REVIEW ARTICLE



Biological approaches of fluoride remediation: potential for environmental clean-up

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Abstract

Fluoride (F), anion of fluorine which is naturally present in soil and water, behaves as toxic inorganic pollutant even at lower concentration and needs immediate attention. Its interaction with flora, fauna and other forms of life, such as microbes, adversely affect various physiochemical parameters by interfering with several metabolic pathways. Conventional methods of F remediation are time-consuming, laborious and cost intensive, which renders them uneconomical for sustainable agriculture. The solution lies in cracking down this environmental contaminant by adopting economic, eco-friendly, cost-effective and modern technologies. Biological processes, viz. bioremediation involving the use of bacteria, fungi, algae and higher plants that holds promising alternative to manage F pollution, recover contaminated soil and improve vegetation. The efficiency of indigenous natural agents may be enhanced, improved and selected over the hazardous chemicals in sustainable agriculture. This review article emphasizes on various biological approaches for the remediation of F-contaminated environment, and exploring their potential applications in environmental clean-up. It further focuses on thorough systemic study of modern biotechnological approaches such as gene editing and gene manipulation techniques for enhancing the plant-microbe interactions for F degradation, drawing attention towards latest progresses in the field of microbial assisted treatment of F-contaminated ecosystems. Future research and understanding of the molecular mechanisms of F bioremediation would add on to the possibilities of the application of more competent strains showing striking results under diverse ecological conditions.

Keywords Fluoride · Bioremediation · Plant growth promoting bacteria

Introduction

Fluoride (F) is a monoatomic anion of fluorine, which comes under halogen group of gases and accounts for 0.3 g kg⁻¹ of the Earth's crust. Being highly electronegative in nature, F exists only in combination with other elements to form compounds and minerals such as fluorspar, cryolite and fluorapatite (Ghosh et al. 2013). Fluoride contamination is a worldwide problem and is anthropogenically added into the environment through airborne as well as non-airborne sources. Airborne industrial sources include aluminium

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Keshav Kant Sahu skeshavkant@gmail.com smelters, brick works and phosphate fertilizer factories, while non-airborne causes originates from spraying F-rich fertilizers in the fields and using contaminated water for irrigation (Ali et al. 2016). Fluoride has been reported to be endemic in 20 countries of the world (Messaitfa 2008). The Union Health and Family Welfare Ministry reported that 19 states of India have high F contamination in their ground water, amongst which Rajasthan occupied the topmost position (24 mg L⁻¹) followed by Andhra Pradesh, Telangana, Uttar Pradesh, Karnataka, Madhya Pradesh, Haryana, Bihar, Chhattisgarh, Maharashtra, Assam, West Bengal and Gujarat (Ali et al. 2016).

Fluoride is considered to be an essential micronutrient for animals, plants and human beings; however, it has proved to exert deleterious effects if exists beyond the permitted values. The permissible amount of F in drinking water is set to 1.5 mg L^{-1} by the World Health Organization (Geneva, Switzerland), while 0.8–1.2 mg L^{-1} exerts the maximum benefit and minimum harmful effect (World Health Organisation 2008). Prolong exposure to F generates devastating effects in

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almost all live forms including humans. Cellular accumulation of F alters the sensitive biochemical processes by interfering with various enzymatic activities, directly or indirectly. It acts as a protoplasmic poison, which can cause various biochemical modifications including oxidative stress and DNA damage resulting into apoptosis and cell cycle alterations (He and Chen 2006; Chouhan et al. 2009). Fluoride may covalently bind to DNA, leading to DNA damage thereby initiating a series of events in the development of chemical carcinogenesis (Zhang et al. 2008). Excessive intake of F in human beings leads to serious health hazards like fluorosis, characterized by various deformities in teeth and skeleton such as dental mottling, osteoporosis and osteosclerosis (Davis 1980). Fluoride has also proved to be a phytotoxic element which interferes with various metabolic pathways in plants. It is absorbed through root system and transported via xylem to different parts, generating many adverse effects.

Fluoride uptake and transport in biological systems occur through the non-ionic diffusion of hydrogen fluoride (HF). The molecules of HF being highly permeant penetrate cell membranes much faster, resulting in a more pronounced intracellular intake and accumulation. The other routes of F intake in cells are considered to be the much demonstrated anion channels and co-transporters (Barbier et al. 2010). Fluoride also combines with other metals such as calcium, aluminium and beryllium to form inorganic complexes for easy penetration through cell membrane. However, the physiological and toxicological effects depend on extracellular metal concentration and the type of ion channels.

Fluoride at micromolar levels promotes cell proliferation, whereas at millimolar concentrations inhibit several enzymes, both in vivo and in vitro. The increasing load of F in the ecosystem produces significant amounts of toxicity which needs to be treated. The many approaches of bioremediation are gradually being accepted as the standard practice over conventional chemical and physical methods as they hold strong potential in removal of F even at low concentrations in a cost-effective and more efficient way.

Toxic effects of fluoride

In plants

Fluoride accumulation in plants is due to its absorption from the air, soil and water, which occurs in two ways: firstly, gaseous and particulate forms are actively absorbed by leaf stomata and to some extent by the cuticle of leaves and secondly, through passive absorption by roots (Baunthiyal and Ranghar 2013). Absorbed F can then be transported to the leaves by the symplastic or apoplastic pathways or via xylematic flow (Elloumi et al. 2005). The uptake is dependent upon several parameters such as soil pH, its activity and composition. A few plants are highly tolerant to F and hence can accumulate up to 4000 μ g g⁻¹ dry mass (DM) without showing any sign of toxicity. However, other plants show toxicity symptoms even at much lower concentration (<20- μ g g⁻¹ DM). Normal concentration of F in plants, growing in nonpolluted soil, is 0.1–15-mg kg⁻¹ DM (Cooke et al. 1976). Plants like *Avena sativa* L., *Acer pseudoplanatus* L. and *Camellia sinensis* L. can accumulate more than the normal concentration of F. Table 1 displays the list of plants and their F accumulation capacity. Fluoride toxicity in plants occurs at three different levels: physiological, biochemical and molecular.

Physiological level

The initial symptoms of F injury are necrosis at the leaf tips, and margins which on prolonged exposure causes notched leaf and its falling-off. Chlorosis is also believed as a sensitive indicator of its toxicity. Severities of these symptoms are influenced by many environmental factors such as concentration of F, distance from source, length of exposure and meteorological conditions (Singh and Verma 2013). It is also reported to adversely affect plant growth and yield responses by causing reductions in growth parameters, i.e. germination percentage, lengths of root and shoot, number and size of leaves, number of flowers, fruit-set percentage and seed-set frequency (Singh et al. 2013).

Biochemical level

The high internal F concentration disturbs almost all the biochemical processes of plants. At micromolar concentration, it promotes cell proliferation, while at millimolar concentration

 Table 1
 Concentration of fluoride in plant materials grown in normal soil (Stevens 1996)

Species	Part	Concentration (mg kg ⁻¹ dry weight)
Avena sativa L.	Grain	6.8
	Straw	14.8
	Root	36.8
	Shoot	40
Triticum aestivum L.	Grain	1.4
	Straw	1.8
Solanum tuberosum L.	Tuber	0.25
	Sprout	10.0
Daucus carota L.	Root	0.21
	Sprout	7.5
Medicago sativa L.	Shoot	10–20
Acer pseudoplatanus L.	Shoot	18
Lotus corniculatus L.	Shoot	15
Dactylis glomerata L.	Shoot	4.6
<i>Camellia sinensis</i> L.	Shoot	67–3062

disrupts, the activities of enzymes involved in several metabolic processes by binding to functional amino acid groups that surrounds their active centre and inhibition of protein secretion and synthesis.

Molecular level

Excess availability of F adversely affects to signalling pathways involved in cell proliferation and apoptosis, regulation of active oxygen species (AOS), patterns of gene expression, etc. (Barbier et al. 2010) (Fig. 1).

In humans and domestic animals

In humans, adequate amount of F is crucial for the integrity of bones and teeth. It improves bone density, enhances the remineralization of enamel and checks dental caries and osteoporosis (Ericsson 1970; Arlappa et al. 2013). On the contrary, excessive intake of F may cause dental fluorosis and other deformities which include long-term frequent joint pain, limited joint movement and limb motor dysfunction. It becomes more challenging for the people suffering from severe health problems like paralysis (Choubisa and Sompura 1996). Fluoride may extensively affect mammalian nervous system because it can penetrate the blood-brain barrier, and excessive ingestion of it can lead to brain-specific metabolic disorders due to the inhibition of several key neuronal enzymes and enhanced rate of lipid peroxidation (Shivarajashankara et al. 2002). Recent studies have shown that F can also be toxic to the immune system, including macrophages.

Fluoride remediation strategies

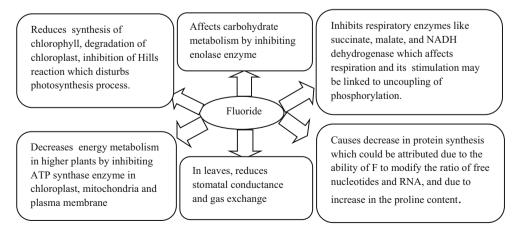
Fluoride is non-biodegradable in nature, therefore causes ecological destruction, gets accumulated in tissues and requires extensive measures for its clean-up (Rose and Marier 1977). Because it is poorly digested by the living organisms (Abdallah et al. 2006), F shows negative effects on plants, animals and human health after entering into the food chain (Stevens et al. 2000). The level of toxicity rendered by the F in different living forms demands effective ways for the ecological removal and remediation of this toxicant. Some of the known and adopted strategies of its removal includes physiochemical and biological approaches.

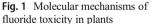
Physiochemical approaches

Several physiochemical procedures such as reverse osmosis (Sehn 2008), electrodialysis (Jayarathne et al. 2014), adsorption (Halder et al. 2015), ion exchange (Rangel et al. 2015), Nalgonda technique, biosorption (Mukherjee and Halder 2016), nanofiltration, etc., have previously been reported for F removal. Although, these methodologies have many advantages, they are allied with numerous negative side effects as well like high initial cost, water desalination, higher waste production, excessive electricity utilization, problem of waste discarding and inadequacy to remove all the pollutants (Gentil and Fick 2016). These problems necessitate to uncover the clean and cost-effective alternative approaches for F removal and remediation.

Biological approaches

Biological methods concerns with the involvement of living organisms like plants and microorganisms or their products for the treatment/removal of toxicants from the environment. The concept of bioremediation is largely exploited in different ways which helps in the conversion of organic contaminants into simpler forms such as carbon dioxide and water by the intensive action of plants/microorganisms as part of their metabolic processes (Sarkar et al. 2006). They have the potency to resist different pollutants by acquiring procedures like bioaccumulation, biotransformation, biosorption, etc. (Mukherjee et al. 2017). Researchers have reported both aerobic and





anaerobic degradation pathways for the mineralization and stabilization of organic contaminants in the atmosphere. Biological treatments are more advantageous because of their cost-effectiveness, operational simplicity and less sludge production. The use of natural resources makes the process environment friendly and highly acceptable even for large-scale treatments (Doble and Kumar 2005).

Phytoremediation

It refers to promising green technology that utilizes plants to clean soil, air and water contaminated with hazardous materials (Kamran et al. 2014). It involves several processes like phytostabilization, phytovolatalization, phytoextraction, phytostimulation and phytodegradation for dealing with the pollutants, amongst which phytostabilization and phytoextraction have already been reported for F remediation.

Phytostabilization

It is mediated by plants that have ability to inhabit the contaminated soil. The direct action of root exudates immobilizes F by trapping in the soil matrices, thereby limiting its solubility. Also, the roots prevent the migration of F due to deflation and erosion by adsorption and precipitation, respectively, within the root zone (Pollard et al. 2014). Abdallah et al. (2006) studied the affinity of grapes to equilibrate F accumulation in its leaf margins by an equivalent Ca accumulation and suggested the entrapping of F as CaF_2 , which was not able to disturb plant metabolism. This study favoured the non-translocation of F through phloem towards lower plant organs.

Fluoride tolerant plants may have the ability to deactivate it better than the sensitive ones. Probable mechanisms includes shifting towards F insensitive metabolic pathways; complexation of F with organic compounds and their removal from the site of enzyme inhibition; reaction with cationic sites; sequestration in vacuoles and translocation to the leaf surfaces, etc. (Stevens 1996; Thijs et al. 2016). Some of the plants are able to decrease internal F by exporting it to the exterior surfaces of leaves (Baunthiyal and Ranghar 2013).

Phytoextraction (phytosequestration)

It involves absorption, translocation and accumulation of contaminants in the aerial parts. The process of extraction takes place through roots, where contaminants are initially absorbed from their immediate environment and then translocated into various harvestable parts (shoots, leaves, etc.) (Petzold et al. 2015). This approach usually involves hyperaccumulators or plants that can accumulate 0.1% or higher of contaminants on DM basis (Baker et al. 2000; Pollard et al. 2014). Plant species having bioconcentration factor (ratio of contaminant concentration in plant root to soil), translocation factor (ratio of contaminant concentration in plant shoot to root) and enrichment factor (ratio of contaminant concentration in plant shoot to soil) values greater than 1 would be considered as hyperaccumulators (Lorestani et al. 2011). A number of plants are known today which are having promising ability to accumulate F (Table 2). However, many plant species have been reported to accumulate F from the soil, but none have been commercially exploited for this purpose. Screening of potential hyperaccumulators for high F tolerance and resistivity is essentially required in the present time period. Rapidly growing high biomass crop with an extensive root system and showing least toxicity could be raised to remediate F from the soil (Baunthiyal and Ranghar 2013). Correspondingly, the aquatic plant species can be easily utilized to remediate F from the polluted water bodies. Baunthiyal and Sharma (2012) demonstrated the potential of eight tree species of semi-arid region for hyperaccumulation of F, out of which Prosopis juliflora (Sw.) DC., was shown to remove significant amount of F from groundwater and soil.

Fluoride tolerant and resistant plants have inherent cellular mechanisms to reduce its toxic effects. In regard, Saini et al. (2012) examined the organ-wise accumulation of F in *P. juliflora*. Their results showed that mentioned species accumulated highest amount of F in roots. The bioaccumulation and translocation factor values were found to be > 1 which illustrated high tolerance and accumulation efficiency of plants for F. In another study, 17 species were exposed to varying concentrations (2.5 to 10 mg L⁻¹) of HF, out of which only three species, *Camellia japonica* L., *Pittosporum tobira* Thunb. and *Saccharum officinarum* L. were able to remove F efficiently. *S. officinarum* showed maximum removal capability, suggesting the activation of some detoxification process in the cell to withstand F concentrations (Santos-Díaz and Zamora-Pedrazaa 2010).

Microbial remediation

Soil rejuvenation after removing its pollutants and contaminants by using bacteria or fungi is termed as microbial remediation. These microbes utilize the contaminants and then degrade them for energy and reproduction.

Microbial remediation is performed by three processes:

- Natural attenuation which takes place naturally with indigenous soil microorganisms
- *Biostimulation* by providing external nutrients, moisture and an ideal pH to the microorganisms present in the soil
- *Bioaugmentation* by the use of externally introduced microorganisms in situations where the naturally occurring microbes go dead because of the intensity of the contamination

Species	Accumulated fluoride	References
Spinacia oleracea L.	1.7 g kg^{-1}	Khandare and Rao 2006
Amaranthus L.	20.9 g kg^{-1}	Khandare and Rao 2006
Abelmoschus esculentus L.	$0.43 \mathrm{g \ kg^{-1}}$	Khandare and Rao 2006
Vachellia tortilis (Forssk.) Galasso and Banfi	592.24 μg g ⁻¹	Baunthiyal and Sharma 2012
Prosopis juliflora (Sw.) DC.	852.01 μg g ⁻¹	Baunthiyal and Sharma 2012
Brassica oleracea L.	12.91 mg mL^{-1}	Patil et al. 2014
Portulaca grandiflora Hook.	22.96 mg mL ^{-1}	Patil et al. 2014
<i>Camellia sinensis</i> L.	1442 mg kg^{-1}	Rangel et al. 2015
Saccharam officinarum L.	1012 mg kg^{-1}	Rangel et al. 2015
Nerium oleander L.	3.7 mg g^{-1}	Khandare et al. 2017
Portulaca oleracea L.	2.4 mg g^{-1}	Khandare et al. 2017
Pogonatherum crinitum (Thunb.) Kunth	2.8 mg g^{-1}	Khandare et al. 2017

Table 2 List of fluoride hyperaccumulator plant species

Bacterial remediation of fluoride

Fluoride affects bacterial cells in many ways: firstly, by directly inhibiting the enzymes; secondly, by forming phosphate analogues like alumino-fluoride and beryllium fluoride complex: and lastly, by forming HF which acts as a transmembrane proton transporter, resulting in uncoupling of oxidative phosphorylation (Poulsen 2011). It has been studied that contamination by atmospheric F causes 80% decrease in soil bacterial biomass (Tscherko and Kandeler 1997). Fluoride mainly exists in combination with proton as HF, which readily cross through the cell membrane and acts as a proton conductor (Chouhan et al. 2012). The contaminant may disrupt the microbial membrane; however, the cell sometimes defends itself by forming an outer covering of hydrophobic layer having solvent efflux pumps over the membrane (Dixit et al. 2015). In spite of the cell deterioration induced by F, microorganisms have developed diverse machinery such as sorption, mineralization, uptake and accumulation, extracellular precipitation, enzymatic transformation and efflux from the cell, for adapting F toxicity (Hussein et al. 2004). These abilities could be sufficiently utilized for successful remediation of F by making it less accessible and toxic for the living forms.

Bacteria can acquire either transient or stable F resistance. Transient resistance is related to the horizontal transfer of plasmids between two cells, while stable F resistance is believed to be due to the chromosomal mutations (Liao et al. 2017). Inhibitory effects of F on intracellular metabolism depend upon the influx of HF which after diffusion dissociates into the H⁺ and F⁻ (Hamilton 1990). This intracellular H⁺ and F⁻ inhibit F-ATPase and enolase enzymes which adversely affects glycolysis pathway and acid tolerance of bacteria. These enzymes are believed to be mutated in the Fresistant bacteria (van Loveren et al. 2008). Also, bacteria can efficiently reduce F concentration because their cell wall is comprised of amines, sulphhydryl, carboxylates and phosphates which are metal-binding groups and help in adherence of the F ions to their surfaces. Certain bacterial species are tolerant to higher concentration of F since they can build their resistance through various mechanisms like bioaccumulation, biotransformation, etc. (Juwarkar and Yadav 2010). Different genera of bacteria have been reported which can tolerate high F concentrations along with the potential to remove it (Table 3). These bacteria can survive in Fcontaminated soil and provides a new opportunity to develop bioremediation techniques which will be a novel way of alleviating F toxicity.

In 2012, Breaker discovered two gene families: crcB and eriC^F which had identical biochemical roles. Both these families encoded F exporters which were directly related to the F resistance in microorganisms. To support this, authors constructed a reporter gene by joining crcB motif RNA (crcB gene) of Bacillus cereus and a homologous eriC motif RNA (eriC gene) of Pseudomonas syringae to a lacZ gene, and transferred this reporter construct into an Escherichia coli after mutating its natural lacZ gene. A high rate of expression was seen when cells were allowed to grow on a medium supplemented with higher concentration of F (Baker et al. 2012; Breaker 2012). Men et al. (2016) observed two *eri*C^F genes in Streptococcus mutans, namely, perA and perB which were also involved in F resistance and has been approved by gene knockout study and gene regulation analysis. Studies revealed that the S. mutans became 100-fold more sensitive to F when both the genes were knocked out. Further, gene regulation studies proved that a single mutation in the promoter region can significantly upregulate the expression of both the genes, thereby conferring F resistance in S. Mutans (Liao et al. 2016; Men et al. 2016; Murata and Hanada 2016).

Table 3 Comparative representation of fluoride removal by bacteria

Species	Fluoride removal (%)	References
Pseudomonas aeruginosa	22.1	Chouhan et al. 2012
Bacillus flexus. NM25	67.45	Pal et al. 2014
Acinetobacter sp. RH5	25.7	Mukherjee et al. 2015
Bacillus cereus	21.91	Banerjee et al. 2016
Providencia vermicola (KX926492)	82	Mukherjee et al. 2017
Ca21-treated live <i>Nostoc</i> sp. (BTA 394)	86	Mukherjee et al. 2017
Nostoc sp. (BTA 394)	73.43	Mukherjee et al. 2017
Immobilized cells of Staphylococcus lentus	92	Mukherjee et al. 2018
Dead cells of Staphylococcus lentus (KX941098)	85.03	Mukherjee et al. 2018

Mycoremediation

Mycoremediation is a method of environmental clean-up which relies on fungi and their enzyme systems. Fungi are metabolically and ecologically diverse organisms which have unique ability to degrade different types of pollutants. Furthermore, these also have ability to form extended mycelial networks, to absorb pollutants as their growth substrates; the process is more likely known as mycosorption. Fungi which are native to the contaminated sites play important role in modification and manipulation of the bioavailability of chemicals, thereby decreasing the allied risks. Their intracellular and extracellular catabolic enzymes make them well suited for bioremediation process which facilitates degradation of a broad range of pollutants and absorption of complex carbohydrates without hydrolysis. Application of fungal biosorbents like Pleurotus ostreatus (Ramanaiah et al. 2007), Aspergillus penicillioides, Mucor racemosus (Prajapat et al. 2010) and Pleurotus eryngii (Amin et al. 2015) for F removal have already been reported. The other fungal species which are used as tools in remediation process includes Trichoderma longibrachiatum, P. ostreatus, Rhizopus arrhizus, Trametes hirsute, T. versicolor, Lentinus edodes, Aspergillus niger, Aspergillus flavus, Aspergillus terreus, Cladosporium resinae and Phanerochaete chrvsosporium (Cecchi et al. 2019). Fungi present the most significant class of soil biomass; however, they have not been exploited to their fullest for the treatment of contaminated environments.

Phycoremediation

Microbial usage for bioremediation has certain limitations such as bacterial sludge produced is of very little use and it also needs a long-standing commitment (Wang et al. 2002). Many of the microorganisms are also associated with the production of toxins and complex end products which are difficult to tackle. For this reason, nowadays phycoremediation has become a potential method for bioremediation (Cho et al. 2011). Algae have the capability of growing in any atmosphere, including domestic, municipal and industrial wastewaters. It could withdraw nutrients for growth from the pollutants; thereby decreasing the level of pollution and contributing to environmental clean-up (Mahapatra et al. 2014). On the other hand, algal biomass has affluent amount of hydrocarbons, polysaccharides and lipids which can be used for the production of biofuel. Many physiochemical and microbial remediation work has previously been done for removing F from soil and water, but little has been reported using microalgae (Bhatnagar and Bhatnagar 2000). There is a recent report on the use of a cyanobacterial strain Starria zimbabweensis for remediation of Fcontaminated waste water. A maximum of 66.6% of F was removed following this strain from synthetic wastewater having 10 mg L^{-1} initial F concentration (Ramachandra et al. 2013; Biswas et al. 2018). Other algal strains which were used as biosorbents were Spirogvra sp.-IO2 and Chara fragilis (Levy and Strauss 1973; Mohan et al. 2007).

Plant-microbe interaction approach

In general, plants are in symbiotic relationship with soil microbes during their growth and development processes. These microorganisms residing in the rhizosphere impose various beneficial effects on the host plant following different mechanisms such as nitrogen fixation and nodulation. These microbes are referred as plant growth-promoting rhizobacteria (PGPR). They are likely to protect the health of plants in an eco-friendly manner, and their interactions with plants are utilized commercially, which have scientific implications for sustainable agriculture (Gouda et al. 2018).

Plant-microbe interaction acts as a low-cost biotechnological approach for ecosystem renaissance and for the F remediation. Most of the F accumulating plants interact with rootassociated microbes to subsist in the toxic and adverse environment (Paul 2007). Efficacy of bioremediation may be partial, particularly when soils are polluted with recalcitrant organic molecules; hence, the combined use of microbes and plants can boost the efficiency (Thijs et al. 2016). Plant roots supply microaerophilic conditions, and their exudates increase the binding and availability of organic contaminants for the soil microbes. This process accelerates the degradability of contaminants by upregulating the microbial metabolism and by altering the soil pH or solubility of contaminants. Possible pathways for enhanced biodegradation of F by root exudates are (1) direct degradation by plant enzymes; (2) secretion of surfactants for enhanced bioavailability of pollutants; (3) utilization of structural analogy and co-metabolic processes and (4) continuous flow of nutrients and energy to stimulate whole process (Martin et al. 2014). Pseudomonas fluorescence has been seen to improve ability of P. juliflora for remediation of F by increasing the mineral content and biomass (Chaudhary and Khan 2016). Bacillus sp. also has capability to remove F from soil and has been reported as PGPR by many researchers (Sheng et al. 2008; Pal et al. 2014). These bacteria reduce F concentration from their surroundings by accumulating it inside their cells. This ability of tolerating high concentration of F may provide opportunity to develop a new bioremediation technique and a novel way of removing this xenobiotic from contaminated soil and water.

Roles of PGPR in phytoremediation

The PGPR promote phytoremediation either directly or indirectly. Direct process comprised of solubilization, bioavailability and final accumulation of contaminants by plants. In contrast, indirect processes involves plant growth enhancement by the production of essential enzymes and volatile organic compounds and by prevention from phytopathogens.

Direct process Microbes present in the soil are involved in different reactions and metabolic processes that occur in the biogeochemical cycling of nutrients, maintenance of soil structure and detoxification of pollutants (Khan et al. 2010). Various PGPR produce siderophores at extreme environmental conditions including scarcity of nutrients or presence of elevated concentrations of pollutants, thus increasing their bio-accessibility in the rhizosphere through a complex reaction (Rajkumar et al. 2012). Low molecular weight organic acids like gluconic, oxalic and citric acids produced by PGPR helps in the mobilization of pollutants such as F, thereby increasing their uptake in plants (Ullah et al. 2015). Under nutrient deficient and adverse conditions, PGPR have the potential to provide essential nutrients to the plants. For instance, nitrogen is about 78% in the atmosphere but is unavailable to plants. Diazotrophic bacteria have the ability to fix atmospheric nitrogen, even under abiotic stresses. Moreover, PGPR play

significant role in mineral uptake, stomatal regulation, osmotic modification and adaptation of root morphology (Compant et al. 2005; Ouzounidou et al. 2006). Biosurfactants are additional important metabolites produced by PGPR that may have the potential to improve F mobilization and phytoremediation. These biosurfactants might form complexes with F like other metals at the soil interface, desorbing it from the soil matrix and thus increasing its solubility and bioavailability for plants (Gadd 2010).

Indirect process Pollutants present in the rhizosphere affect nutrient uptake, consequently reducing plant growth. The PGPR provide tolerance to plants by protecting them from phytopathogens and by contributing to enhanced biomass production. This not only helps the plants to grow and flourish in the contaminated ecosystem but also facilitate them to remove immediate pollutants (Luo et al. 2012). Plants may have altered physiology and biochemistry owing to numerous environmental stresses. In such conditions, many PGPR have the capability of producing phytohormones like indole-acetic acid, cytokinins and gibberellins, thereby improving hormone levels and plant responses to stresses (Glick 2012). Collectively, these bacteria improve plant growth in polluted soil, which in turn facilitates the phytoremediation process.

Technologies to understand plant-microbe interaction during fluoride remediation

Transgenic technologies

The potential of plant-microbe interactions for F remediation can be further improved by accomplishment of genetic modifications both in soil bacteria and in plant species.

Omics approaches

Plant responses towards environmental changes comprise of several means which include changes in gene expression (transcriptome), defence responses and also accumulation of protein(s) that can degrade pollutants or metabolites. Nowadays, omics technologies are widely used to understand the cellular processes, genetic control and signalling networks involved in plant responses to environmental stresses. For this, next-generation sequencing technology in addition to mass spectrometry has made a noteworthy contribution to figure out how plant-microbe interaction occurs and also to resolve problems related to soil remediation (Deshmukh et al. 2014; Rabara et al. 2017). The omics technology also has the potentiality to understand the gene-protein networks involved in bioremediation/phytoremediation and can be further utilized for the treatment of F-contaminated ecosystems. Additionally, they can be pooled with other new diagnostic technologies

such as mathematical modelling or network biology for the better understanding of the biological processes implicated in the plant-microbe interaction.

Transcriptomics and proteomics

Transcriptomic means the study of transcriptome, the complete set of RNA transcripts produced by the genome under specific condition or in specific cells. Comparison of transcriptomes by using high-throughput methods like microarray analysis allows the identification of genes that are differentially expressed in distinct cell population or in response to different treatments and stresses. With the help of transcriptomics, improvement can be done in plants and microbes which could further be used for phytoremediation. Through genetic engineering, manipulation in gene expression pattern can be done in plants responsible for (i) uptake and transportation of different pollutants and (ii) converting pollutants to less toxic and volatile forms (Kotrba et al. 2009; Mosa et al. 2016).

Bacterial cells use various RNA and proteins to reduce F toxicity which could possibly serve as objectives for compounds that increases the uptake or release of F by bacteria. The *crc*B gene is associated with F riboswitches, and thus, compounds that can block the F exporter, encoded by this gene might serve as antimicrobial agent when combined with F. One can exploit components of the F sensory and resistance systems to manipulate cellular concentration of F. For example, F-specific biochemical tools such as F riboswitches, when fused with reporter genes, can act as convenient in vivo sensor of F detection. Similarly, model bacterial species can be genetically altered to make them more sensitive or resistant to F by knocking out or overexpressing exporters or other toxicity mitigation proteins (Baker et al. 2012).

In addition, this technology can identify several metabolic pathways that are modified in response to stresses and can also be exploited in transcriptome profiling to identify upregulated or downregulated genes under stress conditions. A transcriptomic technology is also used to detect and alter the PGPR genes that may be upregulated or downregulated under polluted environmental conditions and can be utilized to enhance the phytoremediation process.

Proteomics is an analytical technique used to identify changes in the protein expression of hyperaccumulator plants and microbes in response to stresses. The utilization of various proteomic tools and technologies has made easy the characterization of plant and environmental interactions, expanding our knowledge of these processes in the future. It has been previously reviewed that in response to various environmental stimuli, plant proteins are involved in establishing many types of plant-bacterial interactions, which includes both, resistance towards pathogenic bacteria and symbiotic relations for nutrient availability (Gamalero and Glick 2011). Chouhan et al. (2012) detected an extra protein band of 30 kDa in *Micrococcus luteus* and three extra bands with molecular weights of 20, 22 and 25 kDa in *Pseudomonas aeruginosa* grown in media having F than grown in F lacking media. Understanding the crucial involvement of these proteins and their detailed profiling by proteomic approach may help in understanding the mechanisms evolved in plants for battling with environmental contaminants. Further, a relative proteomics study of plant-microbe interactions in a polluted environment can discover key players of phytoremediation (Chouhan et al. 2012).

Metagenomics

A number of new encroachments in science and technology have always been adapting in the field of bioremediation for establishment of better environment, and in that progression, metagenomics is considered as one of the better emerging technologies. Metagenomics is a tactical approach to identify microbes at the genomic level. For the detection of specific pollutant degrading microbe, researchers isolate the microbial genome directly from the contaminated sample, sequence and then perform reverse transcriptase real-time PCR (RT-qPCR) to quantify the expression of specific genes involved in the degradation of a pollutant (Deverapalli and Kumawath 2015). Screening and identification of metagenomes from the contaminated environments are essential in this study. Various studies have showed that the interest in the metagenomicsbased bioremediation has been gradually increased and has proved to be the best adaptation which leads to establishment of a non-toxic and pure environment. The approach can be successfully applied for the identification of F-resistant strains and understanding their gene profiles best adapted for F remediation.

Nanotechnology

These days nanotechnology mediated bioremediation is in practice widely. Researchers have shown that nanosized zerovalent ions have the potential to degrade organic pollutants like atrazine, molinate and chlorpyrifos (Zhang 2003; Ghormade et al. 2011). It also affects the uptake of organic pollutants like trichloroethylene in *Populus deltoides* W. Bartram ex Marshall via fullerenes (Ma and Wang 2010).

Amongst various physiochemical methods, adsorption offers potential benefits particularly in places where water supply is limited and in small-scale community level or household level where water treatment is required. This process is less energy intensive and offers other advantages like ease of operation, flexibility, simplicity of design and costeffectiveness (Sarkar et al. 2006; Kumar et al. 2011). Many materials like alumina, iron-based oxides, rare metal oxides, activated carbon, etc., have been tested as possible adsorbents for removal of F from water (Stanic et al. 2014). In recent years, application of nanomaterials in environmental applications is getting more importance due to their advantages such as high reactivity and high surface area to volume ratio (Tang and Lo 2013). Christina and Viswanathan (2015) developed iron oxide nanoparticles which were later immobilized in sodium alginate matrix and studied for their F removal potential. These nanoparticles exhibited a maximum adsorption capacity of ~ 58 mg g⁻¹ F for an initial concentration of 40 mg L⁻¹ F at pH 5. In another study, bacterial-surfactin-mediated hydroxyapatite (Hap) nanoparticle was able to remove F from water. Findings of Bazrafshan et al. (2016) showed copper oxide nanoparticles as effective adsorbent of F.

Nanoparticles of various metals are used for plant growth as well as remediation of silver, gold, titanium, silicon, etc. A report showed that silver nanoparticles amplified the PGPR activity which then increased the root area and length of host plant (Sajid et al. 2006). The root-shoot ratio was also changed with the implementation of silver nanoparticles. The plants irrigated with municipal wastewater had higher activities of peroxidase and catalase which were further augmented by the solicitation of silver nanoparticles. Application of silver nanoparticles also modulated the levels of abscisic acid (34%), indole acetic acid (55%) and gibberellic acid (82%). These also increased the proline (70%), encountered oxidative stress and augmented the bioremediation potential of PGPR for lead, cadmium and nickel (Lippmann et al. 1995; Cardoso et al. 2005). Municipal wastewater needs to be treated with PGPR and silver nanoparticles prior to its use for irrigation. This aims for the better growth of the plant and enhanced bioremediation of toxic substances like F (Khan and Bano 2018). Researches proposed that Sol-Gel synthesized titanium oxide nanoparticles stably attach with PGPRs of plant roots and encourage bacterial performance (Timmusk et al. 2018).

Future prospects

Studies on the approaches of F remediation have provided a greater understanding of the use of different biological agents with unique capabilities in treating F contamination in a more pronounced way. The diversified and unmatching performances of the microorganisms and plant varieties look forward in designing new strategies for improving their efficacy. Further, strain discovery, improvement and selection of new isolates will not only create new avenues in treating F but would also expand towards broader spectrum of action. Formulation of PGP microbial consortia and transgenic strains for F remediation can be accomplished to boost their survivability into the environment for more efficient results (Islam et al. 2013; Ojuederie and Babalola 2017). New recombinant DNA techniques can be used to engineer F-degrading microorganisms with targeted mode of action. Moreover,

developing new methods for regulating gene expression and the use of targeted and random mutagenesis which increases the activity of biodegrading enzymes in these organisms is highly desirable. Besides this, more research is needed to completely understand the metabolic pathways in transgenic plants and microbes so as to discover their impact and side effects when used in bioremediation. Likewise, hyperaccumulator plants can also be genetically enhanced for effective extraction of F from the soil environment by the process of phytoextraction for efficient phytoremediation (Ojuederie and Babalola 2017).

Conclusions

The prominent and solemn effects of F on the biosphere are a matter of concern and require a worldwide attention to deliver solutions for its secure removal from the environment. While many physiochemical techniques have been developed, bioremediation is a safe and innovative method for remediating this toxicant. Phytoremediation with PGPR is an emerging technology that appears to solve problems without any hazardous effects. These soil bacteria transform pollutants into soluble and bioavailable forms, which consequently facilitates phytoremediation. Other than this, chemical and biological synthesis of metal nanoparticles, formation of bacterial consortium and the use of transgenics are gaining momentum for the cautious and efficient removal of F from the environment. Even though the above association proved as beneficial approaches for F handling, more intuitive discovery is necessary to optimize the plant-microbe interaction which requires a well-orchestrated combination of various omics technologies along with skilful implementation of gene editing and other transgenic approaches to introduce foreign genes.

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