RESEARCH ARTICLE

Application of a combined response surface methodology (RSM)‑artifcial neural network (ANN) for multiple target optimization and prediction in a magnetic coagulation process for secondary efuent from municipal wastewater treatment plants

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Abstract

In this study, an enhanced coagulation-focculant process incorporating magnetic powder was used to further treat the secondary effluent of domestic wastewater from a municipal wastewater treatment plant. The purpose of this work was to improve the discharged water quality to the surface water class IV standard of China. A novel approach using a combination of the response surface methodology and an artifcial neural network (RSM-ANN) was used to optimize and predict the total phosphorus (TP) pollutant removal and turbidity. This work was frst evaluated by RSM using the concentrations of coagulant, magnetic powder, and focculant as the controllable operating variables to determine the optimal TP removal and turbidity. Next, an ANN model with a back-propagation algorithm was constructed from the RSM data along with the non-controllable variables, raw TP concentration, and raw water turbidity. Under the optimized experimental conditions (28.42 mg/L coagulant, 623 mg/L magnetic powder, and 0.18 mg/L focculant), the TP and turbidity removal reached 88.79 \pm 5.45% and 63.48 \pm 9.60%, respectively, compared with 83.28% and 59.80%, predicted by the single RSM model, and 87.71 \pm 5.74% and 64.62 \pm 10.75%, predicted by the RSM-ANN model. The treated water were 0.17 \pm 6.69% mg/L of TP and 2.46 \pm 5.09% NTU of turbidity, respectively, which completely met the surface water class IV standard (TP<0.3 mg/L; turbidity<3 NTU). Therefore, this work demonstrated that the discharged water quality was completely improved using the magnetic coagulation process. In addition, the combined RSM-ANN approach could have potential application in municipal wastewater treatment plants.

Keywords Wastewater treatment plant · Magnetic coagulation-flocculation · Phosphorus removal · Secondary effluent · Domestic wastewater · Turbidity removal · Box-Behnken Design · Back-propagation algorithm

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Introduction

Water pollution has become an urgent environmental issue facing our world. The levels of contaminants, especially total phosphorus (TP), have been increasing at a very high rate, and this has led to a signifcant impact on ecological systems as well as on the economy and human health. Approximately 15% of the global population is concerned with the issue of water safety (Katz and Dosoretz, [2008](#page-11-0)). In China, a large amount of industrial and domestic wastewater has been discharged into the environment as a result of decades of rapid economic development and urbanization, and domestic sewage water constitutes approximately 70% of the total wastewater produced by the country. Therefore, it has become an urgent matter to implement intensive wastewater treatment programs to efficiently remove TP from wastewater.

In the past, physical or chemical methods such as adsorption, precipitation, fltration, and oxidation have been popular for treating wastewater (Ho and Babel, [2020\)](#page-11-1). However, there has recently been renewed interest in staged coagulation-focculation-based methods that are cheaper and require less time to operate (Mortadi et al. [2020](#page-12-0)). Coagulation plays an important role in reducing the levels of particulates, synthetic organic carbon, precursors of disinfection by-products (DBPs), and some inorganic and metal ions as well as microorganisms in wastewater to improve its quality (Wang et al. [2021\)](#page-12-1). During conventional coagulation treatment, alum coagulant, which undergoes a hydrolysis reaction to form Al^{3+} species, is the commonly used inorganic coagulant (Santos et al., [2016\)](#page-12-2). It has been shown that aluminum salt-based coagulants are much more efective than iron salt-based coagulants for removing turbidity from wastewater (Kumari and Gupta, [2020](#page-11-2); Rizzo et al. [2008\)](#page-12-3). Aluminum ions tend to attract particles in the wastewater to form larger focs that can then be easily precipitated and removed from the water (Beltran et al. [2009\)](#page-11-3).

In recent years, the use of the polymerized forms of metal coagulants (e.g., poly-aluminum chloride (PAC)) has increased because of their excellent focculating property (Zhou et al. [2021;](#page-12-4) Bogunovi et al., [2021](#page-11-4)). Compared with traditional $\text{Al}_2(\text{SO}_4)_3$, PAC has both strong charge neutralization and bridging abilities for easily attracting particulate matter or colloidal matter to form large flocs (Rizzo et al. [2019](#page-12-5); Usefi and Asadi-Ghalhari, [2019](#page-12-6)). However, organic polymer-based flocculants, such as polyacrylamide (PAM), are widely used in the coagulation process by researchers. The advantages of their excellent flocculation effect and low dosage make them more efective in many cases for wastewater treatment plants (Li et al. [2018](#page-12-7)). PAC and PAM have been widely used in the coagulation–focculation process in WWTPs for decades (Liu et al. [2013a](#page-12-8), [b;](#page-12-9) Wang et al. [2011](#page-12-10); Xu et al. [2020\)](#page-12-11). However, with China's gradually increasing wastewater treatment plant discharge water quality standards, how to further improve the pollutant removal efficiency of the conventional coagulation process has become a hot research topic (Lv et al. [2019](#page-12-12)).

An important factor that can greatly enhance the efectiveness of the coagulation-focculation-based method is to speed up the precipitation process of the foc and its separation from the water, and this might be achieved by adding magnetic powder (Liu et al. [2013a,](#page-12-8) [b](#page-12-9); Kumari et al. Kumari and Gupta, [2020\)](#page-11-2). The presence of magnetic powder could enhance the efectiveness of the separation provided by the coagulation-focculation treatment system. Recently, it has been reported that $Fe₃O₄$ as magnetic seeding has been used to remove micro-pollutants from wastewater with the advantages of accelerated foc sedimentation, less sludge production, and low cost (Huangfu et al. [2017\)](#page-11-5). Furthermore, the magnetic powder may be separated from the waste sludge through a magnetic separator that can be set up on a magnetic coagulation device and then reused in the system (Lv et al. 2020 ; Yao et al. 2014). However, the effectiveness of this method largely depends on the precise amount of magnetic powder added and the exact coagulant used. Many factors can influence the efficiency of the magnetic coagulation-focculation process, such as the amount of coagulant, focculant, magnetic powder, pH, mixing, and hydraulic retention time (Lohwacharin et al. [2014;](#page-12-15) Qasim et al. [2019](#page-12-16)). Thus, the proper optimization of these factors can signifcantly increase the treatment efficiency. It has been reported that the water quality of secondary effluent from municipal wastewater treatment plants is stable (Li et al. [2009](#page-11-6)). Furthermore, the pH values of the water source in this work were calculated to be approximately $7.2 \pm 8.8\%$. Hence, pH was not evaluated as a primary variable in this study.

In traditional methods of optimization, the efect of only one variable is studied at a time. This approach not only wastes precious experimental time, but it is also not possible to observe the overall efect of the variables on responses. RSM is widely used to design experiments and optimize experimental data. It can reduce the required number of experiments and determine the experimental conditions for obtaining the best response (Mohammad et al. [2019\)](#page-12-17). RSM has become a common mathematical model for water treatment because it offers several advantages, such as requiring fewer tests, lower costs, time-saving, as well as having the power to evaluate the interactions among various factors.

However, RSM is difficult to practically apply in wastewater treatment plants (WWTP) due to its inability to incorporate non-controllable variables. Because wastewater treatment plants are diferent from laboratory experiments, the daily water quality fuctuates, and once the infuent water quality has changed signifcantly beyond the original design range of the RSM, it is difficult for the RSM to obtain the optimal process solution and accurately predict the efuent quality. However, an artifcial neural network (ANN) is a computer simulation of the biological neuron transmission process in the biological brain. It can mimic the great capability of learning and evaluating errors (Igwegbe et al. [2019](#page-11-7)). The greatest advantage of an ANN model is that it can handle a large amount of nonlinear and complex data with the addition of non-controllable variables, and it can be constantly updated using the input of new data obtained during the operational process of a WWTP. Thus, the combination of an ANN with the RSM will help to overcome the limitation of the RSM and predict the optimum conditions and accurate removals for any infow water quality.

Previous studies have focused on a comparison between the RSM and ANN models for optimizing and modeling predictions, but these models have rarely been used as a combined approach (Hafeez et al. [2020\)](#page-11-8). In this study, a novel approach of a combined RSM-ANN model was used for multiple target optimization by the magnetic coagulation process. This work aimed to improve the discharge water quality and to evaluate this novel optimization/prediction strategy for WWTP applications in the future.

Materials and methods

Materials

Characteristics of the secondary effluent

Wastewater samples were collected from the secondary effluent of domestic wastewater from a municipal wastewater treatment plant (WWTP) in Wenzhou, China. The initial total phosphorus (TP), pH, total nitrogen, chemical oxygen demand, and turbidity were 0.52 mg/L, 7.1, 14.32 mg/L, 18 mg/L, and 33.40 NTU, respectively. The poly aluminum chloride (PAC), polyacrylamide (PAM), and magnetic powders (Fe₃O₄) were of analytical grade. The PAC and PAM were purchased from the Xinyuan Environmental Protection Company (Henan), and the magnetic powder was acquired from the Kaili Metallurgical Research Institute (Tianjin).

Preparation of the coagulant and focculant

The coagulant, PAC, was prepared as follows. A total of 5 g of PAC was weighed and dissolved in a 200 ml beaker and stirred until the powder was completely dissolved. The volume was placed in a 500 ml volumetric fask, shaken well, and a concentration of a 10 mg/ml PAC solution was prepared on the day of the test. Thereafter, the solution of the focculant and the PAM were prepared in a 500 ml beaker by weighing 0.3 mg of the PAM, and the beaker was placed on a magnetic stirrer and stirred at 1000 rpm/min for 4 h. This was done because PAM dissolves in water as a viscous, colorless solution and is extremely difficult to hydrolyze completely. Finally, the water samples were subjected to both conventional coagulation and magnetic coagulation assays, and the results from the two methods were compared. The chosen method was then optimized.

Design method of the coagulation jar tests

Initial magnetic coagulation test

The experimental method of coagulation was conducted by Jiang et al. ([2014\)](#page-11-9). Our experimental conditions were optimized in terms of the PAC addition order, the stirring rate, and the stirring time. The conditions were obtained by the order of the PAC, the magnetic powder, and the PAM, with stirring at 500 rpm for 45 s, 500 rpm for 45 s, and 80 rpm for 10 min, respectively.

The experimental procedure of this work was conducted as follows. The wastewater sample (1 L) was dispensed into a 2-L beaker, and a total of six such samples were prepared for each test. To each beaker containing the wastewater, a diferent amount of coagulant (PAC) was added, yielding diferent fnal concentrations (10, 20, 30, 40, and 50 mg/L), and the samples were then stirred for 45 s at 500 rpm. Next, the magnetic powder was added to each mixture to a fnal concentration of 500 mg/L, and the mixture was stirred for another 45 s. After that, the focculant (PAM) was added to the mixtures to 0.3 mg/L followed by stirring for 10 min at 80 rpm. The mixtures were allowed to stand for 15 min and 200 ml of the water was withdrawn within 2 cm from the surface. The TP concentrations and turbidity in the samples were then measured to determine the optimum concentration of PAC that resulted in the maximal removal of TP and turbidity. The experiment was repeated using 0.3 mg/L of PAM and the PAC concentration that yielded the maximum TP and turbidity removal rates while varying the concentration of the magnetic powder from 300 to 700 mg/L. Finally, the concentrations of the PAC and magnetic powder that gave the maximal TP and turbidity removal rates were used to determine the optimum concentration of PAM by conducting the test under diferent concentrations of PAM (0 to 0.6 mg/L).

Response surface methodology design

The statistical design of the experimental analysis was performed using the Design Expert software (version 10.0). The Box-Behnken Design (BBD) and RSM were applied to optimize three variables: the coagulant (PAC), the magnetic powder, and the focculant (PAM) Barekati-Goudarzi et al.

[\(2016\)](#page-11-10). Data from the initial test were used as center points to select the ranges of the PAC, the magnetic powder, and the PAM concentrations, which were 10 to 40 mg/L, 300 to 900 mg/L, and 0 to 0.3 mg/L, respectively. The levels and the three variables are shown in Table [1](#page-3-0). The optimized result generated by the RSM was then used to perform an additional experiment as further confrmation. In all the cases, the TP and turbidity removals were analyzed as the response.

Data analysis

The designed RSM model included 15 runs. The modeling and analysis of the experimental data were performed using Design Expert 10.0. To optimize the three variables (PAC, magnetic powder, and PAM concentrations) in the process of magnetic coagulation-focculation, the data were ftted to a second-order polynomial model (Khalid et al. [2019](#page-11-11)). The quadratic equation model for the predicted optimal conditions is expressed by the following equation (Zhang et al. [2018](#page-12-18)).

$$
Y = \beta_0 + \sum_{i=1}^n \beta_i x_i + \sum_{i=1}^n \beta_{ii} x_i^2 + \sum_{i=1}^{n-1} \sum_{j=i+1}^n \beta_{ij} x_i x_j
$$
\n(1)

where *Y*, β_0 , β_i , β_{ii} , and β_{ij} are the predicted response, regression constant coefficient, linear coefficient, quadratic coefficient, and interaction coefficient, respectively; and x_i and x_j are the coded values of the variables.

The graphical data were analyzed using an analysis of variance (ANOVA) to determine the interactions of the different variables and responses. The value of the coefficient of determination (R^2) was used to express the quality of the ft produced by the RSM model (Momeni et al. [2018\)](#page-12-19). The statistical signifcance of the model was assessed using an *F*-test. *P*-values at the level of 95% confdence were used to determine the signifcance of the variables Pambi and Musonge ([2016\)](#page-12-20). A statistical signifcance inconsistency between the predicted and experimental results was considered at a *P*<0.05 level. The smaller the *P*-value, the better the model could ft the data Tetteh and Rathilal ([2018\)](#page-12-21). For each of the tested variables (coagulant, magnetic powder,

Table 1 Experimental range, levels, and coded names of the three variables

Variable	Coded	Level		
		-1	0	$+1$
PAC (mg/L)	Α	10	25	40
Magnetic powder (mg/L)	в	300	600	900
PAM (mg/L)	C	θ	0.15	0.3

and focculant), a 3D surface plot and its respective contour plot were also obtained at the three levels.

Artifcial neural network

A feed forward network, the ANN model, was developed with a back-propagation (BP) algorithm using MATLAB 2014a software. The concentration of PAC, magnetic powder, and PAM were selected as the three process variables for the input layer, whereas the removals of TP and turbidity were chosen as the output layer neurons to build a threelayer BP-ANN model. The experimental data were normalized between 0 and 1. The tansig function and the purelin function were selected as the transfer functions of the hidden layer and output layer, respectively. To determine the optimal network topology, the number of neurons was trained and selected as optimal, after which the neural network with the optimal efect was also selected by varying the number of training sessions. The optimal network model was selected by comparing metrics such as R^2 , the root mean square error (RMSE), and the average absolute deviation (AAD).

Similar to the RSM model, the experimental data were also generated from the BBD design and then used to determine the optimum architecture with network parameters. Then the trainlm function for training the BP-ANN model was then selected. The total data from the experiments were divided into three portions: the training set, the validation set, and the testing set representing 60%, 20%, and 20% of the experimental data, respectively. The aim of the threepart data was to evaluate the performance of the model with respect to the prediction for unseen experimental data that were not trained and to reduce the generalization capability of the model.

Results and discussion

Conventional coagulation/magnetic coagulation process evaluation

The results obtained using the conventional coagulation and magnetic coagulation methods under their best condition were studied to evaluate which process presented better parameter removals under the same settling times. Figure [1](#page-4-0) shows the TP removal and turbidity using the two methods under settling times of 1, 2, 5, 10, 15, and 30 min.

It can be observed, in a general way, that magnetic coagulation presents better parameter removals compared to conventional coagulation. In addition, in assays conducted under diferent settling times, the settling time for magnetic coagulation was much shorter than that of the conventional coagulation process. In terms of conventional coagulation, both the TP and turbidity removals kept increasing with

Fig. 1 Effect of settling time on TP and turbidity removal by conventional and magnetic coagulation

increases in the settling time, and the highest removals were obtained and remained stable at 30 min of settling. Under the same conditions, the magnetic coagulation sedimentation processes were nearly completed at approximately 5 min and remained stable after 15 min. The same behavior applied to magnetic coagulation was reported by Dai ([2017\)](#page-11-12) and Ding et al. ([2021](#page-11-13)). This might be explained by the possibility that the magnetic seed addition formed a magnetic foc with magnetic seeds as the core. The coagulant would hydrolyze and wrap around the magnetic powder particles, and

Table 2 Box-Behnken design with the observed data and predicted responses

this would greatly increase the chance of collision between pollutants and focs, thus increasing the pollutant removal efect while also accelerating the foc settling process. This would provide a huge advantage for engineering applications. Therefore, it can be concluded that the great efect of the magnetic powder addition to enhance coagulation can greatly improve pollutant removal. Furthermore, the use of magnetic powder could also signifcantly reduce the settling time, which plays an important role in the advanced treatment. Thus, the magnetic coagulation process was chosen and further optimized by the RSM.

RSM modeling and optimization

The results obtained from the initial magnetic coagulation test revealed that the PAC, magnetic powder, and PAM concentrations were the primary factors that afected the TP and turbidity removals from the wastewater samples. The TP and turbidity removals were further investigated by performing 15 experimental runs according to the BBD design (Table [2](#page-4-1)), and the optimum conditions were obtained by the RSM (Table S1). The experimental data acquired from the BBD were subsequently used to obtain the predictive values via the RSM and the BP-ANN models.

The values predicted by the RSM displayed good agreement with the actual data, with a coefficient of determination close to one $(R^2 = 0.9922, 0.9920;$ Table [3](#page-5-0)). The results were analyzed using the software Design Expert 10.0, and the regression equations of the second-order model for the TP and turbidity removal rates were obtained. The regression

A: PAC, B: magnetic powder, C: PAM. # Unit of A, B, and C: mg/L

Source	TP removal $(\%)$			Turbidity removal $(\%)$	
	F value		P -value	F value	P-value
Model	70.69	$< 0.0001*$		69.17	$0.0001*$
A-PAC	269.75	$< 0.0001*$		27.62	$0.0033*$
B-Magnetic powder	8.26	0.0348		6.62	0.0499
C-PAM	23.49	0.0047		7.09	0.0448
AB	0.51	0.5077		108.49	$0.0001*$
AC	2.24	0.1949		31.82	0.0024
BC	1.91	0.2250		84.37	$0.0003*$
A^2	139.67	$< 0.0001*$		81.99	$0.0003*$
B ²	125.43	$< 0.0001*$		275.18	$< 0.0001*$
C^2	115.54	0.0001		40.12	$0.0014*$
Lack of Fit	1.02	0.5289#		1.32	0.4576#
Model statistics					
AP	22.678			23.885	
R^2	0.9922			0.9920	
Adjusted R^2	0.9782			0.9777	
Predicted R^2	0.9175			0.9092	

Table 3 ANOVA for the response quadratic model of TP removal and turbidity removal

*Signifcant; #insignifcant

equations are shown in Eq. (2) (2) (2) and Eq. (3) (3) . In addition, the ftting quality of the RSM model was also evaluated by an application of an analysis of variance (ANOVA). The coefficients of the quadratic equations (Eq. (2) (2) and Eq. (3) (3)) and the *p*-values for the TP and turbidity removals were acquired from subsequent experimental data. The responses of the TP and turbidity removals in terms of the three variables are shown in Eq. (2) (2) and Eq. (3) (3) , respectively. Both equations

are second-order polynomial equations, and A, B, and C represent the concentration of the coagulant, the magnetic powder, and the focculant, respectively. The positive number in front of each term represents a positive effect, while the negative number indicates a negative efect. The PAC, magnetic powder, and PAM all exerted a positive efect on the efficiency of TP removal, but only the PAM exerted a positive effect on turbidity removal. Equations (2) and (3) (3) are as follows:

$$
Y_{TP} = 83.04 + 2.41A + 0.42B + 0.71C + 0.15AB
$$

+ 0.31AC + 0.29BC - 2.55A² - 2.42B² - 2.32C² (2)

$$
Y_{turbidity} = 60.38 - 3.28A - 1.61B + 1.66C + 9.21AB + 4.99AC
$$

+ 8.12BC - 8.33A² - 15.26B² - 5.83C² (3)

Optimization of TP removal

Figure [2](#page-5-3) represents the efect of the three factors (A, B, and C), and the interaction between any two factors on the TP removal effect. It can be seen from Fig. [2a](#page-5-3) that when the fxed magnetic powder dosing amount was certain, the removal efficiency of TP gradually increased as the PAC concentration increased from 10 mg/L. When the PAC concentration was raised to approximately 30 mg/L, the removal efficiency of TP reached the peak of the rising trend. In addition, with further coagulant PAC injection, the TP removal efficiency showed a slow decreasing trend, and the removal efect became slightly worse. However, it can be seen that the TP removal at the response value remained above an 86% removal rate, probably because the concentration of PAC was the primary factor in the removal of TP from water, but

Fig. 2 a Response plot showing the efects of PAC and magnetic powder on TP removal and **b** response plots showing the efects of PAM and PAC on the TP removal

when the coagulant was added to reach a certain concentration, the TP removal efect in water no longer improved. The addition of excess PAC would lead to the formation of too many products of PAC hydrolysis, which in turn, can interfere with the adsorption and precipitation of TP, resulting in a lower removal rate Krupińska ([2020](#page-11-14)).

While the PAC dosing concentration was fxed, with an increase in the magnetic powder dosing from 300 to 900 mg/L, the TP removal efficiency reached the highest in the middle of the 600–700 mg/L of magnetic powder dosing, and then it gradually decreased. This was because the magnetic powder itself does not have the hydrolytic properties of a coagulant, and the purpose of adding a magnetic species to water is to easily form a magnetic foc with magnetic powder as the core that can better adsorb suspended matter and make it settle. Therefore, the TP removal rate was improved by appropriately increasing the amount of the magnetic powder injection. In addition, it can be seen from Fig. [2](#page-5-3) that when the concentration of the magnetic powder was 300 mg/L, the PAC concentration of 40 mg/L could only reach a removal rate of approximately 79%. However, with a gradual increase in the concentration of magnetic powder, the corresponding PAC concentration was gradually reduced, and the maximum TP removal increased. This indicated that the magnetic powder could reduce the use of PAC to a great extent, reduce the cost, and make a signifcant improvement in pollutant removal. However, the magnetic coagulation process is performed for good magnetic composite foc to achieve the best TP removal. Thus, the amount of magnetic powder and PAC concentration has an optimal matching ratio. If either the amount of magnetic powder or that of the PAC is too much, this will decrease the pollutant removal efect.

Figure [2b](#page-5-3) shows the interaction between PAC and PAM. It can be seen from the graph that the TP removal changed rapidly followed by the PAC concentration, and then it reached the peak when the PAC was 30 mg/L. In contrast, a change in the PAM concentration had less of an efect on the TP removal rate, with the highest point near 0.15 mg/L. It could be concluded that PAC was the most important infuencing factor on TP removal by magnetic coagulation. This was because PAC hydrolysis produced positively charged Al^{3+} that could effectively adsorb and subsequently complex phosphorus for a better sedimentation process.

Optimization of turbidity removal

Figure [3](#page-6-0) shows the interaction efect of ABC on turbidity removal. Similarly, it shows a circle between two factors in the plane diagram, and this indicates that the interaction effects were significant in terms of turbidity (Fig. $3a$). The turbidity removal rate increased slowly with an increase in the PAC dosage and then decreased rapidly when the concentration of the PAC dosage was certain. In addition, the efect of the magnetic powder on turbidity also displayed the same effect. This is because when the magnetic powder dosage is too large and shows a saturation trend, there will be residual magnetic powder in the water that will be difficult to adsorb with the pollutants and it will drop, thus suspending in the water to interfere with the turbidity removal effect. This also demonstrates the reason for the use of turbidity as an indicator of the magnetic coagulation process.

As shown in Fig. [3b,](#page-6-0) at PAM concentrations less than 0.12 mg/L , it was difficult to obtain satisfactory turbidity removal rate results regardless of the variation in the magnetic powder. This indicated that the removal mechanism of the magnetic mixing coagulation was via adsorption on the adsorption sites of the hydrolysis products of the coagulants and focculants to efectively adsorb pollutants. When the concentration of the polymer focculant was too small, it was difficult to achieve an effective removal effect even if

Fig. 3 a Response plot showing the effects of the PAC and magnetic powder on turbidity removal and b response plot showing the effects of the PAM and magnetic powder on turbidity removal

the amount of magnetic powder was increased. The above results demonstrate that the turbidity value of the effluent can directly refect whether the amount of magnetic powder is too much. More precisely, the amount of magnetic powder is an important factor that affects the turbidity of the effluent.

Optimization of the process parameters

The optimum conditions for both TP and turbidity removal were discovered by using the three variables, namely, concentrations of the PAC, magnetic powder, and PAM. To obtain the optimum conditions for multiple responses, the desired conditions for each response were selected from the options provided (Gadekar and Ahammed [2019\)](#page-11-15). In this study, the desired goal for turbidity removal was set as "maximize," while TP removal was set as "in range." Finally, the optimum conditions were obtained at a PAC of 28.42 mg/L, magnetic powder of 623 mg/L, and PAM of 0.18 mg/L. The predicted TP and turbidity removals were 83.28% and 59.80%, respectively.

ANOVA

The data generated by ANOVA in the RSM model for TP and turbidity removals are summarized in Table [3.](#page-5-0) The model was statistically acceptable as indicated by the high *F* value (70.69, 69.17) and low *p*-value (<0.0001, 0.0001). The lack of ft (LOF) value indicated a smaller variation in the data around the ftted model. Poor ftting of the model to the data would result in an insignifcant LOF. The large value for the lack of fit (Prob $> F = 0.5289, 0.4576$) showed that the *F*-statistic was insignifcant, and this implied that the correlation between the variables and the process responses was signifcant for this quadratic model.

For both TP and turbidity removal, the model revealed a large and signifcant efect exerted by each of the three variables and their interaction. The model signifcance was computed based on the *F*-value and *p*-value. The value of *p* $(< 0.05$) revealed a significant interaction between any two of the three variables, and the smaller the *p* value, the more signifcant the interaction. The model chosen was largely based on the judgment conceived by the experimental design. Variables A, B, and C had significant effects on both removals, and the three variables had signifcant quadratic effects on A^2 , B^2 , and C^2 , respectively (Table [3\)](#page-5-0).

BP‑ANN modeling

Selection of the processing units and the hidden layer neurons

The generalization ability of the BP-ANN model has a great correlation with the number of hidden layers and the neurons of the hidden layers. The more hidden layers and neurons in the hidden layers, the better the generalization capability and the lower the accuracy of the network. The mean square error (MSE) values of diferent neuron numbers in the hidden layer are shown in Table S2, and the neuron number with the smallest MSE was selected to construct the BP-ANN model. Therefore, $3 \times 7 \times 2$ was taken as the optimal architecture for the model (Fig. S1).

Network training and evaluation

To analyze the ftting and predictive performance of the two models, namely, the RSM and the BP-ANN models, four indicators of each model, the root mean squared errors (RSME), the average absolute deviation (AAD), the mean squared error (MSE), and the correlation coefficient (R^2) , were calculated (Table [4\)](#page-7-0). The MSE values of the TP and turbidity obtained from the RSM and the BP-ANN model were 0.23, 1.23, and 4.45, 11.56, respectively. Moreover, larger values of these indicators (the RMSE, AAD, and MSE) were found to be associated with a greater model prediction error, consistent with a previous study Ghritlahre and Prasad [\(2018\)](#page-11-16). The results in Table [4](#page-7-0) show that the RSM model gave lower values for the AAD, MSE, and RSME compared with the BP-ANN model. Thus, from the perspective of these three indicators, the RSM model could be more reliable, yielding a better ft and prediction than the BP-ANN model under that specifc water quality (Ganapathy et al. 2020). However, the value of R^2 indicates good fitness between the experimental and predicted data given by the model. The results depicted that both the RSM (0.9922 for TP and 0.9920 for turbidity) and the BP-ANN $(R^2=0.9461)$ for TP and 0.9461 for turbidity) models both had great ft-ness, as shown in Table [4](#page-7-0), because the R^2 were relatively close to one. In this case, the signifcance of the BP-ANN was demonstrated by the three R^2 values, namely, the training set, the testing set, and the verifed set. The validation $R²$ of the response given by the BP-ANN was 0.96 (Fig. [4](#page-8-0)), which implies a good, stable prediction ability for the TP and turbidity removal by magnetic coagulation under a given range of concentrations.

Table 4 Comparison of the predictive capabilities of the RSM and BP-ANN models

Parameters	TP		Turbidity		
	RSM	BP-ANN	RSM	BP-ANN	
RMSE	0.48	1.11	2.11	3.4	
MSE	0.23	1.23	4.45	11.56	
AAD	2.36	2.48	9.61	9.24	
R^2	0.9922	0.9461	0.9920	0.9461	

The experimental and predicted data given by the two models in terms of the TP and turbidity removals are graphically shown in Fig. $5a$ and [c.](#page-9-0) It can be seen that the actual results are slightly larger than the prediction results produced by the BP-ANN in terms of the TP removal experimental numbers at 6, 12, and 13. However, in terms of turbidity removal, the ANN errors are larger than the RSM at the numbers 3, 6, 11, and 12. This indicates that the RSM is a superior model compared to the single BP-ANN model. The residual values of the two models for the removal rate are shown in Fig. [5b](#page-9-0) and Fig. [5d,](#page-9-0) where the deviation of the residual values for the RSM clearly displays a regular curve compared with that of the BP-ANN model. The smaller fuctuation in the residuals near zero, the more accurate the model prediction. This indicates that the RSM could optimize the process better and had a greater ftting and prediction performance of the pollutant removal under the specifc water qualities compared to the single BP-ANN. However, the limitation of the RSM is its inability to incorporate noncontrollable variables. Thus, by taking the self-learning and updating advantage of the BP-ANN, it is possible to better predict the removal efects by the magnetic coagulation process based on the RSM model. Therefore, the combination of the two models appeared to have a good ft and provide enhanced fexibility, feasibility, and accuracy in practical applications for wastewater treatment plants.

RSM‑ANN modeling and validation

The RSM-ANN model was constructed using the same experimental data that was used for the BP-ANN model with two additional uncontrollable variables, the raw TP concentration, and the raw water turbidity. In addition, the same back-propagation algorithm was applied for the RSM-ANN modeling. The RSM-ANN was tested by varying the number of hidden layers and neurons in order to select the optimum topology architecture by comparing the MSE and the four R^2 values. In terms of R^2 (all), the number of neurons in the hidden layer showed a gradual increase in R^2 (all) when the number of neurons was valued from seven to 10. When the number of neurons exceeded 10, the value of R^2 also decreased gradually. While the training R^2 of the topologies $5 \times 7 \times 2$ and $5 \times 8 \times 2$ were 0.9998 and 1, respectively, and these were higher than the topology $5 \times 10 \times 2$

- \bullet - Experimental data --- RSM predicted data --- BP-ANN predicted data

Fig. 5 a Comparison of the experimental with the predicted value for TP removal by the RSM and BP-ANN models, **b** distribution of the residual value of TP removal, **c** comparison of the experimental with

(0.9997). Validation is the most important indicator in the analysis of network performance. According to the MSE values, the MSE values were smaller when the number of neurons was 10, 12, and 14 and their training R^2 values were all close to 1. However, the values of validation R^2 , testing R^2 , and R^2 (all) were not close to 1 compared to 10 for the number of neurons were 12 and 14. This indicated training overftting (12 and 14), resulting in a decrease in the model prediction performance. Therefore, $5 \times 10 \times 2$ was used as the optimal architecture for the model (Table [5](#page-9-1)). The training and predicted R^2 were 0.9997 and 0.9336 (Fig. [6\)](#page-10-0), respectively, which were closer to one than the predicted R^2 of the RSM (0.9175, 0.9092). Additional experiments conducted under the optimum conditions for TP and turbidity removal were $88.79 \pm 5.45\%$ and $63.48 \pm 9.60\%$, respectively, and the values predicted by the RSM-ANN and the RSM were 87.71 \pm 5.74%, 64.62 \pm 10.75%, 83.28%, and 59.80%, respectively. Furthermore, the treated water qualities of the

the predicted value for turbidity removal by the RSM and BP-ANN models, and **d** distribution of the residual values of turbidity removal

Table 5 MSE values and R^2 of the different topologies of the RSM-ANN model

Topology	MSE	R^2				
		Tr	Val	Test	All	
$5 \times 7 \times 2$	128.1849	0.9998	0.7450	0.7085	0.9040	
$5 \times 8 \times 2$	219.9573	1	0.7041	0.9704	0.9350	
$5 \times 9 \times 2$	102.1649	0.9877	0.8359	0.8382	0.9244	
$5 \times 10 \times 2^*$	100.4588	0.9997	0.9483	0.9336	0.9636	
$5 \times 11 \times 2$	104.7157	0.9676	0.8792	0.7456	0.8860	
$5 \times 12 \times 2$	46.7556	0.9553	0.8522	0.4236	0.8006	
$5 \times 13 \times 2$	257.8313	0.2643	0.6046	0.3512	0.0186	
$5 \times 14 \times 2$	17.3254	0.9850	0.9964	0.7040	0.8223	

Tr, Training; *Val*, validation

*The optimum topology

TP concentration and turbidity values were $0.17 \pm 6.69\%$ mg/L and $2.46 \pm 5.09\%$ NTU, respectively, which completely met the surface water class IV standard. This result indicates that the RSM-ANN provided a better predicted performance than a single RSM model. The RSM-ANN model improves on the limitations of the RSM while allowing for more practical applications for guiding wastewater treatment plants.

Conclusion

In this study, a coagulation process enhanced by $Fe₃O₄$ as the magnetic powder was employed to further treat the secondary effluent from a WWTP to improve the effluent quality. Multiple responses in terms of the TP and turbidity removals were simultaneously optimized and predicted with the help of the RSM and the BP-ANN approaches. The optimum conditions were obtained using the RSM at a PAC of 28.42 mg/L, magnetic powder of 623 mg/L, and PAM of 0.18 mg/L. Both models had good fitness and prediction abilities, while the RSM provided a slightly better prediction performance under known water quality conditions compared to the single BP-ANN model.

A novel approach combining the RSM and BP-ANN was then developed by adding uncontrollable variables. The RSM-ANN model was optimized and validated using the additional experimental data. Under optimum conditions, the TP and turbidity removal obtained were 88.79 \pm 5.45% and 63.48 \pm 9.60%, respectively, and the predictions of the RSM-ANN (87.71 \pm 5.74% for TP and $64.62 \pm 10.75\%$ for turbidity) were closer than that of the single RSM predictions (83.28% for TP and 59.80% for turbidity) to the experimental values. Moreover, the effluent TP and turbidity were $0.17 \pm 6.69\%$ mg/L and $2.46 \pm 5.09\%$ NTU, respectively, which were far below the water quality standard of Class IV. Therefore, the results of the study showed that the magnetic coagulation process could be used as an advanced treatment to improve the discharged water quality from a WWTP, and the combined RSM-ANN could be an efective and powerful approach for predicting TP and turbidity with instant adjustments of the operational variables.

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Author contribution KW established the models, analyzed and interpreted the experimental data, and was a major contributor in writing the manuscript. YM designed the experiments, performed the water quality measurements along with KW, and was a contributor on graphical processing. CW did the literature research. QK downloaded and installed the required software. MZ provided the vehicles for the experimental condition. QW came up with the idea, consulted the literatures, and contacted with the management of the multiple wastewater treatment plants. All authors read and approved the fnal manuscript.

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Data availability The datasets used and/or analyzed during the current study are available from the corresponding author on reasonable request.

Declarations

Ethics approval and consent to participate Not applicable.

Consent for publication Not applicable.

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References

- Barekati-Goudarzi M, Mehrnia MR, Roudsari FP, Boldor D (2016) Rapid separation of microalga Chlorella vulgaris using magnetic chitosan: process optimization using response surface methodology. Particul Sci Technol 34:165–172. [https://doi.org/10.1080/](https://doi.org/10.1080/02726351.2015.1054973) [02726351.2015.1054973](https://doi.org/10.1080/02726351.2015.1054973)
- Beltrán Heredia J, Sánchez Martín J (2009) Removing heavy metals from polluted surface water with a tannin-based focculant agent. J Hazard Mater 165:1215–1218. [https://doi.org/10.1016/j.jhazm](https://doi.org/10.1016/j.jhazmat.2008.09.104) [at.2008.09.104](https://doi.org/10.1016/j.jhazmat.2008.09.104)
- Bogunovi M, Ivanev-Tumbas I, Česen M, Esen M, Sekulić TD, Prodanović J, Tubić A, Heath D, Heath E (2021) Removal of selected emerging micropollutants from wastewater treatment plant effluent by advanced non-oxidative treatment - a lab-scale case study from Serbia. Sci Total Environ 765:142764. [https://doi.](https://doi.org/10.1016/j.scitotenv.2020.142764) [org/10.1016/j.scitotenv.2020.142764](https://doi.org/10.1016/j.scitotenv.2020.142764)
- Dai LY (2017) Application of magnetic coagulation technology in high phosphorus wastewater treatment. Phosphate & Compd. Fertil. 032(010):47–49 ((in Chinese))
- Ding N, Wang X, Jiang L, Geng Y, Dong LM, Liu H (2021) Enhancement of sludge dewaterability by a magnetic feld combined with coagulation/focculation: a comparative study on municipal and citric acid–processing waste-activated sludge. Environ Sci Pollut Res 28(3):1–10.<https://doi.org/10.1007/s11356-021-13278-x>
- Gadekar MR, Ahammed MM (2019) Modelling dye removal by adsorption onto water treatment residuals using combined response surface methodology-artifcial neural network approach. J Environ Manage 231(FEB.1):241–248. [https://doi.org/10.1016/j.](https://doi.org/10.1016/j.jenvman.2018.10.017) [jenvman.2018.10.017](https://doi.org/10.1016/j.jenvman.2018.10.017)
- Ganapathy S, Balasubramanian P, Vasanth B (2020) Comparative investigation of Artifcial Neural Network (ANN) and Response Surface Methodology (RSM) expectation in EDM parameters. Materials Today: Proceedings 46(19):9592–9596. [https://doi.org/](https://doi.org/10.1016/j.matpr.2020.05.499) [10.1016/j.matpr.2020.05.499](https://doi.org/10.1016/j.matpr.2020.05.499)
- Ghritlahre HK, Prasad RK (2018) Application of ANN technique to predict the performance of solar collector systems - a review. Renew Sust Energ Rev 84:75–88. [https://doi.org/10.1016/j.rser.](https://doi.org/10.1016/j.rser.2018.01.001) [2018.01.001](https://doi.org/10.1016/j.rser.2018.01.001)
- Hafeez A, Taqvi SAA, Fazal T, Javed F, Khan Z, Amjad US, Bokhari A, Shehzad N, Rashid N, Rehman S, Rehman F (2020) Optimization on cleaner intensifcation of ozone production using artifcial neural network and response surface methodology: parametric and comparative study. J Clean Prod 252:119833. [https://doi.org/10.](https://doi.org/10.1016/j.jclepro.2019.119833) [1016/j.jclepro.2019.119833](https://doi.org/10.1016/j.jclepro.2019.119833)
- Ho N, Babel S (2020) Bioelectrochemical technology for recovery of silver from contaminated aqueous solution: a review. Environ Sci Pollut Res 2020:1–15. [https://doi.org/10.1007/](https://doi.org/10.1007/s11356-020-10065-y) [s11356-020-10065-y](https://doi.org/10.1007/s11356-020-10065-y)
- Huangfu XL, Ma C, Ma J, He Q, Yang C, Jiang J, Wang Y, Wu Z (2017) Signifcantly improving trace thallium removal from surface waters during coagulation enhanced by nanosized manganese dioxide. Chemosphere 168:264–271. [https://doi.org/10.1016/j.](https://doi.org/10.1016/j.chemosphere.2016.10.054) [chemosphere.2016.10.054](https://doi.org/10.1016/j.chemosphere.2016.10.054)
- Igwegbe CA, Mohammadi L, Ahmadi S (2019) Modeling of adsorption of methylene blue dye on $Ho\text{-}CaWO₄$ nanoparticles using response surface methodology (RSM) and artifcial neural network (ANN) techniques. MethodsX 6:1779–1797. [https://doi.org/](https://doi.org/10.1016/j.mex.2019.07.016) [10.1016/j.mex.2019.07.016](https://doi.org/10.1016/j.mex.2019.07.016)
- Jiang W, Xu H, Wang DS, Xiao F (2014) The characteristics of focs and zeta potential in nano-TiO₂ system under different coagulation conditions. Colloids Surf, A 452:181–188. [https://doi.org/10.](https://doi.org/10.1016/j.colsurfa.2014.03.073) [1016/j.colsurfa.2014.03.073](https://doi.org/10.1016/j.colsurfa.2014.03.073)
- Katz I, Dosoretz CG (2008) Desalination of domestic wastewater efuents: phosphate removal as pretreatment. Desalination 222:230– 242.<https://doi.org/10.1016/j.desal.2007.01.160>
- Khalid AAH, Yaakob Z, Abdullah SRS (2019) Analysis of the elemental composition and uptake mechanism of Chlorella sorokiniana for nutrient removal in agricultural wastewater under optimized response surface methodology (RSM) conditions. J Clean Prod 210:673–686. <https://doi.org/10.1016/j.jclepro.2018.11.095>
- Krupińska I (2020) The efect of the type of hydrolysis of aluminium coagulants on the efectiveness of organic substances removal from water. Desalin Water Treat 186:171–180. [https://doi.org/](https://doi.org/10.5004/dwt.2020.25248) [10.5004/dwt.2020.25248](https://doi.org/10.5004/dwt.2020.25248)
- Kumari M, Gupta SK (2020) A novel process of adsorption cum enhanced coagulation-focculation spiked with magnetic nano adsorbents for the removal of aromatic and hydrophobic fraction of natural organic matter along with turbidity from drinking water. J Clean Prod 244:118899. [https://doi.org/10.1016/j.jclepro.2019.](https://doi.org/10.1016/j.jclepro.2019.118899) [118899](https://doi.org/10.1016/j.jclepro.2019.118899)
- Li C, Ma J, Shen J, Peng W (2009) Removal of phosphate from secondary effluent with Fe_2^+ enhanced by H_2O_2 at nature pH/neutral

pH. J Hazard Mater 166(2–3):891–896. [https://doi.org/10.1016/j.](https://doi.org/10.1016/j.jhazmat.2008.11.111) [jhazmat.2008.11.111](https://doi.org/10.1016/j.jhazmat.2008.11.111)

- Li RH, Guo BY, Sun JZ, Yue QY (2018) Coagulation behavior of kaolin-anionic surfactant simulative wastewater by polyaluminum chloride-polymer dual coagulant. Environ Sci Pollut Res 25:7382– 7390.<https://doi.org/10.1007/s11356-017-1073-0>
- Liu D, Wang P, Wei GR, Dong WB, Hui F (2013a) Removal of algal blooms from freshwater by the coagulation-magnetic separation method. Environ Sci Pollut Res 20:60–65. [https://doi.org/10.1007/](https://doi.org/10.1007/s11356-012-1052-4) [s11356-012-1052-4](https://doi.org/10.1007/s11356-012-1052-4)
- Liu YY, Zhang WJ, Yang XY, Xiao P, Wang DS, Song YQ (2013b) Advanced treatment of effluent from municipal WWTP with different metal salt coagulants: contaminants treatability and foc properties. Sep Purif Technol 120:123–128. [https://doi.org/10.](https://doi.org/10.1016/j.seppur.2013.09.046) [1016/j.seppur.2013.09.046](https://doi.org/10.1016/j.seppur.2013.09.046)
- Lohwacharin J, Phetrak A, Oguma K, Takizawa S (2014) Flocculation performance of magnetic particles with high-turbidity surface water. Water Supply 14(4):609–617. [https://doi.org/10.2166/ws.](https://doi.org/10.2166/ws.2014.015) [2014.015](https://doi.org/10.2166/ws.2014.015)
- Lv M, Zhang Z, Zeng J, Liu JF, Sun MC, Yadav RS, Feng YJ (2019) Roles of magnetic particles in magnetic seeding coagulation-focculation process for surface water treatment. Sep Purif Technol 212:337–343. <https://doi.org/10.1016/j.seppur.2018.11.011>
- Lv M, Li D, Zhang Z (2020) Feng Y (2020) Magnetic seeding coagulation: effect of Al species and magnetic particles on coagulation efficiency, residual Al, and floc properties. Chemosphere 268(1):129363. [https://doi.org/10.1016/j.chemosphere.2020.](https://doi.org/10.1016/j.chemosphere.2020.129363) [129363](https://doi.org/10.1016/j.chemosphere.2020.129363)
- Mahmoodi-Babolan N, Heydari A, Nematollahzadeh A (2019) Removal of methylene blue via bioinspired catecholamine/starch superadsorbent and the efficiency prediction by response surface methodology and artifcial neural network-particle swarm optimization. Bioresour Technol 294:122084. [https://doi.org/10.1016/j.](https://doi.org/10.1016/j.biortech.2019.122084) [biortech.2019.122084](https://doi.org/10.1016/j.biortech.2019.122084)
- Mohammad R, Hamed T, Mohammad RR (2019) Degradation of crystal violet in water solution using post discharge DBD plasma treatment: factorial design experiment and modeling. Chemosphere 232:213–223. <https://doi.org/10.1016/j.chemosphere.2019.05.153>
- Momeni MM, Kahforoushan D, Abbasi F, Ghanbarian S (2018) Using chitosan/chpatc as coagulant to remove color and turbidity of industrial wastewater optimization through RSM design. J Environ Manage 211(APR.1):347–355. [https://doi.org/10.1016/j.jenvman.](https://doi.org/10.1016/j.jenvman.2018.01.031) [2018.01.031](https://doi.org/10.1016/j.jenvman.2018.01.031)
- Mortadi A, Chahid EG, Abderrahmane E, Chahbi M, Moznine, RE (2020) Complex electrical conductivity as a new technique to monitor the coagulation-focculation processes in the wastewater treatment of the textile industry. Water Resour Ind 24:100130. <https://doi.org/10.1016/j.wri.2020.100130>
- Pambi RL, Musonge P (2016) Application of Response Surface Methodology (RSM) In the treatment of final effluent from the sugar industry using Chitosan. Water Pollution 209:209–219. [https://](https://doi.org/10.2495/WP160191) doi.org/10.2495/WP160191
- Qasim M, Park S, Lee J, Kim JO (2019) Evaluation of foc settling velocity models through image analysis for ballasted focculation. Desalin Water Treat 144:370–383. [https://doi.org/10.5004/](https://doi.org/10.5004/dwt.2019.23586) [dwt.2019.23586](https://doi.org/10.5004/dwt.2019.23586)
- Rizzo L, Gennaro AD, Gallo M (2008) Coagulation/chlorination of surface water: a comparison between chitosan and metal salts. Sep Purif Technol 62(1):79–85. [https://doi.org/10.1016/j.seppur.](https://doi.org/10.1016/j.seppur.2007.12.020) [2007.12.020](https://doi.org/10.1016/j.seppur.2007.12.020)
- Rizzo L, Malato S, Antakyali D, Beretsou VG, Đolić MB, Gernjak W, Heath E, Ivancev-Tumbas I, Karaolia P, Ribeiro ARL, Mascolo G, McArdell CS, Schaar H, Silva AMT, Fatta-Kassinos D (2019) Consolidated vs new advanced treatment methods for the removal of contaminants of emerging concern from urban wastewater. Sci Total Environ 655:986–1008. [https://doi.org/10.1016/j.scitotenv.](https://doi.org/10.1016/j.scitotenv.2018.11.265) [2018.11.265](https://doi.org/10.1016/j.scitotenv.2018.11.265)
- Santos TRT, Silva MF, Nishi L, Vieira AMS, Klein MRF, Andrade MB, Vieira MF, Bergamasco R (2016) Development of a magnetic coagulant based on Moringa oleifera seed extract for water treatment. Environ Sci Pollut R 23:7692–7700. [https://doi.org/10.](https://doi.org/10.1007/s11356-015-6029-7) [1007/s11356-015-6029-7](https://doi.org/10.1007/s11356-015-6029-7)
- Tetteh EK, Rathilal S (2018) Evaluation of the coagulation foatation process for industrial mineral oil wastewater treatment using response surface methodology (RSM). Int J Environ Impacts 1(4):491–502.<https://doi.org/10.2495/EI-V1-N4-491-502>
- Usef S, Asadi-Ghalhari M (2019) Optimization of turbidity removal by coagulation and focculation process from synthetic stone cutting wastewater. International J Energy Wat Resour 3:33–41. [https://](https://doi.org/10.1007/s42108-019-00010-2) doi.org/10.1007/s42108-019-00010-2
- Wang JP, Chen YZ, Wang Y, Yuan SJ, Yu HQ (2011) Optimization of the coagulation-focculation process for pulp mill wastewater treatment using a combination of uniform design and response surface methodology. Water Res 45:5633–5640. [https://doi.org/](https://doi.org/10.1016/j.watres.2011.08.023) [10.1016/j.watres.2011.08.023](https://doi.org/10.1016/j.watres.2011.08.023)
- Wang P, Ding S, An G (2021) Removal of disinfection by-product precursors by Al-based coagulants: a comparative study on coagulation performance. J Hazard Mater 420:126558. [https://doi.org/](https://doi.org/10.1016/j.jhazmat.2021.126558) [10.1016/j.jhazmat.2021.126558](https://doi.org/10.1016/j.jhazmat.2021.126558)
- Xu Q, Xiao KK, Wang H, Wu QX, Liang S, Yu WB, Hou HJ, Liu BC, Hu JP, Yang JK (2020) Insight into efects of organic and inorganic phosphorus speciations on phosphorus removal efficiency in secondary effluent. Environ Sci Pollut Res 27(11):11736-11748. <https://doi.org/10.1007/s11356-020-07774-9>
- Yao M, Nan J, Chen T (2014) Efect of particle size distribution on turbidity under various water quality levels during focculation processes. Desalination 354:116–124. [https://doi.org/10.1016/j.](https://doi.org/10.1016/j.desal.2014.09.029) [desal.2014.09.029](https://doi.org/10.1016/j.desal.2014.09.029)
- Zhang X, Yao J, Peng W, Xu W, Li Z, Zhou C (2018) Degradation of dichloroacetonitrile by a UV/peroxymonosulfate process: modeling and optimization based on response surface methodology (RSM). RSC Adv 8(59):33681–33687. [https://doi.org/10.1039/](https://doi.org/10.1039/c8ra07009a) [c8ra07009a](https://doi.org/10.1039/c8ra07009a)
- Zhou G, Wang Q, Li J, Li QS, Xu H, Ye Q, Wang YQ, Shu SH, Zhang J (2021) Removal of polystyrene and polyethylene microplastics using PAC and $FeCl₃ coagulation: performance and mechanism.$ Sci Total Environ 752:141837. [https://doi.org/10.1016/j.scitotenv.](https://doi.org/10.1016/j.scitotenv.2020.141837) [2020.141837](https://doi.org/10.1016/j.scitotenv.2020.141837)

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