



# Production and Characterization of Metastable $ZrO_2$ - $Al_2O_3$ Coatings Obtained by APS + Quench

S. Dosta, I.G. Cano, J.R. Miguel, and J.M. Guilemany

(Submitted July 30, 2007; in revised form January 14, 2008)

**Producing nanostructured materials through metastable phases is an interesting novel route in the field of ceramic materials. Due to their small grain size and uniform structure, these nanostructured bulk materials exhibit very interesting properties. Metastable coatings can be produced starting from microstructured powders through atmospheric plasma spray technique, followed by a quenching route. The initial powders are melted during the spraying and deposited over a substrate that is quenched with liquid nitrogen feeders, producing metastable coatings. The thermal-sprayed coatings have been characterized using x-ray diffraction, scanning electron microscopy, field emission scanning electron microscopy, and energy dispersive spectroscopy. The properties of such coatings have been also studied obtaining promising results.**

**Keywords** APS coatings, ceramic oxide layers, manufacturing, nanostructured coatings, nanostructured materials, processing, properties, thermal and phase stability of coatings

## 1. Introduction

Recent research (Ref 1, 2) in the field of nanostructured ceramic materials has underscored the importance of using feedstock powders with metastable phases. Grain growth is reduced due to the immiscibility of the two phases in the solid state, and the obtained materials exhibit very interesting properties such as higher hardness and toughness (Ref 3).

Plasma-sprayed  $ZrO_2$  coatings have been used as thermal protection for aircraft engines parts such as combustor cans, nozzle and walls, or gas path transition pieces. Moreover, the hardness of  $ZrO_2$  coatings can be improved with the addition of  $Al_2O_3$ . In this study,  $ZrO_2$ - $Al_2O_3$  system has been selected to study how nanostructured coating can be obtained by atmospheric plasma spray (APS) technology following a metastability route.

To obtain nanostructured coatings, one way is to use nano-scaled agglomerated powders but, these feedstock powders are usually very expensive and it is difficult to preserve the nano-scale during the high-temperature process.

Another way consists of obtaining nanostructured coatings through metastable states. These metastable

states can be obtained by APS followed of a cooling process.

During a conventional APS process, the initial powder is melted for its residence in the hot zone of the plasma jet and then the molten particles are deposited over a substrate producing a coating. This results in a combination of phases where metastable and equilibrium phases coexist. Previous researches (Ref 4–6) described the use of thermal spray to obtain coatings which were formed by metastable phases due to the high-thermal gradient involved in the spraying process. Actually these “metastable” coatings are also composed of stable phases (equilibrium phases) while as a totally metastable state can be achieved with an ultrahigh cooling rate (Ref 7, 8). When a heat treatment of the totally metastable coatings is carried out, the existing phases transform to the equilibrium ones producing nanostructured free-standing coatings. It is due to the mutually suppression of grain growth as a direct consequence of the immiscibility of the phases in the solid state.

In this work,  $ZrO_2$ - $Al_2O_3$  coatings were produced by APS attaining a totally metastable state through a quench process. This process used liquid nitrogen supplied by feeders in continuous flow, achieving coatings with thickness of 250  $\mu m$  formed by only metastable phases.

The obtained coatings were characterized using x-ray diffraction (XRD), scanning electron microscopy (SEM), field emission scanning electron microscopy (FESEM), and energy dispersive spectroscopy (EDS). Hardness was analyzed by indentation method and the result showed improvements in relation with the conventional coatings.

## 2. Experimental Procedure

A commercially available  $ZrO_2$ -20 wt.%  $Al_2O_3$  powder has been used as feedstock in the present study

S. Dosta, I.G. Cano, J.R. Miguel, and J.M. Guilemany, Thermal Spray Centre (CPT), Department of Ciència dels Materials i Enginyeria Metal·lúrgica, Universitat de Barcelona, Barcelona, Spain. Contact e-mail: sdosta@ub.edu.

(Tosoh, Japan). This powder contains an addition of 3 mol%  $Y_2O_3$  to stabilize the tetragonal zirconia phase ( $t\text{-ZrO}_2$ ) and it is an agglomerated and sintered powder. The mean particle size of this powder is 57  $\mu\text{m}$ .

To obtain metastable phases, the  $ZrO_2\text{-Al}_2O_3$  powder has been sprayed using an F4 plasma torch and refrigerated layer by layer with two liquid nitrogen feeders (Air Products, S.A.). They were sprayed over a steel 50 mm  $\times$  20 mm  $\times$  5 mm flat substrate (UNS G41350). The substrates were previously degreased with acetone and grit blasted with corundum at 5.6 bar, 45° using a blasting distance of 250 mm. The grit-blasted substrates had a mean roughness ( $R_a$ ) of about 5  $\mu\text{m}$ , being always higher than 4  $\mu\text{m}$ .

The characterization of the samples included SEM-FESEM images and EDS phase analysis. More accurate phase analysis of the starting powders and coatings was performed by XRD.

Cross-sectional microhardness measurements were performed by means of Vickers indentation at 300 gf load, and the indentations were measured with an optical microscope to increase measurement accuracy.

### 3. Results and Discussion

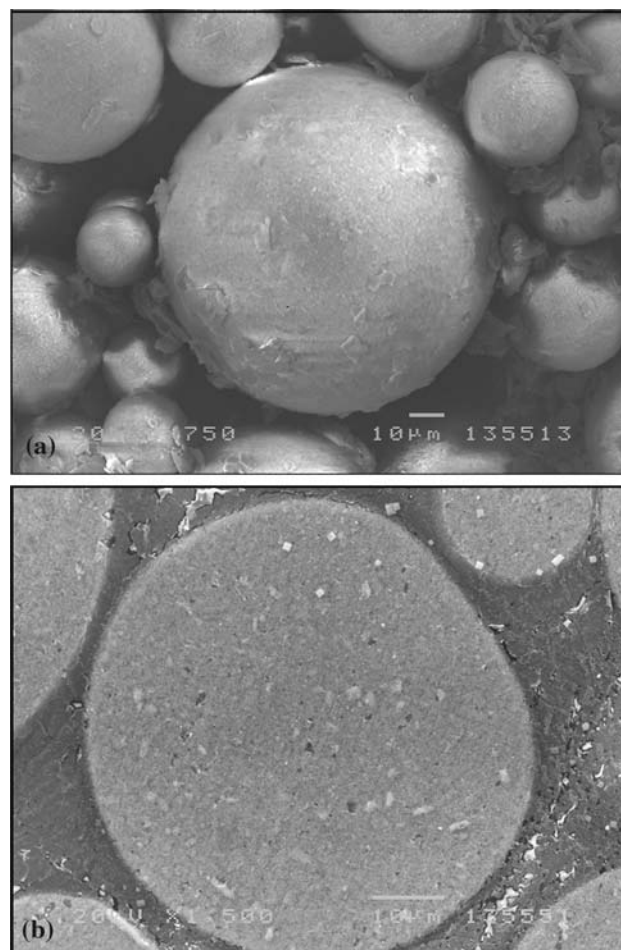
#### 3.1 Feedstock Powder

Figure 1a shows the free-surface micrographs of the initial powder. This powder exhibits a spherical morphology and fine particles of  $Al_2O_3$  are homogeneously distributed over the  $ZrO_2$  matrix as can be seen in the cross-sectional BSE-SEM image (Fig. 1b) where the brighter particles correspond to  $Al_2O_3$ .

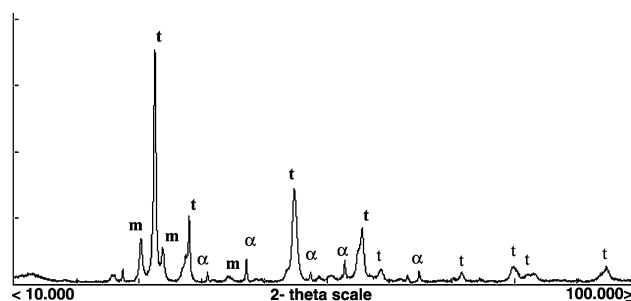
The XRD spectrum is shown in Fig. 2, where the main phase is  $t\text{-ZrO}_2$ . A small amount of monoclinic  $ZrO_2$ , formed due to the partially stabilization with yttria is also observed. The XRD pattern also shows peaks corresponding to  $\alpha\text{-Al}_2O_3$  phase, that is the equilibrium phase at room temperature.

#### 3.2 Coatings

SEM cross-sectional images of the as-sprayed  $ZrO_2\text{-Al}_2O_3$  coatings (using the APS+quenching route) are shown in Fig. 3. It can be observed a homogeneous structure with a thickness around 250  $\mu\text{m}$ . The porosity has been estimated in 15% being a typical value for a coating obtained by APS. Nondifferentiated phases are observed in Fig. 3 as can be corroborated by the XRD pattern (Fig. 3b), that shows no presence of  $Al_2O_3$ . Just a mixture of cubic and tetragonal  $ZrO_2$  phase have been identified. The main phase is the cubic phase, being a significant result because it is a nonequilibrium phase at room temperature. The fact that  $Al_2O_3$  phase is not detected in the XRD pattern could be explained by the formation of amorphous  $Al_2O_3$  or/and some dissolution in the  $ZrO_2$  matrix. The high-cooling rate used in the process can lead to the integration of aluminum in the cubic

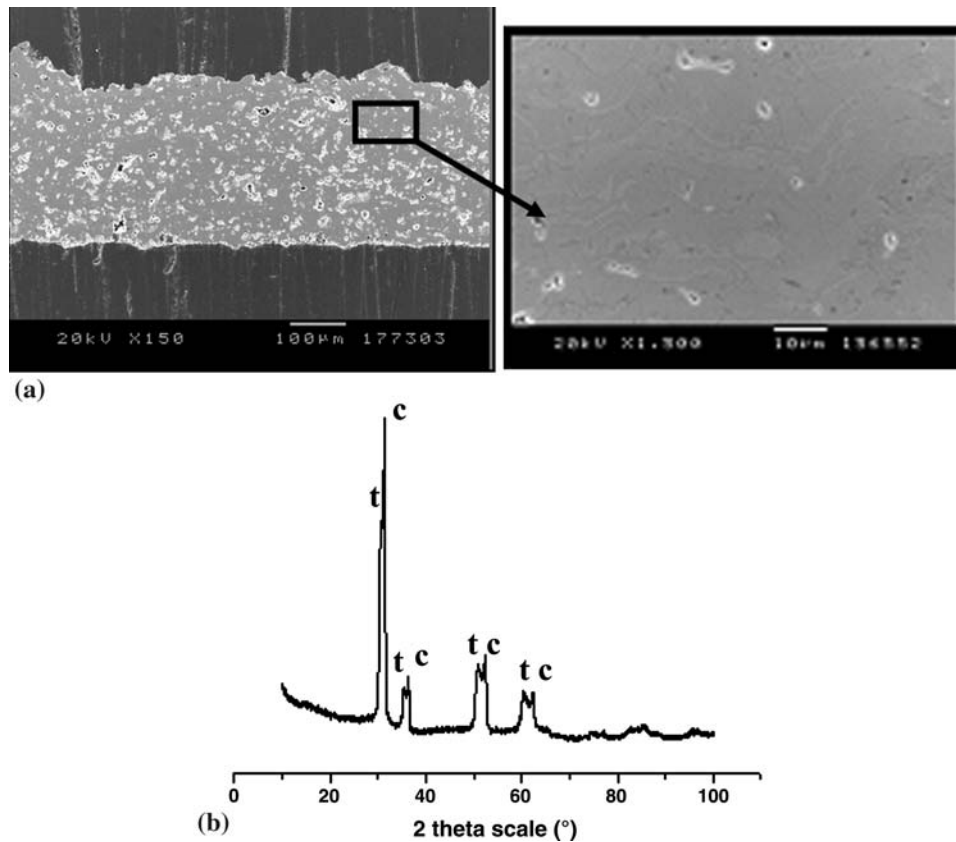


**Fig. 1** SEM (a) free-surface and (b) cross-sectional images of  $ZrO_2\text{-Al}_2O_3$  feedstock powder where the brighter particles correspond to  $Al_2O_3$



**Fig. 2** XRD pattern of  $ZrO_2\text{-}20\text{ wt.}\% \text{Al}_2O_3$  feedstock powder

zirconia cell. When the indexation of the peak corresponding to cubic phase is done, the lattice parameter ( $d \approx 4.93$ ) result is smaller than the standard pattern. This is coherent due to the smaller radius of  $Al^{3+}$  (0.068 nm) than  $Zr^{4+}$  (0.086 nm). Rietveld refinement is being carried out to determinate this mechanism.



**Fig. 3** As-sprayed  $\text{ZrO}_2$ -20 wt.%  $\text{Al}_2\text{O}_3$  coating (a) cross-sectional image and (b) XRD, where *t* means  $\text{ZrO}_2$ -tetragonal phase and *c* means  $\text{ZrO}_2$ -cubic phase of  $\text{ZrO}_2$ -20 wt.%  $\text{Al}_2\text{O}_3$  as-sprayed coating

The as-sprayed coatings were heat treated at 1200 and 1400 °C for 1 h and the resulting materials were characterized by XRD and SEM. After the thermal treatment carried out at 1200 °C, the XRD spectrum only showed the presence of  $\text{ZrO}_2$  (tetragonal) phase, which is predicted by the equilibrium phase diagram when the  $\text{ZrO}_2$  is doped with  $\text{Y}_2\text{O}_3$ . No presence of  $\text{Al}_2\text{O}_3$  was detected as Fig. 4 shows although  $\text{Al}_2\text{O}_3$  precipitates are observed. This can be attributed to be in the first stages of the precipitation phenomena because after the thermal treatment at 1400 °C,  $\alpha$ - $\text{Al}_2\text{O}_3$  equilibrium phase clearly appears (Fig. 5b).  $\text{Al}_2\text{O}_3$  grains have kept their nanometric size even at 1400 °C as can be seen in Fig. 6, where a TEM image shows the nanometric precipitates of  $\text{Al}_2\text{O}_3$  into a  $\text{ZrO}_2$  matrix.

### 3.3 Microhardness Values

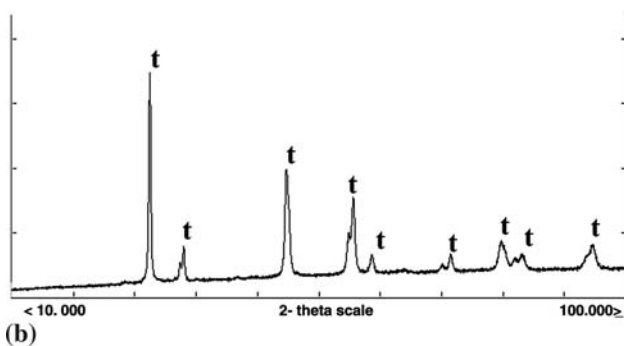
Representative Vickers microhardness values ( $\text{HVN}_{300\text{ gf}}$ ) obtained for the as-sprayed  $\text{ZrO}_2$ -20 wt.%  $\text{Al}_2\text{O}_3$  coating have been  $\text{HVN}_{300\text{ gf}} = 649 \pm 80$ .

Once heat treated at 1200 °C was carried out, the nucleation of  $\alpha$ - $\text{Al}_2\text{O}_3$  and the reduction of the porosity were achieved in the  $\text{ZrO}_2$ - $\text{Al}_2\text{O}_3$  coatings. Thus, the

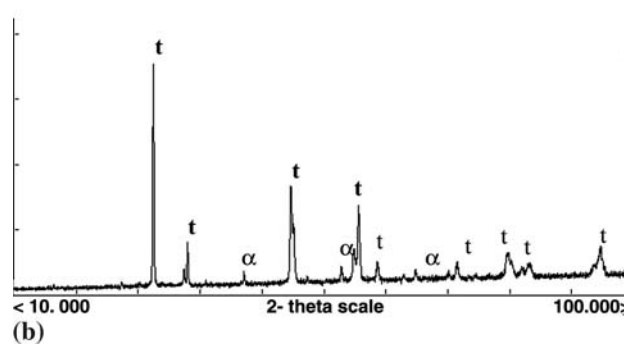
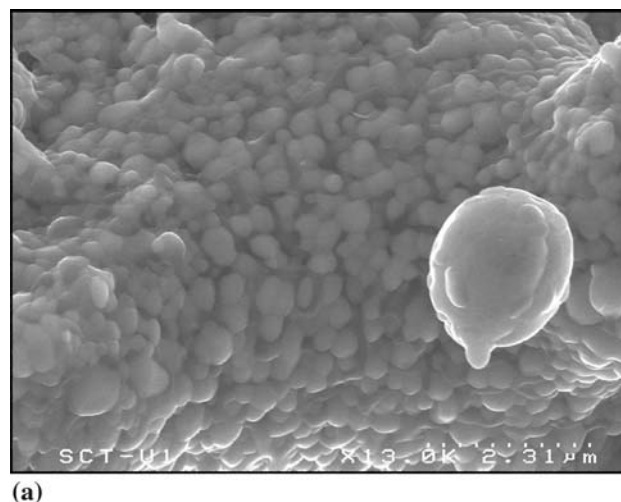
microhardness value increased being  $\text{HVN}_{300\text{ gf}} = 1063 \pm 86$  for the coating heat-treated at 1200 °C (Fig. 7).

Image analysis (Software Matrox Inspector v. 1.71) was used to quantify coatings porosity. Quoted values are an average of 10 areas at the same magnification for each coating. Porosity measured by image analysis was  $16.8\% \pm 0.3$  for the metastable coating, being  $15.5\% \pm 0.5$  and  $11.3\% \pm 0.4$  for the heat-treated coatings at 1200 and 1400 °C, respectively. The pores are randomly scattered throughout the cross section and has been reduced as the treatment temperature increased.

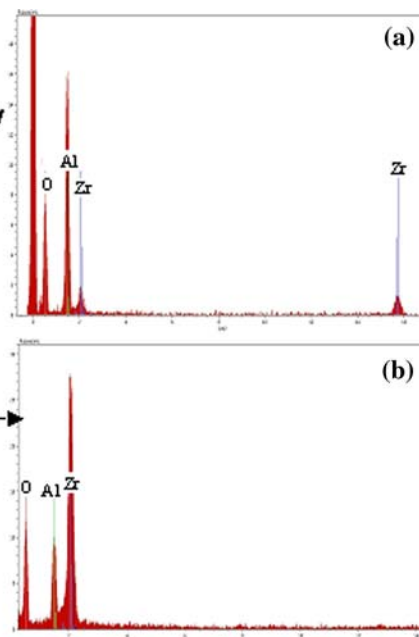
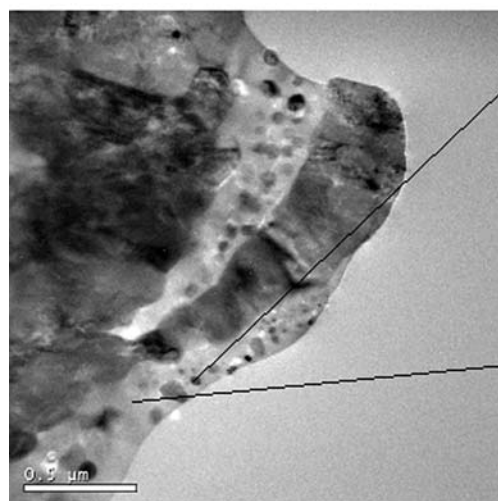
After the heat treatment at 1400 °C, a little grain growth is produced for the  $\text{Al}_2\text{O}_3$  phase and porosity is also reduced.  $\text{HVN}_{300\text{ gf}}$  obtained for the coating heat treated at 1400 °C were  $\text{HVN}_{300\text{ gf}} = 1219 \pm 36$ , which means that the microhardness value was doubled when compared to the as-sprayed coating. Microhardness has increased due to the reduction of porosity, but also for the full precipitation of  $\text{Al}_2\text{O}_3$ . In the XRD pattern of samples treated at 1200 °C, only  $\text{ZrO}_2$  is detected due to the noncomplete precipitation of  $\text{Al}_2\text{O}_3$ . When the total  $\text{Al}_2\text{O}_3$  has precipitated in form of nanograins, the state of maximum hardness is achieved due to the reinforcement mechanisms.



**Fig. 4** (a) Free-surface images and (b) XRD where *t* means  $\text{ZrO}_2$ -tetragonal phase of the  $\text{ZrO}_2$ -20 wt.%  $\text{Al}_2\text{O}_3$  coating treated at  $1200^\circ\text{C}$  for 1 h

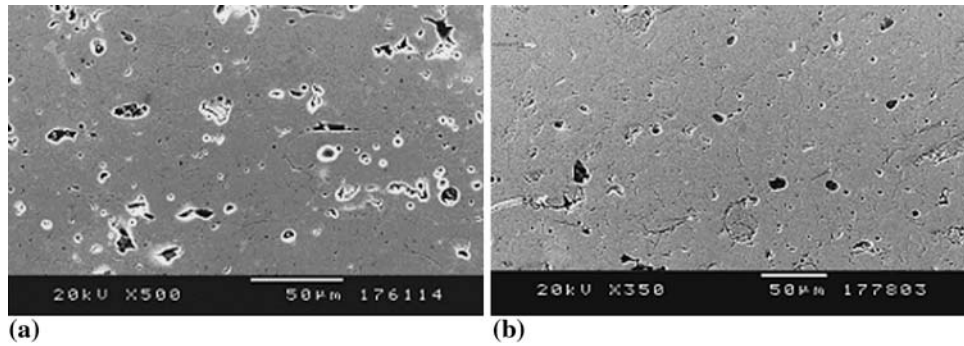


**Fig. 5** (a) Free-surface images and (b) XRD where *t* means  $\text{ZrO}_2$ -tetragonal phase of the  $\text{ZrO}_2$ -20 wt.%  $\text{Al}_2\text{O}_3$  coating treated at  $1400^\circ\text{C}$  for 1 h



**Fig. 6** TEM image of  $\text{ZrO}_2$ -20 wt.%  $\text{Al}_2\text{O}_3$  heat-treated at  $1400^\circ\text{C}$  with EDS analysis corresponding to the nanometric precipitates of  $\text{Al}_2\text{O}_3$  (a) into a  $\text{ZrO}_2$  matrix (b)





**Fig. 7** Cross-sectional images of  $ZrO_2$ -20 wt.%  $Al_2O_3$  heat-treated at (a) 1200 °C and (b) 1400 °C

#### 4. Conclusions

- $ZrO_2$ -20 wt.%  $Al_2O_3$  system has been tested to obtain metastable coatings by atmospheric plasma spraying technology followed of a quenching process.
- As-sprayed coatings exhibited microstructural homogeneity and were composed by a combination of tetragonal and cubic  $ZrO_2$  phases.
- These metastable structures evolved to equilibrium phases after the heat-treatments. Final structures were formed by nanometric grains.
- After the heat-treatments, microhardness values were improved (80%) in relation to values of conventional coatings. This improvement was attributed to the reduction of porosity and the precipitation of  $Al_2O_3$

#### Acknowledgments

The authors wish to thank the University of Barcelona for the financial support for this research, to the UE for the project NAMAMET STRP-001470, to the Generalitat de Catalunya for the project 2005 SGR-00310 and to the

Ministerio de Educación y Ciencia for the project MAT2007-65179.

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