RESEARCH ARTICLE



Effects of cadmium, lead, mercury, chromium, and selenium co-treatment on egg quality and fatty acids

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Abstract

This study aimed to reveal the effect of selenium (Se) and heavy metals (chromium (Cr), cadmium (Cd), lead (Pb), and mercury (Hg)) on the quality, fatty acids, and 13 kinds of ions in the egg yolk and albumen. Four experimental groups were established, including a control group (control; basal diet), Se group (basal diet + Se), heavy metals group (basal diet + Cd $Cl_2 + Pb(NO_3)_2 + HgCl_2 + CrCl_3$), and Se + heavy metal (HM) group (basal diet + Se + CdCl_2 + Pb(NO_3)_2 + HgCl_2 + CrCl_3). Se supplementation significantly increased the experimental egg yolk percentage since Se accumulation mainly occurred in the yolks of the eggs. The Cr content in the yolks of the Se + heavy metal groups decreased at 28 days, while a significant reduction was evident in the Cd and Hg levels of the Se + heavy metal yolks compared to the heavy metal group at 84 days. The complex interactions between the elements were analyzed to determine the positive and negative correlations. Se displayed a high positive correlation with Cd and Pb in the yolk and albumen, while the heavy metals minimally affected the fatty acids in the egg yolk.

Keywords Yolk · Albumen · Selenium · Heavy metals · Fatty acids

Introduction

Selenium (Se) is an essential nutritional micro-element for many organisms and is indispensable for maintaining human health. Se displays antioxidant properties and participates in free radical removal and carcinogenic resistance (Pilarczyk et al. 2019). It is a vital component of glutathione peroxidase (GSH-Px), an antioxidant enzyme mainly responsible for preventing reactive oxygen species (ROS) formation while participating in the metabolic regulation of various hormones (such as thyroid hormones) (Muhammad et al. 2021). Se deficiency is associated with a compromised immune system and increased susceptibility to various disorders, including cancer, cardiovascular disease, diabetes,

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Kehong Liang liangkehong@caas.cn and immunodeficiency (Fisinin et al. 2009; Rayman 2012; Vinceti et al. 2018). The low dietary Se levels evident in regions of China, New Zealand, Europe, and Russia (Kipp et al. 2015) have necessitated the development of Seenriched food, such as supplemented eggs, which is produced by adding Se to layer feed (Liu et al. 2022). These types of products may provide a solution for the widespread dietary Se deficiency.

Studies have shown that adding organic Se to the diets of chickens substantially increases Se accumulation in eggs (Payne et al. 2005; Skrivan et al. 2008). Liu et al. (2020) reported that adding Se to the diets of laying hens increased egg production, while Se yeast supplementation was more successful in raising the Se content in egg yolk than sodium selenite. Urso et al. (2015) indicated that the Se content in egg yolk increased when adding 0.15 to 0.3 mg/kg Se to the diet. Besides increasing the accumulation in eggs, Se also improves antioxidant ability while displaying higher levels in other tissues (Pappas et al. 2006a, 2006b; Liu et al. 2020). Ou et al. (2017) showed that adding Se nanoparticles to the feed of laying hens effectively increased the anti-oxidative protection against deoxynivalenol toxicity. Scheideler et al. (2010) indicated that supplementing the diets of laying hens with vitamin E and Se improved the

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vitelline membrane strength of fresh and aged eggs. Antioxidants, such as vitamin E and Se, provide crucial protection against lipid peroxidation (Pappas et al. 2005). As an anti-oxidative form of the active center of GPX, Se helps to lower cholesterol levels. Therefore, dietary Se supplementation significantly decreases the total egg yolk cholesterol (Muhammad et al. 2021).

Se is also vital in counteracting tagonizing metal toxicity. It restricts the oxidative stress induced by cadmium (Cd), protecting mammals against toxicity (Zhang et al. 2020a, b). Furthermore, Se supplementation reduces lead (Pb) toxicity and lipid peroxidation by increasing the GPX activity (Adel et al. 2016). It may act as an endogenous "stop signal" for the carcinogenic cell signals (such as AP-1 and NFkB activation) induced by arsenic (As), reducing its toxicity (Zeng et al. 2005). Moreover, Se and zinc (Zn) display a synergistic effect in resisting Cd and Pb toxicity. Luo et al. (2020) indicated that Se counteracted Cd, Pb, and chromium (Cr) while exhibiting a synergistic relationship with Zn and Copper (Cu). Cao et al. (2014) showed that Se, Zn, and silicon (Si) significantly decreased Cd, Cr, Cu, and Pb proliferation in heavy metal-contaminated grain with no effect on yield. The effect of Se against mercury (Hg) is complex and depends on several features, such as the form of Hg, the form of Se, the organ, and the dose (Spiller 2018). Se-enriched eggs are produced in over 25 countries (Fisinin et al. 2009). Since eggs represent a dietary staple food worldwide, the Se and heavy metals transferred to eggs via animal feed can easily reach humans. However, no studies are available regarding the combative effect of Se-enriched eggs against heavy metal toxicity. Therefore, this study explores the relationship between Se and other elements, providing a basis for producing eggs safe for consumption and Se-enriched eggs while discussing egg quality and fatty acid content.

Materials and methods

Experimental design and diets

The experimental procedures were approved by the Animal Care and Use Committee of Sichuan Agricultural University. A total of 160 56-week-old Lohmann laying hens were weighed and randomly divided into four groups (ten replicates per treatment, four hens per replicate), namely, the control group fed a basal diet, the Se group fed a basal diet supplemented with 0.4 mg/kg Se yeast, the heavy metal group fed a basal diet supplemented with heavy metals (5 mg/kg CdCl₂ + 50 mg/kg Pb(NO₃)₂ + 3 mg/kg HgCl₂ + 5 mg/kg CrCl₃), and the Se + heavy metal group (Se + HM) fed a basal diet supplemented with Se and heavy metals (5 mg/kg CdCl₂ + 50 mg/kg Pb(NO₃)₂ + 3 mg/kg HgCl₂ + 5 mg/kg CdCl₂ + 50 mg/kg Pb(NO₃)₂ + 3 mg/kg HgCl₂ + 5 mg/kg CdCl₂ + 50 mg/kg Se yeast). The diets were formulated

according to the published NRC (1994) recommendations and the Lohmann hen manual, while the feed was administered in mashed form. The entire feeding trial lasted 84 days. The composition and nutrient levels of the basal diet are presented in Table S1, while the metal ion content of presented in Table S2.

Egg quality determination

Egg samples were collected at the end of the supplementation period to analyze the egg quality. The eggshells were measured using the Egg Shell Force Gauge Model II, while shell strength was determined with an eggshell thickness gauge (Robotmation Co., Ltd, Tokyo, Japan). The color values (brightness (black/white) (L^*), brightness (black/white) (a^*) and chroma value (blue/yellow) (b^*)) were measured using a Minolta colorimeter (Konica Minolta Sensing Inc., Osaka, Japan). The albumen height, yolk color, and Haugh units (HU) were analyzed via an Egg Multi-tester (EMT-7300, Robotmation Co., Ltd, Tokyo, Japan). The yolk index value was calculated as $100 \times$ (yolk height (cm)/yolk diameter (cm)), while the yolk ratio was determined as yolk weight (g)/egg weight (g) \times 100. The albumin ratio was computed as $100 \times$ [albumin weight (g)/egg weight (g)].

Metal element analysis

Here, 1.5 mL nitric acid (HNO₃) was added to 0.5 g of the sample, after which the digestion temperature was raised to 80 °C, where it was maintained for 8 h. The acid was then expelled via the open tank cover at 150 °C. After complete evaporation of the acid, 2 mL HNO₃ was added and digested for a further 2 h at 150 °C. Following digestion, the sample was diluted to 50 mL with deionized water and analyzed using inductively coupled plasma mass spectrometry (ICP-MS). The instrumental parameters of the equipment used were described in previous studies (Zhao et al. 2021). The recovery via ICP-MS validation ranged from 91.6 to 98.3%, indicating the accuracy of the method.

Lipid profile evaluation

The fatty acids in the yolk were analyzed using a technique described by Batkowska et al. (2021). The fatty acid methyl esters were assessed via a standard gas chromatography method (Model 7890A, Agilent Technologies, Palo Alto, CA, USA) using an SP-2560 column (100 m × 0.25 mm ID Supelco Inc., Bellefonte, PA, USA) with a flame ionization detector. The initial column oven temperature was 100 °C, which was increased at 10 °C min⁻¹ to 180 °C, where it was maintained for 6 min. Then, the temperature was increased at 1 °C min⁻¹ to 200 °C, maintained for 20 min, and finally elevated to 230 °C at 4 °C min⁻¹. The injector

and detector temperatures were maintained at 270 $^{\circ}$ C and 280 $^{\circ}$ C, respectively.

Statistical analysis

The results were statistically analyzed using two-way and one-way analysis of variance (ANOVA), followed by post hoc Tukey's test using SPSS version 19.0. The data were expressed as the means \pm standard deviation and were assumed statistically significant at P < 0.05. The relationships between the metal elements were estimated via Pearson's correlation coefficients.

Results and discussion

The effects of Cd, Pb, Hg, Cr, and Se co-treatment on the egg quality

The external and internal qualities of the fresh eggs from the hens provided with different treatment diets are shown in Table 1. The highest yolk percentage and lowest eggshell color were identified in the Se groups (P < 0.05). The absolute nutritional value of the egg yolk was relatively high, while its size directly affected the egg flavor (Zhang et al. 2021). Therefore, eggs with large yolks are more popular among consumers (Fernandez and Andersen 2015). Gao et al. (2021) found that Se supplementation significantly increased egg yolk percentage. Eggshell color is an important indicator affecting the price of eggs (Wei and Bitgood 1990). In addition to biliverdin, protoporphyrin IX has been identified as a main determinator of eggshell color (Miksik et al. 1996). Se accumulation may affect the expression of the synthetic pigment fund, increasing the color of the eggshell. The HUs of the control and Se groups were higher (P < 0.05) than the heavy metal and Se + HM treatment groups. The treatment did not significantly affect the weight and color of the yolk or the strength and thickness of the eggshell (P > 0.05). Mutlu et al. (2021) revealed a significant HU decrease in quails exposed to Pb. Since HUs are vital indicators of protein quality, higher content levels indicate superior internal egg quality. Therefore, Pb exposure may lead to liver cell toxicity and liver protein synthesis damage, reducing albumen protein production. Ovomucin represented the key protein responsible for egg albumen viscosity and elevated HU levels (Omana et al. 2010). However, Kim et al. (2019) indicated no statistical differences between the external and internal egg quality parameters measured in the laying hens exposed to feed containing heavy metals. This variation in results may be due to the different heavy metal doses.

The Se and heavy metals in the egg yolk and albumen

The Se and heavy metal content in the eggs during the laying period is presented in Fig. 1. The Se content in the egg yolks of the control group increased from 27.80 to 46.87 μ g/g, followed by a decrease to 34.07 μ g/g, while displaying higher levels in the Se and Se + HM groups at 6.38-14.88 µg/g and 5.43-12.21 µg/g, respectively. No significant differences (P > 0.05) were evident between the Se content of the heavy metal group at the beginning and end of the feeding period. The experimental results of this study are consistent with those of previous research. For example, Utterback et al. (2005) reported that the Se content in the eggs of hens fed a basal diet decreased as the feeding period was extended. Pan et al. (2011) indicated that the Se content in the eggs exposed to different Se treatments was significantly higher at 18 days than at 14 days and 21 days. The Se content in the albumen of the control, Se, heavy metal, and Se+HM treatment groups increased from 10.89–17.85 µg/g, 21.75–33.61 µg/g, 10.56–17.17 µg/g, and 20.44–31.11 μ g/g, respectively. The mean Se content was higher in the yolk than the albumen in all the treatment groups, which was consistent with research by Pilarzyk et al.

Table 1	The egg quality after
an 84-da	ay feeding period

Parameters	Control $(n = 40)$	Se $(n = 40)$	Heavy metals $(n=40)$	Se + HM (n = 40)
Yolk weight (g)	17.46 ± 0.82	17.42 ± 0.86	17.22 ± 0.80	17.56 ± 0.92
Yolk percentage (%)	$27.08 \pm 0.62 \mathrm{b}$	$28.25 \pm 0.52 \mathrm{a}$	$27.14 \pm 0.67b$	26.89 ± 0.61 b
Yolk color	7.04 ± 1.02	7.85 ± 1.12	6.96 ± 1.27	7.00 ± 1.32
Haugh unit	$90.23 \pm 3.82a$	$91.02 \pm 4.36a$	$84.79 \pm 4.82b$	$84.57 \pm 4.54b$
Albumin height (mm)	$8.40 \pm 1.12a$	7.90 ± 1.21 ab	$7.24 \pm 0.83b$	$6.84 \pm 1.22b$
Eggshell strength (kg/cm ²)	3.90 ± 0.69	3.84 ± 0.72	3.81 ± 0.60	3.52 ± 0.58
Eggshell thickness (mm)	0.36 ± 0.02	0.37 ± 0.02	0.36 ± 0.03	0.37 ± 0.03
Eggshell color ($L^*a^*b^*$)	$74.02 \pm 3.46a$	$68.32 \pm 3.63b$	$74.89 \pm 3.82a$	74.54±3.36a

 $L^*a^*b=L^*-a^*-b^*$ (the lower the value, the darker the egg), $L^*=$ brightness (black/white), $a^*=$ chroma value (green/red), and $b^*=$ chroma value (blue/yellow). Values with different letters indicate significant differences (P < 0.05). Values are expressed as means \pm standard deviation



Fig. 1 The Se, Cd, Cr, Hg, and Pb contents in the eggs. The error bars correspond to standard errors. Different lowercase letters indicate significant differences on different days, while different capital letters indicate significant differences between the treatments

(2019), who found that the yolk displayed higher Se levels than the albumen, presenting a more potent Se source. After Se supplementation, the Se content was two-fold higher than the unsupplemented yolk and albumen, while the increase was statistically significant. Zhang et al. (2020a, b) showed that the Se content of the entire egg increased when adding 0.25 mg/kg Se to the diet.

The Cr content in the yolk of all the experimental groups was affected over time, while the Cd levels remained unaffected, and the Hg content decreased. However, although the Pb levels displayed no differences in the heavy metal and Se + HM groups, they increased in the control and Se groups. The Cd, Pb, and Hg content in the yolk was not affected by dietary Se addition, while the Cr levels in the yolk of the group supplemented with Se for 28 days were higher than in the control group. These findings were consistent with the results of previous studies (Zhao et al. 2021), showing a positive correlation between Cr and Se. Moreover, Se + HM group for 28 days was decreased than the heavy metal group; these could be due to the Se could alleviate the toxic effects of Cr (Liu et al. 2018). The Cr, Cd, and Pb levels in the albumen were affected over time. Compared to the Se group, the Cr content was higher in the Se + HM group at 84 days, while the Pb levels were higher at 28 days, 56 days, and 84 days (P < 0.05), respectively. Previous studies have revealed a correlation between Se and

Table 2 The content of the major, minor, and trace elements in the egg yolk

Element Days		Treatment					T*D
		Control $(n=40)$	Se $(n = 40)$	Heavy metals $(n=40)$	Se + HM (n = 40)		
Major elements							
Na (mg/g)	28	33.49±4.76Aa	39.57 ± 3.96 Aa	34.47 ± 8.28 Aa	35.71±6.74 Aa	ns	ns
	56	35.92±4.79 Aa	31.64±6.74 Ab	33.44±8.55 Aa	34.89±7.28 Aa	ns	
	84	31.37 ± 4.94 Aa	26.23 ± 5.64 Ab	28.40 ± 9.29 Aa	28.00 ± 5.73 Ab	ns	
	Sig	ns	***	ns	*		
Mg (mg/g)	28	9.36 ± 1.59 Ab	$10.55 \pm 0.92 \text{Aa}$	$10.00 \pm 1.87 \text{Aa}$	10.36±2.47Aa	ns	*
	56	11.19±2.15Ab	8.75±1.99Aa	9.99±2.92Aa	9.87 ± 2.29Aa	ns	
	84	14.65 ± 3.10 Aa	10.47 ± 2.29Ba	10.85 ± 2.51 Ba	11.28±3.49Ba	*	
	Sig	***	ns	ns	ns		
K (mg/g)	28	$76.60 \pm 11.95 \text{Ab}$	87.93±7.63Aa	79.86±17.32Aa	84.80 ± 17.12 Aa	ns	*
	56	90.71±16.41Aa	72.03 ± 12.62Ab	82.08±19.06Aa	81.49 ± 18.71Aa	ns	
	84	$12.02 \pm 1.41 \text{Ac}$	$10.22 \pm 2.16 \text{Ac}$	$11.08 \pm 2.45 \text{Ab}$	10.35 ± 1.94 Ab	ns	
	Sig	***	***	***	***		
Ca (mg/g)	28	$14.51 \pm 2.27 Aa$	15.73±1.45Aa	15.19 ± 3.08 Aa	16.48±6.74Aa	ns	ns
	56	18.86±7.33Aa	13.47 ± 3.21 Aa	17.32 ± 8.10 Aa	16.36 ± 6.08 Aa	ns	
	84	15.95 ± 2.30 Aa	13.10±2.77Aa	$14.29 \pm 3.90 \text{Aa}$	13.46 ± 3.01 Aa	ns	
	Sig	ns	ns	Ns	ns		
Minor elements							
Mn (mg/g)	28	$0.09 \pm 0.02 \text{Ab}$	0.12 ± 0.02 Ab	0.10 ± 0.03 Ab	$0.11 \pm 0.05 \text{Ab}$	ns	ns
	56	0.12 ± 0.04 Ab	0.10 ± 0.04 Ab	$0.11 \pm 0.05 \text{Ab}$	0.10 ± 0.03 Ab	ns	
	84	1.43 ± 0.24 Aa	1.06±0.23Ba	1.21±0.35ABa	1.09 ± 0.20 Ba	**	
	Sig	***	***	***	***		
Fe (mg/g)	28	5.04 ± 1.12 Aa	5.32 ± 0.79 Aa	5.18±0.99Aa	5.19 ± 1.21 Aa	ns	ns
	56	5.75 ± 0.98 Aa	4.32 ± 0.93 Bb	5.03±1.37ABa	$4.98 \pm 0.98 \text{ABa}$	*	
	84	2.37 ± 0.54 Ab	$1.63 \pm 0.36 Bc$	1.91±0.51ABb	1.75 ± 0.43 Bb	***	
	Sig	***	***	***	***		
Ni (µg/g)	28	18.90 ± 7.63 Bb	85.70±53.67Aa	73.21 ± 41.05Aa	73.58±47.38Aa	**	*
	56	96.65±79.29Aab	59.02 ± 48.02Aa	27.65±13.72Ab	22.70 ± 3.51 Aa	ns	
	84	138.05±91.27Aa	74.28 ± 43.74Aa	64.57 ± 12.08Aa	70.65 ± 55.09 Aa	ns	
	Sig	*	ns	***	ns		
Cu (mg/g)	28	0.12 ± 0.02 Aa	0.13 ± 0.01 Aa	0.12 ± 0.03 Aa	0.13 ± 0.03 Aa	ns	*
	56	0.14 ± 0.03 Aa	0.11 ± 0.03 Aab	0.13 ± 0.03 Aa	0.12 ± 0.03 Aa	ns	
	84	0.13 ± 0.03 Aa	$0.09\pm0.02\mathrm{Ab}$	$0.10 \pm 0.02 Aa$	0.10 ± 0.03 Aa	ns	
	Sig	ns	**	ns	ns		

Table 2 (continued)

Element	Days	Treatment					T*D
		Control $(n=40)$	Se $(n = 40)$	Heavy metals $(n=40)$	Se + HM (n = 40)		
Zn (mg/g)	28	3.63±0.63Aa	4.03 ± 0.52 Aa	3.89±0.85Aa	$3.66 \pm 0.53 A$	ns	ns
	56	3.94±0.64Aa	3.04 ± 0.59 Bb	3.42 ± 0.77 ABa	$3.43 \pm 0.67 \text{AB}$	*	
	84	3.79±0.67Aa	$3.10 \pm 0.61 \text{Ab}$	3.74 ± 1.08Aa	$3.24 \pm 0.73 A$	ns	
	Sig	ns	**	ns	ns		
Mo (µg/g)	28	23.45 ± 7.21Aa	16.81±3.99Bab	15.12±5.11Ba	$13.63 \pm 3.45B$	**	*
	56	15.53 ± 6.44 Ab	14.38 ± 2.98 Ab	16.59±4.39Aa	$14.95 \pm 4.37 \text{A}$	ns	
	84	18.70±4.64Aab	$20.22\pm6.96\mathrm{Aa}$	18.74 ± 4.84 Aa	$19.54 \pm 9.04 \mathrm{A}$	ns	
	Sig	**	*	ns	ns		
Trace elements							
As (µg/g)	28	0.86±0.51Aa	0.84 ± 0.55 Aa	0.72 ± 0.49 Aa	0.48 ± 0.32 Aa	ns	ns
	56	0.36 ± 0.19 Ab	0.34 ± 0.14 Ab	0.40 ± 0.24 Aa	0.36 ± 0.18 Aa	ns	
	84	0.39 ± 0.10 Ab	0.28 ± 0.11 Ab	$0.36 \pm 0.17 Aa$	0.32 ± 0.14 Aa	ns	
	Sig	*	*	ns	ns		
Co (µg/g)	28	0.45 ± 0.13 Ab	0.75 ± 0.25 Ba	0.69 ± 0.23 Bb	0.48 ± 0.12 Ab	**	**
	56	$0.88 \pm 0.47 \text{Ab}$	0.54±0.22Aa	$0.62 \pm 0.25 \text{Ab}$	0.54 ± 0.23 Ab	ns	
	84	17.92±10.37Aa	8.26±4.94Bb	7.11±1.26Ba	7.28 ± 5.70 Ba	*	
	Sig	***	***	***	***		
V (µg/g)	28	$0.11 \pm 0.05 \text{ABb}$	0.19 ± 0.10 Ab	$0.18 \pm 0.04 \text{Ab}$	$0.07 \pm 0.02 Bb$	*	*
	56	0.26 ± 0.14 Ab	0.19 ± 0.18 Ab	0.20 ± 0.17 Ab	0.37 ± 0.00 Ab	ns	
	84	1.20 ± 0.45 Aa	1.79±0.88Aa	1.89±1.37 Aa	1.03±0.43 Aa	ns	
	Sig	***	***	*	**		

The values are expressed as means \pm standard deviation. The mean values in the same row (different treatments on the same day) with different capital letters indicate significant differences. The mean values in the same column (the same treatment on different days) with different lower-case letters indicate significant differences. -: not detected; ns: not significant, P > 0.05; Sig.: significance; T^{*}D: the significance of the interaction between the treatment and days; *P < 0.05; **P < 0.01; ***P < 0.001

Pb in both humans and animals (Yoo et al. 2002; Zhao et al. 2021). Zhang et al. (2017) reported that Se supplementation increased the Cr, Mn, and Zn contents in the kidneys of chickens. Although the Hg content was not affected over time, differences were apparent between the batches since the Se + HM group differed (P < 0.05) from the others, presenting the lowest values at 28 days (0.09 µg/g), while it was no longer detectable at 56 days and 84 days. Methyl Hg represented the predominant form of Hg in the eggs (Ackerman et al. 2013) and was readily absorbed in the intestine and accumulated in the body. It is considered more toxic than inorganic Hg (Fant et al. 2001). Se minimized the hazardous effect of Hg(II) and MeHg (Luque-Garcia et al. 2013).

The elemental content in the egg yolk and albumen

The content of four major elements (Na, Mg, K, and Ca), six minor elements (Mn, Fe, Ni, Cu, Zn, and Mo), and three trace elements (As, Co, and V) in the yolk (Table 2) and albumen (Table 3) were examined to determine the effect of Se, heavy metals, and their combined treatment

on ion homeostasis and disturbance. The egg yolks of all the batches displayed a significant decrease (P < 0.05) in the levels of K and Fe over time and a significant increase (P < 0.05) in the Mn, Co, and V contents, while the Ca content was not affected. The Na, Cu, Zn, and As content in the Se group declined over time. The Mg content in the control group displayed a significant decrease (P < 0.05), while the Mg, Mn, Fe, and Co levels presented higher values (14.65 mg/g, 1.43 mg/g, 2.37 mg/g, and 17.92 µg/g) at 84 days. The Mg, Ca, Fe, Cu, and Co values in the albumen of all the batches were highest at 56 days, while the Na and K levels declined over time. No significant differences were evident in the Na, K, Ca, Mn, and Cu contents of each batch. Compared with the control group, the three minor elements (Fe, Zn, and Mo) decreased significantly in the other groups at 28 d (P < 0.01). The Mg and Mo levels were reduced in the heavy metal group, while the As content displayed significantly increased levels compared with the control group at 56 d (P < 0.05). The Ni content was significantly lower in the Se + HM group than in the heavy metal group (P < 0.01). Heavy metal poisoning affects the heavy metal content and other

Table 3 The content of the major, minor, and trace elements in the egg albumen

Element	Days	Treatment				Sig	T*D
		Control (n=40)	Se (n=40)	Heavy metals (n=40)	Se + HM (n=40)		
Major elements							
Na (mg/g)	28	126.99±11.37Aa	120.41 ± 11.53 Aa	123.29±9.76Aa	120.00 ± 11.30 Aa	ns	ns
	56	92.93±18.85Ab	90.30 ± 9.47 Ab	84.53±12.31Ac	81.94 ± 12.79Ac	ns	
	84	92.81 ± 14.72Ab	88.36±11.53Ab	97.11±10.38Ab	$95.54 \pm 9.67 \text{Ab}$	ns	
	Sig	***	***	***	***		
Mg (mg/g)	28	7.26 ± 0.89 Ab	$7.17 \pm 0.70 \text{Ab}$	$7.54 \pm 0.76 \mathrm{Ac}$	7.21 ± 0.88 Ab	ns	***
	56	20.45 ± 5.51 Aa	20.75 ± 9.24 Aa	14.67±3.94ABa	12.09±3.28Ba	**	
	84	9.87±1.36Ab	$9.44 \pm 2.30 \text{Ab}$	$10.92 \pm 1.62 \text{Ab}$	$10.43 \pm 2.65 Aa$	ns	
	Sig	***	***	***	**		
K (mg/g)	28	93.59±9.52Aa	91.01 ± 10.14Aa	91.58±9.66Aa	91.08±13.24Aa	ns	ns
	56	$14.88 \pm 3.20 \text{Ab}$	14.90 ± 2.13 Ab	13.50 ± 2.34 Ab	12.51±1.91Ab	ns	
	84	$13.74 \pm 2.30 \text{Ab}$	13.53 ± 2.13 Ab	$14.82 \pm 1.88 \text{Ab}$	$14.54 \pm 2.68 \text{Ab}$	ns	
	Sig	***	***	***	***		
Ca (mg/g)	28	$0.37 \pm 0.20 \text{Ac}$	0.38 ± 0.15 Ac	0.33 ± 0.12 Ac	$0.29 \pm 0.07 \text{Ac}$	ns	ns
	56	2.02 + 0.63Aa	1.93 ± 0.50 Aa	1.63 ± 0.39 Aa	1.53 ± 0.45 Aa	ns	
	84	0.89 ± 0.27 Ab	0.96 ± 0.26 Ab	1.04 + 0.34Ab	1.02 + 0.39Ab	ns	
	Sig	***	***	***	***		
Minor elements	0						
Mn (mg/g)	28	0.01 ± 0.00 Ab	0.01 ± 0.01 Ab	0.01 ± 0.00 Ac	0.02 ± 0.01 Aa	ns	ns
(8,8)	56	0.07 ± 0.04 Aa	0.05 ± 0.02 Aa	0.05 ± 0.02 Aa	0.04 ± 0.02 Aa	ns	
	84	0.04 ± 0.02 Aa	0.05 ± 0.01 Aa	0.03 ± 0.01 Ab	0.05 ± 0.02 Aa	ns	
	Sig	***	***	***	ns	110	
Fe $(m\sigma/\sigma)$	28	0.07 ± 0.03 Ac	0.04 ± 0.02 Bc	0.04 ± 0.01 Bc	0.03 ± 0.02 Bb	**	***
re (mg/g)	20 56	0.07 ± 0.05 Re 0.17 ± 0.06 Bb	$0.04 \pm 0.02 Be$	$0.04 \pm 0.01 Be$ 0.26 ± 0.10 ABa	$0.03 \pm 0.02B0$ $0.23 \pm 0.15ABa$	*	
	84	0.26 ± 0.13 A a	0.19 ± 0.07 Ab	0.20 ± 0.02 Bb	$0.23 \pm 0.15 ABa$	**	
	Sig	***	***	***	***		
Ni (ug/g)	28	33.67 ± 18.75 Ab	34.61 ± 15.55 Ab	$23.23 \pm 8.84 \text{ Ab}$	$23.26 \pm 11.79 \Delta_{2}$	ne	ne
INI (μg/g)	20 56	55.07 ± 18.75 AU 70.05 ± 11.60 BC a	$11673 \pm 3501 \text{ A}_{2}$	$25.25 \pm 0.04A0$	23.20 ± 11.79 Ka 53.10 ± 23.17 Ca	**	115
	84	$60.12 \pm 30.62 \Lambda_{\odot}$	110.75 <u>-</u> 55.91Ma	24.33 ± 6.04 Cb	33.17 ± 23.17 Ca	**	
	04 Sig	09.12 <u>+</u> 39.02Aa **	++.00 <u>+</u> 20.30AB0 ***	24.33 ± 0.94C0 ***	34.32 ± 17.47 Ca		
$C_{\rm H}$ (mg/g)	31g 28	1654 + 410Ab	14 15 + 2 81 4 5	14.47 ± 3.20 Åg	$1306 \pm 2.21 \text{ Ab}$	ne	ne
Cu (llig/g)	20 56	$10.54 \pm 4.19 \text{AU}$	$14.13 \pm 2.01 \text{AU}$	14.47 ± 3.20 AC	$15.90 \pm 5.51 \text{A0}$	ns	115
	90 84	30.30 ± 19.49 Aa	00.07 ± 21.21 Aa	37.03 ± 23.73 Aa	40.71 ± 13.52 Ad	ns	
	04 Sia	34.33 ± 12.33A0	51.94 ± 10.51Aau	52.54±0.46A0	52./1±15.50Aa	118	
7 n (mala)	31g	0.07 + 0.05 Å a	0.02 ± 0.01 Po	0.02 ± 0.01Pb	0.02 + 0.01Ph	**	**
Zii (iiig/g)	20	0.07 ± 0.03 Aa	$0.03 \pm 0.01 \text{Ba}$	$0.05 \pm 0.01 B0$	$0.02 \pm 0.01 B0$		
	30 94	0.04 ± 0.01 Aa	0.07 ± 0.03 AB	0.03 ± 0.02 Aa	0.04 ± 0.02 Aa	ns **	
	04 Sia	0.05 ± 0.05 Aa	0.05 ± 0.02 ABa	0.03±0.00B0 *	0.03 ± 0.01ABab		
Ma (wala)	51g	11S	$\frac{115}{1.75} + 0.74 \text{ AD}_{2}$	1 05 + 0 24DCh	1.00 + 0.20Ch	**	*
Mo (µg/g)	28	2.38 ± 0.79 AD	1.75 ± 0.74 ABa	$1.05 \pm 0.34 BCb$	1.00 ± 0.39 CD	**	4
	30	4.12 ± 0.92 Aa	4.06 ± 1.43 ABa	$2.73 \pm 1.19Ba$	$2.12 \pm 0.88Ba$	Ŧ	
	84 0'	1.32±0.59Ab	1.91±0.86Aa	$1.90 \pm 0.91 \text{Aab}$	1.65 ± 0.78 Aa	ns	
T 1 ·	51g	* * *	ns	Ŧ	Tr Tr		
Trace elements	20	0.00 - 0.00 1	0.41 - 0.224	0.00	0.07		-1-
As (µg/g)	28	0.38 ± 0.29 Aa	0.41 ± 0.23 Aa	0.23 ± 0.0 / Ab	0.27 ± 0.07 Aa	ns	*
	56	0.20 ± 0.06 Ba	0.26 ± 0.04 ABa	0.28 ± 0.0 / Ab	0.30 ± 0.08 Aa	**	
	84	0.36 ± 0.10 Aa	0.33 ± 0.07 Aa	0.38 ± 0.06 Aa	0.32 ± 0.12 Aa	ns	
	Sig	ns	ns	***	ns		

Table 3 (continued)

Element Days	Days	ays Treatment					T*D
		Control (n=40)	Se (n=40)	Heavy metals (n=40)	Se + HM (n=40)		
Co (µg/g)	28	$0.07 \pm 0.02 \text{Ac}$	0.06±0.01Ab	$0.06 \pm 0.02 \text{Ab}$	0.11±0.06Ab	ns	*
	56	10.88±4.61Aa	$18.68 \pm 10.42 \mathrm{Aa}$	13.71±7.46Aa	6.82 ± 4.88 Aa	ns	
	84	$6.40 \pm 2.87 \text{Ab}$	$3.49 \pm 1.60 \text{ABb}$	2.11 ± 0.85 Bb	$3.00 \pm 1.63 \text{ABb}$	*	
	Sig	***	***	***	**		
$V (\mu g/g)$	28	-	-	-	-	-	**
	56	1.67±1.12Aa	1.22 ± 0.55 Aa	1.31±0.49Aa	$1.01 \pm 0.50 \text{Ab}$	ns	
	84	1.18±0.56ABa	1.91 ± 1.00Aa	0.85 ± 0.32 Bb	1.89±0.93Aa	**	
	Sig	ns	ns	*	*		

The values are expressed as means \pm standard deviation. The mean values in the same row (different treatments on the same day) with different capital letters indicate significant differences. The mean values in the same column (the same treatment on different days) with different lower-case letters indicate significant differences. -: not detected; ns: not significant, P > 0.05; Sig.: significance; T^{*}D: the significance of the interaction between the treatment and days; *P < 0.05; **P < 0.01; ***P < 0.001

ion levels, disrupting the mineral elements (Bargellini et al. 2008). Although various in vivo and in vitro studies reported that the absorption, distribution, and retention of Zn, Fe, and Cu were affected by either Se deficiency or supplementation, the results are controversial and require confirmation (Klotz et al., 2003). Li et al. (2018) reported that treating the ovaries of laying hens with Cd reduced the Ca, Ti, Cu, and Zn levels, while significantly decreasing



Fig. 2 The correlation between the elemental content in the yolk and albumen. From pink to red (the values range from 0 to 1) indicate a gradual increase in the positive correlation. From pink to blue (the values range from 0 to -1) indicate a gradual increase in the negative correlation

Table 4 The analysis of the fatty acid composition in the	Fatty acids	Control $(n=40)$	Se $(n = 40)$	Heavy metals $(n=40)$	Se + HM (n = 40)
egg yolk	C4:0	0.10 ± 0.01 d	$0.11 \pm 0.01c$	$0.14 \pm 0.01 \text{b}$	$0.15 \pm 0.01a$
	C14:0	0.30 ± 0.02	0.30 ± 0.02	0.31 ± 0.02	0.32 ± 0.03
	C14:1	0.03 ± 0.01	0.04 ± 0.01	0.04 ± 0.01	0.04 ± 0.01
	C15:0	0.06 ± 0.00	0.06 ± 0.01	0.06 ± 0.00	0.06 ± 0.00
	C16:0	23.59 ± 0.71	23.83 ± 0.48	23.33 ± 0.60	23.45 ± 0.61
	C16:1	1.61 ± 0.17	1.76 ± 0.22	1.68 ± 0.18	1.77 ± 0.23
	C17:0	0.24 ± 0.02	0.23 ± 0.03	0.23 ± 0.01	0.23 ± 0.02
	C17:1	0.12 ± 0.01	0.12 ± 0.01	0.12 ± 0.00	0.12 ± 0.01
	C18:0	11.00 ± 0.43	10.47 ± 0.63	10.54 ± 0.52	10.38 ± 0.54
	C18:1n9t	0.15 ± 0.01	0.15 ± 0.01	0.15 ± 0.01	0.15 ± 0.01
	C18:1n9c	38.04 ± 0.85	37.59 ± 1.04	36.98 ± 0.55	37.48 ± 0.88
	C18:2n6t	0.027 ± 0.003 b	0.029 ± 0.004 ab	$0.028 \pm 0.003 b$	$0.033 \pm 0.006a$
	C18:2n6c	17.47 ± 1.09	17.65 ± 1.37	18.06 ± 0.53	17.78 ± 1.09
	C18:3n6	0.15 ± 0.01	0.14 ± 0.02	0.14 ± 0.01	0.14 ± 0.01
	C18:3n3	0.47 ± 0.04 b	$0.54 \pm 0.05a$	$0.53 \pm 0.03a$	$0.56 \pm 0.06a$
	C20:0	0.04 ± 0.00	0.04 ± 0.00	0.04 ± 0.01	0.04 ± 0.00
	C20:1	0.21 ± 0.02	0.20 ± 0.02	0.20 ± 0.01	0.20 ± 0.01
	C20:2	0.22 ± 0.01	0.22 ± 0.02	0.22 ± 0.01	0.22 ± 0.02
	C21:0	0.04 ± 0.01	0.04 ± 0.01	0.04 ± 0.01	0.04 ± 0.00
	C20:4n6	2.34 ± 0.09	2.28 ± 0.11	2.33 ± 0.11	2.34 ± 0.13
	C20:3n6	0.25 ± 0.02	0.23 ± 0.03	0.26 ± 0.02	0.25 ± 0.02
	C20:5n3	0.04 ± 0.00	0.04 ± 0.00	0.04 ± 0.00	0.04 ± 0.00
	C24:0	1.03 ± 0.06	1.04 ± 0.05	1.01 ± 0.03	1.06 ± 0.09
	C22:6n3	1.16 ± 0.25	1.28 ± 0.14	1.38 ± 0.24	1.20 ± 0.11
	C24:1	$1.29 \pm 0.52b$	$1.59 \pm 0.18b$	$2.15 \pm 0.29a$	$1.95 \pm 0.21a$
	SFA	36.42 ± 0.74	36.13 ± 0.63	35.71 ± 0.61	35.74 ± 0.66
	MUFA	41.46 ± 0.95	41.45 ± 1.07	41.31 ± 0.48	41.71 ± 1.02
	PUFA	21.91 ± 1.10	22.20 ± 1.46	22.76 ± 0.65	22.33 ± 1.09
	ω-3 PUFA	$1.67 \pm 0.26b$	1.87 ± 0.16 ab	$1.95 \pm 0.23a$	1.79 ± 0.09 ab
	ω-6 PUFA	20.24 ± 1.10	20.33 ± 1.43	20.81 ± 0.59	20.54 ± 1.11
	ω-6/ω-3	12.46 ± 2.37	10.95 ± 1.11	10.80 ± 1.31	11.50 ± 0.96

Values with different letters indicate significant differences (P < 0.05). Values are expressed as means ± standard deviation. SFA, saturated fatty acids; MUFA, monounsaturated fatty acids; PUFA, polyunsaturated fatty acids

the Ba content and substantially increasing the Fe, Mo, and Cd levels.

Pearson's correlation analysis was performed to investigate the correlation between the mineral elements. These correlations are presented in Fig. 2. Se was positively correlated with Na, Mg, Ca, Fe, Cu, Cd, and Pb (P < 0.05 for the Mg in the yolk, as well as the Na and Cd in the albumen, and P < 0.01 for the other elements). In the yolk, Se appeared to be correlated positively with Zn, As, and Hg (P < 0.01) and negatively with Mn (P < 0.01). In the albumen, Se was positively correlated with Cr, Mn (P < 0.01), and Mo (P < 0.05), and negatively with K. Pappas et al. (2011) investigated the interaction between Se and the various trace elements involved in the antioxidant system of chicken eggs. Cr was positively associated with the Pb in the albumen and negatively with Hg in the yolk (P < 0.01). Furthermore, Cd was correlated positively with Fe, Zn, and Pb in both the yolk and albumen. A positive correlation was evident between Hg and Cd or Cr in the egg yolk, while Cr was positively associated with Pb in the albumen. The Se displayed a high positive correlation with Cd and Pb in the yolk and albumen in the present study. Previous research involving commercially Se-enriched eggs also revealed that Se was positively correlated with the Pb in the eggs (Zhao et al. 2021). Overall, the complex interactions indicated that both synergistic and antagonistic effects persisted among these elements. The possible protective role of Se and the toxic effect of heavy metals may contribute to changes in the trace elements.

The effect of Cd, Pb, Hg, Cr, and Se co-treatment on fatty acids

The fatty acid profiles in the egg volks of all the experimental groups are presented in Table 4. Nine saturated fatty acids (SFAs) were found (C4:0, C14:0, C15:0, C16:0, C17:0, C18:0, C20:0, C21:0, and C24:0) in the analyzed eggs. Except for butyric acid (C4:0), no significant differences were evident between the other SFAs of the four treatment groups. Moreover, Se addition significantly increased the butyric acid (C4:0) content compared to the group without Se supplementation (Se group vs. the control group, and Se+HM group vs. the heavy metals group). These outcomes were consistent with data reported by Gangdadoo et al. (2018), who indicated that providing broilers with a 0.9-mg/kg Se nanoparticle diet increased the short-chain fatty acids, especially butyric acid. Of the monounsaturated fatty acids (MUFA), oleic acid (C18:1n9c) dominated in all the batches, ranging from 36.98 to 38.04%, with no significant differences between the four treatment groups. The nervonic acid (C24:1) level increased significantly after heavy metal supplementation. Furthermore, Se supplementation increased the linolelaidic acid (C18:2n6t) content, although the impact was lower. The α -linolenic acid (C18:3n3) level was considerably higher in the Se group than in the control group. However, this relationship was not observed in the Se+HM and heavy metal groups, consequently exhibiting a higher ω -3 polyunsaturated fatty acid (PUFA) level than in the control group eggs. These findings corresponded with data obtained by Zudunczyk et al. (2013), who noticed an increase in the C18:3n3 and ω-3 PUFA content of eggs after dietary Se supplementation. This may be connected to the interaction between the PUFAs and Se, possibly via GSH-Px action (Fasiangova & Borilova 2017).

Conclusion

This study investigates the effect of Se and heavy metal (Cr, Cd, Pb, and Hg) interaction on the quality, fatty acid levels, and content of 13 types of ions in eggs. Se supplementation increases the yolk percentage of the eggs, as well as the Se content, which mainly accumulates in the yolk. Moreover, Se reduces the Cr content in the yolk at 28 days and the Cd and Hg levels in the albumen at 84 days. The complex interactions between the elements are analyzed, and both positive and negative correlations between these elements are presented. Se displays a significant positive correlation with Cd and Pb in the yolk and albumen, while heavy metals minimally affect the fatty acids in the yolk.

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experimental processing and data interpreting and wrote the paper; Shiping Bai and Hong Zhu performed part of the experiments, and all authors commented on previous versions of the manuscript. All authors read and approved the final manuscript.

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Data availability The datasets used or analyzed during the current study are available from the corresponding authors on reasonable request.

Declarations

Ethics approval and consent to participate Not applicable

Consent for publication Not applicable.

Conflict of interest The authors declare no competing interests.

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