TECHNICAL ARTICLE





Fabrication and Properties of Ni₃Si-TiC Composites by In Situ Reaction Sintering

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Ni₃Si-based composites with different TiC contents are fabricated by in situ reaction sintering method. The phase composition, microstructure, microhardness, and tribological properties of the composites are studied. The results show that the addition of TiC particles prevents the formation of γ -Ni₃₁Si₁₂ phase, and the composites mainly consist of β_1 -Ni₃Si and TiC phases. The composites have high hardness and relative density, and the values are above 580 HV and 87%, respectively. The friction coefficients are as low as 0.22. The wear rates of the composites are in the order of 10^{-6} mm³ m⁻¹, which are two orders of magnitude lower than that of Ni₃Si alloy. The primary wear mechanism of Ni₃Si alloy is oxidation wear and slight abrasive wear, and the composites with TiC addition are oxidation wear and fatigue wear. The incorporation of TiC particles improves the tribological performance of Ni₃Si alloy significantly. The composite with 20 wt.% TiC has the optimal tribological property.

Keywords metallic matrix composites, sintering, tribology, wear

1. Introduction

Intermetallic alloys have attractive properties including good high-temperature strength, relatively low density, excellent high-temperature oxidation, and corrosion resistance and therefore exhibit a great potential application in harsh environments. The major obstacles to using the alloys are their low ductility at ambient temperature while alloying and microstructural control can significantly improve the ambient-temperature ductility (Ref 1, 2). For these reasons, intermetallic matrix composites (IMCs) are considered to be applied in aeroengine and automotive components, nuclear and power generation equipment, and extreme environments. For example, titanium aluminides-based composites have wide application prospects in the engine component, such as compressor, turbine, and combustor (Ref 3). The relevant studies also show that IMCs have excellent wear, high-temperature oxidation resistance, and corrosion resistance (Ref 4-7). Meanwhile, metallic silicides present better properties than metallic aluminides, and thus Ni₃Si-based composites can be regarded as potential highperformance structural materials.

The typical preparation methods of Ni_3Si materials are arc melting (Ref 8-10), laser cladding (Ref 11-13), and self-

propagating high-temperature synthesis (Ref 14-16). But both high preparation temperature and fast cooling rate lead to the formation of by-products and aggregation of the reinforcements, which occur easily in the composites during the smelting process. Although the preparation temperature of self-propagating high-temperature synthesis is very low, the composition system is very limited. Reaction sintering is a method that can solve these problems effectively during the preparation of the composites. In the meantime, some studies have shown that reaction sintering can be used for the preparation of Ni₃Si-based alloys. Van Dyck et al. produced Ni₃Si-based alloys by reactive powder metallurgy combined with hot isostatic pressing and considered the reaction mechanism and densification process of Ni₃Si alloys (Ref 17). Miura et al. prepared Ni₃(Si,Ti) intermetallic alloy by the spark plasma sintering (SPS) method using elemental powders (Ref 18). The results showed that the sintered alloy had a high relative density in the range of 96-99%, and the high-temperature hardness and yield stress of the sintered alloys were higher than those of the cast alloys. Niu et al. fabricated Ni₃Si alloy from elemental powders by vacuum hot-press sintering method and studied the tribological properties under different conditions (Ref 19-22). These results demonstrated that the alloy possessed lower friction coefficient and wear rate in low vacuum condition as compared to high vacuum and air conditions; the tribological properties of Ni₃Si alloys with Ti addition are better than that of Ti6Al4V alloy under seawater condition; Ni₃Si alloy with Ti addition showed better wear resistance at high temperatures as compared to pure Ni₃Si alloy, and the wear rates were in the magnitude of 10^{-5} mm³/N m. So, it is possible to prepare Ni₃Sibased composites, and the microstructure and properties of the composites should be investigated.

It is well noted that TiC possesses many desirable properties, such as high hardness, low density, high modulus, and high corrosion and wear resistance. Hence, TiC is a suitable candidate for being used as a reinforcement in composites. In addition, the wettability between TiC and Ni is relatively good. In this study, TiC-reinforced Ni₃Si-based composites

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were fabricated by in situ reaction sintering, which had not been reported so far. The formation mechanism of Ni₃Si-based composite was discussed, and the influence of TiC content on microstructures and mechanical properties of Ni₃Si-TiC composites was also studied. The tribological properties of the composites were investigated in detail. The effects of load and TiC content on the wear behaviors were discussed. This research provided a theoretical foundation for the application of the composites in harsh environments.

2. Materials and Methods

Ni₃Si-based composites were prepared by vacuum hotpressed sintering method. The raw materials were commercial powders of nickel (99.8% purity, 325 mesh), silicon (99.5% purity, 300 mesh), titanium (99.5% purity, 325 mesh), boron (98.5% purity, 200-300 mesh), and TiC (99.95% purity, about 80 nm) (in mass percent). The compositions of the composites are listed in Table 1. The addition of Ti is positive to improve the toughness of Ni₃Si alloy. The samples doping with 50 wt. ppm boron is effective to suppress the intergranular fracture due to environmental embrittlement at room temperature (Ref 23). These powders were mixed evenly for 8 h in a planetary ballmill machine (Pulverisette 6, Germany). The mixed powders were placed in graphite molds and were then pressed into cylindrical compacts with dimensions of ϕ 50 mm \times 10 mm using a vacuum hot-press machine (VHP200/20-2000, China). The sintering conditions were in the temperature range of 1100 °C-1200 °C and at a pressure of 30 MPa for 40 min. The sintered samples were homogenized at 800 °C for 4 h.

The phase compositions were identified using a Shimadzu XRD 6000 diffractometer (Japan) equipped with Cu-K_{α} radiation at 40 kV and 30 mA. The 2 θ scan range was from 20 to 80° with a step size of 5°/min. The microstructures and surface morphologies of the composites were studied using a scanning electron microscope (SEM, JEOL JSM-6480, Japan) equipped with an energy-dispersive spectrometer (EDS, Oxford INCA X-Act, UK). The samples were etched for 15 s in the solution of HF and H₂O₂ before the microstructural analysis. The densities of the composites were calculated by Archimedes' principle using deionized water as the immersion medium at room temperature. The relative densities were calculated using the relation:

$$D = \left(\frac{d_m}{d_{th}}\right) \times 100\% \tag{Eq 1}$$

where D is the relative density, d_m is the measured densities, and d_{th} is the theoretical density. The theoretical densities are calculated from the XRD measured lattice parameters. The

 Table 1
 Compositions and sintering temperatures of the composites used in this study

No	Composition	Sintering temperature
1	Ni ₇₅ Si ₂₀ Ti ₅ + 0 wt.% TiC	1100 °C
2	$Ni_{75}Si_{20}Ti_5 + 10$ wt.% TiC	1140 °C
3	$Ni_{75}Si_{20}Ti_5 + 20$ wt.% TiC	1170 °C
4	$Ni_{75}Si_{20}Ti_5 + 30$ wt.% TiC	1180 °C

microhardnesses of the samples were measured using a semiautomatic microhardness tester (HXS-1000 TAC, China) at a load of 0.5 kg and a dwell time of 10 s. The tribological tests were carried out by a ball-on-disk tribometer (HT-1000, China). The test samples were machined to the size of 20 mm diameter and 3 mm thickness. The counterpart ball was the commercial Si_3N_4 ceramic ball with a diameter of 5 mm (hardness 15 GPa). The applied load was in the range of $5 \sim 20$ N, the sliding speed was 0.10 m/s, and the sliding time was 30 min. The volumetric wear rate was calculated using the equation:

$$W = \frac{AL}{S} \tag{Eq 2}$$

where *W* is the wear rate (mm³ m⁻¹), *A* is the cross-section area of the worn track (mm²), *L* is the perimeter of the circular track (mm), and *S* is the total sliding distance (m). The cross sections of the worn tracks were measured using a laser confocal microscope (Olympus, LEXT-OLS4000, Japan). The chemical states of Ni, Si, Ti, and C elements on the worn surface of the composites were examined by an X-ray photoelectron spectroscope (XPS, Thermo Scientific Nexsa, USA). The XPS utilized monochromatic Al-K_{α} radiation at 1486.6 eV.

3. Results

3.1 Microstructure and Mechanical Properties of the Composites

The XRD patterns of the composites with different TiC content are presented in Fig. 1. It can be seen that the sample without TiC addition is composed of β_1 -Ni₃Si and a small amount of γ -Ni₃₁Si₁₂ phase. With the addition of TiC particles, the main phase compositions of the composites are β_1 -Ni₃Si and TiC phase. It should be noted that the peak intensities of γ -Ni₃₁Si₁₂ phase of the composites decrease significantly compared to Ni₃Si alloy, and the intensities are similar in all the composites. According to the phase diagram of the Ni-Si system, the liquid phase can appear at the temperature and liquid phase are conducive to the reactions among the elements;



Fig. 1 XRD patterns of Ni₃Si-TiC composites

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especially, the element of Ti can cause γ -Ni₃₁Si₁₂ phase to transform into Ni₃Si phase (Ref 19). In addition, TiC particles play a role of isolation, which separates the metallic powders in a small area. As the reaction between nickel and silicon occurs, the formed Ni₃₁Si₁₂ phase can further react with the nearby nickel and titanium elements to form the Ni₃Si phase. Therefore, there is almost no γ -Ni₃₁Si₁₂ phase in the composites with the addition of TiC. Compared to the standard spectrum of Ni₃Si, the diffraction peaks of Ni₃Si phase shift to lower angles in all the samples. It is worth noting that x-ray diffraction cannot distinguish between α -Ni and β_1 -Ni₃Si, because both phases have similar structures and lattice parameters.

The microstructures of Ni₃Si-TiC composites are shown in Fig. 2. It is clear that no apparent defect is observed in Ni₃Si alloy, and the microstructure of the alloy is relatively compact, as shown in Fig. 2(a). The microstructure of the alloy consists of two regions: The dark phase is the silicon-rich region, and the composition is 68.51Ni-31.49Si (at.%); the bright phase is the nickel-rich region, and the composition is 81.41Ni-14.69Si-3.90Ti (at.%). Combined with XRD analysis, the dark region is the γ -Ni₃₁Si₁₂ phase which contains almost no titanium. The bright region is a eutectic structure, and the compositions are 76.97Ni-19.03Si-4.00Ti and 86.55Ni-10.98Si-2.47Ti, respectively. According to the ternary phase diagram of Ni, Si, and Ti (Ref 24), the eutectic structure is composed of β_1 -Ni₃Si and α -Ni phases. When 10% TiC is added, the TiC phase is aggregation in the composite (shown in Fig. 2b). The structures of the matrix alloys are more homogeneous as the content of TiC increases (shown in Fig. 2c and d). This is because the higher sintering temperature and the aggregation of TiC promote the reaction among the elements. It should be noted that there are some micropores in the composite with high TiC addition compared to Ni₃Si alloy. But the micropores are mainly concentrated in the TiC phase, and this indicates that the high content of TiC makes the composite hard to be densified.

The relative densities of the composites with different TiC content are shown in Fig. 3. The sample without TiC addition exhibits the highest relative density, and the value is above 98%. In general, the high sintering temperature and high heat release are beneficial to the compaction and densification of the materials during the hot-pressing sintering process. The high relative density of this sample can attribute to the high sintering temperature and high reaction heat of Ni with Si and Ti. With the addition of TiC, Ni₃Si alloy content decreases in the composite, which leads to a decrease in the reaction heat. The lower content of the matrix alloy, the lower heat release. In order to achieve the high relative density of the sintered material, the sintering temperature must be higher. Therefore, the composites with higher content of TiC have higher sintering temperature. The two opposite effects on sintering process result in a maximum value of the relative densities of the composites, and so the relative density of the composite containing 20% TiC is a little higher than those of the composites containing 10 and 30% TiC.

The microhardnesses of the composites are shown in Fig. 4. The results show that the hardness of the composite with 10 wt.% TiC addition is close to that of Ni₃Si alloy. This is because the composite shows poor densification and a lower content of y phase, whereas Ni₃Si alloy exhibits higher densification and high content of γ phase. The hardness of the composites with 20 wt.% TiC addition increases significantly with the further increase in TiC content, which is attributed to the high content of TiC and high relative density. It is worth noting that although the composite with 30% TiC addition presents some micropores (shown in Fig. 2d), it still has the highest hardness. The high sintering temperature makes the reaction of Ni with Si and Ti more completely, and the in situ matrix alloy is compacted during the hot-pressed process. However, there are some defects in the aggregation area of TiC phase that is hard to fully densify. Consequently,



Fig. 2 SEM micrographs of Ni3Si-TiC composites: (a) 0 wt.% TiC, (b) 10 wt.% TiC, (c) 20 wt.% TiC, (d) 30 wt.% TiC



Fig. 3 Relative densities of $Ni_3 Si\mathcal{Si-TiC}$ composites with different TiC content



Fig. 4 Effect of TiC content on the hardness of the composites

the compact matrix alloy provides good bearing capacity, and the composite has a higher hardness.

3.2 Tribological Properties of the Composites

Figure 5 shows the average friction coefficients of the composites with different TiC content. It is observed that the friction coefficients of the composites are all below 0.22, and the composites show good anti-friction performance. The previous study indicates that Ni₃Si materials exhibit high coefficients of friction (about 0.5) (Ref 14). It has also been reported that the oxidation resistance of Ni₃Si is reduced by the addition of Ti (Ref 25, 26). Accordingly, the low friction coefficient may be due to surface oxidation, and the oxide layer prevents direct contact with the surface asperities (Ref 27). The friction coefficient of Ni₃Si alloy decreases with increasing load. The high relative density of Ni₃Si alloy effectively prevents the oxidation gradually increases with the increase of load. Furthermore, the oxide layer reduces the friction



Fig. 5 Variations of the friction coefficients of the composites with the applied load



Fig. 6 Variations of the wear rates of the composites with the applied load

coefficient as the load increases. As compared to Ni_3Si alloy and the composite with 10% TiC, the composite with 20% TiC has a good relative density and low content of the Ni-Si phase, and thus, it presents a higher friction coefficient at the load of 5 N. At high loads, the defects cause severe oxidation, which generates the similar surface contact between the tribopairs. As a result, the friction coefficient of all the composite with TiC addition remains almost constant at high loads. In addition, the friction coefficient of the composites decreases with increasing TiC content at high loads, and this is because of the hardness.

The variations of wear rates of the composites with load at a sliding speed of 0.10 m/s are shown in Fig. 6. All the wear rates increase with increasing load. The wear resistance of the composite is higher than that of Ni₃Si alloy. The sample with 10% TiC has higher wear resistance and lower hardness compared to the alloy, whose wear rates are all in the order of 1×10^{-4} mm³ m⁻¹. The reason is that the high hardness of the TiC phase plays a role in the wear resistance. While the wear

rate of Ni₃Si alloy is the highest among the samples, it is still an order of magnitude lower than that in the previous study (Ref 19). Remarkably, the wear rates of the composites with 20% and 30% TiC are much lower, and all in the order of 1×10^{-6} mm³ m⁻¹. It means that the high content of TiC significantly improves the wear resistance of the composites, which is two magnitudes higher than that of Ni₃Si alloy. The wear rates of the samples decrease roughly with increasing TiC content, and the composite with 20% TiC exhibits the optimal wear resistance.

Figure 7 shows the worn surfaces of the composites without and with 10% TiC at different loads. It can be seen that some fine debris particles and shallow furrows disperse on the worn surface of Ni₃Si alloy at the load of 10 N, as shown in Fig. 7 (a). The oxide particles are squeezed and then compacted to form an oxide layer on the worn surface when the applied load is 20 N (shown in Fig. 7b). The formation of the oxide layer causes the lowest friction coefficient to the alloy at the load. It should point out that the oxide layer is in the process of continuous spallation and reconstruction during the sliding friction. The wear mechanism of the alloy is mainly oxidation wear and slight abrasive wear. The morphologies of the worn surface of the composite with 10% TiC are markedly different from that of Ni₃Si alloy, as shown in Fig. 7 (c) and (d). A gray oxide layer already forms at low load, but the local area appears to peel off in flakes. The oxide layer, of which composition is 15.19Ni-14.15Si-2.03Ti-60.94O-7.69C, contains high content of silicon. The friction coefficients of the composite fluctuate slightly around 0.14 in the test range. The formation of the oxide layer and the exposure of the fresh surface of the composite cause the similar surface contact between the tribopairs and make the friction coefficients almost constant as the load increases. Moreover, there are some pits on the worn surface, and this indicates that the wear mechanisms of the composite are oxidation wear and fatigue wear under both loads.

Figure 8 shows the SEM images of worn surfaces of the composites with 20% and 30% TiC at different loads. The morphologies of both composites are similar to that of the composite with 10% TiC. The oxide layers and some spalling pits are also observed on the worn surface of both samples. The wear mechanisms of both composites are oxidation wear and fatigue wear. Surprisingly, the oxide layers of the two composites have a higher silicon content compared to the composite with 10% TiC. The average compositions of both oxide layers are 8.80Ni-13.99Si-3.06Ti-66.73O-7.82C (20% TiC) and 5.63Ni-14.41Si-4.13Ti-68.35O-7.48C (30% TiC), respectively. However, the silicon contents of the two composites decrease with increasing TiC content, which are contradictory to the above results. A possible reason is that silicon transfers from the Si₃N₄ ball to the oxide layer.

Figure 9 shows the SEM images of worn surfaces of the counterpart ball sliding against different samples. For the Ni₃Si/Si₃N₄ tribopair, it can be seen that there are some shallow grooves and wear debris on the worn surface of Si₃N₄ ball (shown in Fig. 9a). The main wear mechanism is slightly abrasive wear. But when the counterpart ball slides against the composite with 20% TiC addition, the worn surface of the ball appears some spalling (shown in Fig. 9b). The main wear mechanism of the ball is fatigue wear. Consequently, the wear mechanisms of the counterpart balls are consistent with the results of the above analysis.

4. Discussion

Nickel-silicon is a low exothermic system, and the adiabatic temperature of Ni₃Si is 2250 K (Ref 16), which is higher than the melting temperature of the alloy. Some studies have shown that the reaction temperature can be as low as 790 °C (Ref 17). During the sintering process, Ni reacts with Si to form the Ni₂Si



Fig. 7 SEM morphologies of worn surface at different loads: (a) and (b) 0% TiC; (c) and (d) 10% TiC



Fig. 8 SEM morphologies of worn surface at different loads: (a) and (b) 20% TiC; (c) and (d) 30% TiC



Fig. 9 SEM morphologies of worn surfaces of the counterpart balls sliding against different samples at 20 N: (a) 0% TiC, (b) 20% TiC

phase, and the high exothermic heat of the reaction causes local melting in the alloy (Ref 17). The fusion occurs easily in the liquid region, and the liquid Ni-Si phase further reacts with the surrounding Ni to form the γ phase. Meanwhile, the Ti particles migrate and aggregate gradually in the local area. With the formation of the γ phase, the liquid phase is consumed and disappears, so the region of the γ phase is almost free of titanium, as shown in Fig. 2(a). As the sintering temperature rises, the reactions in the Ti-rich region begin to take place near the eutectic points of Ni-Si-Ti (Ref 24). Finally, the eutectic structure of β_1 and α phases forms. The phase composition of Ni₃Si alloy with Ti addition is α -Ni, β_1 -Ni₃Si, and a small amount of γ -Ni₃₁Si₁₂. With the addition TiC particles, the particles do not react with the elemental powders. The particles become gradually agglomeration as the reaction among the elements. The agglomeration of TiC particles separates the blended elemental powders into a small area and prevents the elemental particles from migration and aggregation over an extensive range. Then the Ni and Ti particles can still react with the γ phase and promote the transformation of the γ phase into

the β_1 phase during the sintering process (Ref 28). The main phase compositions of the composites with TiC addition are β_1 -Ni₃Si and TiC phases. Therefore, there is almost no Ti-rich area in the matrix alloys, of which compositions are more homogeneous compared to Ni₃Si alloy. It is worth noting that the diffraction peaks of Ni₃Si phase shift to lower angles in all the samples, and this is due to the partial substitution of Si by Ti in Ni₃Si unit cell (Ref 29). The more the content of solid solution titanium is, the bigger the peak of Ni₃Si phase deviates.

From the above analysis, Ni₃Si-TiC composites are successfully prepared by in situ reaction sintering, and the compositions of the matrix alloys are more uniform than that of Ni₃Si alloy. Besides, the lower wear rates indicate the composites have good adhesion between the matrix and TiC phase. Although the spalling pits are observed (shown in Fig. 8), the spallation only occurs between TiC particles rather than between the matrix and TiC particles. Single TiC particle is firmly embedded in the matrix (shown in Fig. 8d), and this aggravates the wear of the Si₃N₄ ball. The wear debris of Si₃N₄

is compacted on the worn surface, and thus the silicon content in the oxide layer increases significantly.

To verify this hypothesis, an XPS analysis was performed on the worn surface. Figure 10 shows the XPS spectra of Ni 2p and Ti 2p on the worn surface. As can be seen from the figure, the chemical states of Ni are similar in the samples with different TiC addition, as shown in Fig. 10(a). The peaks appeared at 855.9 eV, and 873.5 eV are ascribed to the Ni 2p3/2 and Ni 2p1/2 peaks of oxidized nickel (Ref 30, 31). The distinct peak around 861-862 eV can be ascribed to the shake-up satellite peak of NiO (Ref 32). A weak peak located at 853.1 eV (curve-fitted value) is attributed to nickel silicide (Ref 33, 34). The peaks at 852.5 eV and 869.6 eV are assigned to metallic Ni (Ref 35, 36). There is a shake-up satellite peak of metallic Ni at about 858.4 eV (Ref 37, 38). The results indicate that nickel element exists in the forms of Ni (0) species and NiO species on both worn surfaces. However, the chemical states of Ti are different in the samples with different TiC addition, as shown in Fig. 10(b). For the sample without TiC addition, the peak at 454.1 eV is detected on the worn surface and assigned to metallic titanium (Ref 39). The fitting peak at 454.5 eV corresponds to the binding energies of TiNi (Ref 40, 41), and the peak located around 458.9 eV can be attributed to siliconcoated TiO₂ (Ref 42). For the sample with 20% TiC addition, the peaks at 454.4 eV and 460.4 eV are ascribed to the Ti 2p3/2 and Ti 2p1/2 peaks of TiC (Ref 43). The peak at 455.3 eV corresponds to TiO species (Ref 44), and this is probably due to the partial oxidation during the friction process. In addition, the peaks at 458.4 eV and 464.1 eV are attributed to Ti 2p3/2 and Ti 2p1/2 of TiO₂ in both samples (Ref 45, 46). From the above analysis, it is concluded that it is free of metallic Ti on the worn surface of the composites with 20% TiC. This means that Ti reacts with Ni-Si thoroughly and is not aggregation in the matrix. Therefore, the composition of the matrix alloy is more uniform than that of Ni₃Si alloy.

Figure 11 shows the XPS spectra of Si 2p on the worn surface. The peak at 99 eV is related to Si-Si bonds in both samples. The very low content of pure silicon in the composite with 20% TiC indicates that the Si element has a good reaction with the Ni element. Since the sample without TiC contains almost no nitrogen, the peaks at 101 eV, 101.8 eV, 102.4 eV, and 103.5 eV are all different chemical species of silicon oxides

on the worn surface (Ref 47-49), and they may be related to SiO_x, Si-O-Ti, Si-O-Si, and SiO₂, respectively. The peak at 103.1 eV also indicates the presence of SiO₂ in the sample with 20% TiC (Ref 50). The difference is that the sample with TiC addition contains nitrogen, and thus, the peaks at 101.6-102.3 eV are attributed to Si 2p of Si₃N₄ (Ref 51). It can conclude that the Si₃N₄ species on the worn surface comes from the Si₃N₄ balls. Although the hardness of Si₃N₄ is much higher than those of the composites, TiC possesses higher hardness than Si₃N₄. Parts of TiC particles are firmly embedded in the matrix alloy and cause the wear of the Si₃N₄ ball instead of TiC particles being pulled out. Therefore, the composites with the addition of TiC exhibit excellent wear resistance, and the wear rates of the composites are about two orders of magnitude lower than that of Ni₃Si alloy. Moreover, the debris forms an oxide layer that contains many brittle Si₃N₄ particles. The brittle oxide layer is easy to crack and peel off even at low loads.



Fig. 11 XPS spectra of Si2p on the worn surfaces of the samples without and with 20% TiC addition at 20 N $\,$



Fig. 10 XPS spectra of Ni2p (a) and Ti2p (b) on the worn surfaces of the samples without and with 20% TiC addition at 20 N

5. Conclusions

Ni₃Si-TiC composites were fabricated by in situ reaction sintering. The phase composition, microstructure, and mechanical and tribological properties of the composites were investigated in this work. The main conclusions are summarized as follows:

- (1) Ni₃Si-TiC composites are successfully fabricated by in situ reaction sintering method. The main phase composition of Ni₃Si alloy is α -Ni, β_1 -Ni₃Si, and a small amount of γ -Ni₃₁Si₁₂, but the composites with the addition of TiC are β_1 -Ni₃Si and TiC. The reason for this is that the TiC phase prevents the elemental particles from migration and aggregation over an extensive range, and the Ti and Ni particles can thoroughly react with the γ phase to form β_1 phase in the matrix alloy.
- (2) The prepared composites have high hardness and relative density. The microhardnesses are all above 580 HV and almost increase with the increase of TiC content; the relative densities are still above 92% when the content of TiC is up to 20%. A small number of defects are mainly located in the aggregation area of TiC rather than at the interface between the matrix and TiC particles.
- (3) The friction coefficients of the composites are below 0.22. At high loads, the friction coefficients of the composites with TiC addition are almost constant under different loads, and the value decreases with increasing TiC content. The composites exhibit excellent wear resistance. The wear rates of the composites with high content TiC are two orders of magnitude lower than that of Ni₃Si alloy. The composite with 20% TiC addition shows the optimal wear resistance in the test range, and the wear rate is in the order of 10⁻⁶ mm³ m⁻¹.
- (4) The Si content on the worn surface increases with the increasing TiC content in the composites. This is because the high hardness of TiC causes the wear of Si₃N₄ ball and significantly improves the tribological properties of Ni₃Si-based composites. The primary wear mechanism of Ni₃Si alloy is oxidation wear and slight abrasive wear, and the composites with TiC addition are oxidation wear and fatigue wear.

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