

# Reactive Oxygen Species Formation and Cell Death in Catalase-Deficient Tobacco Leaf Discs Exposed to Paraquat

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**Abstract** In the present work, the response of tobacco (*Nicotiana tabacum* L.) wild-type SR1 and transgenic CAT1AS plants (with a basal reduced CAT activity) was evaluated after exposure to the herbicide paraquat (PQ). Superoxide anion ( $O_2^{\cdot-}$ ) formation was inhibited at 3 or 21 h of exposure, but  $H_2O_2$  production and ion leakage increased significantly, both in SR1 or CAT1AS leaf discs. NADPH oxidase activity was constitutively 57% lower in non-treated transgenic leaves than in SR1 leaves and was greatly reduced both at 3 or 21 h of PQ treatment. Superoxide dismutase (SOD) activity was significantly reduced by PQ after 21 h, showing a decrease from 70% to 55%, whereas catalase (CAT) activity decreased an average of 50% after 3 h of treatment, and of 90% after 21 h, in SR1 and CAT1AS, respectively. Concomitantly, total CAT protein content was shown to be reduced in non-treated CAT1AS plants compared to control SR1 leaf discs at both exposure times. PQ decreased CAT expression in SR1 or CAT1AS plants at 3 and 21 h of treatment. The mechanisms underlying PQ-induced cell death were possibly not related exclusively to ROS formation and oxidative stress in tobacco wild-type or transgenic plants.

**Keywords** Cell death · Heavy metals · Hydrogen peroxide · *Nicotiana tabacum* · Reactive oxygen species

## Introduction

Reactive oxygen species (ROS) are generated in plants during photosynthesis and respiration, and their homeostasis is regulated by the equilibrium between synthesis and detoxification. Plants can cope with ROS by enzymatic and non-enzymatic antioxidant systems.

The major sites of the main ROS ( $O_2^{\cdot-}$  and  $H_2O_2$ ) formation during abiotic stress is still a matter of debate; however, NADPH oxidase, peroxidases, and lipoxygenase have been put forward as putative sources of ROS production during the oxidative burst [1–3].

Catalase (CAT, EC 1.11.1.6) is a tetrameric iron-porphyrin protein that catalyzes the decomposition of  $H_2O_2$  into water and dioxygen. In plants, it is located in microbodies such as peroxisomes and glyoxysomes and is involved in detoxification of  $H_2O_2$  generated by various environmental stresses like high light [4, 5], temperature [6], and metals [7, 8], as well as  $H_2O_2$  produced under physiological conditions. CAT1 is the dominant CAT isoform in leaves, where it serves in the removal of photorespiratory-derived  $H_2O_2$  [9], and CAT1AS plants (CAT-deficient plants) were originally produced to study the role of CAT in plants [10]. Under photorespiratory conditions created by high light exposure, CAT-deficient plants showed no visible disorders (except for a reduced growth compared to WT plants), and in elevated light

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rapidly accumulated  $H_2O_2$ , developed white necrotic lesions on the leaves, stimulated salicylic acid, and ethylene production, and induced the expression of acidic and basic pathogenesis-related proteins with a timing and magnitude similar to the hypersensitive response against pathogens [9, 11, 12].

The active component of the herbicide paraquat, methyl viologen (MV; 1,10-dimethyl-4,4'-bipyridinium dichloride), exerts its phytotoxic effects on plants by transferring electrons from photosystem I to molecular oxygen, resulting in the accumulation of superoxide radicals in chloroplasts [13]. Both its herbicidal and toxicological properties depend on the transformation of bipyridyl residues to monocation radicals that result from an enzymatic one-electron reduction in the presence of NADPH. The radical reacts with oxygen rendering superoxide anion and regenerating the paraquat cation, thus closing a redox cycle. Thus, MV toxicity results from deleterious reactions of cellular components with ROS which can trigger harmful reactions such as DNA alterations, enzyme inactivation, protein degradation, and lipid peroxidation, affecting key components of plant cell metabolism [1, 5, 14]. It has been reported that paraquat can readily penetrate the leaf cuticle, quickly decreasing photosynthetic activity, rupturing membranes, and causing loss of turgor in treated leaves within hours [15, 16].

Esterases (EC 3.1.1.x) represent a diverse group of hydrolases catalyzing the cleavage and formation of carboxyl ester bonds. They are involved in the formation of the cell wall [17], degradation of xenobiotics [18], and signaling [19]. Esterases have been used as markers of viability in combination with other dyes for a long time. Esterase activity depends on cell viability but is independent of membrane integrity [20]. Their connection with development has made them a suitable marker of development in plants.

In the present work, we compared paraquat-induced damage in wild-type and CAT-deficient plants, with the aim to elucidate the importance of the ROS detoxifying system in avoiding PQ toxicity.

## Materials and Methods

### Plant Growth Conditions and Treatments

*Nicotiana tabacum* var. Petit Havana SR1-wild-type and *N. tabacum* CAT1AS (a transgenic line that expresses only 10–30% of wild-type CAT activity in the leaves and only 40% in the roots due to the antisense expression of the *cat1* gene) derived from *N. tabacum* Petit Havana SR1 [12] were used for the experiment. The seeds of the transgenic tobacco

line were kindly provided by Dr. F. Van Breusegem (Gent University, Belgium). Seeds were germinated and grown as described previously [12]. All treatments were performed in a controlled environmental chamber with a relative humidity of 70% and temperature of 24/21°C for day/night period, with a light intensity of  $120 \text{ mmol m}^{-2} \text{ s}^{-1}$ . Unless otherwise indicated, experiments were performed using the fourth or fifth leaf (counting from the bottom) of 6–8-week-old plants. Leaf discs (8 mm diameter) were cut with a cork-borer, put in glass flasks containing 25 ml of the treatment solution (distilled water or  $100 \mu\text{M}$  PQ), and incubated in a rotary shaker for 3 or 21 h, under continuous illumination. After that, leaf discs were washed with distilled water, dried with adsorbent paper, and used for analysis.

### In Situ $O_2^{\cdot-}$ Localization

To estimate  $O_2^{\cdot-}$  content, leaf discs from SR1 or CAT1AS were immersed in a 0.05% solution of nitroblue tetrazolium (NBT) in 50 mM potassium phosphate buffer (pH 6.4), vacuum-infiltrated for 3 min, and illuminated 2 h until appearance of dark spots. Leaf discs were bleached by immersing in boiling ethanol. Before staining with NBT, leaves were immersed for 2 h in  $20 \mu\text{M}$  diphenylene iodonium (DPI, a NADPH oxidase inhibitor), infiltrated, and then treated with PQ. Inhibition of an oxidative burst by DPI is indicative of the involvement of a NADPH oxidase in the process [21].

### In Situ $H_2O_2$ Localization

Leaf discs from SR1 and CAT1AS plants were exposed to PQ to evaluate “in situ”  $H_2O_2$  production by an endogenous peroxidase-dependent staining procedure using 3,3'-diaminobenzidine (DAB) [22]. Leaf segments were immersed in a  $1 \text{ mg ml}^{-1}$  DAB solution, pH 3.8, vacuum-infiltrated for 3 min, incubated at room temperature for 2 h in light until appearance of brown spots to evidence  $H_2O_2$  formation. Leaves were bleached in boiling ethanol. Ascorbic acid was used as antioxidant to confirm that brown spots correspond to  $H_2O_2$  formation.

### Cell Death Detection: Evans Blue Staining, Electrolyte Leakage, and Esterase Activity

To determine changes in viability of cells by PQ treatment, leaf discs were infiltrated with a 0.25% (w/v) aqueous solution of Evans Blue [23] during 15 min at room temperature and then washed twice for 15 min with distilled water to remove the dye in excess. Finally, leaf discs were incubated in distilled water overnight. For quantitative assessment, blue precipitates were quantified by solubilization

with 1% (*w/v*) SDS in 50% (*v/v*) methanol at 50°C for 1 h, and the absorbance was measured at 595 nm.

Cell death was also estimated by measuring ion leakage from leaf discs according to the method described by Shou et al. [24]. Conductivity of treatment solutions was measured at time zero (T0) and following 3 or 21 h of incubation (T1). Then, leaf discs were boiled in the solution at 100°C for 1 h, and the conductivity was measured again (T2), expressing the results as relative conductivity  $[(T1 - T0)/(T2 - T0)] \times 100$ .

Extracts for determination of esterase (EA, EC 3.1.1.x) were prepared using 0.150 g FW of leaf discs in 1.0 ml of 250 mM potassium phosphate buffer pH 8.7 containing 1 mM DTT and 1 g PVP. The homogenates were centrifuged at 10,000×*g* for 15 min. After centrifugation, 150 µl of supernatant (S) was collected and mixed with 2,850 µl of newly prepared phosphate buffer containing 0.0125% FDA (solution in dry acetone). The mixture was incubated at 35°C for 15 min and rapidly placed on ice to stop the reaction. The absorbance was measured at 490 nm as described by Steward et al. [25]. A control containing phosphate buffer instead of supernatant was performed to take into account FDA self-degradation. The amount of acetone did not exceed 1% in the reaction mixture.

#### Thiobarbituric Acid Reactive Substances Determination

Lipid peroxidation was determined as the amount of thiobarbituric acid reactive substances (TBARS) as described by Heath and Packer [26], using 300 mg FW homogenized in 3 ml of 20% (*w/v*) trichloroacetic acid (TCA). To 1 ml of the aliquot of the supernatant, 1 ml of 20% (*w/v*) TCA containing 0.5% (*w/v*) TBA and 100 µl 4% (*w/v*) BHT in ethanol were added. The mixture was heated at 95°C for 25 min and then quickly cooled on ice. The contents were centrifuged at 3,000 rpm for 3 min, and the absorbance was measured at 532 nm. The value for non-specific absorption at 600 nm was subtracted. The concentration of TBARS was calculated using an extinction coefficient of 155 mM<sup>-1</sup> cm<sup>-1</sup>.

#### Subcellular Fractionation and Assay of NADPH Oxidation

Samples (1 g FW) were homogenized in 3.5 ml of HEPES [4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid]-KOH buffer (pH 7.8), containing 250 mM sucrose, 0.5 mM PMSF, 1.0 mM ethylenediaminetetraacetic acid (EDTA), 1.0 mM DTT, and 0.5 mM MgCl<sub>2</sub> (reaction buffer). To obtain microsomes, the homogenates were centrifuged successively at 600×*g* for 15 min, at 2,000×*g* for 20 min, and at 140,000×*g* for 1 h. From this last centrifugation, pellet (microsomal fraction) and a supernatant (cytosolic fraction) were obtained. The final pellet

was resuspended in 500 µL of the reaction buffer and was used to determine the NADPH oxidation rate, following the method used by Van Gestelen et al. [27], monitoring NBT reduction by NADPH at 530 nm for 3 min. NADPH oxidation activity was calculated by taking the difference between the apparent reaction rates with or without SOD (75 U/ml) in the reaction mixture, using an extinction coefficient of 12.8 mM<sup>-1</sup> cm<sup>-1</sup>.

#### Enzyme Preparations and Assays

Extracts for determination of ascorbate peroxidase (APOX, EC 1.11.1.11) and guaiacol peroxidase (GPOX, EC 1.11.1.7) were prepared using 0.2 g FW of leaf discs in 1.5 ml of 50 mM potassium phosphate buffer pH 7.8 containing 1 mM EDTA and 1 g PVP. APOX activity was measured immediately in fresh extracts as described by Nakano and Asada [28]. GPOX activity was determined following the increase in absorbance at 470 nm due to the formation of tetraguaiacol ( $\epsilon$  26.6 mM<sup>-1</sup> cm<sup>-1</sup>) [29]. Homogenates for superoxide dismutase (SOD, EC 1.15.1.1) and CAT activities were prepared under ice-cold conditions from 0.3 to 0.6 g FW of leaf discs homogenized in 3 or 2 ml of extraction buffer (containing 50 mM phosphate buffer (pH 7.8), 0.5 mM EDTA, 1 g polyvinylpyrrolidone (PVP), and 0.5% (*v/v*) Triton X-100) at 4°C for SOD and CAT, respectively. Total SOD activity was assayed as described by Becana et al. [30]. CAT activity was determined in the homogenates by measuring the decrease in absorbance at 240 nm (Hitachi U-2000, Hitachi Ltd, Tokyo, Japan) in a reaction medium containing 50 mM potassium phosphate buffer (pH 7.2) and 2 mM H<sub>2</sub>O<sub>2</sub>. The pseudo-first order reaction constant ( $k' = k \times [\text{CAT}]$ ) of the decrease in H<sub>2</sub>O<sub>2</sub> absorption was determined, and CAT content in picomoles per gram FW was calculated using  $k = 4.7, 107 \text{ M}^{-1} \text{ s}^{-1}$  [31].

#### Western Blot Analysis of CAT

Soluble protein extracts were prepared from leaf discs, and 25 µg of total protein were electrophoresed on 12% SDS-PAGE in a Mini PROTEAN III equipment (Bio-Rad), as described by Laemmli [32]. Following electrophoresis at 4–8°C, proteins were transferred to a nitrocellulose membrane (Amersham Biosciences). To immunodetect CAT, the membrane was incubated at 4–8°C overnight with a polyclonal antibody raised in rabbit against a cottonseed CAT (kindly provided by Dr Trelease from Arizona State University). Bands were detected with an anti-rabbit IgG peroxidase conjugated secondary antibody (Sigma), and 3,3'-diaminobenzidine (DAB) was used as substrate for staining procedure. Membranes were photographed with a Fotodyne equipment and analyzed with GelPro software.

### Determination of Protein Concentration

Protein concentration for all samples was determined by the method of Bradford [33] using bovine serum albumin as a standard.

### Statistics

All data presented are the mean values of two or three independent set of experiments. Each value was presented as means  $\pm$  standard errors of the mean (SE), with a minimum of three replicates. Statistical analysis was carried out by one-way ANOVA using the Tukey test to evaluate whether the means were significantly different, taking one asterisk  $*p < 0.05$  and three asterisks  $***p < 0.001$  as significant.

## Results

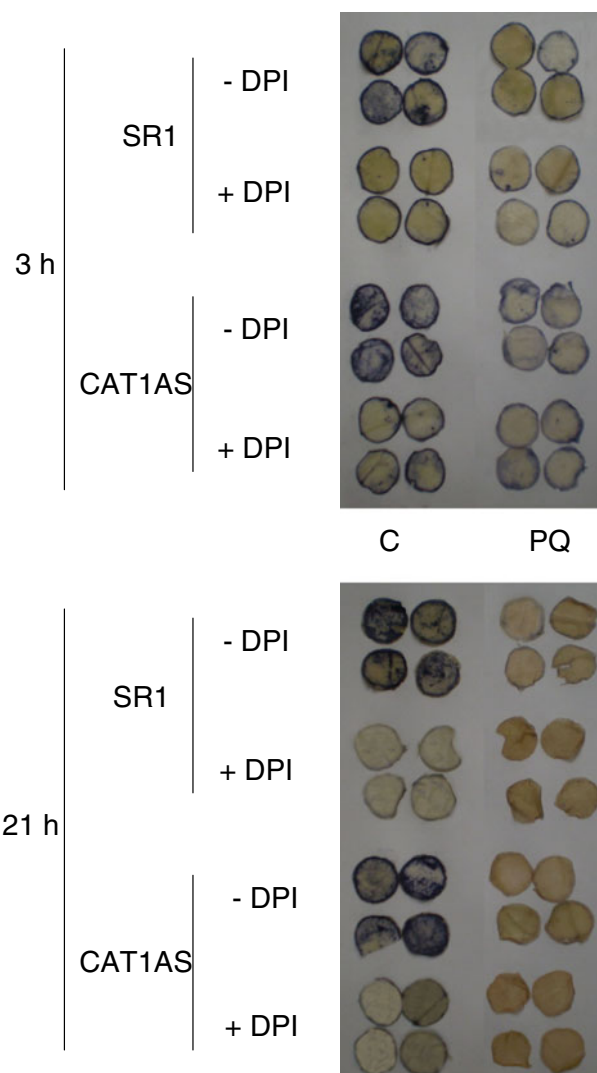
### Measurement of Hydrogen Peroxide and Superoxide Anion Accumulation by NBT and DAB Staining

Superoxide anion was detected by the blue formazan precipitation produced after the reaction with nitroblue tetrazolium. At 3 or 21 h, non-treated SR1 or CAT1AS leaf discs showed similar  $O_2^{\cdot-}$ -dependent formazan deposition that was almost completely reversed when diphenyleneiodonium (DPI), an inhibitor of flavin-containing oxidases like NADPH oxidase, was used (Fig. 1). In PQ-treated leaves, except for a slight formazan formation in CAT1AS leaf discs at 3 h, almost no NBT staining was observed in any of the exposure times, revealing a complete inhibition of  $O_2^{\cdot-}$  formation, either in SR1 or CAT1AS plants.

$H_2O_2$  production was visualized by staining leaf segments with 3,3'-diaminobenzidine (DAB), a histochemical reagent that polymerizes and turns brown in the presence of  $H_2O_2$  (Fig. 2). An evident accumulation of  $H_2O_2$  was observed in leaf discs under PQ treatment, DAB staining being moderately more intense in SR1 than in CAT1AS leaf discs at 3 h of treatment. At 21 h,  $H_2O_2$  content increased compared to 3 h of treatment in both SR1 and CAT1AS leaf discs (Fig. 2). A slight staining was observed in wild-type or transgenic non-treated discs.  $H_2O_2$  accumulation was avoided to different extents by infiltrating leaf discs with ascorbate previously to DAB staining, which demonstrated the specificity of this reaction for  $H_2O_2$  (Fig. 2).

### NADPH Oxidase Activity

The effect of PQ on the generation of  $O_2^{\cdot-}$  through NADPH oxidase activity was measured using microsomal fractions of tobacco leaf discs using the NADPH-dependent NBT

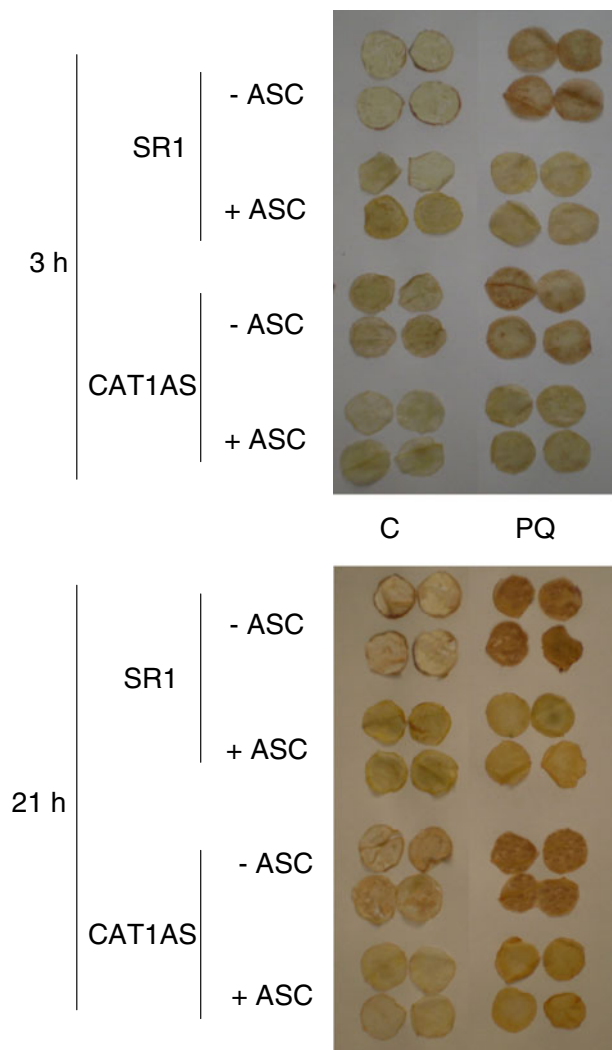


**Fig. 1** Accumulation of  $O_2^{\cdot-}$  in leaf discs of tobacco SR1 or CAT1AS plants. Control and PQ-treated leaf discs were vacuum-infiltrated for 3 min with 0.05% nitroblue tetrazolium (NBT) and incubated 2 h under illumination as described in “Materials and Methods”.  $O_2^{\cdot-}$  formation was revealed by the appearance of blue spots characteristics of formazan deposition. DPI, an inhibitor of flavin-containing oxidases like NADPH oxidase, was used to confirm the involvement of a NADPH oxidase in  $O_2^{\cdot-}$  formation

reduction reaction. NADPH oxidase activity in non-treated transgenic plants was constitutively 57% lower than in control wild-type plants. At 3 h, PQ reduced the enzyme activity to 13% and 18% of the control in SR1 and CAT1AS leaf discs, respectively, whereas at 21 h, NADPH oxidase activity decreased to 59% and 13% of the controls in SR1 and CAT1AS, respectively (Fig. 3).

### Antioxidant Enzymes Activity

Superoxide dismutase activity, thought to be an important source of superoxide anion-derived  $H_2O_2$  in plant cells, did

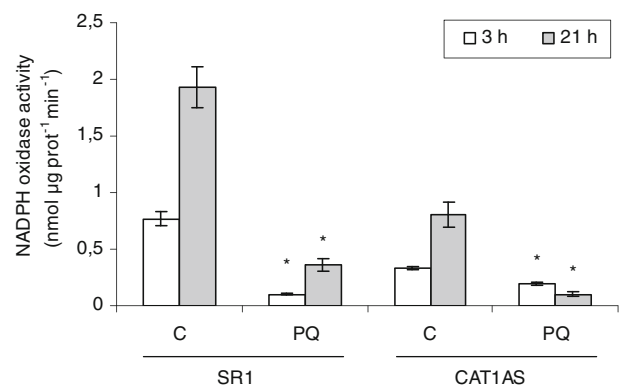


**Fig. 2** Accumulation of  $\text{H}_2\text{O}_2$  in leaf discs of tobacco SR1 or CAT1AS plants. Control and PQ-treated leaf discs were vacuum-infiltrated for 3 min with  $1 \text{ mg ml}^{-1}$  3,3'-diaminobenzidine (DAB) and incubated 2 h under illumination as described in “Materials and Methods”.  $\text{H}_2\text{O}_2$  was revealed by the appearance of brown spots due to DAB polymerization. Ascorbic acid was used as antioxidant to confirm that brown spots correspond to  $\text{H}_2\text{O}_2$  formation

not show changes at 3 h but was significantly reduced at 21 h, in SR1 and CAT1AS leaf discs exposed to PQ, reaching a range of decay from 70% to 55%, respectively ( $p < 0.01$ ) (Table 1).

Analysis of CAT clearly showed that the enzyme activity in non-treated CAT1AS plants was only 15% of that in SR1 plants, thus confirming that the transgenic cultivar conserved the original described protein reduction. PQ treatment reduced CAT activity in a time-dependent manner in SR1 and CAT1AS leaves compared to non-treated leaves, from an average of 50% at 3 h of treatment to 10% at 21 h in both cultivars (Table 1).

The herbicide markedly decreased CAT protein content at 21 h in SR1 leaf discs, but it almost did not affect the



**Fig. 3** NADPH-dependent superoxide production in SR1 or CAT1AS tobacco leaf discs. Leaf discs were exposed to  $100 \mu\text{M}$  PQ for 3 or 21 h. The NADPH-dependent oxidase activity was performed indirectly monitoring NBT reduction at 530 nm according to the description in “Materials and Methods”. Values are the means of three different experiments with three replicated measurements, and bars indicate S.E. Asterisk indicates significant differences ( $p < 0.05$ ) according to Tukey's multiple range test

protein content in CAT1AS plants with respect to the non-treated transgenic leaves. The hardly detectable expression of CAT protein in the SR1 plants at 21 h was in accordance with the enzyme activity by this time. Almost no differences were observed either in SR1 or CAT1AS leaf discs at 3 h of PQ exposure (Fig. 4).

APOX activity was almost completely abolished by PQ at 3 and 21 h of exposure ( $p < 0.001$ ). The total activity of GPOX was also reduced in response to PQ treatment, to 75% and 37% of the controls, in SR1 and CAT1AS, respectively, at 3 h of treatment. Almost no GPOX activity was detected at 21 h of PQ treatment either in SR1 or CAT1AS leaf discs ( $p < 0.001$ ) (Table 1).

**Evidence of Oxidative Damage and Cell Death: Thiobarbituric Acid Reactive Substances Content, Electrolyte Leakage, Evans Blue Staining, and Esterase Activity**

Paraquat-induced lipid peroxidation determined by TBARS content was observed only in wild-type SR1 leaf discs at 21 h of exposure (Fig. 5), where the content of TBARS increased by 52% with respect to non-treated plants.

Ion leakage is considered a cell death marker and can be measured by changes in the conductivity of leaf discs. No symptoms of damage were observed after 3 h of PQ treatment in both SR1 and CAT1AS leaf discs (Fig. 6). Electrolyte leakage significantly increased in leaf discs under 21 h of PQ treatment in SR1 or transgenic plants with increments of 300% and 219%, respectively (Fig. 6). Ascorbic acid almost completely reversed the PQ-induced leakage of electrolytes, suggesting that  $\text{H}_2\text{O}_2$  is playing a key role in PQ-induced membrane damage (data not shown).

**Table 1** Effect of 100  $\mu\text{M}$  PQ on CAT, APOX, GPOX, and SOD activities at 3 and 21 h of treatment

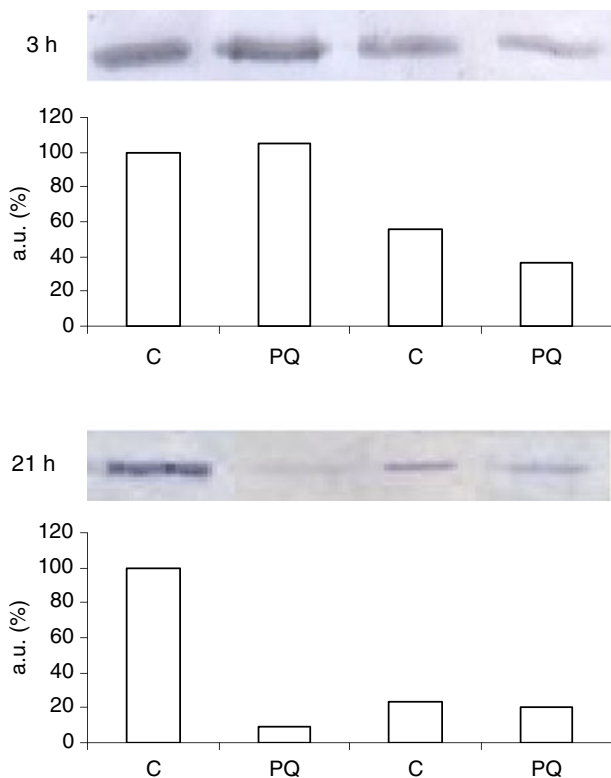
		SR1		CAT1AS	
		C	PQ	C	PQ
CAT ( $\text{pmol g}^{-1} \text{FW s}^{-1}$ )	3 h	0.57 $\pm$ 0.04 <sup>a</sup>	0.28 $\pm$ 0.04 <sup>b</sup>	0.08 $\pm$ 0.03 <sup>a</sup>	0.04 $\pm$ 0.01 <sup>b</sup>
	21 h	1.01 $\pm$ 0.13 <sup>a</sup>	0.11 $\pm$ 0.04 <sup>b</sup>	0.18 $\pm$ 0.04 <sup>a</sup>	0.02 $\pm$ 0.04 <sup>b</sup>
APOX ( $\text{pmol g}^{-1} \text{FW s}^{-1}$ )	3 h	19.14 $\pm$ 1.29 <sup>a</sup>	0.88 $\pm$ 0.85 <sup>b</sup>	13.45 $\pm$ 1.45 <sup>a</sup>	0.39 $\pm$ 0.02 <sup>b</sup>
	21 h	24.97 $\pm$ 7.71 <sup>a</sup>	0.50 $\pm$ 0.12 <sup>b</sup>	47.48 $\pm$ 4.86 <sup>a</sup>	0.60 $\pm$ 0.01 <sup>b</sup>
GPOX ( $\mu\text{mol g}^{-1} \text{FW}$ )	3 h	1,933.9 $\pm$ 347.1 <sup>a</sup>	1,437.9 $\pm$ 142.02 <sup>a</sup>	1,303.08 $\pm$ 27.5 <sup>a</sup>	486.01 $\pm$ 31.5 <sup>b</sup>
	21 h	1,352.91 $\pm$ 368.5 <sup>a</sup>	10.36 $\pm$ 1.74 <sup>b</sup>	1,449.71 $\pm$ 265.1 <sup>a</sup>	12.89 $\pm$ 1.76 <sup>b</sup>
SOD ( $\text{U g}^{-1} \text{FW}$ )	3 h	17.95 $\pm$ 5.39 <sup>a</sup>	18.16 $\pm$ 1.82 <sup>a</sup>	15.91 $\pm$ 3.83 <sup>a</sup>	11.99 $\pm$ 5.47 <sup>a</sup>
	21 h	24.45 $\pm$ 7.23 <sup>a</sup>	7.33 $\pm$ 2.89 <sup>b</sup>	10.93 $\pm$ 4.55 <sup>a</sup>	5.04 $\pm$ 1.98 <sup>b</sup>

Data are the mean $\pm$ SE of two independent experiments, with five replicates for each treatment. Different letters within rows indicate significant differences ( $b, p < 0.05$ ), according to Tukey's multiple range test. One unit of CAT is the amount of the enzyme that oxidized 1  $\mu\text{mol}$  of  $\text{H}_2\text{O}_2$  per minute under the assay conditions. One unit of APOX forms 1 mmol of ascorbate oxidized per minute under the assay conditions. One unit of GPOX is the amount of the enzyme that reduced 1 mmol of  $\text{H}_2\text{O}_2$  per minute under the assay conditions. One unit of SOD is the amount of the enzyme that inhibits the reduction of NBT by 50% under the assay conditions

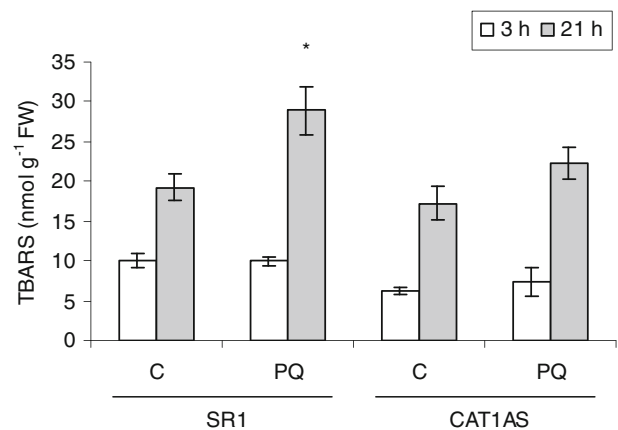
Cell death in tobacco leaf discs exposed to PQ was determined using the Evans blue staining (a dye that

specifically stains dead cells). Surprisingly, no eluted staining was observed in wild-type or transgenic leaf discs exposed for up to 21 h to PQ (Fig. 7).

Esterase activity was measured as another cell death factor because this method is dependent not only on cell viability but also on the metabolic state of the cell [20]. Esterase activity was comparable in non-treated wild-type or transgenic leaf discs at both exposure times. PQ treatment inhibited the enzyme activity to half of that observed in controls at 3 h and an average of 72% at 21 h, in both cultivars (Fig. 8).

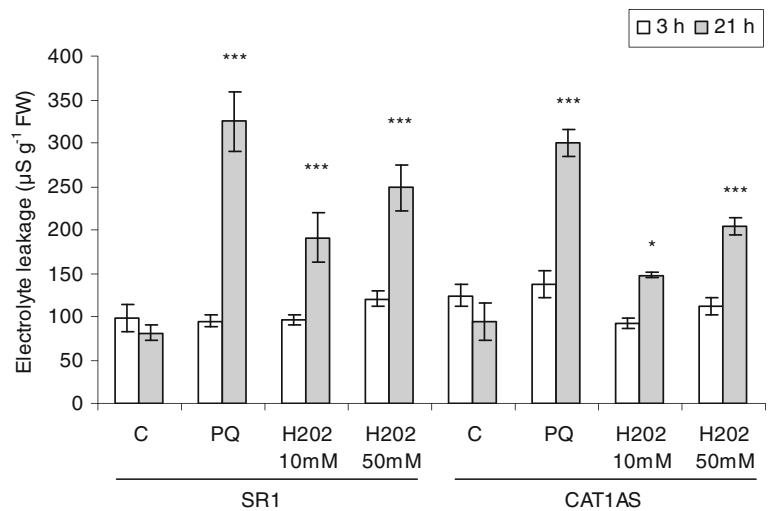


**Fig. 4** Catalase protein expression in leaf discs of tobacco SR1 or CAT1AS plants determined by western blot protein and densitometry analysis. Leaf discs were exposed to 100  $\mu\text{M}$  PQ for 3 or 21 h. CAT expression was revealed with a polyclonal antibody raised in rabbit against a cottonseed CAT as described in “Materials and Methods”. The experiment was repeated three times and a representative image is presented



**Fig. 5** Thiobarbituric acid reactive substances (TBARS) content of tobacco SR1 or CAT1AS plants. Lipid peroxidation was measured in control and PQ-treated leaf discs as the amount of thiobarbituric acid reactive substances according to the description in “Materials and Methods”. Values are the means of two different experiments with three replicated measurements, and bars indicate S.E. Asterisk indicates significant differences ( $p < 0.05$ ) according to Tukey's multiple range test

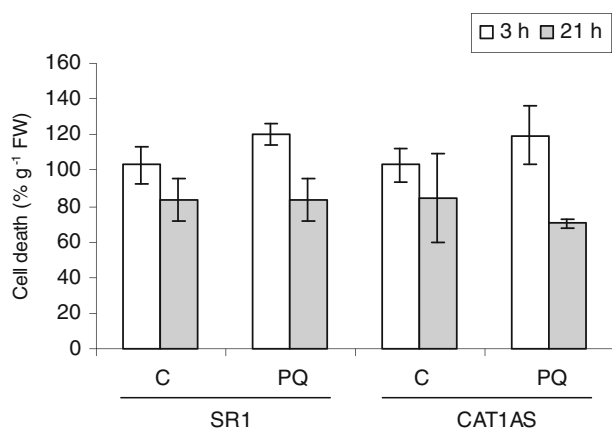
**Fig. 6** Cell death measured as electrolyte leakage in leaf discs of tobacco SR1 or CAT1AS plants, expressed as relative conductivity. Leaf discs were exposed to 100  $\mu\text{M}$  PQ, 10 mM, or 50 mM  $\text{H}_2\text{O}_2$ , for 3 or 21 h as described in “Materials and Methods”. Values are the means of two different experiments with three replicated measurements, and bars indicate S.E. Asterisk indicates significant differences ( $*p < 0.05$ ,  $***p < 0.001$ ) according to Tukey’s multiple range test



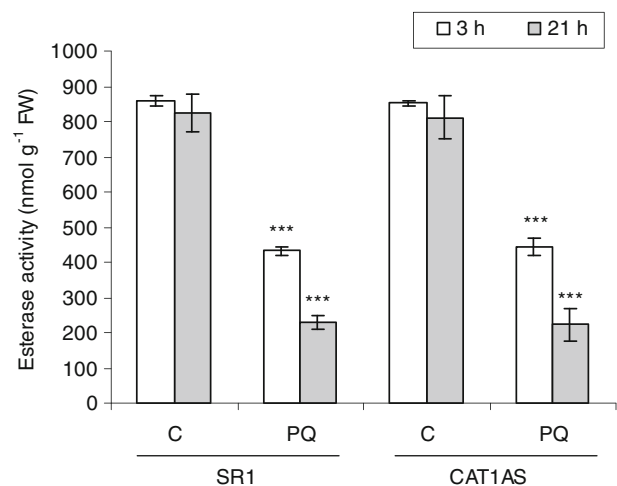
## Discussion

It is well known that PQ causes superoxide-dependent cell death [34] and has been used together with aminotriazole (a CAT inhibitor that leads to  $\text{H}_2\text{O}_2$  accumulation and subsequent cell death [35]) to study ROS involvement in the cell death processes in *Arabidopsis* plants [36]. Considerable controversy about the mechanisms involved in PQ toxicity in animals is well documented [37]. Peixoto et al. [38] demonstrated that  $\text{H}_2\text{O}_2$  formation is not the major factor of PQ toxicity at mitochondrial level in plants and animals and showed that PQ toxicity depends on fatty acid composition of membranes related to peroxidability. Hence, the main objective of this work was to investigate whether CAT-deficient plants had an increased susceptibility to PQ exposure compared to the wild-type SR1 plants.

In either wild-type or transgenic leaves exposed for 3 or 21 h to PQ, NBT staining was completely abolished, and this reduction was related to a PQ-direct inhibition of superoxide generation dependent on a NADPH oxidase. This result was rather surprising considering that PQ is an herbicide which alters the normal electron transport in chloroplasts and consequently, increased ROS formation in plants. In spite of the diminished superoxide anion production, DAB deposition due to  $\text{H}_2\text{O}_2$  accumulation was clearly evident. This accumulation could be the consequence of PQ transfer of 1  $e^-$  to  $\text{O}_2$  to generate  $\text{O}_2^{\cdot-}$  in illuminated cell chloroplasts, which is then converted to  $\text{H}_2\text{O}_2$  either spontaneously or enzymatically via superoxide dismutase and by PQ inhibition of  $\text{H}_2\text{O}_2$ -detoxifying



**Fig. 7** Cell death estimated as Evans blue staining in leaf discs of tobacco SR1 or CAT1AS plants, expressed as percentage of the controls. Leaf discs were exposed to 100  $\mu\text{M}$  PQ for 3 or 21 h as described in “Materials and Methods”. Values are the means of two different experiments with three replicated measurements, and bars indicate S.E.



**Fig. 8** Cell viability measured as esterase activity in leaf discs of tobacco SR1 or CAT1AS plants. Leaf discs were exposed to 100  $\mu\text{M}$  PQ for 3 or 21 h as described in “Materials and Methods”. Values are the means of two different experiments with three replicated measurements, and bars indicate S.E. Asterisks indicate significant differences ( $***p < 0.001$ ) according to Tukey’s multiple range test

enzymes activities (CAT, APOX, and GPOX). ROS could also have been generated by peroxidases [39] as well as amino oxidases in the apoplast and xanthine oxidase [40, 41]. In addition, the increase of catalytic Fe in PQ-treated plants postulated by Iturbe-Ormaetxe et al. [42] could be strengthening PQ toxicity by the generation of hydroxyl radicals [43].

In PQ-treated tobacco leaves, the accumulation of  $H_2O_2$  was strongly correlated to an increase in electrolyte leakage and a decrease in esterases activity, both are markers of cell damage or cell death. However, PQ-treated leaves showed no Evans positive staining and TBARS only increased in SR1 leaves treated with PQ. Montillet et al. [44] demonstrated that a massive  $H_2O_2$  production is necessary to provoke an intense ROS-mediated lipid peroxidation despite that lipid peroxidation does not always accompany cell death processes in CAT1AS tobacco plants.

The lack of correlation of Evans blue staining with the other cell death markers in tobacco leaves could be a consequence of the severe injury of membranes that occurred in leaves under PQ treatment, which in turn could not retain the blue stain inside the cells, giving an underestimated result [25].

The avoidance in the increase in electrolyte leakage mediated by ascorbic acid or CAT, either in the wild-type SR1 or in the transgenic CAT1AS leaf discs (data not shown), clearly suggested the involvement of  $H_2O_2$  in the cell death process, in accordance with the results of Dat et al. [5] who showed that changes in  $H_2O_2$  homeostasis induced cell death in tobacco CAT1AS plants exposed to high light intensities.

CAT deficiency reduces the  $H_2O_2$ -removing capacity of plant cells and, consequently, may lead to a higher steady-state of the other  $H_2O_2$  detoxifying enzymes, as APOX and GPOX [45]. Willekens et al. [9] demonstrated that tobacco transgenic CAT1AS plants activated alternate enzymatic mechanisms for  $H_2O_2$  scavenging to compensate for the shortage of CAT after being transferred to high illumination (300  $mmol/m^2/s$  PPFR) for 48 h, showing a persistent increase in both glutathione peroxidase and ascorbate peroxidase expression but without avoiding necrosis in CAT1AS plants. A similar result was obtained in tobacco SR1 or CAT1AS control leaf discs, where APOX activity was higher in the transgenic cultivar compared to SR1 at 21 h of incubation under light (Table 1). However, PQ treatment inhibited APOX activity in both cultivars contributing to the strong toxicity of the herbicide observed by the electrolyte leakage or cell death at 21 h of treatment. This result is supported by the data of Murgia et al. [46] that highlighted the role of APOX in the fine modulation of  $H_2O_2$  levels.

The overall results obtained in the present work suggested that cell damage was mediated by an increased level of  $H_2O_2$  in PQ-treated tobacco plants, though to different extents depending on the cultivar and time. In our system,  $H_2O_2$  increased independently of the inhibition of NADPH oxidase activity, which suggests other ways of  $H_2O_2$  formation. In chloroplasts under normal physiological conditions, SOD and APOX activities are closely balanced, i.e., the excess of superoxide radicals is eliminated by SOD, generating  $H_2O_2$  that is scavenged through APOX activity. The increase observed in  $H_2O_2$  could not be detoxified in tobacco leaves in this way due to the strong decay in both GPOX and APOX activities. Some experimental evidence has suggested that APOX is more labile than SOD under oxidative stress conditions generated by MV in plants like spinach [47] or tobacco [48]. In the presence of PQ, chloroplasts start to accumulate  $H_2O_2$  within minutes [28], and APOX is the first enzyme to become inactivated after short light exposure [47]. In accordance, in SR1 or CAT1AS PQ-treated leaves, SOD was shown to be more resistant to damage than APOX (Table 1).

The results obtained in the present work suggested that the mechanisms underlying PQ-induced cell death were possibly not related exclusively to ROS formation and oxidative stress in tobacco wild-type or transgenic plants, and these facts are currently under investigation.

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