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Eco-friendly synthesis of substituted tetrahydroquinolines as potential ecdysone receptor agonists

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Abstract

New eco-friendly approaches were proposed for the synthesis of *cis*-cyclopentene-annulated heterocyclic compounds containing a tetrahydroquinoline moiety. For the first time we implemented a one-pot three-component cyclocondensation of aromatic amines (aniline, 5-aminoquinoline, *o*-phenylenediamine), aldehydes, and cyclopentadiene (CPD) in water and in the ionic liquid. The effect of synthesized substituted 3a,4,5,9b-tetrahydro-3*H*-cyclopenta[*c*]quinolines on the reproduction of house fly imago and on the initial stage of offspring ontogeny was evaluated in comparison with the effect of natural insect hormone using biological screening (*Musca domestica*). The most probable factors of stabilization of prepared compounds **7–10** in the active site of the *Heliothis virescens* receptor were identified using AutoDock 4.2, AutoDock Vina, and GOLD Suite molecular docking software. According to the results of three scoring functions, 4-(3-chlorophenyl)-8-fluoro-3a,4,5,9btetrahydro-3*H*-cyclopenta[*c*]quinoline forms the most stable complex with the chosen receptor. The results of bioassays and molecular docking indicate that these compounds may be considered as potential ecdysone agonists.

Keywords Tetrahydroquinoline · Ecdysone agonists · Eco-friendly approaches · Povarov reaction

Introduction

The growing demand for the development of environmentally safe chemical processes more and more often attracts the attention of synthetic chemists focusing on the synthesis of pure chemical compounds for health care, increasing life standards, and environmental safety. The humanistic nature of green chemistry is reflected by its twelve rules, proposed in 1998 by P. Anastas and J. Warner, for the development of new chemical processes ranging from the used feedstock to efficiency and waste safety (Anastas and Eghbali 2010).

Natural and synthetic compounds with a tetrahydroquinoline moiety possess a broad spectrum of biological activities (Ghashghaei et al. 2018; Muthukrishnan et al. 2019), such as anti-inflammatory (Gosmini et al. 2014), antitumor (Chen et al. 2016; Dayal et al. 2020; Hanashalshahaby et al. 2019), antituberculosis (Chavan et al. 2019; Kumar et al. 2011), antifungal (Chander et al. 2016; Chavan et al. 2019; Ozaki et al. 2019), antibacterial (Diaz et al. 2018; Kimura et al. 2019; Martínez et al. 2019; Onyedibe et al. 2021), and antimalarial (Bendale et al. 2007; Van Voorhis et al. 2007) activities. Not long ago, it was found that tetrahydroquinoline derivatives are non-steroidal agonists of *Aedes albopictus* mosquito ecdysone receptor, responsible for their development and reproduction (Kitamura et al. 2014; Ueno et al. 2021; Yokoi et al. 2019), and that they possess a larvicidal activity, which offers prospects for practical use for the development of new pest control agents (Fig. 1).

The most popular and optimal approach to the synthesis of substituted tetrahydroquinolines is the Povarov (aza-Diels–Alder) reaction, which consists in formal [4+2]-cycloaddition of aromatic imines (Schiff bases) to electron-rich olefins catalyzed by Lewis or Brønsted acids (Povarov 1967). The three-component one-pot version of the Povarov reaction proved to be a synthetically convenient and atom-economic method for the preparation of this

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Fig. 1 Tetrahydroquinoline derivatives as potent ecdysone agonist

type of structures (Glushkov and Tolstikov 2008; Tolstikov et al. 2014b). However, the catalysts and solvents used in this reaction (Glushkov and Tolstikov 2008) often suffer from drawbacks such as high cost, water sensitivity, poor availability, and toxicity. This complicates scaling up of the synthesis for both engineering and economic reasons. Therefore, it is necessary to develop modern approaches that would comply with green chemistry principles (Anastas and Eghbali 2010; Van Aken et al. 2006) for the synthesis of tetrahydroquinoline derivatives. The search for cheap acid catalysts free from heavy metals and for green solvents for the Povarov reaction and the synthesis of polysubstituted tetrahydroquinoline derivatives is a relevant task (Petronijević 2017). Ionic liquids (ILs) have occupied a decent place in organic synthesis in recent decades and are postulated as green reagents, owing to their stability, low toxicity (Rogers and Seddon 2003), and reusability (Sheldon 2005). Examples of the use of ILs to promote aza-Diels-Alder reactions with microwave or electrochemical assisted are described (Bortolami et al. 2021; Mert-Balci et al. 2013).

In order to develop eco-friendly approaches to the synthesis of polycyclic compounds containing a tetrahydroquinoline moiety, we implemented a one-pot three-component cyclocondensation of aromatic amines (aniline, 5-aminoquinoline, o-phenylenediamine), aldehydes, and cyclopentadiene (CPD) in water and in ILs. Biological screening of the obtained tetrahydrocyclopenta[c]quinolines was carried out using the *Musca domestica* insect. The results are analyzed in comparison with the data of molecular docking.

Results and discussion

The possibility of synthesizing Schiff bases from aromatic aldehydes and S-methyl, S-benzyl, and S-*n*-octyl-dithiocarbazates and thiosemicarbazides both in water and with lemon juice catalyst was reported previously (Ali et al. 2020). The authors noted that the yield of the target product in the developed eco-friendly procedure was higher than that in the conventional procedure implying heating in ethanol. In another study (Petronijević 2017) describing the syntheses of 3,4-dihydro-2(1*H*)-quinoxalinones and 3,4-dihydro-1,4-benzoxazin-2-ones, diluted (1:10) freshly squeezed lemon juice was used as the biocatalytic medium for the condensation of aromatic amines with enolates.

Analysis of the published protocols stimulated us to implement one-pot cyclocondensation (Povarov reaction) under conditions complying with the green chemistry principles. Previously, tetrahydro-3*H*-cyclopenta[*c*]quinolines were synthesized in acetonitrile or using CF₃CO₂H as the catalyst (Tolstikov et al. 2014b). There are few reported examples of eco-friendly atom-economic one-pot synthesis of polycycles with a tetrahydroquinoline moiety (Li et al. 2015). We accomplished three-component cyclocondensation of 4-fluoroaniline 1 with an equimolar amount of ethyl glyoxylate 3 and a threefold molar excess of cyclopentadiene (CPD) in aqueous lemon juice (1:1), which afforded the target cyclopentene-annulated tetrahydroquinoline 7 in 40% yield within 15 min (Scheme 1). When the reaction mixture was heated to 70°C for 20 min, the yield of the target product considerably increased.

The reaction is diastereoselective, as indicated by the homo-and heterocorrelation ¹H and ¹³C NMR spectra of the cyclocondensation product **7.** Single-crystal X-ray diffraction data for **7** unambiguously prove the structure of rel-($3aR^*$, $4S^*$, $9bS^*$)-4-ethoxycarbonyl-8-fluoro-3a,4,5,9b-tetrahydro-3*H*-cyclopenta[*c*]quinoline (Fig. 2).



 $R = F_{(1, 8, 9)}, H_{(2, 10)}; Ar = 2 - F - C_6 H_4 (4, 10), 4 - C F_3 - C_6 H_4 (5, 8), 3 - C - C_6 H_4 (6, 9)$

Scheme 1 Synthesis of 4-substituted-3a,4,5,9b-tetrahydro-3H-cyclopenta[c]quinolines 7-10



Fig. 2 Molecular structure of rel- $(3aR^*, 4S^*, 9bS^*)$ -4-ethoxycarbonyl-8-fluoro-3a, 4, 5, 9b-tetrahydro-3*H*-cyclopenta[*c*]quinoline 7 in the crystal

An attempt to carry out three-component cyclocondensation of 4-fluoroaniline **1** with aromatic aldehydes **4–6** under the developed conditions resulted in the formation of only Schiff bases (Celik and Kuzu 2019; Kerner et al. 2016).

The synthesis of 4-aryl-3a,4,5,9b-tetrahydro-3*H*-cyclopenta[*c*]quinolines **8–10** was affected when the three-component cyclocondensation was carried out in IL (1-ethyl-3-methylimidazolium tetrachloroaluminate). The reaction of amine **1**, **2** with aromatic aldehyde **4–6** and CPD in 1 mL of IL for 15 min resulted in a quantitative and diastereoselective formation of *cis*-cyclopentene-annulated tetrahydroquinolines **8–10** (Scheme 1). The formation of these products was confirmed by coincidence of the physicochemical assignment results with the data reported previously (Tolstikov et al. 2014b). The spin–spin coupling constant values ³ *J* (3a, 9b)=8, ³*J* (4, 3a)=3.2 Hz (Tolstikov et al. 2014b) of compounds **8**, **9** and ³*J* ((3a, 9b), (4, 3a)=8 Hz of **10**, proved their *cis*-orientation and 3a*R**, 4*S**, and 9b*S** relative configurations of the stereogenic centers.

In order to extend the potential of the cyclocondensation reaction taking place in aqueous lemon juice and ILs, 5-aminoquinoline **11** and *o*-phenylenediamine **12** were used as the amine components. This gave cycloadducts 13a, 13c and 15a, 15b (Scheme 2). The formation of cyclopentene-annulated 4-ethoxycarbonyl-3a,4,5,11btetrahydro-3H-cyclopenta[c]-1,7-phenanthroline **13a** in aqueous lemon juice occurred with a low yield (7%), and heating and increase in the reaction time did not affect the reaction pathway, which led to imine 13b as the major product. A change in the reaction pathway toward the formation of the target product was attained in IL, in which the three-component condensation of *p*-trifluoromethylbenzaldehyde, aminoquinoline 11, and CPD resulted formation of 4-[4-(trifluoromethyl)phenyl]-3a,4,5,11b-tetrahydro-3H-cyclopenta[c][1,7]phenanthroline **13c** (81%). It is noteworthy that the product of the eco-friendly cyclocondensation formed diastereoselectively (dr > 95%) and in a higher yield than under the conditions (CF₃CH₂OH, CF₃COOH, Ar, ~25 °C) developed previously for the one-pot synthesis of cyclopenta[c]1,7-phenanthrolines (65%) (Tolstikov et al. 2014a). The spin-spin coupling constants (J(3a,4)=2.4)and J(3a,11b) = 8,4 Hz) of the vicinal protons at the newly formed C(3a), C(4), and C(11b) asymmetric centers of compound 13c correspond to their *cis* relative positions.

The condensation of *o*-phenylenediamine with ethyl glyoxylate and CPD in aqueous lemon juice afforded Schiff base **14**. The target annulation product was formed in an ionic liquid. The structure molecules **15a**, **b** are characterized by double set of proton and carbon signals of equal intensity in the ¹H and ¹³C NMR spectra, which is indicative of *syn/anti* (1:1, ¹H NMR)-isomer composition (Savchenko et al. 2022). Under these conditions, the yield of the summarized 4,7-bis((4-(trifluoromethyl)phenyl)-3,3a,4,5,6,7,7a,8,10a,12b-decahydrodicyclopenta[*c*,*i*]-1,10-phenanthroline **15a**, **b** was 64%.

Thus, we developed an efficient diastereoselective onepot approach to the three-component cyclocondensation of arylamines, aldehydes, and cyclopentadiene consistent with green chemistry principles.



Scheme 2 Cyclocondensation of 5-aminoquinoline 11 and o-phenylenediamine 12 with aldehydes and cyclopentadiene under eco-conditions

Bioassay

The synthesized compounds **7–10** were tested with respect to 24-h-old image of Sh gen strain (F 182) *Musca domestica*. Once out of the puparium, the adult insects were placed into 300 cm³ cages (three pairs in each cage; two repetitions). The reference compound, 20-hydroxyecdysone (20E), and the test compounds were added to the drinking bowls in concentrations of 1×10^{-8} M during 4 days. Then the solutions in drinking bowls were replaced by pure water, and the insect reproduction was monitored for 10 days by placing standard containers with egg-laying substrate into the cages. The development of laid eggs was observed up to the preparation of larvae for pupation. The control cages and drinking bowls contained pure water.

The assessment of effects of tetrahydroquinolines **7–10** on the adult house fly and the initial stage of offspring ontogeny demonstrated that the test compounds tend to accelerate the reproductive maturation; as a result, the egg-laying period started 2–4 days earlier than that in the control. The most pronounced acceleration was observed in the presence of compounds **10** (Fig. 3). It is noteworthy that this effect was retained when mixtures with 20E were used, and ecdysterone, which has a stimulating activity

toward the house fly imago, completely counterbalanced the adverse effect of the compounds on the fecundity: Compound 8 only completely suppressed reproduction. As a result, the specific fecundity of the females treated with mixtures of 7–10 with 20E did not differ from that for 20E alone (Savchenko et al. 2015). However, this acceleration of the reproduction apparently took a lot of resources of the organism and, consequently, the lifespan of the imago markedly decreased. The decrease of the lifespan by onethird of the normal value was enhanced by combination with 20E.

The beneficial effect of these compounds was also noted for embryogenesis: The hatching of eggs laid by females was almost 100%, while in the control groups, no more than 70–80% of eggs were hatched (Fig. 4). The differences between the compounds were observed in the delayed effects: Only compounds **8** (in mixture with 20E) and **10** accelerated the development of larvae emerging from eggs and induced the arrival of the larval–pupal transformation phase 3–4 days earlier compared to the control.

All of the effects that we detected indicate that compounds 7-10 are incorporated into the regulation of protein biosynthesis in the same way as 20E, that is, via the interaction with the insulin signaling pathway (Cai et al. 2016; Mendes and Mirth 2016). The facts of shortening of the



Fig. 3 Effects of tetrahydroquinolines 7–10 and their mixtures with 20E on the reproductive maturation period of the house fly imago. Red asterisk means significant difference with control (p < 0.05)

Fig. 4 Effects of compounds 7–10 and their mixtures with 20E on the house fly offspring development. Significant difference with control (p < 0.01) noted for variants 7, 9–10 and mixtures 8–10 with 20E

imago lifespan and decrease in the female fecundity, which confirm too fast resource utilization, may be indicative of incorporation of the obtained compounds into the regulation of carbohydrate and lipid metabolism, resulting in disturbance of metabolism synchronization with the energy demand of reproduction (Hou et al. 2015).

Molecular docking

The ecdysone receptor (EcR) has a remarkable ability to structurally adapt to various types of ligands. EcR binds to ecdysteroids, including 20-hydroxyecdysone (20E) and ponasterone A, and non-steroidal synthetic agonists such as dibenzoyl hydrazine (DBH) insecticides, e.g., N'-tertbutyl-N'-(3,5-dimethylphenyl)carbonyl-5-methyl-2,3-dihydro-1,4-benzodioxine-6-carbohydrazide (HWG) (Kitamura et al. 2014) (Fig. 5). The concept of receptor adaptability and plasticity of ligands was considered in a previous study (Billas et al. 2003). The subsequent structural observations provided a rational interpretation of the flexible ability of EcR to bind to and to be activated by chemically and structurally diverse ligands (Browning et al. 2021). Here we report a molecular docking study of EcR adaptability toward binding to potential synthetic agonists of the tetrahydroquinoline type in comparison with known data for native ecdysteroid and DBH-type synthetic agonist (HWG).

The synthesized structures **7–10** were docked into the active site of the target protein that was obtained from the isolated crystal structure of the moth *Heliothis virescens* EcR-USP heterodimer complexed with ponasterone A (PonA) (1r1k) with a resolution of 2.9 Å. For comparison, we also estimated the binding affinity in silico of PonA and the synthetic ligand HWG to the same active site of protein macromolecule EcR. The computing experiment was carried out using the AutoDock 4.2 (www.ebi.ac.uk), AutoDock Vina, and GOLD Suite (www.ccdc.cam.ac.uk) software programs.

Known, that three-component Povarov reaction, catalyzed by Lewis or Brønsted acids between aromatic amine, aldehyde, and CPD affords the preferential formation of *endo/exo,cis*- tetrahydroquinoline and its minor *trans*-isomer (Glushkov and Tolstikov 2008). Although in our eco-compatible experiments we observe the formation of diastereomer I (*endo/exo-cis*) mainly (NMR ¹H, dr [>] 98%), we decided to conduct molecular docking for the probable *trans*-isomer II too. The isomers *endo-cis* I and *exo-cis* I (Fig. 6) have the same NMR spectra, and their specific rotation is equal by zero.

The values of the three scoring functions and the rootmean-square deviations between the solutions obtained for *endo/exo-cis* I and *endo/exo-trans* II isomers are presented in Table 1 and Table 2 SI, respectively. A comparison of the results of molecular docking (Table 2 SI) of *exo-cis* I and *exo-trans* II isomers of the most active compounds 9 and 10 shows high values of the root-mean-square deviation (RMSD \geq 2), which indicates a low probability of validity of the data obtained and makes it possible to exclude the consideration of *exo-cis/trans* isomers in the subsequent study.

In most cases, the coordinates of synthesized structures calculated using two different scoring functions (GOLD Suite and AutoDock Vina) coincided within the error, RMSD = 0.5 Å (Table 1). In the case of estimation of the ligand affinity to the target protein using three scoring functions (GOLD Suite, AutoDock 4.2, and AutoDock Vina), an allowable deviation of the obtained coordinates from one another was observed for stereoisomers 9(I), 9(II), and **10(II)** (Table 1). The best agreement between the docking solutions regarding both the calculated ligand coordinates and binding energies to the protein was found for ligands 9(I), 9(II), and 10(II). The most pronounced differences between the estimated coordinates and scoring functions were observed for ligands 7(I, II) 8(I, II), and 10(I) (Table 1, RMSD \geq 2). For example, the convergence of solutions obtained for ligands 8(I) and 9(II) was demonstrated using discussed docking programs (Fig. 7). This differences in the



Fig. 5 Structural formulas of agonists EcR

Fig. 6 Structures of the stereoisomers I and II of tetrahydroquinolines 7–10



R:CO₂Et, X=F (7); 4-CF₃-C₆H₄, X=F (8); 3-Cl-C₆H₄, X=F (9); 2-F-C₆H₄, X=H.

Table 1GOLD, AutoDock,and AutoDock Vina scoringfunctions of the studied *endo-cis* I and *endo-trans* II isomersof tetrahydroquinolines 7–10and root-mean-square deviationof the obtained coordinates

No.	PLP* fitness	ADT2*, kJ/mol	AVina*, kJ/mol	RMSD*, Å (Gold vs ADT2)	RMSD*, Å (GOLD vs AVina)	RMSD*, Å (ADT2 vs AVina)
7(I)	66.32	- 7.46	- 8.4	≥2	0.387	≥2
8 (I)	62.85	- 8.16	- 8.8	≥2	0.470	≥2
9(I)	66.33	- 9.03	- 10.0	0.384	0.347	0.080
10(I)	66.30	- 8.55	- 8.7	≥2	0.352	$2 \ge$
7(II)	69.27	- 7.58	- 8.3	≥2	0.176	$2 \ge$
8 (II)	83.21	- 8.48	- 9.7	≥2	0.461	$2 \ge$
9(II)	76.38	- 8.77	- 9.7	0.461	0.469	0.202
10(II)	73.22	- 8.14	- 9.7	0.469	0.466	0.086
PonA	98.86	-14.14	- 12.40	0.188	0.188	0.1>
HWG	55.57	- 9.08	- 6.2	≥2	≥2	1.4
20E	97.83	- 13.53	- 10.80	0.132	0.129	0.1>

PLP* fitness-the values of the scoring function of GOLD Suite

ADT2*-the values of the scoring function of AutoDock

AVina*-the values of the scoring function of AutoDock Vina

RMSD*-root-mean-square deviation of the coordinates obtained by two different scoring functions



Fig. 7 Comparison of the solutions obtained for **9**(**I**) and **8**(**II**) using different molecular docking programs: **a 9**(**I**) in GOLD and AutoDock 4.2; **b 8**(**II**) in AutoDock 4.2 and AutoDock Vina

docking results are due to different methods for estimating the steric and energy matching of the ligands to the ligandbinding cavity of EcR inherent in these software programs.

The results of simulation of EcR complexes with compounds **7–10**, obtained using three scoring functions of various types (AutoDock 4.2 (**ADT**), AutoDock Vina (**AV**), and GOLD), are summarized in Table 3 SI. Same Table 3 SI gives detailed description for all factors ensuring effective binding of the test ligands to the receptor active site.

As an example, Fig. 8 shows the docking solution for tetrahydroquinoline 9(I), which resides in the same spatial area of the protein as the cyclic part of the native ligand



Fig.8 Simulated interactions of synthesized ligand 9(I) with amino acid residues in the active site of the target protein (EcR)

ponasterone A. Stabilization of the position of studied ligand in the EcR active site is enhanced by hydrogen bonds and by hydrophobic interactions with the nearby amino acid residues of the H1, H3, H5, and H6 helices and β -sheet (Table 3 SI). In particular, this ligand is stabilized in the active site of the receptor owing to hydrogen bonds with THR346, MET380; hydrophobic interactions with VAL395, ARG387, VAL384, LEU420; and to π - π -interactions with PHE397.

The *endo*-isomers (**I**, **II**) of compounds 7–10 in the in silico obtained EcR complexes are arranged as two closely spaced clusters. For example, Fig. 9 shows the location of the 9(**I**) and 9(**II**) isomers belonging to different clusters relative to ponasterone A and HWG in the active site of target protein. Compounds 7(**I**)–10(**I**) form the first cluster of docking solutions. These ligands occupy an area of the cyclic part of ponasterone A and interact with the amino acid residues of H1, H3, H5, and H6 helices and β -sheet. In the considered case, a high level of steric complementarity with EcR according to the results of three scoring functions was found for ligand 9(**I**) (see Table 1). The docked positions of ligands 7(**I**)–10(**I**) relative to PonA and HWG are illustrated in SI (Fig. 1).

Compounds **7(II)**–**10(II)** form the second cluster of docking simulations, because the positions of these ligands are shifted toward the side chain of PonA in the docked model. One of the aromatic rings of these molecules occupies the same cavity of the binding pocket as the HWG ligand's aromatic moiety (Fig. 10). According to the results of the three computational methods, stereoisomers **9(II)** and **10(II)** show the highest binding affinity to the target protein among the tested products of the second cluster. The docked position of the synthetic ligand (HWG) takes place of the side chain of PonaA in the binding site of EcR (Fig. 9), and it demonstrates a lower level of affinity in silico to the EcR compared to ponasterone A (Table 1).



Fig. 9 The docked positions of ligands 9(I) and 9(II) relative to HWG and PonA in the active site of the protein EcR (AutoDock 4.2, AutoDock Vina, and Gold Suite software): the HWG ligand (gray), the PonA ligand (blue), 9(I) (yellow), 9(II) (red)

Thus, docking results against *Heliothis virescens* EcR-USP heterodimer complexes with the synthesized compounds **7–10** were indicated that all exceed the synthetic HWG ligand in terms of affinity in silico, but they are inferior to natural ecdysteroids. However, the difference in the values of the scoring functions that were obtained for natural ligand and for the considered tetrahydroquinoline molecules indicates the potential strength of the interaction between protein with synthesized ligand.



Fig. 10 The docked position of reference ligand with code HWG and ligands **7(II)–10(II)** in the active site of the protein EcR (1r1k) (AutoDock 4.2, AutoDock Vina, and Gold Suite software): the HWG ligand (gray), **7(II)** (yellow), **8(II)** (magenta), **9(II)** (green), **10(II)** (cyan)

Experimental

General methods

One-dimensional (¹H and ¹³C) and two-dimensional (COSY, NOESY, HSQC, and HMBC) NMR spectra of compounds were recorded on Bruker Avance 400 HD Ascend spectrometer (400.13 MHz for ¹H and 100.62 MHz for ¹³C) and Bruker Avance II 500 HD Ascend spectrometer (500.17 MHz for 1H and 125.77 MHz for ¹³C) using standard Bruker pulse sequences. For NMR data, the chemical shifts are reported in δ (ppm) referenced to residual solvent protons and ¹³C signals in deuterated chloroform or methanol. Coupling constants (J) are expressed in Hertz (Hz). High-resolution mass spectra (HRMS) were measured on an instrument («MaXis impact», Bruker) using a time-of-flight mass analyzer (TOF) with electrospray ionization (ESI). In experiments on selective collisional activation, the activation energy was set at maximum abundance of fragment peaks. A syringe injection was used for solutions in MeCN (flow rate 5 μ L/min). Nitrogen was applied as a dry gas; the interface temperature was set at 180 °C. Column chromatography and TLC were performed using silica gel (< 0.06 mm) and pre-coated silica gel (Silufol plates), respectively; spots were processed by treatment with a solution of 4-hydroxy-3-methoxybenzaldehyde in ethanol, acidified with sulfuric acid. Melting points were determined on Boetius hot-stage microscope. Crystal of the compound 7 mounted on glass fiber was studied with an Xcalibur Gemini Eos automated four-circle diffractometer (graphite monochromator, MoK α radiation, $\lambda = 0.71073$ A, ω -scan mode, $2\theta \max = 62^{\circ}$) at ambient temperature (293–298 K). Collected data were processed using the program CrysAlisPro (CrysAlisPRO 2012). Structures determination was carried out with the OLEX2 program (Dolomanov et al. 2009). The structures were solved by direct methods and refined using the full-matrix least-squares method in the anisotropic approximation for non-hydrogen atoms. All hydrogen atoms are generated using the proper HFIX command and isotropically refined using the riding model. The calculations were performed using the SHELX program package (Sheldrick 2008). The key crystallographic data and X-ray experiment details for compound 7 are presented in Supplementary data. The molecular plots were drawn using Mercury (Macrae et al. 2020). Crystallographic data from compound have been deposited with the Cambridge Crystallographic Data Center as Supplementary Material number CCDC-2210899 (7). Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK. E-mail:deposit@ccdc. cam.ac.uk.

Molecular docking

The crystal structure of the heterodimer EcR/USP of the moth *Heliothis virescens* was considered as a target protein, which was obtained from Protein Data Bank in the form of an heterodimer complex with PonA (1r1k) with a resolution of 2.9 Å (https://www.rcsb.org/). The computational experiment was carried out using the programs AutoDock 4.2 (Goodsell et al. 1990; Khairutdinov et al. 2014; Morris et al. 2009), AutoDock Vina (Gaillard 2018; Trott and Olson 2010), and GOLD Suite (Gaillard 2018; Trott and Olson 2010). Ligands for calculations were selected taking into account the most probable stereoisomerism. The structures of the natural ecdysteroid, 20-hydroxyecdysone, were positioned in the same active center as well as the native ligand of the complex EcR–Ponasteron A (taking into account the stereoisomerism of these biologically active substances).

The geometric parameters of ligands 7-10 were optimized by molecular mechanics by application of the MMFF94 force field using the MarvinSketch program, version 19.19 (www.chemaxon.com), and by the semiempirical PM6 method using the GAUSSIAN09 (Frisch M. J. 2009). The subsequent preparation of the structures of ligands 7-10 and 1r1k macromolecules was carried out using the AutoDockTools (Goodsell et al. 1990; Khairutdinov et al. 2014; Morris et al. 2009) and HERMES (www.ccdc.cam. ac.uk) software. Prior to calculations, all water molecules were removed from the protein. The files with the results were converted to Mol2 and PDBQT formats with addition of the lacking hydrogen atoms and partial atomic charges calculated by the Gasteiger method (Gaillard 2018; Goodsell et al. 1990; Khairutdinov et al. 2014; Morris et al. 2009; Trott and Olson 2010).

In AutoDockTools, a three-dimensional box (26 Å size) was generated, and the test ligands were placed into the box. The position of the reference HWG agonist was taken as the center of the box (www.rscb.org). The range for the ligandbinding cavity of the target protein in the GOLD Suite program was 12 Å; in this case, the position of ponasterone A was taken as the center.

The optimal positions of the ligands in the receptor active site were identified using the Lamarckian genetic algorithm (Fuhrmann et al. 2010; Kerzmann et al. 2008; Morris et al. 2009, 2008; Rurainski et al. 2009) and the Broyden–Fletcher–Goldfarb–Shanno (BFGS) local search algorithm. The docking procedure took into account the crystal structure of the target protein and implied full conformational flexibility of the ligands.

The estimation of the ligand affinity to the target protein in the GOLD program was based on calculation of the piecewise linear potential (CHEMPLP) scoring function. The AutoDock 4.2 and AutoDock Vina scoring functions acted as tools for parameterization of the energy contributions of receptor-ligand interactions according to AutoDock and AutoDock Vina, respectively.

When searching for potentially bioactive conformers using the evaluation function of the AutoDock 4.2 program, the angle of internal rotation around all single bonds in ligands was 30°, and the movement of ligand molecules as a whole in space was also carried out with an angle of 30° relative to the initial conformations. In the AutoDock Vina and GOLD Suite programs, molecular docking was performed with default parameters. Ligand conformations characterized by the minimum value of evaluation functions were taken as the optimal solution for the search of bioactive conformation in the AutoDock 4.2 and AutoDock Vina software. In the GOLD Suite program, the values of the obtained affinity parameters were directly dependent on the steric complementarity and energy correspondence between the studied compounds and the EcR protein. The assessment of the quality of ligand positioning in the active center of EcR was characterized by the RMSD value, which is the standard deviation of the position of the ligand after docking from its native position in the simulated complexes. Docking solutions were clustered based on RMSD = 2.0 Å. The RMSD value estimated by comparing the coordination of ponasterone A calculated by the molecular docking method with its native position in the active center of the receptor was in the range of 0.010-0.180 Å, which indicates the applicability of the evaluation functions of the AutoDock 4.2, AutoDock Vina, and GOLD programs to the modeling of agonists and antagonists of the receptor.

Procedure for lemon juice-catalyzed Povarov reaction

Freshly distilled cyclopenta-1,3-diene (3 mmol, 0.25 mL) and ethyl glyoxylate **3** (1 mmol, 0.1 mL) were added sequentially to a solution of amine **1** (1 mmol, 0.095 mL) or 5-aminoquinoline **11** (1 mmol, 144 mg) in hot lemon juice (5 mL). The mixture was stirred while heating to 70–80 °C for 0.25 h until the amine disappeared (TLC monitoring). The reaction mixture was extracted with CH_2Cl_2 (3×5 mL), the solvent was evaporated, the residue was purified by column chromatography (SiO₂, petroleum ether/ethyl acetate, 10:1) to isolate products **7** and **13a**.

rel-(3aR^{*},4S^{*},9bS^{*})-4-ethoxycarbonyl-8-fluoro-3a,4,5,9b-tetrahydro-3*H*-cyclopenta[*c*] quinoline 7

White solid, Yield: 76%, m.p. 72–74 °C (hexane). ¹H NMR (500 MHz, CDCl₃) δ 6.69–6.77 (m, 2H, H-7, H-9), 6.58–6.61 (m, 1H, H-6), 5.71 (m, 2H, H-1, H-2), 4.24–4.38 (m, 2H, OCH₂), 4.06 (m, 2H, H-4, H-9b), 3.34–3.36 (m, 1H, H-3a), 2.33–2.39 and 2.46–2.52 (m, 2H, H-3), 1.28–1.36 (m,

3H, CH₃). ¹³C NMR (125 MHz, CDCl₃) δ 171.74 (OCO), 156.58 (d, *J*=235 Hz, C-8), 140.09 (C-5a), 133.57 (C-1), 130.37 (C-2), 127.38 (d, *J*=6 Hz, C-9a), 116.49 (d, *J*=7 Hz, C-6), 114.71 (d, *J*=22 Hz, C-7), 113.13 (d, *J*=22 Hz, C-9), 61.25 (OCH₂), 56.81 (C-4), 46.71 (C-9b), 40.41 (C-3a), 32.67 (C-3), 14.26 (CH₃). HRMS (ESI-TOF) *m*/*z* [M⁺] Calculated for C₁₅H₁₆FNO₂: 261.1165; Found: 261.8601.

rel-(3aR*,4S*,11bS*)-4-ethoxycarbonyl-3a,4,5,11b-tetrahydro-3*H*-cyclopenta[c][1,7]phenanthroline 13a

Yellow solid, Yield: 7%, m.p. 114–116 °C (hexane). ¹H NMR (400 MHz, CDCl₃) δ 8.77–8.78 (m, 1H, H-8), 8.15–8.18 (m, 1H, H-6), 7.49–7.51 (m, 1H, H-10), 7.34–7.36 (m, 1H, H-11), 7.26–7.29 (m, 1H, H-7), 5.84 (s, 1H, H-1), 5.64 (s, 1H, H-2), 4.94 (s, 1H, NH), 4.24–4.29 and 4.33–4.37 (m, 2H, CH₂), 4.21 (m, 2H, H-4,11b), 3.39–3.42 (m, 1H, H-3a), 2.31–2.35 and 2.45–2.50 (m, 2H, H-3), 1.31–1.35 (m, 3H, CH₃). ¹³C NMR (100 MHz, CDCl₃) δ 171.94 (CO), 149.42 (C⁸), 147.54 (C^{5a}), 137.93 (C^{9a}), 133.88 (C¹), 130.62 (C¹¹), 130.09 (C²), 128.65 (C⁶), 120.74 (C^{11a}), 120.03 (C¹⁰), 119.66 (C⁷), 118.68 (C^{5b}), 61.44 (CH₂), 55.98 (C⁴), 46.63 (C^{11b}), 41.07 (C^{3a}), 32.21 (C³), 14.30 (CH₃). HRMS (ESI-TOF) *m/z* [M⁺] Calculated for C₁₈H₁₈N₂O₂: 294.1368; Found: 294.1365.

Ethyl-(E)-2-(quinoline-5-ylimino) acetate 13b

Yellow solid, Yield: 76%, amorphous. ¹HNMR (400 MHz, CDCl₃) δ 9.31–9.33 (m, 1H, H-2). 8.51 (d, 1H, *J*=8.4 Hz, H-4), 8.13 (m, 1H, H-10), 8.04–8.08 (m, 1H, H-8), 7.92–7.95 (m, 1H, H-7), 7.73–7.78 (m, 1H, H-3), 7.78 (d, 1H, *J*=7.6 Hz, H-6), 4.27–4.48 (m, 2H, H-12), 1.39 (t, *J*=7 Hz, 3H, H-13). HRMS (ESI-TOF) *m/z* [M+H]⁺ Calculated for C₁₃H₁₂N₂O₂; 228.0898; Found: 229.0985⁻

Diethyl-2,2'-(1,2-phenylenedi(nitrilo)diacetate 14

White solid, Yield: 70%, m.p. 172–174 °C (EtOH). ¹HNMR (400 MHz, MeOD) δ 8.21 (s, 2H, CH=N), 7.84 (d, 2H, J=8.0 Hz, CH-Ar), 7.57–7.61 (m, 2H, CH-Ar), 4.20–4.25 (m, 4H, OCH₂), 1.31 (t, 6H, J=6.8 Hz, CH₃). ¹³C NMR (100 MHz, MeOD₃) δ 169.06 (OC=O), 150.51 (C=N), 132.68 (C-Ar), 128.72 (C-Ar), 123.79 (CAr), 60.99 (OCH₂), 12.96 (CH₃). HRMS (ESI-TOF) m/z [M]⁺ Calculated for C₁₄H₁₆N₂O₄: 276.1110: Found: 276.1115.

General procedure of synthesis 4-aryl-3a,4,5,9b-tetrahydro-3*H*-cyclopenta[*c*]quinolones 8–10

Freshly distilled cyclopenta-1,3-diene (3 mmol, 0.25 mL) and appropriate aldehydes **4–6** (1 mmol) were added sequentially to a solution of amine **1** or **2** (1 mmol) in ionic liquid

(2 mL). The mixture was stirred while room temperature for 0.5 h until the amine disappeared (TLC monitoring). The reaction mixture was treated with water and extracted with CH_2Cl_2 (3×5 mL), the solvent was evaporated, the residue was purified by column chromatography (SiO₂, petroleum ether/ethyl acetate, 10:1) to isolate products **8–10**.

rel-(3aR^{*}, 4S^{*}, 9bS)-4-((4-trifluoromethyl) phenyl)-8-fluoro-3a,4,5,9b-tetrahydro-3*H*-cyclopenta[*c*] quinoline 8

Yellow solid, m.p. 74–76 °C (hexane). ¹HNMR (400 MHz, CDCl₃) δ 7.67 (d, J=8.0 Hz, 2H, H-3', H-5'), 7.59 (d, J=8.4 Hz, 2H, H-2', H-6'), 6.82 (d, J=2.8 Hz, 1H, H-9), 6.75 (d, J=8.4 Hz, 1H, H-7), 6.62 (d, J=8.8 Hz, 1H, H-6), 5.85 (br.s, w_{1/2}=8 Hz, 1H, H1), 5.71 (br.s, w_{1/2}=8 Hz, 1H, H-2), 4.68 (d, J=3.2 Hz, 1H, H-4), 4.13 (d, J=8.8 Hz, 1H, H-9b), 3.03 (d, J=8.8 Hz, 1H, H-3a), 1.82 and 2.62 (d, J=8.8, J=9.2 Hz, 2H, H-3). ¹³C NMR described 11b(Tolstikov et al. 2014b).

rel-(3aR^{*},4S^{*},9bS^{*})-4-((3-chlorophenyl)-8-fluoro-3a,4,5,9b -tetrahydro-3*H*-cyclopenta[*c*]quinoline 9

Yellow solid, m.p. 88–90 °C (hexane). ¹HNMR (400 MHz, CDCl₃) δ 7.52 (s, 1H, H-2'), 7.42 (s, 1H, H-4'), 7.36–7.38 (m, 2H, H-5',6'), 6.86 (d, *J*=2.4 Hz, 1H, H-9), 6.79 (d, *J*=8.4 Hz, 1H, H-7), 6.60–6.62 (m, 1H, H-6), 5.88 (m, 1H, H-1), 5.74 (m, 1H, H-2), 4.59 (d, *J*=3.2 Hz, 1H, H-4), 4.13 (d, *J*=8.8 Hz, 1H, H-9b), 3.03 (d, *J*=8.8 Hz, 1H, H-3a), 2.66 and 1.88 (d, *J*=8.8, *J*=9.2 Hz, 2H, H-3). ¹³C NMR described (Tolstikov et al. 2014b).

rel-(3aR*4S*9bS*)-4-(2-fluorophenyl)-3a,4,5,9b-tetrahydro-3*H*-cyclopenta[c]quinoline 10

Yellow solid, Yield: 79%, m.p.106-108 °C (hexane). ¹HNMR (400 MHz, CDCl₃) δ 7.66 (t, J = 7.2 Hz, 1H, H-6'), 7.28-7.32 (m, 1H, H-4'), 7.21-7.25 (m, 1H, H-5'), 7.08-7.13 (m, 2H, H-7, H-3'), 7.03-7.06 (m, 1H, H-6), 6.82 (t, J=7.2 Hz, 1H, H-8), 6.68 (d, J=7.6 Hz, 1H, H-9), 5.89 (br. s, $w_{1/2} = 8.0$ Hz, 1H, H-1), 5.69 (br. s, $w_{1/2} = 7.6$ Hz, 1H, H-2), 5.02 (br. s, $w_{1/2} = 10$ Hz, 1H, H-4), 4.18 (d, J = 8 Hz, 1H, H-9b), 3.63 (br. s, $w_{1/2} = 16$ Hz, 1H, NH), 3.20 (d, J=8 Hz, 1H, H-3a), 2.63–2.69 and 1.83–1.90 (m, 2H, H-3). ¹³CNMR (100 MHz, CDCl₃) δ 160.14 (d, J = 245 Hz, C-F), 145.53 (C-5a), 134.15 (C-1), 130.19 (C-2), 129.97 (d, J=12 Hz, C-1'), 129.04 (C-7), 128.46 (d, J=8 Hz, C-4'), 127.24 (d, J = 4 Hz, C-6'), 126.34 (C-6), 126.25 (C-9a), 124.22 (d, J=3 Hz, C-5'), 119.47 (C-8), 116.09 (C-9), 115.25 (d, J=21 Hz, C-3'), 51.04 (C-4), 46.19 (C-9b), 43.35 (C-3a), 31.76 (C-3). HRMS (ESI-TOF) *m/z* [M⁺] Calculated for C₁₈H₁₆FN: 265.1266; Found: 265.8655.

rel-(3aR*,4S*,11bS*)-4-[4-(trifluoromethyl) phenyl]-3a,4,5,11b-tetrahydro-3*H*-cyclopenta[c][1,7]phenanthroline 13c

Freshly distilled cyclopenta-1,3-diene (3 mmol, 0.25 mL) and aldehyde 4 (1 mmol, 0.14 mL) were added sequentially to a solution of amine 11 (1 mmol, 144 mg) in ionic liquid (2 mL). The mixture was stirred while room temperature for 0.5 h until the amine disappeared (TLC monitoring). The reaction mixture was treated with water and extracted with CH_2Cl_2 (3×5 mL), the solvent was evaporated, the residue was purified by column chromatography (SiO₂, petroleum ether/ethyl acetate, 10:1) to isolate product 13c. Yellow solid, Yield 81%, m.p. 120-122 °C (hexane). ¹HNMR $(400 \text{ MHz}, \text{CDCl}_3) \delta 8.84 \text{ (d}, J = 3.2 \text{ Hz}, 1\text{H}, \text{H-8}), 8.16 \text{ (d},$ J = 8.0 Hz, 1H, H-6), 7.69 (d, 4H, J = 8.8 Hz, H-2',3',5',6'), 7.59 (d, J=8.8 Hz, 1H, H-10), 7.46 (d, J=8.8 Hz, 1H, H-11), 7.35 (d, J = 8.0 Hz, 1H, H-7), 5.99 (m, 1H, H-1), 5.70 (m, 1H, H-2), 4.84 (d, J=2.4 Hz, 1H, H-4), 4.31 (d, 1H, H-4)J=8.4 Hz, H-11b), 3.14 (d, J=8.4 Hz, 1H, H-3a), 2.69 and 1.88 (d, 2H, J=8.4 Hz, H-3). HRMS (ESI-TOF) m/z [M⁺] Calculated for C₂₂H₁₇F₃N₂: 366.1343; Found: 366.7878, ¹³C NMR described (Tolstikov et al. 2014a).¹⁶

4,7-Bis-(4-(trifluoromethyl)phenyl)-3,3a,4,5,6,7,7a,8,10a,12 b-decahydrodicyclopenta[*c*,*i*]-1,10-phenantroline 15

Freshly distilled cyclopenta-1,3-diene (6 mmol, 0.5 mL) and aldehyde 4 (2 mmol, 0.28 mL) were added sequentially to a solution of amine 12 (1 mmol, 108 mg) in ionic liquid (2 mL). The mixture was stirred while room temperature for 0.5 h until the amine disappeared (TLC monitoring). The reaction mixture was treated with water and extracted with CH_2Cl_2 (3×5 mL), the solvent was evaporated, the residue was purified by column chromatography (SiO₂, petroleum ether/ethyl acetate, 10:1) to isolate product 15. Yield: 64%, yellow solid, m.p. 108–110 °C. ¹HNMR (500 MHz, CDCl₃) δ 7.61–7.66 м (8H, Ar), 6.73 and 6.67 (s, 2H, H-11,12), 5.92 and 5.89 (m, 2H, H-1,10), 5.66 (m, 2H, H-2,9), 4.73 and 4.59 (s, 2H, H-4,7), 4.16 and 4.14 (d, J=9 Hz, 2H, H-10a,12b), 3.11 and 2.98 (m, 2H, H-3a,7a), 2.59, 2.49, 1.97 and 1.80 (m, 4H, H-3,8). HRMS (ESI-TOF) m/z [M+H]⁺ Calculated for C₃₂H₂₆F₆N₂: 553.2073; Found: 553.3938, ¹³C NMR described (Savchenko et al. 2022).

Conclusion

Thus, this paper addresses eco-friendly approaches to the synthesis of new tetrahydroquinoline compounds **7–10**. The biological assays of the products carried out using the model

Musca domestica insect and analysis of molecular docking solutions make it possible to consider the prepared compounds as potential EcR-USP agonists.

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Declarations

Conflict of interest The authors declare no conflict of interest.

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