RESEARCH ARTICLE



Enhanced reducing sugar production by blending hydrolytic enzymes from *Aspergillus fumigatus* to improve sugarcane bagasse hydrolysis

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Abstract

Biomass pretreatment for the production of second-generation (2G) ethanol and biochemical products is a challenging process. The present study investigated the synergistic efficiency of purified carboxymethyl cellulase (CMCase), β -glucosidase, and xylanase from *Aspergillus fumigatus* JCM 10253 in the hydrolysis of alkaline-pretreated sugarcane bagasse (SCB). The saccharification of pretreated SCB was optimised using a combination of CMCase and β -glucosidase (C+ β ; 1:1) and addition of xylanase (C+ β +xyl; 1:1:1). Independent and dependent variables influencing enzymatic hydrolysis were investigated using response surface methodology (RSM). Hydrolysis using purified CMCase and β -glucosidase achieved yields of 18.72 mg/mL glucose and 6.98 mg/mL xylose. Incorporation of xylanase in saccharification increased the titres of glucose (22.83 mg/mL) and xylose (9.54 mg/mL). Furthermore, characterisation of SCB biomass by scanning electron microscopy, X-ray diffraction, and Fourier transform infrared spectroscopy respectively confirmed efficient structural disintegration and revealed the degree of crystallinity and spectral characteristics. Therefore, depolymerisation of lignin to produce high-value chemicals is essential for sustainable and competitive biorefinery development.

Keywords Alkaline pretreatment · Saccharification · Glucose · Xylose · Degree of crystallinity · Spectral characteristics

Introduction

Lignocellulosic biomass is a polymer that has attracted worldwide attention due to its economic potential for renewable feedstock sources for energy production, especially given the rapid depletion of oil reserves and other environmental issues. Lignocellulosic waste materials such as those from agricultural, agro-industrial, and forestry processes are comparatively less expensive than other sources of feedstock (Wang et al. 2018; Velvizhi et al. 2023). Therefore, methods to manipulate the chemical composition of lignocelluloses to combat energy resource depletion are in great demand

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² Department of Biotechnology, National Institute of Technology Andhra Pradesh, Tadepalligudem 534101, India (Dharmaraja et al. 2023). Lignocellulose biomass-derived biofuels can replace fossil fuels due to greater efficiency and yield. The quantity of lignocellulose dry matter available is believed to be 85 to 125 billion tonnes (Zhang et al. 2016). However, chemical complexity makes it difficult to utilise the sugar molecules in these substances. Few processes are able to efficiently convert complex sugars of lignocelluloses into simple fermentable sugars. Complexity is contributed by lignin, hemicellulose, and cellulose fibrils, which are abundant in plant cell walls (Velvizhi et al. 2022). Thus, there is a need for an effective pretreatment procedure that can break down the composite cellulose molecules and allow enzymes to access and hydrolyse these substances (Chaudhary et al. 2023). Enzymatic hydrolysis fragments the crosslinks between lignin and hemicellulose adjoining the cellulose fibres (Hans et al. 2023). This is followed by breaking the hydrogen bonds of the crystalline structure of cellulose, increasing the surface area of the fibres and enhancing permeability (Devi et al. 2022; Kukreti et al. 2023).

The present work focused on developing more costeffective and viable sources of biofuel generation. Sugarcane bagasse (SCB) is a type of biomass that contributes to meeting the world's energy needs and environmental

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sustainability. Worldwide, sugarcane production is ~1.6 billion tonnes per year, which generates ~ 279 million metric tons of SCB (Chandel et al. 2012; Jugwanth et al. 2020). Brazil is reportedly the largest sugarcane producer, followed by India and China (Khoo et al. 2018). SCB is a waste material from the sugarcane industry that has gained the attention of scientists as a prospective source for second-generation biofuel production (Ajala et al. 2021). The chemical composition of SCB is well studied in the literature (Batalha et al. 2015; Santo et al. 2018; Ávila et al. 2018; Hans et al. 2023). Another advantage is the availability of sugarcane by-products from sugar industries and alcohol-producing plants, and the resultant decrease in logistic expenses if biofuel production is established in the same unit (Ferreira-Leitão et al. 2010). Converting lignocellulosic materials into ethanol involves initial pretreatment, enzymatic hydrolysis to yield sugars, fermentation, and fractionation (Aziz et al. 2023). Pretreatment procedures break down the complex organic matter, decrease the quantity of hemicellulose and lignin, and alter the crystalline configuration of cellulose to enhance its susceptibility to enzymatic attack (Chauhan et al. 2023). Without the pretreatment step, the efficiency of converting glucose and xylose into ethanol by enzyme hydrolysis is very low, but high concentrations of xylose and glucose can be liberated with the help of a pretreatment procedure (Guilherme et al. 2015).

There are several investigations on SCB as a biomass source. However, there is a need to assess the use of sugarcane straw, alone or with bagasse, as raw material to produce fermentable sugars for ethanol production. Additionally, developing more efficient enzyme combinations requires detailed information on the explicit enzymes engaged in the breakdown of sugarcane-based organic matter (Huang et al. 2011; Adav et al. 2012; Ávila et al. 2018). Disruption of the complex structure of cellulose-hemicellulose-lignin is an essential step in biorefining biomass. Alkaline pretreatment acts directly on this structure by promoting delignification by breaking down α - and β -alkyl- and aryl-ether bonds, deprotonating phenolic groups liberated from lignin and depolymerising cellulose to eliminate some uronic acids and acetyl groups of xylan chains, consequently enhancing the susceptibility of the substrate to enzymatic breakdown. Moreover, research on the life cycle assessment of secondgeneration biofuel production processes using alkaline pretreatment as an initial step has demonstrated that an alkaline solution can be retrieved and used again, making the process more economically feasible (Rocha et al. 2014).

In recent years, research on the efficient and economically viable enzymatic transformation of lignocellulosic biomass has been the focus of many international studies to overcome constraints associated with the high cost of commercial enzymes. Filamentous fungi mainly produce hydrolytic enzymes, and many studies have sought superior strains with high enzyme activity, especially cellulases and hemicellulases (Santos et al. 2015; de Oliveira et al. 2017; Prajapati et al. 2020). These enzymes play significant roles during the enzymatic saccharification of biomass; the characteristic biospecificity of hydrolytic enzymes for a particular chemical bond is used to produce the corresponding products. In this regard, cellulases and cellobiases (or β -glucosidases) are responsible for breaking down the glycosidic bonds of cellulose to produce glucose while xylanases break down the glycosidic bonds of xylan to release xylose (Ibarra-Díaz et al. 2020).

This study explored the synergistic efficiency of hydrolytic enzymes produced from Aspergillus fumigatus JCM 10253 (Paramjeet et al. 2021; Saroj et al. 2022) in the hydrolysis of alkaline-pretreated SCB. Specifically, we focused on optimising and evaluating the saccharification yield using carboxymethyl cellulase (CMCase) and β-glucosidase, including supplementation of xylanase, on alkaline-pretreated SCB. To our knowledge, this is the first report on blending hydrolytic enzymes obtained from a single strain of A. fumigatus in SCB hydrolysis. Enzymatic hydrolysis was optimised using a response surface methodology (RSM), and five independent variables (temperature, pH, enzyme loading, substrate concentration, and incubation time) were evaluated. This approach can determine the enzymatic production potential and the appropriate proportions of enzymes. Furthermore, untreated, pretreated and enzymatically hydrolysed biomass were characterised using scanning electron microscopy (SEM), X-ray diffraction (XRD), and Fourier transform infrared spectroscopy (FTIR).

Materials and methods

Enzymes

Purified enzymes CMCase (19.5 IU/mL), β -glucosidase (0.083 IU/mL), and xylanase (26.8 IU/mL) from *A. fumigatus* JCM 10253 were used for enzymatic hydrolysis. Their activities were measured and defined as described in previous reports (Paramjeet et al. 2021; Saroj et al. 2022).

Raw material processing

SCB was procured from a local jaggery manufacturing unit in Warangal, Telangana, India. SCB was used as lignocellulosic biomass after juice extraction without washing, and air-dried at room temperature to reduce the moisture content. An electric crusher was used to homogenise SCB and generate fine particles. The consistency of the particle size was maintained between 3 and 5 mm in diameter by sieving. Dried biomass was stored in plastic Ziploc bags at room temperature for further experiments.

Alkaline-based pretreatment

SCB was treated with 0.5 M sodium hydroxide (10% w/v) at 121 °C for 1 h. After treatment, the solid fraction was separated by filtration and subjected to repeated washing with distilled water until the pH was neutralised. The pretreated biomass was dried at 60 °C for 12 h and stored in a desiccator for hydrolysis and physicochemical characterisation.

Enzymatic hydrolysis

Enzymatic hydrolysis was conducted at different temperatures spanning 30-60 °C with an initial substrate concentration of 2.5% (w/v) pretreated dry mass (SCB) and incubated for up to 48 h in 250-mL Erlenmeyer flasks with shaking at 150 rpm. Saccharification was performed in flasks containing 100 mL of 50 mM sodium citrate buffer (pH 4.8), and sodium azide (0.04%) was added to the reaction mixture to prevent microbial contamination. Reactions were treated with different concentrations of substrates (SCB) from 1 to 3.5% (w/v) and enzyme loading from 1 to 3.5 mL (v/v). Two approaches were employed: a mixture of CMCase and β -glucosidase at a 1:1 ratio (% v/v), and a mixture of CMCase, β -glucosidase and xylanase at a 1:1:1 ratio (% v/v). The effect of pH (4, 4.2, 4.6, 4.8, 5, and 5.2) on total reducing sugar (TRS) yield was tested. Samples were obtained at 12, 24, 36, 48, and 72 h following hydrolysis and enzymes were deactivated by heating for 10 min at 100 °C. The resulting solids were separated by centrifugation for 5 min at 5000 rpm and filtered using a 0.22 µm syringe filter (Millipore, Billerica, MA, USA). Total reducing sugar was quantified as the total sum of glucose, cellobiose, arabinose, xylose, and other sugars released in each treatment.

Analytical method

After enzymatic hydrolysis, chromatographic separation of soluble sugars in hydrolysates was performed using high-performance liquid chromatography (HPLC) with a Prominence UFLC instrument (Shimadzu, Kyoto, Japan) equipped with an LC-20AD pump and a refractive index detector RID-10A. The Rezex-RPM-monosaccharide-Lead (II) ion column (Phenomenex, Torrance, CA, USA) was used to detect the sugars. HPLC-grade water was used as the mobile phase under isocratic mode. The temperature of the column oven and detector were set to 80 and 50 °C, respectively. The pump flow rate was 0.6 mL/min for 30 min run time. Chromatograms for each sample and standards were compared with respect to retention time (RT) to quantify reducing sugars. Standards used in analytical procedures were HPLC-grade chemicals from Sigma-Aldrich Co. (St. Louis, MO, USA).

Physicochemical characterisation of feedstock

SEM

SEM analysis was performed using a Carl Zeiss MA15/EVO 18 SEM (Carl Zeiss Microscopy GmbH Carl-Zeiss-Promenade 10 07745, Jena, Germany) instrument to examine the surface morphological characteristics of SCB. Untreated, pretreated and enzymatically hydrolysed SCB samples were fixed with carbon tape supporting an aluminium stub and analysed with acceleration voltage and magnification.

XRD

The cellulose crystalline structure of SCB samples was analysed using a Bruker D8 Advanced X-ray Diffractometer (Bruker AXS GmbH, Germany). Samples were dehydrated at 60 °C in an oven before analysis. Intensities were determined between 10° and 60° of a 2θ scale (scattering angle) with a scanning speed of 0.05°/scan. XRD patterns were used to calculate the cellulose crystalline index, represented as *CrI*, using the following equation (Segal et al. 1959):

Crystallinity Index(*CrI*) =
$$\frac{(I_{002} - I_{am})}{I_{002}} \times 100$$

where I_{002} represents the intensity of the (crystalline phase) 002 peak ($2\theta = 21.8^{\circ}$), and I_{am} represents the intensity of the peak corresponding to the amorphous phase ($2\theta = 18.2^{\circ}$).

FTIR

Changes in the chemical structure of SCB were determined following pretreatment and enzymatic hydrolysis by FTIR analysis using a 3000 Hyperion Microscope and a Vertex 80 FTIR System (Bruker Optics GmbH, Ettlingen, Germany). FTIR spectra of SCB were recorded at 4 cm⁻¹ resolution over a 500 - 4000 cm⁻¹ wavenumber range (Bala and Singh 2019).

Optimisation of enzymatic hydrolysis to generate glucose and xylose using RSM

Design Expert 7.0 software (Stat-Ease, Minneapolis, MN, USA) was used to apply the quadratic model to examine correlations between the quantity of sugars released and five variables governing the hydrolysis with the help of enzymes. The assortment criteria for elements and the range of variables depended on the condition of the operating system, which affected the hydrolysis process by enzymes. For

the dominant complex strategy, variables were fixed on a five-point scale, as depicted in Tables 1 and 2. The central composite design (CCD) comprised five factors, namely temperature (A, 30, 40, 50 and 60 °C), pH (B, 4.6, 4.8 and 5), enzyme loading (C, 2, 2.5, 3 and 3.5 mL), substrate concentration (D, 2, 2.5, 3 and 3.5 g) and incubation time (E, 24, 36, 48, 60 and 72 h), for reducing sugar (glucose and xylose) production from the substrate (pretreated SCB). Two sets of enzymes were prepared to employ RSM. In the first set, only considered on the substrate (set of the first set) of the first set of the first set of the set of the set of the first set of the se

purified CMCase and β -glucosidase (C + β ; 1:1 v/v) were mixed in equal proportion. The second set was prepared by mixing enzymes of the first set combined with purified xylanase (C + β + xyl; 1:1:1 v/v). Fifty runs for each enzyme set were conducted and production of reducing sugar was considered a response activity. The results of the dominant complex strategy for the examined variables (temperature, pH, enzyme loading, substrate concentration and incubation time) are summarised in

Tables 3 and 4. RSM was employed for analysis of variance (ANOVA) to obtain significant statistical parameters. The R^2 value was determined to assess the significance of the model. Second-order polynomial equations were applied to determine the relationships between independent variables to evaluate the predicted response, and 3D graphs were plotted to assess the effects of significant variables on the response. All experiments were performed in triplicate to identify differences in the obtained data, and results are presented as the mean \pm standard deviation.

Table 1 Experimental setup and independent variables used in central composite design for reducing sugar using combination of enzymes $(C+\beta)$

Independent variable	Units	Levels				
		-1	0	+1		
Temperature	°C	30	45	60		
pH		4.6	4.8	5		
Enzyme loading	mL	2	2.5	3		
Substrate concentration	g	2	2.5	3		
Incubation time	h	36	48	60		

Table 2 Experimental setup and independent variables used in central composite design using combination of enzymes $(C + \beta + xyl)$

Independent variables	Units	Levels	Levels			
		-1	0	+1		
Temperature	°C	30	45	60		
pH		4.6	4.8	5		
Enzyme loading	mL	2	2.75	3.5		
Substrate concentration	g	2	2.75	3.5		
Incubation time	h	24	48	72		

Results and discussion

Hydrolytic performance of alkaline-pretreated SCB under different conditions

Effect of temperature

The effects of temperature on enzymatic hydrolysis using cellulases $(C + \beta)$ and supplementation with xylanase $(C+\beta+xyl)$ are shown in Figs. 1a and 2a, respectively. Optimal enzymatic hydrolysis under both conditions was obtained at 50 °C with total reducing sugar of 27.59 mg/mL and 34.65 mg/mL, respectively. At 30 °C and 40 °C, the degree of enzymatic hydrolysis of pretreated SCB was lower than that at 50 °C, and the amount of reducing sugar was lower when the hydrolysis temperature was raised to 60 °C. This behaviour was most likely brought on by higher temperatures increasing enzyme inactivation. Saccharification using a mixture of cellulases from Fusarium incarnatum, A. niger and Trichoderma harzianum achieved maximal reducing sugar production at 50 °C (Pant et al. 2021). Huang et al. (2015) assessed alkali-pretreated bamboo residues and the optimum enzymatic saccharification yield was achieved at 50 °C. Similar studies have been conducted to maximise the enzymatic hydrolysis of lignocelluloses using endo-1,4-β-xylanase at 50 °C (Ge et al. 2014; Yang et al. 2015; Chen et al. 2020).

Effect of pH

pH is a critical factor affecting cellulose and hemicellulose degradation for reducing sugar production. Herein, the maximal yield was obtained at pH 4.8. The total reducing sugar released by hydrolysis using cellulases $(C + \beta)$ and with addition of xylanase $(C + \beta + xyl)$ was 32.13 mg/mL and 40.37 mg/mL, respectively (Figs. 1b and 2b). Recent studies found that pH 4.8 was optimum for hydrolysis of corn stover powder and sugarcane bagasse to produce reducing sugar (Zhang and Wu 2021; Barrameda et al. 2023). Maximum reducing sugar production is typically in the pH range of 4.8-5.0 (Alrumman 2016; Procentese et al. 2017; Sun et al. 2018). pH significantly affects the hydrolytic behaviour of cellulases; the hydrolytic process can only occur after the formation of an enzyme-substrate complex, and pH has a similar influence on both adsorption and hydrolysis, which typically occur at pH 4.8.

Effect of enzyme loading

The effect of enzyme loading on the hydrolysis of pretreated SCB was determined (Figs. 1c and 2c). An increase in enzyme loading from 1 to 3 mL (v/v) in the reaction mixture

Table 3

lable 3 RSM design:	A: Tem-	B: pH	oH C: Enzyme	D: Substrate	E: Incuba-	R1	
variables for reducing sugar yield using combination of	(°C)		loading (mL)	concentration (g)	tion time (h)	Glucose (mg/mL)	Xylose (mg/mL)
enzymes $(C + \beta)$	45	4.8	2.5	2.5	48	20.98	7.85
	30	5	2	2	60	3.04	2.12
	45	4.8	2.5	2	48	17.01	6.04
	45	4.8	2.5	2.5	60	14.83	5.83
	30	5	2	3	36	3.44	2.65
	60	5	3	3	60	5.43	1.93
	45	4.8	2.5	2.5	48	18.71	7.82
	45	4.8	2.5	2.5	48	18.95	7.94
	45	4.8	2.5	3	48	16.32	5.87
	60	5	2	2	36	6.72	2.25
	60	4.8	2.5	2.5	48	8.98	4.85
	45	4.8	2	2.5	48	17.93	6.98
	30	5	3	3	60	4.53	2.34
	60	4.6	2	2	60	4.84	2.12
	45	4.8	3	2.5	48	16.98	5.76
	60	5	3	2	36	4.03	2.19
	30	4.6	3	3	36	5.84	3.22
	30	5	2	2	36	6.94	2.45
	45	4.8	2.5	2.5	48	19.56	6.94
	60	4.6	3	2	60	9.04	2.67
	60	5	2	3	36	4.26	1.02
	60	4.6	3	3	60	6.64	3.11
	30	4.6	2	2	36	6.43	2.35
	30	4.8	2.5	2.5	48	7.93	4.32
	30	5	3	3	36	4.32	2.24
	60	5	3	3	36	4.74	2.02
	60	4.6	3	2	36	5.12	2.35
	60	4.6	2	3	60	5.43	3.23
	30	4.6	3	2	60	3.43	2.63
	60	5	2	2	60	4.61	1.13
	60	4.6	2	3	36	6.03	2.73
	60	4.6	3	3	36	6.54	2.83
	45	4.8	2.5	2.5	48	19.87	7.74
	60	5	3	2	60	5.34	3.04
	30	5	3	2	60	3.74	2.03
	30	4.6	2	3	36	6.94	3.04
	30	4.6	3	2	36	6.84	3.24
	60	4.6	2	2	36	3.35	1.95
	30	5	3	2	36	4.34	2.45
	45	5	2.5	2.5	48	16.63	4 92
	45	4.8	2.5	2.5	48	18.06	6.98
	30	5	2.	3	60	4.11	2.34
	30	4.6	3	3	60	5.43	2.74
	45	4.8	2.5	2.5	48	20.07	8.04
	45	4.8	2.5	2.5	36	15 74	4 93
	30	4.6	2.5	3	60	3.95	2 34
	45	4.0 4.8	2 5	25	48	19.98	7 95
	 60	 5	2.5	3	60	6.04	2 44
	45	5 4.6	2 5	25	48	18.45	2.77 6 58
	+5	4.0	2.3	2.3	+0	10.75	0.50
	30	4.6	2	2	60	5.02	2.14

Table 4RSM design:optimisation of processvariables for reducing sugaryield using combination ofenzymes $(C + \beta + xyl)$

A: Tem-	B: pH	C: Enzyme	D: Substrate	E: Incubation	R1			
(°C)		loading (mL)	concentration (g)	time (h)	Glucose (mg/mL)	Xylose (mg/mL)		
30	5	2	3.5	24	2.87	1.21		
30	4.6	3.5	3.5	72	8.91	4.39		
45	4.8	2.75	2.75	48	25.08	10.65		
60	4.6	2	3.5	24	2.92	1.42		
45	5	2.75	2.75	48	14.07	8.06		
30	5	3.5	3.5	72	8.65	3.64		
30	4.8	2.75	2.75	48	10.4	6.78		
60	5	3.5	3.5	72	6.87	3.02		
45	4.8	2.75	2	48	15.04	8.64		
60	4.6	2	2	24	2.38	1.76		
45	4.8	2.75	2.75	48	23.05	10.09		
60	4.6	3.5	2	24	2.32	1.23		
60	5	2	2	24	1.26	1.02		
60	5	2	3.5	24	3.02	1.72		
45	4.8	2.75	2.75	48	22.53	11.58		
30	5	3.5	3.5	24	2.65	1.02		
60	5	2	2	72	6.02	4.11		
30	5	3.5	2	24	2.67	1.06		
60	5	3.5	2	24	2.02	1.29		
30	4.6	2	2	24	3.09	2.03		
30	4.6	3.5	2	24	2.98	1.39		
60	4.6	2	2	72	8.22	3.04		
60	4.6	2	3.5	72	7.32	3.95		
45	4.8	2.75	2.75	48	24.86	10.02		
45	4.8	2.75	3.5	48	18.09	8.43		
30	4.6	2	2	72	8.34	3.67		
30	5	3.5	2	72	7.98	3.48		
30	5	2	2	72	7.83	3.05		
45	4.8	2.75	2.75	48	23.07	10.37		
60	5	2	3.5	72	5.62	3.01		
45	4.8	3.5	2.75	48	20.86	8.49		
45	4.8	2	2.75	48	18.95	8.07		
45	4.6	2.75	2.75	48	18.32	8.91		
30	5	2	3.5	72	6.47	3.06		
30	4.6	2	3.5	24	2.43	1.46		
60	4.6	3.5	3.5	24	2.98	1.26		
45	4.8	2.75	2.75	24	3.95	4.33		
60	5	3.5	3.5	24	2.52	1.42		
60	4.6	3.5	3.5	72	6.82	3.51		
30	4.6	2	3.5	72	8.65	4.03		
45	4.8	2.75	2.75	72	17.95	8.37		
30	5	2	2	24	2.32	1.05		
45	4.8	2.75	2.75	48	25.07	10.23		
60	4.8	2.75	2.75	48	9.07	7.03		
60	4.6	3.5	2	72	7.69	3.74		
30	4.6	3.5	3.5	24	2.21	1.33		
45	4.8	2.75	2.75	48	23.09	9.98		
60	5	3.5	2	72	6.32	3.47		
30	4.6	3.5	2	72	8.23	4.03		
45	4.8	2.75	2.75	48	20.05	10.07		
				-				

containing pretreated biomass resulted in an increase in total reducing sugar yield to 32.3 mg/mL with enzyme loading (C+ β) at 2.5 mL (v/v). Similarly, enzyme loading (C+ β +xyl) of 3 mL (v/v) achieved a 41.4 mg/mL total reducing sugar yield. Sugar production was not enhanced by an additional increase in enzyme load. Elevated quantities of cellulase have the potential to reverse saccharification by accelerating transglycosylation processes, causing hydrodynamic instability, inappropriate mixing and slurry suspension. The enzyme dose should be kept to a minimum because the expense of cellulase greatly increases the cost of the hydrolysis process (Alrumman 2016).

Effect of substrate concentration

Substrate concentration is another significant parameter affecting total reducing sugar yield during enzymatic hydrolysis. In the present study, pretreated SCB concentration achieved the maximum yield at 2.5 g (w/v %), which produced 32.9 mg/mL of total reducing sugar using cellulases (C+ β). However, adding xylanase (C+ β +xyl) achieved the optimum yield at a substrate concentration of 3 g (w/v %), which produced 43.87 mg/mL of total reducing sugar. In both cases, the optimum conditions for total reducing sugar production were determined after a 48-h incubation (Figs. 1d and 2d).

During saccharification, adding excess substrate may reduce sugar yield through end-product inhibition when combined with limited mass transfer within the reaction mixture, due to the high viscosity of the slurry. Other factors include a lower level of carbohydrate conversion at a high substrate loading and a decrease in the reaction of cellulose material due to a low quantity of enzyme loading. Thus, saccharification degree strongly depended on the enzyme–substrate ratio (Alrumman 2016). Several reports demonstrated an increase in substrate concentration leading to high sugar yield (Akhtar et al. 2001; Alrumman 2016; Pant et al. 2021). In enzymatic hydrolysis, converting polysaccharides to monosaccharides is a more complex process that depends on enzyme–substrate synergism (Rajak and Banerjee 2016).

Effect of incubation time

To enhance the rate of enzymatic hydrolysis, it is essential optimise critical process parameters such as temperature, pH, enzyme loading, substrate to liquid ratio and saccharification time. These process parameters contribute to the enzymatic hydrolysis of lignocelluloses into monomeric sugars (Vimala Rodhe et al. 2011; Sharma et al. 2015). Enzymatic hydrolysis using cellulases (C + β) and adding xylanase (C + β + xyl) achieved a maximum total reducing sugar yield of 34.31 mg/mL and 44.03 mg/ mL, respectively, after 48 h (Figs. 1e and 2e). Furthermore, increasing the duration of saccharification did not enhance total reducing sugar production. Previous studies reported a maximum reducing sugar yield after 48 h of incubation using different pretreated biomass and enzymatic cocktails from fungal sources (Sharma et al. 2017; Pimentel et al. 2021; Shankar et al. 2022; Infanzón-Rodríguez et al. 2023). Furthermore, according to Mahamud and Gomes (2012), factors such as pretreatment of the substrate, enzyme and substrate concentrations, product inhibition and enzyme stability affect the rate and degree of saccharification, with the rate of hydrolysis decreasing rapidly over time as a result.

Physicochemical characterisation of untreated, alkaline-pretreated and enzymatically hydrolysed SCB

SEM analysis

SEM analysis was carried out to study the morphological structure of SCB biomass under various treatments (Fig. 3). The morphology of untreated SCB biomass showed a compressed and rough structure (Fig. 3a). On the other hand, the morphology of alkaline-pretreated SCB biomass showed fragmentary deconstruction of the rigid structure of plant cell walls (Fig. 3b). However, enzymatically hydrolysed SCB biomass using cellulases $(C + \beta)$ and with addition of xylanase $(C + \beta + xyl)$ demonstrated significant disruption and exposure of the fibres (Fig. 3c and d). Hence, unprocessed organic matter was thoroughly packed and firm. After alkaline pretreatment, the composition was altered due to the formation of cavities and the release of hemicelluloses and lignin. This made the structure more open and accessible to enzymes. Several studies have described the morphological structures of lignocellulosic biomass based on SEM analysis (Lv et al. 2013; Chandel et al. 2014; de Oliveira et al. 2017; Pereira et al. 2016). Zhang et al. (2018) and Guilherme et al. (2017) revealed an increase in the surface area after pretreatment of SCB biomass by splitting fibres due to the removal of the lignin fraction and increased biomass digestibility with cellulases.

XRD analysis

After plotting 2θ versus the intensity of the cellulose signals of SCB samples (Fig. 4), the crystallinity index (*CrI*) was calculated from the intensity of the cellulose crystal peak at a 2θ value of 21.8 and the amorphous intensity at a 2θ value of 18.2. The intensities of crystalline cellulose, amorphous crystal, crystallinity index, and % crystallinity index are listed in Table 5. Untreated SCB had a *CrI* of 38.64% while



(e)

Fig. 1 Illustration of enzymatic hydrolysis $(C + \beta)$ effect of **a** temperature, **b** pH, **c** enzyme loading, **d** substrate concentration and **e** incubation time on total reducing sugar production

alkaline-pretreated SCB had a *CrI* of 58.14%. Hydrolysed pretreated SCB using cellulases $(C+\beta)$ had a highest *CrI* of 62.15% while $(C+\beta+xyl)$ hydrolysed pretreated SCB with

xylanase had a *CrI* of 61.66%. The *CrI* of cellulases (C + β) hydrolysed pretreated SCB was 60% higher than that of untreated SCB and 10% higher than alkaline-pretreated SCB.





(b)

(a)





Fig. 2 Illustration of enzymatic hydrolysis ($C + \beta + xyl$) effect of **a** temperature, **b** pH, **c** enzyme loading, **d** substrate concentration and **e** incubation time on total reducing sugar production

In this study, the combined effects of alkaline pretreatment and enzymatic hydrolysis contributed to increased *CrI*. The findings showed that pretreatment of SCB resulted in a greater removal of lignin and hemicellulose. Moreover, it has been previously documented that successful biomass pretreatment increases the amorphous nature of cellulose fibres, increasing their availability for enzymatic hydrolysis (Moura et al. 2018; Nath et al. 2021; Tavares et al. 2024).



Fig. 3 Scanning electron microscopy of SCB biomass: **a** untreated SCB showing rigid surface, **b** alkaline pretreated SCB showing cracked surface, **c** enzymatic $(C+\beta)$ hydrolysed pretreated SCB and **d** enzymatic $(C+\beta+xyl)$ hydrolysed SCB showing cavitation surface

Crystallinity index measures the comparative quantity of cellulose present in organic matter (Joshi et al. 2011; Ahvenainen et al. 2016). This parameter was greater than that of untreated biomass, indicating the partial removal of lignin or hemicelluloses and increased cellulose content in treated samples (Bernardinelli et al. 2015). These results are consistent with previous reports on switchgrass (Karp et al.

2015; Wyman et al. 2011), sugarcane bagasse (Zhang et al. 2018), the upper portions of sugarcane (Sindhu et al. 2013) and sweet sorghum bagasse (Zhang et al. 2011). Thus, alkaline pretreatment of SCB can be employed to improve delignification and cellulose recovery with increased cellulose exposure for effective enzymatic hydrolysis.

FTIR analysis

FTIR spectra of lignocellulosic materials reveal the molecular vibrations associated with peaks corresponding to cellulose, hemicelluloses and lignin. The observed bands and their assignments are summarised in Supplementary Table S1. Figure 5 shows FTIR spectra of untreated, alkaline-pretreated, cellulases $(C + \beta)$, and addition of xylanase $(C + \beta + xyl)$ hydrolysed SCB biomass. The structural characteristics of cellulose and hemicellulose were reflected by a prominent band at $1000 - 1200 \text{ cm}^{-1}$. The -C-O and C-O-C stretching vibrations of cellulose were reflected in peaks at 1053 - 1164 cm⁻¹. The bands at 897 – 898 cm⁻¹ corresponding to β -(1,4)-glycosidic linkage C-O-C stretching vibrations were strengthened by the transition from a crystalline to an amorphous structure of cellulose (Nath et al. 2021). However, the O-H stretching vibrations of hydrogen bonds were almost the same for untreated, pretreated and enzymatic hydrolysed residues based on 3406 - 3460 cm⁻¹ adsorption peaks. Following pretreatment, there were some peaks at 2898 - 2917 cm⁻¹ associated with the C-H stretching vibrations of methylene groups of cellulose (Semwal et al. 2023). In addition to the -C = O bond of the acetyl group in the hemicellulose structure, adsorption peaks at 1714 - 1731 cm⁻¹ were associated with ester linkages between hemicellulose and lignin. Considering the above, we determined that there were very few or no acetyl groups, and that the xylan content decreased following pretreatment. The corresponding bands became less intense due to hemicellulose deacetylation (Semwal et al. 2023; Hans et al. 2023; Kapoor et al. 2015). For untreated SCB, bands at 1514 cm⁻¹ and 1604 cm⁻¹ were attributed to the C = C stretching vibrations of aromatic rings. The lignin content was not determined in alkalinepretreated SCB, which indicates that lignin structures were hydrolysed during pretreatment. These results showed that cellulose and hemicellulose are more exposed following alkaline pretreatment of SCB.

Regression analysis and model fitting

The relative impact on glucose and xylose production is indicated by the regression coefficient. The equation indicates which factor had the greatest effect on glucose and xylose production. After 50 trials, optimisation approaches were made simpler by RSM. Interactions between five independent variables and dependent variables for the response surface quadratic model were assessed to determine the influence of individual and combined variables on glucose and xylose yield. Specifically, ANOVA was performed to explore the variability of each response variable for enzymatic hydrolysis $(C + \beta)$ of pretreated SCB (Supplementary Tables S2 and S3). The outcomes were determined using a second-order polynomial equation and the significance of regression coefficients. The models for quadratic regression for glucose and xylose are conveyed by the second-order polynomial equations, represented in Eqs. (1) and (2), respectively.

$$Glucose (mg/mL) = +18.65 + 0.32 * A - 0.50 * B + 0.096 * C$$

$$+ 4.412E - 003 * D - 0.18 * E + 0.11 * A * B$$

$$+ 0.22 * A * C + 0.10 * A * D + 0.58 * A * E$$

$$- 0.30 * B * C - 0.14 * B * D + 0.042 * B * E \qquad (1)$$

$$+ 0.073 * C * D + 0.28 * C * E + 0.13 * D * E$$

$$- 9.33 * A^{2} - 0.24 * B^{2} - 0.33 * C^{2} - 1.12 * D^{2}$$

$$- 2.50 * E^{2}$$

Xylose (mg/mL) = +7.11 - 0.082 * A - 0.29 * B + 0.10 * C

$$+ 0.086 * D + 7.941E - 003 * E - 0.059 * A * B$$

+ 0.057 * A * C + 3.438E - 003 * A * D
+ 0.17 * A * E - 0.033 * B * C - 0.14 * B * D (2)
+ 0.026 * B * E - 0.11 * C * D + 0.017 * C * E
+ 0.065 * D * E - 1.97 * A² - 0.80 * B² - 0.18 * C²
- 0.60 * D² - 1.17 * E²

A positive symbol placed before every factor in the abovementioned mathematical model signifies a synergistic influence, while an antagonistic effect is indicated by a negative symbol. If a coefficient represents any single factors, it reflects the impact of that particular factor on the glucose and xylose yield. These factors are temperature (A), pH (B), enzyme loading (C), concentration of substrate (D) and incubation time (E). The outcome was obtained with a coefficient representing two factors. The remaining are represented by second-order terms showing interactions between the two factors and the quadratic effect.

The results of ANOVA for enzymatic hydrolysis $(C + \beta + xyl)$ of pretreated SCB are depicted in Supplementary Tables S4 and S5 with respect to the response surface quadratic model. The response functions were applied with the help of regression analysis. Comparison between the five independent variables and the respective responses were outlined by the second-order polynomial equations for glucose and xylose, represented in Eqs. (3) and (4), respectively.

$$Glucose (mg/mL) = +20.58 - 0.39 * A - 0.43 * B + 0.15 * C$$

$$+ 0.13 * D + 2.74 * E - 0.11 * A * B$$

$$- 0.047 * A * C + 0.076 * A * D$$

$$- 0.26 * A * E + 0.17 * B * C + 0.10 * B * D$$
(3)
$$- 0.20 * B * E + 0.049 * C * D + 0.092 * C * E$$

$$- 0.12 * D * E - 8.08 * A^{2} - 1.62 * B^{2} + 2.09 * C^{2}$$

$$- 1.25 * D^{2} - 6.87 * E^{2}$$

Fig. 4 X-ray diffraction spectra of SCB biomass of untreated (red curve), alkaline-pretreated (green curve), $(C+\beta)$ enzymatic hydrolysed (purple curve) and $(C+\beta+xyl)$ enzymatic hydrolysed (blue curve)



Table 5 XRD analysis for crystallinity index (*CrI* %) of untreated, alkaline-pretreated, hydrolysed using cellulases (C+ β) and incorporated with xylanase (C+ β +xyl) SCB biomass

Biomass	<i>I</i> ₀₀₂	I _{am}	CrI (%)	Percentage increase in crystallinity index compared to untreated sample
Untreated	3960	2430	38.64	0.00
Alkaline-pretreated	5160	2160	58.14	50.48
Enzymatic hydrolysed (C + β)	5310	2010	62.15	60.85
Enzymatic hydrolysed $(C + \beta + xyl)$	5790	2220	61.66	59.59

$$Xylose (mg/mL) = +9.78 - 0.020 * A - 0.19 * B$$

$$+ 3.235E - 003 * C - 5.294E
- 003 * D + 1.16 * E + 0.12 * A * B
- 0.058 * A * C - 0.023 * A * D
- 0.065 * A * E + 0.020 * B * C
- 0.028 * B * D - 0.045 * B * E
- 7.188E - 003 * C * D + 0.095 * C * E
+ 3.125E - 004 * D * E - 2.28 * A2
- 0.70 * B2 - 0.90 * C2 - 0.65 * D2
- 2.83 * E2
(4)$$

Correlations between *p* value and respective coefficients were found to be inversely proportional. The models were significant (*p* < 0.0001) and consistent with the quadratic models in terms of correlation between the responses and the independent variables (Supplementary Tables S2–S5). Models were capable of predicting glucose and xylose yields based on *p* values and lack of fit. The higher coefficient (R^2) and adjusted (R^2) values show the accuracy of the models. The coefficient (R^2) was used to derive the fitness of the model, giving 97.5% for glucose and 94.6% for xylose from cellulases (C+ β) hydrolysed SCB biomass, and their adjusted (R^2) values were 95.7% and 90.9%, respectively. In xylanase (C+ β +xyl) hydrolysed SCB, the fitness of the model was determined by coefficients (R^2) of 92.7% and 97.8% for glucose and xylose, respectively, and their adjusted (R^2) values were 87.6% and 96.3%, respectively. For both glucose and xylose, the results indicate that the developed model fitted the experimental data very well.

Model validation

RSM yielded the required theoretical models for releasing glucose and xylose based on a derived set of input variables. To validate the model, a very efficient theoretical model was used, and to observe the interfaces between variables, plots with 3D surfaces were employed, as depicted in Figs. 6 and 7. The interactive effects of variables on the production of reducing sugar were investigated by generating 3D response surface plots against any two independent factors and their corresponding sugar production while maintaining other variables at their central (0) level. Hydrolysis of SCB using cellulases $(C+\beta)$ exhibited an optimal substrate concentration at 2.54% (w/v), pH 4.6, enzyme loading 2.81 mL, temperature 45.2 °C and processing duration 47.9 h. The experimental glucose yield was 18.72 mg/mL, close to the predicted value of 19.04 mg/ mL. Similarly, for xylose, the optimal value of substrate concentration was achieved at 2.54% (w/v), pH 4.7, enzyme loading 2.6 mL, temperature 44.8 °C and processing duration 48 h. The experimental yield of xylose was 6.98 mg/mL, very close to the predicted value of 7.14 mg/mL, as shown in Table 6.



Fig. 5 Fourier transform infrared spectra of SCB biomass comparing **a** untreated, **b** alkaline pretreated, **c** enzymatic hydrolysed $(C+\beta)$ and **d** enzymatic hydrolysed $(C+\beta+xyl)$

Hydrolysis of SCB biomass following addition of xylanase $(C + \beta + xyl)$ had an optimal substrate concentration at 2.79% (w/v), pH 4.7, enzyme loading 3.5 mL, temperature 44.5 °C and processing duration 53 h. The experimental glucose yield was 22.83 mg/mL, very close to the predicted value of 23.13 mg/mL. On the other hand, for xylose yield, the optimal substrate concentration was at 2.75% (w/v), pH 4.7, enzyme loading 2.76 mL, temperature 44.8 °C and processing duration 52.9 h. The maximum yield of xylose was 9.54 mg/mL, very close to the predicted 9.91 mg/mL (Table 7). The observed values of glucose and xylose closely matched the predicted values, and the experimental results under ideal conditions matched the predictions relatively well, providing strong validation for the RSM models. Comparison of experimental and predicted values confirmed that the experimental strategy can efficiently maximise variables in the process optimisation, and examine the significance of individual, combined and cooperative impacts of test variables in the enzyme-assisted hydrolysis process.

Saccharification of dennanath and hybrid napier grass showed an increase in total reducing sugar using cellulases from *A. fumigatus* (CWSF-7) and commercial xylanase (Mohapatra et al. 2018). Moretti et al. (2014) demonstrated the enzymatic hydrolysis of pretreated SCB using endoglucanase, β -glucosidase and xylanase from *Myceliophthora thermophila* M.7.7., with enhanced glucose and xylose yields. A previous study on hydrolysis involving enzymatic cocktails from *A. fumigatus* SCBM6 and *A. niger* SCBM1 showed greater efficiency in depolymerising hemicellulose fractions (de Oliveira Rodrigues et al. 2017). Enzymatic hydrolysis of sugarcane bagasse using filter paper units (FPU), CMCase and β -glucosidase of *A. niger* ITV02 achieved a higher glucose yield (Infanzón-Rodríguez et al. 2023). The study showed that alkaline pretreatment of corn



Fig. 6 Effect of variables on enzymatic hydrolysis $(C + \beta)$ of alkaline pretreated SCB biomass on glucose (a) and xylose (b) yield

stover powder and enzymatic hydrolysis were optimised using RSM to maximise glucose and xylose yields (Zhang and Wu 2021). Moreover, hydrolysates have potential applications in various fields for high-value-added products and advanced biofuels such as butanol. Specifically, xylose can produce xylitol, lactic acid, acetic acid and biodegradable plastics (de Cassia Pereira et al. 2016; de Arruda et al. 2017; Mardawati et al. 2022; Devi et al. 2022; Chauhan et al. 2023). Enzymes are always specific for their particular substrates, and incorporating xylanase increases the overall efficiency of cellulases. Supplementation of xylanases increases the solubility of xylan within substrates (De Figueiredo et al. 2017; Mohapatra et al. 2018). Thus, it is essential to understand the mechanisms within reaction mixtures containing biomass, cellulases and xylanases. The present study revealed that purified cellulases can have a significant effect alongside supplementing xylanase. This strategy can be implemented for efficient enzymatic hydrolysis of different biomasses.



Fig. 7 Effect of variables on enzymatic hydrolysis $(C+\beta+xy)$ of alkaline pretreated SCB biomass on glucose (a) and xylose (b) yield

Table 6Optimum processvariables for maximum reducingsugar yield using combination	Tempera- ture (°C)	рН	Enzyme load- ing (mL)	Substrate conc. (g)	Incubation time (h)	Predicted Glucose (mg/mL)	Experimental
of enzymes $(C + \beta)$	45.2	4.60	2.81	2.54	47.9	19.04 Xylose (mg/mL)	18.72 (±0.96)
	44.8	4.76	2.64	2.54	48.0	7.14	6.98 (±0.81)
Table 7Optimum processvariables for maximum reducingsugar yield using combination	Temperature (°C)	рН	Enzyme load- ing (mL)	Substrate conc. (g)	Incubation time (h)	Predicted Glucose (mg/mL)	Experimental
of enzymes $(C + \beta + xyl)$	44.5	4.7	3.50	2.79	53.0	23.13 Xylose (mg/mL)	22.83 (±0.11)
	44.8	4.7	2.76	2.75	52.9	9.91	$9.54 (\pm 0.48)$

Conclusion

We investigated the saccharification of alkaline-pretreated SCB using hydrolytic enzymes from A. fumigatus JCM 10253. The saccharification optimisation strategy showed that combining CMCase, β-glucosidase and xylanase increased enzymatic hydrolysis efficiency and achieved a higher total reducing sugar yield. The central composite design achieved maximum glucose (22.83 mg/mL) and xylose (9.54 mg/mL) production. Therefore, process optimisation with statistical designs can effectively recover both glucose and xylose sugars from SCB. SEM analysis confirmed the cell wall disintegration of SCB by the alkaline pretreatment, which improved enzyme accessibility and adsorption by surface area. Diffractograms of SCB following different treatments demonstrated the conspicuous expansion of crystallinity in which a combination of pretreated and enzymatic $(C + \beta)$ hydrolysis SCB produced a maximum CrI of 62.15%, compared with 38.64% for untreated SCB and 58.14% for alkaline-pretreated SCB. This indicated that pretreatment of SCB with 0.5 M sodium hydroxide achieved greater removal of lignin and hemicelluloses, and was simultaneously an effective pretreatment for enhancing enzymatic saccharification. FTIR analysis of pretreated SCB revealed more prominent cellulosic peaks than for untreated SCB. Furthermore, alkaline pretreatment and combined enzyme mixtures can be effective and economically beneficial for delignification, cellulose exposure and enhanced enzymatic hydrolysis to produce fermentable sugars, which could be further exploited for biofuel production.

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Author contribution All authors contributed to the study's conception and design. Paramjeet Saroj performed material preparation, data collection and analysis. Paramjeet Saroj wrote the first draft of the manuscript. Korrapati Narasimhulu and Manasa P commented on previous

versions of the manuscript. All authors read and approved the final version of the manuscript.

Data availability The authors declare that the data supporting the findings of this study are available within the article and its Supplementary Information files.

Declarations

Ethics approval and consent to participate Not applicable.

Consent for publication Not applicable.

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