

Use of Fe-C Information as Reference for Alloying Effects on B_S



LINDSAY LEACH, PETER KOLMSKOG, LARS HÖGLUND, MATS HILLERT,
and ANNIKA BORGENTAM

Many empirical equations of the variation of the critical temperature with alloy content of the start of bainite formation in steels are available. They are often obtained by regression analysis of measured values for a large number of alloyed steels, usually with several alloying elements. However, such equations differ considerably, especially when applied to pure Fe-C alloys, which results in large differences between reported effects of individual alloying elements since they have not been based on the Fe-C system as a reference. Apparently, for the first time, an empirical equation is now derived by starting with information from Fe-C alloys and low alloy steels and then adding the effect of each alloying element separately, using information from ternary Fe-C-M alloys. Sets of information from the same alloy content but different carbon contents proved particularly useful. Lines connecting such points are regarded as B_S lines for the respective alloy content and the effect of alloying elements was evaluated from their distance from the B_S line for Fe-C alloys. Only under this condition can coefficients for alloying elements be expected to represent the physical effect of the elements. The resulting equation was tested with about 600 experimental B_S temperatures.

<https://doi.org/10.1007/s11661-019-05371-1>
© The Author(s) 2019

I. INTRODUCTION

BAINITE is one of the microstructures that can form from austenite on cooling. It has long been of considerable practical importance because it may prevent the complete transformation of austenite to martensite. On the other hand, a completely bainitic microstructure may have unique properties and has found practical use, especially in recent times. Alloying elements have been used for controlling the formation of bainite in various applications. Due to various aspects, it has thus been of considerable interest to be able to control the formation of bainite by alloying elements and information on their effects on the start temperature for bainite formation has been of central interest. The start temperature is denoted as B_S and the variation of the B_S temperature with the carbon content at fixed alloy content can be represented by a B_S line.

Several empirical equations for predicting the B_S temperature from the steel composition have been derived from large amounts of experimental information on alloyed steels. The result is usually presented as a linear equation with a coefficient for each element, often

obtained by regression analysis. An early example is the equation by Steven and Haynes^[1] who determined the B_S temperatures of a group of 65 steels and evaluated the coefficients by linear regression analysis.

$$B_S(^{\circ}\text{C}) = 830 - 270\text{C} - 90\text{Mn} - 37\text{Ni} - 70\text{Cr} - 83\text{Mo} \quad [1]$$

The carbon and alloy contents are given in mass pct. In a recent review, Kang *et al.*^[2] listed nine, more recent equations,^[3–11]

$$B_S(^{\circ}\text{C}) = 656 - 57.7\text{C} - 75\text{Si} - 35\text{Mn} - 15.3\text{Ni} - 34\text{Cr} - 41.2\text{Mo} \quad [2]$$

$$B_S(^{\circ}\text{C}) = 718 - 425\text{C} - 42.5\text{Mn} \quad [3]$$

$$B_S(^{\circ}\text{C}) = 844 - 597\text{C} - 63\text{Mn} - 16\text{Ni} - 78\text{Cr} \quad [4]$$

$$B_S(^{\circ}\text{C}) = 720 - 582.63\text{C} + 126.6\text{C}^2 \\ - 91.68\text{Mn} - 66.34\text{Ni} - 31.66\text{Cr} \\ - 42.37\text{Mo} + 9.16\text{Co} - 36.02\text{Cu} - 41.15\text{Ru} \quad [5]$$

$$B_S(^{\circ}\text{C}) = 732 - 202\text{C} + 216\text{Si} - 85\text{Mn} - 37\text{Ni} - 47\text{Cr} - 39\text{Mo} \quad [6]$$

LINDSAY LEACH, PETER KOLMSKOG, LARS HÖGLUND, MATS HILLERT, and ANNIKA BORGENTAM are with the Department of Materials Science and Engineering, KTH Royal Institute of Technology, Brinellvägen 23, 10044 Stockholm, Sweden. Contact e-mail: annbor@kth.se

Manuscript submitted February 8, 2019.
Article published online August 5, 2019

$$B_S(^{\circ}\text{C}) = 711 - 361.9\text{C} + 261.9\text{C}^2 - 28.3\text{Mn} + 43.7\text{Si} \quad [7]$$

$$B_S(^{\circ}\text{C}) = 630 - 45\text{Mn} - 40\text{V} - 35\text{Si} - 30\text{Cr} - 25\text{Mo} - 20\text{Ni} - 15\text{W} \quad [8]$$

$$B_S(^{\circ}\text{C}) = 745 - 110\text{C} - 59\text{Mn} - 39\text{Ni} - 68\text{Cr} - 106\text{Mo} - 17\text{MnNi} + 6\text{Cr}^2 + 29\text{Mo}^2 \quad [9]$$

$$B_S(^{\circ}\text{C}) = 839 - 270[1 - \exp(-1.33\text{C})] - 86\text{Mn} - 23\text{Si} - 67\text{Cr} - 33\text{Ni} - 75\text{Mo} \quad [10]$$

It is striking that the numerical values of the coefficients vary considerably between the equations. For Mn it varies from 35 to 90 and for Ni from 15.3 to 66.34. This may partly be caused by the use of different groups of steels by different authors. It may seem that the resulting equation in such a case can give usable predictions for new steels of similar compositions. The equations also differ appreciably when applied to binary Fe-C alloys as illustrated in Figure 1. To indicate that they apply to binary Fe-C alloys with no alloying elements they will all be denoted as B_S° lines. It is evident that at least a large majority of the authors have not paid any attention to the consequences for Fe-C alloys which may be understandable if their interest was in alloyed steels.

Of course, the evaluated coefficient for an alloying element represents the difference from the properties of the Fe-C alloys according to the same equation. It is thus to be expected that the coefficient for an alloying element differs between equations if their predictions for Fe-C alloys differ. To obtain an equation that predicts realistic effects of individual alloying elements, it is thus necessary to base the equation on a realistic description of binary Fe-C alloys. It may also seem wise to evaluate the coefficient for each alloying element separately because, if two alloying elements occur simultaneously in a large fraction of the alloys, used in the regression analysis, it may be difficult to separate their effects properly. The strategy of the present work was thus to derive an empirical equation by starting with information from binary Fe-C alloys and then evaluating the alloying effects one by one from ternary information rather than applying a regression analysis to all the experimental information in one operation. Possible interactions between alloying elements will be neglected. Even among the above equations, which to a large extent have been based on steels with more than one alloying element, only one, Eq. [9], contains a term for interaction between two elements.

An extensive review of B_S information was recently performed by Kolmskog *et al.*^[12] and they paid special attention to binary Fe-C alloys for which the direct information was very meagre, possibly due to experimental difficulties, *e.g.*, by rapid formation of cementite

and the competition with pearlite. They derived some additional information from the variation of growth rate with the carbon content of alloys. They also included corresponding information on Widmanstätten ferrite, here denoted as W_S , accepting the view that there is no difference between bainitic ferrite and Widmanstätten ferrite.^[13] In addition to classical B_S information, concerning the critical temperature of bainite formation for fixed composition (method T), Kolmskog *et al.* also included experimental information on the critical carbon content of bainite formation at fixed temperature and fixed alloy content (method C). They listed the information separately for the experimental T(temperature) and C(carbon content) methods but a statistical test did not indicate that they concerned critical conditions for different physical processes. The present work was based on their compilation and results from both methods were accepted.

For Fe-C alloys, or for alloyed steels of some fixed alloy content, the information was expected to form B_S lines. In the present work, B_S lines will be a useful means of illustration and C method data will be particularly valuable because sets of such data from various temperatures are often presented for some fixed alloy content and they can thus be illustrated directly as a B_S line. Such lines are more seldom obtained from classical B_S information but two examples are presented in Figure 2. One was obtained from a set of Cr steels with similar alloy content but different carbon contents and another from a set of plain carbon steels.

The straight line was fitted to information from the set of plain carbon steels with an average alloy content of 0.38 pctMn and 0.18 pctSi (mass pct).^[13] This line was marked WB_S because it concerned a combination of data from Widmanstätten ferrite from higher temperatures and bainite from lower. As already mentioned, this combination is based on the assumption that there is no difference between acicular ferrite when it forms primarily, whether it will later be found together with cementite, in an arrangement regarded as bainite, or not. Due to a resulting wide temperature range, it could yield some support of the straight extrapolations of lines in Figure 1 outside their experimental ranges.

The filled squares in Figure 2 were obtained by interpreting TTT diagrams from Lyman and Troiano^[14] for a set of Cr steels with an average alloy content of 3.0 Cr, 0.26 Mn and 0.20 Si (mass pct). That information indicates a strongly curved B_S line and was recently used by van Bohemen^[11] when he derived Eq. [10], which thus predicts a curved B_S° line for binary Fe-C alloys. This may be taken as an illustration that to obtain a reliable empirical equation one should use some information from binary Fe-C alloys. Lines for austenite in para-equilibrium, PE, with ferrite are included in Figure 2 for comparison and were calculated for the respective average alloy content using Thermo-Calc software^[15] and the TCFE8 database.^[16] For the plain carbon steels, there is an increasing distance with lowering temperature between the PE line and experimental information but for the Cr steels there is a large distance at all

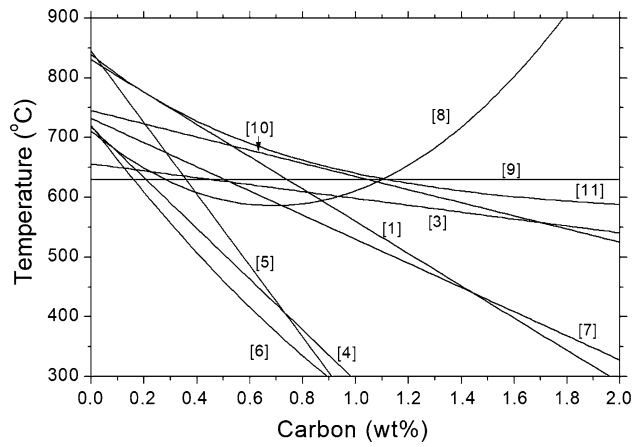


Fig. 1— B_s^0 lines for Fe-C alloys predicted from many empirical equations.

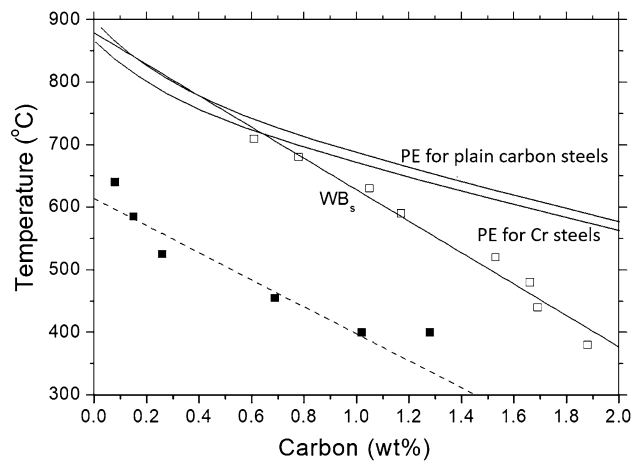


Fig. 2—Filled squares: B_s information from Lyman and Troiano^[14] on Cr steels. Representation by a straight, dashed line shows strong deviation for some points. Open squares: W_s and B_s information from Hillert^[13] on plain carbon steels. The straight solid line is the W_B_s line from a least square fit.

temperatures. In principle, it is this effect of alloying elements that will be examined in the following.

II. ANALYSIS OF EXPERIMENTAL DATA

A. The $W_B_s^0$ Line for Binary Fe-C Alloys

The present work was based on the information compiled by Kolmskog *et al.*^[12] It started by focusing on the binary Fe-C system for which they had made a special effort. The main results are presented in Figure 3 together with the B_s^0 line predicted from Eq. [1] obtained by Steven and Haynes.^[1] Six points by Aaronson^[17] agree well with the B_s^0 line by Steven and Haynes,^[1] which had been published a few years earlier. Nevertheless, there is a lack of documentation of his values and they were not considered when a linear regression analysis was done. It was performed using only the filled symbols in Figure 3 which originate from Johanson,^[18]

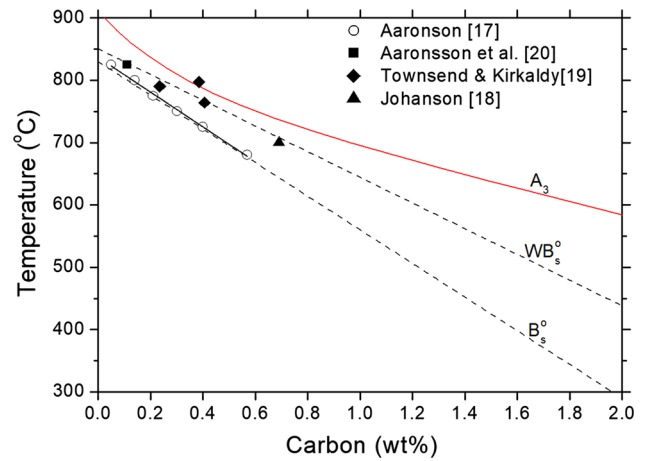


Fig. 3—Experimental information from Fe-C alloys. The dashed B_s^0 line is extrapolated to Fe-C alloys using Eq. [1] by Steven and Haynes.^[1] The dashed $W_B_s^0$ line is the result of the present linear regression analysis of all the filled symbols.

Townsend and Kirkaldy^[19] and Aaronson *et al.*^[20] It gave the following result and the corresponding line is included in Figure 3 as a line denoted as $W_B_s^0$ to emphasize that it concerns Widmanstätten ferrite as well as bainite in binary Fe-C alloys.

$$W_B_s^0(^{\circ}\text{C}) = 850 - 206C \quad [11]$$

The $W_B_s^0$ line in Figure 3 has been extended far outside the experimental range. The extension could possibly be tested using the information from plain carbon steels, presented in Figure 2, by neglecting the effects of their Mn and Si contents. That information covers a much wider range of carbon content. That test will be made later in connection to the assessment of the effect of Mn.

For binary Fe-C alloys, the PE line degenerates to the A_3 line, which is included in Figure 3, and the growing distance from the $W_B_s^0$ line with increasing carbon content demonstrates that the conditions at the interface of the growing acicular ferrite may be rather close to equilibrium at low carbon contents but may deviate markedly at higher carbon contents. According to the strategy of the present work, the $W_B_s^0$ line will now be accepted as the Fe-C reference in the assessments of the effects of alloying elements from ternary information.

B. Analysis of Information from Ternary Systems

The two coefficients for Fe-C alloys in Eq. [11] will now be applied to ternary Fe-C-X systems and the effects of alloying elements will be evaluated for each system. Interactions between elements will be neglected and the resulting equations will thus be linear. As a consequence, it should not be necessary to use information from quaternary alloys but such information will be used when the information from a ternary system is not self-consistent.

Most B_s information is available from separate alloys and effects of alloying elements can then be derived by

linear regression analysis. However, for ternary systems there is sometimes information from a set of alloys with the same alloying element content but varying carbon content. As mentioned in the introduction, the B_S values from such sets should in principle be combined into WB_S lines but experimental scatter may result in a cloud of points around an imagined ideal line. It is then possible to estimate experimental scatter directly. Furthermore, according to a linear approach, the slope of the WB_S line for a fixed alloy content in a T, C diagram is given by the second term in the empirical equation and such WB_S lines for all alloying elements should be straight and parallel to the WB_S^0 line for binary Fe-C alloys. Due to non-linearity in the experimental information, there may be deviations from this rule but they should at least vary in a systematic way. This is thus a second way to check the quality of information. It was decided to rely on such sets of B_S values when available.

1. Fe-C-Mn

Liu *et al.*^[21,22] studied the critical carbon content in a ternary alloy with 2.99 mass pct Mn at a series of temperatures and results are plotted in Figure 4 together with the PE line and the B_S^0 and WB_S^0 lines for Fe-C alloys from Figure 3. The points fall onto a straight, imagined WB_S line, roughly parallel to the B_S^0 or WB_S^0 line. Direct estimate of the vertical distance of that line from the WB_S^0 line yields a Mn coefficient of about -80 K/pct, which may be compared with values from -91.68 to -35 in Eqs. [1] through [10].

For regression analysis, information was added from Borgenstam^[23] who studied a number of Mn contents but only three temperatures for each alloy. In addition, five single B_S values on ternary Fe-C-Mn alloys^[20-26] were accepted. As already explained, information on plain carbon steels in Figure 2 was included and also the five black points in Figure 3 that yielded the WB_S^0 line. All this information is collected in Figure 5 where values from sets of the same alloy content were connected by short straight lines to show their relation. It is interesting that all of the lines for alloys from Borgenstam have

negative slopes as the line for the alloy from Liu *et al.* but the scatter is appreciable.

A linear regression analysis was done with all information in Figure 5. The following equation was thus obtained,

$$WB_S(^{\circ}C) = 854 - 212C - 79Mn \quad [12]$$

This result is illustrated in Figure 6 with WB_S lines calculated for the experimental Mn contents of the sets of alloys with equal alloy contents. The first two coefficients in Eq. [12] agree well with those in Eq. [11], which is demonstrated by the similarity between the 0Mn and WB_S^0 lines. One may thus consider the information from plain carbon steels, which was added to Figure 5, as support for Eq. [11]. The Mn coefficient agrees excellently with the value of -80 K/pct, evaluated directly from the same information when given in Figure 4. It would thus seem attractive to accept all the coefficients as given by Eq. [12]. However, in line with the strategy to start with the binary information, it was instead decided to accept Eq. [11] and evaluate only the Mn coefficient from the regression analysis. It yielded Eq. [13] which gave only a marginal change of the Mn coefficient.

$$WB_S^0(^{\circ}C) = 850 - 206C - 78Mn \quad [13]$$

2. Fe-C-Ni

There is much information from ternary Fe-C-Ni alloys^[18,20,27-30] but few sets with constant alloy content. The information is presented in Figure 7 where the three circles for 1 mass pct Ni from Johanson,^[18] combined with the single cross for 1.06 mass pct Ni from Aaronson *et al.*,^[20] can give a reasonably straight WB_S line parallel to the WB_S^0 line. Imagined slopes for the other alloy contents show the same tendency but with

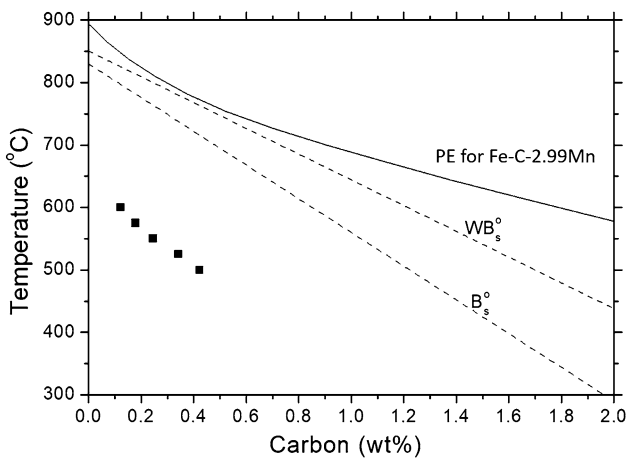


Fig. 4— B_S values from Liu *et al.*^[22] from ternary Fe-C-2.99 Mn alloys with different carbon contents. They are compared with the dashed B_S^0 and WB_S^0 lines from Fig. 3.

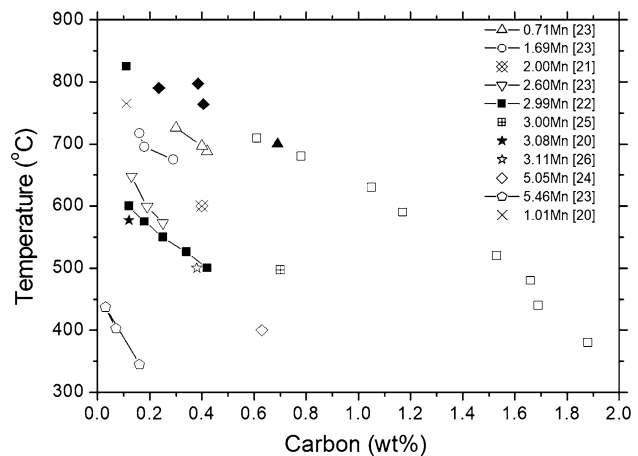


Fig. 5— B_S values from Fig. 4 combined with information from other sets of Fe-C-Mn alloys with constant alloy contents. Comparison is also made with plain carbon steels with the information in Fig. 2 and Fe-C alloys with the information in Fig. 3 but the data by Aaronson^[17] were excluded.

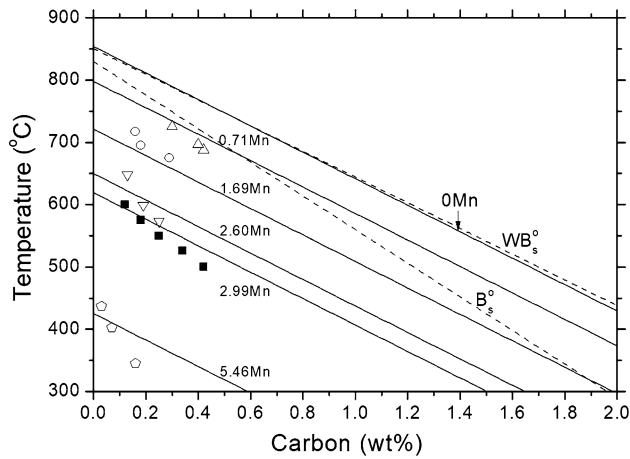


Fig. 6—Linear regression analysis of all information in Fig. 5. Results are presented as WB_S lines for the Mn contents of the experimental information. The result for zero Mn content should ideally be identical to the WB_S^0 line from Fig. 3 for Fe-C alloys. Information on experimental points is given in Fig. 5.

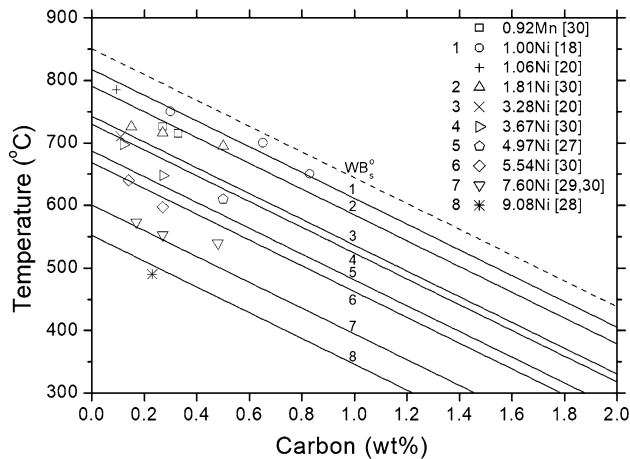


Fig. 7— B_S information from Fe-C-Ni alloys. Symbols for each alloy content should ideally form a WB_S line. That is most evident by combining information on 1.00 and 1.06 mass pct Ni which gives a WB_S line parallel to the WB_S^0 line for Fe-C alloys.

less distinction. A linear regression analysis with the two coefficients from Eq. [11] fixed gave

$$WB_S(^{\circ}C) = 850 - 206C - 33Ni \quad [14]$$

The family of parallel WB_S lines in Figure 7 were calculated from Eq. [13] for the experimental Ni contents. There is a reasonable fit in the whole range of Ni contents but only the information from 1.00 mass pct Ni cover a sufficiently wide range of carbon contents to directly support the evaluated slope of the parallel WB_S lines.

3. Fe-C-Cr

B_S information from Fe-C-Cr alloys is presented in Figure 8 and there are several sources [20,23,27,31-33] and most of them with sets of alloys with the same alloy content but different carbon contents. Straight,

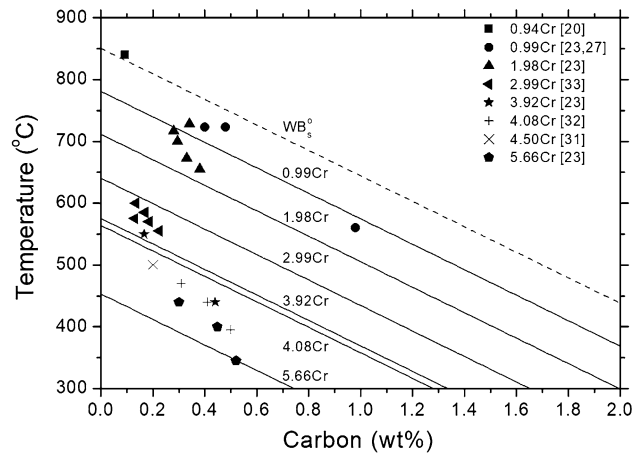


Fig. 8— B_S information from Fe-C-Cr alloys. Data of each kind of symbol should ideally form a B_S line. They often seem to be steeper than the WB_S^0 line for Fe-C alloys.

imagined WB_S lines for these sets would give negative slopes with a tendency to be steeper than the WB_S^0 line derived for binary Fe-C alloys. A linear regression analysis of all the information in this figure was done but with the two first coefficients fixed to the values in Equation 11. The result was

$$WB_S(^{\circ}C) = 850 - 206C - 70Cr \quad [15]$$

The family of parallel WB_S lines in Figure 8 were calculated from Eq. [15] for the experimental Cr contents and seem to represent the information reasonably well, considering the experimental scatter, which is directly noticed for sets of equal alloy content. However, there are obvious cases of deviation.

4. Fe-C-Mo

B_S information from Fe-C-Mo alloys is presented in Figure 9 and there are many sources [20,34-40] and several of them with sets of alloys with the same alloy content but different carbon contents. Straight, imagined WB_S lines for these sets would have slopes that agree better with the WB_S^0 line than the imagined WB_S lines for Fe-C-Cr alloys would do. Examples are the standing triangles and the asterisks. A linear regression analysis with the two coefficients from Eq. [11] fixed gave

$$WB_S(^{\circ}C) = 850 - 206C - 75Mo \quad [16]$$

The family of parallel WB_S lines in Figure 9 were calculated from Eq. [16] for the experimental Mo contents. There is reasonable agreement around 2.5 mass pct Mo but alloys with lower Mo contents fall below their WB_S lines.

5. Fe-C-Si

For ternary Fe-C-Si alloys, there are two sets of B_S values for constant Si content by Pevzner *et al.* [41] and Papadimitriou *et al.* [42] with 2.69 and 3.85 mass pct Si, respectively. The values are plotted in Figure 10 together with B_S^0 , WB_S^0 and A_3 lines for Fe-C alloys

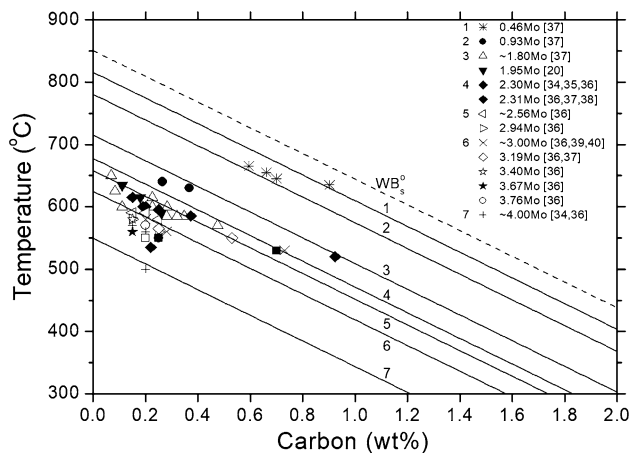


Fig. 9— B_s information from Fe-C-Mo alloys. Each kind of symbol should ideally form a WB_s line.

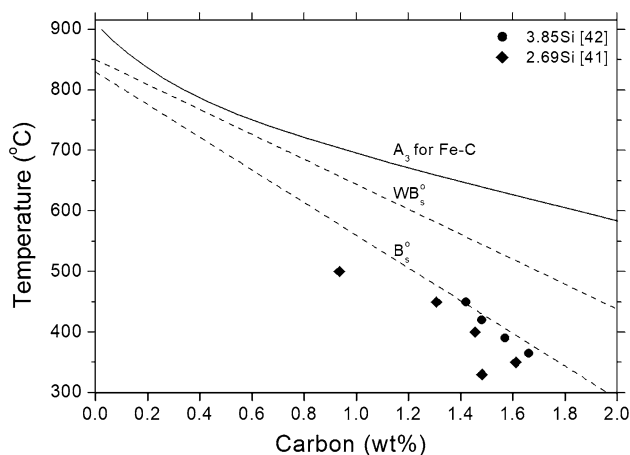


Fig. 10— B_s information on Fe-C-Si alloys from two sets of constant alloy content. Pevzner *et al.*^[41] used 2.69 Si and Papadimitriou *et al.*^[42] used 3.85 Si.

from Figure 3 for comparison. Even though the experimental range of carbon content is limited, the four circles from Papadimitriou *et al.* could define a straight, imagined WB_s line very well and it would be only slightly steeper than the WB_s^0 line for Fe-C alloys. The lowest diamond from Pevzner *et al.* will be neglected because the rejection of carbon from the precipitated ferrite may not be complete due to insufficient annealing time, considering how low that temperature is. The next three diamonds would define a straight, imagined WB_s line with a slope supporting the slope of the first set. However, the two sets are in conflict because they refer to different Si contents and the one with the lower Si content is further away from the WB_s^0 line than the other. The coefficient for Si was evaluated for both from their vertical distance from the WB_s^0 line, yielding -56 and -32 K/pct, respectively. This could indicate that one of the sets has a strong systematic error or both have. The effect of Si cannot be estimated without further information, which might be obtained from quaternary alloys.

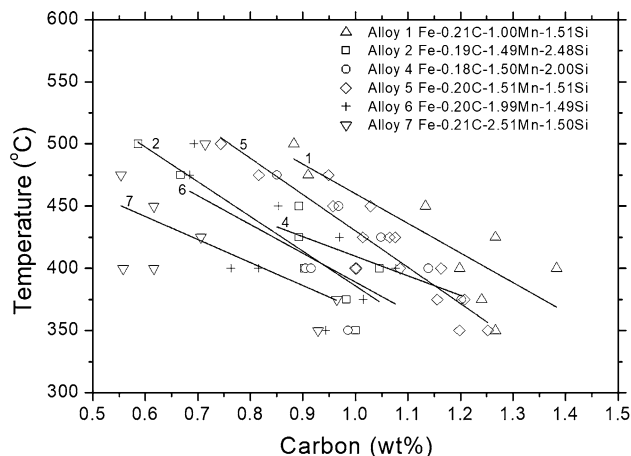


Fig. 11—Combination of B_s values from Fe-C-Mn-Si alloys from Usui *et al.*^[43] and Sugimoto *et al.*^[44,45] There are six alloy contents, each with several carbon contents. Straight lines are least square representations for equal alloy content.

C. Evaluating the Effect of Si from Quaternary Information

Valuable information for evaluating the coefficient for Si may be obtained from Fe-C-Mn-Si alloys since the effect of Mn has now been established. Usui *et al.*^[43] and Sugimoto *et al.*^[44,45] have performed a very thorough study of the critical carbon content for bainite formation for seven alloy contents and over a wide range of temperatures. All the B_s values are presented in Figure 11 and, in principle, each set for a fixed alloy content should form a WB_s line. A least square fit was done on each set to estimate the slope of its WB_s line. The resulting straight lines for six of the alloy contents are included in Figure 11. The information from the third set was excluded because it covered a narrow range of carbon contents and the line was not well defined.

The six lines are not very parallel but point in the same general direction as the WB_s^0 line in Figure 3. A linear regression analysis was done without fixing any coefficient, which was possible because the information now included different contents of both Mn and Si. The following equation was obtained,

$$WB_s(^{\circ}C) = 833-241C-74Mn-36Si \quad [17]$$

The first two coefficients in Eq. [17] represent Fe-C alloys and it is thus surprising but promising that they agree rather well with Eq. [11]. It may seem tempting to accept all coefficients in Eq. [17] but the present strategy was to rely on binary and ternary information as far as possible. The present linear regression analysis was thus repeated with the C and Mn coefficients fixed to the values in Eq. [13]. Only the coefficient for Si was thus evaluated here and the following equation was obtained,

$$WB_s(^{\circ}C) = 850-206C-78Mn-61Si \quad [18]$$

It is interesting that the Si coefficient was almost doubled compared to Eq. [17] although they are both based on the same Si information. The difference is

explained by the difference in the first two coefficients that represent Fe-C alloys. This exemplifies the fact that an error in one coefficient can be partly compensated by accepting an error in another coefficient. Here, it demonstrates the importance of starting an evaluation of coefficients from Fe-C alloys and then continuing to ternary alloys, without revising the two Fe-C coefficients later on. For the same reason, it would not have been wise to accept the Mn value of -74 K/pct in Eq. [17] from quaternary alloys, obtained together with a Si value. That is why not only the two Fe-C coefficients from Eq. [13] were fixed in the last regression analysis, but also the Mn value of -78 K/pct.

It is now a question whether -61 K/pct should be accepted as the coefficient for Si. Comparison with the values from the ten empirical expressions, listed as Eqs. [1] through [10], gave no valuable indication because their values spread from -75 to $+216$ K/pct for Si, which is connected to the large differences between the coefficients for the other elements. Comparison with the information from ternary Fe-C-Si alloys in Figure 10, which yielded two alternatives for the Si coefficient, -32 or -56 K/pct, showed a promising similarity to the second value. It was reported by Pevzner *et al.*^[41] and it may be concluded that the first value, which was reported by Papadimitriou *et al.*^[42] and showed only small scatter, had a strong systematic error. In line with the strategy of the present work, it may seem that one should accept -56 K/pct because it was obtained from a ternary alloy, rather than -61 K/pct based on quaternary information. However, instead of relying on the single ternary study by Pevzner *et al.*, which showed strong scatter, it was decided to accept -61 K/pct from Eq. [18]. On the other hand, the difference between -56 and -61 is small and the choice between them may not be critical. A family of WB_S lines for the six alloy contents were finally calculated from Eq. [18] with the Si coefficient of -61 K/pct and were plotted in Figure 12. They show reasonable agreement with the experimental information but also demonstrate the experimental scatter.

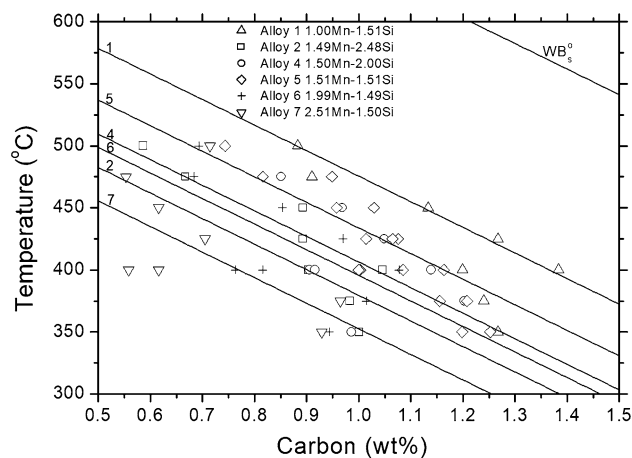


Fig. 12—Result of linear regression analysis of Fe-C-Mn-Si data in Fig. 11 to evaluate the coefficient for Si.

III. DISCUSSION

A. Consequences of Linear Equation

The present work is concerned with the rationalization of available experimental information on the critical condition of bainite formation, usually represented by B_S temperatures. That information has many times been described with linear equations and the coefficients have been derived directly by optimized fitting to a large amount of experimental B_S information and without any interpretation or input from physical understanding or modeling. They may thus be regarded as empirical equations and most of Eqs. [1] through [10] are examples. This was also accepted in the present work and the following empirical equation was obtained by combining the various coefficients obtained in Section IV.

$$WB_S(^{\circ}C) = 850 - 206C - 78Mn - 33Ni - 70Cr - 75Mo - 61Si \quad [19]$$

The aim of the present work was to evaluate the first two coefficients from information on binary Fe-C alloys. Historically, bainite was understood as an acicular microstructure of ferrite and cementite and could thus be regarded as a eutectoid microstructure but different from pearlite, which is a eutectoid with cells of ideally spherical shape. As a eutectoid microstructure, bainite should contain some fraction of cementite and could not be expected to form in alloys without carbon. Nevertheless, all empirical equations, including Eq. [19], predict B_S temperatures for zero carbon. Due to the linear kind of equation, B_S lines have been forced to start from zero carbon content and to be straight and parallel, independent of the alloy content. It is an important question whether available experimental information supports the use of a linear equation. That will be further discussed but there will be no effort to support a starting point from information on carbon-free iron. The first term in Eq. [19] was simply obtained by linear extrapolation from Fe-C alloys. For previous empirical equations, the first two coefficients were obtained by linear extrapolations for alloyed steels and the numerical values depended on the selected group of steels.

Figure 2 displayed an example which seemed to indicate that a B_S line can be far from straight which would imply that the linear equation is not adequate to represent the properties of the B_S function. In that case, there may not be any reason to extend the B_S line back to pure iron. However, a B_S line was now calculated from the new Eq. [19] for the average composition of the Cr steels and that is the dashed line which was already plotted in Figure 2. It appears to be a reasonable representation of three of the experimental data and it is suggested that the impression of a curved WB_S line is caused by major errors in two of the experimental points. It should be mentioned that these data were read from published TTT diagrams which cannot be expected to give very reliable results. It seems reasonable not to

accept a curved B_S line connecting the points for the Cr steels in Figure 2.

B. Combination of B_S and W_S Temperatures

The examination of published empirical equations for B_S temperatures revealed that they make very different predictions of B_S temperatures for the Fe-C alloys and the coefficient for an alloying element would not have the same value in different equations even if they had been fitted to the same B_S temperature. The problem was caused by the lack of B_S information on binary Fe-C alloys because there is an important range of temperature and carbon content where acicular ferrite has been classified as Widmanstätten ferrite. As already mentioned, the coefficients for carbon had thus been obtained by extrapolation from B_S information on alloyed steels. The aim of the present work was to derive an equation that could make reasonable predictions of B_S temperatures in Fe-C alloys. The coefficients for alloying elements could only then have proper physical meaning. This would be possible if one could accept information on the critical temperature of Widmanstätten ferrite, W_S , as if it concerns bainite. For a very long time, there has been an on-going debate regarding the basic relation between bainite and acicular ferrite but it will not be discussed in the present attempt to derive an empirical equation because it was done without any physical modeling. The use of W_S information will instead be justified by well-known microscopic observations. Bainite was originally regarded as an acicular eutectoid mixture of ferrite and cementite although ferrite was observed to be the leading phase. Usually, cementite appears almost simultaneously to acicular ferrite in lower bainite. However, primary plates or laths of ferrite have more time to develop in upper bainite and it will then be more evident that ferrite is the leading phase. In a transition range of temperature, it is just a matter of taste whether to define acicular ferrite, observed in the microscope, as the ferritic constituent of bainite or Widmanstätten ferrite and to denote the critical temperature of formation as B_S or W_S . In the latter case one has reported the results as W_S temperatures. Si as an alloying element in steel will delay the formation of cementite and allow primary ferrite to grow larger before cementite appears and the transition range will move to lower temperatures. For low temperatures, it is now fairly common to regard primary plates or laths of ferrite as bainite even when cementite is very much delayed or not even observed. The B_S temperature is then defined as the critical temperature of the formation of acicular ferrite, whether it appears together with cementite or not. This is a common procedure at low temperatures and in the present work it was extended to higher temperatures where some W_S temperatures have been reported for Fe-C alloys.

The information from plain carbon steels, which was displayed in Figure 2, contains both B_S and W_S information and is represented by a fairly straight line. In Figure 5, it is displayed together with more W_S information and it is seen that all of that could be reasonably represented by a straight line. This seems to support that

the combination of them to yield an empirical equation for prediction of critical values.

C. Test of New Equation

The new Eq. [19] was now tested by application to all the alloys with relevant experimental information compiled by Kolmskog *et al.*^[12] The results are presented in Figure 13 for T method data and in Figure 14 for C method data. The RMS of the deviations was 70 K and 76 K, respectively. When Kolmskog *et al.*^[12] tested the Steven and Haynes equation to the same data they obtained similar results, 72 K and 87 K, respectively. It seems that the new equation is only slightly better than the classical one in applications to alloyed steels in general. On the other hand, it is most probably superior for estimating the effects of individual alloying elements.

Available empirical equations for predicting B_S temperatures of steels are mainly based on information from some selection of commercial steels. Their predictions for Fe-C alloys reveal a considerable scatter, which is not surprising because the analytical descriptions of the binary Fe-C alloys are obtained by extrapolation from alloyed steels. Evidently, the differences in results depend on differences between the selected steels. The considerable scatter is very serious because the evaluated effects of alloying elements are given relative to the particular Fe-C description in each equation. To study alloying effects, it is thus necessary to have access to reliable information on Fe-C alloys and an equation for predicting B_S temperatures must contain a realistic description of binary Fe-C alloys which is presented in this paper.

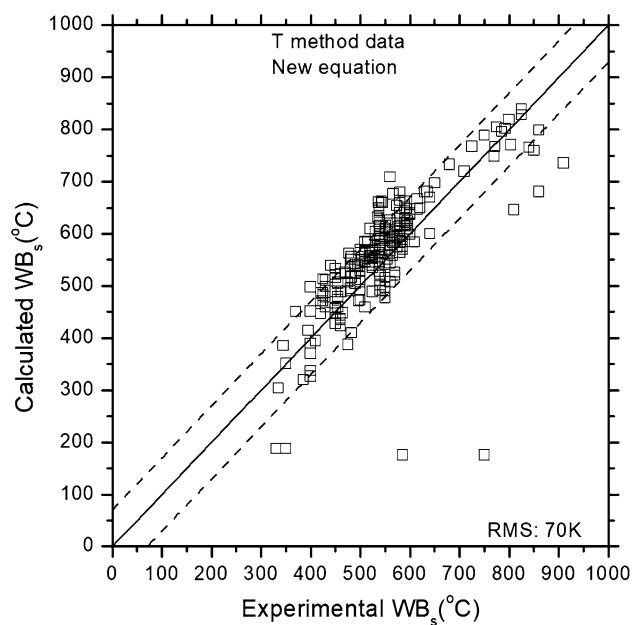


Fig. 13—Comparison of experimental W_S and B_S values from T methods and predicted values from new equation.

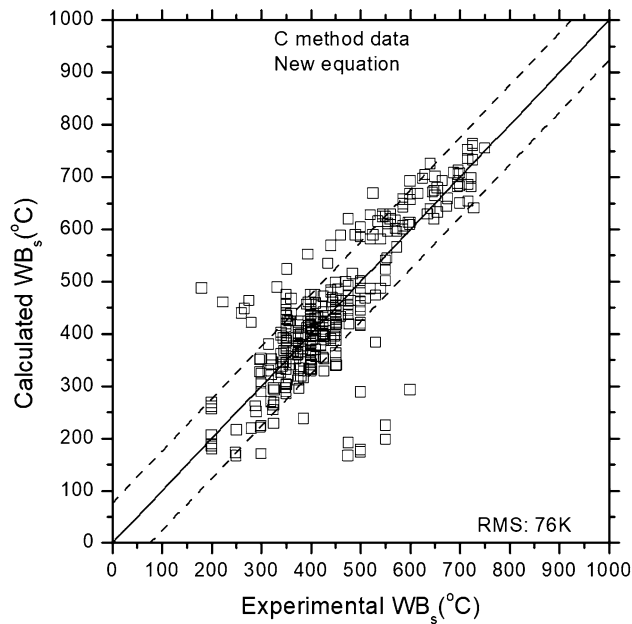


Fig. 14—Comparison of experimental W_S and B_S values from C methods and predicted values from new equation.

IV. SUMMARY

The effect of an alloying element on the B_S temperature can be expressed only relative to the behavior of binary Fe-C alloys. Such information is not easily available and it is generally accepted to study the alloying effect simultaneously for several elements by some kind of regression analysis of information from a considerable number of steels. An equation of the B_S temperature for Fe-C alloys is then obtained together with the effects of alloying elements relative to that equation. However, there have been several such efforts and a comparison of the predicted behavior of binary Fe-C alloys show wide scatter. As a consequence, the effects of individual alloying elements also show wide scatter. On the other hand, the result of such an effort may still be of some practical value if it was based on information from a limited range of alloyed steels and only applied to the same kind of steels. However, it would be difficult to analyze the physical factors behind the effects of individual elements.

The primary aim of the present work was to find an equation of the B_S temperature as function of the carbon content in binary Fe-C alloys but such information was found to be very meagre. The situation was improved by accepting that a single equation can be used for B_S and corresponding information for Widmanstätten ferrite, W_S . The resulting equation had some support from information on plain carbon steels, covering a wide range of temperatures.

Accepting the new equation for Fe-C alloys it was then possible to evaluate the effect of alloying elements one by one from ternary information. The best ternary information was found for Fe-C-Mn alloys. For Fe-C-Si alloys there were two conflicting sets of information and

additional information on Fe-C-Mn-Si alloys was employed in the final decision. Information on Fe-C-Cr, Fe-C-Mo and Fe-C-Ni alloys were also analyzed.

The results were summarized in a linear equation which was tested by application to a large survey of information from Kolmskog *et al.*^[12] It was divided in two groups concerning the measurement of the critical temperature where bainite starts forming, the T method, or the critical carbon content where bainite stops forming, the C method. Those authors used the classical equation by Steven and Haynes^[1] and found that there was not a significant difference between the two groups and most of the scatter was due to experimental scatter. That was now confirmed when the new equation was applied to the two groups of information. It seems that the new equation is only slightly better than the classical one in applications to alloyed steels in general. On the other hand, it is most probably superior for estimating the effects of individual alloying elements. Further improvement is called for and would depend critically on better information on binary Fe-C alloys but also on less experimental scatter for ternary alloys.

ACKNOWLEDGMENTS

Open access funding provided by Royal Institute of Technology. This study was conducted within the VINN Excellence center Hero-m and Competence center Hero-m 2 Innovation. The authors would like to thank Hossein Ehteshami for discussions. The authors acknowledge the financial support provided by VINNOVA, the Swedish Governmental Agency for Innovation Systems, Swedish industry and KTH Royal Institute of Technology.

OPEN ACCESS

This article is distributed under the terms of the Creative Commons Attribution 4.0 International License (<http://creativecommons.org/licenses/by/4.0/>), which permits unrestricted use, distribution, and reproduction in any medium, provided you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made.

REFERENCES

1. W. Steven and A.G. Haynes: *J. Iron Steel Inst.*, 1956, vol. 183, pp. 349–59.
2. S. Kang, S. Yoon, and S.-J. Lee: *ISIJ Int.*, 2014, vol. 54, pp. 997–99.
3. J.S. Kirkaldy and G. Venugopalan: in *Phase Transformations in Ferrous Alloys*, A.R. Marder and J.I. Goldstein, eds., TMS AIME, Warrendale, PA, 1984, p. 125.
4. M. Suehiro, T. Senuma, H. Yada, Y. Matsumura, and T. Ariyoshi: *Tetsu-to-Hagané*, 1987, vol. 73, pp. 1026–33.
5. R.L. Bodnar, T. Ohashi, and R.I. Jaffee: *Metall. Trans. A*, 1989, vol. 20A, pp. 1445–60.
6. J. Zhao and Z. Jin: *Mater. Sci. Technol.*, 1992, vol. 8, pp. 1004–10.

7. T. Kunitake and Y. Okada: *J. Iron Steel Japan*, 1998, vol. 84, pp. 137–41.
8. J.K. Lee, *Prediction of Tensile Deformation Behavior of Formable Hot Rolled steels*. POSCO Technical Research Laboratories Report, POSCO, Pohang, Korea, 1999.
9. J. Zhao, C. Liu, Y. Liu, and D.O. Northwood: *J. Mater. Sci.*, 2001, vol. 36, pp. 5045–56.
10. Y.K. Lee: *J. Mater. Sci.*, 2002, vol. 21, pp. 1253–55.
11. S.M.C. van Bohemen: *Mater. Sci. Techn.*, 2012, vol. 28, pp. 487–95.
12. P. Kolmskog, L. Leach, L. Höglund, M. Hillert and A. Borgenstam: *Comparison of Critical Temperature and Critical Carbon Content of Bainite Formation*. (to be published).
13. M. Hillert: *The Growth of Ferrite, Bainite and Martensite*, Report, Swedish Inst. Metal Research, 1960. Printed in Thermodynamics and Phase Transformations, The Selected Works of Mats Hillert, eds. J. Ågren, Y. Bréchet, C. Hutchinson, J. Philibert and G. Purdy, EDP Science, Les Ulis Cedex, France, 2006.
14. T. Lyman and A.R. Troiano: *Trans. ASM*, 1946, vol. 37, pp. 402–48.
15. J.O. Andersson, T. Helander, L. Höglund, P. Shi, and B. Sundman: *Calphad*, 2002, vol. 26, pp. 273–312.
16. Thermo-Calc: *TCFE8-TCS Steels/Fe-Alloys Database*, v8.0, 2015. <http://www.thermocalc.com>. Accessed 2017.
17. H.I. Aaronson: in *Decomposition of Austenite by Diffusional Processes*, V.F. Zackay and H.I. Aaronson, eds., Interscience, New York, 1962, pp. 387–546.
18. S.E. Johanson, Thesis, KTH, Stockholm 1959. Quoted in Ref. 13.
19. R.D. Townsend and J.S. Kirkaldy: *Trans. ASM*, 1968, vol. 61, pp. 605–19.
20. H.I. Aaronson, P.G. Boswell, and K.R. Kinsman: in *Mechanical Properties and Phase Transformations in Engineering Materials—Earl L. Parker Symposium on Structure Property Relationships* TMS-AIME, Warrendale, PA (1986), pp. 467–73.
21. Z.Q. Liu, G. Miyamoto, Z.G. Yang, and T. Furuhashi: *Acta Mater.*, 2013, vol. 61, pp. 3120–29.
22. Z.Q. Liu, G. Miyamoto, Z.G. Yang, C. Zhang, and T. Furuhashi: *Metall. Mater. Trans. A*, 2015, vol. 46A, pp. 1544–49.
23. A. Borgenstam, KTH Royal Institute of Technology, Stockholm, Sweden, *Unpublished Research*, 2005.
24. A. Hultgren: *Trans. ASM*, 1947, vol. 39, pp. 915–89.
25. M. Hillert: KTH, Royal Institute of Technology, Stockholm, Sweden, *Unpublished Research*, 1970.
26. S.K. Liu, L. Yang, D.G. Zhu, and J. Zhang: *Metall. Mater. Trans. A*, 1994, vol. 25A, pp. 1991–2000.
27. A. Hultgren: *Jernkont. Ann.*, 1951, vol. 135, pp. 403–94.
28. K. Tsuzaki, K. Fujiwara, and T. Maki: *Mater. Trans.*, 1991, vol. 8, pp. 667–78.
29. M. Umemoto and I. Tamura: *Acta Metall.*, 1986, vol. 34, pp. 2235–45.
30. A. Borgenstam and J.M. Ericsson: in *Solid-to-Solid Phase Transformations in Inorganic Materials 2005*, J.M. Howe, D.E. Laughlin, J.K. Lee, U. Dahmen, and W.A. Soffa, eds., TMS, Warrendale, 2005, pp. 105–10.
31. J.V. Bee, R.W.K. Honeycombe, and P.R. Howell: *Metall. Trans. A*, 1979, vol. 10A, pp. 1207–12.
32. H.K.D.H. Bhadeshia, in *Solid-to-Solid Phase Transformations 1981*, H.I. Aaronson, D.E. Laughlin, R.F. Sekerka, C.M. Wayman, eds., TMS-AIME, Pittsburgh, 1981, pp. 1041–48.
33. H. Goldenstein and H.I. Aaronson: *Metall. Trans. A*, 1990, vol. 21A, pp. 1465–78.
34. F.G. Berry and R.W.K. Honeycombe: *Metall. Trans.*, 1970, vol. 1, pp. 3279–86.
35. H. Tsubakino and H.I. Aaronson: *Metall. Trans. A*, 1987, vol. 18A, pp. 2047–60.
36. G.J. Shiflet and H.I. Aaronson: *Metall. Trans. A*, 1990, vol. 21A, pp. 1413–32.
37. W.T. Reynolds, Jr, F.Z. Li, C.K. Shui, and H.I. Aaronson: *Metall. Trans. A*, 1990, vol. 21A, pp. 1433–63.
38. W.T. Reynolds, Jr, S.K. Liu, F.Z. Li, S. Hartfield, and H.I. Aaronson: *Metall. Trans. A*, 1990, vol. 21A, pp. 1479–91.
39. K.M. Wu and M. Enomoto: *Scripta Mater.*, 2002, vol. 46, pp. 569–74.
40. M. Enomoto: *Scripta Mater.*, 2002, vol. 47, pp. 145–49.
41. L.M. Pevzner, T.D. Kubyushina, G.M. Rovenskii, and A.I. Samoilov: *Soviet Metallogr. Treat. Met.*, 1956, vol. 11, pp. 2–20.
42. G. Papadimitriou and J.M.R. Genin: *Mater. Res. Soc.*, 1983, vol. 21, pp. 747–74.
43. N. Usui, K. Sugimoto, E. Nishida, M. Kobayashi, and S. Hashimoto: *CAMP-ISIJ*, 1990, vol. 3, pp. 2013–14.
44. K. Sugimoto, J. Sakaguchi, T. Iida, and T. Kashima: *ISIJ Int.*, 2000, vol. 40, pp. 920–26.
45. K. Sugimoto, K. Nakano, S. Song, and T. Kashima: *ISIJ Int.*, 2002, vol. 42, pp. 450–55.

Publisher's Note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.