ORIGINAL RESEARCH ARTICLE

An Examination of the Tin Pest Phenomenon Over a 10‑Year Period

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Abstract

Primarily in response to the implementation of the Restriction of Hazardous Substances (RoHS) European Directive, tinrich solder alloys and pure tin component surface fnishes have been introduced to printed circuit assembly processes. "Tin pest," a metallurgical phenomenon that can lead to mechanical and electrical degradation of solder, has been associated with tin-rich/pure tin materials. Historically, solder alloy element additions, such as lead or antimony, were used to prevent the formation of tin pest. However, the re-introduction of pure tin component surface fnishes and high-percentage tin solder alloys, driven by the requirement to eliminate lead from electronics products, has allowed tin pest to re-emerge as a potential reliability threat. In this study, a pure tin printed circuit assembly test vehicle was created to assess the possible impact of the tin pest in an actual "real-world" process scenario. Some test vehicles were subjected to a static −40°C temperature soak for a period of 10 years while other test vehicles were subjected to thermal shock. Subsequent to this testing, no indications of tin pest were observed in either set of test vehicles.

Keywords Tin pest · allotropic transformation · lead-free solder alloys

Background

Tin pest is one of the "ancient" history topics of the electronics industry. Historically, tin pest was referred to as tin disease or tin plague. These terms were based on the observations of "infected" tin $(\alpha$ -tin) coming into contact with areas of "uninfected" tin (β-tin) and spreading the infection like a "disease".^{[1](#page-10-0)} The German translation of "tin plague" is "zinn" pest," which English-speaking researchers ultimately began referring to as "tin pest" many years ago. Industry investigations and observations of the tin pest have been published for more than 100 years.^{[2](#page-10-1)} In 1851, Erdmann discovered a change in the tin pipes of a church organ in Zeitz, Germany.^{[3](#page-10-2)} He claimed that the tin pipes appeared crystalline and brittle on the surface, hypothesizing that the modifcation was due to a change in crystal structure caused by vibrations in the organ pipes. In addition to Erdmann's discoveries, Chiavari et al.^{[3](#page-10-2)} also examined ancient tin-based organ pipes, using x-ray difraction and scanning electron microscopy (SEM). Bars of tin stored at −38°C during the Leningrad winter

 \boxtimes Ross Wilcoxon ross.wilcoxon@collins.com of 1867 were reported to have completely disintegrated.^{[4](#page-10-3)} Tin pest has even generated its own urban myth: namely that Napoleon's defeat in Russia has been blamed in part on the disintegration of tin buttons on the soldier's uniforms, thereby reducing their efectiveness in cold weather. Documented archeological evidence has shown that the soldier uniform buttons were actually composed of pewter, copper, or brass and the officer uniform buttons were either silver or gold $(Fig. 1)$ $(Fig. 1)$ $(Fig. 1)$.^{[5](#page-10-4)} The regimental uniform records have no documentation of tin buttons, thus disproving the myth.^{[6](#page-10-5)}

Occurrences of the tin pest phenomenon have not been limited to the historic past. In 2009, Burns^{[7](#page-10-6)} reported a tin pest occurrence on tin-plated copper electrical connectors that were in storage. The connectors were intended to be replacements for installed connectors on an electric transit system. When it was time to replace the installed connectors, the replacements were found to be covered with a powder that was determined to be tin pest. While the connectors were specifed to have a tin coating with 2% bismuth, subsequent analysis of unafected areas on the connector demonstrated that the tin coating only included 0.1% bismuth.

In 2017, tin pest in the US National Defense tin stock-pile was reported.^{[8](#page-10-7)} The tin stockpile was established in the years leading up to World War II to address concerns that the supply for new material from Asia could be cut of. For

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decades, the stockpile has experienced the formation of tin pest, leading to crumbling of the tin. The tin pest is thought to be due to a combination of long-term air pollution and storage in a cold environment. The more than 4000 tons of tin in the stockpile are currently being converted back into 99.99% pure tin through a re-melting process.

In 2019, tin pest was reported on the international space station.^{[9](#page-10-8)} The tin pest was located on a tin-plated copper beryllium grounding strap bracket (Fig. [2](#page-1-1)).

Metallurgy of Tin Pest Phenomenon

Tin pest is caused by an allotropic transformation, which is the occurrence of an element in two or more crystalline forms. Six elements are known to exhibit allotropic transformations at one atmosphere of pressure and in the temperature range of −40°C to +60°C: carbon, cerium, phosphorus, sulfur, yttrium and tin.¹⁰ The tin allotropic transformation is from β-tin ("white" tin, metallic, body-centered tetragonal crystallographic lattice) into α-tin ("gray" tin, semiconducting, face-centered diamond cubic lattice) and occurs at or below 13.2°C. Figure [3](#page-1-2) illustrates the crystallographic structure diferences between the α-tin and β-tin.

The transformation in crystal structure from β-tin (white tin) to α -tin (gray tin) leads to a 27% increase in volume.^{[11](#page-10-10)}

Fig. 1 Fragment of Napoleon's army soldier uniform. Photo credit: Dr. Rimantas Jankauskas, Vilnius University.

This large volume change causes the naturally brittle gray tin to transform into a powder. Figure [4](#page-2-0) illustrates the tin pest allotropic transformation.[12](#page-10-11) Tin's maximum allotropic transformation rate occurs around -40° C.^{[11](#page-10-10)}

Industry investigations have documented another important crystallographic phenomenon associated with tin: the formation of tin whiskers.^{[13](#page-10-12)} Tin whiskers are single crystal flament structures that extrude from tin surfaces and can cause electrical failure of printed wiring assemblies. Other metals such as zinc and cadmium have also been shown to produce whiskers.[14](#page-10-13) The specifc root cause parameter set that induces whisker growth is not yet fully understood by the electronics industry. Figure [5](#page-2-1) illustrates tin whiskers on a pure tin surfaced fnished component. Pure tin materials are defned as tin plating or solder alloys that contain a minimum of 99% tin and 1% of other constituents. Tin whiskers and tin pest phenomena are both related to pure tin materials and have therefore received considerable attention over the past two decades due to the transition to lead-free electronics. However, it is important to recognize that tin pest and tin whiskers are due to two completely diferent mechanisms. Tin pest is the result of a well-understood change in the

Fig. 3 Tin allotropic crystallographic structures.

Fig. 2 Tin pest on an international space station ground strap bracket; (a) full strap, (b) close-up view. Photo credit: NASA.^{[9](#page-10-8)}

Fig. 4 Tin pest transformation example reprinted with permission from K. Sweatman.^{[12](#page-10-11)}

Fig. 5 Tin whiskers on a TSOP component.^{[15](#page-10-14)}

crystallographic structure of tin while tin whiskers are due to a less understood change in the physical shape of the material that appears to be related to local stress redistribution in tin plating. This study focused on tin pest and does not address the topic of tin whiskers.

Industry Literature Review

Tin is a common element utilized in solders and for many years has been alloyed with lead to generate a dependable solder for electronics. With the adoption of the Restriction of Hazardous Substances (RoHS) directive by the European Union (EU) in 2003 and its efective date of July 1, 2006, lead and other hazardous materials are restricted from use in the manufacture of various types of electronics and electrical equipment. As a result of this restriction, there has been continuous research to develop lead-free solders, especially those suited for mission-critical devices undergoing harsh environments. However, the utilization of lead-free solders has launched a resurgence of interest in tin pest, its ability to form in tin rich solders, and concerns for its impact on electronics reliability.

Signifcant investigation has been conducted on tin pest phenomenon using bulk samples of diferent lead-free tin alloys. In a 2001 study conducted by Kariya et al., 16 ingots composed of a Sn-0.5 wt.% Cu solder alloy were held at −18°C temperature for 2 years. After 1.5 years, about 40% of the specimen was transformed, and after 1.8 years, about 70% of the specimen was transformed to α-tin (Fig. [6](#page-3-0)). The researchers concluded that "tin pest could lead to total disintegration of micro-electronic solder joints." They warned that if the tin-copper eutectic system became the prominent lead-free solder, "major ramifcations on the service lifetime of electronic assemblies" could result.

A year later, Plumbridge^{[17](#page-10-16)} published additional investigation results on the Sn-0.5Cu and Sn-3.5Ag solder alloys that illustrated continued tin pest progression of bulk sample transformation (Fig. [7](#page-3-1)). In contrast, a Sn-Zn-3Bi alloy in the investigation had no signs of tin pest formation after 6 years of −18°C exposure. Plumbridge expressed concerns about products using lead-free solders with a design life of 25 years or more. He addressed the fact that many experiments rarely exceed 1 year, and in his study, it took 7 years before tin pest warts were visible in a majority of the solder alloys. He stated that there was, "a complacency that exists among much of the [electronics] industry regarding the signifcance of tin pest formation for reliability of solder ioints."^{[19](#page-10-17)}

While investigations with bulk materials have exposed concerns for tin pest, relatively few industry studies have addressed the likelihood of tin pest formation in actual solder joints. As part of a larger bulk solder tin pest investigation, $Peng¹⁸$ examined two mobile phone circuit boards in which one board used Sn36Pb2Ag solder and the other board had Sn3.8Ag0.7Cu solder. Neither board developed tin pest when subject to −196°C for 50 h, −40°C for 4 years, or −17°C for 1.5 years. Osterman[19](#page-10-17) subjected ball grid array (BGA) and quad fat package (QFP) components soldered with three lead-free solder alloys on test vehicles with immersion tin surface fnishes. The test vehicles were subjected to two diferent storage conditions, −30°C for 29 months or −80°C exposure for 20 months, with inspection intervals of approximately every three months. At the end of testing, none of the samples at either storage temperature exhibited indications of tin pest. Additionally, Plumbridge 20 20 20 did not include any tin pest testing on component solder joints in his research. He indicated however that component solder joint behavior may difer from bulk solder materials since "actual solder joints may be resistant [to tin pest transformation] due to the limited free solder surface available and the constraint of intermetallic compounds" and components did not include any testing on component solder joints in his research. 20

Fig. 6 Tin pest allotropic transformation on bulk solder samples, reprinted with permission from Plumbridge.¹

Fig. 7 Tin pest allotropic transformation on Sn-0.5Cu bulk solder samples, reprinted with permission from Plumbridge.¹⁷

Numerous studies aimed at improving the understanding of the kinetics of tin pest transformation are available in the published literature. Aptekar and Styrkas, 21 Zeng et al., 22 Matvienko and Sidelnikov, 23 23 23 and Burgers and Groen²⁴ used a variety of analytical techniques, including x-ray difraction, SEM, inoculation and temperature manipulation to measure the allotropic transformation for pure tin specimens. These studies have found a wide range of tin pest initiation times, which have been summarized by Styrkas 25 in terms of methodology and conversion percentage. Burgers and Groen²⁴ and Zeng et al.²² suggested that their comprehensive tin kinetics analysis supported the use of an Avrami relation to model the reaction kinetics for the β-to-α transformation.

Kariya et al.,^{[16](#page-10-15)} Plumbridge,¹⁷ Illes,²⁶ Skwarek et al.,^{[27](#page-10-26)} and Di Maio and Hunt^{[28,](#page-10-27)[29](#page-10-28)} investigated the allotropic transformation with a variety of lead-free solder alloys using inoculation and temperature manipulation. Illes^{[26](#page-10-25)} described the general process in three stages. The frst stage is nucleation, in which new phases develop through self-organization or new thermodynamic phases are created. Growth, in which the transformation progresses, is the second stage. The third stage is saturation, in which the transformation comes to an end and a metastable α -Sn state remains. Di Maio and $Hunt^{30}$ $Hunt^{30}$ $Hunt^{30}$ developed a method to track the $\beta \rightarrow \alpha$ transformation through time-lapse photography utilizing 99.99% pure tin. They observed the transformation beginning on the surface of the specimen and then progressing slowly while sequentially transforming small layers of material. Their observations agree with the findings of Kariya et al.¹⁶ Eventually, the top layer disintegrates and the transformation continues to progress further. Both parallel and perpendicular cracks with respect to the propagation direction were observed. Di Maio and $Hunt^{30}$ $Hunt^{30}$ $Hunt^{30}$ concluded that parallel cracks resulted from the stresses generated between transformed and untransformed regions, while perpendicular cracks result from the collisions of two transforming regions.

Equally importantly, Di Maio and $Hunt^{28}$ presented a procedure for monitoring the growth of tin pest utilizing electrical resistance measurements. At −35°C, the resistivity of α-Sn is about one order of magnitude greater than the resistivity of β-Sn at the same temperature. As a result, their procedure allowed them to develop a table indicating the propensity of transformation of seven diferent binary alloys.

There is disagreement in the published literature on the efects of mechanical deformation on lead-free tin alloys as a mechanism for accelerating or decelerating tin pest formation. The general consensus among a majority of the researchers is that cold working accelerates tin pest forma-tion.^{[16,](#page-10-15)[31](#page-10-30)} These researchers suggest that the accelerated transformation rate may be due to accumulated internal strain energy from the mechanical treatment. In turn, the built-up internal strain energy generates an additional driving force for the $\beta \rightarrow \alpha$ transformation. In contrast, Peng^{[18](#page-10-18)} suggested that cold working inhibits tin pest formation. Peng stated that the diference in results may be due to the deformation energy being consumed by stress relaxation and microstructural changes, such as coarsening, which occurred when the samples were stored at room temperature for over 6 months before subjecting the samples to low temperatures.

There is general consensus in the industry literature on what elements promote or retard tin pest formation. Researchers typically fnd that Pb, Bi, and Sb are efective inhibitors of tin pest formation with Bi having been shown to be the most effective.^{[10](#page-10-9),[16](#page-10-15),[20](#page-10-19)[,28](#page-10-27)[,32–](#page-10-31)[36](#page-10-32)} For instance, Plumbridge¹⁷ examined a Sn-8Zn-3Bi alloy for 6 years and no tin pest was observed; however; other bulk tin alloy samples with Pb, Ag, and Cu exhibited at least some tin pest formation. Zn, Al, Mg, and Mn are elements documented by researchers to accelerate tin pest formation. $32,37$ $32,37$ Additionally, MacIntosh, 38 as part of an International Tin Research Institute (ITRI) study, reported on the bulk susceptibility of 44 diferent tin containing materials exposed to -17.8 °C (0°F) and -40 °C (-40 °F) storage conditions for a 10-year period. Overall, only minor tin pest observations were documented, including in eight commercially available solders containing Pb and Sb that showed no indications of tin pest transformation.

Industrial-grade solder alloys have typically had small percentage additions of all three of these elements (Bi, Pb, Sb) that were sufficient to prevent the tin pest phenomenon. However, the introduction of high-percentage tin solder alloys that are lead-free due to the RoHS legislative pressures has reintroduced tin pest questions and concerns. Electronic equipment is used in a wide number of product-use environments with −40°C temperature exposures. Therefore, the fundamental purpose of this investigation was to determine whether a lead-free, high-percentage tin solder alloy would undergo a tin pest reaction over time.

Objective

The primary objective of the investigation was to determine whether tin pest could occur on a pure tin solder alloy on a "real-world" printed circuit assembly that was fabricated using conventional industry surface mount soldering processes and subsequently subjected to test conditions that have been shown to induce tin pest in bulk samples.

Test Procedures

An SMTA Saber evaluation board with an organic solderability preservative (OSP) surface fnish was procured from Practical Components. Components on the test board had two diferent surfaces fnishes: DPaks (also known as TO-252) and resistors (1206 and 1210 sizes) had matte tin surface fnish while small outline integrated circuit (SOIC-16), plastic leaded chip carrier (PLCC-68), and small outline transistor (SOT-23) components had tin/lead surface fnish. While this study focused on the potential increased risk of tin pest in lead-free assembly, including components with tin–lead surface fnish provided a baseline direct comparison in the event that the solder joints with matte tin surface fnish demonstrated substantial tin pest formation. In addition, 912 unpopulated component pads on the circuit boards were printed with solder paste to produce solder samples for tin pest inspection locations. A pure tin solder paste was created from 99.99% pure tin powder (Kester Solders) and Alpha Metals UP-78 paste fux. The constituted solder paste was 89% pure tin powder and 11% paste fux by weight. The test vehicle, components and solder paste materials were specifcally selected to create an assembly with abnormally high tin solder joints. Each test vehicle consisted of a printed circuit assembly populated with two DPak components, one PLCC68 component, three SOIC16 components, 24 SOT23 components, 20 1206 resistors, and 14 1210 resistors in the locations indicated on Fig. [8](#page-5-0). The 912 solder pad deposits were applied in multiple areas on unused test board pads.

Twelve test vehicles in total were assembled at the Collins Aerospace Coralville, Iowa facility. The solder paste was hand-applied onto the test vehicles with a metal squeegee. The components were manually placed on the test vehicles and then processed through a Heller refow oven using a lead-free profle with a peak temperature of 260°C. After reflow soldering, flux residues were removed from the test vehicles with an Electrovert Aquastorm aqueous cleaner using Kyzen SSA chemistry. The boards were then placed in polyester bags to prevent contamination and to facilitate handling.

Test Vehicle Conditioning

The test vehicles were divided into three sets for conditioning:

• Set 1 test vehicles (quantity of two test vehicles) were subjected to thermal shock testing with temperature extremes of −40°C and +40°C, temperature transition time of 60 s, and 20 m dwells at each extreme temperature. Thermal shock test vehicles were subjected to a total

Fig. 8 Tin pest test vehicle (SMTA Saber Evaluation Board): Outlined areas indicated primary component test population regions (SOT-23, SOIC16, PLCC68, 1206, 1210, DPak).

Fig. 9 Optical images of solder joints after 3000 thermal shocks: (a) SOIC16 with tin/lead fnish, (b) DPak with matte tin fnish.

Fig. 10 Optical image of solder pad deposit after 3000 thermal shocks.

Fig. 12 Optical image of solder pad deposits after 10-year static soak @ −40°C.

Fig. 11 Optical images of solder joints after 10-year static soak @ −40°C: (a) SOIC16 with tin/lead fnish, (b) DPak with matte tin fnish.

of 3000 thermal shocks with visual inspection at \times 100 magnifcation approximately every 500 thermal shocks.

- Set 2 test vehicles (quantity of five test vehicles) were subjected to static cold-soak conditioning at −40°C. The static soak test vehicles were inspected using optical microscopy with \times 100 magnification approximately once a week for the initial 3 months and then once every 3 months for the duration of the 10-year study.
- Set 3 test vehicles (quantity of five test vehicles) were also subjected to static cold-soak conditioning at −40°C. However, 100 locations on each of these test vehicles were specifically "inoculated" with α -tin crystals in an effort to encourage allotropic transformation. Set 3 test vehicles were also inspected using optical microscopy with \times 100 magnification approximately weekly for the initial 3 months and then once every 3 months for the duration of the 10-year study.

Inspection Results

The following observations were recorded:

- Set 1 test vehicles: Optical microscopy inspection found no indications of tin pest on any components or solder pad deposits after 3000 thermal shocks had been accu-mulated. Figure [9](#page-6-0) illustrates post-thermal shock components with both tin/lead and matte tin fnishes. Figure [10](#page-6-1) illustrates a solder pad deposit after 3000 thermal shocks.
- Set 2 test vehicles: None of the optical microscopy inspections conducted over the 10-year period of static cold-soak conditions revealed any indications of tin pest. Figure [11](#page-6-2) shows examples of solder joints after 10 years while Fig. [12](#page-6-3) shows solder deposits.

• Set 3 test vehicles: During the first 7 months of static cold-soak conditioning, optical microscopy inspection revealed no indications of tin pest. However, starting with the 10-month inspection period, tin pest transformation was frst observed on "inoculated" locations, and the phenomenon continued to progress over time. Figure [13](#page-7-0) illustrates a single solder pad deposit before and after tin pest was observed, while Fig. [14](#page-8-0) shows the substantial damage incurred on these inoculated solder deposits after 10 years of cold temperature soak.

Analytical Results

The published industry literature contains numerous discussions on the impact of element inhibitors of tin pest formation. A number of analytical analysis methods, e.g., x-ray fuorescence spectroscopy (XRF), inductively coupled plasma (ICP), were conducted to determine the level of trace elements in the tin solder powder used to create the solder paste, the resulting assembly solder joints, and the component fnishes (see Tables [I](#page-8-1) and [II\)](#page-8-2). Table [I](#page-8-1) shows the three component package styles with tin–lead surface fnish (PLCC68, SOIC16 and SOT23) and the three package styles with matte tin surface fnish (SMT 1206, SMT1210 and DPak). Table [II](#page-8-2) includes ICP measurements for SAC305 solder alloy as a reference to compare the trace elements of the solder joints in this study to conventional solder joints.

This information on the tin powder, the resulting assembly solder joints and what can be found in a typical SAC305 solder joint were used to compare/contrast the reported literature results to the current investigation results. The

Fig. 13 Optical image of inoculated solder pad deposit after 10-year static soak @ −40°C: (a) no tin pest at 7 months, (b) tin pest transformation initiated at 10 months.

Fig. 14 Optical image of inoculated solder pad deposit after 10-year static soak @ −40°C: (a) inoculated pad that is missing solder deposit, (b) portion of solder pad deposit that fell off.

Table I XRF analysis of component finishes	Component type	$Sn\%$	Pb%
	PLCC68	76	25
	SOIC ₁₆	83	17
	SOT ₂₃	88	12
	SMT 1206	100	0
	SMT1210	100	0
	DPak	100	0

Table II ICP analysis of tin powder, assembly solder joint and standard SAC305 solder alloy

Values in % by weight

A: Tin powder and solder joint in this study

B: Solder joint in a production board in accordance with IPC-J-STD-006

potential impact on retarding tin pest formation for the measured levels of antimony and bismuth was in general agreement with the published literature.

Statistical Assessment

Assembled test vehicles included three diferent solder confgurations, as the pure tin solder paste was applied to all pads, some of which were assembled with components that had either tin–lead (SnPb) surface fnish or matte tin surface fnish. Each pad on the circuit board represented a potential opportunity for tin pest to be observed. Table [III](#page-9-0) summarizes the number of possible tin pest sites for each solder confguration on a single test board.

The three diferent test sets used in the study did not include the same number of test boards. Also, some of the potential tin pest sites in the Set 3 test population were inoculated (10 solder joints on resistors and 90 solder deposits). Therefore, the number of potential sites for tin pest varied with the test set and solder configuration. Results of observations are summarized in Table [IV,](#page-9-1) which reports results in a format of x/y (z), in which x is the number of sites that exhibited tin pest, y is the total number of possible tin pest sites, and z is the portion of possible sites exhibiting tin pest. This table shows that only those locations that were seeded with α -tin, i.e., were inoculated, exhibited tin pest.

Discussion

This investigation found no evidence of tin pest on noninoculated assembly solder joints that were subjected to either thermal shock or static cold-soak conditioning methods. The only tin pest transformation that was observed over the 10-year exposure period occurred in samples that were intentionally inoculated and subjected to static cold temperature soak. These results agree with **Table III** Test quantities for opportunities to observe tin pest per test board

Table IV Tin pest results—# observed / # opportunities (% of samples exhibiting tin pest)

those of Peng^{[18](#page-10-18)} and Osterman,^{[19](#page-10-17)} who reported no indications of tin pest allotropic transformation on electronic components subjected to cold temperature exposure of 1–9 years. The investigation results also agree with Vetter and Guhl,^{[39](#page-10-35)} who reported no allotropic transformation on solder joints over a 12-month observation period. Vetter and Guhl believed that the lack of transformation in the soldered test vehicles was due to elemental suppression caused by impurities in the solder joints from the solder powder used to fabricate the assembly solder paste and the component/board surface fnishes. The results from this investigation also agree with other industry research citing the effects of elemental suppression. Three elements bismuth, antimony, and lead—have been shown to con-sistently suppress/eliminate the tin pest reaction.^{[2,](#page-10-1)[10,](#page-10-9)[12](#page-10-11),[20](#page-10-19)} This investigation's 10-year continuous −40°C exposure is equal to that of the 10-year ITRI 38 study. In the ITRI study, the tin pest transformation was suppressed by elemental additions in bulk test samples. In contrast, this investigation used actual printed circuit solder joints produced with contemporary printed circuit assembly processes. The lack of tin pest occurrence on components with tin/lead surface fnishes aligns with other industry investigations that

concluded that the presence of lead suppresses the allotropic transformation.

Conclusion

This investigation showed that pure tin assembly solder joints did not undergo the tin allotropic transformation under thermal shock and extended static cold-soak conditioning. Pure tin assembly solder joints that were intentionally "inoculated" with α -tin did exhibit tin pest in comparison to pure tin solder joints that were not inoculated or assembly solder joints that were created with tin/lead surface fnished components. Since the pure tin assembly solder joints did exhibit tin pest only when they were "inoculated" with α -tin, it appears that the inherent low-percentage elemental contributions inherent to actual lead-free solder joints are sufficient to suppress the formation of tin pest on circuit card assemblies.

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Conflict of interest The authors declare that they have no confict of interest.

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