



# Effect of carbon rate and type amended with ammonium or nitrate on nitrous oxide emissions in a strong ammonia oxidation soil

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## Abstract

**Purpose** The work aimed to (1) better understand how C rate and type affecting N<sub>2</sub>O emissions when combined application with different N forms in a strong ammonia oxidation soil, (2) further explore the tradeoff and mechanism of C availability to N transformations, and (3) provide evidence for making targeted N<sub>2</sub>O mitigation measures in these kinds of agricultural soils.

**Materials and methods** The soil was collected from a typical farmer's field in Quzhou, Hebei province, China, classified as calcareous Cambisols (FAO classification system) and characterized with high pH (7.72), low organic C (9.1 g kg<sup>-1</sup>), and a strong ammonia oxidation potential. Two microcosm incubations were conducted in laboratory under 20 °C and 70% WFPS (water-filled pore space), and the soils were amended with different glucose rates (0, 0.5, 1.0, 2.0, 4.0 mg C kg<sup>-1</sup>) in Exp I, and with different C sources (glucose, pectin, starch, cellulose, lignin, straw) in Exp II, meanwhile with NH<sub>4</sub><sup>+</sup> or NO<sub>3</sub><sup>-</sup>-based fertilizer addition. 480 g (equivalent to oven-dried weight) of pre-incubated soil was placed in a glass jar (1165 mL volume) and adjusted to a field bulk density of 1.34 g cm<sup>-3</sup>. All the glass jars were placed randomly and incubated for 15 days under aerobic condition.

**Results and discussion** Alone application of NH<sub>4</sub><sup>+</sup> released more N<sub>2</sub>O (peak at 16.27 μg N kg<sup>-1</sup>) than alone NO<sub>3</sub><sup>-</sup> application. However, combined application of glucose with NO<sub>3</sub><sup>-</sup> could emit more N<sub>2</sub>O than when combined with NH<sub>4</sub><sup>+</sup>, and the highest N<sub>2</sub>O peak value was found in 0.5 g kg<sup>-1</sup> of glucose. When pectin is combined with NH<sub>4</sub><sup>+</sup>, resulted in higher accumulated N<sub>2</sub>O emissions than other C sources, while applying straw and cellulose had negligible effects on N<sub>2</sub>O emissions but enhanced CO<sub>2</sub> emissions under the incubation conditions. Glucose and pectin were identified as two labile C sources, significantly enhanced N<sub>2</sub>O emissions. As expected, C addition could stimulate mineral N immobilization.

**Conclusions** Combined application of C and N enhanced N<sub>2</sub>O and CO<sub>2</sub> emissions albeit to different extents. A higher accumulated N<sub>2</sub>O emission was found when C/N ratio was lower (12.5). The stimulating effect of glucose and pectin as labile C sources with N addition on N<sub>2</sub>O production should be considered, especially in soils with high pH, low organic C, and a strong ammonia oxidation potential. Therefore, it may have the potential to mitigate N<sub>2</sub>O emissions from the field when avoiding applying both N and C especially rich in glucose and pectin together.

**Keywords** Ammonium · Carbon rate and type · C/N ratio · Nitrate · Nitrous oxide emission

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## 1 Introduction

Nitrous oxide (N<sub>2</sub>O) is a potent greenhouse gas (GHG), significantly contributing to a global warming, with about 300-fold higher global warming potential than carbon dioxide (CO<sub>2</sub>) (IPCC 2007), and involved in the destruction of the protective ozone layer (Ravishankara et al. 2009). Nearly 65% of the total anthropogenic N<sub>2</sub>O was emitted from agricultural section globally (IPCC 2007), and in China, up to 75% of N<sub>2</sub>O was emitted from this section (UNEP 2013). Agricultural soils are the largest source of N<sub>2</sub>O production

due to the large use of synthetic nitrogen (N) fertilizers and manure (IPCC 2013). Several biological processes involved in the soil N cycling contribute to  $N_2O$  emissions, including autotrophic/heterotrophic nitrification, heterotrophic denitrification, nitrifier denitrification, and dissimilatory nitrate reduction to ammonium (DNRA) and anammox, and nitrification and denitrification are considered as the major  $N_2O$  sources in soils (Butterbach-Bahi et al. Butterbach-Bahl et al. 2013). Nitrification is the main process of  $N_2O$  production under aerobic, high ammonia, low nitrate concentration conditions, while heterotrophic denitrification-related  $N_2O$  mainly occurs under oxygen-limited and high nitrate conditions (Wrage et al. 2001, 2005; Kool et al. 2010; Baggs 2011; Butterbach-Bahl et al. 2013; Pilegaard 2013; Zhu et al. 2013). Heterotrophic nitrification might also play an important role in pasture soils, and readily degradable carbon (C) addition could stimulate heterotrophic nitrification and thereby promote  $N_2O$  emissions (Lan et al. 2019).

The global consumption of fertilizer N was projected to increase at a rate of 1.4% annually from 2014 to 2018, because of meeting the growing population demand for the greater food production (FAO 2015). The extensive use of fertilizer N would greatly contribute to the apparent elevation in atmospheric  $N_2O$  concentrations (Shcherbak et al. 2014). An emerging body of research shows that N fertilizer type is the key affecting factor on  $N_2O$  emissions under certain  $O_2$  and moisture condition. Greater  $N_2O$  emissions were reported from nitrate ( $NO_3^-$ )-based fertilizers than ammonium ( $NH_4^+$ )-based fertilizers with the high moisture and C soils (Dobbie and Smith 2003; Smith et al. 2012). However, the summary of available measurements from fertilized fields showed that ( $NH_4^+$ )-based fertilizers had greater  $N_2O$  emissions than that of ( $NO_3^-$ )-based fertilizers (Bouwman et al. 2002; Li et al. 2013). It is still not clear of the effects of N forms on  $N_2O$  emissions in the soils characterized with high pH, low organic C, and a strong ammonia oxidation potential, especially simultaneously combined with different C components.

Previous studies have shown that the application of readily degradable C, such as glucose, could inhibit nitrification-related  $N_2O$  by reducing  $NH_4^+$  availability via immobilization (Freppaz et al. 2007; Ma et al. 2016), which also depended on the availability and application rate of external C input (Granli and Bøckman 1994). But  $N_2O$  production from the soil could also be stimulated by the application of readily degradable C possibly due to increasing denitrification (Wrage et al. 2001). Carbon could provide not only energy for soil microbe but also electron acceptor for denitrification (Yokoyama et al. 1992; Rivett et al. 2008; Zhu et al. 2015). The contrasting effects of readily degradable C on nitrification and denitrification might lead to discrepant results on the effect of C application on  $N_2O$  emissions from the soil. The effects of different C components (with different degradable extents) on  $N_2O$

emission and N transformation under aerobic condition, especially simultaneously combined applying with ( $NH_4^+$ ) or ( $NO_3^-$ )-based fertilizer, are incompletely understood, and future research is urgently needed. To date, studies that combined examine the effect of C and N forms on  $N_2O$  emissions are also scarce. How the interaction of C availability and N forms affect  $N_2O$  emissions is still unclear.

In the present study, different forms and doses of C and N were applied to a strong ammonia oxidation soil in order to investigate the interaction of C availability and N forms affecting  $N_2O$  emissions and N transformation. We hypothesized that labile C with N addition would enhance  $N_2O$  emissions, and higher C/N ratio could decrease  $N_2O$  emissions due to enhancing N immobilization. The findings from this study would help us to develop better C and N managements for mitigating  $N_2O$  emissions in field conditions.

## 2 Materials and methods

### 2.1 Site and properties of soil samples

Soil samples were taken from a typical farmer's field where carried out winter wheat-summer maize double cropping system in the last many years at Quzhou research station (36.87°N, 115.02°E), Hebei province, China. The site is under warm temperate and semi-arid monsoon climatic conditions. The long-term (from 1980 to 2010) annual average temperature was 13 °C, and the annual average precipitation was 494 mm. Most rainfall occurred during summer (July to September) accounting for 60% of the total rainfall in a year. The soil is a typical calcareous fluvo-aquic soil belonging to calcareous Cambisols (FAO classification system), which has low  $NH_4^+$  and dissolved C but with high  $NO_3^-$  and pH and is characterized by a high nitrification potential (Wan et al. 2009; Ju et al. 2011; Ju and Zhang 2017). Ten soil cores (0–20 cm) were randomly taken from the field (about half hectare) after summer maize harvest in 2011 to make a composited sample; the sample was stored in an ice box before returning to the laboratory within half a day. The sampled soil was homogenized, sieved (2 mm), air-dried, and stored at room temperature (20 °C) before the incubation commencement.

The soil was light loam containing 39.34% sand, 58.25% silt, and 2.41% clay, with a field bulk density 1.34 g  $cm^{-3}$ , a pH of 7.72 (1:2.5, soil/water), a soil organic carbon concentration of 9.5 g  $kg^{-1}$ , a dissolved organic C of 45.0 mg  $kg^{-1}$ , a soil total N concentration of 0.7 g  $kg^{-1}$ , and  $NH_4^+$  and  $NO_3^-$  concentrations of 0.38 and 31.82 mg  $kg^{-1}$ , respectively. The potential nitrification rate was 204.5 mg  $NO_3^-$ -N  $kg^{-1}$   $day^{-1}$  and measured by the method of Hart et al. (1994).

## 2.2 Experimental design

Two laboratory incubation experiments were conducted. Experiment I aimed to test the effects of different C application rates with N forms on N<sub>2</sub>O emissions and N transformation. It was conducted with two N forms (ammonium sulfate, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and calcium nitrate, Ca(NO<sub>3</sub>)<sub>2</sub>) and four levels of C (glucose) addition. Apart from CK treatment, 40 mg N kg<sup>-1</sup> as NH<sub>4</sub><sup>+</sup> or NO<sub>3</sub><sup>-</sup>, was applied with or without glucose for each treatment. The four glucose rates were 0.5, 1.0, 2.0, and 4.0 g C kg<sup>-1</sup>, respectively, and the corresponding C/N ratios of the applied C and N were 12.5, 25, 50, and 100, respectively. Each treatment had three replicates. Therefore, eight treatments plus three controls (no N or C addition and N addition only) were carried out in the Experiment I as follows:

1. CK (no N and C addition, CK)
2. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (NH<sub>4</sub><sup>+</sup>)
3. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 0.5 g C kg<sup>-1</sup> glucose (NH<sub>4</sub><sup>+</sup> + Glu.0.5)
4. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 1.0 g C kg<sup>-1</sup> glucose (NH<sub>4</sub><sup>+</sup> + Glu.1.0)
5. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 2.0 g C kg<sup>-1</sup> glucose (NH<sub>4</sub><sup>+</sup> + Glu.2.0)
6. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 4.0 g C kg<sup>-1</sup> glucose (NH<sub>4</sub><sup>+</sup> + Glu.4.0)
7. Ca(NO<sub>3</sub>)<sub>2</sub> (NO<sub>3</sub><sup>-</sup>)
8. Ca(NO<sub>3</sub>)<sub>2</sub> + 0.5 g C kg<sup>-1</sup> glucose (NO<sub>3</sub><sup>-</sup> + Glu.0.5)
9. Ca(NO<sub>3</sub>)<sub>2</sub> + 1.0 g C kg<sup>-1</sup> glucose (NO<sub>3</sub><sup>-</sup> + Glu.1.0)
10. Ca(NO<sub>3</sub>)<sub>2</sub> + 2.0 g C kg<sup>-1</sup> glucose (NO<sub>3</sub><sup>-</sup> + Glu.2.0)
11. Ca(NO<sub>3</sub>)<sub>2</sub> + 4.0 g C kg<sup>-1</sup> glucose (NO<sub>3</sub><sup>-</sup> + Glu.4.0)

Experiment II aimed to test the effects of different C types with N addition on N<sub>2</sub>O emissions and N transformation. Six C sources (glucose, pectin, starch, cellulose, lignin and wheat straw) were used with C rate of 0.5 g C kg<sup>-1</sup>. N rate was the same as experiment I as 40 mg N kg<sup>-1</sup> of NH<sub>4</sub><sup>+</sup> or NO<sub>3</sub><sup>-</sup>, respectively. Each treatment had three replicates. Therefore, 12 treatments plus three controls (no N or C addition and N addition only) were carried out in experiment II as follows:

1. CK (no N and C addition, CK)
2. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (NH<sub>4</sub><sup>+</sup>)
3. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 0.5 g C kg<sup>-1</sup> glucose (NH<sub>4</sub><sup>+</sup> + glue)
4. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 0.5 g C kg<sup>-1</sup> pectin (NH<sub>4</sub><sup>+</sup> + pectin)
5. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 0.5 g C kg<sup>-1</sup> starch (NH<sub>4</sub><sup>+</sup> + starch)
6. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 0.5 g C kg<sup>-1</sup> cellulose (NH<sub>4</sub><sup>+</sup> + cellulose)
7. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 0.5 g C kg<sup>-1</sup> lignin (NH<sub>4</sub><sup>+</sup> + lignin)
8. (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> + 0.5 g C kg<sup>-1</sup> straw (NH<sub>4</sub><sup>+</sup> + straw)
9. Ca(NO<sub>3</sub>)<sub>2</sub> (NO<sub>3</sub><sup>-</sup>)
10. Ca(NO<sub>3</sub>)<sub>2</sub> + 0.5 g C kg<sup>-1</sup> glucose (NO<sub>3</sub><sup>-</sup> + glu)
11. Ca(NO<sub>3</sub>)<sub>2</sub> + 0.5 g C kg<sup>-1</sup> pectin (NO<sub>3</sub><sup>-</sup> + pectin)
12. Ca(NO<sub>3</sub>)<sub>2</sub> + 0.5 g C kg<sup>-1</sup> starch (NO<sub>3</sub><sup>-</sup> + starch)
13. Ca(NO<sub>3</sub>)<sub>2</sub> + 0.5 g C kg<sup>-1</sup> cellulose (NO<sub>3</sub><sup>-</sup> + cellulose)
14. Ca(NO<sub>3</sub>)<sub>2</sub> + 0.5 g C kg<sup>-1</sup> lignin (NO<sub>3</sub><sup>-</sup> + lignin)
15. Ca(NO<sub>3</sub>)<sub>2</sub> + 0.5 g C kg<sup>-1</sup> straw (NO<sub>3</sub><sup>-</sup> + straw)

## 2.3 Procedure of the incubation experiments

Fresh soil samples were pre-incubated under 50% WFPS (water-filled pore space) at 20 °C for a week in the dark to stabilize the microbial community and avoid fluctuations of soil respiration in the dry-wet process. A total of 480 g (equivalent to oven-dried weight) of pre-incubated soil was placed in a glass jar (1165 mL volume with 16.3 heights and 9.5 inner diameters) and adjusted to a field bulk density of 1.34 g cm<sup>-3</sup>. The C sources like pectin, starch, cellulose, lignin, and straw were mixed directly with the soils, and the straw was dried and ground (passed through 1 mm mesh) before addition. The solutions of NH<sub>4</sub><sup>+</sup> or NO<sub>3</sub><sup>-</sup> for each treatment were applied evenly to the soil in the glass jar using a syringe to achieve the target soil moisture of 70% WFPS, which was the moisture level as maximum N<sub>2</sub>O emission from the previous studies (Huang et al. 2014). Each glass jar was covered with parafilm, which allowed air exchange but prevented water loss during the incubation. Soil moisture was replenished every 4 days. All the glass jars were placed randomly and incubated for 15 days at 20 °C in the dark. There were 33 glass jars (11 treatments \* 3 replicates) for experiment I and 45 jars (15 treatments \* 3 replicates) for experiment II for gas samplings and the last destructive soil sampling, respectively. Totally, 132 glass jars (11 treatments \* 3 replicates \* 4 sampling times) for experiment I, and 135 glass jars for experiment II (15 treatments \* 3 replicates \* 3 sampling times) were incubated for destructive soil samplings and analysis.

## 2.4 Gas sampling and analysis

Gas samples (20 ml) were collected on days 1, 2, 3, 5, 8, 11, and 15 for N<sub>2</sub>O and CO<sub>2</sub> concentrations analysis. Before gas sampling, parafilm was removed, and the air was allowed to exchange with the ambient air for 5 min. Subsequently, the jars were sealed with rubber seals equipped with a three-way valve, and this time point was defined as the zero point. Gas samples were collected at 0, 10, and 20 min with syringes. Samples were analyzed by a modified gas chromatography (Agilent 6820, Santa Clara, USA) with an ECD detector for N<sub>2</sub>O and TCD detector for CO<sub>2</sub> concentration (Zheng et al. 2008) respectively. The ECD detector was fitted with a <sup>63</sup>Ni and operated at 350 °C. High-purity dinitrogen (99.999%) was used as the carrier gas for N<sub>2</sub>O analysis. Ten percent of CO<sub>2</sub> in pure N<sub>2</sub> was used as a make-up gas for ECD. The GC was calibrated using known concentrations of mixed gas (0.354 ppm N<sub>2</sub>O in pure N<sub>2</sub>). Air conditioning was used in the laboratory during measurements to minimize the tendency of the ECD to drift with changing temperature.

## 2.5 Soil sampling and analysis

Destructive soil sampling was carried out for soil mineral N analysis on days 0, 3, 5, and 15 for experiment I, and on days 0, 3, and 15 for experiment II. Each soil sample was extracted with 1 M KCl (soil: 1 M KCl = 1:5) to determine the concentrations of  $\text{NH}_4^+$  and  $\text{NO}_3^-$  using a continuous flow analyzer (AA3, Seal Analytical, Norderstedt, Germany).

## 2.6 Calculations and statistical analysis

$\text{N}_2\text{O}$  and  $\text{CO}_2$  fluxes were calculated according to the following equation:

$$F = \frac{273}{273 + T} \times M \times 10^{-3} \times 60 \times 24 \times \frac{V}{22.4} \times \frac{dc}{dt} / 0.48$$

where  $F$  is the  $\text{N}_2\text{O}$  ( $\mu\text{g N}_2\text{O-N kg}^{-1} \text{ day}^{-1}$ ) or  $\text{CO}_2$  ( $\text{mg CO}_2\text{-C kg}^{-1} \text{ day}^{-1}$ ) emission rate,  $T$  ( $^{\circ}\text{C}$ ) is the air temperature in the incubation glass jar,  $M$  is the molecular weight of N in molecular  $\text{N}_2\text{O}$  ( $28 \text{ g mol}^{-1}$ ) or molecular weight of C in molecular ( $\text{CO}_2 \text{ g mol}^{-1}$ ), 60 (min) is the number of minutes in an hour, 24 is the number of hours in a day,  $V$  (L) is the volume of the headspace,  $c$  is the concentration of  $\text{N}_2\text{O}$  ( $\text{nL L}^{-1}$ ) or  $\text{CO}_2$  ( $\mu\text{L L}^{-1}$ ),  $t$  (min) is the time from jar sealing,  $dc/dt$  is the change in  $\text{N}_2\text{O}$  ( $\text{nL L}^{-1} \text{ min}^{-1}$ ) or  $\text{CO}_2$  ( $\mu\text{L L}^{-1} \text{ min}^{-1}$ ) concentration per unit of time, 22.4 ( $\text{L mol}^{-1}$ ) is the molecular volume at 101 kPa and 273 K, and 0.48 (kg) is the weight of oven-dried soil in the jar.

Accumulated  $\text{N}_2\text{O}$  and  $\text{CO}_2$  emissions during the whole incubation were derived from the sums of measurement and no-measurement days estimated by linear interpolation (Mosier et al. 2006).

Differences in accumulated  $\text{N}_2\text{O}$  and  $\text{CO}_2$  emissions were analyzed by one-way analysis of variance (ANOVA) and compared by Duncan's multiple range test at the 5% level using IBM SPSS Statistics 20; all the figures were made by Sigmaplot 12.5.

## 3 Results

### 3.1 Effects of C rates amended with ammonium or nitrate on $\text{N}_2\text{O}$ and $\text{CO}_2$ emissions (experiment I)

#### 3.1.1 $\text{N}_2\text{O}$ and $\text{CO}_2$ emissions

Glucose addition significantly stimulated  $\text{N}_2\text{O}$  emissions regardless of applied N forms (Fig. 1). More  $\text{N}_2\text{O}$  produced after glucose addition combined with  $\text{NO}_3^-$  than with  $\text{NH}_4^+$ .  $\text{N}_2\text{O}$  emissions when glucose with  $\text{NH}_4^+$  form reached the peak immediately in the first incubation day but followed by a repaid decline as the incubation proceeded (Fig. 1a). The highest peak value

was found in the low rate of glucose treatment ( $0.5 \text{ g C kg}^{-1}$ ), while the lowest peak value was under the highest glucose rate ( $4 \text{ g C kg}^{-1}$ ).  $\text{N}_2\text{O}$  peak value from  $\text{NH}_4^+$  + Glu.0.5 treatment was about 150 times higher than that from only  $\text{NH}_4^+$ -N addition treatment. However, a different trend was observed when glucose addition with  $\text{NO}_3^-$  form (Fig. 1b). Addition of high rate of glucose ( $4 \text{ g C kg}^{-1}$ ) significantly stimulated  $\text{N}_2\text{O}$  emissions on the first day and decreased afterward, but such response was not observed in lower rates ( $0.5, 1, \text{ and } 2 \text{ g kg}^{-1}$ ) treatments.  $\text{N}_2\text{O}$  fluxes peaked on the second day of the incubation in the  $\text{NO}_3^-$  + Glu.0.5 and  $\text{NO}_3^-$  + Glu.1.0 treatments, and kept consistent during the first 2 days in  $\text{NO}_3^-$  + Glu.2.0 treatment. The  $\text{N}_2\text{O}$  peak values followed the order  $\text{NO}_3^-$  + Glu.0.5 >  $\text{NO}_3^-$  + Glu.1.0 >  $\text{NO}_3^-$  + Glu.4.0 >  $\text{NO}_3^-$  + Glu.2.0.  $\text{N}_2\text{O}$  peak from  $\text{NO}_3^-$  + Glu.0.5 treatment was about 235 times higher than that from only  $\text{NO}_3^-$ -N addition treatment.  $\text{N}_2\text{O}$  flux from  $\text{NO}_3^-$  treatment was similar to CK treatment (from 0 to  $6.7 \mu\text{g N kg}^{-1} \text{ day}^{-1}$ ) during the incubation, suggesting there were little  $\text{N}_2\text{O}$  emissions despite the presence of  $\text{NO}_3^-$  in the soil.

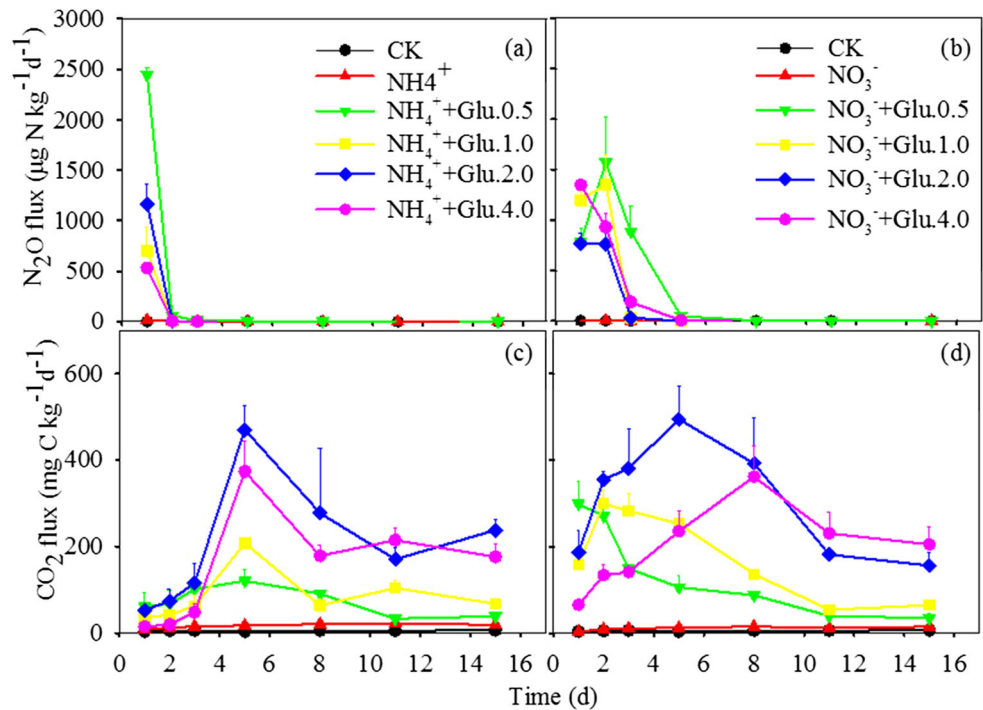
$\text{CO}_2$  fluxes increased with the addition of glucose and  $\text{NH}_4^+$  (Fig. 1c) or  $\text{NO}_3^-$ -forms (Fig. 1d).  $\text{CO}_2$  fluxes of  $\text{NH}_4^+$  + glucose treatments showed an uptrend in the first 5 days of the incubation, then decreasing gradually afterward.  $\text{CO}_2$  flux was the highest in  $2.0 \text{ g C kg}^{-1}$  of glucose rate and increased by about 20-fold compared with only  $\text{NH}_4^+$  addition. As for  $\text{CO}_2$  fluxes in treatments with glucose plus  $\text{NO}_3^-$ -based fertilizer addition (Fig. 1d), glucose addition enhanced  $\text{CO}_2$  fluxes to different degrees, and the order of the  $\text{CO}_2$  peak value was  $\text{NO}_3^-$  + Glu.2.0 >  $\text{NO}_3^-$  + Glu.4.0 >  $\text{NO}_3^-$  + Glu.0.5 >  $\text{NO}_3^-$  + Glu.1.0.

Accumulated  $\text{N}_2\text{O}$  and  $\text{CO}_2$  emissions were regulated by the ratios of external C and N inputs (Fig. 2, Table A1—Electronic Supplementary Material-ESM). The accumulated  $\text{N}_2\text{O}$  emission was the highest when C/N ratio was 12.5 (Fig. 2a) and decreased along with increasing C/N ratio under both N forms. The  $\text{N}_2\text{O}$  emissions were significantly higher ( $P < 0.05$ ) when the N source was  $\text{NO}_3^-$  than  $\text{NH}_4^+$  with equal glucose addition in most treatments except for Glu.2.0 treatment (Table A1 – ESM). The trends of accumulated  $\text{CO}_2$  emissions (Fig. 2b) were different from  $\text{N}_2\text{O}$  emissions, and it increased along with increasing C/N ratio and reached the peak when C/N was 50, and then decreased when C/N ratio was 100 in both N forms. The accumulated  $\text{CO}_2$  emissions were higher but not significantly ( $P > 0.05$ ) when N form was  $\text{NO}_3^-$  than  $\text{NH}_4^+$  with equal glucose addition in most treatments except for Glu.2.0 treatment ( $P < 0.05$ ) (Table A1—ESM).

#### 3.1.2 Dynamics of $\text{NH}_4^+$ and $\text{NO}_3^-$ concentration and N balance

The concentrations of  $\text{NH}_4^+$  decreased after  $\text{NH}_4^+$  and glucose addition in the first 5 days of incubation, indicating high

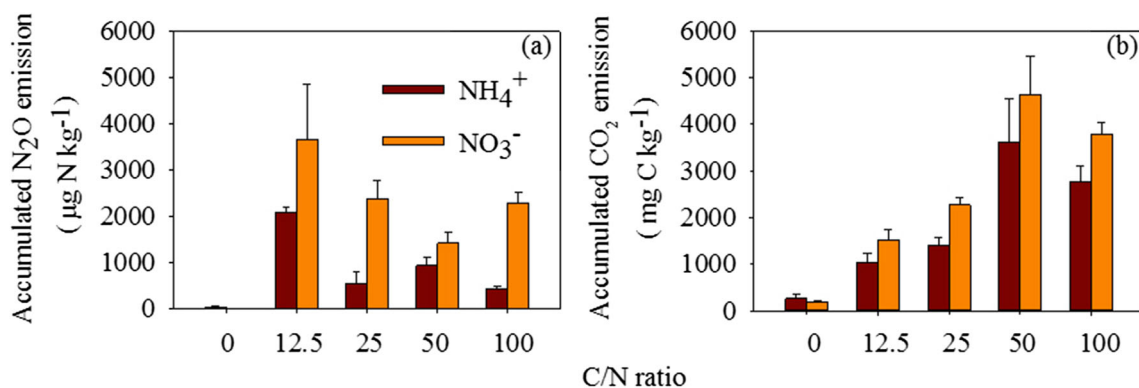
**Fig. 1** N<sub>2</sub>O and CO<sub>2</sub> fluxes from different treatments during the incubation in experiment I. Nitrogen source for **a, c** was (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, for **b, d** was Ca(NO<sub>3</sub>)<sub>2</sub>, and the carbon source was glucose. Error bars are the standard error of three replicates



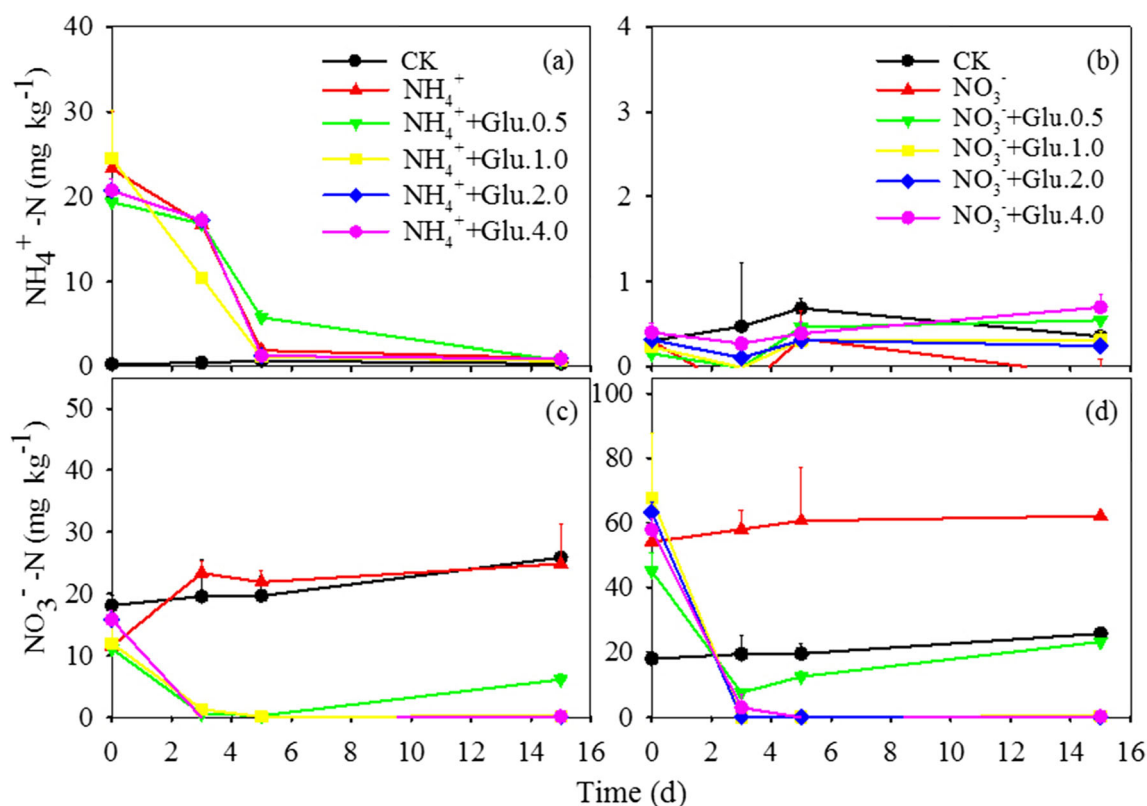
nitrification or immobilization occurred in the soil (Fig. 3a). The concentrations of NO<sub>3</sub><sup>-</sup> generally increased under NH<sub>4</sub><sup>+</sup> treatment without glucose addition (Fig. 3c), but it decreased quickly to zero on the third day of the incubation in all the addition of glucose treatments regardless of how much C added. Few NH<sub>4</sub><sup>+</sup> was detected, and it remained consistently low (below 1 mg N kg<sup>-1</sup>) during the whole incubation in all NO<sub>3</sub><sup>-</sup> treatments (Fig. 3b). The NO<sub>3</sub><sup>-</sup> concentrations were relatively stable during the whole incubation in CK and NO<sub>3</sub><sup>-</sup> treatments, indicating low processes of N transformations in the soil. Whereas with the application of glucose (Fig. 3d), the NO<sub>3</sub><sup>-</sup> concentrations decreased sharply to zero in the first 3 days of incubation in all glucose treatments, indicating high denitrification or immobilization occurred in the soil. The increased trend of NO<sub>3</sub><sup>-</sup> concentrations in NO<sub>3</sub><sup>-</sup> + Glu.0.5

treatment after 3 days showed mineralization again with low dose of glucose addition.

Nitrogen balance, as seen in Table A2 (ESM), reflects soil N transformations and their relationship with N<sub>2</sub>O emissions. The positive N balance could be explained by immobilization of mineral N (NH<sub>4</sub><sup>+</sup>+NO<sub>3</sub><sup>-</sup>) or other gaseous loss such as NH<sub>3</sub> or N<sub>2</sub>, while the negative N balance could be explained by mineralization of organic N in the soil. Different C/N ratios due to different amount of glucose addition affected soil N mineralization and mobilization in various ways. In control (CK) and NO<sub>3</sub><sup>-</sup> treatments, low negative N balance was observed while a low positive N balance was observed in NH<sub>4</sub><sup>+</sup> treatment. However, all treatments with glucose addition had relatively large positive values of N balance, demonstrating high mineral N immobilization or other gaseous losses. With the glucose addition, the N



**Fig. 2** Effects of different C/N ratios on accumulated N<sub>2</sub>O (a) and CO<sub>2</sub> (b) emissions. C/N ratio was calculated by rates of external C and N addition. Error bars are the standard error of three replicates



**Fig. 3** Ammonium and nitrate concentrations of different treatments on sampling days in experiment I. Nitrogen source for **a, c** was  $(\text{NH}_4)_2\text{SO}_4$ , for **b, d** was  $\text{Ca}(\text{NO}_3)_2$ , and carbon source was glucose. Error bars are the standard error of three replicates

balance values of  $\text{NO}_3^-$  form were higher than that of  $\text{NH}_4^+$  form except  $\text{NO}_3^- + \text{Glu.0.5}$  treatment.

### 3.2 Effects of C types amended with ammonium or nitrate on $\text{N}_2\text{O}$ and $\text{CO}_2$ emissions (experiment II)

#### 3.2.1 $\text{N}_2\text{O}$ and $\text{CO}_2$ emissions

The highest  $\text{N}_2\text{O}$  flux was found when pectin is applied with  $\text{NH}_4^+$  (Fig. 4a, b); however, when N form changed to  $\text{NO}_3^-$ , the highest  $\text{N}_2\text{O}$  was emitted from glucose treatment (Fig. 4c, d).  $\text{N}_2\text{O}$  fluxes from all N treatments reached the peak on the first day of the incubation, and followed by a quick decline afterward. The accumulated  $\text{N}_2\text{O}$  emissions are shown in Fig. 5a and Table A3 (ESM). During the whole incubation, total  $\text{N}_2\text{O}$  emissions of pectin and glucose treatments were higher than other C treatments combined with either  $\text{NH}_4^+$  or  $\text{NO}_3^-$ . The accumulated  $\text{N}_2\text{O}$  emissions of other four C treatments followed the order  $\text{NH}_4^+ + \text{lignin} > \text{NH}_4^+ + \text{starch} > \text{NH}_4^+ + \text{straw} > \text{NH}_4^+ + \text{cellulose}$ , and  $\text{NO}_3^- + \text{starch} > \text{NO}_3^- + \text{straw} > \text{NO}_3^- + \text{lignin} > \text{NO}_3^- + \text{cellulose}$ , but the differences were not significant ( $P > 0.05$ ). Accumulated  $\text{N}_2\text{O}$  emissions of  $\text{NO}_3^- + \text{cellulose}$  were the lowest among the different C treatments combined with either  $\text{NH}_4^+$  or  $\text{NO}_3^-$ .

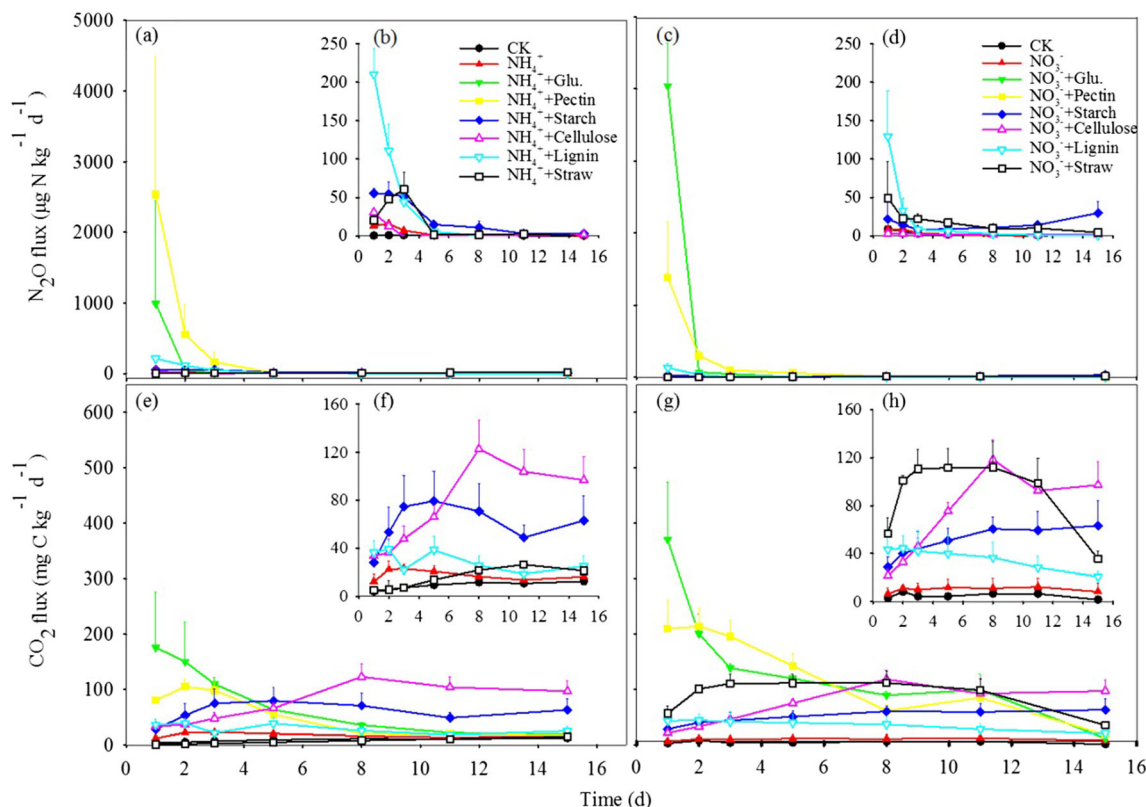
The  $\text{CO}_2$  fluxes reached the peak at different incubation times after different C addition with  $\text{NH}_4^+$  form (Fig. 4e, f). With the

glucose addition,  $\text{CO}_2$  fluxes peaked on the first incubation day up to  $175.8 \text{ mg C kg}^{-1} \text{ day}^{-1}$ , and followed by a decrease trend. Whereas for other C treatments, the  $\text{CO}_2$  emissions peak occurred after 2 days and even after 11 days for the straw treatment, indicating that different C types stimulated the microbial respiration to different degrees. Cellulose and straw addition significantly ( $P < 0.05$ ) increased accumulated  $\text{CO}_2$  emissions during the whole incubation (Fig. 5b; Table A3), and more  $\text{CO}_2$  produced when N form was  $\text{NO}_3^-$  compared to  $\text{NH}_4^+$ .

When applied N as  $\text{NO}_3^-$  form (Fig. 4g, h), the dynamic of  $\text{CO}_2$  fluxes were similar to that of  $\text{NH}_4^+$  treatments. With the addition of glucose and pectin, higher  $\text{CO}_2$  fluxes were observed than other treatments especially for the first 8 days and peaked on the first day of the incubation. The  $\text{CO}_2$  emissions reached the peak in the rest of the C treatments lagged behind the treatments of  $\text{NO}_3^- + \text{Glu}$  and  $\text{NO}_3^- + \text{pectin}$  and lasted longer during the incubation.  $\text{NO}_3^- + \text{Glu}$  and  $\text{NO}_3^- + \text{pectin}$  treatments produced the highest accumulated  $\text{CO}_2$  during the incubation (Table A3). The lowest  $\text{CO}_2$  fluxes were observed in lignin treatment combined with either  $\text{NH}_4^+$  or  $\text{NO}_3^-$ .

#### 3.2.2 Dynamics of $\text{NH}_4^+$ and $\text{NO}_3^-$ concentration and N balance

When the N forms as  $\text{NH}_4^+$ , the soil  $\text{NH}_4^+$  concentrations decreased (Fig. 6a), and while the  $\text{NO}_3^-$  concentrations



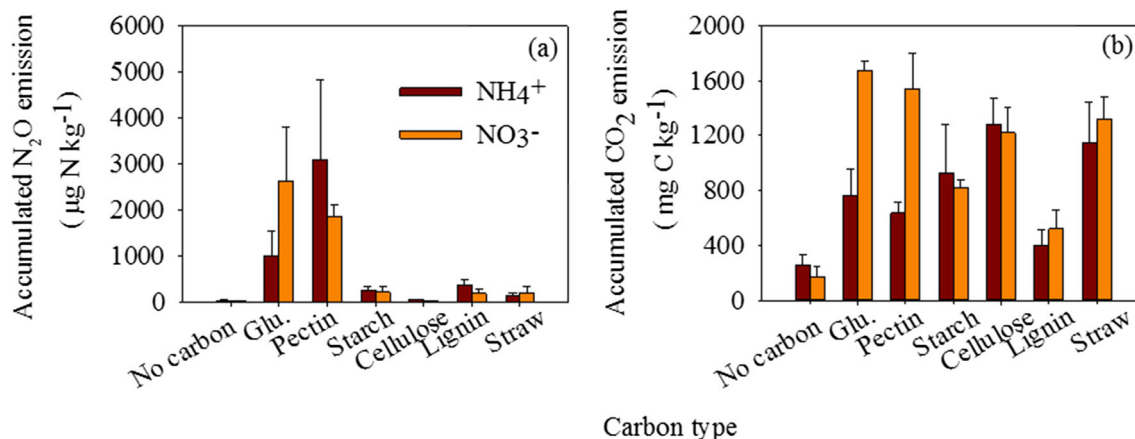
**Fig. 4** N<sub>2</sub>O and CO<sub>2</sub> fluxes of different treatments during the incubation in experiment II. Nitrogen source for (a, b, e, f) was (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, for (c, d, g, h) was Ca(NO<sub>3</sub>)<sub>2</sub>. The insert panel (b, d, f, h) are shown without

NH<sub>4</sub><sup>+</sup>+Glu. and NH<sub>4</sub><sup>+</sup>+Pectin to make the figure more clear, Error bars are the standard error of three replicates

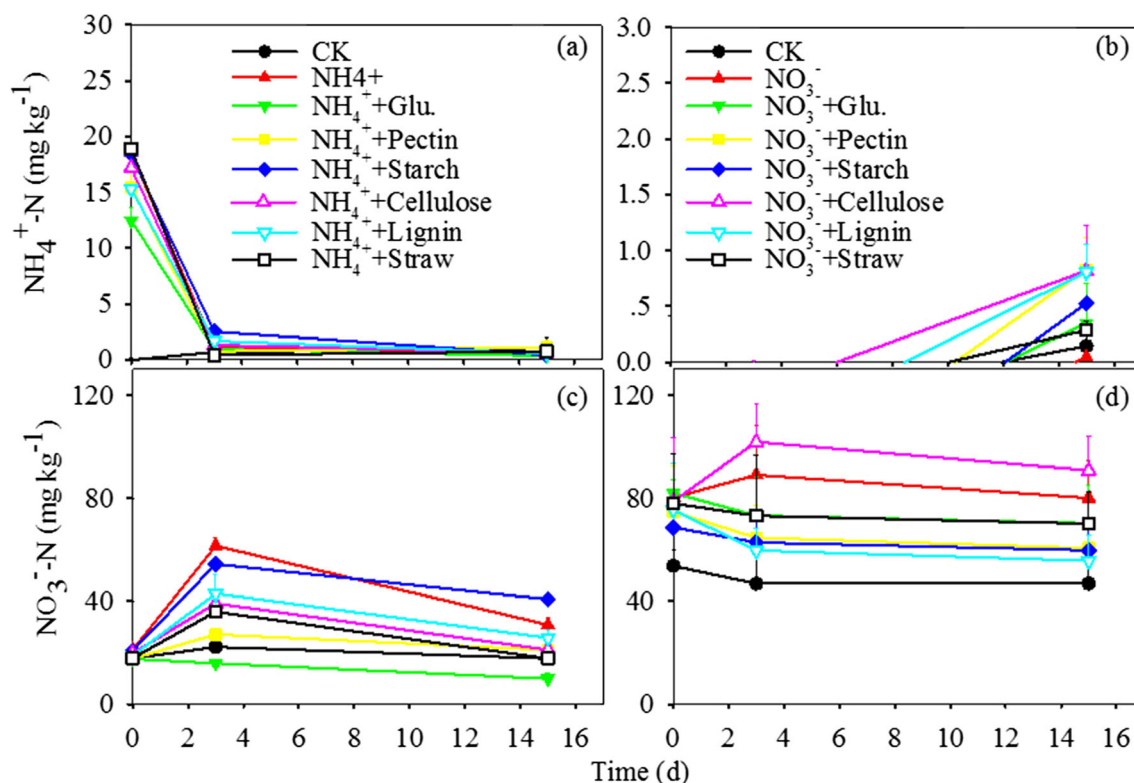
increased before the third day (Fig. 6c), and then slightly decreased till the end of the incubation. The concentrations of NH<sub>4</sub><sup>+</sup> in the NH<sub>4</sub><sup>+</sup> + Glu treatment was the lowest among C addition treatments as glucose might increase the immobilization of NH<sub>4</sub><sup>+</sup> rapidly. At the end of the incubation, there were various amounts of NO<sub>3</sub><sup>-</sup> detected in the treatments with the range of about 10 in NH<sub>4</sub><sup>+</sup> + Glu treatment to about 40 mg N kg<sup>-1</sup> in NH<sub>4</sub><sup>+</sup> + starch treatment. When applied N as NO<sub>3</sub><sup>-</sup> form, the NH<sub>4</sub><sup>+</sup> concentrations were low and had a slight uptrend at the end of the incubation probably due to

mineralization (Fig. 6b). Apart from cellulose addition, NO<sub>3</sub><sup>-</sup> concentrations decreased to different degrees among other C treatments (Fig. 6d).

When applied N form as NH<sub>4</sub><sup>+</sup>, relatively larger values of N balance were observed with the addition of glucose, cellulose, and straw treatments (Table A4—ESM), indicating higher mineral N immobilization or other gaseous losses in these treatments. When applied N form as NO<sub>3</sub><sup>-</sup>, the values of N balance showed positive in all treatments except the relatively high negative N balance only in NO<sub>3</sub><sup>-</sup> + cellulose



**Fig. 5** Effects of different C types on accumulated N<sub>2</sub>O (a) and CO<sub>2</sub> (b) emissions. Error bars are the standard error of three replicates



**Fig. 6** Ammonium and nitrate concentrations of different treatments on sampling days in experiment II; Nitrogen source for (a) and (c) was  $(\text{NH}_4)_2\text{SO}_4$ , for (b) and (d) was  $\text{Ca}(\text{NO}_3)_2$ ; Error bars are the standard error of three replicates

treatment, indicating enhanced mineralization with addition of cellulose.

## 4 Discussion

In the current incubation study, the combined application of C and N significantly enhanced  $\text{N}_2\text{O}$  and  $\text{CO}_2$  emissions, albeit to varying degrees, which was consistent with other field results (Ju et al. 2011; Bao et al. 2012; Cui et al. 2012; Gao et al. 2014). Moreover, different types of C and N had different effects on increasing  $\text{N}_2\text{O}$  and  $\text{CO}_2$  emissions. For example, when applied C source was pectin or glucose, more accumulated  $\text{N}_2\text{O}$  were emitted than other C sources regardless of applied N forms. Several previous studies revealed that soil  $\text{N}_2\text{O}$  production rate could be stimulated by adding readily degradable C, possibly due to increasing denitrification (Wang et al. 2005; Cheng et al. 2012; Ameloot et al. 2016; Mehnaz et al. 2018). In our study,  $\text{NO}_3^-$  concentrations sharply decreased after C and N addition indicating denitrification occurrence or immobilization. C addition could greatly enhance  $\text{O}_2$  consumption and microorganism respiration in the soil, where generated anaerobic microsites which was favorable for denitrification (Huang et al. 2014; Song et al. 2019). We did not think that DNRA could occur because the  $\text{NH}_4^+$  concentration did not increase at the end of the incubation in both experiments. However, the immobilization of  $\text{NO}_3^-$  is

actually a possible process of  $\text{NO}_3^-$  decrease. A meta-analysis also showed that specific exogenous organic C input at a high rate ( $>0.5 \text{ g C kg}^{-1}$ ) or with a high C/N ratio ( $>18$ ) could enhance microbial  $\text{NO}_3^-$  immobilization (Cheng et al. 2017).

Our results indicated that the soils treated with high concentrations ( $>0.5 \text{ g C kg}^{-1}$ ) of glucose produced lower  $\text{N}_2\text{O}$  than  $\text{NH}_4^+ + \text{Glu } 0.5$  during the incubation (Table A1—ESM, Fig. 1). It is possibly because glucose was easy to be used by soil microorganism, and high concentrations might stimulate the growth and activity of heterotrophic microorganisms (Schimel and Bennett 2004; Booth et al. 2005), thus assimilation of mineral N (Burger and Jackson 2003; Myrold and Posavatz 2007), which reduced the  $\text{NH}_4^+$  substrate for nitrification and then reduced related  $\text{N}_2\text{O}$  production as well. And it was also reflected by higher  $\text{CO}_2$  production in this treatment (Table A1). Another reason for lower amounts of  $\text{N}_2\text{O}$  emissions from high application rate of glucose ( $>0.5 \text{ g kg}^{-1}$ ) than  $\text{NH}_4^+ + \text{Glu } 0.5$  was that it might stimulate the anaerobic conditions which enhanced the activity of the enzyme responsible for reduction of  $\text{N}_2\text{O}$  to  $\text{N}_2$  (Murray et al. 2004; Kool et al. 2010).

$\text{NH}_4^+$  and  $\text{NO}_3^-$ -N forms are two main N fertilizer types in crop production, which are also substrates for nitrification and denitrification respectively, and have various effects on  $\text{N}_2\text{O}$  emissions from different soil types. Bouwman et al. (2002) reported that more  $\text{N}_2\text{O}$  was emitted when applied  $\text{NO}_3^-$  than  $\text{NH}_4^+$ -based fertilizers from New Zealand and Germany



grassland soil due to fine texture and high clay content of the soil. However, in the present study, we found opposite results that application of  $\text{NH}_4^+$ -based fertilizer alone produced more  $\text{N}_2\text{O}$  than  $\text{NO}_3^-$ -based fertilizer due to high nitrification capacity of our studied soil (Ding et al. 2015; Ju and Zhang 2017);  $\text{NH}_4^+$ -based fertilizer could be immediately proceeded by nitrifiers and produce large amount of nitrification-related  $\text{N}_2\text{O}$  (Wan et al. 2009; Ju et al. 2011; Ju and Zhang 2017; Zhu et al. 2019a, b).

Different C types had various effects on  $\text{N}_2\text{O}$  and  $\text{CO}_2$  emissions, indicating different interactions between C and N with respect to  $\text{N}_2\text{O}$  and  $\text{CO}_2$  production. In our study, the application of glucose and pectin with  $\text{NH}_4^+$  or  $\text{NO}_3^-$  resulted in the highest  $\text{N}_2\text{O}$  emissions (Table A3—ESM, Fig. 6). Probably because they were more readily degradable C to stimulate growth of soil microorganisms. In contrast, complex C compounds like straw had low ability of  $\text{NO}_3^-$  immobilization and produced lower  $\text{N}_2\text{O}$  and  $\text{CO}_2$  than glucose and pectin addition, which was in line with previous studies (Miller et al. 2008; Huang et al. 2013; Qiu et al. 2013). The soluble C types like glucose and pectin were more efficiently used by soil microorganisms in a soil with pH 7.4 under anaerobic condition (de Catanzaro and Beauchamp 1985). Although our incubation condition was aerobic, the C addition might enhance  $\text{O}_2$  consumption and respiration in soil matrix, which could create anaerobic condition (Song et al. 2019).

The  $\text{CO}_2$  fluxes and its duration are mainly related with the structure and composition of various organic materials. In our study, the straw treatment combined with  $\text{NH}_4^+$  or  $\text{NO}_3^-$  produced small amount of  $\text{CO}_2$  in the first few days but showed sustained production over time as the straw contained much lower soil labile C and required a longer decomposition time. The limited available C and aerobic condition might inhibit denitrification, resulting in less  $\text{N}_2\text{O}$  emissions (Table A3—ESM) and higher final  $\text{NO}_3^-$  (Fig. 6) concentrations in  $\text{NO}_3^-$  + straw treatment. Only readily available C sources (e.g., glucose and pectin) together with either  $\text{NH}_4^+$  or  $\text{NO}_3^-$  stimulated  $\text{N}_2\text{O}$  emissions. Therefore, from our current study under the aerobic conditions, the possible mitigation strategies of  $\text{N}_2\text{O}$  emission would be avoiding the combination addition of N fertilizers and easily accessible C.

Soil  $\text{NH}_4^+$  and  $\text{NO}_3^-$  were involved in many N transformation pathways, including nitrification, denitrification, N immobilization, and mineralization. Nitrification and denitrification depended on  $\text{NH}_4^+$  and  $\text{NO}_3^-$  concentrations and  $\text{O}_2$  status, and soil N mineralization and immobilization were related to the C/N ratio of added substrate (Larsen and McCartney et al. 2000). Higher C/N ratio (27) led to a strong immobilization of N, while lower C/N ratio (15–17) led to mineralization (Bengtsson et al. 2003), which explained soil  $\text{NH}_4^+$  kinetics in the present study (Fig. 2, Table A2—ESM). Moreover, C types could also affect N transformations, and the organic materials with higher C/N ratio (45.5) could have a

slower decomposition rate than that with lower C/N ratio (18.8), which stimulated N immobilization (Trinsoutrot et al. 2000; Moritsuka et al. 2004). The immobilization was greater in soils incubated with glucose amendments than those incubated with more complex C compounds (Magill and Aber 2000). The returned C sources in the field are crop roots, straw, manure or compost, which are composed with different C components. As we did the model experiments in laboratory, we are interested in the effects of different C components on N transformations and  $\text{N}_2\text{O}$  emissions. The results would help us understand the interactions of C and N when applying the integrates of complex OM in situ.

Previous studies regarding application of C and N rates and types on  $\text{N}_2\text{O}$  emissions were almost all under strict anaerobic condition, but our study was aerobic which would be a better representation of real field conditions in upland soils (Ju and Zhang 2017). We still have no direct evidences to show nitrifier denitrification or denitrification in our current treatments (Yang et al. 2017), but measuring  $\text{O}_2$  concentration in soil microsites might be helpful to further explain the mechanisms of  $\text{N}_2\text{O}$  production and differentiate  $\text{N}_2\text{O}$  processes in the soil.

## 5 Conclusions

We observed from the incubation study that the application of mixed C and N substrates enhanced  $\text{N}_2\text{O}$  and  $\text{CO}_2$  emissions albeit to different extents. A higher accumulated  $\text{N}_2\text{O}$  emission was found when C/N ratio was lower (12.5).  $\text{NO}_3^-$  together with glucose treatments produced more  $\text{N}_2\text{O}$  than  $\text{NH}_4^+$ -treated soils. The stimulating effect of glucose and pectin as labile C sources with N addition on  $\text{N}_2\text{O}$  production should be considered, especially in soils with high pH, low organic C, and a strong ammonia oxidation potential. Under field conditions, therefore, avoiding applying both N and C especially rich in glucose and pectin together could be the suitable targeted strategy for mitigating  $\text{N}_2\text{O}$  emissions.

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