RESEARCH ARTICLE



# Source apportionment of  $PM<sub>2.5</sub>$  during different haze episodes by PMF and random forest method based on hourly measured atmospheric pollutant

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# Abstract

Hourly measured PM2.5-bound species, gases, and meteorological data were analyzed by the PMF receptor model to quantify source contributions, and by the random forest to estimate decisive factors of variations of  $PM_{2.5}$ , sulfur oxidation ratio (SOR), and nitrogen oxidation ratio (NOR) during different haze episodes. PM<sub>2.5</sub> variation was influenced by CO (17%), SO<sub>2</sub> (19%),  $NH<sub>3</sub>$  (12%), O<sub>3</sub> (10%), air pressure (P, 9.9%), and temperature (T, 10%) during the whole period. SOR was determined by SO<sub>2</sub> (15%), temperature (T, 9.8%), relative humidity (RHU, 15%), and pondus hydrogenii (pH, 35%), and NOR was influenced by  $NO_{x}$  (19%),  $O_{3}$  (14%), NH<sub>3</sub> (13%), and RHU (15%). Three types of pollution episodes were captured. Process I was characterized by high CO (contributing  $40\%$  of PM<sub>2.5</sub> concentration variation estimated by the random forest) due to coal combustion for heating during winter in northern China. According to the PMF, coal combustion (32%) and secondary sources (38%) were both the most important contributors in the first stage, and then, when the RHU increased to above 80%, the highest contribution was from secondary sources (40%). Process II was during the Spring Festival and was characterized by 8.8  $\mu$ g m<sup>−3</sup> firework contribution. High  $SO_2$  during this process, especially on the CNY's Eve, was observed due to the firework displays, and  $SO_2$ gave a high contribution (24%) to PM<sub>2.5</sub> variation. Process III showed high ions and high RHU in summer with sulfate and nitrate contributing 44% and 22%, respectively. Furthermore, meteorological parameters and NH<sub>3</sub> play a key role on SOR and NOR.

Keywords Source appointment · Random forest · High time-resolved species · Air · Haze

# Introduction

Particulate matter (PM) and gaseous pollutants have led to major environmental problems in the last few decades in southern and eastern Asia (Huang et al. [2014;](#page-9-0) Zhang et al. [2017](#page-11-0)). In the recent year, China has experienced extremely severe and persistent haze pollution, especially in the North China Plain (Wang et al. [2016;](#page-10-0) Cheng et al. [2016;](#page-9-0) Zou et al. [2017\)](#page-11-0). Severe pollution

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is accompanied by extremely poor air quality and poor visibility, which threatens human health (Jerrett et al. [2017;](#page-9-0) Li et al. [2018;](#page-10-0) Almetwally et al. [2020](#page-9-0)). Considerable research has reported that the mechanisms are very complex and differ for different pollution episodes (Wu et al. [2019\)](#page-11-0). However, pollution mechanisms and influence factors during different pollution episodes are still not well understood (Cheng et al. [2016](#page-9-0); Xie et al. [2019\)](#page-11-0). In order to investigate pollution mechanisms, severe pollution episodes were captured and studied by environmentalist.

The origins of PM, which can be emitted from primary sources and formed through chemical reactions from precursors, may rapidly change during haze episode (Notario et al. [2013;](#page-10-0) Zheng et al. [2014;](#page-11-0) Wu et al. [2020](#page-11-0)). So as to perform source apportionment of PM, receptor models based on high timeresolved species data have been used to investigate timely variations of source contributions (Eatough et al. [2008](#page-9-0); Pancras et al. [2013;](#page-10-0) Gao et al. [2016](#page-9-0); Han et al. [2016;](#page-9-0)), which is fundamental to understanding the influence factors of heavy pollution episodes. Meteorologists have confirmed that gaseous pollutants and meteorological parameters play important roles in haze (Kuwata

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et al. [2017;](#page-9-0) Wang et al. [2019a](#page-10-0)). Gaseous pollutants can act as precursors of secondary aerosol (Schelden et al. [2017](#page-10-0); Yao et al. [2018\)](#page-11-0). Adverse meteorological conditions can lead to accumulation of pollutants and favor formation of secondary aerosol (Wang et al. [2019b](#page-11-0)). Thus, gaseous pollution and meteorological parameters are important for  $PM<sub>2.5</sub>$  heavy pollution and formation of secondary aerosol, but their relationships are very complex (Trivedi et al. [2014;](#page-10-0) Wang et al. [2014a](#page-10-0)). Nevertheless, there is nonlinear relationship between the decisive influencing factors (e.g., gaseous pollution and meteorological parameters) and PM2.5 concentration variation. At present, machine learning algorithm would be used to better identify the decisive influencing factors on  $PM_{2.5}$  levels and secondary formation. The algorithm provides novel machine learning–based framework for data analysis. In this study, random forest algorithm, a classification and regression tool (Svetnik et al. [2003](#page-10-0)), is introduced to study the formation mechanism and important influencing factors of air pollutants based on gaseous pollutants and meteorological parameters. In brief, this study has one main contribution that the research attempts to investigate the relationships between PM<sub>2.5</sub> concentrations and the gaseous pollution and between PM<sub>2.5</sub> and meteorological factors by machine learning method from the macroscopic perspective.

Comprehensive observations of chemical species, gases, PM mass concentrations, and meteorological parameters at a 1-h time resolution were conducted to explore influence factors of haze episodes. Different types of pollution episodes were observed during the sampling period. The levels of  $PM_2$ , s-bound species (such as  $NO_3^-$ ,  $Cl^-$ ,  $SO_4^{2-}$ ,  $NH_4^+$ ,  $Ca^{2+}$ ,  $Na^+$ ,  $Mg^{2+}$ ,  $K^+$ , OC, EC), pollutants (such as  $PM_1$ ,  $PM_{2.5}$ ,  $PM_{10}$ , TSP,  $SO_2$ ,  $O_3$ ,  $NH<sub>3</sub>$ , NO<sub>x</sub>, CO), and meteorological parameters (such as wind speed, relative humidity, temperature) were investigated. Positive Matrix Factorization (PMF) would be applied for source apportionment of  $PM_{2.5}$  based on chemical species. Then, random forest method was applied to estimate the impacts of gaseous pollutants and meteorological parameters on  $PM_{2.5}$  concentrations, sulfur oxidation ratio (SOR), and nitrogen oxidation ratio (NOR). Our work used the machine learning algorithm to reveal the characteristics and the reasons for the formation of typical heavy pollution episodes. The severe pollution episodes provide researchers an opportunity to study different mechanisms of pollution. The methods and influence factors reported in this paper are applicable to other emerging economies or developing countries and have significance for efforts to design effective management strategies.

# Methodology

## Sampling site and study period

The high time-resolved measurements of chemical species, gases, and PM mass concentrations were conducted on the rooftop of a five-story building (the Tianiin Environmental Protection Bureau) in downtown Tianjin, which is a municipality directly under the Central Government of China. Tianjin, near the capital of China (Beijing), is in the Bohai Economic Circle. Tianjin is a megacity under rapid industrialization and urbanization in the last few decades and currently has a population of over 14 million and an area of 11,947 km<sup>2</sup>. The sampling site was located in a mixed residential and commercial area surrounded by few direct industrial sources, high vehicular emissions, and construction areas. The observations were conducted in February, March, June, July, August, and September of 2015 and produced data for 4,344 h. The details of sampling have been reported in our previous publication (Tian et al. [2018a](#page-10-0)).

## Sampling methods and instrumentation

To obtain high time-resolved speciate data for  $PM_{2.5}$ , instruments were used to detect ambient  $PM_{2.5}$ . Hourly concentrations of elemental carbon (EC) and organic carbon (OC) were detected through a semi-continuous EC/OC carbon aerosol analyzer (Model-4, Sunset Laboratory Inc., USA) (Dall'Osto et al. [2014](#page-9-0)) using the basic thermal/optical transmittance measurement protocol of the National Institute for Occupational Safety and Health (NIOSH). Two temperature stages were used to determine OC and EC: an aliquot of sample filter  $(2.1 \text{ cm}^2)$  was heated stepwise to 820 °C in a furnace in a non-oxidizing atmosphere (100% He); the oxidizing oven was then cooled to 550 °C, and the filter was again gradually heated to 870 °C in an oxidizing atmosphere (98% He, 2%)  $O_2$ ). Evolved carbon was oxidized to  $CO_2$  and detected by a non-dispersive infrared detector (NDIR) during each temperature step. The split point was quantified as the carbon evolved after the introduction of oxygen but before the point where transmittance became equal to its initial value. Calibration was performed by introducing a known amount of methane into the oven and measuring its constant response. The carbon that evolved before the split point was OC, whereas EC was measured as the carbon evolved after this point but prior to the methane calibration peak (Tiwari et al. [2013\)](#page-10-0).

Hourly concentrations of ionic species in  $PM_{2.5}$ , including  $NO_3^-$ , Cl<sup>−</sup>, F<sup>−</sup>, SO<sub>4</sub><sup>2−</sup>, NH<sub>4</sub><sup>+</sup>, Ca<sup>2+</sup>, Na<sup>+</sup>, Mg<sup>2+</sup>, and K<sup>+</sup>, were measured using a URG 9000D ambient ion monitor (AIM) (Chapel Hill, NC). The AIM separated and analyzed each anion and cation through a particle collection system and ion chromatographs (ICs). The  $PM_{2.5}$  samples were collected at a flow rate of 3 L min<sup>-1</sup> by using a sharp cut cyclone (Manigrasso et al. [2010](#page-10-0)). Then, a liquid diffusion parallel-plate denuder was used to separate the gases from the aerosol samples. The watersoluble compositions of aerosol and gaseous pollutants were collected through four syringes installed into pre-concentrators and then injected into the ICs. Anion detection was conducted in a gradient elution program using a KOH solution at a flow rate of 1.0 mL min<sup>-1</sup>, and the cation analyzer was run with methane sulfonic acid at a flow rate of  $0.5$  mL min<sup>-1</sup>.

Gaseous pollutants included  $SO_2$ ,  $O_3$ , NO, NO<sub>2</sub>, and CO, and PM mass concentrations were of four sizes:  $PM_1$ ,  $PM_2$ ,  $PM_{10}$ , and TSP. Mass concentrations of PMs were continuously measured by a tapered element oscillating microbalance (TEOM) mass sensor. The  $SO<sub>2</sub>$  in the atmosphere was determined by pulsed fluorescence technology. The  $O_3$  was analyzed by a dual cell photometer, a concept adopted by the US NIST as a national standard. Atmospheric NOx was measured by chemiluminescence technology; the analyzer had isolated outputs for  $NO$  and  $NO<sub>2</sub>$  that could be individually calibrated. The CO was quantified based on the absorption of infrared radiation at a 4.6-μm wavelength.

For quality assurance and quality control (QA/QC), the particle stream entered an aerosol super-saturation chamber to increase particle growth to obtain higher efficiencies. After collection through four syringes, the aerosols and the gaseous pollutants were injected into the ICs within 1 h. Anion/cation calibration solutions were used for the calibration of the ICs on the AIM for at least 1 month. The minimum detection limits (MDLs) were as follows: 0.2  $\mu$ g m<sup>-3</sup> (Cl<sup>-</sup>), 0.2  $\mu$ g m<sup>-3</sup> (F<sup>−</sup>), 0.2  $\mu$ g m<sup>-3</sup> (NO<sub>3</sub><sup>−</sup>),  $0.2 \mu g$  m<sup>-3</sup> (NO<sub>2</sub><sup>-</sup>), 0.3 μg m<sup>-3</sup> (SO<sub>4</sub><sup>2</sup><sup>-</sup>), 1.8 μg m<sup>-3</sup> (NH<sub>4</sub><sup>+</sup>), 2.3 μg m<sup>-3</sup> (Ca<sup>2+</sup>), 0.8 μg m<sup>-3</sup> (Mg<sup>2+</sup>), 0.5 μg m<sup>-3</sup> (K<sup>+</sup>), and  $0.6 \mu g$  m<sup>-3</sup> (Na<sup>+</sup>). The EC/OC carbon aerosol analyzer was calibrated each month through a blank punch of a pre-heated quartz fiber filter and standard sucrose solutions (3.2 μg C  $\mu$ <sup>1</sup>). The quartz fiber filter was changed each week during the analysis process. The MDLs were 0.45 μg cm<sup>-2</sup> and 0.06 μg  $\text{cm}^{-2}$  for OC and EC, respectively. The MDLs for PM, NO, NO<sub>2</sub>, SO<sub>2</sub>, CO, and O<sub>3</sub> were as high as 0.1  $\mu$ g m<sup>-3</sup>, 0.40 ppb, 0.40 ppb, 0.5 ppb, 0.04 ppm, and 0.50 ppb, respectively. The flow calibration, gas tightness test, blank filter test, and standard sample calibration were all conducted for QA/QC.

Meteorological parameters, including temperature (T), relative humidity (RHU), air pressure (P), wind speed (WS), and wind direction (DD), used to evaluate the impact of meteorological conditions were acquired from an online database at <http://rp5.by>. The meteorological parameters were available for 2:00 AM, 5:00 AM, 8:00 AM, 11:00 AM, 2:00 PM, 5: 00 PM, and 8:00 PM each day. The meteorological parameters were acquired at the Beimalu Weather Station in downtown Tianjin. It is about 7 km far away from the Tianjin Environmental Protection Bureau Building.

Except for chemical species, gases, and PM mass concentrations, SOR and NOR were also used as follows:

$$
SOR = SO42 / (SO42 + SO2)
$$

$$
NOR = NO32 / (NO32 + NO3)
$$

where  $SO_4^2$  is the S mass in  $SO_4^2$ ,  $SO_2$  is the S mass in  $SO_2$ ,  $NO_3^-$  is the N mass in  $NO_3^-$ , and  $NO_x$  is the N mass in NOx. TS and TN were used to indicate the total emissions of S and N (Chen and Xie [2014\)](#page-9-0). SOR and NOR were used to indicate the reaction of sulfur and nitrogen (Chen and Xie [2014;](#page-9-0) Wang et al. [2018](#page-10-0)).

In this research, the aerosol pH was calculated by utilizing the thermodynamic model ISORROPIA-II (Fountoukis and Nenes [2007](#page-9-0)), which could predict the physical state and composition of atmospheric inorganic aerosols. The ISORROPIA-II model can solve forward problems in which T, relative humidity, and the concentrations of gas + aerosols were known (e.g.,  $NH_3^+NH_4^+$ ), and reverse problems in which T, relative humidity, and the concentrations of aerosol (but not gas) species were known. The pH calculation utilized measurements of NH<sub>3</sub>, NH<sub>4</sub><sup>+</sup>, Na<sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, SO<sub>4</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, and Cl- for Tianjin from February to September of 2015. In this study, the ISORROPIA-II model was run in the forward mode and assumed that the aerosol solutions were metastable, as a high degree of accuracy determined on the basis of measurements of semivolatile partitioning of certain species (e.g., NH<sub>3</sub>/NH<sub>4</sub><sup>+</sup>) (Guo et al. [2015;](#page-9-0) Song et al. [2018](#page-10-0))

# Source apportionment

EPA PMF 5.0 model (Paatero and Tapper [1994;](#page-10-0) Paatero [1997;](#page-10-0) EPA  $2014$ ) was applied for  $PM<sub>2.5</sub>$  source apportionment based on hourly measured  $PM_{2.5}$ -bound chemical species. The PMF is used to solve the mass balance between observed species concentrations X, estimated source profiles F, and estimated contributions G (Ogulei et al. [2006;](#page-10-0) Tian et al. [2014](#page-10-0)):

$$
X_{ij} = \sum_{h=1}^{p} g_{ih} f_{hj} + e_{ij}
$$

where  $x_{ij}$  is the measured concentration of the *j*th species in the *i*th sample,  $f_{hi}$  (g g<sup>-1</sup>) is the estimated species profile of the hth source,  $g_{ih}$  is the estimated contribution by the hth factor to each sample,  $e_{ii}$  is the residuals, and p is the number of factors. Moreover, ME-2 can incorporate prior information such as chemical properties to apportion sources through auxiliary equations (Tian et al. [2018b](#page-10-0)).

In this work, the random forest algorithm, a non-linear, data-driven model, is employed to study the influence factor of  $PM<sub>2.5</sub>$ , NOR, and SOR. Random forest performs treeensemble of Classification and Regression Trees (CART). Regression tree is trained to perform differentiable mapping from the m-dimensional input x to its leaf index (Chen and Guestrin [2016\)](#page-9-0):

$$
f(x) = \omega_q(x) \left( q : \mathbb{R}^m \to T, \omega \in \mathbb{R}^T \right)
$$

Here, f is parameterized by the learned tree structure q (of T leaves) and leaf weights ω. Independent regression trees f can be trained by corresponding tree structure q and leaf weights  $\omega$ .

# Results and discussion

# Concentrations

It was calculated that the average concentrations of  $PM_1$ , PM<sub>2.5</sub>, PM<sub>10</sub>, and TSP were 57, 96, 104, and 152 µg m<sup>-3</sup>, respectively, from February 1, 2015, to September 30, 2015. The  $PM_1/PM_{2.5}$ ,  $PM_{2.5}/PM_{10}$ , and  $PM_{10}/TSP$  ratios were 0.59, 0.92, and 0.70, respectively. These ratios usually varied with different pollution processes and can indicate different pollution mechanisms. As reported by Alastuey et al. [\(2004\)](#page-9-0), the fine particles can be caused by a low proportion of secondary and primary combustion emitted species, and the high coarse part can be caused by the high load of other contributions such as natural and anthropogenic mineral dust.

The important components of  $PM_{2.5}$  include  $NO_3^-$  (13%),  $SO_4^2$ <sup>-</sup> (11%), NH<sub>4</sub><sup>+</sup> (14%), OC (9.5%), EC (2.7%), and Cl<sup>-</sup> (2.5%). And the hourly data were collected from February 1, 2015, to September 30, 2015. The annual average concentration of Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, NH<sub>4</sub><sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>, K<sup>+</sup>, Na<sup>+</sup>, OC, EC, CO, SO<sub>2</sub>, NO<sub>x</sub>, O<sub>3</sub>, and NH<sub>3</sub> were 2.5  $\mu$ g m<sup>-3</sup>, 11  $\mu$ g  $m<sup>-3</sup>$ , 0.40 μg m<sup>-3</sup>, 9.7 μg m<sup>-3</sup>, 13 μg m<sup>-3</sup>, 0.18 μg m<sup>-3</sup>, 0.07 μg m-3, 0.87 μg m-3, 0.77 μg m-3, 9.4 μg m-3, 2.7 μg m-3, 1.3 μg  $m^{-3}$ , 28 μg m<sup>-3</sup>, 56 μg m<sup>-3</sup>, 103 μg m<sup>-3</sup>, and 29 μg m<sup>-3</sup>, respectively. Moreover, RHU, pH, and WD were recorded and their annual mean were 58%, 4.2, and 2.7 m/s, respectively.

# Influence factors on PM2.5, SOR, and NOR

To comprehensively analyze the role of meteorological factors and gaseous pollutant for affecting local  $PM_{2.5}$  concentration, NOR and SOR, we employed random forest model. A number of factors were selected as follows: T, RHU, P, WS, DD, pH, CO,  $SO_2$ ,  $NO_x$ ,  $O_3$ , and  $NH_3$ . By analyzing variables through multiple nonlinear regression, the model quantitatively calculated the relative importance of different factors about  $PM_{2.5}$ , NOR, and SOR. As shown in Table 1.

#### Influence factors on  $PM<sub>2.5</sub>$

Based on the random forest, the results displayed that CO had the most important influence on  $PM<sub>2.5</sub>$  concentration and contributed 17% to  $PM_{2.5}$  concentration variations.  $SO_2$  and  $NO_x$ (totally  $24\%$ ) also strongly linked with  $PM<sub>2.5</sub>$  concentration. It has been reported that in the condition of stable meteorological and enough residence time,  $SO_2$  and  $NO_x$  can be transformed into sulfate and nitrate. P (9.9%) made the great contribution to PM<sub>2.5</sub> concentration among meteorological parameters. It may happen that when there was high pressure, down draft impeded going up of  $PM_{2.5}$  and led to accumulation of particles. T (10%) was also an important factor for  $PM_{2.5}$  concentration variation. Luo et al. [\(2017\)](#page-10-0) noted that air convection relied on temperature, high temperature could lead to dilution of and  $PM_{2.5}$ , and it was

**Table 1** Different influence factors on  $PM_{2.5}$ , SOR and NOR during the whole period

Influence factors	$PM_{2.5}$	SOR.	<b>NOR</b>
CO.	0.17	0.030	0.080
SO <sub>2</sub>	0.19	0.15	0.070
NOx	0.050	0.020	0.19
O <sub>3</sub>	0.10	0.090	0.14
NH <sub>3</sub>	0.12	0.050	0.13
Temperature (T)	0.10	0.098	0.050
air pressure $(P)$	0.099	0.030	0.080
Relative humidity (RHU)	0.067	0.15	0.15
Wind direction (DD)	0.047	0.020	0.040
Pondus hydrogenii (pH)	0.048	0.35	0.060
Wind speed (WS)	0.011	0.010	0.020

on the contrary when temperature reduction, at the same time, high temperature could elevate formation rates of secondary aerosols. RHU (6.7%) and pH (4.8%) could accelerate formation of secondary aerosol and increased the chance of collision and adsorption between particles, which resulted in decreasing  $PM<sub>2.5</sub>$ concentration. DD (4.7%) and WS (1.1%) can also influence on PM2.5 concentration variations. Wind speed could favor particles spread and diffusion of  $PM_{2.5}$  and low  $PM_{2.5}$  concentration (Wang et al. [2019b;](#page-11-0) Karimian et al. [2019\)](#page-9-0). And different speed directions had a significant effect on the spatial distribution of PM2.5 and could change transport of atmosphere pollutants.

## Influence factors on SOR and NOR

The importance analysis of gaseous pollutants  $(CO, SO<sub>2</sub>)$  $NO<sub>x</sub>, O<sub>3</sub>, NH<sub>3</sub>$  and meteorological parameters (T, RHU, P, DD, WS, and pH) on SOR and NOR were also constructed based on random forest method. It was found that pH (35%), SO<sub>2</sub> (15%), RHU (15%), T (9.8%), and O<sub>3</sub> (9%) played important role on SOR. As reported in this work and related work, SOR in the atmosphere was closely related to pH (Fuzzi [1978](#page-9-0); Brimblecombe and Spedding [1972\)](#page-9-0). Sakamoto et al.  $(2004)$  $(2004)$  found that  $O_3$  had noticeable effect on the amount of  $SO<sub>2</sub>$  oxidation. And RHU was significantly positively correlated with SOR and NOR (Yao et al. [2020\)](#page-11-0). In addition, temperature could promote the rate of sulfur oxidation. Meanwhile,  $NO_x$  (19%),  $O_3$  (14%),  $NH_3$  (13%), and RHU (15%) had a greater influence on NOR. Seinfeld and Pandis [\(1998\)](#page-10-0) have found that the formation of nitrogen oxidation was dominated by the following reaction:

 $NO_2 + OH + M \rightarrow HNO_3 + M$  $NO_2 + O_3 \rightarrow NO_3 + O_2$  $NO<sub>2</sub> + O<sub>3</sub> + M \rightarrow N<sub>2</sub>O<sub>5</sub> + M$  $N_2O_5 + H_2O \rightarrow 2HNO_3$  $NH_3 + H_2O \rightarrow NH_4^+ + OH$ 

# <span id="page-4-0"></span>Characteristics of typical pollution episodes

During the sampling period, three typical pollution episodes were captured and selected for further analysis. The concentrations of species in  $PM_{2.5}$  during each episode were introduced into the USEPA PMF 5.0 model for source apportionment, and the gaseous pollutants and meteorological parameters were introduced into the random forest model to explore the influence factors. The factor profiles of three episodes estimated USEPA PMF 5.0 is exhibited in Fig. 1. Coal combustion, traffic emission, and resuspended dust were consistently identified for three pollution episodes. Tian et al. [\(2020\)](#page-10-0) has explored how to better conduct PMF during haze episodes, showing that the PMF performance was poor for some episodes through

whole-based mode (using all data of whole sampling period as input). Thus, the episode-based mode (using data of each episode) was used in this work to conduct PMF. In addition, the consistent results of PMF and random forest based on different datasets demonstrate the reliability of two methods. We add the bootstrapping (BS) to estimate the uncertainty of the PMF solution. BS involves resampling the input dataset, fitting PMF model parameters for this resampled dataset, and then using the variations among these resampled or "bootstrapped" fitted profiles to estimate the uncertainty of the initial PMF solution (Norris et al. [2014\)](#page-10-0). The BS results of each episode are listed in Table [2](#page-5-0). Thus, 4, 6, and 5 factors were selected for episodes I, II and III, respectively, because the BS values were all higher than 75%, indicating the BS uncertainties can be interpreted. The coal combustion was

Fig. 1 The factor profiles of three episodes estimated USEPA PMF 5.0. Episode I is characterized by heavy coal combustion; the five factors, coal combustion, traffic emission, resuspended dust, secondary sources, and other, make different contributions to various components in episode I; episode II is characterized by the Spring Festival; the seven factors, coal combustion, traffic emission, resuspended dust, sulfate, nitrate, firework, and other, make different contributions to various components in episode II; episode III is characterized by high inorganic ions and high RH; the six factors, coal combustion, traffic emission, resuspended dust, sulfate, nitrate, and other, make different contributions to various components in episode III



<span id="page-5-0"></span>Table 2 The bootstrapping (BS) results of three typical pollution episodes

Episode I		Episode II		Episode III	
			4 factors 5 factors 6 factors 7 factors 5 factors 6 factors		
			BS 80-90% 50-90% 75-100% 30-100% 100%		$50 - 100\%$

identified by high loadings of Cl<sup>-</sup>, OC, and EC (Shen et al. [2010\)](#page-10-0). The traffic emission was characterized by OC and EC but low Cl<sup>-</sup> (Liu et al. [2017\)](#page-10-0). The resuspended dust was identified by relatively high weights of  $Ca^{2+}$  and  $Mg^{2+}$  (Shen et al. [2011\)](#page-10-0). Factors associated with  $NO_3^-$ ,  $SO_4^2$ , and  $NH_4^+$  were distinguished as secondary sources or nitrate and sulfate (Guan et al. [2019](#page-9-0)). The OC and EC in the secondary sources may be linked with secondary organic carbon and transport of combustion emissions (Guan et al. [2019\)](#page-9-0).

#### Process I

The first interesting episode was found from February 14, 2015, 0:00 AM to February 15, 2015, 4:00 AM;  $PM_{2.5}$  concentrations ranged from 157 to 313  $\mu$ g m<sup>-3</sup>; then, after a short decrease,  $PM_{2.5}$  returned to relatively higher levels (February 15, 2015, 4:00 PM–February 16, 2015, 12:00 PM). As shown in Fig.  $2(b)$ , the abundances of OC and EC were higher than those for other sampling period. In general, OC and EC were markers of combustion (Chow et al. [2004\)](#page-9-0). According to the PMF results in Fig. [1](#page-4-0) (episode I), for  $PM_{2.5}$  concentration, the average contributions of coal combustion, traffic emission, resuspended dust, and secondary sources were 32%, 17%, 5.5%, and 38%, respectively. As demonstrated in Fig. [2\(a\),](#page-6-0) in the first stage, coal combustion (88 µg m<sup>-3</sup>, 40%) and secondary sources (98  $\mu$ g m<sup>-3</sup>, 41%) were both the most important contributors, and in the second stage, the highest contribution was from secondary sources (63 µg m<sup>-3</sup>, 38%). In this stage, coal combustion had emitted high  $SO_2$ , which formed secondary  $SO_4^2$  under the high RH in the second stage. And the secondary sources were primary pollution source.

Based on the result of random forest in Fig. [2\(c\),](#page-6-0) CO (40%) remained at a high contribution for  $PM_{2.5}$  concentration variation. Previous study noted that CO mainly resulted from the coal combustion for heating during the winter in northern China (Du et al. [2016;](#page-9-0) Wang et al. [2010](#page-10-0)). The results indicated that P (20%) and T (16%) also had a great influence on  $PM_{2.5}$ contribution variation in the heavy pollution process. Therefore, PMF and the random forest results were consistent and both of them demonstrated that this heavy pollution was characterized by coal combustion.

In addition, SOR was analyzed by in putting gaseous pollutants and meteorological parameter in random forest model;

the results are shown in Fig.  $2(d)$ ; it was obvious that, among all the factors,  $SO<sub>2</sub>$  (43%) was the most important factor for SOR. In this pollution process, DD (22%) greatly affected the SOR transformation and it was the second largest factor. A suggested explanation is that wind direction determined gaseous pollutant pathway. Definitely, pH (6%) and RHU (7%) also served as pivotal role. When it came to NOR in Fig.  $2(e)$ ,  $NO<sub>x</sub>$  (24%) and  $O<sub>3</sub>$  (17%) occupied primary important factor for NOR formation variation. Moreover, in this process, CO  $(11\%)$ , DD  $(10\%)$ , and pH  $(12\%)$  had more impact on NOR.

#### Process II

The second interesting episode was observed from February 18, 2015, 12:00 AM to February 21, 2019, 11:00 PM, which covered the Spring Festival. During the period of sampling, the concentration of  $PM_{2.5}$  ranged from 31 to 407  $\mu$ g m<sup>-3</sup>. The chemical compositions of  $PM<sub>2.5</sub>$  during this process were characterized by high  $K^+$  (3.9%) and Cl<sup>-</sup> (4.0%), which were much higher than the average of the entire period (1.0% for  $K^+$ and 2.7% for Cl  $\,$ ). Compared with the episode one, NH<sub>4</sub><sup>+</sup> and SO4 2- were much less; however, both of them were slightly higher than the annual average, respectively (as shown in Fig. [3](#page-7-0) (b)). As reported in our previous work and the literature (Tian et al. [2014;](#page-10-0) Sarkar et al. [2010\)](#page-10-0), Cl and  $K^+$  were considered as firework-related species and could be markers for fireworks.  $K^+$  and Cl<sup>-</sup>, commonly in the form of perchlorate or chlorate, are major components in black powder and act as the main oxidizers during burning. The corresponding chemical equations are  $2KClO_3 = 2KCl + 3O_2$  and  $KClO_4 = KCl + 2O_2$ .

Based on the PMF results in Fig. [1](#page-4-0) (episode II), in process II, for  $PM_{2.5}$  concentration, the average contributions of traffic emission, coal combustion, nitrate, resuspended dust, sulfate, firework, and unknown sources were 18%, 24%, 20%, 6.6%, 14%, 8.8%, and 7.5%, respectively. As demonstrated in Fig. [3\(a\),](#page-7-0) in this process, coal combustion (24  $\mu$ g m<sup>-3</sup>) and nitrate (20 μg m<sup>-3</sup>) were important contributors, followed by traffic emission (18 μg m<sup>-3</sup>), sulfate (13 μg m<sup>-3</sup>), firework (8.6 μg  $m^{-3}$ ), and resuspended dust (6.4 µg m<sup>-3</sup>). It was found that firework was an important pollution source. A further discussion that gaseous pollutants and meteorological parameter had impact on  $PM_{2.5}$  concentration variation by the random forest, as visualized in Fig.  $3(c)$ , the result showed that  $SO<sub>2</sub>$  (24%) kept at a high contribution for heavy pollution. Previous study noted that the number of firework was lighted, and it became a major pollution source in the Spring Festival (Wang et al. [2007;](#page-10-0) Li et al. [2017\)](#page-10-0). CO (12%) and NOx (13%) were also considered major contribution for  $PM<sub>2.5</sub>$  contribution, and both were closely to related to coal combustion emissions and vehicle exhaust (Kota et al. [2014\)](#page-9-0). Besides, P (25%) and pH (11%) also had a great influence on the heavy pollution. Therefore, combining PMF and the random results showed

<span id="page-6-0"></span>

Fig. 2 The abundance and concentration of chemical species, PMF result, and random forest result during process I (episode characterized by heavy coal combustion). a the concentration contribution from coal combustion, traffic emission, resuspended dust, and secondary sources change over time in episode I; b the difference between the average mass percent of

PM<sub>2.5</sub> components in episode I and all year; c gases and meteorological parameters have different contributions to  $PM_{2.5}$  variation in episode I; **d** gases and meteorological parameters have different contributions to SOR in episode I; e gases and meteorological parameters have different contributions to NOR in episode I

that this heavy pollution was characterized by firework combustion.

In terms of the random forest results for SOR in Fig.  $3(d)$ , it was obvious that, among all the gaseous pollutants and meteorological parameters, precursor  $SO<sub>2</sub>$  (27%) was the most important factor for SOR; in this pollution process, P (22%) greatly affected the SOR transformation. Definitely, pH (7%) and RHU (13%) also served as pivotal role. When it came to NOR, as visualized in Fig.  $3(e)$ , NO<sub>x</sub> (35%) occupied primary important factor, followed by P (20%), RHU (10%), and  $O_3$  (8%).

#### Process III

The third pollution episode was characterized by high inorganic ions and high RHU level (51–95%) in summer. The PM<sub>2.5</sub> mass concentrations were at relatively higher levels,

ranging from 81 to 291  $\mu$ g m<sup>-3</sup>. The value of SOR and NOR were 0.55 and 0.27, respectively. As visualized in Fig.  $4(b)$ , high fractions of NO<sub>3</sub><sup>-</sup> (15%) and SO<sub>4</sub><sup>2-</sup> (16%) during the pollution episode can be found and their mass concentrations were much higher than that during other sampling period.

Based on the PMF results, as demonstrated in Fig. [1](#page-4-0) (episode III), the average contributions of traffic emission, coal combustion, resuspended dust, nitrate, sulfate, and other sources for  $PM_{2.5}$  mass concentration were 13%, 12%, 4.1%, 22%, 44%, and 3.9%, respectively. As shown in Fig. [4\(a\),](#page-8-0) sulfate (82 μg m<sup>-3</sup>) and nitrate (42 μg m<sup>-3</sup>) were the most important contributors. Other source contributors were traffic emission (25 μg m<sup>-3</sup>), coal burning (22 μg m<sup>-3</sup>), and resuspended dust  $(7.7 \mu g m^{-3})$ .

According to the results of the random forest in Fig. [4\(c\)](#page-8-0), in process III, it was estimated that the relative importance of CO,  $SO_2$ , NH<sub>3</sub>, and RHU to  $PM_{2.5}$  concentration variation

<span id="page-7-0"></span>

Fig. 3 The abundance and concentration of chemical species, PMF result, and random forest result during process II (episode characterized by the Spring Festival). a The concentration contribution from coal combustion, traffic emission, resuspended dust, sulfate, nitrate, and firework change over time in episode II; b the difference between the average mass percent

of  $PM_{2.5}$  components in episode II and all year; c gases and meteorological parameters have different contributions to  $PM_{2.5}$ variation in episode II; d gases and meteorological parameters have different contributions to SOR in episode II; e gases and meteorological parameters have different contributions to NOR in episode II

were 40%, 13%, 8.3%, and 15%, respectively. Simultaneously, it was calculated that  $SO_2$  (24%), NH<sub>3</sub> (28%), and T (19%) were the main influence factors for SOR. And the results showed that NOR was greatly affected by  $NO<sub>x</sub>$  and  $NH<sub>3</sub>$ , and the importance of both were 12% and 45%, respectively. NOx,  $NH_3$ , and  $SO_2$  mainly came from vehicle exhaust and coal burning (Xie et al. [2005](#page-11-0); Meng et al.  $2011$ ). Under the higher RHU, NH<sub>3</sub> was an important alkaline gas in the atmosphere, and it can react with acid gases  $(SO<sub>2</sub>, NO<sub>x</sub>, etc.)$  to form secondary aerosols (Zhang et al. [2012\)](#page-11-0). This episode can be identified as reaction caused by high RH. The strong influence of RH on the reaction of inorganic ions was worthy of notice. High RH was favorable for the formation of sulfate and nitrate in the aqueous phase and may also improve sulfate and nitrate partitioning into the liquid phase formed by a gas-phase homogeneous reaction of precursors. Alkaline aerosol particles in northern China have been reported to lead to high PH values (Kulshrestha et al. [1998](#page-9-0)), which could also enhance the equilibria reactions. Therefore, relative humidity provided a key catalyst to chemically react to secondary aerosol formation. It was apparent that this episode was a common type of haze in northern China with characteristic features of high inorganic aerosols, high RH, stagnant meteorological conditions with low mixing heights, and large emissions of primary air pollutants causing secondary pollution.

<span id="page-8-0"></span>

Fig. 4 The abundance and concentration of chemical species, PMF result and random forest result during process III (episode characterized by high inorganic ions and high RH). a The concentration contribution from coal combustion, traffic emission, resuspended dust, sulfate, and nitrate change over time in episode III; b the difference between the average mass percent of  $PM_{2.5}$  components in episode III and all year; c gases

and meteorological parameters have different contributions to  $PM_{2.5}$ variation in episode III; d gases and meteorological parameters have different contributions to SOR in episode III; e gases and meteorological parameters have different contributions to NOR in episode III

# **Conclusions**

Hourly measured  $PM<sub>2</sub>$ , s-bound species, gases, and meteorological data were analyzed by the PMF receptor model to quantify source contributions and by the random forest regression method to identify the contribution of gaseous pollution and meteorological parameters on the changes of  $PM<sub>2.5</sub>$  concentrations, SOR, and NOR. It was found that  $CO$ ,  $SO<sub>2</sub>$ , NOx, P, and T played an important role in  $PM_{2.5}$  variation; pH, SO<sub>2</sub>, RHU, and T were the dominating influence factors for SOR; and  $NO_x$ ,  $O_3$ ,  $NH_3$ , and RHU determined the NOR for the whole sampling period. Proposed PMF-random forest method was used to quantify the source contributions to  $PM<sub>2.5</sub>$ .

Three types of pollution episodes were captured. The first pollution episode (process I) was characterized by heavy coal combustion due to heating in the winter in northern China. CO remained at a high level and contributed 40% of PM<sub>2.5</sub> concentration variation. According to the PMF results, coal combustion and secondary sources were the most important contributors in the first stage, and then, the highest contribution was from secondary sources. The second pollution episode (process II) happened during the Spring Festival and was characterized by high K<sup>+</sup> and Cl<sup>-</sup>. Firework combustion contributed 8.8  $\mu$ g m<sup>-3</sup> estimated by PMF. High SO<sub>2</sub> during process II, especially on the CNY's Eve, was observed due to the firework displays, and  $SO_2$  gave a high contribution (24%) to

<span id="page-9-0"></span> $PM<sub>2.5</sub>$  concentration variation. The second pollution episode (process III) was characterized by high inorganic ions and high RH in summer. Sulfate (44  $\mu$ g m<sup>-3</sup>) and nitrate (22  $\mu$ g m<sup>-3</sup>) were the most important contributors. It was estimated that the relative importance of  $CO$ ,  $SO<sub>2</sub>$ , NH<sub>3</sub>, and RHU to PM2.5 concentration variation were 40%, 13%, 8.3%, and 15%, respectively. Through comparing the influence factors for SOR and NOR during episode I in winter and episode III in summer, DD was more important in episode I indicating that this episode was strongly influenced by the transportation; while  $NH<sub>3</sub>$  was more important in episode III indicating alkaline precursor strongly influence the episode in summer.

Authors' contributions Junwei Yang, Zhimei Xiao, and Kui Chen performed the experiments.

Xin Du made the figures. Xin Du and Yingze Tian wrote the paper. All the authors contributed to the submitted version of the manuscript. Yingze Tian and Yinchang Feng acquired the funding.

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Data availability All relevant data are within the manuscript and available from the corresponding author upon request.

## **Declarations**

Ethical approval Not applicable.

Consent to participate All authors participated in this work.

Consent to publish All authors agree to publish.

Conflict of interest The authors declare that they have no conflict of interest.

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