0.37$	3.25±0.21	
	MPU	$1.46 \pm 0.15$	$2.63 \pm 0.22$	$2.92 \pm 0.40$	$3.18 \pm 0.51$	$3.62 \pm 0.30$	
$Blood(\mu g/L)$	NU	$1.76 \pm 0.32$	3.18±0.25	$3.43 \pm 0.29$	3.68±0.17	3.82±0.35	
	GU	2.49±0.19	4.05±0.14	4.29±0.25	4.38±0.32	4.60±0.51	
	SU	2.36±0.28	3.74±0.43	3.79±0.45	4.06±0.27	4.13±0.51	
	MPU	$2.65 \pm 0.35$	4.21±0.25	$4.51 {\pm} 0.39$	$4.69{\pm}0.48$	4.85±0.37	

<sup>a</sup> Non-SLT users

<sup>b</sup> Gutkha users (GU)

<sup>c</sup> Snuff users (SU)

<sup>d</sup> Mainpuri users (MPU)

in biological samples of tongue, cheeks, and pharynx cancer patients as compared to controls consumed SLT products (p < 0.001), as shown in Table 5.

It was reported in previous studies that certain toxic elements were found in SLT products [38] and thus their intake via inhaling (snuff) or ingestion (gutkha and mainpuri) can cause serious diseases, including oral cancer [39]. It was reported that exposure to As via different routes and smoking synergistically increases the risk of lung cancer, bladder cancer, and induction of skin lesions [21, 40]. An experimental study showed that As and cigarette smoke act synergistically to cause DNA damage [41]. At present, millions of people worldwide suffer from chronic As poisoning [42, 43] mainly due to consumption of As-contaminated water and food. Lindberg et al. [26] reported that the As content in tobacco and other gradients (betel nut and betel quid) can further increase the risk of As induced skin lesions among people of As endemic areas.

Several studies have been reported that As is present in measurable quantity in tobacco products, although concentrations are relatively smaller than those of other metals, such as cadmium and lead [44, 45]. The As has been detected in cured or processed tobacco leaves at concentrations of approximately 400 ng/g, at dried basis [46], while in certain SLT products, its concentrations ranging between 130 and 360 ng/g of dry tobacco [44, 45, 47]. The As has been shown to induce carcinogenesis via a wide range of cellular changes including alterations in cell differentiation and proliferation [48, 49]. It was reported by Hayes that As has been found to induce chromosomal aberrations and sister chromatids exchange [48]. Studies have been reported that cells exposed to As have also been shown to increase cellular tyrosine phosphorylation, which is related to the aberrant cell signaling and uncontrolled cell growth associated with cancer development [50–52].

Epidemiologic studies have documented that long-term exposure to As is associated with an increased risk of cancer of the lung, skin, and probably other anatomic sites. The As is also one of major risk factors for black foot disease, a unique peripheral vascular disease identified in endemic areas of arsenicosis in Taiwan, where residents had used high As tainted artesian well water for more than 50 years. Exposure to As causes different types of cancers (head & neck, bladder, lung, skin, kidney, prostate, and liver) as well as cardiovascular disease, diabetes, developmental and reproductive effects [53–56]. It was suggested by Marano et al. and Cox that due to potential mechanisms of As carcinogenicity, its removal from cigarette tobacco might reduce human health risks [57, 58]. Epidemiology studies were reviewed, As biomarker

concentrations in a population representative of the US were evaluated, and a probabilistic risk assessment was undertaken [59, 60].

The International Agency for Research on Cancer (IARC) now regards the betel nut which is part of mainpuri and gutkha, itself known as a carcinogen [61]. It is demonstrating that reactive oxygen species, such as hydroxyl radical, are formed in the human oral cavity during SLT products chewing, and that the activity might cause oxidative DNA damage, which transformed into oral cancer [62].

## Conclusion

The results of this study revealed that the significant differences were observed in As concentration in biological samples of oral cancer patients as compared to noncancerous controls consumed or/not any type of SLT products. The imbalance in As level in oral cancer patients could be due to change of cellular metabolism in the cancer process. It was also observed that the socioeconomic factors may also play a role in higher mortality rates for oral cancer patients, such as poor nutrition, irregular screening, late diagnosis, and unequal access to health care due to poverty because the cost of cancer treatment is very high. Since the role of As in the mechanism of oral cancer development is still unclear, further detailed and comprehensive investigations are necessary.

## References

- Khandekar SP, Bagdey PS, Tiwari RR (2006) Oral cancer and some epidemiological factors: a hospital based study. Indian J Community Med 31:157–159
- Parkin DM, Bray F, Ferlay J, Pisani P (2005) Global cancer statistics, 2002. CA-Cancer J Clin 55:74-108
- Kazi TG, Wadhwa SK, Afridi HI, Kazi N, Kandharo GA, Baig JA, Shah AQ, Kolachi NF, Khan S (2010) Evaluation of cadmium and zinc in biological samples of tobacco and alcohol user male mouth cancer patients. Hum Exp Toxicol 29:221–230
- Rodu B, Cole P (2002) Smokeless tobacco use and cancer of the upper respiratory tract. Oral Surg Oral Med Oral Pathol Oral Radiol Endod 93:511–515
- Accortt AN, Waterbor WJ, Beall C, Howard G (2005) Cancer incidence among a cohort of smokeless tobacco users (United States). Cancer Causes Control 16:1107–1115
- Bouquot JE (1991) Reviewing oral leukoplakia. Clinical concepts for the 1990s. J Am Dent Assoc 122:80–82
- Golia EE, Dimirkou EA, Mitsios EIK (2007) Accumulation of metals on tobacco leaves (primings) grown in an agricultural area in relation to soil. Bull Environ Contam Toxicol 79:158–162
- Stepanov I, Hecht SS (2005) Tobacco-specific nitrosamines and their pyridine-*N*-glucuronides in the urine of smokers and smokeless tobacco users. Cancer Epidemiol Biomarkers Prev 14:885–891

- Arain et al.
- Sari A, Tuzen M (2009) Biosorption of As (III) and As (V) from aqueous solution by macrofungus (*Inonotus hispidus*) biomass: equilibrium and kinetic studies. J Hazard Mater 164:1372–1378
- Abernathy CO, Thomas DJ, Calderon RL (2003) Health and risk assessment of arsenic. J Nutr 133:1536–1538
- Sari A, Uluozlu OD, Tuzen M (2011) Equilibrium, thermodynamic and kinetic investigations on biosorption of arsenic from aqueous solution by algae (*Maugeotia genuflexa*) biomass. Chem Eng J 167: 155–161
- Arain SS, Kazi TG, Afridi HI, Brehman KD, Naeemullah, Shah F, Mughal MA (2014) Arsenic content in smokeless tobacco products consumed by population of Pakistan: related health risk. J AOAC Int 97:1662–1669
- Hossain MF (2006) Arsenic contamination in Bangladesh—an overview. Agric Ecosyst Environ 113:1–16
- Alrmalli SW, Haris PI, Harrington CF, Ayub MA (2005) Survey of arsenic in foodstuffs on sale in the United Kingdom and imported from Bangladesh. Sci Total Environ 337:23–30
- Lin MC, Liao CM (2008) Assessing the risk on human health associated with inorganic arsenic from ground water-cultured milkfish in Southern Taiwan. Food Chem Toxicol 46:701–709
- Baig JA, Kazi TG, Arain MB, Afridi HI, Kandhro GA, Sarfraz RA, Jamal MK, Shah AQ (2009) Evaluation of arsenic and other physico-chemical parameters of surface and ground water of Jamshoro, Pakistan. J Hazard Mater 166:662–669
- Chen YC, Su HJ, Guo YL, Houseman EA, Christiani DC (2005) Interaction between environmental tobacco smoke and arsenic methylation ability on the risk of bladder cancer. Cancer Causes Control 16(2):75–81
- Sari A, Tuzen M (2010) Biosorption of As (III) and As (V) from aqueous solution by lichen (*Xanthoria parietina*) biomass. Sep Sci Technol 45:463–471
- Brehman KD, Kazi TG, Afridi HI, Rafique T, Baig JA, Arain SS, Naeemullah, Panhwar AH, Arain SA (2014) Evaluation of fresh and stored rain water quality in fluoride and arsenic endemic area of Thar Desert, Pakistan. Environ Monit Assess 186:8611–8628
- Huang YK, Huang YL, Hsueh YM, Yang MH, Wu MM, Chen SY et al (2008) Arsenic exposure, urinary arsenic speciation, and the incidence of urothelial carcinoma: a twelve-year follow-up study. Cancer Causes Control 19:829–839
- Steinmaus C, Bates MN, Yuan Y, Kalman D, Atallah R, Rey OA et al (2006) Arsenic methylation and bladder cancer risk in case– control studies in Argentina and the United States. J Occup Environ Med 48(5):478–488
- Byrd DM, Roegner ML, Griffiths JC, Lamm SH, Grumski KS, Wilson R, Lai S (1996) Carcinogenic risk of inorganic arsenic in perspective. Int Arch Occup Environ Health 68:484–494
- Tuzen M, Saygi KO, Karaman I, Soylak M (2010) Selective speciation and determination of inorganic arsenic in water, food and biological samples. Food Chem Toxicol 48:41–46
- Aitio A, Becking G (2001) Arsenic and arsenic compounds. World Health Organization Geneva 2001
- Arain MB, Kazi TG, Baig JA, Jamali MK, Afridi HI, Jalbani N et al (2009) Respiratory effects in people exposed to arsenic via the drinking water and tobacco smoking in southern part of Pakistan. Sci Total Environ 407:5524–5530
- Lindberg AL, Sohel N, Rahman M, Persson LA, Vahter M (2010) Impact of smoking and chewing tobacco on arsenic-induced skin lesions. Environ Health Perspect 118:533–538
- Arain SS, Kazi TG, Arain JB, Afridi HI, Brahman KD, Naeemullah (2014) Preconcentration of toxic elements in artificial saliva extract of different smokeless tobacco products by dual-cloud point extraction. Microchemical 112:42–49
- Zaman K, Pardini RS (1996) An overview of the relationship between oxidative stress and mercury and arsenic. Toxic Subst Mech 15:151–181

- Kessel M, Liu SX, Xu A, Santella R, Hei TK (2002) Arsenic induces oxidative DNA damage in mammalian cells. Mol Cell Biochem 234–235:301–308
- Kitchin KT (2001) Recent advances in arsenic carcinogenesis: modes of action, animal model systems, and methylated arsenic metabolites. Toxicol Appl Pharmacol 172:249–261
- 31. Kitchin KT, Ahmad S (2003) Oxidative stress as a possible mode of action for arsenic carcinogenesis. Toxicol Lett 137:3–13
- IARC (International Agency for Research on Cancer) (2003) IARC Monographs on the evaluation of carcinogenic risks to humans. Some drinking-water disinfectants and contaminants, including arsenic. Lyon
- Elci L, Divrikli U, Soylak M (2008) Inorganic arsenic speciation in various water samples with gf-aas using coprecipitation. Int J Environ Anal Chem 88:711–723
- 34. Uluozlu OD, Tuzen M, Mendil D, Soylak M (2010) Determination of As(III) and As(V) species in some natural water and food samples by solid phase extraction on streptococcus pyogenes immobilized on Sepabeads SP 70 and hydride generation atomic absorption spectrometry. Food Chem Toxicol 48:1393–1398
- Tuzen M, Çıtak D, Mendil D, Soylak M (2009) Arsenic speciation in natural water samples by coprecipitation-hydride generation atomic absorption spectrometry combination. Talanta 78:52–56
- Arain SS, Kazi TG, Afridi HI, Talpur FN, Kazi AG, Brahman KD, Panhwar AH, Arain MS (2014) Scalp hair and blood cadmium levels in association with chewing gutkha, mainpuri, and snuff, among patients with oral cancer in Pakistan. J Oral Pathol Med. doi:10.1111/jop.12283
- 37. Asta J, Guillard E, Tissut M, Gaude T, Ravanel P (2003) Heavy metal transfer from atmosphere to plants. J Phys 107:65–67
- 38. Arain SS, Kazi TG, Arain JB, Afridi HI, Kazi AG, Nasreen S, Brahman KD (2014) Determination of nickel in blood and serum samples of oropharyngeal cancer patients consumed smokeless tobacco products by cloud point extraction coupled with flame atomic absorption spectrometry. Environ Sci Pollut Res 21:12017–12027
- 39. Kazi TG, Wadhwa SK, Afridi HI, Kazi N, Kandharo GA, Baig JA, Shah AQ, Kolachi NF, Arain MB (2010) Interaction of cadmium and zinc in biological samples of smokers and chewing tobacco female mouth cancer patients. J Hazard Mater 176:985–991
- Bates MN, Rey OA, Biggs ML, Hopenhayn C, Moore LE, Kalman D et al (2004) Case–control study of bladder cancer and exposure to arsenic in Argentina. Am J Epidemiol 159(4):381–389
- Hays AM, Srinivasan D, Witten ML, Carter DE, Lantz RC (2006) Arsenic and cigarette smoke synergistically increase DNA oxidation in the lung. Toxicol Pathol 34(4):396–404
- Hughes MF, Beck BD, Chen Y, Lewis AS, Thomas DJ (2011) Arsenic exposure and toxicology: a historical perspective. Toxicol Sci 123:305–332
- Rodriguez-Lado L, Sun G, Berg M, Zhang Q, Xue H, Zheng Q, Johnson CA (2013) Groundwater arsenic contamination throughout China. Science 341:866–868
- Counts ME, Morton MJ, Laffoon SW, Cox RH, Lipowicz PJ (2005) Smoke composition and predicting relationships for international commercial cigarettes smoked with three machine-smoking conditions. Regul Toxicol Pharmacol 41:185–227
- Pappas RS, Stanfill SB, Watson CH, Ashley DL (2008) Analysis of toxic metals in commercial moist snuff and Alaskan iqmik. J Anal Toxicol 32:281–291

- Lugon-Moulin N, Martin F, Krauss MR, Ramey P, Rossi L (2008) Arsenic concentration in tobacco leaves: a study on three commercially important tobacco (*Nicotiana tabacum* L.) types. Water Air Soil Pollut 192:315–319
- International Agency for Research on Cancer (IARC), 2004a. Tobacco smoke and involuntary smoking, In: IARC monographs on the evaluation of carcinogenic risks to humans. Volume 83. <a href="http://monographs.iarc.fr/ENG/Monographs/vol83/index.php">http://monographs.iarc.fr/ENG/Monographs/vol83/index.php</a>>(accessed 15.05.12).
- Hayes RB (1997) The carcinogenicity of metals in humans. Cancer Causes Control 8:371–385
- Leonard A, Gerber GB (1994) Mutagenicity, carcinogenicity and teratogenicity of vanadium compounds. Mutat Res 317:81–88
- Hossain K, Akhand AA, Kato M et al (2000) Arsenite induces apoptosis of murine T lymphocytes through membrane raft-linked signaling for activation of c-Jun aminoterminal kinase. J Immunol 165:4290–4297
- Chen W, Martindale JL, Holbrook NJ, Liu Y (1998) Tumor promoter arsenite activates extracellular signal-regulated kinase through a signaling pathway mediated by epidermal growth factor receptor and Shc. Mol Cell Biol 18:5178–5188
- Silvera SAN, Rohan TE (2007) Trace elements and cancer risk: a review of the epidemiologic evidence. Cancer Causes Control 18: 7–27
- 53. Khlifi R, Olmedo P, Gil F, Hammami B, Chakroun A, Rebai A, Hamza-Chaffai A (2013) Arsenic, cadmium, chromium and nickel in cancerous and healthy tissues from patients with head and neck cancer. Sci Total Environ 452:58–67
- Benbrahim-Tallaa L, Waalkes MP (2008) Inorganic arsenic and human prostate cancer. Environ Health Perspect 116:158–164
- Navas-Acien A, Sharrett AR, Silbergeld EK, Schwartz BS, Nachman KE, Burke TA et al (2005) Arsenic exposure and cardiovascular disease: a systematic review of the epidemiologic evidence. Am J Epidemiol 162:1037–1049
- 56. Navas-Acien A, Silbergeld EK, Streeter RA, Clark JM, Burke TA, Guallar E (2006) Arsenic exposure and type 2 diabetes: a systematic review of the experimental andepidemiological evidence. Environ Health Perspect 114:641–648
- Marano KM, Naufal ZS, Kathman SJ, Bodnar JA, Borgerding MF, Wilson CL (2012) Arsenic exposure and tobacco consumption: biomarkers and risk assessment. Regul Toxicol Pharmacol 64: 225–232
- Cox LA Jr (2009) Could removing arsenic from tobacco smoke significantly reduce smoker risks of lung cancer? Risk Anal 29:3– 17
- Enterline PE, Marsh GM (1982) Cancer among workers exposed to arsenic and other substances in a copper smelter. Am J Epidemiol 116:895–911
- 60. US Department of Health and Human Services (USDHHS) (2012) Guidance for industry, Reporting harmful and potentially harmful constituents in tobacco products and tobacco smoke under Section 904(a)(3) of the Federal Food, Drug, and Cosmetic Act. Draft Guidance. Food and Drug Administration. Center for Tobacco Products
- Warnakulasuriya S, Trivedy C, Peters TJ (2002) Areca nut use: an independent risk factor for oral cancer. Brit Med J 324:799–800
- Hecht SS (2003) Tobacco carcinogens, their biomarkers and tobacco-induced cancer. Nat Rev Cancer 3:733–744