ORIGINAL ARTICLE



# Bioavailability of heavy metals in soil of the Tieguanyin tea garden, southeastern China

Jingwei Sun<sup>1,2,4</sup> · Ruilian Yu<sup>1</sup> · Gongren Hu<sup>1</sup> · Songhe Jiang<sup>1</sup> · Yunfeng Zhang<sup>1,2,4</sup> · Xiaoming Wang<sup>3</sup>

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Abstract The bioavailability of 22 heavy metals was investigated at 19 sampling sites in Tieguanyin tea garden in Anxi County, Fujian Province, southeastern China. Heavy metal concentrations were determined by inductively coupled plasma-mass spectrometry (ICP-MS) and evaluated by geo-accumulation index  $(I_{qeo})$ . Dilute nitric acid extraction was used to evaluate biological activity. Cu, Pb, and Cd were highly bioavailable and most easily absorbed by tea trees. Heavy metal bioavailability in the surface soil was as the ratio of the effective state to the total amount. Cd had the highest  $I_{geo}$  values, and the respective samples and sites were classified as moderately/strongly contaminated. Cd element is considered the main factor of heavy metal pollution in the tea garden in Anxi. The other heavy metals studied were present in lower concentrations; thus, the samples were classified as uncontaminated or slightly contaminated.

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Gongren Hu hugongren@163.com

- <sup>1</sup> College of Chemical Engineering, Huaqiao University, Xiamen 361021, China
- <sup>2</sup> School of Resources and Environmental Science, Quanzhou Normal University, Quanzhou 362000, China
- <sup>3</sup> Analytical Laboratory of Beijing Research Institute of Uranium Geology, Beijing 100029, China
- <sup>4</sup> Fujian Provincial Key Laboratory of Ecology-Toxicological Effects and Control for Emerging Contaminants, Putian, China

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### **1** Introduction

Tieguanyin tea, produced mainly in Anxi County, Fujian Province, China, is a type of Oolong tea. As one of China's ten most famous teas, this type is in high demand by consumers owing to its pleasant aroma and high contents of amino acids, vitamins, minerals, tea polyphenols, and alkaloids in addition to a variety of other nutritional ingredients. The tea industry plays an important role in Quanzhou's economy, culture, and tourism. However, the considerable economic development in recent years has been tainted by wastewater irrigation, atmospheric deposition, industrial waste emissions, and the application of pesticides and chemical fertilizers that have caused soil pollution. Heavy metal pollution, in particular, is becoming an increasingly serious issue. High concentrations of heavy metals in the soil can endanger human health through the food chain, and pose a substantial threat to the ecosystem.

Although it is an important indicator of soil pollution, total heavy metal content is only part of the story; heavy metal migration in soil is complicated by partial absorption of the metals. Therefore, content does not correspond to bioavailability of heavy metals (Alloway 2013; Maderova and Paton 2013). An increasing amount of recent environmental research has been devoted to methods for determining heavy metal bioavailability in soils.

It is widely accepted that the bioavailability of heavy metals is the amount absorbed by organisms at different levels of toxicity. Many studies have shown that although the total content of heavy metals in soil can be used to evaluate pollution status, the bioavailability of heavy metals in a specific environment is closely related to its peculiarities (Gao et al. 2014). At present, single extraction and sequential extraction are the most commonly used methods for extracting heavy metals from soil. In the single extraction protocol, which is simple and effective, chemical reagents such as dilute nitric acid and dilute hydrochloric acid are used to evaluate the mechanism of heavy metal pollution. Many countries have adopted the single extraction method for evaluating heavy metals, whereas others employ multistage extraction. The European Community Bureau of Reference (BCR) sequential extraction method, which is characterized by adequate stability, reproducibility, and precision, has also been extensively applied (Mossop and Davidson 2003).

In this study, we determined the bioavailability of 22 heavy metals at 19 sampling sites in Tieguanyin tea garden, Anxi County. The geo-accumulation index ( $I_{geo}$ ) was used to evaluate the total heavy metal content, and bioavailability was calculated as the ratio of dilute nitric acid single extraction to the total amount of heavy metals. The aim of this study was to provide a scientific basis for effective prevention and control of heavy metal pollution in Tieguanyin tea garden.

## 2 Materials and methods

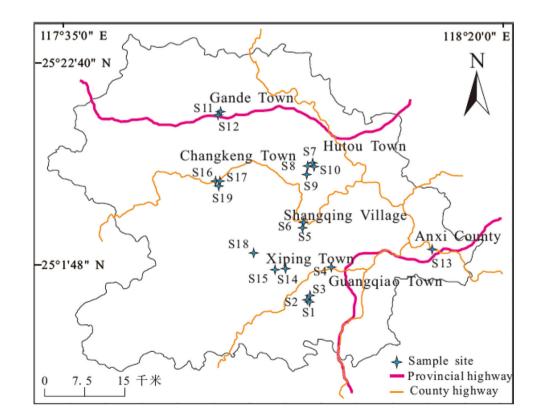
## 2.1 Sampling and preparation

Nineteen soil samples from 0 to 20 cm depth were collected from 19 sites in Tieguanyin tea garden in June and October 2016 (Fig. 1). All samples were carefully stored in clean plastic vessels prior to processing and analysis and were labelled with their sampling location, date, number, and the name of the sample collector. In the laboratory, the soil samples were air-dried in a controlled clean environment, and all litter, roots, animal residues, and other debris were removed. The soil was then spread into a thin layer and repeatedly turned over to accelerate drying. The samples were then ground with an agate mortar and pestle and sieved with a 200-mesh nylon sieve. The material under the sieve was stored in sealed polyethylene bags for future analysis, and the sieve materials were returned to the original bag.

## 2.2 Sample analysis

## 2.2.1 Measurement of total heavy metal content

All soil samples were analyzed by microwave digestion with HCl–HNO<sub>3</sub>–HF (Wysocka and Vassileva 2016). Fe and Mn were analyzed by atomic absorption spectrometer



**Fig. 1** Sites for surface soil sampling in the studied tea garden in Anxi, China

Table 1 Descriptive statistics of heavy metals in soil samples (mg/kg)

|                                   | -     |       | -     |       |       |       |       |       |       |       |       |
|-----------------------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| Metals                            | Li    | Be    | Sc    | V     | Cr    | Ni    | Cu    | Zn    | Rb    | Мо    | Cd    |
| Minimum                           | 10.80 | 0.486 | 7.810 | 31.80 | 11.80 | 4.280 | 7.580 | 54.90 | 35.10 | 0.708 | 0.107 |
| Maximum                           | 38.10 | 5.940 | 21.10 | 127.0 | 48.80 | 18.30 | 26.50 | 639.0 | 194.0 | 4.990 | 0.749 |
| Mean                              | 20.71 | 1.662 | 13.23 | 67.44 | 23.82 | 7.815 | 15.25 | 129.8 | 109.0 | 2.075 | 0.358 |
| Background content (Liu 1995)     | 20.80 | 1.220 | 8.420 | 46.50 | 24.70 | 9.000 | 9.700 | 51.50 | 160.0 | 10.20 | 0.047 |
| Ratio of mean to background value | 1.00  | 1.36  | 1.57  | 1.45  | 0.96  | 0.87  | 1.57  | 2.52  | 0.68  | 0.20  | 7.62  |
| Metals                            | Sb    | Cs    | Ba    | W     | Tl    | Pb    | Bi    | Th    | U     | Sr    | In    |
| Minimum                           | 0.343 | 3.830 | 69.40 | 1.360 | 0.474 | 22.60 | 0.403 | 17.50 | 4.730 | 12.00 | 0.052 |
| Maximum                           | 2.930 | 14.70 | 618.0 | 9.770 | 1.420 | 82.10 | 2.280 | 61.60 | 10.40 | 406.0 | 0.953 |
| Mean                              | 0.810 | 6.749 | 313.5 | 4.344 | 0.883 | 46.82 | 0.970 | 30.99 | 6.699 | 53.42 | 0.185 |
| D 1 1 (T 1 1005)                  | 0.450 | 4.130 | 423.0 | 3.530 | 0.683 | 46.80 | 0.610 | 27.28 | 4.660 | 34.00 | 0.070 |
| Background content (Liu 1995)     | 0.450 | 4.150 | 425.0 | 5.550 | 0.005 | 40.00 | 0.010 | 27.20 | 4.000 | 54.00 | 0.070 |

Table 2 Classes of Igeo values

| I <sub>geo</sub> | $I_{geo} \le 0$                      | $0 < I_{geo} < 1$       | $1 < I_{geo} < 3$                | $3 < I_{geo} < 5$        | $I_{geo} > 5$          |
|------------------|--------------------------------------|-------------------------|----------------------------------|--------------------------|------------------------|
| Class            | 1                                    | 2                       | 3                                | 4                        | 5                      |
| Soil<br>quality  | Uncontaminated/slightly contaminated | Moderately contaminated | Moderately/strongly contaminated | Strongly<br>contaminated | Extremely contaminated |

(AAS; TAS-986, Beijing Purkinje General Instrument Co., Ltd.) and other metals were analyzed by inductively coupled plasma-mass spectrometry (ICP-MS). The following procedure was applied for preparing the blend solutions. About 0.1 g of soil sample and adequate volumes of the spike solutions—4 mL of HNO<sub>3</sub>, 0.1 mL of HCl, and 1 mL of HF—were subsequently added to each Teflon vessel, and sample digestion realized in a closed microwave system. Rhodium was used as the internal standard. The sample solutions and reagent blanks were analyzed for metals by ICP-MS (ELAN9000; PerkinElmer, Massachusetts, USA) at the Chinese Academy of Sciences. Background corrections and matrix interference were monitored throughout the analyses. All experiments involving the soil samples were conducted in duplicate. A

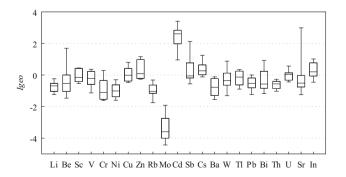


Fig. 2 Box-plots of the geoaccumulation index of 22 heavy metals in the soil

parallel sample analysis of soil GSS-7 series standard (GBW07407) showed 94%-106% recovery of each element.

#### 2.2.2 Measurement of heavy metal speciation

Briefly, 1 g of soil was weighed in a 50-mL plastic centrifuge tube and mixed with 15 mL of 0.5 mol/L HNO<sub>3</sub>. Next, the sample was oscillated at 220 rpm for 24 h and then centrifuged at 4000 rpm for 10 min. The constant volume was adjusted to 50 mL and placed in a refrigerator at 4 °C. The concentrations of heavy metals were determined by ICP-MS.

#### **3** Results and discussion

#### 3.1 Total heavy metal levels

Descriptive statistics of heavy metals in soil samples in Tieguanyin tea garden are shown in Table 1. The ratios of mean to background values greater than 1.00 were Cd (7.62) > In (2.65) > Zn (2.52) > Sb (1.8) > Cs (1.63) > Bi (1.59) > Sr (1.57) = Cu (1.57) = Sc (1.57) > V (1.45) > U (1.44) > Be (1.36) > Tl (1.29) > W (1.23) > Th (1.14). The geological accumulation index method was used for further evaluation.

In 1969, Muller (1969) proposed the utilization of  $I_{geo}$  to assess the pollution of heavy metals in sediments;

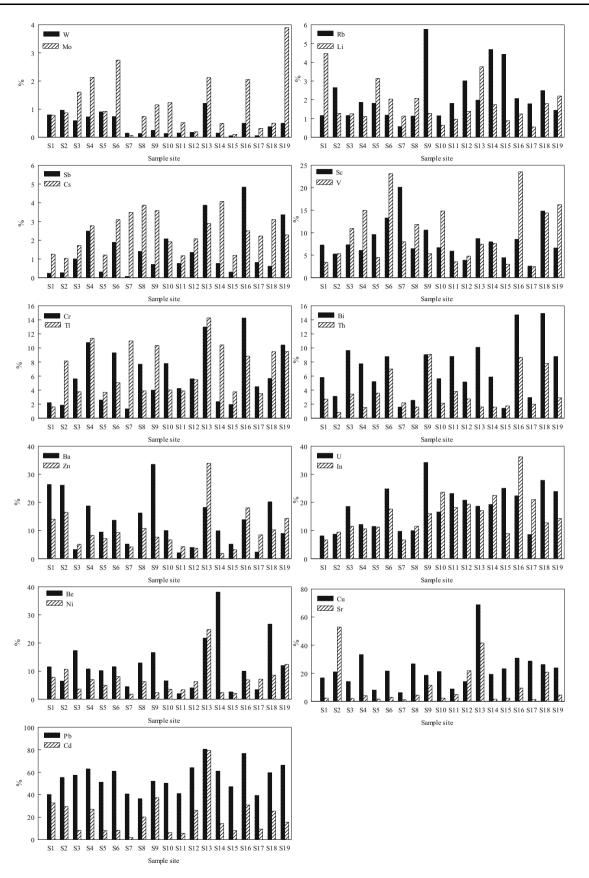


Fig. 3 The ratios of HNO<sub>3</sub>-extractable contents to total contents for heavy metals

subsequently, this method was used to determine soil pollution (Loska and Wiechula 2003). Here,  $I_{geo}$  is calculated by the formula:

$$I_{geo} = log_2[C_n/(1.5 \cdot B_n)]$$

where  $C_n$  is the concentration of a trace element in the soil,  $B_n$  is the geochemical background content (Liu 1995), and 1.5 is the background matrix correction factor owing to lithogenic effects.  $I_{geo}$  was classified into five grades, as shown in Table 2.

The results of  $I_{geo}$  are shown in Fig. 2. Most  $I_{geo}$  classes were comprised of uncontaminated or slightly contaminated soils, with a few moderately contaminated by heavy metals. Overall, Cd had the highest  $I_{geo}$  value, classifying as moderately/strongly contaminated. In previous studies, P fertilizers have been considered the main source of Cd pollution (Gray et al. 1999).

In fact, the various forms of heavy metals present in the acid-extractable and residual phases have different mobility and toxicity (Gao et al. 2014). In this study, extraction of all soil samples was performed with 0.5 mol/L HNO<sub>3</sub> (Zhang et al. 2015), and bioavailability based on the results.

#### 3.2 Bioavailability of heavy metals

The bioavailability of heavy metals can be expressed as the rate of extraction or activation—the ratio of the effective state of heavy metals to the total amount (Hu et al. 2013). Heavy metal bioavailability was calculated as:

 $A_c(\%) = A_{ci}/B_{ci} \times 100\%$ 

where  $A_c$  is the bioavailability of a given trace element in the soil,  $A_{ci}$  the HNO<sub>3</sub>-extractable content of the trace element, and  $B_{ci}$  the total content of the trace element. Higher A<sub>c</sub> indicates increased bioavailability of heavy metals in the ecological environment, which leads to more pronounced negative effects on the ecosystem (Wu et al. 2016). As shown in Fig. 3, the ratios of W, Mo, Rb, Li, Sb, Cs, Cr, Tl, Bi, Th, Sc, and V were less than 25%, indicating that their bioavailability was low. Low bioavailability suggests low activity and thus reduced absorption by the roots of tea plants. The proportions of acid-extractable concentrations of certain heavy metal elements such as Ba, Zn, U, In, Be, and Ni were close to 40% at some sites, indicating the presence of biological activity. The ratios of acid-extracted Cu, Pb, and Cd were higher at the S13 sampling site, and the proportion of the acid-extracted forms of the three heavy metals was almost 80%. These results indicate easy absorption of the heavy metals by the tea plants.

#### **4** Conclusions

Cd had the highest  $I_{geo}$ ; thus, the respective sampling site was classified as moderately/strongly contaminated. This heavy metal was identified as the main pollution factor in the tea garden in Anxi. Fertilizer containing P was the main source of Cd contamination; therefore, its application should be justified and carefully controlled. The dilute nitric acid extraction method was employed to evaluate biological availability. Cu, Pb, and Cd showed high bioavailability and were easily absorbed by tea trees. Although the total amount of Pb and Cu did not indicate severe pollution, their high biological activities warrant considerable attention to their concentrations.

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