**RESEARCH ARTICLE** 



# Mercury and selenium biomagnification in a coastal food web from the Gulf of California influenced by agriculture and shrimp aquaculture

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#### Abstract

The biomagnification of Hg and Se was studied using nitrogen stable isotope analysis during four seasons in a coastal lagoon of the eastern central Gulf of California. This lagoon receives agricultural, municipal, and shrimp aquaculture effluents. The species were categorized into organism groups and presented a significant accumulation of Hg and Se with respect to the sources, while the concentration of both elements in sediment and suspended particulate matter (SPM) was low. Our data confirms the positive transfers (biomagnification factors >1) of Hg and Se in the entire studied food web, and it was structured in five trophic levels across all seasons. Additionally, there were no linear correlations between the molar Se:Hg ratios and the trophic levels of the organism groups. However, the Se:Hg ratios among organism groups were >1, which indicates that there is an excess of Se and that it is not a limiting factor for the detoxification of Hg.

Keywords Mercury · Selenium · Stable isotopes · Biomagnification · Coastal lagoon · Anthropogenic activities

# Introduction

Mercury (Hg) contamination is a worldwide problem given it is a persistent and highly toxic contaminant that occurs ubiquitously in the environment. Its bioaccumulation in marine food webs is several times higher compared to baseline levels, and the greatest concern is related to the transformation of reactive Hg into organic species, which are highly toxic even at low concentrations (Fitzgerald and Lamborg 2005; Dehn et al. 2006). The accumulation of Hg species depends on abiotic (e.g., pH, salinity, redox, and chemical speciation) and biotic factors (e.g., diet, age, sex, the physiological role of each element, and trophic position) (Lavoie et al. 2013). Conversely, selenium (Se) is an essential element for all organisms given it is a cofactor of enzymes (e.g., glutathione peroxidase or thioredoxin reductase); its deficiency results in

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the increased risk of the development of many chronic degenerative diseases, and the action as a protective agent against the toxicity of Hg was proposed by the formation of insoluble Hg–Se complexes (Sørmo et al. 2011; Kehrig et al. 2013). A molecular excess of Se ratio to methyl-mercury (CH<sub>3</sub>Hg) prevents the inhibition of selenoenzyme activities, thereby alleviating Hg–exposure risks (Ralston et al. 2019). However, high levels of Se can be toxic for organisms, and their interactions with Hg may also have synergistic effects depending on the sensitivity of the organ or organism which is determined by the threshold concentrations of both elements (Khan and Wang 2009; Dang and Wang 2011).

Although Hg biomagnification has been usually documented (e.g., Dehn et al. 2006; Jara-Marini et al. 2012; Kehrig et al. 2013; Lavoie et al. 2013), evidence for Se biomagnification is inconsistent with cases of biomagnification (e.g., Kehrig et al. 2013; Økelsrud et al. 2016) and no biomagnification (e.g., Ouédraogo et al. 2015). According to Reinfelder et al. (1998), the principal factors that drive the Hg and Se transference in food webs are the assimilation efficiency, the ingestion rate, the physiological loss, and the dilution by growth; however, Stewart et al. (2004) proposed that the dominant factor controlling this transference may vary under specific conditions.

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The variation in the ratio of stable isotopes of nitrogen  $(\delta^{15}N=^{15}N/^{14}N)$  has been used in ecology to determine the relative trophic position of species as a function of its time-integrated dietary history (Minagawa and Wada 1984). Due to the relatively constant enrichment rate of the heavier isotope (<sup>15</sup>N) across trophic levels (2–4‰), nitrogen isotopic signatures provide valuable information on the trophic position of an organism in a food web (Post 2002). Additionally, the biomagnification of contaminants in food webs may be determined by the correlation between the trophic position of organisms and their contaminant concentrations (Post 2002; Power et al. 2002; Dehn et al. 2006; Jara-Marini et al. 2009, 2012; Økelsrud et al. 2016).

Marine ecosystems associated with the Gulf of California are ecologically and socioeconomically important in Mexico due to their elevated biodiversity, productivity, and commercial fisheries (Páez-Osuna et al. 2017). Specifically, the El Tobari (ET) lagoon is an ecosystem with the highest priority for conservation as a result of its rich biodiversity, high rates of biological productivity, and endemism of invertebrates, fish, and bird species (Aguilar et al. 2008). However, this lagoon is ecologically impacted by discharges of municipal, agriculture, and shrimp farm effluents that contain a myriad of pollutants, and some of the ecological (Martínez-Durazo et al. 2019) and metal accumulation (Jara-Marini et al. 2020) impacts have been previously reported. Approximately 230,000 ha of irrigated agriculture surface in the Yaqui Valley (Gortáres-Moroyoqui et al. 2011; McCullough and Matson 2016) extensively uses a great variety of agrochemicals and constitutes significant metal inputs for the ET lagoon. Furthermore, inputs of Hg and Se from intensive shrimp farming may be significant through the use of artificial feed, fertilizers, and other chemical additives, including acidity correctors and algaecides (Lyle-Fritch et al. 2006). Similar conditions occur in other coastal ecosystems in the ecoregion of the Gulf of California. In this study, we examined the biomagnification of Hg and Se in the food web of the ET lagoon during four typical seasons (summer and autumn 2011 and winter and spring 2012). Our study objectives were to (a) characterize the trophic position of food web species through stable nitrogen isotopic ratios; (b) assess the Hg and Se transfer and biomagnification through the ecosystem food web; and (c) determine the Se:Hg molar ratios in the organisms of the food web.

# Materials and methods

#### Study area

The ET lagoon is located in northwestern Sonora, Mexico, along the eastern central coast of the Gulf of California (Fig.

1). The characteristics and prevailing climatic conditions of the ET lagoon have been described in previous studies (Jara-Marini et al. 2013a, b). This ecosystem plays a crucial ecological role in the support of a variety of endemic and migratory organisms, and for this reason, it was included among the Conservation of Coastal and Oceanic Priority Zones of Mexico (Aguilar et al. 2008). However, the ET lagoon also receives the regular effluent discharges from (a) agriculture (10 irrigation districts ~230,000 ha); (b) municipal untreated and partially treated sewage from Obregon City and other surrounding towns (~375,800 habitants); and )c) aquaculture (1190 ha of shrimp ponds) (Gortáres-Moroyoqui et al. 2011; INEGI 2015; McCullough and Matson 2016).

#### Sampling and processing

Four sample collections were conducted during the summer (August 2011), autumn (October 2011), winter (February 2012), and spring (May 2012) at seven highly biodiverse sites inside the ET lagoon and in one reference site outside of the lagoon (Fig. 1). In each site, samples of subsurface water (<1 m depth, 1 L Nalgene bottles), surface sediment (0–2.5 cm from the top of the sedimentary column), and numerous organisms were collected following standardized sampling protocols. Details of the sampling and processing of the samples were previously reported in Martínez-Durazo et al. (2019) and Jara-Marini et al. (2020).

The surface water samples were filtered through a precleaned and precombusted (500°C, 4 h) glass fiber filter (GF/F), and the total suspended particulate matter (SPM) contents were gravimetrically determined. The organic carbon (Loring and Rantala 1992), carbonates (Rauret et al. 1988), and the Hg and Se contents in bioavailable fractions (chemical sequential extraction scheme of Huerta-Diaz and Morse 1990) were determined in surface sediments. The composited samples of representative species of different organism groups in the ET lagoon food web were identified according to Fischer et al. (1995a, b, c) and Sibley (2000) and processed for isotopic and metal analysis according to Martínez-Durazo et al. (2019) and Jara-Marini et al. (2020). The organism groups and the characteristics of the abiotic samples are listed in Table 1.

#### Hg and Se analyses

Lyophilized tissues of the biological samples ( $0.30 \pm 0.003$  g on dry weight basis) were digested with 5 mL of concentrated HNO<sub>3</sub> and heated in three steps ( $100 \degree C$  for 5 min, then  $120 \degree C$  for 5 min, and finally 140  $\degree C$  for 10 min) using a microwave system (MARS-X). The bioavailable extracts from the sediments and the biological digested tissues were analyzed for Hg and Se using cold vapor and hydride generation (Model VGA 110), respectively,



Fig. 1 Map of the sampling sites in the study area

coupled to an atomic absorption spectrophotometer (Varian, Model SpectrAA-240-FS). Certified reference material (marine sediment PACS-2 and dogfish muscle DORM-2 from the National Research Council Canada) and blanks were used for metal quality control purposes. The resulting agreement with certified values for both elements was higher than 10% in sediment fractions and in the biological tissues. The limits of detection for the Hg and Se analyses were estimated as 0.21 and 0.33 ng g<sup>-1</sup> for the reactive fraction, 0.28 and 0.35 ng g<sup>-1</sup> for the pyrite fraction, and 0.32 and 0.38 ng g<sup>-1</sup> for the biological analysis, respectively.

## Stable isotope analyses

Subsamples of dried SPM, sediment, and biological tissues were processed for the removal of carbonates, after which they were dried and packed into tin cups for N isotope analysis (see details in Martínez-Durazo et al. 2019). The <sup>15</sup>N analyses were performed in the Environmental Isotope Laboratory of the Department of Geosciences (University of Arizona, Tucson, AZ) using a Finnigan Delta Plus XL continuous-flow gasratio Mass Spectrometer Carlo Erba NA 2100 ANCA-NT 20-20 Stable Isotope Analyzer (precision was  $\leq 0.2\%$  in n=100, as estimated by the international standards IAEA-N-1 and IAEA-N-2 from the International Atomic Energy Agency). The isotopic composition ( $\delta^{15}$ N) was expressed as the relative difference between the isotopic ratios in the sample and conventional standards (atmospheric  $N_2$ ):

$$\delta^{15}N (\%) = \left[ \left( R_{sample} / R_{standard} \right) - 1 \right] \times 1000$$
  
where R= <sup>15</sup>N /<sup>14</sup>N.

## Data analyses

Since the  $\delta^{15}$ N and Hg and Se values were non-normally distributed (Shapiro-Wilks and Bartlett tests, p<0.05), Kruskal-Wallis and Student-Newman-Keuls tests (Glantz 2002) were used to compare the isotopic and metal concentrations among seasonal groups. The estimation of the most important trophic relations (carbon and nitrogen sources) was reported previously by Del Rio-Salas et al. (2019) using a stable isotope Bayesian SIAR mixing model (Parnell et al. 2010).

The trophic level (TL) was estimated assuming a <sup>15</sup>N trophic enrichment factor of 2.54% (Vanderklift and Ponsard 2003), according to the following formula (Hobson and Welch 1992; Post 2002):

$$TL = 2 + \left[ \left( \delta^{15} N_{\text{secondary consumer}} - \delta^{15} N_{\text{primary consumer}} \right) / 2.54 \right]$$

where  $\delta^{15}N_{secondary\ consumer}$  is the  $\delta^{15}N$  composition of the secondary or higher consumer, and  $\delta^{15}N_{primary\ consumer}$  is the  $\delta^{15}N$  composition of the baseline consumer (the composition used was zooplankton).

Name/group	Species	Feeding habit	Tissue	Sampling site	Number of pooled organisms	п
Source						
Sediment	-	-	_	1, 2, 3, 4, 5, 6, 7, and	-	64
Suspended particulate matter (SPM) Primary producers	_	_	_	8 1, 2, 3, 4, 5, 6, 7, and 8	_	64
Phytoplankton (seston)	Diatoms (90–98%) and dinoflagellates $(2, 10\%)$	Autotrophic	Whole	1, 2, 4, 5, 6, 7, and 8	n	28
Macroalgae	(2–10%) Ulva intestinalis	Autotrophic	Fronds	1, 2, 4, 5, 6, 7, and 8	n	20
Macroalgae	Ulva lactuca	Autotrophic	Fronds	1, 2, 4, 5, 6, 7, and 8	n	24
Macroalgae	Gracilaria vermiculophylla	Autotrophic	Fronds	1, 2, 4, 5, 6, 7, and 8	n	14
Macroalgae	Spyridia filamentosa	Autotrophic	Fronds	1, 2, 4, 5, and 6	n	16
Mangrove	Rhizophora mangle	Autotrophic	Leaves	4, 5, and 6	6	12
Mangrove	Avicennia germinans	Autotrophic	Leaves	4,5, and 6	6	12
Primary consumers						
Zooplankton	Copepods (95–99%), Chaetognatha $(1-2\%)$ , and zoea $(1-3\%)$	Plankton consumer	Whole	1, 2, 4, 5, 6, 7, and 8	n	28
Clam	Chione gnidia	Filter feeding	Soft	1, 2, 4, 5, 6, 7, and 8	4	28
Clam	Chione fluctifraga	Filter feeding	Soft	1, 2, 4, 5, 6, 7, and 8	6	28
Clam	Anadara tuberculosa	Filter feeding	Soft	2, 4, 5, and 6	3	16
Oyster	Crassostrea cortezienzes	Filter feeding	Soft	4, 5, and 6	10	12
Oyster	Crassostrea gigas	Filter feeding	Soft	4 and 6	5	8
Barnacles	Fistulobalanus dentivarians	Filter feeding	Soft	4, 5, and 6	30	12
Snail	Littoraria aberrans	Filter feeding	Soft	4, 5, and 6	7	18
Snail	Hexaplex erythrostoma	Omnivorous	Soft	1, 2, 4, 5, 6, and 7	5	24
Flathead mullet-juvenile	Mugil cephalus	Omnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	2	14
Secondary consumers						
Shrimp	Litopenaeus vannamei	Omnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	6	28
Crab-juvenile	Callinectes arcuatus	Omnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	4	14
Yellow fin mojarra-juvenile	Gerres cinereus	Omnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	3	10
White grunt-juvenile	Haemulois leuciscus	Omnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	3	10
Snapper-juvenile	Lutjanus argentivensis	Omnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	3	12
Tertiary consumers						
Flathead mullet-adult	Mugil cephalus	Omnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	2	28
Crab-adult	Callinectes arcuatus	Carnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	1	24
Yellow fin mojarra-adult	Gerres cinereus	Carnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	1	20
White grunt-adult	Haemulopsis leuciscus	Carnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	1	20
Snapper-adult	Lutjanus argentivensis	Carnivorous	Whole/muscle	1, 2, 4, 5, 6, 7, and 8	1	20
Quaternary consumers						
Yellow-crowned night heron	Nyctanassa violacea	Carnivorous	Blood	1, 2, and 7	2	10
Clapper rail	Rallus longirostris	Carnivorous	Blood	1, 2, and 7	2	10
Roseate spoonbill	Platalea ajaja	Carnivorous	Blood	1, 2, and 7	2	8
Cormorant	Phalacrocorax brasilianus	Carnivorous	Whole/muscle	1,2, and 7	1	8
Magnificent frigatebird	Fregata magnificens	Carnivorous	Whole/muscle	1, 2, and 7	1	10

 Table 1
 Sediment, suspended particulate matter, and groups of organisms collected in the El Tobari lagoon ecosystem during the four seasons<sup>1,2</sup>

<sup>1</sup> *n*: number total of samples

<sup>2</sup> In parenthesis, average of weight of pooled or unitary samples

The biomagnification factor (BMF) corrected for differences in trophic position (based on  $\delta^{15}$ N) was calculated according to the procedure of Dehn et al. (2006) for the selected predator-prey scenarios as:

$$BMF = (M_{Predator}/M_{Prey})/(\delta^{15}N_{Predator}/\delta^{15}N_{Prey})$$

where  $M_{Predator}$  and  $M_{Prey}$  are the concentrations of each metal (µg  $g^{-1}$  DW) in the predator and prey species, and  $\delta^{15}N_{Predator}/\delta^{15}N_{Prey}$  is the N composition (%) of the predator and prey pair.

The Hg and Se concentrations in the groups of organisms were logarithm-transformed (log-transformed) to normalize the distribution. A simple linear regression model using TL values as the independent variable and element concentrations in organisms as the dependent variables were then performed to establish the trophic relationships between both of these variables. The trophic biomagnification factor (TBMF) for the two studied elements was estimated from the antilogarithm of the slopes of these linear correlations. Seasonal differences of TBMF values for Hg and Se were evidenced using a t-student test ( $\alpha$ =0.05). A TBMF>1 indicates the entire biomagnification of a contaminant in the food web (Mackay et al. 2016). All of the statistical analyses were performed using the NCSS Statistical Software (NCSS 2007).

# Results

#### Hg and Se concentrations in sources and organisms

The concentrations of Hg in the bioavailable (reactive plus pyrite) fraction of sediment and in the SPM on the August (0.10 to 0.11  $\mu$ g g<sup>-1</sup>) and February (0.09 to 0.11  $\mu$ g g<sup>-1</sup>) samples evidenced nonsignificant differences; however, their levels increased significantly (p<0.05) in the May (0.27 to 0.29  $\mu$ g g<sup>-1</sup>) and October (0.26 to 0.30  $\mu$ g g<sup>-1</sup>) samples (Table 2). Significant accumulations of Hg with respect to sediments and SPM were determined in macroalgae species, which varied from 0.84  $\mu$ g g<sup>-1</sup> for *Spyridia filamentosa* in May to 1.57  $\mu$ g g<sup>-1</sup> for *Ulva lactuca* in August (Table 2). The two mangrove species showed similar (p>0.05) Hg concentrations ranging from 0.82  $\mu$ g g<sup>-1</sup> in October to 1.15  $\mu$ g g<sup>-1</sup> in February, while phytoplankton presented the highest accumulation of this element among primary producers and varied from 1.14  $\mu$ g g<sup>-1</sup> in May to 1.69  $\mu$ g g<sup>-1</sup> in August.

The primary consumers showed increases in the levels of Hg with respect to primary producers that were most significant in August (from 1.43  $\mu$ g g<sup>-1</sup> for *Mugil cephalus* juvenile fish to 3.52  $\mu$ g g<sup>-1</sup> for *Callinectes arcuatus* juvenile crab) and February (from 1.66  $\mu$ g g<sup>-1</sup> for *M. cephalus* juvenile fish to 4.28  $\mu$ g g<sup>-1</sup> for *Haemulopsis leuciscus* juvenile fish). Similarly, the accumulation of Hg in secondary and tertiary

consumers was significant (p<0.05) with respect to primary consumers, and this was most evident in August (varying from 2.44  $\mu$ g g<sup>-1</sup> in adults of *M. cephalus* to 5.16  $\mu$ g g<sup>-1</sup> in adults of *C. arcuatus*) and February (varying from 4.02  $\mu$ g g<sup>-1</sup> in adults of *M. cephalus* adult to 5.18  $\mu$ g g<sup>-1</sup> in adult fishes of *Lutjanus argentivensis*). The highest Hg levels were determined in the quaternary consumers (birds) during May, and they ranged from 3.09  $\mu$ g g<sup>-1</sup> in *Platalea ajaja* to 6.44  $\mu$ g g<sup>-1</sup> in *Fregata magnificens*. Among this group of species, *F. magnificens* presented the highest concentrations of Hg during all seasons ranging from 4.19  $\mu$ g g<sup>-1</sup> in October to 6.44  $\mu$ g g<sup>-1</sup> in May.

In addition, Table 2 presents the Se concentrations in sources and organism groups. The concentrations of Se in the sediment (0.07 to 0.19  $\mu g \ g^{-1})$  and SPM (0.09 to  $0.12 \ \mu g \ g^{-1}$ ) varied slightly during the four sampling seasons. The levels of Se in primary producers with respect to sediment and SPM increased significantly (p<0.05), with seasonal variations and ranges from 1.24 to 2.69  $\mu$ g g<sup>-1</sup> in August, from 1.80 to 3.31  $\mu$ g g<sup>-1</sup> in February, from 2.34 to 3.22  $\mu$ g g<sup>-1</sup> in May, and from 2.12 to 2.92  $\mu g g^{-1}$  in October. This group exhibited seasonal variation in their Se accumulation, with the most significant (p<0.05) high levels for macroalgae species (from 3.02 to 3.31  $\mu$ g g<sup>-1</sup>) in February and in the mangrove Avicennia germinans (3.22  $\mu$ g g<sup>-1</sup>) in May. The Se accumulation in primary consumers with respect to primary producers was significant (p<0.05) during all seasons, ranging from 2.29  $\mu$ g g<sup>-1</sup> for *M. cephalus* juveniles during October to 21.2  $\mu$ g g<sup>-1</sup> for the barnacle *Fistulobalanus dentivarians* in August. Some species of this group showed seasonal significant variations in Se content, with the lowest values in juveniles of *M. cephalus* (2.71 and 2.29  $\mu$ g g<sup>-1</sup> in August and May, respectively), C. arcuatus (2.74  $\mu$ g g<sup>-1</sup> in October), and L. argentivensis (2.71  $\mu g g^{-1}$  in May). The secondary and tertiary consumers presented significantly (p<0.05) higher levels of Se than the primary consumers, ranging from 4.81  $\mu$ g g<sup>-1</sup> in adults of *C. arcuatus* in October to 12.9  $\mu$ g  $g^{-1}$  in adults of *L. argentivensis*. Both groups also presented seasonal significant (p<0.05) variations in some species, with the highest values of 9.33  $\mu g g^{-1}$  for *M. cephalus* adults in May, 9.17  $\mu$ g g<sup>-1</sup> for *C. arcuatus* adults in August, and 12.9  $\mu$ g g<sup>-1</sup> for *L. argentivensis* adults in October. The Se accumulation in the quaternary consumers was significantly (p<0.05) higher than in other groups, with seasonal variations in the birds *Rallus longirostris* (from 14.6  $\mu$ g g<sup>-1</sup> in October to 28.3  $\mu$ g g<sup>-1</sup> in August) and *F. magnificens* (from 19.1  $\mu$ g g<sup>-1</sup> in May to 34.8  $\mu g g^{-1}$  in February).

No spatial significant (p>0.05) variations were determined regarding Hg and Se concentrations in sediments nor in organisms. The homogeneous distribution of these elements in the sediments may be related to hydrodynamic conditions of the ET lagoon, given site three is close to the northern mouth and the currents spread contaminants in the ecosystems

Table 2Mercuduring the four set	rry and seleniu easons <sup>1,2</sup> . The	m content (प्र ranges of bio	g $g^{-1}$ , mean $\pm$ magnification	: standard dev 1 factors (BM	/iation) and δ F) and the tre	<sup>15</sup> N (%0) com ophic biomagn	position, in th nification fact	ne organic mat ors (TBMF) fo	ter sources and groups of r both elements are also I	organisms colle presented	cted in the El Tobari	i (ET) lagoon
	Summer: Au	1011 gust 2011		Winter: Feb	ruary 2012		Spring: May	2012	Autumn: Oc	tober 2012	Range	s of BMF <sup>3</sup>
	Hg	Se	δ <sup>15</sup> N	Hg	Se	$\delta^{15}N$	Hg	Se	δ <sup>15</sup> N Hg	Se	δ <sup>15</sup> N Hg	Se
Sediment <sup>4</sup> SPM A. germinans R. mangle	$\begin{array}{l} 0.11^{a}\pm 0.01\\ 0.09^{a}\pm 0.01\\ 0.87^{c}\pm 0.09\\ 0.91^{c}\pm 0.11 \end{array}$	$\begin{array}{c} 0.07^{a}\pm0.01\\ 0.111^{a}\pm0.03\\ 2.69^{d}\pm0.31\\ 1.37^{c}\pm0.12\end{array}$	$\begin{array}{l} 8.75^{a}\pm1.75\\ 10.5^{ab}\pm1.98\\ 12.1^{b}\pm0.29\\ 11.4^{ab}\pm2.13\end{array}$	$\begin{array}{l} 0.10^{a}\pm0.03\\ 0.11^{a}\pm0.03\\ 0.99^{c}\pm0.07\\ 1.15^{c}\pm0.01 \end{array}$	$\begin{array}{l} 0.11^{a}\pm0.02\\ 0.09^{a}\pm0.01\\ 2.91^{d}\pm0.20\\ 3.02^{d}\pm0.08\end{array}$	$\begin{array}{l} 9.40^{a}\pm1.33\\ 8.30^{a}\pm1.99\\ 10.7^{ab}\pm0.14\\ 11.2^{ab}\pm0.64 \end{array}$	$\begin{array}{l} 0.27^{b}\pm0.03\\ 0.26^{b}\pm0.05\\ 0.91^{c}\pm0.03\\ 0.87^{c}\pm0.10 \end{array}$	$\begin{array}{c} 0.19^{a}\pm0.04\\ 0.11^{a}\pm0.02\\ 3.22^{cd}\pm0.13\\ 3.09^{cd}\pm0.29 \end{array}$	$\begin{array}{l} 8.81^a\pm2.02 & 0.29^b\pm0.03\\ 11.9^b\pm1.44 & 0.30^b\pm0.03\\ 9.17^a\pm0.53 & 0.86^c\pm0.06\\ 8.84^a\pm1.26 & 0.82^c\pm0.11 \end{array}$	$\begin{array}{c} 0.19^{a}\pm 0.06\\ 0.12^{a}\pm 0.03\\ 2.12^{c}\pm 0.13\\ 2.92^{c}\pm 0.15\end{array}$	$\begin{array}{l} 8.19^{a}\pm1.74 & -\\ 8.91^{a}\pm2.33 & -\\ 11.5^{ab}\pm1.73 & -\\ 8.41^{a}\pm1.17 & -\end{array}$	1 1 1 1
U. lactuca U. intestinalis	$\pm 0.06$ 1.57 <sup>d</sup> $\pm 0.08$ 1.30 <sup>d</sup> $\pm 0.13$	$1.37^{c} \pm 0.12$ $1.39^{c} \pm 0.14$	$13.7^{bc} \pm 0.42$ $11.1^{ab} \pm 0.85$ MS	$1.13^{\circ} \pm 0.10$ $1.06^{\circ} \pm 0.08$	$3.31^{d} \pm 0.39$ $3.25^{d} \pm 0.32$ $2.25^{d} \pm 0.32$	$12.9^{bc} \pm 1.13$ $11.7^{ab} \pm 0.42$	$0.85^{\circ} \pm 0.16$ $0.95^{\circ} \pm 0.21$	$2.40^{\circ} \pm 0.25$ $2.25^{\circ} \pm 0.18$ $2.27^{d} \pm 0.40$	$10.8^{ab} \pm 0.45 \ 0.99^{c} \pm 0.18$ $10.6^{ab} \pm 0.36 \ NS$ $11.1^{ab} \pm 0.31 \ 10.6^{c} \pm 0.15$	$2.19^{\circ} \pm 0.13$ NS	15.7 <sup>cd</sup> ± 0.45 - NS - 12.1 <sup>bc</sup> ± 1.26	1 1
o. <i>vermicuopnyu</i> S. <i>filamentosa</i> Phytoplankton Zooplankton	$\begin{array}{c} a \text{ NS} \\ 1.38^{d} \pm 0.22 \\ 1.69^{d} \pm 0.09 \\ 2.10^{e} \pm 0.06 \end{array}$	1.24 <sup>c</sup> $\pm$ 0.09 2.41 <sup>cd</sup> $\pm$ 0.24 4.12 <sup>e</sup> $\pm$ 0.24	10.6 <sup>ab</sup> $\pm$ 0.34 9.18 <sup>a</sup> $\pm$ 0.54 11.4 <sup>ab</sup> $\pm$ 0.43	$0.97 \pm 0.06$ NS $1.15^{\circ} \pm 0.08$ $1.96^{d} \pm 0.21$	$\frac{5.25 \pm 0.52}{\text{NS}}$ NS 1.80 <sup>c</sup> ± 0.54 3.07 <sup>d</sup> ± 0.20	NS 8.79 <sup>a</sup> $\pm$ 0.36 11.8 <sup>ab</sup> $\pm$ 0.40	$\begin{array}{l} 0.91 \ \pm \ 0.11 \\ 0.84^{c} \pm \ 0.16 \\ 1.14^{c} \pm \ 0.15 \\ 1.71^{de} \pm \ 0.15 \end{array}$	$3.2.7 \pm 0.49$ $3.46^{d} \pm 0.14$ $2.34^{c} \pm 0.29$ $4.36^{d} \pm 0.20$	11.1 $\pm 0.74$ 1.00 $\pm 0.13$ 11.5 <sup>bc</sup> $\pm 0.32$ 0.9 <sup>c</sup> $\pm 0.16$ 8.33 <sup>a</sup> $\pm 0.09$ 1.26 <sup>cd</sup> $\pm 0.13$ 11.2 <sup>ab</sup> $\pm 0.36$ 1.49 <sup>d</sup> $\pm 0.14$	2.26 $\pm 0.12$ 2.24° $\pm 0.15$ 2.285°° $\pm 0.18$ 3.78° <sup>d</sup> $\pm 0.14$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	- - 27 1.04–1.38
C. gnidia C. fluctifraga A. tuberculosa C. cortezienzes	$2.63^{\text{ef}} \pm 0.30$ $2.59^{\text{ef}} \pm 0.36$ $1.55^{\text{d}} \pm 0.15$ $2.59^{\text{ef}} \pm 0.63$	$\begin{array}{l} 4.02^{e}\pm 0.25\\ 8.65^{h}\pm 0.39\\ 5.60^{f}\pm 0.16\\ 8.43^{gh}\pm \end{array}$	$\begin{array}{c} 13.1^{bc}\pm 0.19\\ 12.0^{b}\pm 0.30\\ 12.1^{b}\pm 0.41\\ 12.1^{b}\pm 0.15\end{array}$	$2.45^{e} \pm 0.32$ $2.60^{e} \pm 0.08$ $2.81^{ef} \pm 0.11$ $2.55^{e} \pm 0.37$	$\begin{array}{l} 4.29^{e}\pm0.16\\ 5.64^{f}\pm0.10\\ 6.34^{g}\pm0.20\\ 5.36^{f}\pm0.22\end{array}$	$\begin{array}{c} 13.0^{bc}\pm 0.24\\ 11.8^{b}\pm 0.29\\ 12.5^{b}\pm 0.34\\ 11.3^{ab}\pm 0.24\end{array}$	$\begin{array}{c} 1.11^{\rm c} \pm 0.18\\ 1.04^{\rm c} \pm 0.16\\ 1.37^{\rm d} \pm 0.11\\ 0.94^{\rm c} \pm 0.13\end{array}$	$8.18^{f} \pm 0.36$ $6.24^{e} \pm 0.32$ $6.34^{e} \pm 0.19$ $6.66^{e} \pm 0.17$	$\begin{array}{c} 13.6^{bc}\pm0.29 \ 1.52^{d}\pm0.12 \\ 12.3^{b}\pm0.49 \ 1.50^{d}\pm0.12 \\ 13.6^{bc}\pm0.24 \ 1.51^{d}\pm0.09 \\ 13.6^{bc}\pm0.24 \ 1.51^{d}\pm0.09 \\ 13.1^{bc}\pm0.34 \ 1.52^{d}\pm0.10 \end{array}$	$4.62^{d} \pm 0.17$ $4.54^{d} \pm 0.25$ $5.17^{de} \pm 0.13$ $5.52^{de} \pm 0.12$	$\begin{array}{c} 13.2^{bc}\pm0.95 \ 0.59-1\\ 12.2^{b}\pm0.47 \ 0.59-1\\ 12.9^{b}\pm0.44 \ 0.74-1\\ 11.4^{ab}\pm0.53 \ 0.52-1\\ \end{array}$	.44 1.12–2.12 .68 1.09–2.75 .72 1.17–2.49 .72 1.41–2.66
C. gigas	$2.55^{\mathrm{ef}} \pm 0.31$	0.27 $7.66^{\mathrm{gh}}\pm$	$12.0^{b} \pm 0.42$	$2.52^{e} \pm 0.33$	$5.07^{\mathrm{f}}\pm0.88$	$13.0^{bc}\pm0.17$	$1.01^{\circ} \pm 0.17$	$7.34^{ef} \pm 0.50$	$12.3^{b} \pm 0.18 \ 1.61^{d} \pm 0.15$	$5.41^{ m de}\pm0.24$	$11.3^{ab} \pm 0.12 \ 0.60 - 1$	.48 1.39-2.44
F. dentivarians L. aberrans H. erythrostoma M. cephalus	$\begin{array}{c} 3.88^{h}\pm0.14\\ 2.84^{f}\pm0.17\\ 2.84^{f}\pm0.12\\ 1.43^{d}\pm0.15\\ 1.43^{d}\pm0.15\end{array}$	$\begin{array}{c} 0.59\\ 21.2^{b}\pm1.94\\ 4.89^{e}\pm0.14\\ 4.89^{e}\pm0.14\\ 2.71^{d}\pm0.34\end{array}$	$\begin{array}{c} 12.1^{b}\pm 0.70\\ 13.0^{bc}\pm 0.19\\ 13.2^{b}\pm 0.21\\ 14.9^{c}\pm 0.36\end{array}$	$\begin{array}{l} 3.23^{f}\pm0.34\\ 1.68^{d}\pm0.07\\ 2.39^{e}\pm0.35\\ 1.66^{d}\pm0.18\end{array}$	$\begin{array}{l} 14.5^{h}\pm0.80\\ 4.93^{ef}\pm0.11\\ 6.33^{g}\pm0.20\\ 4.15^{e}\pm0.23 \end{array}$	$13.6^{bc} \pm 0.24$ $12.7^{b} \pm 0.46$ $13.3^{bc} \pm 0.38$ $15.0^{c} \pm 0.04$	$\begin{split} 1.54^{de} \pm 0.14 \\ 1.46^{de} \pm 0.10 \\ 1.12^{c} \pm 0.16 \\ 1.45^{de} \pm 0.05 \end{split}$	$\begin{array}{l} 16.8^{h}\pm1.18\\ 4.25^{d}\pm0.14\\ 14.2^{h}\pm1.36\\ 4.37^{de}\pm0.27 \end{array}$	$\begin{array}{l} 13.4^{bc}\pm0.18\ 2.79^{ef}\pm0.32\\ 12.2^{b}\pm0.30\ 1.55^{d}\pm0.11\\ 12.9^{b}\pm0.12\ 1.55^{d}\pm0.06\\ 14.5^{c}\pm0.28\ 1.26^{cd}\pm0.14\end{array}$	$\begin{array}{c} 14.2^{h}\pm1.46\\ 4.47^{d}\pm0.13\\ 6.94^{e}\pm0.15\\ 12.29^{c}\pm0.12\end{array}$	$\begin{array}{l} 13.2^{bc}\pm 0.15 \ 0.84 \\ 12.3^{b}\pm 0.14 \ 0.78 \\ 12.3^{b}\pm 0.14 \ 0.78 \\ 13.6^{bc}\pm 0.24 \ 0.83 \\ 0.83 \\ 0.52 \\ 0.52 \end{array}$	.81 3.24-6.66 .18 0.90-1.50 .99 1.03-1.71 .67 0.44-1.07
juvenile L. vannamei C. arcuatus iuvenile	$\begin{array}{c} 2.83^{\rm f} \pm 0.06 \\ 3.52^{\rm g} \pm 0.22 \end{array}$	$5.61^{f} \pm 0.16$ $4.51^{ef} \pm 0.23$	$13.5^{bc} \pm 0.17$ ( 15.1 <sup>c</sup> ± 0.21	$\begin{array}{c} 3.26^{f}\pm 0.42\\ 3.36^{f}\pm 0.27\end{array}$	$4.72^{e} \pm 0.34$ $3.60^{d} \pm 0.24$	$13.5^{bc} \pm 0.44$ $15.0^{c} \pm 0.16$	$1.41^{de} \pm 0.11$ $2.13^{e} \pm 0.12$	$5.23^{de} \pm 0.60$ $3.73^{d} \pm 0.30$	$13.1^{bc} \pm 0.30 \ 1.68^{d} \pm 0.17 \\ 15.6^{cd} \pm 0.36 \ 1.54^{d} \pm 0.16$	$4.99^{d} \pm 0.21$ $2.76^{c} \pm 0.14$	$13.4^{bc} \pm 0.18 \ 0.70{-}1 \\ 14.8^{c} \pm 0.62 \ 0.79{-}1$	.45 1.03–1.34 .35 0.52–0.92
G. cinereus juvenile H. leuciscus	$2.32^{\text{ef}} \pm 0.19$ $2.47^{\text{ef}} \pm 0.18$	$5.11^{\text{ef}} \pm 0.45$ $5.76^{\text{f}} \pm 0.12$	$15.8^{cd} \pm 0.18$ $16.6^{d} \pm 0.44$	$3.75^{g}\pm0.48$ $4.28^{gh}\pm$	$4.06^{\rm e} \pm 0.19$ $4.96^{\rm ef} \pm 0.25$	$15.7^{cd}\pm0.50$ $17.2^{d}\pm0.53$	$1.75^{de} \pm 0.22$ 2.03 <sup>e</sup> $\pm 0.17$	$4.99^{e} \pm 0.21$ $6.89^{e} \pm 0.35$	$15.9^{cd} \pm 0.48 \ 2.35^{ef} \pm 0.13$ $16.0^{de} \pm 0.32 \ 2.19^{e} \pm 0.15$	$3.67^{cd} \pm 0.11$ $4.35^{d} \pm 0.18$	$15.4^{cd} \pm 0.46 \ 0.70^{-1}$ $15.6^{cd} \pm 0.36 \ 0.71^{-1}$	.22 0.64–0.79 .18 0.82–1.08
juvenile L. argentivensis	$2.51^{ef}\pm0.20$	$5.54^{\rm f}\pm0.28$	$16.7^{d}\pm0.14$	$0.84 \\ 2.94^{ m ef} \pm 0.18$	$4.35^{\text{e}}\pm0.15$	$15.4^{cd}\pm0.20$	$1.71^{de}\pm0.17$	$2.57^{\rm c}\pm 0.11$	$14.9^{c} \pm 0.14 \ 2.40^{cf} \pm 0.10$	$3.87^{cd}\pm0.23$	$16.5^d \pm 0.33 \ 0.72{-}1$	.13 0.43-0.81
Juvenue <i>M. cephalus</i> adult <i>C. arcuatus</i> adult	$2.44^{\rm ef} \pm 0.15$ $5.16^{ m i} \pm 0.49$	$7.38^g \pm 0.31$ $9.17^{hi} \pm 0.58$	$\begin{array}{c} 16.5^{d}\pm0.31\\ 16.0^{d}\pm0.10\end{array}$	$4.02^{g} \pm 0.24$ $4.34^{gh} \pm$	$\begin{array}{c} 6.33^g\pm 0.20\\ 5.78^{fg}\pm 0.27\end{array}$	$\begin{array}{c} 16.1^{de}\pm 0.34 \\ 16.1^{de}\pm 0.49 \end{array}$	$\begin{array}{c} 1.93^{de}\pm 0.14\\ 2.41^{f}\pm 0.17\end{array}$	$9.33^{f} \pm 0.72$ $4.90^{d} \pm 0.42$	$ \begin{array}{l} 16.2^{d}\pm 0.54  1.40^{d}\pm 0.14 \\ 16.1^{d}\pm 0.27  2.24^{e}\pm 0.10 \end{array} $	$6.27^{e} \pm 0.12$ $4.81^{d} \pm 0.47$	$\begin{array}{l} 17.0^{d}\pm0.23  2.33{-}1.\\ 16.4^{d}\pm0.59  1.09{-}1 \end{array}$	4.2 15.7–36.4 .53 0.76–1.37
G. cinereus adult H. leuciscus adult L. argentivensis	$\begin{array}{c} 4.13^{\rm gh}\pm\\ 0.17\\ 4.36^{\rm g}\pm0.18\\ 4.27^{\rm gh}\pm\\ 0.21\end{array}$	$\begin{array}{c} 8.31^{gh}\pm\\ 0.26\\ 9.60^{h}\pm0.54\\ 11.2^{h}\pm0.59\end{array}$	$17.3^{d} \pm 0.18$ $17.7^{de} \pm 1.35$ $17.1^{d} \pm 0.14$	$\begin{array}{c} 0.41\\ 4.76^{h}\pm0.18\\ 5.05^{hi}\pm0.20\\ 5.18^{hi}\pm0.15\end{array}$	$7.34^{g} \pm 0.21$ $7.78^{g} \pm 0.22$ $7.86^{g} \pm 0.24$	$\begin{array}{l} 17.1^{de}\pm 0.18\\ 18.0^{e}\pm 0.41\\ 16.7^{de}\pm 0.25 \end{array}$	$2.15^{ef} \pm 0.11$ $2.78^{f} \pm 0.13$ $2.81^{f} \pm 0.18$	$6.48^{e} \pm 0.26$ $6.73^{e} \pm 0.21$ $7.48^{ef} \pm 0.18$	$\begin{split} & 17.6^{de}\pm 0.18 \ 3.32^{fg}\pm 0.11 \\ & 18.3^e\pm 0.38 \ 3.27^{fg}\pm 0.14 \\ & 18.1^e\pm 0.16 \ 3.24^{fg}\pm 0.12 \end{split}$	$\begin{array}{l} 8.62^{f}\pm0.36\\ & 9.60^{f}\pm0.10\\ & 12.9^{g}\pm0.40 \end{array}$	$\begin{array}{l} 17.8^{de}\pm 0.40 \ 1.13-1\\ 17.0^{d}\pm 0.25 \ 1.13-1\\ 17.0^{d}\pm 0.33 \ 1.19-1\end{array}$	.49 0.92–1.30 .53 0.92–1.51 .57 1.09–2.10
N. violacea R. longirostris P. ajaja P. brasilianus	$\begin{array}{c} 0.21 \\ 4.92^{i} \pm 0.21 \\ 4.82^{i} \pm 0.12 \\ 4.88^{i} \pm 0.42 \\ 4.38^{i} \pm 0.20 \end{array}$	$\begin{array}{c} 17.9^{i}\pm1.07\\ 28.3^{j}\pm2.13\\ 33.2^{j}\pm1.81\\ 22.5^{i}\pm1.34\end{array}$	$\begin{array}{l} 16.5^{d}\pm0.44\\ 16.7^{d}\pm0.33\\ 16.3^{d}\pm0.19\\ 18.2^{e}\pm0.13\end{array}$	$\begin{array}{l} 4.90^{h}\pm0.25\\ NS\\ 5.55^{i}\pm0.35\\ NS\\ NS\end{array}$	$\begin{array}{c} 21.0^{i}\pm2.18\\ NS\\ 18.9^{i}\pm0.97\\ NS\\ NS\end{array}$	$\begin{array}{c} 16.7^{de}\pm0.23\\ NS\\ 16.4^{de}\pm0.29\\ NS\end{array}$	$3.20^{f} \pm 0.17$ NS $3.09^{f} \pm 0.16$ $6.24^{g} \pm 0.15$	$23.7^{i} \pm 0.73$ NS $20.3^{i} \pm 1.13$ $20.5^{i} \pm 1.13$	$\begin{array}{l} 17.3^{de}\pm 0.18 \ 3.98^{\mu}\pm 0.18 \\ NS \ 3.71^{\mu}\pm 0.11 \\ 16.5^{d}\pm 0.27 \ 3.95^{\mu}\pm 0.12 \\ 18.6^{e}\pm 0.13 \ NS \end{array}$	$\begin{array}{c} 19.8^{i}\pm1.84\\ 14.6^{h}\pm1.20\\ 28.9^{j}\pm1.61\\ NS\end{array}$		.03 3.36-5.04 .88 3.62-5.23 .88 3.91-7.00 .03 3.50-3.96

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	Summer: Au	ıgust 2011		Winter: Feb	ruary 2012		Spring: May	2012		Autumn: Oct	ober 2012		Ranges of	$BMF^3$
	Hg	Se	δ <sup>15</sup> N	Hg	Se	δ <sup>15</sup> N	Hg	Se	$\delta^{15}N$	Hg	Se	$\delta^{15}N$	Hg	Se
F. magnificens TBMF <sup>5</sup>	$5.20^{i} \pm 0.17$	$23.1^{1} \pm 1.88$	$17.6^{de} \pm 0.85$	$5.68^{i} \pm 0.19$	34.8 <sup>j</sup> ± 3.75 −	$18.2^{e} \pm 0.10$	$6.44^{g} \pm 0.30$	19.1 <sup>hi</sup> ± 1.24 -	$19.2^{e} \pm 0.12$	$(4.19^{g} \pm 0.14)$	$20.6^{i} \pm 1.27$	$19.4^{e} \pm 0.97$	1.22 - 3.04 1.38 - 1.62	3.16–6.91 1.45–1.70
<sup>1</sup> <i>SPM</i> suspender <sup>2</sup> Different letters <sup>3</sup> Estimated from <sup>4</sup> Bioavailable fr <sup>6</sup> <sup>5</sup> Estimated from	d particulate me s by column for their principal actions (reactive the slope of an	atter, <i>NS</i> not sa t each metal de food contribut e + pyrite fract ntilogarithm of	umpled enote statisti tions using t ions; Huerte the slopes o	cal significant he stable isote a-Diaz and M of the linear c	t differences ( ope Bayesian orse 1990) fo	p<0.05) SIAR mixin r Hg and Se g (Hg or Se)	g model analy versus TL. No	sis 5 seasonal dif	Ťerences (p>	0.05)				

 Table 2 (continued)

according to tidal regime. Concerning organisms, the slight spatial variation could be related to the effects of marine currents on metal distribution and to the mobility of species through the ecosystem.

## Trophic transfer of Hg and Se and molar Se:Hg ratios

The log-transformed Hg concentrations versus the trophic levels of the ET lagoon food web presented significant (p<0.05) linear correlations (R values from 0.64 to 0.83; Fig. 2), evidencing increases of Hg levels from primary producers to tertiary consumers. Similarly, significant positive (p<0.05) correlations between log-transformed Se concentrations versus the trophic levels of the ET lagoon were also observed (R values from 0.47 to 0.55). Thus, the increase in Se content through the studied food web was evident.

The TBMF is the slope of the correlation between logarithm-contaminant concentrations versus TL. A TBMF>1 indicates the entire biomagnification of a contaminant in the food web (Mackay et al. 2016). The ranges of TBMF that were determined in the ET lagoon varied from 1.38 to 1.62 (no seasonal differences, p>0.05) for Hg and from 1.45 to 1.70 (no seasonal differences, p>0.05) for Se (Table 2), which is indicative of the biomagnification of both elements. Table 2 also shows the BMF for the different preypredator relations in the studied food web, with ranges from 0.52 to 14.2 for Hg and from 0.52 to 36.4 for Se. There are increases in the BMF through the trophic relationships of the ET lagoon relations, with ranges of 0.52-0.67 to 0.84-1.81 for Hg and of 0.43-0.81 to 3.24-6.66 for Se for primary consumers and from 1.15-2.03 to 1.33-3.88 for Hg and of 3.50-3.96 to 3.91-7.00 for Se for tertiary consumers. Values of BMF>1 are also indicative of a positive transfer of contaminants between the different organism groups (Dehn et al. 2006).

The linear correlations between the molar Se:Hg ratios and the TL of the organism groups of the ET lagoon did not present a defined pattern and were only significant (p<0.05) for the summer (August 2011) season ( $R^2$ =0.30; Fig 2). The tendency in three of the four seasons was to decrease according to the TL increments in the ET lagoon food web. The ranges of molar Se:Hg ratios in the entire food web were 0.36–2.91 for summer (August 2011), 0.42–2.37 for winter (February 2012), 0.20–1.69 for spring (May 2012), and 0.35–1.63 for autumn (October 2012).

# Discussion

## Hg and Se sources in El Tobari lagoon

In the present study, the concentrations of Hg and Se found in the sediment were similar to those reported in mangrove or



Fig. 2 The relationship between Hg and Se contents (logarithm transformed), Se to Hg molar ratio, and trophic levels (TL) in the El Tobari lagoon food web. Lines represent the estimated linear regression between both variables for each sampling season



Fig. 2 continued.

coastal ecosystems around the world; however, they were significantly lower than those reported in urbanized harbors (Haris et al. 2017). These authors also reported levels of Hg lower than those found in this study in an anthropogenically impacted mangrove ecosystem.

The measurements of bioavailable Hg concentrations in sediments were compared to those previously reported in the ET lagoon (0.22–1.38  $\mu$ g g<sup>-1</sup>; Jara-Marini et al. 2013a). These were similar to the recent samples collected during the summer and winter seasons but were significantly lower than those collected during the spring and autumn seasons. Jara-Marini et al. (2013b) also reported enrichment factors for Hg ranging from 2.8 to 12.9, which is classified as strong enrichment and is indicative of the anthropogenic sources of this element (Birch 2017). The seasonal variations of Hg and Se concentrations in the sediment and SPM in the ET lagoon evidenced seasonal and yearly variations probably associated to the anthropogenic inputs of both elements. Specifically, values of bioavailable Hg in the sediments that were collected in the spring and autumn seasons (0.27 and 0.29  $\mu g g^{-1}$ , respectively) did not exceeded the threshold effect level (TEL= 130  $\mu$ g g<sup>-1</sup>) (Birch 2018); this would not represent adverse biological effects. However, the transference of metals and metalloids from sediments may occur even if the levels are considered low (Birch 2017, 2018).

As previously reported, the ET lagoon is an ecosystem that is affected by agriculture activities from the Yaqui Valley, which is one of the most extensive agricultural areas in Mexico and discharges 10 agricultural drains into this ecosystem (Jara-Marini et al. 2013a, b, 2020; Martínez-Durazo et al. 2019). Mercurial fungicides (e.g., methyl-mercury, methoxyethyl mercury, and phenylmercury acetate) were used as a seed preservative between 1950 and 1970 (Gupta and Aggarwal 2007). Although these fungicides were prohibited in the early 1970s, their extensive application resulted in Hg accumulation in soils and posterior mobilization to adjacent coastal ecosystems (Birch 2018) and thus to the potential for transference to aquatic food webs (Calle et al. 2018; Fang et al. 2019).

Diverse studies have reported the application of Se mineral fertilizers with a load from 5 to 20 g ha<sup>-1</sup> to increase the element bioavailability levels in soils (Curtin et al. 2008; Jiang et al. 2015; Garousi 2017). An annual discharge between 1150 and 4600 kg Se year<sup>-1</sup> from the 230,000 ha of the Yaqui Valley during the 1950-1970 period has been estimated. The irrigation in each crop cycle provokes the Se lixiviation and may be an important source of the element to adjacent ecosystems. Santos et al. (2015) reported that agriculturally drained water contains 140-1400 mg Se  $L^{-1}$ , which could explain the accumulation and transfer of this element in the organisms of the ET lagoon by its continued exportation through the drains. However, the presence of a natural source of Se cannot be ignored. There is no precise quantitative information regarding the flows in Yaqui Valley drains for the estimation of this Se load into the ET lagoon.

The ET lagoon possesses an extensive surface of shrimp aquaculture in the north zone (1190 ha) with effluents that are another potential Hg source to this ecosystem. The inputs of this element from intensive shrimp farming may be significant through the use of artificial feed, fertilizers, and other chemical additives, including acidity correctors and algaecides (Lyle-Fritch et al. 2006). Lacerda et al. (2011) estimated Hg emission factors of  $\sim 83.5 \text{ mg ha}^{-1}$  in each culture cycle of a coastal ecosystem in Brazil. Assuming that the same conditions operate in the ET lagoon, the shrimp area ponds represent a load of 99.4 g Hg yr<sup>-1</sup>. Conversely, assuming the calculated Hg emission factors for urban wastewaters (200 mg  $ha^{-1} yr^{-1}$ ) that were reported by Lacerda et al. (2011), the inputs of Hg from Obregon city and other towns located in the basin of the ecosystem (surface of 331,200 ha) represent a load of 66.2 kg Hg yr<sup>-1</sup>. However, a correct and more realistic estimation is required considering that wastewaters are partially treated, and part of these effluents are not entirely discharged to the ET lagoon.

There is no available information for the calculation of the shrimp farm load of Se from shrimp aquaculture. However, by assuming a Se content of 1.22  $\mu$ g Se g<sup>-1</sup> in the shrimp feed (Wang et al., 2019), a feed conversion ratio (wet food added/ wet weight of shrimp harvested) of 1.4 (Lyle-Fritch et al. 2006) and considering that the Se content in the shrimp farming is 2.23  $\mu$ g g<sup>-1</sup> (Silva et al. 2016), it is possible to deduce that the 1190 ha of shrimp ponds in the adjacent shrimp farms receives a Se contribution through the food of 2433 g per cycle. The Se recovery through the harvested shrimp is estimated to be 637 g; therefore, we estimated that 1.8 kg of Se is released per cycle, of which a proportion is discharged to the ET lagoon via shrimp effluents.

## Trophic transfer of Hg and Se through the food web

Many studies have reported the biomagnification of Hg and/or Se in food webs in freshwater (Arcagni et al. 2013; Økelsrud et al. 2016), coastal (Kehrig et al. 2013; Seixas et al. 2014), pelagic (Escobar-Sánchez et al. 2011; Ordiano-Flores et al. 2012; Lavoie et al. 2013), and artic (Dehn et al. 2006; Lavoie et al. 2013) ecosystems. The elevated biomagnification potentials of both elements are caused by their high assimilation efficiencies and ingestion rates and low efflux rates (Reinfelder et al. 1998; Wang 2002), and these variables depend on the biotic factors (e.g., growth rate, species diversity, and length of the food web) and physicochemical conditions of the ecosystems (e.g., pH, salinity, dissolved organic matter, redox potential) (Wang 2002). Lavoie et al. (2013) found that polar and temperate regions have higher Hg trophic magnification slopes than tropical ecosystems, which is explained by the low growth and excretion rates at cold temperatures and the low species diversity (less complex food chains). Since diet dominates Se uptake, trophic relations might be the limiting factor for the biomagnification of this element. Stewart et al. (2004) reported that clams and crustaceans cause the efficient transfer of Se to upper trophic levels by their low excretion rates. Similarly, the predominant species of phytoplankton may limit the positive transfer of Se to predators given the dinoflagellate group has high assimilation efficiencies and ingestion rates and/or low efflux rates, as compared to chlorophytes. Nonetheless, the precise mechanisms are still uncertain (Stewart et al. 2010). Our data indicates that bivalves could be significant sources of Se to higher trophic positions. In addition, the formation of an equimolar Hg–Se complex binding to selenoprotein P may lead to a positive correlation between Hg and Se (Sasakura and Suzuki 1998), which would explain the Hg and Se biomagnification found in the ET lagoon.

The antagonistic effect of Se against Hg toxicity is widely recognized (Ralston et al. 2019). The Hg-Se antagonism is based on the formation of certain Hg-Se compounds that limit the mobility, bioavailability, and affinity to the target sites. Most of the organism groups presented molar Se:Hg ratios >1 during the four seasons, which is indicative of an excess of Se in relation to their essentiality, but this is not a limiting factor for the detoxification of Hg (Sørmo et al. 2011). The effects of Se against Hg toxicity include the redistribution of Hg in the tissues, the competition for binding sites, and the formation of un-reactive Hg-Se complexes (Stewart et al. 2010). It was proposed that Se acts against  $Hg^{2+}$  and methyl-Hg, reducing their toxicity and possible transfer through the food web by the formation of insoluble mercury selenides (Se-Hg complexes) that interrupt the activity of Se-dependent functions (e.g., selenoenzymes) (Sørmo et al. 2011), since they may be less solubilized from the prey and could be absorbed by the predator digestive tract. Wang et al. (2004) reported that the dietary assimilation of Hg<sup>2+</sup> and methyl-Hg by the diatom Thalassiosira pseudonana and the green mussel Perna viridis was not significantly affected by the different Se<sup>6+</sup> and Se<sup>4+</sup> concentrations. In contrast, seleno-Lmethionine significantly inhibited the uptake of methyl-Hg and enhanced the uptake of  $Hg^{2+}$  by the diatoms and the mussels at a relatively low concentration, possibly by the complexation of Hg<sup>2+</sup>.

## Conclusions

The biomagnification of Hg and Se studied during four seasons in a subtropical coastal lagoon (ET) of the eastern central Gulf of California included the characterization of the trophic position of food web species through stable nitrogen isotopic ratios. For this, a total of thirty-three species belonging to five organism groups were examined. The BMF ranges for Hg and Se in the organisms were 0.52–1.81 and 0.90–6.66 for primary consumers (zooplankton); 0.70–1.45 and 0.52–1.34 for secondary consumers (crustaceans); 1.09-14.2 and 0.76-36.4 for tertiary consumers (mainly fishes); and 1.15-3.88 and 3.16-7.00 for quaternary consumers (birds), respectively. The estimated TBMF for the trophic web was 1.38-1.62 and 1.45-1.70 for Hg and Se, respectively. These data confirm the biomagnification of Hg and Se (BMF and TBMF>1) during the four seasons examined. Additionally, it evidences that primary consumers are the main sources of Hg and Se to higher trophic levels. Conversely, although the linear correlations between the molar Se:Hg ratios and the TL of the organism groups of the ET lagoon did not present a defined pattern, the Se:Hg ratios among organism groups were >1, which suggests an excess of Se, and that it is not a limiting factor for the detoxification of Hg.

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Author contribution AMG sampled, processed, and analyzed sediment and biota samples. JGH designed the sampling and sampled, processed, and analyzed avifauna samples. MFSJ interpreted stable isotopes data. FPO interpreted Hg and Se data. MEJM designed the sampling, processed and analyzed sediment and biota samples, performed the dataset, and was a major contributor in writing the manuscript. All authors read and approved the final manuscript.

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**Data availability** The datasets used and/or analyzed during the current study are available from the corresponding author on reasonable request.

#### **Declarations**

Ethics approval and consent to participate All applicable international, national, and/or institutional guidelines for the care and use of animals were followed. This article does not contain any studies with human participants performed by any of the authors. Consent to participate is "Not applicable."

Consent to participate or publication Not applicable.

**Competing interest** The authors declare no competing interests.

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